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# Investigation of aldehyde oxidase and xanthine oxidoreductase in rainbow trout (Oncorhynchus mykiss)

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Thesis submitted in partial fulfilment of the requirements for the Degree of Doctor of Philosophy July 2014

#### Abstract

Molybdo-flavoenzymes (MFEs), aldehyde oxidase (AOX) and xanthine oxidoreductase (XOR) are involved in the oxidation of N-heterocyclic compounds and aldehydes, many of which are environmental pollutants, drugs and vitamins. This biotransformation generally generates more polar compounds that are more easily excreted, thus MFEs have been classed as detoxication enzymes.

To date there has been scant study of the properties, substrate and inhibitor specificities of MFEs in non-mammalian vertebrate organisms. This investigation focuses on MFEs in rainbow trout (Oncorhynchus mykiss) as it belongs to a class of fish that host a single AOX (AOX $\beta$ ) and one XOR.

In this study the substrate specificity of rainbow trout liver AOX and XOR was investigated using HPLC and spectrophotometric assays. AOX in hepatic cytosol was found to be able to catalyse the oxidation of azanaphthalenes belonging to a group of compounds that are environmental pollutants such as phenanthridine, phthalazine and cinchonine. In addition, xenobiotic aromatic aldehydes (vanillin and dimethylaminocinnamaldehyde) and drugs such as allopurinol and pyrazinamide were substrates. Several endogenous vitamins including pyridoxal (vitamin B6), all-trans retinal (vitamin A) and N<sup>1</sup>-methylnicotinamide were also biotransformed by the rainbow trout AOX. In contrast to liver no AOX activity was detectable in kidney and gill tissue. XOR activity in rainbow trout liver was measurable with the endogenous purine xanthine, purine drug metabolites (1methylxanthine and 6-thioxanthine) and N-heterocyclic drugs (allopurinol and pyrazinamide). Unlike mammalian XOR that can utilise both  $NAD^+$  and  $O_2$  as

electron acceptors, trout XOR was exclusively  $NAD^+$ -dependent with no activity being detected with  $O_2$ .

Eadie-Hofstee plots were using to determine the Km and Vmax of rainbow trout AOX and XOR with different substrates and it was found the Vmax of the rainbow trout enzymes were generally lower and Km generally higher than mammalian AOX and XOR.

Inhibitors of mammalian AOX were tested to determine if they could interact with the piscine AOX. Environmental pollutants ( $17\alpha$ -ethinyl estradiol and phenanthridine), an endogenous steroid (estradiol) and drugs (chlorpromazine and menadione) were found to be effective inhibitors and were classed as competitive, non-competitive and uncompetitive respectively using Lineweaver-Burk plots. The drug metabolite, oxipurinol, was a non-competitive inhibitor of rainbow trout XOR. In order to further characterise trout AOX protein purification was carried out. In contrast to mammalian AOX, the piscine enzyme was not thermotolerant at 55°C nor was it inhibited by benzamidine, thus heat treatment and affinity chromatography could not be used as a purification steps. Trout AOX was purified 210-fold using ammonium sulphate fractionation, together with ion exchange and gel filtration chromatography. The native molecular mass of the piscine AOX was

295 kDa, which is similar to mammalian AOXs.

In conclusion this study yields new insight into groups of anthropogenic environmental pollutants, drugs and vitamins that are substrates and inhibitors of an ancestral vertebrate AOX. The toxicological relevance of these findings is discussed.

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#### **ACKNOWLEDGEMENTS**

No project such as can be carried out by only one individual. I would like to express my thanks to God who is the source and origin of all knowledge.

I fell highly privileged to express my heartiest gratitude to the worthy and kind supervisor, Dr Dougie Clarke for his dynamic supervision, constructive criticism and affectionate behaviour throughout this study. His wide knowledge and logical way of thinking has been of great value for me. And his intensive and creative comments have helped me step by step throughout this project.

My sincere deepest gratitude to all my family for their support and encouragement through this work, I would like to thank my colleagues at the school of Applied Sciences who shared great research ideas during my time in Huddersfield University.

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#### List of Abbreviations:

Abbreviation:	Definition:		
2-PY:	N <sup>1</sup> -methyl-2-pyridone-5-carboxamide		
4-PY:	N <sup>1</sup> -methyl-4-pyridone-3-carboxamide		
6-MP:	6-mercaptopurine		
ADP:	Adenosine diphosphate		
AhR:	Aryl hydrocarbon receptors		
ALDH:	Aldehyde dehydrogenase		
AMP:	Adenosine monophosphate		
AOX:	Aldehyde oxidase		
AOH1:	Aldehyde oxidase homologue 1		
ATP:	Adenosine triphosphate		
ATR:	All-trans retinal (vitamin A)		
BSA:	Bovine serum albumin		
CV:	Coefficient of variation		
CYP:	Cytochrome P450		
DAD:	Diode array detector		
DMAC:	Dimethylaminocinnamaldehyde		
DCPIP:	2, 6-dichlorophenol-indophenol		
EBT:	Erochrome black T		
FAD:	Flavin adenine dinucleotide		
GMP:	Guanosine monophosphate		
IMP:	Inosine monophosphate		
HPLC:	High performance liquid chromatography		
H <sub>2</sub> O <sub>2</sub> :	Hydrogen peroxide		
kDa:	kilo Dalton		
K <sub>m</sub> :	Michaelis constant		
LOD:	Limit of detection		

#### Abbreviation: Definition:

LOQ:	Lower limit of Quantification		
MFEs:	Molybdo-flavoenzymes		
MoCo:	Molybdenum cofactor		
MTT:	Dimethylthiazol-diphenyl tetrazolium bromide		
MB:	Methylene blue		
mM:	Millimolar		
NMN:	N <sup>1</sup> -methylnicotinamide		
$NAD^+$ :	Nicotinamide adenine dinucleotide		
INT:	Iodonitro-tetrazolium		
NPAHs:	Nitrogen containing polycyclic aromatic hydrocarbons		
NBT:	Nitro blue tetrazolium		
NADH:	Nicotinamide adenine dinucleotide (reduced form)		
PMS:	Phenazine methosulphate		
PAOX:	Pseudogene aldehyde oxidase		
PBS:	Phosphate buffer saline		
PCR:	Polymerase chain reaction		
RSD:	Relative standard deviation		
TTC:	Tetrazolium chloride		
TAN:	Total ammonia nitrogen		
XDH:	Xanthine dehydrogenase		
XO:	Xanthine oxidase.		
XOR:	Xanthine oxidoreductase		

#### 1. Introduction.

The term used to describe the chemical substances which are foreign to animal life is xenobiotics. These chemical substances include drugs, cosmetics, food additives, environmental pollutants, pesticides, flavouring fragrances and industrial chemicals(Idle and Gonzalez. 2007). Many of these chemical substances access our bodies by the diet, atmosphere, drinking water and drug administration. These substances go through a broad varieties of processes of detoxication of which generally generate compounds that are much less toxic, much more polar, and also quickly excretable (Timbrell. 2009). Of these processes oxidation has an important purpose in the metabolism of such xenobiotics. The microsomal cytochrome P-450 mono-oxygenase plays an important role in this biotransformation process; however the cytosolic enzymes are also involved in this biotransformation process. This thesis is concerned with two cytosolic molybdo-flavoenzymes (MFEs), namely aldehyde oxidase (AOX; EC 1. 2. 3. 1) and xanthine oxidoreductase (XOR; EC 1. 2. 3. 2). In mammals the second of these enzymes, XOR, exists as two forms xanthine dehydrogenase (XDH; EC 1. 17. 1. 4) and xanthine oxidase (XO; EC 1. 17. 3. 2) that are derived from a single gene and both have identical primary amino acid sequence. The native dehydrogenase form that uses NAD<sup>+</sup> as an electron acceptor can be converted to the XO form that uses  $O_2$  as an electron acceptor by cysteine oxidation or limited proteolysis (Nishino and Nishino. 1997, Nishino et al., 2008). The change from the XDH to the XO form is associated with a conformational change in protein structure that prevents the bulky NAD<sup>+</sup> from entering the electron acceptor binding site while the smaller oxygen can enter this site (Nishino and Nishino. 1997, Nishino et al., 2008). The conversion of XDH to XO has only been demonstrated in mammalian species that have specific cysteines (e.g in rats XDH has cysteine 535 and cysteine 992) and lysines (e.g in rats XDH has lysine 754 and lysine 771) that are the site of proteolytic cleavage. The conversion of XDH to XO in vivo that result in the enzyme using molecular oxygen as electron acceptor results in oxygen free radicals being generated which have been shown to be responsible for protein damage, DNA mutation and cell membrane disintegration linked with many diseases such as cancer, liver damage and cardiovascular disease (Nishinaka et al., 2001). These enzymes are present in almost all organisms from bacteria to human (Beedham. 2001, Garattini et al., 2008, Garattini et al., 2009, Pryde et al., 2010, Garattini and Terao. 2011, Garattini and Terao. 2012, Hartmann et al., 2012, Kurzawski et al., 2012, Garattini and Terao. 2013).

#### 1.1. Molybdo-flavoenzyme structure

Both MFEs enzymes, AOX and XOR, are composed of molybdenum cofactor (MoCo), one flavin adenine dinucleotide (FAD) and a pair of non-identical metal sulfur centers [2Fe-2S] (Palmer et al., 1969, Nishino. 1994, Borges et al., 2002, Nishino et al., 2002, Garattini et al., 2008, Pryde et al., 2010, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013). Molybdo-flavoenzymes are homodimers as active forms and are composed from a pair of identical subunits of approximately 150 kDa (Figure 1). Their structure is split into three distinct domains as follows: an 85 kDa C-terminal domain that include Moco along with the substrate binding site, and an N-terminus that includes a 20 kDa domains with a couple of [2Fe-2S] clusters, and a 45 kDa flavin adenine dinucleotide (FAD) joining domain (Kisker. 1997, Hille. 2005, Mendel and Bittner. 2006a, Wollers. 2008, Garattini and Terao. 2012, Garattini and Terao. 2013).



Figure 1: The domain structure of the AOX and XDH enzyme family.

The crystallization of mammalian molybdo-flavoenzymes has provided much information about the domain structure of molybdo-flavoenzymes (Romao et al., 1995, Enroth et al., 2000, Truglio et al., 2002, Pryde et al., 2010, Mahro et al., 2011, Coelho et al., 2012, Garattini and Terao. 2012, Kikuchi et al., 2012, Garattini and Terao. 2013, Mendel. 2013). Both bovine XO and XDH have been determined by X-ray crystallography to be separated into three sub-domains. The first sub-domain (residues 1-165) at the N-terminus has a pair of iron-sulphur cofactors. It is linked to the FAD binding domain (residues 226-531) by a long portion connected with amino acids (residues 166-225) to another linker portion (residues 532-861) this then links to a specific sub-domain containing on the molybdenum cofactor (residues 590-1,332), which is located close to the interfaces of the FAD binding domain and the iron sulphur(Enroth et al., 2000, Kikuchi et al., 2012).

The N-terminus domain binds the two  $Fe_2S_2$  clusters, the center domain include the FAD binding site and the C-terminus domain is the location of this Mo-pterin cofactor (Mendel and Bittner. 2006a, Mendel. 2013).

# **1.2.** Molybdenum cofactor and the catalytic mechanism of molybdo-flavoenzymes

Molybdenum itself is biologically inactive unless bound to a tricyclic pterin compound where it then forms molybdenum cofactor (Moco) (Figure 2). Moco is situated at the active site of AOX and XOR(Mendel and Bittner. 2006b). The pterin structure coordinates to the molybdenum metal by an enedithiolate side chain (Hille. 2005). The molybdo-flavo enzymes, AOX and XOR, contain an active site that is believed to be in a five-coordinate complex with two enedithiolate ligands of the molybdopterin cofactor, one oxo group, one sulfide group, and one hydroxyl or water molecule. This coordination results in a square pyramidal geometric structure and is the site for substrate binding and enzyme inhibition (Brondino et al., 2005).



Figure 2: Molybdenum Cofactor structure in molybdenum enzymes.

The active site structure is attached with the pyranopterin compound through two enedithiolate side chains (green colour) to produce the molybdenum cofactor. Based on Mendel, 2013.

The molybdenum cofactor's purpose is to transfer electrons out of the molybdenum center following the oxidation reaction has taken place. This results in the reduction of molybdenum from Mo (VI) to Mo (IV) (Hille. 2005, Okamoto et al., 2013). The movement of electron flow in MFEs is shown in (Figure 3), in which the Fe/S I and Fe/S II are a pair of iron-sulfur center.

## Mo-pterin — Fe/SI — Fe/SII — FAD Electron movement —

**Figure 3:** The movement of electron flow out from the molybdenum center of molybdoflavoenzymes. According to Okamoto et al., 2013.

The molybdo-flavoenzymes AOX and XOR exhibit two types of reactions at the same time involving the molybdenum cofactor, oxidative and reductive half reactions. The oxidative half reaction at the molybdenum site, and the reduction half reaction at the FAD site are shown in (Figure 4) (Calzei et al., 1995, Hille and Nishino. 1995, Nishino and Okamoto. 2000, Okamoto et al., 2004, Hille. 2005, Brondino et al., 2006, Kitamura et al., 2006, Pryde et al., 2010, Okamoto et al., 2013). Electron donors such as vanillin act at molybdenum site, reducing it from Mo (VI) to the Mo (IV) state. The reducing equivalents transfer directly to the FAD site or via the iron-sulfur center as shown in (Figure 5)(Calzei et al., 1995, Hille and Nishino. 1995, Beedham. 1998, Kitamura et al., 2006, Beedham. 2010).

The reduction half reaction takes place at the FAD site, and the enzyme is reoxidized by interaction with oxygen in case of AOX and XO or with  $NAD^+$  in case of XDH, and  $FADH_2$  is reduced into FAD. The reduction of oxygen leads to formation of superoxide anion and hydrogen peroxide, and these two products

damage the cell, but the cell itself has an enzyme to protect from this attack; for superoxide anion the enzyme superoxide dismutase and for hydrogen peroxide the enzyme hydrogen peroxidase (Kelley et al., 2010). XDH displays partial oxidase activity under conditions in which NAD<sup>+</sup> levels are diminished, such as the ischemic/hypoxic microenvironment encountered in vascular inflammation (Harris and Massey. 1997). This same inflammatory milieu leads to enhanced XO levels and thus increased XO-derived ROS formation, resulting in activation of redoxdependent cell signaling reactions and alterations in vascular function, and reduction of NAD<sup>+</sup> leads to formation of NADH and H<sup>+</sup>. In the presence of an alternative electron donor, MFEs can mediate the reduction of various compounds such as in my example, it reduced the fenthion sulfoxide to its reduced form fenthion as shown in (Figure 4) (Kitamura et al., 2003, Okamoto et al., 2004, Hille. 2005, Kitamura et al., 2006, Beedham. 2010, Pryde et al., 2010). These enzymes can be alternately reduced by the substrate after which re-oxidised by their individual electron acceptors at the FAD site. In vitro, potassium ferricyanide and 2, 6-dichloroindophenol (DCIP) have proven to be effective unnatural electron acceptors for these enzymes that it can be reduced at FAD site (Figure 4) (Krenitsky et al., 1972, Beedham, 1985, Slef and Stadtman, 2000, Kitamura et al., 2006).



Figure 4: Oxidative and reduction half reaction and electron flow out from the molybdenum center of molybdo-flavoenzymes.

According to (Hille and Nishino. 1995, Nishino and Okamoto. 2000, Okamoto et al., 2004, Beedham. 2010, Okamoto et al., 2013).

The mechanism of AOX and XOR inhibitors was proposed by Fridovich and Handler. (1962), Rajagopalan and Handler (1962, 1964 and 1968) and Beedham C (1987) that suggest 4 electrons carriers in molybdenum center of rabbit hepatic AOX and XOR are responsible for the mechanism and reaction with artificial electron acceptors. These studies demonstrated that ferricyanide, methylene blue, phenazine methosulphate react at the iron sulphur cluster and 2, 6 dichlorophenolindophenol (DCPIP) at near to the molybdenum cofactor and that dyes nitroblue tetrazolium (NBT), trinitrobenzensulfonic acid dimethylthiazoland diphenyltetrazolium bromide (MTT) react at the FAD site. Cytochrome c and with oxygen in addition to  $NAD^+$  also react at the FAD center as in seen (Figure 5) (Fridovich and Handler. 1962, Rajagopalan et al., 1962, Rajagopalan and Handler. 1964a, Rajagopalan et al., 1968, Beedham. 1987).



**Figure 5:** Postulated linear electron transfer sequence and site of electron egress to acceptors. Red colour indicates artificial electron acceptor and blue colour indicates natural acceptors.

In molybdo-flavoenzymes reactions the oxygen atom used in the hydroxylation reaction of the carbon center of substrates containing N-heterocycle or aldehyde functional groups is derived from water molecule rather than that from oxygen ( $O_2$ ) itself as shown in (Figure 6) (Hille. 2005).



Figure 6: Oxidation reaction by molybdo-flavoenzymes AOX and XOR.

An example of an aldehyde substrate (vanillin) and its corresponding metabolite a carboxylic acid (vanillic acid) is shown in Figure 7, while another example for an N-heterocycle (1-methylxanthine) and its corresponding metabolite a uric acid (1-methyluric acid) is shown in Figure 8.



Figure 7: Oxygen atom transfer reaction molybdo-flavoenzyme involving an aldehyde.

The oxygen incorporated into the vanillic acid is from a water molecule. According to (Obach. 2004, Arnaud. 2011).

Oxidation reaction catalyzed by molybdo-flavoenzymes AOX and XOR generating an oxidized substrate and also two reducing equivalents (Hille and Sprecher. 1987, Xia et al., 1999, Beedham. 2001, Hille. 2005, Brondino et al., 2006, Okamoto et al., 2013).



Figure 8: Oxygen atom transfer reaction molybdo-flavoenzyme involving an N-heterocycle

# **1.3.** Substrate specificity of aldehyde oxidase and xanthine oxidoreductase

Many different aliphatic compounds such as aromatic aldehydes and also Nheterocyclic compounds are substrates for AOX and XOR that are biotransformed to their carboxylic acid or hydroxylated N-heterocyclic metabolites (Beedham. 2001, Garattini et al., 2003, Garattini et al., 2008, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013). Although the class of substrates of MFEs are similar in their structure, AOX and XOR differ in inhibitor response as well as substrate specificity (Krenitsky et al., 1986, Beedham. 2001, Kitamura et al., 2006, Mendel and Bittner. 2006a, Kundu et al., 2007, Beedham. 2010, Pryde et al., 2010, Garattini and Terao. 2012, Garattini and Terao. 2013). A list of AOX and XOR substrates that illustrate the substrate specificity of the two enzymes are given in Table 1, many of which were utilised in the study detailed in this thesis.

The oxygen incorporated into the 1-methyluric acid is from a water molecule. According to (Obach. 2004, Arnaud. 2011).

Substrates	AOX	XOR	References
Allopurinol (hyperuricemia drug)	++	+++++	(Krenitsky et al., 1972, Krenitsky et al., 1986, Reiter et al., 1990, Yamamoto et al., 1991, Beedham. 2001, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013)
Azaarenes (phenanthridine, phthalazine <sup>**</sup> , acridine, quinoline, isoquinoline, 1- hydroxyisoquinoline, 3- hydroxyisoquinoline, quinoxaline, quinazoline and cinnoline) Environmental pollutants	+++++	- ** ++	(Sorouraddin et al., 2009, Garattini and Terao. 2011, Garattini and Terao. 2012, Rashidi et al., 2013), (Beedham et al., 1990, Obach. 2004, Panoutsopoulos and Beedham. 2004, Kitamura et al., 2006, Barr and Jones. 2011, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013)
Cinchona alkaloid (cinchonine, cinchonidine, quinine and quinidine) Drugs and environmental pollutants	+++++	-	(Palmer et al., 1969, Beedham et al., 1992, Itoh et al., 2006, Al-Tayib. 2009, Liu et al., 2009)
Hypoxanthine ( endogenous purine)	-	+++++	(Krenitsky et al., 1972, Kitamura et al., 2006, Chen et al., 2012, Pasalic et al., 2012, Okamoto et al., 2013)
6-Mercaptopurine (anti-cancer drug)	++	+++++	(Kitamura et al., 2006, Rashidi et al., 2007, Garattini and Terao. 2011, Sorouraddin et al., 2011, Garattini and Terao. 2012)
Methotrexate (anti-cancer drug)	+++++	-	(Johns. 1967, Beedham. 1985, Chladek et al., 1997, Jordan et al., 1999, Kitamura et al., 1999, Beedham. 2001, Kitamura et al., 2006, Liu et al., 2009)
1-Methylxanthine (caffeine metabolite)	-	+++++	(Krenitsky et al., 1972, Reinke et al., 1987, Arnaud. 2011)
N <sup>1</sup> -Methylnicotinamide (endogenous vitamin metabolite)	+++++	+	(Murashige et al., 1966, Krenitsky et al., 1972, Ohkubo and Fujimura. 1978, Kitamura et al., 2006, Tayama et al., 2007, Kitamura et al., 2008, Peretz et al., 2012)
Pyrazinamide (tuberculosis drug)	++	+++++	(Yamamoto et al., 1987, Yamamoto et al., 1991, Shibutani et al., 1999, Kitamura et al., 2006)
Pyridoxal (vitamin B6)	+++++	-	(Stanulovic and Chaykin. 1971, Krenitsky et al., 1972, Kitamura et al., 2006, Garattini and Terao. 2011, Garattini and Terao. 2012, Peretz et al., 2012, Garattini and Terao. 2013)
All-trans retinal (vitamin A aldehyde)	+++++	-	<ul> <li>(Kitamura et al., 2006, Garattini et al., 2008, Terao et al., 2009, Garattini and Terao.</li> <li>2011, Garattini and Terao. 2012, Chen and Reese. 2013, Garattini and Terao. 2013)</li> </ul>
Vanillin (present in food stuffs)	+++++	-	(Beedham. 2001, Obach. 2004, Panoutsopoulos et al., 2005, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013)
Xanthine (endogenous purine)	-	+++++	(Krenitsky et al., 1972, Yamamoto et al., 2007, Chen et al., 2012, Pasalic et al., 2012, Okamoto et al., 2013)

**Table 1:** Examples of mammalian molybdo-flavoenzyme substrates

(+++++ = high activity ++ = intermediate activity + = low activity - = no activity). \*\* Of the azaarenes listed only phthalazine is a substrate for XOR.

With regards endogenous roles of MFEs the best known role of XOR is in endogenous purine degradation in which it is involved in the biotransformation of hypoxanthine into xanthine and into the more polar metabolite uric acid (Figure 9) (Krenitsky et al., 1972, Nishino and Okamoto. 2000, Kitamura et al., 2006, Garattini et al., 2008, Okamoto et al., 2013).



Figure 9: Role of xanthine oxidoreductase in endogenous purine catabolism in mammals.

ADP: adenosine diphosphate, AMP: adenosine monophosphate, ATP: adenosine triphosphate, GMP: guanosine monophosphate, IMP: inosine monophosphate, XDH: xanthine dehydrogenase, XO: xanthine oxidase(Pacher et al., 2006, Chen et al., 2012, Pasalic et al., 2012, Okamoto et al., 2013).

Due to structural similarity to the endogenous substrates, many exogenous methylated xanthines are also substrates for XOR. XOR is involved in the oxidative metabolism of the exogenous food constituent and pharmacological agent caffeine (1, 3, 7 trimethylxanthine). This is shown in Figure 10, where cytochrome P450 1A2 is involved in the initial demethylation steps and XOR is involved in the final oxidation to generate the polar 1-methyluric acid that is excreted(Krenitsky et al., 1972, Hamelin et al., 1994, Streetman et al., 2000, Arnaud. 2011).



Figure 10: The role of xanthine oxidoreductase in caffeine metabolism.

CYP: cytochrome P-450. XOR: xanthine oxidoreductase (Relling et al., 1992, Hamelin et al., 1994, Baselt and Cravey. 1996, Arnaud. 2011).

With regards AOX there are three well documented endogenous substrates. Alltrans retinaldehyde is an endogenous substrate for AOX (Figure 11), that plays an important role in the visual process. It has been suggested that AOX may play an important role in this visual process by biotransformation of all-trans retinaldehyde to its corresponding carboxylic acid metabolite which is retinoic acid, which known as the active form of vitamin A(Beedham. 2001, Kitamura et al., 2006, Garattini et al., 2008, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013). The biotransformations of all-trans retinaldehyde by AOX to its corresponding carboxylic acid metabolite all-trans retinoic acid (RA) was first observed in rabbit liver cytosol, who discovered that the biotransformation activity did not need addition of NAD<sup>+</sup> for its biotransformation process (Tomita et al., 1993, Tsujita et al., 1994, Garattini et al., 2008, Chen and Reese. 2013, Garattini and Terao. 2013).



**Figure 11:** Biotransformation of all-trans retinal to retinoic acid by aldehyde oxidase. Based on. Chen and Reese, 2013.

Another endogenous aldehyde that is an AOX substrate is pyridoxal (vitamin B6). Pyridoxal its biotransformed to its corresponding carboxylic acid metabolite (4pyridoxic acid) by AOX (Figure 12) (Beedham. 2001, Garattini et al., 2003, Kitamura et al., 2006, Garattini et al., 2008, Garattini and Terao. 2011, Garattini and Terao. 2012, Peretz et al., 2012, Garattini and Terao. 2013). Interestingly in mice pyridoxal can be biotransformed by purified mouse aldehyde oxidase AOX1 as well as AOH1 but it is not a substrate for AOH2 (Garattini et al., 2008).



**Figure 12:** Oxidation of pyridoxal to 4-pyridoxic acid by aldehyde oxidase. According to. Peretz et al., 2012.

As well as the oxidation of non-charged N-heterocyclic AOX is also involved in the oxidation of charged N-heterocyclic which have a  $CH=N^+$  group. An example for such a compounds as corresponding metabolite of vitamin B3 (N<sup>1</sup>- methylnicotinamide) that metabolised by AOX and XO (Figure 13) (Murashige et al., 1966, Ohkubo and Fujimura. 1978, Beedham. 1987, Shibata et al., 1988, Beedham. 2001, Sugihara et al., 2006, Kundu et al., 2007, Pryde et al., 2010).


Figure 13: Biotransformation of vitamin B3 in mammals.

The oxidation of N<sup>1</sup>-methylnicotinamide by AO to N<sup>1</sup>-methyl-2-pyridone-5-carboxamide (2-PY) and N<sup>1</sup>-methyl-4-pyridone-5-carboxamide (4-PY). Started with methylation of nicotinamide by nicotinamide methyltransferase according to Pryde et al., 2010.

As well as aldehydes, AOX can also catalyse the oxidation of nitrogen containing polycyclic aromatic hydrocarbons (NPAHs) which have a CH=N- group (Table 1) that are environmental pollutant, an example for these compounds is phthalazine (Figure 14), which its biotransformed to its corresponding metabolite 1-phthalazinone by mammalian AOX(Stubley et al., 1979, Beedham et al., 1990, Obach. 2004, Panoutsopoulos and Beedham. 2004, Barr and Jones. 2011).



**Figure 14:** Oxidation of phthalazine by aldehyde oxidase to 1-phthalazinone (Obach. 2004, Barr and Jones. 2011).

Another example for an nitrogen containing polycyclic aromatic hydrocarbons (NPAHs) AOX substrate is phenanthridine that is biotransformed to more polar metabolite phenanthridinone by AOX (Figure 15) aiding its excretion (Blumer *et al.*, 1977, Beedham. 2001, Kitamura *et al.*, 2006, Sorouraddin *et al.*, 2009). Phenanthridine is an azaarene pollutant found in terrestrial and aquatic environments from industrial activities associated with fossil fuels (Blumer et al., 1977, Balch et al., 1995, Osborne et al., 1997, de Voogt and Laane. 2009).



Phenanthridine

Phenanthridinone

**Figure 15:** Biotransformation of phenanthridine to its metabolite phenanthridinone by AOX. Based on .Sorouraddin et al., 2009.

One important drug substrate for AOX is the pteridine compound methotrexate (4amino-N<sup>10</sup>-methylpteroyl-L-glutamic acid). This anti-leukemia drug is biotransformed by AOX to7-hydroxymethotrexate (Figure 16) (Kitamura et al., 2006, Garattini et al., 2008, Liu et al., 2009, Pryde et al., 2010, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013).



**Figure 16:** Metabolism of methotrexate to its corresponding metabolite 7-hydroxymethotrexate by AOX. According to (Liu et al., 2009, Pryde et al., 2010, Garattini and Terao. 2013).

Another example of a substrate for AOX as well as XOR is 6-mercaptopurine. This purine drug is biotransformed by XO to 6-mercaptopurine-2-one (6-thioxanthine) and 8-hydroxy-6-mercaptopurine and further biotransformation to its final metabolite 6-thiouric acid by both AOX and XOR. 6-methylmercaptopurine which is another metabolite of 6-mercaptopurineis produced by methyltransferase which is also further biotransformation to final metabolite 8-oxo-6-methylmercaptopurine by both XO and AOX (Figure 17) (Beedham. 2001, Kitamura et al., 2006, Rashidi et al., 2007, Sorouraddin et al., 2011, Garattini and Terao. 2013).



6-methylmercaptopurine

8-Oxo-6-methylmercaptopurine

Figure 17: Biotransformation of 6-mercaptopurineby AOX and XOR.

According to (Beedham. 2001, Rashidi et al., 2007, Sorouraddin et al., 2011, Garattini and Terao. 2013)

Both AOX and XOR are also involved in reductive activity with many substrates in the presence of an electron donor in anaerobic condition. (Beedham. 2001, Obach. 2004, Kitamura et al., 2006, Beedham. 2010). The reductive activity of AOX is generally found to be greater than that of XOR with many substrates according to (Beedham. 2001). Tables 2 and 3 summarise the reductive activities of AOX and XOR with different compounds.

AOX substrates	AOX generated metabolites	References	
Acetophenone oxime, salicylaldoxime and benzamidoxime	Corresponding oxo compounds and a ketimine	(Tatsumi and Ishigai. 1987)	
Aromatic and heterocyclic hydroxamic acids (e.g salicylhydroxamic acid, nicotinohydroxamic acid and anthranilhydroxamic acid	Amides (salicylamide, nicotinamide and anthranilamide)	(Sugihara and Tatsumi. 1986, Katsura et al., 1993)	
Fenthion sulfoxide	Fenthion	(Leoni et al., 2008)	
Methyl red and orange	Cleavage of the azo linkage, Dimethyl-p-phenylenediamine and O-amino benzoic acid	(Kitamura and Tatsumi. 1983, Stoddart and Levine. 1992, Miyajima et al., 2000)	
Nicotinamide-N-oxide, imipramine-N-oxide, cyclobenzaprine-N-oxide and s-(-)-nicotine-1-N-oxide	Their parent amines (nicotinamide, imipramine, cyclobenzaprine and nicotine)	(Kitamura and Tatsumi. 1984, Sugihara et al., 1996, Kitamura et al., 2001)	
Nitrated polycyclic hydrocarbons	Hydroxylamines	(Tatsumi et al., 1986)	
Nitrosoamines (N- nitrosodiphenylamine)	Hydrazines (1,1- diphenylhydrazine)	(Tatsumi et al., 1983)	
Sulindac sulfoxide	Sulindac sulfide	(Pay et al., 1980, Duggan. 1981, Kitamura et al., 2001)	
Zonisamide	2-sulfamoylacetylphenol	(Kitamura et al., 2001)	

**Table 2**: Aldehyde oxidase catalyzed reduction of different compounds in mammals.

XOR substrates	XOR generated metabolites	References	
6-bromomethyl-(9H)-purine	6-methylpurine	(Porter. 1990)	
Doxorubicin	Its semiguinone metabolites	(Yee and Pritsos. 1997,	
DONOTUNICI	no semiquinone metabolites	Barnabe et al., 2002)	
Mitomucin C	2.7 diaminomitosono	(Pan et al., 1984, Gustafson	
Mittomychi C	2,7-drammonntosene	and Pritsos. 1992)	
Nitrated polycyclichydrocarbons	Hydroxylamines	(Tatsumi et al., 1986)	
S-(-)-nicotine-1-N-oxide	Nicotine	(Sugihara et al., 1996)	

**Table 3:** Xanthine oxidoreductase catalyzed reduction of different compounds in mammals.

## 1.4. Molybdo-flavoenzymes inhibitors.

For many years, inhibitors of AOX and XOR have been described with varying degrees of detail(Rajagopalan et al., 1962). The inhibitors have been used in many in vitro systems such as partially purified enzyme, recombinant expressed enzyme, liver cytosol and liver slices and across several species most notably rabbit, guinea pig and human. Inhibitors are proposed to affect different domains and prosthetic groups in AOX and XOR (Johns. 1967, Yoshihara and Tatsumi. 1985, Yoshihara and Tatsumi. 1997, Sugihara et al., 1999, Beedham. 2001, Kitamura et al., 2003, Vila et al., 2004, Pacher et al., 2006, Diamond et al., 2010, Pryde et al., 2010, Barr and Jones. 2011).

# 1.4.1. Aldehyde oxidase inhibitors

Examples of AOX inhibitors are menadione, estradiol, chlorpromazine and phenanthridine which are of widely differing chemical structure (Figure 18). Menadione (vitamin  $K_3$ ) is a quinone that was discovered as a specific non-competitive AOX inhibitor by Rajagopalan et al., (1962) in rabbit liver with N<sup>1</sup>-methylnicotinamide as a substrate. Since then many other researchers have confirmed this in a multitude of species summarised in Table 4 and 5 with a variety of AOX

substrates. Estradiol was found to be a non-competitive AOX inhibitor by Johns (1967) in human liver with variety of AOX substrates as summarised in Table 4.



Figure 18: Structures of the aldehyde oxidase inhibitors. According to Merck Index (2014)

Species/tissue	Substrate	Inhibitor	Substrate Concentration	Inhibitor Concentration	Ki or inhibition %	Mode of inhibition	References
Human/liver	Phthalazine and DMAC	Raloxifene	5-100µM	1-10µM	0.8nM	Uncompetitive	(Choughule et al., 2013)
Human/ liver	Phthalazine and vanillin	Menadione, estradiol and 17α-ethinyl estradiol	2μΜ	50µM	0.29µM and 0.57µM respectively	Mixed mode of inhibition, almost uncompetitive	(Obach. 2004)
Human /liver	Phthalazine	Menadione	2μΜ	50µM	0.20µM	Non-competitive	(Obach et al., 2004)
Human /liver	Phthalazine	Chlorpromazine	2μΜ	50µM	0.57µM	competitive	(Obach et al., 2004)
Human /liver	Phthalazine	estradiol, 17α-ethinyl estradiol	2μΜ	50μΜ	0.29µM and 0.57µM	Uncompetitive	(Obach et al., 2004)
Human /liver	Phthalazine	Raloxifene	2μΜ	50µM	0.0029µM	Uncompetitive	(Obach et al., 2004)
Human /liver	Benzaldehyde, phenazine methosulfate (PMS)	Menadione	56 µM	15 μM, 0.08 μM and 5 μM	55%	Non-competitive	(Johns. 1967)
Human /liver	Benzaldehyde, phenazine methosulfate (PMS)	β-estradiol	56 μΜ	15 μM, 0.08 μM and 5 μM	50%	Non-competitive	(Johns. 1967)
Human /liver	Benzaldehyde, phenazine methosulfate (PMS)	Chlorpromazine	56 µM	15 μM, 0.08 μM and 5 μM	51%	Competitive	(Johns. 1967)
Human /liver	2-Hydroxypyrimidine, benzaldehyde	Chlorpromazine	0.5µM	1 mM and 0.4 mM	45%	Competitive	(Johns. 1967)
Human/ liver	2-hydroxy-5- fluropyrimidine	Menadione	0.54µM	25 μΜ	46%	Non-competitive	(Johns. 1967)
Human/ liver	Phthalazine and DMAC	Raloxifene, estradiol, ethinyl estradiol, menadione and chlorpromazine	5-100µM	0.1µm to 5µM	0.29μM, 0.57μM, 0.2μM and 0.57μM	Uncompetitive, non- competitive	(Barr and Jones. 2011, Barr and Jones. 2013)
Human/liver	Phthalazine	Chlorpromazine and estradiol	0.03-25μM (3μM)	50μΜ	0.27 μM and 0.62 μM respectively	Non-competitive and uncompetitive	(Nirogi et al., 2013)

**Table 4:** Summary of the effects of prototypical inhibitors on human aldehyde oxidase.

Spacios/tissue	Substrate	Inhibitor	Substrate	Inhibitor	Ki or inhibition	Mode of	Defenences
Species/tissue	Substrate	minipitor	Concentration	Concentration	%	inhibition	References
Bovine/ liver	N <sup>1</sup> -methylnicotinamide	Phenanthridine	12 mM	0.2 mM	-	Competitive	(Yielding and Tomkins. 1961)
Guinea pig/liver	DMAC	Raloxifene	5-100µM	1-10µM	3.2µM	Uncompetitive	(Choughule et al., 2013)
Guinea pig /liver	Phthalazine	Raloxifene	5-100µM	1-10µM	42nM	Uncompetitive	(Choughule et al., 2013)
Rhesus monkey /liver	All-trans retinal (vitamin A)	Menadione	0.01mM, 0.1mM and 1mM	0.02mM, 0.2mM and 2mM	60%, 28% and 15%	Non-competitive	(Lakshmanan et al., 1964)
Rhesus monkey/liver	DMAC	Raloxifene	5-100µM	1-10µM	22µM	competitive	(Choughule et al., 2013)
Rhesus monkey/liver	Phthalazine	Raloxifene	5-100µM	1-10µM	230 nM	competitive	(Choughule et al., 2013)
Rabbit /liver	Phthalazine	Benzamidine	2μΜ	5mM	-	-	(Stell et al., 1989)
Rabbit / liver	Benzaldehyde, phenazine methosulfate (PMS)	Chlorpromazine	56 $\mu$ M and 0.5 $\mu$ M	5 μΜ	50% and 48%	Competitive	(Johns. 1967)
Rabbit/ liver	Benzaldehyde, 2- hydroxpyrimidine and N <sup>1</sup> -methylnicotinamide	Menadione, chlorpromazine and KCN	2mM	200μM and 500μM, 500μM and 1000μM respectively	78%, 48%, 48% and 18%	Non-competitive	(Dick et al., 2005, Dick et al., 2007)
Rabbit /liver	N <sup>1</sup> -methylnicotinamide	Menadione	1.5 mM	0.1 µM	85%	Non-competitive	(Rajagopalan et al., 1962)
Rabbit /liver	N <sup>1</sup> -methylnicotinamide	Estradiol	5 mM	0.2 µM	62%	Non-competitive	(Rajagopalan et al., 1962)
Rabbit /liver	Phenazine methosulphate, nitro blue tetrazolium (NBT)	Menadione	5 mM	1.6 μΜ	-	Non-competitive	(Rajagopalan and Handler. 1964a)
Rabbit /liver	Phenazine methosulphate, nitro blue tetrazolium (NBT)	Estradiol	5 mM	16 µM	-	Non-competitive	(Rajagopalan and Handler. 1964a)
Sea bream /liver	2-hydroxypyrimidine, N <sup>1</sup> -methylnicotinamide or butyraldehyde	Menadione and β-estradiol	12 mM	0.1 mM	75% and 60%	Non-competitive	(Kitamura et al., 2003)
Sea bream /liver	2-hydroxypyrimidine, N <sup>1</sup> -methylnicotinamide or butyraldehyde	Chlorpromazine	12 mM	0.1 mM	78%	Non-competitive	(Kitamura et al., 2003)

**Table 5:** Summary of the effect of aldehyde oxidase inhibitors with non-human species.

Since then many other researchers have confirmed this in human, rabbit and sea bream liver as summarised in Table 4 and 5 with a variety of AOX substrates. The discovery of the related xenoestrogen oral contraceptive drug 17a-ethinyl estradiol as an uncompetitive AOX inhibitor was made by in Obach 2004 in human liver with phthalazine and vanillin as a substrates. Another non-competitive inhibitor of AOX is the drug chlorpromazine Johns (1967) in a different species with benzaldehyde and phenazine methosulphate as a substrate as summarised in Table 5. Since then many other researchers have confirmed this in a human, rabbit and sea bream liver as summarised in Table 4 and 5 with a variety of AOX substrates. Interestingly the osteoporosis drug raloxifene has been found to have competitive and non-competitive properties as an AOX inhibitor by many researchers such as Barr and Jones 2011, 2013 and Choughule et al. (2013) in human, guinea pig and rhesus monkey liver with phthalazine and DMAC as a substrate (Table 4). The environmental NPAHs, phenanthridine is a competitive inhibitor in bovine liver with N<sup>1</sup>-methylnicotinamide as a substrate (Yielding and Tomkins. 1961). Finally the first researchers to find benzamidine as a specific AOX inhibitor was Stell et al. (1989) in rabbit liver with phthalazine as a substrate as summarised in Table 5.

### **1.4.2.** Oxipurinol as a specific xanthine oxidoreductase inhibitor

With regards XOR inhibitor oxipurinol is the best documented. The inhibitor oxipurinol is generated when the anti-gout substrate drug allopurinol is biotransformed by XOR (Figure 19).



**Figure 19:** Generation of the xanthine oxidoreductase inhibitor oxipurinol from allopurinol. According to Barr and Jones, 2011, Barr and Jones, 2013.

The first researcher to find allopurinol and oxipurinol as a specific XOR inhibitor was by Elion. (1966) in bovine milk, mouse and human liver with xanthine as a substrate. Since then many other researchers have confirmed this in a multitude of species summarised in (Table 6) with xanthine and pterin as XOR substrates. The mode of inhibition with allopurinol and oxipurinol is competitive (Okamoto et al., 2008) and Nishino et al., 2008) (Table 6).

Species/tissue	Substrate	Inhibitor	Substrate Concentration	Inhibitor Concentration	Ki or inhibition %	References
Bovine /milk	Xanthine	Oxipurinol	0.15mM	100µM	1 µM	(Okamoto et al., 2008)
Bovine/ milk	Hypoxanthine and xanthine	Oxipurinol	0.15mM	100µM	0.1 µM	(Nishino et al., 2008, Okamoto et al., 2008)
Bovine/ milk	Xanthine	Allopurinol	0.15mM	3.3 µM	0.9 nM	(Okamoto et al., 2003)
Bovine/ milk	Xanthine	Allopurinol	40 µM	14-28 µM	6.3µM	(Elion. 1966)
Mouse/liver	Xanthine	Allopurinol	40 µM	14-28 μM	1.2 µM	(Elion. 1966)
Human/liver	Xanthine	Allopurinol	40 µM	14-28 µM	1.1 µM	(Elion. 1966)
Human/ liver	Pterin	Allopurinol	0.6-10µM	100µM	86%	(Obach. 2004)
Human/ liver	Xanthine	Allopurinol	0.15mM	0.1-2mM	4 µM	(Krenitsky et al., 1986)
Wistar rat/liver	Pterin	Allopurinol	-	100µM	-	(Derbre et al., 2012)

**Table 6:** Summary of the effects of xanthine oxidoreductase inhibitors allopurinol and oxipurinol with different mammalian species.

### **1.5. Species variation in molybdo-flavoenzyme activity.**

Molluscs, crustaceans, insects, bacteria, fungi and all vertebrate classes have been found to contain the two molybdo-flavoenzymes (AOX and XOR)(Krenitsky et al., 1972, Wurzinger and Hartenstein. 1974, Moura et al., 1976, Woolfolk and Downard. 1977, Pometto and Crawford. 1983, Turner et al., 1987, Glatigny and Scazzocchio. 1995, Romao et al., 1995, Xu and Johnson. 1995, Schrader et al., 1999, Beedham. 2001, Yasuhara et al., 2002, Garattini et al., 2008, Beedham. 2010). Many species differences show marked quantitative effects when using cinchonidine as substrate the AOX activity was lower in monkey than that in rabbits (Fukiya et al., 2010). When is used zebularine as substrate with mouse, monkey and human and it was found that the activity of AOX was highest with mouse than that with monkey and human(Klecker et al., 2006). Sugihara et al. (1997) found when using benzaldehyde and  $N^{1}$ methylnicotinamide as substrates with human and monkey that the AOX activity with human was lower than that with monkey(Sugihara et al., 1997). In comparison, rabbit AOX has lower reductase activity toward zonisamide as a substrate as compared to monkey(Kitamura et al., 2001). When Kitamura and Tatsumi.(1984) used nicotinamide N-oxide as substrate with rabbits, hogs, guinea pig, hamster, rat and mice and it was found that all had similar AOX reductase activity(Kitamura and Tatsumi. 1984). On the other hand, Hirao et al. (1994) found lower AOX reductase activity with rat AOX than that with rabbit AOX when naphthalene 1,2-oxide and also benzo [a] pyrene-4,5-oxide were used as substrates(Hirao et al., 1994).

