

1 **Title**

2 Isolation and characterisation of a new [FeFe]-hydrogenase from *Clostridium perfringens*

3

4 **Authors**

5 Simone Morra¹, Beatrice Mongili¹, Sara Maurelli², Gianfranco Gilardi¹ and Francesca Valetti*¹

6

7 **Affiliations**

8 ¹ Department of Life Sciences and Systems Biology, University of Torino, Torino, Italy.

9 ² Department of Chemistry, University of Torino, Torino, Italy.

10

11 **Running title (max 50 characters)**

12 The [FeFe]-hydrogenase of *Clostridium perfringens*.

13

14 ***Corresponding Author:**

15 Dr. Francesca Valetti. Department of Life Sciences and Systems Biology, University of Torino, Via

16 Accademia Albertina 13, 10123, Torino, Italy. Tel: +390116704646. Fax: +390116704508. E-mail:

17 francesca.valetti@unito.it.

18

19

1 **Abstract (50-250 words)**

2 This paper reports the first characterisation of an [FeFe]-hydrogenase from a *Clostridium*
3 *perfringens* strain previously isolated in our laboratory from a pilot-scale bio-hydrogen plant, that
4 efficiently produces H₂ from waste biomasses. On the basis of sequence analysis, the enzyme is a
5 monomer formed by four domains hosting various iron-sulphur centres involved in electron transfer
6 and the catalytic centre H-cluster. After recombinant expression in *E. coli*, the purified protein
7 catalyses H₂ evolution at high rate of 1645±16 s⁻¹. The optimal conditions for catalysis are in the pH
8 range 6.5-8.0 and at the temperature of 50°C. EPR spectroscopy showed that the H-cluster of the
9 oxidised enzyme displays a spectrum coherent with the H_{ox} state, while the CO inhibited enzyme
10 has a spectrum coherent with the H_{ox}-CO state. FTIR spectroscopy showed that the purified enzyme
11 is composed of a mixture of redox states, with a prevalence of the H_{ox}; upon reduction with H₂,
12 vibrational modes assigned to the H_{red} state were more abundant, while binding of exogenous CO
13 resulted in a spectrum assigned to the H_{ox}-CO state. The spectroscopic features observed are similar
14 to those of the [FeFe]-hydrogenases class, but relevant differences were observed given the
15 different protein environment hosting the H-cluster.

16
17 **Keywords (max 6, alphabetical order, at least 3 not used in the title)**

18 Bio-hydrogen; *Clostridium perfringens*; [FeFe]-hydrogenases; H-cluster; iron-sulphur centres;
19 recombinant expression.

20

21

1 Introduction

2 Hydrogen is a promising energy carrier that may replace or complement fossil fuels. For this
3 purpose, the production of this gas by renewable technologies is necessary, and, in this perspective,
4 the exploitation of microorganisms or enzymes for high rate hydrogen synthesis from low cost
5 substrates has been strongly supported [1-4].

6 In Clostridia, the very last step of H₂ production is catalysed by [FeFe]-hydrogenases, a class of
7 redox enzymes found in several microorganisms, that display exceptionally high turnover rates
8 [5,6].

9 [FeFe]-hydrogenases are characterised by the peculiar catalytic centre H-cluster, an organometallic
10 cofactor composed by a cubane [4Fe4S] sub-cluster and a [2Fe] sub-cluster; the [4Fe4S] centre is
11 coordinated by four conserved cysteines, while the [2Fe] centre is coordinated by a single protein
12 cysteine and by other non protein ligands [7,8].

13 These enzymes have been widely investigated and their application as natural or engineered
14 catalysts has been proposed [9-13]. Surprisingly, despite the large interest and vast availability of
15 different natural enzymes, only few of them have been studied so far [5,14,15].

16 *Clostridium perfringens* is a Gram-positive, anaerobic, spore-forming bacterium that is widely
17 distributed in the environment and is also part of the normal flora of the human and animal
18 intestine. Despite this, it is a relevant pathogen as it can cause infections and food poisoning
19 [16,17].

20 Similarly to other Clostridia, its metabolism is based on anaerobic fermentation: after glycolysis,
21 pyruvate is oxidised via pyruvate-ferredoxin oxidoreductase (PFOR) resulting in the production of
22 lactate, alcohol, acetate and butyrate together with gaseous CO₂ and H₂ [16]. *Clostridium*
23 *perfringens* is often found in anaerobic digestion plants, where it is one of the responsible of H₂
24 generation [18-20].

