

Ultrasound-Promoted Synthesis of 3-(Thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamides and Anticancer Activity Evaluation in Leukemia Cell Lines

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3-(Thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamides were efficiently prepared through a cyclocondensation of thiophenylchalcones with aminoguanidine hydrochloride under ultrasonic conditions in the presence of KOH and ethanol as a green solvent in short reaction times (15-35 min) and good yields (62-95%). All compounds produced were evaluated against the human Jurkat and RS4;11 acute lymphoblastic leukemia cell lines of T- and B-cell origin, respectively, and the K562 myelogenous leukemia cell line. Six compounds presented half maximal inhibitory concentration (IC₅₀) values around 15 μmol L⁻¹ and five compounds presented IC₅₀ values around 40 μmol L⁻¹ for at least one of the three cell lines analyzed. One compound was not significantly cytotoxic, presenting IC₅₀ value > 100 μmol L⁻¹.

Keywords: amidinopyrazole, pyrazoline, cytotoxic activity, leukemia, ultrasonic irradiation

Introduction

Amidine functional group has proven to be an important fragment in compounds with recognized bioactivities. In this sense, several amidines have been prepared and their antifungal,¹ antiprotozoal,² antibacterial,³ anti-HIV,⁴ antithrombotic,⁵ and antidegenerative⁶ potentials have been evaluated. Pentamidine is clinically used for treatment of pneumonia and first stage human African tripanosomiasis.⁷ That drug is on the WHO's List of Essential Medicines for the basic health system.⁸ Moreover, amidine containing molecules have been pointed as prominent prototypes in the search for new anticancer agents.⁹ In addition,

thiophene-2-carboximidamides have shown potent and selective inhibitory activities of nitric oxide synthases for the treatment of human melanoma.¹⁰

In the same context, pyrazoles are recognized as a fundamental class of heterocyclic compounds because of their well-established applicability in several areas as agrochemicals, functional materials and medicines.¹¹ Dihydro-1H-pyrazole derivatives have been reported as antiproliferative agents¹² and inhibitors of vascular endothelium growth factors,¹³ mitotic kinesin spindle protein (KSP),¹⁴ and multidrug resistance protein 1 (MDR1).¹⁵ Pyrazoline derivatives have shown high selectivity against leukemia cell lines (the concentration for 50% of maximal inhibition of cell proliferation, GI₅₀ = 0.69-3.35 μmol L⁻¹) in comparison with eight other tumor cell lines.¹⁶

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Taking into account the valuable pharmacological properties of pyrazoline and amidine scaffolds, we envisioned that hybrid molecules could be active against leukemia cells. Analogous 4,5-dihydro-1H-pyrazole-1-carboximidamides have been already prepared by the cyclocondensation of α,β -unsaturated ketones with aminoguanidine free base under conventional¹⁷ and sonochemical¹⁸ conditions. However, to the best of our knowledge there are not reported studies about the antileukemic potency of these hybrid molecules.

Thus, in our continuous efforts to develop sonochemically promoted reactions in environmentally benign solvents¹⁹ and synthetic methodologies for preparation of heterocyclic compounds²⁰ selected on the basis of their biological activity,²¹ we describe herein a rapid and efficient synthetic method for the preparation of 3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochlorides under ultrasonic conditions. In addition, all compounds synthesized were evaluated against the human Jurkat and RS4;11 acute lymphoblastic leukemia cell lines of T- and B-cell origin, respectively, and the K562 myelogenous leukemia cell line.

Experimental

Chemistry

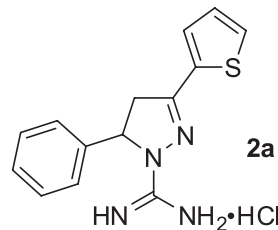
General

3-Aryl-1-(thiophen-2-yl)prop-2-en-1-ones **1a-l** were prepared by us following reported procedures.²² All the chemicals were used without purification as purchased from commercial suppliers. The sonicated reactions were carried out with a microtip probe connected to a 500 W Sonics Vibracell ultrasonic processor operating at 20 kHz at 20% of the maximum power output. Reaction progresses were monitored by gas chromatography (GC). Melting point values were determined in open capillary on an Instrutherm DF-3600 II apparatus and are uncorrected. Infrared spectra (IR) were acquired on a JASCO-4100 spectrophotometer as KBr pellets. ¹H and ¹³C nuclear magnetic resonance (NMR) spectra were acquired on a Bruker DPX400 instrument (400 MHz for ¹H and 101 MHz for ¹³C) in 5 mm sample tubes at 298 K in dimethyl sulfoxide (DMSO-*d*₆) using tetramethylsilane (TMS) as internal reference standard. High resolution mass spectra were recorded on a Bruker microTof (Q-TOF) mass spectrometer in electrospray ionization (ESI) mode.

