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#### Ecotoxicology and Human Environmental Health

## Structural features and pro-inflammatory effects of watersoluble organic matter in inhalable fine urban air particles

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#### 14 ABSTRACT

15	The impact of inhalable fine particulate matter (PM <sub>2.5</sub> , aerodynamic diameter $< 2.5 \ \mu$ m) on public
16	health is of great concern worldwide. Knowledge on their harmful effects are mainly due to studies
17	carried out with whole air particles, being the contribution of their different fractions largely
18	unknown. Herein, a set of urban $PM_{2.5}$ samples were collected during day and nighttime periods
19	in Autumn and Spring, aiming to address the seasonal and day-night variability of water-soluble
20	organic matter (WSOM) composition. In vitro analysis of oxidative and pro-inflammatory
21	potential of WSOM samples was carried out in both acute (24 h) and chronic (3 weeks) exposure
22	setups using Raw264.7 macrophages as cell model. Findings revealed that the structural
23	composition of WSOM samples differs between seasons and in a day-night cycle. Cells
24	exposure resulted in an increase in the transcription of the cytoprotective <i>Hmox1</i> and pro-
25	inflammatory genes <i>II1b</i> and <i>Nos2</i> , leading to a moderate pro-inflammatory status. These
26	macrophages showed an impaired capacity to subsequently respond to a strong pro-
27	inflammatory stimulus such as bacterial lipopolysaccharide, which may implicate a
28	compromised capacity to manage harmful pathogens. Further investigation on aerosol
29	WSOM could help to constrain the mechanisms of WSOM-induced respiratory diseases
30	and contribute to PM <sub>2.5</sub> regulations.

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- 32
- 33

### 34 INTRODUCTION

35	Inhalable fine atmospheric particulate matter (PM $_{2.5}$ , aerodynamic diameter < 2.5 $\mu$ m) is
36	of serious concerns in terms of health effects, including cardiovascular diseases, <sup>1,2</sup> airway
37	damages and lung carcinogenesis,2-4 and adverse neurodevelopmental effects.5-7
38	Oxidative stress, genotoxicity, and inflammation have been suggested to be the central
39	mechanisms by which $PM_{2.5}$ may impair the normal cellular physiological/biochemical
40	processes, resulting in tissues damages and, therefore, facilitating the incidence and
41	development of those adverse health outcomes. <sup>2</sup> Lung cells, such as epithelial cells and
42	alveolar macrophages, are the primary targets of $PM_{2.5}$ -induced oxidative damage and
43	pro-inflammatory effects. <sup>8–10</sup> In the past, these effects have been linked to $PM_{2.5}$ mass
44	concentration; $^{9,11}$ however, evidences indicate that $PM_{2.5}$ chemical composition play an
45	important role in the interaction between fine particles and lung cells membrane. <sup>10,12–17</sup>
46	$PM_{2.5}$ is a complex mixture of inorganic and carbonaceous constituents, whose
47	composition depends on the emission sources (natural and/or anthropogenic), formation
48	process ( <i>i.e.</i> , secondary origin), atmospheric processing, and weather conditions. A small
49	set of chemical species have been linked to the oxidative potential and inflammatory

50	impact of ambient PM <sub>2.5</sub> . These include water-soluble metals ( <i>e.g.</i> , Fe, Ni, Cu, Cr, Mn,
51	Zn, V, and Pb), <sup>10,12,17,18</sup> water-soluble ions ( <i>e.g.</i> , SO <sub>4</sub> <sup>2-</sup> , NO <sub>3</sub> <sup>-</sup> ), <sup>10</sup> solvent-extractable
52	organic components, such as polycyclic aromatic hydrocarbons (PAHs) and PAH nitro-
53	derivatives, polychlorinated biphenyls, organochlorine pesticides, and polybrominated
54	diphenyl ethers, <sup>16,18,19</sup> and wood smoke tracers ( <i>e.g.</i> , levoglucosan and galactosan). <sup>19,20</sup>
55	Although not receiving much attention by the air pollution health research community,
56	water-soluble organic matter (WSOM) of ambient $PM_{2.5}$ have been recently also
57	recognized as capable of mediating reactive oxygen species (ROS) generation. <sup>17,21–26</sup> In
58	Northern Hemisphere midlatitudes, this organic aerosol component represent 10 to 80%
59	of the total particulate organics, <sup>27–30</sup> whereas lower percentage values (up to 13%) have
60	been reported for Southern Hemisphere locations. <sup>31</sup> Based on highly informative off-line
61	analytical techniques, such as multidimensional nuclear magnetic resonance (NMR)
62	spectroscopy and high-resolution mass spectrometry, it has been shown that aerosol
63	WSOM consists of a highly diverse suite of oxygenated compounds, including
64	dicarboxylic acids, keto-carboxylic acids, aliphatic aldehydes and alcohols, saccharides,
65	saccharide anhydrides, aromatic acids, phenols, but also amines, amino acids, organic

66	nitrates, and organic sulfates. <sup>31–40</sup> Although being a major fraction of organic aerosols,
67	the way by which the aerosol WSOM is redox-active or exert inflammation remains
68	unresolved. It contributes greatly to this situation the inherent complexity of aerosol
69	WSOM, with different chemical structures and associated physical properties. The use of
70	chemical assays, such as dithiothreitol (DTT), has suggested important contributions of
71	specific aerosol WSOM fractions, namely of isolated hydrophobic fractions (the so-called
72	"humic-like substances" operational concept), <sup>21,22,25</sup> on the oxidative potential of ambient
73	PM <sub>2.5</sub> . However, no evidence exists on which structural components of aerosol WSOM
74	actually induce oxidative stress and inflammation. An enhanced knowledge of the relative
75	contribution of WSOM substructures to the oxidative and pro-inflammatory potential of
76	this organic aerosol fraction would be useful to understand the net impacts of air
77	particulate organic matter on human health.
78	Within this context, this study aims to establish a relationship between the oxidative and
79	pro-inflammatory potential of aerosol WSOM and their atmospheric concentrations and
80	structural characteristics. A set of $PM_{2.5}$ samples were collected at an urban location,
81	during day and nighttime periods, in Autumn and Spring seasons. With this $PM_{2.5}$

