



Metamorphic CO₂ production in calc-silicate rocks from the eastern Himalaya

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Metamorphic CO₂ production in calc-silicate rocks from the eastern Himalaya

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Abstract

Metamorphic degassing from active collisional orogens may supply a significant fraction of the global solid-Earth derived CO₂ to the atmosphere, thus playing a fundamental role even in today's Earth carbon cycle. The Himalayan belt, a major **active** collisional orogen, is a good candidate for the production of large amounts of metamorphic CO₂ that may influence the long-term climate. Metamorphic CO₂ can be produced during prograde metamorphism of impure carbonate rocks. **However, reliable quantitative modeling of metamorphic CO₂ fluxes from the Himalayan belt requires a good knowledge of the nature, magnitude and distribution of the CO₂-producing processes.**

This study focuses on the metamorphic decarbonation processes occurring during the Himalayan collision, **with special focus on** the distribution of different types of metacarbonate rocks in the Eastern Himalaya **and** their petrographic description. Petrological data about **selected** CO₂-producing reactions **suggest that** the most **common** calc-silicate **rocks are potential great sources of metamorphic CO₂, the molar amount of CO₂ in the fluid ranging between 37 and 44 % in CFMAS-HC lithologies and between 21 and 57 % in NKCFMAS-HC lithologies.** These results contribute to a better understanding of the influence exerted by orogenic processes on climatic changes at global scale.

Keywords: Orogenic CO₂; Climate changes; Decarbonation processes; Himalaya; Metacarbonate rocks.

Introduction and aim of the study

Metamorphic degassing from active collisional orogens is suspected to supply significant amounts of CO₂ to the atmosphere, thus playing a fundamental role in the long-term (>1 Ma) global carbon cycle (e.g. KERRICK & CALDEIRA, 1993; SELVERSTONE & GUTZLER, 1993; BICKLE, 1996; BERNER, 1999; MÖRNER & ETIOPE, 2002; GAILLARDET & GALY, 2008; EVANS, 2011).

A number of parameters could positively influence the extent of metamorphic degassing; among them, of major importance are the occurrence of appropriate lithologies, the size of the orogens and the maximum temperatures reached during their metamorphic evolution. The Himalaya is the **largest** “large-hot” collisional orogen on Earth (BEAUMONT *et alii*, 2010), over 2500 km wide along strike and rich in high grade metamorphic rocks. Moreover, tectonic and erosional processes are still active today. Therefore, the Himalaya is a likely candidate for the production of a huge amount of metamorphic CO₂ that may have contributed to changes in long-term climate of the past and that may still influence the atmospheric composition in the near future (e.g. GAILLARDET & GALY, 2008; GIRAULT *et alii*, 2014).

In a somehow over-simplified conceptual scheme, large metamorphic CO₂ fluxes should have been (and should still be) facilitated by rapid metamorphism of large volumes of metacarbonate rocks, coupled with facile escape of CO₂ to the Earth’s surface. However, the nature, magnitude and distribution of the CO₂-producing processes in Himalaya, is still poorly known. It is therefore **difficult** to build a reliable quantitative modeling of metamorphic CO₂ fluxes in the Himalaya, as well as in any other collisional orogen (e.g. MÖRNER & ETIOPE, 2002).

The key to **understanding** the metamorphic CO₂ flux in the past, is **to investigate** the petrologic **evolution** of the **calc-silicate** CO₂-source rocks currently exposed on the Earth surface. Within the framework of several **projects**, our research group focused on the metamorphic decarbonation processes occurring during the evolution of the Himalayan orogen (ROLFO *et alii*, 2015). Fieldwork activity was combined with petrographic, petrologic, structural, geochronological, geochemical and fluid inclusion studies with the aims of clarifying: (i) abundance and types of CO₂-source rocks, (ii)

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3 nature and rate of CO₂-producing reactions, (iii) nature, composition and amounts of the released
4 CO₂-rich fluids, (iv) nature and distribution of the CO₂ escape-paths toward the Earth's surface, and
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6
7 (v) chronology of metamorphic CO₂-producing reactions occurred in the Himalayas at different
8 structural levels and at different times.
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11 In this paper we present our results on points (i) to (iii): the distribution and petrography of the
12 different types of metacarbonate rocks in the Eastern Himalaya is reported in detail for the first
13 time, and petrological evidence on the nature of the CO₂-producing reactions in different types of
14 calc-silicate rocks is summarized and discussed in the global perspective of the orogenic CO₂-cycle.
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20 21 22 **Field descriptions of metacarbonate rocks in eastern Himalaya**

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24 A number of authors recognized calc-silicate rocks among the most common lithologies in the
25 Greater Himalayan Sequence (GHS), which is the core of the Himalayan metamorphic belt and
26 consists of medium- to high-grade micaschist and paragneiss and of granitic orthogneiss showing
27 pervasive evidence of anatexis (e.g. LOMBARDO *et alii*, 1993; CAROSI *et alii*, 1999; SEARLE *et alii*,
28 2003; GOSCOMBE *et alii* 2006; LONG *et alii*, 2011; MOSCA *et alii*, 2011, 2014; ROLFO *et alii*, in
29 press). However, these lithologies have been understudied so far, mainly because they are
30 notoriously difficult to be studied by means of petrologic modeling.
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35 In the eastern Himalaya, calc-silicate rocks are widespread in the lower and upper structural levels
36 of the GHS (MOSCA *et alii*, 2013). In selected localities, calc-silicate rocks outcrop even at
37 mountain scale, with notable examples occurring for instance in the Everest and Makalu national
38 parks (Fig. 1).
39

