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19 VARIETAL AND PRE-FERMENTATIVE VOLATILES DURING RIPENING OF *VITIS*20 *VINIFERA* CV NEBBIOLO BERRIES FROM THREE GROWING AREAS.

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Flavour analysis of grape is a key step in quality evaluation. The Stir Bar Sorptive Extraction technique (SBSE, 'Twister'®) was used to assess varietal and pre-fermentative volatile accumulation in 'Nebbiolo' berries, from véraison to harvest. Grapes were collected in three vineyards, representing different 'crus' in the cultivation areas of Barolo, Barbaresco and Roero (North-western Italy). Volatile constituents of grapes were identified and quantified by GC-MS.

We demonstrate the influence exerted by the growing location on volatile concentration and profile, as well as on the timing of volatile accumulation. The accumulation of certain classes of compounds, considered favourable for defining berry quality, followed common patterns, and was negatively correlated to that of compounds with herbaceous and grassy notes, such as the C6 compounds.

38 PCA analysis shows that the concentrations of varietal and pre-fermentative volatiles were more 39 effective in separating growing areas than dates of harvest. Grapes from the Barbaresco area, 40 showing higher values of the concentration ratio between favourable and unfavourable 41 compounds throughout ripening, could be statistically separated from grapes from the other areas.

43 Keywords: SBSE/GC-MS, flavours, grapes, principal component analysis.

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45 **1. Introduction**

46 Hundreds of volatiles have been identified in *Vitis vinifera* berries. Compounds with a C6-moiety 47 are products of the lipoxygenase pathway and are the major volatile constituents of varietal and 48 pre-fermentative aroma in many varieties (Gomez, Martinez & Laencina, 1995; Kalua & Boss, 49 2010; Yang et al., 2009). They have grassy, herbaceous odours, generally considered unpleasant 50 in wines if they are present in high concentrations (Baumes, Cordonnier, Nitz & Drawert, 1986). 51 Several other aliphatic and aromatic alcohols and aldehydes, norisoprenoids and terpenes have 52 been identified in grapes (Sefton, Francis & Williams, 1993). Some aliphatic aldehydes such as 53 octanal, decanal and (Z)-2-heptenal have citrus-like odour whereas furfural and benzaldehyde are 54 responsible for almond aroma. Apocarotenoid volatiles (β -ionone and β -damascenone) are among 55 the most important contributors to fruity and floral notes in many fruits, including grapes, 56 together with furanones, like 4-hydroxy-2,5-dimethyl-3(2H)-furanone and lactones (Klee, 2010). 57 As grapes ripen, a number of changes occur, including sugar and flavonoid accumulation, 58 modifications in the content of organic acids, and changes in the concentration of volatile 59 substances. The modifications of the berry volatile composition are still little known in 'neutral' 60 varieties, which represent most of winemaking grapes, as investigations have generally been 61 focused on 'aromatic' varieties, very rich in both free and in glycoconjugated volatiles, terpenes 62 in particular. Neutral grapes, however, possess a number of free and glycosylated volatile organic 63 compounds, generally at lower concentrations respect to the aromatic ones. When grapes are 64 thawed and grinded a number of volatiles develop: these molecules were named 'varietal and 65 pre-fermentative related volatile compounds' (Coehlo, Rocha, Delgadillo & Coimbra, 2006). They have been associated to C6 alcohols and aldehydes (Gomez, Martinez & Laencina, 1995), benzene derivatives, esters, and non-glycosylated monoterpenes and sesquiterpenes (Kalua & Boss, 2010); these compounds are present to such low concentrations that, when instrumental sensitiveness is low, they are seldom detected. This is probably a major reason of the scarcity of published data on the volatile composition of neutral grapes (Salinas, Zalacain, Pardo & Alonso, 2004).

72 Volatile analysis is a key step in grape quality evaluation. Volatiles have traditionally been 73 detected by GC/MS on grape extracts, which requires a prior time- and solvent-consuming 74 preparation. Many methods, using different solvents and extractants have been proposed up to 75 now_(Sefton, Francis & Williams, 1993; Cabrita, Costa Freitas, Laureano & Di Stefano, 2006). In 76 1999 Baltussen, Sandra, David & Cramers proposed a novel extraction technique for aqueous 77 samples based on the use of a magnetic stir bar coated with polydimethylxylosane (PDMS). This 78 technique, based on the concept of 'sorption' is known with the name of Stir Bar Sorptive 79 Extraction (SBSE) and is commercially available under the brand name 'Twister'®. Sorption 80 offers a number of advantages above traditional adsorption such as: 1) it is gentler as analytes are 81 not retained on an active surface, and degradation of unstable compounds is reduced or absent; 2) 82 desorption can be performed at lower temperatures, minimizing the losses of thermolabile 83 compounds; 3) the retaining capacity of PDMS is not influenced by the sample matrix, in 84 particular by the presence of different amounts of water (the main constituent of grape 85 homogenates) or of other analytes, since each solute has his own partitioning equilibrium into 86 PDMS; 4) degradation fragments originating from the PDMS stir bar coating give mass-spectra 87 that can be very easily recognized and discarded (Baltussen, Sandra, David & Cramers, 1999). 88 This technique has been successfully used in the detection of volatiles in many matrices but at the

moment few applications have dealt with grapes (Luan, Mosandl, Munch & Wust, 2005;
Pedroza, Zalacain, Lara, Salinas, 2010). Salinas, Zalacain, Pardo & Alonso (2004) were the first
to propose the use of this technique for detecting volatiles in grape berries.

92 The concentrations of volatiles change during grape ripening and information on these changes, 93 and the factors which induce them, is of high technological importance as it allows harvesting 94 grapes with specific aroma profiles. Little has been published up to now about the evolution of 95 volatiles during neutral grape berry ripening even though several reports deal with the detection 96 of volatiles at harvest. Studies dealing with the detection of grape volatiles during ripening are 97 relatively recent and mostly based on adsorption (Kalua & Boss, 2009; Coehlo, Rocha, Barros, 98 Delgadillo & Coimbra, 2007) or sorption methods (Salinas, Zalacain, Pardo & Alonso, 2004). 99 Another well-known, but little detailed, factor affecting aromatic composition of neutral grapes 100 (and possibly of the derived wines) is the so-called 'terroir' where the vines are grown, each 101 terroir giving rise to specific aromatic notes for a given vine genotype. However, very limited 102 information is available about the influence exerted by the growing location on the volatile 103 composition of neutral grapes (Bureau, Razungles & Baumes 2000; Ji & Dami, 2008).

104 In the present study we report the results of varietal and pre-fermentative volatile assessment 105 from véraison to harvest in berries of the neutral cultivar Nebbiolo, which is widely grown in 106 North-western Italy for the production of premium wines. Volatiles were detected using the 107 SBSE technique on grapes collected in three vineyards, each located in one of three terroirs of 108 Piedmont (Barbaresco, Barolo and Roero). Our aim was to identify individual molecules or 109 classes of molecules able to distinguish growing areas. Moreover, we assessed the timing of 110 accumulation of some class of volatiles in Nebbiolo grapes during ripening. Our results allow i) 111 to gain information regarding the aroma potential of Nebbiolo when harvested at different

developmental stages; ii) to assess the influences exerted by environmental conditions on the timing of flavour accumulation in grapes and iii) to contribute to knowledge about the key steps of flavour biosynthesis in grape berries.

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116 **2. Materials and Methods**

117 2.1 Samples

118 Nebbiolo grapes from three cultivation areas of Piedmont in North-West Italy (Barolo, 119 Barbaresco and Roero) were collected from early véraison to harvest in 2010. The 20/25 year old 120 vines were trained to the vertical trellis system and Guyot pruned. Canopies were routinely and 121 similarly managed during spring and summer accordingly to the standard cultural practices of the 122 cultivation area. In addition, crop load was controlled and standardized with cluster removal in 123 the pre-véraison period. Grapes were collected fortnightly from three homogenous replicate 124 groups of 20 adjacent vines in each vineyard. Berries were detached from the rachis in small 125 groups of 3 to 5 respectively from the upper, middle and bottom part of the cluster, to avoid 126 possible effects of scalar maturity inside the cluster. 250-300 berries were collected from each 127 replicate group of vines and from both sides of the row to overcome possible effects of light 128 exposure and temperature on secondary metabolite accumulation. Berries were stored in portable 129 refrigerators at 5°C and transported to the lab within 4 hrs. In the lab, berries were separated from 130 the rachis with small scissors and a subgroup of 200 berries per replicate was weighed and stored 131 at -80 °C until volatile analysis. Must was obtained by crushing the remaining berries and total 132 soluble solids (TSS) were determined with a digital refractometer (ATAGO, PR-32).

Meteorological data were obtained from three automatic weather stations, each located within a 5km range from one of the vineyards.

