



Technological University Dublin [ARROW@TU Dublin](https://arrow.tudublin.ie/) 

[Articles](https://arrow.tudublin.ie/dubenart) **Dublin Energy Lab** 

2015

# Indoor Characterisation of a Photovoltaic/ Thermal Phase Change Material System

Maria Browne Trinity College Dublin, Ireland

Keith Lawlor Technological University Dublin

Adam Kelly Technological University Dublin

Brian Norton Technological University Dublin, brian.norton@tudublin.ie

Sarah McCormack University of Dublin, Trinity College, sarah.mccormack@tcd.ie

Follow this and additional works at: [https://arrow.tudublin.ie/dubenart](https://arrow.tudublin.ie/dubenart?utm_source=arrow.tudublin.ie%2Fdubenart%2F67&utm_medium=PDF&utm_campaign=PDFCoverPages) 

Part of the [Electrical and Computer Engineering Commons](http://network.bepress.com/hgg/discipline/266?utm_source=arrow.tudublin.ie%2Fdubenart%2F67&utm_medium=PDF&utm_campaign=PDFCoverPages) 

### Recommended Citation

Norton, B. et al. (2015) Indoor Characterisation of a Photovoltaic/ Thermal Phase Change Material System. Energy Procedia 70 ( 2015 ) 163 – 171.. doi:10.1016/j.egypro.2015.02.112

This Article is brought to you for free and open access by the Dublin Energy Lab at ARROW@TU Dublin. It has been accepted for inclusion in Articles by an authorized administrator of ARROW@TU Dublin. For more information, please contact [yvonne.desmond@tudublin.ie, arrow.admin@tudublin.ie,](mailto:yvonne.desmond@tudublin.ie,%20arrow.admin@tudublin.ie,%20brian.widdis@tudublin.ie)

[brian.widdis@tudublin.ie](mailto:yvonne.desmond@tudublin.ie,%20arrow.admin@tudublin.ie,%20brian.widdis@tudublin.ie).<br>@000

SY NC SA









Available online at www.sciencedirect.com



Energy **Procedia** 

Energy Procedia 70 (2015) 163 - 171

## International Conference on Solar Heating and Cooling for Buildings and Industry, SHC 2014

# Indoor characterisation of a photovoltaic/ thermal phase change material system

Maria C. Browne<sup>a,\*</sup>, Keith Lawlor<sup>b</sup>, Adam Kelly<sup>b</sup>, Brian Norton<sup>b</sup> and Sarah J. M<sup>c</sup>Cormack<sup>a</sup>

*a Department of Civil, Structural and Environmental Engineering, University of Dublin, Trinity College, College Green, Dublin 2, Ireland b Dublin Energy Laboratory, Dublin Institute of Technology, Grangegorman, Dublin 7, Ireland*

#### **Abstract**

Photovoltaic/ thermal (PV/T) systems integrate photovoltaic and solar thermal technologies and have the added advantage of producing both electrical and thermal energy simultaneously. This study has been carried out to investigate the performance of a PV/T phase change material (PCM) system under experimental climatic conditions. Electricity is generated by the PV during daylight hours and the heat produced is absorbed and stored in PCM. Water flows through a pipe network within the PCM and absorbs the stored heat in the PCM. In the PV/T-PCM system, water remained at a higher temperature for an extended period of time as well as shifting the time period of available heat when compared to the reference systems.

© 2015 The Authors. Published by Elsevier Ltd. © 2015 The Authors. Published by Elsevier Ltd. This is an open access article under the CC BY-NC-ND license (intp.//creativecommons.org/ncenses/oy-nc-nu/4.0/).<br>Peer-review by the scientific conference committee of SHC 2014 under responsibility of PSE AG (http://creativecommons.org/licenses/by-nc-nd/4.0/).

*Keywords:* Photovoltaic/ thermal systems; phase change material; thermal perfomance

### **1. Introduction**

 $\frac{1}{2}$ 

The annual global energy consumption is increasing by approximately 2% per annum [1]. According to the U.S. Energy Information Administration fossil fuels produce 80% of the energy consumed worldwide [2]. However, fossil fuels are a limited commodity and the advancement of renewable sources of energy is essential in order to maintain a sustainable environment. Renewable energy technologies have been shown to offer a viable solution to addressing the issue of climate change and global warming [3]. For centuries now, solar power has been harnessed using innovative methods to trap the suns energy.

