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# Influence of Cultivation Conditions on the Production of a Thermostable Extracellular Lipase from Amycolatopsis Mediterranei DSM 43304

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1	Influence of cultivation conditions on the production of a thermostable extracellular
2	lipase from Amycolatopsis mediterranei DSM 43304
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## 23 Abstract

24 Among several lipase producing actinomycete strains screened, Amycolatopsis 25 mediterranei DSM 43304 was found to produce a thermostable, extracellular lipase. 26 Culture condition and nutrient source modification studies involving carbon sources, 27 nitrogen sources, incubation temperature and medium pH were carried out. Lipase activity 28 of  $1.37 \pm 0.103$  IU/ml of culture medium was obtained in 96 h at 28 °C and pH 7.5 using 29 linseed oil and fructose as carbon sources and a combination of phytone peptone and yeast 30 extract (5:1) as nitrogen sources. In optimal culture conditions the lipase activity was enhanced 12-fold with a 2-fold increase in lipase specific activity. The lipase showed 31 32 maximum activity at 60 °C and pH 8.0. The enzyme was stable between pH 5.0–9.0 and temperatures up to 60 °C. Lipase activity was significantly enhanced by  $Fe^{3+}$  and strongly 33 inhibited by Hg<sup>2+</sup>. Li<sup>+</sup>, Mg<sup>2+</sup> and PMSF significantly reduced lipase activity, whereas other 34 35 metal ions had no significant effect at 0.01 M concentration. A. mediterranei DSM 43304 36 lipase exhibited remarkable stability in the presence of a wide range of organic solvents at 37 25% (v/v) concentration for 24h. These features render this novel lipase attractive for 38 potential biotechnological applications in organic synthesis reactions. 39

40 Keywords: Amycolatopsis mediterranei; Screening; Organic solvent-tolerant;

41 Thermostable; Lipase

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- 44

## 45 **1. Introduction**

Lipases (triacylglycerol acylhydrolases E.C.3.1.1.3) are serine hydrolases of considerable 46 47 physiological significance and industrial potential [46]. Interest in lipases has greatly 48 increased in recent years, mainly because they present a broad range of biotechnological 49 applications due to their unique characteristics of substrate specificity, regio-specificity and 50 chiral-selectivity [41]. Many microorganisms, including bacteria, yeast, and fungi, have 51 been shown to secrete lipases during their growth on hydrophobic substrates [25]. Among 52 lipases of various origins, those from bacteria show the highest degree of versatility, 53 reactivity and stability in the catalysis of reactions in organic solvents [19]. Lipase 54 applications range from the hydrolysis of fats in wastewaters to the synthesis of chiral 55 pharmaceutical intermediates [42]. These applications often require harsh reaction 56 conditions *e.g.* high temperatures may be required to either favour stereoselectivity or 57 solubilise high melting point lipids [17]. Biocatalysis with lipases is often carried out in 58 organic solvents to promote synthetic reactions by reducing water activity. Many lipases 59 lose activity in organic solvents, and therefore, there is an ongoing interest in lipases that 60 are solvent-tolerant [7]. Despite the advantages of biocatalysis in organic solvent based 61 systems, the catalytic activities of enzymes in these systems are typically much lower than 62 in aqueous solutions [43]. Currently, since microbial lipases do not have the desirable 63 combination of thermostability and stability in both hydrophobic and hydrophilic organic 64 solvents, the search for new lipases is required [31, 33]. There are two main strategies for 65 obtaining lipases with improved properties either protein engineering of currently known 66 lipases [40] or the search for novel lipolytic activities in previously unexplored

microorganisms [11]. As each industrial application requires specific properties of lipases,
there is still an active interest in finding novel lipases for specific applications.
To date, a large number of lipases from filamentous fungi, yeasts and unicellular bacteria
have been extensively studied, both from the biochemical and from genetic point of view

71 [8]. However, despite their high biotechnological potential for the production of secondary

72 metabolites and enzymes, the actinomycetes have not been widely studied for lipase

73 production [1, 30, 50].

74 The present paper deals with the screening of lipase producing actinomycete strains and the 75 culture conditions for optimum enzyme production by a selected strain of *Amycolatopsis* 76 mediterranei DSM 43304. Medium composition, initial pH, temperature, and time of 77 incubation were examined for the optimization of lipase production. Lipase characteristics 78 with respect to the optimal temperature and pH for both activity and stability are examined. 79 We also characterize the activity and stability of the lipase in the presence of various 80 organic solvents, metal ions, detergents and inhibitors. To our knowledge, the present work 81 is the first report of an organic solvent-tolerant lipase from the genus Amycolatopsis.

## 82 **2. Materials and methods**

#### 83 2.1 Materials

Analytical reagent grade chemicals were purchased from commercial sources at the highest
purity. Unless mentioned otherwise, all culture media and chemicals used were from Sigma
(Dublin, Ireland). Phytone peptone was obtained from BBL Microbiology Systems
(Cockeysville, MD, USA) and Bacto-peptone was obtained from Difco Laboratories
(Detroit, MI, USA). Natural oils were purchased from the local retail in Dublin.

89 2.2 Microorganisms

90	Actinomycete strains were obtained from Divisional Culture Collection, School of Biology,
91	Newcastle University, UK.
92	2.3 Maintenance of microorganisms
93	Actinomycete strains were grown on GYM agar slants (g/l: glucose 4.0 g; yeast extract 4.0
94	g; malt extract 10 g; CaCO <sub>3</sub> 2.0 g; agar 12.0 g; pH 7.2). The working stock cultures were
95	maintained and stored on GYM slants at 4 °C.
96	2.4 Culture conditions
97	The composition of basal medium used was (g/l): NaNO <sub>3</sub> 0.5 g; KCl 0.5 g; MgSO <sub>4</sub> 7H <sub>2</sub> O
98	0.5 g; $KH_2PO_4$ 2.0 g; yeast extract 1.0 g; and Bacto-peptone 5.0 g. The pH was adjusted to
99	7.2 with 1 M NaOH or 1 M HCl. Then 1.0% (v/v) olive oil was added. Media were
100	sterilized for 15 min at 121 °C at 15 psi. Submerged microbial cultures were incubated in
101	250 ml Erlenmeyer flasks containing 50 ml of basal medium with 5 ml inoculum on a
102	rotary shaker (130 rpm) at 28 °C.
103	2.5 Rhodamine B agar screening
104	The primary screening for the detection of lipolytic activity on solid media was carried out
105	on rhodamine B agar (RBA) as described by Kouker and Jaeger [29] with some
106	modifications. The growth medium containing 0.9% (w/v) peptone water, 0.25 % (w/v) of
107	yeast extract, 2% (w/v) of agar, was adjusted to pH 7.2, autoclaved and cooled to 60 $^\circ$ C.
108	Then, filter sterilized rhodamine B stock solution (1.0 mg/ml) in distilled water was added
109	to a substrate lipoidal emulsion to yield a final concentration of 0.001% (w/v). The
110	substrate lipoidal emulsion consisted of 1.5% (w/v) olive oil with 0.25% (v/v) Tween 80 in
111	distilled water that was sterilized by autoclaving. The resulting mixture of lipoidal emulsion
112	with growth medium (1:10) was vigorously stirred to emulsify for 15 min. The medium

113 was allowed to stand for 10 min at 60 °C to reduce foaming before pouring 20 ml of

114 medium into plastic petri plates. Fresh RBA plates were spot inoculated with 72 h-old

actinomycete cultures in GYM broth (g/l: glucose 4.0 g; yeast extract 4.0 g; malt extract 10

116 g; pH 7.2) and incubated at 28 °C for 6 days. The plates with visible growth were UV

117 irradiated (350 nm). Lipase production was identified as orange fluorescence under UV

118 light.

## 119 **2.6** Screening in submerged fermentation broth

120 For screening in submerged cultivation conditions, 50 ml of basal medium in 250 ml

121 Erlenmeyer flasks was inoculated with 5 ml, 72 h-old actinomycete culture in GYM broth

122 and incubated at 28 °C on a reciprocal shaker (130 rpm). After 96 h, samples were

123 processed for lipase activity assay. One ml of culture was centrifuged at  $10,000 \times g$ , at 4 °C,

124 for 10 min to obtain a cell free supernatant. The clear supernatant was filtered through 0.2

125 μm filter before lipase activity assay.

