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Chapter

Searching for Metabolic Pathways of Anaerobic Digestion: A Useful List of the Key Enzymes

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Abstract

The general scheme of anaerobic digestion is well known. It is a complex process promoted by the interaction of many groups of microorganisms and has four major steps: hydrolysis, acidogenesis, acetogenesis, and methanogenesis. The aim of the study was to prepare a systematized list of the selected enzymes responsible for the key pathways of anaerobic digestion based on the Kyoto Encyclopedia of Genes and Genomes database resource. The list contains (i) key groups of hydrolases involved in the process of degradation of organic matter; (ii) the enzymes catalyzing reactions leading to pyruvate formation; (iii) the enzymes of metabolic pathways of further pyruvate transformations; (iv) the enzymes of glycerol transformations; (v) the enzymes involved in transformation of gaseous or nongaseous products of acidic fermentations resulting from nonsyntrophic nutritional interactions between microbes; (vi) the enzymes of amino acid fermentations; (vii) the enzymes involved in acetogenesis; and (viii) the enzymes of the recognized pathways of methanogenesis. Searching for the presence and activity of the enzymes as well as linking structure and function of microbial communities allows to develop a fundamental understanding of the processes, leading to methane production. In this contribution, the present study is believed to be a piece to the enzymatic road map of anaerobic digestion research.

Keywords: anaerobic digestion, enzymes, hydrolysis, acidogenesis, acetogenesis, methanogenesis, syntrophy, metabolic pathways

1. Introduction

Anaerobic digestion (AD), whose final products are methane and carbon dioxide, is a common process in natural anoxic environments such as water sediments, wetlands, or marshlands. The environments have to be rich in organic matter and poor with other electron acceptors such as nitrate, compounds containing oxidized forms of metals, and sulfate. AD is also common in landfills and wastewater treatment plants and was used by man to produce biogas from waste biomass as an alternative energy source.

AD is a complex process that requires the metabolic interaction of many groups of microorganisms responsible for four closely related major steps. The first one is hydrolysis of complex organic polymers (e.g., polysaccharides, lipids, proteins) to monomers (sugars, fatty acids, amino acids). The second step is acidogenesis that results in formation of hydrogen and carbon dioxide as well as nongaseous fermentation products, that is, low-molecular-weight organic acids and alcohols. These products are further oxidized to hydrogen, carbon dioxide, and acetate in acetogenic step that involves mainly syntrophic degradation of nongaseous fermentation products. The fourth step is methanogenesis. Three groups of substrates for methane production and three types of methanogenesis); reduction of CO_2 with H_2 or formate and rarely ethanol or secondary alcohols as electron donors (hydrogen-dependent and hydrogen-independent methylotrophic methanogenesis). The two last steps, acetogenesis and methanogenesis, are closely related and involve syntrophic associations between hydrogen-producing acetogenic bacteria and hydrogenotrophic methanogenes (Figure 1) [1–5].

Recently, there has been a rapid development in culture-independent techniques (meta-omics approaches such as metagenomics, metatranscriptomics, metaproteomics, metabolomics) for exploring microbial communities, which have led to a new insight into their structure and function in both natural environments and anaerobic digesters. The current trends involve the combined use of meta-omic approaches and detailed reactor performance data as well as isotope labeling techniques that allow us to develop a fundamental understanding of the processes occurring in AD. Those activities are aimed to improve

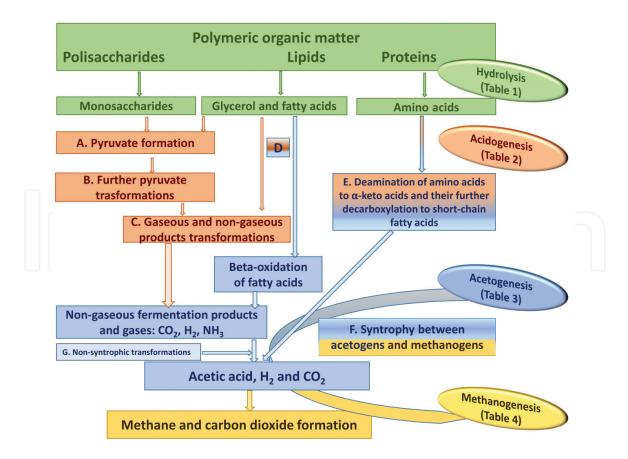


Figure 1.

A scheme of anaerobic digestion of organic matter. Enzymes catalysing specific reactions of AD are presented in **Tables 1–4**. Thus in Figure 1 there are the links to **Tables 1–4**. Furthermore, background colours in the Figure correspond to the background colours of the title rows in the **Tables 1–4**: hydrolysis is indicated in green, acidogenesis in orange, acetogenesis in blue and methanogenesis in yellow. A, B, C, D, E refer to the title rows in **Table 2**; F, G refer to the title rows in **Table 3**.

biogas production and increase the share of renewable energy in total energy consumption [6–9].

Analysis of many studies on metagenomes of microbial communities from anaerobic digesters shows that (i) contribution of methanogens in the methaneyielding microbial communities is relatively small, below 20%; (ii) the most abundant phyla of bacteria are usually *Firmicutes, Bacteroidetes, Proteobacteria*, and *Actinobacteria*; (iii) methanogenic archaea are dominated by acetotrophs or hydrogenotrophs with a certain contribution of methylotrophs; (iv) substrate, operational conditions such as temperature, pH, ammonia concentration, etc. shape the structure, percentage distribution of specific taxons, and functioning of the community of microorganisms; (v) it is important to describe interactions within microbial communities and assign functions in AD steps to specific groups of microbes; and (vi) the majority of sequences are not classified at the genus level confirming that most of the microorganisms are still unrecognized [6, 10–15].

In this contribution, the purpose of the study was to prepare a list of the selected enzymes and their catalyzed reactions, being a specific enzymatic road map of AD metabolic pathways, useful in molecular studies. The available metabolic pathway databases such as KEGG PATHWAY Database [16–18], MetaCyc Metabolic Pathway Database, BioCyc Database Collection [19], and BRENDA—The Comprehensive Enzyme Information System [20] were used to select metabolic pathways dedicated only to AD from hydrolysis to methanogenic steps exerted by microbes.

2. Selected enzymes of anaerobic digestion

Figure 1 shows a scheme of AD and **Tables 1–4** present a summary of the selected enzymes and enzymatic reactions involved in decomposition of organic matter to methane and carbon dioxide. **Tables 1–4** are an extension of **Figure 1**, and in **Figure 1**, there are the links to **Tables 1–4**.

The key groups of hydrolases involved in the process of degradation of organic matter are esterases, glycosidases, and peptidases, which catalyze the cleavage of ester bonds, glycoside bonds, and peptide bonds, respectively (**Table 1**). **Table 1** also includes other classes of hydrolases such as acting on carbon-nitrogen bonds, other than peptide bonds.

