Supplementary material for the article:

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Supplementary Information

Synthesis, characterization and crystal structures of two pentagonal-bipyramidal Fe(III) complexes with dihydrazone of 2,6-diacetylpyridine and Girard's T reagent. Anticancer properties of various metal complexes of the same ligand.

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Content:

 Table S1. Crystal data and structure refinement details for 7 and 8.

 Table S2. Selected bond lengths (Å) and angles (°) for 7 and 8.

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	7	8	
formula	$C_{22}H_{35}FeN_{10}O_4S_3$	D ₄ S ₃ C ₄₇ H ₇₂ Fe ₃ N ₂₃ O ₉ S ₉	
$Fw (g mol^{-1})$	655.63	1559.36	
crystal size (mm)	$0.40\times 0.02\times 0.02$	$0.15\times0.10\times0.05$	
crystal color	green green		
crystal system	triclinic	triclinic	
space group	<i>P</i> –1	<i>P</i> –1	
<i>a</i> (Å)	7.5016(9)	12.1021(3)	
<i>b</i> (Å)	11.8462(10)	15.9085(5)	
<i>c</i> (Å)	18.2010(15)	20.0448(6)	
α (°)	82.959(7)	95.611(2)	
β (°)	89.790(8)	95.470(2)	
γ (°)	79.031(9)	110.266(3)	
$V(Å^3)$	1575.6(3)	3568.24(19)	
Ζ	2	2	
calcd density (g cm ⁻³)	1.382	1.451	
<i>F</i> (000)	686	1618	
no. of collected reflns	14602	33057	
no. of independent reflns	7234	16315	
$R_{\rm int}$	0.1371	0.0453	
no. of reflns observed	2972	10061	
no. parameters	381	896	
$R[I > 2\sigma(I)]^a$	0.0696	0.0694	
wR_2 (all data) ^b	0.1476	0.1891	

 Table S1. Crystal data and structure refinement details for 7 and 8.

Goof, S ^c	0.872	1.044
maximum/minimum residual	+0.64/-0.83	+2.50/-0.92

electron density (e $Å^{-3}$)

 ${}^{a} R = \sum ||F_{o}| - |F_{c}|| \sum |F_{o}|. \ ^{b} wR_{2} = \{\sum [w(F_{o}^{2} - F_{c}^{2})^{2}] / \sum [w(F_{o}^{2})^{2}] \}^{1/2}.$ ${}^{c} S = \{\sum [(F_{o}^{2} - F_{c}^{2})^{2}] / (n/p) \}^{1/2} \text{ where } n \text{ is the number of reflections and } p \text{ is the total number of parameters refined.}$

7			
Fe1–N3	2.193(4)	O1–Fe1–N3	71.37(14)
Fe1–N4	2.208(4)	O1–Fe1–N4	140.86(16)
Fe1–N5	2.199(4)	O1–Fe1–N5	149.71(15)
Fe1–N8	2.031(4)	O1–Fe1–N8	90.44(14)
Fe1–N9	2.049(4)	O1–Fe1–N9	91.86(15)
Fe1–O1	2.079(3)	O1–Fe1–O2	78.00(13)
Fe1–O2	2.061(3)	N8–Fe1–N9	176.71(17)
8			
Fe1–N3	2.181(4)	O1–Fe1–N3	71.81(14)
Fe1–N4	2.200(3)	O1–Fe1–N4	141.57(14)
Fe1–N5	2.199(4)	O1–Fe1–N5	148.46(13)
Fe1–N8	2.039(4)	O1–Fe1–N8	90.42(15)
Fe1–N9	2.025(4)	O1–Fe1–N9	91.52(15)
Fe1–O1	2.092(3)	O1–Fe1–O2	76.81(13)
Fe1–O2	2.086(3)	N8–Fe1–N9	176.70(17)
Fe2-N12	2.197(4)	O3-Fe2-N12	71.69(14)
Fe2-N13	2.206(4)	O3-Fe2-N13	141.58(14)
Fe2N14	2.180(4)	O3-Fe2-N14	148.41(14)
Fe2-N17	2.016(4)	O3-Fe2-N17	89.21(16)
Fe2-N18	2.043(4)	O3-Fe2-N18	92.88(14)
Fe2–O3	2.090(3)	O3–Fe2–O4	77.00(13)
Fe2–O4	2.100(3)	N17-Fe2-N18	176.28(17)
Fe3-N19	2.056(4)	N19–Fe3–O1w	90.45(17)
E-2 01	2.076(4)	NO1 E $_{2}$ O1.	174 99(14)

Table S2. Selected bond lengths (Å) and angles (°) for 7 and 8.