82	sampling scheme, it was intended to apportion the seasonal and day-night variability of
83	WSOM structures to their oxidative and pro-inflammatory potential. The compositional
84	features of aerosol WSOM samples were exploited by liquid-state one-dimensional (1D)
85	and two-dimensional (2D) NMR spectroscopy. The effects of aerosol WSOM on nitric
86	oxide (NO) production, ROS cellular levels, transcription of the inflammatory genes <i>II1b</i>
87	and <i>Nos2</i> , as well as activation of associated signalling pathway NF- $\kappa$ B were assessed
88	in Raw264.7 macrophages. The effect on the transcription levels of <i>Hmox1</i> , a central
89	player in detoxification of electrophilic and oxidative stresses, was also evaluated. The
90	biological assays were carried out considering the effect of both acute (24 h) and chronic
91	(3 weeks) exposures to aerosol WSOM, with the latter at maximal theoretical WSOM
92	doses ( <i>i.e.</i> , considering the average human air intake per hour, and the atmospheric
93	water-soluble organic carbon concentrations in day-night cycles). Investigating the
94	potential impact of aerosol WSOM composition on human health is essential, particularly
95	when experiencing serious pollution conditions. This study set the basis for further
96	understanding of the mechanisms of fine aerosol WSOM-induced respiratory diseases
97	and may contribute to the development of targeted $PM_{2.5}$ regulations.

99

#### 100 MATERIALS AND METHODS

101 Aerosol sampling and extraction of WSOM samples

102 The PM<sub>25</sub> samples were collected on a rooftop (*c.a.* 20 m above the ground) at the 103 campus of University of Aveiro (40°38'N, 8°39'W), which is located about 10 km from the 104 Atlantic coast on the outskirts of the city of Aveiro. The sampling site is impacted by both 105 marine air masses travelling from the Atlantic Ocean and anthropogenic emissions from vehicular transport, residential, and industrial sources.<sup>30,32,41</sup> Episodes of increased PM<sub>2.5</sub> 106 107 and WSOM concentrations are common in this area during colder periods and they can 108 last several days,<sup>29,30,41</sup> allowing the collection of enough amount of aerosol WSOM within relatively short periods of time. The PM2.5 samples were collected from 14 to 29 109 110 November 2016 (Autumn) and 27 March to 04 April 2016 (Spring), during the weekdays 111 (*i.e.*, from Monday to Friday) in two distinct time periods: (i) daytime: 9:40 a.m. to 5:30 112 p.m. (Autumn)/6:30 p.m. (Spring), and (ii) overnight: 5:40 p.m. (Autumn)/6:40 p.m. 113 (Spring) to 9:30 a.m. The PM<sub>2.5</sub> samples were collected on pre-fired quartz-fiber filters

114	(20.3×25.4 cm; Whatman QM-A) with an airflow rate of 1.13 m <sup>3</sup> min <sup>-1</sup> . Additional details
115	on aerosol sampling procedure are available in Section S1, in Supporting Information (SI).
116	After sampling, the filter samples were folded in two, wrapped in aluminum foil and
117	immediately transported to the laboratory, where they were weighted and stored frozen
118	until further analysis. The meteorological data recorded during $PM_{2.5}$ collection is
119	available in Table S1 (SI). The determination of organic carbon (OC) and elemental
120	carbon (EC) in each $PM_{2.5}$ sample was performed by means of a Lab OC-EC Aerosol
121	Analyzer (Sunset Laboratory Inc.) following a thermo-optical method, described in section
122	S2 (SI).
122 123	S2 (SI). An area of 315 cm <sup>2</sup> of each quartz filter was extracted with 150 mL of ultra-pure water
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123 124 125	An area of 315 cm <sup>2</sup> of each quartz filter was extracted with 150 mL of ultra-pure water (18.2 M $\Omega$ cm, filter area to water volume ratio of 2.1 cm <sup>2</sup> mL <sup>-1</sup> ) by mechanical stirring for 5 min followed by ultrasonic bath for 15 min. Each final aqueous slurry was filtered through
123 124 125 126	An area of 315 cm <sup>2</sup> of each quartz filter was extracted with 150 mL of ultra-pure water (18.2 M $\Omega$ cm, filter area to water volume ratio of 2.1 cm <sup>2</sup> mL <sup>-1</sup> ) by mechanical stirring for 5 min followed by ultrasonic bath for 15 min. Each final aqueous slurry was filtered through a hydrophilic polyvinylidene fluoride membrane filter (Durapore®, Millipore, Ireland) of

130	each aqueous extract was measured by means of a Skalar (Breda, Netherlands) San++
131	Automated Wet Chemistry Analyzer, based on a UV-persulfate oxidation method.42
132	After the WSOC extraction, and to obtain sufficient amount of WSOM samples for both
133	NMR and biological assay studies, the aqueous aerosol extracts from each sampling
134	period within each season were batched together, on a total of four pooled WSOM
135	samples representative of day and nighttime conditions in Autumn and Spring seasons.
136	Each pooled WSOM sample was further divided into two aliquots of similar volume: one
137	aimed to NMR analysis and the other to biological assay studies. The aliquots were
138	freeze-dried, and the obtained residues (designated as "whole aerosol WSOM sample")
139	were kept in a desiccator over silica gel until further analysis. Additional details on aerosol
140	WSOM samples processing are available in Figure S1, section S3 (SI).
141	
142	Liquid-state 1D and 2D NMR spectroscopy
143	All NMR spectra were acquired using a Bruker Avance-500 spectrometer operating at
144	500.13 and 125.77 MHz for <sup>1</sup> H and <sup>13</sup> C, respectively, and equipped with a liquid nitrogen
145	cooling CryoProbe Prodigy™. All 1D and 2D spectra were run at 295.1 K, and additional