In the acidogenic stage of AD, the key step is pyruvate formation from carbohydrates (**Table 2**, Part A) or other compounds and further pyruvate transformations toward short-chain fatty acids and ethanol (**Table 2**, Part B). The Part C of the **Table 2** also considers transformation of gaseous and nongaseous products of acidic fermentations, resulting from nonsyntrophic nutritional interaction between bacteria. The Parts D and E present the enzymes of glycerol and amino acid transformations, respectively. The latter requires syntrophic cooperation between microorganisms.

The enzymes catalyzing oxidation of nongaseous products of acidogenesis mainly butyrate, propionate, acetate, lactate, ethanol including the enzymes of reverse electron transfer (process responsible for energy conservation in syntrophically growing acetogens) are shown in **Table 3**.

The enzymes of the three recognized pathways of methanogenesis such as acetotrophic, hydrogenotrophic, and methylotrophic are listed in **Table 4**.

The data were prepared on the basis of detailed analysis of AD research. The enzyme nomenclature comes from the Kyoto Encyclopedia of Genes and Genomes (KEGG) database resource.

Hydrolytic enzyme	Reaction/process	EC number
Esterases	Acting on ester bonds	EC 3.1
Glycosidases	Acting on glycoside bonds	EC 3.2
	Acting on cellulose	
Cellulase; endo-1,4-beta-D- glucanase	Endohydrolysis of $(1 \rightarrow 4)$ -beta-D-glucosidic linkages in cellulose, lichenin, and cereal beta-D-glucans	EC 3.2.1.4
Cellulose 1,4-beta- cellobiosidase (nonreducing end)	Hydrolysis of $(1 \rightarrow 4)$ -beta-D-glucosidic linkages in cellulose and cellotetraose, releasing cellobiose from the nonreducing ends of the chains	EC 3.2.1.91
Beta-glucosidase	Hydrolysis of terminal, nonreducing beta-D-glucosyl residues with release of beta-D-glucose	EC 3.2.1.21
	Acting on hemicellulose	
Endo-1,4-beta-xylanase	Endohydrolysis of (1 \rightarrow 4)-beta-D-xylosidic linkages in xylans	EC 3.2.1.8
Xylan 1,4-beta-xylosidase	Hydrolysis of $(1 \rightarrow 4)$ -beta-D-xylans, to remove successive D-xylose residues from the nonreducing termini	EC 3.2.1.37
Mannan endo-1,4-beta- mannosidase	Random hydrolysis of $(1 \rightarrow 4)$ -beta-D-mannosidic linkages in mannans, galactomannans, and glucomannans	EC 3.2.1.78
Beta-mannosidase	Hydrolysis of terminal, nonreducing beta-D-mannose residues in beta-D-mannosides	EC 3.2.1.25
Alpha-galactosidase	Hydrolysis of terminal, nonreducing alpha-D-galactose residues in alpha-D-galactosides, including galactose oligosaccharides, galactomannans, and galactolipids	EC 3.2.1.22
Alpha-glucuronidase	An alpha-D-glucuronoside + $H_2O \rightarrow$ an alcohol + D-glucuronate	EC 3.2.1.139
Peptidases	Acting on peptide bonds	EC 3.4
	Other hydrolases	
Hydrolases acting on carbor bonds	n-nitrogen bonds, other than peptide	EC 3.5
Hydrolases acting on ether l	ponds	EC 3.3
Hydrolases acting on carbor	n-carbon bonds	EC 3.7
Hydrolases acting on halide	bonds	EC 3.8
Hydrolases acting on phospl bonds	horus-nitrogen	EC 3.9
Hydrolases acting on sulfur- bonds	nitrogen	EC 3.10
Hydrolases acting on carbor bonds	n-phosphorus	EC 3.11
Hydrolases acting on sulfur- bonds	-sulfur	EC 3.12
Hydrolases acting on carbor	n-sulfur bonds	EC 3.13
, 0		

Table 1.The selected enzymes of hydrolytic step of anaerobic digestion [21, 22].

Enzyme	Reaction/process	EC number
A. Pyruva	ate formation from carbohydrates [23]	
Glycolysis (the Embden-Meyerhof-Parnas pathway)	
Hexose kinase	$D\text{-}Glucose + ATP \leftrightarrow D\text{-}glucose\text{-}6\text{-}phosphate + ADP$	EC 2.7.1.1
Phosphoglucose isomerase	$D\text{-}Glucose 6\text{-}phosphate \leftrightarrow D\text{-}fructose 6\text{-}phosphate$	EC 5.3.1.9
Phosphofructose kinase	ATP + D-fructose 6-phosphate \leftrightarrow ADP + D- fructose 1,6-bisphosphate	EC 2.7.1.11
Fructose-bisphosphate aldolase	Fructose-1,6-bisphosphate ↔ dihydroxyacetone phosphate + glyceraldehyde-3-phosphate	EC 4.1.2.13
Triose phosphate isomerase	Glyceraldehyde 3-phosphate \leftrightarrow dihydroxyacetone phosphate	EC 5.3.1.1
Glyceraldehyde-3-phosphate dehydrogenase	D-Glyceraldehyde 3-phosphate + phosphate + NAD ⁺ \leftrightarrow 1,3-bisphosphoglycerate + NADH + H ⁺	EC 1.2.1.12
Phosphoglycerate kinase	1,3-Bisphosphoglycerate + ADP \leftrightarrow 3- phosphoglycerate + ATP	EC 2.7.2.3
Phosphoglycerate mutase	3-Phosphoglycerate \leftrightarrow 2-phosphoglycerate	EC 5.4.2.1
Enolase	2-Phospho-D-glycerate \leftrightarrow phosphoenolpyruvate + H_2O	EC 4.2.1.11
Pyruvate kinase	$Phosphoenolpyruvate + ADP \leftrightarrow pyruvate + ATP$	EC 2.7.1.40
2-Keto-3-deoxy-6-p	hosphogluconate (the Entner-Doudoroff pathway)	
Glucose-6-phosphate dehydrogenase	D-glucose 6-phosphate + NADP ⁺ \leftrightarrow 6-phospho-D-glucono-1,5-lactone + NADPH + H ⁺	EC 1.1.1.49
Phosphogluconate dehydrogenase	6-Phospho-D-gluconate + NAD(P) ⁺ \leftrightarrow 6-phospho- 2-dehydro-D-gluconate + NAD(P)H + H ⁺	EC 1.1.1.43
2-Keto-3-deoxy-6- phosphogluconate aldolase	2-Dehydro-3-deoxy-6-phospho-D-gluconate ↔ pyruvate + D-glyceraldehyde 3-phosphate	EC 4.1.2.14
B. Further transformation	tions of pyruvate—glycolytic fermentations [23-27	7]
Lactate dehydrogenase	$Pyruvate + NADH \leftrightarrow lactate + NAD^{*}$	EC 1.1.1.27
Pyruvate:ferredoxin oxidoreductase, PFOR	Pyruvate + CoA + oxidized Fd \leftrightarrow acetyl-CoA + reduced Fd + CO ₂ + H ⁺	EC 1.2.7.1
NADH:ferredoxin oxidoreductase, NFOR	$\textbf{Oxidized Fd} + \textbf{NADH} \leftrightarrow \textbf{reduced Fd} + \textbf{NAD}^{\star} + \textbf{H}^{\star}$	EC 1.18.1.3
Ferredoxin hydrogenase	2 reduced ferredoxin + 2 $H^+ \leftrightarrow H_2$ + 2 oxidized ferredoxin	EC 1.12.7.2
Phosphotransacetylase	$CoA + acetyl phosphate \leftrightarrow acetyl-CoA + phosphate$	EC 2.3.1.8
Acetate kinase	ATP + acetate \leftrightarrow ADP + acetyl phosphate	EC 2.7.2.1
NAD ⁺ -dependent ethanol dehydrogenase	Acetaldehyde + NADH + $H^+ \leftrightarrow$ ethanol + NAD ⁺ An aldehyde + NADH + $H^+ \leftrightarrow$ a primary alcohol + NAD ⁺	EC 1.1.1.1
Acetaldehyde dehydrogenase	Acetaldehyde + CoA + NAD ⁺ \leftrightarrow acetyl-CoA + NADH + H ⁺	EC 1.2.1.10
Acetyl-CoA acetyltransferase	$\texttt{2-acetyl-CoA} \leftrightarrow \texttt{CoA} + \texttt{acetoacetyl-CoA}$	EC 2.3.1.9
3-Hydroxybutyryl-CoA dehydrogenase	3-Acetoacetyl-CoA + NADPH + $H^+ \leftrightarrow$ 3-hydroxybutanoyl-CoA + NADP ⁺	EC 1.1.1.157
Crotonase 3-OH-butyryl-CoA dehydratase	$\texttt{3-Hydroxybutanoyl-CoA} \leftrightarrow \texttt{crotonoyl-CoA} + \texttt{H}_2\texttt{O}$	EC 4.2.1.55

