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Growth diagram and magnetic properties of hexagonal LuFe₂O₄ thin filmsWenbin Wang,^{1,2} Zheng Gai,^{2,3} Miaofang Chi,² Jason D. Fowlkes,³ Jieyu Yi,¹ Leyi Zhu,⁴ Xuemei Cheng,⁵ David J. Keavney,⁶ Paul C. Snijders,² Thomas Z. Ward,² Jian Shen,^{1,7,*} and Xiaoshan Xu^{2,†}¹*Department of Physics, University of Tennessee, Tennessee 37996, USA*²*Materials Science and Technology Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, USA*³*Center for Nanophase Materials Sciences, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, USA*⁴*Materials Science Division, Argonne National Laboratory, Argonne, Illinois 60439, USA*⁵*Department of Physics, Bryn Mawr College, Bryn Mawr, Pennsylvania 19010, USA*⁶*Advanced Photon Source, Argonne National Laboratory, Argonne, Illinois 60439, USA*⁷*Department of Physics, Fudan University, Shanghai 200433, China*

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A growth diagram of Lu-Fe-O compounds on MgO (111) substrates using pulsed laser deposition is constructed based on extensive growth experiments. The LuFe₂O₄ phase can only be grown in a small range of temperature and O₂ pressure conditions. An understanding of the growth mechanism of Lu-Fe-O compound films is offered in terms of the thermochemistry at the surface. Superparamagnetism is observed in the LuFe₂O₄ film and is explained in terms of the effect of the impurity hexagonal LuFeO₃ (h-LuFeO₃) phase and structural defects.

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I. INTRODUCTION

Multiferroics have attracted great attention recently because of their promising new functionality and intriguing fundamental science. A multiferroic material with a large ferroic polarization, high ordering temperature, and strong coupling between the ferroic orders is ideal for applications. So far, those desired properties have not been realized in a single phase material. Multiferroics like BiFeO₃ where the magnetic and electric orders originate from different parts of the structure have high ordering temperatures but weak coupling between different orders.¹ Other materials, like TbMn₂O₅, exhibiting ferroelectricity due to the broken symmetry caused by the spiral magnetic moment have strong magnetoelectric coupling.² However, here the ordering temperature is very low and the electric polarization is small. LuFe₂O₄ contains layers of Fe₂O₂ with a triangular lattice that are sandwiched by LuO₂ layers. Combined with the mixed valance of Fe, the Fe₂O₂ layers in the triangular lattice form a charge ordered state at $T_{CO} = 320$ K, followed by a ferrimagnetic order at $T_N = 240$ K (Ref. 3). Significant changes in dielectric properties have been observed upon application of a small magnetic field at room temperature.⁴ The relatively high transition temperature, large polarization, high magnetic coercivity, and the strong magnetoelectric coupling make LuFe₂O₄ a unique multiferroic material. Recently, the possibility of fast switching and high tunability of LuFe₂O₄ due to the electronic origin of its charge order was demonstrated.⁵

Compared to the large amount of effort to study bulk LuFe₂O₄, there are only a couple of reported attempts to grow LuFe₂O₄ thin films on α -Al₂O₃ (001) and on Si substrates using pulsed laser deposition (PLD).^{6,7} Liu *et al.* found that the growth of LuFe₂O₄ on α -Al₂O₃ (001) (with a target consisting of a sintered mixture of Lu₂O₃ and Fe₂O₃) needs substrate temperatures as high as 850 °C (Ref. 6). In addition, a significant deviation of the Lu:Fe stoichiometry from 1:2 was observed, which was attributed to different ablation efficiencies of Lu and Fe in the target. This problem was circumvented by enriching the Fe concentration of the target

material. However, as a result, Fe₃O₄ and Fe₂O₃ impurities were introduced as intermediate layers between the LuFe₂O₄ film and the α -Al₂O₃ substrate.

In this paper, we present a comprehensive study on the growth of Lu-Fe-O compound thin films on MgO (111) substrates using pulsed laser deposition (PLD). The experimentally constructed growth diagram shows that the parameter space for growing epitaxial LuFe₂O₄ thin films turns out to be a narrow window of temperature and O₂ pressure, which creates significant experimental difficulties. Based on these results we have gained fundamental understanding of the growth of Lu-Fe-O compound films: the growth temperature needs to be high enough to stabilize the LuFe₂O₄ phase; on the other hand, the loss of Fe at high temperature also produces phases other than LuFe₂O₄. These two effects cause a narrow window of the growth condition producing LuFe₂O₄. Typical LuFe₂O₄ films appear to be superparamagnetic, which is consistent with the fact that the LuFe₂O₄ in the film is epitaxially sandwiched by an impurity phase of hexagonal LuFeO₃ (h-LuFeO₃). The current demonstration of epitaxial growth of LuFe₂O₄ thin films opens up new possibilities for studying multiferroicity of low-dimensional LuFe₂O₄, tuning of its properties, and eventual functionalization.

