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Influence of Basis-set Size on the $X^2\Sigma^+_{1/2}$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$, and $B^2\Sigma_{1/2}$ potential-energy curves, $A^2\Pi_{3/2}$ 2 vibrational energies, and D₁ and D₂ line shapes of Rb+He

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The influence of basis set size on the $X^2\Sigma_{1/2}^+$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ potential energy curves, $A^2\Pi_{3/2}$ vibrational energies, and D1 and D2 line shapes of Rb + He

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The $X^2\Sigma_{1/2}^+$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ potential energy curves for Rb+He are computed at the spin-orbit multi-reference configuration interaction level of theory using a hierarchy of Gaussian basis sets at the double-zeta (DZ), triple-zeta (TZ), and quadruple-zeta (QZ) level of valence quality. Counterpoise and Davidson-Silver corrections are employed to remove basis set superposition error and ameliorate size consistency error. An extrapolation is performed to obtain a final set of potential energy curves in the complete basis set limit (CBS). This yields four sets of systematically improved $X^2\Sigma_{1/2}^+$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ potential energy curves that are used to compute the $A^2\Pi_{3/2}$ bound vibrational energies, the position of the D2 blue satellite peak, and the D1 and D2 pressure broadening and shifting coefficients, at the DZ, TZ, QZ, and CBS level. Results are compared with previous calculations and experimental observation.

INTRODUCTION

The invention of the Optically Pumped Alkali Laser (OPAL)[1–5] has led to renewed interest in the spectroscopy of small concentrations of alkali-metal atoms vaporized in a rare-gas buffer. OPALs are three level systems where the alkali-metal atoms are pumped on the D2 transition, then make a collisionally induced transition from the $^2P_{3/2}$ to the $^2P_{1/2}$ excited atomic states, and then lase on the D1 transition. The buffer gas is used to pressure broaden the alkali D2 line to match the pump bandwidth as well as facilitate the fine structure transition. This has driven recent interest in pressure broadened D1 and D2 line shapes [6–10] of alkali-metal atoms as well as fine structure transition rates [11, 12].

In a previous effort[13] we calculated the broadening and shifting coefficients as a function of temperature for a set of nine OPAL diatomic pairs, each comprising an alkali-metal atom perturbed with a noble-gas atom. These calculations are based on line shape theory of Anderson and Talman[14] that utilizes difference potentials (DP) between the upper and lower potential energy curves (PEC) of the alkali-metal noble-gas pair. The results demonstrated the sensitivity of the broadening and shifting coefficients to the form of the longrange DP, and their insensitivity to the intermediate and short ranges of the DP. While the core of the spectral line shape is dependent primarily on the long range interaction potentials, the line wing and associated satellite features depend strongly on the intermediate and short

range region of the DP, as well as on the dipole transition moments (see Allard *et al.* [15]).

In this paper we first compute a set of systematically improved $X^2\Sigma_{1/2}^+$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ PECs for Rb+He at the spin-orbit multi-reference configuration interaction level of theory using a hierarchy of Gaussian basis sets at the double-zeta (DZ), triple-zeta (TZ), and quadruple-zeta (QZ) level of valence quality. Counterpoise and Davidson-Silver corrections are employed to remove basis set superposition error and correct for size consistency error. An extrapolation is performed to obtain a final set of potential energy curves in the complete basis set (CBS) limit. These systematically improved PECs are then used together with the dipole autocorrelation theory of spectral line shape[15] to study the effect of the systematic improvement of the PECs on various features of the D1 and D2 line shapes of Rb+He.

The discussion is organized as follows. In the section labeled 'Computational Approach' we give the details of the construction of ab-initio based PECs for a hierarchy of basis sets and extrapolation scheme used for computing the PECs in the CBS limit. The characteristics of the PECs and difference potential are discussed in the section labeled 'Results and Discussion', followed by $A^2\Pi_{3/2}$ vibrational levels and their comparison with experientially measured levels. The satellite peak position prediction and its convergence with basis set size is also presented in this section, followed by computation of line broadening and shifting coefficients and the influence of basis set size on its values. Concluding remarks and a summary

are given at the end.

COMPUTATIONAL APPROACH

Working within the framework of the non-relativistic Hamiltonian, an alkali-metal atom in its ground electronic state has one electron in the outer s orbital, resulting in a doublet-S $({}^{2}S)$ ground state. The noblegas atom's ground electronic configuration results in a Singlet-S (^{1}S) state due to lack of unpaired electrons. The resulting Σ molecular state that arises from the alkali-metal atom plus noble-gas atom system is labeled by $X^2\Sigma$ and has A_1 symmetry in C_{2v} point group of the molecular frame. The lowest electronic excitation of the s-electron into p orbital in the alkali-metal atom results in a ${}^{2}P$ electronic state. Combining this with the ground ${}^{1}S$ state of the noble-gas atom results in a $B^2\Sigma$ electronic state with A_1 symmetry and two $A^2\Pi$ degenerate electronic molecular states, one with B_1 symmetry and the other with B_2 symmetry. When the spin-orbit correction is introduced in the Hamiltonian, the $X^2\Sigma$ and $B^2\Sigma$ molecular states become $X^2\Sigma_{1/2}^+$ and $B^2\Sigma_{1/2}^+$, respectively. The $A^2\Pi$ states cease to be degenerate and split into $A^2\Pi_{1/2}$ and $A^2\Pi_{3/2}$ spin-orbit states with $A^2\Pi_{1/2}$ being the lower of the two. These are the four lowest molecular electronic spin-orbit states for the Rb + He diatomic system studied in this work.