Enzyme	Reaction/process	EC number
	$A \rightarrow 2$ NAD+ reduced Fd + butyryl-CoA catalyzed by ase/electron-transfer flavoprotein complex	butyryl CoA
Butyryl-CoA dehydrogenase	A short-chain acyl-CoA + electron-transfer flavoprotein ↔ a short-chain trans- 2,3-dehydroacyl-CoA + reduced electron-transfer flavoprotein	EC 1.3.8.1
Butyryl-CoA dehydrogenase/Etf complex	Butanoyl-CoA + 2 NAD ⁺ + 2 reduced Fd \leftrightarrow Crotonoyl-CoA + 2 NADH + 2 oxidized Fd	EC 1.3.1.109
Phosphotransbutyrylase	Butanoyl-CoA + phosphate \leftrightarrow CoA + butanoyl phosphate	EC 2.3.1.19
Butyrate kinase	Butanoyl phosphate + ADP \leftrightarrow butanoate + ATP	EC 2.7.2.7
PFL—pyruvate formate lyase	Pyruvate + CoA \leftrightarrow acetyl-CoA + formate	EC 2.3.1.54
FHL—formate hydrogen lyase	$Formate \rightarrow H_2 + CO_2$	EC 1.17.99.7
Pyruvate carboxylase	ATP + pyruvate + $HCO_3^- \leftrightarrow ADP$ + phosphate + oxaloacetate	EC 6.4.1.1
Malate dehydrogenase	$Malate + NAD^{+} \leftrightarrow oxaloacetate + NADH + H^{+}$	EC 1.1.1.37
Fumarate hydratase	$Malate \leftrightarrow fumarate + H_2O$	EC 4.2.1.2
Fumarate reductase	Fumarate + a quinol \leftrightarrow succinate + a quinone	EC 1.3.5.4
	$Fumarate + NADH \leftrightarrow succinate + NAD^{+}$	EC 1.3.1.6
Succinyl-CoA synthetase	GTP + succinate + CoA = GDP + phosphate + succinyl-CoA	EC 6.2.1.4
Methylmalonyl CoA mutase	Succinyl-CoA \leftrightarrow (R)-methylmalonyl-CoA	EC 5.4.99.2
Methylmalonyl CoA epimerase	(R)-methylmalonyl-CoA \leftrightarrow (S)-methylmalonyl-CoA	EC 5.1.99.1
Methylmalonyl-CoA decarboxylase	(S)-methylmalonyl-CoA \leftrightarrow propanoyl-CoA + CO ₂	EC 4.1.1.41
Propionate-CoA transferase	Acetate + propanoyl-CoA ↔ acetyl- CoA + propanoate	EC 2.8.3.1
C. Transformation of gaseous ar	nd nongaseous products of acidic fermentations (the examples)	he selected
Transformation of lactate and acc	etate to butyrate, hydrogen, and carbon dioxide ([28 therein)] and cited
Lactate dehydrogenases	(S)-lactate + NAD ⁺ \leftrightarrow pyruvate + NADH + H ⁺	EC 1.1.1.27
	(R)-lactate + NAD ⁺ \leftrightarrow pyruvate + NADH + H ⁺	EC 1.1.1.28
	Lactate + 2 NAD ⁺ + 2 reduced Fd \leftrightarrow pyruvate + 2 NADH + 2 oxidized Fd See Table 3	EC 1.3.1.110

Pyruvate is oxidized to acetyl coenzyme A, which is further routed to acetate and butyrate with hydrogen release. See Part B: Further transformations of pyruvate—glycolytic fermentations

Transformation of ethanol and acetate to butyrate and hydrogen in Clostridium kluyveri [29]		
Acetate kinase	See Part B. Further transformations of pyruvate—	EC 2.7.2.1
Acetyl-CoA acetyltransferase	glycolytic fermentations	EC 2.3.1.9
3-Hydroxybutyryl-CoA dehydrogenase		EC 1.1.1.157
3-Hydroxyacyl-CoA dehydratase		EC 4.2.1.55
Butyryl-CoA dehydrogenase/Etf complex		EC 1.3.1.109