146	details on NMR data acquisition can be found in Section S4, in SI. The dried WSOM
147	samples (7 to 3 mg) were dissolved in deuterated methanol (MeOH- $d_4$ , ~1 mL) and
148	transferred to 5 mm NMR tubes. The identification of functional groups in the NMR spectra
149	was based on their chemical shift relative to solvent (MeOH- $d_4$ ) peak set at $\delta_H$ 3.31 ppm
150	and $\delta_{C}$ 49.0 ppm. The interpretation of the spectral regions and structural assignments
151	were based on the NMR chemical shift data described in the literature for standard
152	organic compounds and for natural organic matter from different environmental
153	matrices, 28, 32, 34, 43, 44 as well as on data generated by NMR simulators software's and
154	databases (namely, Perkin Elmer ChemBioDraw® Ultra 14.0 and nmrdb.org).45
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155	
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155 156 157	Cell culture and treatments Raw 264.7, a mouse leukaemic monocyte macrophage cell line from American Type
155 156 157 158	Cell culture and treatments Raw 264.7, a mouse leukaemic monocyte macrophage cell line from American Type Culture Collection (ATCC number: TIB-71), was cultured in Dulbecco's Modified Eagle

162	overnight and then treated with indicated WSOC concentrations. In prolonged exposure
163	experiments the cells were treated for 21 days, twice a day, with maximal theoretical
164	inhaled doses of WSOC calculated from an mean inspired air volume of 8.64 m <sup>3</sup> per day:
165	for Autumn season experiments, the cells were treated in the morning with 7.8 $\mu g$ of
166	daytime WSOC extract, and at the evening with 16.2 $\mu g$ of nighttime extract; in Spring
167	season experiments, the cells were treated in the morning with 5.6 $\mu g$ of daytime WSOC
168	extract and at the evening with 6.1 $\mu g$ of nighttime extract. The cell treatments in the
169	morning and evening were timed to mimic the schedule of $PM_{2.5}$ samples collection,
170	whereas the cell medium was completely replaced every 2 days.
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171 172 173 174	Nitric oxide (NO) production and <i>in vitro</i> antioxidant activity The production of NO was measured by the accumulation of nitrite in the culture supernatants, using a colorimetric reaction with the Griess reagent as previously

178	WSOM extracts and then 1 $\mu$ g mL <sup>-1</sup> LPS was added following an incubation period of 24
179	h. In prolonged exposure experiments cells were treated for 21 days with WSOM extracts
180	and then stimulated for additional 24h with 1 $\mu g$ mL <sup>-1</sup> LPS. The effect of WSOM extracts
181	on the modulation of LPS-induced cellular oxidative stress was also addressed. Briefly,
182	Raw cells were plated at 0.05x10 <sup><math>^6</math></sup> per well in a µ-Chamber slide (IBIDI GmbH,
183	Germany), allowed to stabilize overnight and then stimulated with 1 $\mu$ g mL <sup>-1</sup> LPS during
184	24 h. The WSOM extracts were added 1 h prior to LPS stimulation. At the end of
185	incubation period, cells were washed three times and then loaded with 5 $\mu M$ $H_2 DCFDA$
186	and 0.5 $\mu$ g mL <sup>-1</sup> Hoechst in Hank's Balanced Salt Solution (HBSS) for 30 min at 37 °C in
187	the dark. Cells were washed three times with HBSS and analyzed with an Axio Observer
188	Z1 fluorescent microscope (Zeiss Group, Oberkochen, Germany) at 63X magnification.
189	
190	Analysis of gene expression by q-PCR

After cell treatment for the indicated times, total RNA was isolated with TRIzol reagent according to the manufacturer's instructions. For analysis of mRNA levels of selected genes, 1 µg of total RNA was reverse-transcribed using the iScript Select cDNA Synthesis

194	Kit and then real-time quantitative PCR (qPCR) reactions were performed using SYBR
195	Green on a Bio-Rad CFX Connect device. The results were normalized using Hprt1 as
196	reference gene and presented as fold change relatively to untreated cells. Primer
197	sequences were designed using Beacon Designer software version 8 (Premier Biosoft
198	International, Palo Alto, CA, USA) and thoroughly tested.
199	
200	Statistical analysis
201	Since aerosol WSOM samples were pooled and analyzed together, it should be
202	mentioned that the cellular responses and PCR data were obtained for each pooled
203	WSOM sample. However, at least three independent biological experiments were carried
204	out for each sample. Results are presented as mean $\pm$ the standard deviation (SD) of the
205	indicated number of experiments. Comparisons between two groups were made by the
206	two-sided unpaired Student's <i>t</i> test and multiple group comparisons by One-Way ANOVA
207	analysis, with a Dunnett's Multiple Comparison post-test. Statistical analysis was
208	performed using GraphPad Prism, version 6 (GraphPad Software, San Diego, CA, USA).
209	Significance levels are as follows: * <i>p</i> <0.05, ** <i>p</i> <0.01, *** <i>p</i> <0.001, **** <i>p</i> <0.0001.

