

Supplementary material for the article:

Vukojević, V.; Djurdjić, S.; Ognjanović, M.; Antić, B.; Kalcher, K.; Mutić, J.; Stanković, D. M. RuO<sub>2</sub>/Graphene Nanoribbon Composite Supported on Screen Printed Electrode with Enhanced Electrocatalytic Performances toward Ethanol and NADH Biosensing. *Biosensors and Bioelectronics* **2018**, *117*, 392–397. <https://doi.org/10.1016/j.bios.2018.06.038>

## **Supplementary material**

### **RuO<sub>2</sub>/graphene nanoribbon composite supported on screen printed electrode with enhanced electrocatalytic performances toward ethanol and NADH biosensing**

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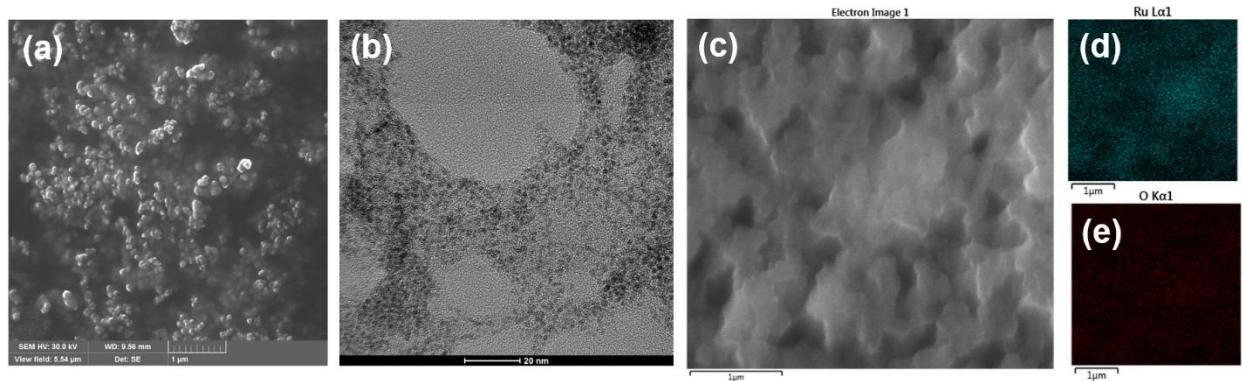
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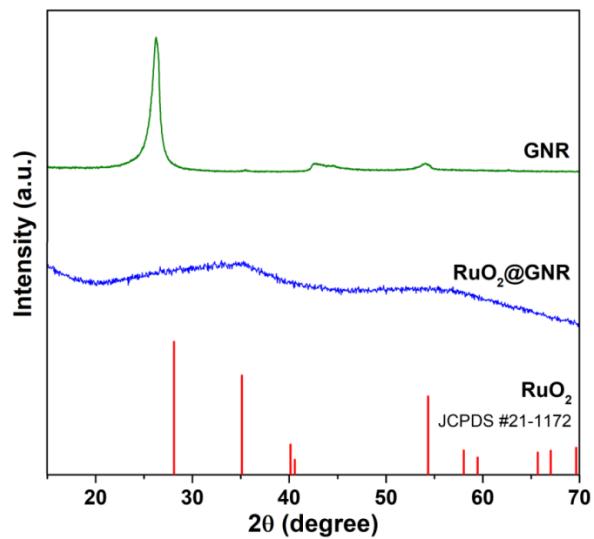
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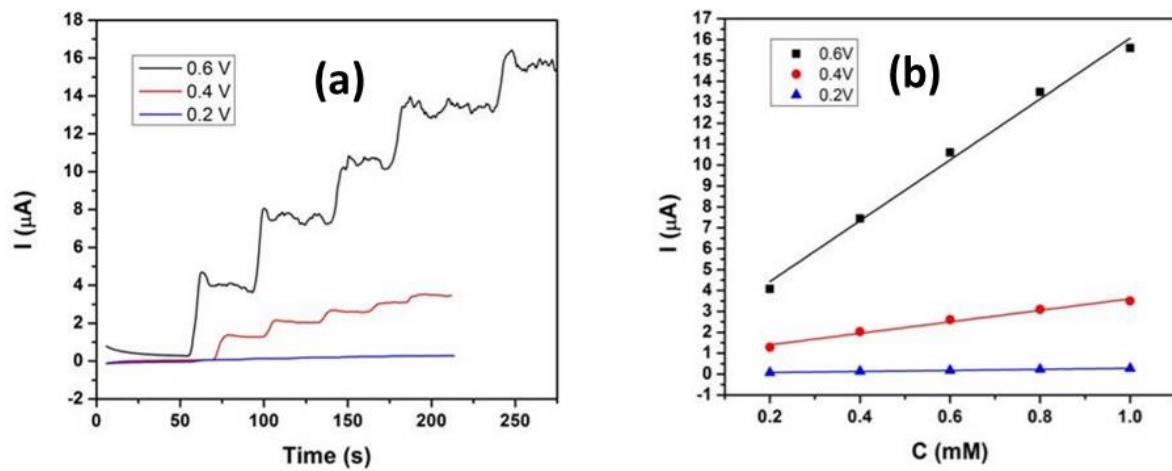
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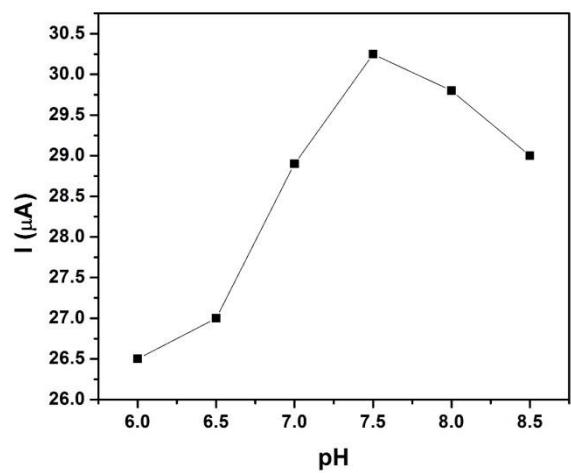
*Figure S1.* (a) FE-SEM micrograph of RuO<sub>2</sub>; (b) TEM micrograph of RuO<sub>2</sub>/GNR composite; (c) FE-SEM micrographs of RuO<sub>2</sub>/GNR nanocomposite; (d) EDS map of ruthenium atom and (e) EDS map of oxygen atom.