135

136 2.2 SBSE methodology

137 Sample preparation for the detection of volatiles involved grinding and homogenization of 138 grapes. A common robot for domestic use was employed to crush berries without breaking seeds; 139 from each sample of 200 homogenized berries, 10 g were diluted to 100 ml with distilled water 140 and 30 μ L of 2-heptanol, 104.38 mg L⁻¹, were added as internal standard for semi-quantification. 141 Samples were left at room temperature for 30 minutes, and during this time lapse they were 142 manually shaken three times; 20 ml of the aqueous grape extract were transferred into a cap-143 screw vial and stirred (700 rpm) with a PDMS stir bar (0.5 film thickness, 10 mm length, 144 Twister®, Gerstel, Mulheim and der Ruhr, Germany) for 6 hours at room temperature (Salinas, 145 Zalacain, Pardo & Alonso, 2004).

146 All grape samples were analyzed within 6.5 hours from grinding and thawing to avoid possible 147 artefacts that may have arisen from reactions due to the acid conditions of the juice, to 148 endogenous enzymes and to other biochemical events, such as fermentation (Kalua and Boss, 149 2008; Salinas, Zalacain, Pardo & Alonso, 2004). However, as crushing grapes induces formation 150 of some compounds that probably are not constitutive of grapes, the detected compounds were 151 named 'varietal and pre-fermentation related volatile compounds' similarly to what was 152 previously proposed by Coehlo and co-workers (2007), thus indicating both compounds 153 constitutive of grapes and compounds which arise in the grape homogenate prior to fermentation. 154 After the stirring step, the bar was picked up from the aqueous grape homogenate, rinsed with 155 distilled water, dried with paper, transferred into a glass thermal desorption tube and introduced 156 into the thermal-desorption unit (TDU, Gerstel, Mulheim and der Ruhe, Germany) in the splitless 157 mode. Thermal desorption was carried out with the following temperature program: 30 °C for 158 0.10 min, ramp rate of 120°C/min to 280°C, and 280°C for 1.00 min. The desorbed analytes were 159 cryo-focused at 0 °C (maintained by the use of liquid CO₂) in a PTV injector (CIS, Gerstel, 160 Germany) for the total desorption time, then ramped at $12^{\circ}C/s$ until $300^{\circ}C$ and held at that 161 temperature for 6.0 min. The analytes were separated on a DB-WAX column (J&W 122-7032; 30 162 m * 0.25 micron * 0.25 mm ID), using He as gas-carrier at a flow rate of 1 ml/min. The GC/MS 163 was an Agilent Technologies, GC 7890A, MS 5975C; the MS ionization energy was set at 70 eV 164 and masses were acquired from 19 to 400 m/z in full scan acquisition mode. The oven GC initial 165 temperature was set at 40 °C for 10 minutes, then at 180 °C with a ramp rate of 2.5 °C/min. 166 Temperature increased to 200°C at 1°C/min and was maintained for 10 minutes. The transfer line 167 temperature was 280 °C. After each desorption the magnetic stir bars were cleaned by immersion 168 in methanol for 24 hours (stirring during the first hour).

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170 2.2.1 Qualitative analysis

171 Volatile compounds were identified comparing mass spectra with the data system library (NIST-172 05a), by comparison with spectra found in literature (NIST Chemistry WebBook, 173 webbook.nist.gov/chemistry/), and/or according with the volatile constituents found in previous 174 studies on Nebbiolo grapes and isolated by means of liquid extraction (Di Stefano, Bottero, 175 Pigella, Borsa, Bezzo & Corino, 1998). In the case of comparison with the data system library, 176 positive characterization was accepted when a compound was identified with a probability higher 177 than 85% in all replicates. Furthermore, for qualitative identification purposes, Kovats indices of 178 identified compounds were calculated using an alkane standard mixture C10-C40 (Sigma-179 Aldrich, St. Louis, MO) as reference for retention times.

181 2.2.2 Semi-quantitative analysis of volatiles

182 Volatile compounds were quantified only when they were present in at least two replicates out of 183 the three of each sample. Concentrations of each identified compound were calculated by 184 comparing each compound peak area response to that of the internal standard; data were 185 expressed as µg equivalents of 2-heptanol per kg of fresh berries. The ratio between favourable 186 and unfavourable compounds was calculated by dividing the sum of the concentrations of 187 volatiles which in the literature are considered favourable to the human senses and that of C6 188 compounds, to whom unpleasant herbaceous notes are attributed. Favourable compounds 189 included aliphatic aldehydes and alcohols other than C6, aromatic aldehydes, esters, terpenes, 190 sesquiterpenes, benzene derivatives, lactones and norisoprenoids.

191

192 2.3 Statistical analysis

A separate extraction and analysis was done from each replicate. For each treatment, the data of the three replicates for each class of detected compounds were averaged and the standard error was calculated. Data collected at comparable phenological stages underwent an analysis of variance (Anova) to find out significant differences among locations (Duncan test for $P \le 0.05$ and $P \le 0.01$). A principal component analysis (PCA) was performed on normalized data; all statistics were carried out with SAS 8.2 for Windows, (SAS Institute, Cary, USA).

199

3. Results

201 Some important meteorological differences were detected among the three growing locations. 202 The vegetative season (April-October) in the Barbaresco area was characterized by a higher 203 cumulated global solar radiation (4200 MJ/m² against 3382 and 3492 MJ/m² in Barolo and Barbaresco, respectively). The vegetative period in the Roero area was cooler and received more
rainfall than in the other two areas (1695 growing degree days base 10 °C, versus 1864 in
Barbaresco and 1939 in Barolo; 626 mm rainfall, versus 561 in Barbaresco and 528 in Barolo).

- 207
- 208 3.1 Varietal and pre-fermentative volatile profiles

209 The varietal and pre-fermentative volatiles of Nebbiolo identified were 43, and could be grouped 210 in 9 chemical classes: aliphatic aldehydes (8 compounds), aromatic aldehydes (2 compounds), 211 aliphatic alcohols (7 compounds), monoterpenes (6 compounds), benzene derivatives (7 212 compounds), esters (7 compounds), lactones (2 compounds), sesquiterpenes (2 compounds) and 213 norisoprenoids (2 compounds; Table 1). C6 compounds (hexanal, (E)-2-hexenal, 1-hexanol, (Z)-214 3-hexenol and (E)-2-hexenol) were identified within aliphatic aldehydes and alcohols. The 215 volatiles of Nebbiolo grapes were mostly common to the three growing areas, except a few 216 compounds that were typical of the Roero vineyard, namely (E, E)-2,4-heptadienal, (E)-2-octen-217 1-ol, the norisoprenoid TDN and y-butyrolactone (Tables 2, 3 and 4). Alpha-terpineol was 218 exclusively found at 19 days post véraison (dpv) in grapes from Barbaresco (Table 2).

The majority of varietal and pre-fermentative volatiles found in Nebbiolo grapes have been previously identified in berries from other *Vitis* varieties except penthyl-hexanoate, butylhexadecanoate, two sesquiterpenes (longicyclene and junipene), and a furanone-type compound (Table 1). The two sesquiterpenes and the furanon-like compound were tentatively identified as such on the basis of NIST 05a; for all of them the probability of matching identification was around 99 %.

225

226 3.2 Accumulation trends of varietal and pre-fermentative volatiles during ripening

227 *3.2.1 Total aldehydes and alcohols*

228 Total aldehydes regularly accumulated throughout ripening in grapes from Barolo whereas in 229 grapes from Barbaresco and Roero they reached a peak of maximum accumulation followed by a 230 decrease (Fig. 1). Total aldehydes were the most abundant compounds from 20days post véraison 231 (dpv), similarly to what was previously detected in neutral Monastrell grapes (Salinas, Zalacain & 232 Pardo, 2004). Their concentration ranged from 30% before véraison to about 70% of total 233 volatiles at the last sampling date. Among the detected aldehydes, C6 aldehydes (hexanal and 234 (E)-2-hexenal) were the most abundant; the other aliphatic aldehydes were generally absent at the 235 first sampling at véraison, but they started to accumulate in the period between 6 and 18 dpv 236 (Tables 2, 3 and 4).

237 Aliphatic alcohols constantly accumulated in grapes from Barbaresco whereas they reached a 238 peak of maximum concentration in grapes from Roero (at 25 dpv) and from Barolo (at 45 dpv, 239 Fig. 1), followed by a decrease. At the first time point, independently from the growing area, 240 aliphatic alcohols (except octanol and (E)-2-octen-1-ol, whose accumulation began later) were 241 already present (Tables 2, 3 and 4). Among aliphatic alcohols, 1-hexanol, (Z)-3-hexenol and (E)-242 2-hexen-1-ol belong to the sub-group of C6 compounds whose concentration decreased during 243 ripening in grapes from Roero (Table 4), whereas it increased in grapes from Barbaresco and 244 Barolo (Tables 2 and 3). C6 aldehydes and alcohols are known to provide the green, grassy notes 245 of many fruits (Klee, 2010). They are formed subsequently to the crushing of berries thanks to 246 the berry constitutive lipoxygenase activity (Gunata, Bayonove, Baumes & Cordonnier, 1985).

247

248 3.2.2 Esters

The total ester content increased during ripening; ester biosynthesis started 25 dpv in grapes from Barbaresco and Barolo, earlier in grapes from Roero (Fig. 1). No ethyl esters were detected whereas methyl- and penthyl- esters of butanoic, nonanoic, decanoic and hexadecanoic acids were identified (Table 1). Among ethyl-esters we exclusively found ethyl-hexadecanoate in trace amounts and exclusively in grapes from the Roero area (Table 4). The fact that no ethyl-esters were detected in homogenates of Nebbiolo grapes could indicate that no fermentation of the juice started during grinding and extraction with the SBSE.

