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2 comparison with experimental observations Pedro Chamorro-Posada¹, Pablo Martín-Ramos², Francisco M. Sánchez-Arévalo³, and 3 Roberto C. Dante⁴ 4 1. Dpto. de Teoría de la Señal y Comunicaciones e IT, Universidad de Valladolid, ETSI 5 Telecomunicación, Paseo Belén 15, 47011 Valladolid, Spain. 6 7 2. EPS, Instituto Universitario de Ciencias Ambientales de Aragón (IUCA), Universidad de Zaragoza, Carretera de Cuarte s/n, 22071, Huesca, Spain. 8 9 3. Departamento de reología y mecánica de materiales, Instituto de Investigaciones en 10 Materiales (IIM), Universidad Nacional Autónoma de México (UNAM), Apdo. Postal 11 70-360, Cd. Universitaria, Mexico City 04510, Mexico. 12 4. R&D department, 2Dto3D S.r.l.s., Via Revalanca 5 (San Firmino), 12036 Revello (CN), Italy. * Corresponding author. Email: pedcha@tel.uva.es Abstract A computational study of the properties of polymeric carbon nitride using molecular

Molecular dynamics simulations of nanosheets of polymeric carbon nitride and

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16 17 dynamics is presented. The analysis of ideal infinite-extent sheets permits to evaluate the 18 effect of temperature on the network of hydrogen bonds responsible for the linkage of the 19 material. The weakening of this binding mechanism at sufficiently high temperatures, 20 together with the inter-layer interactions characteristic of this type of 2D materials, is 21 shown to determine the conformation properties of polymeric carbon nitride at the 22 nanoscale. The results obtained from the simulation of finite samples in the canonical 23 ensemble at varying temperatures are consistent with those from the characterization of 24 our experimentally synthesized samples. Hydrogen bonding between adjacent polymer ribbons leads this process and is the cause of the typical crumpled structure of thismaterial.

27 Keywords: carbon nitride; conformation; molecular dynamics; MOPAC; nanosheets

28 1. INTRODUCTION

Polymeric carbon nitride has attracted large interest as a precursor for the synthesis of super-hard carbon nitride phases (1-10). This material finds many applications in the wide field of sensors due to the combination of its chemical sensitivity with its optical and semiconductor properties (11-13). Also, monolayer polymeric carbon nitride forms a new family of wide-gap semiconducting materials, in which the gap can be tuned by small variations in the stress applied to the layers or by the presence of adatoms (14).

35 The various existing synthesis routes provide predominantly amorphous samples 36 structured at the nanoscale level as graphite-like (15), nanosheets (16-19), nanotubes (20), 37 or nanopillars (21-22). They can also be structured in mesoporous conformations that lead to new applications, like the generation of nanostructured metal nitrides (23). 38 Conformation as single sheets that can be exfoliated from graphitic carbon nitride samples 39 40 (24) or directly grown (25) permits to increase its efficiency as a catalyst. The detailed 41 arrangement of this material is of most relevance for the specific applications and strongly 42 depends on the ambient conditions of the synthesis (16-19) and on other parameters, such 43 as the precursor type (25). Understanding of the detailed supramolecular mechanisms that 44 rule the conformation of this material is thus essential for the optimization of the synthesis 45 procedures and the development of new related materials and applications.

In this work, we perform a study of the conformation properties of polymeric carbon nitride sheets under different ambient conditions using atomistic molecular dynamics (MD) simulations. The analyses address both the general intrinsic properties of ideal infinite-extent 2D polymeric carbon nitride sheets and the evolution in the structural

50 organization of finite size samples; specifically, the routes towards the formation of 51 crumpled nanosheets.

52 MD (26) is a useful tool in the study of the conformational transitions of very large 53 molecular and supramolecular systems like, for instance, protein folding (27). It provides 54 not only the final minimum energy structures, but the MD trajectories also afford the 55 detailed pathways and a valuable insight into the underlying mechanisms determining the 56 evolution. Molecular dynamics simulations have been extensively used in the study of the 57 intrinsic properties of 2D materials, particularly graphene (28-31) and related materials 58 like graphane (32) or multilayer graphene (33). These systems are especially interesting 59 from a theoretical point of view since, according to the Mermin-Wagner theorem (34), 60 long-wavelength fluctuations destroy any crystalline order in pure 2D systems. 61 Nevertheless, 2D membranes can exist in a 3D space, where the displacements are not 62 constrained in two dimensions.

