



## Analysis of Crystallinity and Physical Properties of the Bio-solar Gemstone

Ismail Ismail<sup>1\*</sup>, Akmal Nizar<sup>1</sup>, Mursa<sup>1</sup>, and Zulkarnain Jalil<sup>1</sup>

<sup>1</sup>Department of Physics, Faculty of Mathematics and Natural Science, Universitas Syiah Kuala,  
Banda Aceh 23111, Indonesia

\*Corresponding author email: [ismailab@unsyiah.ac.id](mailto:ismailab@unsyiah.ac.id)

Received: September 17, 2019

Accepted: February 14, 2020

Online : February 14, 2020

**Abstract** – Bio-solar is one of the natural gemstones found in the province of Aceh in Indonesia. Nevertheless, detail information on this gemstone is still unknown. The purpose of this work is to analyze the chemical composition, crystallinity, specific gravity, and hardness of the bio-solar gemstone from Aceh. X-ray diffraction has been used to determine the chemical composition and crystallinity of the sample. The hardness of the sample was measured by using Diamond Selector II. The specific gravity of the sample was determined by the water displacement method. Our results show that the bio-solar gemstone from Aceh is composed of CaO, SiO<sub>2</sub>, MgO, Al<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>, and Ni<sub>2</sub>O<sub>3</sub> phases. It is found that the bio-solar gemstone is a crystalline material, not amorphous. The average crystallite size of this gemstone is 35.3 nm (353 Å). The specific gravity of bio-solar gemstone is found to be 3.09 – 3.34. Its hardness is 3 to 4 Mohs. Our analysis suggests that the bio-solar can be classified as a soft vesuvianite gemstone.

**Keywords:** gemstone, jade, vesuvianite, bio-solar

### Introduction

The gemstone is a mineral crystal which has been mostly used as jewelry. One of the favorite gems is jade (Walker, 1991). Chinese people believe that jade can make good health and long life. Jade price can exceed the price of gold (New York Times, 2010). Meanwhile, natural jade is rare; therefore, its price has become expensive. Consequently, only high-class people use natural jade, and it has become a symbol of nobility and perfection (Desjardins, 2015). For those reasons, people try to search for natural jade everywhere in the world. Jade has been found in several countries, including in Indonesia. Jade has been defined to consist mostly of jadeite or nephrite (Harlow *et al.*, 2005). Jadeite is a pyroxene group that includes Na(Al, Fe)-Si<sub>2</sub>O<sub>6</sub> (Morimoto, 1988; Prewitt *et al.*, 1966; Htein *et al.*, 1994). In the meantime, nephrite is an amphibole group which is composed of Ca<sub>2</sub>(Mg, Fe)5Si<sub>8</sub>O<sub>22</sub>(OH)<sub>2</sub> (Liu *et al.*, 2011). Nephrite can be classified into two types that are nephrite-tremolite and nephrite-antigorite. Serpentine is a type of jade. Sometimes, it is called a serpentine jade. Its color can be green, yellowish-green, or brownish-green. Two kinds of serpentine are antigorite and chrysotile. Vesuvianite is also known as jade. It is often called as California jade or American jade. Vesuvianite is also known as idocrase. Sometimes, idocrase is used for gemstone-quality vesuvianite. According to web mineral data, the chemical composition of vesuvianite gemstone is SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, MgO, CaO, and H<sub>2</sub>O (webmineral.com). However, Groat *et al.* found that the chemical composition of vesuvianite is SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, MgO, CaO, FeO, Fe<sub>2</sub>O<sub>3</sub>, and MnO (Groat *et al.*, 1992).

