

1 **Novel Thermal Barrier Coatings with Hexagonal Boron Nitride**
2 **Additives Resistant to Molten Volcanic Ash Wetting**
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26 **Abstract**

27

28 Yttria-stabilized zirconia thermal barrier coatings (YSZ TBCs) are prone to degradation by atmospheric
29 contaminants such as volcanic ash, posing a serious threat to their durability. Here we demonstrate that
30 hexagonal boron nitride (*h*-BN) additives to YSZ coatings (*h*-BN YSZ), exhibit resistance against volcanic
31 ash interaction at 1250 °C. In preliminary tests performed under vacuum, *h*-BN YSZ coatings demonstrate
32 greatly inhibited wetting against molten volcanic ash drops. Under oxidizing conditions, *h*-BN YSZ provide
33 enhanced wetting and chemical resistance over conventional coatings. This study lays groundwork for the
34 development of novel additives which are physico-chemically resilient against molten volcanic ash wetting.

35

36 **Keywords:** Thermal Barrier Coatings, Hexagonal Boron Nitride, Volcanic ash, Non-wetting, Silica-
37 phobicity.

38

39 1. Introduction

40 With the steady growth of global air traffic, the occurrence rate of volcanic ash plume encounters by
41 aircrafts has soared and the hazards generated by the ingestion of molten silicate deposits into the engines
42 have garnered greater attention leading to deeper investigation in the aviation sector [1–7]. In 2010, the
43 eruption of Eyjafjallajökull in Iceland triggered the largest aviation shutdown in Europe since World War
44 II and caused economic losses estimated at ~2 billion euros [8,9]. Given the catalytic role of air travel in
45 shaping socio-economic forces across the globe, diminishing the potential impact of environmental dust in
46 general, molten calcium magnesium alumino-silicate (CMAS) deposits and volcanic ash in particular on
47 aviation safety has become a research and development imperative [10,11].

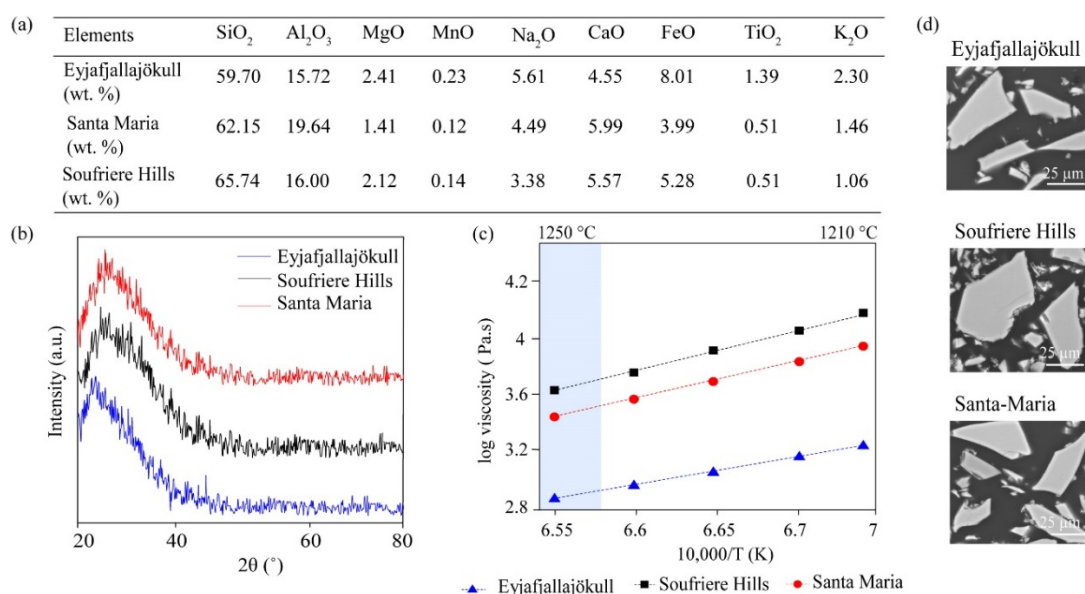
48 When volcanic ash is ingested into a modern jet engine, it may initially erode/abrade compressor blades,
49 leading to a reduction in operational efficiency, the ash subsequently enters the combustion chamber where
50 peak flame temperatures can reach 2000 °C [12]. Such temperatures far exceed the melting temperatures
51 (~900 °C–1300 °C) of the glassy and mineral constituents of volcanic ash [13,14]. As a result, the ingested
52 ash particles liquify (by glass softening and/or mineral melting) and may adhere to, and flow across (i.e.,
53 impaction, spreading and wetting) the surfaces of the engine components (e.g., nozzle guide vanes and high-
54 pressure turbine blades) that are commonly protected by thermal barrier coatings (TBCs) [15–18].

55 Conventional TBCs are porous ceramic coatings (typically made of tetragonal-phase ZrO_2 ceramic
56 stabilized by 6–8 wt.% Y_2O_3 , commonly referred to as YSZ) prepared by atmospheric plasma-spraying
57 (APS) or electron-beam physical vapor deposition (EB-PVD) [19, 20]. Molten volcanic ash can rapidly
58 ingress into the porous coating due to capillary action, thereby cause spallation of the TBCs due to stiffening
59 as a resultant of infiltrated melt solidification during cooling [21]. Simultaneously, a chemical interaction
60 takes place between molten volcanic ash and YSZ coating consisting of the dissolution of initial metastable
61 tetragonal (t') YSZ grains in the molten CMAS followed by precipitation of Y depleted m- ZrO_2 [22]. In the
62 past ten years, it has been suggested that the most promising solution for mitigation of the infiltration of
63 molten volcanic ash into TBCs might be achieved by Gadolinium zirconate TBCs ($Gd_2Zr_2O_7$) [23, 24].
64 $Gd_2Zr_2O_7$ TBCs counteract silicate melt infiltration thermo-chemically by crystallizing the melt and

65 formation of oxy-apatite phase upon reaction [25]. The inclusion However, the deterioration of TBC
 66 primarily emanates due to wetting of molten volcanic ash on the surface [18]. To expand the scope of
 67 potential mitigational measures to counteract molten volcanic ash wetting, we go beyond the state of art of
 68 traditional TBC materials. Here, we introduce the design of a more resilient *h*-BN additives to YSZ with
 69 properties that reduce silicate melt wettability, and thereby inhibit physical infiltration and degradation via
 70 chemical reaction.

71 2. Experimental Procedure

72 2.1 Fresh volcanic ash characterization and homogenization



73 **Fig. 1.** Characterization of volcanic ashes. (a) Bulk chemical composition of volcanic ashes. (b) XRD spectra of the
 74 re-melted volcanic ashes showing only the amorphous silica shoulder, further confirming the amorphous character of
 75 the ash samples. (c) Viscosity of the ash samples over the temperature range 1210 °C–1250 °C. (d) SEM images of re-
 76 melted Eyja, SF and SM ashes.
 77

78 Three fresh volcanic ash samples were utilized for this experiment: Eyjafjallajökull (Eyja), Soufriere Hills
 79 (SH), and Santa Maria (SM). Bulk chemical composition of the volcanic ashes was determined by X-ray
 80 fluorescence spectrometry (XRF) (Fig. 1a). The crystal-bearing natural volcanic ash samples were heated
 81 in a platinum crucible to 1650 °C for 48 hours and were subsequently quenched in air, to produce crystal-
 82 free glasses with a common thermal history. The glasses were then crushed and sieved to produce finely

83 powdered ash. X-ray diffraction (XRD) pattern was measured for the glassy samples of the re-melted ash
84 (GE X-ray diffractometer 3003 TT) (Fig. 1b).