#### **1.5.1.** Vertebrate molybdo-flavoenzymes

The number of MFE genes varies between mammals. It was found that the numbers of active AOX genes in vertebrates are ranged from 1 to 4 (Dickinson. 1971, Lindsley and Zimm. 1992, Terao et al., 2006, Garattini et al., 2008, Garattini et al., 2009, Terao et al., 2009, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013). Some vertebrates such as Glires have four of AOX genes and one XOR gene (rat, mice, pika as well as rabbit)(Garattini et al., 2008, Terao et al., 2009, Garattini and Terao. 2011, Garattini and Terao. 2013, Kurosaki et al., 2013). Mouse AOXs are the best characterised AOX genes and proteins for all vertebrates; Table 7 summarizes the tissue distribution and substrate specificity of mouse AOX. It is interesting that other mammalian species possess a reduced complement of functional MFE genes. Dogs (Canis lupus familiaris) possess two AOX genes (AOX4 and AOX3L1) and one XOR, while bovine (Bos taurus) has three AOX genes (AOX1, AOX4 and AOX3L1) and one XOR(Terao et al., 2006, Kurosaki et al., 2013). Humans have just one functional AOX gene (AOX1) and one XOR gene (Garattini et al., 2008, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013). The chicken (Gallus gallus) possesses two AOX genes (AOX1 and AOX2) and one XOR (Terao et al., 2006, Kurosaki et al., 2013). Cat (Felix catus) has just one functional AOX gene (AOX3L1) and one XOR gene(Garattini and Terao. 2013, Kurosaki et al., 2013), while Cynomolgus monkey(Macaca fascicularis)and Rhesus monkey (Macaca mulatta) have two AOX genes (AOX1 and AOX3L1) and three AOX genes (AOX1,AOX4 and AOX3L1) respectively and one XOR(Garattini and Terao. 2013, Kurosaki et al., 2013). Horse (Equus caballus) has three AOX genes (AOX1, AOX4 and AOX3L1) and one XOR gene ((Kurosaki et al., 2013). Baboon (Papio Hamadryas) has three AOX genes (AOX1, AOX4 and AOX3L1) and only one XOR gene (Kurosaki et al., 2013). Interestingly dogs and cats do not express an AOX enzyme in hepatic tissue unlike other mammalian species studied (Terao et al., 2006, Garattini and Terao. 2013, Kurosaki et al., 2013). A listing of the current knowledge of MFEs AOX and XOR in several non-glire species is given on (Table 8 and 9).

# Table 7: Summary of molybdo-flavoenzymes in mice

The substrate specificity along with tissue distribution of molybdo-flavoenzymes in mice. Based on (Kurosaki et al., 2013).

Old enzyme nomenclature of AOX/XOR	New enzyme nomenclature of AOX/XOR	Tissue distribution	Known substrates	References
AOX1	Aox1	Liver, lung, heart, kidney, testis, brain, eye, spinal cord and spleen	Butyraldehyde, nicotinamide, octanal, phthalazine, all- trans retinaldehyde, pyridoxal, vanillin, 2- hydroxpyrimidine	(Moriwaki et al., 1996, Kurosaki et al., 1999, Wright et al., 1999, Kurosaki et al., 2004, Vila et al., 2004, Garattini et al., 2008, Garattini et al., 2009, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013)
AOH1	Aox3	Liver, lung and testis	Benzaldehyde,butyraldehyde, octanal,phthalazine, phenanthridine, all-trans retinaldehyde, 2-hydroxpyrimidine	(Tomita et al., 1993, Moriwaki et al., 1996, Kurosaki et al., 1999, Wright et al., 1999, Terao et al., 2001, Kurosaki et al., 2004, Vila et al., 2004, Garattini et al., 2008, Garattini et al., 2009, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013)
AOH2	Aox4	Skin, stomach, Haderian gland, sebaceous gland, epidermis,eye, pancreas, brain and esophagus	Phenanthridine, all-trans retinaldehyde	(Holmes. 1979, Sasai and De Robertis. 1997, Terao et al., 2001, Kurosaki et al., 2004, Vila et al., 2004, Garattini et al., 2008, Garattini et al., 2009, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013)
АОНЗ	Aox3L1	Skin, Bowmans gland and olfactory mucosa	Benzaldehyde, octanal, phthalazine, all-trans retinaldehyde, 2-hydroxpyrimidine	(Kurosaki et al., 2004, Vila et al., 2004, Garattini et al., 2008, Garattini et al., 2009, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013)
XOR	Xdh	Liver, muscle, mammary glands, oesophagus, stomach, duodenum, jejunum, lleum, colon, rectum, lung, kidney, heart, small intestine and spleen	Hypoxanthine, xanthine	(Yoshimura and Oka. 1989, Terao et al., 1992, Falciani et al., 1994, Kurosaki et al., 1995, Kuroisaki et al., 1996, McManaman et al., 1999, McManaman et al., 2002, Terao et al., 2006, Garattini et al., 2008, Kurosaki et al., 2013)

Species	AOX	Tissue distribution	References
Baboon (Papio Hamadryas)	AOX1,AOX4 and AOX3L1	Intestine extract. heart, kidney and liver	(Holmes and Vandeberg. 1986, Garattini et al., 2008, Kurosaki et al., 2013)
Bovine (Bos taurus)	AOX1,AOX4 and AOX3L1	Lung, eye, kidney, testis, liver, spleen and ciliary body	(Shimada et al., 1989, Calzei et al., 1995, Beedham. 2001, Terao et al., 2006, Garattini et al., 2008, Kurosaki et al., 2013)
Cat (Felix catus)	AOX3L1	Small intestine and kidney	(Krenitsky et al., 1974, Beedham. 2001, Terao et al., 2006, Garattini et al., 2008, Garattini and Terao. 2013, Kurosaki et al., 2013)
Chicken (Gallus gallus)	AOX1 and AOX2	Liver, Kidney brain, skin, muscle, thymus, trachea, lung, spleen, pancreas, stomach, adrenal gland and testis	(Krenitsky et al., 1974, Nishino and Nishino. 1989, Nishino et al., 1989, Terao et al., 2006, Garattini et al., 2008, Kurosaki et al., 2013)
Cynomolgus monkey (Macaca fascicularis)	AOX1 and AOX3L1	Liver and nasal mucosa	(Kawashima et al., 1999, Sugihara et al., 2000, Diamond et al., 2010, Van Rompay. 2012, Kurosaki et al., 2013)
Dog (Canis lupus familiaris)	AOX4 and AOX3L1	Testis, Nasal mucosa, lacrimal glands, skin and oesophagus	(Krenitsky et al., 1974, Vila et al., 2004, Terao et al., 2006, Garattini et al., 2008, Kurosaki et al., 2013)
Horse (Equus caballus)	AOX1,AOX4 and AOX3L1	Liver	(Seeley et al., 1984, Prichard et al., 1991, Räsänen et al., 1993, Garattini et al., 2008, Kurosaki et al., 2013)
Human (Homo sapiens)	AOX1	Lung, liver and brain	(Krenitsky et al., 1974, Minoshima et al., 1995, Terao et al., 1998, Beedham. 2001, Moriwaki et al., 2001, Peretz et al., 2007, Garattini et al., 2008, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013)
Rhesus monkey (Macaca mulatta)	AOX1,AOX4 and AOX3L1	Liver and nasal mucosa	(Krenitsky et al., 1974, Sugihara et al., 1997, Garattini et al., 2008, Fukiya et al., 2010, Garattini and Terao. 2011, Tawa et al., 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013)

**Table 8**: Summary of aldehyde oxidase genes and tissue distribution in different species

Species	Tissue distribution	References		
Baboon (Papio Hamadryas)	Intestine and liver	((Holmes and Vandeberg. 1986, Garattini et al., 2008, Kurosaki et al., 2013)		
Bovine (Bos taurus)	Heart, liver, mammary gland and milk	(Shimada et al., 1989, Calzei et al., 1995, Beedham. 2001, Terao et al., 2006, Garattini et al., 2008, Kurosaki et al., 2013)		
Cat (Felix catus)	Liver, small intestine and kidney	(Krenitsky et al., 1974, Beedham. 2001, Terao et al., 2006, Garattini et al., 2008, Kurosaki et al., 2013)		
Chicken (Gallus gallus)	Skin, lung, brain, liver, muscle, spinal cord, thymus, trachea, spleen, pancreas, stomach, adrenal gland and testis	(Nishino and Nishino. 1989, Nishino et al., 1989, Terao et al., 2006, Garattini et al., 2008, Kurosaki et al., 2013)		
Cynomolgus monkey (Macaca fascicularis)	Testis, small intestine, brain and liver	(Kawashima et al., 1999, Sugihara et al., 2000, Diamond et al., 2010, Van Rompay. 2012, Kurosaki et al., 2013)		
Dog (Canis lupus familiaris)	Nasal mucosa, skin, spleen, lacrimal glands and oesophagus	(Krenitsky et al., 1974, Vila et al., 2004, Terao et al., 2006, Garattini et al., 2008, Kurosaki et al., 2013)		
Horse (Equus caballus)	Lung, skeletal muscle, liver and small intestine	(Seeley et al., 1984, Prichard et al., 1991, Räsänen et al., 1993, Garattini et al., 2008, Kurosaki et al., 2013)		
Human (Homo sapiens)	Spleen, lung, mammary gland, kidney, liver, stomach, skeletal muscle and milk	(Krenitsky et al., 1974, Minoshima et al., 1995, Terao et al., 1998, Beedham. 2001, Moriwaki et al., 2001, Peretz et al., 2007, Garattini et al., 2008, Garattini and Terao. 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013)		
Rhesus monkey (Macaca mulatta)	Testis, small intestine, brain and liver	(Krenitsky et al., 1974, Sugihara et al., 1997, Garattini et al., 2008, Fukiya et al., 2010, Garattini and Terao. 2011, Tawa et al., 2011, Garattini and Terao. 2012, Garattini and Terao. 2013, Kurosaki et al., 2013)		

**Table 9:** Summary of xanthine oxidoreductase tissue distribution in different species

## **1.5.2.** Molybdo-flavoenzymes in fish

AOX and XOR in fish are responsible for metabolism of various compounds such as aromatic aldehydes and N-heterocyclic compounds (Lakshmanan et al., 1964, Krenitsky et al., 1974, Wurzinger and Hartenstein. 1974, Cleare et al., 1976, Kamiński and Jeżewska. 1985, Tatsumi et al., 1992, Johnson et al., 1993, Schlenk. 1998, Bleeker et al., 2001, Ueda et al., 2001, Ueda et al., 2002, Basha and Rani. 2003, Kitamura et al., 2003, Pandey et al., 2003, Ziegler. 2003, Isamah and Asagba. 2004, Resende et al., 2005, Asagba et al., 2010, Hegazi et al., 2010, Kurosaki et al., 2013). These studies either measured the activity directly by monitoring substrate or product concentrations or used artificial electron acceptors as shown in (Table 10 and Figure 3).

# 1.5.2.1. Aldehyde oxidase activity in fish liver

As described in previous sections AOX activity is principally found in hepatic tissue of mammals(Garattini and Terao. 2013, Kurosaki et al., 2013) and several workers have also demonstrated activity in this tissue in bony fish, among the most primitive vertebrates to have this activity. The first report of AOX in fish was by Lakshamanan et al. (1964) who examined the oxidation of the aldehydes of both vitamin A1 (alltrans retinal) and vitamin A2 (3,4-dehydro-all-trans retinal) to the corresponding acids by aldehyde oxidase enzyme in the liver of several freshwater-fish (Ophiocephalus muralius, Labeo calbasu) and (Cyprinus carpio communis) as shown in (Table 10). The study also found that the enzyme from freshwater-fish livers was inhibited by the AOX inhibitor menadione (Lakshmanan et al., 1964).

Common name. (Latin name)	Substrate	Electron acceptor	AOX activity	References
African catfish	D 111 1		142.0	(Isamah and Asagba.
(Clarias gariepinus)	Benzaldenyde	2,6-DCPIP	143.8	2004)
Electric catfish	Banzaldahuda	2.6 DCPIP	130	(Isamah and Asagba.
(Malapterurus electricus)	Benzaidenyde	2,0-DCFIF	150	2004)
African catfish	Benzaldehyde	2 6-DCPIP	1316.8 <sup>7d</sup> 629.35 <sup>21d</sup>	(Asagha et al. 2010)
(Clarias gariepinus)	Denizateonyte	2,0 0011	1510.0 , 025.55	(1154gou et ull., 2010)
Lesser spangled emperor	Benzaldehvde	MTT	+	(Johnson et al., 1993)
(Lthrinus choerorynchus)				(
<b>Rankin cod</b>	Benzaldehyde	MTT	+	(Johnson et al., 1993)
(Epinepheius multinotatus)				
(Lutianus sebae)	Benzaldehyde	MTT	+	(Johnson et al., 1993)
Sea bream(Pagrus major)	Benzaldehvde	0,	4 35	(Kitamura et al., 2003)
Spangled emperor	Deminiating at	02	1.00	(IIIIIIIIIII et uii, 2000)
(Lethrinus nebulosus)	Benzaldehyde	MTT	+	(Johnson et al., 1993)
Sea bream(Pagrus major)	DMAC	O <sub>2</sub>	7.3	(Kitamura et al., 2003)
Goldfish(Carassius auratus)	2-hydroxypyrimidine	Fenthion sulphoxide	0.18	(Kitamura et al., 2003)
Goldfish(Carassius auratus)	2-hydroxypyrimidine	2-nitrofluorene	+	(Ueda et al., 2001)
<b>Goldfish</b> (Carassius auratus)	2-hydroxynyrimidine	Sodium	++	(Tatsumi et al. 1992)
G L (T	2	nifurstyrenate	0.27	
Sea bream(Pagrus major)	2-hydroxypyrimidine	2-nitrofluorene	0.25	(Ueda et al., 2002)
Sea bream(Pagrus major)	2-hydroxypyrimidine	nifurstvrenate	6.5	(Tatsumi et al., 1992)
<b>Smooth dogfish</b> (Mustelus canis)	6-methylpurine	Ferricyanide	73	(Krenitsky et al., 1974)
Largemouth bass			10	
(Micropterus salmoides)	6-methylpurine	Ferricyanide	10	(Krenitsky et al., 19/4)
Florida pompano	6 mathulnurina	Forriovanida	10	(Kranitsky at al. 1074)
(Trachinotus carolinus)	0-methylpurme	Terricyanide	10	(Kieliitsky et al., 1974)
Sea mullet	6-methylpurine	Ferricvanide	10	(Krenitsky et al., 1974)
(Menticirrhus saxatilis)			17	
Blue fish (Pomatomus saltatrix)	6-methylpurine	Ferricyanide	17	(Krenitsky et al., 1974)
(Scomberonorus maculatus)	6-methylpurine	Ferricyanide	<26	(Krenitsky et al., 1974)
<b>Bluegill</b> (Lepomis macrochirus)	6-methylpurine	Ferricvanide	35	(Krenitsky et al., 1974)
Channel catfish	o meanyipanne			
(Icatalurus punctatus)	6-methylpurine	Ferricyanide	23	(Krenitsky et al., 1974)
Summer flounder	6 mathulnurina	Forriovanida	47	(Kranitsky at al. 1074)
(Paralichthyes dentatus)	0-meuryipurme	Ferricyanide	47	(Kieliitsky et al., 1974)
Goldfish(Carassius auratus)	6-methylpurine	Ferricyanide	27	(Krenitsky et al., 1974)
Norfolk spot	6-methylpurine	Ferricyanide	30	(Krenitsky et al., 1974)
(Leiostomus xanthurus)	~ 1			· · · ·
(Lenomis gibbosus)	6-methylpurine	Ferricyanide	57	(Krenitsky et al., 1974)
Nile tilania juveniles				
(Oreochromis niloticus)	Phthalazine	Ferricyanide	275	(Hegazi et al., 2010)
Northern hog sucker	¥7!11:	0		(Wurzinger and
(Hypentelium nigricans)	vaniiin	$O_2$	+	Hartenstein. 1974)
White sucker	Vanillin	0,	+	(Wurzinger and
(Catostamus commersoni)		02		Hartenstein. 1974)
Northern pike(Esox lucius)	Vanillin	$O_2$	+	(Wurzinger and Hartenstein 1974)
Pumpkinseed sunfish				(Wurzinger and
(Lepomis gibbosus)	Vanillin	$O_2$	++	Hartenstein. 1974)
Walleve (Stizostadion vitroum)	Vanillin	0		(Wurzinger and
wancyc(Suzosteuron vitreum)	v ammin	02	++	Hartenstein. 1974)
Orange-fin labeo	Vitamin A <sub>1</sub> and A <sub>2</sub>	O <sub>2</sub>	67 <sup>A1</sup> . 70.2 <sup>A2</sup>	(Lakshmanan et al.,
(Labeo calbasu)				1904) (Labaharan int
(Ophiocephalus muralius)	Vitamin A <sub>1</sub> and A <sub>2</sub>	$O_2$	100.1 <sup>A1</sup> , 112.3 <sup>A2</sup>	(Lakshmanan et al., 1964)
Common carp			A1 A2	(Lakshmanan et al
(Cyprinus carpio communis)	Vitamin A <sub>1</sub> and A <sub>2</sub>	$O_2$	41.9 <sup>A1</sup> , 42.55 <sup>A2</sup>	1964)

Table 10.Summary of in vitro studies of hepatic aldehyde oxidase activity in fish species.

Abbreviations: AOX activity nmol/min/mg liver, DMAC - p-dimethylaminocinnamaldehyde, 2, 6 DCPIP -2, 6-dichlorophenolindophenol. MTT - Dimethylthiazol-diphenyltetrazolium bromide, A1, A2- vitamin A1 and A2, 7d and 21d (Acclimatised for 14 days in large aquaria then kept in bowls of 20 litres for 7 days and 21 days). Ten years later Krenitsky et al., (1974) examined the activity of AOX with 6methylpurine as a substrate and potassium ferricyanide as electron acceptor in the liver of several fish. The study found that highest activity of AOX with pumpkinseed sunfish (Lepomis gibbosus) and lowest activity with blue fish (Pomatomus saltatrix) as shown in (Table 10) (Krenitsky et al., 1974). In the same years, Wurzinger and Hartenstein, (1974) used vanillin as an AOX substrate with O2 as an electron acceptor to measure AOX activity in livers of several fish species such as Northern hog sucker (Hypentelium nigricans), Northern pike (Esox lucius), pumpkinseed sunfish (Lepomis gibbosus) and walleye (Stizostedion vitreum) (Table 10)(Wurzinger and Hartenstein. 1974). Following these earlier studies it was not till two decades later that there was further research published on fish AOX. Tatsumi et al, (1992) examined the activity of aldehyde oxidase in goldfish (Carassius auratus) and sea bream (Pagrus major) livers using 2-hydroxypyrimidine as aldehyde oxidase substrate and sodium nifurstyrenate as electron acceptor as shown in (Table 10) (Tatsumi et al., 1992). Johanson et al. (1993) examined allozymes of aldehyde oxidase (allelic variants of this enzyme) in the livers of four different species of fish which are commercially farmed in Australia, red emperor (Lutjanus sebae), spangled emperor (Lethrinus nebulosus), lesser spangled emperor (Lethrinus choerorynchus) and Rankin cod (Epinephelus multinotatus). This study used electrophoresis and stained the allozymes in the gels, using benzaldehyde as AOX substrate and hypoxanthine as XOR substrate (Table 10) (Johnson et al., 1993). The electron acceptor dimethyl-diphenyltetrazolium bromide (MTT) was used to visualize electrophoretic variants of AOX and investigate the genetic similarities of these species of fish. The study demonstrated that whilst there was statistically significant variation in allelic frequencies in three of the species, there were no clear geographical groupings of populations. Supporting previous evidence that populations of these fish are extensively connected, even over large geographical distance. Consecutive studies by Ueda et al, (2001 and 2002) examined the activity of aldehyde oxidase in goldfish (Carassius auratus) and sea bream (Pagrus major) livers, using benzaldehyde oxidation as direct assay and 2hydroxypyrimidineas substrate with 2-nitrofluorene as electron acceptor in an anaerobic and indirect assay as shown in (Table 10)(Ueda et al., 2001, Ueda et al., 2002). Kitamura et al, (2003) also examined the activity of aldehyde oxidase with sea bream (Pagrus major) and goldfish (Carassius auratus), while using different aldehyde oxidase substrates such as benzaldehyde, dimethylaminocinnamaldehyde  $N^{1}$ -(DMAC) as direct assays. They also used 2-hydroxypyrimidine, methylnicotinamide or butyraldehyde as electron donor and the organothiophosphate insecticide fenthion as an electron acceptor as shown in (Table 10) (Kitamura et al., 2003). Isamah and Asagba, (2004) and Asagba et al, (2010) tested the AOX activity with benzaldehyde as a substrate with African catfish (Clarias gariepinus) and electric catfish (Malapterurus electricus) in liver using 2, 6-dichlorpheno-indophenol (2, 6-DCPIP) as an artificial electron acceptor as shown in (Table 10)(Isamah and Asagba. 2004, Asagba et al., 2010). While Hegazi et al, (2010). examined the activity of aldehyde oxidase in juvenile Nile tilapia (Oreochromis niloticus) liver using phthalazine as substrate and potassium ferricyanide as an electron acceptor, which reduced into ferricyanide as shown in (Table 10)(Hegazi et al., 2010).

Several fish have been reported to have no AOX activity as shown in (Table 11). As well as in vitro studies, in vivo studies have suggested AOX activity towards NPAHs environmental pollutants exist in fish. Bean et al (1985) demonstrated that quinoline was metabolised to hydroxyquinoline in rainbow trout (Oncorhynchus mykiss) (Bean et al., 1985). Bleeker et al, (2001) examined the metabolism of the environmental

pollutant phenanthridine in carp (Cyprium carpio) liver and bile into its metabolite 6(5H)-phenanthridinone (phenanthridone) suggesting that fish AOX catalysed this biotransformation (Bleeker et al., 2001).

**Table 11:** Summary of fish species that have no detectable hepatic AOX activity. Based on

 Wurzinger and Hartenstein, (1974)

Common name	Latin name	Substrate	Electron acceptor
Stoneroller Minnow	Campostoma anomalous	Vanillin	$O_2$
Carp	Cyprium carpio	Vanillin	O <sub>2</sub>
Cutlips Minnow	Exoglossum maxilingua	Vanillin	$O_2$
Fathead Minnow	Pimephales promelas	Vanillin	$O_2$
Margined Madtom	Noturus insignis	Vanillin	$O_2$
Yellow Perch	Perca flavescens	Vanillin	O <sub>2</sub>

# **1.5.2.2. Extrahepatic aldehyde oxidase activity in fish**

Krenitsky et al., (1974) examined the activity of aldehyde oxidase with 6methylpurine and potassium ferricyanide as electron acceptor. Two Cypriniformes and one Perciformes were tested for extrahepatic tissue distribution in intestine and kidney. All species studied had measurable extrahepatic activity that was found to be two and one fold lower respectively compared with liver, the activity in extrahepatic tissue less than that in hepatic tissue (Table 12) (Krenitsky et al., 1974). Asagba et al, (2010) tested the AOX activity using benzaldehyde as a substrate with a Siluriformes (African catfish) in kidney, gills, muscle and brain using 2,6-dichlorpheno-indophenol (2,6-DCPIP) as an artificial electron acceptor as shown in (Table 12). Interestingly Asagba et al (2010) found that AOX activity decreased in African catfish acclimatised for 14 days in large aquaria then kept in bowls of 20 litres for 7 days and 21 days. Although the reason for this was not given this could be due to stress as two fish of approximately 30 cm were kept in small 20 litre bowl.

<b>Table 12:</b> Summary of extrahepatic and hepatic AOX activity in fish species. Based on	
Krenitsky et al, (1974)*, and Asagba et al, (2010) **.	

Order	Family	Common name	Latin name	Habitat	Organ	AOX
Cypriniformes*	Cyprinidae	Goldfish	Carassius auratus	F/P	liver	27
Cypriniformes*	Cyprinidae	Goldfish	Carassius auratus	F/P	intestine	<14
Cypriniformes*	Cyprinidae	Goldfish	Carassius auratus	F/P	kidney	19
Perciformes*	Sciaenidae	Norfolk spot	Leiostomus xanthurus	S/P	liver	< 30
Perciformes*	Sciaenidae	Norfolk spot	Leiostomus xanthurus	S/P	intestine	19
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	liver 7d	1316.8
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	liver 21d	629.35
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	kidney 7d	640.38
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	kidney 21d	460.48
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	gill 7d	248.3
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	gill 21d	349.7
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	muscle 7d	148.45
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	muscle 21d	107.5
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	brain 7d	275.8
Siluriformes **	Clariidae	African catfish	Clarias gariepinus	F/P	brain 21d	125.6

Activity measured: AOX activity µmol/min/g of tissue. Abbreviations: (S) salt water (marine), and (F) fresh water, (D) Demersal (benthic), (P) Pelagic, 7d and 21d (Acclimatised for 14 days in large aquaria then kept in bowls of 20 litres for 7 days and 21 days)

# 1.5.2.3. Aldehyde oxidase as reductase activity in fish

The reductase activity of AOX has long been established since Krenitsky found it in 1974(Krenitsky et al., 1974). The first report of AOX as a reductase in fish was by Tatsumi et al, (1992). Tatsumi's group examined the activity of aldehyde oxidase as reductase with antibacterial sodium nifurstyrenate as electron acceptor which was reduced into cyano-pentenone, as an anaerobic and indirect reaction in goldfish (Carassius auratus) and sea bream (Pagrus major) livers, using 2-hydoxypyrimidine as an electron donor as shown in (Table 7)(Tatsumi et al., 1992). A decade later Ueda et al, (2001 and 2002) examined the reductase activity of aldehyde oxidase using 2-nitrofluorene which is a nitropolycyclic aromatic hydrocarbon (environmental pollutants found in automobile exhaust) as an electron acceptor. They demonstrated that this was reduced into 2-aminofluorene in goldfish (Carassius auratus) liver, using 2hydroxypyrimidine as electron donor (Table 10) and inhibited by menadione as shown in (Table 10) (Ueda et al., 2001, Ueda et al., 2002). Kitamura et al, (2003)

examined the activity of aldehyde oxidase as reductase using fenthion sulphoxide (an insecticide) as electron acceptor which it reduced into fenthion in goldfish (Carassius auratus), using 2-hydroxypyrimidine as an electron donor as shown in (Table 10) (Kitamura et al., 2003). Collectively these studies on AOX as a reductase in fish liver demonstrated that the enzyme has the capacity to reductively metabolise compounds that are environmental pollutants demonstrating the importance of the study of this important detoxication enzyme in these organisms. In vivo the physiological relevance of reductase activity is not known as these xenobiotic electron acceptors will be competing with a natural endogenous electron acceptor.

## **1.5.2.4.** Xanthine oxidoreductase activity in fish

It was not until 10 years after AOX activity was reported in fish that the first report of XOR activity in fish appeared Krenitsky et al, (1974). Using xanthine as XOR substrate with NAD<sup>+</sup> or ferricyanide as an electron acceptor as shown in (Table 13)Krenitsky et al (1974) found activity in 11 fish species (Krenitsky et al., 1974). Wurzinger and Hartenstein, (1974) also examined the activity of XOR using hypoxanthine as a XOR substrate, using NAD<sup>+</sup> as an electron acceptor in the liver of several fish species as shown in (Table 13) (Wurzinger and Hartenstein. 1974). Cleare et al, (1976) demonstrated XOR activity with several fish liver species such as brill (Scophthalmus rhombus), cod (Gadus morhua), dab (Limanda limanda), lemon sole (Microstomus kitt), mackerel (Scomber scombrus), plaice (Pleuronectes platessa), perch (Perca fluviatilis) and whiting (Merlangius merlangus) using xanthine as a XOR substrate, O<sub>2</sub> or 2,6-dichlorophenol-indophenol (2,6-DCPIP) and NAD<sup>+</sup> as an electron acceptor as shown in (Table 13)(Cleare et al., 1976).

Common name, Latin name	Substrate	Electron acceptor	XO XOR		% XDH	References
Carp (Cyprinus carpio)	Hypoxanthine	$NAD^+$	0	0	++	(Wurzinger and Hartenstein. 1974)
Cutlips minnow (Exoglossum maxilingua)	Hypoxanthine	$NAD^+$	0	0	++	(Wurzinger and Hartenstein. 1974)
Margined madtom (Noturus insignis)	Hypoxanthine	$\mathbf{NAD}^+$	0	-	+	(Wurzinger and Hartenstein. 1974)
<b>Northern hog sucker</b> (Hypentelium nigricans)	Hypoxanthine	$\mathbf{NAD}^+$	0	-	+	(Wurzinger and Hartenstein. 1974)
Stoneroller minnow	Hypoxanthine	$\mathbf{NAD}^+$ and $\mathbf{INT}$	0	0	++	(Wurzinger and Hartenstein.
White sucker(Catostamus commersoni)	Hypoxanthine	NAD <sup>+</sup>	0	-	+	(Wurzinger and Hartenstein.
Yellow perch (Perca flavescens)	Hypoxanthine	$\mathbf{NAD}^+$	0	-	++	(Wurzinger and Hartenstein. 1974)
Brown trout (Salmo trutta)	Purine	$NAD^+$	+	3-9	+	(Resende et al., 2005)
Zebra fish (Danio rerio)	Pteridine	O2 or NAD <sup>+</sup>	+	+	+	(Ziegler. 2003)
African catfish (Clarias gariepinus)	Xanthine	O <sub>2</sub>	+	76 <sup>7d</sup> , 60 <sup>21d</sup>	+	(Asagba et al., 2010)
African catfish (Clarias gariepinus)	Xanthine	O <sub>2</sub>	+	197	+	(Isamah and Asagba. 2004)
Electric catfish (Malapterurus electricus)	Xanthine	O <sub>2</sub>	+	255.8	+	(Isamah and Asagba. 2004)
Channel catfish (Icatalurus punctatus)	Xanthine	Ferricyanide and NAD <sup>+</sup>	<3	100	97	(Krenitsky et al., 1974)
Spanish mackerel (Scomberonorus maculatus)	Xanthine	NAD <sup>+</sup> and Ferricyanide	14	120	88.4	(Krenitsky et al., 1974)
Pumpkinseed sunfish (Lepomis gibbosus)	Xanthine	NAD <sup>+</sup> and Ferricyanide	<2	95	97.9	(Krenitsky et al., 1974)
Sea mullet (Menticirrhus saxatilis)	Xanthine	NAD <sup>+</sup> and Ferricyanide	-	150	100	(Krenitsky et al., 1974)
Norfolk Spot (Leiostomus xanthurus)	Xanthine	NAD <sup>+</sup> and Ferricyanide	<6	180	96.7	(Krenitsky et al., 1974)
Gold fish (Carassius auratus)	Xanthine	NAD <sup>+</sup> and Ferricyanide	8	120	93.4	(Krenitsky et al., 1974)
Largemouth bass (Micropterus salmoides)	Xanthine	NAD <sup>+</sup> and Ferricyanide	<3	410	99.3	(Krenitsky et al., 1974)
Florida pompano (Trachinotus carolinus)	Xanthine	NAD <sup>+</sup>	-	180	100	(Krenitsky et al., 1974)
Summer flounder (Paralichthyes dentatus)	Xanthine	NAD <sup>+</sup> and Ferricvanide	5	210	97.6	(Krenitsky et al., 1974)
Blue fish (Pomatomus saltatrix)	Xanthine	Ferricvanide and NAD <sup>+</sup>	2	38	94.7	(Krenitsky et al., 1974)
Bluegill (Lepomis macrochirus)	Xanthine	NAD <sup>+</sup> and Ferricvanide	<3	180	98.4	(Krenitsky et al., 1974)
Megrim (Lepidorhombus whiffagonis)	Xanthine	O <sub>2</sub> or 2.6-DCPIP	5	0	0	(Cleare et al. 1976)
Pike (Esox lucius)	Xanthine	O <sub>2</sub> or 2,6-DCPIP and NAD <sup>+</sup>	0	123	100	(Cleare et al., 1976)
Whiting (Merlangius merlangus)	Xanthine	O <sub>2</sub> or 2,6-DCPIP and NAD <sup>+</sup>	0	98	100	(Cleare et al., 1976)
Cod (Gadus morhua)	Xanthine	O <sub>2</sub> or 2,6-DCPIP and NAD <sup>+</sup>	0	98	100	(Cleare et al., 1976)
Dab (Limanda limanda)	Xanthine	O <sub>2</sub> or 2,6-DCPIP and NAD <sup>+</sup>	0	221	100	(Cleare et al., 1976)
Brill (Scophthalmus rhombus)	Xanthine	O <sub>2</sub> or 2,6-DCPIP	0	197	100	(Cleare et al., 1976)
Lesser spotted dogfish (Scyliorhinus canicula)	Xanthine	O <sub>2</sub> or 2,6-DCPIP	0	24	100	(Cleare et al., 1976)
Ray (Dipturus canutus)	Xanthine	O2 or 2,6-DCPIP	0	148	100	(Cleare et al., 1976)
Herring (Clupea harengus)	Xanthine	O <sub>2</sub> or 2,6-DCPIP	0	221	100	(Cleare et al., 1976)
White pollack (Pollachius virene)	Xanthine	O2 or 2,6-DCPIP	0	0	100	(Cleare et al., 1976)
Haddock (Melanogrammus aeglefinus)	Xanthine	O2 or 2,6-DCPIP	0	-	100	(Cleare et al., 1976)
Summer flounder (Paralichthyes dentatus)	Xanthine	O2 or 2,6-DCPIP	0	-	100	(Cleare et al., 1976)
Black sole (Solea solea)	Xanthine	O2 or 2,6-DCPIP	0	-	100	(Cleare et al., 1976)
White sole (Solea solea)	Xanthine	O2 or 2,6-DCPIP	0	73	100	(Cleare et al., 1976)
White trout (Salmo trutta)	Xanthine	O2 or 2,6-DCPIP	0	116	100	(Cleare et al., 1976)
Gurnard ( Chelidonichthys cuculus)	Xanthine	O2 or 2,6-DCPIP	0	14	100	(Cleare et al., 1976)
John dory (Zeus faber)	Xanthine	O2 or 2,6-DCPIP	0	-	100	(Cleare et al., 1976)
Lemon sole (Microstomus kitt)	Xanthine	O <sub>2</sub> or 2,6-DCPIP	0	52	100	(Cleare et al., 1976)
Mackerel (Scomber scombrus)	Xanthine	O2 or 2,6-DCPIP	0	44	100	(Cleare et al., 1976)
Plaice (Pleuronectes platessa)	Xanthine	O2 or 2,6-DCPIP	0	79	100	(Cleare et al., 1976)
Perch (Perca fluviatilis)	Xanthine	2,6-DCPIP	0	248	100	(Cleare et al., 1976)
Carp (Cyprinus carpio)	Xanthine or Hypoxanthine	NAD <sup>+</sup> or O <sub>2</sub>	+	-	99.4	(Kamiński and Jeżewska. 1985)
Giant catfish (Wallago attu)	Xanthine	NAD <sup>+</sup> or O <sub>2</sub>	+	528	+	(Pandey et al., 2003)
Nile tilapia juveniles (Oreochromis niloticus)	Xanthine	$NAD^+$	+	120	-	(Hegazi et al., 2010)
Mozambique tilapia	V	N14 D+		1/7		(Destre and D 1 2002)
(Oreochromis mossambicus)	Aanthine	NAD	+	107	+	(Basna and Rani. 2003)

# **Table 13:** In vitro studies of hepatic xanthine oxidoreductase activity in fish species measured using spectrophotometry.

Abbreviations: 2, 6-DCPIP – 2, 6-dichlorophenol-indophenol. MTT - Dimethylthiazol-diphenyltetrazolium bromide, PMS-phenazine methosulphate. INT – iodonitro-tetrazolium, 7d and 21d (Acclimatised for 14 days in large aquaria then kept in bowls of 20 litres for 7 days and 21 days), XOR = sum of activity with  $O_2$  and NAD<sup>+</sup> as electron acceptor, XO = activity with  $O_2$  as electron acceptor.

In addition, Kaminski and Jezewska, (1985) examined the activity of XOR in carp (Cyprinus carpio) liver using xanthine or hypoxanthine as substrate and  $NAD^+$  or  $O_2$ as an electron acceptor as shown in (Table 13) (Kamiński and Jeżewska. 1985). While, Johnson et al, (1993) examined the activity of XO with several fish liver species such as lesser spangled emperor (Lthrinus choerorynchus), rankin cod (Epinephelus multinotatus), red emperor (Lutjanus sebae) and spangled emperor (Lethrinus nebulosus) with hypoxanthine as substrate and dimethylthiazoldiphenyltetrazolium (MTT) as an electron acceptor as shown in (Table 15) (Johnson et al., 1993). Pandey et al, (2003) used xanthine as XO substrate and(PMS) phenazine methosulphate as an electron acceptor to examine the activity of XO in giant catfish (Wallago attu) liver as shown in (Table 13)(Pandey et al., 2003). Basha and Rani, (2003) examined the activity of XDH with Mozambique tilapia (Oreochromis mossambicus) liver, using xanthine as substrate and  $NAD^+$  as an electron acceptor as shown in (Table 13)(Basha and Rani. 2003). In addition, Ziegler, (2003) examined the activity of XOR using pteridine as substrate and O<sub>2</sub> or NAD<sup>+</sup> as an electron acceptor with zebra fish (Danio rerio) liver as shown in (Table 13) (Ziegler. 2003). Isamah and Asagba, (2004) using xanthine as substrate and  $O_2$  as an electron acceptor to examine the activity of XO in several Western Niger-Delta fish liver species such as African catfish (Clarias gariepinus) and electric catfish (Malapterurus electricus) as shown in (Table 13)(Isamah and Asagba. 2004). Resende et al, (2005) examined the activity of XDH in brown trout (Salmo trutta) liver, using purine as substrate and NAD<sup>+</sup> as an electron acceptor, as shown in (Table 13) (Resende et al., 2005). Xanthine as substrate and O2 as an electron acceptor in African catfish (Clarias gariepinus) liver from Delta State, Nigeria as shown in (Table 13) (Asagba et al., 2010). In addition, Hegazi et al, (2010) examined the activity of XDH in Nile tilapia juveniles (Oreochromis niloticus) liver. Using xanthine as substrate and  $NAD^+$  as an electron acceptor, as shown in (Table 13)(Hegazi et al., 2010). Several fish have been reported to have no XOR activity as shown in (Table 14).

Common name	Latin name	substrate	Electron acceptor	Reference
Smooth dogfish	Mustelus canis	Xanthine	$\mathbf{NAD}^{+}$ and Ferricyanide	(Krenitsky et al., 1974)
Northern pike	Esox lucius	Hypoxanthine	$\mathbf{NAD}^{+}$	(Wurzinger and Hartenstein. 1974)
Pumpkinseed sunfish	Lepomis gibbosus	Hypoxanthine	$\mathbf{NAD}^{+}$	(Wurzinger and Hartenstein. 1974)
Walleye	Stizostedion vitreum	Hypoxanthine	$\mathbf{NAD}^{+}$	(Wurzinger and Hartenstein. 1974)

Table 14: Summary of fish species that had no detectable XOR activity.

As well as the spectrophotometric studies above many different researchers also used electrophoresis and histochemical dye methods to investigate the hepatic xanthine oxidoreductase activity in different fish as seen in (Table 15).

Common name, Latin name	Substrate	Electron acceptor	XOR	XDH	References
lesser spangled emperor (Lthrinus choerorynchus)	Hypoxanthine	MTT	+	+	(Johnson et al., 1993)
<b>Rankin cod</b> (Epinephelus multinotatus)	Hypoxanthine	MTT	+	+	(Johnson et al., 1993)
Red emperor (Lutjanus sebae)	Hypoxanthine	MTT	+	+	(Johnson et al., 1993)
Spangled emperor (Lethrinus nebulosus)	Hypoxanthine	MTT	+	+	(Johnson et al., 1993)
Spotted murrel (Channa punctatus)	Xanthine	MTT	-	-	(Sastry and Rao. 1984)
Spotted murrel (Channa punctatus)	Xanthine	MTT	-	-	(Sastry and Rao. 1982)
Barbell (Barbus bocagei)	Xanthine	PMS	-	-	(Peixoto et al., 2013)
Orangefin labeo (Labeo calbasu)	Hypoxanthine	MTT	+	+	(Singh et al., 2010)
White sea bream (Diplodus sargus)	Hypoxanthine	MTT	+	-	(Lenfant. 2003)
Juvenile Chinese sturgeon (Acipenser sinensis)	Hypoxanthine	MTT	+	-	(Feng et al., 2012)
<b>White crappie</b> (Pomoxis annularis) <b>adult</b>	Hypoxanthine	EBT	+	+	(Epifanio and Philipp. 1993)
White crappie (Pomoxis annularis) embryos or larva	Hypoxanthine	EBT	+	-	(Epifanio and Philipp. 1994)
Black crappie (Pomoxis nigromaculatus) adult	Hypoxanthine	EBT	+	+	(Epifanio and Philipp. 1993)
Black crappie (Pomoxis nigromaculatus) embryos or larva	Hypoxanthine	EBT	+	-	(Epifanio and Philipp. 1994)
Atlantic mackerel (Scomber scombrus)	Xanthine	TTC	+	+	(Smith and Jamieson. 1978)
Japanese medaka (Oryzias latipes)	Xanthine	PMS + NBT	+	+	(Nakano and Whiteley. 1965)

**Table 15:** Summary of in vitro studies of hepatic xanthine oxidoreductase activity in fish species using electrophoresis and histochemical dye methods

Abbreviations: TTC – Tetrazolium chloride, EBT – Erochrome black T, MTT - Dimethylthiazol-diphenyltetrazolium bromide, PMS- phenazine methosulphate, NBT- nitro blue tetrazolium.

# 1.5.2.5. Extrahepatic xanthine oxidoreductase activity in fish

Krenitsky et al, (1974) used a spectrophotometric assay, xanthine as XOR substrate and NAD<sup>+</sup> or ferricyanide as electron acceptor to measure extrahepatic XOR activity in several fish species. Two Cypriniformes and one Perciformes were tested for extrahepatic tissue distribution in intestine and kidney XOR activity. All species studied had measurable extrahepatic activity that was found to be 4-fold lower in intestine and 1.5-fold more in kidney compared with liver, the highest activity was found in the kidney of the goldfish (Carassius auratus) (Table 16) (Krenitsky et al., 1974). The activity of extrahepatic XOR tested by Resende et al, (2005)was tested in one Salmoniformes in which the activity found in kidney of brown trout (Salmo trutta) was found to be 2-fold more compared with liver (Table 16)(Resende et al., 2005). Hegazi et al, (2010) also examined the extrahepatic activity of XOR in one Perciformes and found the activity in white muscle of juvenile Nile tilapia (Oreochromis niloticus) 1.3-fold lower compared with liver (Table 16) (Hegazi et al., 2010). Temple et al., (1979) examined the extrahepatic XOR activity in intestine of goldfish (Carassius auratus) with 100% XDH (Temple et al., 1979). Sastry and Rao. (1982 and 1984) found the activity of XOR in different tissue of spotted murrel (Channa punctatus) in kidney, gills, brain, intestine and muscle (Sastry and Rao. 1982, Sastry and Rao. 1984). Hari and Neeraja (2012) examined the extrahepatic activity of XOR in one Cypriniformes and found the activity in the kidney of Common carp (Cyprinus carpio) as XO(Hari and Neeraja. 2012). Nakano and Whiteley (1965) examined the extrahepatic activity of XOR in one Beloniformes and found the activity in intestines of Japanese medaka (Oryzias latipes) (Nakano and Whiteley. 1965).