The research on jade has been conducted continuously. In Turkey, Murat *et al.* characterized the purple stone, which mainly contained SiO<sub>2</sub> (63.54%) and Al<sub>2</sub>O<sub>3</sub> (20.17%), and Na<sub>2</sub>O (6.71%). The specific gravity of this gemstone was found to be 3.24 to 3.42. This Turkish gemstone was classified as jadeite-jade (Murat *et al.*, 2012). In Indonesia, the community continues to explore precious stones, including in the province of Aceh. Some gemstones were discovered in Nagan Raya, Aceh (Tempo, 2015). A previous study classified gemstones

from the region of Aceh in Indonesia as jadeite, nephrite, serpentine, and vesuvianite based on their colors and hardness (Nurul *et al.*, 2014). They suggested that the colors and hardness of the gemstones were influenced by impurity mineral compounds (Nurul *et al.*, 2014). One of a well-known gem from the province of Aceh is called bio-solar. It is so beautiful, as shown in Figure 1. Its color is dark yellow. A recent study showed that this bio-solar gemstone contains 59.8% of CaO, 19.7% of SiO<sub>2</sub>, 11.1% of Fe<sub>2</sub>O<sub>3</sub>, 7.5% of Al<sub>2</sub>O<sub>3</sub>, and 1.3% of NiO. Based on its chemical composition, the bio-solar gemstone from Aceh can be classified as a type of vesuvianite jade (Akmal *et al.*, 2016). However, detail information of this gemstone, such as crystallinity, specific gravity, and hardness, is still unknown. According to the previous study, the vesuvianite gemstone is crystallite, having the tetragonal atomic structure (Anthony *et al.*, 2010). Its values of specific gravity and hardness are in the range 3.32 – 3.43 and 6 – 7 Mohs, respectively (Anthony *et al.*, 2010). The objectives of this study are to understand the chemical composition, crystallinity, specific gravity, and hardness of the bio-solar gemstone. The results are presented in this paper.



Figure 1. Bio-solar gemstone from Nagan Raya, Aceh

### Materials and Methods

The bio-solar gemstone samples were purchased from the Aceh Gemstone Market in Tingkeum, Aceh Besar District, where they are initially from the district of Nagan Raya in the Aceh province, Indonesia. Minerals identification and the crystallite size of bio-solar gemstone were determined by using x-ray diffraction (Shimadzu, D7000). The sample of bio-solar gemstone for x-ray diffraction measurement was crushed into powder. The measurement condition was at room temperature, x-ray wavelength  $\lambda = 1.5406 \text{ \AA}$ , continuous scan, scan range ( $2\theta$ ) from 10 to 80 degrees with each step of 0.02 degrees. The sample size for specific gravity and hardness measurements was 17 mm x 15 mm x 7 mm. Four samples were used in the analyses. The hardness of bio-solar gemstone was measured by using Diamond Selector II. The specific gravity of the bio-solar gemstone was determined by using the water displacement method. Electronic balance with high sensitivity was used in determining the specific gravity.

### Results

Our result of x-ray diffraction (XRD) measurement of the bio-solar gemstone is shown in Figure 2. There were eleven sharp peaks observed in our XRD measurement. The highest peak was seen at the Bragg angle ( $2\theta$ ) of 32.48 degrees. The kind of minerals (a type of oxide) contained in the bio-solar gemstone was determined by comparing the XRD data to JCPDS (Joint Committee on Powder Diffraction Standards).