<sup>\*</sup> Corresponding author. Tel.: 00-353-1-8964561; *E-mail address:* brownema@tcd.ie

The most common of these are photovoltaic solar cells and solar systems which are used to generate electrical and thermal energy. This study presents a novel hybrid PV/T – PCM system which consists of a photovoltaic (PV) solar panel attached to a thermal absorbing material where the system utilizes the thermal energy generated by the PV to heat water. It is charged using waste heat energy generated by the PV. At a later time, the PCM discharges the absorbed heat energy to the water. A pipe network positioned inside the PCM container acts as a heat exchanger, transferring heat from the PCM to the water. The experiment is carried out under the simulated conditions of that in a hot climate as presented in Rehman et al [4].

#### *1.1. Background*

In a hybrid PV/T system both electricity and heat are produced; the heat generated by the PV cell is captured and absorbed by water or air. The incorporation of thermal collectors with PV technology has been seen to increase the overall efficiency of a PV/T system as thermal energy is produced as a by-product of the production of electrical energy [5]. For many PV/T systems, the electrical efficiency is lower than the electrical efficiency of a PV system alone however; the overall energy efficiency is still higher than a PV due to the production of hot air or water.

The assessment of a natural air circulation PV system comparing air and water as a heat transfer fluid with and without glazing showed water circulation proved more efficient than air [6]. Zodag at al. [7] investigated the performance of four groups of PV/T systems of nine different designs. Their thermal and electrical efficiencies were compared in addition to the yearly yield of hot water. It was shown that a sheet and tube was the most feasible design, though was not the most efficient when compared to other systems; it was the ease of manufacture that made it the best option. The fluid channel below the transparent PV was found to have the highest efficiency with an annual average efficiency of 43.5%. A model of a PV/T system using water as a transfer fluid found the thermal and electrical efficiencies could reach 63.65% and 8.45%, respectively [8].

Phase change materials (PCM) absorb thermal energy as latent heat at a constant phase change temperature. At initial heating, PCM heats sensibly and when it reaches the melting/solidification temperature, PCM absorbs latent heat, progressively melting. Melted PCM continues to warm further as it melts heating sensibly, as in Figure 1 [9].



Fig. 1. Graphical representation of the variation of stored heat of a PCM with increasing temperature [9]

The duration and temperature range over which the phase change takes place depends on the mass of the PCM and the thermal conductivity of PCM and any enhanced heat transfer elements therein. There has been considerable research on types of PCM, their applications and physical properties [10-12].

PCM has been extensively investigated in the thermal management of PV. A series of small scale indoor experiments found calcium chloride regulated the front surface temperatures for a prolonged period of time compared to other PCMs tested [13]. Aluminium fins were integrated into a PV/PCM system to investigate the melting process of PCM and the effect of differently spaced fins on the melting fraction of the PCM both theoretically and experimentally using PCM RT27 [14]. A finite element numerical heat transfer model investigated the temperature evolutions, velocities and melt fraction of PCM in a PV/PCM system [15]. Capric-palmitic was found to have the highest temperature regulation with conduction dominating heat transfer.

More recent attention has focused on the provision of integrating PCM with PV/T systems to store and make use of waste heat generated by the PV. A one dimensional energy balance model of a PV/T-PCM system has been simulated and with the inclusion of an applicable PCM the PV output can increase by approximately 9% compared to PV only and an average water temperature rise of 20ºC[16]. A model of a BIPV-PCM system installed in an office façade has been validated using field testing. The maximum electrical efficiency has been shown to reach 10% with thermal efficiencies of 12% [17]. An investigation into the concept of a BIPV-PCM system found the design advantageous due to the reduction of materials required during construction, removal of heat in the summer by PCM and the harnessing of solar energy reduces the need for conventional grid power [18].

The aim of the presented work is to give an evaluation of a novel PV/T-PCM system under conditions of high solar radiation to enhance heat storage and improve the overall efficiency of a PV/T-PCM system. Such a system has been previously investigated under conditions of the maritime island climate of Ireland. It was a found PCM can be used in a PV/T system for the absorption of heat generated by the PV. The heat was stored for an extended period of time compared to a system without the PCM. It was found 12% percent of the energy stored in the PCM was transferred to the water [19; 20].

