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# LIGNIN AND CARBON ISOTOPES COMPOSITION OF SU PARTICULATES BY CAPILLARY GAS CHROMATOGRAPHY OXIDE OXIDATION PRODUCTS

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#### ABSTRACT

Lignin are high molecular weight phenolic polymers vascular plants. As a result of their natural abundal microbial degradation,  $\operatorname{col}\nolimits$  gnminonlay efoads ad in soil and  $\operatorname{col}\nolimits$ matter. The aim of this study was to identify the  $c_1$ particulate fraction in the L-basearlis Baenadch wahred e Lanedicat lignin materials.s Logaracterized by oxidative degrad chosen as the most suitable oxydizing adgeent tet to phroend of extracted with hexane and analyzed by capillary gas provides high sensitivity and precision required for the identity and precision required for the identity of  $\sim$ trace levels of lignin in seawater. A suite of up to relative concentration and plant tissue sopuence esdominator

Keywordisgnin, CuO oxidation, Particulate matter

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## 1. INTRODUCTION

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materials in aquatic a and te vegeta<sup>t</sup>ion

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Organic matter in the Lignin, the second most ab environment, significantl<sub>) naturally</sub> occurring polymer after dynamics of microalgae thr<sub>is</sub> an exclusive and of variabilita<sup>[38</sup>t] bodid is and macromolecule found in vast of organic materials ente<sub>plant</sub>? <sup>28,29</sup> The ligheminved phenols partly produced by some  $organisms,h$  saus phytopl macroalgae and chemoautoplant tissuensd environmental This organic material will degradation with d<sup>[19]</sup> fe Neons molecular level has been widely practice in the set of the Meons molecular level has been widely provided and the set of t terrestrial plants produce a variety of geochemical applica resistant to microbial deg <sub>biomarker</sub> lignin found innts seidimofeten explore the entry of ter the reactivity of organic matter an exclusive and stable p that was ained through the CuO ox proc<sup>[de]</sup>san be dus o analyze lignin  $m \cdot \frac{16}{6}$  $\frac{16}{6}$  $\frac{16}{3}$ ,  $\frac{24}{11}$  analysis at the approach in investigation source and procesoscimeum of the voor

sediment and soil, and at the study of limnology and oceanography<sup>[12,21]</sup>. The  $\delta^{13}C$ compositions of particulate organic carbon (POC) have been investigated in a range of sedimentary environments<sup>[5,8,19,25]</sup>; while carbon isotopic (especially  $^{13}$ C) data on lignin phenols in marine sediments remains sparse<sup>[3,23]</sup>. Therefore, any changes in environmental conditions and vegetation can be deduced from the composition of the organic compounds.

Identification and quantification of eight different lignin structural phenols were hexane, accomplished by comparison of mass spectral sample response to an internal ethyl vanillin standard and an external calibration curve of authentic lignin phenolic  $com $pounds^{[2]}$ . Alkaline CuO oxidation of$ lignin yields a suite of phenols that are Suspended grouped based on plant taxonomy, and that are used to identify woody and nonwoody tissues of vascular plants $[14,27]$ . The concentration of CuO lignin monomers relativized to the total of each variety were evaluated by principal components analysis to identify broad differences in lignin composition.

 $^{13}$ C explain the behavior of transportation The objective of this study was to landfill into the sea using lignin phenol biomarkers combined with carbon isotopes  $(\delta^{13}C$  values) of the suspended particulate fraction in to the Losari Beach and Lae-lae Island.

# **2. MATERIAL AND METHODS**

### **2.1 Materials**

Calibration compounds, CuO, glucose, Fe( $NH_4$ )<sub>2</sub>(SO<sub>4</sub>).6H<sub>2</sub>O, Ethyl vanillin, n-N,O-bis(trimethylsilyl)trifluoro  $(BSTFA)$  mixed with  $1\%$ trimethylchlorosilane (TMCS) and pyridine.