The electronic wave function of Rb + He is first calculated with Multiconfiguration Self Consistent Field (MC-SCF) approach[16, 17]. The rubidium nucleus and core electrons are represented by a spin-orbit effective core potential(SOECP)[18], whereas the Rb atom's valance electrons and the He atom's electrons are treated explicitly. The orbitals for the MCSCF are chosen according to the complete active space (CAS) formulation in C_{2v} point group symmetry. In the ground state asymptotic limit, the 4s and 4p orbitals of Rb and the 1s orbital of He are doubly occupied. These orbitals are optimized in the MCSCF calculation but are left out of the active space. The active space comprises of the singly occupied 5s and unoccupied 5p orbitals of Rb. This active space of one electron in four orbitals results in four molecular reference states (two of symmetry A1, one B1, and one B2). The same active space is used to perform multireference configuration interaction singles and doubles (MRCISD) calculations[19–21] to capture the correlation energy. The effect of inclusion of the doubly occupied He 1s orbital in the active space does not result in any significant difference in the results presented here and therefore we report calculations with four active orbitals. Sizeextensivity error of the correlation energy is corrected by the standard method of Davidson and Silver [22], and the relativistic corrections to the electronic states

are computed using a state-interacting method. In this method the full Hamiltonian is made up of the sum of the non-relativistic electronic Hamiltonian and the spin orbit hamiltonian, $\hat{H}_{el} + \hat{H}_{SO}$. This Hamiltonian is expanded in a basis set of MRCISD wave functions (solutions for \hat{H}_{el}), where we replace the eigenvalues of \hat{H}_{el} with their Davidson corrected counterparts. The result is known as the spin orbit (SO) matrix, and its diagonalization results in the relativistic corrected energies. The SOECP is employed in the calculation of the matrix elements.

We consider a hierarchy of segmented contracted basis sets ranging from double-zeta, triple-zeta to quadruple-zeta quality. For the helium atom, def2-svp, def2-tzvpp, and def2-qzvpp basis sets of Weigend and Ahlrichs[23–26] are used. The "def2" basis sets are improved versions of Stuttgart-Bonn-Koeln pseudopotentials where 'p' and 'pp' refer to smaller and larger sets of polarization functions, respectively. The def2-svp basis set is effectively a double zeta quality contracted basis set of type $(4s,1p) \rightarrow [2s,1p]$. The def2-tzvpp basis set consists of [3s,2p,1d] basis functions, and includes additional polarization function over its def2-tzvp counterpart ([3s,1p]). The def2-qzvpp basis set, $(8s,3p,2d,1f) \rightarrow [4s,3p,2d,1f]$, is augmented with a f-type polarization function.

For the Rb atom, basis sets dhf-svp, dhf-tzvpp, and dhf-qzvpp[23, 27] with associated ECP28MDF effective core potential from Weigend[28] are used. The SOECP descriptor, ECP28MDF, indicates that 28 core electrons are replaced by the pseudopotential; 'M' denotes that neutral atom was used for generating the pseudopotential, and, 'DF' is an abbreviation for "Dirac-Fock relativistic". Such ECPs are sometimes referred to as the relativistic effective core potential which states that the ECP parameters are based on atomic Dirac-Fock theory. The dhf- basis sets are larger and improved over def2- Rb basis sets.

The def2-svp and dhf-svp basis sets are collective labeled as 'DZ' in the remainder of this paper. Similarly, calculations performed with the def2-tzvpp and dhf-tzvpp are labeled as 'TZ', and, those with def2-qzvpp and dhf-qzvpp are labeled as 'QZ'. The calculations of PECs and dipole transition moments have been performed with the Molpro electronic structure program package[29].

Basis Set Error Correction

Weakly interacting systems like Rb+He can suffer from significant basis set superposition error (BSSE). We correct for this using the counterpoise procedure originally developed by Boys and Bernardi [30]. In general, the counterpoise correction to the energy takes the following

form[31],

$$\delta^{CP}(R) = \sum_{i}^{N} e_{i}^{self} - e_{i}^{full}(R), \tag{1}$$

where R is the internuclear separation, N is the number of fragments in the system, e_i^{self} is the energy of the i^{th} fragment in its own basis, and e_i^{full} is the energy of the i^{th} fragment in the full basis set with all the other fragments containing ghost atoms.

All molecular states considered in this work correlate with the helium atom in its ground 1S_0 state, and, the Rb+He molecular ground state $(X^2\Sigma_{1/2}^+)$ correlates with the rubidium atomic ground state, therefore Eq.(1) can be written as,

$$\delta^{CP}_{X^2\Sigma^+_{1/2}}(R) = Rb^{self}_{^2S_{1/2}} - Rb^{full}_{^2S_{1/2}}(R) + He^{self}_{^1S_0} - He^{full}_{^1S_0}(R), \tag{2}$$

where $Rb_{2S_{1/2}}^{self}$ is the energy of the rubidium atom in it's ground $^2S_{1/2}$ atomic state and it's own basis, $Rb_{2S_{1/2}}^{full}(R)$ is the energy of the rubidium atom in the same ground state and in both its own basis and the basis of a ghost helium atom placed at a distance R away from the rubidium. The terms for the energies of the helium atom are analogous. A calculation of the BSSE correction to the energy for the excited $A^2\Pi_{1/2}$ molecular state involves the He atom in its ground 1S_0 state and the rubidium atom in its first excited $^2P_{1/2}$ state,

$$\delta^{CP}_{A^2\Pi_{1/2}}(R) = Rb^{self}_{^2P_{1/2}} - Rb^{full}_{^2P_{1/2}}(R) + He^{self}_{^1S_0} - He^{full}_{^1S_0}(R). \tag{3}$$