40
41 At the outcrop scale, calc-silicate rocks are easy to map and usually show very characteristic
42 deformation styles because of their relatively weak rheological behavior compared to the host
43 quartz-feldspar rich rocks. In hand specimen, calc-silicate rocks vary in appearance according to
44 their modal composition and relative abundance of various Ca-rich minerals (Fig. 2). They are often
45 banded at a centimetre to decimetre scale and the most typical textures include granoblastic
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3 structures. The most abundant minerals are anorthite-rich plagioclase and diopsidic pyroxene.
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5 **Assemblages also include** additional Ca-Mg-Fe-rich silicates (e.g. Ca-amphibole, grossular-rich
6
7 garnet, zoisite), Na-Ca solid solutions (e.g. scapolite) and **locally abundant** K-rich minerals (e.g.
8
9 white mica, biotite, K-feldspar). Garnet- and clinopyroxene- rich / plagioclase-poor varieties of
10
11 calc-silicate rocks with abundant clinopyroxene+plagioclase intergrowths are quite similar to the
12
13 retrogressed eclogites described in eastern Himalaya (LOMBARDO & ROLFO, 2000). However, these
14
15 calc-silicate rocks may be clearly recognized on the **basis** of minerals composition (always Ca-rich),
16
17 occurrence of scapolite and relative abundance of titanite, and should thus not be confused with
18
19 eclogites (see for instance the samples described by CORRIE *et alii*, 2010).
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23 Field data acquired in **over** 10 years in central-eastern Nepal and Sikkim allowed us to distinguish
24
25 two main different modes of occurrence at the different structural levels in the GHS (Fig. 3), **as**
26
27 **preliminary outlined by Rolfo *et alii* (2015):**
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30 (i) in the lower portion of the GHS (GHS-L), calc-silicate rocks generally occur as decimetre to
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32 metre-thick layers or massive boudins (Fig. 4a,b,c) enveloped by the main foliation of the hosting
33
34 medium- to high-grade, locally anatectic, staurolite- and/or kyanite-bearing metapelites (e.g.
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36 GROPPPO *et alii*, 2009; MOSCA *et alii*, 2012);
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38
39 (ii) **in the** structurally **higher portion of the GHS** (GHS-U), calc-silicate rocks are hosted in anatectic
40
41 kyanite-sillimanite- bearing gneisses (i.e. Barun Gneiss, see GROPPPO *et alii* 2012) and often occur as
42
43 tens to hundreds of metres thick, folded or boudinated, levels occasionally associated **with** layers of
44
45 impure marbles (Fig. 4a,d,e).
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48 Where preserved, the original transition between the host paragneiss and the calc-silicate granofels
49
50 is generally gradual and is characterized by the progressive disappearance of biotite and the
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52 appearance of calc-silicate minerals. A banded structure is **often** observed in the calc-silicate rocks,
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54 which is defined by the different modal proportion of the rock-forming minerals in adjacent layers.
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56 These features suggest that the calc-silicate rocks derive from former marly intercalations within a
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58 thick sedimentary sequence.
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3 The calc-silicate rocks are characterized by different mineral assemblages that vary systematically
4 as a function of the structural level and of the protolith composition (Fig. 3); they are described in
5 detail in the next section.
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9 10 11 **Petrography of different varieties of metacarbonate rocks**

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14 The studied samples are medium- to fine-grained and generally have granofelsic structure. An
15 exception includes phyllosilicate-rich schists, and gneisses that have evidence of brittle-to-ductile
16 deformation resulting in local grain size reduction. Two different calc-silicate groups have been
17 recognized, corresponding to different protolith compositions: they can be described in the (1)
18 CFMAS-HC (CaO-FeO-MgO-Al₂O₃-SiO₂-H₂O-CO₂) and (2) NKCFMAS-HC (Na₂O-K₂O-CaO-
19 FeO-MgO-Al₂O₃-SiO₂-H₂O-CO₂) model systems, respectively. The CFMAS-HC calc-silicate rocks
20 are significantly more abundant in the GHS-L than in the GHS-U (Fig. 3); the NKCFMAS-HC
21 calc-silicate rocks are widespread in both the GHS-L and GHS-U, but they occur as thin levels
22 (decimetre- to metre- thick) in the GHS-L whereas they form very thick layers in the GHS-U (up to
23 hundreds of metres thick) (Fig. 4d,e). In both calc-silicate groups, mineral assemblages vary with
24 increasing metamorphic grade from lower to upper structural levels of the GHS.
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42 **1. CFMAS-HC assemblages**

43 The following mineral assemblages can be observed from lower to upper structural levels (and from
44 lower to higher metamorphic grade; Table 1):
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46 *(A) Tremolite-bearing (clinopyroxene and garnet -absent) impure marbles*

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48 These rocks occur as metre-thick layers within two-mica garnet ± staurolite ± kyanite -bearing
49 schists of the GHS-L and have been recognized only sporadically (Fig. 3); their structure vary from
50 granoblastic to slightly foliated. Beside calcite, these marbles are characterized by the presence of
51 tremolite, often occurring as centimetre-long, slightly oriented, poikiloblasts (Fig. 5a) locally
52 replaced by aggregates of talc, and minor graphite.
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3 *(B) Garnet + amphibole ± zoisite -bearing (clinopyroxene -absent) calc-silicate rocks*

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5 These rocks represent thin layers (centimetre- to decimetre- thick) intercalated with two-mica
6
7 garnet-bearing schists and gneisses and minor quartzites of the GHS-L. They are characterized by
8
9 abundant garnet (generally already evident at the outcrop scale), quartz and by the systematic lack
10
11 of calcite and clinopyroxene. Ca-rich plagioclase, zoisite, green amphibole and accessory titanite
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13 are also part of the mineral assemblage (Fig. 5b). Zoisite is locally abundant (up to 30-40 vol%) and
14
15 is replaced by coarse-grained plagioclase (Fig. 5c).

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18 *(C) Garnet + clinopyroxene ± zoisite -bearing calc-silicate rocks*

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20 This assemblage is the most abundant in the CFMAS-HC calc-silicate group. These very common
21
22 rocks occur as thin layers (decimetre-thick) and/or boudins within garnet + K-feldspar + biotite +
23
24 kyanite ± sillimanite anatectic paragneisses and are widespread in the upper structural levels of the
25
26 GHS-L. Commonly banded, these rocks display centimetre-thick alternated layers of different
27
28 mineral modes, compositions and microstructures. These comprise (a) Grt-rich, Cpx-poor (Fig.
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30 5d,e), (b) Cpx-rich, Grt-poor layers. However, the equilibrium assemblage in both layer types
31
32 consists of plagioclase + clinopyroxene + quartz + garnet ± zoisite; calcite is absent. Titanite and
33
34 apatite are ubiquitous as accessory minerals. In layer (a), garnet is grossular-rich (Grs₆₇₋₇₄) and
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36 mainly occurs as large porphyroblasts with a spongy appearance, intimately intergrown with quartz
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38 (Fig. 5d,e) and locally partially replacing clinopyroxene rims. Plagioclase is an almost pure
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40 anorthite (An₉₅₋₉₆). Graphite is abundant and occurs as large flakes concentrated in the plagioclase-
41
42 rich domains (Fig. 5d,e). Relict zoisite is included in both garnet and clinopyroxene. In layers (b),
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44 garnet occurs as: (1) large zoned porphyroblasts with almandine-rich cores and grossular-rich rims
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46 (Grs₇₁₋₇₃); (2) small blebs and/or idiomorphs (Grs₆₇₋₇₆) associated with plagioclase. Clinopyroxene
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48 forms large, weakly zoned crystals, locally intergrown with plagioclase.