## 126 **2.7 Spectrophotometric** *p***-NPP** assay

127 Lipase activity was quantitatively assayed in cell free supernatant using *p*-nitrophenyl 128 palmitate (p-NPP) as substrate. This assay was performed as described by Winkler and 129 Stuckman [55] with some modifications. A stock solution of *p*-NPP was freshly prepared in 130 2-propanol at a concentration of 0.3% (w/v). This solution (Solution A) was subjected to 3 131 min sonication (135W, 42 kHz, Branson 5510E-MT). Then, 900 µl of 1:20 dilution of the 132 substrate stock solution A in solution B (0.1% (w/v) gum arabic, 0.4% (v/v) Triton X100 in 133 distilled water) with 50 µl of appropriate buffer were preincubated for 2 min at the assay 134 temperature before adding 50 µl of enzyme sample. This mixture was incubated at the 135 assay temperature for 10 min, and the reaction was terminated by addition of 2 ml of 0.2 M

136 Na <sub>2</sub> CO <sub>3</sub> solution. Released <i>p</i> -nitrophenol ( <i>p</i> -NP) was immediately determined
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- 137 measuring the absorbance at 410 nm in a Unicam UV-VIS spectrophotometer (Model UV2
- 138 2000E, Cambridge, UK). Appropriate blanks were used to subtract the absorbance
- 139 corresponding to the reaction mixture other than that produced by the specific hydrolysis of
- 140 *p*-NPP. The molar extinction coefficient of *p*-NP ( $\varepsilon_{410nm} = 16,900 \text{ M}^{-1} \text{ cm}^{-1}$ ) was estimated
- 141 from the absorbance of standard solutions of *p*-NP. One international unit of lipase activity
- 142 was expressed as the amount of enzyme liberating 1  $\mu$ M of *p*-NP per minute under the
- 143 conditions of the assay.

## 144 **2.8 Biomass concentration analysis**

- 145 After centrifugation at 10,000  $\times$ g, at 10 °C, for 10 min and washing the pellet in 0.9% (w/v)
- 146 NaCl solution, the pellet of 5 ml suspension sample was dried to a constant weight at 80 °C

147 for 48 h and the dry biomass weight was determined gravimetrically.

#### 148 **2.9 Statistical analysis**

- 149 Data were analyzed using analysis of variance. In all these cases the analyses were
- 150 conducted using SPSS (version 15.0) using the procedure of general linear model (Tukey
- 151 test). The level of tested significance was at  $p \le 0.05$ .

#### 152 **3.** Time course of lipase production by *A. mediterranei* DSM 43304

153 Time course of lipase production was studied in the basal medium using shake flask

- 154 cultures. A 10% (v/v) of 72 h-old inoculum grown in GYM broth was added to 50 ml
- medium, in a 250 ml Erlenmeyer flask and incubated at 130 rpm on a rotary shaker, at 28
- <sup>156</sup> °C. Samples were analyzed at 24 h intervals to determine pH, dry biomass and lipase
- 157 activity in the culture supernatant. The effects of the cultivating conditions in shake flask
- 158 experiments are commonly investigated by subjecting the microorganism to different

159 environmental conditions [56]. In the present work the effect of initial pH and incubation

160 temperature on the culture was studied using shake flask cultures at different temperatures

161 (20–45 °C) and initial values of pH (5.0–9.0). The effect of inoculum size on lipase

- 162 production was investigated by varying culture inoculum size from 2% to 12% of total
- 163 volume. All experiments were carried out at least in triplicate.

#### 164 4. Nutritional factors affecting lipase production by *A. mediterranei* DSM 43304

- 165 The general procedure for cultivation was as follows: 10% (v/v) of 72 h-old inoculum
- 166 grown in GYM broth was inoculated into 50 ml of culture medium and incubated at 28 °C
- 167 for 96 h. The culture was harvested by centrifugation at 10,000  $\times g$ , 4 °C for 10 min. The
- 168 cell free supernatant was filtered (0.2 µm filter, Millipore) before spectrophotometric
- 169 determination of lipase activity. Each experiment was carried out in triplicate.

#### 170 **4.1 Effect of inducers on lipase production**

171 To determine the effects of substrate related compounds, the olive oil in the basal medium

- 172 was substituted with natural oils (jojoba, corn, cottonseed, grapeseed, groundnut, linseed,
- 173 rapeseed, soybean and sunflower) at 1.0% (v/v) concentration.

## 174 **4.2 Effect of surfactants on lipase production**

- 175 The following detergents were added to the basal medium as lipase inducers, replacing
- 176 olive oil, at 0.5% (w/v) concentration: Span 40, Span 65, Span 80, Tween 20, Tween 21,
- 177 Tween 40, Tween 80 and Triton X100.

## 178 **4.3 Effect of carbon source additives on lipase production**

- 179 In order to elucidate the effect of carbon source additives on lipase production, olive oil in
- 180 the basal medium was combined with 1.0% (w/v) of the following sugars: arabinose,

181 dextrin, fructose, galactose, glucose, lactose, mannitol, maltose, maltotetraose, mannose,

182 raffinose, rhamnose, sorbitol, sucrose, starch and xylose.

183 **4.4 Effect of organic nitrogen sources on lipase production** 

- 184 To assess the effects of organic nitrogen source in combination with yeast extract on lipase
- 185 production, Bacto-peptone in the basal medium, was substituted with phytone peptone,
- 186 yeast extract, corn steep liquor, beef extract, skim milk, wheat peptone, fish peptone,
- 187 tryptone, casein hydrolysate, casein, and wheat gluten, each at a concentration of 0.5%
- 188 (w/v).

## 189 **4.5 Effect of phytone peptone and yeast extract on lipase production**

- 190 Phytone peptone with yeast extract was the best nitrogen source for A. mediterranei DSM
- 191 43304 lipase production. The influence of the amount of phytone peptone and yeast extract
- 192 in basal medium was tested by varying their concentrations in the basal medium.

### 193 **4.6 Effect of inorganic nitrogen sources on lipase production**

- 194 In order to assess the effects of inorganic nitrogen sources on lipase production, NaNO<sub>3</sub> in
- 195 the basal medium was substituted with:  $NH_4Cl$ ,  $(NH_4)_2SO_4$ ,  $(NH_4)_2HPO_4$ ,  $NH_4H_2PO_4$ ,
- 196  $CH_3COONH_4$ ,  $NH_4NO_3$  and urea, each at 0.05% (w/v).

## 197 **4.7 Effect of metal ions on lipase production**

198 Metal ions individually and in combination were tested for their effects on optimal lipase

- 199 production in basal medium.  $Mg^{2+}$ ,  $Ca^{2+}$  and  $Fe^{3+}$  salts were incorporated into the basal
- 200 medium at 0.05% (w/v) concentration. Also combinations of  $Mg^{2+}$  and  $Ca^{2+}$ ,  $Ca^{2+}$  and  $Fe^{3+}$ ,
- and  $Mg^{2+}$  and  $Fe^{3+}$  were added at individual final concentration of 0.025% (w/v) to assess
- 202 their cumulative effect on lipase production.

#### 203 5. Characterization of *A. mediterranei* DSM 43304 lipase

#### 204 **5.1 Effect of temperature activity and stability**

205 Lipase activity was measured at various temperatures (30–80 °C) under standard assay

206 conditions. Thermal stability of the enzyme was investigated by preincubating the enzyme

- 207 at various temperatures (30–80 °C) for 3 h. The samples were then assayed for residual
- 208 lipase activity under standard assay conditions.

### 209 **5.2 Effect of pH on activity and stability**

210 The effect of pH on lipolytic activity was determined in the following buffers (all at 50

211 mM): HCl–KCl (pH 2.0), sodium citrate (pH 3.0), succinate–NaOH (pH 4.0 and 5.0),

sodium phosphate (pH 6.0 and 7.0), Tris-HCl (pH 8.0 and 9.0) and glycine-NaOH (pH

213 10.0). The optimum pH obtained was used for investigation of thermostability and other

214 parameters. These buffers were used to determine pH stability of the crude lipase

215 preparation. A mixture (1:1) of crude enzyme and buffers (0.1 M) was incubated for 24 h at

- 216 4 °C and the standard enzyme assay described previously was performed. The residual
- 217 activities were calculated by comparison with the activity in 50 mM Tris–HCl buffer, pH
- 218 8.0, without pre-incubation.
- 219 **5.3 Effect of organic solvents on lipase stability**

The effect of various polar and non-polar organic solvents with different log*P* values on crude lipase stability was investigated. One ml of organic solvent was added to 3.0 ml of cell free supernatant and incubated at 30 °C, while shaking at 200 rpm for 24 h to ensure the continuous mixing of enzyme and the solvents. The enzyme stability was expressed as the remaining activity relative to the control without solvent.

## 225 **5.4 Effect of metal ions and effector molecules on lipase stability**

The effect of metal ions and effector molecules on lipase activity was studied at pH 8.0 by

incubating the enzyme in presence of 1 mM of metal ions  $(Ag^+, Ba^{2+}, Co^{2+}, Li^+, Mn^{2+}, Ni^{2+}, Ni^{$ 

 $228 \qquad Pb^{2+}, Ca^{2+}, Fe^{3+}, Cu^{2+}, Zn^{2+}, Mg^{2+} \text{ and } Hg^{2+}) \text{ and effectors (PMSF, EDTA, SDS, NH_4^+ and Hg^{2+})}$ 

229 urea). Incubation was carried out at 60 °C for 10 min and assayed for lipase activity.