The $A^2\Pi_{3/2}$, and $B^2\Sigma_{1/2}^+$ molecular states both involve the rubidium atom in its $^2P_{3/2}$ excited state, and thus both have the same counterpoise correction to the energy,

$$\begin{split} \delta^{CP}_{A^2\Pi_{3/2}}(R) &= Rb^{self}_{^2P_{3/2}} - Rb^{full}_{^2P_{3/2}}(R) + He^{self}_{^1S_0} - He^{full}_{^1S_0}(R), \\ \text{and } \delta^{CP}_{B^2\Sigma^+_{1/2}}(R) &= \delta^{CP}_{A^2\Pi_{3/2}}(R). \end{split} \tag{4}$$

The total BSSE correction to the $X^2\Sigma_{1/2}^+$ PEC for the DZ, TZ, and QZ basis sets as a function of R is shown in Figure 1 where the size of the BSSE decreases as the valence quality of the basis increases. This trend in BSSE is caused by the more complete nature of the the larger basis sets. It is interesting to note that the BSSE correction exhibited almost no difference when computed using Eqs. 2, 3, or 4 and as a result is essentially independent of the excitation level of the rubidium atom. This occurs because the excited state wave functions are constructed using the same basis functions as the ground state. Thus Equations 2, 3, and 4 all yield the same values of BSSE correction.

The calculated PES are corrected for basis set completeness by extrapolating the energy values to the com-

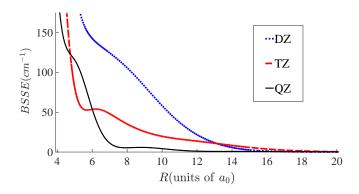


FIG. 1. BSSE in the $X^2\Sigma_{1/2}^+$ PECs calculated at the double zeta(DZ), triple zeta(TZ), quadruple zeta(QZ) basis set levels.

plete basis set limit using a mixed exponential and Gaussian approach of Peterson *et al.* [32],

$$E(X) = E_{\infty} + A \exp{-(X - 1)} + B \exp{-(X - 1)^2}.$$
 (5)

where X is the cardinality of the basis set (2 for DZ, 3 for TZ, etc.), E(X) is the energy at a particular X, and, A, B, and E_{∞} are fitting coefficients. The electronic structure energy at fixed internuclear distances is calculated using DZ, TZ, and QZ quality basis including all the corrections discussed above. The calculated energy values are then used to find the fitting coefficients in Eq. 5 using a least-squares approach. The coefficient E_{∞} yields the extrapolated energy.

RESULTS AND DISCUSSION

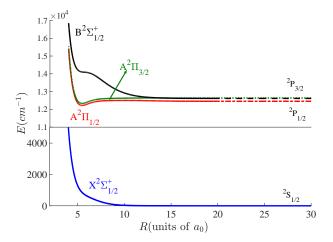


FIG. 2. The Rb + He molecular PECs calculate using multireference configuration interaction approach extrapolated to the complete basis set limit from double zeta(DZ), triple zeta(TZ), and quadruple zeta(QZ)

The $X^2\Sigma_{1/2}^+$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$, and $A^2\Sigma_{1/2}^+$ molecular states of Rb+He computed at MRCISD in the complete

basis set size limit (corrected for basis set superposition errors) are shown in Figure 2. In the asymptotic limit of the ground electronic state, $X^2\Sigma_{1/2}^+$, the molecular energy corresponds to the sum of the ground ${}^2S_{1/2}$ atomic energy of Rb plus the ground ${}^{1}S_{0}$ atomic energy of He. As the internuclear distance (R) decreases the $X^2\Sigma_{1/2}^+$ state is found to be repulsive with a negligible well of about $1 cm^{-1}$, which is very sensitive to the level of theory used in computing the PECs. The asymptotic limit of the $A^2\Pi_{1/2}$ state the molecular energy corresponds to the sum of the ${}^{2}P_{1/2}$ atomic energy of Rb and the ground ${}^{1}S_{0}$ atomic energy of He. As the internuclear separation decreases a small barrier occurs before the $A^2\Pi_{1/2}$ state turns attractive, resulting in a potential well. In the asymptotic limit the $A^2\Pi_{3/2}$ and the $B^2\Sigma_{1/2}^+$ PECs are degenerate and correspond to the ${}^{2}P_{3/2}$ atomic energy of Rb and the ground ${}^{1}S_{0}$ atomic energy of He. As the internuclear separation decreases the molecular states diverge, with the $B^2\Sigma_{1/2}^+$ state becoming repulsive and the $A^2\Pi_{3/2}$ state becoming attractive. The repulsive nature of the $B^2\Sigma_{1/2}^+$ state exhibits a 'shelf' like feature around $R=5-7 \mathring{\tilde{A}}$. The $A^2\Pi_{3/2}$ state exhibits no barrier, and its well is deeper than the $A^2\Pi_{1/2}$ well. Both of these wells equilibrium positions occur at the same value of R. A qualitative explanation of the nature of the non-relativistic molecular electronic states has been outlined by Baylis [33]. In the ${}^{2}\Sigma$ state the Rb electron is mainly in the spherically symmetric, σ molecular orbital, whereas in the ${}^{2}\Pi$ state the alkali electronic wavefunction has π character with node along the internuclear axis allowing the He atom to approach Rb atom closely before the repulsive interaction become dominant.