63 Graphene-like materials can be modelled as polymerized or tethered membranes, which have been subject of intense theoretical investigations (35-37) and which are 64 65 characterized by the fact that bonds linking the membrane sites are unbreakable. Even 66 within the physical constraint of self-avoidance, these 2D membranes typically undergo a 67 transition to a crumpled state (or a tubular state in the presence of anisotropy (38)), 68 whereas a flat state, with strong height fluctuations, is also possible due to the coupling 69 between bending and stretching modes. On the other hand, polymeric carbon nitride sheets 70 are characterized by their binding through a network of relatively weak hydrogen bonds, 71 far from the ideal unbreakable condition, that links the melon linear arrays of the basic 72 heptazine building blocks. Using MD simulations, we find that the morphological 73 properties of the material under study are predominantly determined by the behaviour of 74 the hydrogen bonds under the varying ambient conditions and the inter-planar interactions

between the heptazine units. In spite of the limited size of our simulation cells when compared with the actual spatial extent of the investigated features, the results of the computations are consistent with the experimental results.

78 **2. METHODS**

79 **2.1. Experimental**

80 2.1.1. X-ray diffraction measurements

X-ray diffraction patterns were obtained by means of a powder diffractometer ENRAFNONIUS FR590, detector INEL CPS 120 (BRUKER AXS, The Netherlands), with
monochromator-quartz discriminator, Debey-Scherrer geometry, using Cu Kα radiation,
and glass capillaries for sample mounting. The samples were ground in an agate mortar
and sifted. The measurements always lasted for 1 hour, and crystalline silicon was used
as a standard.

87 2.1.2. TEM and SEM characterization

88 Transmission electron microscopy (TEM) and scanning electron microscopy (SEM)
89 analyses were performed with a JEOL JEM-FS2200 HRP and a JEOL JSM-820 SEM
90 microscope (JEOL, Japan), respectively.

91 2.2. Computational

The MD simulations were performed using the large-scale atomic molecular massively parallel simulator (LAMMPS) (39) using the DREIDING force field (40). This generic model includes parameters for H and the non-metallic main-group elements, plus a few metals. It specifically takes into account the effect of hydrogen bonds, which are particularly relevant for the material under analysis.

97 The data used for the MD simulations were built from the 2D ground state geometry of 98 the PM6 Hamiltonian (41), calculated using MOPAC (42). The DREIDING force field 99 polymer layers used as initial conditions for the MD computations where relaxed using

100 LAMMPS, both as finite size and periodic infinite-extent sheets. The minimized 101 geometries yielded a=19.240 Å, b=13.365 Å and $\gamma=90.000^{\circ}$. The values of b and γ were in excellent agreement with those derived from semi-empirical quantum chemistry 102 103 calculations and previously reported in (19), with a somewhat larger deviation in the value 104 of a. MD optimized geometries displayed very good planarity, very similar to that of the 105 semi-empirical computations. Two nearly square sheets were generated with 7776 and 106 31104 atoms and spanning 13×9 and 24×18 unit cells of the 2D polymer, respectively. Their approximate sizes were 159.8 Å \times 159.5 Å and 319.7 Å \times 319.0 Å, respectively. 107

The MD simulations of infinite 2D sheets were performed in the isothermal-isobaric (NPT) ensemble with periodic boundary conditions. The calculations for finite size systems, were started with well-formed one, three or six nearly square polymer sheets, and were performed in the NVT ensemble. The equations of motion used in LAMMPS were those of Shinoda, Shiga, and Mikami (43) and the time integration schemes closely followed those derived in (44). Thermostating and barostating were implemented using Nosé-Hoover algorithm.

- 115 3. RESULTS AND DISCUSSION
- 116 **3.1. MD Simulations**

117 The properties of single sheets of polymeric carbon nitride are first analysed. These are 118 the basic building blocks of three-dimensional arrangements (15, 19) and also the desired 119 final products for many applications. These arrangements can potentially be obtained by 120 exfoliation of more complex materials or by direct growth. The study of the response of 121 infinite sheets to changes in temperature and pressure permit to quantify, in particular, the 122 behaviour of the hydrogen bond network as the temperature grows. The results obtained 123 are consistent with subsequent studies on the processes responsible for the morphological 124 organization of the material under treatment at high temperatures.

