

1 Article

2 **Inhalable spray-dried chondroitin sulfate**
3 **microparticles: Effect of different solvents on particle**
4 **properties and drug activity**5 **Susana Rodrigues^{1,2}, Ana M. Rosa da Costa³, Noelia Flórez-Fernández^{1,2,4}, María Dolores Torres⁴,**
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17 **Abstract:** Spray-drying stands as one of the most used techniques to produce inhalable
18 microparticles, but several parameters from both the process and the used materials, affect the
19 properties of the resulting microparticles. In this work we describe the production of drug-loaded
20 chondroitin sulfate microparticles by spray-drying, testing the effect of using different solvents
21 during the process. Full characterisation of the polymer and of the aerodynamic properties of the
22 obtained microparticles are provided envisaging an application in inhalable tuberculosis therapy.
23 The spray-dried microparticles successfully associated two first-line antitubercular drugs (isoniazid
24 and rifabutin) with satisfactory production yield (up to 85%) and drug association efficiency (60% –
25 95%). Ethanol and HCl were tested as co-solvents to aid the solubilisation of rifabutin and
26 microparticles produced with the former generally revealed the best features, presenting a better
27 ability to sustainably release rifabutin. Moreover, these presented aerodynamic properties
28 compatible with deep lung deposition, with aerodynamic diameter around 4 µm and fine particle
29 fraction of approximately 44%. Finally, it was further demonstrated that the antitubercular activity
30 of the drugs remained unchanged after encapsulation independently of the used solvent.

31 **Keywords:** chondroitin sulfate, isoniazid, inhalable microparticles, rifabutin, solvents, spray-drying

32

33 **1. Introduction**

34 Polysaccharides have been frequently used as matrix materials of inhalable microparticles aimed
35 at lung drug delivery and spray-drying is a predominant technique applied in their production.
36 Natural materials show the general advantage of having higher probability to provide
37 biocompatibility and biodegradability, rendering them attractive for pharmaceutical applications.
38 Chondroitin sulfate (ChS) is a natural polymer commonly found on proteoglycans in several tissues,
39 including the lungs [1]. It is composed of alternating sulfated *N*-acetylgalactosamine and glucuronic
40 acid residues, both referred to be recognised by macrophage receptors [2, 3], which provides an
41 important advantage in macrophage targeting strategies [4]. Tuberculosis is a respiratory disease
42 characterised by the intracellular accumulation of the infectious agent (*Mycobacterium tuberculosis*) in
43 the alveolar macrophages. Direct lung delivery of the antibiotics could, thus, bring advantages over
44 the conventional oral therapy, as it co-localises drugs and pathogenic agents in the same

45 compartment. Inhalable carriers have, however, to be optimised to reach the alveoli, where
46 macrophages are located.

47 According to the World Health Organization (WHO), combined therapy of tuberculosis is
48 mandatory to potentiate the treatment efficiency [5], and isoniazid (INH) and rifabutin (RFB) are two
49 of the used first-line drugs [6]. They display different characteristics, namely regarding molecular
50 weight and aqueous affinity. In this regard, INH is a 137 g/mol hydrophilic molecule, while RFB has
51 847 g/mol and is hydrophobic, thus requiring organic or acidic solvents to dissolve. In parallel, the
52 literature reports that the use of different solvents in the spray-drying of individual drugs has an
53 effect on particle size, density, shape and surface appearance. Specifically, it is described that
54 hydrophilic drugs processed by spray-drying may generate smaller particle sizes if surrounded by
55 solvents with higher water content [7]. In turn, hydrophobic drugs lead to smaller and more porous
56 particles when processed in more organic solvents [8]. To our knowledge, the comparison of the effect
57 of different solvents in a mixture of two drugs of opposite character (one hydrophilic and the other
58 hydrophobic), in the same formulation is not reported.

59 Bearing in mind the usefulness of a single formulation of inhalable microparticles conjugating
60 two antitubercular drugs, following the WHO requirements on tuberculosis therapy, this work
61 focused on studying the effect of using different co-solvents in the spray-drying process. Specifically,
62 ethanol and hydrochloric acid (HCl) were tested as co-solvents, along with water, in a spray-drying
63 process envisaging the production of ChS microparticles loaded with INH and RFB. The molecular
64 mass distribution was evaluated for ChS and ChS-based microparticles, and a rheological study
65 performed, in order to determine the effect of spray-drying on the polymer properties. The effect of
66 the microencapsulation process on the antimicrobial activity of the drugs was also evaluated,
67 encompassing the effect of using different solvents. Finally, the respirability of the produced dry
68 powders was characterised to ascertain the effect of the solvents on the aerodynamic properties, and
69 to establish the potential of the carriers for an inhalable strategy to treat tuberculosis.

70 2. Materials and Methods

71 2.1. Materials

72 ChS was acquired from Creative Biomart (USA). Acetic acid, INH, Tween 80®, phosphate buffer
73 saline (PBS) tablets pH 7.4, thiazolyl blue tetrazolium bromide (MTT), dimethyl sulfoxide (DMSO)
74 and HCl were purchased from Sigma-Aldrich (Germany). RFB was supplied by Chemos (Germany).
75 Hydroxypropylmethylcellulose (HPMC) capsules (size 3) were received from Capsugel (Colmar,
76 France). The RS01 Dry Powder Inhaler was a kind gift from Plastiapi S.p.a. (Italy). Middlebrook 7H9
77 (M7H9) and oleic acid, albumin, dextrose and catalase (OADC) supplement were purchased from
78 Remel (Lenexa, USA). Ultrapure water (MilliQ, Millipore, UK) was used throughout. All other
79 chemicals were reagent grade.

