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# Transpiration and moisture evolution in packaged fresh horticultural produce and the role of integrated mathematical models: A review

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1 **Transpiration and moisture evolution in packaged fresh**  
2 **horticultural produce and the role of integrated**  
3 **mathematical models: A review**

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50  
51 **Abstract**

52 Transpiration has various adverse effects on postharvest quality and the shelf-life of fresh fruit  
53 and vegetables (FFV). If not controlled, the water released through this process results in  
54 direct mass loss and moisture condensation inside packaged FFV. Condensation represents a  
55 threat to the product quality as water may accumulate on the product surface and/or packaging  
56 system, causing defects in external appearance and promoting growth of spoilage  
57 microorganisms. Thus, moisture regulation is extremely important for extending FFV shelf-  
58 life. This review focuses on transpiration phenomenon and moisture evolution in packaged  
59 fresh horticultural produce. It provides recent information on various moisture control  
60 strategies suitable for packaging of fresh horticultural produce. It also provides an evaluation  
61 on the role and application of integrative mathematical modelling in describing water  
62 relations of FFV for packaging design, as well as, an overview of models reported in  
63 literature.

64  
65 **Keywords:** Moisture loss; packaging; humidity control; mathematical modelling; fresh  
66 produce; condensation

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## 71 **1. Introduction**

72 Fresh horticultural produce are highly perishable commodities, as they remain metabolically  
73 active even after harvest. Fresh produce continues to lose water due to transpiration and  
74 respiration process. This turns produce shelf-life into a race against the clock for growers,  
75 processors, and retailers to maintain quality and reduce food loss (Mahajan, Caleb, Singh,  
76 Watkins, & Geyer, 2014). This water loss is usually associated with economic loss since it  
77 causes a decrease in saleable mass, due to shrivelling of the product (Caleb, Mahajan, Al-  
78 Said, & Opara, 2013; Veraverbeke, Verboven, Van Oostveldt, & Nicolai, 2003b). In addition,  
79 moisture loss of the fresh produce can accumulate on the product surface and/or packaging  
80 system, causing defects in external appearance and promoting growth of spoilage  
81 microorganisms (Kang & Lee, 1998; Linke & Geyer, 2013). This leads to quality  
82 deterioration and flavour loss. Hence, it is important to remove or avoid moisture  
83 condensation on the product in order to maintain quality and prevent the growth of spoilage-  
84 causing microorganisms (Powers & Calvo, 2003).

85 According to Fonseca, Oliveira, and Brecht (2002) the goals of postharvest technology are to  
86 maintain freshness quality and reduce losses in the postharvest value chain of fresh fruit and  
87 vegetables (FFV). Temperature control and modification of atmosphere are important factors  
88 to extend a products shelf life (Fonseca et al., 2002). Nevertheless, besides these two factors  
89 the control of storage or in-package relative humidity (RH) is of critical importance (Tano,  
90 Oulé, Doyon, Lencki, & Arul, 2007). For example, Rux et al. (2015) investigated the  
91 transpiration behaviour of mushroom under different temperature and RH, and determined the  
92 effect of salt embedded humidity-regulating tray on in-package humidity and condensation  
93 behaviour. The authors reported that the humidity-regulating tray absorbed part of the water  
94 vapour produced by mushroom during the 6 d of storage, but its regulatory capacity was not  
95 efficient to avoid in-package moisture condensation. Therefore, understanding the  
96 physiological response of individual fresh horticultural produce towards optimum  
97 packaging/storage system design with adequate humidity control is one of the keys to  
98 achieving the postharvest technology goals.

99 Furthermore, mathematical modelling plays an important role in predicting the physiological  
100 response of FFV under different storage conditions. Mathematical models offer the possibility  
101 to describe characteristic changes in biological systems as a function of different  
102 environmental conditions, without the need to access these conditions in real time  
103 (Castellanos & Herrera, 2015). This makes it possible to optimise packaging design under  
104 different storage conditions for FFV (Kang & Lee, 1998), and to estimate the packaging

105 requisites for specific fresh produce (Caleb et al., 2013; Sousa-Gallagher, Mahajan, &  
106 Mezdad, 2013).

107 In this context, the aim of this article is to provide a comprehensive review regarding the  
108 transpiration phenomenon and moisture evolution inside packaged fresh horticultural produce.  
109 The role and application of integrative mathematical modelling in describing water relations  
110 of fresh horticultural produce for packaging design is discussed. In addition, an overview of  
111 the various moisture control strategies, mathematical models reported in literature, and future  
112 prospects is presented.

113

## 114 **2. Transpiration phenomenon in fresh horticultural produce**

115 Transpiration is a critical physiological process for FFV (Xanthopoulos, Athanasiou, Lentzou,  
116 Boudouvis, & Lambrinos, 2014). Once separated from the mother plant, FFV cannot replace  
117 water from the plant and/or soil and depend on their own water content for transpiration and  
118 organic substrate for respiration (Caleb et al., 2013). Transpiration phenomenon involves  
119 three main stages: i) moisture is transported as liquid and vapour from intercellular spaces to  
120 and through the skin of the product; ii) moisture is evaporated from the outer surface layer of  
121 the product; and iii) convective mass transfer of the moisture to the surroundings (Becker &  
122 Fricke, 2001; Veraverbeke, Verboven, Van Oostveldt, & Nicolai, 2003a). In terms of plant  
123 physiology there are four FFV components involved in the transpiration process this include:  
124 a) intercellular air spaces, through where water vapour diffuses inside the FFV; b) cuticle,  
125 responsible for the transpiration in which liquid water moves to the cell walls on the cuticle  
126 side of epidermal cells; where it can evaporate and the vapour is then diffused across the  
127 cuticle; c) stomata, through where water vapour diffuses in order to reach the boundary layer;  
128 and, d) boundary layer, which is located at the leaf surface and is the final component  
129 encountered by diffusing water vapour (Nobel, 2009).

130 Transpiration is driven by a concentration difference and can be described in terms of water  
131 activity differences across the membrane, moisture concentration and water vapour pressure  
132 differences between a product's surface and its surrounding (Becker & Fricke, 2001;  
133 Veraverbeke et al., 2003a, 2003b). Based on this definition, there should theoretically be no  
134 potential for transpiration phenomenon at 100% RH (i.e. saturated storage condition) and  
135 constant temperature since there is no water vapour pressure difference. However, this is not  
136 the case for saturated conditions as transpiration occurs due to the heat generated by the  
137 respiration process (Becker & Fricke, 1996; Sastry, Baird, & Buffington, 1977; Tano,

138 Kamenan, & Arul, 2005). Recently, Mahajan et al. (2016) investigated the moisture loss  
139 behaviour of three different FFV and a dummy evaporation sphere stored at 13 °C, 100% RH.  
140 Results showed that despite water vapour saturation the three tested products lost mass at  
141 100% RH, while no mass was lost from the evaporating sphere. These results agree with the  
142 hypothesis that respiratory heat can significantly influence moisture evolution from FFV  
143 under saturated conditions. This implies that transpiration in packaged fresh produce  
144 continues where water vapour saturation is commonly observed. It also indicates that the  
145 transpiration process under saturated conditions is a complex process that involves different  
146 heat components including respiratory heat generated by the product; evaporative cooling  
147 effect on the product's surface; convective heat transfer between the product and its  
148 surrounding environment.

149

## 150 **2.1. Potential effect on postharvest quality of fresh horticultural produce**

151 Transpiration phenomenon causes both water loss and evolution of free water from FFV,  
152 which may lead to formation of moisture condensation on the surface of product and/or  
153 packaging material. The free water, also known as moisture, facilitates the growth of fungal  
154 and bacterial pathogens (Holcroft, 2015; Linke & Geyer, 2013). Water loss results in direct  
155 mass loss, shrivelling, gloss reduction, limpness and wilting of horticultural produce. As the  
156 produce continues to lose water, its appearance, quality, shelf life, profitability, and consumer  
157 appeal diminishes (Holcroft, 2015; Thompson, Mitchell, Rumsay, Kasmire, & Crisosto,  
158 1998).

159 Water loss affects FFV in different degrees. According to Holcroft (2015), leafy vegetables  
160 wilt after approximately 3 - 5% of water loss, while for nectarines shrivelling occur after 19%  
161 of water loss. There is extensive literature stating the maximum permissible water loss (%) for  
162 a wide range of FFV (Kays & Paull, 2004; Robinson, Browne, & Burton, 1975; Thompson et  
163 al., 1998). For instance, the maximum permissible mass loss for grape and nectarine is 5%  
164 and 21%, respectively (Kays & Paull, 2004). For summer squash the permissible mass loss is  
165 24%, while for broccoli and carrot with leaves it is 4% (Thompson et al., 1998). Also, fresh  
166 produce response to transpiration such as biochemical, microbiological, and physiological  
167 changes contribute to quality degradation. These responses are usually temperature dependent  
168 and affect transpiration of FFV and low RH can raise transpiration damage leading to  
169 dehydration, increased respiratory intensity, and loss of product quality (Castellanos &

170 Herrera, 2015). Therefore, optimum temperature and RH should be maintained for each  
171 product in order to extend shelf-life and maintain products quality.

172

## 173 **2.2. Transpiration measurement**

174 Water loss from FFV, also known as moisture loss or transpiration phenomena, is often  
175 expressed as the percentage change in mass of the original or initial product mass. The  
176 quantity of water loss over a given period of time is considered as the water loss rate, also  
177 referred to as rate of moisture loss or transpiration rate (TR) (Maguire, Banks, & Opara,  
178 2001). Calculation of the TR based on moisture loss per unit time is the most used and  
179 reported method to describe transpiration phenomenon in fresh horticultural produce (Caleb et  
180 al., 2013; Castellanos & Herrera, 2015; Mahajan, Oliveira, & Macedo, 2008a; Shirazi &  
181 Cameron, 1993; Sousa-Gallagher et al., 2013).

182 However, there are two main possible approaches to calculate TR of fresh produce. The first  
183 approach is by gravimetric measurement of change in product mass over time. The second  
184 approach is based on theoretical determination of TR, via the Fick's law of diffusion. It is  
185 worth mentioning that the gravimetric measurement of TR is used by many authors to find  
186 other parameters, such as the transpiration coefficient and/or tissue and boundary layer  
187 resistance that better describes the transpiration phenomenon (Linke, 1997; Sastry &  
188 Buffington, 1983; Thompson et al., 1998).