25 In this work, the [FeFe]-hydrogenase CpHydA was studied for the first time. The gene CPF_2655,
26 encoding for CpHydA, was cloned from the strain *Clostridium perfringens* SM09, recently isolated
27 in our laboratory from a pilot-scale bio-hydrogen plant in Torino (Italy) which was particularly
28 efficient in H₂ production from waste biomasses [20,21].

29 Previous works showed that the gene CPF_2655 is expressed during H₂ production [20] and that it
30 is directly involved into the metabolic pathway [22].

31 Here, the new enzyme was recombinantly expressed in *E. coli* in the active form, purified and
32 characterised both in its functional and spectroscopic features.

1 **Materials and methods**

2 **Gene cloning**

3 A DNA fragment of 2126 bp containing the entire coding sequence of CPF_2655 was amplified by
4 PCR from the genome of *Clostridium perfringens* SM09 [20] using the following primers:
5 ATGGCGTTGAAGAAGCAAAG and AACCGTTTTTCATCCATGAGC. For this purpose, the
6 proof-reading KOD Hot Start DNA polymerase (Merck) was used following the manufacturer's
7 instructions.

8 Subsequently, the 5' of the gene was modified by PCR to insert an NdeI site and the 3' was
9 modified to add a sequence coding for Strep-tagII and a XhoI site using the following primers:
10 AACCATATGAATAAAATAATAATCAATGATAAGACTATCG and
11 ATCTCGAGTTATTTTTCAAATTGAGGATGACTCCAATTTTTTTTATATTTTCATGTGTAAT
12 AACTCATGAG.

13 The PCR fragment (1760 bp) was digested with NdeI and XhoI (Thermo Scientific) and ligated into
14 the empty expression vector pECr1 [23] to give the new vector pECPF2655. The T4 DNA Ligase
15 (Thermo Scientific) was used. The gene cloned into the final expression vector has been fully
16 sequenced.

17 The sequence has been deposited into the GenBank database and assigned the accessions KP115260
18 (DNA) and AJQ21778 (protein).

19 A homology model of the 3D structure was built by the Swiss-Model server
20 (<http://swissmodel.expasy.org/>) using Cpl 3C8Y x-ray structure as template.

21 The sequence alignment was performed using MultAlin (<http://multalin.toulouse.inra.fr/multalin/>).

22 The accessions of the other [FeFe]-hydrogenase sequences are: CaHydA (NP_346675), Cpl
23 (AAA23248), DdH large subunit (1HFE) and CrHydA1 mature form (AAL23572).

24

25 **Recombinant expression and purification**

26 Recombinant expression in *E. coli* Rosetta2(DE3) was performed as previously described [24].

27 Briefly, the vector pECPF2655 (harbouring CpHydA gene and CaHydE maturation gene) was co-
28 transformed with pCaFG (harbouring CaHydF and CaHydG maturation genes) [23]. Cells were
29 aerobically grown in Terrific Broth medium supplemented with 2 mM ammonium ferric citrate as a
30 source of iron, 200 µg/mL carbenicillin, 50 µg/mL streptomycin and 34 µg/mL chloramphenicol.

31 When the OD₆₀₀ reached 0.4, the culture was supplemented with 2 mM cysteine, 25 mM sodium
32 fumarate and 0.5% w/v glucose and the expression was induced by the addition of 1.5 mM IPTG.

33 After the induction, cells were incubated over night under pure argon flow to maintain anaerobic
34 conditions in a water bath at 30°C. To prevent oxygen inactivation of the active CpHydA all the

1 following manipulations were performed into a glove box (Plas Labs) under an anaerobic
2 hydrogen/nitrogen atmosphere; before use, all solutions were vacuumed, equilibrated with the
3 anaerobic atmosphere and supplemented with 2-20 mM sodium dithionite.
4 Purification of CpHydA was carried out under strict anaerobic conditions by affinity
5 chromatography using Strep-Tactin Superflow high capacity cartridges (IBA) and following
6 manufacturer's instructions. The enzyme was eluted and stored in 100 mM Tris·HCl, 150 mM NaCl
7 pH 8.0 supplemented with 2 mM sodium dithionite and 2.5 mM desthiobiotin.
8 Coomassie-stained SDS-PAGE was used to determine the purity and the molecular weight of the
9 purified enzyme. Protein concentration was assayed with Bradford assay using bovine serum
10 albumin as standard.