The shoulder like structure observed on the $B^2\Sigma_{1/2}^+$ has been the subject of many investigations[34–37]. Pascale and Vandeplanque[34] have shown that the $B^2\Sigma_{1/2}^+$ changes from a purely repulsive form (as calculated by Baylis [33]), and converges, to a surface with shoulder when coupling of $B^2\Sigma_{1/2}^+$ state with other neighboring states is included. The shoulder structure is also observed in the Multichannel Quantum Defect Theory (MQDT) potential surface calculations[35–37], wherein, the atomic Hamiltonian consists of ion core (Rb+) and Rydberg electron and their interaction with the He atom. The excited electron experiences a pure Coulomb potential when the electron radial distance from the ion core is sufficiently large, allowing channel mixing effects of the short-range interactions.

Figure 3 shows that the $X^2\Sigma_{1/2}^+$ of Rb + He becomes less repulsive as the basis set size is increased. The $X^2\Sigma_{1/2}^+$ PEC exhibits a very shallow well of less than $1 \ cm^{-1}$. A comparison of the well depth of molecular ground state and its comparison with other theoretical estimates is presented in Table I. Previous studies[38] observed a significantly deeper well $(D_e \approx 10 \ cm - 1)$ for

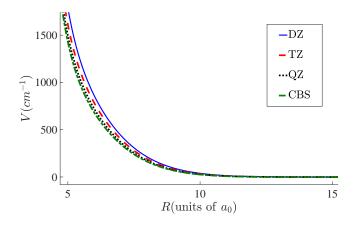


FIG. 3. The $X^2\Sigma_{1/2}^+$ PECs calculated at the DZ, TZ, QZ, and CBS basis set levels.

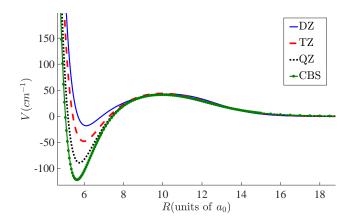


FIG. 4. The $A^2\Pi_{1/2}$ PECs calculated at the DZ, TZ, QZ, and CBS basis set levels.

the ground state for Rb + He, but those calculations did not correct the electronic structure calculation for BSSE. As seen in Table I, current values of r_{min} are fairly close to earlier calculations and all agree to within 10% of each other. The dissociation energies exhibit a wider range of values and reflect the various methods and basis sets used for the calculations. Figure 4 and Figure 5 show the convergence of the $A^2\Pi_{1/2}$ and $A^2\Pi_{3/2}$ states, respectively. The wells in both of these excited states strongly depends on the quality of the basis set. Increasing the basis set size results in a deeper well, smaller equilibrium position, and lower anharmonicity. Note that the barrier in the $A^2\Pi_{1/2}$ state changes very little with increasing basis set size. The convergence of the $B^2\Sigma_{1/2}^+$ PEC is shown in Figure 6 where the shoulder becomes lower in energy and more pronounced as the basis set increases in size. It is interesting to note that the difference in energy between surfaces computed at the DZ and TZ levels in Figure 4 and Figure 5 is comparable to the difference in energy between surfaces computed at the TZ and QZ lev-

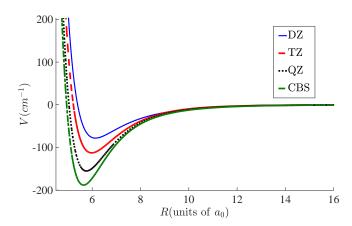


FIG. 5. The $A^2\Pi_{3/2}$ PECs calculated at the DZ, TZ, QZ, and CBS basis set levels.

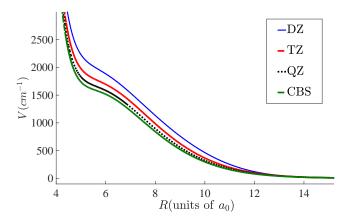


FIG. 6. The $B^2\Sigma_{1/2}^+$ PECs calculated at the DZ, TZ, QZ, and CBS basis set levels.

els. This suggests that the basis set extrapolation may not have completely converged and some improvement may be expected by performing the extrapolation using surfaces computed at the 5Z level.

The transition dipole moments of Rb + He are also calculated for a range of internuclear separation. The transition dipole moment for molecular states is assumed to be equal to that of the Rb 2P $-^2S$ transition in the asymptotic limit. Since these calculations are performed in C_{2v} point group symmetry the B_1 - B_2 (transition between ${}^{2}\Pi$ states) transitions are forbidden while the transition between all other states are allowed for a certain component of the dipole moment. Transitions between states of the same symmetry A_1 - A_1 (the $^2\Sigma$ ground and excited states) are possible for the z component of the dipole moment, which is parallel to the C_2 axis. Transitions between different symmetry states are possible for the x component of the dipole moment $(A_1-B_1;$ the Σ ground state and one component of ${}^{2}\Pi$ state), and for y component of the dipole moment $(A_1$ - B_2 , $^2\Sigma$ states

TABLE I. Well depths D_e (cm^{-1}) and equilibrium positions r_{min} (a_0) for the electronic energy states of Rb+He. We also include the barrier height and position for the $A^2\Pi_{1/2}$ state. We compare our results to other theoretical calculations. Note that the pseudopotential calculations by Pascale [39] and the DFT calculations by Zbiri and Daul [40] report energies for the $A^2\Pi$ curve and are listed under both the $A^2\Pi_{1/2}$ and $A^2\Pi_{3/2}$ columns for ease of comparison.

