

# Evaluation of Point Source Pollutants impact on Water Quality and Self-purification Capacity of Abay River in Ethiopia

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## Abstract

This study was conducted to evaluate influence of point source pollutants on self-purification capacity of Abay River within 8 kms from Lake Tana to Sebatamit. For assessment of the overall water quality level, the water quality index was used. The river's self-cleaning potential was assessed based on physicochemical parameters as indicators and using the modified Streeter-Phelps model after selecting the best reaeration coefficient through velocity and depth measurements. The outcome of the study showed that TDS, EC, BOD<sub>5</sub>, Ammonia, temperature, pH, and salinity were found within the permissible limits of EPA. The turbidity was higher than the WHO limit at sampling point two and DO value was below the WHO and EPA limit at sampling points two, three, four, five and six. The result shows that the critical distance and time in which the river attains its self-purification capacity is at 1.3, 2.8 and 0.3 kms from Bahirdar Textile, Bahir Dar municipal and Habesha Tannery Effluent mixing points for 1.68, 0.82 and 0.48 hour, respectively. This showed that the river have good purification capacity at a moment even though the point sources are discharging an effluent which is not properly treated.

**Keywords:** Physicochemical parameters, Reaeration coefficient, Water Quality Index

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## 1 INTRODUCTION

Water contamination is caused by discharging of waste water effluent into the fresh water systems of inadequately treated sewage. This can contribute to damage to the environment of aquatic ecosystems and issues of public health (Ogundiran and Fawole, 2014). Contaminants that cause water pollution include a wide range of chemicals, bacteria, and physical changes such as high temperatures (Ogundiran and Fawole, 2014). More than half of the world's major rivers are degraded, according to the world water commission for the 21st century. They are therefore threatening human health and destroying the habitats around them (Nyasulu, 2012). Diseases such as typhoid fever, cholera and other intestinal diseases are caused due to drinking of contaminated water (Danquah, 2010). Waste disposal and management are major challenges, which confront urban centers throughout the world. The habit of untreated sewage discharges into surface water bodies is popular in many developing countries. This is due to the expansion of industrialization and high urbanized societies with lack of waste management facilities. In Ethiopia, the wastewater effluents from different sources are joining the river without and partially treatment. In Addis Abeba, Ethiopia 40 percent of streams are contaminated by various wastewater discharges into rivers (Yohannes and Elias, 2017), (Beyene et al., 2009). Abay River is one of the water bodies in Ethiopia which is affected by discharge of untreated effluent. Bahir Dar textile, Habesha Tannery and Bahir Dar Tannery discharge their effluents without treatment and with partially treatment directly into the Abay River which is the source of Blue Nile River.

Many studies conducted in different parts of the world showed a considerable river water quality change due to point sources (Van der Hoven et al., 2017). Abrehet et al have reported that the pollution of the water body by textile industry effluents discharged into the stream such as the textile industry effluent showed considerable negative effects on the water quality of the streams and human health (Mehari et al., 2015). Assefa Wosenie et al stated that effluents from tannery highly affect the micro invertebrates and industrial effluents and toxic substances that have a number of adverse effects on the water bodies. The level of damage at the downstream intermediate site was high and at the upstream site it was very little. This high downstream weakness made water undesirable for domestic, agricultural and esthetic uses (Mehari et al., 2015).

Natural water systems self-purification is a difficult process that often requires simultaneous physical, chemical and biological processes. The water is purified in the sense that the concentration of waste materials has been reduced mostly by means of biodegradation processes. Therefore, this process is very closely tied with the dissolved oxygen content and all the sources and sinks of oxygen in a river (Taseiko et al.). Several problems have arisen because of the reduction in the self-cleaning capacity of streams. Dissolved oxygen (DO) is one of the vital for microorganisms in water bodies and is an important indicator of the aquatic ecosystem's health. A particular

ecosystem DO can differ as a function of many interlinked dynamic parameters such as, the presence of organic contaminants and marine species biological activities. It is therefore important and timely to apply models that can effectively predict the DO of a river water influenced by contamination from the point source (Omole and Longe, 2008).

