

Structure and binding analysis of *Polyporus squamosus* lectin in complex with the Neu5Ac α 2-6Gal β 1-4GlcNAc human-type influenza receptor

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Glycan chains that terminate in sialic acid (Neu5Ac) are frequently the receptors targeted by pathogens for initial adhesion. Carbohydrate-binding proteins (lectins) with specificity for Neu5Ac are particularly useful in the detection and isolation of sialylated glycoconjugates, such as those associated with pathogen adhesion as well as those characteristic of several diseases including cancer. Structural studies of lectins are essential in order to understand the origin of their specificity, which is particularly important when employing such reagents as diagnostic tools. Here, we report a crystallographic and molecular dynamics (MD) analysis of a lectin from *Polyporus squamosus* (PSL) that is specific for glycans terminating with the sequence Neu5Ac α 2-6Gal β . Because of its importance as a histological reagent, the PSL structure was solved (to 1.7 Å) in complex with a trisaccharide, whose sequence (Neu5Ac α 2-6Gal β 1-4GlcNAc) is exploited by influenza A hemagglutinin for viral adhesion to human tissue. The structural data illuminate the origin of the high specificity of PSL for the Neu5Ac α 2-6Gal sequence. Theoretical binding free energies derived from the MD data confirm the key interactions identified crystallographically and provide additional insight into the relative contributions from each amino acid, as well as estimates of the importance of entropic and enthalpic contributions to binding.

Keywords: GLYCAM / PSL / influenza viral adhesion / lectin / molecular dynamics

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Introduction

The term “sialic acids” refers to a family of N- or O-substituted derivatives of neuraminic acid, a 9-carbon acidic sugar, the most common of which is 5-*N*-acetylneuraminic acid (Neu5Ac). They are expressed abundantly in eukaryotes and are frequently located at the termini of the glycan chains of glycoproteins and gangliosides on cell surfaces (Kelm and Schauer 1997). Because of their exposure on cell surfaces, glycans terminating in sialic acids are often involved in protein-mediated cell–cell interactions, which are essential for normal cell development (Kelm and Schauer 1997). Concomitantly, bacterial (Lehmann et al. 2006) and viral (Shinya et al. 2006) pathogens have evolved to express surface proteins [adhesins and hemagglutinins (HAs)] specific for mediating pathogen adhesion to sialylated glycans on host tissue. The ability to characterize the sialylation patterns of glycans is therefore of considerable relevance in determining the receptor specificities of pathogens. In addition to the vast diversity in glycan structures introduced by sequence and linkage, structural heterogeneity among sialylated glycans can arise from variations in the glycosidic linkage positions associated with the Neu5Ac residues, which may be α 2-3 or α 2-6, principally to galactose (Gal) or *N*-acetylgalactosamine (GalNAc) residues, or α 2-8 or α 2-9 to adjacent sialic acid residues (Angata and Varki 2002; Varki NM and Varki A 2007). Sialic acids may themselves be derivatized by, for example, *O*-acetylation or by replacement of the *N*-acetyl moiety with *N*-glycolyl. Thus, sialic acid-binding proteins can display a range of possible binding modes (Kelm and Schauer 1997).

A notable example of an interaction between sialylated glycans and a pathogen is that between the HA protein found on the surface of influenza A virus, which targets the host glycan receptor sequence Neu5Ac α 2-3Gal or Neu5Ac α 2-6Gal, or sometimes both, depending on the influenza strain. The specificity of HA largely determines species specificity, with human infective strains recognizing the Neu5Ac α 2-6Gal epitope, avian the α 2-3 and swine both α 2-3 and α 2-6 (Suzuki et al. 2000). The preference of human influenza HA for the Neu5Ac α 2-6Gal epitope arises from mutations in the avian HA sequence (Gagneux et al. 2003). Remarkably, as few as two point mutations in avian HA may be sufficient to induce this specificity shift, as seen in the case of the 1918 “Spanish Flu” pandemic (Lewis 2006).

Lectin histochemistry has been pivotal in characterizing the distribution of α 2-3- and α 2-6-linked sialic acids in host epithelial tissue samples, by exploiting the binding of *Sambucus nigra* agglutinin (SNA-I; Shibuya et al. 1987) to the α 2-6 linkage and *Maackia amurensis* agglutinin (Wang and Cummings 1988) to the α 2-3 (Baum and Paulson 1990; Shinya et al. 2006). Such data underpin our understanding of influenza transmission models.

Lectin histochemistry has also been employed to characterize the presence and distribution of aberrant host glycosylation (Boland et al. 1991; Sata et al. 1991; Benallal et al. 1995; Vierbuchen et al. 1995; Yamashita et al. 1995; Murayama et al. 1997), which can be a hallmark of several diseases, including tumor metastasis (Wang 2005). In establishing correlations between glycan modifications and disease states, it is particularly crucial to employ lectins with precisely known and appropriate specificities. In human colorectal cancer, for example, abnormal levels of Neu5Ac α 2-6Gal-containing glycans can result from the up-regulation of the α 2-6 sialyltransferase (ST6GalI; Dall'Olio et al. 2000). In this example, a diagnostic or a histochemical reagent, such as the lectin SNA-I is not ideal as it can recognize both the ST6GalI product, as well as the Neu5Ac α 2-6GalNAc sequence, which is a product of sialyltransferases ST6GalNAcI or ST6GalNAcII.

The *Polyporus squamosus* lectin (PSL) has specificity for Neu5Ac α 2-6Gal over the Neu5Ac α 2-3 sequence, but unlike SNA-I, does not bind to the mucin-derived *O*-linked Neu5Ac α 2-6GalNAc sequence (Shibuya et al. 1987; Mo et al. 2000; Toma et al. 2001) and, therefore, has important advantages in histochemical analyses. Due to the potential importance of PSL as a tool in biomedical and cancer research (Lehmann et al. 2006) and to determine the molecular basis for its high specificity, we undertook to solve its crystal structure in complex with a human-type influenza epitope Neu5Ac α 2-6Gal β 1-4GlcNAc (6'-SLN; Stevens et al. 2006) and to conduct a comprehensive computational analysis to quantify the per-residue binding energy contributions that determine the specificity of PSL.