General procedure for the ultrasound-promoted synthesis of 5-aryl-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamides (**2a-l**)

To a 50 mL vial containing a solution of 3-aryl-1-(thiophen-2-yl)prop-2-en-1-ones **1a-l** (1 mmol) in ethanol (15 mL), the aminoguanidine hydrochloride (0.22 g, 2 mmol) and KOH (0.11 g, 2 mmol) were added. The reaction mixture was sonicated for the time indicated in Table 1 and the reaction temperature reached 55-60 °C after 10 minutes. The resulting solution was cooled to room temperature and acidified using 10% HCl (10-15 mL). The salts were extracted with chloroform (3 × 20 mL) and the combined organic layer was dried over anhydrous magnesium sulfate. Removal of solvent under vacuum afforded a crude material. The pure products **2a-l** were obtained as amorphous solids with yields of 62-95% after recrystallization from ethyl acetate and drying in a desiccator.

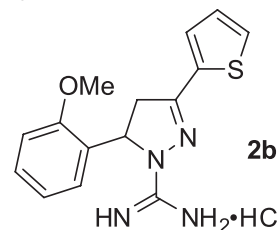
5-Phenyl-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochloride (**2a**)



Yellowish solid; mp 266-268 °C; IR (KBr) ν / cm^{-1} 3337, 3115, 1609, 1430; ¹H NMR (400 MHz, DMSO-*d*₆) δ 7.97 (bs, 4H, C(NH₂)₂), 7.86 (d, 1H, *J* 5.0 Hz, T-H*), 7.57 (d,

1H, *J* 3.6 Hz, T-H), 7.39 (d, 2H, *J* 7.6 Hz, Ph-H), 7.35-7.32 (m, 1H, Ph-H), 7.25 (d, 2H, *J* 7.4 Hz, Ph-H), 7.20-7.17 (m, 1H, T-H), 6.02 (dd, 1H, *J* 2.4, 11.0 Hz, H_x[†]), 4.11 (dd, 1H, *J* 11.2, 17.8 Hz, H_m[†]), 3.36 (dd, 1H, *J* 2.7, 17.8 Hz, H_a[†]); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 153.2, 153.0, 139.6, 132.7, 132.0, 131.3, 129.0, 128.2, 125.3, 60.4, 44.4; HRMS *m/z*, [M + H]⁺ calcd. for C₁₄H₁₅N₄S: 271.1017; found: 271.1015. *T-H: thiophene hydrogens; [†]H_a, H_m and H_x: pyrazolyl ring hydrogens.

5-(2-Methoxyphenyl)-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochloride (**2b**)

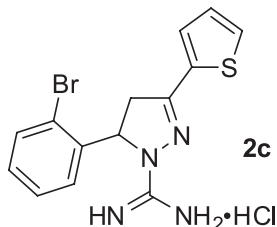


Yellowish solid; mp 287-288 °C; IR (KBr) ν / cm^{-1} 3255, 3190, 3103, 1618, 1423, 1243; ¹H NMR (400 MHz, DMSO-*d*₆) δ 7.85 (bs, 4H, C(NH₂)₂), 7.82 (dd, 1H, *J* 0.9, 5.0 Hz, T-H), 7.54

(dd, 1H, *J* 0.8, 3.6 Hz, T-H), 7.34 (m, 1H, Ph-H), 7.16 (dd, 1H, *J* 3.8, 5.0 Hz, T-H), 7.11 (d, 1H, *J* 8.2 Hz, Ph-H), 6.97-6.89 (m, 2H, Ph-H), 5.87 (dd, 1H, *J* 3.2, 11.1 Hz, H_x), 4.04 (dd, 1H, *J* 11.1, 17.7 Hz, H_m), 3.83 (s, 3H, Ph-OCH₃), 3.24 (dd, 1H, *J* 3.2, 17.7 Hz, H_a); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 156.2, 153.8, 152.9, 132.8, 131.8, 131.1, 129.6, 128.2, 126.3, 124.9, 120.3, 111.8, 57.2, 55.7, 43.3;

HRMS m/z , $[M + H]^+$ calcd. for $C_{15}H_{17}N_4OS$: 301.1123; found: 301.1148.