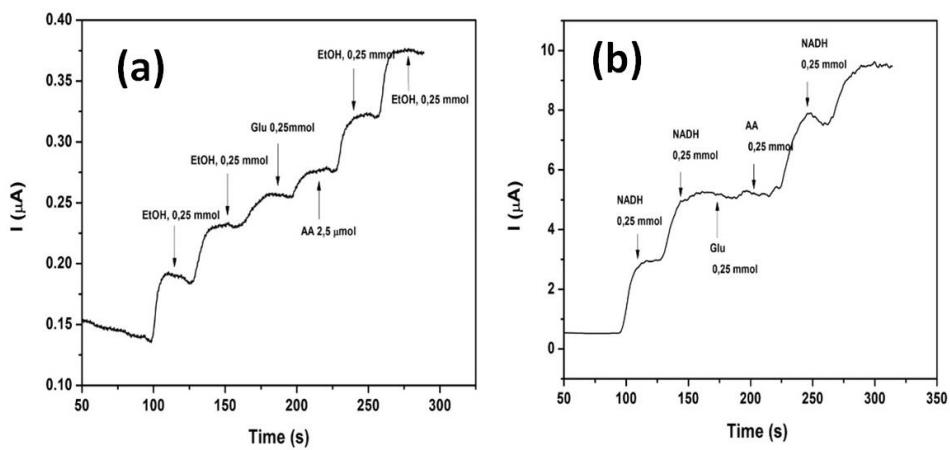
*Figure S2.* XRD patterns of GNR and RuO<sup>2</sup>/GNR composite.



*Figure S3.* (a) Chronoamperometric responses of NADH at different potentials; (b) Calibration curves obtained for successive addition of 0.2 mmol NADH at different operating potential with  $\text{RuO}_2$ -GNR/SPCE working electrode.



*Figure S4.* Effect of different pH of phosphate buffer supporting electrolyte.



*Figure S5.* (a) Effect of possible interfering compounds for the detection of ethanol; (b) Effect of possible interfering compounds for the quantification of NADH. All experiments are done under previously optimized experimental conditions.

*Table S1.* Obtained parameters for NADH sensing at different working potentials.

Potential	Range	LOD (mmol/l)	Slope	Correlation factor
+ 0.2V	0.2 – 1 mmol	0.0964 mmol/l	0.2515	0.9965
+ 0.4V	0.2 – 1 mmol	0.1371 mmol/l	2.75	0.9929
+ 0.6 V	0.2 – 1 mmol	0.0962 mmol/l	14.54	0.9965

*Table S2.* Results obtained for the determination of ethanol content in the three Serbian schnapps.

Sample	Expected (%)	Found (%)	Recovery (%)
Plumb	45	45.4	101
Apricot	40	40.9	102
Pear	38	39.1	103