Table 16: Extrahepatic XOR	activity in fish
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Order	Family	Common name	Latin name	Habitat	Organ	XO	XOR	% XDH
Cypriniformes *	Cyprinidae	Goldfish	Carassius auratus	F/P	liver	8	120	93.4
Cypriniformes *	Cyprinidae	Goldfish	Carassius auratus	F/P	Intestine	3	33	91
Cypriniformes ****	Cyprinidae	Goldfish	Carassius auratus	F/P	Intestine	-	-	100
Cypriniformes *	Cyprinidae	Goldfish	Carassius auratus	F/P	Kidney	11	190	94.2
Cypriniformes †††	Cyprinidae	Common carp	Cyprinus carpio	S/P	Kidney	42.6	-	-
Perciformes *	Sciaenidae	Norfolk spot	Leiostomus xanthurus	S/P	liver	<6	180	96.7
Perciformes *	Sciaenidae	Norfolk spot	Leiostomus xanthurus	S/P	Intestine	<4	90	<95.6
Perciformes ***	Cichlidae	Nile tilapia juveniles	Oreochromis niloticus	F/P	liver	+	160	99.4
Perciformes ***	Cichlidae	Nile tilapia juveniles	Oreochromis niloticus	F/P	White muscle	-	120	-
Perciformes †	Sciaenidae	Spotted murrel	Channa punctatus	S/P	liver	6.2	-	-
Perciformes†	Sciaenidae	Spotted murrel	Channa punctatus	S/P	kidney	8.6	-	-
Perciformes †	Sciaenidae	Spotted murrel	Channa punctatus	S/P	gills	2.9	-	-
Perciformes†	Sciaenidae	Spotted murrel	Channa punctatus	S/P	intestine	2.7	-	-
Perciformes†	Sciaenidae	Spotted murrel	Channa punctatus	S/P	brain	8.2	-	-
Perciformes †	Sciaenidae	Spotted murrel	Channa punctatus	S/P	muscles	5.5	-	-
Perciformes ††	Sciaenidae	Spotted murrel	Channa punctatus	S/P	liver	3.7	-	-
Perciformes ††	Sciaenidae	Spotted murrel	Channa punctatus	S/P	kidney	5.7	-	-
Perciformes ††	Sciaenidae	Spotted murrel	Channa punctatus	S/P	gills	4.2	-	-
Perciformes ††	Sciaenidae	Spotted murrel	Channa punctatus	S/P	intestine	8.0	-	-
Perciformes ††	Sciaenidae	Spotted murrel	Channa punctatus	S/P	brain	5.0	-	-
Perciformes ††	Sciaenidae	Spotted murrel	Channa punctatus	S/P	muscles	3.8	-	-
Salmoniformes **	Salmonidae	Brown trout	Salmo trutta	F/P	liver	+	3-9	+
Salmoniformes **	Salmonidae	Brown trout	Salmo trutta	F/P	Kidney	+	1.5-5	+
Beloniformes ††††	adrianichthyidae	Japanese medaka	Oryzias latipes	S/P	Liver	-	-	+
Beloniformes ††††	adrianichthyidae	Japanese medaka	Oryzias latipes	S/P	Gut	-	-	+

Abbreviations: F-freshwater fish, S- saltwater fish, P- pelagic. Activity measured: XDH and XO activity nmol/min/mg of tissue, XO urate formed with  $O_2$ , XOR urate formed with  $O_2$  and NAD<sup>+</sup>. Based on Krenitsky et al, (1974)\*, Resende et al, (2005) \*\*, Hegazi et al,(2010) \*\*\*, Temple et al, (1979) \*\*\*\*, (Sastry and Rao 1984) †,(Sastry and Rao 1982)††, (Hari and Neeraja 2012)†††and (Nakano and Whiteley 1965)††††.

# **1.5.3.** Molybdo-flavoenzymes as pollution monitoring in fish

As AOX and XOR biotransform substrates to more polar compounds, several research groups have measured the activity of these enzymes in animals exposed to pollutant to determine if they are increased or decreased and could be used as pollution biomarkers. Basha and Rani, (2003) examined induction of enzymes in fish in response to oxidative stress caused by heavy metals (such as cadmium, zinc, mercury and copper) which are released into ponds, lakes and rivers in industrial effluents in India. They discovered significant elevations of detoxification enzymes including XO in the liver and kidney of the freshwater teleost, tilapia (Oreochromis mossabicus), using xanthine as substrate and  $NAD^+$ , iodophenyl-tetrazolium chloride (INT) as electron acceptor. As a result of cadmium toxicity and it was found that the XOR increased ~1.8 fold in liver and in kidney(Table 17) (Basha and Rani. 2003). Pandey et al. (2003) examined biomarker responses including XO in several tissues of the freshwater giant catfish (Wallago attu), obtained from two different sites, which varied in levels of pollution, along the river Yamuna in India (see Table 17 and 18), Water from the site, was more heavily polluted as a result of increased industrial activity and chemical use, was found to have higher pH and lower level of dissolved oxygen compared to the less polluted site. Using xanthine as substrate and phenazine methosulphate (PMS) as electron acceptor, the researchers found that XO activity was significantly higher in fish taken from the heavily polluted area and used this evidence to propose that such XO be used future monitoring of aquatic ecosystem pollution. They found that activity was increased 2.47-fold in liver, 4.49-fold in gill and 2.11fold in kidney(Pandey et al., 2003). Interestingly Sastry and Rao (1982 and 1984) examined the pollution effect of mercuric chloride with spotted murrel (Channa punctatus) liver and found no change on XOR activity by using xanthine as substrate

(Table 19). In addition the activity of hepatic XOR tested with heavy metal pollution with barbell (Barbus bocagei)found no change on its activity (Peixoto et al., 2013) (Table 17).

Common name	Latin name	Pollutant	Fold increase/ decrease	References	
African catfish	Clarias gariepinus	Heavy metal	2.51 increase	(Isamah and Asagba. 2004)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	No change	(Asagba et al., 2010)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	No change	(Asagba et al., 2010)	
Electric catfish	Malapterurus electricus	Heavy metal	2.37 increase	(Isamah and Asagba. 2004)	
Giant catfish	Wallago attu	Heavy metal	2.47 increased	(Pandey et al., 2003)	
Mozambique tilapia	Oreochromis mossabicus	Cadmium (5 mg/L 15 days)	1.86 increase	(Basha and Rani. 2003)	
Nile tilapia juveniles	Oreochromis niloticus	TAN (5 mg/L 70 days)	2.9 increase	(Hegazi et al., 2010)	
Nile tilapia juveniles	Oreochromis niloticus	TAN (10 mg/L 70 days)	3.5 increase	(Hegazi et al., 2010)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/liter)	No change	(Sastry and Rao. 1984)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/liter)	No change	(Sastry and Rao. 1982)	
Barbell	Barbus bocagei	Heavy metal	No change	(Peixoto et al., 2013)	

Table 17: Hepatic OR activity as a biomarkers for pollution in fish.

Abbreviations: 7d for 7 days exposure, 21d for 21 day exposure, 7d and 21d (acclimatised for 14 days in large aquaria then kept in bowls of 20 litres for 7 days and 21 days), TAN – Total ammonia nitrogen. Using xanthine as substrate.

Common name	Latin name	Pollutant	Fold increase/ decrease	Organ	References	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	1.42 decreased	Kidney	(Asagba et al., 2010)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	1.99 decreased	Kidney	(Asagba et al., 2010)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	2.33 decreased	Gill	(Asagba et al., 2010)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	2.28 decreased	Gill	(Asagba et al., 2010)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	No change	Muscle	(Asagba et al., 2010)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	No change	Muscle	(Asagba et al., 2010)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	No change	Brain	(Asagba et al., 2010)	
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	No change	Brain	(Asagba et al., 2010)	
Giant catfish	Wallago attu	Heavy metal	2.11 increased	Kidney	(Pandey et al., 2003)	
Giant catfish	Wallago attu	Heavy metal	4.49 increased	Gill	(Pandey et al., 2003)	
Mozambique tilapia	Oreochromis mossabicus	Cadmium (5 mg/L 15 days)	1.84 increase	Kidney	(Basha and Rani. 2003)	
Nile tilapia juveniles	Oreochromis niloticus	TAN (5 mg/L 70 days)	2.25 increase	White muscle	(Hegazi et al., 2010)	
Nile tilapia juveniles	Oreochromis niloticus	TAN (10 mg/L 70 days)	2.6 increase	White muscle	(Hegazi et al., 2010)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Kidney	(Sastry and Rao. 1984)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	2 increase	Gills	(Sastry and Rao. 1984)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Intestine	Sastry and Rao.1984)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Brain	(Sastry and Rao. 1984)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Muscle	Sastry and Rao.1984)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Kidney	(Sastry and Rao. 1982)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Kidney	(Sastry and Rao. 1982)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Kidney	(Sastry and Rao. 1982)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Kidney	(Sastry and Rao. 1982)	
Spotted murrel	Channa punctatus	Mercuric chloride (3 µg/litre)	No change	Kidney	(Sastry and Rao. 1982)	
Common carp	Cyprinus carpio	Ammonia (5 mg/L 7 days)	2 decrease	Liver	(Hari and Neeraja. 2012)	

**Table 18:** Extrahepatic XOR activity as biomarkers for pollution in fish.

Abbreviations: 7d for 7 days exposure, 21d for 21 day exposure, 7d and 21d (acclimatised for 14 days in large aquaria then kept in bowls of 20 litres for 7 days and 21 days), TAN – Total ammonia nitrogen. All activity was measured using xanthine as substrate.

Isamah and Asagba, (2004) examined XO and AOX activity in the liver of electric catfish (Malapterurus electricus) and African catfish (Clarias gariepinus), using xanthine and benzaldehyde as substrate respectively, taken from two different sites along the Warri river in the Western Niger-Delta (see Table 19). Different levels of pollution as result of various effluent entries into the river were present at the two sample sites. It was found that XO may act as a marker of stress due to pollution by potentially increasing with elevated levels of contaminants and it found the activity of XO increased 2.37-fold and AOX activity increased 1.46-fold in electric catfish (Malapterurus electricus) and in African catfish (Clarias gariepinus) AOX activity increased 2.35-fold and XO activity increased 2.51-fold (Table 17 and 19) (Isamah and Asagba. 2004). Asagba et al. (2010) examined induction of antioxidant defences in fish in response to oxidative stress caused by heavy metals such as cadmium, which are released into ponds, lakes and rivers in industrial effluents. They examined the effect of the exposure to cadmium over time on the activities of AOX in the liver, kidney and gills and xanthine oxidase in the liver of the catfish (Clarias gariepinus), using benzaldehyde and xanthine as substrate respectively(see Table 19). Asagba et al. (2010) was found the cadmium inhibited the activity of AOX and xanthine XOR in the liver, gill and kidney of catfish. They found the activity of AOX in liver decreased 35.7-fold, kidney with 7 day exposure decreased 2.23-fold and 21 day exposure decreased 1.62-foldand gills with 7 day exposure decreased 3.28-fold and 21 day exposure decreased 1.88-fold, while XO in kidney with 7 day exposure decreased 1.42-fold and 21 day exposure decreased 1.99-fold decreased and gills with 7 day exposure decreased 2.33-fold and 21 day exposure decreased 2.28-fold (Asagba et al., 2010). Hegazi et al. (2010) examined the induction of AOX and XO in liver with phthalazine and xanthine as substrates respectively in liver and white muscle of

Common name	Latin name	Pollutant	Substrates	Fold increase/ decrease	Organ	References
African catfish	Clarias gariepinus	Heavy metal	Benzaldehyde	2.35 increase	liver	(Isamah and Asagba. 2004)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	Benzaldehyde	35.7 decrease	liver	(Asagba et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	Benzaldehyde	1.81decrease	liver	(Asagba et al., 2010)
Electric catfish	Malapterurus electricus	Heavy metal	Benzaldehyde	1.46 increase	liver	(Isamah and Asagba. 2004)
Nile tilapia juveniles	Oreochromis niloticus	TAN (5 mg/L 70 days)	Phthalazine	2.1 increase	liver	(Hegazi et al., 2010)
Nile tilapia juveniles	Oreochromis niloticus	TAN (10 mg/L 70 days)	Phthalazine	2.5 increase	liver	(Hegazi et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	Benzaldehyde	2.23 decrease	kidney	(Asagba et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	Benzaldehyde	1.62 decrease	kidney	(Asagba et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	Benzaldehyde	3.25 decrease	gill	(Asagba et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	Benzaldehyde	1.88 decrease	gill	(Asagba et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	Benzaldehyde	1.15 decrease	muscle	(Asagba et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	Benzaldehyde	0.91 decrease	muscle	(Asagba et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 7 days)	Benzaldehyde	1.69 increased	brain	(Asagba et al., 2010)
African catfish	Clarias gariepinus	Cadmium (5 mg/L 21 days)	Benzaldehyde	1.27 decrease	brain	(Asagba et al., 2010)

Table 19: Hepatic and extrahepatic AOX activity as a biomarkers for pollution in fish.

Abbreviations: 7d for 7 days exposure, 21d for 21 day exposure, 7d and 21d (acclimatised for 14 days in large aquaria then kept in bowls of 20 litres for 7 days and 21 days). TAN – Total ammonia nitrogen.

Nile tilapia juveniles (Oreochromis niloticus) in response to oxidative stress caused by chronic ammonia exposure (Table 18 and 19), which reduces growth rate performance causes gill hyperplasia, and induces hyperexcitability, coma, convulsions and finally death. The activity of hepatic XO was increased 2.9-fold and 3.5-fold following exposure to 5mg/L and 10mg/L of ammonia respectively. The activity of XO in white muscle increased 2.25-fold and 2.6-fold following exposure to 5mg/L and 10mg/L of allowing exposure to 5mg/L a
muscle and found no change in to its activity. While Hari and Neeraja (2012) examined the activity of XO with ammonia in Common carp (Cyprinus carpio) kidney and found 2-fold decrease of its activity.

### **1.5.4.** Genetic and structural properties of molybdo-flavoenzymes in fish.

It was initially believed that the developmental model organism, zebra fish (Danio rerio) did not have MFE genes as computer searches of whole genomes failed to find them initially, while the genome in the fishes Fugu rubripes and also Poecilia reticulata had at the least the XOR gene (Garattini et al., 2003).Later it was found the MFEs genes were present in all fish genomes examined including zebra fish as seen in table 20 with (Kurosaki et al., 2013). Using genome sequencing data, Kurosaki et al., (2013) reconstructed the evolution of MFEs by predicting the structures of MFEs genes and pseudogenes. Their work demonstrated that bony fish are most primitive vertebrate with an AOX gene (AOX $\alpha$ ), coming from the duplication of an ancestral XDH. They also showed that during fish evolution duplication of AOX $\alpha$  into AOX $\beta$ and pseudogenization of the ancestral AOX $\alpha$  occurred (Figure 20, 21)(Kurosaki et al., 2013). An orthologs of the piscine AOX $\beta$  was observed to be preserved in amphibians and it is likely that this is the precursors of reptilian, avian and mammalianAOX1 found in the liver of most higher organisms including human where AOX1 it is the sole AOX enzyme, as in Table 7, 8, 9 and Figure 22. On the other hand amphibian AOX $\gamma$  is a duplication of AOX $\beta$  and the likely ancestor of avian and reptilian AOX2 that, in turn, gave rise to mammalian AOX3L1 (Figure 22). Subsequent gene duplications generated the two mammalian genes, AOX3 and AOX4 (Figure 22). The evolution of vertebrate AOX genes is thus dominated by duplication ,pseudogenization and deletion events (Kurosaki et al., 2013).Figure 22 shows how AOX and XDH genes evolved in different species of vertebrates. Table 20 summarises our current knowledge of AOX and XDH genes and pseudogenes in fish, amphibians, reptiles, birds and mammals (Kurosaki et al., 2013).



Figure 20: Summary of AOX and XDH genes in fish. Based on Kurosaki et al., 2013. The green colour indicates XDH genes, the yellow colour indicates AOX $\beta$  genes and red colour indicates AOX $\alpha$ genes. The numbers above the genes indicate the exons present and the crosses (X) indicate inactive pseudogenes that have exons missing.

		Fish	Comments
	Jawless fish Cartilaginous fish	Lamprey (Petromyzon marinus) Little skate( <i>Leucoraja erinacea</i> ) Australian ghost shark (Callorhinchus milii)	No AOX
			It is proposed ΑΟΧα evolved first, based on similarity of ΑΟΧα gene to XDH
$- \frac{XDH}{- AOX\beta} - \frac{AOX\beta}{- AOX\beta} - \frac{AOX\beta}{-$	Bony fish Cypriniformes Semionotiformes	Zebrafish ( <i>Danio rerio</i> ) Spotted gar ( <i>Lepisosteus oculatus</i> )	AOXβ evolved from AOXα
	Bony fish Salmoniformes Coelacanthiformes	Atlantic salmon ( <i>Salmo salar</i> ) Coelacanth ( <i>Latimeria chalumnae</i> )	AOXα pseudogene : inactive gene
- <mark>ΧDΗ</mark> - - ΑΟΧβ-	Tetraodontiformes Gasterosteiformes Cyprinodontiformes Beloniformes Perciformes	Fugu (Takifugu rubripes) Pufferfish (Tetraodon nigroviridis) Stickleback (Gasterosteus aculeatus) Platy (Xiphophorus maculatus) Medaka (Oryzias latipes) Nile tilapia (Oreochromis niloticus)	AOXα deleted from genome

**Figure 21:** Proposed evolution of AOX and XDH genes in fish. Based on Kurosaki et al., 2013. The green colour indicates XDH genes, the yellow colour indicates AOX $\beta$  genes and red colour indicates AOX $\alpha$  genes. The curved arrows indicate duplication events and the crosses (X) indicate inactive pseudogenes that have exons missing.



**Figure 22:** Summary of evolution of AOX and XDH genes in fishes, amphibians, reptiles, birds and mammals. Based on Kurosaki et al., 2013. The green colour indicates XDH genes, the yellow colour indicates AOX $\beta$  and AOX1 genes, red colour indicates AOX $\alpha$  genes, blue colour indicates (AOX2) and the purple colour indicates (AOX2 gene). The curved arrows indicate duplication events and the crosses (X) indicate inactive pseudogenes that have exons missing.[  $\rightarrow$  ] Indicates direction of transcription.

Organism	Class	Order	Family	Common name (Species name)	MFE genes present
Jawless fish	Hyperoartia	Petromyzontiformes	Petromyzontidae	(Betromyzon morinys)	XDH
(Agnauna) Cartilaginous			~	(Petromyzon marinus) Little skate	
fish (Chondrichthyes)	Chondrichthyes	Rajiformes	Rajidae	(Leucoraja erinacea)	XDH
		Chimaeriformes	Callorhinchidae	Australian ghost shark (Callorhinchus milii)	XDH
	Actinopterygii	Cypriniformes	Cyprinoidea	Zebrafish (Danio rerio)	ΑΟΧα, ΑΟΧβ,ΧDΗ
		Semionotiformes	Lepisosteidae	Spotted gar (Lepisosteus oculatus)	ΑΟΧα, ΑΟΧβ,ΧDΗ
Bony Fish		Tetraodontiformes	Tetraodontidae	Fugu (Takifugu rubripes)	ΑΟΧβ,ΧDΗ
(Osteichthyes)				Pufferfish(Tetraodon	ΑΟΧβ,ΧDΗ
		Gasterosteiformes	Gasterosteidae	Stickleback(Gasterosteus	ΑΟΧβ,ΧDΗ
		Cyprinodontiformes	Poeciliidae	Platy (Xiphophorus maculatus)	ΑΟΧβ,ΧDΗ
		Beloniformes	Adrianichthyidae	Medaka (Oryzias latipes)	ΑΟΧβ,ΧDΗ
		Perciformes	Cichlidae	Nile tilapia (Oreochromis niloticus)	ΑΟΧβ,ΧDΗ
		Salmoniformes	Salmonidae	Atlantic salmon (Salmo salar)	ΡΑΟΧα, ΑΟΧβ,ΧDΗ
	Sarcopterygii	Coelacanthiformes	Coelacanthidae	Coelacanth (Latimeria chalumnae)	ΡΑΟΧα, ΑΟΧβ,ΧDΗ
Amphibians	Amphibia	Anura	Pipidae	African frog (Xenopus tropicalis)	ΑΟΧβ,ΑΟΧγ,ΡΑΟΧμ,ΧDΗ
	Sauropsida	Testudines	Emydidae	Western painted turtle (Chrysemys picta bellii)	AOX1,AOX2,XDH
	Sauropsida	Testudines	Trionychidae	Chinese softshell turtle(Pelodiscus sinensis)	AOX1,AOX2,XDH
Reptiles	Sauropsida	Squamata	Iguanidae	Anole lizard (Anolis carolinensis)	AOXô,AOXɛ,AOX1,AOX2,XDH
	Sauropsida	Archosauria	Crocodylidae	American alligator (Alligator mississippiensis)	ΑΟΧδ,ΑΟΧε,ΑΟΧξ,ΑΟΧ1,ΑΟΧ2
	Sauropsida	Archosauria	Crocodylidae	Saltwater crocodile(Crocodylus porosus)	ΑΟΧδ,ΑΟΧε,ΑΟΧξΑΟΧ1,ΡΑΟΧ2
	Aves	Galliformes	Phasianidae	Chicken(Gallus gallus) Turkey(Meleagris gallopavo)	AOX1,AOX2,XDH
D:	Aves	Psittaciformes	Psittacidae	Budgerigar(Melopsittacus undulatus)	AOX1,AOX2,XDH
Biras	Aves	Passeriformes	Estrildidae	Zebra finch (Taeniopygia guttata)	AOX1,XDH
	Aves	Anseriformes	Anatidae	Mallard duck (Anas platyrhynchos)	AOX1,XDH
	Mammalia	Monotremata	Ornithorhynchidae	Platypus(Ornithorhynchus anatinus)	AOX1,AOX3,AOX3L1,AOX4,XDH1, PXDH2, PXDH 3, PXDH 4
	Mammalia	Didelphimorphia	Didelphinae	Opossum (Monodelphis domestica)	AOX1,AOX3,AOX3L1,AOX4,XDH
	Mammalia	Dasyuromorphia	Dasyuridae	Tasmanian devil(Sarcophilus harrisii)	AOX1,AOX3,AOX3L1,AOX4,XDH
	Mammalia	Diprotodontia	Macropodidae	Wallaby (Macropus eugenii)	AOX1,AOX3,AOX3L1,AOX4,XDH
	Mammalia	Afrosorcida	Tenrecidae	Lesser hedgehog tenrec(Echinops telfairi)	AOX1,AOX3,AOX3L1,XDH
	Mammalia	Proboscidea	Elephantidae	African elephant (Loxodonta Africana)	AOX1,AOX3,AOX3L1,AOX4,XDH
Mammals	Mammalia	Hyracoidea	Procaviidae	Rock hyrax (Procavia capensis)	AOX1,AOX3,AOX3L1,AOX4,XDH
	Mammalia	Cingulata	Dasypodidae	Armadillo (Dasypus novemcinctus)	PAOX1,AOX3,AOX3L1,AOX4,XDH
	Mammalia	Pilosa	Megalonychidae	Two toed sloth (Choloepus hoffmanni)	PAOX1, AOX3,AOX3L1,AOX4,XDH
	Mammalia	Chiroptera (Megachiroptera)	Pteropodidae	Flying fox (Pteropus vampyrus)	PAOX1, PAOX 3, PAOX 4,AOX3L1.XDH
	Mammalia	Chiroptera (Microchiroptera)	Vespertilionidae	Little brown bat (Myotis lucifigues)	PAOX3,AOX1,AOX3L1,AOX4,XDH
	Mammalia	Perissodactyla	Equidae	Horse(Equus caballus)	PAOX3,AOX1,AOX3L1,AOX4,XDH
	Mammalia	primates	Hominidae	Human (Homo sapiens)	PAOX3, PAOX311, AOX1, XDH

Table	20:	AOX	and	XOR	genes	in f	fish,	amphibians,	reptiles,	birds	and mammal	S

Based on (Kurosaki et al., 2013). Abbreviation. P=pseudogene.

## **1.6.** Purification of aldehyde oxidase and xanthine oxidoreductase enzymes.

MFEs have been purified from many different species with different methodologies as summarised in the (Tables 21, 22 and 23). Tables 21 and 22 summarise purification protocols for AOX from diverse number of organisms. Reviewing the literature on the subject reveals that the initial step of purification of AOX is generally subcellular fractionation to generate cytosol where the enzyme resides. A popular subsequent purification method used for AOX is heat precipitation. This method can be a useful step in the early stages of a purification scheme involving more heat stable proteins. It is generally achieved by incubating the protein mixture at known temperature for 10 minute, then cool in ice for 15 minute, then centrifugation at high centrifuge for 15 minute. These results in the denaturation of the unwanted less stable proteins in the mixture that are removed by precipitation. By this principle there can often by a 2-fold purification of a heat stable protein in the supernatant fraction (Bonner. 2007). The thermal stability of mammalian and Drosophila AOX has allowed this to be used successfully in 15 out of 18 purification protocols. Temperatures used for these purification protocols range from 50-68 °C with 50-55°C being the most common temperature used (Table 21 and 22). Interestingly heat treatment (50 - 70 °C) is also used in the purification of related MFEs, XOR (Table 23).

The next step commonly used in many AOX purification protocols is ammonium sulphate fractionation (Table 21 and 22). In this method the solubility of proteins is increased at low salt concentrations and is decreased at high salt concentrations. When a salt such as ammonium sulphate is added to a protein solution, the ions in the ammonium sulphate have a greater charge density than the proteins, they attract the

Species	Tissue	Enzyme	Purification steps	Molecular mass	Substrates	Fold purification a, b	Yield %	References
<b>Guinea pig</b> (Cavia porcellus)	Liver	AOX	Cytosol, 30-50% $(N_{H4})_2SO_4$ , DEAE- cellulose, FMN – sepharose 4B, Sephacryl S-300	Native PAGE 300 kDa SDS PAGE 148 kDa	2-hydroxypyrimidine, N <sup>1</sup> -methylnicotinamide,	120 <sup>a</sup>	26	(Yoshihara and Tatsumi. 1985)
Hamster (Hamster)	Liver	AOX	Cytosol, heat at 50 °C, 30-50% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , DEAE-cellulose, phenyl- Toyopearl, TSK-gel G3000SW <sub>XL</sub>	Native PAGE 288 kDa SDS PAGE 145 kDa	Benzaldehyde	89 <sup>a</sup>	12	(Sugihara et al., 1999)
Mouse (Mus musculus)	Liver	AOH1	Cytosol, heat at 55 °C, 50% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Benzamidine sepharose 6B, MonoQ	SDSPAGE 147 kDa Zymogram MALDI-TOF MS	Phthalazine	173 <sup>a</sup>	1.7	(Terao et al., 2001)
<b>Mouse</b> (Mus musculus)	Liver	AOH2	Cytosol, heat at 55 °C, 50% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Benzamidine sepharose 6B, MonoQ	SDS PAGE 150 kDa MALDI-TOF MS Zymogram	All-trans retinaldehyde	346 <sup>a</sup>	5	(Terao et al., 2009)
<b>Mouse</b> (Mus musculus )	Nasal epithelial mucosa	AOH3	Cytosol, heat at 55 °C, 50% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , MonoQ	SDS PAGE 150 kDa MALDI-TOF MS Zymogram western blot	Phthalazine, All-trans retinaldehyde, Benzaldehyde, 2-hydroxypyrimidine, Octanal	18 <sup>a</sup>	26	(Kurosaki et al., 2004)
Mouse (Mus musculus )	Liver	AOX	Cytosol, 30-50% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Benzamidine sepharose 6B, DEAE- cellulose	Native PAGE 265 kDa SDS PAGE 138 kDa	Benzaldehyde, DMAC, 2-hydroxypyrimidine	140 <sup>a</sup>	-	(Yoshihara and Tatsumi. 1997)
<b>Mouse</b> (Mus musculus)	Liver	AOX1	Cytosol, heat at 55 °C, 50% (NH <sub>4</sub> ) <sub>2</sub> SO4, Benzamidine sepharose 6B, MonoQ	SDS PAGE 150 kDa MALDI-TOF MS Zymogram, western blot	Phthalazine	627 <sup>a</sup>	3	(Vila et al., 2004)
Rat (Rattus norvegicus)	Liver	AOX	Crude extract, heat at 55 °C, 60% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , HTP column, Benzamidine sepharose 6B, Sephadex G-25 column	Native PAGE 285 kDa SDS PAGE 150 kDa	DMAC	945 <sup>b</sup>	13	(Maia and Mira. 2002)

Abbreviation: p-dimethylaminocinnamaldehyde - DMAC a = from cytosol, b= from crud extract (homogenate)

 Table 21: Summary of purification protocols for aldehyde oxidase from rodent species.

Species	Tissue	Enzyme	Purification steps	Molecular mass	Substrates	Fold purification a, b	Yield %	References
Bovine (Bos taurus)	Liver	AOX	Cytosol, heat at 60 °C, 40% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Fast-Q,Sephacryl, Benzamidine sepharose 6B	SDS PAGE 150 kDa	Phenanthridine	307 <sup>a</sup>	3	(Calzei et al., 1995)
<b>Chicken</b> (Gallus gallus)	Kidney	AOX1	Cytosol, heat at 55 $^{\circ}$ C, 40% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Benzamidine Sepharose 6B, MonoQ	SDS PAGE 150 kDa MALDI MS	All-trans retinaldehyde	-	-	(Terao et al., 2006)
Drosophila (Drosophila melanogaster)	Flies	AOX	Cytosol, heat at 68 °C, 55-75% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Sephadex G200, DEAE -Cellulose	Native PAGE 280 kDa	Benzaldehyde	481 <sup>a</sup>	18	(Andres. 1976)
Drosophila (Drosophila melanogaster)	Flies	AOX	Crude extract, CM-cellulose, 55-75% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , DEAE –Cellulose, hydroxylapatite		Benzaldehyde	361 <sup>b</sup>	16	(Dickinson. 1970)
Sea bream (Pagrus major)	Liver	AOX	Cytosol, heat at 50 °C, 30-50% ((NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , DEAE-cellulose		Benzaldehyde	33 <sup>a</sup>	5.6	(Kitamura et al., 2003)
<b>Sea bream</b> (Pagrus major)	Liver	AOX	Cytosol, heat at 45 °C, 30-45% $(NH_4)_2SO_4$ , DEAE-cellulose column chromatography		Benzaldehyde, 2-nitrofluorene 2-hydroxypyrimidine	-	-	(Ueda et al., 2002)
Pig (Sus scrofa)	Liver	AOX	Crude extract, heat at 55 $^{\circ}$ C, 60% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Acetone precipitation, Alumina Cý, DEAE- cellulose, preparative electrophoresis	Disc gel electrophoresis 270 kDa	N <sup>1</sup> -methylnicotinamide	115 <sup>b</sup>	6.8	(Felsted et al., 1973)
Rabbit (Oryctolagus cuniculus)	Liver	AOX	Crude extract, heat at 55 $^{\circ}$ C, 60% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Acetone precipitation, Calcium phosphate gel, DEAE-cellulose, preparative electrophoresis	Disc gel electrophoresis 260 kDa	N <sup>1</sup> -methylnicotinamide	118 <sup>b</sup>	14.4	(Felsted et al., 1973)
Rabbit (Oryctolagus cuniculus)	Liver	AOX	Crude extract, heat at 58 °C, 50% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , 50% Acetone precipitation, DEAE- Sephacel,HA-Ultrogel, Sephacryl S-300, MonoQ	Native PAGE 290 kDa SDS PAGE 144 kDa EPR spectroscopy	1-methylnicotinamide	-	-	(Turner et al., 1995)
Rabbit (Oryctolagus cuniculus)	Liver	AOX	Crude extract, heat at 55 °C, 50% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Benzamidine sepharose 6B, MonoQ	SDS PAGE 150 kDa MALDI-TOF MS, Zymogram, western blot	Phthalazine	400 <sup>b</sup>	64	(Stell et al., 1989)

Abbreviation: a = from cytosol, b= from crude extract (homogenate)

**Table 22:** Summary of purification protocols for aldehyde oxidase from non-rodent species.

Species	Tissue	Enzyme	Purification steps	Molecular mass	Substrates	Fold purification a, b	Yield %	References
<b>Chicken</b> (Gallus gallus)	Liver	XDH	Crude extract, heat at 70°C, 30% $(NH_4)_2SO_4$ , Sephadex G25, DEAE -Sephadex	Native PAGE 300 kDa	Xanthine + $NAD^+$	-	-	(Rajagopalan and Handler. 1967)
Drosophila (Drosophila melanogaster)	Flies	XDH	Crude extract, heat at 68°C, 30-50% $(NH_4)_2SO_4$ , Sephadex G200, DEAE - Cellulose	Native PAGE 300 kDa Isoxanthopterin + NAD <sup>+</sup>		481 <sup>b</sup>	18	(Andres. 1976)
Drosophila (Drosophila melanogaster)	Flies	XDH	Crude extract, 30-55% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Acetone 25-50%, DEAE –Cellulose, Sephacryl- HR200, Superose 12, MonoQ	SDS PAGE 150 kDa Pterin+NAD <sup>+</sup>		2840 <sup>b</sup>	21	(Hughes et al., 1994)
<b>Carp</b> (Cyprinus carpio)	Liver	XOR	Cytosol, 30-60% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Sephadex G-25, DEAE-Sephacel		Xanthine, Hypoxanthine + NAD <sup>+</sup>	-	-	(Kamiński and Jeżewska. 1985)
<b>Pig</b> (Sus scrofa)	Liver	XO	Crude extract, heat at 55 $^{\circ}$ C, 60% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Acetone precipitation, Alumina Cỷ, DEAE- cellulose		Xanthine	83 <sup>b</sup>	3	(Felsted et al., 1973)
Rabbit (Oryctolagus cuniculus)	Liver	XO	Crude extract, heat at $55C^{\circ}$ , $60\%$ (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Acetone precipitation.		Xanthine	98 <sup>b</sup>	19	(Felsted et al., 1973)
Rat (Rattus norvegicus)	Liver	XO	Crude extract, heat at $55^{\circ}$ C, $60\%$ (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , HTP, Q-Sepharose F.F.	Native PAGE 300 kDa SDS PAGE 150 kDa	Xanthine	1167 <sup>b</sup>	19	(Maia and Mira. 2002)
Rat (Rattus norvegicus)	Liver	XOR	Crude extract, heat at 65 $^{\circ}$ C, 50% (NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> , Benzamidine Sepharose 6B, Sephadex G-50	SDS PAGE 150 kDa	Xanthine $+ NAD^+$	199 <sup>b</sup>	5.7	(McManaman et al., 1996)
Rat (Rattus norvegicus)	Liver	XO	Crude extract, DEAE bath, 43% acetone precipitation, DEAE Sephacel pool, HA Ultrogel	SDS PAGE 137 kDa	Xanthine	781 <sup>b</sup>	19	(Engerson et al., 1987)
Rat (Rattus norvegicus)	Liver	XDH	Crude extract, DEAE bath, 43% acetone precipitation, DEAE Sephacel pool, HA Ultrogel	SDS PAGE 137 kDa	Xanthine+NAD <sup>+</sup>	1060 <sup>b</sup>	47	(Engerson et al., 1987)

Abbreviations: a = from cytosol, b= from crude extract (homogenate)

Table 23: Summary of xanthine oxidoreductase purification from different species

water molecules away from the surfaces of the proteins. This exposes hydrophobic patches on the proteins causing them to interact with each other forming inter-protein complexes. When a protein complex becomes too large it comes out of solution as a precipitate (Bonner. 2007). In all 18 protocols for AOX purification, ammonium sulphate fractionation has been used as an enrichment step prior to chromatography (Table 21 and 22). Sometimes a single ammonium sulphate fractionation is used but some protocols use two steps of ammonium sulphate fractionation to generate a narrower precipitation range (Table 21 and 22). Ammonium sulphate fractionation with 30-50 % saturation being the most common ammonium sulphate fractionation used.

With regards chromatography steps, four types of chromatography media have been used in the purification of AOXs namely ion exchange, gel filtration, hydrophobic interaction and affinity chromatography.

For AOX purification anion exchange media is the preferred ion exchange chromatography reagent used to separate negatively charged proteins. The molecular basis of this chromatography is that the resin in anion exchange columns contains cationic (positively charged) sites to bind proteins with a net negative charge. The bound proteins are then eluted by gradually increasing the salt concentration of the eluting buffer. A typical elution used is a gradient of 0 - 1M NaCL(Bonner. 2007). Anion exchange has been used in 10 out of 18 AOX purification protocols with DEAE-cellulose, 7 out of 18 AOX purification protocols with Mono-Q and 1 out of 18 AOX purification protocols with FAST-Q media being used. Anion exchange has been used in the purification of the related enzyme XOR (Table 23).

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Gel filtration that separates molecules on the basis of their differences in size and shape have been used in 4 out of 18 AOX purification protocols (Table 21 and 22). Affinity chromatography that separates proteins on the basis of a reversible interaction between a protein or group of proteins and a specific ligand have been used in 9 out of 18 AOX purification protocols (Table 21 and 22). 8 out of 18 AOX purification protocols were with benzamidine sepharose 6B media being used on the basis of benzamidine being a competitive inhibitor of AOX (Table 21 and 22) (Stell et al., 1989).

A literature survey (Table 21 and 22) reveals that to obtain a homogeneous preparation of AOX as assessed by SDS-PAGE or mass spectrometry generally the enrichment factor from liver is 89 - 2840 folds (Table 21 and 22). In order to achieve this yields range from 1.7 - 64 % (Table 21 and 22) liver AOX.

With regards substrates used to monitor the purification of AOX the most popular are 2-hydroxypyrimidine,  $N^1$ -methylnicotinamide, benzaldehyde, phthalazine, all-trans retinaldehyde, octanal, p-dimethylaminocinnamaldehyde (DMAC), phenanthridine and 2-nitrofluorene (Table 21 and 22). In these assays which are generally spectrophotometric some investigations do not directly measure substrate or product. These indirect assays use artificial electron acceptors such as dichloroindophenol (DCIP) and ferricyanide (Table 21 and 22).

With regards determining the native molecular mass of AOX, gel filtration has been used in 23 out of 28 protocols to determine this native molecular mass of purified enzyme preparations. These range from 212 - 300 kDa (Tables 21, 22 and 23).

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#### **1.7.** Aims of the project.

As indicated in the introduction the study of MFEs in fish has received much less attention than their mammalian counterparts. Of particular interest for aquatic organisms is a group of nitrogen containing polycyclic aromatic hydrocarbons (NPAHs) and their metabolites that are found as aquatic environmental pollutants (Blumer et al., 1977, Balch et al., 1995, Osborne et al., 1997, de Voogt and Laane. 2009) are AOX substrates that are cytotoxic, embryotoxic, teratogenic, genotoxic and carcinogenic (Southworth et al., 1979, Balch et al., 1995, Jung et al., 2001, Barron et al., 2004, Peddinghaus et al., 2012, Brinkmann et al., 2014). These include AOX substrates such as phenanthridine, quinoline, isoquinoline, hydroxyjsoquinolines, cinchonine, phthalazine, hydroxyphthalazines, quinazoline, hydroxyquinazolines and acridine (Rajagopalan and Handler. 1964b, Palmer et al., 1969, Krenitsky et al., 1972, McCormack et al., 1978, Stubley et al., 1979, Stubley and Stell. 1980, McMurtrey and Knight. 1984, Beedham et al., 1992, Robertson and Bland. 1993, Rashidi. 1996, Schofield et al., 2000, Beedham. 2001, Al-Tayib. 2009, Pryde et al., 2010, Barr and Jones. 2013, Mu et al., 2014).

Despite NPAHs being important aquatic pollutants to date only one of these, phthalazine, has been investigated as an AOX substrate in a fish species (Nile tilapia juveniles (Oreochromis niloticus)) (Hegazi et al., 2010).

The species focussed on in this research project and aims of this study are as follows: Rainbow trout (Oncorhynchus mykiss) was the focus of this study for several reasons. Rainbow trout is a member of the Salmonidae family, are widely distributed fish species cultured in regions with cold and cool water temperatures around the world (Harvey. 2004). They are important food and sport fish and the most cultivated cold freshwater fish in the US, Canada and European countries (Wolf and Rumsey. 1985). Due to its economic importance and ease of availability of tissue it is one of the most rigorously studied fishes in many research areas including carcinogenesis, toxicology, comparative immunology, disease ecology, physiology and nutrition (Behnke. 1992). With regards MFEs rainbow trout belongs to a class of fish Salmonidae, that have a single active AOX gene (AOX $\beta$ ) that is the ancestral AOX gene that gave rise to AOX1 the predominant hepatic AOX in rodents and the only AOX in humans(Kurosaki et al., 2013). Despite the aforementioned reasons for studying AOX in a Salmoniforme there has been no study to date (Tables 10, 11 and 12).

(a) The initial aim was to compare the biotransformation of endogenous and exogenous substrates by AOX and XOR in rainbow trout, rat and human.

(b) Up until now there has been no published work on any fish AOX and XOR with Km and Vmax measurements. An aim was therefore to determine the Km and Vmax for MFEs in rainbow trout and compare it with results in rat and human using spectrophotometric and HPLC assays.

(c) An additional aim of this project was to determine if environmental pollutants, drugs and estrogens might inhibit AOX in fish.

(d) As there has been no report of the purification of any piscine AOX to homogeneity, another goal was to purify the AOX enzyme from rainbow trout liver to examine its molecular characteristics.

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#### **2.0.** Materials and methods.

All reagents and chemicals were obtained from Fisher Scientific and Sigma / Aldrich Chemical Company Ltd, Poole, UK. Solvents and mobile phase reagents were obtained from different companies but were all for HPLC grade purity. The control of substances hazardous to health (COSHH) risk assessment for all reagents and chemicals were prepared before experiments were initiated.

Male Wistar rats (Rattus norvegicus) and male New Zealand white rabbit liver cytosol were obtained from the biomedical services unit, University of Dundee (Dr D.J.Clarke). Pooled human liver cytosol was obtained from Invitrogen Ltd. Immature rainbow trout (Oncorhynchus mykiss) liver was obtained from Pennine Trout Farm and Fishery, Calderbrook Road, Littleborough OL15 9HL, UK.

#### 2.1. In vitro analysis of molybdo-flavoenzyme activity.

#### **2.1.1. Preparation of cytosol.**

For the preparation of cytosol all steps were carried out at  $0 - 4^{\circ}$ C. Approximately 10 g of liver, kidney and gill were weighed out and homogenised in 40 ml of cold buffer (0.25 M sucrose, 10 mM Tris HCL pH 7.4) with a motor-driven tissue mortar fitted with a Teflon pestle (clearance 0.15-0.23mm, speed 10000 rpm and 5 minute duration) to produce a 25% w/v homogenate. The homogenate was then centrifuged at 4°C for 15 minutes at 10,000 xg, to pellet out the nuclear/mitochondrial fractions of the homogenate. The supernatant was then removed and centrifuged for a further 60 minutes at 4°C at 105,000 xg to obtain the cytosolic fraction. This fraction was then collected and separated into 0.5 ml aliquots and stored at -80°C.

#### 2.1.2. Gel filtration of cytosol.

Gel filtration of cytosol was carried out in order to remove endogenous small molecules, such as endogenous substrates and inhibitors, which may interfere with enzyme assays. A PD-10 gel filtration column was used (GE Healthcare Bio-Science, UK). Prior to use the column was allowed to drain before being equilibrated with 25 ml (5 column volumes) 50 mM Tris HCL pH 7.4. 2.5 ml of cytosol was then loaded onto the column and eluted with 3.5 ml of 50 mM Tris HCL pH 7.4, the protein containing fraction was then collected, pooled and aliquoted into 0.5 ml fractions in 1.5 ml polypropylene Eppendorf tubes to avoid repeated freeze/thawing and stored at - 80°C.

#### 2.2. Protein determination.

The amount of protein in each sample was calculated using a modification of the method described by Smith et al. using bovine serum albumin (BSA) as standard (Smith et al., 1985). The bicinchoninic acid (BCA) based assay is available as a kit from Sigma-Aldrich Co.

#### 2.2.1. Bicinchoninic acid reagents.

a) 1000 ml solution containing bicinchoninic acid (40 g), sodium carbonate (Na<sub>2</sub>CO<sub>3</sub>) (8 g), sodium tartrate (Na<sub>2</sub>C<sub>4</sub>H<sub>4</sub>O<sub>6</sub>)(16 g) and sodium bicarbonate (NaHCO<sub>3</sub>) (9.5 g) in 0.1N NaOH (final pH 11.25).Purchased from Sigma- Aldrich Co.

b) 25 ml solution containing 4% w/v copper sulphate pentahydrate (CuSO<sub>4</sub>.  $5H_2O$ ).For preparing of a set protein standards a 1mg/ml stock solution of bovine serum albumin (BSA) was prepared. This was then used to make the dilutions for the calibration curve as shown in (Table 24).