Rb+He	$X^{2}\Sigma_{1/2}^{+}$	$A^{2}\Pi_{1/2}$		$A^{2}\Pi_{3/2}$
	well	well	barrier	well
D_e				
current work	-0.9	-122.0	41.2	-188.4
Hirano et al. [41]	-	-102.1	26.5	-176.8
Zbiri and Daul [40]	-	-276	-	-276
Pascale [39]	-	-134	-	-134
Blank et al. [38]	-8.7	-95.9	20.0	-159.1
r_{min}				
current work	14.4	5.7	10.0	5.7
Hirano et al. [41]	-	6.1	10.0	6.1
Zbiri and Daul [40]	-	6.1	-	6.1
Pascale [39]	-	6.25	-	6.25
Blank et al. [38]	12.5	5.9	10.4	5.9

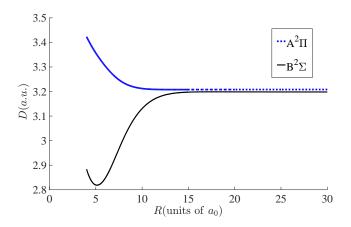


FIG. 7. Dipole transition moments between excited states and the ground state calculated at the MRCISD level of theory. The dipole moment labeled, $A^2\Pi$, is applicable for either $A^2\Pi_{1/2}$ or $A^2\Pi_{3/2}$ state as the choice of excited state. The dipole moment labeled $B^2\Sigma$ is appropriate for choosing the $B^2\Sigma_{1/2}^+$ state as the excited state.

and B_2 component of ${}^2\Pi$ state). It is observed that in the asymptotic limit (as expected), the transition moments $\langle {}^2\Pi(B_1)|\mu_x|^2\Sigma(A_1)\rangle$, $\langle {}^2\Pi(B_2)|\mu_y|^2\Sigma(A_1)\rangle$, and, $\langle {}^2\Sigma(A_1(2))|\mu_z|^2\Sigma(A_1(1))\rangle$ are equal in magnitude to the Rb ${}^2P-{}^2S$ transition.

A difference of the order of about 6% is observed between transition moments calculated at MCSCF level and MRCI level of theory. No significant difference is observed between transition moments at MRCISD and SOCI levels therefore only MRCISD transition moment results are plotted in Figure 7. The dipole transition mo-

TABLE II. Convergence for the Rb+He $A^2\Pi_{3/2}$ vibrational energy levels (in $cm^{-1}).$

=				
E	DZ	TZ	QZ	CBS
E_0	27.7	34.2	42.7	49.0
E_1	61.7	82.2	105.2	122.2
E_2	76.1	107.7	140.2	165.3
E_3	78.6	112.8	153.8	184.2
E_4	-	-	155.8	188.6

TABLE III. $Rb + He \ A^2\Pi_{3/2}$ vibrational energy level differences (in cm^{-1}) for $\Delta\nu = 1$ compared to experiment and two other theoretical calculations.

ΔE	This Work	Exp[42]	Theory[38]	Theory[41]
$E_1 - E_0$	73.2	65.8(3)	55.5	60.5
$E_2 - E_1$	43.1	43.7(2)	33.1	39.2
$E_3 - E_2$	19.0	23.2(7)	17.7	18.2
$E_4 - E_3$	4.4	8.8(6)	9.6	11.9
$E_5 - E_4$	-	-	4.5	7.9

ment between the $A^2\Pi_{1/2}$ state to the ground $X^2\Sigma_{1/2}^+$ ground state is the same as the dipole transition moment between the $A^2\Pi_{3/2}$ and the ground state. In Figure 7 we refer to this moment simply as $A^2\Pi$. The transition moment between the $B^2\Sigma_{1/2}^+$ and the ground state is referred to as $B^2\Sigma$. The dipole transition moments between all the excited states and the ground state show a small variation during the interaction. Such variations can effect the intensity of spectral line shapes induced by non-resonant collisions in the vicinity of the line wing[15].

Vibrational energy levels of the $A^2\Pi_{3/2}$ state are calculated by representing the nuclear Hamiltonian in a finite set of harmonic oscillator basis functions and diagonalizing the resulting matrix [38]. The vibrational levels calculated for potential energy surfaces with different basis

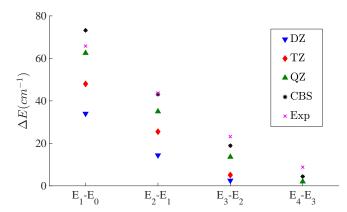


FIG. 8. Convergence of $A^2\Pi_{3/2}$ vibrational energy level differences (in cm^{-1}) as a function of basis set compared with know experimental values.

sets are listed in Table II. We note that as the well depth increases in depth with increasing basis set size an additional vibrational state is observed from TZ to QZ level of basis set. In Table III the vibrational energy level difference is compared with experimental values and other theoretical calculations. It is interesting to note that in the present calculations the $A^2\Pi_{3/2}$ well depth is larger compared to those calculated by Hirano et.al., but due to the difference in the shape of the PEC for $A^2\Pi_{3/2}$ we only observe five vibrational states compare to six as estimated by previous theoretical calculations. This can also be seen when comparing the difference between vibrational energy levels which are in very good agreement with experimentally measured levels as shown in Figure 8.