The Abay River, which is source of the Blue Nile River, is the main surface water supply for over 20 million people in Ethiopia, and over 300 million societies at downstream part of the Nile Stream (Yitayew and Melesse, 2011). There have been several efforts to improve the reliability of river water and attempts to drain wastewater treatment plants from point source. Such efforts by wastewater treatment plants and industrial wastewater treatment plants, however, are not yet fully implemented and operated in the River Basin, where point sources can be a major cause of contamination for Abay River. The overall objective of this analysis was to evaluate point source pollutants impact on Abay River water self-cleaning potential at the head of the Blue Nile River (Abay River in Ethiopia). The specific goals were to identify the influence of pollution sources on Abay River's quality of water and to measure the Abay River's pollution status using the water quality index. This was selected because of water quality index model simplifies and show water quality statuses in short way (Bora and Goswami, 2017; Dutta et al., 2018; Elshemy and Meon, 2016).

## 2 MATERIALS AND METHODS

### 2.1 Description of Study area

Abay River is situated in Ethiopia's National Regional State of Amhara. The study area is located on the southern shore of the heart-shaped Tana Lake (Figure 1), the source of the river Blue Nile, around 565 km northwest of Addis Ababa. It has 1801 m above sea level and 11°38' N latitude and 37°01' E longitude. The Nile basin comprises more than 10 percent of the landmass of Africa in more than 11 countries including Ethiopia, Sudan and Egypt. Almost all those countries water source is depend on Nile River. Except 20 percent of amount of water of Egypt, the remain is contributing from Ethiopian highlands (Conniff et al., 2013).