# **2.2 Study areas sampling and sample preparation**

particulate matter was collected from the Losari Beach and Lae-lae Island, Makassar, South of Sulawesi (Fig. 1). Five liters of water were collected using a niskin bottle at the depth of 5 m below sea surface, refrigerated and transported to the laboratory where they were passed through pre-weighed GF/F 0.7 µm. The filters were dried (60  $^{\circ}$ C,  $\sim$ 24 h) and reweighed, vielding



**Figure 1.** Sampling locations in the Losari Beach (Red colour) and the Lae-lae Island (green colour)

200-500 mg of suspended particulate matter (SPM). This sample was analysed for stable carbon isotope composition and lignin content.

# **2.3 Elemental and carbon isotope analyses**

Organic carbon (OC) and total nitrogen (TN) were measured using an elemental analyzer (Thermo Fisher Scientific) after acid treatment with 1 N HCl until no bubbles were observed. The sample pre-treatment steps for carbon isotope analysis. Aliquots of powdered suspended particulate matter from the acid treatment were assayed using a carbon analyzer and/or elemental analyzer. Analyses were carried out using an EA-IRMS (elemental analyzer–isotope ratio mass spectrometry). Carbon isotopic compositions are expressed as  $\delta^{13}C$ , in parts The pr thousand (‰), relative to the VPDB standard.

**2.4 Lignin phenol analysis**

Eight lignin phenols were quantified for (BSTFA) all samples and included three vanillyl

phenols (vanillin, acetovanillone and vanillic and three syringyl phenols (syringaldehyde, acetosyringone and syringic acid). Additionally, two cinnamyl phenols (p-coumaric acid and ferulic acid) were quantified solely for the freshwater samples.

Lignin analysis was carried out using the cupric oxide (CuO) oxidation method  $[10]$ , Table 1. Precisely 0.1 g sediment of SPM samples,  $0.5$  g Fe(NH<sub>4</sub>)<sub>2</sub>(SO<sub>4</sub>).6H<sub>2</sub>O and 1.0 g CuO powder were added with 2 N NaOH into Teflon containers with stainless steel jackets at 175 $\mathrm{^{\circ}C}$  in a drying oven for 3 h and manually shaken every hour. The oxidation products are cooled then added 10 ml of 1 N NaOH and centrifuged. This step was repeated twice. The supernatants were pooled and acidified to pH 1 with 6 N HCl. The supernatants were then extracted three times with 10 ml hexane.

Excess solvent dried off with a gentle stream of  $N_2$ . The product was diluted with equal volume of pyridine and the silylating reagent bis-trimethylsilyltrifluoroacetamide with  $1\%$  trimethylchlosilane (TMCS) (as catalyst). The product, with

<b>Original</b> <b>Technique</b>	<b>Modification</b>	<b>Reference for modification</b>	<b>This Study</b>	<b>Notes</b>
N/A	to reaction vessel	Addition of glucose Hernes <i>et al</i> (2002); Louchouarn <i>et</i> Glucose <i>al</i> (2010); Kaiser <i>et al</i> (2012); Jex <i>et al</i> $(2014)$		superoxidation Minimizes effects in low organic matter
Ethyl vanillin as Cinnamic acid quantification standard	quantification (internal standard)	as Opsah et al (1999); Hernes and Ethyl vanillin Benner (2002); Louchouarn et al (2010)		Eliminates steps, significant ethyl vanillin can be lost during laboratory processing in samples with limited matrix
$\rm ^{o}C$ reaction <i>temperature</i> )	<b>Oksidation at 150</b> Oksidation at 155 Goni and <b>vessel</b> vessel temperature) (2012)	(external $\degree$ C (internal reaction Louchouarn <i>et al</i> (2010); Loh <i>et al</i> (external reaction representative	Hedges (1992); Oksidation at 175 <sup>°</sup> C More bomb Teflon temp)	accurate and temperature control
Ethyl extraction	eter Ethyl extraction	acetate Goni and Montgomery (2000); Hexan extraction Louchouarn et al (2010); Spencer <i>et al</i> (2010); Jex <i>et al</i> (2014)		Substitutes highly explosive for safer reagent, reagent slightly higher extraction efficiency
with Flame GC GC - <b>Ionization Detector</b> Spectrometry	with Detector	Mass Opsahl and Benner; Staniszewski GC <i>et al</i> (2001); Louchouarn <i>et al</i> Spectrometry Detector eliminates co-elution problems (2010)	with	Mass Increased sensitivity and