230 Residual lipase activity was calculated as a percentage of that without metal ions/effectors.

## 231 6. Results and discussion

## 232 **6.1 Screening of strains for lipase production**

233 Figure 1 shows rhodamine B agar screening for lipase production by actinomycete strains.

234 Except for Amycolatopsis coloradensis DSM 44225 and Streptomyces aureoverticillatus

NRRL B-3326, all actinomycete strains showed brilliant pink-red/orange fluorescence on

236 UV irradiation of RBA plates. Lipase screening in basal medium using shake flask cultures

showed presence of lipolytic activity in cell free supernatants prepared from actinomycete

culture broths (Table 1). Of 18 actinomycetes strains, 7 showed more than 0.05 IU/ml at 96

h. Among these, A. rubida DSM 44637, S. rochei DSM 40231 and S. griseus subsp.

240 griseus DSM 40236 produced the highest activities at  $0.149 \pm 0.017$  IU/ml,  $0.141 \pm 0.004$ 

241 IU/ml and  $0.116 \pm 0.010$  IU/ml, respectively.

242 Actinomycetes are Gram-positive bacteria with a remarkable genetic repertoire for

243 producing secondary metabolites and enzymes. Despite their high biotechnological

potential, the actinomycetes have not been widely investigated for lipase activity [8, 13,

245 50]. Only a few studies have been reported on their esterase and lipolytic activities [6].

Large et al. described lipase activity associated with the cells of different *Streptomyces* sp.,

which were found to be induced and enhanced by the presence of a lipid substrate in the

fermentation medium [30]. Gandolfi et al. carried out carboxylesterase screening of

*Streptomyces* strains on solid media using tributyrin, triolein and Tween 60 as substrate and
evaluated their cell bound and extracellular hydrolytic activities [18]. Cardenas et al.

251 isolated novel actinomycetes and fungal strains and demonstrated their lipolytic activity by

employing screening techniques on solid and liquid media using agar plates supplemented

with emulsified olive oil and tributyrin [8]. The present investigation identified a number of

actinomycetes producing significant extracellular lipolytic activities.

## 255 6.2 Effect of culture conditions on lipase production by A. mediterranei DSM 43304

256 The preliminary characterization of extracellular lipases from actinomycete strains in terms

257 of pH optimum, temperature optimum and thermostability identified the lipase from *A*.

258 *mediterranei* DSM 43304 as the most thermostable and analysis by zymography indicated

259 presence of a single lipase in extracellular culture broth (data not shown). Therefore, A.

260 *mediterranei* DSM 43304 was selected for further characterization and various culture

261 parameters were studied to improve lipase production.

#### 262 **6.2.1 Time course of lipase production**

263 It has been reported that the lipase synthesis of S. exfoliatus M11 and S. coleicolor A3(2) is 264 growth phase dependent [47]. Therefore, the time course of lipase synthesis in basal 265 medium by A. mediterranei DSM 43304 was monitored by measurement of lipase activity, 266 dry biomass and pH. Figure 2 shows that substantial lipase production commenced at 24 h 267 and reached a maximum at 96 h. Further incubation lead to increase in lipase activity with 268 a slow decrease in the lipase specific activity (data not shown). Loss of lipase specific 269 activity may be due to secretion of other proteins at the late logarithmic phase leading to an 270 apparent decrease in lipase specific activity. Swift et al. reported that once cell densities 271 have reached certain threshold level, generally in the late logarithmic phase, the expression

- of genes encoding exoproteins and secretion system is induced [49]. There was a shift in
- pH from 7.20  $\pm$  0.02 to 8.02  $\pm$  0.057 during the first 24 h of incubation, rising to pH 8.96  $\pm$
- 0.11 at 96 h.

## 275 **6.2.2 Effect of inoculum size on lipase production**

- 276 Low inoculum density may give insufficient biomass causing reduced product formation,
- whereas a higher inoculum may produce too much biomass leading to poor product
- formation [37]. Increased enzyme production was observed with the increase in inoculum
- size and showed maximum enzyme activity ( $0.105 \pm 0.003$  IU/ml) and biomass ( $5.8 \pm 0.6$
- 280 mg/ml) production at 96 h with 10% inoculum (Table 2). Further increases in inoculum size
- resulted in decreased enzyme synthesis, probably due to nutrient limitation.

#### 282 **6.2.3 Effect of incubation temperature on lipase production**

- 283 Temperature may affect lipase production [34]. Submerged fermentation was carried out at
- 284 20–45 °C with 10% inoculum for 96 h to evaluate the effect of incubation temperatures on
- growth and enzyme production. Maximum enzyme and biomass production was observed
- in the mesophilic range at 28–35 °C (Figure 3). The optimal temperature determined for
- 287 lipase production by *A. mediterranei* DSM 43304 (28 °C) is comparable to those of *S.*
- 288 *erythraea*, *S. clavuligerus* [30] and *Psedomonas aeruginosa* PseA [45].

## 289 **6.2.4 Effect of initial medium pH on lipase production**

- 290 The pH of the culture broth was found to be one of the most critical environmental
- 291 parameters affecting the growth and enzyme production by *A. mediterranei* DSM43304.
- 292 The results showed maximum biomass and lipase production at initial medium pH of 7.5
- 293 (Figure 4). Lipase activity dropped significantly at alkaline and acidic pH of 9.0 and 5.0,
- respectively. The optimum pH (7.5) is close to optimum pH of 7.0 for lipase production by

295 *Candida* sp. [52] but is lower than that of other reported lipase-producing organisms.

296 *Bacillus mycoides* showed optimal lipase production at pH 8.0. [53] whereas, maximal

lipase production by *A. terreus* was observed at an initial medium pH of 9.0 [22]. The pH

change observed during growth of the organism may affect the enzyme stability in the

299 medium [23].

#### 300 6.2.5 Effect of inducers on lipase production

301 Natural oils as carbon sources had different effects on lipase production. The results (Table 302 3) indicate that all lipidic sources supported lipase activity, ranging from  $0.014 \pm 0.001$ 303 IU/ml to  $0.128 \pm 0.004$  IU/ml. The highest lipase production ( $0.128 \pm 0.004$  IU/ml) and 304 biomass (16.0  $\pm$  1.4 mg/ml) was found using linseed oil followed by sunflower oil (0.104  $\pm$ 305 0.004 IU/ml). Soybean oil gave the lowest lipase activities. There are very few examples in 306 literature where linseed oil has been used as a lipid source for lipase production. Linseed oil 307 was tested as one of the oils for lipase production by Burkholderia cepacia [44] and in case 308 of P.camembertii Thom PG-3 it was found to be the second best inducer for lipase 309 production [51]. Natural oils such as soybean, corn, sunflower, olive, palm and cotton seed 310 oils, amongst others, are cited as inducers of lipase production, comprising at times, the 311 sole source of carbon in the medium [34, 38, 44, 51]. 312 Surfactants as lipase inducers in the medium did not enhance lipase production when 313 compared to olive oil (Table 3). Similar effects were reported on lipase production in 314 Rhizopus sp. BTNT-2 [3] and Yarrowia lipolytica [15]. Surfactants do not always increase lipolytic enzyme production [35] and their effect varies with microorganism, surfactant 315 316 type and its concentration [15, 52]. However, the addition of surfactant to the culture

317 medium has been shown to increase the secretion of lipolytic enzymes in a number of

318 microorganisms, attributable to alteration of cell permeability leading to increased protein

319 secretion or to surface effects on cell bound enzymes [52]. A wide variety of surfactants

320 like Tweens, Triton, SDS, PEG and gum arabic have been studied by different investigators

321 [15, 35, 52].

#### 322 **6.2.6 Effect of carbon source additives on lipase production**

323 The effect of of carbon source in the basal medium on lipase production by A. mediterranei 324 DSM43304 is shown in Table 4. Lipase production and biomass was significantly enhanced 325 with the addition of sugars to the basal medium in most cases. Fructose gave the highest 326 activity of  $0.467 \pm 0.007$  IU/ml with biomass of  $11.2 \pm 0.9$  mg/ml, whereas lactose 327 produced the highest biomass of  $14.9 \pm 0.4$  mg/ml with lipase activity of  $0.399 \pm 0.021$ 328 IU/ml. Lipase production was increased 3-fold with lactose, sorbitol, maltose and xylose as 329 carbon source additives. However, addition of arabinose, dextrin, and sucrose had no 330 significant effect on lipase production and showed significant decrease in biomass. In 331 contrast, addition of rhamnose showed increased biomass with no significant improvement 332 in lipase activity. A range of different carbon sources (i.e. carbohydrates, alcohols, acids, 333 lipids) have been reported to support both growth of lipolytic enzyme producers and 334 lipase/esterase production [4, 22]. Although lipidic carbon sources seem to be generally 335 essential for obtaining a high enzyme yield, some authors have indicated good results in the 336 absence of fats and oils [34], whereas in some cases, a mixture of compounds has been 337 proposed as optimum carbon source [14, 18]. Similar to the present study, fructose as 338 carbon source was found to significantly increase lipase activity in Issatchenkia orientalis 339 [11]. Lactose was reported to support high to moderate lipase production in various

340 microorganisms [26], which is similar to the present study where lactose led to a significant341 increase in lipase production.