80 2.2. Polymer purification

81 ChS from Creative Biomart (USA) was not of pharmaceutical grade and contained 5.52% (w/w)
82 of protein on its composition. A purification of ChS was, thus, performed by ethanol precipitation.
83 To do so, commercial ChS was solubilised in water at 5% (w/w, 200 mL) and poured over 250 mL of
84 ethanol. This was left to rest at 4 °C overnight. After that, the dispersion was centrifuged (22 000 g, 4
85 °C, 1 h). The polymer was recollected from the pellet, freeze-dried and stored in a desiccator for
86 further use. All the experiments described below were performed with purified ChS.

87 2.3 Microparticle production

88 ChS microparticles were prepared by spray-drying, either unloaded or containing an association
89 of the antitubercular drugs INH and RFB. ChS was used at 2% (w/v) in all cases and the drug-loaded
90 microparticles were produced at ChS/INH/RFB mass ratio of 10/1/0.5. While INH was solubilised in
91 water, the hydrophobic character of RFB required the use of co-solvents. Ethanol 70% (v/v) and HCl
92 0.01 M were tested for this end and the obtained RFB solution was then added to the previously

93 formed ChS/INH solution. When ethanol was used, final water/ethanol ratio of 80/20 (v/v) was
94 applied in the spraying solution, while HCl was used at a final concentration of 0.002 M.

95 Microparticles were produced from both solutions using a laboratory mini spray-dryer (Büchi
96 B-290, Büchi Labortechnik AG, Switzerland) operating in open mode and equipped with a high-
97 performance cyclone. The operating parameters were: inlet temperature: 175 ± 2 °C, aspirator setting:
98 90%, feed rate: 0.7 ± 0.1 mL/min, and spray flow rate: 473 L/h. These conditions resulted in outlet
99 temperature of 110 ± 2 °C. After spray-drying, microparticles were collected, placed in a dark flask
100 and stored inside a desiccator until further use.

101 The spray-drying yield was calculated by gravimetry, comparing the total amount of solids
102 initially added with the resultant weight of collected microspheres.

103 2.4. Characterisation of ChS-based microparticles

104 The morphology of microparticles was characterised by field emission scanning electron
105 microscopy (FESEM; FESEM Ultra Plus, Zeiss, Germany). Dry powders were placed onto metal plates
106 and 5 nm thick iridium film was sputter-coated (model Q150T S/E/ES, Quorum Technologies, UK)
107 on the samples before viewing.

108 Microparticle size was estimated as the Feret's diameter and was directly determined by optical
109 microscopy (Microscope TR 500, VWR international, Belgium) from the manual measurement of 300
110 microparticles ($n = 3$).

111 The particle size distribution was determined by laser light scattering (Spraytec, Malvern
112 Panalytical, UK). To do so, approximately 15 mg of dry powder were dispersed in 15 mL of 2-
113 propanol and sonicated for 5 min. The volume-based size distribution was characterised and the
114 particle sizes below which 10%, 50% and 90% of the spray lies determined, being expressed as $Dv(10)$,
115 $Dv(50)$ and $Dv(90)$. From these values, span was calculated as follows:

$$\text{Span} = \frac{Dv(90) - Dv(10)}{Dv(50)} \quad (1)$$

116 The analyses were carried out in triplicate with an obscuration threshold of 10%.

117 Tap density (g/cm^3) was determined using a tap density tester (Densipro 250410, Deyman,
118 Spain), by measuring the volume of a known weight of powder before and after tapping, respectively
119 ($n = 3$). The determination of tap density involved tapping the sample until no further reduction of
120 powder volume was observed (average of 180 taps).

121 2.5. Determination of drug association and release

122 To determine the drug content, drug-loaded microparticles were incubated in 0.1 M acetic acid,
123 under magnetic stirring for 30 min, thus ensuring complete dissolution of microparticles. After
124 filtration ($0.45 \mu\text{m}$), quantification was performed by high performance liquid chromatography
125 (HPLC, Agilent 1100 series, Germany). A LiChrospher® 100 RP-18 ($5 \mu\text{m}$) column of 4 mm i.d. \times 250
126 mm length with security guard cartridge (CS, Germany) was used. Detection was performed by a
127 diode array detector at 275 nm. Mobile phase consisted in a mixture of 20 mM phosphate buffer pH
128 = 4 (A) and acetonitrile (B), flowing at a rate of 1.0 mL/min. Gradient started as 95:5 (A:B) for the first
129 5 min and changed to 30:70 at 8 min, remaining as such until 14 min. It returned then to initial
130 conditions at 16 min, for a total run time of 17 min. Under these conditions, retention times of INH
131 and RFB were 4.0 and 12.0 min, respectively. A linear calibration plot for INH and RFB was obtained
132 over the range of 10-400 $\mu\text{g}/\text{mL}$ ($n = 3$). Drug association efficiency (AE) and microparticle (MP)
133 loading capacity (LC) were estimated as follows ($n = 3$):

$$\text{AE} (\%) = (\text{Real amount of drug on MP} / \text{Theoretical amount of drug on MP}) \times 100, \quad (2)$$

$$\text{LC} (\%) = (\text{Real amount of drug on MP} / \text{Weight of MP}) \times 100, \quad (3)$$

134 Drug release was determined in PBS pH 7.4 added of 1% (v/v) Tween 80®. The assay respected
135 sink conditions, with the maximum amount of drug being always below 30% of maximum solubility

136 [9]. INH solubility was considered 274.0 ± 4.8 mg/mL [10], while that of RFB was 0.496 mg/mL [11].
 137 A determined amount of microparticles (30 mg) was incubated with the medium (10 mL), at 37 °C,
 138 under mild shaking (100 rpm, orbital shaker OS 20, Biosan, Latvia). Samples (1 mL) were periodically
 139 collected and the amount of each drug quantified as indicated above ($n = 3$). Similarity factor (f_2) was
 140 used to compare the release profile of the different drugs and formulations (similar profiles present
 141 a f_2 value not lower than 50) [12].