Enzyme	Reaction/process	EC number
Acetate CoA-transferase	Acyl-CoA + acetate \leftrightarrow a fatty acid anion + acetyl-CoA	EC 2.8.3.8
Reductive carbon monoxide dehyd	rogenase/acetyl-CoA synthase pathway (reductive C [30]	CODH/ACS)
NADP-dependent formate dehydrogenase	$CO_2 + NADPH \leftrightarrow formate + NADP^*$	EC 1.17.1.10
Formyltetrahydrofolate synthetase	ATP + formate + tetrahydrofolate \leftrightarrow ADP + phosphate + 10-formyltetrahydrofolate	EC 6.3.4.3
Methenyltetrahydrofolate cyclohydrolase	10-Formyltetrahydrofolate \leftrightarrow 5,10- methenyltetrahydrofolate + H ₂ O	EC 3.5.4.9
NADP-dependent methylenetetrahydrofolate dehydrogenase	5,10-Methenyltetrahydrofolate + NADPH + $H^+ \leftrightarrow$ 5,10-Methylenetetrahydrofolate + NADP ⁺	EC 1.5.1.5
Ferredoxin-dependent methylenetetrahydrofolate reductase	5,10-Methylenetetrahydrofolate + 2 reduced Fd + 2 H ⁺ \leftrightarrow 5-methyltetrahydrofolate + 2 oxidized Fd	EC 1.5.7.1
5,10-Methylenetetrahydrofolate reductase	5,10-Methylenetetrahydrofolate + NAD(P)H + H ⁺ \leftrightarrow 5-methyltetrahydrofolate + NAD(P) ⁺	EC 1.5.1.20
5-Methyltetrahydrofolate: corrinoid/iron–sulfur protein Co- methyltransferase	$\begin{array}{l} [Co(I) \ corrinoid \ Fe-S \ protein] \ + \ 5-\\ methyltetrahydrofolate \ \leftrightarrow \ [methyl-Co(III) \\ corrinoid \ Fe-S \ protein] \ + \ tetrahydrofolate \end{array}$	EC 2.1.1.258
Carbon monoxide dehydrogenase	$CO_2 + 2 \text{ reduced Fd} + 2 \text{ H}^+ \leftrightarrow CO + H_2O + 2$ oxidized Fd	EC 1.2.7.4
CO-methylating acetyl-CoA synthase	CO + CoA + [methyl-Co(III) corrinoid Fe-S protein] \leftrightarrow acetyl-CoA + [Co(I) corrinoid Fe-S protein]	EC 2.3.1.169
D. 0	Glycerol transformations [31, 32]	
	Oxidative pathway	
Glycerol dehydrogenase	Glycerol + NAD ⁺ ↔ glycerone (dihydroxyacetone) + NADH + H ⁺	EC 1.1.1.6
Dihydroxyacetone kinase	ATP + glycerone \leftrightarrow ADP + glycerone phosphate	EC 2.7.1.29
For further	reactions, see Part A: Pyruvate formation	
	Reductive pathway	
Glycerol dehydratase	Glycerol \leftrightarrow 3-hydroxypropionaldehyde + H ₂ O	EC 4.2.1.30
1,3-Propanediol dehydrogenase	3-Hydroxypropionaldehyde + NADH + H $^+ \leftrightarrow$ 1,3-propanediol + NAD $^+$	EC 1.1.1.202
T A	mino acids fermentations [33–37]	

Syntrophy with H₂-scavenging microorganism: amino acid degradation involves NAD(P)- or FADdependent deamination of amino acids to the corresponding α -keto acids by amino acid dehydrogenases (EC 1.4.1.X): RCH(NH₄⁺)COO⁻ + H₂O \rightarrow RCOCOO⁻ + NH₄⁺ + H₂ and further conversion of α -keto acids via oxidative decarboxylation to fatty acids: RCOCOO⁻ + H₂O \rightarrow RCOO⁻ + CO₂ + H₂ [33]

Without syntrophy with H₂-scavenging microorganism: **Stickland Reaction**—coupled oxidation-reduction reactions between suitable amino acids (coupled deamination of amino acids); one member of the pair is oxidized (dehydrogenated) and the other is reduced (hydrogenated) [34], for example,

Alanine and glycine: alanine + 2 glycine + $3H_2O \rightarrow 3 \text{ acetate}^- + 3NH_4^+ + HCO_3^- + H^+$ Valine and glycine: valine + 2 glycine + $3H_2O \rightarrow \text{ isobutyrate}^- + 2 \text{ acetate}^- + 3NH_4^+ + HCO_3^- + H^+$ Leucine and glycine: leucine + 2 glycine + $3H_2O \rightarrow \text{ isovalerate}^- + 2 \text{ acetate}^- + 3NH_4^+ + HCO_3^- + H^+$

Enzyme	Reaction/process	EC number
-	ydrogenases catalyzing deamination of amino acid orresponding α-keto acids [33]	s to the
Aspartate dehydrogenase	L-aspartate + H ₂ O + NAD(P) ⁺ \leftrightarrow oxaloacetate + NH ₃ + NAD(P)H + H ⁺	EC 1.4.1.21
Valine dehydrogenase	L-valine + H_2O + NADP ⁺ \leftrightarrow 3-methyl-2- oxobutanoate + NH ₃ + NADPH + H ⁺	EC 1.4.1.8
Alanine dehydrogenase	L-alanine + H_2O + $NAD^+ \leftrightarrow$ pyruvate + NH_3 + $NADH$ + H^+	EC 1.4.1.1
Leucine dehydrogenase	L-leucine + H_2O + NAD+ \leftrightarrow 4-methyl-2- oxopentanoate + NH ₃ + NADH + H ⁺	EC 1.4.1.9
Key en	zymes of Stickland reaction [34–36]	
Glycine	e reductase GR pathway (grd operon)	
Glycine reductase	Glycine + phosphate + reduced thioredoxin + $H^+ \leftrightarrow$ acetyl phosphate + NH_3 + oxidized thioredoxin + H_2O	EC 1.21.4.2
Acetate kinase	Acetyl phosphate + ADP \leftrightarrow acetate + ATP	EC 2.7.2.1
Proline	reductase PR pathway (<i>prd</i> operon)	
D-proline reductase (dithiol)	D-proline + dihydrolipoate ↔5-aminopentanoate (5-aminovalerate) + lipoate	EC 1.21.4.1
	Others examples [33]	
Serine dehydratase	$\begin{array}{l} \text{L-serine} \leftrightarrow pyruvate + \text{NH}_3 \mbox{ (overall reaction)} \\ (1a) \mbox{ L-serine} \leftrightarrow 2\text{-aminoprop-2-enoate} + H_2O \\ (1b) \mbox{ 2-Aminoprop-2-enoate} \leftrightarrow 2\text{-iminopropanoate} \\ (spontaneous) \\ (1c) \mbox{ 2-Iminopropanoate} + H_2O \leftrightarrow pyruvate + \text{NH}_3 \\ (spontaneous) \end{array}$	EC 4.3.1.17
Threonine dehydratase	L-threonine \leftrightarrow 2-oxobutanoate + NH ₃ (overall reaction) (1a) L-threonine \leftrightarrow 2-aminobut-2-enoate + H ₂ O; (1b) 2-Aminobut-2-enoate \leftrightarrow 2-iminobutanoate (spontaneous) (1c) 2-Iminobutanoate + H ₂ O \leftrightarrow 2-oxobutanoate + NH ₃ (spontaneous)	EC 4.3.1.19
Detailed pathways of	glutamate fermentation via 3-methylaspartate [37]	
Glutamate mutase (methylaspartate mutase)	L-glutamate ↔ L-threo-3-methylaspartate	EC 5.4.99.1
Methyl aspartase	L-threo-3-methylaspartate \leftrightarrow mesaconate (2-methylfumarate) + NH ₃	EC 4.3.1.2
Mesaconase (2-methylmalate dehydratase)	2-Methylfumarate + H2O \leftrightarrow (S)-2-methylmalate	4.2.1.34
Citramalate lyase	 (2S)-2-hydroxy-2-methylbutanedioate ↔ acetate + pyruvate (S)-2-methylmalate = 2-hydroxy-2-methylbutanedioate 	4.1.3.22
For further transform	ations of pyruvate to acetate and butyrate, see Part B.	
For further trans	formations of pyruvate to propionate, see Part B.	
Detailed pathway of g	glutamate fermentation via 2-hydroxyglutarate [37]
Glutamate dehydrogenase	L-glutamate + H_2O + $NAD^+ \leftrightarrow 2$ -oxoglutarate +	1.4.1.2