Final protein concentration	Volume BSA stock solution	Distilled water
(mg/mL)	(µl)	(µl)
0	0	100
0.2	20	80
0.4	40	60
0.6	60	40
0.8	80	20
1.0	100	0

Table 24. Preparation of protein calibration standards

#### 2.2.2. Bicinchoninic acid methodology.

a) The required amount of protein determination reagent was prepared by adding 1 part  $CuSO_4$ . 5H<sub>2</sub>O 4% w/v solution to 50 parts bicinchoninic acid solution.

b) Bovine serum albumin standards or 1 in 20 dilution gel-filtered cytosol (20  $\mu$ l cytosol + 380  $\mu$ l PBS) were prepared in duplicate.

c) 300  $\mu$ l of freshly protein determination reagent was added to 15  $\mu$ l of diluted cytosol or protein standard and vortexed.

d) All tubes were incubated at 37°C for 30 minutes.

e) The absorbance at 562 nm was then measured on a microplate reader and distilled water was used to zero the instrument.

#### **2.3.** Preparation of substrate stocks.

5 mM individual stock solution of AOX and XOR substrates and corresponding metabolites were prepared with appropriate solvent as indicated in (Table 25). After preparing the stock solutions, working standard solutions of 1 mM were prepared by dilution with water.

Substrate and product (5mM stock solution)	Comment	AOX and XOR	Solvent/treatment
Allopurinol	Hyperuricemia drug	AOX/XOR	10 mM NaOH /sonicating
Oxipurinol	Allopurinol metabolite	AOX/XOR	10 mM NaOH /sonicating
All-trans retinal	Vitamin A	AOX	Absolute ethanol
Retinoic acid	Vitamin A metabolite	AOX	Absolute ethanol
Cinchonine	Anti-malarial drug	AOX	dissolve in minimum volume of 0.1M HC then dH <sub>2</sub> O added
DMAC	Colouring dye	AOX	Absolute ethanol
6-mercaptopurine	Anti-cancer drug	AOX/XOR	10 mM NaOH /sonicating
6-thiouric acid	6-mercaptopurine metabolite	AOX/XOR	10 mM NaOH /sonicating
1-methylxanthine	Purine metabolism	XOR	10 mM NaOH /sonicating
1-methyl uric acid	1-methylxanthine metabolite	XOR	10 mM NaOH /sonicating
N <sup>1</sup> - methylnicotinamide	Endogenous vitamin metabolite	AOX	dH <sub>2</sub> O /sonicating
Phenanthridine	Environmental pollutant	AOX	Absolute ethanol
Phenanthridone	Phenanthridine metabolite	AOX	Absolute ethanol
Phthalazine	Environmental pollutants	AOX	Absolute ethanol
Phthalazinone	Phthalazine metabolite	AOX	Absolute ethanol
Pyrazinamide	Tuberculosis drug	AOX/XOR	10 mM NaOH /sonicating
Pyridoxal	Vitamin B6	AOX	dissolve in minimum volume of absolute ethanol then dH <sub>2</sub> O added
6-pyridoxic acid	Pyridoxal metabolite	AOX	dissolve in minimum volume of absolute ethanol then dH <sub>2</sub> O added
Vanillin	Food constituent	AOX	dH <sub>2</sub> O /heated to 70°C
Vanillic acid	Vanillin metabolite	AOX	dH <sub>2</sub> O /heated to 70°C
Xanthine	Purine metabolism	XOR	10 mM NaOH /sonicating
Uric acid	Xanthine metabolite	XOR	10 mM NaOH/Sonicating

**Table 25:** List of stock substrates, products and solvent used to prepare them.

Abbreviation: DMAC – dimethylaminocinnamaldehyde.

#### 2.4. Spectrophotometric determination of aldehyde oxidase activity.

AOX assays were performed with 40µl of gel filtered cytosol, 10µl of 1mM of substrate (dimethylaminocinnamaldehyde (DMAC), phenanthridine or vanillin) and 50µl Tris HCL pH 7.4 at 37°C. Spectrophotometric molybdo-flavoenzyme assays were conducted by using a microplate spectrophotometer (BioTek) at 37°C. All assays

were executed in triplicate in 100  $\mu$ l reaction volumes. All cytosol samples were frozen and thawed just once, and the spectrophotometric data were collected at 5 second intervals for 3 to 5 minutes utilizing Gen5<sup>TM</sup> software on the Windows XP PC connected to the microplate reader's spectrophotometer (BioTek), using the appropriate wavelength and molar absorption coefficient for each substrate / product as summarised in (Table 26).

**Table 26:** Wavelength and molar absorption coefficients used for the spectrophotometric measurement of aldehyde oxidase activity

Substrates	Principle of assay	Wavelength nm	Molar absorption, E (M <sup>-1</sup> cm <sup>-1</sup> )	Reference
DMAC	Disappearance of substrate	398	30500	(Kurth and Kubiciel. 1984)
Phenanthridine	Appearance of product	322	6400	(Johnson et al., 1984)
Vanillin	Disappearance of substrate	310	8854	(Panoutsopoulos and Beedham. 2004)

Abbreviation: DMAC – dimethylaminocinnamaldehyde.

# 2.5. Spectrophotometric determination of xanthine oxidoreductase activity.

The oxidation of xanthine to the uric acid and 1-methylxanthine to the 1-methyluric acid was assayed by monitoring the appearance of uric acid or 1-methyluric acid at 295 nm with the electron acceptors NAD<sup>+</sup> and O<sub>2</sub> (Waud and Rajagopalan. 1976). A summary of the assay conditions is given in (Table 27). The specific activity was calculated using the molar extinction coefficient for uric acid and 1-methyluric acid which is  $9,600 \text{ M}^{-1} \text{ cm}^{-1}$ .

хо	activity a	Issay	XO activity blank			
Reagent	Volume	Final concentration	Reagent	Volume	Final concentration	
Gel-filtered cytosol	60 µ1		Gel-filtered cytosol	60 µ1		
100 mM Tris HCL		20 mM Tris HCL	100 mM Tris			
pH 8.0	20 µ1	pH 8.0	HCL, pH 8.0	20 µl	30 mM	
1.5 mM (X or 1-MX) in 10 mM NaOH	10 µ1	0.15 mM (X or 1MX) 1mM NaOH	10 mM NaOH	10 µl	1 mM	
dH <sub>2</sub> O	10µ1		dH <sub>2</sub> O	10µ1		
XOF	<b>R</b> activity	assay	XOR activity blank			
Reagent	Volum e	Final concentration	Reagent	Volume	Final concentration	
Gel-filtered cytosol	60 µl		Gel-filtered cytosol	60 µl		
100 mM Tris HCL		20 mM Tris HCL	100 mM Tris			
pH 8.0	20 µl	pH 8.0	HCL, pH 8.0	20 µ1	20 mM	
$6.7 \text{ mM NAD}^+$	10 µ1	$0.67 \text{ mM NAD}^+$	$6.7 \text{ mM NAD}^+$	10 µl	0.67 mM	
1.5 mM (X or 1-MX) in 10 mM NaOH	10 µl	0.15 mM (X or 1MX) 1mM NaOH	10 mM NaOH	10 µ1	1 mM	

 Table 27: Summary of xanthine oxidoreductase assay setup.

Abbreviation; Xanthine(X) and 1-methylxanthine (1-MX)

The total XOR activity was calculated following subtraction of the XOR blank. This gave the total XOR activity with both NAD<sup>+</sup> and  $O_2$  as electron acceptors and was the total activity of the enzyme in the sample as both a dehydrogenase and oxidase. The XO activity was calculated following subtraction of the XO blank. This gave the activity with  $O_2$  as an electron acceptor. In order to calculate the XDH activity in the sample, the XO activity was subtracted from the XOR activity to give the activity with NAD<sup>+</sup> as electron acceptor only. The percentage XDH activity in the sample was calculated by dividing the XDH activity by the total XOR activity and multiplying by 100. Similarly the percentage XO activity was calculated by dividing the XO activity by 100.

#### 2.6. Determination of specific activity from spectrophotometric data.

The number of nmoles of substrate biotransformed per minute was calculated using the absorbance change observed per minute using the relationship :c = A/EI where c =concentration (mM), A = absorbance, I = path length of absorbing solution in cm, E = molar extinction coefficient. The light path for 100µl in a 96-well microplate was 0.681cm. Specific activity was then calculated as the number of nmoles of substrate converted per minutes per mg of protein. The protein concentration of the sample was measured as described in the section 2.2.

#### **2.7. Determination of optimum conditions of molybdo-flavoenzymes.**

As enzyme activity changes according to the reaction conditions changes in pH, temperature and substrate concentration were investigated.

#### 2.7.1. Determination of optimum pH of buffer.

The pH dependence of AOX and XDH were analyzed by assay of the enzymes at 37°C for 5 minutes at different pH values 5.6, 6.5, 7, 7.5 and 8.0 utilizing sodium phosphate buffers (50 mM final concentration).

#### 2.7.2. Temperature dependence of AOX and XOR activities.

This assay was performed to study the effects of incubation temperature on reaction rates of enzyme activity at different incubation temperature. The aliquots of filtered cytosol were assayed at 25, 30, 37, 40, 45, 55°C and 60°C in a microplate reader with a final DMAC concentration of 100 $\mu$ M as substrate of AOX and a final xanthine concentration of 100 $\mu$ M as substrate of XOR and 0.67mM NAD<sup>+</sup> as an electron acceptor. As summarized in (Tables 26 and 27).

#### 2.7.3. Heat stability study.

The aliquots of filtered cytosol were incubated at 55°C for 15, 30 and 60 minute in a thermocycler machine (PCR machine) with a heated lid. The heated lid of the PCR machine prevented evaporation and losses due to refluxing. The cytosols were removed from heating and cooled in ice for 5 min and cooled cytosol centrifuged at 13,000 xg for 5 min and any precipitate discarded. The assays were performed on the supernatant fraction using standard assay conditions for AOX activity with DMAC, as summarized in section 2.4.

## 2.8. Assay of molybdo-flavoenzyme activity using high pressure liquid chromatography.

All aldehydes and N- heterocyclic compounds mentioned before in section 2.3 (Table 25) were purchased from Sigma Aldrich and reagents required for the preparation of the HPLC mobile phases were all of HPLC grade purity from the same company.

The analysis of aldehyde and N-heterocyclic compounds and their corresponding metabolites were carried out by reverse phase HPLC. The system used comprised of a Beckman Coulter System Gold<sup>TM</sup> 127 Solvent HPLC Module (dual pump) and a programmable UV detector (module 166) or a programmable diode array detector (module 168) along with injector designed with a 20 µl sample loop and auto sampler. The analytical columns used were either: a Kromasil (25 cm X 4. 6 mm, 5µm; RP-C18), Hypersil ODS (25 cm X 4. 6 mm, 5µm; RP-C18), LiChrosphere® (25 cm X 4 mm, 5µm; RP-C18) or Spherisorb® (5µm CNRP 4.6 X 15 cm; RP-C18). A guard column with same stationery phase as the main chromatography column was used upstream in the analytical column. These columns were from Phenomenex, UK and Capital HPLC limited, UK. The mobile phases were filtered through a 0.45µm filter

before use. Isocratic and gradient methods were utilized to develop the optimal conditions for the separation of analytes and the most highly sensitive method for the determination of the different compounds.

5 mM individual stock solutions of AOX and XOR substrates and corresponding metabolites were prepared within appropriate solvent as described in (Table 25) in section 2.3. After preparing the stock solutions 1mM working standard solutions were prepared by dilution along with water. Quantification was done using five standards prepared from stock solutions with distilled water, 0.01, 0.02, 0.03, 0.05 and 0.1mM. 20µl of these compounds were injected onto the HPLC instrument starting with the lowest concentration to give a range of 0.2 to 2 nmol, as documented in figures 1- 5 in the Appendix 3. All stock's solutions were stored at  $4^{0}$ C for up to 3 months with the exception of vanillin and N<sup>1</sup>-methylnicotinamide, which were stored at  $-20^{0}$ C.

Calibration standards injected on the same day were used to calculate intra-day accuracy and precision. This is a measure of the distribution of individual measurements around the mean. This parameter was assessed by repeated analysis of the same solution and expressed as the relative standard deviation (RSD) otherwise known as the coefficient of variation (CV). % RSD for retention time was calculated by dividing the standard deviation by the mean of retention time of three replicates in the same day (Miller and Miller. 2005, Ghassan et al., 2010). The accuracy of the assay was calculated by comparison of the nominal analyte concentration to the actual concentration obtained from the linear regression line within the concentration investigated (0.01, 0.02, 0.03, 0.05, 0.1mM) (Accuracy = nominal concentration / actual concentration x 100). By using slope, intercept and the correlation coefficient (RSQ or  $r^2$ ) for the linearity of the calibration curve was determined for each analyte. The lower limit of quantification (LOQ) and limit of detection (LOD) were

determined for each analyte. According to international conference on harmonisation (ICH) guidelines, the LOD is defined as the lowest concentration of an analyte in a sample that can be detected but not quantified. The LOQ is the lowest amount of analyte in a sample that can be quantitatively determined with suitable precision and accuracy under the standard operational conditions of the method (ICH. 1994). The lower LOQ and LOD of samples was defined as the lowest added analyte concentration that can be measured with a low RSD and an accuracy of  $100 \pm 20\%$ that can be discriminated significantly from the basal concentration of analyte. The ICH has listed two options available to calculate both the LOD and the LOQ of an assay. One of these options are expressed as a concentration at a specified signal to noise ratio, usually 3:1 and 10:1 for the signal to noise ratio for LOD and LOQ respectively (ICH. 1994). The LOD and the lower of LOQ were calculated from the mean of the slope (S) and standard deviation of the intercept of three calibration curves using the linear regression method. The LOD and LOQ are calculated according to the following equations: The LOD = 3.3 (SD/S) and the LOQ = 10(SD/S) (ICH. 1994, Miller and Miller. 2005). All mobile phases and HPLC methods for assay of MFE activity for allopurinol, 6-mercaptopurine, 1-methylxanthine, xanthine, pyridoxal, all-trans retinal, vanillin, cinchonine, N<sup>1</sup>-methylnicotinamide, phenanthridine, phthalazine and pyrazinamide are summarized in (Tables 28 and 29).

Analytes	Chromatograph column	Mobile phases	Gradient/ Isocratic	Flow rate ml/min	Wavelength	Reference
Cinchonine	Hypersil ODS 5µm (25 cm x 4.6 mm, C18)	A:0.4% w/v CH <sub>3</sub> COONH <sub>4</sub> , pH 3.27 B:Acetonitrile 35% v/v.	Isocratic	1.0	248 nm UV detector	(Beedham et al., 1992)
Pyrazinamide	Kromasil 5µm (25cm x 4.6mm, C18) with guard column 5µm	A: 20 mM KH <sub>2</sub> PO <sub>4</sub> , pH 5.2 B:Acetonitrile 5% v/v	Isocratic	0.8	268 nm UV detector	(Kraemer et al., 1998)
Pyridoxal	Kromasil 5µm (25cm x 4.6mm, C18) with guard column 5µm	A: 60 mM Na <sub>2</sub> HPO <sub>4</sub> , pH 6.5 B: Methanol 25% v/v.	Isocratic	0.8	315 nm UV detector	Modified from(Talwar et al., 2003)
All-trans retinal	Kromasil 5µm (25cm x 4.6mm, C18) with guard column 5µm	A: 30 mM CH <sub>3</sub> COONH <sub>4</sub> , pH 4.5 B: Acetonitrile 70% v/v.	Isocratic	1.0	378 nm diode array detector	Modified from(Huang and Ichikawa. 1997)
Vanillin	Kromasil 5µm (25cm x 4.6mm, C18) with guard column 5µm	A: 0.22 M KH <sub>2</sub> PO <sub>4</sub> , pH 2.9 B: Acetonitrile 15% v/v.	Isocratic	1.5	285 nm UV detector	(Panoutsopoulos and Beedham. 2004)

 Table 28: Isocratic HPLC protocols for the measurement of MFE activity

Analytes	Chromatograph column	Mobile phases	Gradient/ Isocratic	Flow rate ml/min	Wavelength	Reference
Allopurinol and oxipurinol	Kromasil 5μm (25cm x 4.6mm, C18) with guard column 5μm	A: water H <sub>2</sub> O with addition of 0.1% v/v of glacial acetic acid. B: Acetonitrile.	Gradient see table 30	Flow rate was varied according to table 31	280 nm UV detector	This study
6-mercaptopurine and 6-thioxanthine	Kromasil 5μm (25cm x 4.6mm, C18) with guard column 5μm	A: water H <sub>2</sub> O with addition of 0.1% v/v of glacial acetic acid. B: Acetonitrile.	Gradient see table 31	Flow rate was varied according to table 32	280 nm UV detector	This study
N <sup>1</sup> - methylnicotinamide (2PY and 4PY)	Water Spherisorb® 5um (CNRP 4.6 x 150 mm; C18)	A: 0.1% v/v of formic acid B: Acetonitrile	Gradient see table 32 0.8		254 nm UV detector	(Szafarz et al., 2010)
N <sup>1</sup> - methylnicotinamide (2PY and 4PY)	Kromasil 5μm (25cm x 4.6mm, C18) with guard column 5μm	A: water H <sub>2</sub> O with addition of 0.1% v/v of glacial acetic acid. B: Acetonitrile	GradientFlow rate was variedsee table 30according to table 31		280 nm UV detector	This study
Phenanthridine and 6(5H)-phenanthridone	Kromasil 5μm (25cm x 4.6mm, C18) with guard column 5μm	A: water H <sub>2</sub> O B: Acetonitrile	Gradient see table 33	1.5	254 nm UV detector	Modified from(LaVoie et al., 1985)
Phthalazine and phthalazinone	Kromasil 5µm (25cm x 4.6mm, C18) with guard column 5µm	A: water H <sub>2</sub> O with addition of 0.1% v/v of glacial acetic acid. B: Acetonitrile.	Gradient see table 31	Flow rate was varied according to table 32	280 nm UV detector	This study
Xanthine and 1-methylxanthine	Kromasil 5μm (25cm x 4.6 mm, C18) with guard column 5μm	A: water H <sub>2</sub> O with addition of 0.1% v/v of glacial acetic acid. B: Acetonitrile	Gradient see table 31	Flow rate was varied according to table 32	280 nm UV detector	Modified from(Begas et al., 2007)
Pyrazinamide 5-hydroxypyrazinamide	Kromasil 5µm (25cm x 4.6mm, C18) with guard column 5µm	A: water H <sub>2</sub> O with addition of 0.1% v/v of glacial acetic acid. B: Acetonitrile	Gradient see table 31	Flow rate was time programmed with the variation in the solvent which was as follows see table 32	268 nm UV detector	This study

Abbreviations: (2-PY) -N<sup>1</sup>-methyl-2-pyridone-5-carboxamide, (4-PY)-N<sup>1</sup>-methyl-4-pyridone-5-carboxamide

**Table 29:** Gradient HPLC protocols for the measurement of MFE activity

Time	Solvent A (%)	Solvent B (%)	Flow (mL/min)
0	100	0	0.8
16	100	0	0.8
16	85	15	1.2
20	85	15	1.2
20	100	0	0.8
24	100	0	0.8

**Table 30:** HPLC program for the separation of allopurinol, N1-methylnicotinamide and their metabolites

Solvent A: water H<sub>2</sub>O with addition of 0.1% v/v of glacial acetic acid, Solvent B: Acetonitrile.

**Table 31:**HPLC program for the separation of 6-mercaptopurine, 1-methylxanthine, xanthine, phthalazine, pyrazinamide and their metabolites

Time	Solvent A (%)	Solvent B (%)	Flow (mL/min)
0	100	0	0.8
7	100	0	0.8
7	100	0	1.5
15	90	10	1.5
15	90	10	1
19	85	15	1
22.5	85	15	1
22.5	100	0	1.2
25	100	0	1.2

Solvent A: water H<sub>2</sub>O with addition of 0.1% v/v of glacial acetic acid, Solvent B: Acetonitrile.

**Table 32:**HPLC program for the separation of N<sup>1</sup>-methylnicotinamide and its metabolites on a polar cyan column

Time	Solvent A (%)	Solvent B (%)	Flow (mL/min)
0	55	45	0.8
1	61	39	0.8
2	67	33	0.8
3	72	28	0.8
4	78	22	0.8
5	84	16	0.8
6	90	10	0.8
9	90	10	0.8
9	55	45	0.8
15	55	45	0.8

Solvent A: 0.1% v/v of formic acid, Solvent B: Acetonitrile.

Table 33:HPLC	program for the se	eparation of	phenanthridine and 6	(5H) phenanthridone
	1 0	1		× / I

Time	Solvent A (%)	Solvent B (%)	Flow (mL/min)
0	95	5	1.5
1	90	10	1.5
6	50	50	1.5
12	0	100	1.5
15	0	100	1.5
15	90	10	1.5
20	90	10	1.5

Solvent A: water H<sub>2</sub>O, Solvent B: Acetonitrile.

### **2.8.1. HPLC analysis of the in vitro metabolism of molybdo-flavoenzyme substrates.**

HPLC assays of AOX and XOR were performed with a final volume of 100µl as following, 50µl of gel filtered cytosol, 10µl of 1mM stock of substrates (vanillin, pyridoxal, all-trans retinal, phenanthridine, phthalazine, N<sup>1</sup>-methylnicotinamide, methotrexate, cinchonine, xanthine, 1-methylxanthine and 6-mercaptopurine and 40µl (40 mM Tris HCL pH 7.4) at 37°C. XDH assays were similar with the exception that  $30\mu$ l of (40 mM Tris HCL pH 7.4) was used and 10µl of 6.7mM NAD<sup>+</sup> was added in addition to the substrate. Reactions were terminated with the addition of 60µl of acetonitrile and samples centrifuged at 12000 xg prior to chromatography of the supernatant on HPLC.

Validation that a compound was a substrate for AOX or XOR was achieved using enzyme specific inhibitors and electron acceptors.

#### 2.9. Determination of $K_m$ and $V_{max}$ for molybdo-flavoenzymes.

The rate of oxidation of varying concentrations of substrate of AOX and XOR were monitored such that linear reaction rates were observed and used for Km and Vmax calculations. The volume of cytosol added was varied to provide suitable reaction rates over the range of substrate concentrations used. The substrates concentrations used were 25, 40, 55, 70, 85 and 100  $\mu$ M for DMAC, vanillin, phenanthridine, xanthine, 1-methylxanthine, pyridoxal, all-trans retinal, allopurinol, phthalazine and 6mercaptopurine. Specific activity was then calculated as described in section 2.6. Eadie Hofstee plots V/[S] versus [V] were then plotted (Figure 23). The intercept on the ordinate axis is Vmax and the slope is -K<sub>m</sub>, the line of best fit through the points on plot was determined using linear regression by least squares method using Microsoft Excel (Microsoft Office).



Figure 23: Eadie- Hofstee plot.

# 2.10. Determination of the mode of inhibition and inhibitor constant (Ki)

The effects of the AOX and XOR inhibitors were studied as summarized in (Table 34) using DMAC and xanthine in spectrophotometric assay. The stock inhibitors were prepared in different solvents as follows. A stock solution of menadione was prepared by dissolving 1mg in 1ml of dimethyl sulphoxide (DMSO). Estradiol,  $17\alpha$ -ethinyl estradiol and phenanthridine stock solutions were prepared by dissolving 1mg in 1ml of ethanol.  $5\mu$ M,  $10\mu$ M and  $50\mu$ M inhibitor stock solutions were prepared from these by dilution in water and one tenth of these were added to assays to give final concentration 0.5, 1 and  $5\mu$ M inhibitor. The XOR inhibitor oxipurinol was prepared as a stock solutions were prepared from this by dilution in water and one tenth of these were added to assays to give final stock solutions were prepared from this by dilution in water and one tenth of these were added to assays to give final stock solutions were prepared from these were added to assays to give final stock solutions were prepared from this by dilution in water and one tenth of these were added to assays to give final stock solutions were prepared from this by dilution in water and one tenth of these were added to assays to give final concentration of 25 $\mu$ M, 50 $\mu$ M and 100 $\mu$ M oxipurinol.

AOX inhibitor assay					
Substrate	stock concentration	Inhibitor	Final concentration in assay	Mode of inhibition of	
	iii assay			enzyme	
		Menadione	0.5, 1and 5µM	Non-competitive	
		Estradiol	0.5, 1and 5µM	Uncompetitive	
DMAC 25, 40, 55 85 and 100	25, 40, 55, 70, 85 and 100µM	$17\alpha$ -ethinyl estradiol	0.5, 1and 5µM	Uncompetitive	
		Phenanthridine	0.5, 1and 5µM	Competitive	
		Chlorpromazine	0.5, 1and 5µM	competitive	
XOR inhibitor assay					
Substrate	stock	Inhibitor	Final concentration	Mode of inhibition	
	concentration in assay		in assay	of mammalian enzyme	
Xanthine $NAD^+$	25, 40, 55, 70, 85 and	Oxipurinol	25, 50 and 100µM	Non-competitive	
	100µM				
	$NAD^+ 0.67mM$				

Table 34: Summary of	AOX and XOR	inhibitors assays
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Abbreviation; DMAC - dimethylaminocinnamaldehyde.

Determination of the mode of inhibition and subsequently the inhibitor constant was achieved by use of the method described for the determination of Michaelis-Menten constants with the addition of inhibitor to each cuvette. The initial rate of oxidation at varying substrate concentrations was measured alone and in the presence of a fixed concentration of inhibitor. This procedure was then repeated in the presence of a second inhibitor concentration. The type of inhibition was then determined by examination of double reciprocal plots of both non-inhibited and inhibited data (Cook and Cleland. 2007) and the inhibitor constant (Ki) was calculated by using this equation for the exhibited competitive inhibition. Ki = Km [I]/ (Km app-Km) where: Km = uninhibited Km, Km app= apparent inhibited Km and [I] = inhibitor concentration and for exhibited uncompetitive and non-competitive inhibition using this equation. Ki = [Vmax app][I]/Vmax, where Vmax = uninhibited Vmax, Vmax app= apparent inhibited Vmax and [I] = inhibitor concentration. To determine Km and Vmax for these calculations Lineweaver-Burk plots1/V versus 1/[S] were plotted. The intercept on the ordinate axis is 1/Vmax and the slope is Km/Vmax. From these plots the mode of inhibition was also determined. (Figure 24) illustrates the different plots expected for competitive (Figure 24 A), uncompetitive (Figure 24 B) and non-competitive (Figure 24 C).



Figure 24: Lineweaver-Burk plots with different types of inhibitors.

#### 2.11. Purification of aldehyde oxidase

#### **2.11.1.** Chromatography instruments media and concentrators used

The instrument used was a GE Healthcare AKTA FPLC System with dual pump and fraction collector operated with Unicorn 4.11 Software. Chromatography media used was Q Sepharose 30/100, Superose 6 prep grade XK 16/70, MonoQ 5/50 GL, benzamidine Sepharose 6B and Mono P HR 5/20 were purchased from VWR International Ltd .UK. The centrifugal concentrators used were Amicon Ultra-15, Ultracel-PL Membrane, purchased from Millipore (UK) Ltd. All mobile phases were of FPLC grade purity and were filtered through a 0.45µm filter before use. Isocratic and gradient chromatography methods were using to develop the optimal conditions for the purification of enzymes from original sample.

#### **2.11.2.** Buffers used in purification protocols.

The following buffers were used in the course of the purification:

Buffer A: 20mM Tris-HCL, 1mM cysteine, 5% (v/v) glycerol. The pH of this buffer was varied depending on chromatographic conditions required.

Buffer B: 20mM Tris-HCL, 1mM cysteine, 5% (v/v) glycerol, 0.3-0.6M KCL. The pH of this buffer was varied depending on chromatographic conditions required.

Buffer C: 0.075M Tris HCL pH 9.3.

Buffer D: 10% (v/v) polybuffer 96, adjusted to pH 6.0 with glacial acetic acid.

Buffer E: 0.1mM glycine pH 9.0, 0.1M NaCL.

Buffer F: 0.1mM glycine pH 9.0, 0.1M NaCL, 20mM benzamidine.

Buffer G: 20mM Tris-HCL pH8.0, 1mM cysteine, 5% (v/v) glycerol, 0.1M NaCL.

All buffers were prepared from ultrapure reagents dissolved in Milli Q grade water and filtered through 0.45µM filters (Millipore, U.K.).

#### 2.11.3. Purification of AOX

Purification of enzyme activity was monitored throughout the purification procedure with DMAC as substrate. Protein elution was monitored at 280nm with an on-line spectrophotometric detection cell in the FPLC instrument. All procedure was carried out at  $0 - 4^{\circ}$ C. (Figure 25) summarises the protein purification protocols used.

#### 2.11.3.1. Preparation of cytosol.

Cytosol was prepared as previously described in section 2.1.1. with 20 to 70 g of fresh liver.

#### 2.11.3.2. Heat treatment step.

Cytosol was subjected to heat treatment in a water bath at 55°C for 15 minutes and after that cooled on ice for 10 minute then centrifuged at 10,000 xg for 15 minutes. The supernatant was then removed and used for subsequent experiments.

#### 2.11.3.3. Ammonium sulphate fractionation step.

Cytosol or heat treated supernatant derived from cytosol were subjected to ammonium sulphate fractionation as follows. Finely ground ammonium sulphate was added slowly to the sample with stirring to a concentration of 13.82g/100ml. This solution was stirred for a further 20 minutes and then centrifuged at 15,000 xg for 15 minutes to pellet protein precipitated by a 25% ammonium sulphate saturation. The supernatant was then subjected to a further fractionation so that the final ammonium sulphate concentration was 14.91g/100ml (60% saturation) and after stirring and centrifuging as in the first ammonium sulphate fractionation step the pellet was resuspended in approximately 5ml of buffer A.



Figure 25: AOX purification protocols.

The sample was then dialysed against three changes of 100 volumes of buffer A over a 24 hour period and the dialysate centrifuged at 15,000 xg for 15 minutes to remove any undissolved material. The sample was then subjected to chromatography for further purification.

#### **2.11.3.4.** Q-Sepharose anion exchange fast flow chromatography.

In this case the pH of buffer A and B was varied from pH 8.0 to pH 8.6 in different chromatographic runs. The dimensions of the Q-Sepharose column were 2.5 x 55cm and the flow rate was 10ml/minute. Following ammonium sulphate fractionation 10ml of the sample was applied to the Q-Sepharose column which had previously been equilibrated with buffer A. Unbound protein was eluted with 140ml of buffer A and protein retained on the column was eluted with 150ml of a gradient of 0.6M KCL in buffer B. The fractions containing AOX activity were pooled and concentrated to 4-5ml by ultrafiltration using Amicon Ultra-15 centrifugal concentrator. Following this anion exchange chromatography step, pooled fractions were further purified by affinity chromatography or chromatofocusing or size exclusion followed by a further anion chromatographic step on MonoQ, as outlined in (Figure 25).

#### 2.11.3.5. Affinity chromatography on benzamidine Sepharose 6B.

Benzamidine Sepharose 6B was used as an affinity matrix to bind AOX (Figure 25). In procedure 1 the AOX-active pooled fractions from the anion exchange columns Q-Sepharose were dialysed in 100 volumes of buffer E, with three changes over 24 hours and applied to a 15ml bed volume benzamidine Sepharose 6B column (2.5 x 10cm) at a flow rate was 0.5ml/minute. After application the flow was stopped for 30 minutes to facilitate binding of AOX to the benzamidine-ligands. Unbound material was then eluted at a flow rate of 1ml/minute by washing the column with 6 ml of buffer E. Bound AOX was eluted by sequential batch elution with 10ml of 5mM benzamidine,
10ml of 10mM benzamidine and 10ml of 20mM benzamidine in buffer F. The column was regenerated for further use by washing the column with 20 ml of 2M NaCL to remove any protein still bound and the re-equilibrated in buffer E.

#### 2.11.3.6. Chromatofocusing on Mono P HR 5/20.

The pooled fractions from the Q-sepharose anion exchange step were passed through a Sephadex G25 column (PD 10; 9.1ml bed volume) equilibrated with buffer C for buffer exchange prior to chromatofocusing. 2ml of protein of the resultant sample was then applied to a Mono P HR 5/20 column fitted to an AKTA FPLC system which had previously been equilibrated with buffer C. The column was then washed with two column volumes of buffer C at flow rate of 1ml/minute followed by 35ml of buffer D to generate a pH gradient from pH 9.3 to pH 6.0 over 30ml. Any material still bound to the column was then removed by a 2M NaCL elution.

#### 2.11.3.7. Gel filtration chromatography.

A Superose 6 16/70 column equilibrated with buffer G was used as a step to separate proteins on the basis of their differences in their native molecular mass (Figure 25). In procedure 4, 0.5ml of the AOX active concentrated fraction from Q-sepharose anion exchange chromatography was applied to the Superose column and eluted at a flow rate of 1ml/minute with 140ml of buffer G using 0.1M NaCL as isocratic elution. The column was then equilibrated with the buffer G for the next run.

#### 2.11.3.8. MonoQ chromatography.

This was used as a final step for the purification of AOX as it has proved a suitable medium for the final purification step of AOX in other species (Table 22). In thiscase the pH buffer A and B were varied from pH 8.0 to pH 8.6 in different chromatographic runs, the column dimensions were 5 x 50 mm and the flow rate was 1ml/minute. Following equilibration with 2 bed volumes of buffer A, 0.3ml of concentrated protein sample containing AOX was eluted by washing the column with 2ml of buffer A to remove unbound protein and bound protein was eluted with 20ml of a linear gradient of 0-0.3M KCL in buffer B. Active AOX fractions eluted by the salt were pooled and concentrated to 1-2ml by ultrafiltration using Amicon Ultra-15 membrane. The column was then regenerated for further use by washing the column with one bed volume of each of the following: 1M NaCL, 1.5M NaOH, and 20% (v/v) ethanol. The column was the re-equilibrated in buffer A.

#### 2.11.4. Sodium dodecyl sulphate polyacrylamide gel electrophoresis

Sodium dodecyl sulphate polyacrylamide gel electrophoresis (SDS-PAGE), a technique for separating of proteins according to their mass was used to monitor the purification of AOX(Laemmli. 1970). All SDS-PAGE gels and reagents were from Life Technologies Ltd, UK. The material for SDS-PAGE is described in the (Table 35) and the protocols used according to Life Technology (www.lifetechnologies.com).

Materials				
Gel used	NuPAGE® Novex® 3-8% Tris-Acetate Gel 1.5 mm			
<b>Running buffer</b>	NuPAGE® Tris-Acetate SDS			
Sample buffer	NuPAGE® LDS			
Ladder	HiMark <sup>™</sup> Unstained Protein Standard			
<b>Denaturing agent</b>	DTT			
Antioxidant agent	NuPAGE® Antioxidant (N,N-dimethylformarmide)			

 Table 35: SDS-polyacrylamide gel electrophoresis reagents (SDS- PAGE)

Abbreviations: LDS- Lithium dodecyl sulphate, DTT-Dithiothreitol, concentration DTT 5% (w/v), MOPS SDS- 3-(N-morpholino) propanesulfonic acid, concentration MOPS 0.6M, SDS 2% and pH 8.4±0.1.

### 2.11.4.1. Staining of SDS polyacrylamide gels

For the staining of SDS PAGE gels Coomassie blue and silver staining methods were used. All materials used purchased from Sigma Aldrich Company UK. The summary of the two methods are given in (Table 36).

Staining	Materials	Method		
	For 500ml of	1) Stain gels with gentle shaking for at least 30		
	Coomassie blue stain.	minutes with Coomassie blue stain heated to 60°C		
	<b>1</b> ) 1.25g Serva G blue.	or two hours at room temperature.		
	<b>2</b> ) 228ml methanols.			
Coomassie	<b>3</b> ) 46ml acetic acid.	2) Destain overnight with 10% v/v methanol, 15%		
blue	<b>4</b> ) 226ml distilled $H_2O$	v/v acetic acid, 75% v/v distilled $H_2O$ .		
	Stir for 15 minute.			
	5) Filter through			
	general grade whatman			
	filter paper.			
Reference	(Laemmli. 1970)			
Staining	Materials	Method		
	1) Oxidiser, 3.4mM	1) Soak gel 2 times for 10 minutes in 40%		
	potassium dichromate	methanol+10% acetic acid (200ml/gel).		
	with 0.0032N Nitric	2) Soak gel 2 times for 10 minutes with 10%		
	Acid.	ethanol+5% acetic acid(200ml/gel).		
	2) Silver nitrate	3) Soak gel for 10minutes in oxidiser (200ml/gel at		
	solution	RT).		
	0.012M silver nitrate.	4) Wash with distilled $H_2O$ three times for 10		
Silver		minutes or until gel colourless (200ml/gel at RT).		
	3) Developing agent	5) Soak for 30 minute in 0.012M silver nitrate		
	0.28M sodium	(200ml/gel) at RT.		
	carbonate+0.5ml/L	<b>6</b> ) Wash with distilled $H_2$ Ofor two minutes.		
	formalin.	7) Develop three times with developing reagent for		
		1, 5 and 3minutes respectively (200ml/gel at RT).		
		8) Stop with 5% v/v acetic acid for 5 minutes, then		
		immerses gel in deionised $H_2O$ .		
Reference	(De Moreno et al., 1985)			

Table 36: Summary of Coomassie Blue and Silver stains used for protein visualisation of SDS-PAGE.

Abbreviations: room temperature (RT)

# **2.11.4.2.** Determination of native molecular mass using Superose 6 chromatography.

The gel filtration marker kit for protein molecular masses 29,000-700,000 Da was obtained from Sigma Aldrich UK. A summary and concentration of the standards used in the calibration curve for the Superose 6 16/70 chromatography is shown in (Table 37). The buffer used for prepare the standard proteinswas20mM Tris HCl pH8.0, 1mM cysteine and 5% (v/v) glycerol with 0.1M NaCL. 1 ml of each prepared standard was injected onto the FPLC starting with the highest molecular mass standard Dextran blue 2,000,000 Da first followed by protein mixture containing all proteins listed in (Table 37). The method and sample preparation was according to (Whitaker. 1963, Andrews. 1964, Marshall. 1970)noting that the Blue Dextran was not mixed with other standards because many proteins bind to dextran. Blue dextran and the protein standards were applied in 1 ml volume and eluted with a 1 ml/minute flow rate.

Standard protein	Approximate Molecular	Recommended
Standard protein	mass	concentration
Carbonic anhydrase from bovine erythrocytes	29000 Da	3mg/1ml
Albumin, bovine serum	66000 Da	10mg/1ml
Alcohol dehydrogenase from yeast	150000 Da	5mg/1ml
β-Amylase from sweet potato	200000 Da	4mg/1ml
Apoferritin from horse spleen	443000 Da	10mg/1ml
Thyroglobulin, bovine	669000 Da	8mg/1ml
Blue Dextran	2000000 Da	2mg/1ml

Table 37: Summary of gel filtration standards used for native molecular mass determination

Absorbance was recorded at 280nm and a graph was drawn by plotting the logarithms of molecular mass of all standards versus to the elution volume of the standards. This graph was used to calculate the molecular mass of AOX.

### 2.12. Statistical analysis.

### 2.12.1. ANOVA-test

Each experiment was repeated at least three times. Statistical analysis was carried out using Microsoft Excel 2010 software. The result of each experiment was determined as the mean  $\pm$  standard deviation. Data were then analysed for significance using one way ANOVA. Differences with p values  $\leq 0.05$  were considered statistically significant.

#### **3. Results.**

The section of results is divided into 6 main sections as follows. Firstly sections 3.1, 3.2 and 3.3 describe the quantification of molybdo-flavoenzyme activity in rainbow trout, human and rat using spectrophotometric and HPLC assays. In sections 3.4 and 3.5 the determination of the kinetic parameters and comparison different inhibitors of AOX and XOR in different species are described. The final section 3.6 covers the purification of AOX from rat and trout.

#### 3.1. Quantification of molybdo-flavoenzyme activity

A number of spectrophotometric assays were used to measure the relative reaction rates of the AOX and XOR. Three spectrophotometric assays measured the rate of biotransformation of compounds vanillin two aldehyde and dimethylaminocinnamaldehyde (DMAC) and one N-heterocyclic AOX substrate phenanthridine with molecular oxygen as an electron acceptor. A further spectrophotometric XOR assay was performed using endogenous substrate xanthine and the caffeine metabolite, 1-methylxanthinewith  $O_2$  or with  $NAD^+$  plus  $O_2$  as electron acceptors. These assays were carried out with  $20 - 40 \mu l$  of gel filtered cytosol at 37°C as described in materials and methods section 2.4 to give linear reaction kinetics. As the majority of MFEs substrates and products cannot be quantified by spectrophotometry, HPLC assays were used to measure MFEs oxidation rates for 13 substrates. In order to optimise the experimental conditions for the assays a series of experiments were performed to assess the optimal pH and temperature. In addition prior to performing HPLC assays the methods were tested for accuracy and precision.

## **3.1.1.** Effect of pH on trout, rat and human molybdo-flavoenzymes activity.

In order to test the effect of pH on trout, human and rat AOX these were assayed under standard assay condition, except that the reaction pH was varied between 5.7 and 8.0 as described in section 2.7.1. The maximum activity of AOX was found to be at pH 7.0 with all three species as shown (Figure 26).



Figure 26: Effect of pH on of trout, human and rat AOX activity.

AOX activity was measured at 37°C. 0.1 mM DMAC was used as the substrate in AOX assay. Each point represents the mean  $\pm$  S.D. of three experiments with liver cytosol from ~3 animals.

In order to test the effect of pH on of trout, human and rat XO and XDH these were assayed under standard assay condition, except that the reaction pH was varied between 5.7 and 8.0 as described in section 2.7.1 and 2.8.1. The maximum activities of XO and XDH were found to be pH 7.0 with the three species tested as shown (Figure 27a and b).





Figure 27: Effect of the pH on of trout, human and rat XOR activity.

The (a) XO activity was measured at 37°C with 0.15 mM xanthine and  $O_2$  as electron acceptor. (b) XDH activity was measured at 37°C with 0.15 mM xanthine and 0.67 mM NAD<sup>+</sup> plus  $O_2$  as electron acceptor. Each point represents the mean  $\pm$  S.D. of three experiments with pooled cytosol from ~3 animals.

## **3.1.2.** Effect of temperature on trout, rat and human molybdo-flavoenzymes activity.

In the initial assays 37°C was used for assays but in order to determine if this was the optimum temperature to use a series of experiments were performed. In order to test effect of temperature on the activity of trout, rat and human AOX this was measured as at different incubation temperatures as described in section 2.7.2. The rate of reaction of trout AOX with DMAC increase gradually with increasing temperature from 25°C to 37°C then decreased gradually until it fell to < 10 % at 55°C and 60°C compared to the activity at 37°C as shown in (Figure 28). The activity at 60°C was ~ 10% obtained at 37°C. The rate of reaction of human and rat AOX with DMAC increase gradually as shown in (Figure 28) and the activity of human and rat at 60°C was ~ 60% obtained at 37°C respectively. As the maximum reaction rate was obtained at 37°C for AOX enzymes, this temperature was therefore selected for all assays.



Figure 28: Effect of the assay incubation temperature on of trout, human and rat AOX activity.

Enzyme activity was measured with 0.1 mM DMAC was used as the substrate in AOX assays. Each point represents the mean  $\pm$  S.D. of three experiments with pooled cytosol from 3 animals.

In order to test the effect of temperature on the activity of trout, human and rat XOR this was measured at different incubation temperatures. The rate of reaction of human and rat XO with xanthine and  $O_2$  as an electron acceptor. Gradually increasing with increase temperature from 25°C to 37°C then decrease gradually as shown in figure 29a and the activity of human and rat at 60°C was ~ 60% obtained at 37°C respectively. The rate of reaction of trout XOR with xanthine and NAD<sup>+</sup> with  $O_2$  as electron acceptors. Increasing gradually with increasing temperature from 25°C to 37°C then decreased gradually until fell to near zero at 55°C and 60°C as shown in (Figure 29b). The activity at 60°C was ~ 5% of that obtained at 37°C. The rate of reaction of human and rat XDH with xanthine and NAD<sup>+</sup> increase gradually with increasing temperature from 25°C to 37°C then gradually decrease as shown in (Figure 29b) the activity of human and rat at 60°C was ~ 60% obtained at 37°C respectively. As the maximum reaction rate was obtained at 37°C for XOR enzyme, this temperature was therefore selected for all assays.



Figure 29: Effect of the assay incubation temperature on of trout, human and rat XOR activity.

Enzyme activity was measured with (a) 0.15 mM xanthine and  $O_2$  as an electron acceptor. (b) 0.15 mM xanthine and 0.67 mM NAD<sup>+</sup> plus  $O_2$  as an electron acceptors. Each point represents the mean  $\pm$  S.D. of three experiments with pooled cytosol from ~3 animals.