142 2.6. Molecular mass distribution

143 High performance size exclusion chromatography (HPSEC) was selected to define the molar
 144 mass profile of ChS, both in the form of polymer and microparticles (unloaded). To perform the
 145 analyses, the HPLC equipment (Agilent 1100 series, Germany) was supplied with two columns (300
 146 \times 7.8 mm) in series (TSKGel G3000PWXL and TSKGel G2500PWXL, Tosoh Bioscience, Stuttgart,
 147 Germany), along with a PWX-guard column (40 \times 6 mm). Samples were filtered (0.45 μ m) and
 148 analysed using a refractive index (RI) detector, under the following conditions: 70 °C, Milli-Q water
 149 as mobile phase and flow rate of 0.4 mL/min. Dextrans (DX) with molecular weight ranging from 12
 150 to 80 kg/mol (Fluka, MO, USA) were used as standards. Analyses were performed at least in
 151 duplicate.

152 2.7. Rheology: Steady-state shear measurements

153 A rheological study was performed on ChS polymer, unloaded ChS microparticles and
 154 ChS/INH/RFB microparticles prepared with either ethanol or HCl. Aqueous dispersions at 10 g/L
 155 were prepared in all cases, dispersing the corresponding amount of sample in distilled water, under
 156 stirring for 15 min, at room temperature. Steady-shear flow curves in terms of apparent viscosity vs.
 157 shear rate were conducted on a controlled-stress rheometer (MCR 302, Paar Physica, Austria) using
 158 a plate-plate geometry (25 mm diameter, 0.5 mm gap). Flow measurements were obtained by
 159 decreasing and, then, increasing shear rate, following a logarithmic ramp to assess the presence of
 160 hysteresis. All trials were made at 25 °C and were controlled by a Peltier system (± 0.01). In all cases,
 161 aqueous dispersions were sealed with light paraffin oil to avoid water loss during measurements and
 162 were rested for 10 min in the measurement system to allow sample structural equilibration. All
 163 experiments were performed at least in triplicate.

164 2.8. Aerodynamic characterisation of microparticles using an Andersen cascade impactor

165 HPMC size 3 capsules were filled with 30 mg of ChS/INH/RFB dry powder, either prepared with
 166 ethanol or HCl. The content of three capsules was discharged in each aerodynamic test using the
 167 medium resistance RS01® inhaler (Plastiap SpA, Italy). The experiments were performed in triplicate.
 168 The device was connected to the Andersen cascade impactor (ACI, Copley Scientific, UK) which
 169 operates at 60 L/min, ensuring a pressure drop of 4 kPa through the device. This was activated for 4
 170 s in order to let 4 L of air passing through the system, thus complying with the standard procedure
 171 described by USP 38 and Ph.Eur.8 [13, 14].

172 ACI separates particles according to their aerodynamic diameter and it was assembled using the
 173 appropriate adaptor kit for the 60 L/min air flow test. Cut-offs of the stages (-1 to 6) at the air flow
 174 rates adopted in this work are reported in Table 1. A glass fiber filter (Whatman, Italy) was placed
 175 right below stage 6 to collect particles with diameter lower than that of stage 6 cut-off.

176

177 **Table 1.** Cut-off aerodynamic diameter (μ m) for stages of Anderson cascade impactor (ACI) used at
 178 60 L/min.

179	Stage -1	Stage -0	Stage 1	Stage 2	Stage 3	Stage 4	Stage 5	Stage 6
180	8.60	6.50	4.40	3.20	1.90	1.20	0.55	0.26

181 The plates of the impactor were coated with a thin layer of ethanol containing 1% (w/v) Tween
182 20® to prevent particle bounce. The drugs were recovered from the apparatus with water/acetonitrile
183 mixture (50/50, v/v) and quantified by HPLC (Agilent 1200 series, Germany), as described above.

184 The quantification of drugs deposited inside the impactor enabled the calculation of different
185 aerodynamic parameters. The emitted dose (ED) is the amount of drug ex-device, considered the total
186 amount of drug collected in the impactor (induction port, stages -1 to 6 and filter). The metered dose
187 (MD) is the amount of drug loaded in the capsules, determined by adding all the drug collected in
188 the impactor and in the inhaler (device, induction port, stages -1 to 6 and filter). The mass median
189 aerodynamic diameter (MMAD) was determined by plotting the cumulative percentage of mass less
190 than the stated aerodynamic diameter on probability scale versus aerodynamic diameter on
191 logarithmic scale. The fine particle dose (FPD) corresponds to the mass of drug particles with
192 aerodynamic diameter lower than 5 µm calculated using the particle size distribution equation
193 obtained from the ACI analysis. The fine particle fraction (FPF) is the ratio between FPD and MD.