11

12 **Hydrogenase activity assay**

13 Hydrogenase activity (hydrogen evolution) was tested as previously described [25]. Briefly, 10 mM
14 dithionite-reduced methyl viologen was used as artificial electron donor at 37°C in 100 mM
15 Tris·HCl, 150 mM NaCl pH 8.0, unless otherwise indicated. The evolution of H₂ was quantified by
16 gas chromatography using an Agilent Technologies 7890A instrument equipped with purged
17 packed inlet, Molesieve 5A column (30 m, ID 0.53 mm, film 25 mm) and thermal conductivity
18 detector; argon was used as carrier gas.

19 For the study of the pH dependance, the following buffering agents were used: 50 mM MES·NaOH
20 (pH range 5.2–6.5) or 50 mM ACES·NaOH (pH range 6.5–7.5) or 50 mM Tris·HCl (pH range 7.5–
21 9.0); the total ionic strength was fixed to 50 mM with NaCl. For the study of the temperature
22 dependance, a refrigerated water bath was used to set up the temperature in the range between 0 and
23 70°C. **The relative activity was calculated as the ratio between the activity of each experimental
24 point and the maximum activity.** The activation energy was calculated by an Arrhenius plot.

25 **Oxygen sensitivity was determined as previously described [26].**

26

27 **EPR spectroscopy**

28 Purified CpHydA was anaerobically concentrated by ultrafiltration using Amicon Ultra 0.5 mL 30K
29 MWCO (Millipore); the buffer was exchanged to remove traces of dithionite and 5% v/v glycerol
30 was added. The enzyme was treated with thionine in a molar excess of 7.8 to obtain the oxidised
31 sample; the final CpHydA concentration was 0.2 mM. The CO-treated sample was obtained by
32 flushing the oxidised sample with CO twice for 30 seconds on ice.

33 X-band continuous wave (CW) EPR spectra were recorded on a Bruker EMX spectrometer
34 equipped with a cylindrical cavity and operating at a 100 kHz field modulation. The experimental

1 parameters were as follows: microwave power 10 mW, modulation amplitude 0.2 mT, temperature
2 77K. Typical measurements were carried out on a 60 μ L frozen solution in a quartz tube with
3 internal diameter 2 mm. EPR spectra were simulated using the Easyspin tool package [27].
4

5 **FTIR spectroscopy**

6 For the characterisation of the H-cluster, purified CpHydA was anaerobically concentrated up to 1
7 mM. The untreated sample was measured immediately after concentration; **the fully oxidised**
8 **sample was obtained by addition of thionine (8 mM)**; the H₂-treated sample was obtained by
9 flushing H₂ three times for 1 minute on ice; the CO-treated sample was obtained by flushing CO
10 twice for 30 seconds on ice.

11 The spectra were acquired at room temperature using a Bruker Tensor 27 FT-IR spectrometer
12 (Bruker Instruments). The sample chamber was purged with 99.9999% pure nitrogen gas.

13 For the characterisation of the H-cluster, a transmission cell (CaF₂ window; 50 μ m pathlength) was
14 used. Spectra were acquired with a resolution of 2 cm^{-1} accumulating 256 scans. The baseline
15 correction was obtained using the Opus 6.0 software (Bruker Instruments) by the concave
16 rubberband algorithm.

17

1 **Results and discussion**

2 **Cloning, recombinant expression and purification**

3 In order to produce large amount of active CpHydA, the gene CPF_2655 was cloned into the vector
4 pECr1 [23], replacing the gene HydA1, and resulting into the new vector pECPF2655 (Fig. 1A).

5 When co-transformed with the vector pCaFG [23], this vector is suitable for the recombinant
6 expression in *E. coli* [23]. The two vectors allow the co-expression of CpHydA with the maturases
7 HydE, HydF and HydG from *Clostridium acetobutylicum*, that allow the assembly of the H-cluster
8 within the [FeFe]-hydrogenase, under strict anaerobic conditions, to avoid the inactivation of the
9 enzyme by atmospheric oxygen [28].

10 CpHydA was purified by affinity chromatography with a typical yield of 1.5 mg per litre of culture.
11 The purified protein (Fig. 1B) has the expected molecular weight of ~64.7 kDa.