The paper is organized as follows. Section II describes the experimental conditions used in this work. Section III presents the experimental results including the growth diagram, structural characterizations, and magnetism. Explanations of the observed growth diagram and magnetism of the films are proposed in Sec. IV.

II. EXPERIMENTAL CONDITIONS

Lu-Fe-O compound thin films were grown using PLD with a KrF ($\lambda = 248$ nm) laser. The energy density of the laser is 2.5 J cm⁻² with a repetition rate of 1 Hz. The target-substrate distance is 3.5 cm. The thickness of the films grown in this study is approximately 100 nm. The substrates are MgO (111) single crystals annealed in O₂ for 24 hours at 1100 °C. The target material used is polycrystalline LuFe₂O₄, whose

properties are verified using powder x-ray diffraction (XRD) and a superconducting quantum interference device (SQUID). After growth, the sample heating is turned off so that the sample cools to below 200 °C at the same pressure as that of the growth condition within 5 minutes. The substrates were clamped on a heater with a Pt foil in between. The sample temperature was measured by a pyrometer using emissivity of 0.3. In principle, all the growth parameters described above will have to be scanned and optimized to realize the growth of high quality LuFe_2O_4 thin films. In this work, we are more focused on elucidating the mechanism of the growth. Therefore, fine scans of the substrate temperature and the O_2 pressure were carried out to map out the growth diagram involving the growth of more than one hundred samples, while all the other parameters were kept constant.

The growth was monitored using the reflection high energy electron diffraction (RHEED) from the heating up of the substrates before the growth to the cooling down after the growth. The RHEED images of the films in the paper were all taken right after the growth at the growth temperature. Transmission electron microscopy (TEM) work was carried out on a Cs-corrected FEI Titan 80/300-kV TEM/STEM. TEM specimen was prepared with traditional mechanical polishing followed by ion milling. High resolution Z-contrast images were acquired using 300KV and a beam size of 0.7 Å. The magnetic properties of the films were measured using SQUID. The subtraction of background is done by assuming that the magnetization saturates in a large field at 300 K.

III. RESULTS

A. Growth diagram

In this work, we start from the ternary phase diagram of the bulk Lu-Fe-O system, a section of which is shown in Fig. 1(a) at 1200 °C (Refs. 8 and 9). This system belongs to the D-type of lanthanoid-Fe-O compounds for which there are four stable three-element phases: LuFe_2O_4 (A) $\text{Lu}_2\text{Fe}_3\text{O}_7$ (B), LuFeO_3 (perovskite or P), and $\text{Lu}_3\text{Fe}_5\text{O}_{12}$ (garnet or G).¹⁰ In principle, one way to form a single LuFe_2O_4 phase is to keep atomic ratio Lu:Fe = 1:2 and vary the O_2 pressure, shown as a thick dashed line in Fig. 1(a).

To elucidate the mechanism of the growth of Lu-Fe-O compound films, we carried out fine scans of the substrate temperature and the O_2 pressure to map out the growth diagram. Figure 1(b) is the resulting experimental growth diagram. The important observations can be summarized as follows. (1) In the low temperature region the growth follows more or less the behavior predicted by the bulk phase diagram Fig. 1(a): at high pressure, the existing phases are LuFeO_3 , Lu_2O_3 , and h- LuFeO_3 ; when the pressure is decreased, the Fe_3O_4 phase starts to appear. This is consistent with the fact that LuFe_2O_4 and $\text{Lu}_2\text{Fe}_3\text{O}_7$ phases are not stable at low temperature.¹¹ (2) In the high temperature region, the growth deviates strongly from the thick dashed line in the bulk phase diagram Fig. 1(a) in that the Lu:Fe stoichiometry differs dramatically from that of the polycrystalline LuFe_2O_4 target. As shown in Fig. 2(a), the LuFe_2O_4 phase is not observed in the films grown at 1050 °C. The dominant phase is always Lu_2O_3 (L).¹² The concentration of LuFeO_3 rises with increasing O_2