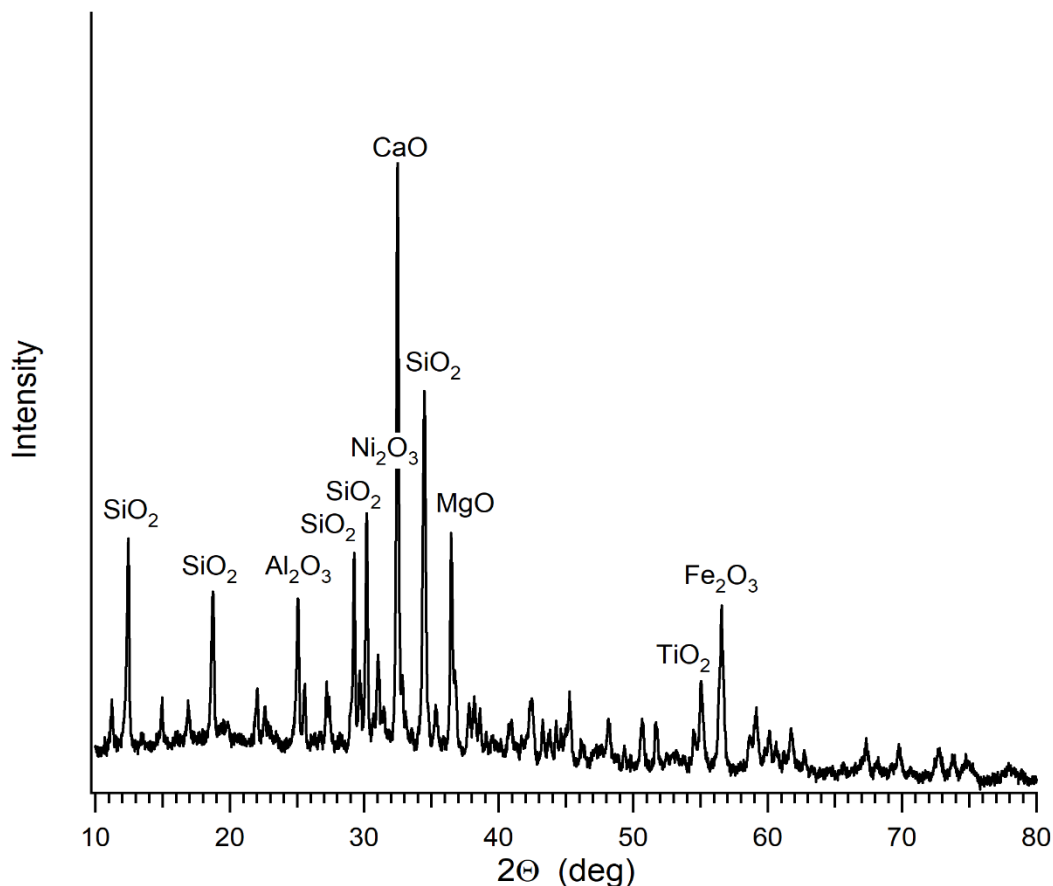


Figure 2. X-ray diffraction data from the bio-solar gemstone

The comparisons between our XRD data and JCPDS data are shown in Table 1. By comparing the experimental data to JCPDS data, we found that the highest peak at 32.48 degrees is the CaO phase. The second highest peak was observed at 34.46 degrees, which is the SiO<sub>2</sub> phase. Some other peaks having SiO<sub>2</sub> periods were found at the Bragg angles of 30.18 degrees, 29.24 degrees, 12.46 degrees, and 18.74 degrees. The MgO, Fe<sub>2</sub>O<sub>3</sub>, Al<sub>2</sub>O<sub>3</sub>, and TiO<sub>2</sub> phases were observed at the Bragg angles of 36.48 degrees, 56.56 degrees, 25.56 degrees, and 55.06 degrees, respectively. There was also a peak seen at 31.02 degrees, which is the Ni<sub>2</sub>O<sub>3</sub> phase. As shown in Table 1, our experimental data are in good agreement with the JCPDS data.

