



2-furyl(phenyl)methanol isolated from *Atractylis gummifera* rhizome exhibits anti-leishmanial activity



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ABSTRACT

We report for the first time the isolation of 2-furyl(phenyl)methanol (**5**) from the chloroform extracts of the *Atractylis gummifera* roots. *A. gummifera* is a thistle belonging to the Asteraceae family that produces the *ent*-kaurane diterpenoid glycoside atractyloside (ATR). ATR (**1**) was isolated and chemically modified to obtain its aglycone atractyligenin (**2**) and the methylated derivatives ATR-OMe (**3**) and genine-OMe (**4**). The compounds **1–5** were structurally characterised and evaluated against the intracellular amastigote, cultured within macrophages, and the extracellular promastigote of *Leishmania donovani*, the protozoan parasite responsible for the highly infective disease visceral leishmaniasis, which is fatal if untreated. The 2-furyl(phenyl)methanol **5** exhibited notable activity against the promastigote.

1. Introduction

Leishmaniasis (L) is a neglected tropical disease (NTD), affecting 12 million people in almost 90 countries and causing 50,000 deaths each year [1–3]. Leishmaniasis is generally grouped into three clinical forms, including cutaneous (CL), mucocutaneous and visceral leishmaniasis (VL) [4]. CL is the most common variant and, although self-healing, can create permanent disfiguring scars [5]. CL is prevalently caused by *Leishmania amazensis*, *Leishmania major*, *Leishmania braziliensis*, *Leishmania mexicana* and *Leishmania panamensis* [6]. On the other hand, VL (also known as *kala azar*), caused by *Leishmania donovani* and *Leishmania infantum*, is the most severe form of the infection and, if not treated, might result in widespread destruction of mucous membranes, visceral and haemopoietic organs, and ultimately death [7].

Although vaccines against the *Leishmania* parasites are currently being developed, the use of chemotherapeutic agents, such as miltefosine, panoromycin and pentavalent antimonials (sodium stibogluconate and meglumine antimoniate), remains, by far, the most effective treatment option [1,8–10]. Anti-leishmanial medicines with the exception of miltefosine are administered intravenously over prolonged periods of time. The rise of antimonial drug-resistant parasites, decades-old first-line drugs and low patient compliance, due to long drug regimens and unpractical administration routes, has led to high rates of treatment failure and relapses [7,11].

As a result, there is an urgent need to access and identify novel and

diverse chemical scaffolds to support and improve the process of developing novel anti-leishmanial medicines. To this end, natural products continue to be a remarkable source of therapeutic leads [12] and tripterpenoids and *ent*-kaurane diterpenoid glycosides, commonly isolated from some Asteraceae and Annonaceae species, were shown to be active *in vitro* against trypanosomatids, such as *Trypanosoma cruzi* and *T. brucei* [13–15].

This finding prompted us to evaluate against protozoan parasites of the genus *Leishmania* (also belonging to the Trypanosomatida order) both aqueous and chloroform extracts of *Atractylis gummifera*, a thistle belonging to the Asteraceae family that grows throughout the Mediterranean region and is widely known for its acute toxicity [16]. The *A. gummifera* rhizome contains atractyloside **1** (ATR), a glycoside consisting of the tetracyclic *ent*-kaurane diterpene atractyligenin **2** attached through a β -linkage to the anomeric carbon of a D-(+)-glucose residue (Fig. 1) [17]. ATR exerts its biological activity by inhibiting the oxidative phosphorylation in the mitochondria of hepatocytes and proximal tubular epithelial cells, and has been extensively investigated for its pharmacological and chemical properties [18–20].

To our knowledge, *A. gummifera* metabolites have not been tested to date against *Leishmania* parasites and here we wish to report on the isolation, characterisation and anti-leishmanial activity of known plant active principles atractyloside (**1**) and atractyligenin (**2**), their methylated derivatives ATR-OMe (**3**) and atractyligenin-OMe (**4**), and 2-furyl(phenyl)methanol (**5**) that was found in the thistle rhizome's