#### **3.1.3.** Validation of the accuracy and precision of HPLC assays

The precision of the various HPLC assays was determined by injecting five concentrations (0.01, 0.02, 0.03, 0.05 and 0.1 mM) of all AOX and XOR substrates and the products listed in materials and methods section 2.8., three times on the same day to determine the intra-day variation. The intra-day variation assessed as the coefficient of variation in peak area ratio. The relative standard deviation (%RSD) calculation for retention time was described in section 2.8. The lower the value is the better the assay performance. At lower level it was below 3.5% for 3 replicates, see Appendix3 (Table 1). The accuracy of the assay was calculated by comparison of the nominal analyte concentration to the actual concentration obtained from the linear regression line within the concentration range investigated (0.01 mM to 0.1 mM), see Appendix3(Table 2). Appendix3 (Table 2) showed that the intra-day values ranged between 95.1-100.8%. The precision of all concentrations of analyte was approximately less than  $\pm$  5%. The calibration graphs obtained with these HPLC methods were linear over the concentration range used and the  $(r^2)$  values were 0.92 to 0.99 for each analyte when plotted for the mean peak area for three injections against the actual concentration (Figures 1- 5 in Appendix 3). This demonstrates that these HPLC methods are precise and repeatable and can be successfully used to analyse MFE catalysed biotransformations in this study. In addition to accuracy and precision tests, the limit of detection (LOD) and the limit of quantification (LOQ) were calculated for each of the analytes in section 2.5.2.3. The LOD of all substrates and products were ranged between 0.1-5.4 nmol and LOQ was ranged from 0.9 to 94 nmol. From these results, it can be concluded that the HPLC methods were sufficiently sensitive to detect low concentrations of substrates and products of molybdo-flavoenzyme catalysed reactions enzymes. The results of linearity, limit of detection and limit of quantification are presented in Appendix3 (Table 3).

#### 3.2. Screening of molybdo-flavoenzyme activities.

Following optimisation and validation of the assays, AOX and XOR activities in rainbow trout, Wistar rat and human cytosols were assessed using a multitude of spectrophotometric and HPLC assays prior to scrutinising the Km and Vmax for different substrates.

## **3.2.1.** AOX activity with exogenous aldehydes in trout tissue and mammalian liver cytosol.

The substrates used to evaluate AOX activity in this research included the exogenous food flavouring agent vanillin (4-hydroxy-3-methoxybenzaldehyde) and artificial dye (DMAC). dimethylaminocinnamaldehyde (Figures 30 and 31) illustrate spectrophotometric assays with the two aldehyde substrates vanillin and DMAC. This demonstrated that liver cytosols from all three species had measurable AOX activity towards these two aromatic aldehydes. In contrast rainbow trout kidney and gill cytosols had no measurable activity towards these substrates (data not shown). In order to check unequivocally if there was no AOX activity towards vanillin in kidney and gill cytosols an HPLC assay was used. This showed that even following a 120 minute incubation with this substrate there was no vanillic acid formed in contrast to liver cytosol (Figure 32). (Figure 33) shows an HPLC chromatogram for all liver cytosol from trout, rat and human with vanillin.



**Figure 30:** Spectrophotometric assays of aldehyde oxidase activities using rainbow trout, Wistar rat and human liver cytosol at 37°C with AOX substrate DMAC. For assays using 50  $\mu$ L rainbow trout [red], 30  $\mu$ L Wistar rat [green] and 20  $\mu$ L human [black] liver cytosol. 100  $\mu$ M final substrate concentrations were used. For analysis conditions see section 2.4.



**Figure 31:** Spectrophotometric assays of aldehyde oxidase activities using rainbow trout, Wistar rat and human liver cytosol at  $37^{\circ}$ C with AOX substrate vanillin. For assays using rainbow trout [red], Wistar rat [green] and human [black] liver cytosol. For all assays 100  $\mu$ M final substrate concentrations were used. For analysis conditions see section 2.4.



Time (minutes)

**Figure 32:** HPLC analysis of the in vitro biotransformation of vanillin by rainbow trout cytosol. (a) Liver, (b) kidney and (c) gill liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.22 M KH2PO4: acetonitrile system (85: 15) as mobile phase. The wavelength of the detection was 285 nm.



Figure 33: HPLC analysis of the in vitro biotransformation of vanillin by liver cytosol.

HPLC chromatogram of the in vitro oxidation of vanillin to vanillic acid by (a) rainbow trout, (b) Wistar rat and (c) human liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.22 M KH<sub>2</sub>PO<sub>4</sub>: acetonitrile system (85: 15) as mobile phase. The wavelength of the detection was 285 nm.

### **3.2.2. HPLC analysis of AOX activity with endogenous vitamins and vitamin metabolites.**

When the endogenous vitamin B6 (pyridoxal) was incubated with rainbow trout, Wistar rat and human liver cytosol its metabolite pyridoxic acid was generated with cytosols from all three species (Figure 34).



**Figure 34:** HPLC analysis of the in vitro biotransformation of endogenous compound pyridoxal by liver cytosol from different species. HPLC chromatogram of the in vitro oxidation of pyridoxal to pyridoxic acid by (a) rainbow Trout, (b) Wistar rat, (c) human liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) using methanol: 60 mM Na<sub>2</sub>HPO<sub>4</sub> (25: 75) as mobile phase. The wavelength of the detection was 315 nm.

When all-trans retinal (vitamin A) was incubated with rainbow trout, Wistar rat and human liver cytosol, the AOX metabolite retinoic acid was generated with cytosols from all three species (Figure 35).



**Figure 35:** HPLC analysis of the in vitro biotransformation of endogenous compound alltrans retinal by liver cytosol from different species. HPLC chromatogram of the in vitro oxidation of all-trans retinal to retinoic acid by (a) rainbow Trout, (b) Wistar rat, (c) human liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 30 mM CH<sub>3</sub>COONH<sub>4</sub>: acetonitrile (30: 70) as mobile phase. The wavelength of the detection was 378 nm.

When liver cytosol from rainbow trout, Wistar rat and human were incubated with  $N^{1}$ methylnicotinamide (NMN)two products (Figure 36) were observed with retention times of (9 minute)  $N^{1}$ -methyl-2-pyridone-5-carboxamide (2-PY) and (12 minute)  $N^{1}$ methyl-4-pyridone-5-carboxamide (4-PY) (Szafarz et al., 2010).



**Figure 36:** HPLC analysis of the in vitro biotransformation of N<sup>1</sup>-methylnicotinamide by liver cytosol from different species. HPLC chromatogram of the in vitro oxidation of N<sup>1</sup>-methylnicotinamide to (2-PY and 4-PY) by (a) rainbow Trout, (b) Wistar rat, (c) human liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Spherisorb; CNRP 5  $\mu$ m) and eluted with a 0.8 % v/v formic acid: acetonitrile gradient as described in section 2.5.3 as mobile phase. The wavelength of the detection was 254 nm.

### **3.2.3.** Spectrophotometric and HPLC analysis of AOX activity with exogenous N-heterocyclic compounds.

#### 3.2.3.1. Phenanthridine.

When the prototypical N-heterocyclic AOX substrate phenanthridine was used as a substrate in a spectrophotometric assay (Beedham. 2001) (Figure 37) only Wistar rat and human liver cytosol had activity. Cytosols prepared from trout liver, kidney and gill had no detectable activity (Figure 37). As there is evidence that at least one fish, the carp, generates phenanthridone from phenanthridine in vivo, an HPLC assay was used to determine if any activity could be detected. This demonstrated that rainbow trout liver cytosol AOX was able to generate measurable phenanthridone (Figure 39). In contrast there was no detectable phenanthridone formed with extended incubation with rainbow trout kidney or gill cytosol (Figure 38). Rat and human liver cytosol were also able to carry out this biotransformation (Figure 39).



**Figure 37:** Spectrophotometric assays of aldehyde oxidase activities using rainbow trout, Wistar rat and human liver cytosol at  $37^{\circ}$ C with AOX substrate phenanthridine. For assays using rainbow trout [red], Wistar rat [green] and human [black] liver cytosol. For all assays 100  $\mu$ M final substrate concentrations were used. For analysis conditions see section 2.4



**Figure 38:** HPLC analysis of the in vitro biotransformation of phenanthridine by trout cytosol. (a) Liver, (b) kidney and (c) gill trout cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with water: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 254 nm.



**Figure 39:** HPLC analysis of the in vitro biotransformation of phenanthridine by rainbow trout, Wistar rat, and human liver cytosol. (a) rainbow trout, (b) Wistar rat, (c) human liver cytosol. HPLC chromatogram of the in vitro oxidation of phenanthridine to phenanthridone. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with water: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 254 nm.

### 3.2.3.2. Phthalazine.

When the N-heterocyclic phthalazine was incubated with rainbow trout, Wistar rat and human liver cytosol, all were found to be capable of biotransforming it to the AOX product phthalazinone (Figure 40).



Time (minutes)

**Figure 40:** HPLC analysis of the in vitro biotransformation of phthalazine by rainbow trout, Wistar rat and human liver cytosol. (a) rainbow trout, (b) Wistar rat, (c) human liver cytosol. HPLC chromatogram of the in vitro oxidation of phthalazine to phthalazinone Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.

#### 3.2.3.3. Cinchonine.

When anti-malaria drug N-heterocycle cinchonine was incubated with rainbow trout, Wistar rat and human liver cytosol, a metabolite of cinchonine was detected by HPLC with all samples (Figure 41) that had a retention time (12 minute) indicative of  $2^{2}$ cinchoninone (Beedham et al., 1992).



#### Time (minutes)

**Figure 41:** HPLC analysis of the in vitro biotransformation of cinchonine to 2-cinchoninone by rainbow trout, Wistar rat and human liver cytosol. (a) rainbow trout, (b) Wistar rat, (c) human liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.4% w/v CH<sub>3</sub>COONH<sub>4</sub>: acetonitrile (65: 35) as mobile phase. The wavelength of the detection was 248 nm.

#### 3.2.3.4. Pyrazinamide.

In mammals the tuberculosis drug pyrazinamide is biotransformed by AOX and XOR. Pyrazinamide was incubated with rainbow trout, Wistar rat and human liver cytosol with  $O_2$  or with NAD<sup>+</sup> as electron acceptor and HPLC analysis were carried out. After incubation with rainbow trout liver cytosol with  $O_2$  or with NAD<sup>+</sup> (Figure 42). The incubation of rainbow trout liver cytosol with pyrazinamide in the presence of  $O_2$  as an electron acceptor for 240 minute incubation time did not show the production of any metabolite peak as seen in (Figure 42a). In contrast, the incubation of trout liver cytosol with pyrazinamide with NAD<sup>+</sup> as an electron acceptor generated 5hydroxypyrazinamide (Figure 42b). In contrast Wistar rat and human liver cytosol displayed activity with both  $O_2$  and NAD<sup>+</sup> as electron acceptor (Figure 43 and 44), indicating that in these species pyrazinamide was a substrate for both oxidase and dehydrogenase enzymes.



**Figure 42:** HPLC analysis of the in vitro biotransformation of pyrazinamide by rainbow trout liver cytosol at 37°C. Incubation (a) with  $O_2(b)$  with NAD<sup>+</sup> as an electron acceptor. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 20 mM KH<sub>2</sub>PO<sub>4</sub>: acetonitrile (95: 5) as mobile phase. The wavelength of the detection was 268 nm.



**Figure 43:** HPLC analysis of the in vitro biotransformation of pyrazinamide by Wistar rat and human liver cytosol at 37°C. Incubation with  $O_2$  as an electron acceptor. (a) Wistar rat, (b) human liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 20 mM KH<sub>2</sub>PO<sub>4</sub>: acetonitrile (95: 5) as mobile phase. The wavelength of the detection was 268 nm.



**Figure 44:** HPLC analysis of the in vitro biotransformation of pyrazinamide by Wistar rat and human liver cytosol at 37°C.Incubation with NAD <sup>+</sup>as an electron acceptor. (a) Wistar rat, (b) human liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 20 mM KH<sub>2</sub>PO<sub>4</sub>: acetonitrile (95: 5) as mobile phase. The wavelength of the detection was 268 nm.

The rainbow trout AOX activity were tested in liver, kidney and gill cytosol and it was found exclusively in the liver using HPLC and spectrophotometric assays. Hepatic aldehyde oxidase in cytosols of rainbow trout, human and rat (Table 38) were able to catalyse the oxidation of a range of MFE substrates to varying degrees. These included the exogenous aldehydes (DMAC and vanillin), endogenous vitamin (all-trans retinal, pyridoxal and N<sup>1</sup>-methylnicotinamide), substrates NPAH environmental pollutants (phenanthridine, phthalazine and cinchonine), and drug (pyrazinamide). The activity of rainbow trout with DMAC as AOX substrate in spectrophotometric assay were found lower compared to that other species (Table 38). The activity of Wistar rat with vanillin as AOX substrate was greater compared than that with human and trout respectively. The activity of human with endogenous vitamin substrates (all-trans retinal, pyridoxal and N<sup>1</sup>-methylnicotinamide) was greater than that with Wistar rat and trout respectively (Table 38). The lowest activity found of NPAH environmental pollutants (phenanthridine, phthalazine and cinchonine) was in trout compared with that of Wistar rat and human and the highest activity found was with human for phthalazine as AOX substrate and the highest activity with Wistar rat was for phenanthridine and cinchonine (Table 38). The highest activity found for pyrazinamide as substrate was with human and the lowest activity was found with trout (Table 38).

Substrate	Species	Activity
	Rainbow trout	+++
DMAC	Wistar rat	++++
	Human	+++++
	Rainbow trout	++
Vanillin	Wistar rat	+++++
	Human	+++
	Rainbow trout	++
Pyridoxal	Wistar rat	++++
	Human	+++++
All-trans retinal	Rainbow trout	++
	Wistar rat	++++
	Human	+++++
	Rainbow trout	+
N <sup>1</sup> -methylnicotinamide	Wistar rat	++++
	Human	+++++
	Rainbow trout	+
Phenanthridine	Wistar rat	+++++
	Human	++++
	Rainbow trout	++
Phthalazine	Wistar rat	++++
	Human	+++++
	Rainbow trout	++
Cinchonine	Wistar rat	+++++
	Human	++++
	Rainbow trout	+
Pyrazinamide	Wistar rat	+++++
	Human	++++

**Table 38:** Summary of in vitro studies of hepatic aldehyde oxidase activity in different species using HPLC and spectrophotometric assays with different substrates

(+++++ = ultra-high activity,+++ = high activity, +++ = intermediate activity, ++ = low activity, + = very low activity).

### **3.3.** Spectrophotometric and HPLC analysis of molybdo-flavoenzyme activity with endogenous purine and purine analogue substrates.

#### **3.3.1.** Xanthine and 1-methylxanthine.

For XOR activity two substrates were assessed using spectrophotometry. These were the endogenous purine catabolite xanthine and the major caffeine metabolite 1-methylxanthine. (Figure 45) illustrates spectrophotometric assays with the xanthine demonstrating that liver cytosols from Wistar rat and human had measurable XOR activity with and without NAD<sup>+</sup>. In contrast rainbow trout liver cytosol only had measurable XOR activity in presence of NAD<sup>+</sup> (Figure 45b). Similar spectrophotometric results were obtained with 1-methylxanthine (data not shown). Rainbow trout kidney and gill cytosols had no measurable activity towards either of these substrates.

In order to check unequivocally if there was no XDH activity towards xanthine and 1methylxanthinein kidney and gill cytosols a sensitive HPLC assay was used, this showed that even following a 120 minute incubation with these substrates there was no uric acid metabolite formed in contrast to liver cytosol (Figures 46 and 47).Reflecting the results from the spectrophotometric assay even after an extended 2 hour incubation no uric acid metabolite was formed with  $O_2$  as an electron acceptor. When xanthine and 1-methylxanthine were incubated with rainbow trout liver cytosol (Figures 48 and 49) there were appreciable levels of uric acid produced with NAD<sup>+</sup> as electron acceptor. This was in contrast to that found when rat and human liver cytosol was incubated with these substrates where both electron acceptors were utilised resulting in uric acid metabolite being formed with both the endogenous and exogenous purine (Figures 50, 51, 52 and 53).



**Figure 45:** Spectrophotometric assays of xanthine oxidoreductase activities using rainbow trout, Wistar rat and human liver cytosol at  $37^{\circ}$ C with XOR substrates. (a) xanthine oxidase activity assay. (b) xanthine oxidoreductase activity assay. Rainbow trout [red], Wistar rat [green] and human [black] liver cytosol. For all assays 100  $\mu$ M final xanthine concentrations were used. For the XOR assay 670  $\mu$ M NAD+ was used as the electron acceptor. For analysis conditions see section 2.5.



**Figure 46:** HPLC analysis of the in vitro biotransformation of xanthine + NAD<sup>+</sup> by rainbow trout cytosol. The chromatogram(a) Liver, (b) Kidney and (c) Gill trout cytosol with xanthine and NAD<sup>+</sup>. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times .Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 47:** HPLC analysis of the in vitro biotransformation of 1-methylxanthine + NAD<sup>+</sup> by rainbow trout cytosol. The chromatogram (a) Liver, (b) Kidney and (c) Gill trout cytosol with xanthine and NAD<sup>+</sup>. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times .Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.

**(a)** 



**Figure 48:** HPLC analysis of the in vitro biotransformation of xanthine by rainbow trout liver cytosol. The chromatogram (a) HPLC analytes generated by rainbow trout liver cytosol with xanthine in presence of  $O_2$  as an electron acceptor. The chromatogram (b) HPLC analytes generated by rainbow trout liver cytosol with xanthine and NAD<sup>+</sup>. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times .Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 49:** HPLC analysis of the in vitro biotransformation of 1-methylxanthine by rainbow trout liver cytosol at 37 °C. The chromatogram (a) HPLC analytes generated by rainbow trout liver cytosol with 1-methylxanthine in the presence of  $O_2$  as an electron acceptor. The chromatogram (b) HPLC analytes generated by trout liver cytosol with 1-methylxanthine and NAD<sup>+</sup> as an electron acceptor Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.


**Figure 50:** HPLC analysis of the in vitro biotransformation of xanthine by Wistar rat liver cytosol. The chromatogram (a) HPLC analytes generated by rat liver cytosol with xanthine and  $O_2$  as electron acceptor. The chromatogram (b) HPLC analytes generated by Wistar rat liver cytosol with xanthine and NAD<sup>+</sup>. The amount of NAD<sup>+</sup> added to incubation 675µM. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 51:** HPLC analysis of the in vitro biotransformation of 1-methylxanthine by Wistar rat liver cytosol at  $37^{0}$ C. The chromatogram (a) HPLC analytes generated by Wistar rat liver cytosol with 1-methylxanthine and O<sub>2</sub> as an electron acceptor. The chromatogram (b) HPLC chromatogram for incubation of Wistar rat liver cytosol with 1-MX and NAD<sup>+</sup> as an electron acceptor. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times .Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 52:** HPLC analysis of the in vitro biotransformation of xanthine by human liver cytosol. The chromatogram (a) HPLC analytes generated by human liver cytosol with xanthine and  $O_2$  as electron acceptor. The chromatogram (b) HPLC analytes generated by human liver cytosol with xanthine and NAD<sup>+</sup>. The amount of NAD<sup>+</sup> added to incubation 675µM. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 53:** HPLC analysis of the in vitro biotransformation of 1-methylxanthine to 1methyluric acid by human liver cytosol at  $37^{0}$ C. The chromatogram (a) HPLC analytes generated by human liver cytosol with 1-methylxanthine using O<sub>2</sub> as an electron acceptor. The chromatogram (b) HPLC analytes generated by human liver cytosol with 1-methylxanthine (1-MX) and NAD<sup>+</sup>. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times .Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.

When the liver cytosol from all three species was incubated with the XOR inhibitor oxipurinol and AOX inhibitor menadione, oxipurinol gave 100% inhibition while the menadione did not have any effect (Figures 54, 55, 56, 57, 58 and 59), this illustrates that XOR is responsible for biotransformation of xanthine and 1-methylxanthine to their metabolites in these species.







**Figure 54:** HPLC analysis of the in vitro biotransformation of xanthine by rainbow trout liver cytosol with and without enzyme inhibitors. The chromatogram of xanthine, NAD<sup>+</sup> plus (a) oxipurinol and (b) menadione. Red line with inhibitor and black line without inhibitor. The amount of inhibitor added to incubation its  $100\mu$ M. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 55:** HPLC analysis of the in vitro biotransformation of 1-methylxanthine by rainbow trout liver cytosol with and without xanthine oxidoreductase and aldehyde oxidase inhibitors. HPLC chromatogram for incubation of trout liver with (1-MX), NAD<sup>+</sup> with (a) oxipurinol, (b) menadione. Red lines with inhibitors and black lines without inhibitors. The amount of inhibitors added to incubation its 100µM. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 56:** HPLC analysis of the in vitro biotransformation of xanthine by Wistar rat liver cytosol with and without enzyme inhibitors. HPLC chromatogram for incubation of Wistar rat liver cytosol with xanthine and (a) oxipurinol, (b) menadione. Red lines with inhibitors and black lines without inhibitors. The amount of inhibitor added to incubation its 100 $\mu$ M. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 57:** HPLC analysis of the in vitro biotransformation of 1-methylxanthine by Wistar rat liver cytosol with and without enzyme inhibitors. HPLC chromatogram for incubation of rat with 1-MX and (a) oxipurinol, (b) menadione. Red lines with inhibitors and black lines without inhibitors. The amount of inhibitors added to incubation its 100 $\mu$ M. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 58:** HPLC analysis of the in vitro biotransformation of xanthine by human liver cytosol with and without enzyme inhibitors. The chromatogram of human liver cytosol with xanthine and (a) oxipurinol, (b) menadione. Red lines with inhibitors and black lines without inhibitors The amount of inhibitor added to incubation its 100 $\mu$ M. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 59:** HPLC analysis of the in vitro biotransformation of 1-methylxanthine by human liver cytosol with and without enzyme inhibitors. HPLC chromatogram for incubation of human liver with 1-MX and (a) oxipurinol, (b) menadione. Red lines with inhibitors and black lines without inhibitors. The amount of inhibitors added to incubation its 100 $\mu$ M. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.

#### **3.3.3.** 6-mercaptopurine.

6-mercaptopurine (6-MP) is a thiopurine anti-leukemia drug in which carbon 6 of purine has been substituted with a sulphur atom. In mammals 6-MP is oxidized to 6thiouric acid via 6-thioxanthine or 8-oxo-6-mercaptopurine by AOX or XOR (Figure 17) although there is some controversy over the in vitro studies (Krenitsky et al., 1972, Hall and Krenitsky. 1986, Rashidi et al., 2007). In order to validate the assay in this study, rabbit liver cytosol was incubated with 6-MP and the products generated analysed by HPLC. This demonstrated that when rabbit cytosol was incubated with 6-MP with O<sub>2</sub> as sole electron acceptor it was biotransformed to 6-thiouric acid via 6thioxanthine (Figure 60). This confirmed that the HPLC procedure and incubation conditions had successfully detected and resolved the expected 6-MP metabolites confirming the lack of detectable activity in rainbow trout liver cytosol towards this substrate was not due to experimental conditions. According to the in vitro experiments with rainbow trout, Wistar rat and human in this study, trout liver cytosol incubated with 6-MP with and without NAD<sup>+</sup> as an electron acceptor for 60 minute incubation time no peak of 6-thiouric acid formed (Figure 61). In contrast Wistar rat and human liver cytosol incubated with 6-mercaptopurine is a substrate for either XOR or AOX using  $O_2$  or NAD<sup>+</sup> as electron acceptors (Figures 62 and 63). On the other hand as in the (Figures 64 and 65), in the Wistar rat and human liver cytosol incubated with XOR inhibitor oxipurinol and AOX inhibitor menadione it was found that oxipurinol produced 100% inhibition while menadione did not have any effect. This shows that XOR was responsible for biotransformation of 6-mercaptopurine (6-MP) to 6-thiouric acid. Due to the 6-MP not being a substrate for rainbow trout it was decided to test 6-thioxanthine as a substrate for either XOR or AOX of rainbow trout liver cytosol without and with NAD<sup>+</sup> as electron acceptors (Figure 66). On the other hand, as in (Figure 67), trout liver cytosol incubated with XOR inhibitor oxipurinol and AOX inhibitor menadione found that oxipurinol produced 100% inhibition while the menadione does not have any effect, this shows that XOR is responsible for biotransformation of 6-thioxanthine to 6-thiouric acid.



**Figure 60:** HPLC analysis of the in vitro biotransformation of 6-mercaptopurine by rabbit liver cytosol. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



#### Time (minutes)

**Figure 61:** HPLC analysis of analytes produced following the in vitro incubation of 6-mercaptopurine with rainbow trout liver cytosol at 37°C. HPLC chromatogram (a) of analytes generated following incubation of 6-MP with trout liver cytosol and  $O_2$  as an electron acceptor. HPLC chromatogram (b) of analytes generated following incubation of 6-MP with trout liver cytosol and  $NAD^+$  as an electron acceptor. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 62:** HPLC analysis of analytes produced following the in vitro incubation of 6mercaptopurine with Wistar rat liver cytosol at 37°C. HPLC chromatogram (a) of analytes generated following incubation of 6-MP and  $O_2$  as an electron acceptor. HPLC chromatogram (b) of analytes generated following incubation of 6-MP and NAD<sup>+</sup>. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times .Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 63:** HPLC analysis of the in vitro biotransformation of 6-mercaptopurine by human liver cytosol. HPLC chromatogram (a) for incubation of human liver with 6-MP and  $O_2$  as an electron acceptor. HPLC chromatogram (b) for incubation of human liver with 6-MP and NAD<sup>+</sup> as an electron acceptor. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 64:** HPLC analysis of the in vitro biotransformation of 6-mercaptopurine by Wistar rat liver cytosol with and without enzyme inhibitors. HPLC chromatogram for incubation of Wistar rat with 6-MP and (a) oxipurinol, (b) menadione. Red lines with inhibitors and black lines without inhibitors. The amount of inhibitor added to incubation its 100 $\mu$ M. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 65:** HPLC analysis of the in vitro biotransformation of 6-mercaptopurine by human liver cytosol with and without enzyme inhibitors. Human liver with 6-MP and (a) oxipurinol, (b) menadione. Red lines with inhibitors and black lines without inhibitors. The amount of inhibitor added to incubation its 100 $\mu$ M. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 66:** HPLC analysis of the in vitro biotransformation of 6-thioxanthine by rainbow trout liver cytosol at 37°C. HPLC chromatograms (a) generated by rainbow trout liver cytosol with 6-thioxanthine and  $O_2$ as electron acceptor. HPLC chromatograms (b) generated by Trout liver cytosol with 6-thioxanthine and NAD<sup>+</sup> as electron acceptor. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



Time (minutes)

**Figure 67:** HPLC analysis of the in vitro biotransformation of 6-thioxanthine by rainbow trout liver cytosol with and without enzyme inhibitors. The chromatogram of rainbow trout with 6-thioxanthine and (a) oxipurinol, (b) menadione. Red lines with inhibitors and black lines without inhibitors. Times at the left hand side of the chromatograms indicate incubation times. The amount of inhibitor added to incubation its 100µM. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient system as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.

#### 3.3.4. Allopurinol.

When liver cytosol from rainbow trout, Wistar rat and human were incubated with hyperuricemia drug allopurinol the two mammalian species exhibited activity with both the  $O_2$  and  $NAD^+$  as electron acceptor indicative of both oxidase and dehydrogenase activity (Figures 68 and 69). In contrast the piscine XOR enzyme was only active as a dehydrogenase as no activity was present even after prolonged incubation times without NAD<sup>+</sup> figure 71.



**Figure 68:** HPLC analysis of the in vitro biotransformation of allopurinol to oxipurinol by Wistar rat liver cytosol at 37°C. (a) HPLC chromatograms of incubation of Wistar rat and allopurinol with  $O_2$  as an electron acceptor. (b) HPLC chromatograms of incubation of Wistar rat and allopurinol with NAD<sup>+</sup> as an electron acceptor. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 69:** HPLC analysis of the in vitro biotransformation of allopurinol by human liver cytosol at 37°C. (a) HPLC chromatograms of incubation of human and allopurinol with  $O_2$  as an electron acceptor. (b) HPLC chromatograms of incubation of human and allopurinol with NAD<sup>+</sup> as an electron acceptor. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5 µm) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.



**Figure 70:** HPLC analysis of the in vitro biotransformation of allopurinol by rainbow trout liver cytosol at 37°C. (a) HPLC chromatograms of analytes from incubation of rainbow trout with allopurinol using  $O_2$  as an electron acceptor. (b) HPLC chromatograms from analytes from incubation of rainbow trout and allopurinol with NAD<sup>+</sup> as an electron acceptor. Times at the left hand side of the chromatograms indicate incubation times. HPLC chromatograms are offset on the vertical axis to allow comparison between different incubation times. Analytes were injected onto a C-18 column (Kromasil 5  $\mu$ m) and eluted with a 0.1% v/v glacial acetic acid: acetonitrile gradient as described in section 2.5.3 as mobile phase. The wavelength of the detection was 280 nm.

## **3.4.** Determination of the kinetic parameters for aldehyde oxidase and xanthine oxidoreductase in trout, rat and human

Km and Vmax were calculated for the oxidation of exogenous aldehydes, endogenous vitamin and purines and exogenous N-heterocycles as substrates for AOX and XOR as described in methods section 2.8. Mean values of Km, Vmax and substrate efficiency (Ks = Vmax/Km) of AOX and XOR were determined using Eadie-Hofstee plots that gave good correlation coefficients ( $r^2$ )> 0.97 Appendix 3 (Figures 6 – 25). The advantage of this linearization method is that it tends to distribute the points more evenly so that it gives equal weight to data points in any range of substrate concentration or reaction rate.

#### **3.4.1. Exogenous aldehydes as substrates for AOX**

The Km and Vmax were calculated for two exogenous aromatic aldehyde compounds, vanillin and dimethylaminocinnamaldehyde (DMAC) by AOX.

The results obtained in the present study shows that DMAC and vanillin were good substrates of trout AOX. The enzyme kinetic parameters for AOX for all species are listed in table 38.  $K_m$  values for DMAC and vanillin were  $21.5 \pm 1.05 \mu$ M and  $53.2 \pm 2.1 \mu$ M respectively for trout liver cytosols.  $V_{max}$  values obtained  $0.69 \pm 0.13$  nmol/min/mg protein and  $1.51 \pm 0.18$  nmol/min/mg protein respectively. The  $K_m$  values for DMAC in trout were approximately 4 and 7-fold higher than that in rat and human. The  $K_m$  value of vanillin in the trout was approximately 5 and 10-fold higher compared with that in rat and human respectively in spectrophotometric assay and Vmax value was similar for both mammalian species. The substrate efficiency (Ks) value of DMAC and vanillin in the trout was lower compared with that in rat and human respectively see Appendix3(figures 6 - 25) (Table 39).

Species	Substrates	${K_m}^{**}(\mu M)$	V <sub>max</sub> (nmol/min/mg protein)	$\label{eq:ks} \begin{split} Ks &= (V_{max} / \ K_{m)} \\ (ml/min/mg \\ protein) \end{split}$
Rainbow trout	DMAC	$21.5 \pm \mathbf{1.05^a}$	$0.69 \pm 0.13^{\mathrm{a}}$	0.032
	Vanillin	$53.2\pm2.11^{\rm a}$	$1.51 \pm 0.18^{\mathrm{a}}$	0.028
	Vanillin	$54.2\pm0.64$	$\boldsymbol{1.89 \pm 0.27}$	0.035
	Pyridoxal	$121.9\pm3.69$	$1.41 \pm 0.69$	0.012
	All-trans retinal	$124.7\pm2.27$	$\textbf{2.22} \pm \textbf{0.62}$	0.017
	N <sup>1</sup> -methylnicotinamide	$\textbf{273.9} \pm \textbf{2.89}$	$1.63 \pm 1.33$	0.006
	Phenanthridine	$55.0 \pm 1.63$	$\boldsymbol{1.78 \pm 0.55}$	0.032
	Phthalazine	$61.2 \pm 1.41$	$\textbf{2.02} \pm \textbf{0.46}$	0.033
	Cinchonine	174.7 ± 2.55	$1.63 \pm 1.33$	0.009
Wistar rat	DMAC	$5.4 \pm 1.62^{\mathrm{a}}$	$2.24 \pm \mathbf{0.07^a}$	0.415*
	Vanillin	$9.7 \pm 1.56^{a}$	$3.77 \pm 0.21^{a}$	0.388
	Vanillin	$10.2\pm0.55^*$	$\boldsymbol{5.97 \pm 0.05}$	0.585*
	Pyridoxal	$46.4\pm2.51$	$6.28 \pm 0.59$	0.135
	All-trans retinal	$\textbf{30.9} \pm \textbf{1.23}$	$5.16\pm0.56$	0.166
	N <sup>1</sup> -methylnicotinamide	$156.2\pm3.68$	$\textbf{5.92} \pm \textbf{0.68}^{*}$	0.037
	Phenanthridine	$5.60\pm0.65^*$	$\textbf{2.43} \pm \textbf{0.14}$	0.433*
	Phthalazine	$10.5\pm0.55$	$\textbf{5.79} \pm \textbf{0.08}$	0.551
	Cinchonine	37.3 ± 1.99	$4.53 \pm 0.52$	0.121
Human	DMAC	$3.4 \pm 1.21^{a}$	$3.14 \pm 0.07^{\mathrm{a}}$	0.923
	Vanillin	$5.5\pm1.83^{\rm a}$	$4.65 \pm 0.10^{a}$	0.845
	Vanillin	8.1±0.36	$6.29 \pm 0.04^{*}$	0.776*
	Pyridoxal	$\textbf{34.1} \pm \textbf{1.97}^{*}$	$9.16 \pm 0.56$	0.268
	All-trans retinal	$12.6\pm0.58$	$8.32 \pm 0.06$	0.660*
	N <sup>1</sup> -methylnicotinamide	$139.5\pm2.86$	$6.69 \pm 0.74$	0.047
	Phenanthridine	$3.2\pm0.23^{\rm a}$	$4.16 \pm 0.14^{a}$	1.300
	Phthalazine	9.8 ± 0.59*	$\textbf{9.43} \pm \textbf{0.07}^{*}$	0.962*
	Cinchonine	$43.8 \pm 1.62$	$8.69 \pm 0.53$	0.198

**Table 39:** Kinetic constants for trout, rat and human liver AOX substrates inspectrophotometric and HPLC assays

Values shown are the mean  $\pm$  S.D. The typical results are from three independent experiments. (\*P< 0.05 one way ANOVA are marked with asterisks) compared with the rainbow trout, \*\*n = 3. (a)-Spectrophotometric assays,

## 3.4.2. Endogenous vitamin aldehydes as substrates for AOX

Two endogenous vitamin aldehydes, pyridoxal (vitamin B6), all-trans retinal (vitamin A) and  $N^1$ -methylnicotinamide (NMN) (a vitamin B3 metabolite) were used as AOX substrates. The Km value of endogenous vitamin was measured using an HPLC assays with trout, rat and human liver cytosols, Km, Vmax and Ks values are summarized in (Table 39).

### **3.4.2.1. Vitamin B6 (pyridoxal)**

The Km value of trout AOX with pyridoxal its  $121.9 \pm 3.6 \,\mu$ M was approximately 3fold more than that for mammalian AOX with Wistar rat and human of  $46.4 \pm 0.55$  $\mu$ M and  $34.1 \pm 1.97 \,\mu$ M respectively (Table 39). The Vmax value of trout AOX with pyridoxal found to be  $1.41 \pm 0.69$  nmol/min/mg protein and was approximately 6 and 9-fold lower than that Wistar rat and human respectively. The substrate efficiency (Ks) value of trout AOX was lower compared than that of human and rat respectively (Table 39).

## **3.4.2.2.** Vitamin A (all-trans retinal)

The Km value of trout AOX with all-trans retinal was  $124.7 \pm 2.27 \mu M$  was approximately 4-fold greater than that measured with Wistar rat with all-trans retinal and ~ 10 fold more than that of human AOX (Table 39). The Vmax value of trout AOX with all-trans retinal was  $2.22 \pm 0.62$  nmol/min/mg protein and was approximately 3 and 4-fold lower than that Wistar rat and human respectively. The substrate efficiency (Ks) value of trout with all-trans retinal was lower than that of rat and human respectively (Table 39).

## **3.4.2.3.** N<sup>1</sup>-methylnicotinamide

The Km value of N<sup>1</sup>-methylnicotinamide was measured and it was found to be 273.9  $\pm$  2.89  $\mu$ M with trout AOX, 156.2  $\pm$  3.68  $\mu$ M with Wistar rat AOX and 139.5  $\pm$  2.86  $\mu$ M with human AOX. The activity of trout AOX with N<sup>1</sup>-methylnicotinamide approximately 1.5-fold more than that of mammalian AOX with N<sup>1</sup>- methylnicotinamide table 38. The Vmax value of Wistar rat and human with N<sup>1</sup>- methylnicotinamide was approximately 5 and 6-foldgreaterrespectivelythan the piscine AOX. The substrate efficiency (Ks) value of trout AOX with N<sup>1</sup>- methylnicotinamide was lower than that both mammalian species (Table 39).

### 3.4.3. Exogenous N-heterocycles as substrates for AOX

The Km, Vmax and Ks values were calculated for three exogenous NPAHs that are environmental pollutants (phenanthridine, phthalazine and cinchonine) are shown in (Table 39).

#### **3.4.3.1.** Phenanthridine

Phenanthridine oxidation by trout cytosol was below detection limits of the spectrophotometric assay at all substrate concentrations used. However the activity was measurable with a sensitive HPLC assay. The Km value of exogenous N-heterocyclic phenanthridine with trout it was  $55.0 \pm 1.63\mu$ M. On other hand, the Km value of mammalian enzyme Wistar rat and human with exogenous N-heterocyclic phenanthridine were  $5.60 \pm 0.65 \mu$ M,  $3.20 \pm 0.23 \mu$ M respectively (Table 39). The Km of the piscine AOX with phenanthridine was approximately 10 times greater than that with mammalian AOX. The Vmax value of Wistar rat and human with phenanthridine was approximately 2 and 4 times respectively more than that with trout AOX. The substrate efficiency (Ks) value of Wistar rat and human with phenanthridine was greater than that with trout AOX (Table 39).

#### **3.4.3.2.** Phthalazine

The Km value of exogenous N-heterocyclic phthalazine with trout it was  $61.2 \pm 1.41$  µM. On other hand, the Km value of mammalian enzyme Wistar rat and human with exogenous N-heterocyclic phthalazine was  $10.50 \pm 0.55$  µM and  $9.8 \pm 0.59$  µM (Table 39). The Km of the piscine AOX with exogenous N-heterocyclic phthalazine was approximately 6 times higher than that with rat and human AOX activity. The Vmax value of trout AOX with phthalazine found to be  $2.02 \pm 0.46$  nmol/min/mg protein and was approximately 2 and 4 times lower than that Wistar rat and human respectively. The substrate efficiency (Ks) value of trout AOX with phthalazine was lower compared with Wistar rat and human respectively (Table 39).

#### 3.4.3.3. Cinchonine

The Km value of exogenous N-heterocyclic cinchonine with rainbow trout AOX approximately 4-fold higher than that two mammalian AOX with cinchonine (Table 39). The Vmax value of Wistar rat and human with cinchonine was approximately 4 and 8-fold greater than that trout AOX. The substrate efficiency (Ks) value of human AOX was greater compared than that Wistar rat and trout respectively (Table 39).

# **3.4.4. Endogenous and exogenous purines and purine analogs as substrates for XOR**

The Km, Vmax and Ks values of purines and purine analogs were measured using spectrophotometric and HPLC assays with rainbow trout, Wistar rat and human as in (Table 40).

#### 3.4.4.1. Xanthine.

The Km value for  $O_2$ -dependent xanthine oxidase assay for human was approximately 2-fold lower than that of rat in spectrophotometer and HPLC assay (Table 40). The Km value NAD-dependent xanthine dehydrogenase of with trout in spectrophotometric assay was 2-fold and 3-fold higher than that Wistar rat and human respectively (Table 40). The Km value of NAD-dependant xanthine dehydrogenase with trout in HPLC assay were 3 times higher than that with Wistar rat and human (Table 40). The NAD-dependant Vmax value for xanthine dehydrogenase with Wistar rat and human was approximately 5 and 7-fold greater than that O<sub>2</sub>-dependant respectively (Table 40). The substrate efficiency (Ks) value of NAD-dependent xanthine dehydrogenase with trout in spectrophotometer was lower compared with that in Wistar rat and human respectively (Table 40).

#### 3.4.4.2. 1-methylxanthine.

The NAD-dependent  $K_m$  value for 1-methylxanthine as an XDH substrate for trout was approximately 2-fold higher than that of other species as in (Table 40). The O<sub>2</sub>dependant Km value for 1-methylxanthine as substrate for both mammals was similar, and the NAD-dependant Vmax value for 1-methylxanthine dehydrogenase with Wistar rat and human was approximately 5 and 8-fold greater than the O<sub>2</sub>dependantactivity respectively (Table 40). The NAD-dependent substrate efficiency (Ks) value for 1-methylxanthine as an XDH substrate for trout was lower than that of other species (Table 40). The  $O_2$ -dependant substrate specificity (Ks) value for 1methylxanthine as substrate for human was greater compared with that Wistar rat (Table 40).