85 The viscosity of the molten volcanic ash as a function of temperature was estimated using multi-component
86 viscosity evaluation model, developed by Giordano *et.al* [26] (Fig. 1c). The re-melted volcanic ash samples
87 were optically investigated using a scanning electron microscope (SEM) (Fig. 1d). SEM and XRD results
88 confirmed that the ash samples were homogeneous. The onset of melting was estimated using the rhyolite-
89 MELTS software (Onset of melting: Eyja: 1151 °C, SM: 1179 °C, SH: 1130 °C) [27].

90 2.2 Fabrication of the TBCs

91 We employed APS method to produce novel coatings consisting of *h*-BN (20 wt.%) and 7YSZ (80 wt.%)
92 (referred to as *h*-BN YSZ) onto an alumina substrate ($\text{Al}_2\text{O}_3 > 96\%$). 7 wt. %YSZ powder with a particle
93 size of less than 63 μm (Fujimi) and boron nitride powder with a particle size of less than 45 μm (Alfa
94 Aesar) were used as feedstock. Alumina substrates were chosen in this work instead of traditional superalloy
95 substrates to facilitate isothermal heat treatment of the system above 1200 °C. Conventional YSZ coatings
96 were fabricated by thermal spraying (Table 1), for comparative tests *versus h*-BN YSZs with same spray
97 conditions (referred to as YSZ). The coatings were ultrasonically cleaned using de-ionized water post
98 fabrication, and stored in a desiccator in order to avoid any surficial contamination.

100 Table 1. Plasma spray parameters.

Arc Voltage (V)	Carrier gas (Ar- L·min ⁻¹)	Arc current (A)	Spray distance (mm)	Powder feed rate (g·min ⁻¹)
70	40	800	100	0.8

101

102 2.3 Wettability measurement and Thermal aging experiments

103 The volcanic ash was compacted into cylindrical pellets with radius and height of 1 mm by pressing the ash
104 into a cylindrical die. The resulting cylindrical ash compact was ejected from the die onto the TBC substrate.
105 Wettability was assessed using sessile drop method by monitoring silhouette transition of the ash compact
106 in a Dataphysics OCA 25-HTV 1800 optical dilatometer, at a temperature of 1250 °C, with a heating rate 5
107 °C min⁻¹ and a hold duration of 1 hour. The experiments were performed in atmospheric conditions and

108 under vacuum conditions, the latter using a turbomolecular pump ($< 10^{-5}$ Pa). The images were captured at
109 60 frames per second (fps), using a charge coupled device. The contact angle was determined to quantify
110 the wettability of molten volcanic ash on TBCs, from the binary converted real time pictures of the molten
111 drop at 1250 °C. A localized methodology of enclosing the contours of the binary drop allows us to evaluate
112 the equilibrium contact angle- θ_c based on enclosed B-spline, using the drop snake plugin in image J software
113 [28]. The extent of spreading of molten volcanic ash on the coatings post thermal aging was evaluated by
114 the spreading area. Thermal aging process was performed using EM301 Hesse instruments furnace. Ash
115 pellets with radius and height of 1 mm were loaded onto the geometrical centre of the coatings, and they
116 were thermally aged at 1250 °C with a heating rate of 10 °C min⁻¹ for a soak-time of 3–48 hrs and cooled at
117 10 °C min⁻¹.

118 **2.4 Micro-hardness measurement**

119 The microhardness of the coatings was measured using Vickers micro-indentation technique
120 (Wilsson VH-1202). The coatings were sectioned across the cross-section and standard
121 metallographic techniques were employed to polish the samples, for Vickers indentation and
122 characterization of the infiltration depth of molten volcanic ash into the coatings. 6 micro-indents
123 were placed across the cross-section of the coatings, at different depths (50, 75, 100 and 150 μ m),
124 with a load of 300 gmf. and a hold-time of 10 seconds. The Vickers micro-hardness (H_v) was
125 evaluated by using the following equation:

$$126 \quad H_v = 0.102 \frac{F}{S}$$

127 where F is the indentation load (N) and S is the indented surface area (mm²).

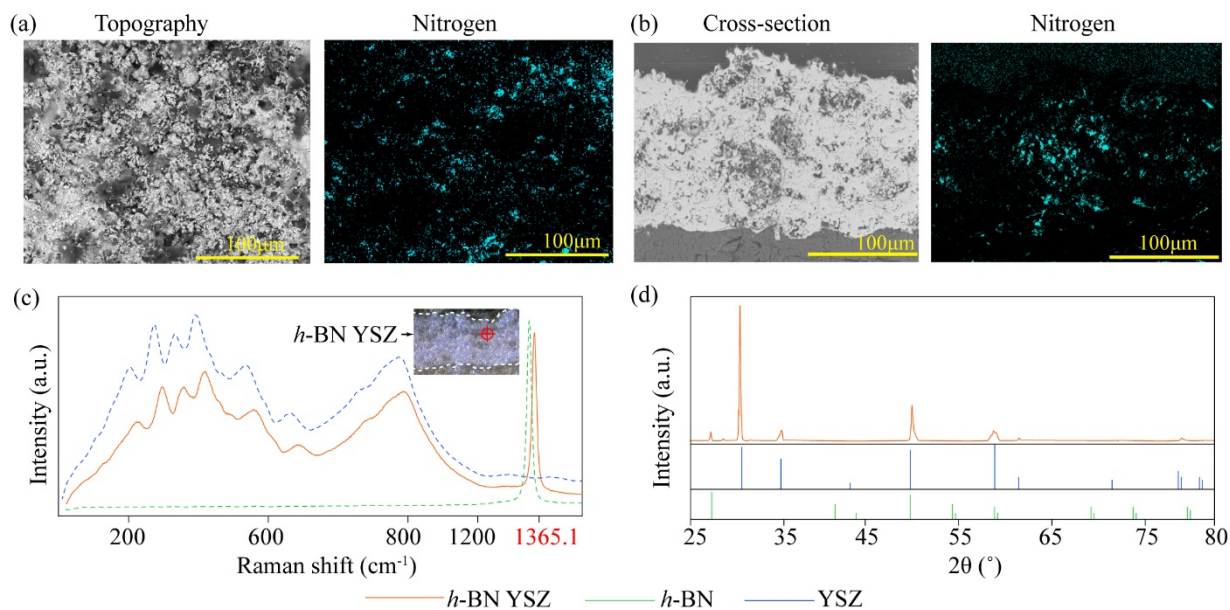
128 **2.5 Characterization Procedures**

129 Standard metallographic techniques were employed to polish the samples and observe the cross-section.
130 Surface roughness (R_a) of the coatings was evaluated using a Keyence VK-X1000 3D-laser scanning
131 microscope. Raman spectroscopic analysis was performed using a (Horiba Xplora) system, with a
132 wavelength of 532 nm. The samples were subsequently investigated by SEM and energy dispersive

133 spectroscopic (EDS) analysis to measure the maximum infiltration depth. The average porosity was
134 estimated across the cross section of infiltrated samples by utilizing ImageJ software.