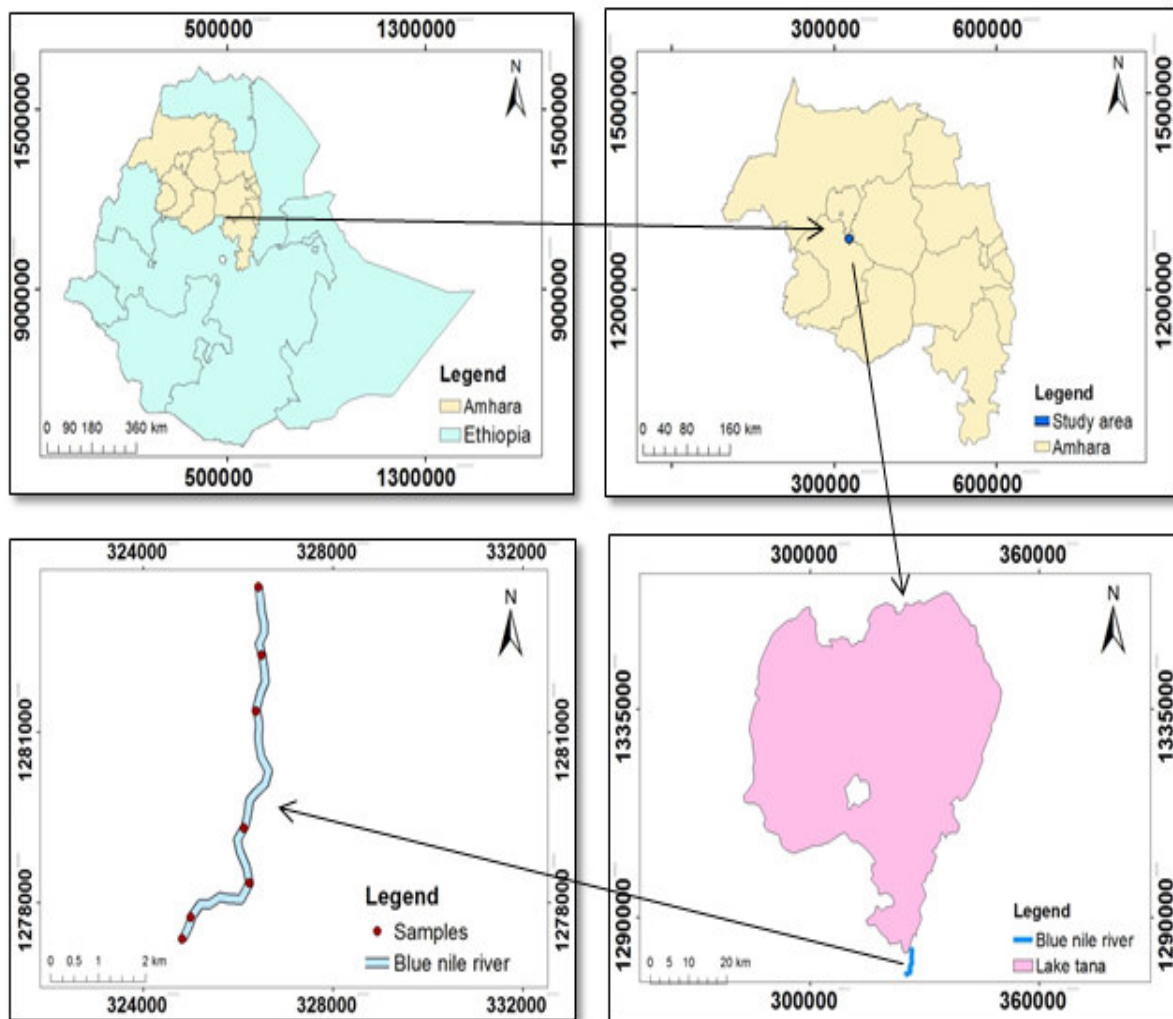


Figure 1 showed that a) Ethiopia, b) Amhara Regional State, c) Tena Lake and d) Abay River (Blue Nile)

Table 1 location and relative distances of sampling points

SP	Relative Distance (km)	X	Y	Z	Description
SP1	0	11°36'24.07"	37°24'28.37"	1795	Upstream site after River Abay leave Lake Tana.
SP2	1.3	11°35'45.09"	37°24'30.90"	1799	Textile effluent Mixed with Abay River.
SP3	2.3	11°35'12.93"	37°24'27.2"	1790	Between textile and ditch
SP4	4.8	11°34'6.66"	37°24'18.53"	1799	Municipal ditch mixed with Abay River
SP5	7.6	11°33'15.04"	37°23'42.48"	1791	Around Bahir dar tannery effluent mix with Abay River
SP6	7.85	11°33'3.17"	37°23'35.78"	1789	Habesha tannery effluent mixed with Abay river
SP7	8	11°33'2.75"	37°23'36.66"	1789	The most downstream Of sampling sites.
E1		11°35'43.13"	37°24'24.18"		Bahir Dar textile effluent
E2		11°34'7.25"	37°24'17.87"		Bahir Dar municipal ditch effluent
E3		11°33'5.24"	37°23'31.81"		Habesha tannery effluent

## 2.2 Selection of Water and wastewater sampling points

The samples were collected from 10 sampling sites including 3 point sources for wastewater effluent discharge labeled (E1 to E3) and 7 sampling sites (SP1 to SP7) along the segment of the river before and after effluent mixing. The three point sources (E1 to E3) were from Bahir Dar textile effluent, Bahir Dar municipal effluent and Habesha Tannery effluent, respectively.

## 2.3 Sample collection and Water Quality parameters

From each sampling point 3 wastewater samples were collected at a 3 points along the width (edge, center, and edge) 3 times. Samples were taken from 3 sections of the stream to minimize inaccuracies that can be made during data collection and also samples were taken at a depth of about 20 cm just under the water surface (Ibrahim et al., 2019; Shrivastava et al., 2015). Measurements of physico-chemical water quality parameters from the samples were carried out using laboratory analyzes in accordance with the procedures for standard water and wastewater testing methods described for water quality analysis (Shrivastava et al., 2015). The parameters such as BOD (Biological Oxygen Demand) and Turbidity were assessed in the laboratory of the Bahir Dar University Institute of Technology, Bahir Dar University. At Amhara Design and Supervision Work Enterprise Laboratory, ammonia – nitrogen (NH<sub>3</sub>-N) was measured. Temperature, PH, EC (electrical conductivity), DO (dissolved oxygen), TDS (total dissolved solids), and salinity were also measured using the field meter YSI 556 Multi Prob (Fernández-Gómez et al., 2012).