Some important differences were found in the type of accumulated esters: grapes from the Barolo area were characterized by the presence of ethyl-hexanoate, methyl-3-OH-butanoate and methylnonanoate (Table 3); conversely, grapes from Barbaresco accumulated methyl-decanoate and butyl-hexadecanoate (Table 2). At harvest, ester total concentrations were similar in grapes from Barbaresco and Barolo whereas quantities detected in Roero grapes were significantly lower (Fig. 1; Table 4).

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263 3.2.3 Terpenes and sesquiterpenes

264 The trend of total terpene accumulation in Nebbiolo showed important differences among 265 vineyards. The biosynthesis of terpenes started before véraison in all vineyards but peaks of 266 maximum accumulation were reached in different moments depending on the growing area (Fig. 267 1). The importance and the sensorial impact of terpenes in floral grapes is well known (Gunata, 268 Bayonove, Baumes & Cordonnier, 1985). Terpenes possess very pleasant 'sweet' and 'floral' 269 aromas and a very low olfactory threshold that allows them to be easily recognizable even at very 270 low concentration. In non-floral cv Monastrell, Salinas and co-workers (2004) reported the 271 existence of a series of non-glycosilated terpenes; in Nebbiolo grapes we identified D-limonene, α -terpineol (a distinctive marker found exclusively in grapes from Barbaresco at 19 dpv, Table 2), (*Z*)-citral, β -citronellool and (*E*)-geranylacetone (Tables 2, 3 and 4); in agreement with Salinas and co-workers (2004) no linalool was detected, showing that the existence of trace amounts of this terpene as varietal volatile should be more properly attributed to artefact formation after the activation of constitutive grape glycosidases, which were not activated with the method we used.

277 Sesquiterpenes possess important biological roles as attractant of insects for pollination or as 278 defence molecules against fungi. The most known sesquiterpene molecule investigated up to now 279 in viticulture/enology is rotundone, associated to the pepper aroma of Shiraz wines (Wood, et al., 280 2008); although little is known about the implications on odour and taste induced by other 281 sesquiterpene molecules in grapes and wines, some sesquiterpenes have recently been described 282 in cv Baga (Rocha, Delgadillo & Coimbra, 2006), and in Cabernet Sauvignon and Riesling 283 (Kalua & Boss, 2010). Sesquiterpene accumulation trend was very characteristic as no 284 sesquiterpenes were detected before véraison; accumulation started since 20 dpv in fruit from 285 Barolo and Barbaresco, and earlier, around véraison, in grapes from Roero (Fig. 1). These 286 observations are in line with those previously reported by Coelho and co-workers (2006) in cv 287 Baga where sesquiterpenes were detected from 14 dpv onwards and with the fact that Vitis 288 vinifera sesquiterpene synthase transcript were exclusively detected during the last phases of 289 berry ripening (Lücker, Bowen & Bohlmann, 2004).

290

291 *3.2.4 Norisoprenoids*

The importance of norisoprenoids in grape aroma is well known; they have been intensively studied as glycosides in many varieties but they have also been reported as components of varietal and pre-fermentative flavours (Salinas, Zalacain, Pardo & Alonso, 2004; Coehlo, Rocha, 295 Barros, Delgadillo & Coimbra, 2007; Kalua & Boss, 2010). Salinas and co-workers (2004) 296 detected vomifoliol, 3-oxo- α -ionol and β -ionone whereas Coehlo and co-workers (2007) detected 297 β -damascenone and β -ionone and Kalua & Boss (2010) exclusively β -ionone at precise moments 298 of ripening. In Nebbiolo grapes, except for trace amounts of TDN found in grapes from Roero at 299 24 and 38 dpv (Tab. 4), we exclusively detected β -ionone (Table 2, 3 and 4), similarly to 300 Pedroza, Zalacain, Lara and Salinas (2010) and Kalua & Boss (2010) did. β-ionone biosynthesis 301 started before véraison and increased throughout ripening, peaking at 10 to 20 dpv depending on 302 the cultivation site. From 20 dpv onwards, norisoprenoid concentration decreased in all 303 vineyards. The incidence of norisoprenoids over total varietal and pre-fermentative volatile 304 amount was very low, ranging from 0.014 to 2.6 %.

- 305
- 306 *3.2.5 Benzene derivatives and lactones*

Benzene derivatives were synthesized already before véraison; their accumulation trend was greatly influenced by the growing area (Fig. 1). At harvest, benzene derivatives accounted for 9.6 % in grapes from Barbaresco, for 5.3 % in grapes from Barolo and for 7.3 % in grapes from Roero. Benzyl alcohol was exclusively detected in grapes from Barolo at harvest (Table 3). The accumulation of benzene derivatives peaked 20 dpv in grapes from Barbaresco and Roero, whereas a completely different behaviour characterized grapes from Barolo (Fig. 1).

313 A furanone-type compound was found in grapes from all three vineyards; Roero grapes were also 314 characterized by the accumulation of γ -butyrolactone (Table 2, 3 and 4). The biosynthesis of 315 lactones started before véraison; similar trends were detected in grapes from Barolo and Roero (a 316 progressive slow increase of concentrations) whereas a peak of maximum accumulation was 317 reached 45 dpv in grapes from Barbaresco (Fig. 1).

318

319 3.2.6 Total varietal volatiles and favourable/unfavourable volatile compound ratios

320 Total varietal volatiles showed an accumulation trend during ripening in grapes from Barolo and 321 Barbaresco whereas a phase of plateau since 15 dpv characterised berries from the Roero area 322 that globally accumulated lower quantities of varietal and pre-fermentative volatiles (Fig. 1). 323 Total volatile maximum concentration was reached at about 45 dpv in grapes from Barbaresco 324 and Barolo, earlier in grapes from Roero (Fig. 1). However, at harvest total amounts of volatiles 325 in Roero grapes were similar to those of berries from the other two areas (Table 4). In grapes 326 from Barolo a final concentration reduction was detected before harvest and this was due to the 327 concomitant reduction of aliphatic alcohol, ester and terpene concentrations (Fig. 1). In Barolo 328 and Barbaresco grapes, the maximum volatile varietal content occurred in a very short period, 329 about 45 dpv, coincident with the highest values of the favourable/unfavourable compound ratio 330 (Fig. 1). The ratio between favourable and unfavourable varietal and pre-fermentative volatiles, 331 calculated since 20 dpv, when phenological stages were comparable among the studied 332 vineyards, showed values ranging from 0.2 to 4.2. It was constantly over 1 in grapes from 333 Barbaresco due to the lower amounts of C6 compounds and to the progressive accumulation of 334 favourable compounds.

335

336 *3.2.7 PCA analysis*

To clarify possible relations among the different classes of compounds and identify general parameters able to distinguish growing locations, we performed a principal component analysis (PCA) on nine variable normalized data (Table 5); with the first three principal components (Prin) the model proposed justified 89 % of the total variance (Table 5). According to the eigenvalues, 5 variables were associated to Prin 1, namely the percentages of norisoprenoids,
volatile total concentrations, the percentages of benzene derivatives, total aldehydes and
sesquiterpenes. One variable (the percentage of total terpenes) was associated to the second Prin;
alcohols were positively associated to the third Prin, although with an eigenvalue slightly lower
than 1 (0.97; Tab. 5).

346 By observing the distribution of individuals in the x-y-z graph (Fig. 2) we noticed that 347 exclusively the first date of sampling was discriminated on Prin 1; the PCA model proposed was 348 effective in the discrimination of samples from the Barbaresco area (except the first sampling), all 349 individuals being well associated to the positive values of Prin 2 and thus characterized by higher 350 terpene content incidence over total concentrations respect to grapes from the other locations. 351 Moreover, samples from the Roero area were well grouped around nil values of Prin 2 (except the 352 first sampling); the third principal component, that alone justified the 10 % of total variance, was 353 not so effective in individual discrimination even though it partially discriminated some 354 Barbaresco and Barolo cases associated to its positive values (Fig. 2).

As to variable correlation (Fig. 2) we found that the percentage of total aldehydes was negatively correlated with that of benzene derivatives (R = -0.96), norisoprenoids (R = -0.80) and terpenes (R = -0.86); these last two in particular are classes of compounds that largely contribute to positive notes of berry aroma. Moreover, benzene derivatives, terpenes and norisoprenoids were positively correlated with each other, with correlation coefficients (R) always higher than 0.7.