189

### 190 **2.2.1. Gravimetric approach**

191 The most commonly reported method for measuring TR is by the gravimetric approach, also  
192 known as the mass loss approach, which involves periodically weighing the produce at a  
193 given temperature and RH. TR can be directly calculated per unit surface area ( $TR_s$ ) (Eq. 1)  
194 and/or per unit of initial mass ( $TR_m$ ) (Eq. 2) of the produce:

$$195 \quad TR_s = \frac{M_i - M_t}{t \cdot A_s} \quad (1)$$

$$196 \quad TR_m = \frac{M_i - M_t}{t \cdot M_i} \quad (2)$$

197 where  $M_i$  is the initial mass of the product;  $M_t$  is product mass at a determined time (t); and  $A_s$   
198 is the initial surface area of the product. Usually  $TR_s$  is commonly expressed in  $\text{mg cm}^{-2} \text{h}^{-1}$  or  
199  $\text{mg cm}^{-2} \text{s}^{-1}$  and  $TR_m$  in  $\text{g kg}^{-1} \text{h}^{-1}$ ,  $\text{mg kg}^{-1} \text{h}^{-1}$  or  $\text{mg kg}^{-1} \text{s}^{-1}$ .

200 Different experimental methods have been reported for the measurement of TR by the mass  
201 loss approach (Fig. 1). In some setups, the balance was located outside the experimental  
202 container, which limits continuous measurement of product mass loss. In these cases the

203 product has to be taken out of the container to be measured and opening of the container can  
 204 result in disturbance of internal atmosphere and RH if it is not carried out with caution  
 205 (Xanthopoulos et al., 2014). In the experiment conducted by Kang and Lee (1998), the  
 206 chamber was equipped with gas control to maintain the desired oxygen (O<sub>2</sub>) and carbon  
 207 dioxide (CO<sub>2</sub>) concentration in order to incorporate the effect of modified atmosphere as one  
 208 of the parameters of TR for apples and minimally processed cut vegetables. A novel setup  
 209 was considered by Mahajan et al. (2016) in their study. The authors included an additional  
 210 infrared temperature sensor to monitor the products' surface temperature and a sensor for the  
 211 surrounding environmental conditions.

212

### 213 **2.2.2. Theoretical approach**

214 It is well established that transpiration can be visualised as the interaction between a driving  
 215 force for mass loss and resistance (Becker & Fricke, 1996, 2001; Leonardi, Baille, &  
 216 Guichard, 2000; Sastry, 1985; Sastry & Buffington, 1983). This interaction is expressed  
 217 mathematically as:

$$218 \quad TR_m = k_t \cdot (P_s - P_\infty) \quad (3)$$

219 where  $TR_m$  is transpiration rate, mass basis (mg kg<sup>-1</sup>s<sup>-1</sup>);  $k_t$  is transpiration coefficient assumed  
 220 constant for a specific product (mg kg<sup>-1</sup> s<sup>-1</sup> MPa<sup>-1</sup>);  $P_s$  is water vapour pressure at the  
 221 evaporating surface of the product (MPa); and  $P_\infty$  is ambient water vapour pressure (MPa). In  
 222 this mathematical equation the driving force for transpiration is represented by ( $P_s - P_\infty$ ),  
 223 which is also known as the water vapour pressure deficit (VPD), and the resistance  
 224 represented by the inverse of the transpiration coefficient ( $k_t$ ). The  $k_t$  can be divided into two  
 225 terms, as follows:

$$226 \quad \frac{1}{k_t} = \frac{1}{k_s} + \frac{1}{k_a} \quad (4)$$

227 where  $k_s$  is skin mass transfer (transpiration) coefficient (mg kg<sup>-1</sup> s<sup>-1</sup> MPa<sup>-1</sup>) and  $k_a$  is air film  
 228 mass transfer (mg kg<sup>-1</sup> s<sup>-1</sup> MPa<sup>-1</sup>), also known as convective mass transfer coefficient or  
 229 external mass transfer coefficient. Combining Eq. 3 with Eq. 4 yields:

$$230 \quad TR_m = \frac{P_s - P_\infty}{\frac{1}{k_s} + \frac{1}{k_a}} \quad (5)$$

231 What differ among authors in using Eq. 5, are the factors and assumptions that are considered  
 232 important or negligible in order to calculate  $k_s$  and  $k_a$ . In Sastry and Buffington (1983), these  
 233 coefficients were represented by  $k_s = \frac{\tau}{\delta\varphi}$  and  $k_a = \frac{1}{h_d}$ , where  $\delta$  is the diffusion coefficient  
 234 of water vapour in air;  $\tau$  the product skin thickness;  $\varphi$  is fraction of product surface covered by



235 pores; and  $h_d$  is convective mass transfer coefficient. In contrast, Fockens and Meffert (1972)  
236 expressed skin mass transfer coefficient as  $k_s = \frac{\xi_1 \beta}{R_D T}$  and air film mass transfer as  $k_a =$   
237  $\frac{\xi_2}{1/\beta + \mu s/\delta}$ , where  $\xi_1$  is a fraction of surface behaving as a free water zone (non-dimensional);  $\beta$   
238 is a convective mass transfer coefficient ( $\text{m s}^{-1}$ );  $R_D$  is a universal gas constant ( $\text{J kg}^{-1}\text{°C}^{-1}$ );  $T$   
239 is the ambient temperature ( $\text{°C}$ );  $\xi_2$  fraction of surface behaving as porous membrane (non-  
240 dimensional);  $\mu$  is resistance factor (non-dimensional);  $s$  is skin thickness (m); and  $\delta$  is  
241 diffusion coefficient of water vapour in the air ( $\text{m}^2 \text{s}^{-1}$ ).

242 Different ranges of transpiration coefficients are shown in Table 1. Limitations of using  
243 transpiration coefficients are that they are restricted to certain range of experimental  
244 conditions; and often product specific. For example, there is a significant difference in  
245 transpiration coefficient of carrot ranging from 106 to 3250  $\text{mg kg}^{-1} \text{s}^{-1} \text{MPa}^{-1}$ , based on  
246 various assumptions adopted in the calculation (Linke & Geyer, 2001). Also, different  
247 experimental methods are used for determining the transpiration coefficient, which results in  
248 different values even for the same product (Sastry & Buffington, 1983). However, Eq. 3 is a  
249 simple mathematical equation that can be used to predict the TR of a specific product. In  
250 order to use this equation details on transpiration coefficient of the specific product and the  
251 calculated water pressure difference between the FFV and surrounding environment are  
252 required. To determine the ambient water vapour pressure, psychrometric charts, which relate  
253 temperature, RH and water vapour pressure can be used.

254 A similar approach to determine the TR of FFV is by the use a known tissue and boundary  
255 layer resistance. Figure 2 presents the dynamics of water loss rate during the postharvest  
256 storage of FFV in this approach at constant heat and mass transfer conditions, and under pre-  
257 defined experimental conditions. The first section is characterised by the atmospheric  
258 evaporation of free surface water from the product. In this case the intensity of transpiration is  
259 solely dependent on the boundary layer resistance. However, when free water is no longer on  
260 the surface, water is transported from inside the produce to the surface, but with an additional  
261 resistance due to internal membranes, called tissue resistance. This additional resistance is  
262 evident by the decrease in the slope of water loss rate over time as shown in the second  
263 section. At this point, the water potential of the produce is also reduced, as shown in the third  
264 section (Linke, 1997). The reduction in water potential is important because the flow of liquid  
265 and/or gaseous water out of a produce, tissue or plant cell, as well as the rate of water  
266 movement directly depends on the water potential gradient between the produce, tissue, or  
267 plant cell and the surroundings (Gomez Galindo, Herppich, Gekas, & Sjöholm, 2004: Nobel,

268 2009). Water potential can be defined as the free energy of water within the respective  
269 system, such as produce, tissue, plant cell, or solution compared to that of pure water (Rodov  
270 et al., 2010). Thus, water potential is indicative of the true water deficit of a system  
271 (Herppich, Mempel, & Geyer, 1999). In addition, in plant physiology, water potential is  
272 generally accepted as the best parameter to describe actual tissue water status (Herppich,  
273 Mempel, & Geyer, 2001).

274 In this approach the resistances in the water vapour pathway can be determined by using a  
275 modified Fick's law in terms of resistances, as shown in Eqs. 6 and 7, while taking into  
276 consideration the conditions presented in section 1 and 2 (Fig. 2).

$$277 \quad TR_s = \frac{x_p - x_A}{r_B + r_T} \quad (6)$$

278 where  $TR_s$  is transpiration rate, area basis ( $\text{mg cm}^{-2} \text{s}^{-1}$ );  $x_p$  is volume related water content of  
279 air in the intercellular spaces in the centre of the produce ( $\text{mg cm}^{-3}$ );  $x_A$  is volume related  
280 water content of the air unaffected by the produce ( $\text{mg cm}^{-3}$ );  $r_B$  is boundary layer resistance  
281 in the water vapour pathway ( $\text{s cm}^{-1}$ ); and  $r_T$  is tissue resistance in the water vapour pathway  
282 ( $\text{s cm}^{-1}$ ), which includes tissue and skin of the fruit or vegetable. However, the tissue  
283 resistance approach becomes negligible when produce surface is wet and therefore the  
284 following equation is valid:

$$285 \quad TR_s = \frac{x_{ps} - x_A}{r_B} \quad (7)$$

286 where  $x_{ps}$  is the water content of the air at the produce surface,  $\text{mg cm}^{-3}$  (Fig. 3). Tissue  
287 resistance is determined by the nature of the plant tissue, which is exclusively dependent on  
288 the internal properties of the product, such as the water activity and sugar. Other factors  
289 influencing tissue resistance of horticultural produce include pre-harvest conditions and  
290 postharvest handling practices (Linke, 1997).

291 On the other hand, the boundary layer resistance is determined by the form of FFV epidermal  
292 layer. It is dependent on external parameters such as shape, dimensions, and surface structure  
293 of the product, as well as environmental conditions such as air flow conditions and surface  
294 temperature of the produce. For the determination of the boundary layer resistance the water  
295 loss rate has to be measured under natural convection. Once boundary layer resistance is  
296 known, tissue resistance can be determined by Eq. 6, as long as the centre of the produce is  
297 water saturated. In Table 2 it is possible to visualise different tissue resistance found by Linke  
298 and Geyer (2000). The boundary layer resistance for single produce items at unrestricted  
299 natural convection and room temperatures was in the range between 1 and 4  $\text{s cm}^{-1}$  for small

300 and bigger FFV, respectively. Both theoretical approaches for estimating TR, via transpiration  
301 coefficient or tissue resistance, have specific limitations due to the different values found in  
302 the literature. However, they are very useful tools to calculate the TR of FFV since no  
303 experimental data is required.

