

THE UNIVERSITY of EDINBURGH

Edinburgh Research Explorer

Photoelectrocatalytic disinfection of water and wastewater

Citation for published version:

Venieri, D, Chatzisymeon, E, Politi, E, Sofianos, SS, Katsaounis, A & Mantzavinos, D 2013, Photoelectrocatalytic disinfection of water and wastewater: Performance evaluation by qPCR and culture techniques' Journal of Water and Health, vol. 11, no. 1, pp. 21-29. DOI: 10.2166/wh.2012.208

Digital Object Identifier (DOI):

10.2166/wh.2012.208

Link: Link to publication record in Edinburgh Research Explorer

Document Version: Early version, also known as pre-print

Published In: Journal of Water and Health

General rights

Copyright for the publications made accessible via the Edinburgh Research Explorer is retained by the author(s) and / or other copyright owners and it is a condition of accessing these publications that users recognise and abide by the legal requirements associated with these rights.

Take down policy The University of Edinburgh has made every reasonable effort to ensure that Edinburgh Research Explorer content complies with UK legislation. If you believe that the public display of this file breaches copyright please contact openaccess@ed.ac.uk providing details, and we will remove access to the work immediately and investigate your claim.



1	REFERENCE: D. Venieri, E. Chatzisymeon, E. Politi, S.S. Sofianos, A. Katsaounis, D. Mantzavinos,
2	Photoelectrocatalytic disinfection of water and wastewater: performance evaluation by qPCR and culture
3	techniques, Journal of Water and Health 11(1) (2013) 21-29. doi:10.2166/wh.2012.208
4	
5	
6	Photoelectrocatalytic disinfection of water and wastewater: performance evaluation by qPCR
7	and culture techniques
8	
9	
10	Danae Venieri ^{1*} , Efthalia Chatzisymeon ¹ , Eleonora Politi ¹ , Spiridon S. Sofianos ¹ , Alexandros
11	Katsaounis ² , Dionissios Mantzavinos ¹
12	
13	¹ Department of Environmental Engineering, Technical University of Crete, Polytechneioupolis,
14	GR-73100 Chania, Greece
15	² Department of Chemical Engineering, University of Patras, GR-26504 Patras, Greece
16	
17	Short title: Photoelectrocatalytic disinfection of water and wastewater

^{*} Corresponding author:

Email: danae.venieri@enveng.tuc.gr; Tel.: +302821037801; Fax: +302821037858

18 Abstract

Photoelectrocatalytic oxidation (PEC) was evaluated as a disinfection technique using water and 19 secondary treated wastewater spiked with Escherichia coli and Enterococcus faecalis. PEC 20 experiments were carried out using a TiO₂/Ti-film anode and a zirconium cathode under simulated 21 solar radiation. Bacterial inactivation was monitored by culture and quantitative PCR (qPCR). 22 Inactivation was enhanced increasing the duration of the treatment and decreasing bacterial 23 population and the complexity of the aqueous matrix. E. coli cells were reduced approximately 6 24 25 orders of magnitude after 15min of PEC treatment in water at 2V of applied potential and an initial concentration of 10⁷ CFU/mL; pure photocatalysis led to about 5 log reduction, while 26 27 electrochemical oxidation alone resulted in negligible inactivation. The superiority of PEC relative to PC can be attributed to a more efficient separation of the photogenerated charge carriers. 28 Regarding disinfection in mixed bacterial suspensions, E.coli was more susceptible than E.faecalis 29 at a potential of 2V. The complex composition of wastewater affected disinfection efficiency, 30 yielding lower inactivation rates compared to water treatment. qPCR yielded lower inactivation 31 32 rates at longer treatment times than culture techniques presumably due to the fact that the latter do not take into account the viable but not culturable state of microorganisms. 33 34

35 Keywords: *E. coli* \cdot *E. faecalis* \cdot Disinfection \cdot qPCR \cdot TiO₂ \cdot Voltage

36 INTRODUCTION

The growing demand for clean water and wastewater, free from disease-causing pathogenic 37 microorganisms, poses new challenges in the development of effective disinfection techniques. 38 Over the last decades, advanced oxidation processes (AOPs) have been recognised as an emerging 39 group of techniques, highly effective for the decomposition and mineralization of organic 40 compounds in aqueous samples (Malato et al. 2009; Chong et al. 2010). TiO₂ photocatalysis is an 41 important member of AOPs and its benefits regarding water and wastewater disinfection have been 42 demonstrated with respect to Escherichia coli, Staphylococcus aureus and Enterococcus faecalis 43 (Chen et al. 2010; Chatzisymeon et al. 2011; Rémy et al. 2011; Venieri et al. 2011). In most 44 photocatalytic applications the catalyst is employed as slurry of fine particles in a photochemical 45 reactor, resulting in certain difficulties, such as post-reaction catalyst recovery and low quantum 46 efficiencies (Koivunen & Heinonen-Tanski 2005; Egerton et al. 2006; Marugán et al. 2006). These 47 problems may be addressed by immobilization of TiO₂ on a conducting support and application of a 48 potential bias, so as to reduce the recombination of charge carriers, which is the main limitation of 49 the process photonic efficiency (Marugán et al. 2009; Egerton 2011). 50