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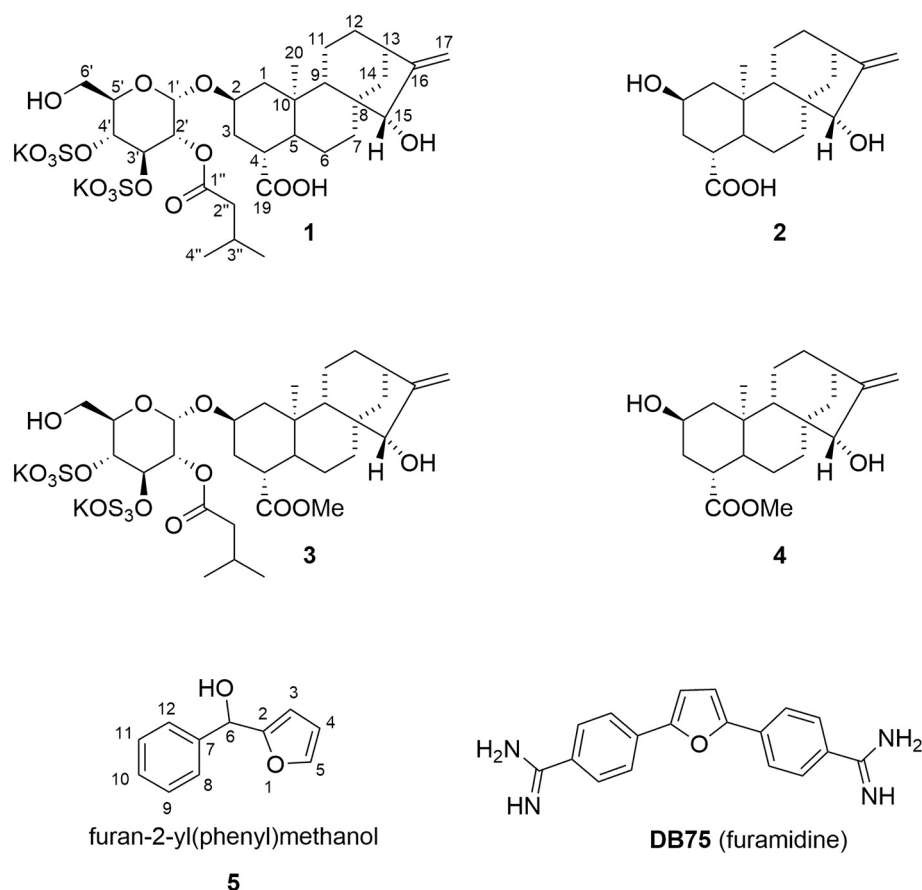


Fig. 1. Structures of atractyloside (1), its aglycone atractyligenin (2), their methyl ester derivatives 3 and 4, and the 2-furyl(phenyl)methanol (5), which was isolated from the CHCl₃ extracts of *A. gummifera* roots. Furamidine (DB75) is a pentamidine analogue bearing a core 2,5 disubstituted furan ring.

Table 1
¹H- and ¹³C-NMR data for compounds 5^a.

Position	δ_{H} (J in Hz)	δ_{C}
2	–	141.5
3	6.78 d (3.3), 1H	115.6
4	6.54 dd (3.3, 1.8)	111.4
5	7.71 d (1.3)	144.8
6	5.64 d (4.4)	62.8
7	–	135.7
8/12	7.49 d (7.4) 2H	126.4
9/11	7.38 t (7.5) 2H	128.4
10	7.32–7.30 m 1H	127.8

^a Recorded in DMSO-*d*₆, 600 and 125 MHz.

chloroform extracts (Fig. 1). The compounds structures were established on the basis of spectroscopic methods, including 1D- and 2D-NMR techniques (¹H- and ¹³C-NMR spectra and COSY, HSQC, NOESY and HMBC experiments) and high-resolution mass spectrometry (HRMS). The luciferase assay was used to screen the compounds (1–5) against the two structural variants of *Leishmania donovani*, i.e., the amastigote non-motile form, which is found in the mononuclear phagocytes and circulatory systems of humans, and the promastigote (extracellular and motile) form, which is found in the alimentary tract of sand flies [21].

2. Results and discussion

2.1. Chemistry

The crushed rhizome of *A. gummifera* was treated with an aqueous-

acetone solution in order to isolate atractyloside 1 (ATR), according to published methods [22,23]. ATR was isolated in good yield and its NMR spectra and MS analysis were in accordance with the literature data [17]. The aqueous-acetone rhizome mixture was then extracted with chloroform and the organic fraction was concentrated and subjected to column chromatography over normal phase silica gel and semi-preparative HPLC to afford 2-furyl(phenyl)methanol 5 as a pale yellow amorphous solid. The molecular formula of 5 was established as C₁₁H₁₀O₂ by HR-ESI-MS, which showed the quasi molecular ion peak [M + H]⁺ at *m/z* 175.0755 (calculated for 174.0681). The IR spectrum showed absorptions for hydroxyl and aromatic (mono-substituted) groups at 3366 and 749 cm⁻¹, respectively.