304

## 305 **2.3. Factors affecting transpiration**

### 306 **2.3.1. Intrinsic factors**

307 Fresh produce shape and size, expressed as surface area-to-volume or surface area-to-mass  
308 ratios, are major factors affecting the  $TR_m$ , especially the boundary layer resistance. Products  
309 with large surface area to mass ratios provide a considerable contact area with surrounding  
310 atmosphere. For example, horticultural products, such as leafy green vegetables and  
311 cauliflowers have higher  $TR_m$ , when compared to spherical produce such as oranges and  
312 tomatoes with lower surface area (Sastry, 1985). Similarly, morphological and anatomical  
313 characteristics of the FFV also have significant effect on TR, specifically on the tissue  
314 resistance. Surface structure for each FFV is unique and those which contain skin and/or a  
315 waxy coating such as apple, provide extra layers of resistance and therefore the water loss rate  
316 in this product is lower than for products without these structures such as mushroom (Sastry,  
317 1985). The skin of FFV acts as a barrier to diffusion of water vapour (Maguire et al., 2001).

318 Purity level of water content in FFV can also affect the TR of the product. Water content in  
319 most FFV contains dissolved/soluble solids (i.e. total soluble solids). Literature has  
320 extensively shown that total soluble solids of FFV significantly differs (Beckles, 2012;  
321 Mahmood, Anwar, Abbas, Boyce, & Saari, 2012). Thus, vapour pressure at the evaporating  
322 surface is determined by Raoult's law and is a little lower than the saturation water vapour  
323 pressure at the same temperature (Sastry, 1985). This effect is also known as the vapour  
324 pressure lowering effect since it causes a reduction in VPD and directly affects the TR.

325 Additionally, physiological condition, such as the maturity stage in fresh produce after harvest  
326 has been shown to significantly influence on TR. In general, immature and over mature fruit  
327 transpires more rapidly than optimally mature fruit due to the permeability of the skin to water  
328 vapour (Mishra & Gamage, 2007; Sastry, 1985). The developmental stages of the fruit  
329 therefore directly affect the tissue resistance of the product. However, factors are often  
330 eliminated as a variable on mathematical models of transpiration due to lack of a reliable  
331 quantitative maturity index (Sastry & Buffington, 1983).

332

### 333 **2.3.2. Extrinsic factors**

334 Impacts of factors such as temperature and RH on TR of fresh horticultural produce have been  
335 extensively investigated over the last decade. Mahajan et al. (2008a) found that by increasing  
336 the RH in the storage containers for whole mushrooms from 76% to 96%, TR decreased by  
337 87% at 4 °C, whereas decreasing the temperature from 16 °C to 4 °C decreased the TR by  
338 61% at 96% RH. Caleb et al. (2013) also showed that by increasing RH inside storage  
339 containers for pomegranate arils from 76% to 96%, decreased TR by 83.5% at 5 °C, while  
340 decreasing the temperature from 15 °C to 5 °C, TR decreased by 68.9%. Xanthopoulos et al.  
341 (2014) reported that the TR for grape tomatoes increased with temperature from 15 °C to 20  
342 °C, while it decreased for RH 80% to 92%. These studies showed that humidity is the variable  
343 with the greatest effect on TR, and the magnitude of TR decrease is product dependent.  
344 Aguirre, Frias, Barry-Ryan, and Grogan (2009) expressed the visual quality of mushroom  
345 stored under different temperatures and humidity using VPD instead of the RH to avoid the  
346 interaction between temperature and RH. Although VPD is a conventional variable for  
347 refrigeration technology, package designers and food technologists usually employ the RH.

348 Airflow around fresh produce and/or through the packaged product, also have a significant  
349 influence on TR. Baltaci, Linke, and Geyer (2010) measured the water loss rate of artificial  
350 fruits (water filled evaporating spheres) inside a plastic box in three layers under natural  
351 convection and forced airflow (0.8 m s<sup>-1</sup>). The authors showed that differences in TR were  
352 dependent on the produce position inside and airflow. They also found that TR was higher  
353 under forced airflow than under natural convective conditions. Air movement around the  
354 product prevents the development of a microenvironment with high-humidity build-up  
355 (Sastry, 1985), and this decreases the resistance of the air films to mass transfer.

356 Physical conditions and surface injuries such as cuts, bruises and scratches on the skin surface  
357 of FFV, tend to increase the TR, as they reduce the tissue resistance due to modification of the  
358 skin (Holcroft, 2015; Maguire et al., 2001). FFV have 2 to 3 times higher TR after harvest  
359 when compared to the steady state values due to the physical injuries caused by detachment  
360 from the mother plant (Sastry et al., 1977). However, during the storage period once the  
361 injuries are healed TR reduces to a lower and relatively steady value (Sastry, 1985).

362 Also, heat removed from the evaporating surface during transpiration causes a lowered  
363 surface temperature and therefore a decreased vapour pressure at the surface, reducing  
364 transpiration (Becker & Fricke, 1996). This effect, also known as evaporative cooling, is more  
365 noticeable at high water vapour pressure differences. In this situation evaporation has a

366 considerable effect on the driving force and consequently on transpiration (Sastry, 1985).  
367 However, respiration increases the product's surface temperature because of heat generation  
368 and this increases water vapour pressure at the surface, increasing transpiration (Becker &  
369 Fricke, 1996). This effect, also referred to as respiratory heat generation, is usually low for  
370 moderate water vapour pressure but can grow into a dominant factor at RH close to saturation.  
371 The respiration phenomena produces an additional mass loss due to carbon loss but it is  
372 considered negligible (Sastry, 1985).

373

### 374 **3. Moisture evolution in packaged fresh horticultural produce**

375 Packaging of FFV leads to accumulation of moisture in the headspace as it acts as an  
376 additional barrier for moisture transfer. The main source of this moisture is the product itself,  
377 however, temperature fluctuations along the supply chain also plays an important role for  
378 moisture evolution and condensation (Powers & Calvo, 2003). Factors affecting moisture  
379 transfer and RH in packaged fresh produce are water vapour permeability of the packaging  
380 films, transpiration and respiration of product, and storage conditions (Lu, Tang, & Lu, 2013).  
381 Therefore, selection of appropriate packaging materials is one of the essential steps for  
382 achieving optimum humidity conditions in packaged fresh produce.

383 The optimum humidity levels vary in each product, yet in order to reach the maximal  
384 postharvest life span it should be taken into account (Ben-Yehoshua & Rodov, 2002). For  
385 most FFV the storage conditions should be within 85% and 98% RH. Nonetheless, for  
386 products such as garlic and onion storage at RH higher than 70 to 75% at optimum  
387 temperatures results in excessive water absorption leading to rooting, mould development and  
388 sprouting (Rodov, Ben-Yehoshua, Aharoni, & Cohen, 2010). In the review by Paull (1999)  
389 the possible effects of temperature and RH on fresh commodity quality was extensively  
390 discussed. The author also provided a detailed summary of optimum RH and temperature as  
391 well as shelf life for a wide range of FFV.

392 Current modified atmosphere packaging (MAP) designs consider the respiration rate of  
393 products as the only important parameter when selecting target gas barrier properties.  
394 However, besides in-package gas composition, it is also essential to take into consideration  
395 the in-package humidity level. In order to avoid moisture condensation and accelerated  
396 growth of spoilage microorganisms (Caleb et al., 2013; Mahajan et al., 2014; Song, Lee, &  
397 Yam, 2001). The in-package humidity is determined by transpiration and respiration of the  
398 fresh produce and water vapour permeability of the packaging material. Most polymeric

399 materials (polyethylene, polypropylene or polyvinyl chloride) used in MAP have lower water  
400 vapour permeability relative to the TR of fresh produce (Rux et al., 2016; Song et al., 2001).  
401 This leads to further development of MAP into a modified atmosphere and humidity package  
402 (MAHP) system, since evaporated water molecules from the produce are not effectively  
403 transmitted across the packaging film and prevail within the package. Hence, the challenge of  
404 designing an effective MAHP system is finding a solution to design optimal atmosphere and  
405 lessen the risk of in-package moisture condensation while still keeping produce mass loss as  
406 low as possible.

407

### 408 **3.1. Moisture condensation dynamics**

409 Condensation is the process in which water vapour turns into liquid form as a result of  
410 temperature differences (Joyce & Patterson, 1994). The temperature at which this process  
411 occurs is known as the dew point temperature (Holcroft, 2015). Condensate will be formed on  
412 any product that is at or below the dew point temperature of the surrounding air. For every  
413 temperature and RH combination at constant pressure, there is a specific and measurable dew  
414 point temperature and in order for condensation to appear the temperature has to fall only by a  
415 fraction of a degree (Joyce & Patterson, 1994). Therefore, dew point measurement is a very  
416 useful parameter to anticipate moisture condensation and develop control measures. It can be  
417 measured directly by means of special sensors or calculated from temperature and humidity  
418 following the known laws of psychrometry. Condensation inside packaged fresh produce  
419 occurs when water molecules evaporated from the product surface do not transmit through the  
420 packaging film and stay within the package (Fig. 4). Horticultural produce specific shape,  
421 dimension and surface structure, as well as environmental parameters such as storage  
422 temperature, RH, and air flow conditions around the produce have a direct impact on the  
423 intensity of condensation process (Rodov et al., 2010).

424 Condensation inside packages of FFV represents a threat to the product quality and safety. It  
425 is almost inevitable to avoid moisture condensation in the entire postharvest supply chain due  
426 to temperature fluctuations. However, there are some recommendations that can be taken into  
427 account in order to minimise the condensation this include: i) storage of the product under  
428 strict temperature control; ii) maintenance of a continuous cold chain; iii) perform packaging  
429 operation under cold condition; iv) temperature conditioning of the packaging material; v)  
430 cool the product to above dew point temperature until they are packed and then cool it to the  
431 desired storage temperature; and, v) faster warming of cold fruit in order to reduce the time  
432 that the produce is wet (Holcroft, 2015).