Results and discussion

Molecular structure

The crystallographic asymmetric unit of PSL bound to 6'-SLN contains a dimer. An analysis of protein interfaces using the Protein Interfaces, Surfaces and Assemblies (PISA) service (Krissinel and Henrick 2007) suggested that both dimeric and tetrameric quaternary structures of PSL are stable in solution. In this crystal structure, the root mean-square deviation (RMSD) of the C α positions calculated over 275 equivalent residues in each protomer is 0.48 Å. The dimer is cylindrical in shape, and the hydrophobic interfacial contact between the two protomers takes place via a third helix in the C-terminal domain formed by residues 208–226 (Figure 1, left panel). Ten inter-domain hydrogen bonds are formed, and 12% of the solvent accessible surface of the individual protomers is buried upon dimerization.

Each protomer is made up of an N-terminal ricin-type B domain (residues 1–153) and a C-terminal domain (154–286) forming an α / β -fold. The B domain comprises and contains three evolutionarily conserved (Q/N-x-W)₃ motifs that are found in the α (1–57), β (58–104) and γ (105–153) subdomains, respectively (Hazes 1996). The overall three-dimensional ricin-type β -trefoil fold of PSL and its related lectins is highly conserved (Figure 1, right panel). The C-terminal domain is characterized by a central five-stranded β -sheet that is flanked by three α -helices and topped by a short strand (Supplementary Material, Figure S1). It shows high fold similarity to its closest relative, the Gal α 1-3Gal-binding agglutinin from the mushroom *Marasmius oreades* agglutinin (MOA) (Grahm et al. 2007).

The PSL crystal structure shows well-ordered electron density for the 6'-SLN ligand in the difference density map ($F_o - F_c$) at the carbohydrate-binding site of the β subdomain (Figure 2), whereas no supporting ligand density was seen in the canonical α and γ subdomains.

Alignment of the sequences of PSL and MOA using *Sequoia* (Bruns et al. 1999), combined with the manual binding motif identification based on ligand proximity

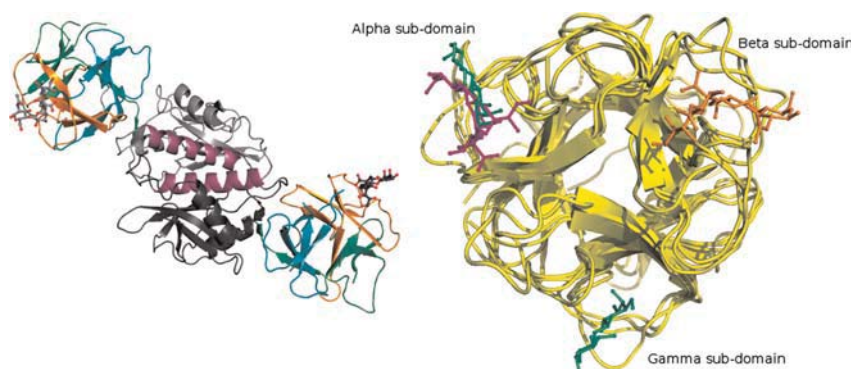


Fig. 1. (Left panel) Quaternary structure of PSL. The two protomers are shown in cartoon mode and the bound ligand (6'-SLN) is shown in ball-and-stick mode. (Right panel) The ricin-type β -trefoil lectins structurally related to PSL and their respective carbohydrate-binding sites. The figure shows that the N-terminal domain of PSL bound to Neu5Ac α 2-6Gal β 1-4GlcNAc and its secondary structural overlay against SRC bound to Neu5Ac α 2-6Gal β 1-4Glc (6'-SL, PDBID: 2DS0) and the two trefoil domains of SNA-II bound to Gal β 1-4GlcNAc (PDBID: 3CA4).

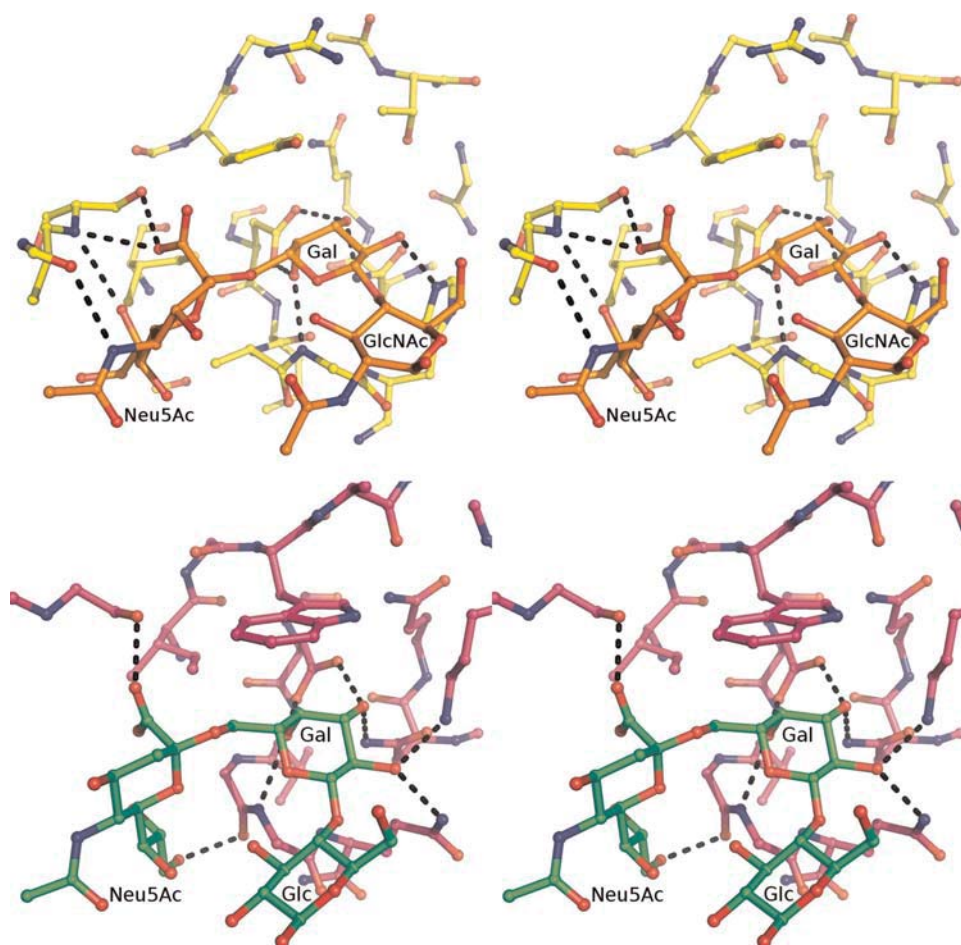


Fig. 4. (Top panel) Stereo view of the hydrogen-bonding network in the 6'-SLN-PSL complex. (Bottom panel) Stereo view of the hydrogen-bonding network in the Neu5Ac α 2-6Gal β 1-4Glc (6'-SL)-SRC complex. For a detailed interaction analysis of the hydrogen-bonding network in the 6'-SLN-PSL complex refer to Table I and Supplementary Table S1.