135 3. Results and Discussion

136 3.1 TBC Characterization



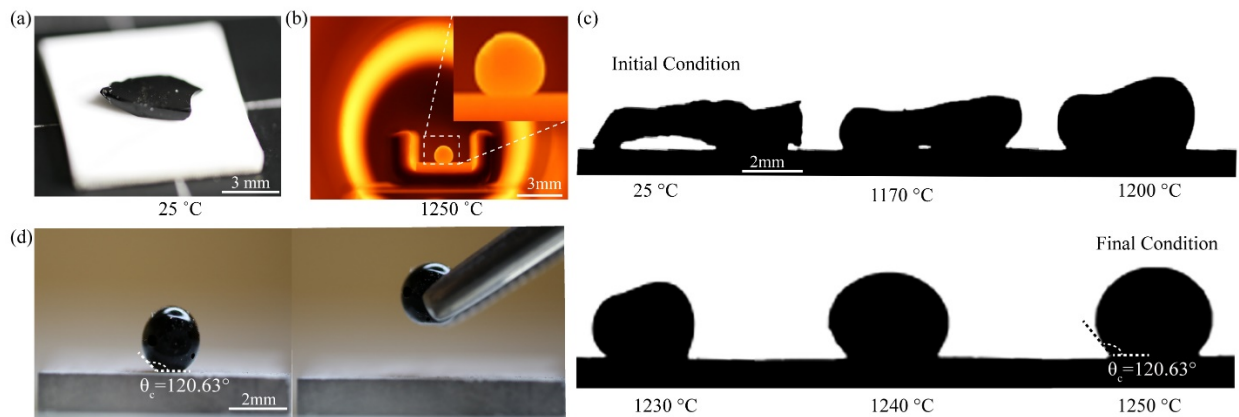
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138 **Fig. 2.** Characterization of *h*-BN YSZ. (a) SEM image of *h*-BN YSZ coating surface and the corresponding distribution
139 of nitrogen obtained by EDS. (b) Cross-sectional SEM image of *h*-BN YSZ coating and the corresponding nitrogen
140 distribution obtained by EDS. (c) Raman spectra acquired at the cross-section of YSZ, *h*-BN YSZ coating and cross-
141 section of *h*-BN substrate. (d) X-ray diffraction pattern of the *h*-BN YSZ TBC indicates the presence of *h*-BN and *t*-
142 YSZ, as compared to standard *h*-BN (crystallography open database (COD): 2016170) and *t*-YSZ (COD: 1521477)
143 XRD profile data.

144 As-sprayed YSZ and *h*-BN YSZ coatings consist of a lamellar structure formed by coarse splat like
145 microstructure. The average porosity of *h*-BN YSZ is ~29% , which is slightly higher than that of YSZ
146 coatings (~22 %). Surface roughness (R_a) values of *h*-BN YSZ and YSZ coatings are $7.3 \pm 1.02 \mu\text{m}$ and 6.8
147 $\pm 0.18 \mu\text{m}$, respectively (See Fig. S1). R_a of TBCs play an important role in the wetting of molten volcanic
148 ash and it was ensured that R_a was uniform across the surface (Fig. S2). The quality of the *h*-BN additives
149 in the *h*-BN YSZ coatings might be compromised by oxidation in air at high temperatures (~1100°C) [29].
150 Thus, particular attention was paid to characterizing any potential oxidation. Coating mapping using
151 electron dispersive spectroscopy identified a small fraction of nitrogen (N), scattered onto and within the
152 coating, suggesting minor oxidation (Figs. 2a and 2b). Raman spectral mapping of the coating revealed a

153 peak at 1365 cm^{-1} (Fig. 2c) and XRD providing unambiguous evidence for the presence of pure (non-
 154 oxidized) *h*-BN particulates in the coating. XRD of the coating further indicates the presence of both *h*-BN
 155 (peak: 27°) and tetragonal YSZ (*t*-YSZ) (peak range: $34.6\text{--}35.8^\circ$), as expected in a well fabricated *h*-BN
 156 YSZ TBC (Fig. 2d). Hence, we must conclude that the oxidation of *h*-BN in the coating remains minor
 157 during our experiments.

158 3.2 Wetting resistance of *h*-BN YSZ against molten volcanic ash

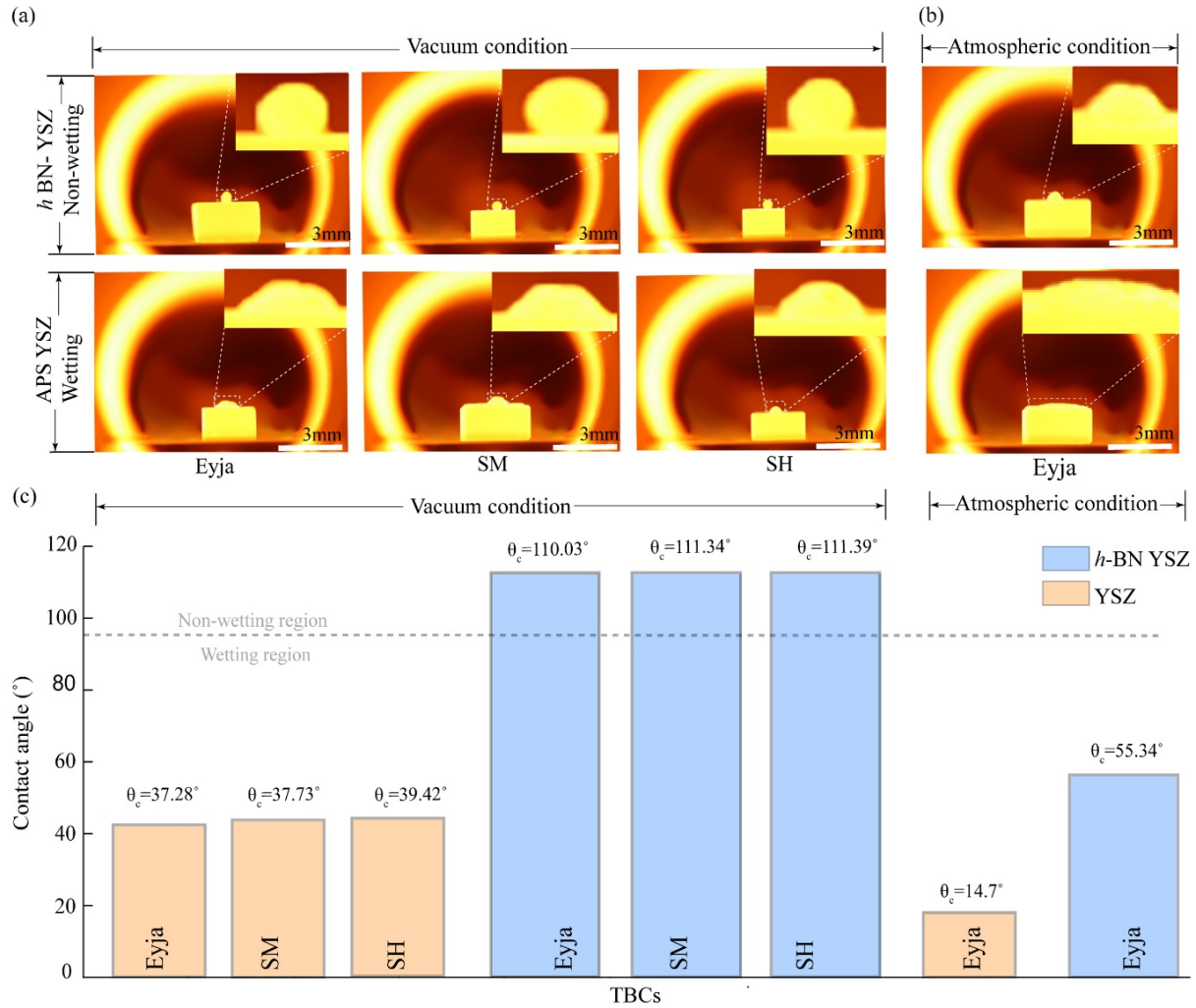
159 Hexagonal boron nitride (*h*-BN) is currently regarded (to best of our knowledge) as the sole refractory
 160 ceramic material exhibiting a non-wettability of molten Si inferred to be due to its covalently-bonded
 161 structure, involving honeycomb-structured sheets bound by weak ionic inter-planar attractions [30–32]. To
 162 evaluate the wettability of *h*-BN with molten volcanic ash, a single Eyja glass shard was loaded onto *h*-BN
 163 substrate (Fig. 3a) and the sample assembly was heated to $1250\text{ }^\circ\text{C}$ (Figs. 3b and 3c). Once cooled to ambient
 164 room temperature, it was observed that solidified Eyja glass exhibited non-wetting and non-sticking
 165 properties on *h*-BN substrate (Fig. 3d and Movie S1). The non-wetting of molten Eyja ash on the *h*-BN
 166 substrate could be attributed to surficial reaction process, by the formation of Si_3N_4 layer [33].