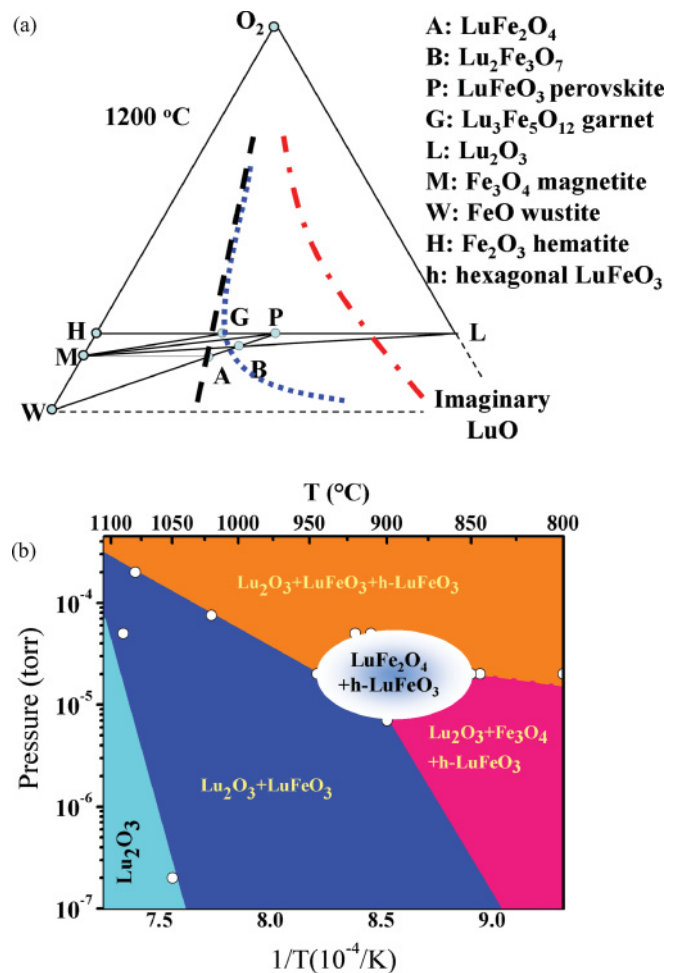


FIG. 1. (Color online) (a) A part of the phase diagram of the bulk Lu-Fe-O ternary system at 1200 °C. The thick dashed line represents the expected growth of the bulk Lu-Fe-O compound when the Lu:Fe stoichiometry is kept as 1:2 at 1200 °C (Refs. 8 and 9). The dash-dotted line depicts the actual growth of Lu-Fe-O compound film at high temperature while the dotted line indicates the growth of at optimal temperature for LuFe_2O_4 film in this work. (b) The experimental growth diagram of the Lu-Fe-O compound thin films on MgO (111) substrates. The subset of data points that define the boundaries are shown as small circles.

pressure. At high enough O_2 pressure, h- LuFeO_3 compounds start to form.¹¹ The formation of Lu-Fe-O compounds in the films qualitatively follows the dash-dotted line in Fig. 1(a). (3) Only in the small range of pressure and temperature indicated by the elliptical area in Fig. 1(b) is the growth of LuFe_2O_4 the most effective. In this case, the growth follows qualitatively the dotted line in Fig. 1(a). Typical XRD data are displayed in Fig. 2(c) showing both LuFe_2O_4 and h- LuFeO_3 , indicating a deviation of Lu:Fe stoichiometry from that of the target even in this narrow window.^{13,14}

B. Structural characterization

The combination of *in situ* structural characterization using RHEED and *ex situ* characterization by XRD allows assignment of the epitaxial relation between the existing phases and the substrates. The results are given in Table I.

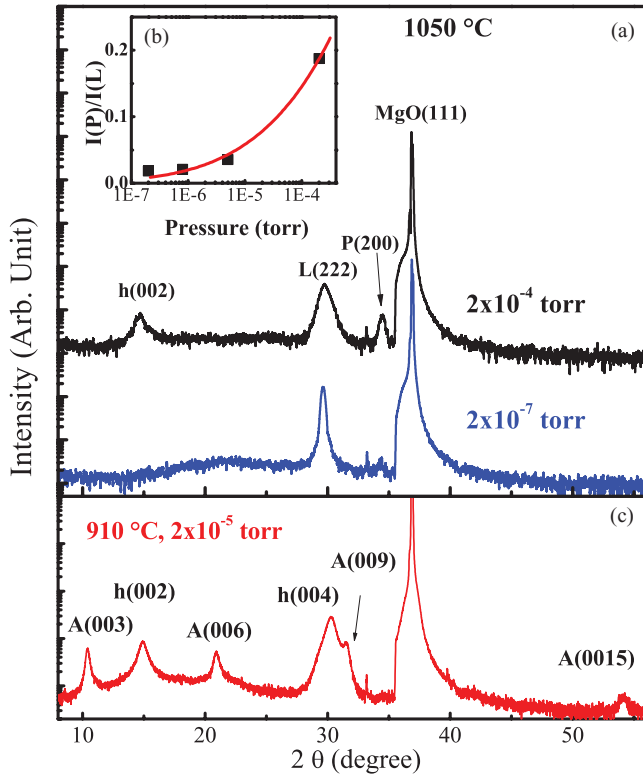


FIG. 2. (Color online) XRD data of the Lu-Fe-O films. (a) XRD data of the films grown at $T = 1050$ °C for two different O_2 pressures. (b) The ratio between the intensity of the P phase ($LuFeO_3$) $I(P)$ and L phase (Lu_2O_3) $I(L)$ as a function of the O_2 pressure at $T = 1050$ °C. The line is a fit with the thermochemistry model (see text). (c) Typical XRD data of a $LuFe_2O_4$ film that shows the $LuFe_2O_4$ c axis to be perpendicular to the substrate surface as expected. Also present is an h - $LuFeO_3$ as an impurity phase.