194 The recovery (%) is the percentage of MD versus the labelled dose. The recovery ranged within
195 86-91% in all the experiments, being thus coincident with the requisites of the pharmacopeia [13, 14].

196 2.9. *In vitro* determination of antitubercular activity

197 The *in vitro* antitubercular efficacy of microparticles was evaluated against *Mycobacterium bovis*
198 bacillus Calmette-Guérin (BCG). The Minimum Inhibitory Concentration (MIC) value of free and
199 microencapsulated INH and RFB was determined by the microdilution method [15]. ChS/INH/RFB
200 microparticles and the powders of free drugs were exposed to UV light for 10 min to provide
201 sterilisation. Then, a solution of each dry powder was prepared at 1 mg/mL and diluted to the desired
202 drug concentrations. First, 1 mg/mL stock solution of INH was prepared by dissolving the drug in
203 the M7H9 supplemented medium. RFB was previously solubilised in DMSO (1 mg/mL) and then
204 diluted with M7H9 broth. Drug stock solutions were mixed to reach concentrations corresponding to
205 drug loadings. Two-fold dilutions of the antibiotics/formulations were performed to obtain final
206 concentrations of RFB from 0.001 to 0.125 µg/mL and of INH from 0.008 to 1 µg/mL. A 1/10 dilution
207 of a McFarland 1.0 turbidity standard suspension of *M. bovis* BCG was inoculated (20 µL of inoculum
208 in 180 µL of medium or test solution) with a multichannel pipette, delivering approximately 10⁴ CFU
209 per well. The outside lane of wells (a frame-like) were filled with sterile distilled water to avoid the
210 evaporation of microplate content. The plates were covered with the lid, sealed with parafilm and
211 incubated at 37 °C (Binder, USA) for 7 days. Afterwards, 30 µL of MTT sterile solution were added
212 to each well, followed by 4 h of incubation at 37 °C. Then, 50 µL of DMSO were added into the wells
213 to solubilise the tetrazolium blue crystals that were produced, which is proportional to the growth
214 rate of mycobacteria. The absorbance was measured by spectrophotometry (Infinite M200, Tecan,
215 Austria) at 540 nm. The assays were performed in triplicate. The MIC value was considered the lowest
216 that inhibited mycobacteria growth by 95 – 100% [16, 17].

217 2.10. Statistical analysis

218 The student t-test and the one-way analysis of variance (ANOVA) with the pairwise multiple
219 comparison procedures (Holm-Sidak method) were performed to compare two or multiple groups.
220 All analyses were run using Sigmaplot (version 12.5) and differences were considered to be
221 significant at a level of $p < 0.05$.
222

223 3. Results and discussion

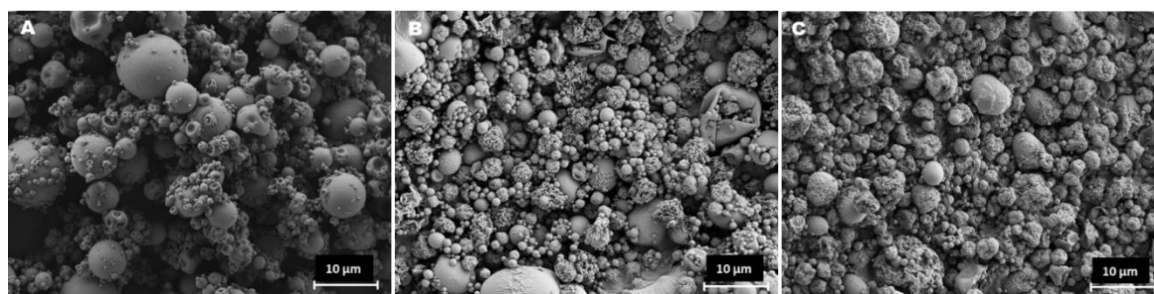
224 3.1. Preparation and characterisation of ChS microparticles

225 ChS undergone an initial step of purification in order to maximise polymer content and avoid
226 the presence of proteins. The purified polymer was used to prepare microparticles loaded with a
227 combination of INH and RFB. As far as we know, this is the first time that ChS is being proposed as
228 platform for the inhalable therapy of tuberculosis. The microparticles were prepared by spray-drying,

229 resulting in yields of approximately 85% (Table 2), which were deemed very satisfactory [18]. INH is
230 a hydrophilic drug, but RFB has hydrophobic character, thus requiring solvents other than water to
231 solubilise. Despite spray-drying permits processing a suspension, not requiring the solubilisation of
232 all components of the spraying dispersion, it was decided to provide the complete solubilisation of
233 both drugs in order to ensure a higher level of homogeneity in the final product. Ethanol and HCl
234 were, then, tested as co-solvents of RFB, being mixed with water.

235 Disregard of the used solvent, the obtained particles generally exhibited a spherical shape, as
236 observed in Figure 1. Moreover, the association of the drugs translated into the production of more
237 wrinkled and corrugated microparticles (Figures 1B and 1C) comparing with the unloaded
238 counterparts (Figure 1A). Objectively, the use of different solvents had no effect on microparticle
239 morphology. The literature reports already several formulations of ChS microparticles, although
240 none used spray-drying and different co-excipients were included in all cases. Nevertheless, it was
241 interesting to notice that the spherical shapes and smooth surfaces observed in the unloaded
242 microparticles reported herein, were common characteristics [19-21]. The microphotographs suggest
243 that the size of drug-loaded particles tends to decrease upon drug association, having a higher
244 number of small particles comparing with unloaded particles. Comparing the Dv_{50} of the produced
245 microparticles, a significant decrease from 9.6 μm of the unloaded particles to 4.1 μm of the drug-
246 loaded counterparts was observed ($p < 0.05$, Table 2). This size range is reported as adequate to
247 potentiate phagocytosis by macrophages [22], which could be advantageous in the treatment of
248 intracellular diseases or vaccination approaches [23, 24].