## 342 6.2.7 Effect of nitrogen sources on lipase production

343 Nitrogen sources, including organic nitrogen and inorganic nitrogen sources play an 344 important role in the biosynthesis of lipase [51]. Lipase production and biomass were found 345 to be highest with phytone peptone ( $0.131 \pm 0.021$  IU/ml) and yeast extract ( $0.104 \pm 0.005$ 346 IU/ml) as organic nitrogen source followed by Bacto-peptone ( $0.071 \pm 0.002$  IU/ml) and 347 cornsteep liquor  $(0.061 \pm 0.006 \text{ IU/ml})$  (Table 5). Significant biomass was produced with 348 beef extract ( $8.4 \pm 0.7 \text{ mg/ml}$ ) and tryptone ( $8.1 \pm 0.9 \text{ mg/ml}$ ) as nitrogen source but the 349 lipase activities obtained were low. Wheat gluten gave the lowest lipase  $(0.017 \pm 0.001)$ 350 IU/ml) yield followed by casein and casein hydrolysate. Similar studies were carried out by 351 other investigators to find the best nitrogen source for lipase production [3, 22]. Soybean 352 meal was reported to be the best nitrogen source for lipase production by *P. camembertii* 353 PG-3 [51]. Higher lipase production was reported using yeast extract as nitrogen source for 354 Saccharomyces cerevisiae [48]. Contrary to the present results, casein and corn gluten were 355 the best sources for lipase production by A. terreus [22]. 356 Phytone peptone with yeast extract was identified as the best nitrogen sources for lipase 357 production by A. mediterranei DSM 43304. The effect of different concentrations of 358 phytone peptone and yeast extract on lipase production is shown in Table 5. There was no 359 significant difference in terms of lipase activities and biomass production by different

360 concentrations of phytone peptone and yeast extract in the basal medium. By contrast, a

361 similar study identified a combination of 0.6% tryptone and 0.2% yeast extract as the best

362 nitrogen source for lipase production by *Bacillus* sp. strain 42 [16]. The effect of different

363 inorganic nitrogen sources showed  $(NH_4)_2HPO_4$  as the best inorganic nitrogen source 364 producing maximum lipase activity followed by NaNO<sub>3</sub> (Table 6). In agreement with the 365 present observation, higher specific activity of Burkholderia cepacia lipase was observed in 366 presence of (NH4)<sub>2</sub>HPO<sub>4</sub> as inorganic nitrogen source followed by KNO<sub>3</sub> [44]. Among the 367 inorganic nitrogen sources tested for Psedomonas sp. G6, NaNO<sub>3</sub> supported the best lipase 368 production [27]. Urea showed the lowest lipase and biomass yield followed by  $NH_4Cl$ . 369 Urea was also reported to be inhibitory for lipase synthesis by P. camembertii Thom PG-3 370 [51].

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## **6.2.8 Effect of metal ions on lipase production**

372 Besides physical and nutritional parameters, metal ions may play an important role in lipase 373 production [39]. The effect of different metal ions and metal ion combinations on lipase 374 production is shown in Table 4. Metal ions did not make any significant difference to lipase 375 activity and biomass production. Contrary to the present results, iron was reported to be critical for the production of lipase by *Psedomonas* sp. G6 [27] and Ca<sup>2+</sup> was reported to 376 377 have strong stimulatory effect on extracellular lipase production by P. fluorescens 2D [36]. Similarly,  $Ca^{2+}$  in presence of  $Mg^{2+}$  was reported to produce a significant increase in lipase 378 379 production by Burkholderia cepacia [44].

## 380 **6.2.9 Lipase production in modified medium**

381 The extent of improvement in lipase activity yield by fermentation using the optimal

382 culture parameters was investigated. The modified culture medium for lipase production by

383 A mediterranei DSM 43304 consisted of 1% (v/v) linseed oil, 1% (w/v) fructose, 0.05%

384 (w/v) (NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub>, 0.05% (w/v) KCl, 0.05% (w/v) MgSO<sub>4</sub>7H<sub>2</sub>O, 0.2% (w/v) KH<sub>2</sub>PO<sub>4</sub>,

385 0.5% (w/v) phytone peptone, 0.1% (w/v) yeast extract, initial pH 7.5 and the culture was

incubated at 28 °C for 96 h. The lipase activity produced increased from  $0.108 \pm 0.002$ 

387 IU/ml ( $3.80 \pm 0.14$  IU/mg) in the basal medium to  $1.372 \pm 0.103$  IU/ml ( $7.27 \pm 0.52$ 

388 IU/mg) in the modified medium. Thus, a 12-fold increase in lipase activity yield and 2-fold

increase in lipase specific activity was achieved by employing newly formulated productionmedium.

#### **391 7. Characterization of lipase**

#### **392 7.1 Effect of temperature on activity and stability**

393 The effect of temperature on the activity and stability of lipase is shown in Figure 5. The 394 lipase from A. mediterranei DSM 43304 was found to be quite thermostable with a 395 temperature optimum of 60 °C at pH 8.0. It retained 90% of activity at 60 °C after 3 h 396 incubation and had a half-life of more than 30 min at 70 °C. High thermal stability of 397 lipolytic activity detected in this study may be useful for several industrial applications. 398 Bacterial lipases generally have temperature optima in the range 30–60 °C. However, 399 reports exist of bacterial lipases with optima both in lower and higher ranges [24]. Thermal 400 stability of a lipase is obviously related to its tertiary structure and is influenced by 401 environmental factors such as pH and the presence of metal ions [62]. A few thermostable 402 lipases from actinomycetes have been reported. S. fradiae var. k11 was reported to produce 403 a proteolysis-resistant lipase having a temperature optimum of 55 °C [60] and a 404 thermostable lipase, most active at 50–60 °C, was purified from S. rimosus [1]. Two 405 different lipases from S. coelicolor A3(2) were characterized showing temperature optima 406 at 20-30 °C (SCO1725) and 45-55 °C (SCO7513) [12]. 407 7.2 Effect of pH on activity and stability

408 Figure 6 shows the effect of pH on lipolytic activity and stability. The lipase was found to 409 be active over the pH range 5.0–8.0. Maximal lipolytic activity was at pH 8.0 and showed 410 retention of 88%, 98%, 95% and 89% activity at pH 5.0, 6.0, 7.0 and 9.0, respectively. The 411 pH stability profile showed the highest stability at pH 8.0 and 9.0 with 96% residual 412 activity. The activity was also stable at pH 6.0 and 7.0 showing 94% residual activity after 413 24 h. The stability of the enzyme in acidic and alkaline pH suggests its usefulness in 414 industrial applications. Generally, lipases of bacterial origin have neutral or alkaline pH 415 optima with the exception of P. fluorescence SIK W1 lipase which has an acidic pH 416 optimum of 4.8 [2]. Most of the lipases characterized from actinomycetes so far show 417 alkaline pH optima. S. rimosus lipase was reported to show a pH optimum range of 9.0-418 10.0 [1] and S. fradiae var. k11 lipase was most active at pH 9.8 [60]. Similarly, S. 419 coelicolor A3(2) lipases showed pH optima between 7.5–10.0 [12]. 420 7.3 Effect of organic solvents on lipase stability 421 The effect of various organic solvents on the stability of A. mediterranei DSM 43304 lipase 422 is shown in Table 6. The lipase exhibited considerable stability in the presence of polar 423 solvents ( $\log P < 0.3$ ) as well as non-polar hydrophobic solvents ( $\log P 0.85-6.6$ ) with 424 significant activation observed in most cases. The highest degree of activation was in 425 benzene and toluene. Lipase was significantly activated after 1 h in *p*-xylene and *n*-hexane 426 showing a 61.9% and 56.7% increase in activity, respectively. Further incubation for 24 h 427 led to 28.2% and 45% decrease in activation indicating destabilizing effects of these 428 solvents. The lipase was least stable in pyridine and showed 49.8% residual activity after 24 429 h. Similarly, a destabilizing effect was seen with t-butanol, dodecane and DMSO. It has 430 been shown that many enzymes retain activity in organic solvents [59] and have interesting

431 catalytic properties such as higher thermostability and altered stereoselectivity [54]. Despite 432 many advantages of enzymatic reactions in organic solvents, in most cases the catalytic 433 activity in organic solvents is orders of magnitude lower than in aqueous systems [10] 434 because of diffusional limitations, changes in protein flexibility, or destabilization of the 435 enzyme [28]. A. mediterranei DSM 43304 lipase was activated and stable in DMF, 436 methanol, ethanol, 2-propanol and acetone. Though solvent stability has been reported for a 437 few actinomycete lipases [5, 8, 32], the high stability and indeed, activation in polar 438 solvents like methanol and ethanol has rarely been observed [8]. S. rimosus lipase was 439 reported to show stimulation of activity upon addition of 2.5% (v/v) 1,4-dioxane, THF, 440 acetone and DMF [32]. Similarly, S. coelicolor lipase was found to be stable with different 441 water-miscible solvents and showed 50% increase in activity in acetone after 18 h and 442 similar to the present study, significant loss of activity was observed with DMSO, while in 443 contrast, DMF led to almost complete loss of activity [5]. It is reported that polar solvents 444 strip off the essential water molecules from the active site of enzymes [21]. For this reason, 445 use of polar solvents is avoided and hydrophobic solvents are more often employed in non-446 aqueous enzymology [20]. The polar solvent tolerant lipases therefore appear promising for 447 catalysis in low water media. This property is a novel attribute of A. mediterranei DSM 448 43304 lipase, which has been reported in few cases [5, 32, 61]. The high stability and 449 activation of A. mediterranei DSM 43304 lipase in a wide range of polar and non-polar 450 organic solvents may make it useful for practical applications in synthetic 451 biotransformation reactions.

