# Oxa-bowls: formation of exceptionally stable diozonides with novel, $\mathbf{C}-\mathbf{H} \cdots \mathrm{O}$ hydrogen bond directed, solid state architecture 

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Several tetraquinane-based diolefins on ozonization furnish unusually stable diozonides of 'bowl-like' topology and their solid state structure reveals interesting architecture woven through a network of unique C-H...O interactions.

We have been engaged in the synthesis of a range of novel, high symmetry, 'bowl-shaped' molecular entities, composed of a carbocyclic base and hetero-atom speckled rim. ${ }^{1}$ In the context of these endeavours, we became interested in an octacyclic, hexaoxa-bowl $1, \mathrm{C}_{14} \mathrm{H}_{14} \mathrm{O}_{6}$, of $C_{2 \mathrm{v}}$ symmetry, which we planned to access from the hexaaldehyde 2, through a cascade of intramolecular acetalization protocols. ${ }^{1}$ Compound 2, in turn, was proposed to be prepared from the $C_{2 \mathrm{v}}$ tetraquinane 3 via ozonolysis or equivalent oxidative cleavage of the disubstituted double bonds (Scheme 1). During our studies of 1, we observed that several derivatives of $\mathbf{3}$ on ozonization furnish remarkably stable diozonides in a highly stereoselective manner. Two of these tetraquinane derived diozonides readily furnished single crystals and their X-ray crystallographic analysis shows a fascinating packing pattern and supramolecular arrangement, sustained by a network of $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds. ${ }^{2}$ To the best of our knowledge, only one stable diozonide [derived from hexamethyl (Dewar) benzene] is known ${ }^{3 a}$ so far, although several stable mono-ozonides have been reported recently. ${ }^{3 c, d}$ The crystal structure ${ }^{3 a, b}$ of the diozonide from hexamethyl (Dewar) benzene has also been determined, but the authors did not observe any significant intermolecular contacts.
Tetraquinanes 3, 4 and their oxa-analogues 5-8 reported earlier by us, ${ }^{4}$ on ozonolysis in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $-78{ }^{\circ} \mathrm{C}$ directly furnished the corresponding diozonides 9-14 after the removal of the solvent and could be readily crystallized (Scheme 2). All the diozonides were stable at room temperature (for months) and showed little decomposition on either moderate heating or on exposure to $\mathrm{Me}_{2} \mathrm{~S}$ and triphenylphosphine under standard conditions (stirring in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution). The gross structures of

MMX minimised Structure

$3 \mathrm{R}=\mathrm{CHO}$ etc.






2

Scheme 1

$3 R=\mathrm{COOMe}$
$4 R==0$

$5 \mathrm{X}=\mathrm{O}, \mathrm{R}=\mathrm{H}$ $6 \mathrm{X}=\mathrm{CH}_{2}, \mathrm{R}=\mathrm{H}$ $7 \mathrm{X}=\mathrm{CH}_{2}, \mathrm{R}=\mathrm{Ph}$ $8 \mathrm{X}=\mathrm{C}=\mathrm{O}, \mathrm{R}=\mathrm{H}$





11 (68\%)
12 (44\%)
13 (47\%)
14 (57\%)

Scheme 2
9-14 were fully revealed through the incisive analysis of their ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data. ${ }^{5}$ However, the bowl-like syn,synstereochemistry present in 9-14, with the two oxa-bridges protruding inwards could not be delineated unambiguously, although the MMX calculations showed this arrangement to be the most stable $\ddagger$ Therefore, recourse was taken to X-ray crystallography and two of the diozonides 9 and 14 furnished single crystals suitable for structure determination.
The diozonide 9 crystallized in space group $C 2 / c$ and an ORTEP perspective (Fig. 1)§ revealed the stereochemistry as syn,syn with the O1 and O1a oxa-bridges tilting significantly inwards ( O 1 and O 1 a distance being $2.62 \AA$ ), and this orientation is facilitated by intramolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds ${ }^{2}$ between the bridgehead H 1 and H 1 a atoms and the O 4 a