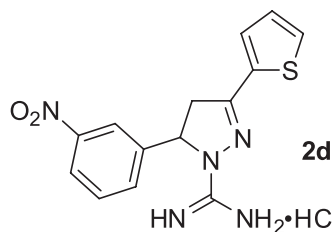
5-(2-Bromophenyl)-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochloride (**2c**)



Yellowish solid; mp 285-288 °C; IR (KBr) ν / cm^{-1} 3290, 3117, 1603, 1423; ^1H NMR (400 MHz, DMSO- d_6) δ 8.03 (bs, 4H, $\text{C}(\text{NH}_2)_2$), 7.86 (d, 1H, J 4.9 Hz, T-H), 7.57 (d, 1H, J 3.4 Hz, T-H), 7.55

(d, 1H, J 8.1 Hz, Ph-H), 7.48 (m, 1H, Ph-H), 7.38 (t, 1H, J 7.9 Hz, Ph-H), 7.22-7.18 (m, 2H, T-H and Ph-H), 6.04 (dd, 1H, J 2.8, 11.2 Hz, H_x), 4.11 (dd, 1H, J 11.2, 17.9 Hz, H_m), 3.42 (dd, 1H, J 2.9, 17.9 Hz, H_x); ^{13}C NMR (101 MHz, DMSO- d_6) δ 153.2, 153.1, 135.2, 133.4, 132.6, 132.3, 132.0, 131.3, 129.8, 128.1, 127.6, 126.7, 58.6, 43.2; HRMS m/z , $[M + H]^+$ calcd. for $C_{14}H_{14}\text{BrN}_4\text{S}$: 349.0122; found: 349.0129.

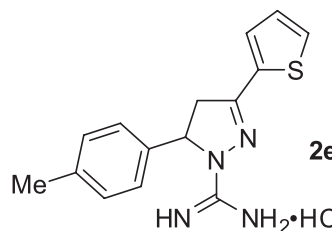
5-(3-Nitrophenyl)-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochloride (**2d**)



Yellowish solid; mp 288-291 °C; IR (KBr) ν / cm^{-1} 3363, 3284, 3204, 3103, 1611, 1524, 1423, 1344, 847; ^1H NMR (400 MHz, DMSO- d_6) δ 8.22 (ddd,

1H, J 0.9, 2.2, 8.2 Hz, Ph-H), 8.10 (t, 1H, J 1.9 Hz, Ph-H), 7.97 (s, 4H, $\text{C}(\text{NH}_2)_2$), 7.87 (dd, 1H, J 1.1, 5.0 Hz, T-H), 7.74 (t, 1H, J 8.0 Hz, Ph-H), 7.65 (d, 1H, J 7.9 Hz, Ph-H), 7.55 (dd, 1H, J 1.1, 3.7 Hz, T-H), 7.20 (dd, 1H, J 3.7, 5.0 Hz, T-H), 6.07 (dd, 1H, J 3.2, 11.3 Hz, H_x), 4.16 (dd, 1H, J 11.4, 18.0 Hz, H_m), 3.48 (dd, 1H, J 3.2, 18.0, H_a); ^{13}C NMR (101 MHz, DMSO- d_6) δ 153.2, 152.8, 148.0, 141.5, 132.5, 132.2, 132.0, 131.5, 130.9, 128.2, 123.2, 120.7, 59.6, 44.3; HRMS m/z , $[M + H]^+$ calcd. for $C_{14}H_{14}\text{N}_5\text{O}_2\text{S}$: 316.0868; found: 316.0878.

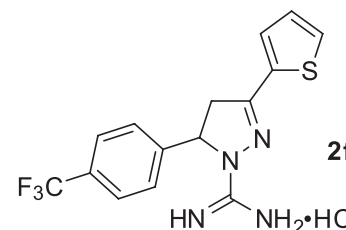
5-(4-Methylphenyl)-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochloride (**2e**)



Yellowish solid; mp 278-280 °C; IR (KBr) ν / cm^{-1} 3255, 3096, 1611, 1430; ^1H NMR (400 MHz, DMSO- d_6) δ 7.91 (bs, 4H, $\text{C}(\text{NH}_2)_2$), 7.84 (d,

1H, J 5.0 Hz, T-H), 7.56 (d, 1H, J 3.6 Hz, T-H), 7.21-7.17 (m, 3H, T-H and Ph-H), 7.13 (d, 2H, J 8.1 Hz, Ph-H), 5.92 (dd, 1H, J 3.0, 11.1 Hz, H_x), 4.07 (dd, 1H, J 11.2, 17.9 Hz, H_m), 3.33 (dd, 1H, J 3.0, 17.7 Hz, H_a), 2.28 (s, 3H, Ph- CH_3); ^{13}C NMR (101 MHz, DMSO- d_6) δ 153.1, 153.0, 137.6, 136.6, 132.8, 131.9, 131.2, 129.5, 128.2, 125.3, 60.2, 44.3, 20.6; HRMS m/z , $[M + H]^+$ calcd. for $C_{15}H_{17}\text{N}_4\text{S}$: 285.1174; found: 285.1179.