#### **2. Experimental set-up**

The experimental set-up is similar to that presented in [20], however this experiment assesses three systems, PV/T-PCM, PV/T and PV, under higher temperature conditions; as shown in Figure 2(a) and presented in Table 1.

System	<b>PCM</b>	Pipe network	Container	Panel
PV/T-PCM	Yes	Yes	Yes	Yes
PV/T	No	Yes	Yes	Yes
PV (reference)	No	No	No	Yes

Table 1. List of systems fabricated



Fig. 2. (a) Experimental set-up (b) Heat mat

A stainless steel container of dimensions 1m x 0.471m x 0.065m was attached to each PV and PV/T system as in Figure 3(a). The container encloses a copper pipe network which allows for the flow of water through both the PV/T- PCM and PV/T systems as in Figure 3(b). The container of the PV/T-PCM system holds the PCM. The PV acted as a reference.



Fig. 3. (a) Sealed container (b) pipe network in position in container

Wire wound silicon rubber heat mats were used to simulate the temperature on the back of a solar panel from approximately 9am to 6pm simulating similar temperatures to that of a desert climate based on data recorded during the month of July[4]. The PV panels were removed for the duration of the experiment so the heat mats were in direct contact with the stainless steel container as in Figure 2(b).

The initial temperature of the heat mat was approximately 20ºC at 9am, this was increased incrementally to approximately 66ºC at 1pm and again the temperature was reduced in small steps to approximately 25ºC at 5.30pm. The heat mat was installed on the  $9<sup>th</sup>$ ,  $10<sup>th</sup>$  and  $11<sup>th</sup>$  of September. The remaining days were tested under ambient conditions to investigate the energy storage potential and energy dissipation within the test systems.

Temperature variation throughout each system was recorded using calibrated T-type and K-type thermocouples with a maximum deviation of  $\pm 0.5^{\circ}$ C. These thermocouples were placed at the back of the PV, back of the container and inside the container. The temperature of the water in the pipe network and cylinders were also recorded. The data was recorded on Delta-T and Agilent 34972A data acquisition systems.

PCM used was a fatty acid eutectic, capric- palmitic (CP). It has been found to have a melting point of 22.4°C; other thermophysical properties are reported in Table 2. The container required 22 kg of PCM. This was carried out by melting the capric and palmitic acid in required weight mixtures and melting the PCM using a series of hot plates. It was necessary to keep all PCM in liquid form at the time of pouring into the container; this ensured there was no segregation of PCM within the container.

Table 2. Properties of capric: palmitic acid

Property	Capric: palmitic acid	
Melting point $(^{\circ}C)$	22.4	
Heat of fusion $(kJkg^{-1})$	195	
Thermal conductivity $(Wm^{-1}K^{-1})$	0.143	
Specific heat capacity $(kJkg^{-1}K^{-1})$	2.4	

The ambient temperatures simulated for a hot climatic region  $[4]$  on the 9<sup>th</sup>, 10<sup>th</sup> and 11<sup>th</sup> of September are presented in Figure 4. The latter three days,  $12^{th}$ ,  $13^{th}$  and  $14^{th}$  of September are the environmental conditions of a maritime climate in Dublin. Ambient temperature reaches highs of 19 $^{\circ}$ C and solar radiation exceeding 800W/m<sup>2</sup> on the days tested.



Fig. 4. Environmental conditions for the period of the experiment

#### **3. Results and discussion**

#### *3.1. Charging and discharging*

The temperature inside the container of the PV/T-PCM system is lower than that of the PV/T system due to the storage of heat by the PCM. There is a lag in the temperature increase of the inside of the container in the PV/T-PCM system compared to that of the PV/T system as the PCM is initially storing the heat, this is seen in Figure 5 when the temperature of the inside of the container of PV/T-PCM system is slower to increase than the temperature of the PV/T system and water. The lag is evident for the entire experiment suggesting the PCM is initially storing heat prior to increasing in temperature despite the climatic conditions.



Fig. 5. Temperature differential due to the PV/T-PCM and PV/T systems against time



Fig. 6. Daily heat storage potential of PV/T-PCM and PV/T systems with the period of available heat against time

The PCM in the PV/T-PCM system extends the time for which the heat is available as it absorbs latent heat and charges during the day. The PV/T system allows for immediate availability of the heat gained by the system as the peak is at approximately midday when the simulated temperatures are at their highest. The PCM shifts the time for which the heat is used as the heat gain is stored until discharging of the PCM occurs.