These modifications have led to the development of photoelectrocatalysis (PEC), which 51 consists of the application of a constant bias anodic potential usually to a TiO₂-based thin film 52 acting as a photoanode. The photoinduced electrons are then continuously extracted from the anode 53 by an external electrical circuit to be injected into the cathode (Sirés & Brillas 2011). The constant 54 current density or bias potential applied to the semiconductor electrode promotes the efficient 55 separation of electron-hole pairs and accelerates the production of photogenerated oxidizing species 56 onto the catalyst surface (Martínez-Huitle & Brillas 2009). In this view, it is not surprising that PEC 57 has recently found a wide range of environmental applications, including degradation of organic 58 pollutants and pharmaceuticals, detoxicifation of aqueous samples, wastewater remediation and 59 microbial inactivation (Fraga et al. 2009; Martínez-Huitle & Brillas 2009; Nissen et al. 2009; 60 Egerton 2011; Frontistis et al. 2011; Li et al. 2011; Sirés & Brillas 2011). 61

PEC seems to be a promising and efficient tool, regarding the effective inactivation of 62 pathogens which are contained in water and wastewater, contributing in the control of waterborne 63 diseases. The bactericidal function of this technique can be attributed to the oxidation properties of 64 photocatalytically generated active oxygen species (AOS), which cause damage to cellular 65 membrane and further destruction of bacterial structure (Li et al. 2011). The studies referred to PEC 66 as means of disinfection highlight the importance of certain parameters like applied voltage, 67 bacterial concentration, treatment time and the aqueous matrix, which are considered determining 68 factors of microbial inactivation (Martínez-Huitle & Brillas 2009). Generally, the overall evaluation 69 of PEC, as a method for disinfection, is mainly performed with the use of a faecal bacterial 70 71 indicator, applying conventional microbiological protocols, based colony-forming units (CFU) counts on selective culture media (Egerton et al. 2006; Philippidis et al. 2010; Egerton 2011; Li et 72 al. 2011). Yet, the environment of PEC treatment may be considered rather stressful for microbial 73 populations, inducing their entrance to the viable but not culturable (VBNC) state. Within this state, 74 bacteria demonstrate metabolic activity and maintain their pathogenic features but they are not 75 recoverable in culture media, leading to false negative results (Muela et al. 2008). The application 76 of molecular methods surpasses this limitation, based on nucleic acid identification and 77 quantification. Among them quantitative PCR (qPCR) has proven to be sensitive yielding accurate 78 79 quantitative results and allowing for new approaches in waterborne pathogen research (Lee et al. 2006; Shannon et al. 2007). 80

81 The aim of this work was to investigate the inactivation of faecal indicators in water and 82 wastewater by means of photoelectrocatalytic oxidation, applying conventional cultural methods 83 and qPCR. This was implemented using representative strains of faecal indicator bacteria, namely 84 *Escherichia coli* and *Enterococcus faecalis*. The effect of operational conditions such as bacterial 85 concentration, potential, treatment time, and aqueous matrix were also examined.

87 MATERIALS AND METHODS

88

89 Bacterial strains and wastewater

90 The bacterial strains used as reference faecal indicators in the present study were Gramnegative E. coli ATCC 23716 and Gram-positive E. faecalis ATCC 14506 (American Type Culture 91 Collection, Rockville, Md. USA). Suspensions of bacterial indicators were prepared in sterile 92 distilled water, which was used as sample for water disinfection experiments. The concentration of 93 94 bacterial cells in the suspension was estimated measuring its optical density at 600 nm (Shimadzu UV1240 spectrophotometer). The bacterial concentrations used in this study were within the range 95 of 10^4 - 10^8 CFU/mL. Plate counts were also performed for accurate bacterial count. 96 Wastewater disinfection experiments were carried out in real wastewater collected from the 97 outlet of the activated sludge unit (prior to chlorination) of the municipal wastewater treatment plant 98 of Chania, W. Crete, Greece. Its main characteristics were determined according to Standard 99 Methods (1999) as follows: the chemical oxygen demand and dissolved organic carbon were 26 and 100 101 7.8 mg/L, respectively, the concentration of chlorides, sulfates, nitrates, nitrites, bicarbonates and total solids were 222.1, 60.3, 25.9, 57.1, 182.1 and 7 mg/L, respectively, while the pH was 7.8. 102 Sterilization of the sample was performed prior to any photoelectrocatalytic treatment in order to 103 104 adjust the initial bacterial concentration at the desired level.