12

13 **Sequence analysis**

14 CpHydA is a monomeric [FeFe]-hydrogenase phylogenetically classified in the cluster A2 with a
15 modular structure M3 (Fig. 1C), according to the most recent literature [6].

16 The enzyme is composed by four domains that contain iron sulphur clusters (Fig. 1D): a [2Fe2S]
17 plant ferredoxin-like (yellow), a [4Fe4S] coordinated by three cysteines and one histidine (red), a
18 2[4Fe4S] bacterial ferredoxin-like (green) and the H-domain containing the catalytic centre H-
19 cluster (blue). The various iron sulphur centres are involved in the electron transfer from the redox
20 partner to the H-cluster, where the reduction of protons into H₂ is catalysed.

21 Alignment with other known [FeFe]-hydrogenase sequences (Fig. 2) allows the identification of the
22 key residues in CpHydA: fifteen cysteines and the histidine responsible for the binding of the four
23 accessory iron sulphur clusters in the N-terminal and C300, C355, C497 and C501, responsible for
24 the binding of the H-cluster in the C-terminal. The latter residues are embedded in the signature
25 sequences L1, L2 and L3 [6,29], which are strognly conserved in all known [FeFe]-hydrogenases.
26 Other relevant residues that can be observed are: M353, K358 and M495, that are essential for non-
27 covalent interactions with the H-cluster [30] and E279, E282, C299 and S319 that are essential for
28 the proton transfer [25,31].

29 Also, the protein sequence of CpHydA is 69% identical to CpI and 31% identical to DdH.

30

31 **Functional characterisation of CpHydA**

32 The purified CpHydA shows the typical catalytic behaviour for [FeFe]-hydrogenases and it is able
33 to evolve H₂ from reduced methyl viologen at a rate of 1645±16 s⁻¹. The hydrogen evolution rate of

1 CpHydA is within the range for [FeFe]-hydrogenases [8], and it is particularly high in comparison
2 to other recombinant enzymes such as CrHydA1, CpI and CaHydA [24,32,33].
3 Purified CpHydA was inactivated by oxygen with a 50% loss of activity after 2.8 minutes exposure
4 to air. This value is within the range of 2-5 minutes determined for other [FeFe]-hydrogenases [26].
5 The pH dependance of the H₂ evolution rate (Fig. 3A) showed a broad peak with maximum activity
6 between 6.5 and 8.0. In comparison to other [FeFe]-hydrogenases, this feature is very similar to that
7 observed in CaHydA where the maximum is at pH 8.0 [25]. It is also similar to that observed in CpI
8 where the maximum is at pH 6.3 and the activity decreases faster at higher pH, but it is much
9 different from CpII where the maxima are at pH 5.8 and 9.1 [26]. The pH dependance is also
10 different from CrHydA1, where the maximum is at pH 6.9 and the bell shape of the curve is much
11 narrower [28].
12 The H₂ evolution activity (Fig. 3B) increases exponentially with temperature, reaches a maximum
13 at 50°C and is mainly lost at 70°C. The temperature maximum is similar to other [FeFe]-
14 hydrogenases, such as CpI (~50°C) [26] and CrHydA1 (60°C) [28], and it is coherent with the
15 growth temperature of *Clostridium perfringens*, which spans from 15 to 50°C, with optimum at
16 45°C [34].
17 The calculated activation energy for H₂ evolution by CpHydA is 50.8±2.1 kJ/mol, which is similar
18 to those calculated for CrHydA1 (55.1 kJ/mol) [35] and CpI (61 kJ/mol) [36].
19 The catalytic activity of CpHydA at relatively high temperatures suggests a good thermal stability
20 of the enzyme that makes it suitable for biotechnological applications.

21

22 **Spectroscopic characterisation of CpHydA**

23 EPR and FTIR spectroscopies were used to characterise the structure, composition, geometry of the
24 catalytic centre H-cluster in different redox states. In particular, EPR was used to investigate the
25 electronic structure and the chemical environment, while FTIR gave information about the
26 vibrational modes of the H-cluster ligands CO and CN.