Table 1. Improvements in lignin phenol methodology from the original technique<sup>[10]</sup>

silylating reagent, was derivatized by heating at 90 $\degree$ C for 10 min and then allowed to cool. Immediately after cooling, the derivatized sample was analyzed for lignin-derived phenols using a gas chromatography mass spectrometer (GC–MS). Hewlett–Packard 5890 fitted with an SE30 capillary column (30 m, 0.25mm internal diameter, SUPELCO). The carrier gas was helium. The temperature of the GC was increased from 100 °C to 200 °C by 5 °C per minute and was  $10.2$ held for 10 min. The temperature was then increased from 200  $^{\circ}$ C to 300  $^{\circ}$ C by 20 $^{\circ}$ C per minute and was held at 300 °C for min. Both injector and detector temperatures at  $300 \degree C$ . The equilibration time was 2 min, and the split ratio was 50:1. The average reproducibility of individual lignin phenols based on duplicate analyses of a same sample was 10% (coefficient of variation).

# **3. RESULTS AND DISCUSSIONS 3.1 Elemental and carbon**

The magnitude of the particulate organic material from the landfill, has helped in tracking the extension of waste on the coast of Losari beach and Lae-lae islands. Percentage of Ctot and Ntot on suspense particle materials at Losari beaches is 0.602 and 0.033, while on Lae-lae island is 0.685 and 0.041. Ctot percentages are more abundant than Ntot, reflecting the situation of most terrestrial organic matter emptying into the ocean.

Spatial variation has implied that the distribution of organic matter from anthropogenic activities has more significant effect than natural processes. The value of C/N ratio in Losari beach and Lae-lae islands are 18.1 and 19.3, respectively, indicated that the organic material is derived from terrigenous. This is due to the inorganic particulate organic material that survived in deep transport to the observation distance,

prolonging the pre-oxidized and more resistant to microbial degradation, thus providing the possibility in detecting the contribution of terrigenous organic matter.

Use of  $\delta^{13}$ C values to distinguish sources of organic matter such as marine and significant differences were observed for the suspended samples. The  $\delta^{13}C$  values of Losari Beach and the Lae-lae Island ranges from -7,861‰ to -9,658‰ and -8,762‰ to - 10,231‰. The  $\delta^{13}$ C values suggested that there is a conservative mixing between terrestrial and aquatic sources derived organic matter throughout the entire core. Therefore, the  $\delta^{13}$ C values of SPM in the Losari beach and the Lae-lae Island indicate a small contribution of aquatic plant species.

## **3.2 Lignin-derived phenols**

Eight "characteristic" lignin-derived phenolic monomers were detected in high concentrations in the suspended particulate fraction in the Losari Beach and Lae-lae Island, reflecting both the high abundance of lignin as a component of terrestrial plant biomass and the preferential accumulation of woody plant fragments (which have a high lignin content) from the estuary.

Contributions of three lignin index phenols (V, S and C) indicating that influxes of the lignin. V-phenols were the most dominant CuO-oxidation products of lignin, followed by S-phenols and C-phenols in the lower parts of the core. In marine samples, the cinnamyl phenol concentrations are not usually reported because of significant nonlignin sources $^{[13,26]}$ .

Lignin phenol ratios (S/V and C/V values) might be changed prior to deposition by a selective degradation of lignin structural units during the transfer of terrigenous organic matter from the water column to sediment deposits<sup>[16,17]</sup>. Syringyl

phenols (S) are exclusively found in angiosperms and cinnamyl phenols (C) are solely found in non-woody tissue, so ratio values of these phenols to ubiquitos vanillyl phenols (V) can differentiate relative contribution OM sources from angiosperm and gymnosperm plant types (S:V) and non woody and woody tissues  $(C:V)^{[26]}$ . The C/V values higher than 0.20 and S/V values higher than 0.40 are indicative of non-woody and angiosperm tissues<sup>[6,7,11]</sup>. Hence, the S/V  $(0,14)$  and  $C/V$   $(0.01)$  ratios in the Losari beach and the S/V values (0.13) and C/V  $(0.03)$  in the Lae-lae island, indicate the  $[1]$ presence of woody and gimnosperm plant tissues.