249



250

251 **Figure 1.** Microphotographs of unloaded ChS microparticles (A) and ChS/INH/RFB microparticles
252 produced with water-ethanol (B) and water-HCl (C) as solvents, as obtained by scanning electron
253 microscopy. Scale bar is 10 μm .

254 The tap density was determined to be around 0.5 – 0.6 g/cm^3 for the dry powders corresponding
255 to unloaded microparticles and microparticles produced with HCl, decreasing to 0.3 g/cm^3 for
256 microparticles produced with ethanol. A lower density has been correlated with less cohesive
257 powders and better flowability [25], comprising an obvious advantage. INH was associated to
258 microparticles with an efficiency around 95%, independently of the used solvent. However, although
259 not very accentuated, a significant difference was observed in RFB association, which was 59% when
260 ethanol was used *versus* 67% when HCl was the solvent ($p < 0.05$). The loading capacity was 8.2% for
261 INH, varying within 2.6% and 2.9% for RFB, values that compare with the theoretical 8.6% for INH
262 and 4.3% for RFB.

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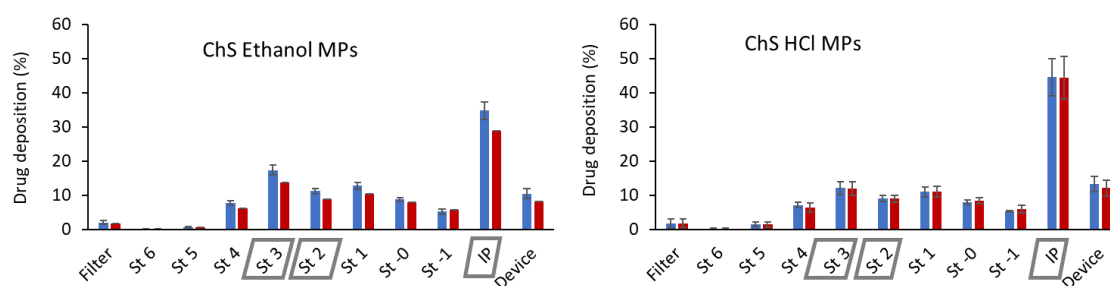
267 **Table 2.** Production yield (PY), median volume particle size (Dv50), tap density, drug association
 268 efficiency (AE) and loading capacity (LC) of ChS microparticles (mean \pm SD, n = 3). Different letters
 269 indicate statistical significant difference within the same parameter.

Microparticles	PY (%)	Dv50 (μm)	Span	Tap density (g/cm^3)	AE%	LC%
ChS	73.3 \pm 4.4 ^a	9.6 \pm 0.2 ^c	2.0 \pm 0.0 ^e	0.50 \pm 0.01 ^g	n. a.	n. a.
ChS/INH/RFB (ethanol)	83.4 \pm 1.0 ^b	4.1 \pm 0.1 ^d	2.9 \pm 0.1 ^f	0.32 \pm 0.03 ^h	INH: 94.9 \pm 5.7 ^j	INH: 8.2 \pm 0.5 ^m
					RFB: 59.0 \pm 6.9 ^k	RFB: 2.6 \pm 0.3 ⁿ
ChS/INH/RFB (HCl)	84.9 \pm 1.1 ^b	4.1 \pm 0.0 ^d	3.1 \pm 0.1 ^f	0.58 \pm 0.02 ⁱ	INH: 94.6 \pm 4.0 ^j	INH: 8.2 \pm 0.3 ^m
					RFB: 67.6 \pm 1.4 ^l	RFB: 2.9 \pm 0.1 ^o

270 AE: association efficiency, ChS: chondroitin sulfate, INH: isoniazid, LC: loading capacity, PY: production yield,
 271 RFB: rifabutin, n.a.: not applicable

272
 273 Figure 2 shows the stage-by-stage deposition profiles of both drugs encapsulated in the tested
 274 microparticles. The similarity of the profiles (INH *vs* RFB) indicates that the two drugs were equally
 275 distributed in the various stages. This suggests the adequacy of spray-drying as technique to produce
 276 microparticles with drug combination, resulting in homogeneous composition independently of
 277 particle sizes, which led to co-deposition of drugs. A statistically significant difference in the
 278 deposition of both formulations was found in the induction port, and in stages 2 and 3 ($p < 0.05$). This
 279 translated into differences in the calculated aerodynamic characteristics, which are displayed in Table
 280 3. The dose emitted from the inhaler was very satisfactory in all cases, reaching 90%. This is indicative
 281 of the suitability of ChS to be used as microparticle matrix material in spray-drying, producing
 282 microparticles with good flowing capacity with any of the tested solvents. Similar MMAD values (3.8
 283 – 4.0 μm) were obtained for both formulations. This aerodynamic diameter fits the range considered
 284 suitable to reach the respiratory zone, which is 1–5 μm [26, 27], thus being adequate for a tuberculosis
 285 therapy approach. Moreover, FPF of 34% - 44% was determined, indicating that this fraction of the
 286 microparticles has the aerodynamic characteristics for a deep lung deposition.