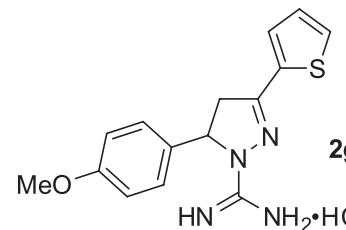
5-(4-Trifluoromethylphenyl)-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochloride (**2f**)



Yellowish solid; mp 295-298 °C; IR (KBr) ν / cm^{-1} 3370, 3290, 3068, 1603, 1423, 1315-1113, 839; ^1H NMR (400 MHz, DMSO- d_6) δ 8.14 (bs,

4H, $\text{C}(\text{NH}_2)_2$), 7.88 (d, 1H, J 4.7 Hz, T-H), 7.79 (d, 2H, J 8.2 Hz, Ph-H), 7.60 (d, 1H, J 3.1 Hz, T-H), 7.52 (d, 2H, J 8.1 Hz, Ph-H), 7.21-7.19 (m, 1H, T-H), 6.27 (dd, 1H, J 2.4, 11.0 Hz, H_x), 4.19 (dd, 1H, J 11.3, 17.8 Hz, H_m), 3.45 (dd, 1H, J 2.8, 17.9 Hz, H_a); ^{13}C NMR (101 MHz, DMSO- d_6) δ 153.1, 153.0, 143.9 (q, J 1.3 Hz), 132.5, 132.0, 131.3, 128.7 (q, J 32.0 Hz), 128.1, 126.3, 125.8 (q, J 3.8 Hz), 123.9 (q, J 272.1 Hz), 60.0, 44.2; HRMS m/z , $[M + H]^+$ calcd. for $C_{15}H_{14}\text{F}_3\text{N}_4\text{S}$: 339.0891; found: 339.0881.

5-(4-Methoxyphenyl)-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochloride (**2g**)

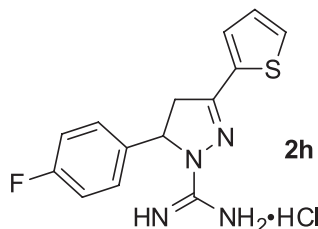


Yellowish solid; mp 296-299 °C; IR (KBr) ν / cm^{-1} 3334-3132, 1611, 1438, 1243, 839; ^1H NMR (400 MHz, DMSO- d_6) δ 7.85

(dd, 1H, J 1.0, 5.0 Hz, T-H), 7.76 (bs, 4H, $\text{C}(\text{NH}_2)_2$), 7.56 (dd, 1H, J 1.1, 3.7 Hz, T-H), 7.19 (dd, 1H, J 3.7, 5.0 Hz, T-H), 7.16-7.09 (m, 2H, Ph-H), 7.01-6.89 (m, 2H, Ph-H), 5.79 (dd, 1H, J 2.4, 11.0 Hz, H_x), 4.05 (dd, 1H, J 11.1, 17.8 Hz, H_m), 3.73 (s, 3H, Ph- OCH_3), 3.41-3.26 (m, 1H, H_a); ^{13}C NMR (101 MHz, DMSO- d_6) δ 159.2, 153.3, 152.7, 132.8, 132.0, 131.5, 131.3, 128.2, 126.7, 114.4, 60.0, 55.2, 44.4; HRMS m/z , $[M + H]^+$ calcd. for $C_{15}H_{17}\text{N}_4\text{OS}$: 301.1123; found: 301.1132.

5-(4-Fluorophenyl)-3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamide hydrochloride (**2h**)

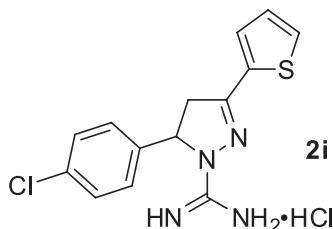
Yellowish solid; mp 243-245 °C; IR (KBr) ν / cm^{-1} 3348-3103, 1596, 1416, 1229, 832; ^1H NMR (400 MHz,



DMSO-*d*₆ δ 7.93 (s, 4H, C(NH₂)₂), 7.86 (dd, 1H, *J* 1.1, 5.0 Hz, T-H), 7.57 (dd, 1H, *J* 1.1, 3.7 Hz, T-H), 7.33-7.21 (m, 4H, Ph-H), 7.19 (dd, 1H, *J* 3.7,

5.0 Hz, T-H), 5.94 (dd, 1H, *J* 2.9, 11.2 Hz, H_x), 4.09 (dd, 1H, *J* 11.2, 17.9 Hz, H_m), 3.53-3.27 (m, 1H, H_a); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 163.1, 160.6, 153.0 (d, *J* 34.2 Hz), 135.8 (d, *J* 2.7 Hz), 132.7, 132.1, 131.4, 128.2, 127.6 (d, *J* 8.3 Hz), 115.9 (d, *J* 21.7 Hz), 59.7, 44.4; HRMS *m/z*, [M + H]⁺ calcd. for C₁₄H₁₄FN₄S: 289.0923; found: 289.0936.