## 2.4 Assessment of the water quality using Water Quality Index (WQI)

Water quality index is a model used to quantify the general quality status of particular sampling station taken from lakes and rivers [13]. Essentially, the WQI is a compilation of a number of parameters which could be used to determine the overall quality of the river (Jahin et al., 2019). The Water Quality Index value can be found by multiplying Xr value and the unit weight (Wi).  $WQI = \sum W_i \times X_r$ . Kumar et al, 2012 was used Xr value based on the scale of different water quality parameters as shown in Tables 1 below (Balan et al., 2012).

Table 2 Rating Scale for Calculation of WQI (Sources: (Kumar et al, 2012))

	Ranges				
pH	7.0-8.5	8.6-8.7	8.8-8.9	9.0-9.2	>9
DO	>7.0	6.8-6.9	6.7-6.8	6.5-6.7	<6.5
EC	0-75	75.1-150	150.1-225	225.1-300	>300
TDS	0-375	375.1-750	750.1-1125	1125.1-1500	>1500
Xr	100	80	60	40	0
Pollution status	Clean	Slightly Pollution	Medium Pollution	Excess pollution	Severe pollution

DO (Dissolved Oxygen), EC (electrical conductivity) and TDS (total dissolved solids)

Water quality status of river can be classified as excellent to bad based on water quality index. According to Kumar et al, 2012 the water quality status can classified based on water quality index values as excellent ( 90-100), good (70-90), medium(50-70) and bad (25-50) (Balan et al., 2012).

## 2.5 Methods used for Self-purification of Abay River

### 2.5.1 Selection of Reaeration Coefficient Models

According to Lazorchak, J.M., Klemm, D.J., Peck, D.V. (1999) in order to select the best appropriate reaeration coefficient average velocity and depth should be taken between 20 intervals (Nel, 2008). Based on this the average velocity and depth were measured at 25 points along the river width for each sampling points except SP4 due to its high depth. At each sampling points, velocity and depth were measured using current meter. The river cross-section was determined by tape across its width and the stream discharge was calculated using the method of Velocity-Area. Using Handheld GPS unit and Google Earth Pro, the overall segment distance and the geographic position of each sampling point was collected. The performance of the model was evaluated by inserting the Ka value and the amount of dissolved oxygen (DO) predicted. The observed and simulated data was used to check the performance of the model. To select the appropriate Ka values different Ka values was tried by inserting on the DO model (Hill et al., 2003).

$$D = D_0 e^{-K_a t} + \frac{K_d L_0}{K_a - K_r} (e^{-K_r t} - e^{-K_a t}) + \frac{K_n L_{no}}{K_a - K_n} (e^{-K_n t} - e^{-K_a t}) + \frac{S}{k_a H} (1 - e^{-k_a x/u}) + \frac{R-P}{k_a} (1 - e^{-k_a x/u})$$

Equation 1

All bathymetric data gathered by travelling by boat and on foot were entered into an equation1. The best fitted reaeration coefficient (Ka) was selected by comparing the results found by inserting Ka values of Street-Phleps, Agummba, O'Connor and Dobbins in equation 1. For this particular place Street-Phleps reaeration coefficient was used in the study area because of its successful prediction of DO value. The re-aeration coefficient proposed by Streeter et al (Streeter et al., 1936) was defined as:

$$k_a = 5.026 \frac{U^{0.969}}{H^{1.673}} \quad \text{Equation 2}$$

Agunwamba et al. (2007), working on the Amadi Creek, proposed a ka defined as (Agunwamba et al., 2006):

$$k_a = 11.6325 \frac{U^{1.0954}}{H^{0.0016}} \quad \text{Equation 3}$$

O'Connor and Dobbins, 1958 (O'Connor and Dobbins, 1958)

$$k_a = 3.9 \frac{U^{0.5}}{H^{1.5}} \quad \text{Equation 4}$$

For all equations 2, 3 and 4; U (average velocity in m/s) and H (Average Depth in m) and those values were measured by the help of current meter.

The performance of the model was evaluated by inserting the Ka value and the amount of dissolved oxygen (DO) was predicted. The observed and simulated data was used to check the performance of the model. To select the appropriate Ka values different Ka values was tried by inserting on the DO model.