433 Gottschalk, Linke, Mészáros, and Farkas (2007) developed a model that predicts the  
434 condensation and transpiration process on a single fruit under varying ambient conditions  
435 along storage time. The model was validated using eight fruits in an open container. Linke  
436 and Geyer (2013) determined the condensation dynamics and intensity within plastic film  
437 packaging for fruit under fluctuating external temperatures. Using packages of plums as a test  
438 case, the authors showed that moisture condensation process occurred with time-delayed and  
439 superimposed varying intensities on the surface of the fruit, inner film surface, and inner tray  
440 walls (Fig. 5). Moisture condensation in the inner film surface was mainly influenced by flow  
441 conditions, external temperature amplitude, and in the inner air volume. On the contrary,  
442 moisture condensation on fruit surface was caused primarily by temperature amplitude and  
443 cycle time. In summary, for the studied cycle time of 240 min, the condensate remained for  
444 53%, 51% and 42% of the cycle time on the inner wall of the tray, plum surface and  
445 underneath film, respectively. Further detailed investigations are needed to evaluate and  
446 simulate moisture condensate formation via integrative mathematical modelling. Such model  
447 can be developed using water vapour related characteristics of packaging materials (water  
448 vapour permeability, macro and micro perforations), and physiological characteristics of  
449 product (respiration and transpiration) as well as external storage environment (temperature,  
450 humidity and air flow).

451

## 452 **3.2. Moisture condensation control strategies**

### 453 **3.2.1 Moisture absorbers**

454 This involves the use of various hygroscopic substrates or substances to attract and hold water  
455 molecules from the surrounding environment. Desiccant and papers pads are used to wrap  
456 fresh produce in order to mitigate moisture accumulation (Ozdemir & Floros, 2004). The use  
457 of these salts and polyols packages offers an alternative way to avoid moisture condensation  
458 inside the package. It has been shown to have beneficial effect on the shelf life of FFV by  
459 reducing microbial growth and preserving colour attributes. Mahajan, Rodrigues, Motel, and  
460 Leonhard (2008b) also developed a moisture absorber. Fast absorbing moisture absorbers  
461 such as calcium chloride ( $\text{CaCl}_2$ ), potassium chloride (KCl) and sorbitol were mixed with a  
462 slow absorbing desiccant such as bentonite in different proportions. Overall results showed  
463 that the appearance of mushrooms improved when 5 g of mixed desiccant was packed in 250  
464 g of mushroom punnet compared to those packed without desiccant.

465 Similarly, Azevedo, Cunha, Mahajan, and Fonseca (2011) designed desiccants with calcium  
466 oxide (CaO), sorbitol, and  $\text{CaCl}_2$  in a range of 0.2 - 0.6 g of desiccant mass in varying

467 proportions. The change in moisture content of each of the mixed desiccants was measured at  
468 regular intervals up to 5 d at 10 °C. Results showed that optimised desiccant mixture, which  
469 contained 0.5, 0.26 and 0.24 g g<sup>-1</sup> of CaO, CaCl<sub>2</sub> and sorbitol, respectively, and had a moisture  
470 holding capacity of 0.813 g water g<sup>-1</sup>. Additionally, absorption of excess moisture from the  
471 headspace, keeps RH inside the package low (Shirazi & Cameron, 1992). Also, the use of  
472 desiccants for FFV with high water activity might lead to excessive moisture loss. Hence,  
473 careful application of desiccants based on detailed research is needed.

474

### 475 **3.2.2 Perforated films**

476 Micro-perforated packaging films are commonly used in fresh produce packaging to enhance  
477 O<sub>2</sub> and CO<sub>2</sub> gas permeability and control moisture around FFV. Such packaging films have  
478 the advantage to avoid in-package anaerobiosis and therefore may extend the shelf-life and  
479 maintain quality of FFV (Jo et al., 2013; Hussein, 2015). Almenar et al. (2007) studied the  
480 behaviour of strawberries packaged with two continuous and three micro-perforated films  
481 (with different gas permeability) with the purpose of obtaining equilibrium atmospheres of  
482 diverse compositions. Results showed that micro-perforated films with one and three holes  
483 provided adequate CO<sub>2</sub> and O<sub>2</sub> equilibrium concentrations. However, micro-perforated films  
484 do not allow for effective diffusion of water vapour into the environment leading to saturated  
485 humidity, moisture condensation and deterioration of fresh packaged horticultural produce  
486 (Rodov et al., 2010).

487 Perforations in a polymeric film is based on a compromise principle since perforations affect  
488 the film's permeability to O<sub>2</sub> and CO<sub>2</sub> to a higher extend than to water vapour. With macro-  
489 perforated packaging films, it is nearly impossible to achieve MA equilibrium, and prevent  
490 excessive mass loss and shrivelling of FFV. In ideal packaging, the humidity level should be  
491 low enough to prevent moisture condensation but sufficiently high enough to reduce product  
492 mass loss, while also having an optimal atmosphere (Rodov et al., 2010).

493

### 494 **3.2.3 Individual shrink-wrapping**

495 Individual shrink wrapping (ISW) is a passive form of MAP in which a polymer film with  
496 selective permeability to CO<sub>2</sub>, O<sub>2</sub>, ethylene and water is used to pack individual fresh produce  
497 in order to maintain its freshness (Dhall, Sharma, & Mahajan, 2012; Megías et al., 2015). The  
498 main advantages of this technology are reduced mass loss, minimised fruit deformation,  
499 reduced chilling injuries and decay (Dhall et al., 2012). Rodov et al. (2010) reported that  
500 shrink wrapping is also efficient in controlling moisture condensation due to a very small



501 headspace volume and negligible temperature differences between the product and the film  
502 surface.

503 Rao, Rao, and Krishnamurthy (2000) studied the effect of MAP and shrink wrapping on the  
504 shelf life of cucumber. Results showed that shrink wrapping with polyethylene film can  
505 extend the shelf life of cucumber for up to 24 d at 10°C. Megías et al. (2015) studied the  
506 effect of ISW on the postharvest performance of refrigerated fruit from two zucchini cultivars  
507 that differ in their sensitivity to cold storage. Results indicated that ISW zucchini packaging  
508 led to improved tolerance to chilling simultaneously with a decrease in oxidative stress,  
509 respiration rate and ethylene production. Despite the positive results, this approach is limited  
510 to spherical or cylindrical products (e.g. cucumber) because if any part of the product is not in  
511 contact with the film then it will lead to moisture accumulation (Rodov et al., 2010).

512

### 513 **3.2.4 Enhanced water vapour permeable films**

514 Various polymers have been developed with relatively high permeability towards water  
515 vapour compared to the commonly used polymeric films such as polypropylene or  
516 polyethylene. These include co-extruded and bio-degradable polymeric films with enhanced  
517 water vapour permeability. Co-extruded films consist of blends of different hydrophilic  
518 polyamides with other polymeric and non-polymeric compounds. The different blends allow  
519 manufacturing materials varying in water vapour permeability, in accordance with required  
520 in-package RH levels (Rodov et al., 2010).

521 As an example, Aharoni et al. (2008) used a co-extruded packaging film Xtend® (StePac,  
522 Tefen, Israel) and reported that Xtend® can effectively modify both atmospheric composition  
523 and RH inside packaging containing various FFV. Similarly, cellulose-based NatureFlex™  
524 (Innovia films, Cumbria, UK) polymeric films also held a good potential for application in  
525 packaging of fresh produce as it has a very high water permeability ( $200 \text{ g m}^{-2} \text{ d}^{-1}$  at 25 °C  
526 and 75% RH) as against the conventional polypropylene film with  $0.8 \text{ g m}^{-2} \text{ d}^{-1}$  water  
527 permeability (Sousa-Gallagher et al., 2013). Also, water vapour transmission rate (WVTR) of  
528 cellulose based NatureFlex™ polymeric films has been shown to increase with the increase  
529 RH. Therefore, care must be taken in designing fresh produce packages, as excessively high  
530 water permeability can lead to higher product moisture and mass loss.

531

### 532 **3.2.5 Humidity-regulating trays**

533 Singh, Saengerlaub, Stramm, and Langowski (2010) reported on the application of humidity-  
 534 regulating trays incorporated with varying concentrations of sodium chloride (NaCl) for fresh  
 535 mushrooms. In this study, different percentages of NaCl were introduced into the polymer  
 536 matrix of the film from which trays were produced. The authors found that the amount of  
 537 water vapour absorbed by the tray is directly proportional to the percentage of salt  
 538 incorporated in the trays. Rux et al. (2015) also reported the use of humidity-regulating trays  
 539 for mushrooms. Trays were produced with NaCl (18% on a weight basis) between the outer  
 540 barrier layer (polypropylene) and the inner sealing layer (polypropylene/ ethylene vinyl  
 541 alcohol/ polyethylene). Results showed that humidity-regulating tray maintained a stable RH  
 542 (93%) inside the package and it absorbed 4.1 g of water vapour within 6 d at 7 °C and 85%  
 543 RH storage condition. Yet the absorbed water vapour was not enough to prevent water  
 544 condensation in the package headspace.

545 Furthermore, Rux et al. (2016) optimised the humidity-regulating tray from a thermoformed  
 546 multilayer structure: polyethylene (outside)/foamed hygroscopic ionomer (active layer) with 0  
 547 (T-0 tray) or 12 (T-12 tray) wt.-% NaCl/hygroscopic ionomer (sealing layer, inside). The  
 548 amount of water absorbed was 7.6 and 13.2 g by T-0 and T-12 trays respectively, which  
 549 indicates that the moisture absorbed by the tray was directly proportional to the amount of salt  
 550 incorporated into the tray matrix. The addition of salt into polymer matrix of packaging tray  
 551 represents a novel approach to control in-package humidity for fresh produce. However,  
 552 further optimisation via mathematical modelling is required for product specific needs.

553

#### 554 **4. Application of integrative mathematical modelling concept**

555 A packaging system for FFV consists of a respiring produce fully enclosed in a tray type  
 556 package lidded with permeable film. Changes in the amount of water vapour content inside  
 557 the package will be dependent on transpirational water loss from the product, water vapour  
 558 transmitted through the packaging film and the water vapour absorbed by the active moisture  
 559 control system. As a result the following unsteady-state mass balance equation may be used to  
 560 describe the rate of change of water vapour in the headspace as a function of time:

$$561 \quad \left\{ \begin{array}{c} \text{Water vapour evolution} \\ \text{in a package} \end{array} \right\} = \left\{ \begin{array}{c} \text{Transpirational water loss} \\ \text{from the product} \end{array} \right\} - \left\{ \begin{array}{c} \text{Water vapour transfer} \\ \text{through packaging film} \end{array} \right\} - \left\{ \begin{array}{c} \text{Water vapour absorbed by the active} \\ \text{moisture control system} \end{array} \right\} \quad (8)$$

562 There is a wealth of published information on modelling of moisture evolution in fresh  
 563 produce (Lu et al., 2013; Mahajan et al., 2016; Rennie & Tavoularis, 2009; Song et al., 2001),  
 564 yet no systematic study has been conducted to bring all the theoretical models together in a  
 565 ready to use format. Hence, the sub-sections below present an overview of published models

566 related to product transpiration, water vapour permeation in perforated packaging system and  
567 active moisture control systems.