211

#### 212 RESULTS AND DISCUSSION

#### 213 Water-solubility of fine urban organic aerosols

The ambient concentrations of PM<sub>2.5</sub>, OC, EC, and WSOC follow the same seasonal 214 215 trend, with the highest levels being found during Autumn (Table 1, and Figures S2 to S5 216 in SI). This seasonal trend has been quite well documented in this and other 217 regions,<sup>28,30,33,37</sup> although an opposite trend has been observed in North America.<sup>47,48</sup> In 218 Autumn, the median PM<sub>2.5</sub> concentration increased 1.5 fold during nighttime compared 219 with daytime, which is likely due to a lower mixing height and more stable atmospheric 220 conditions during the nighttime as well as increased emissions from residential heating 221 sources. These events could also explain the 3.3 and 3.6 fold increase in OC and WSOC 222 concentrations, respectively, during nighttime. In Spring, the diurnal variations of the 223 median OC and WSOC concentrations were evident, but the median concentration values 224 of these carbonaceous fractions only increased 1.3 and 1.5 fold, respectively, during 225 nighttime, whereas no major difference was observed for the PM<sub>2.5</sub> levels. Interestingly,

226	the OC-day/OC-night, EC-day/EC-night, and WSOM-day/WSOM-night ratios increased
227	from a range of 0.34 – 0.71, 0.14 – 0.72, and 0.40 – 0.65, respectively, between the $27^{th}$
228	and $29^{th}$ of March, to $1.2 - 2.7$ , $0.8 - 1.8$ , and $1.1 - 2.1$ , respectively, between $30^{th}$ of
229	March and 6 <sup>th</sup> of April (Figures S3 to S5, SI). This increase could be explained by a
230	decrease in the OC, EC and WSOC concentrations during nighttime in the second half of
231	the sampling period. During this period, the median values of temperature and maximum
232	wind velocity raised from 13 to 15°C and 2.7 to 4.0 m s <sup>-1</sup> , respectively, leading to a more
233	turbulent atmosphere and less emissions from anthropogenic activities related to house
234	heating and, therefore, to a decrease in the nighttime OC, EC and WSOC concentrations.
235	Furthermore, it is also likely that an additional OC source was present in daytime Spring
236	samples, which cannot be explained solely by changes in primary emissions in view of
237	the decrease of EC concentrations compared with Autumn. Considering that photo-
238	oxidative capacity of atmosphere might be enhanced in Spring, <sup>28,29</sup> in situ production of
239	secondary OC or chemical aging of insoluble primary organics could likely contribute to
240	this additional OC. This enhanced photochemical activity can also explain the higher
241	WSOC/OC ratios of Spring samples compared to those of Autumn samples.

# 244 Day-night variability of aerosol WSOM features

245	Figure 1(A) shows the <sup>1</sup> H NMR spectra of pooled aerosol WSOM samples representative								
246	of day and nighttime conditions in Autumn and Spring seasons. These <sup>1</sup> H NMR spectra								
247	exhibit a remarkable similarity to those of other urban aerosol WSOM samples, <sup>28,30–32</sup>								
248	comprising a complex overlapping profile with broad bands superimposed by a relatively								
249	small number of sharp peaks. For a further understanding of these spectral profiles, a								
250	quantitative integration of the four main regions assigned to different types of non-								
251	exchangeable organic hydrogen [Figure 1(A)] was performed in order to assess the								
252	abundance of each functionality in WSOM samples. As depicted in Figure 1(B), and								
253	regardless of the sampling period, the saturated aliphatic protons (H–C) are typically the								
254	most important component, followed by unsaturated (H–C–C=) and oxygenated (H–C–O)								
255	aliphatic protons, and a less contribution from aromatic protons (Ar-H). Comparison								
256	between daytime and nighttime WSOM samples are expected to vary somewhat within								
257	both seasons, due to the distinctly different emission sources and atmospheric oxidation								

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258	conditions. However, in Spring, very little day-night variability was observed for the NMR									
259	structural signatures identified in the WSOM samples. On the other hand, in Autumn, the									
260	aliphatic H–C structures exhibit a distinct maximum (45%) for day aerosol WSOM and a									
261	minimum (34%) for the night sample. Additional differences between day and night									
262	WSOM samples in Autumn are found in the spectral regions associated with protons									
263	bound to oxygenated aliphatic (H–C–O) and aromatic (Ar–H) structures, whose									
264	contributions are higher for WSOM collected overnight (29% and 11%, respectively) than									
265	in WSOM collected in daytime (19% and 6.8%, respectively). The occurrence of strong									
266	H-C-O and aromatic signatures overnight in Autumn may be associated with the									
267	contribution of fresh biomass burning emissions for house heating under low air									
268	temperature conditions. <sup>28,30,31</sup> Moreover, the presence of an intense sharp resonance at									
269	$\delta$ <sup>1</sup> H 5.3 ppm [Figure 1(A)] attributed to protons bound to anomeric carbons [O–C(H)–O],									
270	such as those of anhydrosugars (e.g., levoglucosan and mannosan), which are known									
271	molecular markers of wood burning emissions, 30,32 further confirms the presence of									
272	smoke particles during this period.									