360

361 **4. Discussion**

362 The flavour differences among 'Barbaresco', 'Barolo' and 'Roero' DOCG wines are well-known363 to wine consumers and they have also been characterized through GC/MS analysis and

364 multivariate statistical tools (Marengo, Aceto & Maurino, 2001). At the last sampling date (that 365 was coincident with harvest for viticulturists) significant differences among the three vineyards 366 were found not only for esters but also for aliphatic alcohols, terpenes, benzene derivatives and 367 total volatile concentration (Table 6). Grapes from the Barbaresco area were distinguishable from 368 those from Barolo and Roero and very often they showed similarity with those from the Barolo 369 area. At the wine level Marengo, Aceto & Maurino (2001) pointed out that wines that underwent 370 a more prolonged ageing (Barbaresco and Barolo) were associated with each other whereas they 371 were quite distant from wines, such as those from Roero, that underwent a reduced ageing period; 372 differences were tied to a series of compounds but the most implicated compounds were esters. In 373 this study we show that Barbaresco and Barolo grapes accumulated more esters than Roero 374 grapes, thus the differences previously detected at the wine level could have a previous origin in 375 grapes rather than be exclusively due to the numerous fermentative reactions that induce ester 376 formation during winemaking and ageing.

377 As shown from the interpretation of the PCA model, grapes from the Barbaresco area, except 378 those of the first sampling date, were well separated on the second principal component axis, 379 representing the incidence of terpenes over total volatiles; previous studies (Bureau, Razungles & 380 Baumes 2000) reported that artificial bunch shading, by reducing illumination of clusters, 381 decreased the concentration of free terpenols and norisoprenoids respect to sun-exposed berries. 382 The Barbaresco growing area was characterized by a higher cumulated global radiation, and by 383 the accumulation of higher quantities of varietal and pre-fermentative terpenes and 384 norisoprenoids, confirming the fundamental role of light on of the biosynthesis of these 385 secondary metabolites.

386 Elaboration of the PCA model proposed suggests that the first sampling date was distinguished 387 from the others, and associated to the positive values of the first principal component axis (Fig. 388 2), whereas the successive sampling dates were not discernable anymore, suggesting that at 389 véraison the biosynthesis of varietal volatiles was not influenced by the growing area, but 390 differences among areas occurred later. Thus the steps of flavour biosynthesis more influenced by 391 climatic conditions seem to be successive to véraison, even though nothing can be said about the 392 steps before véraison as we started sampling at véraison; the only study dealing with the 393 accumulation of varietal volatiles before véraison concluded that a crucial step of volatile 394 biosynthesis occurs around 4 weeks after flowering in Cabernet Sauvignon and Riesling (Kalua & 395 Boss, 2010). Our data suggest that at véraison there are no crucial steps in Nebbiolo grape flavour 396 biosynthesis but they take place later and the growing area largely influences them.

397 The evolution of volatile classes during ripening suggests some further general patterns of 398 biosynthesis. The negative correlation between aldehydes (on average 91 % represented by C6 399 compounds) on one side and terpenes, norisoprenoids and benzene derivatives on the other, could 400 be an analytical evidence of the fact that the accumulation of 'negative' or 'positive' varietal 401 volatile is antithetical. The positive correlation terpenes/norisoprenoids (R = 0.70) could be 402 explained as an analytical evidence of the known common steps of monoterpenes and carotenoids 403 (the precursor of norisoprenoids) biosynthetic pathways (Mathieu, Wirth, Sauvage, Lepoutre, 404 Baumes & Gunata, 2009).

The use of crushed grapes together with prolonged time of SBSE/grape homogenate contact have probably promoted the formation of C6 compounds as it was the case in other studies using SBSE or SPME techniques and implying grinding. C6 compounds accounted for 14 to 71 % of total varietal volatiles with increasing values from véraison to harvest in Barolo and Barbaresco

409 grapes; in Roero grapes the incidence of C6 compounds over total was already high (49 %) at the 410 first sampling and it increased to 69 % at harvest. At harvest, independently from the cultivation 411 area, C6 compounds represented the major class of grape volatiles, consistent with previous 412 findings (Kalua & Boss, 2009; Kalua & Boss, 2010; Yang et al., 2009). Temporal analysis of the 413 ratio between favourable and unfavourable (C6 aldehydes and alcohols) compounds shows that in 414 grapes from Barbaresco and Barolo the optimum values were reached at 42 and 44 dpv, 415 respectively, thus about 15 days before the date of effective harvest. Oliveira, Faria, Sá, Barros & 416 Araújo (2006) discovered the effectiveness of C6 compounds, particularly of the ratio (E)-3-417 hexenol/(Z)-3-hexenol in discriminating wine origin in monovarietal wines. In Nebbiolo we 418 found, besides hexanol, (Z)-3-hexenol and (E)-2-hexen-1-ol, and we were thus able to calculate 419 their ratio, which at harvest was 0.6 for Roero, 1.2 for Barolo and 1.3 for Barbaresco grapes. The 420 classificatory capacity of this ratio is thus evident and once more an association between 421 Barbaresco and Barolo grapes on one side and Roero grapes on the other, emerges. Nevertheless 422 further studies will confirm the capability of ratios such as this to discriminate growing locations. 423 Calculation of this or other ratios could represent an important start point for finding in the wide 424 range of varietal and pre-fermentative volatiles some target compounds that could be considered 425 as a possible 'fingerprints', not only of the growing locations but also of the variety. Further 426 studies are ongoing in our lab on other varieties and over more years.

427

428 4. Conclusions

429 The SBSE was effectively used for the first time to detect varietal and pre-fermentative volatiles430 in grapes of cv Nebbiolo, an important neutral Italian variety, base cultivar for the production of

Barbaresco, Barolo and Roero wines. The influence exerted by the growing location on volatileconcentration and profile was analysed, as well as on the timing of volatile biosynthesis.

433 One of the main aims of this study was to verify the existence of differences in the volatile profile 434 of the same genotype, grown in different areas or terroirs. The results show that, even in a 435 relatively homogenous geographic area like southern Piedmont, the growing locations induce 436 differences in the concentration and accumulation trend of several classes of compounds, in 437 particular esters. On the contrary, certain classes of compounds such as sesquiterpenes and 438 norisoprenoids showed no differences in the pattern of accumulation in the different areas. 439 Varietal and pre-fermentative volatiles seemed to be more effective in separating growing areas 440 than dates of sampling: as a matter of fact grapes from the Barbaresco area, showing higher 441 values of the favourable/unfavourable compound ratio throughout ripening, were separated from 442 grapes from the other areas.

443 The concentrations of certain classes of compounds such as terpenes, norisoprenoids and benzene 444 derivatives were generally correlated among them so that this opens the possibility to follow the 445 'aromatic ripening' of grapes by following the evolution of a unique class of compounds.

Further studies, over more years and also on global aroma potential, i.e. including glycoconjugate volatile detection, will be necessary to better identify Nebbiolo varietal volatiles, to deepen knowledge about volatile concentrations and profiles and to investigate their implications on wine quality as influenced by the growing area.

450

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- 458
- 459 References
- 460 Baltussen E., Sandra P., David F., & Cramers C.A. (1999). Stir Bar Sorptive Extraction (SBSE),
- 461 a novel extraction technique for aqueous samples: theory and principles. *J. Microcolumn*462 *Sep, 11,* 737-747.
- Baumes R., Cordonnier R., Nitz S., & Drawert F. (1986). Identification and determination of
 volatile constituents in wines from different vine cultivars. *Journal of the Science of Food and Agriculture*, *37*, 927-943.
- Bureau S. M., Razungles A. J. & Baumes R.I. (2000). The aroma of Muscat of Frontignan
 grapes: effect of the light environment of vine or bunch on volatiles and glycoconjugates.
- 468 *Journal of the Science of Food and Agriculture*, 80, 2012-2020.
- Cabrita M. J., Costa Freitas A. M., Laureano O., & Di Stefano R. (2006). Glycosidic aroma
 compounds of some Portuguese grape cultivars. *Journal of the Science of Food and Agriculture*, 86, 922-931.
- 472 Caven-Quantrill D.J., & Buglass A.J. (2007). Determination of volatile organic compounds in
 473 English vineyard grape juices by immersion stir bar sorptive extraction-gas
 474 chromatography/mass spectrometry. *Flavour Fragrance Journal.*, 22, 206–213.

475	Coelho E., Rocha S. M., Delgadillo I., & Coimbra M. A. (2006). Headspace-SPME applied to
476	varietal volatile components evolution during Vitis vinifera L. cv. 'Baga' ripening.
477	Analitica Chimica Acta, 563, 204–214.

- 478 Coelho E., Rocha S.M., Barros A.S., Delgadillo I., & Coimbra M.A. (2007). Screening of variety-
- 479 and pre-fermentation related volatile compounds during ripening of white grapes to define
 480 their evolution profile. *Analitica Chimica Acta*, 597, 257-264.
- 481 Di Stefano R., Bottero S., Pigella R., Borsa D., Bezzo G., & Corino L. (1998). Precursori d'aroma
 482 glicosilati presenti nelle uve di alcune cultivar a frutto colorato. *L'Enotecnico*, 63-74.
- Gomez E., Martinez A., & Laencina J. (1995). Changes in volatile compounds during ripening of
 some grape varieties. *Journal of the Science of Food and Agriculture*, 67, 229-233.
- 485 Gunata Y.Z., Bayonove C.L., Baumes R.L., & Cordonnier R.E. (1985). The aroma of grapes.