## 452 **7.4 Effect of metal ions and effectors on lipase stability**

453 The effect of different metal salts and effector molecules on A. mediterranei DSM 43304 lipase is shown in Table 7. The crude lipase was strongly inhibited by  $Hg^{2+}$  losing 80% of 454 activity in 10 min suggesting it is able to alter enzyme conformation as has been reported 455 for other lipases [9].  $Li^+$  and  $Mg^{2+}$  also significantly reduced enzyme activity. The other 456 457 metal ions tested did not produce adverse effect on the activity of the enzyme. Compared to the control,  $Ca^{2+}$  did not cause significant reduction in lipase, even though  $Ca^{2+}$  has been 458 reported to stabilize lipolytic activity [9, 20].  $Cu^{2+}$  has been reported to be a strong inhibitor 459 of lipase activity [19, 26], but no inhibition was observed with  $Cu^{2+}$  at a concentration of 1 460 mM under the conditions tested. Similarly, S. fradiae var. k11 lipase was reported to retain 461 99.4% lipase activity in the presence of 1 mM  $Cu^{2+}$  for 30 min [60].  $Cu^{2+}$  did not 462 significantly reduce the lipolytic activity of SCO1725 and SCO7513 from S. coelicolor 463 464 A3(2) [12]. The lipase activity was unaffected by the metal-chelating agent EDTA 465 indicating that A. mediterranei DSM 43304 lipase is probably not a metalloenzyme. The 466 activities of lipases from S. rimosus and S. fradiae var. k11 have also been reported to be 467 unaffected by EDTA [1, 60]. The effect of a serine inhibitor PMSF, at 1 mM concentration 468 gave a 15% reduction in lipase activity possibly suggesting the presence of a hydrophobic 469 lid hindering access to the catalytic site [12]. The extracellular lipase from S. rimosus was 470 found to be marginally affected by 1 mM PMSF showing retention of 90% of residual activity [1]. Similarly, 89–91% residual activity was retained in presence of 1 mM PSMF 471 by S. coelicolor A3(2) lipases [12]. The lipase showed 88% retention of activity in presence 472 473 of 1 mM urea. SDS was found not to affect lipase activity. In agreement with the present study lipase activity was reported to be stable with 79% residual activity in the presence of 474 475 1 mM SDS by *B. cepacia* lipase [57], whereas *Yarrowia lipolytica* lipase showed

476 significant inhibition of lipase activity with 1 mM SDS [58]. Contrary to the present results,

477 lipase from *B. thermoleovorans* CCR11 was reported to be completely inhibited by 1 mM

478 SDS in 1 h [9].

#### 479 **8.** Conclusions

480 A lipolytic actinomycete strain, Amycolatopsis mediterranei DSM 43304 was identified as 481 producing an extracllular organic solvent-tolerant thermostable lipase. A significant 12-fold 482 increase in lipase activity and 2-fold increase in lipase specific activity was achieved by 483 employing the newly formulated culture medium. The extracellular lipase was highly stable 484 in a broad range of polar and non-polar organic solvents, with maximum enhancement of 485 activity in non-polar solvents. The properties of thermostability, stability and enhancement 486 of activity in organic solvents, and stability with a broad range of effector molecules shows 487 that this lipase may be a promising enzyme for applications in a range of *in vitro* 488 biotransformation reactions.

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Strain	Rhodamine B agar <sup>*</sup>	Lipase activity <sup>**</sup> (IU/ml)
Amycolatopsis coloradensis DSM 44225		$0.046 \pm 0.001^{abc}$
<i>A. amakusaensis</i> NRRL B-3351	+	$0.059 \pm 0.004^{ m bc}$
A. fastidiosa DSM 43855	+	$0.040 \pm 0.002^{ m abc}$
A. mediterranei DSM 43304	+	$0.086 \pm 0.012^{\rm d}$
A. rubida DSM 44637	+	$0.149 \pm 0.017^{ m f}$
A. sulphurea DSM 46092	+	$0.062 \pm 0.008^{\rm cd}$
Nocardia araoensis DSM 44729	+	$0.031 \pm 0.002^{a}$
N. higoensis DSM 44732	+	$0.037 \pm 0.001^{\rm abc}$
N. kruckzakiae DSM 44877	+	$0.033 \pm 0.003^{a}$
Streptomyces amquistii NRRL B-1685	+	$0.046 \pm 0.001^{abc}$
S. griseus subsp. griseus DSM 40236	+	$0.116 \pm 0.010^{\circ}$
S. coelicolor A3(2)	+	$0.033 \pm 0.001^{a}$
S. annulatus NRRL B-2000	+	$0.053 \pm 0.001^{abc}$ $0.051 \pm 0.002^{abc}$
S. arabicus NRRL B-1733	+	$0.031 \pm 0.002^{a}$ $0.030 \pm 0.001^{a}$
S. aurantiogriseus NRRL B-5416	+	$0.038 \pm 0.006^{\mathrm{abc}}$
<i>S. rochei</i> DSM 40231	+	$0.030 \pm 0.000$ $0.141 \pm 0.004^{ m ef}$
<i>S. aureoverticillatus</i> NRRL B-3326	_	$0.034 \pm 0.004^{\mathrm{ab}}$
S. althioticus NRRL B-3981	+	$0.037 \pm 0.004^{ab}$
letter were not significantly different ( $p \le$ *The RBA plates, after 6 days incubation	0.05; by Tukey test) at 28 °C, were exposed to UV (3	
letter were not significantly different ( $p \le$ *The RBA plates, after 6 days incubation	0.05; by Tukey test) at 28 °C, were exposed to UV (3 scence (–).	50 nm) to detect fluorescen
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letter were not significantly different ( $p \le$ *The RBA plates, after 6 days incubation a orange fluorescence (+); no orange fluores	0.05; by Tukey test) at 28 °C, were exposed to UV (3 scence (–).	50 nm) to detect fluorescene
orange fluorescence (+); no orange fluores	0.05; by Tukey test) at 28 °C, were exposed to UV (3 scence (–).	50 nm) to detect fluorescene
letter were not significantly different ( $p \le$ *The RBA plates, after 6 days incubation a orange fluorescence (+); no orange fluores	0.05; by Tukey test) at 28 °C, were exposed to UV (3 scence (–).	50 nm) to detect fluorescene
letter were not significantly different ( $p \le$ *The RBA plates, after 6 days incubation a orange fluorescence (+); no orange fluores	0.05; by Tukey test) at 28 °C, were exposed to UV (3 scence (–).	50 nm) to detect fluoresce

## **Table 1.** Lipase screening of actinomycete strains in basal medium:

712	Inoculum volume (%, v/v)	Lipase (IU/ml)	Biomass (mg/ml)	Final pH
713	2	$0.055 \pm 0.005^{a}$	$0.7\pm0.4^{\mathrm{a}}$	$8.66\pm0.07$
714	4	$0.064 \pm 0.002^{ab}$	$1.0 \pm 0.3^{a}$	$8.82\pm0.06$
715	6	$0.079 \pm 0.010^{abc}$	$1.5 \pm 0.1^{a}$	$8.60 \pm 0.08$
716	8	$0.085 \pm 0.009^{bcd}$	$4.8\pm0.4^{\mathrm{b}}$	$8.76 \pm 0.04$
717	10	$0.105 \pm 0.003^{d}$	$5.8 \pm 0.6^{b}$	$8.77 \pm 0.03$
718	12	$0.102 \pm 0.005^{\rm cd}$	$5.6 \pm 0.7^{\rm b}$	$8.73 \pm 0.04$
719	* The shake flask experiments we			
720	**Data are means $\pm$ standard devia	ations of three determinatio	ns. Column data followed	by the same
721	superscript letter were not signific	cantly different ( $p \le 0.05$ ; b	y Tukey test).	
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**Table 2.** Effect of inoculum level on lipase production:

Inducer	Lipase	Biomass	Final pH
	(IŪ/ml)	(mg/ml)	-
Natural oils (1.0%, v/v)			
Control <sup>*</sup>	$0.081 \pm 0.004^{\rm f}$	$6.0\pm2.8^{\mathrm{ab}}$	$7.90\pm0.03$
Jojoba	$0.071 \pm 0.003^{\text{ef}}$	$4.0\pm1.4^{\mathrm{a}}$	$8.84\pm0.07$
Corn	$0.052 \pm 0.002^{ m cd}$	$4.5\pm3.5^{ab}$	$8.71 \pm 0.04$
Cottonseed	$0.060 \pm 0.007^{ m de}$	$4.5\pm0.7^{\rm ab}$	$8.70\pm0.28$
Grapeseed	$0.046 \pm 0.001^{ m bc}$	$2.5\pm0.7^{\mathrm{a}}$	$8.67\pm0.03$
Groundnut	$0.037 \pm 0.001^{\rm b}$	$4.5\pm0.7^{\rm ab}$	$8.64\pm0.06$
Linseed	$0.128 \pm 0.004^{ m h}$	$16.0 \pm 1.4^{b}$	$7.01 \pm 0.01$
Rapeseed	$0.013 \pm 0.004^{\rm a}$	$2.5\pm0.7^{\mathrm{a}}$	$8.66 \pm 0.18$
Soybean	$0.014 \pm 0.001^{\rm a}$	$1.5\pm0.7^{\mathrm{a}}$	$8.70\pm0.42$
Sunflower	$0.104 \pm 0.004^{ m g}$	$11.5 \pm 2.1^{\rm bc}$	$7.81\pm0.07$
Surfactants (0.5 %, w/v)			
Control <sup>*</sup>	$0.125 \pm 0.006^{e}$	$5.5\pm0.1^d$	$7.26 \pm 0.03$
Span 40	$0.031 \pm 0.003^{ab}$	$1.6\pm0.8^{\mathrm{ab}}$	$8.71 \pm 0.03$
Span 65	$0.034 \pm 0.001^{\rm b}$	$1.6 \pm 0.1^{\mathrm{ab}}$	$8.68 \pm 0.04$
Span 80	$0.063 \pm 0.002^{\rm d}$	$4.7\pm0.8^{cd}$	$8.10 \pm 0.21$
Tween 20	$0.056 \pm 0.003^{ m cd}$	$1.5\pm0.7^{\mathrm{ab}}$	$8.75\pm0.03$
Tween 21	$0.040 \pm 0.004^{\rm bc}$	$0.4\pm0.1^{\mathrm{a}}$	$6.99\pm0.01$
Tween 40	$0.053 \pm 0.003^{ m cd}$	$1.2\pm0.1^{\mathrm{ab}}$	$8.74\pm0.03$
Tween 80	$0.067 \pm 0.007^{ m d}$	$3.2 \pm 0.1^{\rm bc}$	$8.54\pm0.03$
Triton X-100	$0.016 \pm 0.002^{\mathrm{a}}$	$0.2\pm0.1^{\mathrm{a}}$	$7.26 \pm 0.06$
The shake flask experiments	were performed for 96 h at 2	28 °C, 130 rpm. Data are	means $\pm$ standard

## **Table 3**. Effect of inducers on lipase production:

765 different ( $p \le 0.05$ ; by Tukey test).

766 Control refers to the basal medium with 1% (v/v) olive oil as lipase inducer.

//-

Sugar additive/metal ion	Lipase activity (IU/ml)	Biomass (mg/ml)	Final pH
Sugar additive (1.0%, w/v)			
Contol <sup>*</sup>	$0.132 \pm 0.007^{ab}$	$7.3\pm0.5^{abc}$	$8.66 \pm 0.01$
Arabinose	$0.158\pm0.013^{abc}$	$4.6\pm0.3^{a}$	$8.90\pm0.08$
Dextrin	$0.194 \pm 0.042^{bcd}$	$4.7\pm0.2^{\mathrm{a}}$	$8.57\pm0.07$
Fructose	$0.467 \pm 0.007^{ m j}$	$11.2\pm0.9^{\text{gh}}$	$8.65\pm0.04$
Galactose	$0.349\pm0.007^{fghi}$	$12.7\pm0.2^{\mathrm{hi}}$	$8.81 \pm 0.03$
Glucose	$0.317\pm0.019^{efgh}$	$10.7\pm0.1^{defgh}$	$8.69 \pm 0.03$
Lactose	$0.399\pm0.021^{hij}$	$14.9\pm0.4^{\mathrm{i}}$	$8.72\pm0.03$
Mannitol	$0.343\pm0.035^{fghi}$	$10.9\pm0.6^{\text{fgh}}$	$8.64 \pm 0.03$
Maltose	$0.424 \pm 0.071^{ij}$	$8.2 \pm 0.4^{bcdef}$	$8.79\pm0.01$
Maltotetraose	$0.237 \pm 0.007^{cde}$	$8.7 \pm 0.6^{cdef}$	$8.49\pm0.03$
Mannose	$0.266\pm0.003^{defg}$	$8.2 \pm 0.2^{bcde}$	$8.64 \pm 0.03$
Raffinose	$0.350\pm0.007^{ghi}$	$8.1\pm0.5^{bcd}$	$8.71 \pm 0.03$
Rhamnose	$0.211 \pm 0.014^{bcd}$	$10.6\pm0.2^{\text{efgh}}$	$8.73\pm0.06$
Sorbitol	$0.410\pm0.006^{hij}$	$9.7 \pm 1.3^{\rm cdefg}$	$8.66 \pm 0.04$
Sucrose	$0.253 \pm 0.014^{cdef}$	$10.2 \pm 1.8^{\text{defgh}}$	$8.88 \pm 0.03$
Starch	$0.078 \pm 0.007^{\rm a}$	$5.7\pm0.1^{ab}$	$8.76 \pm 0.03$
Xylose	$0.417 \pm 0.008^{ij}$	$9.8\pm0.5^{cdefg}$	$8.78\pm0.04$
Metal ions (%, w/v)			
Control <sup>**</sup>	$0.096 \pm 0.009^{a}$	$6.1 \pm 0.4^{a}$	$8.70 \pm 0.10$
$Fe^{3+}$ (0.05%)	$0.109 \pm 0.005^{a}$	$6.9\pm0.5^{\mathrm{a}}$	$8.62 \pm 0.07$
$Ca^{2+}(0.05\%)$	$0.102 \pm 0.001^{a}$	$5.8\pm0.4^{\rm a}$	$8.75\pm0.06$
$Mg^{2+}$ (0.025%) + Fe <sup>3+</sup> (0.025%)	$0.097 \pm 0.003^{a}$	$4.8\pm0.6^{\rm a}$	$8.60 \pm 0.07$
$Fe^{3+}$ (0.025%) + $Ca^{2+}$ (0.025%)	$0.104 \pm 0.001^{a}$	$6.1 \pm 1.8^{a}$	$8.82 \pm 0.04$
$Mg^{2+}(0.025\%) + Ca^{2+}(0.025\%)$	$0.092 \pm 0.004^{a}$	$5.8\pm0.3^{\rm a}$	$8.76 \pm 0.03$

**Table 4**. Effect of sugar additives and metal ions on lipase production:

804 The shake flask experiments were performed for 96 h at 28 °C, 130 rpm. Data are means ± standard

805 deviations of three determinations. Column data followed by the same superscript letter were not significantly

806 different ( $p \le 0.05$ ; by Tukey test).

807 \* Control refers to the basal medium in the absence of any sugar additive.

- 808 \*\* Control with 0.05% (w/v) of  $Mg^{2+}$  in the basal medium.