125 3.1.1. Infinite sheets

126 The intrinsic properties of ideal, infinite extent, 2D arrangements were addressed using 127 periodic boundary conditions in the NPT ensemble. Final trajectory equilibrium states for 128 simulations at zero pressure and T=100 K and T=700 K are displayed in Fig. 3. The results 129 show how a finite temperature induced height fluctuations in the sheet, similarly to the 130 case of graphene (32-33). The average absolute deviations from the surface level h are 131 shown in Fig. 2(a) for the range between 100 K and 1000 K. The results for the two sheets 132 of different size were very similar, with the same approximate linear dependence of h in 133 that range. As opposed to graphene, these height fluctuations have an important effect on 134 the structural integrity of the layer through their impact on the hydrogen bonds network, 135 as it can be appreciated by comparing subplots (a) and (b) with (c) and (d), respectively, 136 in Fig. 1. This can be clearly appraised in the dependence of the energy contributed by 137 hydrogen bonds normalized with the total number of heptazine units in the simulation 138 domain shown in Fig. 2(b). These results show some dependence with the size of the 139 sample under study. The contribution of the hydrogen bond network to lower the energy 140 of the system decreases as temperature is increased and part of these bonds are broken. 141 Further, the rate of change of this energy with T saturates at higher temperatures due to 142 the finite number of total hydrogen bonds.

The structural integrity of other non-hydrogen-bonded 2D materials is preserved under very large in-plane uniaxial or biaxial stress, resulting in a net change of the potential energy in the membrane (32). In the polymeric carbon nitride presented herein, simulations conducted even at low temperature show that increasing the applied finite uniaxial or biaxial in-plane stress eventually results in the collapse of the 2D polymer at a finite time that depends on the applied pressure. The atom displacements in the onset of the instabilities resemble those associated to the low frequency lattice modes of the 2D

150 polymer (19), which may play a key role in the nascent growth of the instabilities.

151 3.1.2. Finite size samples

152 The simulations performed on finite size samples show that, in general, the dynamic 153 trajectories leading to their final conformations are driven by two main effects: the 154 previously described weakening of the hydrogen bonds network as temperature increases, 155 plus the plane-to-plane interactions that take place after polymer sections are loosened or 156 released from their hydrogen bond restraints. These two combined effects produce the 157 folding and piling of the material layers in different manners, determining the final 158 arrangement of the material. The detailed evolution depends not only on the ambient 159 conditions, but also on the amount and the characteristics of the initially interacting 160 material.

161 The conformation dynamics of a finite single sheet of polymeric carbon nitride were 162 analysed from computations performed in the canonical ensemble. The final states of the 163 calculations of 500 psec evolutions of a layer with 13×9 unit cells in a volume of 164 $200 \times 200 \times 200$ Å³ at different temperatures are shown in Fig. 3. In all cases, equilibration 165 of the structures took place within the first 100 psec.

166 For the lowest temperatures in the sequence, 500 K and 600 K, the sheet essentially 167 kept its integrity. At 500 K only a very small defect in the 2D structure was found, 168 localized in one of the corners of the sheet, and corresponded to the disordered two-layer 169 piling of a section spanning very few heptazine units. At 600 K, a defect (a bilayer stripe) 170 could be identified. The dynamics of its formation, starting with the folding of one of the 171 corners and the sliding of the feature along the 2D structure, revealed a very interesting 172 balance between the inter-heptazine interaction and its hindrance by the hydrogen bonds 173 at this temperature.

174 At 700 K, the repetition of the folding and propagation of defect stripes produced the 175 packing of several layers. This disordered and non-homogeneous piling, in turn, induced 176 some curvature in the formed multilayer structures. As the temperature was increased, the 177 folding of the structure and the packing of multiple layers was facilitated by the 178 debilitation of the network of hydrogen bonds. The route to the layer piling then included 179 the break-up of stripes of material and their layer-to-layer interaction. As the hydrogen 180 bond network weakened with the temperature growth, the prominence of inter-planar 181 interaction increased and pilings with larger numbers of layers were produced.