167
 168 **Fig. 3.** Non-wetting behavior of hexagonal-boron nitride (*h*-BN) thin film substrate. (a) Eyja glass shard before thermal
 169 treatment on *h*-BN thin film substrate. (b) Real time photograph of the molten Eyja at $1250\text{ }^\circ\text{C}$, in-vacuo condition. (c)
 170 Sequential In-situ silhouette morphological transition of the irregular shard to a sphere from $25\text{ }^\circ\text{C}$ to $1250\text{ }^\circ\text{C}$. (d)
 171 Photographs showing the non-wetting behavior (left) and non-adhesion property (right) of solidified melt on the surface
 172 of *h*-BN substrate.

173 To evaluate the wettability of *h*-BN YSZ coating with molten volcanic ash, preliminary wetting experiments
 174 were performed under vacuum, for a duration of 1 hour. We find that the three volcanic ash samples evolved
 175 to form non-wetting, near-spherical molten droplets with high contact angles $\theta_c \sim 110^\circ$ when resting on the

176 *h*-BN YSZ, whereas they significantly wetted the surface of YSZ coating, evidenced by low contact angles
 177 $\theta_c \sim 40\text{--}55^\circ$ (Figs. 4a and 4c; Movie S2). The results demonstrate that *h*-BN YSZ coating exhibit non-
 178 wettability to molten volcanic ash under vacuum.



179
 180 **Fig. 4.** Wettability of volcanic ash on conventional YSZ and novel *h*-BN YSZ coatings. (a) Photographs of the three
 181 volcanic ash droplets with variable wetting degrees on the surface of *h*-BN YSZ and YSZ coatings in-vacuo at 1250 °C.
 182 (b) Photographs of molten Eyja ash droplets on the *h*-BN YSZ and YSZ coatings at 1250 °C, in oxidizing condition.
 183 (c) Comparison of the contact angles resultant from the wetting of molten volcanic ash onto *h*-BN YSZ and
 184 conventional YSZ coatings in vacuum and atmospheric oxidizing conditions at 1250 °C.

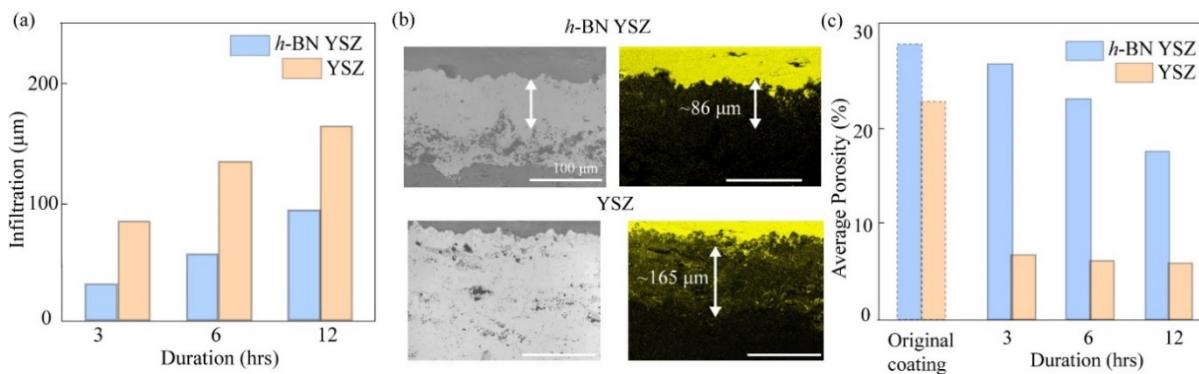
185 Bearing in mind the potential oxidation of *h*-BN, we further determined the wetting properties of the
 186 coatings with Eyja ash under oxidizing atmospheric conditions. Eyja ash was chosen due to the lowest
 187 melting point and lowest viscosity at 1250 °C compared with the other volcanic ash samples. Here, 1-hour
 188 hold-time tests were performed and *in-situ* images at 1250 °C reveal that the melt completely spread onto
 189 the surface of YSZ coating ($\theta_c = 14.7^\circ$). Spreading of molten Eyja ash on *h*-BN YSZ (Figs. 4b and 4c) was

190 much more limited, exhibiting a contact angle $\theta_c = 55.34^\circ$, more than thrice that of conventional YSZ coating
191 (Fig.4c and Movie S3). These results demonstrate that the novel *h*-BN YSZ coatings have an increased
192 resilience to wetting by molten volcanic ash compared to YSZ coatings, under atmospheric conditions. In
193 order to examine the influence of R_a on the wetting, *h*-BN YSZ of $R_a = 3.43 \pm 0.37 \mu\text{m}$ was fabricated (Fig.
194 S3) and θ_c was found to be 51.5° , post wetting of molten Eyja ash for a soak-time of 1 hour (Movie S4).
195 Thereby, it was inferred that the variation in R_a of *h*-BN YSZ coating did not influence the degree of wetting
196 on the surface. To resolve the extent to which molten volcanic ash spreads onto the TBCs, the spreading
197 area of the melt was evaluated. We note that the onset of colour fading was observed by the formation of a
198 faint-brown ring/ corrosion ring (Fig. S4) [34]. The corrosion ring formed post 3 hrs for the YSZ coating,
199 whereas it was observed post 12 hrs for the *h*-BN YSZ coating, that is 4 times longer. Surficial reaction as
200 a result of complete wetting of the molten Eyja ash ($\theta_c \sim 0$), is evident from the formation of the corrosion
201 ring, which is visible for a soak time of 3 hours on YSZ coating, in contrast to 12 hrs long term exposure of
202 *h*-BN YSZ coatings.