287



288

289 **Figure 2.** Stage-by-stage deposition profiles of isoniazid (blue) and rifabutin (red) inside the Andersen
 290 cascade impactor after aerosolisation of chondroitin sulfate (ChS) microparticles (MPs) with RS01
 291 high resistance inhaler operated at 60 L/min (values are mean \pm SD, n = 3). Grey boxes in X axis
 292 represent $p < 0.05$ comparing the same stage between the two formulations. IP: induction port; St:
 293 stage.

294

295 Despite having very similar MMAD, microparticles produced using ethanol as solvent provided
 296 a lower deposition in the induction port and higher deposition on stages 2 and 3, in the sequence of
 297 the determined higher FPD, enabling deep lung delivery of 1.5 mg of RFB and 3.1 mg of INH. The
 298 presence of ethanol apparently benefited the formulation, resulting in a less cohesive powder with

299 better flowing properties. Despite the slightly lower drug association efficiency, the resulting
 300 aerodynamic properties provide increased drug accumulation in the respiratory zone comparing
 301 with that evidenced by microparticles produced with HCl.

302

303 **Table 3.** Aerodynamic characteristics of ChS/INH/RFB (10/1/0.5, w/w) microparticles (mean ± SD, n =
 304 3).

Microparticles	Powder Emitted dose (%)	Drug	MMAD (µm)	GSD (µm)	FPD (mg)	FPF (%)
ChS/INH/RFB (ethanol)	90.9 ± 1.0 ^a	INH	3.8 ± 0.1 ^b	1.9 ± 0.1 ^c	3.1 ± 0.3 ^d	43.7 ± 2.4 ^g
		RFB	3.9 ± 0.1 ^b	2.0 ± 0.1 ^c	1.5 ± 0.1 ^e	42.6 ± 1.7 ^g
ChS/INH/RFB (HCl)	89.6 ± 1.8 ^a	INH	4.0 ± 0.2 ^b	2.0 ± 0.0 ^c	2.9 ± 0.1 ^d	35.0 ± 1.7 ^h
		RFB	4.0 ± 0.3 ^b	2.1 ± 0.1 ^c	1.2 ± 0.1 ^f	34.0 ± 3.7 ^h

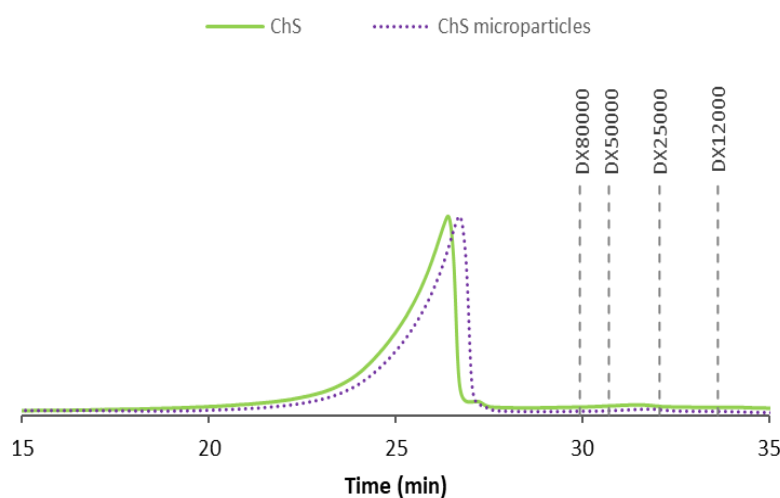
305 FPD: Fine particle dose; FPF: Fine particle fraction; GSD: Geometric standard deviation; MMAD: Mass median
 306 aerodynamic diameter.

307

308 3.2. Impact of spray-drying on polymer characteristics

309 Spray-drying was the technique selected to produce the microparticles, as it permits tailoring
 310 their properties for deep lung delivery. Nevertheless, it was deemed important to study the impact
 311 of the process on the polymer properties. The molecular mass distribution of ChS was investigated
 312 as raw material and after the spray-drying process (unloaded microparticles), and the obtained
 313 profiles are exhibited in the Figure 3. In both samples, the molecular mass was greater than 80 kDa,
 314 which was the highest standard available. The peaks of the two samples evidenced similar
 315 distribution profiles and, although showing a slight difference, the molecular mass was considered
 316 similar. Rani et al. (2017) isolated chondroitin sulfate from the cartilage of chicken keel bone and
 317 reported the peak of molecular mass at 100 kDa [28].

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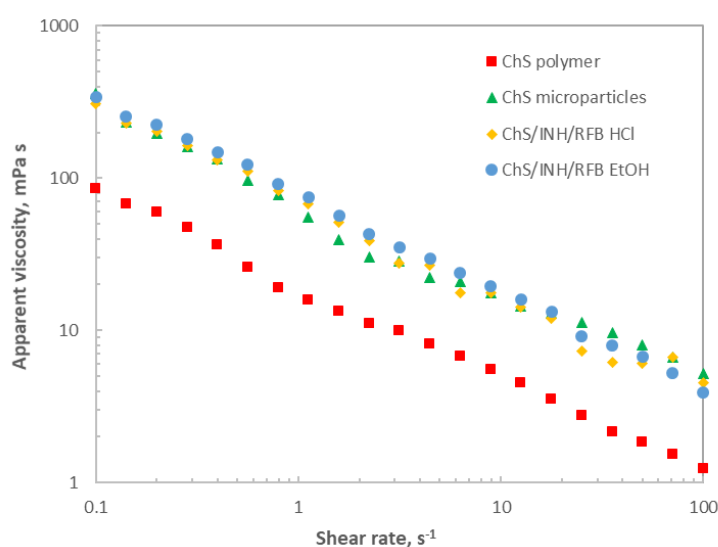
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320 **Figure 3.** High performance size exclusion chromatography (HPSEC) profiles of chondroitin sulfate
 321 (ChS) as raw material, represented by continuous line, and after processing by spray-drying
 322 (unloaded microparticles), illustrated by dotted line.