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54 In summary, microstructural evidence suggest that garnet grew at the expense of zoisite,
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56 clinopyroxene and calcite, the latter being only locally observed as inclusion in garnet. Notably,
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58 coarse-grained graphite has been interpreted as precipitated from a H₂O-CO₂ fluid released through
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3 decarbonation reactions during prograde and/or early retrograde metamorphic evolution (GROPPO *et*
4 *alii*, 2013).

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7 *(D) Clinopyroxene -bearing (garnet -absent) calc-silicate rocks*

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9 These rocks are locally observed as decimetre- to metre-thick layers within garnet + K-feldspar +
10 biotite + sillimanite ± kyanite anatectic paragneisses of the GHS-U, locally associated **with** type (F)
11 of the NKCFMAS-HC group. They are characterized by a very simple mineral assemblage
12 consisting of quartz, plagioclase and clinopyroxene (Fig. 5f,g), ± very minor calcite, and by the
13 systematic lack of garnet.
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23 **2. NKCFMAS-HC assemblages**

24 The following mineral assemblages can be observed from lower to upper structural levels (and from
25 lower to higher metamorphic grade) (Table 1):
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29 *(A) Muscovite and/or phlogopite -bearing impure marbles*

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31 These rocks crop out as thin (decimetre- to metre thick) layers intercalated **with** GHS-L two-mica
32 garnet ± kyanite schists, and locally **with** type (B) lithologies of the NKCFMAS-HC group and type
33 (A) of the CFMAS-HC group. The **schistose texture is defined by** alignment of muscovite and/or
34 phlogopite (Fig. 6a,b). Quartz, minor plagioclase, rare tremolite and graphite also occur in the main
35 mineral assemblage. Titanite and tourmaline are present as accessory minerals.
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43 *(B) Muscovite and/or biotite + calcite -bearing phylladic micaschists*

44 Although not very abundant, this lithology is quite **representative** and **is** probably the most **common**
45 product of medium-grade metamorphism of the former marly protolith. It occurs as thin (decimetre-
46 to metre- thick) layers within two-mica garnet -bearing schists, at the lowermost structural levels of
47 the GHS-L, immediately above the LHS two-mica augen-gneiss (Ulleri orthogneiss). These rocks
48 are characterized by a mixed silicatic and carbonatic assemblage, mainly consisting of quartz,
49 plagioclase, muscovite, biotite and calcite in different modal amounts (Table 1, Fig. 6c,d). Relict
50 porphyroclastic ankerite is locally observed, partially replaced by calcite. Biotite often shows a
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3 greenish-brownish pleochroism (Fig. 6c). Calcite is widespread in the matrix, where it is locally
4 concentrated in millimetre-thick discontinuous domains alternating to the silicate-rich portions (Fig.
5 6d). Rare phlogopite + actinolite + calcite -rich calc-silicate granofels are associated with these
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7 phylladic micaschists and probably represent the metamorphic products of Mg-Fe rich marly
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9 intercalations.
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14 *(C) Muscovite and/or biotite + zoisite + scapolite + garnet -bearing micaschists*

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16 These rocks have been only rarely observed, most probably because at the outcrop scale they
17 resemble normal two-mica + garnet -bearing schists and they can be identified as calc-silicate rocks
18 only under the microscope. They occur as thin (centimetre- to decimetre- thick) layers associated
19
20 with (or at the same structural level than) type (B) of the NKCFMAS-HC group. The most common
21
22 feature of these rocks is the presence of abundant Ca-rich plagioclase, zoisite and/or scapolite (i.e.
23
24 calc-silicate minerals) in addition to the two-mica + garnet assemblage commonly found in the
25
26 metapelites at these structural levels (Fig. 6e,f). Microstructural relationships suggest that zoisite is
27
28 replaced by Ca-rich plagioclase, which is in turn replaced by scapolite. Calcite is absent, probably
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30 because it has been completely consumed during metamorphism. Titanite is a common accessory
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32 mineral.
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39 *(D) Biotite and/or K-feldspar + scapolite -bearing calc-silicate gneisses*

40 Similarly to the previous type (C), these rocks are not easy to be identified in the field.
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42 Macroscopically, they resemble the common biotite + garnet -bearing gneisses to whom they are
43 associated; calc-silicate minerals can be identified only under the microscope. They occur as metre-
44
45 thick layers within the biotite + K-feldspar + garnet + kyanite ± sillimanite gneisses, locally
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47 anatectic, cropping out at the uppermost structural levels of the GHS-L, and are often associated
48
49 with the type (E) of the NKCFMAS-HC calc-silicate group. These rocks generally show a gneissic
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51 texture due to abundance of biotite which is mostly concentrated in discontinuous layers; a typical
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53 character is the presence of scapolite in an otherwise common mineral assemblage consisting of
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3 quartz, plagioclase, biotite, \pm K-feldspar, \pm clinozoisite/epidote (Fig. 6g,h). Titanite is ubiquitous;
4
5 calcite is absent.

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7 *(E) Biotite and/or K-feldspar + scapolite + clinozoisite + amphibole + clinopyroxene -bearing*
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9 *calc-silicate gneisses*

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11 Often associated with the type (D) of the NKCFMAS-HC calc-silicate group, these calc-silicate
12
13 gneisses are quite easy to be recognized in the field due to the presence of abundant green
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15 amphibole and minor clinopyroxene. The presence of this amphibole in the equilibrium assemblage
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17 is the most common feature of this calc-silicate type (Fig. 6i,j). The abundant biotite is in some
18
19 places replaced by K-feldspar (Fig. 6i). The calc-silicate assemblage includes plagioclase,
20
21 clinozoisite and minor scapolite (Fig. 6i,j). Clinopyroxene is less abundant than amphibole. Titanite
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23 is the most common accessory mineral.
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27 *(F) K-feldspar + scapolite + clinopyroxene \pm zoisite \pm calcite -bearing calc-silicate rocks*
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30 This assemblage is the most abundant among the NKCFMAS-HC calc-silicate group. These very
31
32 common rocks occur as thick (tens to hundreds of metres-thick) layers within the GHS-U garnet +
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34 biotite + K-feldspar + sillimanite \pm kyanite anatectic gneisses (Barun gneisses); calcite-rich layers
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36 (impure marbles) are locally intercalated with predominant calcite-poor layers (calc-silicate rocks).
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38 The main mineral assemblage consists of K-feldspar + clinopyroxene + scapolite \pm zoisite \pm calcite
39
40 \pm minor quartz (Fig. 6k,l). Relict biotite often occurs within clinopyroxene and/or is replaced by K-
41
42 feldspar. Clinopyroxene (Di_{67-72}) is often partially replaced by retrograde green Ca-amphibole \pm
43
44 epidote \pm calcite (Fig. 6k), whereas scapolite ($eqAn_{66-69}$) is locally partially to completely replaced
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46 by aggregates of plagioclase + calcite and/or it is overgrown by coarse-grained epidote (Fig. 6k,l).
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48 In addition to the ubiquitous titanite, a strongly pleochroic allanite and a bluish to colorless
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50 tourmaline locally occur, whereas graphite is always absent.
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56 **Metamorphic evolution and CO₂-producing processes in calc-silicate rocks**
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3 The nature of the CO₂-producing reactions in the most abundant types of each calc-silicate group
4 (i.e. type (C) representative of the CFMAS-HC group and type (F) for the NKCFMAS-HC group)
5 has been petrologically investigated in the appropriate model systems using isobaric T-X(CO₂)
6 phase diagram sections and/or pseudosections, and phase diagram projections in which fluid
7 composition is not explicitly constrained (for a full description of methods and results, see GROPP
8 *et alii*, 2013; GROPP *et alii*, submitted). Independently from the model system, the relevant CO₂-
9 producing reactions are either “truly” univariant reactions (i.e. effectively univariant in the P-T
10 space and corresponding to isobaric invariant points in the T-X(CO₂) space) or isobaric univariant
11 reactions (i.e. divariant in the P-T space). In the following, the main results of these petrologic
12 studies are summarized.
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27 ***CFMAS-HC group: type C (Grt + Cpx ± Zo-bearing) calc-silicate rocks***