Details of oligosaccharide binding

The binding mode of the oligosaccharide is defined principally by hydrogen bonds between the terminal two residues (Neu5Ac α 2-6Gal) and the protein, with a total of eight hydrogen bonds shared between the glycosyl residues and the protein (Table I). As expected from earlier studies (Tateno et al. 2004), the majority of the interactions involve residues in the β subdomain; however, it is notable that residues Asn30 and Ser32 from the α subdomain interact with atoms N5 and O1A of the Neu5Ac residue (Figure 4, top panel). The *N*-acetylglucosamine (GlcNAc) residue does not interact with the lectin directly, but through mediating water molecules Wat454 and Wat542. Water molecules Wat542, Wat510, Wat467 and Wat348 hydrogen bond to both the lectin and the oligosaccharide, thus forming an extensive hydrogen bond network (Supplementary Material, Table S1). In addition to the hydrogen bond contacts, the ligand orientation is further stabilized by a hydrophobic stacking interaction between the pyranose ring of the Gal residue and the aromatic side chain of Tyr87. Stacking interactions are particularly prevalent in carbohydrate-protein complexes

(Bush et al. 1999), and in the case of galactosyl, residues are seen in the binding of oligosaccharides to galectins (Ford et al. 2003; Di Lella et al. 2009).

To augment the crystallographic data, a 10 ns molecular dynamics (MD) simulation in explicit water was performed on the N-terminal ricin B-type domain (residues 1–153) of PSL (Δ PSL) in complex with 6'-SLN. By monitoring the fluctuations in the inter-residue distances, data from the MD simulation provide additional insight into the strengths of these interactions. In general, the MD data (Table I) confirmed that the hydrogen bonds observed crystallographically were also populated in solution at room temperature; however, the magnitudes of the deviations in the interatomic distances, as well as the percentage occupancies of the hydrogen bonds, indicate that the four hydrogen bond interactions with the Gal residue are in general more robust than those formed with the Neu5Ac.

The three-bond α 2-6 linkage is relatively flexible due to the presence of the ω -angle, which can potentially populate three distinct rotamers (Wolfe 1972), referred to as the *gauche*-*trans* (*gt*, $\omega_{O5-C5-C6-O6} \approx 60^\circ$), *gauche*-*gauche* (*gg*, $\omega \approx -60^\circ$) and

trans-gauche (tg , $\omega \approx 180^\circ$) rotamers (Kirschner and Woods 2001). For the Neu5Ac α 2-6Gal sequence in the PSL structure, as well as in the complex with SRC and in all reported complexes with influenza HA, such as PDB IDs 3HTQ (avian HA; Lin et al. 2009), 1RVZ (human HA; Gamblin et al. 2004), 1RVT (swine HA; Gamblin et al. 2004), the ω -angle adopts exclusively the gt orientation (Table II). This is also the most populated rotamer present in solution as indicated by NMR (Poppe et al. 1992), and as confirmed by 200 ns MD simulations here (Supplementary Material, Figure S3). The overall conformation of the ligand in PSL is equivalent to that seen in all reported HA crystal structures containing human

flu receptor ligands (Figure 5), although the mode of binding is not equivalent. In all cases, the ψ -angles populate the *trans*-rotamer, whereas the φ -angle adopts only one of the two rotamers typically observed for the Neu5Ac α 2-6Gal linkage in solution (Poppe et al. 1992).

Table I. Key interactions between the 6'-SLN and PSL residues from the X-ray structure and MD simulation

	Interatomic distance ^a	
	X-ray	MD (10 ns)
Direct H-bonds		
Neu5Ac(O1A):Ser32(O γ)	2.56	2.64 \pm 0.12 (100%) ^b
Neu5Ac(O1A):Ser32(N)	3.07	3.22 \pm 0.25 (87%)
Neu5Ac(O8):Ser32(N)	3.33	3.15 \pm 0.21 (94%)
Neu5Ac(N5):Asn30(O)	2.59	3.00 \pm 0.24 (96%)
Gal(O3):Asp72(O δ 2)	2.67	2.60 \pm 0.09 (100%)
Gal(O4):Asp72(O δ 1)	2.64	2.59 \pm 0.14 (100%)
Gal(O2):His76(N2)	2.60	2.86 \pm 0.08 (99%)
Gal(O3):Asn94(N δ 2)	2.94	2.93 \pm 0.16 (100%)
Aromatic stacking interactions ^c		
$r_{\text{Gal}}:\text{Tyr87}$	4.46	4.45 \pm 0.01
$\theta_{\text{Gal}}:\text{Tyr87}$	157.64	158.4 \pm 4.90

^aValues are expressed in Å. The electron density for chain A was the better ordered among the two NCS-related protomers; therefore, representative H-bond distances were calculated using this protomer.

^bFor the MD data, the distances were averaged over 10 ns and H-bond length (± 1 SD) and occupancy assessed for periods during which the distance between the heavy atoms was < 3.5 Å.

^cDistance (r) in Å between the centroid (c) of the galactosyl ring and the tyrosine phenyl ring, and angle (θ) between the normal to the tyrosine ring plane and the vector between the centroid c , in degrees (Ford et al. 2003).