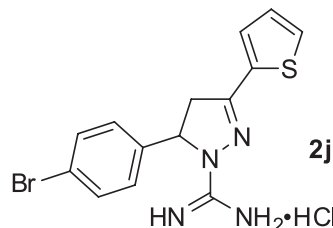
5-(4-Chlorophenyl)-3-(thiophen-2-yl)-4,5-dihydro-1*H*-pyrazole-1-carboximidamide hydrochloride (**2i**)



Yellowish solid; mp 257-260 °C; IR (KBr) ν / cm⁻¹ 3070, 1606, 1425, 825; ¹H NMR (400 MHz, DMSO-*d*₆) δ 7.98 (bs, 4H, C(NH₂)₂), 7.85 (d,

1H, *J* 4.8 Hz, T-H), 7.56 (d, 1H, *J* 3.1 Hz, T-H), 7.47 (d, 2H, *J* 8.2 Hz, Ph-H), 7.27 (d, 2H, *J* 8.3 Hz, Ph-H), 7.20-7.18 (m, 1H, T-H), 6.01-5.99 (m, 1H, H_x), 4.10 (dd, 1H, *J* 11.2, 17.8 Hz, H_m), 3.40-3.35 (m, 1H, H_a); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 153.2, 152.9, 138.5, 132.9, 132.6, 132.1, 131.4, 129.0, 128.2, 127.4, 59.8, 44.3; HRMS *m/z*, [M + H]⁺ calcd. for C₁₄H₁₄ClN₄S: 305.0628; found: 305.0629.

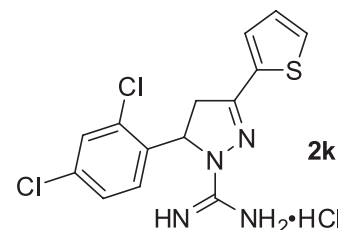
5-(4-Bromophenyl)-3-(thiophen-2-yl)-4,5-dihydro-1*H*-pyrazole-1-carboximidamide hydrochloride (**2j**)



Yellowish solid; mp 265-266 °C; IR (KBr) ν / cm⁻¹ 3269, 3190, 3103, 1611, 1430; ¹H NMR (400 MHz, DMSO-*d*₆) δ 8.01 (bs,

4H, C(NH₂)₂), 7.85 (d, 1H, *J* 5.0 Hz, T-H), 7.60 (d, 2H, *J* 8.3 Hz, Ph-H), 7.56 (d, 1H, *J* 3.6 Hz, T-H), 7.22 (d, 2H, *J* 8.4 Hz, Ph-H), 7.20-7.18 (m, 1H, T-H), 6.03 (dd, 1H, *J* 2.6, 11.1 Hz, H_x), 4.11 (dd, 1H, *J* 11.3, 17.8 Hz, H_m), 3.38 (dd, 1H, *J* 2.7, 17.8 Hz, H_a); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 153.0, 152.9, 138.8, 132.5, 131.9, 131.8, 131.2, 128.1, 127.6, 121.3, 59.7, 44.1; HRMS *m/z*, [M + H]⁺ calcd. for C₁₄H₁₅N₄SBr: 349.0122; found: 349.0141.

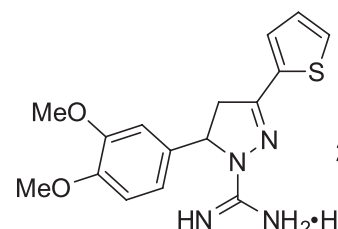
5-(2,4-Dichlorophenyl)-3-(thiophen-2-yl)-4,5-dihydro-1*H*-pyrazole-1-carboximidamide hydrochloride (**2k**)



Yellowish solid; mp 290-293 °C; IR (KBr) ν / cm⁻¹ 3334, 3175, 3103, 1611, 1423, 825; ¹H NMR (400 MHz, DMSO-*d*₆) δ 8.00 (bs,

4H, C(NH₂)₂), 7.85 (dd, 1H, *J* 1.1, 5.0 Hz, T-H), 7.56 (dd, 1H, *J* 1.1, 3.7 Hz, T-H), 7.55-7.53 (m, 1H, Ph-H), 7.48-7.47 (m, 1H, Ph-H), 7.38 (t, 1H, *J* 7.9 Hz, Ph-H), 7.21-7.18 (m, 2H, T-H and Ph-H), 6.02 (dd, 1H, *J* 3.1, 11.2 Hz, H_x), 4.10 (dd, 1H, *J* 11.3, 17.9 Hz, H_m), 3.42 (dd, 1H, *J* 3.1, 17.9 Hz, H_a); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 153.1, 152.9, 142.0, 132.5, 132.0, 131.2, 131.2, 131.0, 128.4, 128.1, 124.1, 121.9, 59.6, 44.2; HRMS *m/z*, [M + H]⁺ calcd. for C₁₄H₁₃Cl₂N₄S: 339.0238; found: 339.0237.