Satellite peak

The emission spectra of Rb+He is of great interest for DPAL laser applications. The D2 line of rubidium when perturbed by helium has been observed to have a satellite peak at 735nm[6]. The location of this peak can be predicted by collisional line shape models operating in the quasistatic limit[14] using the equation for the intensity measured relative to line center,

$$I(\omega) \propto \sum_{c} R_c^2 |D(R_c)|^2 \left| \frac{\mathrm{d}(\Delta V)}{\mathrm{d}R} \right|_{R_c}^{-1} \times n_{Ng} \exp\left(-\frac{X^2 \Sigma_{1/2}^+(R_c)}{k_B T}\right), \tag{6}$$

where $\mathrm{D}(R_c)$ is the transition dipole matrix element, n_{Ng} is the concentration of the noble gas, k_B is Boltzmann's constant, T is the absolute temperature, ΔV is the relevant difference potential (described below), $X^2\Sigma_{1/2}^+(R_c)$ is the ground state PEC, and $R_c(\omega)$ are Condon points given by the solutions to the equation $\Delta V(R_c) = \hbar \omega [43]$. Here it is assumed that the concentration of the rubidium gas is low relative to n_{He} , and the line broadening occurs only as a result of Rb + He collisions.

In order to use Eq. (6) to evaluate the satellite peak of the rubidium D2 line, we must first identify the relevant difference potential. The D2 line involves both the difference of the $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ with the ground state since both of these excited states have asymptotic limits that correspond to the $^2P_{3/2}$ state of the rubidium atom[13]. However, it is only the difference potential from the $B^2\Sigma_{1/2}^+$ that influences the satellite peak position. This difference potential can be explicitly written as,

$$\Delta V = \left(B^2 \Sigma_{1/2}^+(R) - X^2 \Sigma_{1/2}^+(R) \right) - \left(\lim_{R \to \infty} B^2 \Sigma_{1/2}^+(R) - \lim_{R \to \infty} X^2 \Sigma_{1/2}^+(R) \right).$$
 (7)

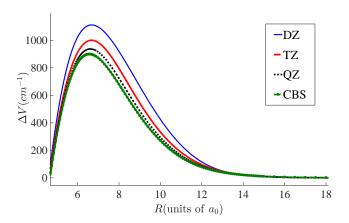


FIG. 9. The difference potential between the $B^2\Sigma_{1/2}^+$ and $X^2\Sigma_{1/2}^+$ PECs calculated at the double zeta(DZ), triple zeta(TZ), quadruple zeta(QZ), and extrapolated (CBS) basis set levels.

TABLE IV. Position (nm) of the satellite on the D2 line of rubidium perturbed by helium.

	DZ	TZ	QZ	CBS	Experiment[6]
satellite position	716	722	725	727	735

The difference potentials calculated by Equation 7 at different levels of basis set are presented in Figure 9. It is primarily the shoulder of the $B^2\Sigma_{1/2}^+$ PEC discussed above which leads to the extremum depicted in Figure 9. The lowering of the shoulder that occurs as the valence quality of the basis increases causes a corresponding reduction of the maximum energy of this extremum. The value of energy at which the extremum occurs corresponds directly to the frequency at which the satellite peak will appear. Table IV tabulates the results of these positions calculated at the DZ, TZ, QZ, and CBS level and a systematic improvement in the predicted satellite line positions is observed. In regards to the satellite line position, preliminary calculations (not presented here) show that all electron basis set effects the $B^2\Sigma_{1/2}^+$ surface the most, over the internuclear separation range of 4-10 Bohr. The $B^2\Sigma_{1/2}^+$ surface is higher in energy (and altered shape) when compared with the PEC calculated with core potentials, resulting in larger value of difference potential. This results in blueshift of satellite line position compared to the core potential predictions. A systematic study covering a hierarchy of all-electron basis sets followed by complete basis set extrapolation is required to make any conclusive inferences about the effect of core potential on satellite line position predictions.

Broadening and shifting coefficients

The broadening and shifting coefficients of Rb perturbed by a buffer gas of He atoms have been calculated using difference potentials derived from the DZ, TZ, QZ, and CBS PECs, as described in the previous sections. We calculate the temperature dependent coefficients using the Anderson Talman (AT) theory of spectral line broadening [13] and present the results in Table V. We observe a systematic monotonic increase in line broadening and line shift predictions as a function of basis set size. In the case of the D1 line, this trend is explained by a close examination of the $A^2\Pi_{1/2} - X^2\Sigma_{1/2}^+$ DP in the asymptotic limit, as shown in Figure 10. As the basis set size is increased, the value of internuclear separation at which the DP reaches its asymptotic value moves to larger internuclear separation (moves outward). The broadening and shifting coefficients using AT theory can be expressed as the sum of an effective hard-sphere contribution and a long-range contribution, where, hardsphere contribution is observed to have a very small effect on shifting coefficients[13]. The hard-sphere contribution increases with basis set size because the DP falloff moves to larger internuclear separation and also the long-range contribution increases because the asymptotic energy value increases with the basis set size. The net result of these contributions results in the monotonic trend observed in line broadening and line shift. The D1 line broadening values at 343K deviate from experimentally measured values by approximately 50-60\% and the deviation is in the range of approximately 10-20% for 394K and 450K. The D1 line broadening value at 394K computed with TZ basis set is within 6% of previous theoretical estimate of Blank and Weeks[13], which is to be expected as the basis set used in their calculation was of similar TZ quality. The calculated D1 line shift values exhibit smaller deviation when compared with experimentally measured values - underestimating for smaller basis set and overestimating in the CBS limit. The observation for the D2 line is analogous, though it is more complicated because it depends on both the $A^2\Pi_{3/2} - X^2\Sigma_{1/2}^+$ and $B^2\Sigma_{1/2}^+ - X^2\Sigma_{1/2}^+$ DPs. Results in Table V are also compared with broadening and shifting coefficients computed using fully quantum-mechanical calculations that employ the Baranger theory of collisional line broaden-

CONCLUSIONS

A hierarchy of segmented contracted basis sets at the DZ, TZ, and QZ level of valence quality are used to compute the $X^2\Sigma_{1/2}^+$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ potential energy curves of Rb+He. The calculations are performed at the MRCISD level where the Rb nucleus and core electrons are represented by a SOECP[18] and the Rb atom's valance electrons and the He atom's electrons

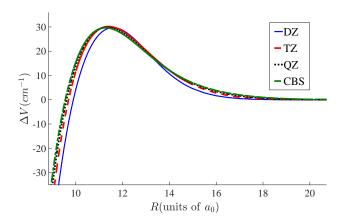


FIG. 10. The difference potential between the $A^2\Pi_{1/2}^+$ and $X^2\Sigma_{1/2}^+$ PECs calculated at the double zeta(DZ), triple zeta(TZ), quadruple zeta(QZ), and extrapolated (CBS) basis set levels.