105

106 **Disinfection experiments**

Photocatalytic experiments were performed using a solar radiation simulator (Newport, model 96000) equipped with a 150 W xenon ozone-free lamp and an Air Mass 1.5 Global Filter (Newport, model 81094), simulating solar radiation reaching the surface of the earth at a zenith angle of 48.2°. According to the spectral irradiance data given by the manufacturer, simulated solar radiation contains about 5% UV-A radiation, and 0.1% UV-B radiation, while the filter cuts radiations with wavelengths lower than 280 nm. The incident radiation intensity on the

113	photochemical reactor in the UV region of the electromagnetic spectrum was measured
114	actinometrically using 2-nitrobenzaldehyde (Sigma-Aldrich) as the chemical actinometer (Willett &
115	Hites 2000; Galbavy <i>et al.</i> 2010) and it was found to be 5.8×10^{-7} einstein/(L×s), which corresponds
116	to an irradiance of 1.31×10^{-2} W/m ² . Reactions took place in an open, double-walled, cylindrical
117	glass vessel under continuous stirring.
118	Constant potentials (2 and 5V) were applied between the TiO_2/Ti anode and the zirconium
119	cathode using a galvanostat-potensiostat (Amel Instruments, model 2053). Details concerning the
120	experimental setup, as well as the preparation and characterization of the TiO_2/Ti -film are given
121	elsewhere (Frontistis et al. 2011). In brief, the anode surface was 5.8 cm ² and had 75:25
122	anatase:rutile phase composition. The reactant mixture had a liquid holdup of 60 mL. The reaction
123	temperature was maintained constant at 25±1°C with a temperature control unit. All experiments
124	were conducted in the presence of 1% (w/v) Na_2SO_4 aqueous solution serving as the electrolyte.
125	Water samples were processed applying (a) photoelectrocatalytic treatment, (b)
126	photocatalytic treatment (PC: solar radiation alone) and (c) electrochemical oxidation (EO:
127	application of voltage only in dark conditions). Photoelectrocatalytic treatment was also applied for
128	disinfection of wastewater samples. All disinfection experiments were performed in triplicate. In a
129	typical experiment, the inoculated aqueous solution was loaded to the cell, left in the dark under
130	stirring for 20 min in order to equilibrate and then exposed to solar radiation and/or potential. At
131	specific time intervals about 3 mL of the reaction solution were withdrawn. Half of the quantity was
132	immediately analyzed with respect to viable bacterial cells applying conventional culture method
133	and the rest was used for DNA extraction and PCR amplification.
134	

135 Culture method

The detection and enumeration of *E. coli* and *E. faecalis* in the solution were performed
using the serial dilution streak plate procedure. The media used in the study were HiCrome
Coliform Agar (HiMedia Laboratories) and Slanetz & Bartley medium (OXOID) for *E. coli* and *E.*

faecalis, respectively. Incubation took place at 37°C. Bacterial counts were performed after 20-24 h
and 48 h for *E. coli* and *E. faecalis* determination, respectively.

141

142 **DNA extraction**

Genomic DNA was extracted performing chemical lysis and phenol/chloroform/isoamyl 143 alcohol (25:24:1) extraction. Namely, the cells were spun for 2 min and were lyzed for 1 h at 37°C 144 with 300 µL of lysozyme lysis buffer (100 mM NaCl, 500 mM Tris [pH 8], lysozyme 10 mg/mL) 145 and 3 µL of 20 mg/ml proteinase K. Then, 200 µL of SDS lysis buffer (100 mM NaCl, 500 mM Tris 146 [pH 8], 10% [w/v] SDS) were added, followed by incubation at 65°C for 10 min. The solution was 147 148 extracted with 750 µL of chloroform/isoamyl alcohol (24:1), spun, and the aqueous phase was reextracted with phenol/chloroform/isoamyl alcohol (25:24:1). Ethanol purification step was 149 performed and the quantity and purity of all DNA samples were determined measuring their 150 absorbance at 260 nm and estimating the ratio of absorbance values at 260 nm and 280 nm, 151 respectively. 152

153

154 Monitoring of *E. coli* and *E. faecalis* through qPCR

The SYBR green method was chosen for the quantification of both bacterial indicators using 155 the StepOne Plus System (Applied Biosystems Inc., Foster City, CA, USA). The gadAB gene was 156 used as a target for *E. coli* quantification through qPCR, and the primers were as follows: forward 157 primer 5'-GCG TTG CGT AAA TAT GGT TGC CGA-3' (gadrt-1) and reverse primer 5'-CGT 158 CAC AGG CTT CAA TCA TGC GTT-3' (gadrt-2) (Chen et al. 2006). The product size for this 159 primer set is 305 bp. The set of primers for *E. faecalis* detection and quantification was designed 160 according to the sequences of 16S rRNA, which are available in GenBank (Bartosch et al. 2004). 161 The primer pair contained a forward primer 5'-AACCTACCCATCAGAGGG -3' (Efs130F) and a 162 reverse primer 5'-GACGTTCAGTTACTAACG -3' (Efs490R), which yield a 360 bp PCR product. 163 Triplicate PCR reactions were carried out with Quantimix Easy SYG Kit (Biotools) to a final 164

volume of 20 µL. The mixed qPCR solution contained 2XPCR master mix, 0.5 mM of each primer, 165 30 nM of reference dye (Rox) all diluted to the final volume of the reaction mixture with 166 DNase/RNase free water. PCR reactions were carried out at a temperature profile of 10 min initial 167 168 denaturation at 95°C, followed by 40 cycles each of denaturation at 95°C for 0.5 min, annealing at 57°C for 1 min, and extension at 72°C for 0.5 min. Deionized water and DNase-treated E. faecalis 169 and E. coli served as negative controls. Also, melt curve analyses were conducted from 55 to 95°C. 170 To determine the detection sensitivity of the qPCR assay, a series of 10-fold diluted pure 171 culture genomic DNA of each bacterium was tested for qPCR amplification and cycle threshold 172 (C_T). Standard curves were generated with *E. coli* and *E. faecalis* DNA. 173