Table 1. The comparison of XRD data from bio-solar gemstone and JCPDS data

No	2θ (degrees)		Intensity	Phase	hkl
	Experiment	JCPDS			
1	32.48	32.300	100	CaO	111
2	34.46	34.645	66	SiO <sub>2</sub>	642
3	30.18	30.248	49	SiO <sub>2</sub>	533
4	36.48	36.932	45	MgO	111

5	12.46	12.923	45	SiO <sub>2</sub>	220
6	29.24	29.150	42	SiO <sub>2</sub>	620
7	18.74	18.315	36	SiO <sub>2</sub>	400
8	25.56	25.839	35	Al <sub>2</sub> O <sub>3</sub>	012
9	56.56	56.575	33	Fe <sub>2</sub> O <sub>3</sub>	211
10	55.06	55.134	28	TiO <sub>2</sub>	202
11	31.02	31.575	24	Ni <sub>2</sub> O <sub>3</sub>	002

Our XRD results (see Figure 2) revealed that the bio-solar gemstone is crystallite, not amorphous. It is a well-ordered crystal. The crystallite size of the bio-solar gemstone can be calculated by using equation (1) (Suryanarayana *et al.*, 1998).

$$D = \frac{k \cdot \lambda}{\beta \cdot \cos(\theta)} \quad (1)$$

Where,  $k = 0.95$  (constant),  $\lambda = 0.15406$  nm (the wavelength of x-ray used in the experiment),  $\theta$  = the Bragg angle in degrees,  $\beta$  = the full width at half maximum (FWHM) in radian,  $D$  = the crystallite size of sample in nm.

The full width at half maximum was calculated by fitting the Lorentzian function to the experimental XRD data, as shown in Figure 3. By using the equation (1) above, the crystallite size of the CaO phase is found to be 45.97 nm (459 Å). The crystallite size of SiO<sub>2</sub> is 33.77 nm (337 Å), which is smaller than the crystallite size of CaO. The crystallite size of MgO is 38.39 nm (384 Å). The crystallite size of Al<sub>2</sub>O<sub>3</sub> is 42.99 nm (429 Å), which is almost the same as the crystallite size of CaO. However, the crystallite sizes of Fe<sub>2</sub>O<sub>3</sub> and TiO<sub>2</sub> are rather small, 23.80 nm (238 Å), and 27.02 nm (270 Å), respectively. The crystallite size of phases contained in the bio-solar gemstone is listed in Table 2.