486 Localisation and evolution of free and bound fractions of some grape aroma components

- 487 cv Muscat during first development and maturation. *Journal of the Science of Food and*488 *Agriculture*, *36*, 857-862.
- Ji T. J., & Dami I. E. (2008). Characterization of free flavour compounds in Traminette grape and
 their relationship to vineyard training system and location. *Journal of Food Science*,
 73(4), C262-C267.
- Kalua C.M,. & Boss P.K. (2008). Sample preparation optimization in wine and grapes: dilution
 and sample/headspace volume equilibrium theory for headspace solid-phase
 microextraction. *Journal Chromatography A*, *1192*, 25-35.
- Kalua C.M., & Boss P.K. (2009). Evolution of volatile compounds during the development of
 Cabernet Sauvignon grapes (*Vitis vinifera* L). *Journal of Agriculture and Food Chemistry*,
 57, 3818-3830.

- Kalua C.M., & Boss P.K. (20r0). Comparison of major volatile compounds from Riesling and
 cabernet Sauvignon grapes (*Vitis vinifera* L.) from fruitset to harvest. *Australian Journal of Grape and Wine Research*, *16*, 337-348.
- 501 Klee H. (2010). Improving the flavor of fresh fruits: genomics, biochemistry, and biotechnology.
 502 *New Phytologist*, *187*, 44-56.
- La Guerche S., de Dauphin B., Pons M., Blancard D., & Darriet P. (2006). Characterization of
 some mushroom and earthy off-odors microbially induced by the development of rot on
 grapes. *Journal of Agriculture and Food Chemistry*, *54*, 9193-9200.
- Luan F., Mosandl A., Munch A., & Wust M. (2005). Metabolism of geraniol in grape berry
 mesocarp of *Vitis vinifera* L. Cv. Scheurebe: demonstration of seteroselective reduction,
 E/Z-isomerization, oxidation and glycosylation. *Phytochemistry*, *66*, 295-303.
- 509 Lücker J., Bowen P., & Bohlmann J. (2004). Vitis vinifera terpenoid cyclases: functional
- 510 identification of two sesquiterpene synthase cDNAs encoding (+)-valencene synthase and
- 511 (-)-germacrene D synthase and expression of mono- and sesquiterpene synthases in
- 512 grapevine flowers and berries. *Phytochemistry*, 65, 2649-2659.
- 513 Marengo E., Aceto M., & Maurino V. (2001). Classification of Nebbiolo-based wines from
- 514 Piedmont (Italy) by means of solid-phase microextraction–gas chromatography–mass
 515 spectrometry of volatile compounds. *Journal of Chromatography A*, 943, 123–137.
- 516 Matthieu S., Wirth J., Sauvage F.X., Lepoutre JP., Baumes R., & Gunata Z. (2009).
- 517 Biotransformation of C-13 norisoprenoids and monoterpenes by a cell suspension culture
 518 of cv Gamay (*Vitis vinifera*). *Plant Cell Tissue Organ Culture*, 97: 203-213.
- 519 Oliveira J.M., Faria M., Sá F., Barros F., & Araújo I.M. (2006). C6.alcohols as varietal markers
 520 for assessment of wine origin. *Analytica Chimica Acta*, *563*, 300-309.

- 521 Pedroza M.A., Zalacain A., Lara J.F., & Salinas M.R. (2010). Global grape aroma potential and
 522 its individual analysis by SBSE-GC-MS. *Food Research International*, *43*, 1003-1008.
- 523 Ruberto G., Renda A., Amico V., & Tringali C. (2008). Volatile components of grape pomaces
- from different cultivars of Sicilian *Vitis vinifera* L. *Bioresource Technology*, *99*, 260–268.
- 525 Salinas M.R., Zalacain A., Pardo F., & Alonso G.L. (2004). Stir Bar Sorptive Extraction applied
- to volatile constituents evolution during Vitis vinifera ripening. *Journal of Agriculture and Food Chemistry*, 52, 4821-4827.
- Sefton M.A, Francis L., & Williams P.J. (1993). The volatile composition of Chardonnay juices:
 a study by flavor precursor analysis. *American Journal of Enology and Viticulture, 44*, 359-370.
- Sotiroudis G., Melliou E., Sotiroudis T. G., & Chinou I. (2010). Chemical analysis, antioxidant
 and antimicrobial activity of three greek cucumber (*Cucumis sativus*) cultivars. *Journal of Food Biochemistry*, 34, 61–78.
- 534 Wood C., Siebert T. E., Parker M., Capone D. L., Elsey G. M., Pollnitz A. P., Eggers M., Meier
- M., Vössing T., Widder S., Krammer G., Sefton M. A., Herderich M. J. (2008). From
 wine to pepper: rotundone, an obscure sesquiterpene, is a potent spicy aroma compound. *Journal of Agriculture and Food Chemistry*, *56*, 3738-3744.
- 538 Yang C.X., Wang Y.J., Liang Z.C., Fan P.G, Wu B.H., Yang L., Wang Y.N., & Li S.H. (2009).
- 539 Volatiles of grape berries evaluated at the germplasm level by headspace-SPME with GC540 MS. *Food Chemistry*, *114*, 1106-1114.
- 541 Yang C., Wang Y., Wu B., Fang J. & Li S. (2011). Volatile compounds evolution of three table
 542 grapes with different flavour during and after maturation. *Food Chemistry*, *128*, 823-830.
- 543

544 CAPTIONS

Table 1 - Varietal and pre-fermentative volatile compounds found in *Vitis vinifera* cv Nebbiolo
grapes by SBSE/GC-MS.

547

Table 2 – Evolution of varietal and pre-fermentative volatiles in *Vitis vinifera* cv Nebbiolo grapes (as $\mu g g^{-1}$ fresh berries) from the Barbaresco area, during ripening. Mean values of three replicates \pm standard errors. Dpv = days post-véraison; SSC = soluble solid content; ND = not detected.

552

Table 3 – Evolution of varietal and pre-fermentative volatiles in *Vitis vinifera* cv Nebbiolo grapes (as $\mu g g^{-1}$ fresh berries) from the Barolo area, during ripening. Mean values of three replicates ± standard errors. Dpv = days post-véraison; SSC = soluble solid content; ND = not detected.

556

Table 4 – Evolution of varietal and pre-fermentative volatiles in *Vitis vinifera* cv Nebbiolo grapes (as μ g g⁻¹ fresh berries) from the Roero area, during ripening. Mean values of three replicates ± standard errors. Dpv = days post-véraison; SSC = soluble solid content; ND = not detected.

560

Table 5 - Eigenvectors of the examined variables on the three principal components (Prin 1, Prin
2 and Prin 3). Eigenvalues of the three Prins and their contribution to total variance. In bold
letters the variables associated to the appropriate Prin.

564

Table 6 – Analysis of variance of averages of the main classes of varietal and pre-fermentative
volatiles of *Vitis vinifera* cv Nebbiolo grapes from three different growing locations at

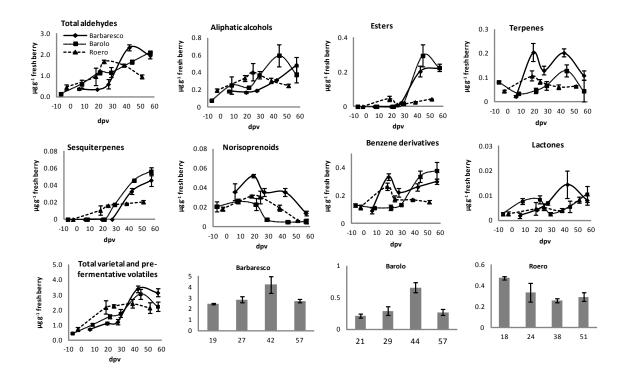
567 comparable phenological stages (1 = 19 days post véraison, dpv; 2 = 27 dpv; 3 = 41 dpv; 4 = 55 568 dpv). Within the same phenological stage, averages were subjected to the analysis of variance 569 and means were separated by the Duncan's test. Means followed by different letters are 570 significantly different for P \leq 0.05 (lowercase letters) and P \leq 0.01 (uppercase letters).

571

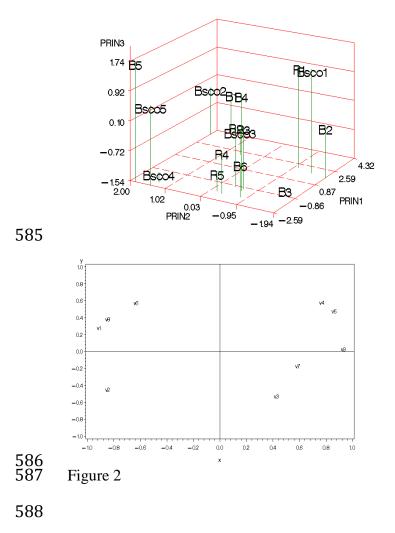
Figure 1 – Accumulation trend of the main classes of varietal and pre-fermentative volatiles in *Vitis vinifera* cv Nebbiolo grapes from three different growing areas in Piedmont (North-West Italy). Average values of three replicates \pm standard errors. Histograms of the favourable/unfavourable compound ratio at similar phenological stages.

576

Figure 2 – Distribution of individuals on the x-y-z axis. Bsco = Barbaresco; B = Barolo; R = 578Roero. The numbers after the acronym indicating the growing area stand for the different picking times. x-y plot of the nine variables used to run the principal component analysis. See footnote of Table 5 for variable identification.