With the second seco	Species	Substrates	K <sub>m</sub> **(µM)	V <sub>max</sub> (nmol/min/mg protein)	$Ks = V_{max} / K_m$ (ml/min/mg
Rainborn         100 ± 100         100 ± 100         000 ± 100           Kanthine+NAD         15 % ± 0.58         0.77 ± 0.03         0.048           1-MX+NAD'         17.6 ± 1.23*         0.52 ± 0.14*         0.029           1-MX+NAD'         18.2 ± 0.55         0.51 ± 0.05         0.028           6-Thioxanthine+O_2         39.9 ± 0.85         1.47 ± 0.30         0.026           6-Thioxanthine+NAD'         30.9±1.52         5.01 ± 0.98         0.099           Allopurinol +NAD'         63.2 ± 0.98         0.56 ± 0.75         0.008           Pyrazinamide +NAD'         209.5 ± 2.40         1.21 ± 0.74         0.005           Xanthine         9.2 ± 0.45         1.27±0.17         0.138           Xanthine+NAD'         5.5 ± 1.45*         4.27 ± 0.87*         0.776*           Xanthine+NAD'         6.4 ± 0.81         5.17 ± 0.15*         0.807           1-MX +ND'         9.2 ± 0.62*         5.31± 0.10*         0.577*           Xanthine+NAD'         8.2 ± 0.62         5.13 ± 0.95         0.625           6-MP + O_2         29.1 ± 0.85         1.61 ± 0.06*         0.056           1-MX+ND'         15.3 ± 0.25         3.68±0.45         0.149           Allopurinol +ND'         15.3 ± 0.50*         1.97±0		Xanthine+NAD <sup>+</sup>	$135 \pm 19^{a}$	$0.56 \pm 0.15^{a}$	o 041
Hamman HAD         Los 0.00         0.000         0.000           1-MX+NAD*         17.6 ± 1.23*         0.52 ± 0.14*         0.029           1-MX+NAD*         18.2 ± 0.55         0.51 ± 0.05         0.028           6-Thioxanthine + 0.2         39.9 ± 0.85         1.67 ± 0.30         0.026           6-Thioxanthine + 0.2         30.9 ± 0.85         1.67 ± 0.30         0.026           6-Thioxanthine + 0.2         ND         ND         ND           Pyrazinamide + 0.2         ND         ND         ND           Pyrazinamide + 0.2         ND         ND         ND           Pyrazinamide + NAD*         209.5 ± 2.40         1.21 ± 0.74         0.005           Xanthine         13.2 ± 1.7*         1.69 ± 0.11*         0.138           Xanthine         5.5 ± 1.45*         4.27 ± 0.87*         0.776*           Xanthine+NAD*         5.5 ± 1.45*         4.27 ± 0.87*         0.776*           Xanthine+NAD*         5.2 ± 0.62*         5.31 ± 0.10*         0.557*           1-MX + O_2         10.7 ± 0.56*         1.01 ± 0.04         0.094           1-MX + ND*         9.2 ± 0.62*         5.31 ± 0.10*         0.625           6-MP + ND*         24.60.85*         3.68±0.45         0.149      <	Rainbow trout	Xanthine+ $NAD^+$	$15.8 \pm 0.58$	$0.30 \pm 0.13$	0.041
Rainbow trout         I-MAT/ND2         I-I-MAT/ND2         I-I-I-I-I-I-I-I-I-I-I-I-I-I-I-I-I-I-I-			$17.6 \pm 0.00$	$0.77 \pm 0.03$	0.029
Rainbow trout         Instrum         Instrum         Instrum         Instrum         Instrum         Instrum           6 - Thioxanthine + NAD'         39.9 ± 0.85         1.67 ± 0.30         0.026           6 - Thioxanthine + NAD'         30.9 ± 0.85         5.01 ± 0.39         0.009           Allopurinol + NAD'         63.2 ± 0.98         0.56 ± 0.75         0.008           Pyrazinamide + O <sub>2</sub> ND         ND         ND           Pyrazinamide + NAD'         209.5 ± 2.40         1.21 ± 0.74         0.005           Xanthine         13.2 ± 1.7*         1.69 ± 0.11*         0.128           Xanthine         9.2 ± 0.45         1.27±0.17         0.138           Xanthine+NAD'         5.5 ± 1.45*         4.27 ± 0.87*         0.776*           Xanthine+NAD'         6.4 ± 0.81         5.17 ± 0.15*         0.807           1-MX + NAD'         8.2 ± 0.62         5.13 ± 0.10*         0.577*           1-MX+NAD'         9.2 ± 0.62*         5.13 ± 0.95         0.625           6 -MP + O <sub>2</sub> 20.1 ± 0.85         1.61 ± 0.06*         0.055           6 -MP + NAD'         24.6±0.85*         3.68±0.45         0.149           Allopurinol + NAD'         15.3 ± 1.46         1.30 ± 1.33*         0.007 <t< td=""><td>1-MX+NAD<sup>+</sup></td><td><math>17.0 \pm 1.25</math> 18.2 ± 0.55</td><td><math>0.52 \pm 0.14</math></td><td>0.029</td></t<>		1-MX+NAD <sup>+</sup>	$17.0 \pm 1.25$ 18.2 ± 0.55	$0.52 \pm 0.14$	0.029
trout         6-Thioxanthine+NAD*         30.9±1.52         5.01±0.98         0.009           Allopurinol +NAD*         63.2±0.98         0.55±0.75         0.008           Pyrazinamide +O2         ND         ND         ND           Pyrazinamide +NAD*         209.5±2.40         1.21±0.74         0.005           Pyrazinamide +NAD*         209.5±2.40         1.21±0.74         0.005           Xanthine         9.2±0.45         1.27±0.17         0.138           Xanthine+NAD*         5.5±1.45*         4.27±0.87*         0.776*           Xanthine+NAD*         6.4±0.81         5.17±0.15*         0.807           1-MX +O2         10.7±0.56*         1.01±0.04         0.094           1-MX +O2         10.7±0.56*         1.01±0.04         0.097           1-MX +ND*         8.2±0.62         5.13±0.95         0.625           6-MP +O2         29.1±0.85         1.61±0.06*         0.055           6-MP+NAD*         24.6±0.85*         3.68±0.45         0.149           Allopurinol +NAD*         15.3±1.05         3.53±0.27         0.230*           Pyrazinamide +NAD*         92.2±1.20*         4.25±1.06         0.046           Allopurinol +N2         92.2±1.20*         1.41±0.01*         0.470		6-Thiovanthine + O.	$39.9 \pm 0.85$	$1.67 \pm 0.30$	0.026
Wistar rat         Non- Allopurinol +NAD <sup>+</sup> 63.2 ± 0.98 63.2 ± 0.98         0.56 ± 0.75         0.008 ND           Pyrazinamide + O <sub>2</sub> ND         ND         ND         ND           Pyrazinamide + O <sub>2</sub> ND         ND         ND         ND           Yanthine         13.2 ± 1.7 <sup>+</sup> 1.69 ± 0.11 <sup>+</sup> 0.128           Xanthine         9.2 ± 0.45         1.27±0.17         0.138           Xanthine+NAD <sup>+</sup> 5.5 ± 1.45 <sup>+</sup> 4.27 ± 0.87 <sup>+</sup> 0.776 <sup>+</sup> Xanthine+NAD <sup>+</sup> 6.4 ± 0.81         5.17 ± 0.15 <sup>+</sup> 0.807           1-MX + O <sub>2</sub> 10.7 ± 0.56 <sup>+</sup> 1.01 ± 0.04         0.094           1-MX+NAD <sup>+</sup> 9.2 ± 0.62 <sup>+</sup> 5.31± 0.10 <sup>4</sup> 0.577 <sup>+</sup> 1-MX+NAD <sup>+</sup> 8.2 ± 0.62         5.13 ± 0.95         0.625           6-MP + O <sub>2</sub> 29.1 ± 0.85         1.61 ± 0.06 <sup>+</sup> 0.055           6-MP + NAD <sup>+</sup> 24.6±0.85 <sup>+</sup> 3.68±0.45         0.149           Allopurinol + NAD <sup>+</sup> 24.6±0.85 <sup>+</sup> 3.68±0.45         0.007           Pyrazinamide + O <sub>2</sub> 163.3 ± 1.46         1.30 ± 1.33 <sup>+</sup> 0.007           Pyrazinamide + ND <sup>+</sup> 22.2 ± 1.20 <sup>+</sup> 4.25 ± 1.06         0		6-Thioxanthine+NAD <sup>+</sup>	30 9+1 52	$5.01 \pm 0.98$	0.099
Humphini (h)         Bold 1600         Bold 1600         Bold 1600         Bold 1600           Pyrazinamide + Q2         ND         ND         ND         ND           Pyrazinamide + Q2         ND         ND         ND         ND           Yughtan (h)         209.5 ± 2.40         1.21 ± 0.74         0.005           Xanthine         9.2 ± 0.45         1.27±0.17         0.138           Xanthine+NAD*         5.5 ± 1.45"         4.27 ± 0.87"         0.776*           Xanthine+NAD*         6.4 ± 0.81         5.17 ± 0.15*         0.807           1-MX + O2         10.7 ± 0.56*         1.01 ± 0.04         0.094           1-MX + ND*         9.2 ± 0.62*         5.31± 0.10*         0.577*           1-MX + NAD*         8.2 ± 0.62         5.13 ± 0.95         0.625           6-MP + O2         29.1 ± 0.85         1.61 ± 0.06*         0.055           6-MP + ND*         24.6±0.85*         3.68±0.45         0.149           Allopurinol + NAD*         15.30±1.05         3.53±0.27         0.230*           Pyrazinamide + O2         163.3 ± 1.46         1.30 ± 1.33*         0.007           Pyrazinamide + O2         163.3 ± 1.26*         1.41 ± 0.01*         0.470           Xanthine         5.5 ± 0.		Allopurinol $+NAD^+$	$63.2 \pm 0.98$	$0.56 \pm 0.75$	0.008
Pyrazinamide +NAD*         209.5 ± 2.40         1.21 ± 0.74         0.005           Xanthine         13.2 ± 1.7*         1.69 ± 0.11*         0.128           Xanthine         9.2 ± 0.45         1.27±0.17         0.138           Xanthine+NAD*         5.5 ± 1.45*         4.27 ± 0.87*         0.776*           Xanthine+NAD*         6.4 ±0.81         5.17 ± 0.15*         0.807           1-MX + O_2         10.7 ± 0.56*         1.01 ± 0.04         0.094           1-MX + NAD*         8.2 ± 0.62         5.31 ± 0.10*         0.577*           1-MX+NAD*         8.2 ± 0.62         5.13 ± 0.95         0.625           6-MP + O_2         29.1 ± 0.85         1.61 ± 0.06*         0.055           6-MP + ND*         24.6±0.85*         3.68±0.45         0.149           Allopurinol + O_2         20.9 ± 0.76         1.19 ± 0.08         0.056           Allopurinol + ND*         15.30±1.05         3.53±0.27         0.230*           Pyrazinamide + O_2         163.3 ± 1.46         1.30 ± 1.33*         0.007           Pyrazinamide + NAD*         92.2 ± 1.20*         4.25 ± 1.06         0.046           Xanthine         5.5 ± 0.50*         1.97±0.05         0.358           Xanthine         5.7 ± 0.30         8.18 ± 0.85<		$\frac{1}{2}$ Pyrazinamide + O <sub>2</sub>		ND	ND
Number of the second		Pyrazinamide +NAD <sup>+</sup>	209.5 + 2.40	$1.21 \pm 0.74$	0.005
Human $1.52 \pm 1.7$ $1.07 \pm 0.17$ $0.138$ Xanthine $9.2 \pm 0.45$ $1.27\pm 0.17$ $0.138$ Xanthine+NAD* $5.5 \pm 1.45^{*}$ $4.27 \pm 0.87^{*}$ $0.776^{*}$ Xanthine+NAD* $6.4 \pm 0.81$ $5.17 \pm 0.15^{*}$ $0.807$ $1-MX + O_2$ $10.7 \pm 0.56^{*}$ $1.01 \pm 0.04$ $0.094$ $1-MX + NAD^{*}$ $9.2 \pm 0.62^{*}$ $5.31 \pm 0.10^{*}$ $0.577^{*}$ $1-MX + NAD^{*}$ $8.2 \pm 0.62$ $5.13 \pm 0.95$ $0.625$ $6-MP + O_2$ $29.1 \pm 0.85$ $1.61 \pm 0.06^{*}$ $0.055$ $6-MP + NAD^{*}$ $24.6\pm 0.85^{*}$ $3.68\pm 0.45$ $0.149$ Allopurinol + $O_2$ $20.9 \pm 0.76$ $1.19 \pm 0.08$ $0.065$ Allopurinol + $NAD^{*}$ $15.30\pm 1.05$ $3.53\pm 0.27$ $0.230^{*}$ Pyrazinamide + $NAD^{*}$ $92.2 \pm 1.20^{*}$ $4.25 \pm 1.06$ $0.046$ Xanthine $3.0 \pm 0.65^{*}$ $1.41 \pm 0.01^{*}$ $0.470$ Xanthine $3.0 \pm 0.65^{*}$ $1.41 \pm 0.01^{*}$ $0.470$ Xanthine $3.0 \pm 0.65^{*}$		Yanthina	$132 \pm 17^{a}$	$1.60 \pm 0.11^{a}$	0.128
Wistar rat $3.2 \pm 0.43$ $1.215.17$ $0.010^{\circ}$ Wistar ratXanthine+NAD* $5.5 \pm 1.45^{\circ}$ $4.27 \pm 0.87^{\circ}$ $0.776^{\circ}$ Number NAM* $5.5 \pm 1.45^{\circ}$ $4.27 \pm 0.87^{\circ}$ $0.776^{\circ}$ Number Number rat $1.MX + 0.2$ $10.7 \pm 0.56^{\circ}$ $1.01 \pm 0.04$ $0.094$ $1.MX + NAD*$ $9.2 \pm 0.62^{\circ}$ $5.31 \pm 0.10^{\circ}$ $0.577^{\circ}$ $1.MX + NAD*$ $8.2 \pm 0.62$ $5.13 \pm 0.95$ $0.625$ $6-MP + 0_2$ $29.1 \pm 0.85$ $1.61 \pm 0.06^{\circ}$ $0.055$ $6-MP + ND*$ $24.6\pm 0.85^{\circ}$ $3.68\pm 0.45$ $0.149$ Allopurinol + $0_2$ $20.9 \pm 0.76$ $1.19 \pm 0.08$ $0.056$ Allopurinol + $NAD^{\circ}$ $15.30 \pm 1.05$ $3.53\pm 0.27$ $0.230^{\circ}$ Pyrazinamide + $0_2$ $163.3 \pm 1.46$ $1.30 \pm 1.33^{\circ}$ $0.007$ Pyrazinamide + NAD* $92.2 \pm 1.20^{\circ}$ $4.25 \pm 1.06$ $0.466$ Xanthine $3.0 \pm 0.65^{\circ}$ $1.41 \pm 0.01^{\circ}$ $0.470$ Xanthine $5.5 \pm 0.50^{\circ}$ $1.97\pm 0.05$ $0.358$ Xanthine $5.2 \pm 0.50^{\circ}$ $1.97\pm 0.05$ $0.358$ Xanthine $5.5 \pm 0.50^{\circ}$ $1.97\pm 0.05$ $0.358$ Xanthine+NAD* $1.9 \pm 0.20^{\circ}$ $5.48 \pm 0.07^{\circ}$ $2.884^{\circ}$ Xanthine+NAD* $2.4 \pm 0.35$ $7.25 \pm 0.17^{\circ}$ $3.021^{\ast}$ $1.MX + 0_2$ $7.0 \pm 0.58^{\circ}$ $1.02 \pm 0.15$ $0.145$ $1.MX + NAD^{+}$ $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^{\circ}$ $1.MX + NAD^{+}$ $5.7 \pm 0.30$ $8.$		Vanthina	$13.2 \pm 1.7$ $9.2 \pm 0.45$	1.07 ± 0.11	0.138
Wistar ratXanthine+NAD* $6.4 \pm 0.81$ $5.17 \pm 0.15^{*}$ $0.070^{*}$ Wistar rat $1.MX + O_2$ $10.7 \pm 0.56^{*}$ $1.01 \pm 0.04$ $0.094$ $1.MX + NAD^{*}$ $9.2 \pm 0.62^{*}$ $5.31 \pm 0.10^{*}$ $0.577^{*}$ $1.MX + NAD^{*}$ $8.2 \pm 0.62^{*}$ $5.31 \pm 0.10^{*}$ $0.577^{*}$ $1.MX + NAD^{*}$ $8.2 \pm 0.62^{*}$ $5.31 \pm 0.95^{*}$ $0.625^{*}$ $6-MP + O_2$ $29.1 \pm 0.85^{*}$ $1.61 \pm 0.06^{**}$ $0.055^{*}$ $6-MP + NAD^{*}$ $24.6 \pm 0.85^{**}$ $3.68 \pm 0.45^{*}$ $0.149^{*}$ Allopurinol + O_2 $20.9 \pm 0.76^{*}$ $1.19 \pm 0.08$ $0.056^{*}$ Allopurinol + NAD^{*} $15.30 \pm 1.05^{*}$ $3.53 \pm 0.27^{*}$ $0.230^{*}$ Pyrazinamide + NAD^{*} $92.2 \pm 1.20^{**}$ $4.25 \pm 1.06^{*}$ $0.046^{*}$ Number NAD^{*} $92.2 \pm 1.20^{**}$ $4.25 \pm 1.06^{**}$ $0.046^{**}$ Xanthine $3.0 \pm 0.65^{**}$ $1.41 \pm 0.01^{**}$ $0.470^{**}$ Xanthine $5.5 \pm 0.50^{**}$ $1.97 \pm 0.05^{**}$ $0.358^{**}$ Xanthine $5.5 \pm 0.50^{**}$ $1.97 \pm 0.05^{**}$ $0.358^{**}$ Xanthine $5.5 \pm 0.50^{**}$ $1.97 \pm 0.05^{**}$ $0.358^{**}$ Xanthine+NAD^{*} $2.4 \pm 0.35^{**}$ $7.25 \pm 0.17^{**}$ $3.021^{**}$ $1-MX + O_2$ $7.0 \pm 0.58^{**}$ $1.00 \pm 0.15^{**}$ $1.435^{**}$ $6-MP + O_2$ $25.7 \pm 0.51^{*}$ $1.73 \pm 0.04^{**}$ $0.067^{**}$ $6-MP + NAD^{*}$ $20.80 \pm 0.41^{*}$ $6.62 \pm 0.64^{**}$ $0.318^{**}$ $6-MP $		$\mathbf{X}_{anthine} + \mathbf{N} \mathbf{\Lambda} \mathbf{D}^{+}$	$5.5 \pm 1.45^{a}$	$1.27\pm0.17$	0.776*
Wistar ratI $MX + O_2$ $10.7 \pm 0.50^{\circ}$ $5.0.7 \pm 0.10^{\circ}$ $5.0.7 \pm 0.10^{\circ}$ Wistar rat $1.MX + O_2$ $10.7 \pm 0.56^{\circ}$ $1.01 \pm 0.04$ $0.094$ $1.MX + NAD^{\circ}$ $9.2 \pm 0.62^{\circ}$ $5.31 \pm 0.10^{\circ}$ $0.577^{\circ}$ $1.MX + NAD^{\circ}$ $8.2 \pm 0.62$ $5.13 \pm 0.95$ $0.625$ $6-MP + O_2$ $29.1 \pm 0.85$ $1.61 \pm 0.06^{\circ}$ $0.055$ $6-MP + NAD^{\circ}$ $24.6 \pm 0.85^{\circ}$ $3.68 \pm 0.45$ $0.149$ Allopurinol + O_2 $20.9 \pm 0.76$ $1.19 \pm 0.08$ $0.056$ Allopurinol + NAD^{\circ} $15.30 \pm 1.05$ $3.53 \pm 0.27$ $0.230^{\circ}$ Pyrazinamide + NAD^{\circ} $92.2 \pm 1.20^{\circ}$ $4.25 \pm 1.06$ $0.046$ Xanthine $3.0 \pm 0.65^{\circ}$ $1.41 \pm 0.01^{\circ}$ $0.470$ Yaratinamide + NAD^{\circ} $92.2 \pm 1.20^{\circ}$ $4.25 \pm 1.06$ $0.046$ Xanthine $3.0 \pm 0.65^{\circ}$ $1.97 \pm 0.05$ $0.358$ Xanthine + NAD^{\circ} $1.9 \pm 0.20^{\circ}$ $5.48 \pm 0.07^{\circ}$ $2.884^{\ast}$ Xanthine+NAD^{\circ} $2.4 \pm 0.35$ $7.25 \pm 0.17^{\circ}$ $3.021^{\circ}$ I-MX + O_2 $7.0 \pm 0.58^{\circ}$ $1.02 \pm 0.15$ $0.145$ I-MX + NAD^{\circ} $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^{\circ}$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^{\circ}$ $0.067$ $6-MP + NAD^{\circ}$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + O_2 $16.7 \pm 0.62^{\circ}$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD^{\circ} $152.25$ $5.52 \pm 1.07^{\circ}$ $0.368^{\ast}$ Pyrazinamid		$\mathbf{X}_{anthine} + \mathbf{N} \mathbf{A} \mathbf{D}^{+}$	5.5 ± 1.45	4.27 ± 0.87 5 17 + 0 15*	0.807
Wistar rat $1-MX + NAD^+$ $9.2 \pm 0.62^{\circ}$ $1.M1 \pm 0.04^{\circ}$ $0.054^{\circ}$ Wistar rat $1-MX + NAD^+$ $8.2 \pm 0.62^{\circ}$ $5.31 \pm 0.10^{\circ}$ $0.577^{\circ}$ $1-MX + NAD^+$ $8.2 \pm 0.62^{\circ}$ $5.31 \pm 0.95^{\circ}$ $0.625^{\circ}$ $6-MP + O_2$ $29.1 \pm 0.85^{\circ}$ $1.61 \pm 0.06^{\circ}$ $0.055^{\circ}$ $6-MP + NAD^+$ $24.6 \pm 0.85^{\circ}$ $3.68 \pm 0.45^{\circ}$ $0.149^{\circ}$ Allopurinol + $O_2$ $20.9 \pm 0.76^{\circ}$ $1.19 \pm 0.08^{\circ}$ $0.056^{\circ}$ Allopurinol + NAD^+ $15.30 \pm 1.05^{\circ}$ $3.53 \pm 0.27^{\circ}$ $0.230^{\circ}$ Pyrazinamide + $O_2$ $163.3 \pm 1.46^{\circ}$ $1.30 \pm 1.33^{\circ}$ $0.007^{\circ}$ Pyrazinamide + $O_2$ $163.3 \pm 1.46^{\circ}$ $1.30 \pm 1.33^{\circ}$ $0.007^{\circ}$ Pyrazinamide + $NAD^+$ $92.2 \pm 1.20^{\circ}$ $4.25 \pm 1.06^{\circ}$ $0.470^{\circ}$ Xanthine $3.0 \pm 0.65^{\circ}$ $1.41 \pm 0.01^{\circ}$ $0.470^{\circ}$ Xanthine $5.5 \pm 0.50^{\circ}$ $1.97 \pm 0.05^{\circ}$ $0.358^{\circ}$ Xanthine $2.4 \pm 0.35^{\circ}$ $7.25 \pm 0.17^{\circ}$ $3.021^{\circ}$ Human $1-MX + O_2$ $7.0 \pm 0.58^{\circ}$ $1.02 \pm 0.15^{\circ}$ $0.145^{\circ}$ $1-MX + NAD^{+}$ $6.6 \pm 1.05^{\circ}$ $8.31 \pm 0.96^{\circ}$ $1.259^{\circ}$ Human $1-MX + NAD^{+}$ $5.7 \pm 0.30^{\circ}$ $8.18 \pm 0.85^{\circ}$ $1.435^{\circ}$ $6-MP + O_2$ $25.7 \pm 0.51^{\circ}$ $1.73 \pm 0.04^{\circ}$ $0.067^{\circ}$ $6-MP + NAD^{+}$ $20.80 \pm 0.41^{\circ}$ $6.62 \pm 0.64^{\circ}$ $0.318^{\circ}$ Allopurinol + NAD^{+} $15 \pm 0.25^{\circ}$ $5.52 \pm 1.07^{\circ}$ <th></th> <td>1-MX + 0</td> <td><math>10.7 \pm 0.51</math></td> <td><math>1.01 \pm 0.04</math></td> <td>0.094</td>		1-MX + 0	$10.7 \pm 0.51$	$1.01 \pm 0.04$	0.094
Wistar rat $1-MX+NAD^+$ $8.2 \pm 0.62$ $5.13 \pm 0.95$ $0.625$ $6-MP + O_2$ $29.1 \pm 0.85$ $1.61 \pm 0.06^+$ $0.055$ $6-MP + NAD^+$ $24.6 \pm 0.85^*$ $3.68 \pm 0.45$ $0.149$ Allopurinol + $O_2$ $20.9 \pm 0.76$ $1.19 \pm 0.08$ $0.056$ Allopurinol + NAD^+ $15.30 \pm 1.05$ $3.53 \pm 0.27$ $0.230^*$ Pyrazinamide + $O_2$ $163.3 \pm 1.46$ $1.30 \pm 1.33^*$ $0.007$ Pyrazinamide + NAD^+ $92.2 \pm 1.20^*$ $4.25 \pm 1.06$ $0.046$ Xanthine $5.5 \pm 0.50^*$ $1.97 \pm 0.05$ $0.358$ Xanthine $5.5 \pm 0.50^*$ $1.97 \pm 0.05$ $0.358$ Xanthine+NAD^+ $1.9 \pm 0.20^a$ $5.48 \pm 0.07^a$ $2.884^*$ Xanthine+NAD^+ $1.9 \pm 0.20^a$ $5.48 \pm 0.07^a$ $2.884^*$ Xanthine+NAD^+ $1.9 \pm 0.20^a$ $5.48 \pm 0.07^a$ $2.884^*$ Xanthine+NAD^+ $2.4 \pm 0.35$ $7.25 \pm 0.17^*$ $3.021^*$ I-MX + O_2 $7.0 \pm 0.58^*$ $1.02 \pm 0.15$ $0.145$ I-MX+NAD^+ $6.6 \pm 1.05^a$ $8.31 \pm 0.96^a$ $1.259^*$ Human $1-MX+NAD^+$ $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^*$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ $6-MP + NAD^+$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + O_2 $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD^+ $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + NAD^+ $90.7 \pm 1.00$ $6.41 \pm 0.94$ $0.070$ <th></th> <td>1-MX+NAD<sup>+</sup></td> <td><math>9.2 \pm 0.62^{a}</math></td> <td><math>5.31 \pm 0.04</math></td> <td>0.577*</td>		1-MX+NAD <sup>+</sup>	$9.2 \pm 0.62^{a}$	$5.31 \pm 0.04$	0.577*
rat $6 \cdot MP + O_2$ $29.1 \pm 0.82$ $1.61 \pm 0.06^+$ $0.025^ 6 \cdot MP + NAD^+$ $24.6 \pm 0.85^+$ $3.68 \pm 0.45^ 0.149^-$ Allopurinol + $O_2$ $20.9 \pm 0.76^ 1.19 \pm 0.08^ 0.056^-$ Allopurinol + NAD^+ $15.30 \pm 1.05^ 3.53 \pm 0.27^ 0.230^+$ Pyrazinamide + $O_2$ $163.3 \pm 1.46^ 1.30 \pm 1.33^+$ $0.007^-$ Pyrazinamide + NAD^+ $92.2 \pm 1.20^+$ $4.25 \pm 1.06^ 0.046^-$ Xanthine $3.0 \pm 0.65^a^ 1.41 \pm 0.01^a^ 0.470^-$ Xanthine $5.5 \pm 0.50^+$ $1.97 \pm 0.05^ 0.358^-$ Xanthine $5.5 \pm 0.50^+$ $1.97 \pm 0.05^ 0.358^-$ Xanthine+NAD^+ $1.9 \pm 0.20^a^ 5.48 \pm 0.07^a^ 2.884^+$ Xanthine+NAD^+ $1.9 \pm 0.20^a^ 5.48 \pm 0.07^a^ 2.884^+$ Xanthine+NAD^+ $2.4 \pm 0.35^ 7.25 \pm 0.17^+$ $3.021^+$ $1-MX + O_2$ $7.0 \pm 0.58^+$ $1.02 \pm 0.15^ 0.145^ 1-MX + NAD^+$ $6.6 \pm 1.05^a^ 8.31 \pm 0.96^a^ 1.259^+$ $1-MX + NAD^+$ $5.7 \pm 0.30^ 8.18 \pm 0.85^ 1.435^+$ $6 - MP + O_2$ $25.7 \pm 0.51^ 1.73 \pm 0.04^+$ $0.067^ 6 - MP + NAD^+$ $20.80 \pm 0.41^ 6.62 \pm 0.64^ 0.318^-$ Allopurinol + O_2 $16.7 \pm 0.62^+$ $1.73 \pm 0.05^ 0.103^-$ Allopurinol + NAD^+ $15 \pm 0.25^ 5.52 \pm 1.07^+$ $0.368^+$ Pyrazinamide + NAD^+ $90.7 \pm 1.00^ 6.41 \pm 0.94^ 0.070^-$	Wistar rat	$1 MX + NAD^+$	$82 \pm 0.62$	5 13 + 0 95	0.625
Human         Intervent         I		$6-MP + O_2$	$29.1 \pm 0.85$	1.61 + 0.06*	0.025
Allopurinol + $O_2$ $20.9 \pm 0.76$ $1.19 \pm 0.08$ $0.056$ Allopurinol + NAD* $15.30\pm 1.05$ $3.53\pm 0.27$ $0.230^*$ Pyrazinamide + $O_2$ $163.3 \pm 1.46$ $1.30 \pm 1.33^*$ $0.007$ Pyrazinamide + NAD* $92.2 \pm 1.20^*$ $4.25 \pm 1.06$ $0.046$ Xanthine $3.0 \pm 0.65^a$ $1.41 \pm 0.01^a$ $0.470$ Xanthine $5.5 \pm 0.50^*$ $1.97\pm 0.05$ $0.358$ Xanthine+NAD* $1.9 \pm 0.20^a$ $5.48 \pm 0.07^a$ $2.884^*$ Xanthine+NAD* $2.4 \pm 0.35$ $7.25 \pm 0.17^*$ $3.021^*$ $1-MX + O_2$ $7.0 \pm 0.58^*$ $1.02 \pm 0.15$ $0.145$ $1-MX + NAD^*$ $6.6 \pm 1.05^a$ $8.18 \pm 0.85$ $1.435^*$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ $6-MP + NAD^*$ $20.80\pm 0.41$ $6.62\pm 0.64$ $0.318$ Allopurinol + O_2 $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD* $15\pm 0.25$ $5.52\pm 1.07^*$ $0.368^*$ Pyrazinamide + O_2 $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$ Pyrazinamide + NAD* $90.7 \pm 1.00$ $6.41 \pm 0.94$ $0.070$		6-MP+NAD <sup>+</sup>	24.6±0.85*	3.68±0.45	0.149
Humania Poly         Humania Poly<		Allopurinol + O <sub>2</sub>	$20.9 \pm 0.76$	$1.19 \pm 0.08$	0.056
Pyrazinamide + $O_2$ 163.3 ± 1.461.30 ± 1.33*0.007Pyrazinamide + NAD <sup>+</sup> 92.2 ± 1.20*4.25 ± 1.060.046Xanthine $3.0 \pm 0.65^a$ $1.41 \pm 0.01^a$ 0.470Xanthine $5.5 \pm 0.50^*$ $1.97 \pm 0.05$ 0.358Xanthine+NAD <sup>+</sup> $1.9 \pm 0.20^a$ $5.48 \pm 0.07^a$ 2.884*Xanthine+NAD <sup>+</sup> $1.9 \pm 0.20^a$ $5.48 \pm 0.07^a$ 2.884*Xanthine+NAD <sup>+</sup> $2.4 \pm 0.35$ $7.25 \pm 0.17^*$ $3.021^*$ 1-MX + $O_2$ $7.0 \pm 0.58^*$ $1.02 \pm 0.15$ $0.145$ 1-MX+NAD <sup>+</sup> $6.6 \pm 1.05^a$ $8.31 \pm 0.96^a$ $1.259^*$ 1-MX+NAD <sup>+</sup> $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^*$ 6-MP + $O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ 6-MP+NAD <sup>+</sup> $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + $O_2$ $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD <sup>+</sup> $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + $O_2$ $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$		Allopurinol +NAD <sup>+</sup>	15.30±1.05	3.53±0.27	0.230*
Pyrazinamide +NAD+ $92.2 \pm 1.20^*$ $4.25 \pm 1.06$ $0.046$ Xanthine $3.0 \pm 0.65^a$ $1.41 \pm 0.01^a$ $0.470$ Xanthine $5.5 \pm 0.50^*$ $1.97 \pm 0.05$ $0.358$ Xanthine+NAD+ $1.9 \pm 0.20^a$ $5.48 \pm 0.07^a$ $2.884^*$ Xanthine+NAD+ $2.4 \pm 0.35$ $7.25 \pm 0.17^*$ $3.021^*$ $1.MX + O_2$ $7.0 \pm 0.58^*$ $1.02 \pm 0.15$ $0.145$ $1.MX + NAD^+$ $6.6 \pm 1.05^a$ $8.31 \pm 0.96^a$ $1.259^*$ $1-MX + NAD^+$ $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^*$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ $6-MP + NAD^+$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + O_2 $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD^+ $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + O_2 $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$ Pyrazinamide + NAD^+ $90.7 \pm 1.00$ $6.41 \pm 0.94$ $0.070$		Pvrazinamide + O <sub>2</sub>	$163.3 \pm 1.46$	1.30 ± 1.33*	0.007
Xanthine $3.0 \pm 0.65^a$ $1.41 \pm 0.01^a$ $0.470$ Xanthine $5.5 \pm 0.50^*$ $1.97 \pm 0.05$ $0.358$ Xanthine+NAD <sup>+</sup> $1.9 \pm 0.20^a$ $5.48 \pm 0.07^a$ $2.884^*$ Xanthine+NAD <sup>+</sup> $2.4 \pm 0.35$ $7.25 \pm 0.17^*$ $3.021^*$ 1-MX + O2 $7.0 \pm 0.58^*$ $1.02 \pm 0.15$ $0.145$ 1-MX+NAD <sup>+</sup> $6.6 \pm 1.05^a$ $8.31 \pm 0.96^a$ $1.259^*$ 1-MX+NAD <sup>+</sup> $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^*$ 6-MP + O2 $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ 6-MP+NAD <sup>+</sup> $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + O2 $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD <sup>+</sup> $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + O2 $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$ Pyrazinamide + NAD <sup>+</sup> $90.7 \pm 1.00$ $6.41 \pm 0.94$ $0.070$		Pyrazinamide +NAD <sup>+</sup>	92.2 ± 1.20*	$4.25 \pm 1.06$	0.046
Human $3.0 \pm 0.05$ $1.41 \pm 0.01$ $1.000$ Xanthine $5.5 \pm 0.50^{*}$ $1.97 \pm 0.05$ $0.358$ Xanthine+NAD <sup>+</sup> $1.9 \pm 0.20^{a}$ $5.48 \pm 0.07^{a}$ $2.884^{*}$ Xanthine+NAD <sup>+</sup> $2.4 \pm 0.35$ $7.25 \pm 0.17^{*}$ $3.021^{*}$ 1-MX + O <sub>2</sub> $7.0 \pm 0.58^{*}$ $1.02 \pm 0.15$ $0.145$ 1-MX+NAD <sup>+</sup> $6.6 \pm 1.05^{a}$ $8.31 \pm 0.96^{a}$ $1.259^{*}$ 1-MX+NAD <sup>+</sup> $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^{*}$ 6-MP + O <sub>2</sub> $25.7 \pm 0.51$ $1.73 \pm 0.04^{*}$ $0.067$ 6-MP+NAD <sup>+</sup> $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + O <sub>2</sub> $16.7 \pm 0.62^{*}$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD <sup>+</sup> $15 \pm 0.25$ $5.52 \pm 1.07^{*}$ $0.368^{*}$ Pyrazinamide + O <sub>2</sub> $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$		Vanthina	$3.0 \pm 0.65^{a}$	$1.41 \pm 0.01^{a}$	0.470
Human $1.9 \pm 0.20^{\circ}$ $1.9 \pm 0.20^{\circ}$ $1.9 \pm 0.03^{\circ}$ $1.040^{\circ}$ Human $1.9 \pm 0.20^{\circ}$ $5.48 \pm 0.07^{\circ}$ $2.884^{*}$ Human $1.MX + O_2$ $7.0 \pm 0.58^{*}$ $1.02 \pm 0.15$ $0.145$ $1-MX + NAD^{+}$ $6.6 \pm 1.05^{\circ}$ $8.31 \pm 0.96^{\circ}$ $1.259^{*}$ Human $1-MX + NAD^{+}$ $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^{*}$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^{*}$ $0.067$ $6-MP + O_2$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + $O_2$ $16.7 \pm 0.62^{*}$ $1.73 \pm 0.05$ $0.103$ Allopurinol + $O_2$ $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$ Pyrazinamide + NAD^{+} $90.7 \pm 1.00$ $6.41 \pm 0.94$ $0.070$		Yanthine	$5.0 \pm 0.05$ 5.5 + 0.50*	1.41 ± 0.01	0.358
HumanHo 2 0100 $0.40 \pm 0.01$ $0.40 \pm 0.01$ Xanthine+NAD* $2.4 \pm 0.35$ $7.25 \pm 0.17^*$ $3.021^*$ $1-MX + O_2$ $7.0 \pm 0.58^*$ $1.02 \pm 0.15$ $0.145$ $1-MX+NAD^+$ $6.6 \pm 1.05^a$ $8.31 \pm 0.96^a$ $1.259^*$ $1-MX+NAD^+$ $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^*$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ $6-MP+NAD^+$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + O_2 $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD^+ $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + O_2 $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$	Human	Xanthine+NAD <sup>+</sup>	$1.9 \pm 0.20^{a}$	$5.48 \pm 0.07^{a}$	2.884*
Human $1.4 \pm 0.00^{\circ}$ $1.4 \pm 0.00^{\circ}$ $1.4 \pm 0.00^{\circ}$ $1.4 \pm 0.00^{\circ}$ Human $1MX + NAD^+$ $6.6 \pm 1.05^a$ $8.31 \pm 0.96^a$ $1.259^*$ Human $1-MX + NAD^+$ $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^*$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ $6-MP + NAD^+$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + $O_2$ $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol + $AD^+$ $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + $O_2$ $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$		Xanthine+NAD <sup>+</sup>	$2.4 \pm 0.35$	$7.25 \pm 0.17^*$	3.021*
Human $1.01 \pm 0.2^{\circ}$ $1.05 \pm 0.00^{\circ}$ $1.02 \pm 0.00^{\circ}$ $1.02 \pm 0.00^{\circ}$ Human $1-MX+NAD^+$ $6.6 \pm 1.05^a$ $8.31 \pm 0.96^a$ $1.259^*$ $1-MX+NAD^+$ $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^*$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ $6-MP+NAD^+$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + $O_2$ $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol+NAD^+ $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + $O_2$ $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$ Pyrazinamide + NAD^+ $90.7 \pm 1.00$ $6.41 \pm 0.94$ $0.070$		$1-MX + O_2$	7.0 + 0.58*	$1.02 \pm 0.15$	0.145
Human $1-MX+NAD^+$ $5.7 \pm 0.30$ $8.18 \pm 0.85$ $1.435^*$ $6-MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ $6-MP+NAD^+$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + $O_2$ $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol+NAD^+ $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + $O_2$ $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$		1-MX+NAD <sup>+</sup>	$6.6 \pm 1.05^{a}$	$8.31 \pm 0.96^{a}$	1.259*
Human $6 - MP + O_2$ $25.7 \pm 0.51$ $1.73 \pm 0.04^*$ $0.067$ $6 - MP + NAD^+$ $20.80 \pm 0.41$ $6.62 \pm 0.64$ $0.318$ Allopurinol + O_2 $16.7 \pm 0.62^*$ $1.73 \pm 0.05$ $0.103$ Allopurinol + NAD^+ $15 \pm 0.25$ $5.52 \pm 1.07^*$ $0.368^*$ Pyrazinamide + O_2 $153.1 \pm 2.05$ $1.07 \pm 0.37$ $0.006$ Pyrazinamide + NAD^+ $90.7 \pm 1.00$ $6.41 \pm 0.94$ $0.070$		1-MX+NAD <sup>+</sup>	$5.7 \pm 0.30$	$8.18 \pm 0.85$	1.435*
6-MP+NAD+20.80±0.416.62±0.640.318Allopurinol + $O_2$ 16.7 ± 0.62*1.73 ± 0.050.103Allopurinol+NAD+15±0.255.52± 1.07*0.368*Pyrazinamide + $O_2$ 153.1 ± 2.051.07 ± 0.370.006Pyrazinamide + NAD+90.7 ± 1.006.41 ± 0.940.070		<b>6-MP</b> + <b>O</b> <sub>2</sub>	25.7 ± 0.51	$1.73 \pm 0.04*$	0.067
Allopurinol + $O_2$ 16.7 $\pm$ 0.62*1.73 $\pm$ 0.050.103Allopurinol+NAD+15 $\pm$ 0.255.52 $\pm$ 1.07*0.368*Pyrazinamide + $O_2$ 153.1 $\pm$ 2.051.07 $\pm$ 0.370.006Pyrazinamide +NAD+90.7 $\pm$ 1.006.41 $\pm$ 0.940.070		6-MP+NAD <sup>+</sup>	20.80±0.41	6.62±0.64	0.318
Allopurinol+NAD <sup>+</sup> 15 $\pm$ 0.25         5.52 $\pm$ 1.07*         0.368*           Pyrazinamide + O <sub>2</sub> 153.1 $\pm$ 2.05         1.07 $\pm$ 0.37         0.006           Pyrazinamide +NAD <sup>+</sup> 90.7 $\pm$ 1.00         6.41 $\pm$ 0.94         0.070		Allopurinol + O <sub>2</sub>	16.7 ± 0.62*	$1.73\pm0.05$	0.103
Pyrazinamide + O2         153.1 ± 2.05         1.07 ± 0.37         0.006           Pyrazinamide +NAD <sup>+</sup> 90.7 ± 1.00         6.41 ± 0.94         0.070		Allopurinol+NAD <sup>+</sup>	15±0.25	5.52± 1.07*	0.368*
Pyrazinamide +NAD <sup>+</sup> 90.7 $\pm$ 1.00         6.41 $\pm$ 0.94         0.070		Pyrazinamide + O <sub>2</sub>	$153.1 \pm 2.05$	$1.07 \pm 0.37$	0.006
		Pyrazinamide +NAD <sup>+</sup>	$\textbf{90.7} \pm \textbf{1.00}$	$6.41 \pm 0.94$	0.070

 Table 40:
 Kinetic constants for trout, rat and human liver XOR substrates in spectrophotometric and HPLC assays

The values shown are the mean  $\pm$  S.D. The typical results are from three independent experiments. (\*P< 0.05 one way ANOVA are marked with asterisks) compared with the rainbow trout, \*\*n = 3. (a)-Spectrophotometric assays.

#### **3.4.4.3.** Allopurinol

The NAD-dependant Km values for allopurinol as an XDH substrate with trout was approximately 4-fold greater than that both mammalian species (Table 40). The O<sub>2</sub>-dependant Km value for allopurinol as XO substrate with Wistar rat and human are similar as in (Table 40). The NAD-dependant Vmax value for allopurinol as XDH substrate with Wistar rat and human was approximately 3 and 5-fold more than that O<sub>2</sub>-dependant Vmax value respectively (Table 40). The NAD-dependant substrate efficiency (Ks) values for allopurinol as an XDH substrate with trout was lower compared than that with Wistar rat and human respectively (Table 40). The O<sub>2</sub>-dependant substrate efficiency (Ks) values for allopurinol as an XDH substrate with trout was lower with the trout with Wistar rat and human respectively (Table 40). The O<sub>2</sub>-dependant substrate efficiency (Ks) value for allopurinol as XO substrate with human was lower than that with Wistar rat (Table 40).

#### 3.4.4.4. 6-mercaptopurine (6MP).

The NAD-dependant Km value for 6-MP as an XDH substrate with both mammalian are similar as in (Table 40). The NAD-dependant Vmax value for 6-MP with Wistar rat and human was approximately 3 and 6-fold greater than that  $O_2$ -dependant Vmax value (Table 40). The NAD and  $O_2$ - dependent substrate efficiency (Ks) value for 6-MP with Wistar rat was lower than that with human (Table 40).

#### **3.4.4.4.** 6-thioxanthine.

The NAD-dependant Vmax value for 6-thioxanthine as an XOR substrate with trout was approximately 2-fold greater than that  $O_2$ -dependant Vmax value (Table 40). The  $O_2$ -dependant substrate efficiency (Ks) value for 6-thioxanthine as an XOR substrate with trout was lower than that with NAD-dependent (Table 40).Due to time constraints activities were not measured for the mammalian species.

#### **3.4.3.4.** Pyrazinamide

With the anti-tuberculosis drug pyrazinamide no activity was found with  $O_2$  as an electron acceptor with rainbow trout cytosol. In contrast, pyrazinamide was biotransformed in the presence of  $O_2$  with the both mammalian enzymes with Km values for both Wistar rat and human being similar as in (Table 40). When NAD<sup>+</sup> was used as an electron acceptor both Wistar rat and human had approximately 4 and 6-fold enhanced Vmax respectively and the trout cytosol had measurable NAD-dependent activity. The Km value for the trout enzyme was approximately 2-fold higher than two mammalian enzymes (Table 40). The NAD-dependent substrate efficiency (Ks) value of trout with pyrazinamide as XDH substrate was found to be lower compared with that Wistar rat and human respectively (Table 40). The  $O_2$ -dependent substrate efficiency (Ks) value for both mammalians with pyrazinamide as substrate was similar (Table 40).

## **3.5.** Comparison of menadione, benzamidine and oxipurinol as molybdo-flavoenzyme inhibitors in different species.

Inhibition of AOX was determined with menadione and benzamidine, whereas inhibition of XOR was determined with oxipurinol (Figure 72). Enzyme activity was measured spectrophotometrically and inhibitor concentrations used were from 0.5  $\mu$ M to 5  $\mu$ M menadione and from 25 $\mu$ M to 100 $\mu$ M for benzamidine and oxipurinol respectively. As seen in (Figure 71a), the AOX inhibitor 5 $\mu$ M menadione strongly inhibited AOX in trout liver with approximately 80% inhibition, while with rat and human enzymes were inhibited ~ 90%. Another AOX inhibitor tested was benzamidine which has been shown to be an inhibitor of AOX in rabbit and rat (Stell et al., 1989, Maia and Mira. 2002) and has been found to be a good affinity ligand for

the purification of AOXs in rabbit, rat, mouse, bovine and chicken (Tables 21 and 22). Benzamidine inhibition is pH dependent with higher pH giving stronger inhibition. Whereas 100 $\mu$ M benzamidine inhibited rat and human AOX ~ 80% at pH 7.4 and by ~ 90% for both rat and human at pH 9.0, in contrast 100 $\mu$ M benzamidine did not inhibit trout AOX either pH (Figures 71b and 71c). The XOR inhibitor oxipurinol at 100 $\mu$ M inhibited XDH in trout, rat and human liver cytosol with ~ 60%, ~ 92% and~ 95% inhibition respectively.