Molecular mechanics Poisson–Boltzmann surface area analysis of binding energies

Based on the stability of the MD simulation of the 6'-SLN– Δ PSL complex, as indicated by the ability of the MD simulation to reproduce the intermolecular interactions associated with the complex (Table I), the MD data were subjected to molecular mechanics Poisson–Boltzmann surface area (MM-PBSA) energy analysis. Such an analysis enables both a global and a per-residue quantification of the energetics responsible for the specificity of the interaction (Tables III and IV). The binding energy was decomposed into contributions from direct electrostatic interactions, polar and non-polar desolvation and van der Waals contacts, employing the MM-PBSA method (Luo et al. 2002).

The ability to assign binding energies on a per-residue basis is a powerful feature of this type of computational analysis as it permits not only the quantification of the strengths of the crystallographically observed interactions but also objectively identifies all energetically significant interactions. In contrast to hydrogen bonds, favorable van der Waals contacts or favorable desolvation-free energies are features that are difficult to quantify from a purely structural perspective. Moreover, by computing the binding energy as the difference between the energies of the protein–oligosaccharide complex and those of the free protein and oligosaccharide, an estimate of the binding free energy is obtained that cannot be deduced from any analysis of the complex alone. Lastly, by performing the energy calculations on multiple configurations extracted from the MD simulation rather than for the single configuration observed crystallographically, the computed

Table II. Average dihedral angles^a for the glycosidic linkages in 6'-SLN bound to PSL, as well as for the same linkages in glycans bound to influenza HA, and free in solution

Angle	PSL		Influenza HA		Free ligand in solution	
	X-ray ^a	MD ^b (10 ns)	H1 ^c	H3 ^d	NMR (Poppe et al. 1992)	MD (200 ns)
α (2-6) glycosidic linkage						
φ	56.2	54.4 (11)	64.2	54.5	–58/–169 (0.9:0.1) ^e	–60/178 (0.95:0.05)
ψ	161.1	174.8 (11)	179.1	–161.2	–176/100 (0.8:0.2) ^e	–166/81 (0.92:0.08)
ω	72.9	60.7 (9)	69.8	69.0	60/–60/180 (0.71:0.20:0.08)	69/–54/169 (0.79:0.08:0.14)
β (1-4) glycosidic linkage						
φ	–87.8	–78.8 (11)	–60.0	–	–83 ^e	–76/–145 (0.96:0.04)
ψ	–153.8	–133.7 (17)	–123.9	–	–126 ^e	–124/70 (0.97:0.03)

^aValues in degrees, averaged over two NCS-related protomers in the asymmetric unit. The inter-residue torsion angles are defined as follows: for the β (1 \rightarrow 4) linkage $\varphi = \text{O5-C1-O4'-C4'}$ and $\psi = \text{C1-O4'-C4'-C5'}$; for the α (2 \rightarrow 6) linkage $\varphi = \text{C1-C2-O6'-C6'}$ and $\psi = \text{C2-O6'-C6'-C5'}$; $\omega = \text{O6'-C6'-C5'-O5'}$.

^bAverage values and standard deviations (in parentheses) computed from a 10 ns trajectory.

^cAverage values from the 6'-SLN ligand conformations from the following PDB structures: 3HTQ (avian HA; Lin et al. 2009), 1RVZ (human HA; Gamblin et al. 2004), 1RVT (swine HA; Gamblin et al. 2004).

^dAverage values for the ligands in PDB structure 1MQN (avian HA; Ha et al. 2003).

^eAverage values and rotamer populations (in parentheses) determined from the scalar J-coupling and NOE data reported for Neu5Ac α 2-6Gal β 1-4Glc (Poppe et al. 1992), employing a rotational isomeric state analysis (DeMarco and Woods 2009), based on states isolated from the MD simulation of the free oligosaccharide. The φ angle has been altered by -120° to comply with the $\varphi = \text{C1-C2-O6'-C6'}$ convention used here.

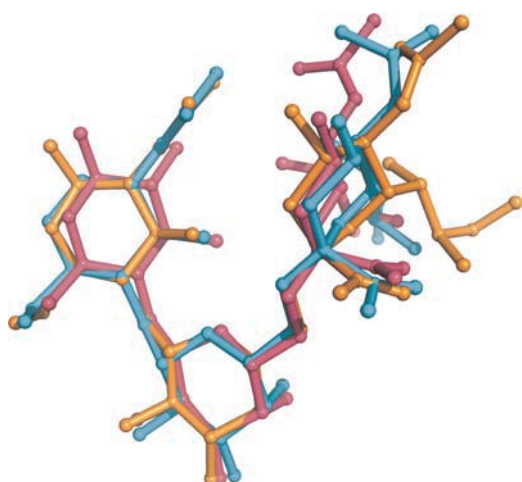


Fig. 5. Superimposition of the ligand from PSL with that from influenza A H1 HA (PDBID:1RVZ) and SRC (Gamblin et al. 2004). Although some disorder in the Neu5Ac residues is present, all the interglycosidic torsion angles populate the same canonical rotamers (Table II).

Table III. Estimated average energetic contributions to $\Delta G_{\text{Binding}}$ computed from an MM-PBSA analysis of the 10 ns MD data for the 6'-SLN-PSL complex relative to the free protein and ligand

Average energy component ^a	Energy (kcal mol ⁻¹)
$\langle \Delta E_{\text{Electrostatic}} \rangle$	-67.1 ± 0.4^b
$\langle \Delta G_{\text{Polar desolvation}} \rangle$	58.2 ± 0.4
Total electrostatics $\langle \Delta G_{\text{Polar}} + \Delta E_{\text{Electrostatic}} \rangle$	-8.9 ± 0.2
$\langle \Delta E_{\text{van der Waals}} \rangle$	-31.4 ± 0.1
$\langle \Delta G_{\text{Nonpolar desolvation}} \rangle$	-3.1 ± 0.0
$\langle \Delta G_{\text{Binding}} \rangle$	-43.5 ± 0.1

^aAveraged over 10 ns.

^bStandard error of the mean.

binding energies represent converged values that may be characterized with statistically significant error estimates.