- 582 Figure 1



	Compound	KI	Reference	
Alishatia aldahudaa	Hexanal ¹	1064	Kolua & Daga 2010	
Aliphatic aldehydes		1064	Kalua & Boss, 2010	
	(E)-2-hexenal ¹	1213	Kalua & Boss, 2010	
	octanal	1291	Kalua & Boss, 2010	
	(Z)-2-heptenal	1319	Caven-Quantrill & Buglass, 2007	*
	(<i>E,E</i>)-2,4-heptadienal	1485	Yang et al., 2011	
	decanal	1497	Salinas et al., 2004	
	(E)-2-nonenal	1530	Ruberto et al., 2008	
	(<i>E,Z</i>)-2,6-nonadienal	1580	Sotiroudis et al., 2009 in cocumber	
Aromatic aldehydes	furfural	1457	Caven-Quantrill & Buglass, 2007	
	benzaldehyde	1510	Sefton et al., 1993	
Aliphatic alcohols	1-butanol	1159	Caven-Quantrill & Buglass, 2007	
	1-hexanol ¹	1363	Kalua & Boss, 2010	
	(Z)-3-hexenol ¹	1387	Sefton et al., 1993	
	(E)-2-hexen-1-ol ¹	1410	Coehlo et al., 2007	
	2-ethyl-hexanol	1501	Caven-Quantrill & Buglass, 2007	
	1-octanol	1570	Yang et al., 2009	
Tomonoo	(<i>E</i>)-2-octen-1-ol	1628	La Guerche et al., 2006 from <i>Botrytis cinerea</i> infected grapes	*
Terpenes	D-limonene	1085	Coehlo et al., 2007	*
	B-cyclocitral	1608	Coehlo et al., 2006	
		1648	Caven-Quantrill & Buglass, 2007	
	(E)-geranylacetone	1662	Salinas et al., 2004	
	α-terpineol	1707	Sefton et al., 1993	
	β-citronellol	1781	Coehlo et al., 2007	
Benzene derivatives	acetophenone	1640	Caven-Quantrill & Buglass, 2007	
	benzyl alcohol	1686	Sefton et al., 1993	*
	benzothiazole	1898	Sefton et al., 1993	
	phenol	1831	Sefton et al., 1993	
	2-phenoxy ethanol (rose ether)	2105	Caven-Quantrill & Buglass, 2007	
	para-buthyl-cresol	2053	Sefton et al., 1993	
	trimethyl-tetrahydro- benzofuranone	2124	Caven-Quantrill & Buglass, 2007	
Esters	ethyl-hexanoate	1240	Yang et al., 2009; Ruberto et al., 2008	*
	methyl-3-OH-butanoate	1483	Yang et al.[6]	*
	methyl-nonanoate	1495		*
	penthyl-hexanoate	1515		
	methyl-decanoate	2035		*
	methyl-hexadecanoate	2061	Caven-Quantrill & Buglass, 2007	
	butyl-hexadecanoate	2253		*
Lactones	y-butyrolactone	1609	Sefton et al., 1993	*
	furanone type compound	1710		
Sesquiterpenes	sesquiterpene 1	1480		
Cooquitorpolito	sesquiterpene 2	1551		
Norisoprenoids	TDN	1735	Sefton et al., 1993 after H ⁺ hydrolysis	*
		1100	Salinas et al., 2004	

Table 1 KI = Kovats index. (1) Compounds attributable to the sub-group of C6 compounds. Compound identification was performed by comparison with mass spectra given by

590 591 NIST05a library or with available spectra in literature. Compounds followed by the asterisk were detected exclusively in grapes from one vineyard. Sesquiterpenes 1 and 2 were

tentatively identified as longicyclene and junipene, respectively.

	SSC (°Brix)	6 dpv 11.3	19 dpv 18.4	27 dpv 19.3	42 dpv 24.1	57 dpv 22.0
Aliphatic aldehydes	hexanal	0.0895 ± 0.0104	0.0038 ± 0.0008	0.0089 ± 0.0040	0.6815 <u>+</u> 0.1051	0.6677 ± 0.0403
	(E)-2-hexenal	0.1595 ± 0.0328	0.1136 ± 0.0130	0.3979 ± 0.2153	1.1076 ± 0.2763	0.9895 ± 0.0547
	octanal	0.0029 ± 0.0016	ND	0.0074 ± 0.0032	0.0064 ± 0.0004	0.0317 ± 0.0112
	(Z)-2-heptenal	0.0655 ± 0.0225	0.0626 ± 0.0205	0.0528 ± 0.0066	0.0167 ± 0.0050	ND
	(E,E)-2,4-heptadienal	ND	ND	ND	ND	ND
	decanal	ND	ND	ND	0.0111 ± 0.0001	0.0273 ± 0.0086
	(E)-2-nonenal	ND	ND	ND	0.0838 ± 0.0102	0.0832 ± 0.0140
	(E,Z)-2,6-nonadienal	ND	ND	0.0043 ± 0.0024	0.0504 ± 0.0132	0.0440 ± 0.0093
Aromatic aldehydes	furfural	0.0660 ± 0.0153	0.1737 ± 0.0386	0.1295 ± 0.0033	0.3737 ± 0.2831	0.0974 ± 0.0278
	benzaldehyde	0.0039 ± 0.0008	0.0054 ± 0.0000	0.0033 ± 0.0011	0.0098 ± 0.0017	0.0230 ± 0.0142
Aliphatic alcohols	1-butanol	0.0159 ± 0.0016	0.0047 ± 0.0020	0.0043 ± 0.0008	0.0197 ± 0.0161	0.0447 ± 0.0127
	1-hexanol	0.0580 ± 0.0174	0.0881 ± 0.0109	0.0634 ± 0.0035	0.1318 ± 0.0182	0.1213 ± 0.0097
	(Z)-3-hexenol	0.0607 ± 0.0071	0.0299 ± 0.0023	0.0208 ± 0.0021	0.0607 ± 0.0069	0.1342 ± 0.0434
	(E)-2-hexen-1-ol	0.0423 ± 0.0041	0.0382 ± 0.0066	0.0753 ± 0.0136	0.0700 ± 0.0014	0.1090 ± 0.0231
	2-ethyl-hexanol	0.0068 ± 0.0010	0.0092 ± 0.0002	0.0103 ± 0.0021	0.0195 ± 0.0014	0.0570 ± 0.0104
	1-octanol	ND	ND	0.0164 ± 0.0091	0.0108 ± 0.0008	0.0218 ± 0.0075
	(E)-2-octen-1-olo	ND	ND	ND	ND	ND
Terpenes	isomenthol	ND	ND	ND	0.0071 ± 0.0007	0.0070 ± 0.0015
·	D-limonene	0.0085 ± 0.0016	0.0235 ± 0.0071	0.0162 ± 0.0043	0.0126 ± 0.0030	0.0233 ± 0.0060
	β-cyclocitral	ND	0.0550 ± 0.0138	0.0168 ± 0.0139	ND	ND
	a-terpineol	ND	ND	0.0234 + 0.0017	ND	ND
	β-citronellol	ND	0.0181 ± 0.0074	0.0162 ± 0.0050	ND	ND
	(E)-geranylacetone	0.0147 ± 0.0018	0.1087 ± 0.0175	0.0575 ± 0.0133	0.1836 ± 0.1250	0.0792 ± 0.0186
Benzene derivatives	acetophenone	0.0588 ± 0.0141	0.1245 ± 0.0066	0.0629 ± 0.0077	0.1279 ± 0.0145	0.1610 ± 0.0104
	benzyl alcohol	ND	ND	ND	ND	ND
	benzothiazole	ND	0.0153 ± 0.0028	0.0356 ± 0.0248	0.0145 ± 0.0005	0.0116 ± 0.0027
	phenol	0.0119 ± 0.0017	0.0238 ± 0.0037	0.0135 ± 0.0011	ND	ND
	2-phenoxy ethanol (rose ether)	0.0163 ± 0.0081	0.1272 ± 0.0026	0.0823 ± 0.0101	0.0789 ± 0.0059	0.0865 ± 0.0122
	para-buthyl-cresol	ND	0.0159 ± 0.0007	0.0104 ± 0.0026	0.0413 ± 0.0004	0.0424 ± 0.0071
	trimetil-tetrahydro-benzofuranone	0.0061 ± 0.0011	0.0276 ± 0.0035	0.0198 ± 0.0009	ND	ND
Esters	ethyl-hexanoate	ND	ND	0.0058 ± 0.0008	ND	ND
	methyl-3-OH-butanoate	ND	ND	ND	ND	ND
	methyl-nonanoate	ND	ND	ND	ND	ND
	penthyl-hexanoate	ND	ND	ND	0.0143 ± 0.0024	0.0119 ± 0.0061
	methyl-hexadecanoate	ND	ND	ND	0.0390 ± 0.0022	0.0331 ± 0.0150
	butyl-hexadecanoate	ND	ND	ND	0.1024 ± 0.0023	0.0927 ± 0.0163
Lactones	γ-butyrolactone	ND	ND	ND	ND	ND
	furanone type compound	0.0020 ± 0.0120	0.0043 ± 0.0001	0.0700 ± 0.0006	0.0354 ± 0.0216	0.0082 ± 0.0018
Sesquiterpenes	sesquiterpene 1	ND	ND	ND	0.0082 ± 0.0004	0.0073 ± 0.0011
	sesquiterpene 2	ND	ND	ND	0.0245 ± 0.0026	0.0385 ± 0.0068
Norisoprenoids	TDN	ND	ND	ND	ND	ND
	β-ionone	0.0359 ± 0.0088	0.0524 ± 0.0015	0.0357 ± 0.0040	0.0358 ± 0.0034	0.0141 ± 0.0025