28 The P-T evolution of the studied samples was derived from that of the hosting GHS-L anatectic
29 paragneiss (GROPP *et alii*, 2009), which experienced peak P-T conditions of 780–800°C and 10.5–
30 11.0 kbar, followed by decompression to ca. 8.5–9.0 kbar associated with moderate cooling to ca.
31 750°C.
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38 The petrological results, described in detail by GROPP *et alii* (2013), define the P-T-X^{fluid} regime
39 during the metamorphic evolution of the studied calc-silicate rocks (Fig. 7a,c). The relevant CO₂-
40 producing reactions occurred during prograde heating up to peak-T of about 800°C (at about 10-11
41 kbar) and/or early decompression. These reactions involved the growth of grossular-rich garnet
42 (Gr_{S67-81}) in equilibrium with quartz at the expenses of zoisite, clinopyroxene and calcite (*isobaric*
43 *univariant reaction 24b* in Fig. 7a), and in equilibrium with plagioclase and quartz at the expenses
44 of zoisite, clinopyroxene and calcite (“truly” univariant reaction U26c in Fig. 7c). The modelled
45 reactions are fully consistent with the observed microstructures and measured mineral
46 compositions. Both these reactions released a CO₂-rich fluid (XCO₂=0.32-0.44 and XCO₂=0.37-
47 0.48, respectively; Fig. 7a,c).
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3 The systematic occurrence of large and abundant graphite flakes suggests that graphite may have
4 precipitated from the CO₂-rich C–O–H fluid internally produced through the garnet-forming
5 reactions. Two possible graphite precipitation mechanisms may be envisaged: (i) T decrease and/or
6 P increase; (ii) X^{fluid} variation to enter the graphite stability field. Hydration reactions at constant P
7 and T would represent a mechanism **capable of changing** the composition of the fluid in an
8 internally buffered system; the widespread occurrence of hydrated minerals associated with graphite
9 support the possibility that precipitation of graphite was triggered by hydration reactions. Moreover,
10 GROPPO *et alii* (2013) demonstrated that the observed graphite modes (ca. 1 vol%) are comparable
11 with those directly related to the estimated amount of decarbonation, strongly suggesting that the
12 studied calc-silicate rocks behaved as a closed system during their prograde and early retrograde
13 evolution.

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29 ***NKCFMAS-HC group: type F (Kfs + Scp + Cpx ± Zo ± Cal –bearing) calc-silicate rocks***

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31 **The GHS-U scapolite-bearing calc silicates experienced isobaric heating at ca. 9 kbar to peak**
32 **temperature conditions of 780-800°C** followed by nearly isothermal decompression to ca. 7 kbar
33 (Fig. 7d), **as defined by the hosting anatectic paragneiss** (Barun Gneiss: GROPPO *et alii*, 2012).

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Thermodynamic modeling (Groppo *et alii*, submitted) demonstrates that the observed
microstructures in this rock type reflect either isobaric univariant assemblages or isobaric invariant
assemblages. The CO₂-rich fluid was released through **several** prograde reactions, depending on the
local bulk composition, involving the growth of clinopyroxene, K-feldspar, and zoisite and the
consumption of scapolite (eqAn=0.64-0.67), quartz and calcite. **The modeled reactions are fully**
consistent with the observed microstructures and measured mineral compositions. These CO₂-
producing reactions are “truly” univariant reactions **(i.e. they correspond to isobaric invariant points**
in the isobaric T-XCO₂ sections) (Fig. 7b, d).

The first univariant curve crossed during the prograde evolution is the *clinopyroxene-forming*
reaction U20a in Fig. 7d (Cal + Pl + Qtz + Tr + Zo = Di + Scp + F). This reaction was crossed at

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3 about 590°C and released a fluid relatively rich in CO₂ (XCO₂ = 0.21-0.28). At higher T (ca.
4 650°C), both the *zoisite- and K-feldspar-producing reaction U10d* (Qtz + Cal + Mu + Scp = Zo + Pl
5 + Kfs + F) (Fig. 7d) and the *clinopyroxene- and K-feldspar-producing reaction U33c* (Cal + Phl +
6 Qtz + Scp + Zo = Di + Kfs + Pl + F) (Fig. 7d) were crossed at ca. 650°C (Fig. 7d), both releasing a
7 CO₂-rich fluid (XCO₂ = 0.51-0.57).
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12 These preliminary results demonstrate that: (i) Kfs + Scp + Cpx ± Zo ± Cal -bearing calc-silicate
13 rocks may act as a CO₂-source during prograde heating, releasing internal-derived CO₂-rich fluids;
14 (ii) the role of phases such as zoisite, scapolite and plagioclase solid solutions must be taken in
15 account whenever a quantitative and reliable estimate of the CO₂-producing metamorphic processes
16 in orogenic zones is sought.
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27 **Discussion and conclusions**

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29 In the eastern Himalaya, calc-silicate rocks are widespread in the lower and upper structural levels
30 of the GHS. Gathering of field data, coupled with new petrographic observations indicate that two
31 main groups of calc-silicate assemblages reflect differences in the protolith composition: CFMAS-
32 HC assemblages, more abundant in the GHS-L, and NKCFMAS-HC assemblages, volumetrically
33 more significant in the GHS-U. Many of these assemblages, especially those equilibrated at lower
34 temperatures and still containing abundant phyllosilicates, are not easy to recognize in the field and
35 probably have been considerably overlooked in the past. These assemblages do not usually contain
36 calcite, because calcite was completely consumed during prograde metamorphism; nevertheless, it
37 is proved that they played an important role in the orogenic-CO₂ cycle.
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49 The petrologic studies summarized in this paper demonstrate that both CFMAS-HC and
50 NKCFMAS-HC calc-silicate groups act as a significant CO₂-source during prograde heating and/or
51 early decompression near peak conditions, releasing internal-derived CO₂-rich fluids through
52 garnet-forming (CFMAS-HC) and clinopyroxene + K-feldspar -forming and scapolite -consuming
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(NKCFMAS-HC) reactions. The molar amount of CO₂ in the fluid ranges between 37 and 44 % in CFMAS-HC lithologies and between 21 and 57 % in NKCFMAS-HC lithologies.