The direct electrostatic contribution from ligand binding ($-67.1 \text{ kcal mol}^{-1}$) was largely offset by the energy ($58.2 \text{ kcal mol}^{-1}$) required to desolvate the polar residues in the binding interface. This is not unexpected given that the electrostatic interactions arise predominantly from hydrogen bonds involving hydroxyl groups in the glycan; hydrogen bonds which, depending on the polarity of the amino acid side chain involved, can be effectively equivalent in strength to those formed with water molecules (Tschampel and Woods 2003). Notably, the data in Table III lead to the conclusion that although hydrogen bonds are essential for binding specificity, a larger net contribution ($-31.4 \text{ kcal mol}^{-1}$) arises from van der Waals contacts. Similar relative contributions have been observed for several carbohydrate-protein complexes (Woods and Tessier 2010).

Per-residue binding energies

The overall binding energy between the protein and the carbohydrate arises from contributions from all the interacting atoms, and thus it is possible computationally to quantify the fractional contributions to the binding energy made by each residue in the protein and in the carbohydrate. Experimentally, similar per-residue affinity data could be obtained for the protein by performing alanine-scanning mutagenesis followed by affinity measurements; however, this would be extremely demanding to undertake.

As expected, all the amino acids that form hydrogen bonds with the ligand (Ser32, Asn30, Asp72, His76 and Asn94) were identified as forming strong electrostatic interactions in the MM-PBSA analysis. These interactions equate to an average hydrogen bond strength of $0.7 \text{ kcal mol}^{-1}$. Notably, several residues that do not form direct interactions were also identified as key for ligand binding, and in some cases (Ile85 and Ala74), the non-specific contacts contributed more to the

Table IV. Contribution to the interaction energies from each glycan residue, as well as for all key^a protein residues

Residue	$\langle \Delta E_{\text{Direct electrostatic}} \rangle$	$\langle \Delta G_{\text{Polar desolvation}} \rangle$	Total _{Electrostatic}	$\langle \Delta E_{\text{van der Waals}} \rangle$	$\langle \Delta G_{\text{Non-polar desolvation}} \rangle$	$\langle \Delta G_{\text{Binding}} \rangle$
Glycan						
GlcNAc	2.9	-1.5	1.4	-2.3	-0.3	-1.2
Gal	-30.2	19.6	-10.6	-5.0	-1.3	-16.9
Neu5Ac	-6.3	11.5	5.2	-8.4	-1.7	-4.9
Total	-33.6	29.6	-4.0	-15.7	-3.2	-22.9
Protein						
Ser32	-9.5	5.5	-4.1	-0.1	-0.1	-4.3
Lys31	-24.3	24.3	-0.1	-2.7	-0.3	-3.1
Gly75	-3.0	2.0	-0.9	-1.7	-0.3	-3.0
Tyr87	-1.0	1.8	0.8	-3.1	-0.3	-2.6
Ile85	-1.1	0.9	-0.2	-1.8	-0.1	-2.1
His76	-3.4	2.7	-0.6	-1.2	-0.2	-2.1
Ala74	-2.2	1.8	-0.3	-1.5	-0.1	-2.0
Asn94	-1.3	0.0	-1.3	-0.4	0.0	-1.8
Asp72	-4.0	1.2	-2.8	2.1	0.0	-0.7
Asn30	-1.2	2.1	0.8	-1.2	-0.2	-0.6
Asp34	15.8	-12.9	2.9	-0.2	0.0	2.8
Subtotal	-35.2	29.4	-5.8	-11.7	-1.9	-19.3
Percent of Total ^b	104.8	102.8	116.6	74.5	83.5	84.5

^aAny residue identified in the hydrogen-bonding analysis or that contributes $>1 \text{ kcal mol}^{-1}$ to $\langle \Delta G_{\text{Binding}} \rangle$.

^bRelative to the total contributions from all of the residues in the protein.

binding energy than that was contributed by hydrogen bond residues (Asn94, Asp72 and Asn30). In addition, the MM-PBSA analysis indicates that the aromatic stacking interaction between Tyr87 ranks this residue as fourth most critical to binding to a net binding energy of $-2.6 \text{ kcal mol}^{-1}$. Approximately 90% of the stacking energy comes from van der Waals contacts, with only 10% predicted to arise from desolvation. This analysis, therefore, suggests that dispersion forces (Morales et al. 2008), and not the hydrophobic effect (Di Lella et al. 2009), are the driving force behind the observed carbohydrate–aromatic stacking.

From the perspective of the carbohydrate, there is no experimental method that provides per-residue binding energy estimates. The computational per-residue binding energy data for the 6'-SLN- Δ PSL system (Table IV) indicate that $\sim 95\%$ of the strength of binding comes from interactions with the disaccharide Neu5Ac-Gal, with the sialic acid accounting for 20% of the total binding affinity, with negligible contribution from the GlcNAc residue. These data are consistent with the structural observation that the GlcNAc residue does not make significant direct contacts with the protein. Further, the data indicate that although the protein might retain Gal-binding affinity, the removal of the Neu5Ac residue would be expected to decrease the affinity by almost 6 kcal mol^{-1} . This prediction is consistent with the greatly attenuated affinity of non-sialylated ligands for PSL (Tateno et al. 2004). However, this estimate neglects any consideration of the significant entropic penalties associated with binding a flexible glycan (see following section).

Role of entropy and enthalpy in ligand binding

Given the known plasticity of glycans, it is common that carbohydrate binding is accompanied by an entropic penalty. Glycan flexibility is readily characterized by monitoring the motions of the interglycosidic torsion angles (φ , ψ and ω), and it is particularly convenient to employ changes in the torsional motions of these angles in the free and bound states to quantify any change in conformational entropy that may occur upon binding.

The conformation of 6'-SLN bound to PSL is indistinguishable from that of the predominant conformation in solution (Table II) and essentially identical to that seen in complexes with the lectin SRC and all influenza HA examples to date. This might suggest that there should be little entropic penalty incurred upon binding. However, the data for 6'-SLN indicate that an entropic penalty of $\sim 3.65 \text{ kcal mol}^{-1}$ is incurred upon binding to PSL (Table V). Of this total penalty, $2.68 \text{ kcal mol}^{-1}$ (73%) arises from stiffening of the three-bond

Neu5Ac α 2-6Gal linkage, with the largest contribution, $1.05 \text{ kcal mol}^{-1}$, arising from the restricted motion of the ω torsion angle. The remaining $0.95 \text{ kcal mol}^{-1}$ comes from restrictions of the rotational degrees of freedom of the β 1-4 linkage. Taking conformational entropy into account, the MM-PBSA contribution to binding, arising exclusively from the Neu5Ac group, is reduced from $-5.9 \text{ kcal mol}^{-1}$ to approximately $-2.25 \text{ kcal mol}^{-1}$.

