Towards a novel microfluidic device for synthesis of gold nanoparticles for drug delivery Loughborough University

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1.Motivation and challenges:

- Gold nanoparticles (AuNPs) are bio-compatible and exhibit unique size dependent optical, physical and electromagnetic properties [1]. As such, AuNPs are increasingly used in various industries and research areas, such as electronics, sensing, catalysis and drug/gene delivery [2].
- AuNPs are relatively easier and inexpensive to synthesise using chemical reduction between a suitable gold salt and a reducing agent.
- Unlike the most common batch processes, microfluidic approach provides improved control over the size and the polydispersity of the synthesised AuNPs and many more advantages such as possibility to use expensive/toxic materials in small volumes, integration of several unit operations in a single device [3] etc.
- Main limitation of synthesis of AuNPs in microfluidic devices is the reactor fouling which need to be addressed when considering scale-up processes.

2. Microfluidic device:

Co-flow glass capillary device was fabricated in the laboratory by coaxially centring a round capillary with specifically tapered injection orifice with a square capillary and fixed on a microscopic slide using epoxy glue. Specially prepared hypodermic needles were placed on delivery points as shown in the Figures 1 and 2.

5. Experimental results:







Figure 1: Photograph of the co-flow glass capillary microfluidic device on the inverted microscope

Figure 2: Co-flow glass capillary microfluidic device and the flow pattern inside the device

3. Experimental setup and the reaction:

Effect of injection orifice size (100 µm, 180 µm, 240 µm), ascorbic acid flow rate and the pH of ascorbic acid stream (pH 3.0±0.2 & pH 10.3±0.2) on particle size and the polydispersity of synthesised AuNPs were investigated to determine optimum conditions to synthesise smaller



Polydispersity Index Vs Ascorbic acid flow rate in different injection orifice sizes of 100, 180 and 240 µm when the gold precursor flow rate was 15 ml/hr and initial pH of ascorbic stream was 10.2



Figure 8: TEM and SEM images of synthesised AuNPs with different parameters (guide; injection orifice diameter – PVP status – initial pH of ascorbic acid) (a) TEM image 180µm– 40K PVP-pH around 10 (b) 240µm-with40KPVP-pH around 3 (c) 240µm–40KPVP-pH around 10 (d) 180µm-40KPVP-pH around 3.

Figure 6: (a) Particle size Vs Ascorbic acid flow rate (b) Figure 7: (a) Particle size Vs Absorbance peak wavelength (b) PDI Vs Absorbance peak wavelength of synthesised AuNPs with different pH of ascorbic acid stream



Figure 9: Observations done by the naked eye regarding the reactor fouling control. By using PVP in the gold precursor stream and elevating pH of ascorbic acid stream, reactor fouling was controlled successfully.

6. Conclusions:

Successfully synthesised 40 - 80nm AuNPs with very low polydispersity using smaller injection orifice sizes, higher ascorbic acid flow rates and higher pH of ascorbic acid stream.

Successfully minimise the reactor fouling by using PVP in the gold precursor stream and changing the pH of ascorbic acid stream to 10.3±0.2.

7. Future work:

Use of 3 phase droplet based microfluidic device to prevent reactor fouling as well as improve mixing of reactants to achieve better results on size and polydispersity of AuNPs.

Investigations on new 3D printed microfluidic device design [6].

Encapsulation of a drug using AuNPs and a polymer and investigations of drug administration using a gene gun [7].





| REFERENCES: | AKNOWLEDGEMENT: | CONTACT INFORMATION |
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| Roduner, E. (2006). Size matters: why nanomaterials are different. Chemical Society Reviews, 35(7), 583–592. Ali, M.E. et al., 2012. Gold nanoparticle sensor for the visual detection of pork adulteration in meatball formulation. Journal of Nanomaterials, 2012, p.1. Chithrani, D.B. et al., 2010. Gold nanoparticles as radiation sensitizers in cancer therapy. Radiation research, 173(6), pp.719–728. Lee, JS., 2010. Recent progress in gold nanoparticle-based non-volatile memory devices. Gold Bulletin, 43(3), pp.189–199. Stuchinskaya, T. et al., 2011. Targeted photodynamic therapy of breast cancer cells using antibody-phthalocyanine-gold nanoparticle conjugates. Photochemical & photobiological sciences Official journal of the European Photochemistry Association and the European Society for Photobiology, 10(5), pp.822–831. Tsuno-yama, H., Ichikuni, N. & Tsukuda, T., 2008. Microfluidic synthesis and catalytic application of pvp-stabilized, 1 nm gold clusters. Langmuir, 24(20), pp.11327–11330. DeMello, A.J., 2006. Control and detection of chemical reactions in microfluidic systems. Nature, 442(7101), pp.394–402. (a) Tyagi, H., Kushwaha, A., Kumar, A., & Aslam, M. (2011). pH-Dependent Synthesis of Stabilized Gold Nanoparticles Using Ascorbic Acid. International Journal of Nanoscience, 10(04n05), 857–860. (b) Vasilescu, A., Sharpe, E., & Andreescu, S. (2012). Nanoparticle-Based Technologies for the Detection of Food Antioxidants. Current Analytical Chemistry, 8(4), 495–505. Wagner, J., Kirner, T., Mayer, G., Albert, J., & Köhler, J (2004). Generation of metal nanoparticles in a microchannel reactor. Chemical Engineering Journal, 101(1-3), 251–260. Martino, C. et al., 2014. A 3D-printed microcapillary assembly for facile double emulsion generation. Lab on a chip, 14(21), pp.4178–82 Zhang, D., Das, D.B. & Rielly, C.D., 2013. Potential of microneedle-assisted micro-particle delivery by gene guns: a review. Drug delivery, 7544(8), | Loughborough University for funding this project LMCC for the TEM and SEM images Technical staff of the Department of Chemical Engineering for their training and assistance Dr Simon Tuplin for the assistance on 3D printing the new de- vice Tony Eyre for designing the microscopic stage for the new 3D printed device | m.v.bandulasena@lboro.ac.uk Department of Chemical Engineering Loughborough University Leicestershire, LE11 3TU, UK WWW.lboro.ac.uk/departments/chemical |