#### *3.2. Heat Storage Potential (HSP)*

The heat storage potential (HSP) was calculated as

$$
HSP = \int_{t=t_0}^{t=t_n} T_{inside\ container} dt - \int_{t=t_0}^{t=t_n} T_{water\ inlet} dt
$$

HSP calculates the amount of heat transfer available to the water at the inlet. Under simulated conditions, HSP in the PV/T-PCM system can be seen to be up to 50% larger than the HSP of the PV/T system, as in Figure 6. This is due to the latent heat capacity of the PCM and its ability to charge and discharge under certain temperatures. Although, the values of HSP are not as high under maritime climatic conditions the amount of energy stored by the PV/T-PCM system still approximately 33% larger than the PV/T system.

For the first three days of the experiment when higher ambient temperatures are simulated [4], PCM increases the heat availability period by almost 100%, up to 20 hours; where the PV/T system has a heat availability period of just over 10 hours. During the latter days under ambient conditions, the period of available heat is reduced in both systems; this is due to the reduction in the heat input. However the PVT/-PCM system still allowed for an extended period of available heat of up to 14 hours compared to a system with no PCM of 8 hours, a 75% increase.

#### *3.3. Water temperature*



Fig. 7. Temperature differential of the water inlet and outlet in PV/T-PCM and PV/T

The temperature difference of the water in the PV/T compared to the PV/T system as illustrated in Figure 7 is a result of the charging and discharging of the PCM. The daily temperature variation of the water in the PV/T-PCM system shows the heat generated by the PV is being stored by the PCM during the day (approximately 11:00-18:00 hours) as there is little temperature difference between the incoming and outgoing water. As ambient temperature decreases the PCM discharges heat gained to the water as illustrated when the temperature of the water increases (approximately between 18:00-7:00).

The immediate availability of heat in the PV/T system can be seen as the temperature of the water peaks at approximately midday, allowing for no storage or extended time of availability. Although, the temperature of the water is not reaching the highs of PV/T-PCM system

The improvement in the potential of the PV/T-PCM system under simulated conditions is evident during the first three days of the experiment as the temperature difference between the inlet and outlet is reaching highs of approximately 8ºCcompared to that of maritime conditions reaching highs of approximately 4ºC[20].

#### **4. Conclusion**

The PV/T-PCM system has been shown to increase its heat storage potential compared to a PV/T system under simulated conditions. The time for which the heat was stored was nearly double that of a PV/T system under the same conditions. Similarly, the heat gain by the water in the PV/T-PCM system was approximately 6<sup>o</sup>C degrees higher than that of a PV/T system.

The performance of a PV/T-PCM system can be seen to dramatically improve compared to a maritime climate like that of Ireland. In a higher temperature climate the heat storage potential increased by 100%. The amount of heat available to the systems has been shown to increase compared to that of a maritime climate. The length of time for which heat is available has been shown to increase by up to 100% compared to a system without PCM, thereby extending the potential time of use.

#### **Acknowledgements**

The authors would like to acknowledge the Higher Education Authority through GREP funding, Science Foundation Ireland (07/RFP/ENEF719) through their Research Frontiers Program, the Research Support Unit at Dublin Institute of Technology and COST Action TU0802 Next generation cost effective phase change materials for increased energy efficiency in renewable energy systems in buildings (NeCoEPCM) and COST Action TU1025 Building Integration of Solar Thermal Systems (BISTS).

We would also like to thank Mr. Colm Gillen and Mr. Thomas Kiely of Dublin Institute of Technology, Kevin Street and Ms. Alba Reig, Mr. Ilario Severino, Mr. Parvat Patil, Dr. Kevin Ryan and Mr. Gordon O' Brien of Trinity College Dublin for their advice and assistance for the duration of the project.