α 2-6 vs α 2-3 specificity

The origin of the preference of PSL for the Neu5Ac α 2-6Gal sequence over the α 2-3 was elucidated by superimposing the ring atoms of the Gal residue in the Neu5Ac α 2-3Gal disaccharide, created and geometry minimized using the online carbohydrate builder Glycosides and Glycoproteins in AMBER (GLYCAM)-Web (<http://www.glycam.com>; Woods Group 2005–2011) on the Gal residue in the Neu5Ac α 2-6Gal sequence in the crystal structure. Both of the experimentally observed orientations of the φ -glycosidic angle ($\varphi = \text{C1-C2-O3'-C3'}$; Siebert et al. 1992; Kiddle and Homans 1998; Milton et al. 1998) associated with the α 2-3 linkage were included in the modeling and examined for interactions with the protein surface. In each idealized case ($\varphi = 180^\circ$, -60° , 60°), notable steric clashes with the protein surface were observed, in particular with the backbone sequence associated with residues Gly92 and Tyr87 (Figure 6A–C). As the C2-O3' bond is perpendicular to the protein surface, it is apparent that indeed no orientation of the α 2-3 linkage could fit into the PSL-binding site without requiring severe deformation of the protein surface.

Gal vs GalNAc (HexNAc) specificity

The molecular basis for the Gal/GalNAc specificity was identified by superimposing the ring atoms of a GalNAc residue, created and geometry minimized using the online carbohydrate builder GLYCAM-Web (Woods Group 2005–2011), onto the coordinates of the Gal residue in the binding site. In such an orientation, an obvious steric collision appeared to be present between the NAc group of the GalNAc and the side chain of His76 (Figure 6D). This side chain participates in hydrogen bonding with Gln112 and so would not be free to escape the collision with a GalNAc simply by reorienting its position. Notably, the MM-PBSA data suggest that His76 also contributes to the overall affinity for the Gal residue. By structural analogy to GalNAc, neither a 4-linked GlcNAc nor a 4-linked ManNAc residue would be predicted to be tolerated in the PSL-binding site.

Conclusions

The role of the α , β and γ subdomains of the PSL β -trefoil in ligand binding has been investigated previously using site-directed mutagenesis (Tateno et al. 2004). The results indicated that the β subdomain was the most probable carbohydrate-binding site in the N-terminal domain since it contained all the residues required both for forming the hydrophobic core of the ricin-like trefoil and those involved in binding a carbohydrate, namely, Asp72, Tyr87, Gln95 and Trp97. Sequence alignment and ligand proximity analysis of the PSL, SRC and MOA

Table V. Contributions to the conformational-binding entropy (kcal mol^{-1})

	6'-SLN	Neu5Ac α 2-6Gal linkage	Gal β 1-4GlcNAc linkage
$-T S_{\text{Total}}$	3.65	—	—
$-T S_{\varphi\psi\omega}$	—	2.68	—
$-T S_{\varphi\psi}$	—	1.10	0.95
$-T S_{\varphi}$	—	0.68	0.35
$-T S_{\psi}$	—	0.73	0.51
$-T S_{\omega}$	—	1.05	—

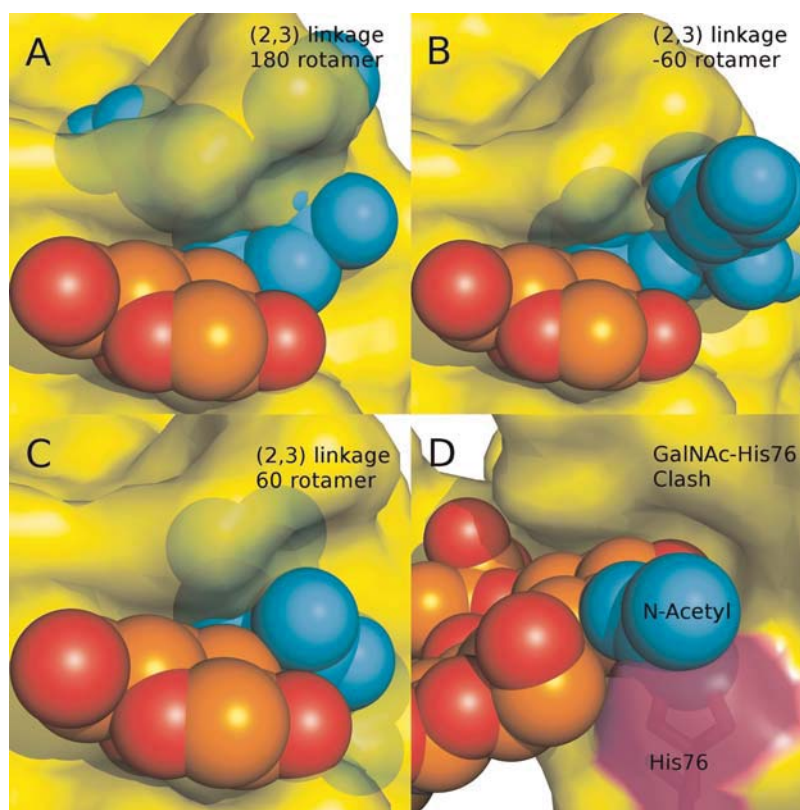


Fig. 6. (A–C) Models of the Neu5Ac α 2-3Gal β -1-4GlcNAc trisaccharide bound to PSL, indicating that no rotamers of the 2-3 linkage can be adopted that avoid collisions with the PSL surface. (D) Orientation of a GalNAc-containing trisaccharide in the binding site of PSL, indicating a collision with His76 (magenta).

lectins (Figures 3 and 4) now indicate that the α subdomain has altered in PSL so as to provide key interactions with the Neu5Ac residue in the oligosaccharide bound in the β subdomain. This adaptation has apparently forced the abandonment of the original α subdomain-binding motif, which is consistent with conclusions drawn from the initial sequence analysis that the α subdomain in PSL lacks residues deemed essential for Gal-binding affinity (Mo et al. 2000; Tateno et al. 2004). Previously, a monomeric form of PSL was found to agglutinate rabbit erythrocytes, indicating the presence of two binding sites, which led to the conclusion that the γ domain, which still contains some of the residues essential for Gal binding, may still retain affinity for Gal β 1-4GlcNAc-terminating sequences (Tateno et al. 2004). Our proximity analysis indeed suggests that the γ domain likely retains Gal-binding affinity. However, the binding data from the MM-PBSA analysis suggest that any such interactions are markedly weaker than those for 6'-SLN-terminating sequences.