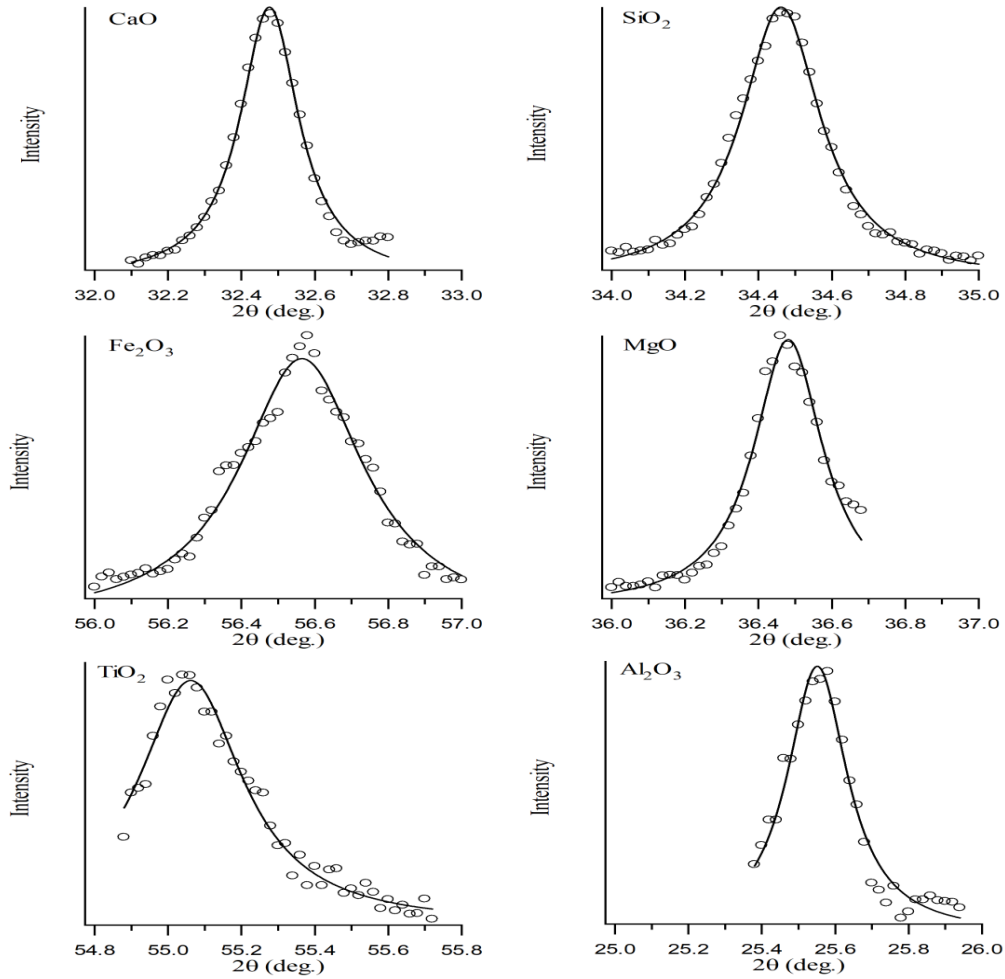


Figure 3. The comparison of the XRD data (circles) and calculated the best fit of Lorentzian function (solid lines) for the bio-solar gemstone

Table 2. The crystallite size of the bio-solar gemstone

No	Phase	2θ (deg)	FWHM (deg)	D (nm)
1	CaO	32.48	0.19	45.97
2	SiO <sub>2</sub>	34.46	0.26	33.77
3	MgO	36.48	0.23	38.39
4	Fe <sub>2</sub> O <sub>3</sub>	56.56	0.40	23.80
5	Al <sub>2</sub> O <sub>3</sub>	25.56	0.20	42.99
6	TiO <sub>2</sub>	55.06	0.35	27.02
The average crystallite size (nm)				35.33

The specific gravity (SG) and hardness of the bio-solar gemstone have been measured. Our measurement results showed that the specific gravity of the bio-solar gemstone is in the range of 3.09 – 3.34. The hardness of the bio-solar gemstone is found to be in the range of 3 to 4 Mohs.

### Discussion

The comparison between minerals contained in the bio-solar gemstone with some other gemstones is listed in Table 3. Mark  $\checkmark$  indicates the type of oxides contained in the gemstones.

Table 3. The comparison of bio-solar with selected gemstones

Type of oxide	Bio-solar (this study)	Jadeite (Prewitt <i>et al.</i> )	Nephrite Actinolite (Evans <i>et al.</i> )	Nephrite Tremolite (Ballinaro <i>et al.</i> )	Serpentine Clinochrysotile (Pollastri <i>et al.</i> )	Serpentine Antigorite (Brindley <i>et al.</i> )	Vesuvianite (Groat <i>et al.</i> )
SiO <sub>2</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
Al <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	-	-	-	-	$\checkmark$
Na <sub>2</sub> O	-	$\checkmark$	-	-	-	-	-
MgO	$\checkmark$	-	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
CaO	$\checkmark$	-	$\checkmark$	$\checkmark$	-	-	$\checkmark$
FeO	-	-	$\checkmark$	-	-	-	$\checkmark$
Fe <sub>2</sub> O <sub>3</sub>	$\checkmark$	-	-	-	-	$\checkmark$	$\checkmark$
TiO <sub>2</sub>	$\checkmark$	-	-	-	-	-	$\checkmark$
MnO	-	-	-	-	-	-	$\checkmark$
Ni <sub>2</sub> O <sub>3</sub>	$\checkmark$	-	-	-	-	-	-