A potentially very important result of our petrologic modeling is that, if the system remains closed, fluid-rock interactions may induce hydration of the calc-silicate assemblages (mainly plagioclase and clinopyroxene) and in-situ graphite precipitation, thereby removing carbon from the fluid. This seem to be the case for the graphite-bearing CFMAS-HC calc-silicate rocks in the GHS-L, which likely behaved as a closed-system during their prograde and early retrograde evolution, although their volume abundance within the hosting paragneiss is relatively low.

The interplay between these two contrasting processes – i.e. production of metamorphic CO₂-rich fluids vs. carbon sequestration through graphite precipitation – must be taken in account when dealing with a global estimate of the role exerted by decarbonation processes on the orogenic CO₂-cycle.

Extending these results in the global perspective of the orogenic CO₂-cycle will require additional studies to **quantitatively** define the **volumetric amount of the different** varieties of CO₂-source rocks within the GHS throughout the Himalayan orogen, **as well as within other orogens**. Moreover, precise geochronology on the peak assemblages resulting from the CO₂-producing reactions in the different types of calc-silicate rocks will help to constrain the possible occurrence of CO₂ **major influxes** in the past.

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5 review by C. Mottram.
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FIGURES CAPTIONS

Fig. 1 - Well known occurrences of calc-silicate rocks at large scale in eastern Himalaya: **(a)** the world-famous ridge between Nupse and Lhotse in upper Khumbu Himal shows very clear yellowish bands (arrow) embedded and deformed within dark rocks of the Everest Series; **(b)** the same bands are clearer when seen from the east, from the upper Barun Valley at the Makalu Base camp; **(c)** in the lower Barun Valley of East Nepal, the lower portion of the left valley side is carved among calc-silicate rocks, gently dipping to the left (west) along a distance of several kilometres; **(d)** in central Sikkim, Pandim (6691 m) west face reveals huge volumes of calc silicate rocks embedded within whitish leucogranites and other dark lithologies of the GHS.

Fig. 2 - Calc-silicate rocks structures, according to their modal composition and the relative abundance of various Ca-rich minerals. **(a)** Banded calc-silicate rock, Samiti Lake, Sikkim, NE India. **(b)** Centimetre-thick bands with white granoblastic plagioclase (Pl) and green diopsidic pyroxene (Di), Lower Barun Valley, eastern Nepal. **(c)** tremolitic amphibole (Am) bearing calc-silicate rock, Chyamtang, Upper Arun Valley, eastern Nepal. **(d)** Ca-garnet / Ca-Mg-pyroxene rich calc-silicate rock, Chyamtang, Upper Arun Valley, eastern Nepal. **(e)** grossular bearing garnetite boudin (Grt), Tashigaon, Arun Valley, eastern Nepal.

Fig. 3 – Simplified geological map of the central-eastern sector of the Himalayan belt (modified from GOSCOMBE *et alii*, 2006; DASGUPTA *et alii*, 2004; MOSCA *et alii*, 2012; MOSCA *et alii*, 2013) showing sample locations subdivided in the CFMAS-HC and NKCFMAS-HC groups, as discussed in the text. White dotted lines are the geotraverses investigated since 2004. The double-dashed grey line is the approximate political boundary between Nepal to the south, west, China (Tibet) to the north, India (Sikkim) to the east. Inset shows the location of the study area (white rectangle) in the framework of the Himalayan chain. MCT: Main Central Thrust; MFT: Main Frontal Thrust; MBT: Main Boundary Thrust; STDS: South Tibetan Detachment System; E: Everest; K: Kangchenjunga; M: Makalu.

Fig. 4 - Schematic cross-section across the eastern Himalaya (modified from GOSCOMBE *et alii*, 2006, and SEARLE *et alii*, 2008) highlighting the structural location and field occurrence of CFMAS-HC (**b,c**) and NKCFMAS-HC (**c,d**) calc-silicate rocks. The arrows in (b) and (c) indicate the calc-silicate boudins. Selected mineral isograds relevant for the predominant lithologies are also shown. Abbreviations and colours of the different calc-silicate groups as in Fig. 3.

Fig. 5 - Representative microstructures of CFMAS-HC calc-silicate rocks. (**a**) Type A: tremolite-bearing impure marble. Crossed Polarized Light (XPL). (**b,c**) Type B: garnet + amphibole + zoisite-bearing calc-silicate rock. The detail in (c) shows plagioclase replacing zoisite. Plane Polarized Light (PPL). (**d,e**) Type C: garnet + clinopyroxene calc-silicate rock. Garnet-rich layer, characterized by garnet + quartz intergrowths and by the presence of abundant graphite flakes (d: PPL; e: XPL). (**f,g**) Type D: clinopyroxene-bearing calc-silicate rock (f: PPL; g: XPL).

Fig. 6 - Representative microstructures of NKCFMAS-HC calc-silicate rocks. (**a,b**) Type A: muscovite + phlogopite impure marble (a: XPL; b: PPL). (**c,d**) Type B: muscovite + biotite + calcite-bearing phyllitic micaschist. Note the unusual greenish-brownish pleochroism of biotite and the presence of discontinuous calcite domains intercalated to the siliciclastic layers (c: PPL; d: XPL). (**e,f**) Type C: muscovite + biotite + scapolite + garnet micaschist (e: PPL; f: XPL). (**g,h**) Type D: biotite + scapolite-bearing calc-silicate gneiss (g: PPL; h: XPL). (**i,j**) Type E: K-feldspar + scapolite + epidote + amphibole + clinopyroxene calc-silicate gneiss. Note the coexistence of amphibole and clinopyroxene (i: PPL; j: XPL). (**k,l**) Type G: K-feldspar + scapolite + clinopyroxene + calcite calc-silicate granofels (k: PPL; l: XPL).

