

*25th International Symposium on Analytical and Environmental Problems***DIRECT PHOTOLYSIS OF FUMONISIN B₁ IN AQUEOUS MEDIUM****Ivana Jevtić¹, Sandra Jakšić², Maria Uzelac³, Biljana Abramović³✉**¹*Higher Medical and Business-Technological School of Applied Studies, Šabac, Hajduk Veljkova 10, Šabac, Serbia*²*Scientific Veterinary Institute „Novi Sad“, Rumenački put 20, Novi Sad, Serbia*³*University of Novi Sad Faculty of Sciences, Department of Chemistry, Biochemistry and Environmental Protection, Trg Dositeja Obradovića 3, 21000 Novi Sad, Serbia*✉ *biljana.abramovic@dh.uns.ac.rs***Abstract**

Fumonisin (FBs), secondary metabolites produced by fungi of the genus *Fusarium* [1], represent a major threat as potential organic pollutants. The FBs have been detected in different types of water [2], as well as produced them by fungi in untreated surface water [3]. Advanced Oxidation Processes (AOPs), have a high potential for water purification, including the removal of hazardous substances and pathogens from different types of water. AOPs are based on physicochemical processes that produce mainly hydroxyl radicals ($\bullet\text{OH}$), representing primary oxidants, which can lead to complete mineralization of pollutants. These processes can be initiated by UV or solar radiation. The photocatalytic degradation has become a powerful method for degradation and transformation of aflatoxin B₁ [4], zearalenone [5], and deoxynivalenol [6] into harmless substances. In this paper, we have investigated optimization of high performance liquid chromatography with fluorescence detector method for monitoring the stability of fumonisins B₁ (FB₁), B₂ (FB₂), and B₃ (FB₃) solutions as well as the efficiency of FB₁ degradation using direct photolysis under UV and solar radiation in ultrapure water. It was found that the sensitivity and separation of the FB₁ peak from *o*-phthalialdehyde–2-mercaptoethanol (used for derivatization) was optimally at isocratic elution using the MeOH–NaH₂PO₄ mobile phase, at a ratio of 75 : 25 (v/v). When studying the efficiency of direct photolysis of 1.47×10^{-6} mol/dm³ solution of FB₁ it was found that after 180 min of irradiation degradation efficiency was 88% using UV and 76% using solar radiation at pH 8.2. Also, the effect of pH in the range from 4.0 to 10.0 on the efficiency of direct photolysis of FB₁ was examined.

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