From the RHEED image, one can measure the in-plane lattice constant for the grown film. If three-dimensional (3D) island growth occurs, the RHEED images correspond to the diffraction pattern of the transmitted electron beam which contains more structural information. Figure 3 shows the RHEED images of the MgO (111) substrates and the $LuFe_2O_4$ films with the electron beams directed along MgO [1-10] or MgO [11-2]. The strong $LuFe_2O_4$ (003), (006), and (009) peaks observed in Fig. 2(b) indicate that the epitaxial relation is $LuFe_2O_4$ [001]// MgO [111], which is expected because both faces have three-fold rotational symmetry. The streaky RHEED patterns in Figs. 3(c) and 3(d) suggest quasi-two-dimensional (2D) growth of $LuFe_2O_4$. The in-plane lattice constants of the film can be calculated from the separation

TABLE I. Morphology and epitaxial relations of compounds grown on MgO (111) substrates.

Compound	Morphology	Epitaxial relation
$LuFe_2O_4$	Quasi 2D	[001]// MgO [111], [100]// MgO [1-10]
Lu_2O_3	Quasi 3D	[111]// MgO [111], [1-10]// MgO [1-10]
$LuFeO_3$	3D	[100]// MgO [111], [001]// MgO [1-10]
Fe_3O_4	3D	[111]// MgO [111], [1-10]// MgO [1-10]
h - $LuFeO_3$	Quasi 2D	[001]// MgO [111], [1-10]// MgO [1-10]

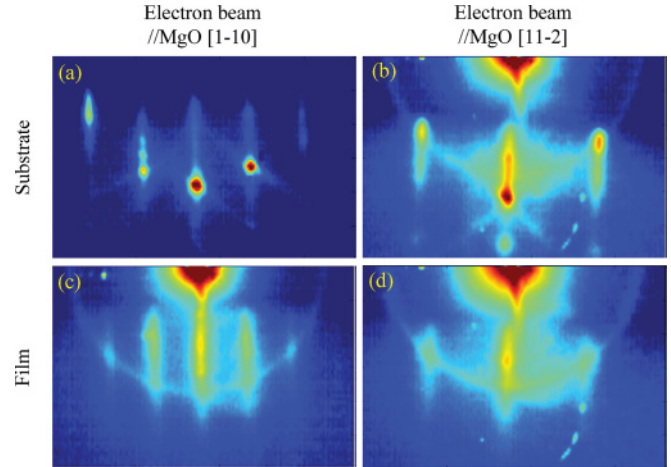


FIG. 3. (Color online) RHEED images corresponding to the patterns of the MgO (111) substrate taken with the electron beam along (a) [1-10] and (b) [11-2] directions, and to the pattern of $LuFe_2O_4$ film taken with the electron beam along (c) MgO [1-10] and (d) MgO [11-2] directions. All panels have the same scale.

of the streaks calibrated by the RHEED pattern of the MgO substrates. It is consistent with the $LuFe_2O_4$ lattice constant 3.44 Å within the experimental uncertainty of 2%. Hence, the in-plane epitaxial relation is $LuFe_2O_4$ [100]// MgO [1-10]. This is unexpected from the point of view of lattice matching, which predicts $LuFe_2O_4$ [100]// MgO [11-2] because a $\sqrt{3} \times \sqrt{3}$ supercell of $LuFe_2O_4$ with a 30° rotation along the [001] direction has less than 0.1% mismatch with a 1×1 of MgO (111) surface.¹⁵ Contrasting with the apparent quasi-2D growth of $LuFe_2O_4$, Lu_2O_3 forms quasi-3D structures on the substrate.¹⁵ However, the RHEED pattern suggests a face centered cubic structure with a lattice constant half of that of bulk Lu_2O_3 . The detailed structure is not clear at present. At low pressure and low temperature, the RHEED signal is dominated by the diffraction pattern of 3D Fe_3O_4 islands along the [11-2] direction, with the [111] direction perpendicular to the substrate surface. This is consistent with the XRD data. The lattice constants are the same as that of bulk Fe_3O_4 within the experimental uncertainty of 2%.

High resolution transmission electron microscopy (HRTEM) reveals the detailed structure of the $LuFe_2O_4$ films. As shown in Fig. 4(a), a layered structure of the film is obvious with some variation at different locations. The fast Fourier transforms (FFT) of the HRTEM image at different locations confirm the epitaxial relation observed from RHEED images: the FFT of the substrate [Fig. 4(d)] indicates the reciprocal lattice of MgO viewed from [11-2] direction. The FFT of the majority of the film [Fig. 4(c)] is consistent with the reciprocal lattice of $LuFe_2O_4$ viewed from the [1-10] direction, while at some locations [Fig. 4(b)] it suggests h - $LuFeO_3$ viewed from the [100] direction. These two phases $LuFe_2O_4$ and h - $LuFeO_3$ were further confirmed by direct observation using atomic-resolution Z-contrast imaging, which is shown in Figs. 4(e) and 4(f). The LuO_2 -FeO-FeO- LuO_2 ordering in the $LuFe_2O_4$ phase and the LuO_2 -FeO- LuO_2 ordering in the h - $LuFeO_3$ phase are clearly observed.