27 The EPR spectra (Fig. 4) are consistent with spectra reported for other well studied [FeFe]-
28 hydrogenases. The oxidised sample (Fig. 4A, Table 1) is characterized by a rhombic EPR spectrum
29 with **g** tensor components $g_1=2.0892$, $g_2=2.0363$, $g_3=1.9954$ typical for the H_{ox} state of the H-
30 cluster [15,37-45]. The simulation analysis also indicates a minor contribution of an axial EPR
31 pattern with **g** values $g_1 = 2.0755$ and $g_2 = g_3 = 2.0080$. This spectroscopic feature has been
32 assigned to a CO inhibited state of the H-cluster due to the release of CO molecules from damaged
33 H-cluster [38,46,47]. Indeed upon treatment with CO the EPR spectrum of the H_{ox} state is
34 completely converted into the typical axial spectrum of the H_{ox}-CO state (Fig. 4B).

1 Transmission FTIR spectra (Fig. 5) show absorption peaks that are similar to signals previously
2 observed in other [FeFe]-hydrogenases, both in terms of the wavenumber and the relative intensity.
3 On this basis, the peaks observed here were assigned to the H_{ox} , H_{ox} -CO and H_{red} state of [FeFe]-
4 hydrogenases (Table 2). When the enzyme is only concentrated without further treatment
5 (untreated), it shows a number of peaks that are due to a mixture of different redox states, with a
6 prevalence of the H_{ox} state. The fully oxidised enzyme was obtained by thionine treatment and
7 shows only peaks of the H_{ox} state. Treatment with H_2 results in the enrichment of peaks that can be
8 assigned to the H_{red} state, while treatment with CO results in a very homogeneous spectrum with
9 peak shifts that are in good agreement with previous studies on other [FeFe]-hydrogenases in the
10 H_{ox} -CO state.

11 Despite the expected similarities between CpHydA and the other [FeFe]-hydrogenases, some
12 relevant differences could be observed. For example, reduction with H_2 do not cause the appearance
13 of signals assignable to the H_{sred} state, which was recently characterised in CrHydA1 [44,45,49];
14 this might be due to the effect of the accessory iron sulphur centres (the so called F-clusters) that are
15 present in CpHydA and absent in CrHydA1. Consistently, this intermediate has never been
16 described before in Cpl. Moreover, the CN signals in the CO-treated sample showed an unusual
17 behaviour: they are very close and appear as a single large peak that was fitted with two peaks with
18 maximum at 2091.1 and 2088.3 cm^{-1} .

19 In conclusion, in this work the [FeFe]-hydrogenase CpHydA from *Clostridium perfringens* SM09
20 was characterised for the first time. The enzyme has sequence similarity with other enzymes of the
21 class. The recombinantly expressed purified enzyme is properly folded and can catalyse H_2
22 evolution at high rates ($1645 \pm 16 s^{-1}$). The catalytic centre H-cluster is correctly inserted and
23 coordinated and it can be readily oxidised with thionine to obtain the H_{ox} state, which can be
24 converted into the H_{ox} -CO state by binding the inhibitor CO, as determined by EPR spectroscopy.
25 Also FTIR confirmed that the H-cluster was properly bound and the typical H_{ox} , H_{ox} -CO and H_{red}
26 signals could be observed.

27 The high turnover rates for hydrogen evolution and the thermal stability and optimum activity at
28 50°C of this newly characterised [FeFe]-hydrogenase make it an excellent catalyst for
29 biotechnological devices and sustainable processes for clean energy production.
30

31

1 **Acknowledgements**

2 The authors are grateful to Dr. P.W. King for the generous gift of the plasmids for expression
3 system in *E. coli* and to Prof. E. Giamello and Prof. M. Chiesa for EPR instrument support and
4 critical manuscript review.

5 This work was supported by “RICERCA LOCALE” 2012 and 2013 from the University of Torino
6 and, partially, by projects BIOH2 and HyStrEM (E.U. Structural Funds N.1083/2006 F.E.S.R.
7 2007-2013). S.M. is the recipient of an “Assegno di ricerca cofinanziato” from the University of
8 Torino.