568

#### 569 **4.1. Moisture evolution due to transpiration**

570 There are two approaches commonly used for the mathematical modelling of the transpiration  
571 phenomena. The first is based on the diffusion equations of Fick's law (Leonardi et al., 2000;  
572 Maguire et al., 2001), and the second approach is based on heat and mass balances (Kang &  
573 Lee, 1998; Lu et al., 2013; Song, Vorsa, & Yam, 2002). The model presented by Sastry  
574 (1985) is the most basic form of a transpiration model:  $TR = k_i (P_s - P_\infty)$ . This model was  
575 applied primarily to storage situations where steady state conditions prevailed and the key  
576 assumption was that temperature of product evaporating surface is the same as its surrounding  
577 environment. However, an error is observed in the model at saturated environments (i.e. VPD  
578 = 0.0) as discussed previously. Therefore, a more complex diffusion model is required to  
579 predict transpiration under saturated and stagnant air flow conditions as observed inside  
580 packaged fresh produce.

581 Non-linear models for estimating TR based on Fick's first law of diffusion have been reported  
582 in the literature, but very little work has been developed in this area, especially for the  
583 prediction of TR under MAP systems. There are at least two major reasons why the  
584 mathematical modelling of TR for MAP systems are not well developed this includes: i)  
585 modelling of this phenomena needs a complete understanding of the dynamic interactions  
586 between permeation through the packaging film and evaporation on produce surface as a  
587 result of the heat released from respiration; and, ii) existing models are limited to cooling  
588 process and bulk storage, which may not be suitable for MAP systems (Song et al., 2002).

589 It is noteworthy to mention that the difference between a  $TR_m$  and  $TR_s$  model is the unit of the  
590  $k_s$  coefficient. Some authors prefer to use it in terms of mass basis (Caleb et al., 2013; Sousa-  
591 Gallagher et al., 2013) since it is easier to determine the mass of product than its surface area,  
592 this makes it a more convenient unit (Sastry, 1985). Other authors emphasised on the  
593 significance of expressing transpiration per unit area (Linke, 1997; Xanthopoulos et al.,  
594 2014), because the area-based transpiration coefficient is not dependent on product mass. An  
595 alternative is the use of an area-based transpiration coefficient combined with a statistically  
596 determined correlation between surface area and mass for a specific FFV. This approach  
597 combines the accuracy of the area based coefficient with the convenience of a quick  
598 calculation of the product surface area from the mass.

599 Other approach for modelling TR is based on heat and mass balance between the produce and  
600 storage atmosphere and is also shown in Table 3. Kang and Lee (1998) developed a  
601 transpiration model to predict moisture loss of fresh produce under ambient and controlled  
602 atmosphere conditions. In this model the sum of heat energies transferred through natural  
603 convection from surrounding air and generated from respiration inside the produce was  
604 assumed to be supplied for evaporating moisture on produce surface. Song et al. (2002)  
605 proposed a respiration-transpiration model by applying simultaneous heat and mass transfer  
606 principles to known physiological behaviour of fresh produce in MAP. Their model applied  
607 the assumption that temperature inside the package was equal to the temperature on the  
608 surface of the produce and therefore external heat was negligible. Lu et al. (2013) developed a  
609 model for transpiration based on mass change of water vapour. Their model considered;  
610 respiratory heat generated by produce, heat absorbed by produce, heat absorbed by gas around  
611 the produce, heat absorbed by the package and heat change caused by gas transmission across  
612 the package.

613 Mathematical models for transpiration, which takes into consideration the various factors  
614 affecting TR, are important tools. They help select targeted package designs with optimum  
615 WVTR and help estimate fresh produce shelf life (Kang & Lee, 1998). Models that do not  
616 take into account all of the factors can in some cases be satisfactory, but may result in large  
617 errors in other cases (Sastry, 1985). However, models that take into account too many factors  
618 become complex with limited application flexibility, since some of the parameters may be  
619 product specific or not easily measurable. For instance skin thickness, pore fraction in the  
620 skin, geometry, thermal diffusivity, and surface cellular structure are factors not easily  
621 measured and/or determined (Kang & Lee, 1998). Therefore, an extremely detailed model  
622 might not be as useful and convenient as a well-designed simple model (Tanner, Cleland,  
623 Opara, & Robertson, 2002). Thus, the development of a successful and accurate mathematical  
624 model for transpiration depends on the parameters considered and the assumptions made. In  
625 addition, respiration plays an important role on the transpiration phenomena for packaged  
626 produce and it is important to take this into account when developing a TR model. Both  
627 Fick's law and heat and mass transfer approach can incorporate this parameter.

628

#### 629 **4.2. Water vapour permeation in perforated packaging systems**

630 Mathematical modelling of mass transfer through perforated packaging is commonly used and  
631 has been extensively reported in the literature. A detailed review on perforation mediated  
632 packaging systems was recently published by Hussein, Caleb, and Opara (2015). An example

633 of the application of mathematical modelling for perforated packaging system can be found in  
 634 the study reported by Fishman, Rodov, and Ben-Yehoshua (1996). The authors developed a  
 635 mathematical model to study the influence of film perforations on water vapour flux through  
 636 the perforated film (Eq. 9):

$$637 \quad F_w = \alpha (H_A - H) \left[ \frac{SP_w}{L} + \frac{\pi R_h^2 N D_w}{L + R_h} \right] \quad (9)$$

638 where  $F_w$  is the water flux ( $\text{m}^3 \text{h}^{-1}$ );  $\alpha$  is water vapour concentration under saturation vapour  
 639 pressure which depends on temperature (non-dimensional);  $H_A$  is RH in the ambient  
 640 atmosphere (non-dimensional);  $H$  is RH (non-dimensional);  $S$  is film area ( $\text{m}^2$ );  $P_w$  is water  
 641 vapour permeability coefficient of the film found from film specifications ( $\text{m}^2 \text{h}^{-1}$ );  $L$  is film  
 642 thickness (m);  $\pi$  is 3.14 (non-dimensional);  $R_h$  is radius of perforation (m);  $N$  is number of  
 643 pores (non-dimensional); and  $D_w$  is the diffusion coefficient of water vapour in air ( $\text{m}^2 \text{h}^{-1}$ ).

644 The overall model showed that perforation had more effects on  $\text{O}_2$  concentration than  
 645 on RH. Although this model was designed for mango fruit; the proposed equations could still  
 646 be valid for other commodities if appropriate transpiration coefficients are inserted. Ben-  
 647 Yehoshua, Rodov, Fishman, and Peretz (1998) applied the model developed by Fishman et al.  
 648 (1996) and evaluated the effects of perforation on MAP with bell peppers and mangoes. The  
 649 results showed that perforating the film affects  $\text{O}_2$  and  $\text{CO}_2$  concentrations as well as moisture  
 650 condensation, but not the in-package RH. Lee, Kang, and Renault (2000) developed a model  
 651 for estimating changes in the atmosphere and humidity within perforated packages of fresh  
 652 produce. The model was based on mass balances of  $\text{O}_2$ ,  $\text{CO}_2$ , nitrogen gas ( $\text{N}_2$ ), and water  
 653 ( $\text{H}_2\text{O}$ ) and included respiration, transpiration and terms for gas and water vapour transfer  
 654 through perforations and films. The water vapour exchange rate through the film was  
 655 modelled based on Fick's law. Similarly, Techavises and Hikida (2008) developed a model  
 656 based in Fick's law that included atmospheric gas ( $\text{O}_2$ ,  $\text{CO}_2$  and  $\text{N}_2$ ) and water vapour  
 657 exchanges in MAP with perforations. The proposed model showed good prediction of gas  
 658 concentrations and RH when compared with experimental results. The differential equation  
 659 used to obtain the volumetric changes inside a perforated MAP of respiring produce for water  
 660 vapour is presented (Eq. 10):

$$661 \quad \frac{dV_H(t)}{dt} = n_p D_H + A_f K_H (P_H - P_T \frac{V_H(t)}{V_T(t)}) \quad (10)$$

662 where  $n_p$  is number of perforations (non-dimensional);  $D_H$  is effective permeability of one  
 663 perforation to water vapour ( $10^{-6} \text{m}^3 \text{h}^{-1} \text{kPa}^{-1}$ );  $A_f$  is surface area of the film package ( $\text{m}^2$ );  
 664  $K_H$  is water vapour transpiration rate of film to water vapour ( $10^{-6} \text{m}^3 \text{m}^{-2} \text{h}^{-1} \text{kPa}^{-1}$ );  $P_H$  is  
 665 partial pressure of water vapour outside the package (kPa);  $P_T$  is total pressure inside the

666 package (kPa), equal to 101.325 kPa;  $V_T(t)$  is total volume of gases inside the package at time  
667  $t$  ( $10^{-6} \text{ m}^3$ ) and effective permeability ( $D_H$ ) is a function of perforation diameter ( $d$ ) in mm:

$$668 \quad D_H = 2.98 \times 10^{-2} d^2 + 5.37 \times 10^{-1} d + 8.22 \times 10^{-1} \quad (11)$$

669 The authors reported that Eq. 10 is valid for water and atmospheric gases in a temperature  
670 range of 5 to 25 °C and for film thickness smaller than 0.025 mm.

