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Spring 2020

CHEM 221-004: Analytical Chemical Methods

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Chemistry: *Spring 2020 Course Syllabus*

[NJIT Academic Integrity Code](#): All Students should be aware that the Department of Chemistry & Environmental Science (CES) takes the University Code on Academic Integrity at NJIT very seriously and enforces it strictly. This means that there must not be any forms of plagiarism, i.e., copying of homework, class projects, or lab assignments, or any form of cheating in quizzes and exams. Under the University Code on Academic Integrity, students are obligated to report any such activities to the Instructor.

COURSE INFORMATION

CHEM 221 Course Description: The objective of this course is to teach the students the science of chemical measurements. Often, scientists and engineers are faced with the question, what is in a given sample, and how much? Chemical measurement or analytical chemistry is the systematic study of methods which can be used to answer these questions.

This course has been designed for students taking their first laboratory course in quantitative methods of chemical analysis. The topics covered fall under the general category of Analytical Chemistry. By the end of the semester, the students are expected to have acquired fundamental knowledge to prepare them to embark upon a rational approach to qualitative and quantitative analysis. After reviewing the large number of topics and experiments that could have been included here, we have selected a small set of diverse experiments that we believe will provide fundamental knowledge of various popular and powerful analytical techniques. These experiments include data evaluation, gravimetric, complexometric/volumetric/potentiometric titration, and UV-Vis molecular spectrophotometry.

It will be to the student's advantage to continue to read and reread the chapters in their textbooks on laboratory techniques throughout the semester. There are also some practical aspects of chemical analysis that are best learnt during the experiments. For example, you will learn that one should not use a graduated cylinder for accurate volumetric measurements, and that solids are best dissolved in a beaker covered with a watch glass. The laboratory professor or teaching assistant (TA) will usually explain and/or demonstrate these and many more techniques which have been used successfully over the years.

Students are encouraged to ask questions before it is too late and the mistakes have already been committed.

Number of Credits: 2

Prerequisites: CHEM 222

Course-Section and Instructors

Course-Section	Instructor
CHEM 221-004 (CRN 15348 , Wednesday, 8:30AM - 12:05PM, Tier 208) Office hours: Wednesday 4 pm or by appointment.	Dr. Omowunmi "Wunmi" Sadik Email: sadik@njit.edu Telephone: 973-596-2833 Office: Tiernan 151

E-Mail: All E-mail to me should start with CHEM 221 in the subject so that it can be filtered appropriately. Any e-mail pertaining to your academic standing (i.e., grades) must be sent from your NJIT account. Anonymous e-mail will not be read.

Lab manual is required.

Reference Textbook: *Quantitative Chemical Analysis*, 9th Ed., D.C. Harris, WH Freeman, NY, 2016, ISBN-10: 1-4641-3538-X

Secondary reference: *Fundamentals of Analytical Chemistry*, ninth edition, by Douglas A. Skoog, Donald M. West, F. James Holler, and Stanley R. Crouch, Brooks/Cole 2014, ISBN-10 0-495-55832-X.

POLICIES

All CES students must familiarize themselves with, and adhere to, all official university-wide student policies. CES takes these policies very seriously and enforces them strictly.

Grading Policy: The final grade in this course will be determined as follows:

Seven labs (each 10%)	80%
Two quizzes (each 5%)	5%
Exam	10%
Lab behavior and lab cleanness	5%

Your final letter grade in this course will be based on the following tentative curve:

A	90-100	C	70-75
B+	86-89	D	60-69
B	80-85	F	<60
C+	76-79		

Makeup Exam Policy: There will normally be **NO MAKE-UP QUIZZES OR EXAMS** during the semester. In the event that a student has a legitimate reason for missing a quiz or exam, the student should contact the Dean of Students office and present written verifiable proof of the reason for missing the exam, e.g., a doctor's note, police report, court notice, etc. clearly stating the date AND time of the mitigating problem. The student must also notify the CES Department Office/Instructor that the exam will be missed so that appropriate steps can be taken to make up the grade.

Cellular Phones: All cellular phones and other electronic devices must be switched off during all lab times. Such devices must be stowed in bags during exams or quizzes.

ADDITIONAL RESOURCES

Chemistry Tutoring Center: Located in the Central King Building, Lower Level, Rm. G12. Hours of operation are Monday - Friday 10:00 am - 6:00 pm. For further information please click [here](#).

Accommodation of Disabilities: Office of Accessibility Resources and Services (*formerly known as Disability Support Services*) offers long term and temporary accommodations for undergraduate, graduate and visiting students at NJIT.

If you are in need of accommodations due to a disability please contact Chantonette Lyles, Associate Director at the Office of Accessibility Resources and Services at 973-596-5417 or via email at lyles@njit.edu. The office is located in Fenster Hall Room 260. A Letter of Accommodation Eligibility from the Office of Accessibility Resources Services office authorizing your accommodations will be required.

For further information regarding self-identification, the submission of medical documentation and additional support services provided please visit the Accessibility Resources and Services (OARS) website at:

- <http://www5.njit.edu/studentsuccess/disability-support-services/>

Course Outline

Week	Date	Topic	Assignment
1	Jan 22	Check in/Safety Lecture	
2	Jan 29	Lab #1: Techniques in Preparing Solutions	
3	Feb 5	Lab#2: Volumetric Glassware, Statistics, and Spreadsheet Exercise	
4	Feb 12	Lab#3: Determination of sulfate as barium sulfate	
5	Feb 19	Lab#3: Determination of sulfate as barium sulfate	
6	Feb 26	Lab #4: Percentage of Na ₂ CO ₃ in a sample	
7	Mar 4	Lab #4: Percentage of Na ₂ CO ₃ in a sample	
8	Mar 11	Lab #5: Determination of hardness of water	
9	Mar 18	Spring break	
10	Mar 25	Lab #6: Potentiometric titration of an acid mixture	
11	April 1	Lab #6: Potentiometric titration of an acid mixture	
12	Apr 8	Lab #7: Determination of trace iron using UV-Visible spectrophotometry	
13	Apr 15	Lab #8: Spectrophotometry of a two component mixture	
14	Apr 22	Make up	
15	Apr 29	Exam	
16	May 6	Check out	