Enzyme	Reaction/process	EC number
2-Hydroxyglutarate dehydrogenase	(S)-2-hydroxyglutarate + acceptor \leftrightarrow 2- oxoglutarate + reduced acceptor	1.1.99.2
Glutaconate (2-hydroxyglutarate) CoA-transferase	Acetyl-CoA + (E)-glutaconate \leftrightarrow acetate + glutaconyl-1-CoA	2.8.3.12
2-Hydroxyglutaryl-CoA dehydratase	(R)-2-hydroxyglutaryl-CoA \leftrightarrow (E)-glutaconyl-CoA + H ₂ O	EC 4.2.1.167
Glutaconyl-CoA decarboxylase	4-Carboxybut-2-enoyl-CoA \leftrightarrow but-2-enoyl-CoA + CO ₂	4.1.1.70

Table 2.The selected enzymes of acidogenic step of anaerobic digestion. A, B, C, D, and E refer to the processes indicatedin Figure 1.

Enzyme	Reaction/process	EC number
F. Acetogenesis dependent on	syntrophic relations between microorganis	ms
$\frac{\text{dehydrogenase/acetyl-CoA}}{\text{Acetate}^- + 4\text{H}_2\text{O} \rightarrow 2 \text{ HCO}_3^- + 4\text{H}_2 + \text{H}^+, 4\text{A}_2 + \text{H}^+, 4\text{H}_2 + \text{H}^+, 4\text{H}_$	Clostridium ultunense—oxidative carbon mosynthase pathway (oxidative CODH/ACS): $\Delta G^{0'} = + 104.6 \text{ kJ/mol}$, with the H ₂ consuming = -31.0 kJ/mol [38]	
NADP-dependent fo	ormate dehydrogenase	See Table 2,
Formyltetrahydr	ofolate synthetase	Part C
Methenyltetrahydro	folate cyclohydrolase	_
NADP-dependent methylenet	etrahydrofolate dehydrogenase	_
Ferredoxin-dependent methy	lenetetrahydrofolate reductase	_
5,10-Methylenetetra	hydrofolate reductase	_
5-Methyltetrahydrofolate:corrinoid/ir	on-sulfur protein Co-methyltransferase	_
Carbon monoxie	de dehydrogenase	_
CO-methylating a	cetyl-CoA synthase	_
•	idation has yet to be confirmed. Direct intersp t excluded (Westerholm et al., 2016)	ecies electron
Acetate oxidation coupled to reduction of	on by <i>Geobacter sulfurreducens:</i> f fumarate to succinate ($\Delta G^{o'} = -249$ kJ per m ds via reactions of the citric acid cycle [39]	ol acetate),
Acetat	e kinase	See Table 2,
Phosphotr	ansacetylase	Part B
	Citric acid cycle	
Citrate synthase	$\begin{array}{l} Acetyl\text{-}CoA + H_2O + oxaloacetate \leftrightarrow citrate \\ + CoA \end{array}$	EC 2.3.3.1
Aconitase	$Citrate \leftrightarrow isocitrate \text{ (overall reaction)}$	EC 4.2.1.3
Isocitrate dehydrogenase (NADP ⁺ - dependent)	Isocitrate + NADP ⁺ \leftrightarrow 2-oxoglutarate + CO ₂ + NADPH + H ⁺	EC1.1.1.42
2-Oxoglutarate:ferredoxin oxidoreductase	2-Oxoglutarate + CoA + 2 oxidized Fd = succinyl-CoA + CO ₂ + 2 reduced Fd + 2 H ⁺	EC 1.2.7.3
Succinyl-CoA:acetate CoA-transferase	Succinyl-CoA + acetate \leftrightarrow acetyl-CoA + succinate	EC 2.8.3.18