174

175 RESULTS AND DISCUSSION

176

Efficiency of photoelectrocatalysis in relation to photocatalysis and electrochemical oxidation The decrease of bacterial population in water samples was investigated under three distinct conditions: (a) photoelectrocatalytic treatment (PEC: simultaneous application of solar radiation and 2V bias potential); (b) photocatalytic treatment (PC: solar radiation alone); (c) electrochemical oxidation (EO: application of voltage only in dark conditions).

The results, summarized in Fig. 1, show that the application of a 2V potential during PEC increased 182 the extent of disinfection compared to PC. For instance, PEC led to a 5.9 log reduction after 15 min, 183 with the respective values of PC and EO being 4.9 and 0.2. However, both PEC and PC resulted in 184 an approximately 7 log reduction after 120 min of treatment, while in the absence of radiation 185 bacterial inactivation was inadequate (i.e. 1.5 log reduction); this clearly shows that EO is not a 186 suitable disinfection process at the conditions under consideration as it is not able to achieve total E. 187 coli inactivation in aqueous suspensions. Control runs were also performed in the absence of 188 radiation and potential showing no change in bacterial population after 120 min of contact time 189 190 (data not shown).

191	Illumination of a semiconductor-electrolyte interface with photons having energy greater
192	than its band gap energy generates electron-hole pairs at the anode electrode surface. The
193	simultaneous application of a bias positive to the flat-band potential produces a bending of the
194	conduction and valence bands which, in turn, causes a more effective separation of the
195	photogenerated carriers within the space charge layer (Morrison 1980). The potential gradient
196	forces the electrons towards the cathode, thus leaving the photogenerated holes to react at the anode
197	with H ₂ O and/or OH ⁻ to yield hydroxyl radicals, i.e.:
198	Anode (working electrode):

199
$$\operatorname{TiO}_2 + h\nu \rightarrow \operatorname{TiO}_2 - e_{cb}^{-} + \operatorname{TiO}_2 - h_{vb}^{+}$$
 (1)

200
$$\operatorname{TiO}_2 - h_{vb}^+ + H_2O_s \rightarrow \operatorname{TiO}_2 - \operatorname{OH}_s + H^+$$
 (2)

201
$$\operatorname{TiO}_2 - h^+_{vb} + OH^-_s \rightarrow TiO_2 - OH_s$$
 (3)

202 $\text{TiO}_2 - e_{cb}^- + \text{TiO}_2 - h_{vb}^+ \rightarrow \text{recombination}$

203 *Cathode (counter electrode):*

$$204 \qquad 2H_2O + 2e^- \rightarrow H_2 + 2OH^- \tag{5}$$

where the subscripts cb and vb denote the conduction and valence bands, respectively, h^+ and $e^$ denote the photogenerated holes and electrons, respectively, while the subscript s refers to species adsorbed onto the photoanode surface.

(4)

Solar radiation has the potential to inactivate bacterial cells. Moreover, in the presence of a 208 catalyst, disinfection is mainly achieved by the action of oxidative radicals released from irradiated 209 TiO₂ (Chen et al. 2010). PEC under sunlight radiation has better performance than PC since the 210 application of potential is believed to suppress the rate of electron-hole recombination, thus 211 212 enhancing photocatalytic rates (Baram et al. 2009). The holes which have been left by electrons are 213 accelerated into the semiconductor surface, where they are available to react with organisms in the water. The obtained results are in accordance to other studies which highlight PEC as an efficient 214 disinfection method, in terms of inactivating mainly E. coli and Gram-negative bacteria (Philippidis 215 et al. 2010; Rahmawati et al. 2011). Generally, the benefits of electric field enhancement have been 216

demonstrated not only for coliforms but also for *Clostridium perfringens* spores and the recalcitrant 217 Cryptospiridium parvum (Egerton et al. 2006).