5-(3,4-Dimethoxyphenyl)-3-(thiophen-2-yl)-4,5-dihydro-1*H*-pyrazole-1-carboximidamide hydrochloride (**2l**)



Yellowish solid; mp 287-291 °C; IR (KBr) ν / cm⁻¹ 3298, 3139, 2830, 1618, 1430; ¹H NMR (400 MHz, DMSO-*d*₆) δ 7.84

(dd, 1H, *J* 1.0, 5.0 Hz, T-H), 7.80 (bs, 4H, C(NH₂)₂), 7.56 (dd, 1H, *J* 1.0, 3.6 Hz, T-H), 7.19 (dd, 1H, *J* 3.7, 5.0 Hz, T-H), 6.96 (d, 1H, *J* 8.3 Hz, Ph-H), 6.92 (d, 1H, *J* 1.9 Hz, Ph-H), 6.67 (dd, 1H, *J* 1.9, 8.3 Hz, Ph-H), 5.78 (dd, 1H, *J* 2.9, 11.1 Hz, H_x), 4.05 (dd, 1H, *J* 11.1, 17.8 Hz, H_m), 3.75 (s, 3H, Ph-OCH₃), 3.73 (s, 3H, Ph-OCH₃), 3.40-3.36 (m, 1H, H_a); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 153.2, 152.8, 148.9, 148.7, 132.8, 131.8, 131.8, 131.8, 131.1, 128.2, 116.9, 112.3, 110.1, 60.3, 55.6, 44.4; HRMS *m/z*, [M + H]⁺ calcd. for C₁₆H₁₉N₄O₂S: 331.1229; found: 331.1252.

In vitro cytotoxicity assay (MTT assay)

In vitro cytotoxicity assay was carried out on three different human leukemia cell lines: K562, erythroleukemia cells bearing the t(9;22)(q34;q11)-derived BCR/ABL1 fusion gene; Jurkat, T-cells acute lymphoblastic leukemia (ALL) cells; and RS4;11, B-cell precursor ALL cells bearing the t(4;11)(q21;q23)-derived KMT2A/AFF1 fusion gene. The cells were maintained in RPMI-1640 (Cultilab) supplemented with 10% fetal bovine serum (FBS; Cultilab) and penicillin/streptomycin, at 37 °C and 5% CO₂. For the cytotoxicity assay, compounds **2a-l** were dissolved in

DMSO in order to obtain a stock solution of 20 mmol L⁻¹. Further dilutions were made in complete culture medium immediately before use. K562 and Jurkat cells were seeded at 3 × 10⁴ cells *per well* while RS4;11 cells were seeded at 4 × 10⁴ cells *per well* in a 96-well plate. After that, 20 μL of 10-fold serial dilutions of compounds **2a-l** or vehicle (DMSO at a final concentration of 0.5%) were added to each well, in triplicate, at final concentrations of 0.1, 1.0, 10 and 100 μmol L⁻¹. Culture plates were kept at 37 °C and 5% CO₂ for 48 h, then, cell viability was measured by adding 20 μL of the MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) reagent (Sigma-Aldrich) at 5 mg mL⁻¹. After 4 h, the precipitated formazan crystals were dissolved by the addition of 100 μL of an acid sodium dodecyl sulfate solution (10% SDS, 0.01 mol L⁻¹ HCl). Following overnight incubation, absorbance was measured at 570 nm for the MTT reaction and 620 nm as reference on scanning. Viability was calculated as a percentage of viable cells at different test concentrations relative to the control (vehicle-treated) cells. The concentration of compounds **2a-l** that resulted in 50% inhibition of cell growth was calculated as the half maximal inhibitory concentration (IC₅₀) by constructing a dose-response curve using GraphPad Prism 5 software (GraphPad Software, Inc., La Jolla, CA).

