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### Effects of Oxide Additives on the Microstructure of Surrogate Nuclear Fuels

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# microstructure of surrogate nuclear fuels Riley C. Winters<sup>1,2</sup>, Adrianna E. Lupercio<sup>1,2</sup>, Cayden Doyle<sup>1,2</sup>, Andrew T. Nelson<sup>3</sup>, Brian J. Jaques<sup>1,2</sup>

• As-received CeO<sub>2</sub> and MnO<sub>2</sub> powders were processed in a high energy planetary ball mill (HEPBM) to mix, reduce particle sizes, and to incorporate Mn<sup>+</sup> into the

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An increasing demand for nuclear energy requires reliable and efficient fuels. Release of fission gases in fuels leads to reduced thermal conductivity in the fuel-cladding gap and thus fuel reliability and efficiency(Fig.1.)[1].

**Fig.1.** Cross sectional schematic of nuclear fuel-cladding system [2]. Larger grain sizes lead to improved fission product retention (Fig.2.). This study analyzes the impact of manganese dioxide  $(MnO<sub>2</sub>)$  on the microstructure of cerium dioxide  $(CeO<sub>2</sub>)$ , a surrogate for uranium dioxide  $(UO<sub>2</sub>)$ .

## **I. BACKGROUND Motivation for Research**

**Fig.2.** SEM micrograph of the microstructure of  $UO<sub>2</sub>$  with average grain size ~10 µm [3].

- A surrogate is used for  $UO<sub>2</sub>$  due to [4]:
	- Reduced radiation exposure
	- Decreased costs
	- Increased timeliness of experiments
- $CeO<sub>2</sub>$  is used as a surrogate for  $UO<sub>2</sub>$  due to [4]:
	- Common cubic fluorite crystal structure
	- Similar melting temperature
	- Similar thermophysical properties



- $CeO<sub>2</sub>$  lattice (Fig.3. and Fig.4.).
- profile in Fig. 5.



Fig.3. Milled 500 ppm  $MnO<sub>2</sub>$ doped  $CeO<sub>2</sub>$  powder. Pure, 1000, and 2500 ppm samples were also fabricated.



1200 °C and 1400 °C profiles were also tested.



# **II. EXPERIMENTAL Materials Synthesis**

• Milled powder was pressed using a dual action die into right cylinder pellets at 150 MPa then sintered with the



**Fig.4.** HEPBM motion schematic.

**Fig.5.** 1500 °C Sintering profile used for pure and doped samples.

### **Characterization Techniques**

• X-ray diffraction (XRD) for phase purity and dopant

• A lanthanum hexaboride (LaB<sub>6</sub>) standard was used in the  $MnO<sub>2</sub>$ -doped samples to identify

- incorporation analysis
	- instrumentation shifting.
- microstructural analysis
- analysis

• Scanning electron microscopy (SEM) for

• Energy dispersive spectroscopy (EDS) for chemical

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![](_page_1_Picture_849.jpeg)

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