The present study of bio-solar gemstone contains CaO, SiO<sub>2</sub>, MgO, Al<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>, and Ni<sub>2</sub>O<sub>3</sub>. Meanwhile, jadeite contains SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, and Na<sub>2</sub>O. The mineral composition of bio-solar is quite different than that of jadeite. Thus, the bio-solar is not a type of jadeite. This situation is the same for nephrite actinolite, nephrite tremolite, serpentine clinochrysotile, and serpentine antigorite. Groat *et al.* found that vesuvianite contains SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, MgO, CaO, FeO, TiO<sub>2</sub>, and Fe<sub>2</sub>O<sub>3</sub> (Groat *et al.*, 1992). All minerals included vesuvianite are found in the bio-solar gemstone, except FeO and MnO. However, the amount of FeO and MnO are small; a particular sample (vesuvianite from Mexico) has 0.06% of MnO and 0.68% of FeO (Groat *et al.*, 1992). There is a peak of Ni<sub>2</sub>O<sub>3</sub> observed in our XRD data. It was noted NiO (1.24%) in bio-solar gemstone by using X-Ray Fluorescent (XRF) (Akmal *et al.*, 2016). There is a small discrepancy between our result and Akmal's result, which could be due to the sensitivity of XRD and XRF instruments.

The size of crystallite reflects the crystallinity that defines the degree of long-range order in a material which affects the properties of a material. The ideal crystallite size is usually between 500 – 1000 nm. However, the crystallite size is often much smaller than that because of the imperfection of material such as impurity, defect, dislocation (Suryanarayana *et al.*, 1998). The study of the crystallite size of the gemstone is very limited in the literature. For example, the crystallite size of opal gemstone from California USA was found to be 15 nm (150 Å) (Sanders, 1975). The crystallite size of amethyst gemstone from Central Kalimantan was 14 nm (Suastika *et al.*, 2017). A recent study found that the crystallite size of gold nanoparticles was 18 – 24 nm (Ogundare *et al.*, 2019). As shown in Table 3, CaO is the most significant crystallite size among all other phases in bio-solar that is 45.97 nm (459.7 Å). The average crystallite size of bio-solar is 35.33 nm (353 Å), which is far below the ideal

size, which could be due to impurity and other imperfection in bio-solar. However, it is larger than the crystallite sizes of amethyst, opal, and gold nanoparticles.

The specific gravity (SG) of the bio-solar gemstone is compared to other gemstones, as shown in Table 4. The SG of bio-solar is 3.09 – 3.34, which is smaller than the SG of jadeite gemstones (Leaming, 1978; Anthony *et al.*, 2010; Franz *et al.*, 2014). The SG of nephrite actinolite is 3.03 – 3.24 that is the same as the SG of the bio-solar gemstone. Meanwhile, the chemical composition of bio-solar is different than that of nephrite actinolite (see Table 3). The SG of bio-solar is higher than the SGs of serpentine clinochrysotile and serpentine antigorite. The bio-solar gemstone is slightly smaller than the SG of vesuvianite from the previous study by Anthony *et al.* (2010). Nonetheless, the chemical composition of bio-solar is about the same as that of vesuvianite gemstone of the prior research by Groat *et al.* (1992).

Table 4. The comparison of the SG and hardness of bio-solar gemstone and other gemstones.

Kind of Gemstone	Specific Gravity (SG)	Hardness (Mohs)
Jadeite (Anthony <i>et al.</i> , 2010)	3.24 – 3.43	6 – 7
Jadeite (Franz <i>et al.</i> , 2014)	3.25 – 3.40	-
Jadeite (Leaming, 1978)	3.24 – 3.43	-
Nephrite actinolite (Anthony <i>et al.</i> , 2010)	3.03 – 3.24	5 – 6
Nephrite actinolite (Leaming <i>et al.</i> , 1978)	2.95 – 3.01	-
Nephrite tremolite (Anthony <i>et al.</i> , 2010)	2.99 – 3.03	5 – 6
Serpentine clinochrysotile (Anthony <i>et al.</i> , 2010)	2.53	2.5
Serpentine antigorite (Anthony <i>et al.</i> , 2010)	2.65	2.5 – 3.5
Vesuvianite (Anthony <i>et al.</i> , 2010)	3.32 – 3.43	6 – 7
Bio-solar (present study)	3.09 – 3.34	3 – 4