9

10

1 **References**

- 2 [1] Levin, D.B., Pitt, L., Love, M. (2004) *Int. J. Hydrogen Energy* **29**, 173-185.
- 3 [2] Hallenbeck, P.C., Ghosh, D. (2012) *J. Environ. Manage.* **95**, S360-S364.
- 4 [3] Krupp, M., Widmann, R. (2009) *Int. J. Hydrogen Energy* **34**, 4509-4516.
- 5 [4] Yasin, N.H.M., Mumtaz, T., Hassan, M.A., Rahman, N.A.A. (2013) *J. Environ. Manage.*
6 **130**, 375-385.
- 7 [5] Vignais, P.M., Billoud, B. (2007) *Chem. Rev.* **107**, 4206-4272.
- 8 [6] Calusinska, M., Happe, T., Joris, B., Wilmotte, A. (2010) *Microbiology* **156**, 1575-1588.
- 9 [7] Fontecilla-Camps, J.C., Volbeda, A., Cavazza, C., Nicolet, Y. (2007) *Chem. Rev.* **107**, 4273-
10 4303.
- 11 [8] Winkler, M., Esselborn, J., Happe, T. (2013) *Biochim. Biophys. Acta* **1827**, 974-985.
- 12 [9] Cracknell, J.A., Vincent, K.A., Armstrong, F.A. (2008) *Chem. Rev.* **108**, 2439-2461.
- 13 [10] Morra, S., Valetti, F., Sadeghi, S.J., King, P.W., Meyer, T., Gilardi, G. (2011) *Chem.*
14 *Commun.* **47**, 10566-10568.
- 15 [11] Woolerton, T.W., Sheard, S., Chaudhary, Y.S., Armstrong, F.A. (2012) *Energy Environ.*
16 *Sci.* **5**, 7470-7490.
- 17 [12] King, P.W. (2013) *Biochim Biophys Acta* **1827**:949-957.
- 18 [13] Happe, T., Hemschemeier, A. (2014) *Trends Biotechnol.* **32**, 170-176.
- 19 [14] De Lacey, A.L., Fernández, V.M. (2007) *Chem. Rev.* **107**, 4304-4330.
- 20 [15] Lubitz, W., Ogata, H., Rüdiger, O., Reijerse, E. (2014) *Chem. Rev.* **114**, 4081-4148.
- 21 [16] Shimizu, T., Ohtani, K., Hirakawa, H., Ohshima, K., Yamashita, A., Shiba, T., Ogasawara,
22 N., Hattori, M., Kuhara, S., Hayashi, H. (2002) *Proc. Natl. Acad. Sci.* **99**, 996-1001.
- 23 [17] Li, J., Adams, V., Bannam, T.L., Miyamoto, K., Garcia, J.P., Uzal, F.A., Rood, J.I.,
24 McClane, B.A. (2013) *Microbiol. Mol. Biol. Rev.* **77**, 208-233.
- 25 [18] Skillmann, L.C., Bajsa, O., Santhanam, B., Kumar, M., Ho, G. (2009) *Water Res.* **43**, 3281-
26 3291.
- 27 [19] Wirth, R., Kovács, E., Maróti, G., Bagi, Z., Rákhely, G., Kovács, K.L. (2012) *Biotechnol.*
28 *Biofuels* **5**, 41.
- 29 [20] Morra, S., Arizzi, M., Allegra, P., La Licata, B., Sagnelli, F., Zitella, P., Gilardi, G., Valetti,
30 F. (2014) *Int. J. Hydrogen Energy* **39**, 9018-9027.
- 31 [21] La Licata, B., Sagnelli, F., Boulanger, A., Lanzini, A., Leone, P., Zitella, P., Santarelli, M.
32 (2011) *Int. J. Hydrogen Energy* **36**, 7861-7865.
- 33 [22] Kaji, M., Taniguchi, Y., Matsushita, O., Katayama, S., Miyata, S., Morita, S., Okabe, A.
34 (1999) *FEMS Microbiol. Lett.* **181**, 329-336.