Enzyme	Reaction/process	EC number
Succinate dehydrogenase	succinate + a quinone \leftrightarrow fumarate + a quinol	EC 1.3.5.1
Fumarate hydratase	$(S)\text{-malate} \leftrightarrow fumarate + H_2O$	EC 4.2.1.2
Malate dehydrogenase	(S)-malate + NAD ⁺ \leftrightarrow oxaloacetate + NADH + H ⁺	EC 1.1.1.37
Butyrate ⁻ + $2H_2O \rightarrow 2$ acetate ⁻	oxidation by <i>Syntrophomonas wolfei</i> : $^{-} + 2H^{+} + 2H_2$, $\Delta G^{0'} = + 48.3$ kJ/mol, with the H ₂ containing the H ₂ contained by $\Delta G^{0'} = -17.3$ kJ/mol [4]	onsuming
CoA transferase	Butyrate + acetyl-CoA ↔ butyryl- CoA + acetate	EC 2.8.3.9
Butyry	l-CoA dehydrogenase	See Table 2
Crotonase-3-C	DH-butyryl-CoA dehydratase	Part B
3-Acetyl	-CoA acetyltransferase	
Hydroxybu	tyryl-CoA dehydrogenase	
Pho	sphotransacetylase	_
	Acetate kinase	-
	nd cytochrome, NADH:hydrogenase/formate-deh -1 complex), Rnf (proton-translocating ferredoxin oxidoreductase) [40]	
	e oxidation by Syntrophobacter wolinii:	
Propionate ⁻ + $3H_2O \rightarrow acetate^-$ + H	·	H ₂ consuming
$Propionate^- + 3H_2O \rightarrow acetate^- + H_meth$	e oxidation by <i>Syntrophobacter wolinii</i> : $1CO_3^- + H^+ + 3H_2$, $\Delta G^{0'} = + 76.0$ kJ/mol, with the 1	See Table 2
$\begin{array}{c} Propionate^- + 3H_2O \rightarrow acetate^- + H_2O \rightarrow acetate + H_2O \rightarrow$	e oxidation by <i>Syntrophobacter wolinii</i> : $HCO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol}$, with the hanogen, $\Delta G^{0'} = -22.4 \text{ kJ/mol}$ [4]	-
Propionate ⁻ + $3H_2O \rightarrow acetate^-$ + H meth Pyr Mal	e oxidation by <i>Syntrophobacter wolinii</i> : $HCO_3^- + H^+ + 3H_2$, $\Delta G^{0'} = + 76.0$ kJ/mol, with the baanogen, $\Delta G^{0'} = -22.4$ kJ/mol [4] puvate carboxylase	See Table 2
Propionate $+ 3H_2O \rightarrow acetate + H_meth$ Pyr Mal	e oxidation by Syntrophobacter wolinii: $HCO_3^- + H^+ + 3H_2$, $\Delta G^{0'} = + 76.0$ kJ/mol, with the baanogen, $\Delta G^{0'} = -22.4$ kJ/mol [4] uvate carboxylase ate dehydrogenase	See Table 2
Propionate $+ 3H_2O \rightarrow acetate + H_meth$ Pyr Mal Fu	e oxidation by Syntrophobacter wolinii: $HCO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol}, \text{ with the I}$ hanogen, $\Delta G^{0'} = -22.4 \text{ kJ/mol} [4]$ wate carboxylase ate dehydrogenase marate hydratase	See Table 2
Propionate $+ 3H_2O \rightarrow acetate + H_meth$ Pyr Mal Fu Succinate dehydrogenase	e oxidation by Syntrophobacter wolinii: $HCO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol}, \text{ with the I}$ hanogen, $\Delta G^{0'} = -22.4 \text{ kJ/mol [4]}$ twate carboxylase ate dehydrogenase marate hydratase marate reductase Succinate + a quinone \leftrightarrow fumarate + a	See Table 2 , Part B EC 1.3.5.1 See Table 2 ,
Propionate $+ 3H_2O \rightarrow acetate + H_meth$ Pyr Mal Fu Succinate dehydrogenase Succi	e oxidation by Syntrophobacter wolinii: $HCO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol, with the Hanogen, \Delta G^{0'} = -22.4 \text{ kJ/mol [4]}uvate carboxylaseate dehydrogenasemarate hydratasemarate reductaseSuccinate + a quinone ↔ fumarate + aquinol$	See Table 2 Part B EC 1.3.5.1
Propionate $+ 3H_2O \rightarrow acetate + H meth$ Pyr Mal. Fu Succinate dehydrogenase Succi	e oxidation by Syntrophobacter wolinii: $ICO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol}, with the Haanogen, \Delta G^{0'} = -22.4 \text{ kJ/mol} [4]ruvate carboxylaseate dehydrogenasemarate hydratasemarate reductaseSuccinate + a quinone \leftrightarrow fumarate + aquinolinyl-CoA synthetase$	See Table 2 Part B EC 1.3.5.1 See Table 2
$\begin{array}{c} Propionate^- + 3H_2O \rightarrow acetate^- + H \\ meth \\ Pyr \\ Mal. \\ Fu \\ Succinate dehydrogenase \\ Succi \\ Methyl \\ Methyl \\ \end{array}$	e oxidation by Syntrophobacter wolinii: $HCO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol}, with the I hanogen, \Delta G^{0'} = -22.4 \text{ kJ/mol} [4]truvate carboxylaseate dehydrogenasemarate hydratasemarate reductaseSuccinate + a quinone \leftrightarrow fumarate + aquinolinyl-CoA synthetaseImalonyl CoA mutase$	See Table 2 Part B EC 1.3.5.1 See Table 2
Propionate ⁻ + 3H ₂ O → acetate ⁻ + H meth Pyr Mal Fu Succinate dehydrogenase Succi Methyl Methylm	e oxidation by Syntrophobacter wolinii: $ICO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol, with the Hanogen, \Delta G^{0'} = -22.4 \text{ kJ/mol [4]}uvate carboxylaseate dehydrogenasemarate hydratasemarate reductaseSuccinate + a quinone ↔ fumarate + aquinolinyl-CoA synthetaseImalonyl CoA mutasenalonyl CoA epimerase$	See Table 2 Part B EC 1.3.5.1 See Table 2
Propionate ⁻ + 3H ₂ O → acetate ⁻ + H meth Pyr Mal Fu Succinate dehydrogenase Succi Methyl Methylma Propionate oxidation coupled with encoded by cytochrome c homol	e oxidation by Syntrophobacter wolinii: $ICO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol, with the Hanogen, \Delta G^{0'} = -22.4 \text{ kJ/mol [4]}uvate carboxylaseate dehydrogenasemarate hydratasemarate reductaseSuccinate + a quinone ↔ fumarate + aquinolinyl-CoA synthetaseImalonyl CoA mutasenalonyl CoA decarboxylase$	See Table 2 Part B EC 1.3.5.1 See Table 2 Part B Part B
Propionate ⁻ + 3H ₂ O → acetate ⁻ + H meth Pyr Mal Fu Succinate dehydrogenase Succi Methyl Methylm Methylm Propio Propionate oxidation coupled with encoded by cytochrome c homol dehydrogenases, hydrogen Six syntrophy-specific functional domains found in the genomes of t butyrate- or propionate-oxidizing	e oxidation by Syntrophobacter wolinii: $ICO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol, with the Hanogen, \Delta G^{0'} = -22.4 \text{ kJ/mol [4]}uvate carboxylaseate dehydrogenasemarate hydratasemarate reductaseSuccinate + a quinone ↔ fumarate + aquinolmyl-CoA synthetaseImalonyl CoA mutasenalonyl CoA epimeraselonyl-CoA decarboxylasenate-CoA transferasea reverse electron transfer that involves menaquinogous genes, cytochrome b:quinone oxidoreductasenases including confurcating [FeFe]-hydrogenasesInterPro number$	See Table 2 , Part B EC 1.3.5.1 See Table 2 , Part B Part B
Propionate ⁻ + 3H ₂ O → acetate ⁻ + H meth Pyr Mal Fu Succinate dehydrogenase Succi Methyl Methylma Propionate oxidation coupled with encoded by cytochrome c homol	e oxidation by Syntrophobacter wolinii: $ICO_3^- + H^+ + 3H_2, \Delta G^{0'} = + 76.0 \text{ kJ/mol, with the Hanogen, \Delta G^{0'} = -22.4 \text{ kJ/mol [4]}uvate carboxylaseate dehydrogenasemarate hydratasemarate hydratasemarate reductaseSuccinate + a quinone ↔ fumarate + aquinolinyl-CoA synthetaseImalonyl CoA mutasenalonyl CoA decarboxylasenate-CoA transferasea reverse electron transfer that involves menaquinogous genes, cytochrome b:quinone oxidoreductasenases including confurcating [FeFe]-hydrogenasesInterPro numberhe$	See Table 2 Part B EC 1.3.5.1 See Table 2 Part B Part B