Figure 71: Inhibition of trout, human and rat aldehyde oxidase by menadione and benzamidine.

(a) Menadione (b) benzamidine at pH 7.4(c) benzamidine at pH 9.0. AOX assays were carried out with 100 $\mu$ M DMAC final concentration. The bar graph shows the mean  $\pm$  S.D.(\*) signify a statistically significant difference (one way ANOVA p.  $\leq$  0.05) compared with the control. The typical results are from three independent experiments.







(a) oxipurinol. XO assays were carried out with 100 $\mu$ M xanthine final concentration (b) oxipurinol. XOR assays were carried out with 100 $\mu$ M xanthine final concentration and670 $\mu$ M NAD<sup>+</sup> final concentration as electron acceptor. The bar graph shows the mean  $\pm$  S.D. (\*) signify a statistically significant difference (one way ANOVA p.  $\leq$  0.05) compared with the control. The typical results are from three independent experiments.
## **3.5.1.** Determination of Ki for molybdo-flavoenzyme inhibitors in trout liver cytosol.

In order to determine the Ki for AOX and XOR inhibitors with the trout liver molybdo-flavoenzymes a series of experiments were conducted with menadione, ethinyl estradiol, estradiol, chlorpromazine, phenanthridine and oxipurinol with  $25\mu$ M to  $100\mu$ M substrate concentration and  $0.5\mu$ M to  $5\mu$ M inhibitor concentration except oxipurinol concentration from  $25\mu$ M to  $100\mu$ M. Lineweaver-Burk plots from this data were used to determine the inhibitor constant Ki. (Figures 73 to 84) show the effect of these inhibitors on trout liver AOX and XDH. (Table 41) summarises the inhibition constants (Ki) and mode of inhibition of different inhibitors with trout AOX and XDH.



Figure 73: Saturation kinetics plot for menadione inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu M$  to  $100\mu M$  and menadione concentration ranged from  $0.5\mu M$  to  $5\mu M$ . The typical results are from three independent experiments.



Figure 74: Lineweaver-Burk plot for menadione inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate varied from  $25\mu M$  to  $100\mu M$  and menadione concentration ranged from  $0.5\mu M$  to  $5\mu M$ . The typical results are from three independent experiments.



Figure 75: Saturation kinetics plot for estradiol inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu$ M to  $100\mu$ M and estradiol concentration ranged from  $0.5\mu$ M to  $5\mu$ M. The typical results are from three independent experiments.



Figure 76: Lineweaver-Burk plot for estradiol inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu$ M to  $100\mu$ M and estradiol concentration ranged from  $0.5\mu$ M to  $5\mu$ M. The typical results are from three independent experiments.



**Figure 77:** Saturation kinetics plot for  $17\alpha$ -ethinyl estradiol inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu$ M to  $100\mu$ M and  $17\alpha$ ethinyl estradiol concentration ranged from  $0.5\mu$ M to  $5\mu$ M. The typical results are from three independent experiments.



**Figure 78:** Lineweaver-Burk plot for  $17\alpha$ -ethinyl estradiol inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu M$  to  $100\mu M$  and  $17\alpha$ ethinyl estradiol concentration ranged from  $0.5\mu M$  to  $5\mu M$ . The typical results are from three independent experiments.



Figure 79: Saturation kinetics plot for chlorpromazine inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu$ M to  $100\mu$ M and chlorpromazine concentration ranged from  $0.5\mu$ M to  $5\mu$ M. The typical results are from three independent experiments.



**Figure 80:** Lineweaver-Burk plot for chlorpromazine inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu M$  to  $100\mu M$  and chlorpromazine concentration ranged from  $0.5\mu M$  to  $5\mu M$ . The typical results are from three independent experiments.



**Figure 81:** Saturation kinetics plot for phenanthridine inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu M$  to  $100\mu M$  and phenanthridine concentration ranged from  $0.5\mu M$  to  $5\mu M$ . The typical results are from three independent experiments.



**Figure 82:** Lineweaver-Burk plot for phenanthridine inhibition with DMAC as an AOX substrate in trout liver cytosol.

The concentration of DMAC as aldehyde oxidase substrate was varied from  $25\mu M$  to  $100\mu M$  and phenanthridine concentration ranged from  $0.5\mu M$  to  $5\mu M$ . The typical results are from three independent experiments.



**Figure 83:** Saturation kinetics plot for oxipurinol inhibition with xanthine as an XOR substrate in trout liver cytosol.

The concentration of xanthine + NAD<sup>+</sup> as xanthine dehydrogenase substrate was varied from  $25\mu M$  to  $100\mu M$  with 6.7mM NAD<sup>+</sup> and oxipurinol concentration ranged from  $25\mu M$  to  $100\mu M$ . The typical results are from three independent experiments.



**Figure 84:** Lineweaver-Burk plot for oxipurinol inhibition with xanthine as an XOR substrate in trout liver cytosol.

The concentration of xanthine + NAD<sup>+</sup> as xanthine dehydrogenase substrate was varied from  $25\mu M$  to  $100\mu M$  with 6.7mM NAD<sup>+</sup> and oxipurinol concentration ranged from  $25\mu M$  to  $100\mu M$ . The typical results are from three independent experiments.

AOX inhibitor	Ki (µM)**	Mode of inhibition
Menadione	$0.17\pm0.03$	Non-competitive
Estradiol	$0.34\pm0.05$	Uncompetitive
Ethinyl estradiol	0.44±0.05	Uncompetitive
Chlorpromazine	0.76 ±0.06	Competitive
Phenanthridine	0.77±0.07	Competitive
XOR inhibitor	Ki (µM) **	Mode of inhibition
Oxipurinol	0.37±0.06	Non-competitive

**Table 41:** Summary of inhibition constant (Ki) and mode of inhibition of different inhibitors with trout AOX and XDH.

The values shown are the (average  $\pm$  S.D<sup>\*\*</sup>), obtained from three separate liver cytosol fractions for each trout.

It was found that all AOX inhibitors had a Ki values between  $0.1\mu$ M to  $1\mu$ M with the lowest Ki value with menadione. The weakest AOX inhibitors found were phenanthridine and chlorpromazine. Lineweaver-Burk plot analysis revealed that the inhibitors had different modes of inhibition. Menadione was a non-competitive inhibitor, while estradiol and  $17\alpha$ -ethinyl estradiol were uncompetitive inhibitors. In contrast, chlorpromazine and phenanthridine were competitive inhibitors. Despite its structural similarity to XOR substrates xanthine and hypoxanthine, oxipurinol was found to be a non-competitive inhibitor.

#### **3.6.** Purification of aldehyde oxidase.

Purification of AOX from rat and fish was monitored throughout the purification procedure with DMAC as substrate. Protein elution was monitored at 280nm with an on-line spectrophotometric detection cell in the FPLC instrument. All procedures were carried out at 0 - 4°C. For preliminary experiments AOX was purified from Wistar rat as known protocols existed in published literature.

#### **3.6.1.Purification of AOX from Wistar rat liver.**

AOX was purified from Wistar rat liver with a series of purification steps as described in material and method section 2.11. (Figure 25), procedure 1.The first step of purification was heat treatment of liver cytosol at 55°C for 15 minutes and this step which gave~ 80% yield with a 2-fold purification as seen in (Table 42). The second step of purification was ammonium sulphate precipitation. 0-25%, 50-60% and 70-100% ammonium sulphate fractions had no activity. In contrast the 25-50% fraction had DMAC oxidase activity. This fraction retained 78 % yield of the initial cytosolic activity and there was an overall 3-fold purification of the enzyme. The third step of purification was Q-sepharose anion exchange chromatography which gave a 37 % yield and 12-fold purification as seen in (Table 42) and (Figure 85) and which gave two peaks. The later eluting DMAC oxidase activity peak had activity towards the both substrates DMAC and phenanthridine.



**Figure 85:** Chromatography of heat treated ammonium sulphate fractionated rat liver cytosol on a Q-sepharose column at pH 8.0. Using 20mM Tris HCL pH8.0 with 1mM cysteine and 5% (v/v) glycerol. Following elution of unbound proteins a linear 0 - 600 mM KCL gradient was applied. Fractions were collected at 1ml/minute and assayed for DMAC oxidase activity. Protein was monitored at 280 nm with an online detector. The result was typical from three independent experiments.

Following concentration using an Amicon concentrator, the peak eluting with high salt (DMAC oxidase peak 2) was further purified by affinity chromatography using benzamidine sepharose 6B as described in materials and methods and following elution of unbound proteins, AOX containing fractions were batch eluted with 10ml of 5mM benzamidine. Pooled and dialysed AOX-active fractions from this step were then subjected to chromatography on a Mono-Q 5/10 anion exchange column (Figure 86). Overall this last step in the purification gave a 22% yield and 227-fold purification of AOX from Wistar rat liver cytosol as seen in (Table 42).



**Figure 86:** Mono-Q 5/10 anion exchange chromatography column of pooled fractions from affinity chromatography step. Using 20mM Tris HCL pH8.6 with 1mM cysteine and 5% (v/v) glycerol. Following elution of unbound proteins a linear 0 - 300 mM KCL gradient was applied. Fractions were collected at 5ml/minute and assayed for DMAC oxidase activity. Protein was monitored at 280 nm with an online detector. The result from three independent experiments.

A summary of purification of Wistar rat AOX is given in (Table 42).

Purification stage	Volume [ml]	Total protein [mg]	Total activity [nmol/min]	Specific activity [nmol/min/mg protein]	Yield %	Purification fold
Cytosol preparation	36.6	7593.3	21109.4	2.78	100	1
Heat treatment at 55°C	26.9	3037.3	16887.5	5.56	80	2
25-50% Ammonium sulphate precipitation	6.8	1895.1	16676.3	8.80	79	3
Qsepharose ion exchange chromatography	1.5	235.8	7873.8	33.39	37.3	12
Benzamidine Sepharose 6B affinity chromatography	0.5	32.2	5973.9	185.15	28.3	67
MonoQ ion Exchange chromatography	0.3	7.4	4686.3	631.83	22.2	227

 Table 42: Summary of Wistar rat AOX purification.

The data shown is from a typical experiment.

### 3.6.2. SDS PAGE analysis of Wistar rat AOX

The SDS PAGE was used as described in materials and methods section 2.11.4 to analyse purified fractions figure 87 demonstrates that the purified preparations contained several polypeptide bands < 66kDa indicating that the enzyme had undergone proteolysis.



Figure 87: SDS PAGE of purified hepatic AOX from Wistar rat liver.

The lane [1] cytosol 20µg, lane [2] purified AOX 5µg from Mono Q, lane [3] molecular mass markers.

The principle and procedure of SDS PAGE and staining as described in method and material section 2.10.1 table 14. The result obtained represents a typical result from three independent purification experiments.

#### 3.6.3. Purification of AOX from rainbow trout liver.

AOX was purified from rainbow trout liver with a series of purification steps as described in materials and methods section 2.11. (Figure 25), procedure 4.In order to determine if heat treatment that is used for the initial purification of mammalian AOX (Table 21 and 22) could be used for the trout enzyme the thermal stability of the trout enzyme was tested at 55°C for 15-60 minutes as described in section 2.7.3. These experiments with the trout enzyme indicated that it was much more heat labile than the rodent or human enzyme with 80% activity lost after a brief 15 minute incubation at the standard temperature used for many mammalian AOX purifications (Figure 88) and (Table 21 and 22). This indicated that this procedure could not be used as a purification step for the piscine enzyme.





Enzyme activity was measured at 37°C, pH 7.5 with 0.1 mM DMAC as the substrate in AOX assays. The bar graph shows the AOX activity mean  $\pm$  S.D.(\*) signify a statistically significant difference (one way ANOVA p.  $\leq$  0.05) compared with the control of three experiments with pooled cytosol from ~3 animals.

Rather than use heat treatment as for the mammalian enzyme the first step used for purification of the trout enzyme was ammonium sulphate precipitations. 0-25%, 50-60% and 70-100% ammonium sulphate precipitations had no AOX activity. In contrast, the 25-50% fraction had ~90% of AOX activity of the initial cytosol and there was an overall 3-fold purification of the enzyme. The 25-50% ammonium sulphate fraction was then subjected to Q-Sepharose anion chromatography. Both buffer pH and salt gradient used were optimised in a series of experiments to obtain good resolution of the AOX-active fraction. (Figure 89) illustrates a typical chromatographic profile obtained using pH 8.6 and a 75 ml 0-1.2M KCL gradient. Following selective pooling of AOX-active fractions this gave a51-fold and 15.4% yield purification as seen in (Table 43 and Figure 89).



**Figure 89:** Chromatography of ammonium sulphate fractionated trout liver cytosol on a Q-sepharose column at pH 8.6.

Proteins were dialysed in 20mM Tris HCL pH8.6 with 1mM cysteine and 5% (v/v) glycerol and injected onto column. Following elution of unbound proteins a linear 0 - 1200 mM KCL gradient was applied. Fractions were collected at 1ml/minute and assayed for DMAC oxidase activity. Protein was monitored at 280 nm with an online detector. The result shown is representative from three independent experiments.

The third step of purification step was Superose 6 gel filtration chromatography that gave 14.6% yield and 70-fold purification as seen in (Table 43 and Figure 90).



**Figure 90:** Superose 6 16/70 gel filtration chromatography column of pooled active fractions from Q-sepharose chromatography step.

The proteins were chromatographed using 20mM Tris HCL pH8.0, 100mM NaCL with 1mM cysteine and 5% glycerol. Fraction were collected at 1ml/minute and assayed for DMAC oxidase activity. Protein was monitored at 280 nm with an online detector. The result shown is representative from three independent experiments.

The fourth step of the purification was using a Mono-Q 5/10 anion exchange chromatography of the AOX-active fraction collected from the gel filtration chromatography step. This resulted in the chromatogram in (Figure 91). This last step in the purification resulted in an overall 12% yield and 210-fold purification of AOX from rainbow trout liver cytosol as seen in (Table 43).

The summary of the purification of rainbow trout AOX is given in the (Table 43).



**Figure 91:** Mono-Q 5/10 anion exchange chromatography column of pooled active fractions from gel filtration chromatography step.

The proteins were chromatographed using 20mM Tris HCL pH 8.6 with 1mM cysteine and 5% (v/v) glycerol. Following elution of unbound proteins a linear 0 - 300 mM KCL gradient was applied. Fractions were collected at 1ml/minute and assayed for DMAC oxidase activity. Protein was monitored at 280 nm with an online detector. The result shown is representative from three independent experiments.

Purification stage	Volume [ml]	Total protein [mg]	Total activity [nmol/min]	Specific activity [nmol/min/mg protein]	Yield %	Purification fold
Cytosol preparation	27.4	1138.91	1161.7	1.02	100	1
25-50% ammonium sulphate fractionation	5.3	344.03	1073.4	3.12	92.4	3
Q-sepharose anion exchange chromatography	1.5	3.38	178.9	52.96	15.4	51
Superose 6 gel filtration chromatography	0.5	2.36	169.6	71.76	14.6	70
Mono Q anion Exchange chromatography	0.3	0.71	149.8	214.61	12.9	210

**Table 43:** Summary of rainbow trout AOX purification.

The data shown is from a single representative experiment from typical data from three independent experiments.

# **3.6.4.** Determination of the native molecular mass of rainbow trout liver AOX using Superose 6 gel filtration chromatography.

In order to determine the native molecular mass of rainbow trout liver AOX a Superose 6 chromatography column (16 x 70 cm) was calibrated using protein molecular mass standards with the same buffer used for the AOX chromatography. The elution volume of each of the standards is shown in (Table 44) and a calibration graph was constructed from this data (Figure 92).

**Table 44:** Elution volume and relative molecular mass of the proteins standards on Superose 6 16/70 gel filtration chromatography

Molecular mass marker	Elution volume [ml]	Relative molecular mass [kDa]
Thymoglobulin	65	669
Apoferritin	72.5	443
β-amylase	77.5	200
Alcohol dehydrogenase	81	150
Bovine serum albumin	85	66
Carbonic anhydrase	94.6	29



Figure 92: Calibration curve for determination of molecular mass of AOX by gel filtration chromatography.

As the elution volume of the peak of rainbow trout liver AOX activity in Superose 6 16/70 was 75ml, the native molecular mass of the fish AOX was calculated from the equation ( $y = 1E+09e^{-0.113x}$ ) derived from the calibration curve. The mass calculated was 295 kDa. This suggests that the enzyme was a homodimer of ~ 150 kDa subunits as has been shown in mammals, however the SDS-PAGE gel analysis did not have any band of this mass. Instead the final purified preparation was composed of many polypeptides less than 150 kDa. This suggests that the trout AOX retained activity in a proteolysed form. Such as phenomenon is also found with proteolysed mammalian XOR that also retains activity and an overall native mass similar to the intact enzyme estimated by gel filtration chromatography (Engerson et al., 1987).

#### **3.6.5. SDS PAGE analysis of rainbow trout AOX.**

The SDS PAGE used its described in method and material section 2.11.4 to analyse purified fractions (Figure 93) demonstrates that the purified preparations contained several polypeptide bands < 55kDa. This suggests that the enzyme had undergone significant proteolysis as the expected single subunit mass was 150 kDa.



Figure 93: SDS PAGE of purified hepatic AOX from rainbow trout liver.

The lane [1] cytosol 20µg, lane [2] purified AOX5µg from Mono Q, lane [3] molecular mass markers.

The principle and procedure of SDS PAGE and staining as described in method and material section 2.10.1 table 14. The result obtained represents a typical result from three independent purification experiments.

As well as the aforementioned methods outlined above two others were tested. In order to determine if chromatofocusing could be used as a purification step following concentration of the fraction from Q-sepharose anion exchange column, chromatofocusing was performed on a Mono P HR 5/20 column with 0.075M Tris HCL pH 9.3 and a pH 9.3 to pH 6.0 gradient was generated with 10% (v/v) polybuffer 96, adjusted to pH 6.0 with glacial acetic acid. Following chromatography fractions were assayed for DMAC oxidase activity. No activity was found in any fraction. In addition as benzamidine sepharose is frequently used as an affinity matrix for the purification of mammalian AOX this was tested with partially purified trout liver AOX. This was unsuccessful and not entirely unexpected as benzamidine was not an effective inhibitor of the trout enzyme (Figure 71).

#### 4. Discussion.

This discussion section follows the order of the results section.

# 4.1. Quantification of molybdo-flavoenzyme activities in trout, human and rat.

Both spectrophotometric and HPLC assays were used for the quantification of molybdo-flavoenzyme activities. For oxidation reactions, substrates in this study were classified into two structure types: aldehydes, which undergo oxidation to carboxylic acids and aromatic azaheterocyclic compounds, which undergo oxidation to lactams. Prior to testing with a number of substrates, the optimal conditions of the activities of both molybdo-flavoenzymes were studied in trout, human and rat using spectrophotometric assays with dimethylaminocinnamaldehyde (DMAC) for AOX and xanthine for XOR as substrates. The effect of pH on trout AOX and XOR was assayed under standard assay condition for trout, human and rat. The maximum activity of AOX and XOR with all species was achieved at a similar pH optimum of pH 7.0 as shown (figures 26 and 27). The pH optimum characteristics are similar to the pH-dependent kinetic study of XOR by Choi et al. who demonstrated that the steady-state and reductive half-reaction kinetics of the enzyme exhibits a bell-shaped profile with the  $pK_{as}$  of 6.6 and 7.4. These were assignable respectively to an active site base and the first ionization of substrate (Choi et al., 2004). In contrast, Stripe and Della Corte (1969) found that the optimum pH of rat liver XO pH 8.0 (Stirpe and Della Corte. 1969). The optimum pH for AOX from trout, human and rat was observed in the present study is similar to that found by Stoddart and Levine (1992) who found the optimal pH of rabbit AOX was 7.5. (Stoddart and Levine. 1992). In order to test the effect of temperature the activity of AOX and XOR of trout was measured at different incubation temperatures for 5 minutes as described in section

2.7.2. The rate of reaction of trout AOX and XOR gradually increased with increase temperature from 25°C up to 37°C, and the activity dramatically decreased after 45°C with less than 10% remaining at 60°C (Figure 28). In contrast the rate of reaction of human and rat AOX and XOR gradually increased with increase temperature from 25°C up to 37°C, then the activity decreased gradually until 60°C with approximately 50% activity remaining at highest temperature. These results resemble those with most animal enzymes that become denatured at temperatures above 40°C (Figure 28). This present result showed that the assay temperature of 37°C is suitable for use with the trout, human and rat MFE. The contrast in the thermolability of the piscine and mammalian AOXs may reflect differences in the primary amino acid sequence of mammalian AOXs that may have increased hydrophobic, ionic and hydrogen bonding that are known to stabilise proteins at elevated temperatures. At present it is not known which amino acid residues are involved in this.

As well as optimising incubation conditions all HPLC protocols for analytes were tested for method linearity, accuracy and precision were checked. Regression for calibration  $(r^2)$  standards injected were more than 0.997 for all analytes tested. Accuracy for the method was more than 95% and precision < 3.5%, calculated by using three injections on the same day as calibration standards (intraday) (Tables 1 and 2 in Appendixes 3). For the linearity for the results, the lower limit of Quantification (LOQ) and limit of detection (LOD) were determined for each analyte (Table 3 in Appendix 3). According to (Miller and Miller. 2005) these methods have acceptable limits for both accuracy and precision and were therefore used to analyzed samples from this study.

#### **4.1.1. Exogenous aldehydes as substrates for AOX**

Two exogenous aromatic aldehyde compounds, vanillin and (DMAC), were used as substrates of AOX. Enzyme activity of AOX was measured spectrophotometrically using two different substrates (DMAC and vanillin) with O<sub>2</sub> as electron acceptor as follows. These were chosen for the following reasons. DMAC is a specific substrate for rat AOX (Kurth and Kubiciel. 1984, Moriwaki et al., 1998, Maia and Mira. 2002). Vanillin has also been shown to be a good substrate with a high affinity for AOX (Panoutsopoulos and Beedham. 2004).

#### **4.1.1.1. Vanillin and DMAC**

Vanillin is an aromatic aldehyde found in food material as an additive and it can be biotransformed to its carboxylic metabolite with  $O_2$  as an electron acceptor primarily by AOX. In the presence of NAD<sup>+</sup>, aldehyde dehydrogenase (ALDH) plays a significant role (Beedham. 2001, Obach et al., 2004, Sahi et al., 2008, Pryde et al., 2010, Garattini and Terao. 2013). Only two studies have directly measured the kinetics of a substrates disappearance or product generation spectrophotometrically on mammals, mollusks, crustaceans, insects and fish AOX. Wurzinger and Hartenstein (1974) used vanillin as substrate of fish and mammalian AOX with  $O_2$  as an electron acceptor in a spectrophotometric assay measuring vanillic acid production (Wurzinger and Hartenstein. 1974). Using (DMAC) and vanillin substrates of fish AOX with  $O_2$ as an electron acceptor (Wurzinger and Hartenstein. 1974, Maia and Mira. 2002, Kitamura et al., 2003, Choughule et al., 2013).

 $K_m$  and  $V_{max}$  were determined from the Eadie- Hofstee plot of V versus V/[S] as described in the section materials and methods using simple rapid spectrophotometric assays. The lowest  $K_m$  value for an AOX substrate was found with DMAC being 21.5  $\pm$  1.05  $\mu$ M (mean  $\pm$  SD) with trout and 3.4  $\pm$  1.2  $\mu$ M with human and 5.4  $\pm$  1.6  $\mu$ M

with rat. In agreement with the present result Li et al. (2009) found that the DMAC is an excellent substrate for rat liver AOX with a low  $K_m$  value (9.6  $\mu M)$  (Liu et al., 2009). The results of this study indicated that DMAC is a better substrate of trout AOX than vanillin with a  $K_m$  value 2-fold higher (53.2  $\pm$  2.1  $\mu M$  ). The  $K_m$  values for DMAC in trout were approximately 4 and 7-fold higher than that in rat and human. The K<sub>m</sub> value of vanillin in the trout was approximately 5 and 10-fold higher compared with that in rat and human respectively in spectrophotometric assay and Vmax value was similar for both mammalian species (see Appendix 3 Figures 6 - 24) (Table 39). The K<sub>m</sub> value for trout AOX with vanillin was approximately two times more than that found guinea pig liver AOX that has a  $K_m$  value of 29  $\mu M$ (Panoutsopoulos et al., 2005). Obach, (2004) found vanillin is an excellent substrate for human liver AOX with a low K<sub>m</sub> of 2.7 µM (Obach. 2004). Vanillin has also been used to study species variation of aldehyde oxidase occurrence in 79 species. Wurzinger and Hartestein (1974) found vanillin to be a substrate of aldehyde oxidase from mollusks, crustaceans, insects and fish. Panoutsopoulos and Beedham (2004) incubated guinea pig AOX with vanillin and as shown by HPLC its metabolite vanillic acid formed quickly (Panoutsopoulos and Beedham. 2004). In this study trout liver, kidney and gill cytosol were incubated with vanillin and products analysed by HPLC. With trout kidney and gill cytosol no vanillic acid peak formed, while when incubated with trout liver cytosol the vanillic acid peak formed (Figure 34a). The trout kidney and gill lack AOX activity and the activity are found only in the liver. The lack of measurable AOX activity in kidney and gill contrasts that found by Krenitsky et al 1974 who measured AOX in the kidney of a Cypriniformes goldfish (Carassius auratus) and Asagba et al 2010 who found AOX activity in the kidney and gill of a Siluriforme African catfish (Clarias gariepinus). The Km of vanillin with trout liver AOX determined using HPLC was found to be  $57.20 \pm 0.6 \mu$ M, which is ~ two times more than that found with human and rat AOX (Table 39).

#### 4.1.2. Endogenous vitamin aldehydes as substrates for AOX

Two endogenous vitamin aldehydes, pyridoxal (vitamin B6), all-trans retinal (vitamin A) and the vitamin B3 metabolite  $N^1$ -methylnicotinamide (NMN) were used as aldehyde substrates of AOX.

#### **4.1.2.1. Vitamin B6 pyridoxal**

Vitamin B6 is a water-soluble compound that contains a pyridine ring that is present in nature as several different forms such as pyridoxal, pyridoxaine (PN), pyridoxamine and the active form is pyridoxal 5'-phosphate (PLP) (Fitzpatrick et al., 2007). PLP is the coenzymatically active form of vitamin B6 and plays an important role in maintaining the biochemical homeostasis of the body (Meister. 1990, Snell. 1990). There are more than 100 PLP-dependent enzymes in a cell that perform essential roles in various metabolic pathways including amino acid metabolism (such as amino acid synthesis and degradation), fatty acid metabolism (such as the synthesis of polyunsaturated fatty acids) and carbohydrate metabolism (such as the breakdown of glycogen) (Mooney et al., 2009). The preferred degradation route from PLP to 4pyridoxic acid involves the dephosphorylation of PLP by a phosphatase followed separately by the actions of AOX and  $\beta$ -nicotinamide adenosine dinucleotidedependent dehydrogenase (Schwartz and Kjeldgaard. 1951, Stanulovic and Chaykin. 1971). When pyridoxal was incubated with trout, rat and human cytosol pyridoxic acid was produced (Figures 35) with Km value for trout AOX  $~121.9 \pm 3.6 \ \mu\text{M},$  rat AOX Km value of 46.4  $\pm$  2.51  $\mu$ M and human AOX Km value of 34.1  $\pm$  1.97  $\mu$ M respectively (Table 39). The Km value of trout AOX with pyridoxal is three times

more than that with human and rat AOX activity with pyridoxal. The Vmax value of trout AOX with pyridoxal found to be  $1.41 \pm 0.69$  nmol/min/mg protein and was approximately 6 and 9-fold lower than that Wistar rat and human respectively.

#### 4.1.2.2. Vitamin A all-trans retinal

Another potential substrate of AOX of physiological importance is all-transretinaldehyde which it oxidised to retinoic acid (Haung et al., 1999). Retinoic acid is a key regulator of the homoeostasis of keratinized epithelia and a recognized morphogen of the vertebrate organisms (Chambon. 1996), although its role in fish is less well understood. All-trans retinal congeners are components of rhodopsin visual system in all vertebrate and invertebrates including fish (Lee et al., 1996, Seki et al., 1998). Interestingly the all-trans retinal determination network that is responsible for controlling eye development is one of the most extensively studied gene regulatory networks in both invertebrate and vertebrate species (Datta et al., 2011). The first report of fish AOX being involved in the biotransformation of all-trans retinal was by Lakshamanan et al. (1964) who examined the oxidation of the aldehydes of both vitamin A1 (all-trans retinal) and vitamin A2 (3, 4-dehydro-all-trans retinal) to the corresponding acids by AOX enzyme in the liver of several freshwater-fish (Ophiocephalus muralius, Labeo calbasu and Cyprinus carpio communis) as shown in (Table 10). The study also found that the enzyme from these freshwater fish livers was inhibited by the AOX inhibitor menadione (Lakshmanan et al., 1964). Although the genes and proteins that control eye development are fully understood proteins that might regulate the levels of all-trans retinal and retinoic acid in fish are poorly understood. In this study, it was important to establish if trout AOX is capable of oxidizing all-trans retinaldehyde to its metabolite retinoic acid like vertebrate AOX (Huang and Ichikawa. 1994, Haung et al., 1999, Kitamura et al., 2006, Garattini et al.,

2008, Terao et al., 2009, Garattini and Terao. 2011, Garattini and Terao. 2012). Incubation of trout cytosol with all-trans retinaldehyde generated retinoic acid (Figure 36a). This indicated that all-trans retinal is a substrate for trout AOX. These finding are consistent with Lakshamanan et al. (1964) who first observed the role of AOX in the oxidation of all-trans retinaldehyde to retinoic acid in fish by liver AOX (Lakshmanan et al., 1964) and Huang and Ichikawa, (1994) and Tsujita et al. (1994) who first observed the role of AOX in the oxidation of all-trans retinaldehyde to retinoic acid without NAD<sup>+</sup> in rabbit by liver cytosol AOX (Huang and Ichikawa. 1994, Tsujita et al., 1994). The Km value for trout, rat and human with all-trans retinal was determined using HPLC and it was found that Km value of trout AOX with alltrans retinal of 124.7  $\pm$  2.27  $\mu$ M, rat AOX Km value of 30.9  $\pm$  1.23  $\mu$ M and human AOX Km value of  $12.6 \pm 0.58 \,\mu$ M (Table 39). The Km value of trout AOX with alltrans retinal was ten and four times more than that human and rat AOX activity with all-trans retinal respectively. The Vmax value of trout AOX with all-trans retinal was  $2.22 \pm 0.62$  nmol/min/mg protein and it found to be approximately 3 and 4-fold lower than that Wistar rat and human respectively (Table 39). The Km values found in this study were of a similar magnitude to those found by Schumann et al., 2009 it found the Km value of all-trans retinaldehyde (vitamin A) with mouse AOX1 of  $55.8 \pm 8.8$ µM (Schumann et al., 2009).

## 4.1.2.3. N<sup>1</sup>-methylnicotinamide (NMN)

Another vitamin tested in this study was  $N^1$ -methylnicotinamide (NMN) which is often used as a marker for aldehyde oxidase (Felsted and Chaykin. 1967, Stanulovic and Chaykin. 1971, Beedham. 1987, Sugihara et al., 2006). NMN is formed from nicotinamide by nicotinamide N-methyltransferase which is widely distributed in animals (Yan et al., 1997). Many findings are consistent with the role of AOX in the conversion of NMN to N<sup>1</sup>-methyl-2-pyridone-5-carboxamide (2-PY) and N<sup>1</sup>-methyl-4-pyridone-5-carboxamide (4-PY) in the final steps of the nicotinamide degradation pathway (Felsted and Chaykin. 1967, Stanulovic and Chaykin. 1971, Beedham. 1987, Sugihara et al., 2006, Peretz et al., 2012). Previously Kitamura et al, (2003) examined the activity of AOX from sea bream (Pagrus major) and goldfish (Carassius auratus), while using N<sup>1</sup>-methylnicotinamide as electron donor and organothiophosphate insecticide fenthion sulphoxide as an electron acceptor as shown in (Table 10)(Kitamura et al., 2003). Rather than measure NMN metabolites Kitamura's study measured fenthion. In this study when NMN was incubated with trout, rat and human cytosol two product peaks were found (Figures 36) suggesting that there was AOX activity towards this compound. Based on what is found in mammals it is assumed that these might be  $N^1$ -methyl-2-pyridone-5-carboxamide (2-PY) and  $N^1$ -methyl-4pyridone-5-carboxamide (4-PY) (Figure 36). The metabolism of NMN to 2-PY and 4-PY has been reported to be catalysed by AOX in a number of mammals including humans (Felsted and Chaykin. 1967, Stanulovic and Chaykin. 1971, Sugihara et al., 2006). NMN showed different metabolic profiles in mice and humans by AOX in vivo and in vitro; for example the major product was 2-PY in humans, whereas in mice were 2-PY and 4-PY (Kitamura et al., 2008). The ratio of 2-PY/4-PY from NMN differs and varies among species (Kitamura et al., 2006, Garattini et al., 2008, Kitamura et al., 2008, Garattini and Terao. 2011). Krenitsky et al. (1986) reported that the N<sup>1</sup>-methylnicotinamide is substrate of human liver XO with very high Km value of 48 mM (Krenitsky et al., 1986). In contrast, Krenitsky et al. (1972) reported that Nmethylnicotinamide is an excellent substrate of rabbit liver AOX with low Km value of 1 mM (Krenitsky et al., 1986). Hall and Krenitsky (1986) reported that the Nmethylnicotinamide is a substrate of rabbit AOX with Km value of 7.9 mM (Hall and

Krenitsky 1986). In contrast, Rajagopalan and Handler (1964) incubated the N<sup>1</sup>methylnicotinamide with rabbit liver AOX and found that the Km value at pH 10 was 80  $\mu$ M and at pH 7.8 was 300  $\mu$ M respectively. These studies measured N<sup>1</sup>methynicotinamide indirectly using 2, 6-dichlorophenolindol (2, 6-DCPIP) as electron acceptor. In this study for the first time N<sup>1</sup>-methylnicotinamide oxidation by AOX was assayed in trout liver cytosol using sensitive a HPLC method as a new direct assay for evaluating the AOX activity in trout. This indicated that  $N^{1}$ methylnicotinamide is a substrate for trout AOX. The Km value of N<sup>1</sup>methylnicotinamide measured was found of 273.9  $\pm$  2.89  $\mu$ M with trout AOX, 156.2  $\pm$ 3.68  $\mu$ M with Wistar rat AOX and 139.5  $\pm$  2.86  $\mu$ M with human AOX and Vmax 1.63  $\pm$  1.33 nmol/min/mg protein, 5.92  $\pm$  0.68 nmol/min/mg protein and 6.69  $\pm$  0.74 nmol/min/mg protein with trout, rat and human respectively as in (Table 39). The Km of trout AOX with  $N^1$ -methylnicotinamide ~ 1.5-fold more than that of mammalian AOX with N<sup>1</sup>-methylnicotinamide. The Km measured in this study were of the same magnitude as that found with N<sup>1</sup>-methylnicotinamide with mouse AOX3 128.5  $\pm$  5.8 µM (Coelho et al., 2012, Mahro et al., 2013). The Vmax value of Wistar rat and human with N<sup>1</sup>-methylnicotinamide was approximately 5 and 6-fold greater respectively than the piscine AOX (Table 39).

#### 4.1.3. Exogenous N-heterocycles as substrates for AOX.

Four exogenous N-heterocycles, phenanthridine, phthalazine, pyrazinamide, cinchonine were used as N-heterocycles substrates of AOX.

#### 4.1.3.1. Phenanthridine.

Phenanthridine is an NPAH pollutant found in terrestrial and aquatic environments from industrial activities associated with fossil fuels (Blumer et al., 1977). NPAHs are therefore a cause for concern both to humans and aquatic wildlife. Several papers reported that the phenanthridine is a good substrate and has a very high affinity towards AOX (Stubley and Stell. 1980, Rashidi. 1996, Beedham. 2001). In the AOX spectrophotometric assays one N-heterocyclic compound, phenanthridine was used as a substrate. Enzyme activity of AOX was measured spectrophotometrically using phenanthridine as substrate with  $O_2$  as electron acceptor as follows. These were chosen for the following reasons, phenanthridine has been shown to be a good specific substrate for monitoring mammalian AOX activity in many studies (Lake et al., 2002, Pirouzpanah et al., 2006, Sorouraddin et al., 2008). Phenanthridine is a specific substrate of mammalian AOX (Table 1). At present it is not known whether AOX in aquatic species such as fish can metabolise phenanthridine to a more polar metabolite (phenanthridone) aiding its excretion. However, Bleeker et al, (2001) examined the metabolism of the environmental pollutant phenanthridine in carp (Cyprium carpio) liver and bile into its metabolite 6(5H)-phenanthridinone (phenanthridone) suggesting that fish AOX catalyses this biotransformation. NPAHs are derivatives of polycyclic aromatic hydrocarbons in which one carbon atom is substituted by a nitrogen atom. Phenanthridine has been identified in air, marine and freshwater environments, and in ground water. High environmental concentrations in the field result from human activities such as combustion of fossil fuels. The presence of the electronegative nitrogen atom makes NPAHs more water soluble than their homocyclic analogues; this increase their mobility and bioavailability in aquatic environments (Bleeker et al., 2001). The activity with phenanthridine was not detectable with trout liver cytosol using the standard spectrophotometric assay. It was a good substrate with both rat and human in my study and with a Km value of  $5.6 \pm 0.65 \mu$ M with rat and Km value of  $3.2 \pm 1.23\mu$ M with human. This agreement with phenanthridine as a specific substrate of AOX has been used in several papers and Km previously has been estimated were < 1  $\mu$ M with rabbit and guinea pig liver AOX enzyme and 6  $\mu$ M and 14  $\mu$ M with rat and human liver AOX respectively (Stubley and Stell. 1980, Rashidi et al., 1997, Beedham. 2001). In this study trout liver, kidney and gill cytosol incubated with phenanthridine and product analysed by HPLC, with trout kidney and gill cytosol no 6(5H)-phenanthridinone (phenanthridone) metabolite peak found. This indicated no AOX activity in trout kidney and gill, while in trout liver cytosol the 6(5H)-phenanthridinone (phenanthridone) metabolite formed (Figure 39a) with a Km value of 55.0 ± 1.63  $\mu$ M. The Km values of trout AOX with phenanthridine it found seven and six times more than that human and rat AOX activity respectively. The Vmax value of Wistar rat and human with phenanthridine was approximately 2 and 4 times respectively more than that with trout AOX (Table 38).In contrast, Hartmann et al., (2012) it found the Km value of phenanthridine with pure human AOX of 3.9 ± 0.8  $\mu$ M. On other hand, Coelho et al., (2012) and Mahro et al., (2013) it found the Km value of phenanthridine with mouse AOX3 32.3 ± 1.4  $\mu$ M and 32 ± 1  $\mu$ M respectively (Coelho et al., 2012, Mahro et al., 2013).

#### 4.1.3.2. Phthalazine.

Phthalazine an aromatic heterocyclic compounds which have a CH=N- group, which its biotransformed to its corresponding metabolite 1-phthalazinone by mammalian AOX (Stubley et al., 1979, Beedham et al., 1990, Obach. 2004, Panoutsopoulos and Beedham. 2004, Barr and Jones. 2011). In this study phthalazine was incubated with trout, rat and human liver cytosol and its metabolite was formed (Figure 40). The Km value determined for trout, rat and human AOX with phthalazine was  $61.2 \pm 1.41 \mu$ M,  $10.5 \pm 0.55 \mu$ M and  $9.80 \pm 0.59 \mu$ M respectively (Table 39). The Km value for trout AOX activity was therefore six and two times more than that human and rat AOX respectively. The Vmax value of trout AOX with phthalazine found to be  $2.02 \pm 0.46$  nmol/min/mg protein and was approximately 2 and 4 times lower than Wistar rat and human respectively (Table 39).study by Beedham et al., (1995) which incubated human, rabbit, guinea pig and baboon liver cytosol with phthalazine, determined the Km for human AOX was41µM (Beedham et al., 1995). This result is similar with trout and was ~ 4-fold higher than that with rat and human in this study. Barr and Jones (2011 and 2013) incubated human liver cytosol with phthalazine, identified its metabolite formed 1-phthalazinone using HPLC and determined the Km of human AOX with phthalazine and it found it to be  $8.0 \pm 0.4 \mu M$  (Barr and Jones. 2011, Barr and Jones. 2013). This result is similar what was found with human cytosol in this study. On other hand Hartmann et al., (2012) measured the activity of phthalazine spectrophotometrically with pure human AOX using 2,6-dichlorophenol-indophenol (2,6-DCIP) as electron acceptor and it found the Km value of phthalazine with human AOX of  $1.3 \pm 0.3 \mu$ M; this result was ~ 10-fold lower than human in this study due to use of an artificial electron transport. On other hand Schumann et al., 2009 found the Km value of phthalazine with mouse AOX1 of  $11.4 \pm 4.0 \,\mu$ M (Schumann et al., 2009) and this agrees with the Km value of phthalazine with human in this study. In contrast Coelho et al., 2012 and Mahro et al., 2013 found the Km value of phthalazine with mouse AOX3 was  $1.4 \pm 0.2 \,\mu$ M (Coelho et al., 2012, Mahro et al., 2013); this result was ~ 10-fold lower than human and rat in this study.

#### 4.1.3.3. Cinchonine.

Cinchonine is a cinchona alkaloid a widely used in the treatment of malaria (Hunter. 1988). All four cinchona alkaloids, quinine, quinidine, cinchonine and cinchonidine, are oxidised by AOX to the 2'-quinolone derivatives in mammals (Palmer et al., 1969, Stubley et al., 1979, Beedham et al., 1992, Zhao and Ishizaki. 1997, Al-Tayib. 2009). When cinchonine was incubated with trout, rat and human cytosol a metabolite of

cinchonine was produced (Figures 41) suggesting AOX plays an role in the metabolism of cinchonine in these species. The Km value of cinchonine with rainbow trout AOX was ~ 4-fold higher than that of two mammalian AOXs (Table 39).The Vmax value of Wistar rat and human AOX with cinchonine was approximately 4 and 8-fold greater than that of trout AOX (Table 39). A study by Fukiya et al., 2010, found the Km value of cinchonidine with rabbit of  $124 \pm 2.44 \mu M$  (Fukiya et al., 2010) this was similar to that found with trout and was ~ 3-fold higher than that found with rat and human AOX in this study.

#### 4.1.3.4. Pyrazinamide.

Pyrazinamide is a drug used to treat tuberculosis that is metabolised to 5hydroxypyrazinamide by the action of XOR and AOX in mammals (Moriwaki et al., 1993, Mehmedagic et al., 1997, Mehmedagic et al., 2002, Kitamura et al., 2006). When pyrazinamide was incubated with trout in the presence of  $O_2$  as an electron acceptor there was no production of 5-hydroxypyrazinamide (Figure 42a), but when pyrazinamide was incubated with trout in the presence of  $NAD^+$ , 5hydroxypyrazinamide was produced (Figures 42b). This was in contrast with that found when pyrazinamide was incubated with both rat and human liver cytosol as 5hydroxy pyrazinamide was formed both in presence of NAD<sup>+</sup> and without of NAD<sup>+</sup>(Figures 44a, 44b, 45a and 45b). The Km value for the NAD<sup>+</sup>-dependent trout enzyme was approximately 2-fold higher than that two mammalian enzyme (Table 40). With the addition of  $NAD^+$  to the assay the Vmax of the rat and human activities were increased about 4 and 6-fold respectively than when oxygen was the sole electron acceptor. This result was consistent with finding in previous in vitro study, which confirmed that both AOX and XOR can oxidise pyrazinamide to 5hydroxypyrazinamide in humans and rats (Yamamoto et al., 1987, Moriwaki et al.,

1993). As in (table 38) the AOX substrate specificity is different from different substrate with the same species as well as with different species tested in this study. For more affinity of substrate to trout AOX found DMAC the best followed with vanillin and the lowest affinity found with N<sup>1</sup>-methylnicotinamide as it have high Km value and lower substrate efficiency (Ks) value. On other hand, the substrate specificity with two mammalian species rat and human AOX have high specificity with DMAC and vanillin with lowest Km value and highest substrate efficiency (Ks) value and lowest specificity with N<sup>1</sup>-methylnicotinamide as high Km value and lowest specificity with N<sup>1</sup>-methylnicotinamide as high Km value and lowest substrate efficiency (Ks) value. These different in specificity due to differentiation in amino acid sequence with different species as mentioned with (Pryde et al., 2010) it used modelling of different AOX substrates to expect its reaction with enzyme and measured by using X-ray crystallography to find the activity of different substrates with the AOX.