The acid/aldehyde ratios of three lignin phenol groups have been used to characterize diagenetic alteration in a variety of<br>geochemical samples. The acid-aldebyde [2] geochemical samples. The acid:aldehyde ratios of V- and S-phenols (i.e. vanillic acid/vanillin (Ad/Al)v and syringic acid/syringaldehyde (Al/Ad)s) are indicators of the diagenetic alteration of lignin<sup>[4,10,11,22]</sup>. (Ad/Al)v is a better indicator of lignin diagenesis than (Ad/Al)s because there are usually higher levels of vanillyl oxidation [3] products<sup>[11]</sup>. The values of  $(Ad/A)$ <sub>V</sub> in the Losari Beach and the Lae-lae Island are 0.17 and 0.15, respectively.

#### **4. CONCLUSIONS**

Lignin phenol ratio in the Losari Beach and Lae-lae island explain the presence of woody and gimnosperm plant tissues. The value of C/N and  $\delta^{13}$ C ratios, spatially has [5] implied that the distribution of organic matter from anthropogenic activities. The inclusion of organic matter in the coastal waters explains that the groundwater organic particulate matter that survived in transportation to the outermost distance,

prolongs the pre-oxidized and more resistant to microbial degradation.

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### **REFERENCES**

- Benner, R., Kaiser, K., 2011. Biological and photochemical transformations of amino acids and lignin phenols in riverine dissolved organic matter. Biogeochemistry 102, 209-222.
- [2]` Crow, S.E., Lajtha, K., Filley, T.R., Swanston, C.W., Bowden, R.D., Caldwell, B.A., 2009. Sources of plant derived carbon and stability of organic matter in soil: implications for global change. Global Change Biology 15, 2003-2019.
- $Culp$  R. (2012) Compound specific radiocarbon content of lignin oxidation products from the Altamaha river and Coastal Georgia. Nucl. Instr. Meth. B,
- [4] Dittmar T, Lara RJ.Molecular evidence for lignin degradation in sulphatereducing mangrove sediments (Amazonia, Brazil). Geochim Cosmochim Acta 2001;65(9):1417 – 1428.
- [5] Drenzek N.J., Montluc\_on D.B., Yunker M.B., Macdonald R.W and Eglinton T (2007) Constraints on the origin of sedimentary organic carbon in the Beaufort Sea from coupled molecular  $^{13}$ C and  $^{14}$ C measurements. Mar. Chem. 103, 146–162.
- [6] Goñi,M. A., and K. A. Thomas. 2000. Sources and transformations of organic matter in surface soils and sediments from a tidal estuary (North Inlet, South Carolina, U.S.A). *Estuaries* **23:** 548– 564.
- [7] Gordon, E.S., Goñi, M.A., 2003. Sources and distribution of terrigenous organic matter delivered by the Atchafalaya River to sediments in the northern Gulf of Mexico. Geochimica Et Cosmochimica Acta 67 (13), 2359– 2375.
- [8] Gustafsson O¨ ., van Dongen B. E., Vonk J. E., Dudarev O. V. and Semiletov I. P. (2011) Widespread release of old carbon across the Siberian Arctic echoed by its large rivers. Biogeosciences 8, 1737–1743.
- [9] Hansell, D.A., Carlson, C.A., 2001, Biogeochemistry of Total Organis Carbon and Nitrogen in the Sargasso Sea: Control by Convective Overtum, Deep Sea Res. Pt. II, 48(8-9), 1649- 1667.
- [10] Hedges, J.I., Ertel, J.R., 1982. Characterization of lignin by gas capillary chromatography of cupric oxide oxidation products. Analytical Chemistry 54, 174–178.
- [11] Hedges, J.I., Clark, W.A., Quay, P.D., Richey, J.E., Devol, A.H., Santos, U.M., 1986. Compositions and fluxes of particulate organic material in the Amazon River.Limnology and Oceanography 31, 717–738.
- [12] Hernes, P.J., Benner, R., 2006. Terrigenous organic matter sources and reactivity in the North Atlantic Ocean and a comparison to the Arctic and Pacific oceans. Marine Chemistry 100, 66–79.
- [13] Hernes, P.J., Robinson, A.C., Aufdenkampe, A.K., 2007. Fractionation of lignin during leaching and sorption and implications for matter ''freshness''. Geophysical Research Letters 34, L17401. doi:10.1029/2007GL031017.
- [14] Hyodo F., Kuwae M., Sasaki N., Hayashi R., Makino W., Kusaka S., Tsugeki N.K., Ishida S., Ohtsuki H., Omoto K., Urabe J. (2017). Variations in lignin-derived phenols in sediments of Japanese lakes over the last century and their reation to watershed vegetation. Organic Geochemistry 103. 125-135
- [15] Jex, C.N., Pate, G.H., Blyth, A.J., Spencer, R.G., Hernes, P.J., Khan, S.J., Baker, A., 2014. Lignin biogeochemistry: from modern processes to Quaternary Archives. Quaternary Science Reviews 87. 46- 59.
- [16] Loh, P.S., Chen, C.A., Anshari, G.Z., Wang, J.T., Lou, J.Y., Wang, S.L., 2012. A comprehensive survey of lignin geochemistry in the sedimentary organic matter along the Kapuas River (West Kalimantan, Indonesia).
- [17] Louchouarn, P., Opsahl, S., Benner, R., 2000. Isolation and quantification of dissolved lignin from natural waters using solid-phase extraction and GC/MS. Analytical Chemistry 72, 2780–2787.
- [18] May, C.L, Koseff, J.R., Kucas, L.V., Cloem, J.E., Schoellhamer, D.H., 2003. Effects of Spatial and Temporal Variability of Turbidy on Phytoplankton Blooms. Mar. Ecol. Prog. Ser. 254, 111-128.
- [19] Nagao, *et al*., 2010. Carbon Isotopes and Lignin composition of POC in a