203 **3.3 Infiltration resistance of *h*-BN YSZ against molten volcanic ash**

204 The infiltration of molten Eyja in the coatings after thermal aging was determined by the elemental
205 distribution of Si, mapped across the cross-section of the coatings. θ_c made by molten Eyja on the surface
206 of *h*-BN YSZ was 43° during a soak time of 3 hrs and 16° during a soak time of 6 hrs at 1250°C (See Fig.
207 S5), whereas molten Eyja thoroughly wetted the surface of YSZ coatings for a soak-time of 3 hrs at 1250°C
208 (See Fig. S6). θ_c of thermally soaked *h*-BN YSZ was higher than conventional YSZ, this was related to its
209 minimal infiltration as seen in Fig 5a. Previous studies on the modifications of the microstructure of TBCs
210 suggested that a minor difference in θ_c , had a significant impact on the infiltration depth of molten CMAS
211 into TBCs [35]. For a soak-time of 12 hrs, the maximum infiltration depth of *h*-BN YSZ ($86 \mu\text{m}$) was twice
212 the infiltration depth of 6hrs thermally aged *h*-BN YSZ ($43 \mu\text{m}$) (Figs. 5a and 5b). Previous research on
213 Alumina doped YSZ has also documented shallow penetration of molten CMAS into a highly porous coating
214 [36]. Here, we observe that the lower wetting propensity of molten ash on *h*-BN YSZ versus YSZ coating
215 had a significant impact on the infiltration depth, presumably owing to the reactive wetting of *h*-BN and

216 molten silicate. The inter-relationship between thermal aging and variation in contact angle was observed
 217 as the molten volcanic ash completely wetted the surface of 12 hrs thermally aged *h*-BN YSZ coating from
 218 16° of 6 hrs thermally aged *h*-BN YSZ coating. However, the lower sub-region of the 12 hrs thermally aged
 219 *h*-BN YSZ coating was unaffected by infiltration and the influence of oxidation could be minor (Fig. 5b),
 220 due to the presence of N in the underlying unaffected layer (See Fig. S7). It should be noted that the
 221 infiltration of molten volcanic ash results in the reduction of porosity of the coatings, and increases the
 222 thermal conductivity[37]. From Fig. 5c it is observed that the average porosity of 12-hrs thermally aged *h*-
 223 BN YSZ coating was 3.6 times higher than APS coating, this might be the reason for the minimal influence
 224 of oxidation to the underlying unaffected layer of *h*-BN YSZ coating. However, to conclude regarding the
 225 oxidation aspect of the novel coatings, an in-depth thermo-chemical analysis of *h*-BN YSZ and molten
 226 silicate interaction is required.

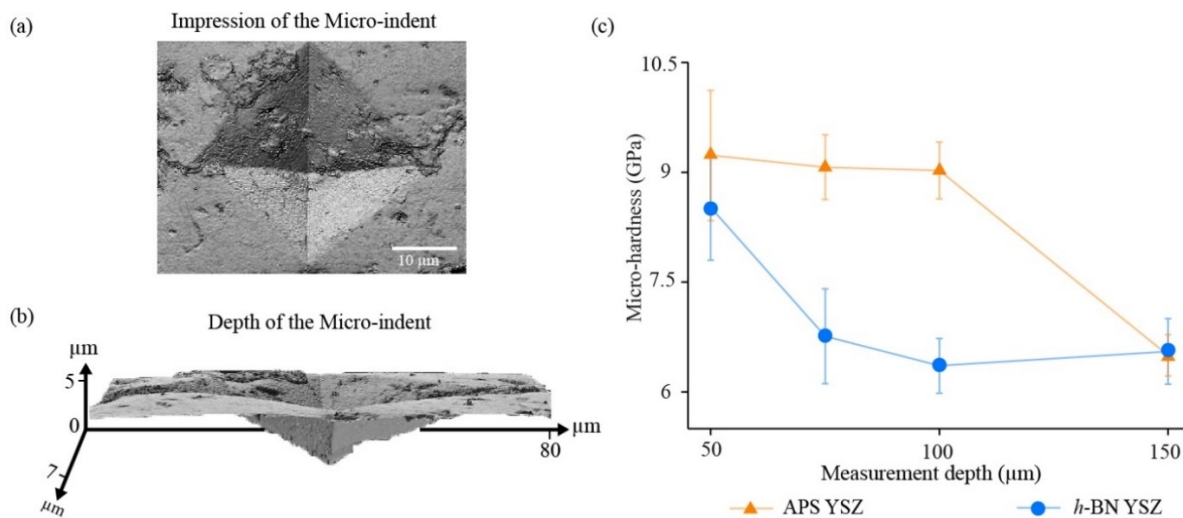


227
 228 **Fig. 5.** Infiltration of molten of Eyja ash into the coatings at atmospheric condition. (a) Maximum infiltration depth
 229 (μm) of molten volcanic ash into each TBC. (b) Si-Elemental maps of *h*-BN YSZ and YSZ coating for soak-time of
 230 12 hrs at 1250 °C. (c) Average porosity (%) of as-sprayed coatings and coatings infiltrated by molten volcanic ash.

231 The coatings were further subjected to thermal aging for 48 hours, and it was observed that there was
 232 no residual glass present on the surface of the coatings and molten Eyja completely reacted with the coatings
 233 (see Fig. S8). It was evident that the conventional YSZ coatings completely reacted with molten Eyja.
 234 Whereas *h*-BN YSZ coating was prone to the reaction in an uneven manner, with an increase in density of
 235 globular Si-based reaction products. The reason behind this phenomenon is still unclear and our future work
 236 will be directed to produce uniform layered *h*-BN YSZ coating structure, in order to investigate this behavior.

237 3.4 Mechanical properties

238 Micro-hardness values of the coatings subjected to infiltration by molten Eyja, for a soak time of three hours
 239 was evaluated. The average micro-hardness values of as-sprayed conventional YSZ (6.55 ± 1.2 GPa) and
 240 *h*-BN YSZ coatings (6.52 ± 0.5 GPa) were nearly the same. The micro-hardness of molten Eyja infiltrated
 241 coatings was measured across the cross-section, tracking the infiltration depths at 50, 75, 100 and 150 μm
 242 respectively. At an infiltration depth of 50 μm (example: Fig. 6a and 6b), it was evident that both
 243 conventional YSZ (9.23 ± 0.89 GPa) and *h*-BN YSZ (8.51 ± 0.72 GPa) coatings exhibited high micro-
 244 hardness values (Fig. 6c), due to the infiltration of molten Eyja. At 75 μm , conventional YSZ coating (9.06
 245 ± 0.44 GPa) exhibited higher micro-hardness values than *h*-BN YSZ coating (6.76 ± 0.56 GPa), this
 246 corresponds to the presence of molten Eyja in conventional YSZ coating, as the micro-hardness value of
 247 solidified Eyja glass shard (Fig. S9) was measured to be 5.19 ± 0.6 GPa.



248
 249 **Fig. 6.** Micro-hardness of the coatings. (a) Indent impression of *h*-BN YSZ coating. (b) Depth of indent impression
 250 on *h*-BN YSZ coating. (c) Change in the micro-hardness corresponding to the measurement depth

251 The infiltration of molten Eyja into the coating and further solidification led to the increase in micro-
 252 hardness value (accompanied by stiffening), which results in the mechanical degradation of the coating [38,
 253 39]. *H*-BN YSZ coatings exhibit better stiffening resistance than the conventional YSZ coatings, as wetting
 254 of the coatings play a governing role in the infiltration of molten volcanic ash thereby influencing the micro-
 255 hardness value.