### 2.5.2 Determination of the oxygen deficit and critical time

To evaluate self-purification capacity modified Streeter-Phelps model were used. To get the distance where can be the river attain its self-purification capacity and the critical dissolved oxygen values equation 5 and 6 were proposed by Gotovtsev et al (Gotovtsev, 2010).

$$t_c = \frac{1}{k_a - k_r} \ln \left[ \frac{k_a}{k_r} \left( 1 - \left( \frac{D_0}{k_d L_0} (k_a - k_r) \right) \right) \right] \quad \text{Equation 5}$$

$$D_c = \frac{k_r L_0}{k_a} \left[ \frac{k_a}{k_r} \left( 1 - \frac{D_0}{L_0} \left( \frac{k_a - k_r}{k_a} \right) \right) \right]^{\frac{k_r}{k_a - k_r}} \quad \text{Equation 6}$$

The saturation dissolved oxygen is dependent on temperature and was calculated as follows:

$$S_s = 14.652 - 0.41022T + 0.007991T^2 - 0.000077774T^3 \quad \text{Equation 7}$$

Equation 7 gives the concentration  $S_s$  (expressed in  $g/m^3$ ) as function of the temperature in  $^{\circ}C$ .

### 3 RESULT AND DISCUSSION

#### 3.1 Results of different Water quality parameters

Table 3 Water quality Results obtained during data collection (Average  $\pm$  SD)

Sampling Points	Temperature( $^{\circ}C$ ) (n=2)	EC( $\mu S/cm$ ) (n=2)	TDS(mg/L) (n=2)	pH (n=2)
SP1	23.53 $\pm$ 0.34	203.0 $\pm$ 3.7	131.98 $\pm$ 47.53	7.23 $\pm$ 0.58
SP2	25.0 $\pm$ 2.3	203.3 $\pm$ 53.4	132.13 $\pm$ 34.36	7.69 $\pm$ 0.37
SP3	24.1 $\pm$ 1.66	197.1 $\pm$ 68.99	128.13 $\pm$ 41.48	7.08 $\pm$ 0.35
SP4	24.72 $\pm$ 1.2	188.6 $\pm$ 60.74	122.61 $\pm$ 30.39	7.06 $\pm$ 0.39
SP5	24.76 $\pm$ 0.42	202.6 $\pm$ 51.07	131.7 $\pm$ 34.45	7.29 $\pm$ 0.83
SP6	25.72 $\pm$ 1.2	203.5 $\pm$ 51.08	132.26 $\pm$ 33.65	7.28 $\pm$ 0.73
SP7	24.54 $\pm$ 0.42	201.6 $\pm$ 55.48	131.07 $\pm$ 35.88	6.96 $\pm$ 0.27

Table 4 Water quality Results obtained during data collection (Average  $\pm$  SD)

Sampling Points	Turbidity(NTU)	DO(mg/L)	BOD(mg/L)	Ammonia (mg/L)	Salinity
SP1	20.62 $\pm$ 17.22	6.5 $\pm$ 0.3	25 $\pm$ 3.5	0.13	0.1
SP2	31.23 $\pm$ 8.07	3.37 $\pm$ 0.18	30 $\pm$ 1.4	0.06	0.1
SP3	20.39 $\pm$ 6.94	3.91 $\pm$ 0.15	23 $\pm$ 2.1	0.03	0.1
SP4	17.2 $\pm$ 2.18	2.85 $\pm$ 0.32	40 $\pm$ 7.1	0.02	0.1
SP5	12.72 $\pm$ 2.83	4.2 $\pm$ 0.06	27 $\pm$ 3.5	0.26	0.1
SP6	14.75 $\pm$ 2.72	3.3 $\pm$ 0.10	38 $\pm$ 5.6	0.18	0.1
SP7	10.45 $\pm$ 3.16	5.0 $\pm$ 0.03	15.5 $\pm$ 0.71	0.04	0.1

Table 5 Effluent quality

NO.	pH	Temperature ( $^{\circ}C$ )	TDS (mg/L)	EC( $\mu S/cm$ )	Salinity	BOD <sub>5</sub> (mg/L)	DO(mg/L)
E1	8.4 $\pm$ 0.8	21.8 $\pm$ 1.7	370 $\pm$ 36.8	569 $\pm$ 56.6	0.11	36 $\pm$ 1.4	3.8 $\pm$ 0.6
E2	6.7 $\pm$ 0.1	19.8 $\pm$ 1.7	604.6 $\pm$ 77.1	930 $\pm$ 118.7	0.65	27 $\pm$ 4.2	3.7 $\pm$ 0.2
E3	7.6 $\pm$ 0.6	21 $\pm$ 1.4	2300 $\pm$ 148.5	3538.5 $\pm$ 228.4	2.5	70 $\pm$ 14	3.3 $\pm$ 0.1