# <FIGURE 1 here>

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274	The compositional day-night variability of aerosol WSOM samples within the two seasonal											
275	periods were further ascertain using 2D NMR spectroscopy (Figures S6 to S8, in SI). The											
276	<sup>1</sup> H- <sup>13</sup> C HSQC NMR spectra of all WSOM samples reveal several important <sup>1</sup> H- <sup>13</sup> C											
277	correlations in three major regions of chemical environments (Figures S6 and S7), but											
278	with very different relative intensities: aliphatic ( $\delta_H$ 0.4–3.6 ppm/ $\delta_C$ 10–45 ppm,											
279	represented by H–C and H–C–C=), <i>O</i> -alkyl ( $\delta_H$ 3.6–6.0 ppm/ $\delta_C$ 50–107 ppm, including											
280	anomeric carbons), and aromatic ( $\delta_H$ 6.5–8.5 ppm/											
281	$\delta_C$ 107–160 ppm) regions. The structural assignments of the 2D NMR cross peaks within											
282	these three chemical shift areas were further carried out based on spectral data from											
283	previous 2D NMR studies of aerosol WSOM samples.30,32,34,49 The main structural											
284	findings are shown in the expanded aliphatic, $O$ -alkyl, and aromatic regions of the <sup>1</sup> H- <sup>13</sup> C											
285	HSQC NMR spectra in Figures S9 to S20 in SI. Overall, 15 polyfunctional aliphatic and											
286	aromatic substructures were identified in this study as being common to all aerosol											
287	WSOM samples, whereas 4 aromatic substructures were typical of aerosol WSOM											
288	sample collected overnight in Autumn (Figure S21, in SI). The presence of these 4 typical											
289	aromatic substructures [structures (18) to (21) in Figure S21] in overnight Autumn WSOM											

290	sample confirm the notable influence of biomass burning emissions into the aerosol									
291	WSOM characteristics during this period. <sup>30,32</sup> Five carbohydrate-like structures [structures									
292	(11) to (15), in Figure S21] were consistently found in all aerosol WSOM samples, but									
293	particularly more prominent in samples collected overnight in Autumn. Except for									
294	trehalose [structure (13) in Figure S21], usually referred as a tracer for the resuspension									
295	of surface soil and unpaved road dust, <sup>50</sup> the presence of the anhydrosugars levoglucosan									
296	and mannosan [structures (11) and (12), respectively, Figure S21] and disaccharides									
297	maltose and sucrose [structures (14) and (15), respectively, Figure S21] further confirm									
298	that biomass burning is an important contributor to aerosol WSOM collected overnight in									
299	Autumn. <sup>30,32</sup> The presence in all WSOM samples of NMR fingerprints assigned to DMA <sup>+</sup> ,									
300	DEA <sup>+</sup> and MSA [structures (7) to (9) in Figure S21], and terephthalic acid [structure (16)									
301	in Figure S21] also pinpoint to the contribution of, respectively, marine organic aerosols <sup>30</sup>									
302	and oxidized aromatic hydrocarbons from urban traffic emissions <sup>30,32</sup> to aerosol WSOM									
303	samples. Despite relatively uniform seasonal distribution of aliphatic structures among all									
304	aerosol WSOM samples, there are actually important differences in the aromatic and ${\cal O}$ -									
305	alkyl composition of nighttime aerosol WSOM samples in Autumn [summarized in Figure									

306	1(C)]. These differences in WSOM chemical composition are expected to exert dissimilar										
307	contribution on the oxidative and pro-inflammatory effects of air organic particles.										
308	Cytotoxicity of aerosol WSOM and impact on macrophages ROS and NO production										
309	Understanding the interaction between airway cellular populations and atmospheric air										
310	particles, as well as the mechanisms through which their constituents cause inflammation										
311	and cellular redox imbalances is of major importance. Therefore, the cytotoxicity of										
312	aerosol WSOM over macrophages was firstly assessed in the WSOC concentration range										
313	of 1 to 100 $\mu g$ mL $^{\text{-1}},$ allowing to conclude that none caused a significant decrease in cell										
314	viability (Section S7, Figure S22). Treatments with LPS or LPS + WSOM are also devoid										
315	of significant impact on macrophages viability (Section S7, Figure S23). As NO is a key										
316	molecule in inflammation and to the macrophage capacity to destroy invading pathogens,										
317	the ability of WSOM to induce NO production and to modulate the LPS-induced										
318	production in these cells was also analyzed. As shown in Figure S24(A), Section S8, day										
319	and night WSOM Autumn samples at higher concentrations slightly induce NO										
320	production. This is in agreement with previous reports showing small increases in NO										
321	release by macrophages treated with particle or water-soluble fraction of $PM_{2.5}$ . <sup>51,52</sup>										

322	Furthermore, it can be perceived a different trend between the two WSOM samples: while								
323	the day Autumn WSOM extract causes a concentration dependent increase in NO								
324	production, the night extract has an opposite behavior. We hypothesize that the night								
325	Autumn WSOM may contain organic compounds that induce NO, but also other								
326	constituents that counteract this effect. Regarding the effects of Spring WSOM samples,								
327	only day samples significantly induce NO release [Figure S24(A)]. Discrepancies between								
328	the biological effects of $PM_{2.5}$ and its WSOM fraction collected during day or night are still								
329	poorly documented, but the impact of seasonal variations in $PM_{2.5}$ chemical composition								
330	( <i>e.g.</i> inorganic ions, elements and PAHs) and cytotoxicity has been extensively								
331	covered. <sup>53–55</sup>								
332	Interestingly, the effects of WSOM samples on the LPS-induced NO production was								
333	significantly inhibited by all the samples at the concentration of 75 $\mu g$ mL $^{-1}$ [Figure								
334									
	S24(B)]. This effect was not due to a direct NO scavenging activity (Figure S25), but								
335	S24(B)]. This effect was not due to a direct NO scavenging activity (Figure S25), but instead it may be the result of a modulation of inducible nitric oxide synthase (iNOS)								
335 336									