219

218

220 Effect of initial bacterial concentration

The influence of the initial E. coli concentration on PEC disinfection ability is presented in 221 Fig. 2. Conventional culture technique (Fig. 2a) showed that the rate of PEC disinfection was 222 inversely proportional to E. coli concentration in water samples. Total inactivation was achieved in 223 relatively short treatment time (i.e. within approximately 15 min) only when bacterial inoculum 224 contained 10⁵ CFU/mL, while at higher concentrations residual E. coli cells were determined even 225 after 60 min. For instance and at an initial concentration of 10^8 CFU/mL, a 5.5 log reduction was 226 achieved within 30-60 min of treatment, beyond which no further inactivation was recorded. 227

SYBR green qPCR was performed to detect the potential presence of live non-culturable 228 bacterial cells. Our effort was to evaluate the consistency of E. coli cells and taking into account 229 that cultivability is not synonymous of viability (Rémy et al. 2011). gPCR showed different periods 230 of microbial inactivation (Fig. 2b) compared to plate counts. Inactivation rates were quite similar 231 when initial genome copies were of 10^7 and 10^5 /mL, as there was no enhancement of the process 232 after 40 min of treatment. During PEC, cell injury and loss of viability are achieved as cellular 233 membrane is destructed through direct contact with the photoanode (Li et al. 2011). Furthermore, 234 radiation induces DNA lesions, damaging nucleic acids and making them functionless. The 235 hydroxyl radical (and other AOS), directly generated by this process, is the main cause of DNA 236 destruction, which in turn leads to cell death (Sinha & Häder 2002; Gogniat & Dukan 2007). 237 According to our results, even after 120 min of treatment, genome copies were detected and 238 239 quantified in the water sample, reaching a plateau without any further decrease. This profile of genome copies in the reaction mixture could be attributed to the detection sensitivity of the method. 240 qPCR is acknowledged as a reliable and sensitive molecular method, whose detection limit may be 241 in the region of 100 fg of E. coli genomic DNA (Lleo et al. 2005; Shannon et al. 2007). However, 242

in the present study environmental samples were processed which may inhibit qPCR or restrict its 243 detection limits. On the other hand, qPCR seems to be valuable in detecting the non-culturable 244 bacterial strains, which are induced under the stressed conditions of PEC. The so-called viable but 245 246 non culturable (VBNC) state of microorganisms is very common when dealing with environmental samples or disinfection techniques, making plating counts inadequate for accurate bacterial 247 enumeration. Bacteria in this state retain their metabolic activity and pathogenic features, posing 248 danger for public health, while they are not recoverable in standard culture media. On the contrary, 249 250 molecular biology methods are capable of detecting these strains contributing in a more reliable microbial evaluation of environmental samples (Wéry et al. 2008). Nevertheless, certain attention 251 252 should be paid on designing PCR experiments, including the amplicon length of the reactions, since it is strongly correlated to the recorded disinfection efficiency. In the present study, SYBR green 253 PCR reactions vielded a 305 bp product for *E. coli* identification and quantification, resulting in 254 reliable and acceptable evaluation of PEC disinfection efficiency (Süb et al. 2009). Also, other 255 points under consideration concerning qPCR is that reactions can be affected by nucleic acid 256 contamination, leading to false positive results and the formation of primer - dimers. Some of the 257 possible sources of contamination are cross-contamination between samples, contamination from 258 laboratory equipment and carryover contamination of amplification products and primers from 259 previous PCRs. Therefore, appropriate measures should be taken to increase the reactions 260 specificity and all experiments should be performed with extra care to minimize contamination 261 risks. 262

263

264 Effect of mixed bacterial populations

In a set of experiments, an attempt was made to evaluate the disinfection efficiency of PEC at 2V in water samples containing two bacterial populations. In this sense, sterile water samples were inoculated with either 10^7 CFU/mL *E. coli* or 10^7 CFU/mL *E. faecalis* or both. The results from culture technique and qPCR are summarized in Fig. 3.

Screening the inactivation profiles of the tested bacterial strains individually, E. coli seems 269 to be less resistant than E. faecalis at the experimental conditions in question. This outcome, which 270 is more obvious in qPCR results, was quite expected and in agreement with other studies dealing 271 272 with the inactivation of various bacterial populations (Cho et al. 2011; Li et al. 2011). Enterococci are Gram positive bacteria, possessing a thick peptidoglycane cell wall and an additional outer 273 membrane containing two lipid bilayers, which provide them high complexity and potential of 274 preserving their viability during PEC treatment. On the other hand, E. coli cells require longer 275 276 treatment periods so as to achieve equal inactivation rates. In this sense, attention should be paid to the differences among main aquatic microbial indicators, which may lead to different PEC 277 278 inactivation efficiency (Cho et al. 2011).

When E. coli and E. faecalis were inoculated together in water samples, the extent of 279 inactivation for either strain was lower than that of their individual treatment. This may be 280 attributed partially to the elevated initial bacterial concentration, whose influence on PEC has been 281 discussed previously (Fig. 2). Other possible explanations would include the presence of 282 competitive microorganisms and the interaction amongst them and/or the competition for AOS 283 between the bacteria and the organic by-products, released to the solution by the inactivated 284 bacteria (Baram et al. 2009). In the present study, although PEC seems to be capable of inactivating 285 the used faecal indicators to a certain extent, the residual cells raise concerns about the suitability of 286 the method for disinfection of complex samples, containing various bacterial populations. 287

These findings are more obvious in qPCR results (Fig. 3b), which reveal longer periods required for bacterial decrease. In the present study, considerable genome copies/mL were recorded after 90 min of treatment, which is in contrast to plate counts. These extended periods could reflect the time necessary to mutate DNA to a point that can no longer be assayed. Cells may have been completely inactivated long before that point. On the other hand, concerns are raised about the suitability of the PCR method to establish a correlation between amplified bacterial DNA and viable *E. coli* and *E. faecalis*. However, because the half-life of the DNA released in the environment is

considered to be very short owing to the presence of numerous nucleases it might be deduced thatthe DNA detected is that contained in non-culturable cells (Lleo *et al.* 2005).