323

324 A rheological study was also performed, evaluating the rheological behaviour of the polymer
 325 before and after spray-drying and further determining the effect of associating the drugs. Figure 4
 326 displays the steady shear flow curves of the tested aqueous dispersions (10 g/L) at 25 °C. In all cases,
 327 shear-thinning behaviour was observed, with the apparent viscosity decreasing with increasing shear
 328 rate (about 2 decades). The apparent viscosity drop can be explained by the alignment of long chain
 329 molecules with each other at the highest shear rates, leading to easier flows. This behaviour is
 330 characteristic of pseudoplastic fluids, which may be described by the power law model, and the
 331 viscosity trend is consistent with a typical non-entangled polymer behaviour in the dilute regime
 332 [29]. At a fixed shear rate, it can be clearly observed that ChS polymer exhibited the lowest apparent
 333 viscosity over the tested range. After processing by spray-drying, higher apparent viscosity values
 334 were observed (about 2-fold), suggesting an effect of the process on polymer characteristics. Since
 335 slight differences were identified in the HPSEC profiles, this behaviour suggests that some aggregates
 336 of the proper polymer chains may have formed during microparticle processing, which could involve
 337 higher flow resistance and, consequently, higher apparent viscosities. The samples corresponding to
 338 post-spray-drying products, all registered similar behaviour, indicating an absence of effect of both
 339 drug association and used solvents. Taking into account the envisaged application as inhalable
 340 tuberculosis therapy, this behaviour may help microparticles to maintain their “particulate” shape
 341 long enough to be phagocytosed prior to dissolution. It should be further highlighted that no
 342 hysteresis was observed in the tested samples, with the consequent advantage from the industrial
 343 point of view.

344



345

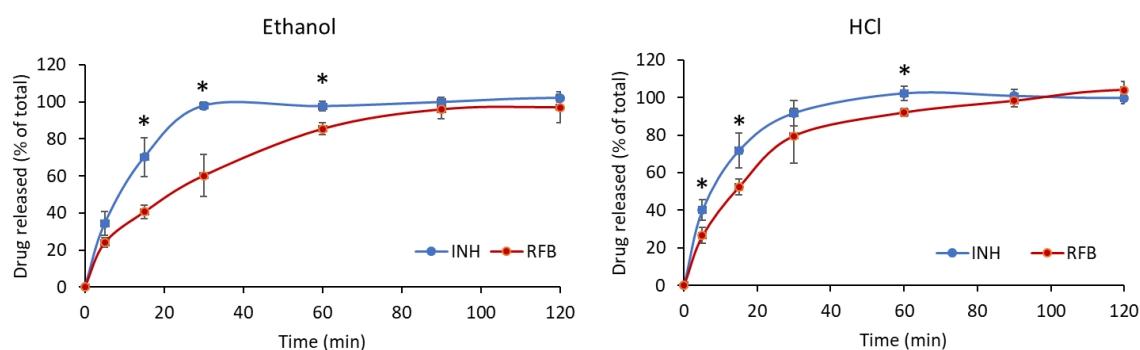
346 **Figure 4.** Steady shear flow curves for aqueous dispersions of chondroitin sulfate (ChS) polymer, ChS
 347 unloaded microparticles and ChS/INH/RFB microparticles produced with ethanol (EtOH) or HCl, at
 348 polymer concentration of 10 g/L and 25 °C.

349 3.4. INH and RFB release from ChS microparticles

350 Considering an application in tuberculosis therapy, drug release profiles were determined in
 351 PBS pH 7.4 added of 1% Tween 80®. The pH resembles that of the lung lining fluid [30] and the
 352 addition of Tween 80® intends to mimic the surfactant content [31]. The results are displayed in Figure

353 5 and evidence rapid release of the drugs, as 100% release was registered within 60 min for INH and
 354 120 min for RFB, in both formulations. The surface irregularities of microparticles, commented in
 355 section 3.1., certainly contributed to the rapid release, due to the increased contact with the release
 356 medium. Additionally, the high solubility of the polymeric matrix was another contributing factor,
 357 with the hydrophilic character of ChS resulting in rapid dissolution of the particle matrix and
 358 permitting a fast release of the drugs. Nevertheless, RFB tends to exhibit slower release than INH (p
 359 < 0.05). The similarity factor f_2 was used to compare the release of both drugs in each formulation
 360 and also the release of each drug in the two different formulations. The comparison between INH
 361 and RFB resulted in f_2 factor of 30 for microparticles produced with ethanol and 43 for microparticles
 362 produced with HCl. As f_2 values are lower than 50, it is concluded that differences exist in the release
 363 profiles of both drugs in both microparticle formulations. This is possibly a consequence of the higher
 364 molecular weight of RFB, along with its hydrophobicity, which contrasts with the hydrophilic
 365 character of INH.