Table 2

		-7 dpv	8 dpv	21 dpv	29 dpv	44 dpv	57 dpv
	SSC (°Brix)	6.6	14.8	21.4	22.6	24.5	24.5
Aliphatic aldehydes	hexanal	0.0231 ± 0.0049	0.1167 ± 0.0425	0.4282 ± 0.0470	0.3551 ± 0.1121	0.5293 ± 0.0201	0.8161 ± 0.1412
	(E)-2-hexenal	ND	0.4245 ± 0.1363	0.6484 ± 0.0360	0.7062 ± 0.1248	0.8282 ± 0.0144	0.8863 ± 0.1343
	octanal	0.0035 ± 0.0004	0.0265 ± 0.0131	0.0231 ± 0.0024	0.0056 ± 0.0000	0.0546 ± 0.0136	0.0816 ± 0.0639
	(Z)-2-heptenal	0.1122 ± 0.0346	0.0381 ± 0.0058	0.0355 ± 0.0098	0.0159 ± 0.0008	0.0000 ± 0.0000	0.0000 ± 0.0000
	(E,E)-2,4-heptadienal	ND	ND	ND	ND	ND	ND
	decanal	ND	ND	ND	0.0055 ± 0.0001	0.0749 ± 0.0423	0.1361 ± 0.1149
	(E)-2-nonenal	ND	ND	ND	0.0465 ± 0.0003	0.0535 ± 0.0113	0.0484 ± 0.0267
	(E,Z)-2,6-nonadienal	ND	ND	0.0230 ± 0.0126	0.0078 ± 0.0018	0.0271 ± 0.0137	0.0133 ± 0.0008
Aromatic aldehydes	furfural	ND	ND	ND	0.0057 ± 0.0010	0.0552 ± 0.0091	0.0834 ± 0.0158
	benzaldehyde	0.0049 ± 0.0025	0.0020 ± 0.0003	0.0062 ± 0.0009	0.0052 ± 0.0008	0.0253 ± 0.0089	0.0210 ± 0.0114
Aliphatic alcohols	1-butanol	0.0100 ± 0.0057	ND	ND	0.0156 ± 0.0008	0.0405 ± 0.0036	0.0430 ± 0.0297
	1-hexanol	0.0123 ± 0.0007	0.0936 ± 0.0491	0.1136 ± 0.0064	0.1288 ± 0.0194	0.1561 ± 0.0366	0.1668 ± 0.0240
	(Z)-3-hexenol	0.0536 ± 0.0068	0.0548 ± 0.0161	0.0390 ± 0.0132	0.0469 ± 0.0009	0.1628 ± 0.0485	0.2631 ± 0.1761
	(E)-2-hexen-1-ol	ND	0.0970 ± 0.0419	0.0697 ± 0.0025	0.1537 ± 0.0214	0.1671 ± 0.0217	0.1725 ± 0.0722
	2-ethyl-hexanol	0.0009 ± 0.0009	0.0068 ± 0.0005	0.0018 ± 0.0001	0.0138 ± 0.0001	0.0462 ± 0.0151	0.0850 ± 0.0664
	1-octanol	ND	0.0011 ± 0.0011	0.0025 ± 0.0025	0.0096 ± 0.0003	0.0245 ± 0.0065	0.0496 ± 0.0443
	(E)-2-octen-1-olo	ND	ND	ND	ND	ND	ND
Terpenes	isomenthol	ND	ND	ND	0.0034 ± 0.0023	0.0088 ± 0.0001	0.0100 ± 0.0072
•	D-limonene	0.0087 ± 0.0038	0.0073 ± 0.0030	0.0079 ± 0.0026	0.0291 ± 0.0066	0.0151 ± 0.0016	0.0637 ± 0.0433
	β-cyclocitral	ND	ND	ND	ND	ND	ND
	a-terpineol	ND	ND	ND	ND	ND	ND
	β-citronellol	ND	0.0038 ± 0.0022	0.0134 ± 0.0029	0.0025 ± 0.0004	0.0000 ± 0.0000	0.2493 ± 0.2460
	(E)-geranylacetone	0.0735 ± 0.0076	0.0238 ± 0.0037	0.0262 ± 0.0055	0.0329 ± 0.0085	0.1053 ± 0.0210	0.0867 ± 0.0700
Benzene derivatives	acetophenone	0.0566 ± 0.0058	0.0434 ± 0.0010	0.0573 ± 0.0099	0.0592 ± 0.0022	0.1775 ± 0.0190	0.1826 ± 0.1302
	benzyl alcohol	ND	ND	ND	ND	ND	0.0236 ± 0.0149
	benzothiazole	0.0007 ± 6.89E-05	0.0016 ± 0.00064	0.0020 ± 0.00088	0.0065 ± 0.0005	0.0125 ± 0.0018	0.0169 ± 0.0094
	phenol	0.0104 ± 0.0019	0.0095 ± 0.0008	0.0093 ± 0.0019	0.0141 ± 0.0036	0.0127 ± 0.0127	0.0424 ± 0.0275
	2-phenoxy ethanol (rose ether)	0.0564 ± 0.0098	0.0480 ± 0.0018	0.0298 ± 0.0050	0.0467 ± 0.0015	0.0946 ± 0.0128	0.0876 ± 0.0581
	para-buthyl-cresol	ND	0.0034 ± 0.0005	0.0067 ± 0.0016	0.0080 ± 0.0003	0.0392 ± 0.0063	0.0256 ± 0.0153
	trimetil-tetrahydro-benzofuranon	0.0077 ± 0.0038	0.0071 ± 0.0014	0.0089 ± 0.0007	ND	ND	ND
Esters	ethyl-hexanoate	ND	ND	ND	ND	ND	ND
	methyl-3-OH-butanoate	ND	ND	ND	ND	0.0791 ± 0.0385	0.0207 ± 0.0113
	methyl-nonanoate	ND	ND	ND	ND	0.0843 0.0201	0.0550 0.0432
	penthyl-hexanoate	ND	ND	ND	0.0176 ± 0.0028	0.104 ± 0.0096	
	methyl-hexadecanoate	ND	ND	ND	ND	0.0285 ± 0.0096	0.0085 ± 0.0041
	butyl hexadecanoate	ND	ND	ND	ND	ND	ND
Lactones	y-butyrolactone	ND	ND	ND	ND	ND	ND
	furanone type compound	0.0026 ± 0.0008	0.0077 ± 0.0011	0.0085 ± 0.0015	0.0027 ± 0.0007	0.0056 ± 0.0024	0.0106 ± 0.0027
Sesquiterpenes	sesquiterpene 1	ND	ND	ND	0.0023 ± 0.0006	0.0019 ± 0.0007	0.0017 ± 0.0012
	sesquiterpene 2	ND	ND	ND	0.0149 ± 0.0000	0.0438 ± 0.0012	0.0543 ± 0.0436
Norisoprenoids	TDN	ND	ND	ND	ND	ND	ND
	ß-ionone	0.0209 ± 0.0053	0.0261 ± 0.0025	0.0230 ± 0.0064	0.0075 ± 0.0002	0.0051 ± 0.0005	0.0057 ± 0.0022
	p-ionone	0.0209 ± 0.0053	0.0261 ± 0.0025	0.0230 ± 0.0064	0.0075 ± 0.0002	0.0051 ± 0.0005	0.0057 ± 0.002