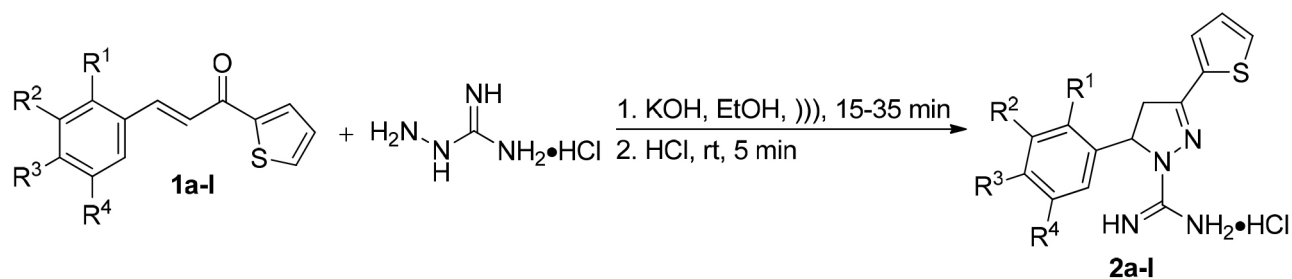
Results and Discussion

Chemistry

The 4,5-dihydro-1*H*-pyrazole-1-carboximidamides (**2a-l**) were synthesized in good yields by the ultrasound-assisted cyclocondensation reaction between compounds **1a-l**, prepared as described in literature,²² and aminoguanidine hydrochloride in the presence of KOH using ethanol as a green solvent (Table 1). The time required for the completion of the reaction depends on the nature of the groups attached to the benzene ring and it was determined by monitoring the consumption of the carbonyl compound by gas chromatography (GC) in intervals of 5 minutes. After the total consumption of the starting materials **1a-l** the reaction solution was acidified with HCl in order to obtain the salts of the products. As showed in Table 1, the reaction tolerate electron-withdrawing and electron-donating groups. In general, substrates that contain electron-withdrawing substituents such as 3-NO₂ (Table 1, entry 4) and 4-CF₃ (Table 1, entry 6) gave products in shorter reaction times but in lower yields.

The structure of the carboximidamides were confirmed by IR, ¹H and ¹³C NMR and HRMS. The IR spectra showed

Table 1. Cyclocondensation reactions between **1a-l** and aminoguanidine hydrochloride in the presence of KOH for the preparation of carboximidamides **2a-l**



entry	Product	R ¹	R ²	R ³	R ⁴	time ^a / min	Yield ^b / %
1	2a	H	H	H	H	20	88
2	2b	OMe	H	H	H	25	80
3	2c	Br	H	H	H	30	70
4	2d	H	NO ₂	H	H	15	65
5	2e	H	H	Me	H	35	80
6	2f	H	H	CF ₃	H	15	62
7	2g	H	H	OMe	H	25	86
8	2h	H	H	F	H	30	73
9	2i	H	H	Cl	H	30	71
10	2j	H	H	Br	H	30	75
11	2k	Cl	H	Cl	H	30	66
12	2l	H	OMe	OMe	H	25	83

^aSonication time; ^byields of the isolated compounds.

sets of absorption bands in accordance with the proposed structures. In the ^1H NMR spectra of the compounds, the characteristic signals of the AMX coupling system between pyrazolyl ring hydrogens, H_a , H_m and H_x , were observed as sets of three doublets of doublets in the δ 3.48-3.24, 4.19-4.04, and 6.27-5.78 ppm regions, respectively. Besides, the formation of the salts was confirmed by the presence of broad singlets in the range of δ 8.14-7.76 ppm due to the presence of four equivalent hydrogens attached to nitrogens atoms in the carboximidamidyl moieties. The analysis of the ^{13}C NMR and HRMS spectra also confirmed the structure of the products.

Although the mechanism of the cyclocondensation reaction between aminoguanidine and α,β -unsaturated ketones was not yet experimentally established, the literature shows a possible explanation based on the hard and soft acids and bases (HSAB) concept and quantum chemical studies on the aminoguanidine.^{17,23} Accordingly, a proposed mechanism is shown in Figure 1. Firstly, KOH neutralizes the aminoguanidine salt to give aminoguanidine in its free form, which can exist as two tautomers. Thus Aza-Michael-type addition from the internal imine nitrogen to the β -position of **1** leads to adduct **I**. In the next step, an intramolecular nucleophilic attack of the primary amino nitrogen to the carbonyl leads to intermediates **II**, which is dehydrated to give the product **III**. Finally, the addition of HCl produces the salt **2**.