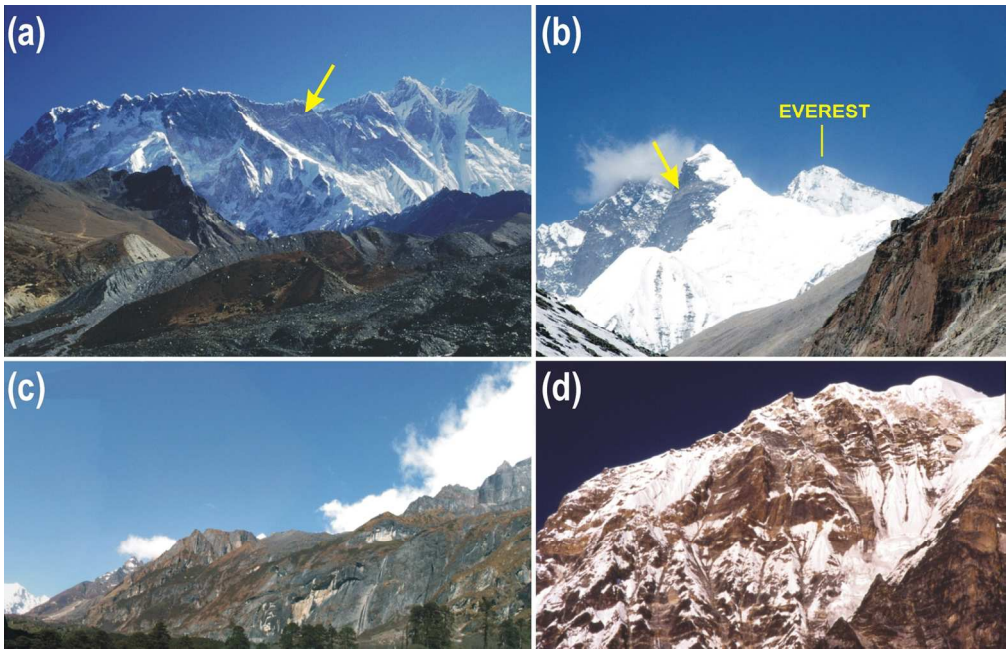
Fig. 7 – (**a,b**) Simplified isobaric T-X(CO₂) phase diagrams sections calculated for activity-corrected end-members, for CFMAS-HC (type C) calc-silicate rocks (a) and NKCFMAS-HC (type F) calc-silicate rocks (b) from the GHS-L and GHS-U, respectively. The composition of the solid solutions (a: garnet; b: scapolite and plagioclase) changes continuously along the isobaric univariant curves. Solid large points are isobaric invariant points; empty points are isobaric singular points (i.e. points at which the stoichiometric coefficient of one phase vanishes, such that this phase behaves