671 Rennie and Tavoularis (2009) also developed a space and time dependent mathematical  
672 model for perforation-mediated MAP. The authors considered respiration, transpiration,  
673 condensation, heat transfer (evaporative, convective, and conductive), and convective and  
674 diffusive transport of  $\text{O}_2$ ,  $\text{CO}_2$  and  $\text{N}_2$  and  $\text{H}_2\text{O}$  through the Maxwell-Stefan diffusion and the  
675 convection mass balance model (Eq. 12):

$$676 \quad \rho \frac{\partial \omega_{\text{H}_2\text{O}}}{\partial t} + \nabla \cdot \left( -\rho \omega_{\text{H}_2\text{O}} \sum_{j=1}^n D_{ij} (\nabla x_{\text{H}_2\text{O}} + (x_{\text{H}_2\text{O}} - \omega_{\text{H}_2\text{O}}) \frac{\nabla p}{p}) \right) = -\rho \omega_{\text{H}_2\text{O}} \cdot u \quad (12)$$

677 where  $\rho$  is the gas mixture density ( $\text{kg m}^{-3}$ );  $t$  is time (s);  $\omega_{\text{H}_2\text{O}}$  is  $\text{H}_2\text{O}$  mass fraction (non-  
678 dimensional);  $D_{ij}$  is the  $ij$  component of multicomponent Fick diffusivity ( $\text{m}^2 \text{ s}^{-1}$ );  $x_{\text{H}_2\text{O}}$  is the  
679 mole fraction of water (non-dimensional);  $p$  is the total gas mixture pressure (Pa); and  $u$  is the  
680 velocity vector ( $\text{m s}^{-1}$ ). Their model can be used for steady-state as well as for transient  
681 analysis of MAP in a wide range of conditions and is valid to model  $\text{H}_2\text{O}$  transport in the  
682 ambient storage environment, the perforations and in the headspace.

683 Li, Li, and Ban (2010) reported a model applicable to non-perforated and micro-perforated  
684 MAP films which simulates changes in concentrations of various gases, such as  $\text{O}_2$ ,  $\text{CO}_2$ ,  
685 ethylene ( $\text{C}_2\text{H}_4$ ) and  $\text{H}_2\text{O}$  inside MAP films over time based on Fick's law of diffusion.  
686 While, Mahajan, Rodrigues, and Leflaive (2008c) developed a mathematical model to  
687 describe the changes in  $WVTR$  as a function of perforation diameter, length and storage  
688 temperature in perforation-mediated MAP:

$$689 \quad WVTR = 2.28 D^{1.72} L^{-0.72} e^{-\frac{12.62}{RT_s}} \quad (13)$$

690 where  $D$  is the perforation diameter (mm),  $L$  is the perforation length (mm),  $R$  is the universal  
691 gas constant ( $0.008314 \text{ kJ mol}^{-1} \text{ K}^{-1}$ ) and  $T_s$  is the storage temperature (K). These studies  
692 present the potential role and application of integrated models in the design of perforation-  
693 mediated MAP systems for FFV. Their findings also highlight that research needs to develop  
694 more flexible and robust models.

695

### 696 **4.3. Active moisture control systems**

697 A possible solution to control humidity involves the use of moisture absorbers. In this case the  
 698 package design requires, in addition to packaging specifications, the selection of appropriate  
 699 desiccants and specification of the amount to be used. This respiration-transpiration model  
 700 presented by Song et al. (2002) was thus developed into the new model presented by Song et  
 701 al. (2001). The new model introduced the moisture sorption behaviour of the absorbent ( $m$ ) as  
 702 follows:

$$703 \quad m = k_{sa} m_{ab}(P_i - P_{ab}) \quad (14)$$

704 where  $m$  is moisture absorption rate of the absorbent ( $\text{kg h}^{-1}$ );  $k_{sa}$  is the absorbent mass  
 705 transfer coefficient that can be experimentally determined absorbent mass transfer coefficient  
 706 ( $\text{kg}_{\text{water}} \text{kg}_{\text{dry matter}}^{-1} \text{h}^{-1} \text{atm}^{-1}$ );  $m_{ab}$  is mass of dried absorbent (kg);  $P_i$  is water vapour pressure  
 707 inside the package containing absorbent (atm); and  $P_{ab}$  is water vapour pressure on the surface  
 708 of the absorbent (atm). Additionally,  $P_{ab}$  is a function of moisture sorption characteristics of  
 709 absorbents and can be estimated (Eq. 15):

$$710 \quad P_{ab} = P_{sp} a_w \quad (15)$$

711 where  $P_{sp}$  is saturated water vapour pressure at constant temperature (atm) and  $a_w$  is the water  
 712 activity of the moisture absorbent (non-dimensional), which can be experimentally  
 713 determined as a function of moisture content. The modified model considered moisture  
 714 sorption characteristics of absorbent and mass transfer coefficient between adsorbent and  
 715 package headspace. The model was successfully validated with blueberries using two  
 716 commercial desiccants, Sanwet (Hoechst Celanese, USA) and Xylitol (Sigma, USA).  
 717 Although the model predictions were in agreement with experimental data obtained, the  
 718 amount of condensation inside the packages was not quantified. Therefore, it is not possible to  
 719 optimise the amount of absorber needed to absorb the excess moisture inside the packages.

720 Furthermore, Mahajan et al. (2008b) investigated the kinetics of moisture absorption for  
 721 mixed desiccant ( $\text{CaCl}_2$ , KCl and sorbitol) at 4, 10, and 16 °C, at different humidity levels  
 722 (76, 86 and 96%). Change in moisture content of the mixed desiccant with respect to storage  
 723 time was fitted to a Weibull distribution model (Eq. 16).

$$724 \quad M_t = M_\infty \left[ 1 - e^{\left(\frac{-t}{\beta}\right)} \right] \quad (16)$$

725 where  $M_t$  is the moisture absorbed (g) at a determined time  $t$  (d);  $M_\infty$  is moisture holding  
 726 capacity at equilibrium (g); and  $\beta$  is the kinetic parameter, which defines the rate of moisture  
 727 uptake process and it represents the time (d) needed to accomplish 63 % of the moisture  
 728 uptake process. The moisture holding capacity was found to be dependent on RH, which

729 increased from 0.51 to 0.94 g water g<sup>-1</sup> desiccant when RH was increased from 76 to 96%.  
730 Similarly, Rux et al. (2016) used a Weibull distribution to fit the moisture uptake data  
731 obtained from the individual humidity-regulating trays. The authors found that packaged  
732 produce with absorbers lost more mass than control samples. Their findings emphasised the  
733 importance of selecting the appropriate and correct amount of moisture absorber in order to  
734 prevent excessive mass loss and shrivelling of packaged product.

735

## 736 **5. Conclusion and future research needs**

737 Harvested horticultural produce are transported from farm to the final consumer. This process  
738 involves many challenges since the product continues both metabolic and physiological  
739 activities after harvest. Thus, strict control of temperature and RH along the supply chain and  
740 storage are decisive factors for maintaining quality of FFV. These factors govern the  
741 respiration and transpiration processes and consequently degradation of organic substrates and  
742 moisture loss. Appropriate packaging of FFV, under optimum storage conditions, offers a  
743 possibility to slow down the physiological processes and extend storage life. However, the  
744 control of moisture evolution inside packaged horticultural products is complicated due to  
745 numerous factors (intrinsic and extrinsic) and the complexity of their interactions. Therefore,  
746 application of integrated mathematical models for water relations presents a possible solution;  
747 to integrate different factors affecting moisture evolution inside packaged horticultural  
748 products. This is vital in order to match the high physiological product requirements and the  
749 mass balance of a packaging system in terms of water vapour inside and outside the package.  
750 It will provide a guiding tool for all the role players in food packaging industry on package  
751 system optimisation such as selection of packaging film, produce amount, package  
752 dimensions, perforation, and moisture control strategies; thereby eliminating the “pack-and-  
753 pray” approach commonly adopted by the food packaging industry.

754

## 755 **Nomenclature**

756  $TR_s$  transpiration rate per unit surface area (mg cm<sup>-2</sup>h<sup>-1</sup> or mg cm<sup>-2</sup>s<sup>-1</sup>)  
757  $TR_m$  transpiration rate per unit of initial mass (g kg<sup>-1</sup>h<sup>-1</sup>, mg kg<sup>-1</sup>h<sup>-1</sup> or mg kg<sup>-1</sup>s<sup>-1</sup>)  
758  $RH$  relative humidity (%)  
759  $M_i$  initial mass of the product (mg, g or kg)  
760  $M_t$  product mass at a determined time (mg, g or kg)  
761  $A_s$  initial surface area of the product (cm<sup>2</sup> or m<sup>2</sup>)  
762  $t$  time (s, h or d)



763	$k_t$	transpiration coefficient ( $\text{mg kg}^{-1}\text{s}^{-1} \text{MPa}^{-1}$ )
764	$P_s$	water vapour pressure at the evaporating surface of the product (MPa)
765	$P_\infty$	ambient water vapour pressure (MPa)
766	$k_s$	skin mass transfer coefficient ( $\text{mg kg}^{-1} \text{s}^{-1} \text{MPa}^{-1}$ )
767	$k_a$	air film mass transfer ( $\text{mg kg}^{-1} \text{s}^{-1} \text{MPa}^{-1}$ )
768	$\delta$	diffusion coefficient of water vapour in air ( $\text{m}^2\text{s}^{-1}$ )
769	$\tau, s$	product skin thickness (m)
770	$\varphi$	fraction of product surface covered by pores (non-dimensional)
771	$h_d, \beta$	convective mass transfer coefficient ( $\text{m s}^{-1}$ )
772	$\zeta_1$	fraction of surface behaving as a free water zone (non-dimensional)
773	$R_D, R$	universal gas constant ( $\text{J kg}^{-1}\text{°C}^{-1}$ )
774	$T$	ambient temperature ( $\text{°C}$ )
775	$\zeta_2$	fraction of surface behaving as porous membrane (non-dimensional)
776	$\mu$	resistance factor (non-dimensional)
777	$x_P$	volume related water content of air in the intercellular spaces in the centre of the
778		produce ( $\text{mg cm}^{-3}$ )
779	$x_A$	volume related water content of air unaffected by produce ( $\text{mg cm}^{-3}$ )
780	$r_B$	boundary layer resistance in the water vapour pathway ( $\text{s cm}^{-1}$ )
781	$r_T$	tissue resistance in the water vapour pathway ( $\text{s cm}^{-1}$ )
782	$x_{ps}$	water content of the air at the produce surface ( $\text{mg cm}^{-3}$ )
783	$F_w$	water vapour flux through the perforated film ( $\text{m}^3 \text{h}^{-1}$ )
784	$\alpha$	water vapour concentration under saturation vapour pressure (non-dimensional)
785	$H_A$	relative humidity in the ambient atmosphere (non-dimensional)
786	$H$	relative humidity (non-dimensional)
787	$S, A_f$	surface area of the film ( $\text{m}^2$ )
788	$P_w$	water vapour permeability coefficient of the film ( $\text{m}^2 \text{h}^{-1}$ )
789	$L$	film thickness (m)
790	$\pi$	3.14 (non-dimensional)
791	$R_h$	radius of perforation (m)
792	$N$	number of pores (non-dimensional)
793	$D_w$	diffusion coefficient of water vapour in air ( $\text{m}^2 \text{h}^{-1}$ )
794	$V_H(t)$	volume of water vapour inside the package at a determined time ( $10^{-6} \text{m}^3$ )
795	$n_p$	number of perforations (non-dimensional)
796	$D_H$	effective permeability of one perforation to water vapor ( $10^{-6}\text{m}^3\text{h}^{-1} \text{kPa}^{-1}$ )