338	production in treated macrophages (Figure 2). These results differ from several reports
339	where atmospheric particulate matter (PM) were shown to exert pro-oxidative effects. <sup>17,56–</sup>
340	<sup>58</sup> These discrepancies may be explained by the differences in the material actually used
341	since most of the available studies employ the whole air PM, or instead it could be due to
342	considerable differences in the chemical composition of WSOM samples. The pro-
343	oxidative characteristics of PM can be therefore mainly attributed to direct physical
344	interactions with the particles themselves, or to their content in elements such as Cu, Cr,
345	Pb, Co, Ni, or even WSOC. <sup>17,58</sup> It should be mentioned that the concentration of water-
346	soluble elements and metals in the aqueous $PM_{2.5}$ extracts here studied are lower than
347	those identified in the literature as exerting biological effects (Section S9, Table S2), thus
348	supporting the assumption that the effects observed in this work are mainly due to an
349	exposure to aerosol WSOM.
350	<figure 2="" here=""></figure>
351	The inhibitory effect on the capacity of macrophages to produce ROS reported in this
352	study is paradoxical: it may be beneficial given that it limits an inflammatory reaction;

353 however, it can compromise the efficacy of macrophages to destroy harmful

354 microorganisms during an infection by limiting the oxidative burst.

355

356

357 Effects of aerosol WSOM on the modulation of macrophage inflammatory status 358 Based on the results obtained thus far, the potential anti-inflammatory activity of aerosol 359 WSOM samples was also explored. For this, the impact of WSOM samples on the 360 transcription Nos2 (pro-inflammatory Hmox1 of 111b. genes) and (anti-361 inflammatory/cytoprotective), or in the modulation of their LPS-induced transcription was 362 evaluated. As shown in Figure S26, Section S10, Autumn and Spring WSOM samples 363 present a similar profile, although with different magnitudes. Hmox1 transcription is 364 induced either by day or night samples in a dose dependent way. This observation agrees with previous reports where PAHs present in PM (fine and coarse) were shown to induce 365 366 *Hmox1* transcription.<sup>2,59,60</sup> *Hmox* is a detoxifying enzyme that is expressed in response to oxidative and electrophilic stresses as the result of the activation of the Keap/Nrf2 367 signaling pathway.<sup>61</sup> Therefore, the observed increases indicate that WSOM samples contain 368

369	electrophilic compounds, of which substructures (1) and (8) to (10) in Figure 1(C) might be									
370	suitable candidates.									
371	Regarding the modulation of <i>II1b</i> and <i>Nos2</i> transcription, once again Autumn and Spring									
372	WSOM samples presented a similar profile, with moderate increases being induced									
373	(Figure S26). These results are in agreement with previous studies where the exposure									
374	to $PM_{2.5}$ was shown to increase the transcription of <i>II1b</i> and <i>Nos2</i> genes in the same cell									
375	model, causing a moderate pro-inflammatory effect. <sup>60,62</sup> In this study, when comparing									
376	day and night WSOM samples, there are clear differences: while for day WSOM, the									
377	increase in samples concentration is followed by an increase in the transcription of									
378	referred genes, the opposite occurs for night samples (Figure S26). This explains the									
379	lower production of NO in cells treated with higher concentrations of night WSOM									
380	samples [Figure S24(A)]. Regarding the cells treated with LPS, although all the tested WSOM									
381	samples have slight pro-inflammatory properties, they downregulate the LPS-induced									
382	inflammatory state by increasing <i>Hmox1</i> and reducing <i>Il1b</i> and <i>Nos2</i> transcription (Figure 3).									
383	Similar profiles were found when the levels of these proteins were analyzed by western blot									
384	(Section S11, Figure S27). The increase in <i>Hmox1</i> transcription and respective protein expression									
385	in addition to the intrinsic antioxidant characteristics of the WSOM samples support their strong									
386	capacity to prevent LPS-induced oxidative stress. In turn, the decrease in the LPS-induced Nos2									

387 transcription (Figures 3) resulted in a downregulation of the iNOS protein levels (Figure S27), 388 which can explain the observed decrease of LPS-induced NO production in cells pre-treated with 389 WSOM extracts. This capacity to limit LPS-induced inflammatory status was not due to 390 impairment of NF-kB nuclear translocation (Section S12, Figure S28). Given that MAPKs signaling pathways were also shown to modulate PM-induced inflammation,<sup>63,64</sup> the effects of 391 392 WSOM extracts on the phosphorylation levels of p38, JNK and ERK (Section S13, Figure S29) 393 were also assessed. While ERK was not affected and JNK marginally activated, LPS-induced 394 activation of p38MAPK signaling cascade was downmodulated by exposure to extracts, 395 particularly to nighttime WSOM samples (Figure S29). This impairment of p38 MAPK activation 396 may be in part responsible for the observed decrease in LPS-triggered inflammatory status. 397 Additionally, contributing to this impaired capacity of macrophages to mount an adequate 398 inflammatory response, the extracts may be causing alterations in the cellular cytoskeleton, 399 impairing the traffic of secretory vesicles and the release of cytokines, in a process similar to the 400 one recently described by Longhin and collaborators.<sup>65</sup> 401 <FIGURE 3 here>

Given that all experiments were performed with relatively high WSOC concentrations
within an acute exposure setup, it was decided to analyze the effects of chronic exposure
by culturing cells during 3 weeks with maximal theoretical aerosol WSOM doses. For this,
the average human air intake per hour and the atmospheric WSOC concentrations during
day and nighttime conditions were taken into account for administrating maximal