The comparison between the hardness of bio-solar with some selected gemstone from previous studies is shown in Table 4. The hardness of bio-solar is 3 – 4 Mohs, which is much smaller than the hardness of jadeite. The hardness of bio-solar is also lower than the hardness of nephrite. However, the hardness of bio-solar is larger than the hardness of serpentine. The previous study reported that the hardness of vesuvianite is 6 – 7 Mohs (Anthony *et al.*, 2010). The hardness of bio-solar gemstone is significantly lower than that of vesuvianite from the previous study.

Meanwhile, we have shown above that the chemical composition of the bio-solar is about the same as that of vesuvianite. Previous work showed that the impurity could affect the hardness of the crystal (Verma *et al.*, 2014). Sayan *et al.* found that impurities influenced the hardness of NaCl crystal. The K ion impurity increased the hardness of crystal. But, the Ni ion impurity decreased the hardness of NaCl crystal (Sayan *et al.*, 2001). Based on these studies by Verma ((2014) and Sayan (2001), we speculate that the hardness of bio-solar gemstone is low compared to the regular vesuvianite because of impurities in it such as Ni<sub>2</sub>O<sub>3</sub> which was observed by our XRD data.

## Conclusion

Our x-ray diffraction analysis showed that the bio-solar gemstone from Aceh Indonesia contains CaO, SiO<sub>2</sub>, MgO, Al<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>, and Ni<sub>2</sub>O<sub>3</sub> phases. The chemical composition of the bio-solar gemstone is about the same as that of vesuvianite. The bio-solar gemstone is crystallite. Its average crystallite size is 35.3 nm (353 Å). The specific gravity of bio-solar gemstone is 3.09 – 3.34, which is about the same as those of vesuvianite.

These results suggest that the bio-solar gemstone from Aceh province in Indonesia is a vesuvianite gemstone. However, its hardness is 3 – 4 Mohs, which is slightly soft compared to the regular vesuvianite.