Enzyme	Reaction/process EC number
FDH accessory protein—tightly connected with FDH	IPR006452
CapA—a membrane-bound complex, a protein involved in capsule or biofilm formation that may facilitate syntrophic growth (<i>also present in acetate-oxidizers</i>)	IPR019079
FtsW, RodA, SpoVE—membrane- integrated proteins involved in membrane integration, cell division, sporulation, and shape determination	IPR018365
Ribonuclease P involved in tRNA maturation	IPR020539
Functional domains involved in electron transfer identified by [42]	InterPro number
Cytoplasmic FDH	IPR027467, IPR006655, IPR006478, IPR019575, IPR001949
Extracytoplasmic FDH	IPR006443
Formate transporter	IPR000292, IPR024002
Fe-Fe hydrogenase	IPR004108, IPR009016, IPR003149, IPR013352
NiFe hydrogenase	IPR001501, IPR018194
Rnf complex: 2 reduced Fd + NAD ⁺ + H ⁺ + Na ⁺ \leftrightarrow 2 oxidized Fd + NADH + Na ⁺ (EC 1.18.1.8)	IPR007202, IPR010207, IPR026902, IPR010208, IPR004338, IPR011303, IPR007329
Ech complex: 2 reduced Fd + NADP ⁺ + H ⁺ \leftrightarrow 2 oxidized Fd + NADPH (EC 1.18.1.2)	IPR001750, IPR001516, IPR001694, IPR006137, IPR001268, IPR012179, IPR001135
Etf alpha, Etf beta, Bcd (Butyryl-CoA dehydrogenase): see Table 2 , Part B (EC 1.3.1.109)	IPR014731, IPR012255, IPR006089, IPR009075, IPR006092, IPR006091, IPR013786, IPR009100
Cytochromes:	IPR023155, IPR024673
c cIII	IPR020942, IPR002322 IPR016174, IPR000516
b561	IPR001199
b5	
DUF224 protein complex	IPR003816, IPR004017, IPR023234
	on by Desulfovibrio vulgaris:
	$\Delta G^{0'} = -8.8 \text{ kJ/mol with the H}_2 \text{ consuming methanogen,}$ = -74.2 kJ/mol [43]
Lactate deh	
Pyruvate:ferredox	in oxidoreductase Part B
Phosphate ace	tyltransferase
Acetate	kinase
Alcohol deh	ydrogenase
Qmo complex, cytochrome formate dehydrogenases	e electron transfer that involves the membrane-bound es, hydrogenases (Coo, Hyn, Hyd, Hys), , menaquinone, membrane-bound Qrc omplex [43, 44]

Enzyme		Reaction/process	EC number
Ethanol + H ₂ C	$0 \rightarrow acetate^- + H^+ + 2H_2, \Delta 0$	Con by <i>Pelobacter carbinolicus</i> $G^{0'} = + 9.6 \text{ kJ/mol with the H}_2 \text{ consuming m}$ G' = -56 kJ/mol [4]	ethanogen,
	NAD ⁺ -dependent eth	nanol dehydrogenase	See Table 2,
	Acetaldehyde dehydro	ogenase (acetylating)	— Part B
Nonacetylating a dehydrogenase	cetaldehyde	An aldehyde + NAD ⁺ + $H_2O \leftrightarrow a$ carboxylate + NADH + H^+	EC 1.2.1.3
	Phosphotra		See Table 2 , — Part B
	nslocating ferredoxin:NAD	rse electron transfer that involves membran ⁺ oxidoreductase, formate dehydrogenases, ng hydrogenases [1, 45]	
G. Acet	ogenesis independent on	syntrophic relations between microorgan	isms
Ethanol oxidat	ion by Acetobacterium wo	odii: 2 ethanol + 2 $CO_2 \rightarrow 3$ acetate—75.4	kJ/mol [46]
Bifunctional acet: dehydrogenase	aldehyde-CoA/alcohol	$\begin{array}{l} \mbox{Ethanol} + \mbox{NAD}^{+} \rightarrow \mbox{acetaldehyde} + \mbox{NADH} \\ & + \mbox{H}^{+} \\ \mbox{acetaldehyde} + \mbox{NAD}^{+} + \mbox{CoA} \rightarrow \mbox{acetyl-} \\ \mbox{CoA} + 2 \mbox{NADH} + \mbox{H}^{+} \\ \mbox{Ethanol} \mbox{ is oxidized to acetyl-CoA in a two-} \\ \mbox{step reaction by a bifunctional acetylating} \\ \mbox{ethanol/aldehyde dehydrogenase} \end{array}$	[EC:1.2.1.10 1.1.1.1]
Ace	tyl-CoA is transformed to a	acetate with the release of ATP	See Table 2 , Part B
Reduction of fer	redoxin by NADH by rever Rnf co	rse electron flow in a reaction catalyzed by mplex	See Part F
Carbon o	lioxide is reduced to acetate	e via the Wood-Ljungdahl pathway	See Table 2 , Part C
Lactate o	xidation by Acetobacteriu	<i>m woodii</i> : 2 lactate $ ightarrow$ 3 acetate—61 kJ/mo	ol [47]
Lactate dehydrog	enase	Lactate + 2 NAD ⁺ + 2 reduced Fd ↔ pyruvate + 2 NADH + 2 oxidized Fd The enzyme uses flavin-based electron confurcation to drive endergonic lactate	EC 1.3.1.110
		oxidation with NAD ⁺ as oxidant at the expense of simultaneous exergonic electron flow from reduced ferredoxin to NAD ⁺	
Pyruvate is tran	sformed to acetyl-CoA and	further to acetate with the release of ATP	See Table 2 , Part B
Reduction of fer	redoxin by NADH by rever Rnf co	rse electron flow in a reaction catalyzed by mplex	See Part F

 Table 3.

 The selected enzymes of acetogenic step of anaerobic digestion. F and G refer to the processes indicated in Figure 1.