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3 as a reactant on one side of the reaction and as a product on the other side of the reaction). The
4 relevant isobaric univariant reactions (e.g. 24b) and invariant points (e.g. I26') are numbered and
5 reported in the insets (the prefix I denotes the isobaric invariant points). Isobaric univariant curves
6 labeled with the same number but with different suffix (e.g. 24a, 24b and 24c in (a)) are
7 characterized by one or more singular points: this means that the different portions of these curves
8 involve the same phases, but some of these phases behave alternatively as a reactant or as a product.
9 Coloured points and curves refer to the reactions relevant to explain the microstructures observed in
10 the studied calc-silicate rocks. The composition of the fluid circulating during the metamorphic
11 evolution of the studied calc-silicate rocks is highlighted in the rectangular coloured boxes. (c,d)
12 Simplified mixed-volatile P–T projections in the system CF(M)AS-HC (c) and NKC(F)MAS-HC
13 (d), respectively, with a selection of the univariant curves “seen” by the studied samples (see
14 GROPPPO *et alii*, 2013 and GROPPPO *et alii*, submitted for further details). Each univariant curve
15 corresponds to an isobaric invariant point (see dotted arrows connecting the isobaric invariant
16 points in (a) and (b) with the correspondent univariant curves in (c) and (d)). The relevant
17 univariant reactions (e.g. U26c) are numbered and reported in the insets (the prefix U denotes the
18 “truly” univariant reactions in the mixed volatile P-T projections). Univariant curves labeled with
19 the same number but with different suffix (e.g. 26a, 26b, 26c and 26d in (c)) are characterized by
20 one or more singular points: this means that the different portions of these curves involve the same
21 phases, but some of these phases behave alternatively as a reactant or as a product. The
22 compositions of the solid solutions (c: garnet; d: scapolite and plagioclase) and of the fluid phase (c
23 and d) change continuously along the univariant curves. Solid small points indicate the variation of
24 fluid (c,d), garnet (c), plagioclase and scapolite (d) compositions along each univariant curve;
25 empty points are singular points. The variations of fluid (F: X(CO₂)), garnet (X_{Grs}), plagioclase
26 (X_{An}) and scapolite (eqAn) compositions along each univariant curve are also reported. The P–T
27 evolution inferred from the hosting anatectic paragneiss (c: GROPPPO *et alii*, 2009; d: GROPPPO *et alii*,
28 2012) is reported with grey arrows. Coloured points and curves refer to the reactions relevant to
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3 explain the microstructures observed in the studied calc-silicate rocks. The composition of the fluid
4 circulating during the metamorphic evolution of the studied calc-silicate rocks is highlighted in
5 the rectangular coloured boxes.
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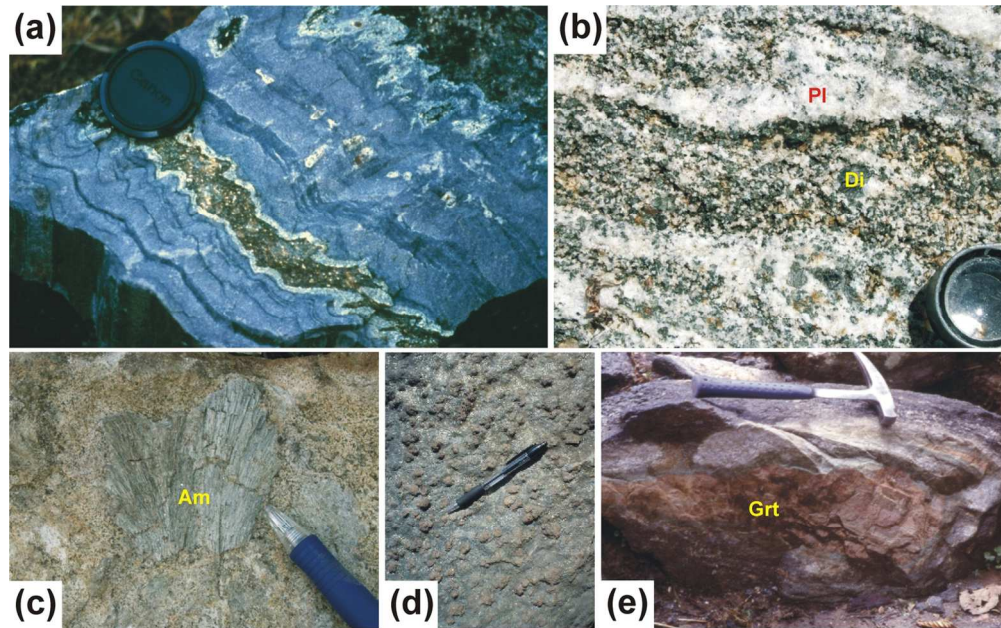
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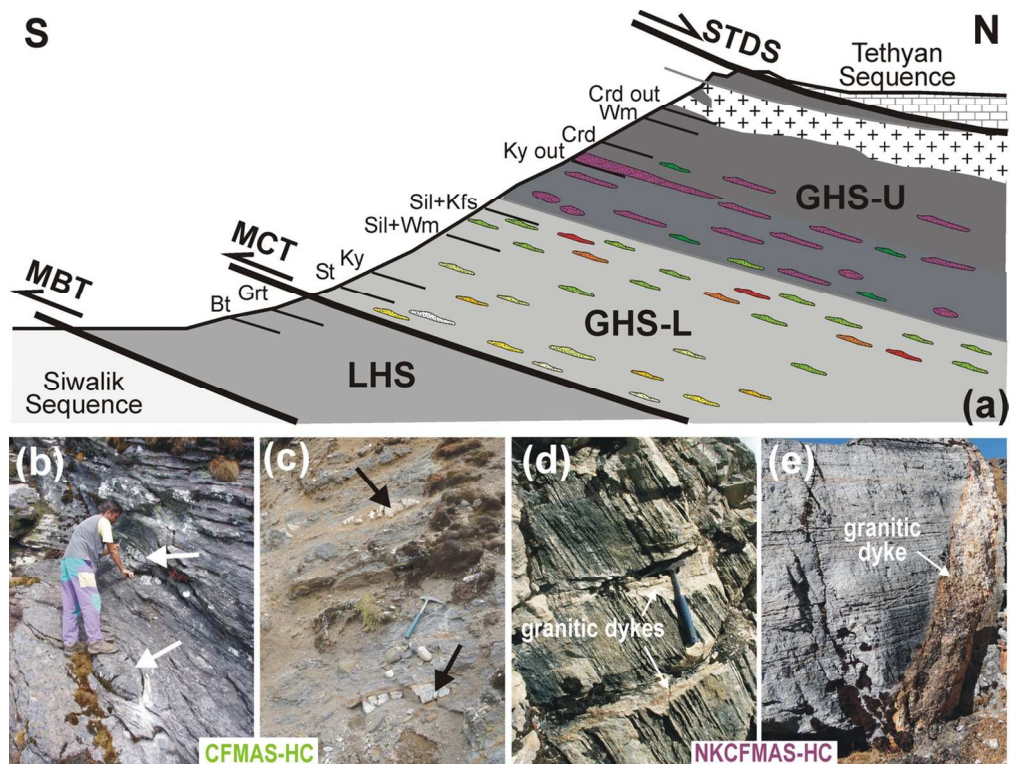
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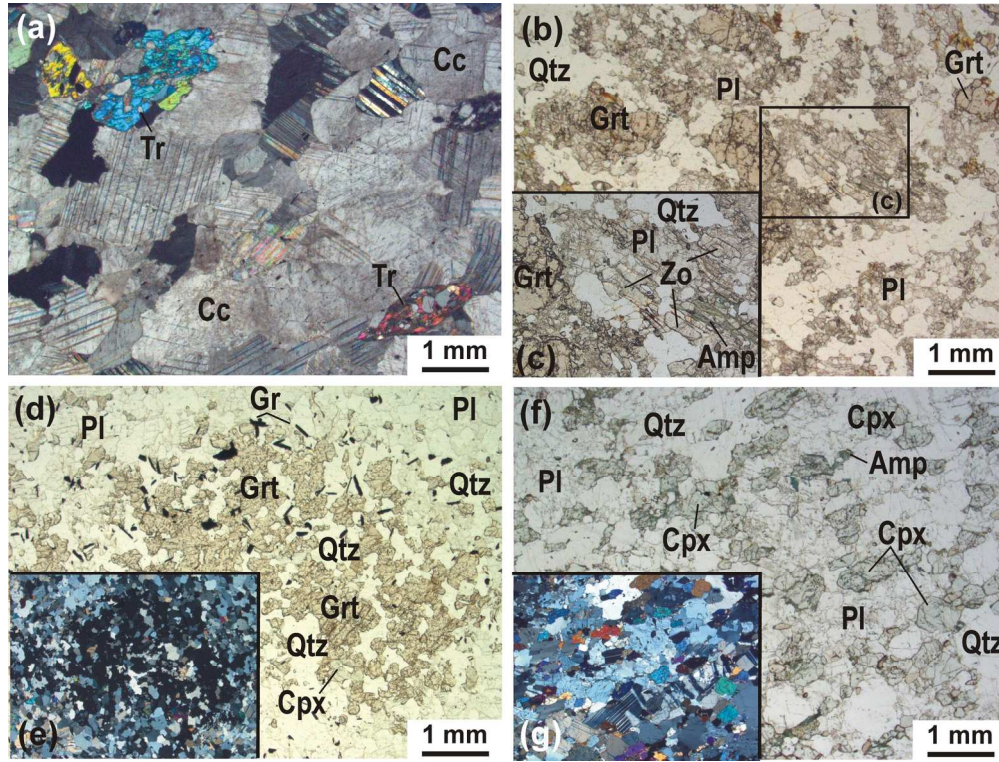
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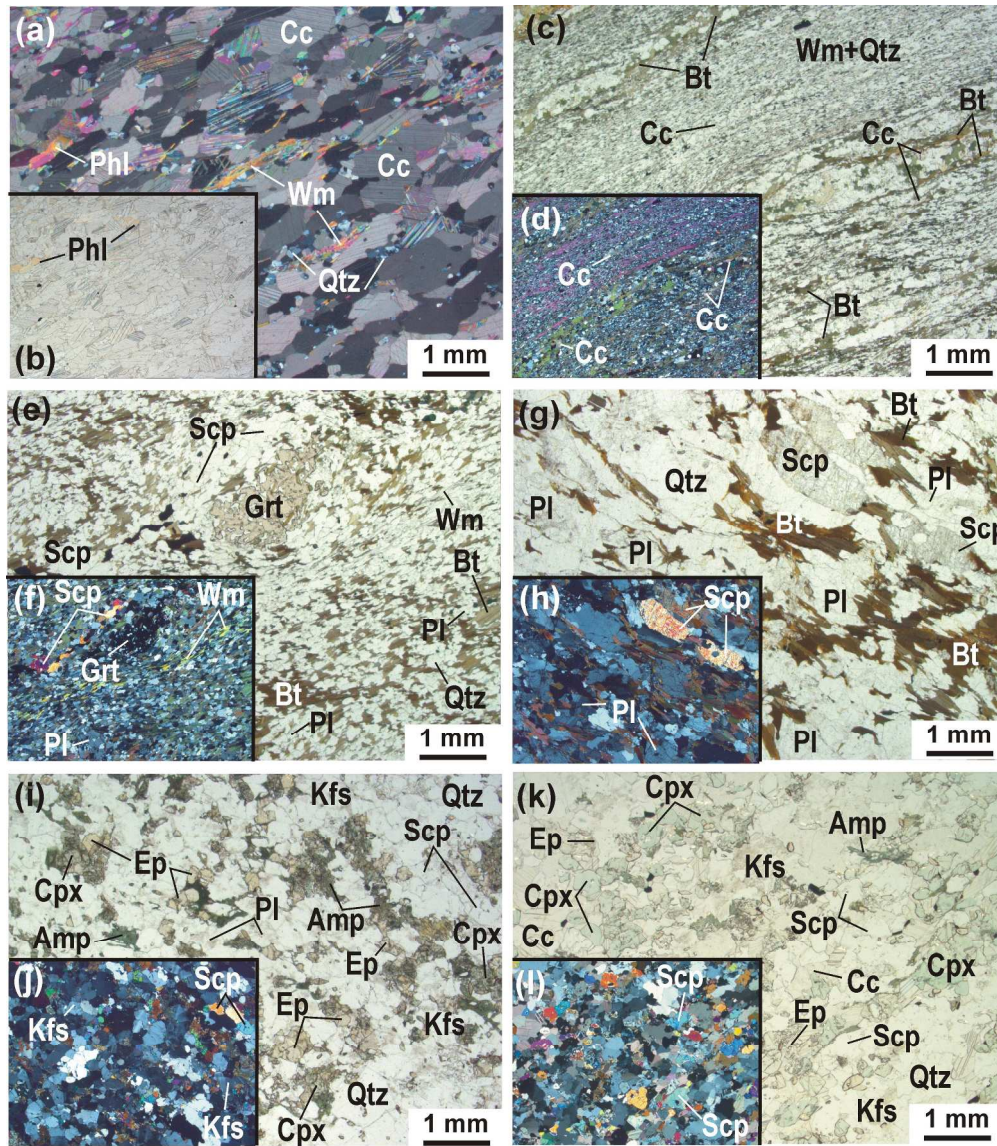
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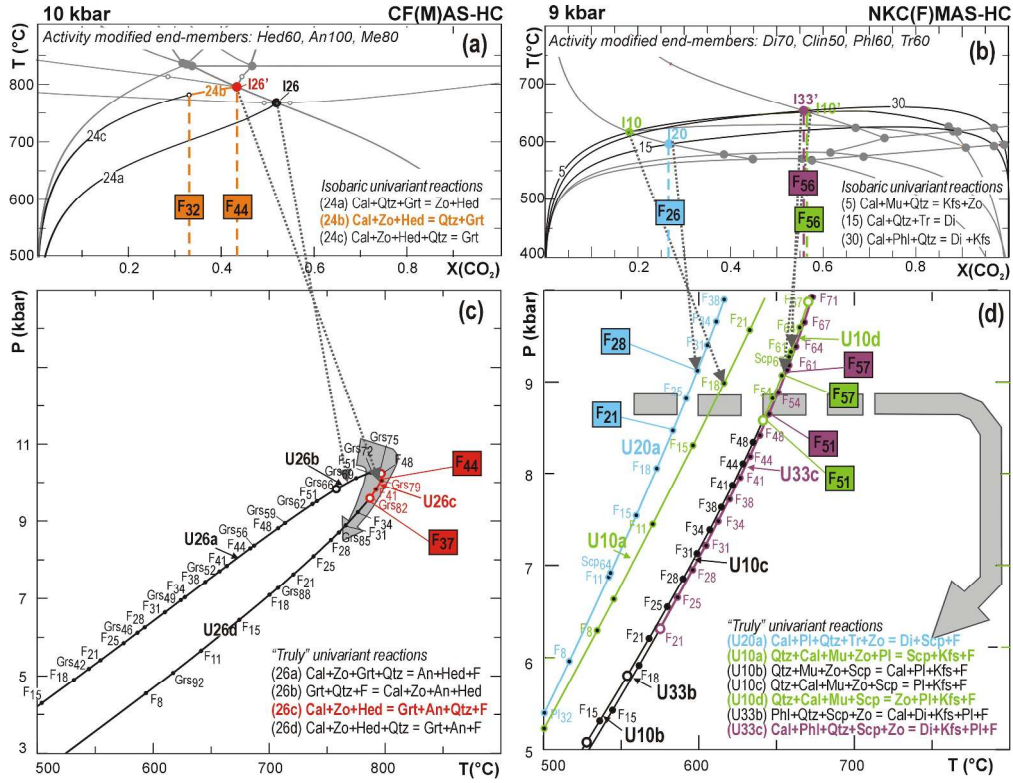
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Table 1 - Equilibrium assemblages and mineral abundances for each of the calc-silicate types