366 When comparing the release of each drug from both formulations, no differences were observed
 367 for INH, as this follows a similar release pattern from both microparticles, which is reinforced by an
 368 f_2 factor of 64. In contrast, Figure 5 evidences a difference in RFB release profile when the two
 369 microparticle formulations are compared. In this context, microparticles produced with ethanol
 370 provided somewhat slower release comparing with those produced with HCl ($p < 0.05$). A f_2 factor of
 371 46 confirms the dissimilar profiles. Considering the results of the rheological study, no differences in
 372 the characteristics of the polymer were found comparing the two spray-dried microparticle
 373 formulations that could justify this behaviour. Therefore, one possible explanation for this effect
 374 could be that HCl remaining in the formulation after spray-drying facilitated RFB solubilisation
 375 afterwards, by providing a certain degree of protonation [32]. Another reason could be the existence
 376 of a different pattern of dispersion of RFB within the microparticle matrix in each formulation. If RFB
 377 is located more at the surface in microparticles produced with HCl, comparing with those produced
 378 with ethanol, this could justify the observed profiles. Nevertheless, further studies would be needed
 379 to study this possibility.



380

381 **Figure 5.** - *In vitro* release of isoniazid (INH) and rifabutin (RFB) from ChS/INH/RFB (10/1/0.5, w/w)
 382 microparticles in PBS pH 7.4-Tween 80®, at 37 °C (mean \pm SD, $n = 3$). * $p < 0.05$ comparing INH and
 383 RFB at a given time point.

384 The encapsulation of the same pair of drugs was reported in other works that used different
 385 microparticle matrix materials. The comparison of release profiles supports the conclusion that the
 386 matrix rules the release pattern. Although ChS is a hydrophilic polymer, it was able to provide slower
 387 release of these drugs in comparison with microparticles composed of fucoidan, another hydrophilic
 388 polymer, which has shown total release of INH and RFB in 30 min [33]. Locust bean gum, in turn,
 389 provides more viscous solutions and, thus, sustained the release of INH until 120 min and RFB up to
 390 240 min [34], while low molecular weight chitosan microparticles released completely INH and RFB
 391 in 120 min [35]. Also chitosan, but with higher molecular weight, provided more sustained release of
 392 INH, up to 40-60 h, which was demonstrated to be influenced by the polymer properties and

393 crosslinking degree [36]. In fact, polymeric particles devoid of crosslinking agents typically show a
394 faster release of the associated drugs [37].

395 It should be stressed that this assay does not mimic *in vivo* conditions, as it is well known that
396 the alveolar epithelium is covered by a thin layer (~0.2 µm) of lung lining fluid [38]. Therefore,
397 microparticles are expected to be only partially in contact with the fluid, and not immersed in it, as
398 in this assay. Consequently, we anticipate that, in *in vivo* conditions, these microparticles will show
399 slower drug release kinetics, allowing microparticle internalisation by macrophages before particle
400 dissolution and complete drug release.

401

402 3.5. Antibacterial activity of drugs after microencapsulation

403 One of the most commonly administered vaccines worldwide and the only one against
404 tuberculosis is based on *M. bovis* BCG, a live attenuated vaccine. BCG vaccine is commonly used in
405 experiments as a surrogate for virulent *M. tuberculosis* [39]. For this reason and because BCG is a slow
406 growing strain with similarities to *M. tuberculosis*, it was selected to perform this assay, determining
407 the MIC value of INH and RFB and the effect of spray-drying on this parameter. MIC value of free
408 and encapsulated drugs was determined by exposing *M. bovis* BCG during 7 days to different
409 concentrations of free drugs, association of free INH and RFB (at INH/RFB = 2/1, to mimic the drug
410 ratio present in the microparticles) and to the two different microparticle formulations. The MIC
411 value observed for the free drugs was 0.125 µg/mL for INH and 0.004 µg/mL for RFB, indicating
412 higher sensitivity of *M. bovis* to RFB than to INH. The literature reports variable values, in
413 dependency of the bacterial strains and determination methods that are used, but the attained values
414 were similar to those reported in other studies [39, 40]. When a combination of both drugs was
415 analysed, RFB ruled the inhibition effect and the same 0.004 µg/mL concentration (0.004 RFB and
416 0.008 INH) led to inhibition of the growth of *M. bovis*. Drug-loaded microparticles evidenced the same
417 MIC value observed for the free drugs in combination, disregard of the formulation. This provides a
418 dual indication: 1) that the spray-drying process does not affect the antitubercular effect of the drugs
419 and 2) that the used solvent has no effect on this property of microparticles.

420 5. Conclusions

421 ChS microparticles were able to successfully associate a combination of INH and RFB, with
422 efficiency between 60% and 95%, either using ethanol or HCl as co-solvent in the spray-drying
423 process. The microencapsulation process induced some alterations in the rheological properties of
424 the polymer, which was independent of the used solvent. In any case, the antitubercular efficacy of
425 the drugs remained unchanged after encapsulation. Despite both the produced formulations of
426 microparticles revealed similar aerodynamic properties and *in vitro* respirability, the dry powder
427 prepared using ethanol as co-solvent showed a better ability to sustain the release of RFB. Regarding
428 the objective of associating INH and RFB, two drugs of opposite aqueous affinity, in the same
429 formulation of ChS microparticles aimed at lung delivery, the use of ethanol as co-solvent was
430 considered to be advantageous.

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