Although the intensity of the XRD peaks originating from the h - $LuFeO_3$ phase seems comparable to that of $LuFe_2O_4$

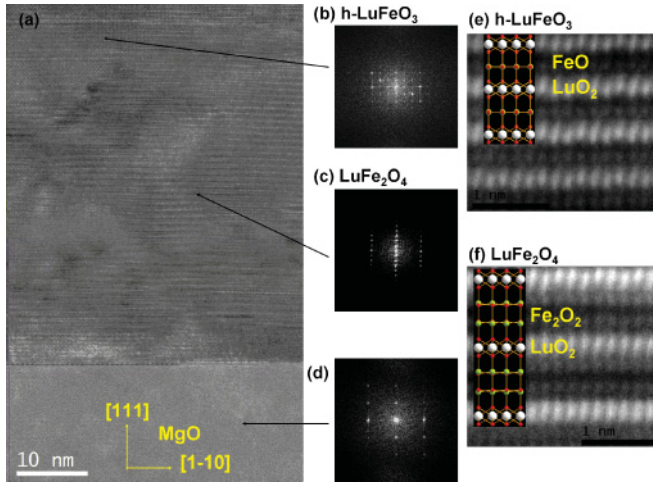


FIG. 4. (Color online) (a) Typical HRTEM image around the interface. (b)–(d) The Fourier transforms of various positions of (a), where (d) is from MgO substrate and (c) is from the majority of the film, and (b) is from the small top left part of the image. (e) and (f) are the atomic-resolution Z-contrast images corresponding to (b) and (c), respectively.

phase, the actual dominant phase is still LuFe₂O₄ due to the lower x-ray scattering cross section of the LuFe₂O₄ as compared with that of the h-LuFeO₃ phase. This is consistent with the low population of the h-LuFeO₃ phase in the HRTEM image. In addition, the RHEED patterns of h-LuFeO₃ and LuFe₂O₄ are supposed to be different according to their structures.^{16–18} The fact that the observed RHEED patterns do not show any indication of h-LuFeO₃ within the detection limit also suggests a low concentration of the h-LuFeO₃ phase in the films.

C. Magnetic properties

Ferrimagnetism, large magnetization, and giant coercivity are of the key properties of LuFe₂O₄ (Refs. 19–21). This makes the study of the magnetic properties of LuFe₂O₄ films critical. As shown in Fig. 5(a), little hysteresis is observed for these LuFe₂O₄ films, in contrast to the bulk.¹⁹ When magnetization is plotted against magnetic field over temperature (H/T), the data of 12, 50, 100, and 150 K fall on top of each other [Fig. 5(b)], indicating superparamagnetic behavior.^{22–24}

Assuming superparamagnetism to be present, one can calculate the magnetic moment from the slope of the low field magnetization data with

$$\frac{dM}{dH} = \frac{N\mu^2}{3kT}\mu_0, \quad (1)$$

where μ , N , μ_0 , and k are the moment of the superparamagnetic domains, number of the domains per unit volume, the vacuum permeability, and the Boltzmann constant, respectively.^{23–25} The magnetic moments normalized to their maximum value as a function of temperature are plotted in Fig. 5(c), which follow the temperature dependence of the bulk saturation magnetization closely, suggesting that the Neel temperature of the films is not very different from the bulk value of 240 K.

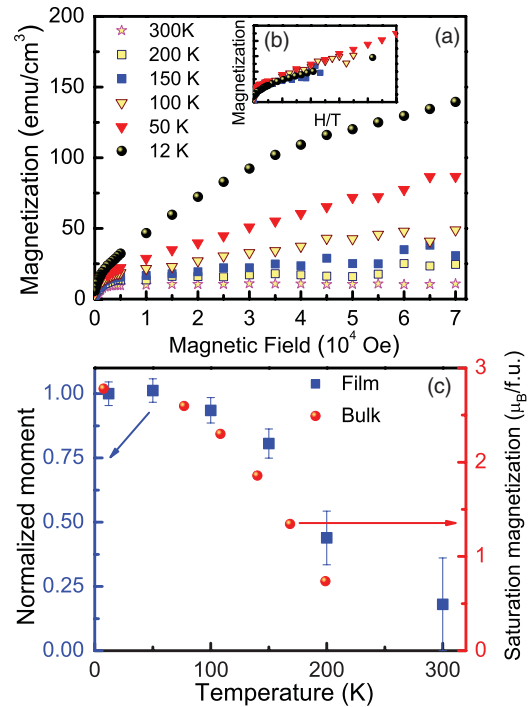


FIG. 5. (Color online) Magnetic properties of a typical LuFe₂O₄ film. (a) The field dependence of the magnetization at various temperatures. (b) Magnetization plotted against magnetic field over temperature (H/T). (c) The temperature dependence of the magnetic moment of the superparamagnetic phase normalized to the maximum value and the bulk saturation magnetization from Ref. 18. The magnetic field is perpendicular to the plane of the film.

IV. DISCUSSION

A. Growth diagram

The most surprising observation of the growth diagram is that the Lu-Fe-O compound formation at high temperature deviates strongly from the Lu:Fe stoichiometry of the target. Here we propose an explanation in terms of competition between nucleation and desorption of adatoms and its dependence on temperature and supersaturation.