## **4.1.4.** Endogenous and exogenous purines and purine analogs as substrates for XOR

#### 4.1.4.1. Xanthine.

Kinetic analyses of XOR enzymes revealed that the endogenous purine xanthine is the preferred substrate for XOR(Moriwaki et al., 1993). In this study using the standard spectrophotometric assay the Km value of rat and human for XO with xanthine using  $O_2$  as an electron acceptor was  $13.2 \pm 1.7 \mu$ M (mean  $\pm$  SD) and  $3.0 \pm 0.65 \mu$ M (mean  $\pm$  SD), respectively, while the Km value of rat and human for XDH with xanthine using NAD<sup>+</sup> as an electron acceptor was  $5.5 \pm 1.45 \mu$ M (mean  $\pm$  SD) and  $1.9 \pm 0.2 \mu$ M (mean  $\pm$  SD), respectively. The NAD-dependant Vmax value for xanthine dehydrogenase with Wistar rat and human was approximately 5 and 7-fold greater than that O<sub>2</sub>-dependant respectively (Table 40). In contrast the trout XOR enzyme
only displayed activity in the presence of NAD<sup>+</sup> as an electron acceptor as no activity with O<sub>2</sub> was detected even after prolonged 120 minute incubations using a sensitive HPLC assay (Figure 48 b). The Km value of trout for XDH with xanthine using NAD<sup>+</sup> as an electron acceptor was 2-3-fold higher than the mammalian enzymes (13.5  $\pm$  1.9  $\mu$ M (mean  $\pm$  SD) (Table 40). This Km value is similar to that obtained by Kaminski and Jezewska (1985) who found that the Km of the carp (Cyprinus carpio) enzyme was 24.7  $\mu$ M. The Km value obtained with the piscine enzyme using xanthine as a XDH substrate was about 2.5 and 6-fold higher than those obtained with rat and human respectively. The activity of XOR in the three species was inhibited by the XOR inhibitor oxipurinol (Figures 54, 56 and 58). These finding agree with Kaminski and Jezewska, (1985) who observed that XDH of carp (Cyprinus carpio) converted xanthine to uric acid only with NAD<sup>+</sup> present and reported that allopurinol is a potent inhibitor of carp (Cyprinus carpio) XOR (Kamiński and Jeżewska. 1985). On other hand, Schumann et al., (2009) examined the activity of xanthine with mouse XDH and it measured its Km value and it found of  $59 \pm 10 \ \mu M$  (Schumann et al., 2009)this result was ~ 4-fold greater than trout and was~ 10-fold greater than rat and was ~ 25fold higher than human in this study. The Km value of mouse XDH was ~ 10-fold greater than rat which may be due to species variation in molybdo-flavoenzyme activity. Krenitsky et al. (1986) reported that the xanthine is an substrate of human liver XO that have low Km value of 7.0 µM (Krenitsky et al., 1986). This result is similar to that with found with human in this study.

The lack of detectable XO activity is similar to that found by other investigators with fish XOR (Krenitsky et al., 1974, Wurzinger and Hartenstein. 1974, Cleare et al., 1976, Kamiński and Jeżewska. 1985). However in their assays a much less sensitive spectrophotometric assay was used which might not detect XO activity. The current study which used a 120 minute incubation and a sensitive HPLC assay therefore unequivocally proves no XO activity exists with trout liver XOR. These results contrast with these by a number of investigators that have seemingly measured XO activity (Sastry and Rao. 1984, Basha and Rani. 2003, Pandey et al., 2003, Asagba et al., 2010, Hegazi et al., 2010, Hari and Neeraja. 2012). In these studies where XO has been apparently measured it important to note that either crude homogenates or crude post nuclear or mitochondrial supernatants were used and additionally gel filtered was not carried out to remove small molecules that may interfere with the assay. This would therefore result in erroneous measurement of XOR activities. As the study describe in this thesis and other studies have used cytosol that has been gel filtered do not detect any activity with O<sub>2</sub> as an electron acceptor it is likely that studies that detect XO with crude non-gel filtered extracts gave false positive results. This is backed up by the fact that all studies that have used the correct procedure with nonmammalian species have consistently not found measurable XO activity. These included fish, amphibians, reptiles and birds where the enzyme exists as a nonconvertible XDH form (Kamiński and Jeżewska. 1985, Frost et al., 1986, Zakrzewska and Jeźewska. 1989, Nishino et al., 2008). In addition XORs in non-mammalian species lack the cysteines required for the XO conversion. Protein sequence information from chicken and Drosophila XOR revealed that these NAD-dependant XDHs lack a cysteines (cysteine 535 and cysteine 992) necessary for this XO conversion (Nishino and Nishino. 1997, Nishino et al., 2008). Similarly fish XORs also lack these cysteines (D. J. Clarke personal communication).

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### 4.1.4.2. 1-methylxanthine.

Previous studies with bovine milk xanthine oxidase demonstrated it catalyses the oxidation of 1-methylxanthine, 7-methylxanthine and theophylline, to varying degrees with the 1-methylxanthine being the best substrate (Krenitsky et al., 1972). In this study the caffeine metabolite 1-methylxanthine was tested as a substrate for trout, rat and human XOR. Trout cytosolic XOR was also able to convert 1-methylxanthine to 1-methyluric acid when NAD<sup>+</sup> was present as an electron acceptor (Figure 49b), but no product was formed without NAD<sup>+</sup> (Figure 49a). The Km value of trout for XDH with 1-methylxanthine using NAD<sup>+</sup> as an electron acceptor was  $17.6 \pm 1.23 \mu M$  (mean  $\pm$  SD) (Table 40). The Km value obtained with the piscine enzymes using 1methylxanthine as XDH substrate was about 2-fold higher than those obtained with human and rat respectively (Table40). The NAD-dependant Vmax value for 1methylxanthine dehydrogenase with Wistar rat and human was approximately 5 and 8-fold greater respectively than the O<sub>2</sub>-dependant (Table 40). The Km value of rat and human XOR in my study with 1-methylxanthine using NAD<sup>+</sup> as an electron acceptor were similar being 9.2  $\pm$  0.62  $\mu$ M (mean  $\pm$  SD) and 8.6  $\pm$  1.05  $\mu$ M (mean  $\pm$  SD), respectively (Table 40). 1-methylxanthine oxidation was inhibited in the presence of XOR inhibitor oxipurinol in all three species tested (Figures 55a, 57a and 59a). These results agree with Reinke et al. (1987) who found the XDH of rat can catalyse the oxidation of 1-methylxanthine to 1-methyluric acid (Reinke et al., 1987).

### 4.1.4.3. Allopurinol

The exogenous purine, allopurinol, is a prodrug that is metabolised to oxipurinol by the action of both AOX and XOR (Reiter et al., 1990, Ichida et al., 1998, Pacher et al., 2006, Yu et al., 2009). When allopurinol was incubated with trout liver cytosol with  $O_2$  as an electron acceptor no activity was detected indicating it was not a

substrate for AOX or XO in this species. Similar to the two xanthine substrates when NAD<sup>+</sup> was included in the incubation allopurinol was biotransformed to oxipurinol indicating that trout XDH was capable of this biotransformation. In contrast when the two mammalian liver cytosols were incubated with allopurinol, oxipurinol was formed with O<sub>2</sub> and NAD<sup>+</sup> (Figures 67 and 68) indicating AOX/XO and XDHs involvement in this drug biotransformation as has been previously shown by others (Reiter et al., 1990, Yamamoto et al., 1991, Moriwaki et al., 1993). The Km value of allopurinol with trout cytosol in presence of NAD<sup>+</sup> as an electron acceptor was  $63.2 \pm 0.98 \ \mu M$ with a Vmax value of  $2.06 \pm 0.75$  nmol/min/mg protein, in contrast, the rat and human liver cytosol metabolise allopurinol as XO in presence of O<sub>2</sub> as an electron acceptor and with Km value of 20.9  $\pm$  0.76  $\mu M$  and 16.7  $\pm$  0.05  $\mu M$  respectively, and as XDH in presence of NAD<sup>+</sup> as an electron acceptor and with Km value of  $15.30 \pm 1.05 \,\mu$ M and  $15.0 \pm 0.25 \,\mu$ M respectively with rat and human. The Km value of trout XDH was four times more than that of rat and human XDH activity (Table 40). The NADdependant Vmax value for allopurinol as XDH substrate with Wistar rat and human was approximately 3 and 5-fold respectively more than that O<sub>2</sub>-dependant Vmax value(Table 40). The Km results are in higher than that reported by Krenitsky et al. (1986) who reported that the allopurinol is an substrate of human liver XO with a low Km value of 1.7 µM (Krenitsky et al., 1986).

### 4.1.4.4. 6-mercaptopurine (6MP)

In the present study, the anti-cancer drug 6-mercaptopurine (6-MP) was incubated with the cytosolic fraction of trout, rat and human. Both rat and human cytosolic fraction biotransformed 6-MP directly to 6-thiouric acid in the presence of  $NAD^+$  or without  $NAD^+$  (Figures 62a, 62b, 64a and 64b) and this activity was inhibited by the XOR inhibitor oxipurinol (Figures 63a and 65a) but was not inhibited by menadione an AOX inhibitor (Figures 63b and 65b). This suggests XOR was the major enzyme responsible for 6-MP to 6-thiouric acid conversion in the two mammalian species. The NAD-dependant Km value for 6-MP as an XDH substrate with both mammalian are similar as in (Table 40). The NAD-dependant Vmax value for 6-MP with Wistar rat and human was approximately 3 and 6-fold greater than that O<sub>2</sub>-dependant Vmax value (Table 40).In contrast no metabolites were formed when 6-MP was incubated with trout liver cytosol with or without NAD<sup>+</sup> (Figure 60) indicating this was not a substrate for trout AOX or XOR.

In order to determine if the structurally related metabolite of 6-MP, 6-thioxanthine was also not a substrate for the trout enzyme; 6-thioxanthine was incubated with trout liver cytosol with and without NAD<sup>+</sup>. 6-thiouric acid was formed with both  $O_2$  and NAD<sup>+</sup> as electron acceptors, with enhanced activity with NAD<sup>+</sup> (Figure 66). This activity was inhibited by the XOR inhibitor oxipurinol but not by the AOX inhibitor menadione (Figure 67). The NAD-dependant Vmax value for 6-thioxanthine as an XOR substrate with trout was approximately 3-fold greater than that  $O_2$ -dependant Vmax value (Table 40). Due to time constraints 6-thioxanthine was not used as a substrate for the mammalian species. The substrate specificity and species differences between 6-mercaptopurine and 6-thioxanthine in trout, rat and human XOR may be due to amino acid differences at the active site of the enzymes.

## **4.2.** Determination of kinetic parameters for aldehyde oxidase and xanthine oxidoreductase inhibitors for trout, rat and human.

In this study preliminary experiments were performed with three well documented inhibitors of AOX and XOR. In this study the AOX inhibitor benzamidine did not have any effect with trout liver cytosol at two pH 7.4 and 9.0 as seen in (Figures 71a and 71b), while with human and rat liver cytosol it resulted in 80-90% inhibition using 100 $\mu$ M benzamidine. When the prototypical AOX inhibitor menadione was used the AOX activity was inhibited 80-92% in the three species tested using a 5  $\mu$ M final concentration (Figure 70). The XOR inhibitor oxipurinol inhibited the two mammalian XORs 90-95% whereas the trout XOR was inhibited ~ 60% with 100 $\mu$ M final concentration of this inhibitor (Figure 72).

In order to further understand the mode and kinetics of inhibition with trout AOX and XOR this was scrutinised with six different inhibitors and the mode of inhibition and Ki determined using Lineweaver-Burk plots. Menadione had a Ki of  $0.17 \pm 0.03\mu$ M and mode of inhibition non-competitive inhibition as seen in (Figures 73 and 74) and (Table 41). The inhibition effect of estradiol and  $17\alpha$ -ethinyl estradiol as inhibitors with trout AOX was uncompetitive with Ki values of  $0.34 \pm 0.05 \mu$ M and  $0.44 \pm 0.05 \mu$ M respectively (Figures 75, 76, 77 and 78). The mode of inhibition and Ki value of chlorpromazine as AOX inhibitor with trout was also tested and a Ki value of  $0.76 \pm 0.06 \mu$ M was obtained and mode of inhibition as competitive (Figures 79 and 80), the environmental pollutant NPAH phenanthridine was found to be competitive inhibitor of trout AOX with a Ki value of  $0.77 \pm 0.7 \mu$ M as in (Figures 81 and 82) which is 3-fold higher than that found with bovine AOX (Table 5). When the XOR inhibitor oxipurinol was tested with trout XOR the mode of inhibition was non-competitive

inhibition with Ki value of 0.37  $\pm$  0.06  $\mu M$  as in (Figures 84 and 85) as similar that it found with bovine in introduction (Table 6).

In conclusion all six different inhibitors tested displayed similar modes of inhibition in trout to that found in mammals (Table 4, 5 and 6).

# **4.3.** Purification of AOX from rainbow trout and Wistar rat liver cytosol.

Prior to the purification of AOX from rainbow trout and Wistar rat liver a literature survey was carried out to establish the best methodologies to use to purify AOX.

AOX has been purified from many different species with different methodologies as summarised in the (Tables 21 and 22). Reviewing the literature on the subject reveals that the initial step of purification of AOX is generally subcellular fractionation to generate cytosol where the enzyme resides. A popular subsequent purification method used for AOX is heat precipitation as mammalian AOXs are thermotolerant. In some purification protocols buffers used to purify AOX contain dithiothreitol and cysteine to protect proteins containing sulfhydryl groups. (Rajagopalan and Handler. 1967, Felsted et al., 1973, Andres. 1976, MacGibbon et al., 1979, Yoshihara and Tatsumi. 1985, Stell et al., 1989, Hughes et al., 1994, Calzei et al., 1995, Turner et al., 1995, McManaman et al., 1996, Yoshihara and Tatsumi. 1997, Sugihara et al., 1999, Terao et al., 2001, Maia and Mira. 2002, Kitamura et al., 2003, Kurosaki et al., 2004, Vila et al., 2004, Terao et al., 2006, Terao et al., 2009, Mahro et al., 2011). Prior to initiating experiments to purify AOX from the piscine species, in order to check procedures were operating successfully initially AOX was purified from Wistar rat liver cytosol using procedure 1 in materials and methods section (Figure 25). Heat treatment, Qanion exchange chromatography, affinity chromatography sepharose using benzamidine 6B Sepharose column and MonoQ anion exchange chromatography successfully gave a 227-fold purification and a 22 % yield. The fold purification was about 4-fold lower than Maia and Mira (2002) found. When AOX was purified from rainbow trout a different procedure was used, as the piscine enzyme was more heat liable than that rodent enzyme with 80% activity lost after a brief 15 minute incubation at the 55°C standard temperature used for mammalian AOX purification (Tables 21 and 22 and Figure 88). The heat treatment step was therefore omitted from the procedure for purification of AOX from rainbow trout. In addition the affinity chromatography step was also not useful for purification of AOX from rainbow trout, due to the benzamidine not being an effective inhibitor of rainbow trout AOX (Figure 71). The alternative procedure used to purify the AOX from rainbow trout liver started with the second step of the purification of AOX from rat liver, this was ammonium sulphate precipitation that successfully purified the piscine enzyme 3-fold. Subsequently a series of chromatography steps were carried out including Qsepharose anion exchange chromatography, gel filtration chromatography on Superose 6 16/70 and the final step was MonoQ anion exchange chromatography. This resulted in a 210-fold purification and 12.9 % yield. The native molecular mass of rainbow trout AOX estimated using gel filtration was found to be similar to its mammalian, avian and insect counterparts being ~ 295 kDa (Table 22). SDS PAGE indicated the trout AOX had undergone extensive proteolysis as several low molecular mass polypeptides were apparent at less than the expected monomeric mass of the enzyme of 150 kDa. This result is in agreement with other researchers that purified AOX from another fish the Sea bream (Pagrus major) that also suggested proteolysis had occurred during purification (Kitamura et al., 2003). Despite this the procedure used in this laboratory had a much enhanced success compared with Kitamura et al. (2003) in that the partially purified AOX from Sea bream liver was only purified 33-fold with a 5.5 % yield. A possible solution to the problems of purification of piscine AOX might be to add a protease inhibitors such as phenylmethylsulfonyl fluoride (PMSF).

Another solution to characterising purified AOX in fish is to use a recombinant protein expression system in which piscine AOX cDNA is expressed in bacteria and subsequently purified using a simple protocol that would avoid the proteases that are found in animal liver. Such a system has been used successfully to study mouse, monkey, rat and human AOX (Kane et al., 2006, Schumann et al., 2009).

#### 4.4. Conclusions.

This study aimed to determine whether a fish rainbow trout (Oncorhynchus mykiss) can biotransform AOX and XOR substrates like their mammalian counterparts. To date there has been no study of AOX in any Salmoniforme despite their commercial and scientific importance.

Rainbow trout AOX and XOR activities were tested in liver, kidney and gill cytosol and it was found exclusively in the liver using HPLC and spectrophotometric assays.

Hepatic cytosols of rainbow trout, human and rat were able to catalyse the oxidation of a range of MFE substrates to varying degrees. These included the exogenous aldehydes (DMAC and vanillin), endogenous vitamin substrates (all-trans retinal, pyridoxal and N<sup>1</sup>-methylnicotinamide), NPAH environmental pollutants (phenanthridine, phthalazine and cinchonine), and drugs (allopurinol and pyrazinamide). The exogenous purine and purine analogues xanthine, 1methylxanthine, 6-mercaptopurine, 6-thioxanthine were also biotransformed by XOR in all species tested (Table 1). The study also demonstrated that the XOR activity of rainbow trout was exclusively NAD<sup>+</sup>-dependant activity unlike that in the two mammalian species tested (Table 40). When Eadie-Hofstee plots were used to determine kinetic values of rainbow trout MFE activities with different substrates and compared with human and rat and it was found that the Km was higher and the Vmax lower in rainbow trout. This study therefore demonstrated that a piscine AOX has the ability to metabolise several NPAH environmental pollutants and is capable of the oxidative biotransformation of several endogenous vitamin substrates like its mammalian counterparts. With regards to the physiological role of NPAH biotransformation by AOX this is an important detoxication mechanism as several studies in vivo and ex vivo have demonstrated that NPAHs are harmful. AOX

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substrates phenanthridine, phthalazine, isoquinoline and acridine together with several benzoacridine and dibenzoacridine have been shown to be cytotoxic in cultured cell lines such as H4IIE rat hepatoma and mouse embryonic carcinoma P19/A15 cell lines (Sovadinova et al., 2006, Beníšek et al., 2011). Bleeker et al (1999) also demonstrated that NPAH compounds acridine, benzo[c]acridine, benzo[a]acridine, quinoline, benzo[f]quinoline, benzo[h]quinoline and phenanthridine were toxic to Chironomus riparius (midge) (Bleeker et al., 1999). Benzoquinoline-diones have also been shown to be teratogenic to cricket embryos resulting in extra heads and legs (Walton et al., 1983). Many of the harmful effects of NPAHs have also been documented in fish species. Baylock et al., (1985) found that acridine was toxic to fathead minnows (Pimephales promelas) (Blaylock et al., 1985), whilst acridine and quinoline have also been shown to cause embryotoxicity in rainbow trout (Salmo gairdneri=Oncorhynchus mykiss) (Black et al., 1983), largemouth bass (Micropterus salmoides) (Black et al., 1983) and zebrafish (Danio rerio) (Peddinghaus et al., 2012). As well as in vivo studies ex vivo work using the fish hepatoma PLHC-1 cell line (derived from Poeciliopsis lucida) has also shown that benzo[a]acridine and dibenz[a, h]acridine are cytotoxic (Jung et al., 2001). Very recently research utilising the rainbow trout liver cell line RTL-W1 has shown that acridine is genotoxic in micronucleus assays (Brinkmann et al., 2014). With regards compounds that interfere with AOX biotransformation several compounds were tested in these studies that have physiological importance. Endogenous steroids (estradiol) and drugs (menadione and chlorpromazine) and exogenous environmental pollutants (phenanthridine and 17aethinyl estradiol) were inhibitors of the piscine AOX enzyme. These results may therefore have important physiological consequences for fish that live in an aquatic environment exposed to pollutants such as NPAHs, estrogens and drugs(Furlong and Carpenter. 1982, Jobling et al., 1996, Schultz et al., 2001, Schultz et al., 2003, Vulliet and Cren-Olivé. 2011, Brulik et al., 2013)as it may have effects on vitamin A, B3 or B6 metabolism (Spinneker et al., 2007, Van der Sluijs et al., 2011). With regards compounds that effect XOR biotransformation the compounds tested were of less physiological importance to a fish, but the results reveal species similarities and differences exist. Whereas oxipurinol was a potent inhibitor of XOR in all species tested, in contrast to the mammalian AOXs benzamidine did not inhibit the piscine AOX.

In order to further characterise trout AOX protein purification was carried out. In contrast to mammalian AOX, the piscine enzyme was not thermotolerant at 55°C nor was it inhibited by benzamidine, thus heat treatment and affinity chromatography could not be used as a purification steps. Trout AOX was purified 210-fold using ammonium sulphate fractionation, together with ion exchange and gel filtration chromatography. The native molecular mass of the piscine AOX was 295 kDa, which is similar to mammalian AOXs.

In conclusion this study yields new insight into groups of anthropogenic environmental pollutants, drugs and vitamins that are substrates and inhibitors of piscine AOX that is an ancestral vertebrate AOX.

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## 4.5. Recommendations for future work.

This study could be expanded to analyse a wider range of xenobiotics including drugs that are potential piscine MFE substrates. A particular focus could be NPAHs which are well documented aquatic environmental pollutants (Blumer et al., 1977, Balch et al., 1995, Osborne et al., 1997, de Voogt and Laane. 2009) that are cytotoxic, embryotoxic, teratogenic, genotoxic and carcinogenic (Southworth et al., 1979, Balch et al., 1995, Jung et al., 2001, Barron et al., 2004, Peddinghaus et al., 2012, Brinkmann et al., 2014).

Another set of compounds that could be tested are MFEs reductase substrates such as acetophenone oxime, salicylaldoxime, benzamidoxime, aromatic and heterocyclic hydroxamic acids, fenthion sulfoxide, sulindac sulfoxide and zonisamide (Table 2) some of which are found in aquatic environment.

Another area that could be researched is the induction of MFEs by NPAHs.

Although cytochrome P450 have been well documented to be induced by PAHs including NPAHs in mammals (Sovadinova et al., 2006, Beníšek et al., 2011) the effect of such compounds on MFEs have received scant attention. There have been the only two studies with two environmental polycyclic aromatic hydrocarbons (PAHs) demonstrating that such compounds can induce MFEs. Johnson et al., (1984) found that oral administration of the NPAH phthalazine (50mg/kg/day) to female rabbits caused an 2 to 3-fold increase in the activity of the hepatic AOX and XOR, whereas there was no effect on microsomal cytochrome P-450 activity (Johnson et al., 1984). Sugihara et al., (2001) demonstrated that 2, 3, 7, 8-tetrachlorodibenzo-p-dioxin (TCDD) induced hepatic XOR and AOX in liver of mice about 3-fold and 1.5-fold respectively (Sugihara et al., 2001). This suggests that induction of MFEs in fish

might be a useful monitor for NPAH pollution to complement the use of CYP1A induction in environmental monitoring (Stagg et al., 2000, Basu et al., 2001, Tom et al., 2001). In order to determine the extent of NPAHs induction of MFEs in fish, investigations could be conducted by treating fish in laboratory conditions with a range of NPAHs to determine if MFE induction occurs. MFEs could also be measured in fish from polluted environments to determine if there is a correlation between NPAHs levels and MFEs activity.

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## 6.0. Appendices.

## 6.1. Appendix 1.

Table 1: Fish	1 species	order	and	l family.	

Common name	Scientific name	Order	Family	Habitat
African Catfish	Clarias Gariepinus	Siluriformes	Clariidae	F/P
Blue Fish	Pomatomus Saltatrix	Perciformes	Pomatomidae	S/D
Bluegill	Lepomis Macrochirus	Perciformes	Centrarchidae	F/D
Brill	Scophthalmus Rhombus	Pleuronectiformes	Scophthalmidae	S/D
Black Sole	Solea Solea	Pleuronectiformes	Soleidae	S/D
Brown Trout	Salmo Trutta	Salmoniformes	Salmonidae	F/P
Carp	Cyprium Carpio	Cypriniformes	Cyprinidae	F/P
Cutlips Minnow	Exoglossum Maxilingua	Cypriniformes	Cyprinidae	F/P
Common Carp	Cyprinus Carpio Communis	Cypriniformes	Cyprinidae	F/D
Channel Catfish	Icatalurus Punctatus	Siluriformes	Ictaluridae	F/P
Cod	Gadus Morhua	Gadiformes	Gadidae	S/D
Dab	Limanda Limanda	Pleuronectiformes	Pleuronectidae	S/D
Electric Catfish	Malapterurus Electricus	Siluriformes	Malapteruridae	F/D
Fathead Minnow	Pimephales Promelas	Cypriniformes	Cvprinidae	F/P
Florida Pompano	Trachinotus Carolinus	Perciformes	Carangidae	S/P
Goldfish	Carassius Auratus	Cypriniformes	Cyprinidae	F/P
Great Snakehead	Ophiocephalus Muralius	Perciformes	Channidae	F/P
Gurnard	Chelidonichthys Cuculus	Scornaeniformes	Triglidae	S/D
Horring	Clupea Harangua	Cluneiformes	Clupeidae	S/D S/D
Herfing	Malanogrammus Acclefic	Codiformere	Cadidae	3/r
Haddock	Melanogrammus Aeglennus	Gadiformes	Gadidae	S/D
John Dory	Zeus Faber	Denoiformes	Contronshidoo	5/P
Largemouth Bass	Micropterus Salmoides	Canabarbinifarman	Centrarchidae	F/P
Lesser Spotted Doglish	Microstomus Kitt	Disuronactiformes	Blouropostidos	5/P
Lemon Sole	Microstomus Kitt	Cilerife mese	Letaluarida	5/D
Margined Madlom	Noturus Insignis	Baraiformas	Ciablidae	F/P
Mozambique Thapia		Perchonnes		F/P
Mackerel	Scomber Scombrus	Perciformes	Scombridoe	S/P
Megrim	Lepidorhombus Whiffagonis	Pleuronectiformes	Scophthalmidae	S/D
Northern Hog Sucker	Hypentelium Nigricans	Cypriniformes	Catostomidae	F/D
Norfolk Spot	Leiostomus Xanthurus	Perciformes	Sciaenidae	5/P
Northern Pike	Esox Lucius	Esocitormes Densiferrates	Cishlidae	F/D
Nile Thapla Juvennes	Ureochromis Niloticus	Comminiferences	Cicilidae	F/P
Orange-Fin Labeo	Labeo Calbasu	Cypriniformes	Cyprinidae	F/P
Pumpkinseed Sunfish	Lepomis Gibbosus	Perciformes	Centrarchidae	F/P
Perch	Perca Fluviatilis	Perciformes	Percidae	F/D
Pike	Esox Lucius	Esociformes	Esocidae	S/D
Plaice	Pleuronectes Platessa	Pleuronectiformes	Pleuronectidae	S/D
Ray	Dipturus Canutus	Rajiformes	Rajidae	S/P
Rankin Cod	Epinephelus Multinotatus	Perciformes	Serranidae	S/D
Red Emperor	Lutjanus Sebae	Perciformes	Lutjanidae	S/D
Smooth Dogfish	Mustelus Canis	Carcharhiniformes	Triakidae	S/P
Spangled Emperor	Lethrinus Nebulosus	Perciformes	Lethrinidae	S/P
Stoneroller Minnow	Campostoma Anomalous	Cypriniformes	Cyprinidae	F/P
Spanish Mackerel	Scomberonorus Maculatus	Perciformes	Scombridae	S/D
Sea Mullet	Menticirrhus Saxatilis	Perciformes	Sciaenidae	S/D
Summer Flounder	Paralichthyes Dentatus	Pleuronectiformes	Paralichthyidae	S/D
White Sucker	Catostamus Commersoni	Cypriniformes	Catostomidae	F/D
Walleye	Stizostedion Vitreum	Perciformes	Percidae	F/P
Whiting	Merlangius Merlangus	Gadiformes	Gadidae	S/P
White Pollack	Pollachius Virene	Gadiformes	Gadidae	S/P&D
White Sole	Solea Solea	Pleuronectiformes	Soleidae	S/D
White Trout	Salmo Trutta	Salmoniformes	Salmonidae	S/D
Yellow Perch	Perca Flavescens	Perciformes	Cichlidae	F/P
Zebra Fish	Danio Rerio	Cypriniformes	Cyprinids	F/P

Abbreviations: F-freshwater fish, S- saltwater fish, P- pelagic, D-demersal.

## 6.2. Appendix 2.



**Figure 1:** Diagram showing pump flow rate (red line) and pump gradient (blue line) change for allopurinol and oxipurinol.



**Figure 2:**Diagram showing pump flow rate (red line) and pump gradient (blue line) change for purines, phthalazine and pyrazinamide.



Figure 3:Diagram showing pump flow rate (red line) and pump gradient (blue line) change for  $N^1$ -methylnicotinamide.



Figure 4:Diagram showing pump flow rate (red line) and pump gradient (blue line) change for phenanthridine and 6(5H)-phenanthridone.

## 6.3. Appendix 3.

**Table 1:%** RSD of retention time of three replicates was the same day of calibration standardfor intra- day precision.

%RSD for retention time (n=3)								
Analyte	0.01mM	0.02mM	0.03mM	0.05mM	0.1mM			
Allopurinol	0.61	1.23	2.26	0.22	1.28			
Cinchonine	1.66	1.68	0.15	2.11	0.74			
Hypoxanthine	1.11	1.36	0.39	0.78	0.75			
6-Mercaptopurine	2.51	0.68	0.39	1.11	0.17			
Methotrexate	0.97	1.01	1.12	1.19	0.27			
1-Methylxanthine	1.61	1.25	1.86	1.02	0.61			
1-Methyluric acid	0.72	0.98	1.02	1.23	0.88			
N <sup>1</sup> -methylnicotinamide	1.12	2.13	0.52	0.27	1.66			
Oxipurinol	0.87	2.05	1.83	0.59	1.18			
Phenanthridine	2.25	3.17	1.13	0.22	0.78			
Phenanthridone	2.41	2.14	2.17	1.21	0.26			
Phthalazine	0.87	2.21	3.21	1.34	1.19			
1-Pthalazone	2.11	1.06	1.27	0.31	0.25			
Pyridoxal	1.41	0.18	0.24	0.89	1.86			
Pyridoxic acid	2.14	0.77	0.26	1.13	0.66			
Pyrazinamide	1.12	1.51	0.57	1.42	1.29			
6-Thioxanthine	1.51	2.22	0.35	0.16	2.32			
Uric acid	1.22	1.53	0.52	1.37	2.26			
Vanillin	1.02	2.23	0.42	0.25	1.16			
Vanillic acid	0.77	1.61	0.89	0.35	0.59			
Xanthine	1.21	0.98	0.19	0.11	1.79			

The % RSD for retention time between injections is in the range of 0.09-3.31.

Mean ± SD of accuracy (%) (n=3)								
	Nominal concentration							
Analyte	0.01 mM	0.02 mM	0.03 mM	0.05 mM	0.1 mM			
Allopurinol	98.2 ± 1.2	99.2 ± 1.1	$97.1 \pm 0.7$	98.5 ± 1.3	98.5 ± 1.7			
Cinchonine	97.5 ± 1.5	98.3 ± 0.8	97.8 ± 1.4	$105.6\pm0.2$	96.9 ± 1.2			
Hypoxanthine	99.6 ± 1.1	99.5 ± 1.7	99.7 ± 2.1	99.5 ± 1.1	99.6 ± 1.9			
6-Mercaptopurine	98.2 ± 0.3	97.2 ± 1.6	95.5 ± 1.7	98.8 ± 1.1	99.2 ± 1.7			
Methotrexate	99.3 ± 1.3	97.1 ± 1.7	$\textbf{98.9} \pm \textbf{0.5}$	96.1 ± 2.4	98.9 ± 1.8			
1-Methylxanthine	98.5 ± 2.5	$100.3\pm0.9$	99.8 ± 2.3	$\textbf{98.1} \pm \textbf{0.7}$	99.9 ± 0.2			
1-Methyluric acid	95.1 ± 1.1	97.3 ± 0.4	96.2 ± 0.9	99.9 ± 0.2	96.7 ± 0.7			
N <sup>1</sup> -methylnicotinamide	95.1 ± 0.7	98.5 ± 0.4	99.9 ± 0.4	$100.3\pm0.7$	96.8 ± 1.4			
Oxipurinol	99.2 ± 2.1	99.1 ± 1.1	100.4 ± 2.2	96.6 ± 1.9	98.8 ± 2.1			
Phenanthridine	100.6 ± 1.9	99.8 ± 1.4	99.2 ± 0.3	99.1 ± 1.3	$100.8 \pm 1.1$			
Phenanthridone	99.5 ± 0.7	97.5 ± 0.2	99.5 ± 1.7	<b>99.5 ± 1.7 99.5 ± 1.1</b>				
Phthalazine	99.1 ± 1.3	98.7 ± 2.2	95.2 ± 1.7	98.9 ± 1.5	99.8 ± 1.5			
1-pthalazone	98.6 ± 1.5	$100.4\pm0.2$	98.7 ± 1.8	99.5 ± 2.1	95.6 ± 2.3			
Pyridoxal	<b>98.2 ± 0.7</b>	97.9 ± 1.6	95.9 ± 2.3	99.9 ± 0.6	$100.2 \pm 1.7$			
Pyridoxic acid	98.4 ± 1.5	$95.4\pm0.7$	99.1 ± 2.4	98.2 ± 1.8	99.2 ± 1.3			
Pyrazinamide	$100.5\pm0.4$	97.3 ± 1.3	99.9 ± 1.1	99.3 ± 1.2	98.3 ± 2.2			
6-Thioxanthine	$100.2\pm0.2$	$98.4 \pm 2.3$	98.5 ± 1.1	99.8 ± 1.9	98.2 ± 1.3			
Uric acid	95.6 ± 1.7	99.7 ± 1.1	99.7 ± 1.2	98.3. ± 1.9	$97.2\pm0.7$			
Vanillin	97.1 ± 1.2	99.1 ± 1.1	98.8 ± 1.1	95.2 ± 1.7	98.5 ± 0.7			
Vanillic acid	97.1 ± 1.7	98.5 ± 1.8	$100.5 \pm 0.4$	97.5 ± 1.3	99.1 ± 0.9			
Xanthine	97.4 ± 1.2	99.6 ± 1.3	$100.2 \pm 0.1$	<b>99.8</b> ± 0.7	99.6 ± 2.1			

**Table 2:**% Accuracy calculated for intra-day analysis of compounds used in HPLC analysis in this study.

Mean area of three injections in the same day of calibration standards (n=3)										
Analyte	0.01 (mM)	0.02 (mM)	0.03 (mM)	0.05 (mM)	0.1 (mM)	Slope	Intercept	RSQ	LOD (nmol)	LOQ (nmol)
Allopurinol	14726	29588	49432	96833	188236	1926912	-4388	1.00	3	80
Cinchonine	99785	151658	197633	390061	889728	8843599	-21381	0.990	0.2	7
Hypoxanthine	39196	64887	109794	187645	378529	1902	- 3135	0.998	5.4	65
6-Mercaptopurine	39438	79514	139407	288320	591803	6091107	-23441	0.994	4	70
Methotrexate	128570	171092	276739	493204	998593	9945386	-3388	0.997	0.6	0.9
1-Methylxanthine	134653	265966	408825	648214	1250231	6228	15347	0.994	4.3	52
1-Methyluric acid	307169	539389	835522	1431411	2952992	14769	- 22718	0.998	1.7	57
N <sup>1</sup> -methylnicotinamide	44454	91846	169578	308302	661872	6764775	-24091	0.996	1	50
Oxipurinol	23394	47904	76786	129660	233588	2714	2357371	1.00	1	50
Phenanthridine	571679	992255	1180187	2130945	4222854	41430984	66235	0.997	2	80
Phenanthridone	132015	260381	397202	689057	1322123	13294923	1474	0.999	2	70
Phthalazine	253827	565518	887197	1506345	2906879	29355401	-7478	0.999	2	70
1-pthalazone	157996	290877	404934	712217	1524442	15199887	-16689	0.998	2	60
Pyridoxal	28160	45701	77618	112825	202247	1944019	7967	0.992	2	70
Pyridoxic acid	31017	50616	87722	138318	230686	2304745	9060	0.987	2	90
Pyrazinamide	70181	148691	217140	331506	640131	6337483	12796	0.998	4	80
6-Thioxanthine	51175	92429	134115	259511	485648	4904335	-1172	0.997	2	70
Uric acid	482658	1095004	1494725	2563976	4955041	24777	30879	0.999	3.5	90
Vanillin	54118	146188	211879	378212	734053	7447551	-6589	0.999	2	60
Vanillic acid	38737	99515	181284	281098	553041	5617796	-4343	0.997	0.1	20
Xanthine	123857	253547	406308	642818	1253249	6273	7542	0.995	4.6	94

**Table 3:** Regression calculation based on mean area of three replicates in the same day of calibration standards against the actual concentration.



Figure 1:Standard curves for substrates and products used in this study



Figure 2:Standard curves for substrates and products used in this study



Figure 3:Standard curves for substrates and products used in this study



Figure 4:Standard curves for substrates and products used in this study



Figure 5: Standard curves for substrates and products used in this study



**Figure 6:** Eadie- Hofstee plot of AOX and XDH activities in trout.(a) AOX-catalyzed oxidation of DMAC. (b) AOX-catalyzed oxidation of vanillin. (c) XDH-catalyzed oxidation of xanthine. (d) XDH-catalyzed oxidation of 1-methylxanthine. The substrates concentrations were 25-100  $\mu$ M of AOX and XDH substrates. The typical results are from three independent experiments.



**Figure 7:** Eadie- Hofstee plot of AOX activities in rat.(a) AOX-catalyzed oxidation of DMAC. (b) AOX-catalyzed oxidation of vanillin. (c) AOX-catalyzed oxidation of phenanthridine. The substrates concentrations were 25-100 µM of AOX substrates. The typical results are from three independent experiments.



**Figure 8:** Eadie- Hofstee plot of XOR activities in rat. (a) XO-catalyzed oxidation of xanthine. (b) XDH-catalyzed oxidation of xanthine. (c) XDH-catalyzed oxidation of 1-methylxanthine. The substrates concentrations were 25-100 µM of XOR substrates. The typical results are from three independent experiments.



**Figure 9:** Eadie- Hofstee plot of AOX activities in human. (a) AOX-catalyzed oxidation of DMAC. (b) AOX-catalyzed oxidation of vanillin. (c) AOX-catalyzed oxidation of phenanthridine. The substrates concentrations were 25-100  $\mu$ M of AOX substrates. The typical results are from three independent experiments.



**Figure 10:** Eadie- Hofstee plot of XOR activities in human. (a) XO-catalyzed oxidation of xanthine. (b) XDH-catalyzed oxidation of xanthine. (c) XDH-catalyzed oxidation of 1-methylxanthine. The substrates concentrations were 25-100 µM of XOR substrates. The typical results are from three independent experiments.



**Figure 11:** Eadie- Hofstee plot of AOX activities in rainbow trout. (a) AOX-catalyzed oxidation of vanillin. (b) AOX-catalyzed oxidation of pyridoxal. (c) AOX-catalyzed oxidation of all-trans retinal. (d)AOX-catalyzed oxidation of N<sup>1</sup>-methylnicotinamide. The substrates concentrations were 25-100  $\mu$ M of AOX substrates. The typical results are from three independent experiments.



**Figure 12:** Eadie- Hofstee plot of AOX activities in rainbow trout. (a) AOX-catalyzed oxidation of phenanthridine. (b) AOX-catalyzed oxidation of phthalazine. (c) AOX-catalyzed oxidation of cinchonine. The substrates concentrations were  $25-100 \,\mu$ M of AOX substrates. The typical results are from three independent experiments.



Figure 13: Eadie- Hofstee plot of XOR activities in rainbow trout. (a) XDH-catalyzed oxidation of xanthine. (b) XDH-catalyzed oxidation of 1-methylxanthine. (c) XDH-catalyzed oxidation of allopurinol. (d) XO-catalyzed oxidation of 6-thioxanthine. The substrates concentrations were 25-100  $\mu$ M of XOR substrates. The typical results are from three independent experiments.





**Figure 14:** Eadie- Hofstee plot of XOR activities in rainbow trout. (a) XDH-catalyzed oxidation of 6-thioxanthine. (b) XDH-catalyzed oxidation of pyrazinamide. The substrates concentrations were 25-100  $\mu$ M of XDH substrates. The typical results are from three independent experiments.



**Figure 15:** Eadie- Hofstee plot of AOX activities in rat. (a) AOX-catalyzed oxidation of vanillin. (b) AOX-catalyzed oxidation of pyridoxal. (c) AOX-catalyzed oxidation of ATR. (c) AOX-catalyzed oxidation of NMN. The substrates concentrations were  $25-100 \,\mu$ M of AOX substrates.



**Figure 16:** Eadie- Hofstee plot of AOX activities in rat. (a) AOX-catalyzed oxidation of phenanthridine. (b) AOX-catalyzed oxidation of phthalazine. (c) AOX-catalyzed oxidation of cinchonine. (d) AOX-catalyzed oxidation of pyrazinamide. The substrates concentrations were  $25-100 \,\mu$ M of AOX substrates.



**Figure 17:** Eadie- Hofstee plot of XOR activities in rat. (a) XO-catalyzed oxidation of xanthine. (b) XDH-catalyzed oxidation of xanthine. (c) XO-catalyzed oxidation of 1-MX. (d) XDH-catalyzed oxidation of 1-MX. The substrates concentrations were 25-100 µM of AOX substrates.



**Figure 18:** Eadie- Hofstee plot of XOR activities in rat. (a) XO-catalyzed oxidation of 6-MP. (b) XDH-catalyzed oxidation of 6-MP. (c) XO-catalyzed oxidation of allopurinol. (d) XDH-catalyzed oxidation of allopurinol. The substrates concentrations were 25-100  $\mu$ M of AOX substrates.



Figure 19: Eadie- Hofstee plot of XDH activities in rat. XDH-catalyzed oxidation of pyrazinamide. The substrates concentrations were 25-100  $\mu$ M of XDH substrates. The typical results are from three independent experiments.



**Figure 20:** Eadie- Hofstee plot of AOX activities in human. (a) AOX-catalyzed oxidation of vanillin. (b) AOX-catalyzed oxidation of pyridoxal. (c) AOX-catalyzed oxidation of ATR. (d) AOX-catalyzed oxidation of NMN. The substrates concentrations were  $25-100 \,\mu$ M of AOX substrates.



**Figure 21:** Eadie- Hofstee plot of AOX activities in human. (a) AOX-catalyzed oxidation of phenanthridine. (b) AOX-catalyzed oxidation of phthalazine. (c) AOX-catalyzed oxidation of cinchonine. (d) AOX-catalyzed oxidation of pyrazinamide. The substrates concentrations were  $25-100 \,\mu$ M of AOX substrates.



**Figure 22:** Eadie- Hofstee plot of XOR activities in human. (a) XO-catalyzed oxidation of xanthine. (b) XDH-catalyzed oxidation of xanthine. (c) XO-catalyzed oxidation of 1-MX. (d) XDH-catalyzed oxidation of 1-MX. The substrates concentrations were 25-100 µM of AOX substrates.



**Figure 23:** Eadie- Hofstee plot of XOR activities in rat. (a) XO-catalyzed oxidation of 6-MP. (b) XDH-catalyzed oxidation of 6-MP. (c) XO-catalyzed oxidation of allopurinol. (d) XDH-catalyzed oxidation of allopurinol. The substrates concentrations were 25-100  $\mu$ M of AOX substrates.



**Figure 24:** Eadie- Hofstee plot of XOR activities in human. XDH-catalyzed oxidation of pyrazinamide. The substrates concentrations were 25-100  $\mu$ M of AOX substrates. The typical results are from three independent experiments.



**Figure 25:** Eadie- Hofstee plot of purified AOX activities in rainbow trout. AOX-catalyzed oxidation of (a) vanillin, (b) pyridoxal and (c) all-trans retinal. The substrates concentrations were  $25-100 \,\mu$ M of AOX substrates. The typical results are from three independent experiments.