297

298 Disinfection in real wastewater

PEC was also employed to disinfect biologically treated effluents taken just before the chlorination step. Sterilized samples were inoculated with approximately 10^7 CFU/mL or 10^4 CFU/mL *E. coli* (Fig. 4). In this set of experiments, a higher value of applied potential (5V) was chosen to highlight its effect on bacterial inactivation. Given that (i) the aqueous matrix was real wastewater, which is generally considered as a "complex" sample containing various organic and inorganic components and (ii) raising the anodic potential enhances photocatalytic rates (Baram *et al.* 2009), runs were performed at 5V.

As seen in Fig. 4a, total E. coli inactivation occurred in almost 15 min when the initial cell 306 density was 10⁴ CFU/mL and this increased to 90 min at 10⁷ CFU/mL. The degree of disinfection 307 was also determined by means of qPCR, estimating bacterial genome copies remained in the 308 aqueous solution after treatment (Fig. 4b). Comparing findings from both procedures there was a 309 significant contrast, as already seen in previous experiments conducted in the present study. qPCR 310 showed that even after 90 min of PEC treatment at 5V potential, genome copies of the bacterial 311 indicator decreased by only 5 orders of magnitude, while colonies of E. coli were not recoverable at 312 the same time. Apart from factors such as the detection limit of the method or VBNC cells, which 313 have been mentioned previously, an important parameter under consideration is the particulate 314 matter present in wastewater. This aids in the resistance of microorganisms to disinfection, as it 315 may interfere by physically shielding bacterial cells and protecting the integrity of the contained 316 DNA. Therefore, the extent of DNA damage is limited, which inevitably results in high numbers of 317 genome copies detected by qPCR, even after long periods of treatment. In the present study, 318 considerable genome copies/mL were recorded, highlighting the importance of wastewater 319 composition and its bacterial content with respect to PEC efficiency. In this sense, plate counts 320

should be matched to qPCR results in order to perform a reliable and complete evaluation of

322 disinfection techniques, such as PEC and other photocatalytic treatments.

Furthermore, part of the photogenerated AOS may be wasted to attack the organic carbon of the wastewater (about 8 mg/L which typically consists of highly resistant humic-type compounds and biomass-associated products) and/or scavenged by bicarbonates, sulfates and chlorides (whose collective concentration is about 460 mg/L) rather than inactivate pathogens. This could be overcome increasing AOS concentration through raising the applied potential (Baram *et al.* 2009) and would possibly justify the nearly identical *E. coli* profiles shown in Figs. 2a and 4a at an initial concentration of 10^7 CFU/mL.

330

331 Conclusions

Photocatalytic inactivation of *E. coli* in water samples is enhanced applying a positive potential
 on TiO₂/Ti-films under simulated solar radiation. At the conditions in question, initial bacterial
 density plays an important role, as it affects adversely the required treatment time for complete
 microbial inactivation.

E. faecalis are more resistant than *E. coli* bacteria during PEC treatment in water samples.
 Furthermore, the disinfection efficiency is directly affected by the sample composition and the
 contained mixed bacterial populations. In complex samples, bacteria are more physically
 protected, resulting in lower inactivation rates.

There was a considerable contrast between colony counts and qPCR results, concerning the
 required time for total bacterial inactivation. qPCR data revealed longer periods required for
 complete bacterial inactivation, compared to the time estimated by culture method. Stressed
 conditions generated during PEC treatment may induce VBNC state of bacteria, which could
 explain the discrepancy between the applied quantitative methods. Additionally, in order to
 establish the accuracy and reliability of qPCR certain factors should be under consideration.

- These include detection limits of the method, amplicon size, primer sequence, cycling conditions
- 347 and inhibitors present in the processed sample.
- For large scale applications, the use of sunlight as renewable energy source would undoubtedly
- be beneficial to induce photocatalytic reactions, as well as provide the necessary electric field,
- thus promoting the principle of sustainable development.
- 351