182 When similar computations were performed on a planar system spanning 26×18 unit 183 cells, this larger sheet seemed to be more robust against the folding process at 700 K, with 184 morphological modifications that were limited to a somewhat smaller fraction of the total 185 sheet surface than in the previous case. This can be explained by the fact that the initiation 186 of the dynamical evolution eventually leading to the stacking and crumpling of the sheet 187 would be facilitated by the irregularities in the system and the processes would become 188 less likely in the more homogeneous system. At higher temperatures (900 K), the initial 189 breakup of the hydrogen links between 1D chains and their inter-planar interaction led to 190 the formation of a thread of 1D chains stacked in the direction perpendicular to the 191 heptazine planes.

Fig. 4 shows the results obtained from the equilibration in the NVT ensemble of three 26×18 sheets, which were initially widely separated and with a random disposition. The computations were performed at different temperatures and, in all cases, the duration of the simulation is 500 psec. Equilibrium was reached within the first 200 psec. At the lower temperatures, 500 K and 700 K, the evolution was dominated by the interaction between the well-formed layers, which resulted in their piling in an essentially three-layer system. When the temperature was increased to 900 K, the thermally-driven sheet undulations broke parts of the hydrogen-bond networks, loosening portions of the initial sheets. The subsequent plane-to-plane interactions of the fragmented material resulted in the formation of an irregular flake with a curved profile. The number of layers at some locations of this disordered cell was larger than three.

203 The results derived from the evolution in the canonical ensemble were very similar 204 when the initial conditions consisted of six identical, well separated and placed at random 205 sheets. The time required to reach an equilibrium state in this case was significantly larger. 206 As in the previous case of three sheets in the starting conditions, the evolution at the lower 207 temperatures produced the piling of the initially well-formed layers. The system dynamics, in this regime, were dominated by the layer-to-layer interactions. At 900 K, the 208 209 thermally induced shaking and break-up of the initial sheets and the following layer-to-210 layer interaction of the resulting fragments finally produced a curved and disordered flake 211 with a number of stacked layers that, at some places, was larger than the initial figure of 212 six. Fig. 5 shows the resulting flake after being annealed by cooling it at low temperature. The top view in Fig. 5(a) displays a turbostratic arrangement of the layers of material 213 214 within the flake, consistent with the experimental observations. The side view of Fig. 5(b) 215 shows the curvature of the flake and the existence of stacked layers that can clearly exceed 216 the number of six even at the edge of the flake. There are portions of the flake interior that 217 display a larger numbers of layers.

Fig. 6 shows the radial distribution function (RDF) of the atoms in the flake processed at 900 K. The RDF of a single sheet of material at very low temperature is also plotted for comparison purposes. This second curve has been arbitrarily scaled to facilitate its contrast. The main peaks due to the in-layer interatomic distances in the single sheet of material are also clearly distinguishable in the sample that has been processed at high temperature. Nevertheless, the disorder in the flake is evident from the broadening of these peaks. In the figure, the peak at 3.21 Å in the signal corresponding to the sample heated to 900 K has been highlighted. Even though there are nearby contributions from in-layer distances, the height of this peak relative to that of the low temperature single-sheet curve is much higher than in the neighbouring peaks. Therefore, this feature can be predominantly attributed to the interlayer stacking. The separation obtained from the calculations was in excellent agreement with the measured values at the corresponding processing temperature.

3.2. X-ray diffractometry (XRD), scanning (SEM), and transmission electron
 microscopy (TEM) observations

A marked corrugation of the polymeric carbon sheets, synthesized from melamine cyanurate (discussed in (19)), was confirmed at temperatures above 600 K. Particles with tubular shape, formed by crumpling of flake particles, were observed at synthesis temperatures around 700 K (Fig. 7).

In fact, at 900 K, the crumpled particles, formed not only tubular particles but also globular ones, as tubules were further shrunk up to form globules with holes (Fig. 7).

This characteristic should be linked to H-bonding, and -as simulations showed- borders
would be more instable and tend to crumple, transforming the flakes in either tubular or
globular particles.

It is noteworthy to point out that this folding effect is coupled to a tighter interlayer distance upon synthesis temperature increase, indicating that -concomitant to the borders' folding- the stacking order increases, as shown in Fig. 8.