The residence time τ_{ad} of an adsorbed atom is given by

$$\tau_{ad} = \frac{1}{\nu} \exp\left(\frac{E_{des}}{kT}\right), \quad (2)$$

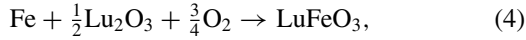
where ν is the vibrational frequency and E_{des} is the desorption energy. Clearly, the residence time of an adatom is shorter at high temperature due to the higher desorption rate. The observed loss of Fe atoms suggests a smaller desorption energy (higher desorption rate) for Fe atoms. At low temperature, because $\exp(\frac{E_{des}}{kT})$ is large for both Lu and Fe adatoms, the Lu:Fe stoichiometry can be close to that of the target.

The nucleation speed of deposited adatoms is

$$J_{nuc} \propto \left(\frac{\Delta\mu^*}{T}\right)^{1/2} \exp\left(-\frac{\kappa}{\Delta\mu^*kT}\right), \quad (3)$$

where $\Delta\mu^*$ is the effective supersaturation (molar bulk Gibbs free energy change with surface energy consideration), while κ is proportional to the square of the edge energy of the nuclei per unit length.²⁶ Therefore, at the high temperature limit,

the nucleation speed decreases with temperature and a high supersaturation favors a high nucleation speed. Consider the reaction²⁷



which takes place under thermodynamic equilibrium during the annealing time in between the laser pulses, the supersaturation of O_2 is related to the O_2 pressure as

$$\Delta\mu_{\text{O(ad)}}^* = \Delta\mu_0^*(T) + \frac{3}{4}N_A kT \ln(P_{\text{O}_2}), \quad (5)$$

where N_A is the Avogadro constant. Equation (5) suggests that higher O_2 pressure always corresponds to larger supersaturation, resulting in faster nucleation and better Lu:Fe stoichiometry.

Combining Eqs. (5) and (3), one has the analytical relation between the nucleation speed and the O_2 pressure

$$J_{\text{nuc}} \propto \left[\frac{\Delta\mu_0^*(T) + \frac{3}{4}N_A kT \ln(P_{\text{O}_2})}{T} \right]^{1/2} * \exp \left[-\frac{\kappa}{\Delta\mu_0^*(T)kT + \frac{3}{4}N_A \ln(P_{\text{O}_2})(kT)^2} \right]. \quad (6)$$

Figure 2(b) shows the XRD intensity (peak area) of the LuFeO_3 phase relative to Lu_2O_3 [$I(P)/I(L)$] as a function of the O_2 pressure at 1050 °C. Assuming that the nucleation speed is proportional to the XRD intensity, one can fit experimental data with Eq. (6). The result shows that $\Delta\mu^*(T = 1050 \text{ °C}) = 269 \text{ kJ mol}^{-1}$, similar to the bulk value found as $\Delta\mu_0 = \Delta H^0 - T\Delta S^0 = 258.2 \text{ J mol}^{-1}$, taking the $\Delta H^0 = -41.8 \text{ kJ mol}^{-1}$ and $\Delta S^0 = -121.4 \text{ J mol}^{-1} \text{ K}^{-1}$, and $T = 1050 \text{ °C}$ (Ref. 28).

In the above analysis, the assumptions we made are as follows: (1) the nucleation speed is proportional to the XRD intensity; (2) at high temperature the thermodynamic equilibrium is reached during the annealing between the laser pulses determines the growth. These assumptions appear to be valid because the thermochemical parameters extracted from the model quantitatively agree with those from the literature. In other words, the growth of Lu-Fe-O at 1050 °C can be described using equilibrium thermodynamics, presumably due to the thermodynamic equilibration that occurs in between the laser pulses. Here the competition between the desorption and nucleation determines the Lu:Fe stoichiometry. When the temperature is high enough, the time scales of the nucleation and desorption are comparable. In this case, change of nucleation speed (due to the change of supersaturation which is a function of O_2 pressure) affects the Lu:Fe stoichiometry dramatically.

Based on this analysis, we expect the optimal growth conditions for LuFe_2O_4 films to be a narrow temperature and pressure window considering the necessary high temperature for the stability of LuFe_2O_4 phase that sets a lower limit, and the different desorption speed of Lu and Fe adatoms which sets an upper limit to the temperature. As we have shown in Sec. III, this is indeed what has been observed in our experiments.

