

Influence of contact angle and temperature on evaporation of droplets

V. Starov,¹ S. Semenov,¹ R. Rubio²

¹*Department of Chemical Engineering, Loughborough University, Loughborough, Leicestershire, LE11 3TU, UK, S.Semenov@lboro.ac.uk*

²*Chimica fisica I, Univ Complutense, Ciudad Universitaria, Madrid, 28040, Spain*

It was shown experimentally earlier, that the rate of change of the volume of an evaporating droplet depends linearly on the radius of the droplet base [1, 2]:

$$\frac{dV(t)}{dt} = -\alpha L(t), \quad (1)$$

where t is time, $V(t)$ is the droplet volume, α is a proportionality constant, $L(t)$ is the radius of the droplet base. The latter means proportionality of total flux of vapour from the surface of a droplet, J , to the radius of the droplet base, L .

Computer simulations were carried out to verify the above experimental dependency (1), as well as to investigate the dependency of proportionality coefficient α on contact angle, θ , and average temperature, T_{av} , of the droplet surface.

The following phenomena were taken into account in our simulations: a heat conduction in the substrate, liquid droplet and surrounding air; the latent heat of vaporisation; vapour diffusion in the surrounding air; viscous flow in a droplet caused by thermocapillary tangential stress.

Convection in the air was neglected because experiments [3] have shown that there is no influence of a forced convection in the surrounding air on the evaporation rate. The Soret effect also was neglected due to small range of temperatures (less than 3 degrees) in the system under investigation.

The simulation has revealed that the proportionality coefficient α substantially depends on the contact angle, θ , as well as directly proportional to the difference of saturated and ambient vapour concentrations, $C_{sat}(T_{av}) - C_{\infty}(T_{\infty})$, at the droplet surface and in ambient air respectively. The results of simulations are in an agreement with the analytical solution obtained by R.G. Picknett and R. Bexon [4]. The only influence of the average temperature, T_{av} , of the droplet surface on α is the change of saturated vapour concentration $C_{sat}(T_{av})$.

This research is supported by MULTYFLOW, EU project, and Engineering and Physical Sciences Research Council, UK.

[1] K.S. Lee, C.Y. Cheah, R.J. Copleston, V.M. Starov, K. Sefiane, *Colloids and Surfaces A* **2008**, 323, 63 – 72.

[2] S. David, K. Sefiane, L. Tadrif, *Colloids and Surfaces A* **2007**, 298, 108 – 114.

[3] V. Starov, K. Sefiane, *Colloids and Surfaces A* **2009**, 333, 170 – 174.

[4] R.G. Picknett, R. Bexon, *Journal of Colloid and Interface Science* **1977**, 61 (2), 336 – 350.