Fig. 1 ORTEP plot of 9


Fig. 2 Crystal packing in 9, view down $a$ axis. Selected C-H‥O interactions: $\left[\mathrm{H} \cdots \mathrm{O}(\AA), \mathrm{C}(\mathrm{H}) \cdots \mathrm{O}(\AA), \mathrm{CH} \cdots \mathrm{O}\left({ }^{\circ}\right)\right]$ : (i) (intra, $\left.\mathrm{C} 1-\mathrm{H} 1 \cdots \mathrm{O} 4\right)$ 2.21, 2.93, 129.5; (ii) (inter, C4-H4‥O2) 2.58, 3.34, 134.4; (iii) (inter, C8-H8C‥O1) 2.60, 3.18, 119.5 .


Fig. 3 ORTEP plot of 14
and O 4 carbonyl groups of the ester moieties ( $\mathrm{H} \cdots \mathrm{O}$ distance $2.21 \AA, \mathrm{C} \cdots \mathrm{O} 2.93 \AA$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ angle of $129.5^{\circ}$ ) (Fig. 1). The packing pattern shows that the bowl-like molecules of 9 are stacked one over the other along the $b$ axis in a columnar arrangement (Fig. 2). In the $a b$ plane there are arrays of unidirectional bowls, each engaged in four $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (two below and two above) to nearest neighbours through H 4 , H4a (ring junction protons on the diquinane moiety) and O2a, O2 oxygens of the peroxo bridge ( $\mathrm{H} \cdots \mathrm{O}$ distance $2.58 \AA$, $\mathrm{C} \cdots \mathrm{O} 3.34 \mathrm{~A}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ angle of $134.4^{\circ}$ ) along the $b$ axis. The adjacent $a b$ planes consist of bowls growing in opposite directions and are held by weak $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ contacts between the Hc protons of the ester methyls and O 1 oxa-bridges of the inversion related bowls (Fig. 2).

The oxa-bowl 14 crystallizes in space group $P \subset a 2_{1}$ and the molecules are tightly packed as indicated by its high density ( $1.681 \mathrm{Mg} \mathrm{m}^{-3}$ ). An ORTEP diagram (Fig. 3)§ once again revealed the syn,syn stereochemistry but the two oxa-bridges are now far apart ( O 1 to O 4 distance is $4.22 \AA$ ), perhaps due to repulsion within the congested concave surface with O7. The bowls are held by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds ( $\mathrm{H} \cdots \mathrm{O}$ distance 2.36 $\AA, \mathrm{C} \cdots \mathrm{O} 3.28 \AA$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ angle of $\left.155.1^{\circ}\right)^{2}$ between H 10 (ring junction proton on the diquinane moiety) and O 7 ether oxygen, along the $c$ axis, defining a ribbon-like pattern (Fig. 4). There is a bowl inversion pattern along the $a$ axis and these bowls are held through a $\mathrm{C} 1-\mathrm{H} 1 \cdots \mathrm{O} 5$ contact. Along the $b$ axis the bowls are held by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds between the H 7 ( $\alpha$ to the ether oxygen) and the O8 carbonyl oxygen ( $\mathrm{H} \cdots \mathrm{O}$ distance $2.57 \AA, \mathrm{C} \cdots \mathrm{O} 3.49 \AA$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ angle of $156.6^{\circ}$ ) to generate another ribbon-like pattern (Fig. 4). It is noteworthy that in the structures of both 9 and $\mathbf{1 4}$, the least acidic diquinane ring junction protons are involved in the $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds and this is perhaps a consequence of the bowl-like topology present in these molecules.

In summary, we have described the characterization of several 'bowl-like' diozonides of exceptional stability and their X-ray structure reveals a novel architecture sustained through a network of $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds.