The different water quality parameters tested and evaluated based on standards was presented in table 3, 4 and 5. BOD<sub>5</sub> (mg/L) and DO (mg/L) from Habesha Tannery were above the permissible limit of WHO and EPA, whereas pH, temperature and BOD<sub>5</sub> from textile and municipal ditch were within the permissible limit of EPA and WHO for surface water course. Except effluent concentration from textile factory, the amount of total dissolved solids (mg/L) from Habesha tannery and municipal ditch was higher than the allowable limit of 500 mg/L. But the amount of electrical conductivity showed that all point sources effluent contains higher than the allowable limit of WHO (300) for surface water. The NH<sub>3</sub>-N values for all surface water sampling stations were lower than the WHO standard of 30 mg / L and the EPA limit of 20 mg / l for surface water(Shrivastava et al., 2015). The amount of ammonia concentration was high at SP5. It can conclude that the tanneries and animal activities increase the amount of ammonia concentration in this sampling point. Generally, the concentration of ammonia at all sampling points was within a permissible limit of EPA (< 20 mg/l) for surface water(Paul, 2011). Except at the sampling point of SP2 the turbidity of all sampling sites were within WHO guide line for surface water (30 mg/L) (Falkowski et al., 1980).The turbidity at SP2 was due to car washing and textile wastewater effluent join the river around that place. At the time of sampling, the mean water temperature was 24.5  $^{\circ}C$  which is within the WHO and EPA standard. Sampling points SP2and SP6 recording the relatively higher temperature due to the entry of waste water effluent from Bahir Dar Textile Factory and Habesha Tannery effluent.

The river water exhibited a near neutral pH (7.06-7.96) and this value found within the interval of WHO and EPA (6.5-9) for drinking water. Awomeso, J., et al (2010) showed that when the pH value is within the WHO interval, micro-organisms that help break down biological waste could stay alive in such a neutral environment and can easily degrade the biological waste water entering the river(Orebiyi et al., 2010). The conductivity apparently increased at sampling stations SP2 (203.3  $\mu S/cm$ ), SP5 (202.61  $\mu S/cm$ ) and SP6 (203.48  $\mu S/cm$ ), respectively due to Textile industry and Habesha Tannery wastewaters effluent. The values of electrical conductivity from the point sources were above permissible limit defended by WHO and EPA. However, this value on the river was within the permissible limit of surface water course of 300 $\mu S/cm$ . The TDS content of water samples obtained at selected stations ranged from 122,61 mg / L to 132,26 mg / that were below the



WHO maximum standard value of 500 mg / L (Paul, 2011). Increasing TDS and conductivity showed that the water is contaminated by domestic and industrial sewage discharges.

The amount of BOD concentration at sampling points SP2, SP4 and SP6 was high (Table 4). This showed that Textile factory at SP2, Municipal wastewater at SP4 and Habesha Tannery factory at SP6 have adverse impact on the river water pollution. Especially municipal wastewater ditch covers several parts of the town and collected different wastewater that can be affected water quality. All sampling station contains the amount of BOD above permissible limit defined by WHO and EEPA for river discharge (< 40 – 50 mg/L). Except the DO values of SP1 and SP7, DO value of Abay River was found below the desired value (5 mg/L) as per the WHO guidelines for surface water courses. In SP2, SP4, and SP6 sample stations, the lowest DO values were observed. As it could have observed when water moves steadily away from upstream (SP1) there were successive changes in pollution. As a result, the river's self-purification could disrupt the predicted usual spoon-shaped DO curve in several places along the stream.