797	$K_H$	water vapour transpiration rate of film to water vapour ( $10^{-6}\text{m}^3 \text{m}^{-2} \text{h}^{-1} \text{kPa}^{-1}$ )
798	$P_H$	partial pressure of water vapour outside the package (kPa)
799	$P_T$	total pressure inside the package (kPa)
800	$V_T(t)$	total volume of gases inside the package at a determined time ( $10^{-6} \text{m}^3$ )
801	$d, D$	perforation diameter (mm)
802	$\rho$	gas mixture density ( $\text{kg m}^{-3}$ )
803	$\omega_{\text{H}_2\text{O}}$	$\text{H}_2\text{O}$ mass fraction (non-dimensional)
804	$D_{ij}$	ij component of multicomponent Fick diffusivity ( $\text{m}^2\text{s}^{-1}$ )
805	$x_{\text{H}_2\text{O}}$	mole fraction of $\text{H}_2\text{O}$ (non-dimensional)
806	$\omega_j$	mass fraction of $\text{H}_2\text{O}$ (non-dimensional)
807	$p$	total gas mixture pressure (Pa)
808	$u$	velocity vector ( $\text{m s}^{-1}$ )
809	$L$	perforation length (mm)
810	$T_s$	storage temperature (K)
811	$m$	moisture absorption rate of the absorbent ( $\text{kg h}^{-1}$ )
812	$k_{sa}$	absorbent mass transfer coefficient ( $\text{kg}_{\text{H}_2\text{O}} \text{kg}_{\text{dry matter}}^{-1} \text{h}^{-1} \text{atm}^{-1}$ )
813	$m_{ab}$	is mass of dried absorbent (kg);
814	$P_i$	is water vapour pressure inside the package containing absorbent (atm)
815	$P_{ab}$	is water vapour pressure on the surface of the absorbent (atm)
816	$P_{sp}$	saturated water vapour pressure at constant temperature (atm)
817	$a_w$	is the water activity of the moisture absorbent (non-dimensional)
818	$M_t$	is the moisture absorbed (g) at a determined time (days)
819	$M_\infty$	is moisture holding capacity at equilibrium (g)
820	$B$	kinetic parameter (non-dimensional)

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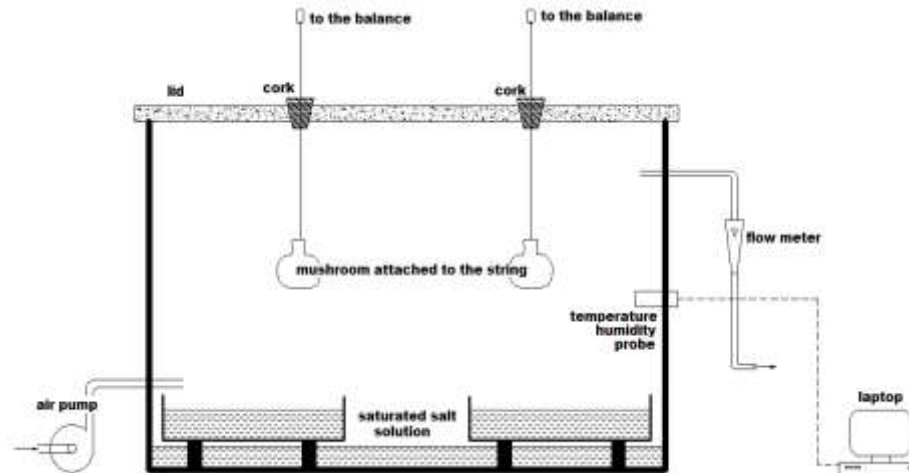
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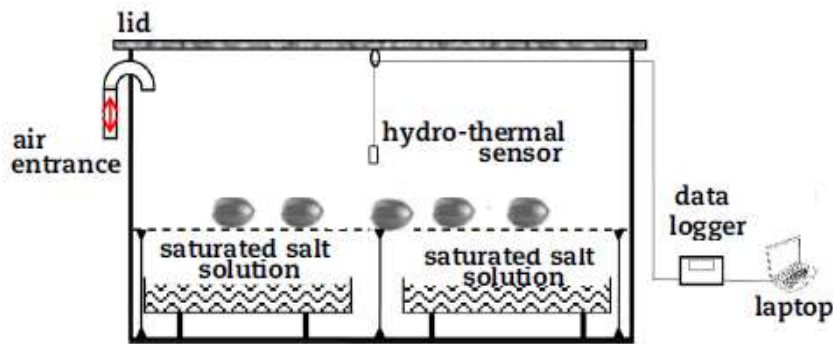
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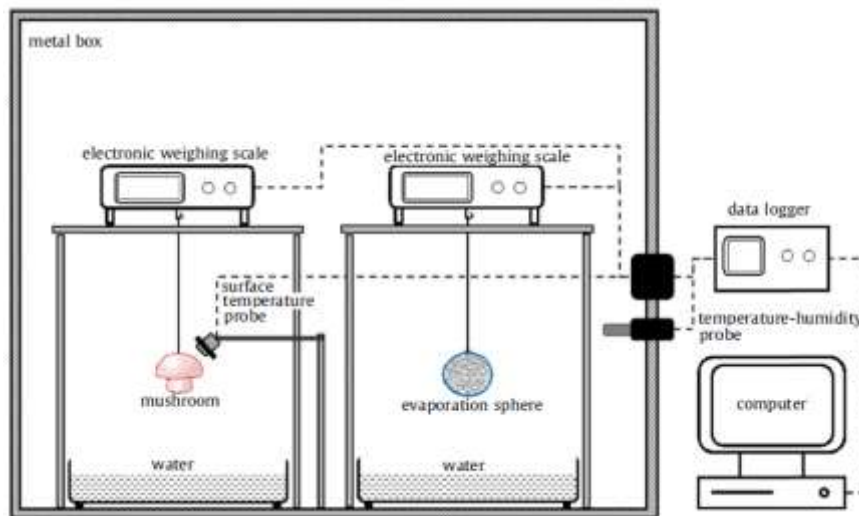
(A)



(B)

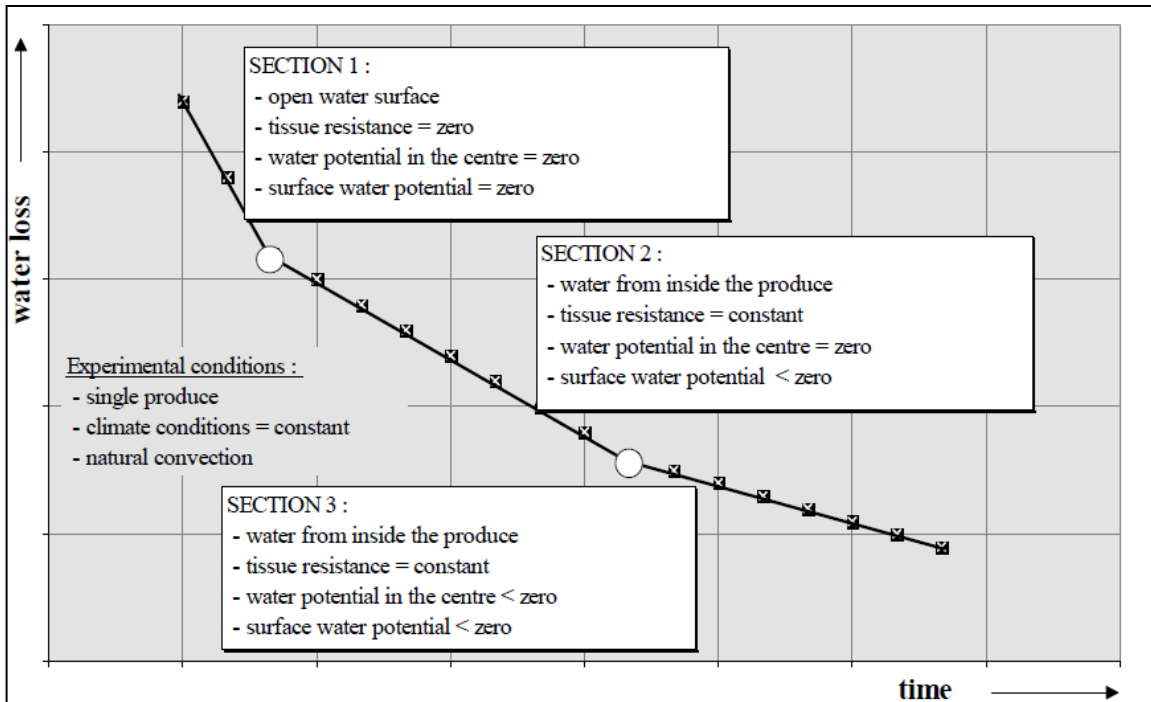


(C)

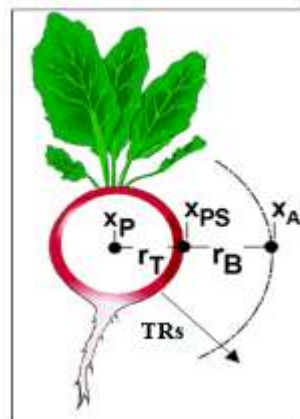


**Fig. 1.** Schematic representation of a typical experimental setup for used for non-continuous (A and B) and continuous (C) measurement of produce mass loss (Adopted from Mahajan et al. (2008a), Xanthopoulos et al. (2014), and Rux et al. (2015), respectively)