References

353	Baram, N., Starosvetsky, D., Starosvetsky, J., Epshtein, M., Armon, R. & Ein-Eli, Y. 2009
354	Enhanced inactivation of <i>E. coli</i> bacteria using immobilized porous TiO ₂
355	photoelectrocatalysis. Electrochim. Acta. 54, 3381-3386.
356	Bartosch, S., Fite, A., Macfarlane, G. T. & McMurdo, M. E. T. 2004 Characterization of bacterial
357	communities in feces from healthy elderly volunteers and hospitalized elderly patients by
358	using real-time PCR and effects of antibiotic treatment on the fecal microbiota. Appl. Environ.
359	Microbiol. 70, 3575-3581.
360	Chatzisymeon, E., Droumpali, A., Mantzavinos, D. & Venieri, D. 2011 Disinfection of water and
361	wastewater by UV-A and UV-C irradiation: application of real-time PCR method.
362	Photochem. Photobiol. Sci. 10, 389-395.
363	Chen, Y. C., Higgins, M. J., Maas, N. A. & Murthy, S. N. 2006 DNA extraction and Escherichia
364	coli quantification of anaerobically digested biosolids using the competitive touchdown PCR
365	method. Water. Res. 40, 3037-3044.
366	Chen, C. Y., Wu, L. C., Chen, H. Y. & Chung, Y. C. 2010 Inactivation of Staphylococcus aureus
367	and Escherichia coli in water using photocatalysis with fixed TiO ₂ . Water. Air. Soil. Pollut.
368	212 , 231-238.
369	Cho, M., Cates, E. L. & Kim, J. H. 2011 Inactivation and surface interactions of MS-2
370	bacteriophage in a TiO ₂ photoelectrocatalytic reactor. <i>Water. Res.</i> 45 , 2104-2110.
371	Chong, M. N., Jin, B., Chow, C. W. K. & Saint, C. 2010 Recent developments in photocatalytic
372	water treatment technology: A review. Water. Res. 44, 2997-3027.
373	Egerton, T. A., Kosa, S. A. M. & Christensen, P. A. 2006 Photoelectrocatalytic disinfection of E.
374	coli suspensions by iron doped TiO ₂ . Phys. Chem. Chem. Phys. 8, 398-406.
375	Egerton, T. A. 2011 Does photoelectrocatalysis by TiO ₂ work? J. Chem. Technol. Biotechnol. 86,
376	1024-1031.

377	Fraga, L. E., Anderson, M. A., Beatriz, M. L. P. M. A., Paschoal, F. M. M., Romão, L. P. & Zanoni,
378	M. V. B. 2009 Evaluation of the photoelectrocatalytic method for oxidizing chloride and
379	simultaneous removal of microcystin toxins in surface waters. Electrochimica. Acta. 54,
380	2069-2076.
381	Frahm, E. & Obst, U. 2003 Application of the fluorogenic probe technique (TaqMan PCR) to the
382	detection of Enterococcus spp. and Escherichia coli in water samples. J. Microbiol. Methods.
383	52, 123-131.Chong, M. N., Jin, B., Chow, C. W. K. & Saint, C. 2010 Recent developments in
384	photocatalytic water treatment technology: A review. Water. Res. 44, 2997-3027.
385	Frontistis, Z., Daskalaki, V. M., Katsaounis, A., Poulios, I. & Mantzavinos, D. 2011
386	Electrochemical enhancement of solar photocatalysis: Degradation of endocrine disruptor
387	bisphenol-A on Ti/TiO ₂ films. Water. Res. 45, 2996-3004.
388	Galbavy, E. S., Ram, K. & Cort Anastasio, C. 2010 2-Nitrobenzaldehyde as a chemical actinometer
389	for solution and ice photochemistry. J. Photochem. Photobiol. A-Chem. 209, 186-192.
390	Gogniat, G. & Dukan, S. 2007 TiO ₂ Photocatalysis causes DNA damage via fenton reaction-
391	generated hydroxyl radicals during the recovery period. Appl. Environ. Microbiol. 73, 7740-
392	7743.
393	Koivunen, J. & Heinonen-Tanski, H. 2005 Inactivation of enteric microorganisms with chemical
394	disinfectants, UV irradiation and combined chemical/UV treatments. Water. Res. 39, 1519-
395	1526.
396	Lee, D. Y., Shannon, K. & Beaudette, L. A. 2006 Detection of bacterial pathogens in municipal
397	wastewater using an oligonucleotide microarray and real-time quantitative PCR. J. Microbiol.
398	Methods. 65, 453-467.
399	Li, G., Liu, X., Zhang, H., An, T., Zhang, S., Carroll, A. R. & Zhao, H. 2011 In situ
400	photoelectrocatalytic generation of bactericide for instant inactivation and rapid
401	decomposition of Gram-negative bacteria. J. Catal. 277, 88-94.
402	Lleo, M. M., Bonato, B., Tafi, M. C., Signoretto, C., Pruzzo, C. & Canepari, P. 2005 Molecular vs 17

