

D-Region Model Study for Udaipur*

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Received 28 April 1982; revised received 29 October 1982

On the basis of a simplified ion-chemical scheme a D-region model is prepared. Multifrequency absorption results measured at Udaipur match with the electron density profile obtained from the scheme. The results are compared with positive ion and electron concentrations measured at the same solar zenith angle position over Thumba. The scheme used is slightly different from that of the existing ones in both negative and positive ion sides.

In the study of mesosphere the six-ion scheme of Mitra and Rowe¹ has led to the establishment of reactions leading up to $\text{NO}^+ \text{H}_2\text{O}$. Using this scheme, models matching with the multifrequency A_1 absorption for Udaipur (lat., $24^\circ 35' \text{N}$; long., $73^\circ 42' \text{E}$) have been prepared in this study. The results have been compared with positive ions measured over Thumba² and different electron density (N_e) profiles of low latitudes.

The scheme used in the present study is the modified form of Mitra and Rowe scheme¹. This modification has been dealt with by Mitra³.

The steady state continuity equations for all the positive and negative ions have been framed and solved simultaneously with an arbitrary initial value of N_e . The equations are then iterated until plasma neutrality condition is achieved. The various parameters used in the analysis are as follows.

Rate coefficients, recombination coefficients and photodetachment coefficients which have been used in the present study are given in Table 1. The values of photodetachment coefficients γ_2 and γ_3 of X^- and Y^- category ions have been adjusted by trial and error, keeping in view the following. (i) The values of λ (ratio of negative ions to electrons) should match the experimental values or theoretical estimate of others such as Rowe *et al.*⁴ and (ii) the values of γ_2 should be much higher than γ_3 . This is due to the estimates of the photodetachment coefficients made for individual negative ions of X^- and Y^- categories⁵. Values less than 10^{-15} for the rate coefficient K_{11} have been used in this study. It is found that the appropriate value of K_{11} is $10^{-17} \text{ cm}^3 \text{ sec}^{-1}$. Further, the value of the lumpsum rate coefficient K_6 converting $\text{NO}^+ \text{H}_2\text{O}$ to $\text{H}^+(\text{H}_2\text{O})_n$ has been adjusted to match the concentration of $\text{NO}^+ \text{H}_2\text{O}$ with the observed one.

Chakrabarty *et al.*⁶ suggested the value of this rate coefficient to be $3.3 \times 10^{-10} \text{ cm}^3 \text{ sec}^{-1}$, but it is found that with this value the concentration of $\text{NO}^+ \text{H}_2\text{O}$ does not match the measured one. The appropriate value of K_6 has been found to be $5 \times 10^{-11} \text{ cm}^3 \text{ sec}^{-1}$. The values of recombination coefficient of water cluster positive ions with electron K_{25} have been computed as suggested by Mitra³.

Table 1—List of Reaction Rates Depicted in the Scheme

Reaction rates	Reference No.
$K_1 = 2 \times 10^{-31} (300/T)^{4.4} [M]$	18
$K_2 = \frac{1.5 \times 10^{+6}}{T^{5.4}} [M] \exp[-2450/T]$	19
$K_3 = 7 \times 10^{-30} (300/T)^3 [M]$	19
$K_4 = 10^{-9}$	20
$K_5 = 10^{-9}$	20
$K_6 = 5 \times 10^{-11}$	
$K_7 = 2.8 \times 10^{-30} (300/T)$	21
$K_8 = 3 \times 10^{-10}$	21
$K_9 = 1.5 \times 10^{-9}$	21
$K_{10} = 4.4 \times 10^{-10}$	22
$K_{11} < 10^{-15}$	23
$K_{12} = 1.4 \times 10^{-10}$	21
$K_{13} = 5 \times 10^{-11}$	21
$K_{14} = 1.4 \times 10^{-29} (300/T) \exp(-600/T)$	24
$K_{15} = 10^{-31}$	25
$K_{16} = 1.5 \times 10^{-10}$	20
$K_{17} = 2 \times 10^{-10}$	21
$K_{18} = 6 \times 10^{-10}$	20
$K_{19} = 10^{-30}$	3
$K_{20} = 1.1 \times 10^{-11}$	20
$K_{21} = 10^{-7}$	26
$K_{22} = 4 \times 10^{-7} (300/T)$	26
$K_{23} = 2 \times 10^{-7} (300/T)$	26
$K_{24} = 2.3 \times 10^{-6} (205/T)$	21
$K_{25} = 10^{-5} \text{ to } 10^{-6}$	26
$\gamma_1 = 0.33$	3
$\gamma_2 = 0.5, 0.2 \text{ and } 0.05 \text{ at } 60, 65 \text{ and } 70 \text{ km, respectively}$	
$\gamma_3 = 10^{-2}, 5 \times 10^{-3} \text{ and } 10^{-3} \text{ at } 60, 65 \text{ and } 70 \text{ km, respectively}$	

* Paper presented at the National Space Sciences Symposium, Bangalore during 3-6 Feb. 1982.

Table 2—Experimental and Computed Total Absorption for Different N_e Profiles at Various Wave Frequencies

Frequency MHz	Experimental total absorption at $\chi = 30^\circ$ dB	Total absorption (dB) for			
		Present model (Profile No. 3)	Rocket profile ¹² (Profile No. 4)	IRI model ¹⁴ (Profile No. 5)	Statistical ¹³ (Profile No. 6)
2.3	45.31	45.05	57.50	49.85	31.86
2.5	40.51	42.94	53.44	47.16	31.54
2.8	38.59	39.62	43.61	47.10	30.43
3.0	36.22	33.87	39.83	40.39	29.42

this ionization dominates over other ionizations in the height range 60-70 km. If we include this ionization, the computed N_e (profile 2) becomes much higher than that expected in this height range. It looks that Parameswaran¹⁵ has overestimated this ionization. If this ionization is reduced by a factor of 10, then the agreement with other profiles in this height range is quite good (profile 3). This ionization slightly improves the disagreement in the NO^+ ion concentration also, but the concentration of O_2^+ ion is not affected. The disagreement in computed and measured O_2^+ ion may not be very serious, since in the mass-spectrometer measurements 32 positive ions are recorded which may be sulphur ions as suggested by Narcisi¹⁶.

Experimental measurements of multi-frequency ionospheric absorption of radiowaves are in progress at Udaipur since 1972 using A_1 technique. For the present study, those measurements have been taken which were made during moderate activity period and at $\chi = 30^\circ$. These are shown in Table 2 along with the computed values. Total ionospheric absorption of the radio waves of frequencies 2.3, 2.5, 2.8 and 3.0 MHz has been computed using Sen-Wyller theory¹⁷ for different N_e profiles shown in Fig. 5. These frequencies have been found to be reflected from the E-region. Therefore, the model D-region N_e profile (No. 3) has been smoothly joined with the Chapman E-region profile with $f_0E = 3.5$ MHz which is appropriate under the same conditions at this latitude. It may be seen that the measured values of absorption agree very well with the computed values for the model N_e profile (No. 3).

The authors are grateful to Dr A P Mitra, Director, National Physical Laboratory (NPL), New Delhi and

to Dr D K Chakrabarty, Physical Research Laboratory, Ahmedabad, for some useful discussions. They are also thankful to Dr A K Saha, NPL, New Delhi, for providing the computer outputs of ion-pair production rates and electron density profiles from IRI (1978) model. One of the authors (TCB) is also thankful to the University Grants Commission, New Delhi, for financial assistance in the form of a teacher fellowship.

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