407	theoretical WSOC quantities in day-night cycles. As demonstrated in Figure 4(A), cells
408	exposed to aerosol WSOM exhibit a slight increase in the transcription of Hmox1, II1b,
409	and Nos2 genes. When these cells chronically exposed to WSOM samples were then
410	treated with LPS, the <i>Hmox1</i> and <i>Nos2</i> slightly increased, but the mRNA levels of <i>II1b</i>
411	were significantly down modulated [Figure 4(B)]. While this small increase observed in
412	<i>Nos2</i> transcription does not reach statistical significance it reveals an opposite tendency
413	to the one observed in acute experiments. We hypothesize that this may be mainly due
414	the highly different applied WSOC concentrations and to the presence in the extracts of
415	pro and anti-inflammatory compounds. In acute experiments the concentration of
416	compounds with anti-inflammatory properties may reach a level enough to impair LPS-
417	triggered signaling pathways, namely those regulating Nos2 transcription. In contrast, in
418	prolonged exposure experiments, the cells were continuously treated with very low
419	amounts of WSOM extracts and the compounds with pro-inflammatory properties may
420	prime cells rendering them more responsive to posterior LPS stimulation. Therefore,
421	prolonged exposure to even very small concentrations of water-soluble organic

422 compounds present in PM<sub>2.5</sub>, can cause a decrease in the capacity of macrophages to

423 respond to a subsequent inflammatory stimulus.

424 <FIGURE 4 here>

425 In summary, the set of data provided by 1D and 2D NMR analysis indicates that the fine 426 aerosol WSOM samples hold similar functional groups; however, they differ in terms of 427 their relative distribution both between seasons and in a day-night cycle. This study also 428 highlights that the compositional features of aerosol WSOM samples correlates with their 429 ability to induce a moderate inflammatory status in macrophages, which at long-term may 430 compromise their capacity to mount an effective inflammatory response required to 431 manage harmful pathogens. Therefore, continuous and prolonged exposure to aerosol 432 water-soluble organic compounds could result in increased susceptibility to respiratory 433 infections. For a better understanding of the overall ability of fine aerosol WSOM to exert pro-inflammatory effects, further studies should be conducted involving a larger data set 434 435 of different WSOM samples, while effectively segregating the contributions from different WSOM constituents. 436

439 ASSOCIATED CONTENT

440	Supporting Information. Experimental procedure for PM <sub>2.5</sub> sampling, meteorological										
441	information collected during each sampling campaign, experimental details for OC and										
442	EC analysis, average ambient concentrations of the main aerosol carbon fractions, 2D										
443	NMR data acquisition and 2D NMR spectral assignments of the aerosol WSOM samples,										
444	water-soluble elements and metals concentrations in PM <sub>2.5</sub> samples, impact of aerosol										
445	WSOM samples on macrophages viability, NO production and scavenging activity,										
446	modulation of macrophage inflammatory status, protein levels of iNOS, HMOX1, and IL-										
447	$1\beta$ , activation of pro-inflammatory NF- $\kappa$ B signaling pathways, and modulation of MAPKs										
448	signaling pathways. This material is available free of charge via the Internet at										
449	http://pubs.acs.org.										

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- 457 The manuscript was written through contributions of all authors and all authors have given
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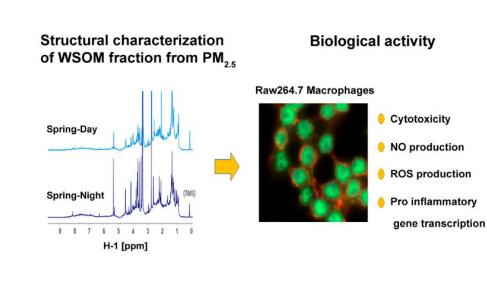
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**TOC Art** 



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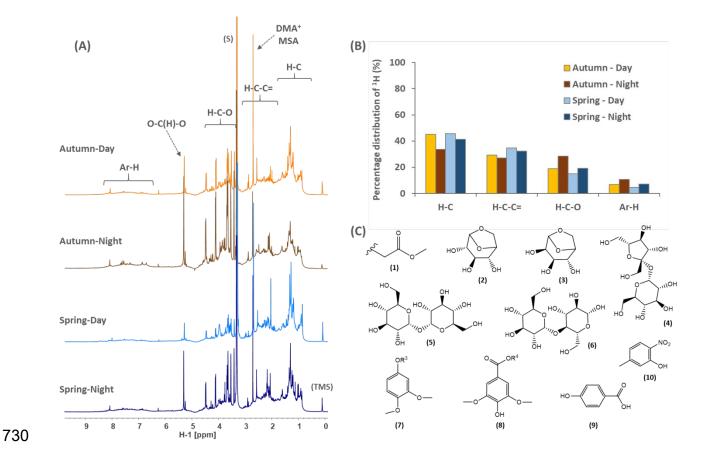
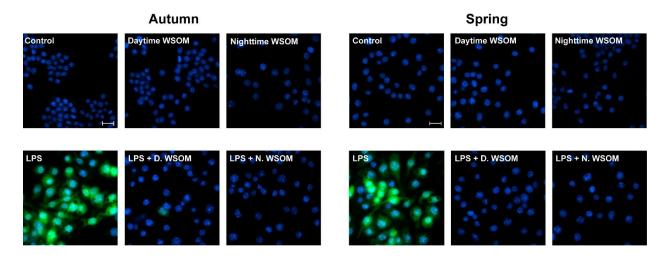