CFMAS-HC calc-silicate group															
	Cal	Qtz	Pl	Zo/Ep	Scp	Wm	Bt	Kfs	Amp	Cpx	Grt	Gr	Ttn	Turm	Aln
(A)	x x x x x	- / x	- / x						x x			x	x		
(B)		x x x	x / x x x	x / x x x					x / x x		x x x		x		
(C)		x x x	x / x x x	x / x x						x x / x x x	x x / x x x	x	x		
(D)		x x x	x x	- / x						x x x			x		
NKCFMAS-HC calc-silicate group															
	Cal	Qtz	Pl	Zo/Ep	Scp	Wm	Bt	Kfs	Amp	Cpx	Grt	Gr	Ttn	Turm	Aln
(A)	x x x x x	x x	- / x			x / x x	x / x x		- / x			- / x	x	x	
(B)	x / x x x	x x x	x / x x			x x / x x x	x x / x x x						x		
(C)		x x x	- / x x	- / x x	- / x x	x / x x	x x / x x x				x		x		
(D)		x x x	x / x x	- / x	x / x x		x x / x x x	- / x x					x		
(E)		x x / x x x	x / x x	- / x x	- / x		- / x x x	x / x x x	x x x	- / x			x		
(F)	- / x x	- / x	x / x x	- / x	x / x x x			x / x x x		x x / x x x			x	x	x

Mineral modes are qualitatively reported (- = not present; x = scarce; x x x x x = very abundant). Retrograde phases are not indicated.