- 403 culture methods for the detection of bacterial faecal indicators in groundwater for human use.
 404 *Lett. Appl. Microbiol.* 40, 289-294.
- Malato, S., Fernandez-Ibanez, P., Maldonado, M. I., Blanco, J. & Gernjak, W. 2009
 Decontamination and disinfection of water by solar photocatalysis: Recent overview and
- 407 trends. *Catal. Today.* **147**, 1-59.
- Martínez-Huitle, C. A. & Brillas, E. 2009 Decontamination of wastewaters containing synthetic
 organic dyes by electrochemical methods: A general review. *Appl. Catal. B: Environ.* 87,
 105-145.
- Marugán, J., Hufschmidt, D., Sagawe, G., Selzer, V. & Bahnemann, D. 2006 Optical density and
 photonic efficiency of silica-supported TiO₂ photocatalysts. *Water. Res.* 40, 833-839.
- 413 Marugán, J., Christensen, P., Egerton, T. & Purnama, H. 2009 Synthesis, characterization and
- 414 activity of photocatalytic sol–gel TiO₂ powders and electrodes. *Appl. Catal. B: Environ.* 89,
 415 273-283.
- 416 Morrison, S. R. 1980 Electrochemistry at semiconductors and oxidized metal electrodes. Plenum
 417 Press.
- Muela, A., Seco, C., Camafeita, E., Arana, I., Orruño, M., López, J. A. & Barcina, I. 2008 Changes
 in *Escherichia coli* outer membrane subproteome under environmental conditions inducing
 the viable but nonculturable state. *FEMS. Microbiol. Ecol.* 64, 28-36.
- 421 Nissen, S., Alexander, B. D., Dawood, I., Tillotson, M., Wells, R. P. K., Macphee, D. E. & Killham,
- 422 K. 2009 Remediation of a chlorinated aromatic hydrocarbon in water by
- 423 photoelectrocatalysis. *Environ. Pollut.* **157**, 72-76.
- 424 Philippidis, N., Nikolakaki, E., Sotiropoulos, S. & Poulios, I. 2010 Photoelectrocatalytic
- 425 inactivation of *E. coli* XL-1 blue colonies in water. *J. Chem. Technol. Biotechnol.* 85, 1054426 1060.
- 427 Rahmawati, F., Kusumaningsih, T. & Hastuti, A. 2011 Photo- and electro-catalysis for solar
- 428 disinfection of *Escherichia coli*-contaminated water using Ag–TiO₂/graphite. *Toxicol*.

- 429 *Environ. Chem.* **93**, 1602-1612.
- 430 Rémy, S. P., Simonet, F., Cerda, E. E., Lazzaroni, J. C., Atlan, D. & Guillard, C. 2011
- 431 Photocatalysis and disinfection of water: Identification of potential bacterial targets. *Appl.*432 *Catal. B: Environ.* 104, 390-398.
- 433 Shannon, K. E., Lee, D. Y., Trevors, J. T. & Beaudette, L. A. 2007 Application of real-time
- quantitative PCR for the detection of selected bacterial pathogens during municipal
 wastewater treatment. *Sci. Total. Environ.* 382, 121-129.
- 436 Sinha, R. P. & Häder, D. P. 2002 UV-induced DNA damage and repair: a review. *Photochem*.
 437 *Photobiol. Sci.* 1, 225-236.
- 438 Sirés, I. & Brillas, E. 2011 Remediation of water pollution caused by pharmaceutical residues based
- 439 on electrochemical separation and degradation technologies: A review. *Environ. Int.* doi:
- 440 10.1016/j.envint.2011.07.012
- 441 Standard Methods for the Examination of Water and Wastewater 1999 20th edition. American
 442 Public Health Association, Washington DC.
- Süb, J., Volz, S., Obst, U. & Schwartz, T. 2009 Application of a molecular biology concept for the
 detection of DNA damage and repair during UV disinfection. *Water. Res.* 43, 3705-3716.
- 445 Venieri, D., Chatzisymeon, E., Gonzalo, M. S., Rosal, R. & Mantzavinos, D. 2011 Inactivation of
- 446 *Enterococcus faecalis* by UV and solar irradiation in water and wastewater: Culture

techniques never say the whole truth. *Photochem. Photobiol. Sci.* doi: 10.1039/c1pp05198a

- 448 Wéry, N., Lhoutellier, C., Ducray, F., Delgenès, J. P. & Godon, J. J. 2008 Behaviour of pathogenic
- and indicator bacteria during urban wastewater treatment and sludge composting, as revealed
 by quantitative PCR. *Water. Res.* 42, 53-62.
- Willett, K. L. & Hites, R. A. 2000 Chemical actinometry: using o-nitrobenzaldehyde to measure
 light intensity in photochemical experiments. *J. Chem. Educ.* 77, 900-902.
- 453
- 454

455	Figure	captions
455	riguic	captions

450	4	J	6
-----	---	---	---

457 Figure 1

458 *E. coli* inactivation in water samples during PEC, PC and EO. Conditions: 10^7 CFU/mL initial

459 bacterial concentration; 2V potential.

460 **Figure 2**

- 461 *E. coli* inactivation in water samples containing various initial bacterial concentrations during PEC
- 462 (2V potential), assessed by (a) the culture technique and (b) qPCR.

463 Figure 3

- 464 E. coli and E. faecalis inactivation in water samples during PEC (2V potential), assessed by (a) the
- 465 culture technique and (b) qPCR.
- 466 Figure 4
- 467 *E. coli* inactivation in real wastewater during PEC (5V potential), assessed by (a) the culture
- technique and (b) qPCR.