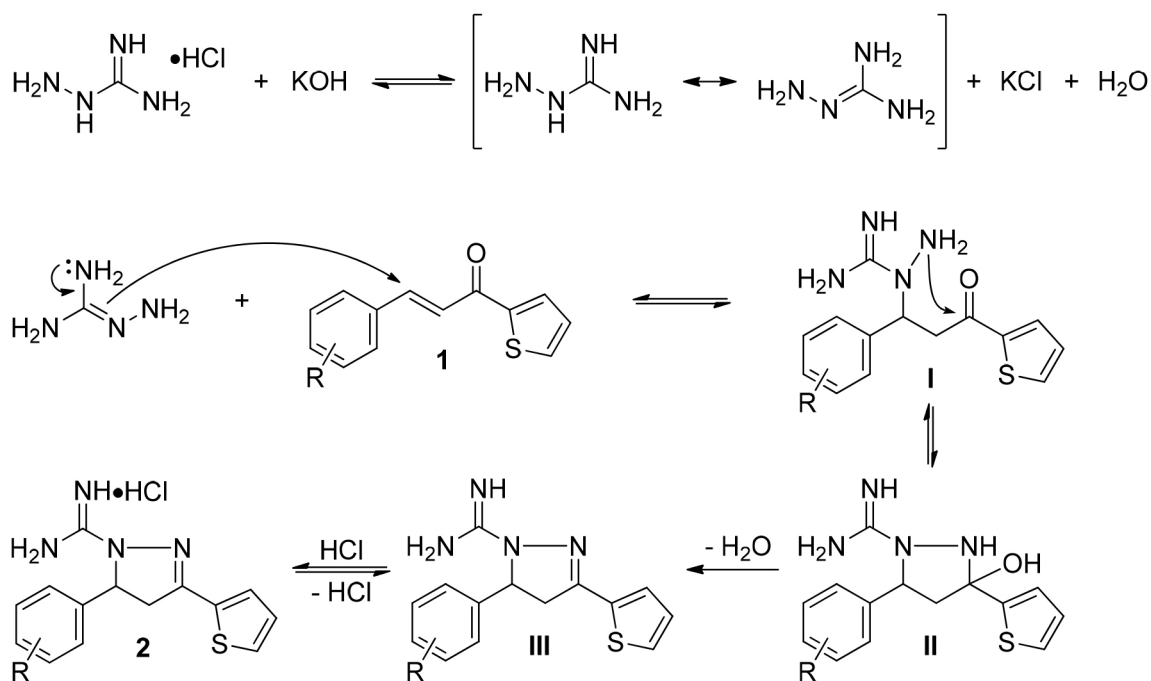


Figure 1. Proposed mechanism for the cyclocondensation reaction between 3-aryl-1-(thiophen-2-yl)prop-2-en-1-ones and aminoguanidine hydrochloride in the presence of KOH.

In vitro evaluation of antileukemia activity

To gain insight on structure-activity relationship, twelve pyrazoline derivatives (**2a-l**) had their cytotoxic activity evaluated against three different human leukemia cell lines (Table 2). Compounds **2b**, **2c**, **2f**, **2i**, **2j** and **2k** showed the best cytotoxic activity, with IC_{50} value around $15 \mu\text{mol L}^{-1}$ for at least one of the three leukemia cell lines analyzed. Compounds **2a**, **2d**, **2e**, **2g** and **2h** showed intermediate cytotoxic activity, with IC_{50} around $40 \mu\text{mol L}^{-1}$. Finally, compound **2l** did not present a significant cytotoxic activity, with IC_{50} value $>100 \mu\text{mol L}^{-1}$.

The *in vitro* cytotoxic activity of pyrazolines is in the same range to that of several anti-leukemia drugs used in the clinic: 6-mercaptopurine (20.9 and $233.4 \mu\text{mol L}^{-1}$, median IC_{50} values for a panel of seven T-cell and fifteen B-cell precursor ALL cell lines, respectively), 6-thioguanine (4 and $11.8 \mu\text{mol L}^{-1}$), dexamethasone (82.7 and $0.19 \mu\text{mol L}^{-1}$), and methylprednisolone (38.6 and $69.8 \mu\text{mol L}^{-1}$).²⁴ In conclusion, the cytotoxic activity of most pyrazolines is comparable to that of drugs used in leukemia treatment.

Conclusions

Ultrasound irradiation was efficient to promote the cyclocondensation reaction between thiophenylchalcones

Table 2. Cytotoxic activity of compounds **2a-1** against different leukemia cell lines

entry	Compound	IC ₅₀ ^a / (μmol L ⁻¹)		
		Jurkat	RS4;11	K562
1	2a	29.5	70.5	21.26
2	2b	10.2	29.4	5.9
3	2c	21.5	20.5	14.7
4	2d	29.5	46.4	33.4
5	2e	34.2	28.7	24.8
6	2f	24.7	17.3	17.4
7	2g	28.6	47.1	40.0
8	2h	42.4	51.5	36.2
9	2i	37.9	16.7	30.4
10	2j	22.0	14.4	23.7
11	2k	21.3	16.2	5.7
12	2l	> 100	> 100	> 100

^aThe drug concentration resulting in a 50% of maximal inhibition of cell proliferation (as measured by MTT staining).

and aminoguanidine hydrochloride in the presence of KOH, furnishing a series of twelve 3-(thiophen-2-yl)-4,5-dihydro-1H-pyrazole-1-carboximidamides in short reaction times (15-35 min) and good yields (62-95%). All compounds prepared were evaluated against the human Jurkat and RS4;11 acute lymphoblastic leukemia cell lines of T- and B-cell origin, respectively, and the K562 myelogenous leukemia cell line. Six compounds presented IC₅₀ values around 15 μmol L⁻¹ and five compounds presented IC₅₀ values around 40 μmol L⁻¹ for at least one of the three cell lines analyzed. One compound was not significantly cytotoxic presenting IC₅₀ value > 100 μmol L⁻¹.

Supplementary Information

Supplementary information (¹H NMR, ¹³C NMR and HRMS spectra and dose-response curves) is available free of charge at <http://jbcs.sbq.org.br> as a PDF file.

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