256 4. Conclusion

257 In order to counteract the propensity of ingested, molten volcanic ash droplets to wet and infiltrate TBCs
258 and cause damage to jet engines, we have developed and tested new *h*-BN YSZ coatings which exhibit
259 enhanced resistant behavior against molten volcanic ash wetting. Employing controlled laboratory testing,
260 we first demonstrated that pure *h*-BN substrates have silica-phobic properties under vacuum conditions at
261 1250 °C. Further, we have fabricated novel coatings containing 80 wt.% YSZ and 20 wt.% *h*-BN, and tested
262 their ability to resist molten volcanic ash droplets, compared to conventional YSZ coating. Under vacuum
263 conditions, this new coating repelled volcanic ash droplets, preventing infiltration of the substrate, unlike
264 the conventional coating. Under oxidizing conditions, *h*-BN YSZ coating appeared more resilient as we find
265 that molten ash only partially wets the substrate *i.e.* the molten ash spread less and more slowly on *h*-BN
266 YSZ *versus* conventional TBCs. *H*-BN YSZ coating exhibited better stiffening resistance than conventional
267 YSZ coating, due to their resilience against rapid wetting and infiltration of molten volcanic ash. The
268 thermo-mechanical aspect of degradation and potential challenges like the prevalence of water vapor in a
269 gas-turbine environment on *h*-BN YSZ, will be discussed in the future. This work extends the scope for
270 physicochemically resilient “silica-phobic” TBCs with *h*-BN additives, which possess the potential for
271 mitigating risks associated with volcanic ash ingestion into jet engines. Further investigations employing
272 such TBC solutions against other atmospheric contaminants, as well as testing of its performance under
273 severe thermal cycling conditions are underway.

274 **Data availability**

275 The data supporting the findings of this study are available from the corresponding author upon reasonable
276 request.

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Novel Thermal Barrier Coatings with Hexagonal Boron Nitride Additives Resistant to Molten Volcanic Ash Wetting

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Supporting Information

Movie Descriptions

Movie S1. Morphological silhouette transition of Eyja glass shard on *h*-BN substrate from room temperature to 1250 °C, performed in vacuum conditions.

Movie S2. Silhouette transition of Eyja, SM and SH ash pellets on conventional YSZ and *h*-BN YSZ coatings from room temperature to 1250 °C, performed in vacuum conditions.

Movie S3. Silhouette transition of Eyja ash pellet on conventional YSZ and *h*-BN YSZ coatings from room temperature to 1250 °C, performed in oxidizing conditions.

Movie S4. Influence of *h*-BN YSZ surface roughness on wetting of Eyja ash on the coating surface.

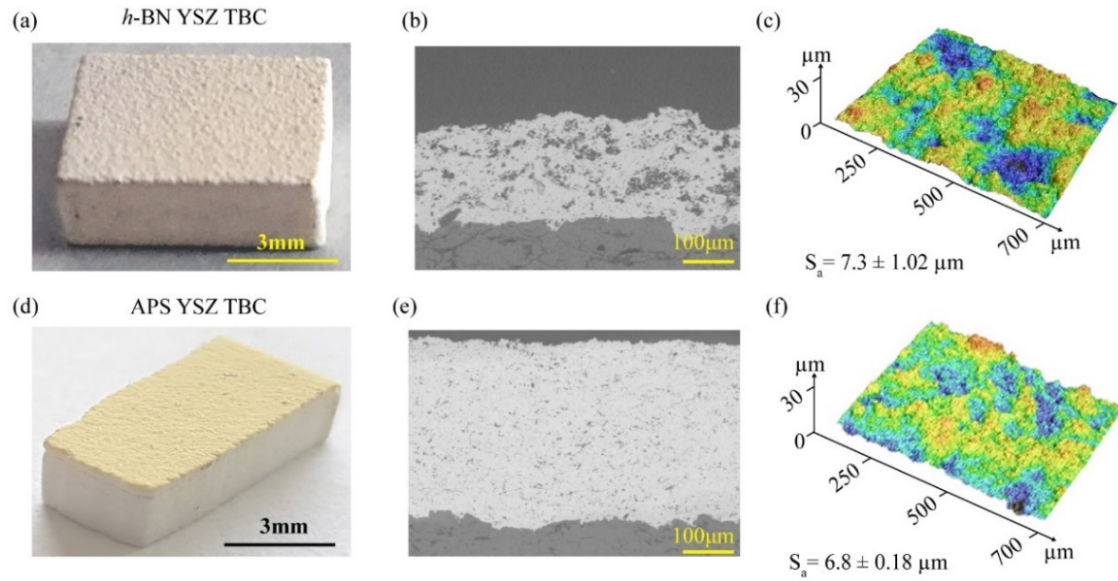


Fig. S1. Properties of *h*-BN YSZ and conventional APS YSZ coatings. (a) Photograph of *h*-BNYSZ coating (6×3 mm) fabricated by thermal spay deposition on an alumina substrate. (b) Cross-sectional SEM image of *h*-BN YSZ coating (thickness: $\sim 200\mu\text{m}$). (c) Surface roughness profile of *h*-BN YSZ coating topography evaluated by confocal laser microscopy. (d) Photograph of conventional APS YSZ coating (6×3 mm) on an alumina substrate. (e) Cross-sectional SEM image of APS YSZ coating (thickness: $\sim 400\mu\text{m}$) (f) Surface roughness profile of APS YSZ coating topography evaluated by confocal laser microscopy.

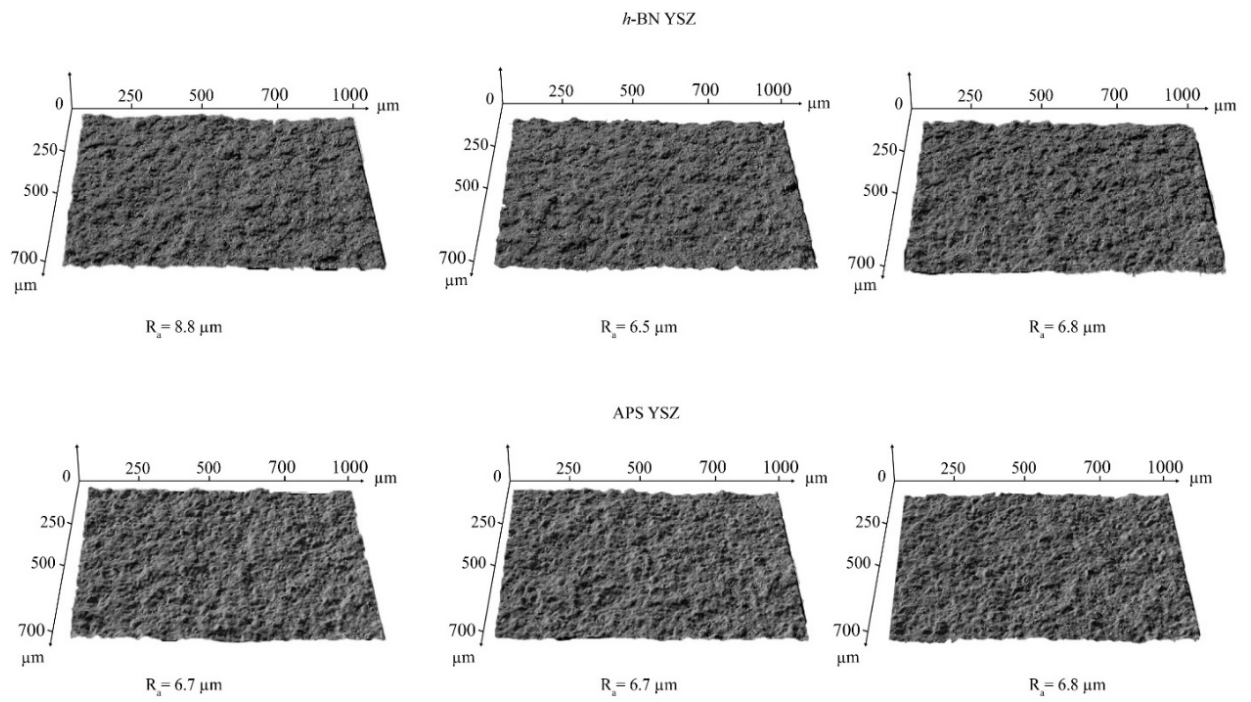
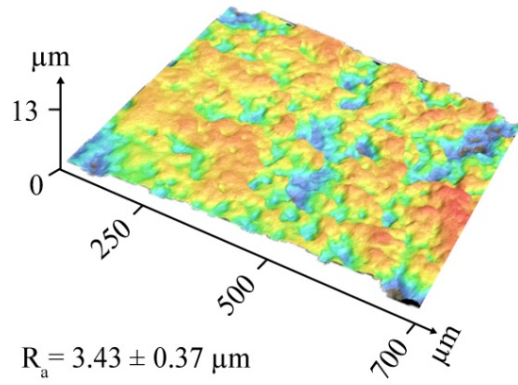


Fig. S2. Surface profile across *h*-BN YSZ and APS YSZ coatings.