small river in Bekanbeushi Moor, northern Japan. *Nuclear Instruments and Methods in Physics Research B 268,* 1098-1101.

- [20] Novo-Uzal, E., Pomar, F., Gómez Ros, L.V., Espiñeira, J.M., Ros Barceló, A., 2012.Evolutionary history of lignins. In: Lise, J., Catherine, L. (Eds.), Advances in Botanical Research. Academic Press, pp. 309–350 (Chapter 9).
- [21] Opsahl, S. and Benner R., 1999. Major flux of terrigenous dissolved organic matter through the Arctic Ocean. *Limnol. Oceanogr.,* 44(8), 2017–2023.
- [22] Orem, W.H., Colman, S.M., Lerch, H.E., 1997. Lignin phenols in sediments of Lake Baikal, Siberia: application to paleoenvironmental studies. Organic Geochemistry 27, 153–172.
- [23] Otto A. and Simpson M. J. (2005) Degradation and preservation of vascular plant-derived biomarkers in grassland and forest soils from western Canada. Biogeochemistry 74, 377–409.
- [24] Otto, A., Simpson, M., 2006. Evaluation of CuO oxidation parameters for determining the sources and stage of lignin degradation in soil. Biogeochemistry 80, 121-142.
- [25] Sarkar. A, Timothy R. Filley, S. Bera., 2015. Carbon isotopic composition of lignin biomarkers: Evidence of grassland over the Gangetic plain during LGM. Quaternary International 355. 194-201.
- [26] Spencer, R.G.M., Aiken. G.R., Dyda, R.Y. Butler, K.D., Bergamaschi, B.A., Hernes, P.J., 2010. Comparison of XAD with other dissolved lignin isolation techniques and a compilation of analytical improvements for the

analysisi of lignin in aquatic settings. Organic Geochemistry 41. 445-453.

- [27] Tareq, S.M., Tanaka, N., Ohta, K., 2007. Isotopes and lignin signature in tropical peat core: An approach to reconstruct past vegetation and climate changes. Tropics 16, 131-140.
- [28] Wakeham S. G., Canuel E. A., Lerberg E. J., Mason P., Sampere T. P. and Bianchi T. S. (2009) Partitioning of organic matter in continental margin sediments among density fractions. Mar. Chem. 115, 211–225.
- [29] Weng, J.K., Chapple, C., 2010. The origin and evolution of lignin biosynthesis. New Phytologist 187, 273–285.