A detailed computational analysis of the interaction between 6'-SLN and PSL provides a structural explanation for all the known binding specificities. In addition, the computational data lead to the conclusion that although specificity is conferred by direct hydrogen bonds between the oligosaccharide and the protein, non-specific van der Waals contacts contribute significantly to the total affinity. As in all similar analyses of carbohydrate-protein binding to date, the strength of direct electrostatic interactions is largely attenuated by desolvation free energy (Woods and Tessier 2010).

Lastly, in all co-crystal structures of influenza HAs with human-type receptor glycans, the glycan displays a characteristic three-dimensional shape (Chandrasekaran et al. 2008) equivalent to that seen in the 6'-SLN-PSL complex, suggesting that PSL will be particularly well suited to histological investigation-related influenza adhesion. Notably, this glycan conformation is also the preponderant conformation seen in solution, and yet there remains a 3.6 kcal mol⁻¹ conformational entropic penalty to the binding of 6'-SLN to PSL, confirming that even for less-flexible glycans, such as 6'-SLN, entropy plays a significant role in determining binding affinity. This observation may be of particular relevance in the emerging area of glycomimetic drug design (Ernst and Magnani 2009).

Experimental

X-ray crystallography

A sample of 6'-SLN was purchased from V-Labs, Covington, LA, U.S.A. Recombinant PSL, with a molecular weight of 31 kDa, was cloned, expressed in *Escherichia coli* as an active and soluble form (Tateno et al. 2004), purified and lyophilized. The recombinant lectin was reconstituted in a buffer containing 10 mM phosphate, pH 7.2, and 150 mM NaCl, purified using gel filtration chromatography on a Sephadex G-200 column and concentrated to ~ 15 mg mL⁻¹. About 24 h prior to the crystallization trials, PSL was combined with the

oligosaccharide 6'-SLN at a molar ratio of 2:1 (oligosaccharide:lectin). The microbatch-under-oil method (Chayen et al. 1992) was initially used to assess the solubility profile of the protein, choice of precipitant, pH and temperature. Small, needle-shaped crystals were obtained from polyethylene glycol 8000 (PEG 8000) precipitant solutions, and a stepwise optimization led to diffraction quality crystals. Cubic-shaped crystals were grown at 20°C from hanging drops composed of 2 µL of the lectin/ligand solution and 2 µL of precipitant. The drops were equilibrated against 22–25% PEG 8000 and 100 mM sodium 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid at pH 6.9–7.2.

The crystals were cryoprotected with PEG 400, and low-temperature diffraction data were collected on cryoprotected crystals at the IMCA-CAT beamline 17-ID at the Advanced Photon Source synchrotron ring at the Argonne National Lab, Argonne, IL. Data acquisition was accomplished using the ADSC Q210 Charge-coupled device with one crystal-rotation axis (φ) and a computer-controlled sample-to-detector distance. The data were recorded as 360 non-overlapping images with a 0.5° oscillation step and a 3 s exposure time. The HKL-2000 suite of programs (Otwinowski and Minor 1997) was used for data processing and scaling. Intensities were converted to structure factors using the program TRUNCATE from the CCP4 (Collaborative Computational Project Number 1994) suite of packages. The Matthew's coefficient was 2.86 Da Å⁻³, and the calculated solvent content in the asymmetric unit was 57%. The crystal parameters and data collection statistics are summarized in Table VI.

Earlier cDNA cloning had revealed that PSL contained a ricin-like B chain (QxW)₃ domain (Hazes 1996) at its

N-terminus. Mutant constructs with single amino acid substitutions in the putative sugar-binding sites had also indicated that the β and γ subdomains were the most probable sugar-binding sites (Tateno et al. 2004). A list of structures was obtained for use as possible models for molecular replacement (MR) using sequences annotated by structure (Milburn et al. 1998). The highest sequence identity (38%) relative to PSL was the Gal α 1-3Gal-binding lectin MOA (Winter et al. 2002). The deposited crystal structure of MOA in complex with a trisaccharide (2IHO) was generously provided by Professor Ute Krengel and coworkers a few months before release by the PDB (Grahm et al. 2007). The PSL structure was solved as a single solution using a monomer of MOA as the MR model and with the program Phaser (McCoy et al. 2007) from the PHENIX (Adams et al. 2002) software suite. Model bias was reduced using density modification (Cowtan 1994) and the ReSOLVE (Terwilliger 2003) package was used for initial rebuilding. This was followed by iterative manual rebuilding and macromolecular refinement using Refmac5 (Murshudov et al. 1997) and the visualization and refinement software packages O (Jones et al. 1991) and Coot (Emsley and Cowtan 2004). The carbohydrate chains were located and refined, followed by location and refinement of water molecules. The final model consists of two polypeptide chains of 285 amino acid residues each, two 6'-SLN molecules and 535 water molecules. Loop residues 26–29 and 123–125 and Met1 were not located in electron density maps and flagged as disordered or missing residues. Glu204 is located in type IV β -turn and is an outlier in the Ramachandran plot (main chain RSCC = 0.98).