Enzyme	Reaction/process	EC number
	I—coenzyme M, H-S-CoB—coenzyme B, H₄MP ₀—5'deazaflavin, H₄SPT—tetrahydrosarcinapter	
Hyd	lrogenotrophic pathway	
Formylmethanofuran dehydrogenase	$CO_2 + MFR + 2 \text{ reduced } Fd + 2H^+ \leftrightarrow \text{ formyl-}$ $MFR + H_2O + 2 \text{ oxidized } Fd$	EC 1.2.7.12
Formylmethanofuran-H₄MPT formyltransferase	Formyl-MFR + $H_4MPT \leftrightarrow MFR$ + formyl- H_4MPT	EC 2.3.1.101
Methenyl-H ₄ MPT cyclohydrolase	Formyl-H ₄ MPT + H ⁺ \leftrightarrow methenyl- H ₄ MPT + H ₂ O	EC 3.5.4.27
F_{420} -dependent methylene-H ₄ MPT dehydrogenase	$\begin{array}{c} \mbox{Methenyl-H_4MPT} + \mbox{reduced } F_{420} \leftrightarrow \\ \mbox{methylene-H_4MPT} + \mbox{oxidized } F_{420} \end{array}$	EC 1.5.98.1
H₂-forming methylene-H₄MPT dehydrogenase	$\begin{array}{l} Methenyl\text{-}H_4MPT + H_2 \leftrightarrow methylene-\\ H_4MPT + H^{*} \end{array}$	EC 1.12.98.2
F ₄₂₀ -dependent methylene-H ₄ MPT reductase	$\begin{array}{l} Methylene{-}H_4MPT \ \ \ \ reduced \ F_{420} \leftrightarrow CH_3 \ \ \\ H_4MPT \ \ \ \ \ oxidized \ F_{420} \end{array}$	EC 1.5.98.2
Methyl-H ₄ MPT:coenzyme M methyl- transferase	Coenzyme M + methyl-H ₄ MPT + 2 Na+/in \leftrightarrow 2-methyl-coenzyme M + 2 Na+/out + H ₄ MPT	EC 2.1.1.86
Methyl-CoM reductase	$CH_3\text{-}S\text{-}CoM + H\text{-}S\text{-}CoB \leftrightarrow CoM\text{-}S\text{-}S\text{-}CoB + CH_4$	EC 2.8.4.1
Heterodisulfide reductase	$\label{eq:com-S-S-CoB} \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	EC 1.8.98.1
A	Acetotrophic pathway	
Acetate kinase-phosphotransacetylase system in <i>Methanosarcina</i> ; acetate thiokinase in <i>Methanosaeta</i>	$Acetate + CoA \leftrightarrow acetyl-CoA + H_2O$	EC 2.7.2.1 EC 2.3.1.8 EC 6.2.1.1
CO-methylating acetyl-CoA synthase	Acetyl-CoA + a [Co(I) corrinoid Fe-S protein] \leftrightarrow CO + CoA + [methyl-Co(III) corrinoid Fe-S protein]	EC 2.3.1.169
5-Methyltetrahydrosarcinapterin: corrinoid/iron-sulfur protein Co- methyltransferase	[Methyl-Co(III) corrinoid Fe-S protein] + tetrahydrosarcinapterin ↔ a [Co (I) corrinoid Fe-S protein] + 5- methyltetrahydrosarcinapterin	EC 2.1.1.245
Anaerobic carbon monoxide dehydrogenase	$CO + H_2O + 2 \text{ oxidized } Fd \leftrightarrow CO_2 + 2 \text{ reduced}$ $Fd + 2 H^+$	EC 1.2.7.4
Methyl H₄SPT: coenzyme M methyltransferase	$CH_3 H_4SPT + H-S-CoM \leftrightarrow CH_3-S-CoM + H_4SPT$	EC 2.1.1
Methyl-CoM reductase	$CH_3\text{-}S\text{-}CoM + H\text{-}S\text{-}CoB \leftrightarrow CoM\text{-}S\text{-}S\text{-}CoB + CH_4$	EC 2.8.4.1
Heterodisulfide reductase	$\begin{array}{l} CoM\text{-}S\text{-}S\text{-}CoB + dihydromethanophenazine} \\ \leftrightarrow CoB + CoM + methanophenazine \end{array}$	EC 1.8.98.1
M	ethylotrophic pathway	
Methanol:corrinoid protein Co- methyltransferase	Methanol + Co(I) corrinoid protein \leftrightarrow Methyl-Co(III) corrinoid protein + H ₂ O	EC 2.1.1.90
[Methyl-Co(III) corrinoid protein]: coenzyme M methyltransferase	Coenzyme M + Methyl-Co(III) corrinoid protein \leftrightarrow 2-(methylthio)ethanesulfonate + Co(I) corrinoid protein	EC 2.1.1.246

Enzyme	Reaction/process	EC number
Methylamine:corrinoid protein Co- methyltransferase	Methylamine + [Co(I) methylamine-specific corrinoid protein] ↔ a [methyl-Co(III) methylamine-specific corrinoid protein] + NH ₃	EC 2.1.1.248
Dimethylamine:corrinoid protein Co- methyltransferase	Dimethylamine + [Co(I) dimethylamine- specific corrinoid protein] ↔ a [methyl- Co(III) dimethylamine-specific corrinoid protein] + methylamine	EC 2.1.1.249
Trimethylamine:corrinoid protein Co- methyltransferase	Trimethylamine + a [Co(I) trimethylamine- specific corrinoid protein] ↔ a [methyl- Co(III) trimethylamine-specific corrinoid protein] + dimethylamine	EC 2.1.1.249
[Methyl-Co(III) methylamine-specific corrinoid protein]:coenzyme M methyltransferase	[Methyl-Co(III) methylamine-specific corrinoid protein] + CoM ↔ methyl-CoM + a [Co(I) methylamine-specific corrinoid protein]	EC 2.1.1.247
Methyl-CoM reductase	$CH_3\text{-}S\text{-}CoM + H\text{-}S\text{-}CoB \leftrightarrow CoM\text{-}S\text{-}S\text{-}CoB + CH_4$	EC 2.8.4.1
Heterodisulfide reductase	$\begin{array}{l} CoM\text{-}S\text{-}S\text{-}CoB \ + \ dihydromethan ophenazine} \\ \leftrightarrow CoB \ + \ CoM \ + \ methan ophenazine \end{array}$	EC 1.8.98.1

Table 4.

The selected enzymes of methanogenic step of anaerobic digestion [48, 49].

3. Conclusion

Biomass conversion to methane and carbon dioxide is the effect of complex interactions between microorganisms. These processes occur due to the microbial enzymatic machinery involved in specific metabolic pathways. Meta-omic analyses of microbial communities involved in AD reveal (i) dependence of microbial communities on the type of feedstock and operational conditions and (ii) describe interactions within microbial communities and ecophysiological functions of the specific taxa. Searching for the gene presence, gene expression, and protein expression, as well as linking structure and function of microbial communities, allows to develop a fundamental understanding of AD. This chapter is believed to contribute to the studies on the enzymatic road map of anaerobic digestion. However, it is only the tip of the iceberg of processes occurring in the microbial cells/microbial communities.

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Conflict of interest

The authors declare that there are no conflicts of interest.

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