TABLE V. Pressure broadening and shift rates. In the table of rates below the Broadening (γ) and Shift (δ) rates are given in cm^{-1}/cm^{-3} .

	, em .					
T(K)	Basis Set	Ι)1	Ι	D2	
		$\gamma \ (\times 10^{-20})$	$\delta \ (\times 10^{-21})$	$\gamma \ (\times 10^{-20})$	$\delta (\times 10^{-3})$	
343	DZ	1.401	4.876	1.152	0.0832	
	TZ	1.492	5.41	1.181	0.45	
	QZ	1.53	6.02	1.198	0.90	
	CBS	1.554	6.403	1.21	1.18	
	$\operatorname{Exp}\left[6\right]$	0.954	5.45	1.01	0.24	
394	DZ	1.460	5.496	1.22	0.0103	
	TZ	1.558	6.104	1.252	0.050	
	QZ	1.595	6.741	1.27	0.0965	
	CBS	1.62	7.15	1.28	1.26	
	$\operatorname{Exp}[44]$	1.29	6.41	1.36	0.504	
	Theory [45]	1.07	-7.89	1.45	-1.54	
	Theory [13]	1.47	7.89	1.35	1.5	
450	DZ	1.51	6.11	1.29	0.124	
	TZ	1.62	6.8	1.32	0.55	
	QZ	1.65	7.46	1.34	1.03	
	CBS	1.68	7.89	1.35	1.34	
	Exp [46]	1.33	6.42	0.972	2.74	

are treated explicitly. Basis set superposition error is removed through a counterpoise correction and size consistency error is accommodated with a Davidson-Silver correction. Potential energy curves computed at the DZ, TZ, and QZ level are extrapolated to obtain $X^2\Sigma_{1/2}^+$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ potential energy curves in the CBS limit. The DZ, TZ, QZ, and CBS potential energy curves are used to compute $A^2\Pi_{3/2}$ vibrational energies, the position of the D2 satellite peak, and the broadening and shifting of the D1 and D2 lines. The potential energy surfaces calculated in the present work are

provided as supplementary material.

As expected, the size of the counterpoise correction to the potential energy curves becomes smaller as the valence quality of the basis increases, with the primary effect of reducing the depth of the small $X^2\Sigma_{1/2}^+$ well at $R=14.4(a_0)$ in the CBS limit. The effect of the counterpoise correction on the $A^2\Pi_{1/2}$ and $A^2\Pi_{3/2}$ wells is counterbalanced by an increase in well depth as the valence quality of the basis increases, and the difference between the $A^2\Pi_{3/2}$ vibrational energies converges toward the experimental result.

Since BSSE is nearly identical for the $X^2\Sigma_{1/2}^+$, $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ potential energy curves at all levels of valence quality, difference potentials used to compute line shape features are not influenced by the counterpoise correction. On the other hand, line shape features are strongly influenced by the valence quality of the basis used to compute the potential energy curves and the corresponding difference potentials. For example, the shoulder on the $B^2\Sigma_{1/2}^+$ potential energy curve is significantly lowered as the valence quality increases. This causes the maximum energy in the difference potential between the $B^2\Sigma_{1/2}^+$ and $X^2\Sigma_{1/2}^+$ potential energy curves, to corre-2SP ondingly decrease. This maximum energy corresponds , to the position of the blue shifted satellite peak and it follows that the D2 satellite peak becomes less blue shifted and approaches the experimental value as the valence quality increases. As with the $A^2\Pi_{3/2}$ vibrational energy differences, the D2 satellite peak is best predicted using the CBS potential energy surface.

Anderson-Talman line broadening theory is used together with difference potentials between the ground $X^2\Sigma_{1/2}^+$ potential energy curve and the excited $A^2\Pi_{1/2}$, $A^2\Pi_{3/2}$ and $B^2\Sigma_{1/2}^+$ potential energy curves to compute the broadening and shifting coefficients of the D1 and D2 lines. As with the D2 satellite peak, the counterpoise correction makes no contribution to the difference potentials and variations in broadening and shifting coefficients are determined by valence quality. As valence quality increases, the D1 broadening coefficient diverges away from the experimental measurement, the D1 shifting coefficient first approaches and then diverges from the experimental measurement, the D2 shifting coefficient does appear to converge to the experimental measurement, and the D2 shifting coefficient first approaches and then diverges from the experimental measurement. As noted in Ref.[13], both the broadening and shifting coefficients are highly sensitive to small changes in the asymptotic form of the difference potential and, as seen with these results, an accurate calculation of line shape coefficients remains a significant challenge.