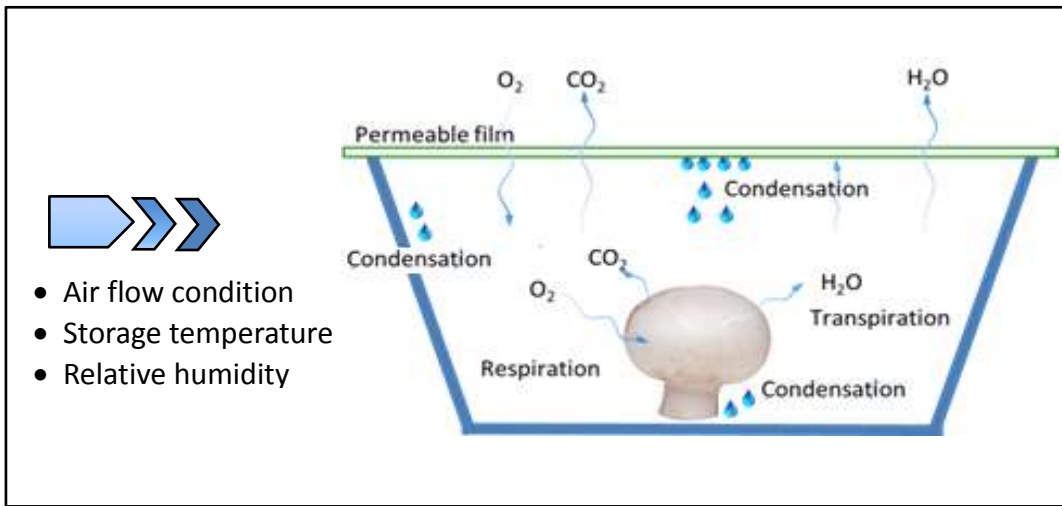




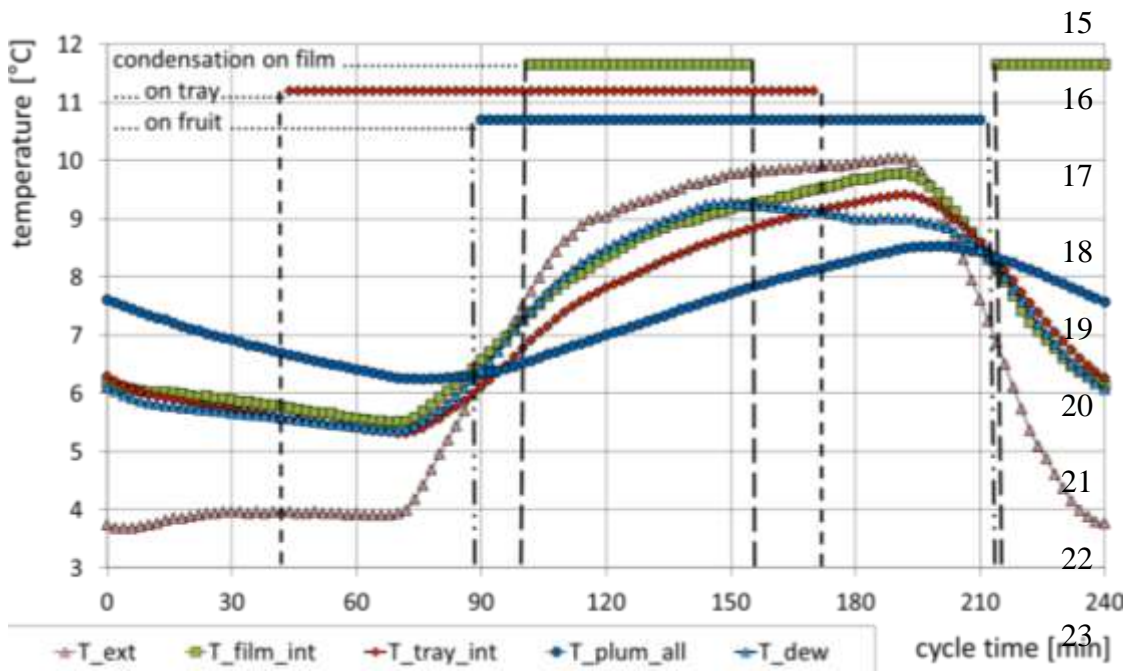
**Fig. 2.** Sections describing the typical water loss of fruit and vegetables during postharvest storage (Adopted from Linke, 1997)



**Fig. 3.** Basic relations for calculating tissue and boundary layer resistances (Adopted from Linke, 1998)



**Fig. 4.** Condensation in packaged fresh produce and environmental parameters impacting the condensation process



**Fig. 5.** Condensation dynamics in plastic film packaging containing fresh plums

(Adopted from Linke & Geyer, 2013)

29 **Table 1.** Range of transpiration coefficients for some fresh fruit and vegetables

<b>Fruit</b>	<b><math>k_t</math> (mg kg<sup>-1</sup> s<sup>-1</sup> MPa<sup>-1</sup>)</b>	<b>Vegetables</b>	<b><math>k_t</math> (mg kg<sup>-1</sup> s<sup>-1</sup> MPa<sup>-1</sup>)</b>
Apple	16 - 100	Potato	2 - 171
Pear	10 - 144	Onion	13 - 123
Grapefruit	29 - 167	Tomato	71 - 365
Orange	25 - 227	Cabbage	40 - 667
Grapes	21 - 254	Lettuce	680 - 8750
Plum	110 - 221	Leek	530 - 1042
Lemon	139 - 229	Carrot	106 - 3250
Peach	142 - 2089	Celery	104 - 3313

30 Source: Thompson et al., 1998 compiled from Sastry et al., 1977

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32 **Table 2.** Tissue resistance of single fresh fruit and vegetables after harvest at natural convection

<b>Fruit</b>	<b>Tissue Resistance (<math>r_T</math>, s cm<sup>-1</sup>)</b>	<b>Vegetables</b>	<b>Tissue Resistance (<math>r_T</math>, s cm<sup>-1</sup>)</b>
Strawberries	3 - 23	Radish tubers	0.25 - 1.5
Plums	23 - 38	Carrots (without leaves)	1 - 6
Apples	170 - 320	White asparagus	11 - 12.5
		Bell peppers	35 - 80

33 Source: Linke and Geyer, 2000; Linke and Geyer, 2001

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49 **Table 3.** Summary of transpiration rate models applied for various horticultural commodities under different storage conditions and their limitations.

Proposed model equation	Unit	Storage conditions	Product	TR Range	Limitation	Reference
$\frac{Q_r W + h A (T - T_p)}{\lambda}$	kg h <sup>-1</sup>	T: 0 RH: 100	Apple	18.4 <sup>2</sup> (normal air) 5.7 <sup>2</sup> (1% O <sub>2</sub> , 1% CO <sub>2</sub> ) 8.7 <sup>2</sup> (3% O <sub>2</sub> , 3% CO <sub>2</sub> )	Model was not validated; not tested in MAP (tested in controlled atmosphere)	Kang and Lee, 1998
		T: 10 RH: 82	Fresh-cut onion Fresh-cut green onion	447 <sup>2</sup> (normal air) 363 <sup>2</sup> (normal air)		
$\frac{Q_r W + W C_s \frac{dT_{sp}}{dt}}{\lambda}$	kg h <sup>-1</sup>	T: 15, 25 RH: 10, 60	Blueberry	NG	T inside the package was considered equal to the T <sub>s</sub>	Song et al., 2002
$\rho \cdot K_i \cdot (a_{wi} - a_w) \cdot (1 - e^{-aT})$	mg cm <sup>-2</sup> h <sup>-1</sup>	T: 4, 10, 16 RH: 76, 86, 96	Mushrooms	0.14 - 2.5 <sup>1</sup>	Model not tested in MAP; does not consider RR	Mahajan et al., 2008
$K_i \cdot (a_{wi} - a_w) \cdot (1 - e^{-aT})$	g kg <sup>-1</sup> 24h <sup>-1</sup>	T: 5, 10, 15 RH: 76, 86, 96	Pomegranate arils	48 - 698 <sup>2</sup>	Model not tested in MAP; does not consider RR	Caleb et al., 2013
	g kg <sup>-1</sup> h <sup>-1</sup>	T: 5, 10, 15 RH: 76, 86, 96	Strawberries	240 - 1160 <sup>2</sup>	Model does not consider RR	Sousa-Gallagher et al., 2013
$K_i \cdot e^{\left[-\frac{E_a}{R} \left(\frac{1}{T} - \frac{1}{T_r}\right)\right]} \cdot (a_{wi} - a_w) \rho \cdot K_i \cdot e^{\left[-\frac{E_a}{R} \left(\frac{1}{T} - \frac{1}{T_r}\right)\right]} \cdot (a_{wi} - a_w)$	g kg <sup>-1</sup> h <sup>-1</sup> mg cm <sup>-2</sup> h <sup>-1</sup>	T: 10, 15, 20 RH: 70, 80, 92	Grape tomato	18 - 107 <sup>2</sup> 0.012 - 0.058 <sup>1</sup>	Model not validated; does not consider RR	Xanthopoulos et al., 2014
$K_i \cdot (a_{wi} - a_w) \cdot (1 - e^{-aT}) + 8.6 RR_{CO2,r} \cdot e^{\frac{-E_a}{R} \left[\frac{1}{(T+273)} - \frac{1}{(Tr+273)}\right]}$	mg kg <sup>-1</sup> h <sup>-1</sup>	T: 13 RH: 100	Mushrooms Strawberries Tomato	713 <sup>2</sup> 122 <sup>2</sup> 17.6 <sup>2</sup>	Model was not validated	Mahajan et al., 2016

50 <sup>1</sup> mg cm<sup>-2</sup> h<sup>-1</sup> (area based); <sup>2</sup> mg kg<sup>-1</sup> h<sup>-1</sup> (mass based); T is temperature (°C), RH is relative humidity (%), RR is respiration rate Q<sub>r</sub>-respiration heat of produce; W-  
51 produce weight; h-convective heat transfer coefficient; A-produce surface area; T<sub>s</sub>-surrounding temperature; T<sub>p</sub>-produce temperature; λ-latent heat of moisture  
52 evaporation/vaporization; C<sub>s</sub> is specific heat of the produce, T<sub>sp</sub> product surface temperature; ρ-water density; K<sub>i</sub>-mass transfer coefficient; a<sub>w</sub>-water activity of the container; a<sub>wi</sub>-  
53 water activity of the commodity; a-coefficient; E<sub>a</sub>-activation energy; R-universal gas constant; T<sub>r</sub>-reference temperature; RR<sub>CO2,ref</sub>-respiration rate of the product at T<sub>r</sub> and 8.6 is the  
54 conversion factor for obtaining TR from the respiratory heat generation, NG is not given.  
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