B. Magnetic properties

The observation of superparamagnetism in the LuFe_2O_4 films is unusual considering the bulk magnetic properties of LuFe_2O_4 : an easy axis along the [001] direction with anisotropy energy as large as 100 K/spin and gigantic coercivity (9 T at 4 K).^{19–21,29–32} These unique bulk properties were attributed to the significant contribution of orbital magnetic moments ($0.8 \mu_B/\text{f.u.}$) plus the collective freezing of magnetic domains with the size of approximately 100 nm in the Fe_2O_2 layer and 30 nm along the [001] direction.^{19,25} The following scenario may explain the reduction of coercivity qualitatively: the structure of LuFe_2O_4 and h- LuFeO_3 both consist of layers of triangular lattice that can be epitaxial to each other nicely. For LuFe_2O_4 , the stacking is $\text{Fe}_2\text{O}_2/\text{LuO}_2$ while for h- LuFeO_3 , FeO layers replace Fe_2O_2 layers.^{16,17} From XRD data, one can see the coexistence of both LuFe_2O_4 and h- LuFeO_3 phases. HRTEM indicates that the LuFe_2O_4 layers are divided into clusters (much smaller than the magnetic domain size in bulk) by the h- LuFeO_3 layers and defects. According to a recent study, h- LuFeO_3 is weakly ferromagnetic (i.e., much less magnetic than LuFe_2O_4) (Refs. 33 and 34). Therefore, when these LuFe_2O_4 clusters are much smaller than the dimensions of the magnetic domain in the bulk, one expects to see a reduction in coercivity. On the other hand, given the large anisotropy energy 100 K/spin, the observed hysteresis is too small even for clusters having a size as small as a few nanometers. Further study on the microscopic magnetic structure is needed to understand the difference between the bulk and films.

V. CONCLUSION

In conclusion, we studied the growth dynamics of LuFe_2O_4 films on MgO (111) substrates and constructed the growth diagram. According to our understanding, application of the correct thermochemistry is the key to preferential formation of the LuFe_2O_4 phase: (1) at low temperature, LuFe_2O_4 is not a thermodynamically stable phase; (2) at high temperature, the Lu:Fe stoichiometry is off by so much due to the faster desorption of Fe adatoms that LuFe_2O_4 can not be formed; (3) in a narrow range of substrate temperature and O_2 pressure, LuFe_2O_4 dominates the grown phases with some h- LuFeO_3 phase epitaxially sandwiched in between due to the loss of Fe atoms. Superparamagnetism is observed in the film of LuFe_2O_4 containing h- LuFeO_3 impurities. The extracted Neel transition temperature is similar to that of bulk.

This work reveals the growth mechanism of Lu-Fe-O compound thin films, paving the way to the growth of high quality LuFe_2O_4 thin films and offers an approach to tuning their properties. This will be critical for future applications using LuFe_2O_4 , a unique multiferroic material with large polarizations, high ordering temperatures, and strong magnetoelectric coupling.

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¹J. Wang, J. B. Neaton, H. Zheng, V. Nagarajan, S. B. Ogale, B. Liu, D. Viehland, V. Vaithyanathan, D. G. Schlom, U. V. Waghmare, N. A. Spaldin, K. M. Rabe, M. Wuttig, and R. Ramesh, *Science* **299**, 1719 (2003).

²N. Hur, S. Park, P. A. Sharma, J. S. Ahn, S. Guha, and S. W. Cheong, *Nature (London)* **429**, 392 (2004).

³N. Ikeda, H. Ohsumi, K. Ohwada, K. Ishii, T. Inami, K. Kakurai, Y. Murakami, K. Yoshii, S. Mori, Y. Horibe, and H. Kito, *Nature (London)* **436**, 1136 (2005).

⁴M. A. Subramanian, T. He, J. Z. Chen, N. S. Rogado, T. G. Calvarese, and A. W. Sleight, *Adv. Mater.* **18**, 1737 (2006).

⁵X. Y. Xu, K. Seal, X. S. Xu, I. Ivanov, C. H. Hsueh, N. Abu Hatab, L. F. Yin, X. Q. Zhang, Z. H. Cheng, B. H. Gu, Z. Y. Zhang, and J. Shen, *Nano Lett.* **11**, 1265 (2011).

⁶J. Liu, Y. Wang, and J. Y. Dai, *Thin Solid Films* **518**, 6909 (2010).

⁷J. Rejman, T. Dhakal, P. Mukherjee, S. Hariharan, and S. Q. Witanachchi, *Mater. Res. Soc. Symp. Proc.* **1199**, F03-22 (2011).

⁸M. S. V. Kumar, K. Kuribayashi, and K. Kitazono, *J. Mater. Res.* **23**, 2996 (2008).

⁹T. Sekine and T. Katsura, *J. Solid State Chem.* **17**, 49 (1976).

¹⁰T. Katsura, T. Sekine, K. Kitayama, T. Sugihara, and N. Kimizuka, *J. Solid State Chem.* **23**, 43 (1978).

¹¹N. Kimizuka, A. Yamamoto, H. Ohashi, T. Sugihara, and T. Sekine, *J. Solid State Chem.* **49**, 65 (1983).

¹²The peak at 30.27° in Fig. 2(c) is assigned as h(004) based on its proximity to the calculated peak position (Ref. 15) of h(004) (30.52°) and the fact that TEM results are consistent with coexistence of h-LuFeO₃ and LuFe₂O₄. The peak at 29.74° of the upper curve in Fig. 2(a) is assigned as L(222) based on its proximity to the calculated peak position (Ref. 15) of L(222) (29.87°) and the observed peak position of L(222) (29.65°) of the lower curve in Fig. 2(a). There may be some intensity of h(004) buried in the shoulder of the assigned L(222) peak of the upper curve in Fig. 2(a). However, the dominant contribution should be from L(222) diffraction.