### 3.2 Water Quality index

The results of the WQI for the sampling points showed that the water quality of the Abay River is impaired by the point sources of pollution, except for the sampling points SP1 upstream point and SP7 the most downstream portion. According to the water quality index-sampling site SP2, SP3, SP5 and SP6 were ranked as a medium polluted sites due to Bahir Dar Tannery and Habesha Tannery effluents entered at SP5 and SP6. Due to the wastewater effluents from Bahir Dar Municipal, sampling site SP4 was badly polluted site. The Weighted factor for different parameters as shown on table 6 was calculated based on Akkaraboyina et al, 2012 (Akkaraboyina and Raju, 2012).

Table 6 Weight factor of different parameters

Parameters	Permissible Limit(Xi) (WHO for surface water)	1/Xi	Wi =(k/Xi)
pH	9	0.111	0.35
DO	5	0.2	0.63
EC	300	0.003	0.01
TDS	500	0.002	0.01
Sum		0.316 (k <sub>i</sub> )	

$$\text{Where, } k = \frac{1}{\sum_{i=1}^n \left(\frac{1}{X_i}\right)} = \frac{1}{\sum_{i=1}^4 0.316} = 3.164$$

Table 7 Water quality index of sampling points of Abay River

WQI	Sp1	Sp2	Sp3	Sp4	Sp5	Sp6	Sp7
Xr(pH)	100	100	100	100	100	100	100
Xr(DO)	80	40	40	0	40	40	80
Xr(EC)	60	60	60	60	60	60	60
Xr(TDS)	100	100	100	100	100	100	100
Wi(pH)	0.351	0.351	0.351	0.351	0.351	0.351	0.351
Wi(DO)	0.632	0.632	0.632	0.632	0.632	0.632	0.632
Wi(EC)	0.01	0.01	0.01	0.01	0.01	0.01	0.01
Wi(TDS)	0.0063	0.0063	0.0063	0.0063	0.0063	0.0063	0.0063
WQI = $\sum (W_i * X_r)$	86.89	61.61	61.61	36.33	61.61	61.61	86.89
Water quality status	Good	Medium	Medium	Bad	Medium	Medium	Good

### 3.3 Reaeration Coefficient Modeling

O'Connor and Dobbins was predicted DO values about 80.24%, Streeter-Phelps predicts 82.62% and Agunwamba et al predicts about 71.4%. Therefore, the best reaeration coefficient that predicts the amount of dissolved oxygen for the river was using Streeter – Phelps reaeration coefficient model.

### 3.4 Self-Purification Capacity

The minimum amount of dissolved oxygen was observed around the point sources and value was gradually increased when away from the point sources. The concentrations of pollutants from the point sources were above the permissible limit defined by world health organization. But except some parameters like BOD the pollutants which join to the river they dispersed, diluted and found within the acceptable level.

## 4 CONCLUSION

The pollutants effluent concentration generated from point sources were above permissible limit of World health

organization. The amount of dissolved oxygen that showed to as self-purification capacity was below the recommended limit of World health organization of 4 mg/L at all the mixing point of point sources with stream. Similar to that water quality index results all sampling points except sampling points one and seven the rest sampling points showed that the river is affected by the point sources. According to Streeter-Phelps model prediction the effluent from Bahir Dar textile, Habesha Tannery and Bahir Dar municipal was purified by the river. When it is tray to evaluate the point sources the segment of the river were divide in to three reaches based on the location of those point sources. Reaches 1 was include SP1, SP2 and SP3, reach2 includes SP4and SP5 and reach3 includes SP6 and SP7. The Streeter-phleps model predicted self-purification of the river distance from the source at reach one to be 1.3 km from Bahir Dar Textile mixing point at 1.68 hr. The observed oxygen also showed at this reach the river attained its self-purification capacity. At reach2 the self-purification capacity was expected to be at 2.8 km from Bahir Dar municipal wastewater effluent mixing point at 0.82 hr. In the final reach reach3, the self-purification capacity was predicted at 0.3k m after mixing of Habesha Tannery effluent to the river flow at 0.48hr. The amount of oxygen found in reach 2 was very lower than the other reaches. This showed that the municipal wastewater effluent highly affect the river water quality and interrupting the self-purification capacity. It can be concluded from the findings that the river currently has a good capacity for self-purification, even though the point sources discharge an effluent that is not properly treated and interrupt the river's capacity for self-purification. If water pollution is not properly monitored, the scale of pollutants from point sources will increase in the future.

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