Computational simulations

MD simulations were performed on free 6'-SLN and on the N-terminal carbohydrate-binding domain (residues 1–153, Δ PSL) of the PSL–6'-SLN complex, both in explicit water, employing the assisted model building and energy refinement (AMBER) force field (Case et al. 2005) with the PARM99SB (Hornak et al. 2006) parameters selected for the protein and counter ions and the GLYCAM06 (Kirschner et al. 2008) parameters for the carbohydrate. Long-range electrostatic interactions were treated with the Particle-Mesh Ewald method (Darden et al. 1993; Sagui and Darden 1999) using a cutoff of 9 and 1.2 Å grid spacing. Methyl and acetyl groups were added to the protein structure to terminate chains with unresolved residues. The most significant impact of these missing sequences was the need to apply restraints on the C α positions to ensure the protein was maintained in the correct overall fold during the MD simulation. Four Na⁺ ions were added to neutralize the net charge on the system. The PSL–6'-SLN system was then centered in a cubic box (70 Å side) of TIP3P water molecules (Jorgensen et al. 1983). Crystallographic water molecules were retained in the simulation. Water and ion positions were minimized and equilibrated for 500 ps (NVT), whereas the protein and carbohydrate non-hydrogen atoms were restrained with a force constant of 1000 kJ mol⁻¹ nm⁻². An additional 500 ps of equilibration under NPT conditions was then performed with only the non-hydrogen atoms in the protein restrained. Finally, an equilibration under NPT conditions with only the C α atoms of the protein

Table VI. X-ray data and refinement statistics

Data collection	
Space group	P2 ₁ 2 ₁ 2
Wavelength (Å)	1.00
Unit cell parameters <i>a</i> , <i>b</i> and <i>c</i> (Å)	115.8, 59.4, 103.2
Resolution (Å)	38.5 to 1.7
Data completeness (%)	99.1 (92.8) ^a
Observed reflections	451179
Unique reflections	79179
Multiplicity	5.7
<i>R</i> _{sym}	6.2 (23.8) ^a
$\langle I/\sigma(I) \rangle$	26.3 (2.7) ^a
Refinement	
Number of protein atoms	4318
Number of water molecules	632
Number of ligand atoms	92
RMSD bond length ^b (Å)	0.023
RMSD bond angles ^b (°)	1.98
Average temperature (B) factors (Å ²)	
All atoms	28.9
Main chain	26.3
Side chain and water molecules	31.5
Ligand	53.2
Ramachandran statistics (%)	
Core and additionally allowed regions	99.4
Disallowed regions	0.6
Crystallographic <i>R</i> _{work}	18.5 (24.2) ^a
Crystallographic <i>R</i> _{free}	21.7 (24.4) ^a

^aValues in parentheses are for the highest resolution shell (1.76–1.7 Å).

^bRMSDs from restraint targets (Engl and Huber 1991).

restrained was performed before a production run of 10 ns was collected under the same conditions. For the simulation of free 6'-SLN, one Na⁺ ion was added to neutralize the net charge the oligosaccharide, which was then centered in a cubic box (30 Å side) of TIP3P water molecules (Jorgensen et al. 1983). Heating and equilibration were then performed as for the protein–oligosaccharide complex, before a production run of 200 ns was collected. All MD simulations were performed with version 4.04 of Groningen machine for chemical simulations (GROMACS; Hess et al. 2008). The AMBER/GLYCAM force field parameters were imported into GROMACS by means of the amb2gmx script, written and distributed by Eric J. Sorin (www.chemistry.csulb.edu/ffamber).

MM-PBSA calculations

Employing 1000 snapshots extracted at 100 ps intervals from the MD trajectory of the 6'-SLN–ΔPSL complex, the binding energy was computed directly from the energies of the binding reaction components [Equation (1)]:

$$\Delta G_{\text{Binding}} = \Delta G_{\text{Complex}} - \Delta G_{\text{Protein}} - \Delta G_{\text{Carbohydrate}} \quad (1)$$

The approach taken follows closely that reported for other carbohydrate–protein complexes (Bryce et al. 2001; Ford et al. 2003; Kadirvelraj et al. 2006). The free energies of the components were approximated by separating the energies into molecular mechanical (electrostatic and van der Waals), polar and apolar solvation and entropic contributions [Equation (2)]:

$$\Delta G = \Delta E_{\text{MM}} - T \Delta S_{\text{MM}} + \Delta G_{\text{Desolvation}} \quad (2)$$

Before the analysis, the water molecules were removed from the solvated trajectory. The energy contribution from solvation was then obtained through application of the Poisson–Boltzmann implicit solvation model.

Entropy calculations

Conformational entropy penalties were estimated using the Karplus–Kushick approach (Karplus and Kushick 1981), in which the relative entropies were derived from the ratios of the determinants of the covariance matrices associated with the glycosidic torsion angles for the bound (σ_B) and free (σ_F) oligosaccharides [Equation (3)]. In the case of carbohydrates, it is particularly appropriate to focus on the conformational entropy associated with the interglycosidic torsion angles (Bryce et al. 2001; Kadirvelraj et al. 2006).

$$\Delta S_{\text{Conformational}} = \frac{1}{2} k_B \ln \left(\frac{\sigma_B}{\sigma_F} \right) \quad (3)$$

Due to the absence of loop residues 26–29 and 123–125 in the crystallographic data, we were unable to perform an analysis of the configurational entropy change associated principally with the stiffening of the protein that occurs upon oligosaccharide binding (Ford et al. 2003).

Supplementary data

Supplementary data for this article is available online at <http://glycob.oxfordjournals.org/>.

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Conflict of interest

None declared.

Atomic coordinates

The atomic coordinates and structure factor amplitudes for PSL has been deposited in the RCSB Protein Data Bank (Bernstein et al. 1977) under PDB ID 3PHZ.

Abbreviations

AMBER, assisted model building and energy refinement; RMSD, root mean-square deviation; Gal, galactose; Galβ1-4GlcNAc, lactosamine; GlcNAc, *N*-Acetyl glucosamine; GLYCAM, Glycosides and Glycoproteins in AMBER; GROMACS, Groningen machine for chemical simulations; HA, hemagglutinin; MD, molecular dynamics; MOA, *Marasmius oreades* agglutinin; MM-PBSA, molecular mechanics Poisson–Boltzmann surface area; MR, molecular replacement; Neu5Ac, sialic acid/5-*N*-Acetylneuraminic acid; PEG, polyethylene glycol; PSL, *Polyporus squamosus* lectin; SNA, *Sambucus nigra* agglutinin; SRC, sialic acid-recognizing lectin; ST6Gal, α2-6 sialyltransferase; 6'-SLN, Neu5Acα2-6Galβ1-4GlcNAc.

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