¹³According to the calculated x-ray powder diffraction intensity of LuFe₂O₄ (Ref. 15), the (0012) peak is two orders of magnitude weaker than the (0015) peak, which explains why the (0015) peak is observed while the (0012) peak is not.

¹⁴Several phases do not show up in the XRD data including FeO and Lu₃Fe₅O₁₂. The fact that there are also no RHEED patterns related to these phases suggests that their population is minimal if not zero. The absence of those phases and the presence of the h-LuFeO₃ which is unstable in bulk may be related to the epitaxial preference.

¹⁵See Supplemental Material at <http://link.aps.org/supplemental/10.1103/PhysRevB.85.155411> for more detailed information of structural parameters, epitaxial relations, analysis of magnetic properties, thermo-chemistry, and representative XRD for each of the growth region in the phase diagram.

¹⁶Y. B. Qin, H. X. Yang, Y. Zhang, H. F. Tian, C. Ma, Y. G. Zhao, R. I. Walton, and J. Q. Li, *J. Phys.: Condens. Matter* **21**, 015401 (2009).

¹⁷H. X. Yang, Y. Zhang, Y. B. Qin, C. Ma, H. F. Tian, and J. Q. Li, *Physica Status Solidi B-Basic Solid State Physics* **247**, 870 (2009).

¹⁸E. Magome, C. Moriyoshi, Y. Kuroiwa, A. Masuno, and H. Inoue, *Jpn. J. Appl. Phys.* **49**, 09ME06 (2010).

¹⁹J. Iida, Y. Nakagawa, S. Takekawa, and N. Kimizuka, *J. Phys. Soc. Jpn.* **56**, 3746 (1987).

²⁰W. Wu, V. Kiryukhin, H. J. Noh, K. T. Ko, J. H. Park, W. Ratcliff, P. A. Sharma, N. Harrison, Y. J. Choi, Y. Horibe, S. Lee, S. Park, H. T. Yi, C. L. Zhang, and S. W. Cheong, *Phys. Rev. Lett.* **101**, 137203 (2008).

²¹K.-T. Ko, H.-J. Noh, J.-Y. Kim, B.-G. Park, J.-H. Park, A. Tanaka, S. B. Kim, C. L. Zhang, and S.-W. Cheong, *Phys. Rev. Lett.* **103**, 207202 (2009).

²²At 12 K, the remanence is 7.5 emu cm⁻³ and the coercive field is approximately 300 Oe. This can be attributed to the existence of a trace amount of Fe₃O₄ islands because their temperature dependences are very similar to those of the samples with larger Fe₃O₄ populations (data not shown) and much larger remanence.

²³C. P. Bean and J. D. Livingston, *J. Appl. Phys.* **30**, S120 (1959).

²⁴X. S. Xu, S. Y. Yin, R. Moro, and W. A. de Heer, *Phys. Rev. B* **78**, 054430 (2008).

²⁵Here the field should be high enough to saturate the impurity moment (>300 Oe).

²⁶I. V. Markov, *Crystal Growth for Beginners: Fundamentals of Nucleation, Crystal Growth, and Epitaxy* (World Scientific, Singapore, 1995).

²⁷This analysis is only sensible at high temperature where desorption is significant. At high temperature, there are basically only two reactions left: Eq. (4) and Fe + $\frac{1}{2}$ Lu₂O₃ + $\frac{3}{4}$ O₂ → h-LuFeO₃ (i.e., other Lu-Fe-O phases are not thermodynamically stable at high temperatures). Unfortunately the thermochemical properties of h-LuFeO₃ are not available. Therefore, only the analysis based on Eq. (4) can give us a degree of understanding of the growth.

²⁸M. S. V. Kumar, K. Nagashio, T. Hibiya, and K. Kuribayashi, *J. Am. Ceram. Soc.* **91**, 806 (2008).

²⁹M. H. Phan, N. A. Frey, H. Srikanth, M. Angst, B. C. Sales, and D. Mandrus, *J. Appl. Phys.* **105**, 07E308 (2009).

³⁰S. Park, Y. Horibe, Y. J. Choi, C. L. Zhang, S. W. Cheong, and W. Wu, *Phys. Rev. B* **79**, 180401 (2009).

³¹F. Wang, J. Kim, Y.-J. Kim, and G. D. Gu, *Phys. Rev. B* **80**, 024419 (2009).

³²M. H. Phan, N. A. Frey, M. Angst, J. de Groot, B. C. Sales, D. G. Mandrus, and H. Srikanth, *Solid State Commun.* **150**, 341 (2010).

³³A. R. Akbashev, A. S. Semisalova, N. S. Perov, and A. R. Kaul, Eds. *Appl. Phys. Lett.* **99**, 122502 (2011).

³⁴Our ongoing work on pure h-LuFeO₃ films is also consistent with weak ferromagnetism.