Therefore, the increase in the number of stacked sheets and the larger interlayer interactions would be linked to the folding effect that produces crumpled particles. In a previous research, it was found that the layers behaved as independent ones (i.e., as a 2D material) below a synthesis temperature of 800 K (around 650 °C). Above this

temperature, the folding effect increased the number of stacked layers as well as the order along the c direction. In summary, the diminution of the interaction on the layer plane implicates a stronger interplanar H-bond interaction.

252 This corrugation effect could be conveniently used to produce globular, spherical or 253 tubular particles. Moreover, this effect of folding seemed to affect THz absorption in a 254 considerable way (19), so that it could be used as a sensor of pressure in special 255 environments. Hydrogen bond in carbon nitride (obtained from urea) was experimentally 256 investigated by Hu et al. (X) and they concluded that melem oligomers were formed and 257 arranged in layers. In spite of that study we suggest that the polymeric structure is compatible with the crumpled morphology. Hydrogen bonding weakening between 258 259 adjacent polymer ribbons is the main cause of the typical crumpled structure of this 260 material, especially observed at high synthesis temperatures (ca. 900 K). This process 261 makes the material more sensitive to the external environment through the possibility of 262 interacting by hydrogen bonding with molecules such as CO₂, etc.

263 **4. CONCLUSION**

A computational study of the morphological properties of a carbon nitride polymer was performed at the nanoscale using molecular dynamics. In spite of the fact that this survey was restrained by the limited size of the simulation domains, a good agreement with the experimental results was found. The synthesized samples consisted of crumpled nanosheets formed by a turbostratic arrangement of polymer layers, which is consistent with the small-size molecular dynamics simulations at the corresponding processing temperatures.

The analysis of the computer simulations results have provided a clear picture of the mechanisms that would lead to such arrangement. One determining factor is the weakness of the hydrogen bonding network that keeps together the 2D polymer and that determines

274 many of its physical and chemical properties(45). This fact had been previously noted, for 275 instance, in relation with the calculation of the electronic band-gap of this material. Due 276 to the very small interactions between heptazine units, the calculated band-gap for the 2D 277 system is very similar to that of the monomer (19). At sufficiently high temperatures, 278 molecular agitation can partly release this relatively weak bonding network and permit the 279 conformational change in the material. The second element is the effect of the laver-to-280 layer interaction, typical of 2D carbon materials, that drives such change. Hydrogen 281 bonding weakening between adjacent polymer ribbons is the main cause of the typical 282 crumpled structure of this material, especially observed at synthesis temperatures above 283 ca. 900 K. However, this property makes this material more sensitive to the external 284 environment, and thus more suitable to be used as a catalyst and a sensor.

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Fig. 1. Equilibrium polymer sheets at zero pressure obtained using periodic boundary
conditions. Left column results ((*a*) and (*c*)) correspond to a period of 13×9 unit cells;
right column ones ((*b*) and (*d*)) correspond to a period of 26×18 unit cells. Top row results
((*a*) and (*b*)) are for a temperature of T=100 K and lower row ones ((*c*) and (*d*)) for T=700
K.



425 Fig. 2. (*a*) Average height fluctuations and (*b*) hydrogen bond contribution to the system
426 energy per heptazine unit averaged over a simulation run after equilibration.



Fig. 3. Final trajectory states after 500 psec evolution in the NVT ensemble at different
temperatures. The initial condition in all cases is a ground state, approximately square,
sheet spanning over 13×9 unit cells of the 2D polymer.



433

434 Fig. 4. Initial (*t*=0) and equilibrium final trajectory states at different temperatures after435 the evolution of three sheets in the canonical ensemble.

(a) (b)

437

Fig. 5. A flake obtained by annealing the final equilibrium geometry after evolution in the
canonical ensemble at 900 K starting from six sheets of polymeric carbon nitride. (*a*) top
view, (*b*) side view of the flake edge. Plots have been produced using VMD software (46).





- 444 sheet of material at low temperature.



- 458 Fig. 7. TEM and SEM images of polymeric carbon nitride obtained at 700 and 900 K from
- 459 melamine cyanurate pyrolysis.
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462 Fig. 8. Variation of the interlayer distance and stacked sheets as a function of the synthesis463 temperature (19).