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- W. Krupke, R. Beach, V. Kanz, and S. Payne, Optics Letters 28, 2336 (2003).
- [2] R. J. Beach, W. F. Krupke, V. K. Kanz, and S. A. Payne, J. Opt. Soc. Am. B 21, 2151 (2004).
- [3] B. V. Zhdanov, T. Ehrenreich, and R. J. Knize, Opt. Commun. 260, 696 (2006).
- [4] R. H. Page, R. J. Beach, V. K. Kanz, and W. F. Krupke, Opt. Lett. 31, 353 (2006).
- [5] B. V. Zhdanov, J. Sell, and R. J. Knize, Electron. Lett. 44, 582 (2008).
- [6] W. S. Miller, C. A. Rice, G. D. Hager, M. D. Rotondaro, H. Berriche, and G. P. Perram, J. Quant. Spectrosc. Radiat. Transfer 184, 118 (2016).
- [7] W. S. Miller, C. A. Rice, and G. P. Perram, J. Quant. Spectrosc. Radiat. Transfer 206, 151 (2018).
- [8] G. Pitz, D. Wertepny, and G. Perram, Phys. Rev. A 80, 062718 (2009).
- [9] G. Pitz, C. Fox, and G. Perram, Phys. Rev. A 82, 042502 (2010).
- [10] G. D. Hager, G. E. Lott, A. J. Archibald, L. Blank, D. E. Weeks, and G. P. Perram, J. Quant. Spectrosc. Radiat. Transfer 147, 261 (2014).
- [11] C. D. Lewis and D. E. Weeks, J. Phys. Chem. A 121, 3340 (2017).
- [12] B. Eshel, J. A. Cardoza, D. E. Weeks, and G. P. Perram, Phys. Rev. A 95, 042708 (2017).
- [13] L. Blank and D. E. Weeks, Phys. Rev. A 90, 022510 (2014).
- [14] N. Allard and J. Kielkopf, Reviews of Modern Physics 54, 1103 (1982).
- [15] N. F. Allard, A. Royer, J. F. Kielkopf, and N. Feautrier, Phys. Rev. A 60, 1021 (1999).
- [16] H.-J. Werner and P. J. Knowles, J. Chem. Phys. 82, 5053 (1985).
- [17] P. J. Knowles and H.-J. Werner, Chem. Phys. Lett. 115, 259 (1985).
- [18] I. S. Lim, P. Schwerdtfeger, B. Metz, and H. Stoll, J. Chem. Phys. 122, 104103 (2005).

- [19] P. J. Knowles and H.-J. Werner, Chem. Phys. Lett. 145, 514 (1988).
- [20] H.-J. Werner and P. J. Knowles, J. Chem. Phys. 89, 5803 (1988).
- [21] P. J. Knowles and H.-J. Werner, Theor. Chim. Acta 84, 95 (1992).
- [22] E. R. Davidson and D. W. Silver, Chem. Phys. Lett. 52, 403 (1977).
- [23] F. Weigend and R. Ahlrichs, Phys. Chem. Chem. Phys. 7, 3297 (2005).
- [24] A. Schaefer, H. Horn, and R. Ahlrichs, J. Chem. Phys. 97, 2571 (1992).
- [25] A. Schaefer, C. Huber, and R. Ahlrichs, J. Chem. Phys. 100, 5829 (1994).
- [26] F. Weigend, F. Furche, and R. Ahlrichs, J. Chem. Phys. 119, 12753 (2003).
- [27] T.Leininger, A.Nicklass, W.Kuechle, H.Stoll, M.Dolg, and A.Bergner, Chem. Phys. Lett. 255, 274 (1996).
- [28] F. Weigend and A. Baldes, J. Chem. Phys. 133, 174102 (2010).
- [29] H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby, M. Schütz, et al., "Molpro, version 2012.1, a package of ab initio programs," (2012), see http://www.molpro.net.
- [30] S. F. Boys and F. Bernardi, Mol. Phys. 19, 553 (1970).
- [31] D. Asturiol, M. Duran, and P. Salvador, J. Chem. Phys. 128, 144108 (2008).
- [32] K. A. Peterson, D. E. Woon, and J. Thorn H. Dunning, J. Chem. Phys. 100, 7410 (1994).
- [33] W. E. Baylis, J. Chem. Phys. **51**, 2665 (1969).
- [34] J. Pascale and J. Vandeplanque, J. Chem. Phys. 60, 2278 (1974).
- [35] N. Y. Du and C. H. Greene, J. Chem. Phys. 90, 6347 (1989).
- [36] E. de Prunelé, Phys. Rev. A 36, 3015 (1987).
- [37] N. Y. Du, Interacting Rydberg Channels in Diatomic Molecules, Ph.D. thesis, Louisiana State University (1989).
- [38] L. Blank, G. S. Kedziora, and D. E. Weeks, J. Chem. Phys. 136, 124315 (2012).
- [39] J. Pascale, Phys. Rev. A 28, 632 (1983).
- [40] M. Zbiri and C. Daul, J. Chem. Phys. 121, 11625 (2004).
- [41] K. Hirano, K. Enomoto, M. Kumakura, Y. Takahashi, and T. Yabuzaki, Phys. Rev. A 68, 012722 (2003).
- [42] M. Mudrich, F. Stienkemeier, G. Droppelmann, P. Claas, and C. P. Schulz, Phys. Rev. Lett. 100, 023401 (2008).
- [43] J. Szudy and W. Baylis, Physics Reports 266, 127 (1996).
- [44] M. Rotondaro and G. Perram, J. Quant. Spectrosc. Radiat. Transfer 57, 497 (1997).
- [45] R. D. Loper, Collisional broadening and shift of D1 and D2 spectral lines in atomic alkali-vapor noble-gas system, Ph.D. thesis, Air Force Institute of Technology (2013).
- [46] E. Roueff and A. Suzor, J. Phys. France 35, 727 (1974).