(a) Surface profile



(b) Spreading behavior

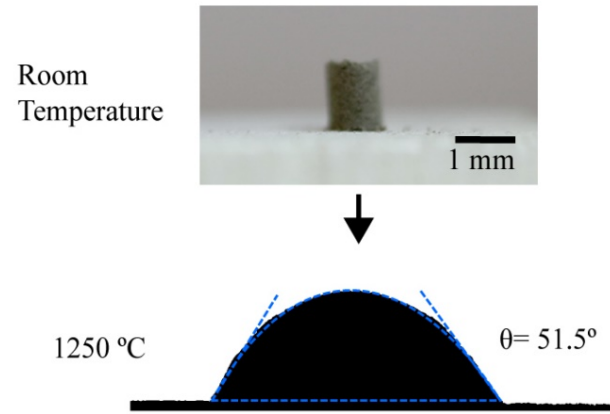


Fig. S3. Influence of R_a on wetting (a) R_a value of h-BN YSZ (b) spreading behavior.

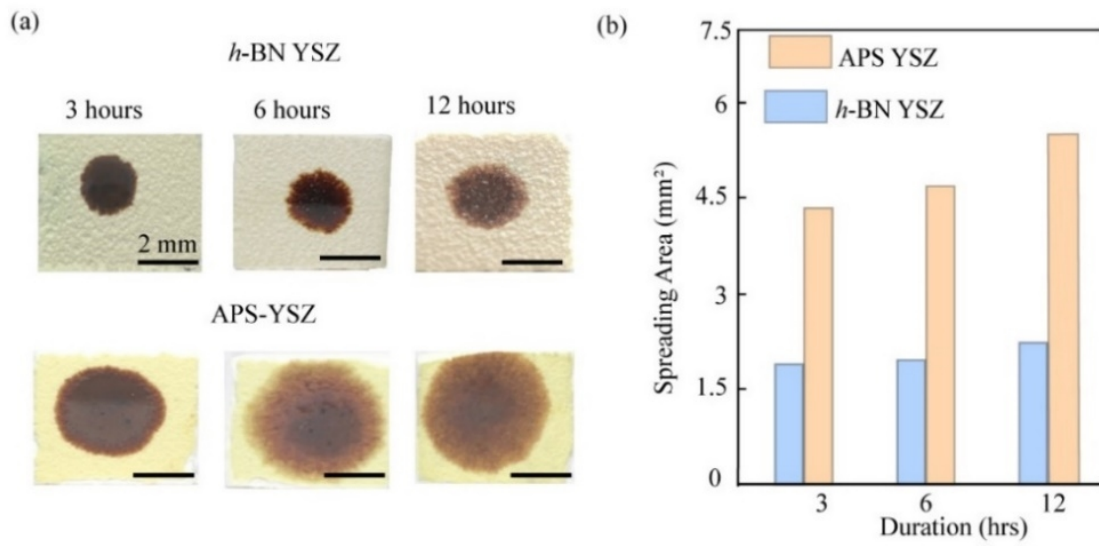


Fig. S4. Extent of Spreading of Eyja ash on the coatings under atmospheric condition. (a) Spreading of molten ash on the surface of *h*-BN YSZ (upper row) and conventional APS TBCs (lower row) at 1250 °C. (b) Comparison of the extent of spreading on the coatings.

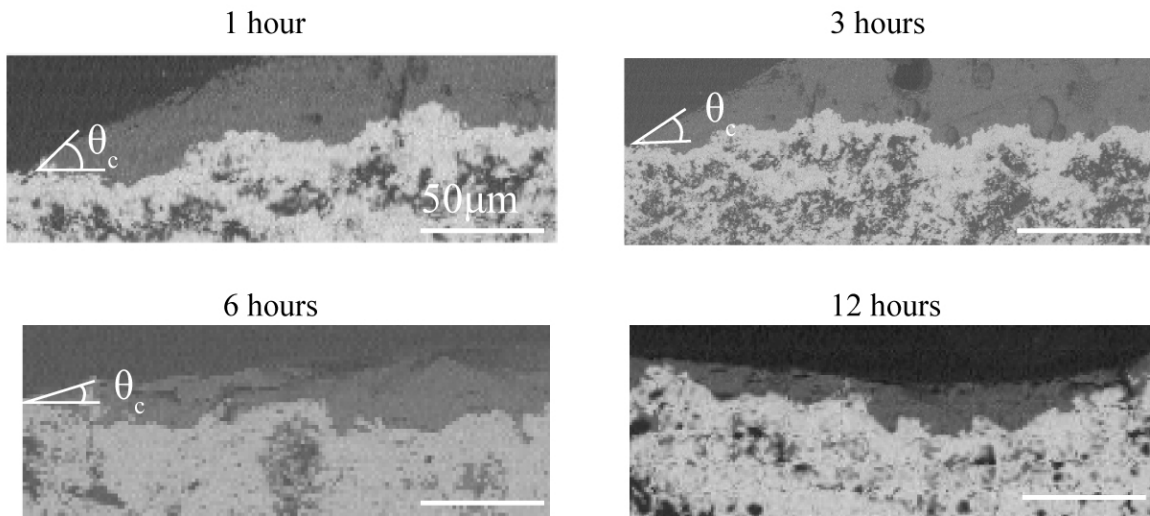


Fig. S5. Contact angle of molten Eyja ash on *h*-BN YSZ post thermal soak time of 1, 3, 6 and 12 hours.