816	Table 5. Effect of nitrogen sources on lipase production:	on lipase production:		
817	Nitrogen source	Lipase activity	Biomass	Final pH
818		(IU/ml)	(mg/ml)	
819	<b>Organic nitrogen source</b> (0.5%, w/v)			
820	Control*	$0.071 \pm 0.002^{ m d}$	$6.6\pm1.9^{ m cd}$	$8.84\pm0.03$
821	Phytone peptone	$0.131 \pm 0.021^{\circ}$	$8.8\pm0.4^{ m d}$	$8.72 \pm 0.01$
822	Yeast extract	$0.104\pm0.005^{\rm e}$	$9.1\pm0.1^{ m d}$	$8.69\pm0.03$
823	Corn steep liquor	$0.061\pm0.006^{\rm cd}$	$7.6\pm0.6^{ m d}$	$8.77\pm0.03$
824	Beefextract	$0.046\pm0.004^{ m bcd}$	$8.4\pm0.7^{ m d}$	$8.90\pm0.17$
825	Skim milk	$0.042\pm0.001^{\rm abc}$	$4.7\pm0.1^{ m bc}$	$8.75\pm0.17$
826	Wheat peptone	$0.030 \pm 0.002^{ m ab}$	$6.6\pm0.4^{ m cd}$	$8.94\pm0.03$
827	Fish peptone	$0.026 \pm 0.006^{ m ab}$	$1.0\pm0.1^{ m a}$	$8.63\pm0.03$
828	Tryptone	$0.024 \pm 0.002^{ m ab}$	$8.1\pm0.9^{ m d}$	$8.64\pm0.03$
829	Casein hydrolysate	$0.023 \pm 0.002^{ m ab}$	$3.6\pm0.3^{ m ab}$	$8.59\pm0.01$
830	Casein	$0.022 \pm 0.001^{\mathrm{ab}}$	$1.0\pm0.1^{ m a}$	$8.82\pm0.03$
831	Wheat gluten	$0.017 \pm 0.001^{ m a}$	$2.0\pm0.4^{ m ab}$	$8.78\pm0.04$
832	Different % combination of PP and YE			
833	$0.5\% \text{ PP} + 0.1\% \text{ YE}^{**}$	$0.096 \pm 0.005^{\mathrm{a}}$	$6.7\pm0.4^{a}$	$8.67\pm0.03$
834	0.4%  PP + 0.2%  YE	$0.102 \pm 0.001^{a}$	$7.4\pm0.3^{ m a}$	$8.77 \pm 0.04$
835	0.3%  PP + 0.3%  YE	$0.100 \pm 0.001^{a}$	$6.6\pm0.6^{a}$	$8.87\pm0.07$
836	0.2%  PP + 0.4%  YE	$0.104 \pm 0.002^{\mathrm{a}}$	$8.9\pm0.5^{\mathrm{a}}$	$8.74\pm0.01$
837	0.1%  PP + 0.5%  YE	$0.096 \pm 0.002^{a}$	$6.8\pm0.8^{\mathrm{a}}$	$8.78\pm0.10$
838	0.6% PP	$0.092\pm0.007^{\rm a}$	$6.9\pm0.7^{\mathrm{a}}$	$8.73\pm0.01$
839	0.6%  YE	$0.097 \pm 0.001^{ m a}$	$6.8\pm0.6^{a}$	$8.81\pm0.03$
840	Inorganic nitrogen source (0.05%, w/v)			
841	Control ***	$0.098\pm0.001^{ m de}$	$5.1\pm0.1^{ m b}$	$8.96\pm0.03$
842	NH4CI	$0.071 \pm 0.001^{\mathrm{ab}}$	$2.5\pm0.7^{ m ab}$	$8.82\pm0.04$
843	$(NH_4)_2SO_4$	$0.078\pm0.002^{\mathrm{abc}}$	$3.0\pm1.4^{ m ab}$	$9.04\pm0.06$
844	$(NH_4)_2HPO_4$	$0.104\pm0.003^{\circ}$	$5.7\pm0.5^{ m b}$	$8.78\pm0.03$
845	$\rm NH_4H_2PO_4$	$0.090\pm0.002^{\mathrm{cde}}$	$4.0\pm1.4^{ m ab}$	$9.00\pm0.03$
846	CH <sub>3</sub> COONH <sub>4</sub>	$0.086\pm0.001^{\rm bcd}$	$3.6\pm0.8^{ m ab}$	$9.14\pm0.06$
847	NH4NO <sub>3</sub>	$0.090\pm0.005^{ m cde}$	$4.9\pm0.5^{ m ab}$	$9.02\pm0.04$
848	Urea	$0.062 \pm 0.011^{ m a}$	$1.5\pm0.7^{\mathrm{a}}$	$9.13\pm0.18$
849	The shake flask experiments were performed for 96 h at 28 °C, 130 rpm. Data are means ± standard deviations of three	ned for 96 h at 28 °C, 130 rJ	pm. Data are means $\pm$ st	iandard deviations of three

determinations. Column data followed by the same superscript letter were not significantly different ( $p \le 0.05$ ; by Tukey test). 850

 $^*$  Control refers to the basal medium with 0.5% (w/v) Bacto-peptone as organic nitrogen source. 851

\*\* Refers to control containing 0.5% (w/v) of phytone peptone (PP) and 0.1% (w/v) of yeast extract (YE) in the basal medium. 852

 $^{***}$  Refers to control containing 0.05 % (w/v) of NaNO3 as inorganic nitrogen source in basal medium. 853

Organic solvent	logP	Relative activity	(%)
C	C	After 1 h	After 24 h
Control		$100.0 \pm 1.2$	$100.0 \pm 0.6$
DMSO	- 1.3	$109.4 \pm 2.3$	$89.6\pm3.7$
DMF	-1.04	$131.3 \pm 6.3^{a}$	$137.6\pm0.9$
Methanol	-0.76	$129.5\pm2.8^{\rm a}$	$136.5\pm0.6$
2-propanol	-0.28	$129.5\pm4.3^{\rm a}$	$112.6 \pm 2.5^{\circ}$
Ethanol	-0.24	$109.6\pm2.0$	$115.9 \pm 3.4$
Acetone	-0.23	$131.7\pm5.2^{\rm a}$	$134.6 \pm 2.8$
t-Butanol	0.35	$96.3\pm5.8$	$75.4 \pm 3.4$
Pyridine	0.64	$72.4\pm3.5^{\rm a}$	$49.8\pm2.8$
Diethyl ether	0.85	$116.8 \pm 2.6^{a}$	$136.9 \pm 1.8$
Benzene	2.0	$169.7\pm2.0^{\rm a}$	$169.1 \pm 1.2$
Chloroform	2.0	$113.8\pm6.3$	$134.1 \pm 2.2$
Toluene	2.5	$153.3 \pm 3.6^{\rm a}$	$157.2 \pm 2.8$
Carbon tetrachloride	2.64	$137.2\pm3.7^{\rm a}$	$117.6\pm0.9$
Cyclohexene	2.86	$138.6 \pm 2.3^{\rm a}$	$109.3\pm2.2$
<i>p</i> -Xylene	3.1	$161.9\pm2.0^{\rm a}$	$133.7 \pm 2.2^{\circ}$
<i>n</i> -Hexane	3.5	$156.7 \pm 3.2^{\rm a}$	$111.7 \pm 4.3$
Dodecane	6.6	$115.5 \pm 3.3^{\rm a}$	$68.0\pm3.9$
Petroleum ether	$\approx 3.0$	$140.5\pm2.6^{\rm a}$	$123.7 \pm 2.2$

 Table 6. Stability of A. mediterranei DSM 43304 lipase in organic solvents:

Values represent the mean of three replicates.

<sup>a</sup> Significantly different ( $p \le 0.05$ ; by Tukey test) with respect to the control. 

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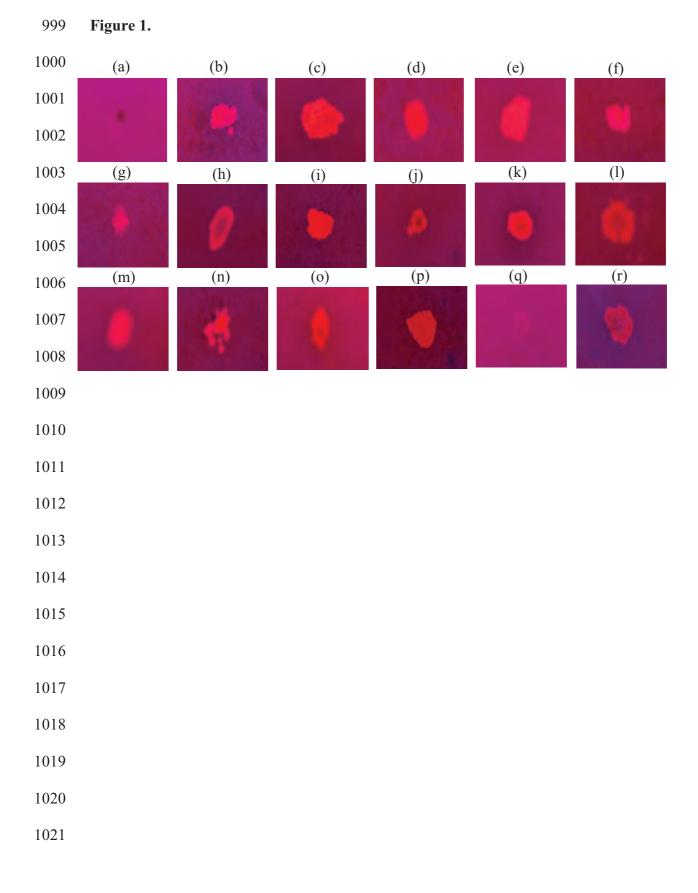
002	Motolion/offector(1 mM)	$\mathbf{P}$ alative activity $(0/)$
903	Metal ion/effector (1 mM)	Relative activity (%)
904	Control	$100.0 \pm 1.0$
905	$Ag^+$ Ba <sup>2+</sup>	88.2 ± 5.5
906	$\operatorname{Ba}^{2^+}$	$99.1 \pm 1.0$
907	$Ca^{2+}_{2+}$	91.1 ± 4.5
908	$\operatorname{Co}^{2^+}_{2^+}$	91.1 ± 0.6
909	$Cu^{2+}$	$100.2 \pm 5.2$
910	$Fe^{3+}$	$115.2 \pm 7.1^{a}$
911	$\mathrm{Hg}_{_{+}}^{2+}$	$20.0 \pm 3.2^{a}$
912	$Li^+_{2+}$	$79.2 \pm 2.6^{a}$
913	Mg <sup>2+</sup> Mn <sup>2+</sup>	$81.3 \pm 3.5^{a}$
914	$Mn^{2+}$	$112.9 \pm 5.2$
915	Ni <sup>2+</sup>	$95.2 \pm 1.9$
916	Pb <sup>2+</sup>	$97.0 \pm 0.6$
917	$Zn^{2+}$	$95.9 \pm 1.6$
918	$\mathrm{NH_4}^+$	$103.8 \pm 1.8$
919	EDTA	$103.6 \pm 1.6$
920	SDS	$100.0 \pm 2.9$
921	Urea	$88.1 \pm 2.9$
922	PMSF	$85.3 \pm 3.2^{a}$
923		ted in the presence of various compounds at 60 °C for 10 min.
	I I I I	T T T T T T T T T T T T T T T T T T T
924	Values represent the mean of three	e replicates.
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925	<sup>a</sup> Significantly different ( $p \le 0.05$ ;	by Tukey test) with respect to the control.
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**Table 7.** Effect of metal ions/effectors on stability of *A. mediterranei* DSM 4334 lipase:

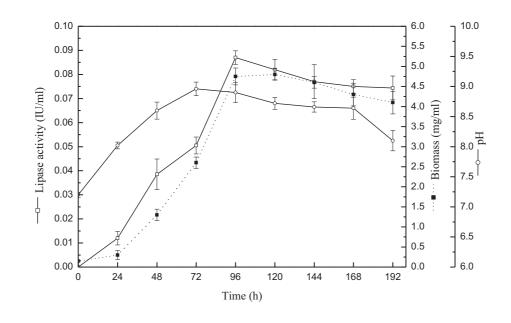
- 953 **Figure captions:**
- 954 **Figure 1**.
- 955 Screening for lipolytic actinomycete strains on rhodamine B agar.
- 956 **a.** *Amycolatopsis coloradensis* DSM 44225; **b.** *A. amakusaensis* NRRL B-3351; **c.** *A.*
- 957 fastidiosa DSM 43855; d. A. mediterranei DSM 43304; e. A. rubida DSM 44637; f. A.
- 958 sulphurea DSM 46092; g. Nocardia araoensis DSM 44729; h. N. higoensis DSM 44732; i.
- 959 *N. kruckzakiae* DSM 44877; j. *Streptomyces amquistii* NRRL b-1685, k. S. griseus subsp.
- 960 griseus DSM 40236; I. S. coelicolor A3(2); m. S. annulatus NRRL B-2000; n. S. arabicus
- 961 NRRL B-1733; o. S. aurantiogriseus NRRL B-5416; p. S. rochei DSM 40231; q. S.
- 962 aureoverticillatus NRRL B-3326; r. S. althioticus NRRL B-3981
- 963 **Figure 2**.
- 964 Time course of lipase production by *A. mediterranei* DSM 43304. Basal medium was
- seeded with 10% (v/v) inoculum and incubated at 28 °C for 192 h at 130 rpm. Samples
- 966 were withdrawn at 24 h intervals to monitor biomass ( $\blacksquare$ ), pH ( $\circ$ ) and lipase activity ( $\square$ ) in
- 967 cell-free supernatants.
- 968 **Figure 3**.
- 969 Effect of incubation temperature on lipase production (□) and biomass (■) during growth of
- 970 A. mediterranei DSM 43304 in shake flask cultures after 96 h of incubation at temperatures
- 971 varying from 20–45 °C.
- 972 **Figure 4**.
- 973 Effect of initial pH on lipase production  $(\Box)$  and biomass  $(\blacksquare)$  in shake flask cultures after 96
- h of cultivation with initial medium pH varying from 5.0 to 9.0.
- 975

976	Figure	5.
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- 977 Effect of temperature on *A. mediterranei* DSM 43304 lipase activity (**■**) and stability (**□**).
- 978 Assay conditions: 30–90 °C, 50 mM Tris–HCl buffer pH 8.0. Thermal stability of the
- enzyme was studied by incubating the enzyme at various temperatures (30, 40, 50, 60, 70,
- 980 80 and 90 °C) for 3 h. Residual activity (%) at each temperature was calculated relative to
- 981 that at 0 h as 100%.
- 982 **Figure 6**.
- 983 Effect of pH on *A. mediterranei* DSM 43304 lipase activity (■) and stability (□).
- 984 For stability studies, residual activities were measured after 24 h incubation at 4 °C in the
- 985 presence of different buffers: HCl–KCl (pH 2.0), sodium citrate (pH 3.0), succinate–NaOH
- 986 (pH 4.0 and 5.0), sodium phosphate (pH 6.0 and 7.0), Tris-HCl (pH 8.0 and 9.0) and
- 987 glycine–NaOH (pH 10.0), all buffers at 50 mM concentration. Assay conditions: 60 °C,
- 988 Tris-HCl buffer, pH 8.0, 50 mM. The activities were compared to the activity determined
- 989 in 50 mM Tris–HCl buffer, pH 8.0 without pre-incubation.
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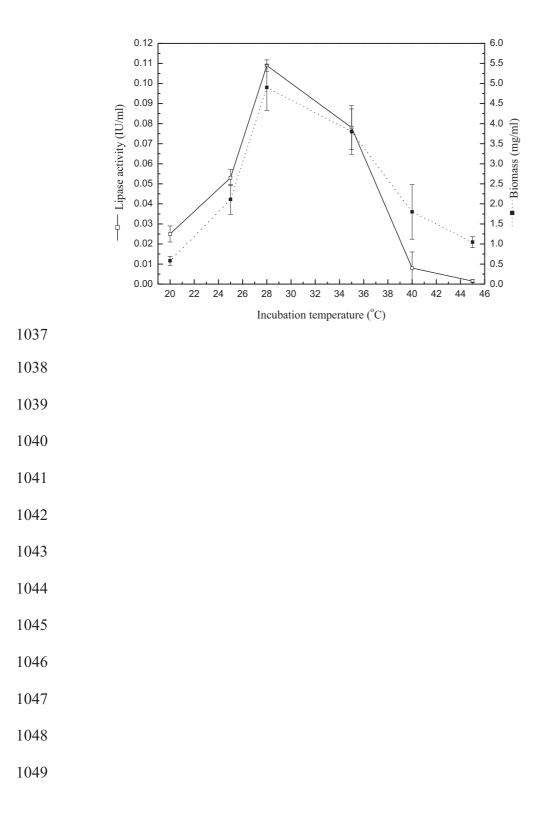


**Figure 2.** 

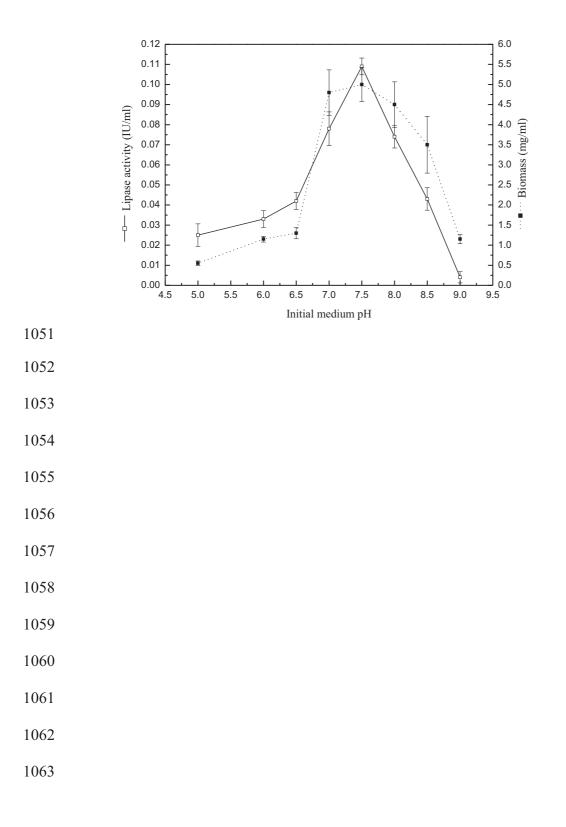




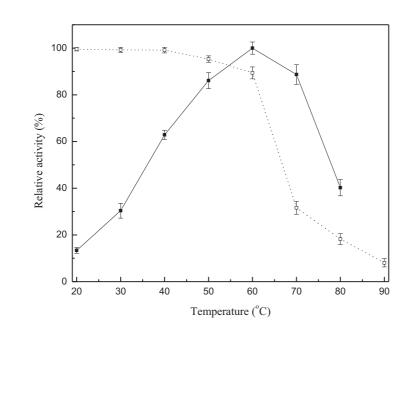
**Figure 3.** 



**Figure 4.** 



## **Figure 5.**



## **Figure 6.**