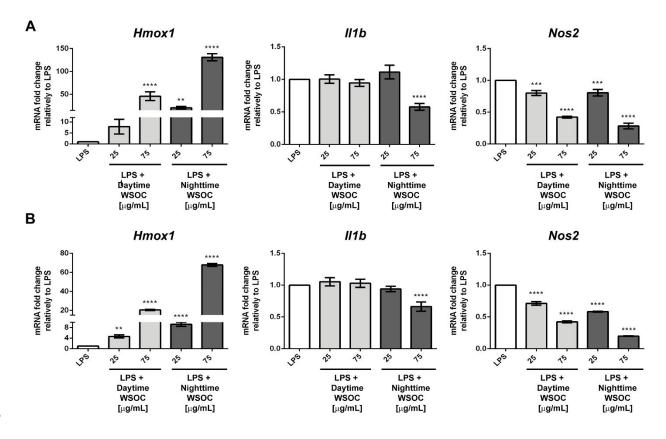
Figure 1. (A) Liquid-state <sup>1</sup>H NMR spectra of aerosol WSOM samples representative of day and nighttime conditions in Autumn and Spring seasons, (B) percentage distribution of <sup>1</sup>H NMR in each aerosol WSOM sample, and (C) aliphatic, carbohydrate and aromatic substructures identified in aerosol WSOM collected overnight in Autumn. See Figures S9 and S10 (in SI) for spectral assignments and identity of aromatic substituents (R<sup>3</sup> and R<sup>4</sup>). Four spectral regions are identified in <sup>1</sup>H NMR spectra (A): H–C, H–C–C=, H–C–O, and Ar–H. NMR resonances assigned to dimethylammonium (DMA<sup>+</sup>), methanesulfonic acid (MSA), and

- 738 protons bound to anomeric carbons [O–C(H)–O] are also identified. Additional resonance
- signals: solvent (S) MeOH- $d_4$ , and tetramethylsilane (TMS) 0.03% (v/v).

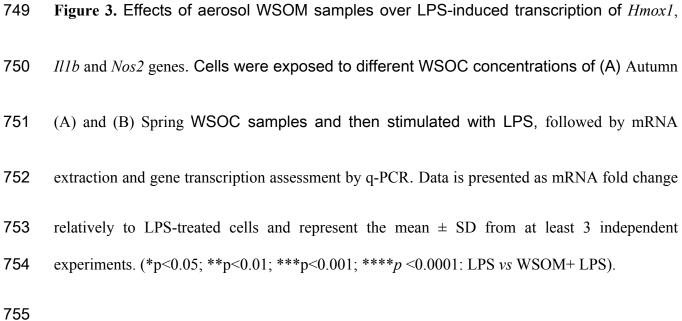


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**Figure 2.** Effect of aerosol WSOM samples on macrophage ROS production and modulation of LPS-induced oxidative stress. Cells were cultured in the indicated conditions and the ROS production was assessed with H<sub>2</sub>DCFDA (green), a ROS-sensitive fluorescent probe. Hoechst (blue) was used to label the nuclei. Images representative of different fields were acquired at a magnification of 63x (scale bar =  $20 \mu m$ ).



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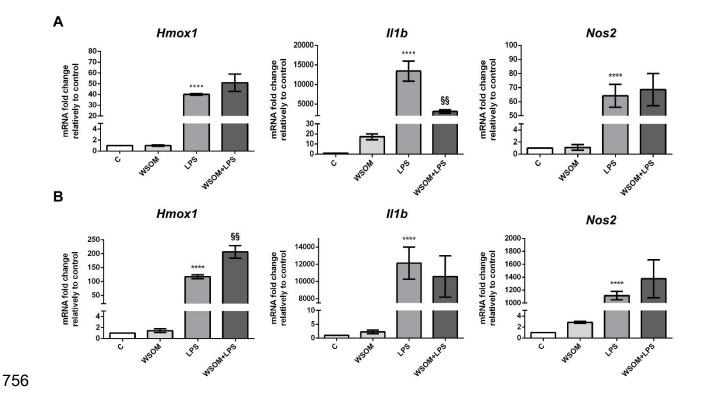
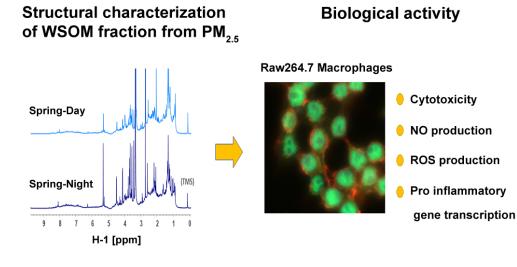


Figure 4. Impact of prolonged exposure to maximal theoretical doses of aerosol WSOM
samples in the (A) gene transcription and (B) LPS-induced gene transcription of *Hmox1*, *IIIb*, and *Nos2*. Cells were exposed for 3 weeks to maximal theoretical quantities of WSOC
extracts, followed by mRNA extraction and gene transcription assessment by q-PCR. Data is
presented as mRNA fold change relatively to control or to LPS-treated cells and represent the
mean ± SD from at least 3 independent experiments. [\*\*\*\*p<0.0001: Control (C) *vs* treatments;
\$%p<0.01: LPS *vs* WSOM+LPS].

**Table 1.** Ambient concentrations (maximum – minimum; median) of urban PM<sub>2.5</sub>, OC,

767 E	C, WSOC, and WSOC/OC in each sampling perio	bd.
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Season	Sampling Period	ΡΜ <sub>2.5</sub> (μg m <sup>-3</sup> )	ОС (µg C m <sup>-3</sup> )	EC (μg C m <sup>-3</sup> )	WSOC (μg C m⁻³)	WSOC/OC (%)
A <b>t</b>	Day	60 – 14; 32	14 – 1.6; 4.1	2.2 – 0.3; 1.3	5.6 – 0.2; 1.3	39 – 12; 34
Autumn	Night	75 – 10; 47	20 – 1.5; 14	5.0 – 0.3; 3.0	6.5 – 0.5; 4.7	34 – 29; 33
	Day	30 – 5.1; 21	6.2 – 0.75; 3.3	1.2 – 0.0; 0.4	2.7 – 0.43; 1.1	57 – 32; 42
Spring	Night	29 – 3.2; 22	6.3 – 1.3; 4.3	1.4 – 0.3; 0.9	2.6 – 0.54; 1.6	44 – 35; 41



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