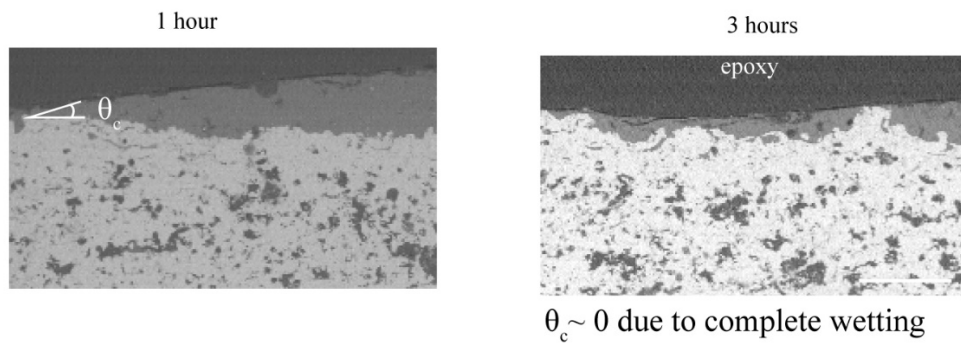


Fig. S6. Contact angle of molten Eyja ash on *h*-BN YSZ post thermal soak time of 1 and 3 hours

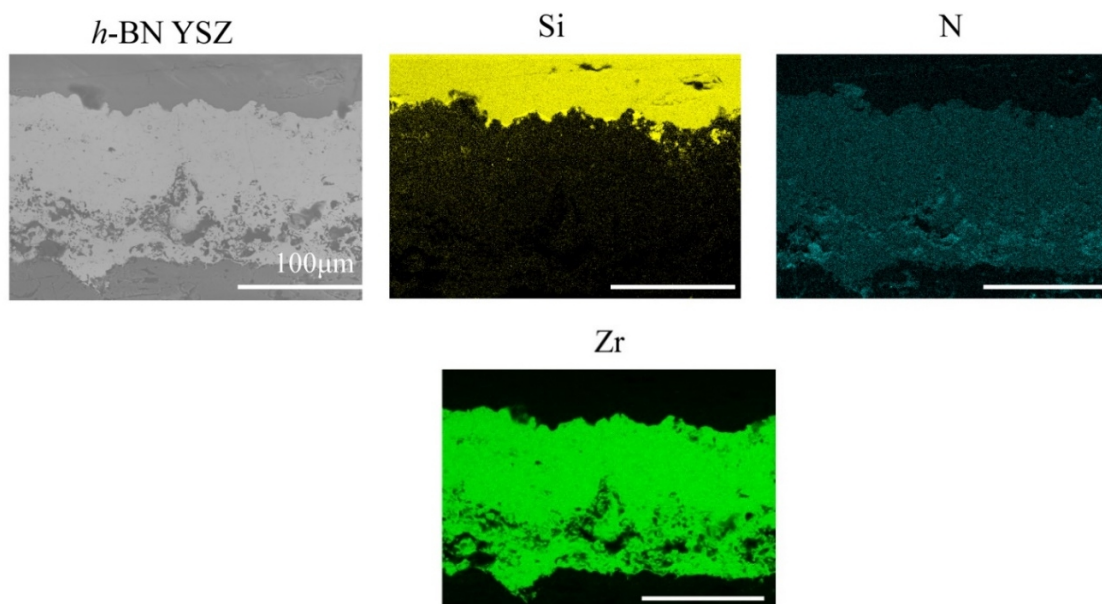
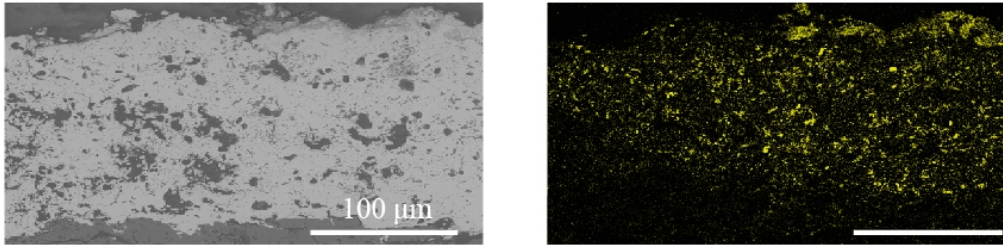


Fig. S7. EDS elemental maps of 12hrs thermally soaked *h*-BN YSZ coating.

h-BN YSZ



YSZ

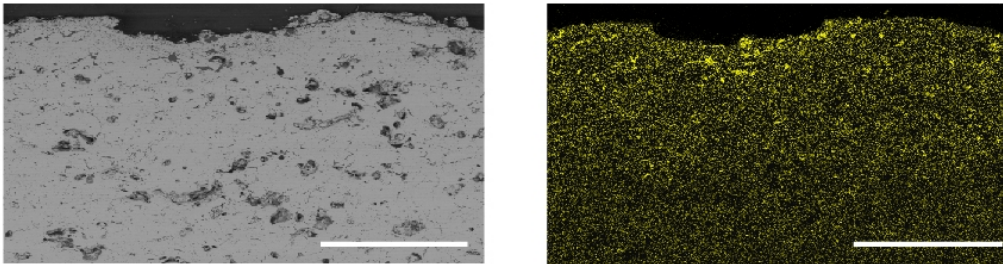


Fig. S8. Si elemental maps of 48hrs thermally soaked *h*-BN YSZ and conventional YSZ coatings.

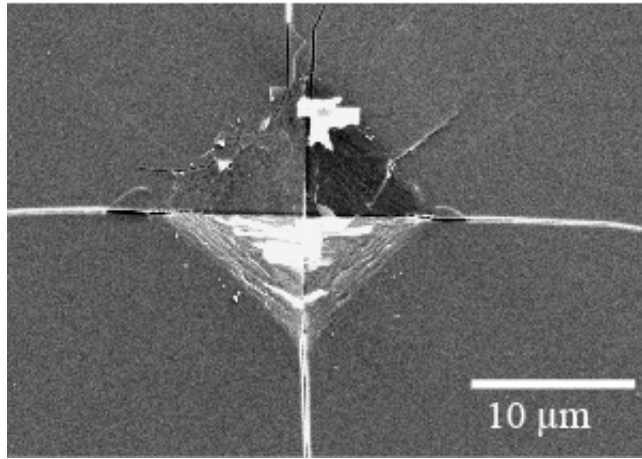
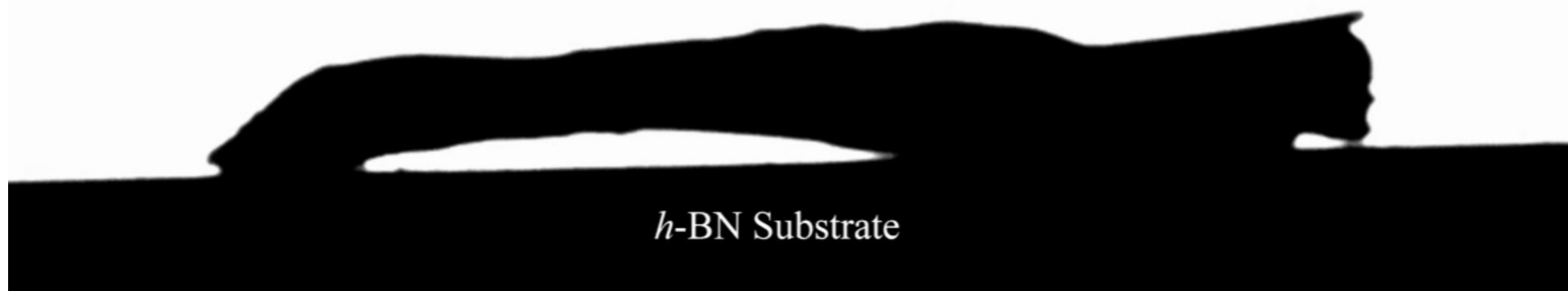


Fig. S9. Impression of the indent on the Eyja glass shard.

Temperature : 25°C
Time : 1 min

1 mm



Temperature : 25°C

Time : 1 min

1mm

Eyjafjallajökull

YSZ



h-BN YSZ



Santa Maria

YSZ



h-BN YSZ



Soufriere Hills

YSZ



h-BN YSZ



Temperature : 25°C
Time : 1 min
Volcanic ash : Eyjafjallajökull

1 mm

YSZ

h-BN YSZ



h-BN YSZ

Temperature : 25°C

Time : 1 min

Volcanic ash : Eyjafjallajökull