Fig. 4 Crystal packing in 14, view down $b$ axis. Selected $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions: $\left[\mathrm{H} \cdots \mathrm{O}(\AA), \mathrm{C}(\mathrm{H}) \cdots \mathrm{O}(\AA), \mathrm{CH} \cdots \mathrm{O}\left({ }^{\circ}\right)\right]:(\mathrm{i})($ inter, $\mathrm{C} 1-\mathrm{H} 1 \cdots \mathrm{O})$ 2.42, 3.05, 121.4; (ii) (inter, C7-H7‥O8) 2.57, 3.49, 156.6; (iii) (inter, C10-H10…O7) 2.36, 3.28, 155.1.

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## Notes and References

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$\ddagger$ The diozonides 9 - $\mathbf{1 4}$ can in principle occur in three diastereoisomeric forms, viz. syn, syn (with the two oxa-bridges directed within the diquinane moiety), anti,anti (with the two peroxy-bridges directed within the diquinane moiety) and syn,anti (with one oxa- and one peroxy-bridge directed within the diquinane moiety).
$\S X$-Ray data for 9: $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{O}_{10}, M=370.30$, colourless crystals from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, monoclinic, space group $C 2 / c, a=11.534(6), b=9.174(9)$ and $c$ $=14.825(19) \AA, \beta=101.12(8)^{\circ}, V=1539(3) \AA^{3}, Z=4, D_{c}=1.598$ $\mathrm{Mg} \mathrm{m}^{-3}, T=293 \mathrm{~K}, F(000)=776, \mu(\mathrm{Mo}-\mathrm{K} \alpha)=0.135 \mathrm{~mm}^{-1}$, crystal dimensions $0.23 \times 0.11 \times 0.15 \mathrm{~mm}$. Data were collected on Enraf-Nonius MACH-3 diffractometer, graphite-monochromated Mo-K $\alpha$ radiation ( $\lambda=$ $0.71073 \AA$ ), by the $\omega$ scan method in the range $2 \leqslant \theta \leqslant 25^{\circ}, 1351$ unique reflections $\left[R_{\mathrm{int}}=0.0379\right]$, of which 1211 had $F_{\mathrm{o}}>4 \sigma\left(F_{\mathrm{o}}\right)$, were used in all calculations. At final convergence $R_{1}[I>2 \sigma(I)]=0.0447, w R_{2}=$ 0.1197 for 119 parameters, GOF $=1.074, \Delta \rho_{\max }=0.297$ e $\AA^{-3}, \Delta \rho_{\min }=$ $-0.251 \mathrm{e}^{\AA^{-3}}$. The data were reduced using XTAL (ver. 3.4), solved by direct methods, refined by full-matrix least-squares on $F^{2}$ with the non-H atoms anisotropic, and H atoms isotropic. ${ }^{6}$ For 14: $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{O}_{8}, M=270.19$, colourless crystals from ethyl acetate-hexane, orthorombic, space group Pca2 (no. 29), $a=15.547(3), b=7.584(1)$ and $c=9.052(10) \AA, V=$ $1067.3(13) \AA^{3}, Z=4, D_{\mathrm{c}}=1.681 \mathrm{Mg} \mathrm{m}^{-3}, T=293 \mathrm{~K}, F(000)=560$, $\mu(\mathrm{Mo}-\mathrm{K} \alpha)=0.147 \mathrm{~mm}^{-1}$, crystal dimensions $0.20 \times 0.11 \times 0.13 \mathrm{~mm}$. Data were collected as above in the range $2 \leqslant \theta \leqslant 30^{\circ}, 1648$ unique reflections $\left[R_{\text {int }}=0.0\right]$, of which $741 \mathrm{had} F_{\mathrm{o}}>4 \sigma\left(F_{\mathrm{o}}\right)$, were used in all calculations. At final convergence $R_{1}[I>2 \sigma(I)]=0.0672, w R_{2}=0.1724$ for 117 parameters, $\mathrm{GOF}=1.046, \Delta \rho_{\max }=0.294 \mathrm{e} \AA^{-3}, \Delta \rho_{\min }=-0.283$ e $\AA^{-3}$. The structure was solved as above with all O atoms anisotropic, C and H atoms isotropic. ${ }^{6}$ CCDC 182/918.

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