#### **References**

- [1] Mason, J. E. (2007). World energy analysis: H2 now or later? *Energy Policy, 35*(2), 1315-1329.
- [2] International Energy Agency. (2013). *International Energy Outlook 2013*. Washington D.C.
- [3] Arent, D. J., Wise, A., & Gelman, R. (2011). The status and prospects of renewable energy for combating global warming. *Energy Economics, 33*(4), 584-593.
- [4] Rehman, S., & El-Amin, I. (2012). Performance evaluation of an off-grid photovoltaic system in Saudi Arabia. *Energy, 46*(1), 451-458.
- [5] Abdul Hamid, S., Yusof Othman, M., Sopian, K., & Zaidi, S. H. (2014). An overview of photovoltaic thermal combination (PV/T combi) technology. *Renewable and Sustainable Energy Reviews, 38*(0), 212-222.
- [6] Tripanagnostopoulos, Y., Nousia, T., Souliotis, M., & Yianoulis, P. (2002). Hybrid photovoltaic/thermal solar systems. *Solar Energy, 72*(3), 217-234.
- [7] Zondag, H. A., de Vries, D. W., van Helden, W. G. J., van Zolingen, R. J. C., & van Steenhoven, A. A. (2003). The yield of different combined PV-thermal collector designs. *Solar Energy, 74*(3), 253-269.
- [8] Wu, S.-Y., Zhang, Q.-L., Xiao, L., & Guo, F.-H. (2011). A heat pipe photovoltaic/thermal (PV/T) hybrid system and its performance evaluation. *Energy and Buildings, 43*(12), 3558-3567.
- [9] Günther, E., Hiebler, S., Mehling, H., & Redlich, R. (2009). Enthalpy of Phase Change Materials as a Function of Temperature: Required Accuracy and Suitable Measurement Methods. *International Journal of Thermophysics, 30*(4), 1257-1269.
- [10] Zalba, B., Marín, J. M., Cabeza, L. F., & Mehling, H. (2003). Review on thermal energy storage with phase change: materials, heat transfer analysis and applications. Applied Thermal Engineering, 23(3), 251-283.
- [11] Sari, A., Sari, H., & Önal, A. (2004). Thermal properties and thermal reliability of eutectic mixtures of some fatty acids as latent heat storage materials. *Energy Conversion and Management, 45*(3), 365-376.
- [12] Sharma, A., Tyagi, V. V., Chen, C. R., & Buddhi, D. (2009). Review on thermal energy storage with phase change materials and applications. *Renewable and Sustainable Energy Reviews, 13*(2), 318-345.
- [13] Hasan, A., McCormack, S. J., Huang, M. J., Norton, B. (2010). Evaluation of phase change materials for thermal regulation enhancement of building integrated photovoltaics. Solar Energy, 84(9), 1601-1612.
- [14] Huang, M. J., Eames, P. C., Norton, B., & Hewitt, N. J. (2011). Natural convection in an internally finned phase change material heat sink for the thermal management of photovoltaics. *Solar Energy Materials and Solar Cells, 95*(7), 1598-1603.
- [15] Sarwar, J., McCormack,S.J., Huang, M.J., Norton, B., . (2014). Numerical investigation and experimental validation of thermal behaviour of organic phase change materials for temperature regulation enhancement of photovoltaics. International Journal of Heat and Mass *Transfer*, 24.
- [16] Malvi, C. S., Dixon-Hardy, D. W., & Crook, R. (2011). Energy balance model of combined photovoltaic solar-thermal system incorporating phase change material. *Solar Energy, 85*(7), 1440-1446.
- [17] Aelenei, L., Pereira, R., Goçalves, H., & Athienitis, A. (2014). Thermal Performance of a Hybrid BIPV-PCM: Modeling, Design and Experimental Investigation. *Energy Procedia, 48*(0), 474-483.
- [18] Yin, H. M., Yang, D. J., Kelly, G., & Garant, J. (2013). Design and performance of a novel building integrated PV/thermal system for energy efficiency of buildings. *Solar Energy, 87*(0), 184-195.
- [19] Browne, M. C., McCormack, S. J., O' Farrell, K., Candanedo, L. M., & Norton, B. (2013) *Preliminary results of a photovoltaic/ phase*  change material thermal system proceedings of the 28th PVSEC Photovoltaic Solar Energy Conference, Paris, France 22nd - 26th September.
- [20] Browne, M., McCormack, S., & Norton, B. (2014) *Experimental Results of a Photovoltaic/Thermal Phase Change Material System* proceedings of the 13th World Renewable Energy Congress, London 3-7th August.