- 1 [23] King, P.W., Posewitz, M.C., Ghirardi, M.L., Seibert, M. (2006) *J. Bacteriol.* **188**, 2163-
2 2172.
- 3 [24] Yacoby, I., Tegler, L.T., Pochekailov, S., Zhang, S., King, P.W. (2012) *PLoS ONE* **7**,
4 e35886.
- 5 [25] Morra, S., Girando, A., Di Nardo, G., King, P.W., Gilardi, G., Valetti, F. (2012) *PLoS ONE*
6 **7**, e48400.
- 7 [26] Adams, M.W.W., Mortenson L.E. (1984) *J. Biol. Chem.* **259**, 7045-7055.
- 8 [27] Stoll, S., Schweiger, A. (2006) *J. Magn. Reson.* **178**, 42-55.
- 9 [28] Happe, T., Naber, D. (1993) *Eur. J. Biochem.* **214**, 475-481.
- 10 [29] Meyer, J. (2007) *Cell. Mol. Life Sci.* **64**, 1063-1084.
- 11 [30] Knörzer, P., Silakov, A., Foster, C.E., Armstrong, F.A., Lubitz, W., Happe, T. (2012) *J.*
12 *Biol. Chem.* **286**, 38341-38347.
- 13 [31] Cornish, A.J., Gärtner, K., Yang, H., Peters, J.W., Hegg, E.L. (2011) *J. Biol. Chem.* **286**,
14 38341-38347.
- 15 [32] Kuchenreuther, J.M., Grady-Smith, C.S., Bingham, A.S., George, S.J., Cramer, S.P., Swartz,
16 J.R. (2010) *PLoS ONE* **5**, e15491.
- 17 [33] Kim, S., Lu, D., Park, S., Wang, G. (2012) *Int. J. Hydrogen Energy* **37**, 15833-15840.
- 18 [34] Brynestad, S., Granum, P.E. (2002) *Int. J. Food Microbiol.* **74**, 195-202.
- 19 [35] Roessler, P., Lien, S. (1984) *Plant Physiol.* **75**, 705-709.
- 20 [36] Kleiner, D., Burris, R.H. (1970) *Biochim. Biophys. Acta* **212**, 417-427.
- 21 [37] Adams, M.W.W. (1987) *J. Biol. Chem.* **262**, 15054-15061.
- 22 [38] Bennett, B., Lemon, B.J., Peters, J.W. (2000) *Biochemistry* **39**, 7455-7460.
- 23 [39] Albracht, S.P.J., Roseboom, W., Hatchikian, E.C. (2006) *J. Biol. Inorg. Chem.* **11**, 88-101.
- 24 [40] Silakov, A., Reijerse, E.J., Albracht, S.P.J., Hatchikian, E.C., Lubitz, W. (2007) *J. Am.*
25 *Chem. Soc.* **129**, 11447-11458.
- 26 [41] von Abendroth, G., Stripp, S., Silakov, A., Croux, C., Soucaille, P., Girbal, L., Happe, T.
27 (2008) *Int. J. Hydrogen Energy* **33**, 6076–6081.
- 28 [42] Kamp, C., Silakov, A., Winkler, M., Reijerse, E.J., Lubitz, W., Happe, T. (2008) *Biochim.*
29 *Biophys. Acta* **1777**, 410-416.
- 30 [43] Silakov, A., Wenk, B., Reijerse, E., Lubitz, W. (2009) *Phys. Chem. Chem. Phys.* **11**, 6592-
31 6599.
- 32 [44] Adamska, A., Silakov, A., Lambertz, C., Rüdiger, O., Happe, T., Reijerse, E., Lubitz, W.
33 (2012) *Angew. Chem. Int. Ed.* **51**, 11458-11462.

- 1 [45] Mulder, D.W., Ratzloff, M.W., Shepard, E.M., Byer, A.S., Noone, S.M., Peters, J.W.,
2 Broderick, J.B., King, P.W. (2013) *J. Am. Chem. Soc.* **135**, 6921-6929.
- 3 [46] Roseboom, W., De Lacey, A.L., Fernandez, V.M., Hatchikian, E.C., Albracht, S.P.J. (2006)
4 *J. Biol. Inorg. Chem.* **11**, 102-118.
- 5 [47] Silakov, A., Wenk, B., Reijerse, E., Albracht, S.P.J., Lubitz, W. (2009) *J. Biol. Inorg. Chem.*
6 **14**, 301-313.
- 7 [48] Chen, Z., Lemon, B.J., Huang, S., Swartz, D.J., Peters, J.W., Bagley, K.A. (2002)
8 *Biochemistry* **41**, 2036-2043.
- 9 [49] Silakov, A., Kamp, C., Reijerse, E., Happe, T., Lubitz, W. (2009) *Biochemistry* **48**, 7780-
10 7786.

11

1 Tables

2

3 Table 1. Spin-Hamiltonian parameters of the H_{ox} and H_{ox-CO} states of H-cluster of CpHydA
 4 extracted by the computer simulations of the CW EPR spectra reported in Figure 4 in comparison to
 5 selected data on other [FeFe]-hydrogenases. *nd* = not determined.