596 Table 3

	SSC (°Brix)	-3 dpv 12.6	18 dpv 19.9	24 dpv 20.6	38 dpv 23.6	51 dpv 24.3
Aliphatic aldehydes	hexanal	0.097 ± 0.044	0.432 ± 0.064	0.422 ± 0.054	0.776 ± 0.134	0.566 ± 0.136
,, ,	(E)-2-hexenal	0.218 ± 0.075	0.772 ± 0.232	0.963 ± 0.141	0.899 ± 0.116	0.908 ± 0.174
	octanal	ND	0.018 ± 0.007	0.007 ± 0.001	0.005 ± 0.002	0.004 ± 0.000
	(Z)-2-heptenal	0.042 ± 0.002	0.062 ± 0.009	0.039 ± 0.006	0.025 ± 0.009	0.015 ± 0.005
	(E,E)-2,4-heptadienal	ND	0.006 ± 0.003	0.005 ± 0.001	0.004 ± 0.001	0.000 ± 0.000
	decanal	ND	0.033 ± 0.022	0.008 ± 0.001	0.005 ± 0.003	0.006 ± 0.000
	(E)-2-nonenal	ND	0.045 ± 0.023	0.035 ± 0.018	0.059 ± 0.017	0.039 ± 0.002
	(E,Z)-2,6-nonadienal	ND	0.034 ± 0.017	0.036 ± 0.011	0.026 ± 0.010	0.036 ± 0.001
Aromatic aldehydes	furfural	ND	0.038 ± 0.017	0.018 ± 0.014	0.016 ± 0.012	0.018 ± 0.01
· · · · · · · , · · ·	benzaldehyde	ND	0.026 ± 0.009	0.013 ± 0.002	0.014 ± 0.003	0.006 ± 0.000
Aliphatic alcohols	1-butanol	0.011 ± 0.003	0.008 ± 0.004	0.020 ± 0.002	0.007 ± 0.004	0.014 ± 0.002
	1-hexanol	0.073 ± 0.009	0.105 ± 0.015	0.191 ± 0.068	0.163 ± 0.020	0.108 ± 0.012
	(Z)-3-hexenol	0.066 ± 0.006	0.087 ± 0.016	0.048 ± 0.004	0.023 ± 0.007	0.031 ± 0.000
	(E)-2-hexen-1-ol	0.034 ± 0.006	0.089 ± 0.030	0.110 ± 0.035	0.075 ± 0.016	0.059 ± 0.007
	2-ethyl-hexanol	0.001 ± 0.001	0.024 ± 0.006	0.014 ± 0.001	0.015 ± 0.001	0.016 ± 0.00
	1-octanol	ND	0.013 ± 0.003	0.014 ± 0.001	0.013 ± 0.001	0.013 ± 0.00
	(E)-2-octen-1-olo	ND	0.009 ± 0.005	0.011 ± 0.002	0.006 ± 0.001	0.007 ± 0.000
erpenes	isomenthol	ND	0.003 ± 0.003 0.003 ± 0.002	0.004 ± 0.001	0.000 ± 0.001 0.009 ± 0.004	0.007 ± 0.000
cipelles	D-limonene	0.029 ± 0.003	0.003 ± 0.002 0.023 ± 0.013	0.004 ± 0.002 0.023 ± 0.004	0.003 ± 0.004 0.019 ± 0.010	0.020 ± 0.003
	β-cyclocitral	ND	0.023 ± 0.013 0.012 ± 0.002	0.023 ± 0.004 0.008 ± 0.001	0.015 ± 0.010 0.006 ± 0.001	0.020 ± 0.000
	a-terpineol	ND	ND	ND	ND	ND
	β-citronellol	ND	0.005 ± 0.003	0.005 ± 0.000	0.005 ± 0.001	0.004 ± 0.00^{-1}
	(E)-geranylacetone	0.016 ± 0.002	0.063 ± 0.003	0.003 ± 0.000 0.044 ± 0.001	0.023 ± 0.001	0.033 ± 0.002
	acetophenone	0.010 ± 0.002 0.054 ± 0.006	0.089 ± 0.003	0.044 ± 0.001 0.067 ± 0.010	0.023 ± 0.003 0.069 ± 0.008	0.067 ± 0.002
Benzene derivatives	benzyl alcohol	ND	0.003 ± 0.013 ND	ND	0.003 ± 0.008 ND	ND
	benzothiazole	ND	0.007 ± 0.001	0.007 ± 0.001	0.009 ± 0.000	0.003 ± 0.00^{-1}
	phenol	0.010 ± 0.001	0.007 ± 0.001 0.042 ± 0.024	0.007 ± 0.001 0.013 ± 0.001	0.003 ± 0.000 0.014 ± 0.001	0.003 ± 0.007 0.027 ± 0.007
	2-phenoxy ethanol (rose ether)	0.043 ± 0.010	0.042 ± 0.024 0.266 ± 0.152	0.075 ± 0.001	0.062 ± 0.001	0.042 ± 0.007
	para-buthyl-cresol	ND	0.025 ± 0.015	0.008 ± 0.000	0.002 ± 0.000 0.009 ± 0.001	0.042 ± 0.002 0.010 ± 0.002
	trimetil-tetrahydro-benzofuranone	0.011 ± 0.001	0.020 ± 0.013 0.040 ± 0.027	0.000 ± 0.000 0.010 ± 0.001	0.003 ± 0.001 0.008 ± 0.001	0.006 ± 0.000
sters	ethyl-hexanoate	ND	ND	ND	0.000 ± 0.001 ND	ND
201013	methyl-3-OH-butanoate	ND	ND	ND	ND	ND
	methyl-nonanoate	ND	ND	ND	ND	ND
	penthyl-hexanoate	0.000 ± 0.000	0.040 ± 0.020	0.011 ± 0.003	0.025 ± 0.004	0.028 ± 0.00^{-1}
	methyl-hexadecanoate	0.000 ± 0.000 ND	0.040 ± 0.020 ND	0.011 ± 0.003 ND	0.025 ± 0.004 ND	
	butyl-hexadecanoate	ND	ND	ND	ND	0.002 ± 0.000 0.013 ± 0.000
aatanaa		ND	ND			0.013 ± 0.00; ND
actones	γ-butyrolactone			0.001 ± 0.000	0.002	
	furanone type compound	0.003 ± 0.001	0.005 ± 0.002	0.005 ± 0.001	0.004 ± 0.001	$0.008 \pm 0.00^{\circ}$
Sesquiterpenes	sesquiterpene 1	ND	0.002 ± 0.001	0.003 ± 0.000	0.002 ± 0.001	0.004 ± 0.000
le ste en seu e tale	sesquiterpene 2	ND	0.009 ± 0.005	0.014 ± 0.001	0.016 ± 0.001	0.016 ± 0.00
Vorisoprenoids	TDN	ND	ND	0.007 ± 0.000	0.006 ± 0.001	ND
	β-ionone	0.018 ± 0.002	0.032 ± 0.002	0.023 ± 0.006	0.013 ± 0.003	0.006 ± 0.000

Table 4

		Prin 1	Prin 2	Prin 3
	V1	0.28	0.48	0.09
	V2	0.18	0.60	0.12
601	V3	-0.39	0.15	0.09
602	V4	0.30	0.41	-0.25
603	V5	-0.39	0.08	-0.17
604	V6	0.39	-0.29	-0.08
605	V7	-0.23	0.11	-0.35
606	V8	-0.40	0.04	0.06
607	V9	0.02	-0.05	0.85
608	V10	0.34	-0.35	-0.13
609	Eingenvalues	5.25	2.13	1.16
610	Total variance	0.52	0.21	0.11
611				

V1 = total varietal and pre-fermentative volatile concentration of Vitis vinifera cv Nebbiolo; V2-V9 = percentages of esters, terpenes, sesquiterpenes, benzene derivatives, C6 compounds, other aliphatic aldehydes and alcohols, norisoprenoids, lactones, respectively; V10 = ratio between C6 and the sum of esters, terpenes, sesquiterpenes and norisoprenoids.

Table 5

		Total aldehydes		А	iphatic alcohols			Esters			Terpenes	
	Barbaresco	Barolo	Roero	Barbaresco	Barolo	Roero	Barbaresco	Barolo	Roero	Barbaresco	Barolo	Roero
1	0.36 b	1.16 a	1.47 a	0.17 bB	0.23 bAB	0.34 aA	0.00 b	0.00 b	0.04 a	0.21 aA	0.05 bB	0.11 bAB
2	0.60 b	1.16 ab	1.55 a	0.19	0.37	0.41	0.02	0.02	0.01	0.13 a	0.07 b	0.08 ab
3	2.34	1.66	1.83	0.31	0.60	0.30	0.20 aAB	0.30 aA	0.03 bB	0.20	0.13	0.06
4	1.96	2.11	1.60	0.49 a	0.37 ab	0.25 b	0.22 aA	0.23 aA	0.04 bB	0.11 a	0.05 b	0.07 ab
		Sesquiterpenes		Ber	zene derivatives		N	orisoprenoids			Lactones	
	Barbaresco	Barolo	Roero	Barbaresco	Barolo	Roero	Barbaresco	Barolo	Roero	Barbaresco	Barolo	Roero
1	0.00	0.00	0.01	0.33 aA	0.11 bB	0.27 aA	0.05 aA	0.02 bB	0.03 bAE	0.004	0.008	0.005
2	0.00 b	0.02 aA	0.02 aA	0.22	0.13	0.18	0.04 a	0.01 b	0.03 a	0.007 a	0.003 b	0.005 ab
3	0.03 bB	0.05 aA	0.02 cC	0.26 ab	0.34 a	0.17 b	0.04 aA	0.01 bB	0.02 cB	0.015 a	0.006 ab	0.004 b
4	0.05 aA	0.06 aA	0.02 bB	0.30 a	0.38 a	0.16 b	0.01	0.01	0.01	0.008	0.011	0.008
						-						
	Total varietal a	and pre-fermentat	ive volatiles									

	Total Tarlotal a	na pro ronnonaa	10 Tolatiloo						
	Barbaresco	Barolo	Roero						
1	1.13	1.58	2.18						
2	1.21 bB	1.77 abAB	2.28 aA						
3	3.41	3.08	2.43						
4	3.15 a	2.24 b	2.15 b						
			•						
Table 6									