## References

- Akmal, N., Mursal, and Ismail. 2016. Minerals identification of bio-solar gemstone from Aceh. *Journal of Aceh Physics Society*, 5 (2): 22-26.
- Anthony, J. W., Bideaux, R. A., Bladh, K. W., and Nichols, M. C., 2010. *Handbook of Mineralogy*, Mineralogical Society of America, Chantilly, VA 20151-1110, USA.
- Ballirano, P., Andreozzi, G. B., and Belardi, G. 2008. Crystal chemical and structural characterization of fibrous tremolite from Susa Valley, Italy, with comments on potential harmful effects on human health. *The American Mineralogist*, 93: 1349-1355.
- Brindley, G. W., and Knorring, O. V., 1954. A new variety of antigorite (ortho-antigorite) from the UNST Sutherland Islands. *The American Mineralogist*, 39: 794 – 804.
- Desjardins J., 2015. The History of Jade: The Emperor's Stone. Available at <https://www.visualcapitalist.com/the-history-of-jade-the-emperors-stone/> (accessed 18 February 2020).
- Evans, B. W., and Yang, H., 1998. Fe-Mg order-disorder in tremolite–actinolite–Ferro-actinolite at ambient and high temperature. *American Mineralogist*, 83: 458–475.
- Franz, L., Sun, T. T., Hanni, H. A., de Capitani, C., Thanasuthipitak, T., Atichat, W. 2014. A comparative study of jadeite, omphacite and kosmochlor jades from Myanmar, and suggestions for a practical nomenclature. *The Journal of Gemmology*, 34(3): 210-229.
- Groat, L. A., Hawthorne, F. C., Ercit, T. S., 1992. The chemistry of vesuvianite. *The Canadian Mineralogist*, 30: 19 – 48.
- Harlow, G.E., Sorensen, S.S., 2005. Jade (nephrite and jadeite) and serpentine: metasomatic connections. *International Geological Review*, 47: 113–146.
- Htein, W., Naing, A. M. 1994. Mineral and chemical compositions of jadeite jade of Myanmar. *Journal of Gemology*, 24: 269–276.
- Leaming, S.F., 1978. Jade in Canada, Geological Survey of Canada Commission, 78-19: 1-59.
- Liu, Y., Deng, J., Shi, G., Yui, T. F., Zhang, G., Abuduwayiti, M. 2011. Geochemistry and petrology of nephrite from Alamas, Xinjiang, NW China. *Journal of Asian Earth Sciences*, 42: 440 – 451.
- Morimoto, N. J. 1988. Nomenclature of pyroxenes. *American Mineralogist*, 73: 1123-1133.
- Murat, H., Basevirgen, Y., Chamberlain, S.C., 2012. Gem Turkish purple jade: Geological and mineralogical characteristics. *Journal of African Earth Science*, 63: 48 – 61.
- New York Times. 2010. Available at [https://www.nytimes.com/2010/09/21/world/asia/21jade.html?\\_r=2&hp&](https://www.nytimes.com/2010/09/21/world/asia/21jade.html?_r=2&hp&) (accessed 14 February 2020).
- Nurul, A., Muchlis, Halimi, K., Nufus, H., Maysura, Z., Simatupang, M. Z., 2014. Classification of jades (giok) Beutong Aceh based on mineral composition. *Journal Natural*, 14 (2): 19-22.
- Ogundarea, O. D., Akinribideac, O. J., Adetunjib, A. R., Adeoye, M. O., and Olubambi, P. A., 2019. Crystallite size determination of thermally deposited Gold Nanoparticles. *Procedia Manufacturing*, 30: 173–179.
- Pollastri, S., Perchiazzi, N., Lezzerini, M., Plaisier, J. R., Cavallo, A., Dalconi, M.C., Gandolfi, N. B., Gualtieri, A. F. 2016. The crystal structure of mineral fibres: Chrysotile. *Periodico di Mineralogia*, 85: 249- 259.
- Prewitt, C. T., and Burnham, C. W., 1966. The crystal structure of jadeite. *The American Mineralogist*, 51: 956 – 975.
- Suryanarayana, C. and Norton, M. G. 1998. *X-Ray Diffraction A Practical Approach*. Plenum Press, New York.
- Sayan, P. and Ulrich, J. 2001. Effect of Various Impurities on the Hardness of NaCl Crystals. *Cryst. Res. Technol.*, 36: 1253–1262.
- Sanders, J. V. 1975. Microstructure and Crystallinity of Gem Opals. *American Mineralogist*, 60: 749-757.
- Suastika, K. G., Yuwana, L., Hakim, L., Darmaji and Khusnul, D. 2017. Characterization of Central Kalimantan's Amethysts by Using X-Ray Diffraction. *IOP Conf. Series: Journal of Physics: Conf. Series*, 846: 012024.



- Tempo, 2015. Available at <https://gaya.tempo.co/read/655532/giok-aceh-20-ton-dievakuasisisinya-masih-di-hutan/full&view=ok> (accessed 15 February 2020).
- Walker, J. 1991. Jade: A Special Gemstone. In: Keverne R. (eds) Jade. Springer, Boston, MA.
- Webmineral. Available at <http://webmineral.com/data/Vesuvianite.shtml#.XkX7us4zbIV> (accessed 18 February 2020).
- Verma, A. K., Chaturbuj Ojha, C. and Shrivastava, A. K. 2014. Effect of impurities on the hardness of alkali halide single crystals. AIP Conference Proceedings, 1591: 1242.