6

Enzyme	H_{ox}			H_{ox-CO}			Ref.
	g_1	g_2	g_3	g_1	g_2	g_3	
CpHydA	2.0892 ± 0.0005	2.0363 ± 0.0005	1.9954 ± 0.0005	2.0755 ± 0.0005	2.008 ± 0.005	2.008 ± 0.005	This work
CrHydA1	2.102	2.040	1.998	2.052	2.007	2.007	[42]
CaHydA	<i>nd</i>	<i>nd</i>	<i>nd</i>	2.075	2.009	2.009	[41]
CpI	2.097	2.039	1.999	2.072	2.006	2.006	[37]
DdH	2.10	2.04	2.00	2.06	2.00	2.00	[39]
	2.100	2.040	1.997	2.065	2.007	2.001	[43,47]

7

8

9

10

11

12 Table 2. FTIR signals: summary of the peaks (wavenumber in cm^{-1}) observed in CpHydA in
 13 comparison to selected works on other [FeFe]-hydrogenases in the H_{ox} , H_{ox-CO} and H_{red} states. *nd* =
 14 not determined.

15

Enzyme	H_{ox}			H_{ox-CO}				H_{red}		Ref.
	CNs	COs	$\mu-CO$	CNs	CO_{exo}	COs	$\mu-CO$	CNs	COs	
CpHydA	2087, 2080	1968, 1944	1800	2091, 2088	2013	1971, 1967	1806	2066, 2039	1897	This work
CpI	2086, 2072	1971, 1948	1802	2095, 2077	2017	1974, 1971	1810	<i>nd</i>	<i>nd</i>	[48]
CrHydA1	2088, 2072	1964, 1940	1800	2092, 2084	2013	1970, 1964	1810	2083, 2070	1935, 1891, 1793	[49]
DdH	2093, 2079	1965, 1940	1802	2096, 2088	2016	1971, 1963	1810	2079, 2041	1965, 1916, 1894	[46]

16

17

1 **Figure legends.**

2

3 **Figure 1.** CpHydA cloning, purification and model structure. A) Map of the expression vector
4 pECPF2655. B) Coomassie-stained SDS-PAGE of the purified CpHydA; molecular weights are in
5 kDa. C) Scheme of the modular domains, where yellow cross = [2Fe2S] plant ferredoxin-like; red
6 diamond = [4Fe4S] coordinated by three cysteines and one histidine; green oval = 2[4Fe4S]
7 bacterial ferredoxin-like; blue rectangle = H-domain. D) Homology model of CpHydA structure.

8

9 **Figure 2.** Sequence alignment of CpHydA with other selected [FeFe]-hydrogenases. CaHydA =
10 *Clostridium acetobutylicum* hydrogenase A; CpI = *Clostridium pasteurianum* hydrogenase I; DdH
11 = *Desulfovibrio desulfuricans* hydrogenase; CrHydA1 = *Chlamydomonas reinhardtii* hydrogenase
12 A1. Grey shaded residues are fully conserved; black shaded residues are cysteines or histidines that
13 coordinate the iron sulfur centres. Squared regions are the conserved signature motifs L1, L2 and
14 L3 that include the H-cluster coordinating cysteines.

15

16 **Figure 3.** Hydrogen evolution characterisation. A) pH dependance; the maximum activity was at
17 pH 8 and was $1645 \pm 16 \text{ s}^{-1}$. B) Temperature dependance; the maximum activity was at 50°C and
18 was $1941 \pm 22 \text{ s}^{-1}$. The activity was assayed by gas chromatography using reduced methyl viologen
19 as artificial electron donor.

20

21 **Figure 4.** Experimental (solid lines) and computer simulations (dotted lines) X-band CW EPR
22 spectra of CpHydA frozen solutions in the A) H_{ox} and B) $\text{H}_{\text{ox}}\text{-CO}$ states of the H-cluster. The spin
23 Hamiltonian parameters extracted from the computer simulations are reported in Table 1. Asterisks
24 in spectrum A indicate the features of the $\text{H}_{\text{ox}}\text{-CO}$ state present as an impurity.

25

26 **Figure 5.** A) Transmission FTIR spectra of CpHydA after different treatments: untreated, thionine
27 oxidised, H_2 -reduced and CO-treated. The H_2 -reduced spectrum has been magnified by a factor of
28 2. B) Summary of the FTIR signals assigned to the H_{ox} , $\text{H}_{\text{ox}}\text{-CO}$ and H_{red} states.

29