The ¹H-NMR spectrum of 5 displayed characteristic up-field protons of a mono-substituted benzene ring at δ_{H} 7.49 (d, *J* = 7.4 Hz, H-8, H-12), 7.38 (t, *J* = 7.5 Hz, H-9, H-11) and 7.32–7.30 (m, H-10) (Table 1). The most up-field singlet at δ_{H} 7.71 (d, *J* = 1.3 Hz, H-5) correlated with signals at δ_{H} 6.54 (dd, *J* = 3.3, 1.8 Hz, H-4) and 6.78 (d, *J* = 3.3 Hz, H-3) in ¹H-¹H COSY and NOESY experiments, and this was indicative of a 2-substituted furan ring.

One signal observed at δ_{H} 5.64 (d, *J* = 4.4 Hz, H-6), which HSBC correlated with δ_{C} 62.8, suggested the presence of an oxygenated sp³ methine bridge connecting the phenyl group with the furan ring. The complete linkage of the phenyl ring and its connection with the furan moiety via the methane bridge was resolved by the key HMBC correlations from H-6 (δ_{H} 5.64) to C-8/12 (δ_{C} 126.4), C-7 (δ_{C} 135.7) and C-2 (δ_{C} 141.5), from H-3 (δ_{H} 6.78) to C-4 (δ_{C} 111.4) and C-5 (144.8), and from H-9/11 (δ_{H} 7.38) to C-8/12 (δ_{C} 126.4) and C-10 (δ_{C} 127.8) (Fig. 2).

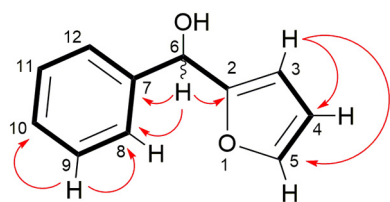


Fig. 2. Key HMBC (in red) and COSY (in bold) correlations of **5**. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

2.2. Biology

Compounds **1–5** were tested for cytotoxicity against transgenic firefly luciferase-expressing *Leishmania donovani* amastigote and promastigote, and bone marrow-derived macrophages from Balb/C mice. ATR **1**, its derivatives **2–4** and the crude chloroform extracts of *A. gummifera* were incubated at different concentrations (0.001–10 mM) with *L. donovani* promastigote and amastigote, cultured intracellularly within macrophages, for 72 h (**Fig. 3**). Residual ATP levels, or luciferase activity, against exogenous luciferin (0.1 µg/mL) were determined using a luminometer and the viability of *L. donovani*, expressed as percent of the no-drug control experiments, was subsequently derived.

Atractyloside **1** and its genine **2** did not exhibit any anti-leishmanial or cytotoxic activity against macrophages, conversely, they appeared to stimulate the growth of the parasites in a dose-dependent fashion as illustrated in **Fig. 3a** (open squares, red lines). However, their methyl carboxylate derivatives **3** and **4** were found to inhibit the growth of *L. donovani* promastigote with IC₅₀ (the dose that kills 50% of cells) values of 5 mM and 0.23 mM, respectively (**Table 2**). Interestingly, the crude rhizome chloroform extracts showed anti-leishmanial activity with growth inhibitory concentrations of 0.059 and 0.88 mM against the promastigote and amastigote forms, respectively (**Table 2**). This notable finding prompted us to analyse the composition of the chloroform extracts of the plant root and 2-furyl(phenyl)methanol **5** was identified as the chemical agent responsible for anti-parasitic activity. The newly isolated compound **5** was retested on its own and showed remarkable activity against the *L. donovani* promastigote (0.029 mM), with relatively low toxicity against macrophages and high selectivity index (SI = 345). Compound **5** retained biological activity, albeit at high concentration, also against the amastigote form of the parasite at 1 mM (SI = 10).

Chiral chromatography analysis of the 2-furyl(phenyl)methanol **5** isolated from the chloroform extract of the *A. gummifera* rhizome revealed the presence of two enantiomers. The configuration of the stereoisomers was assigned by comparing the HPLC elution order with previously reported data [24]. The first peak, which eluted at 25.4 min,

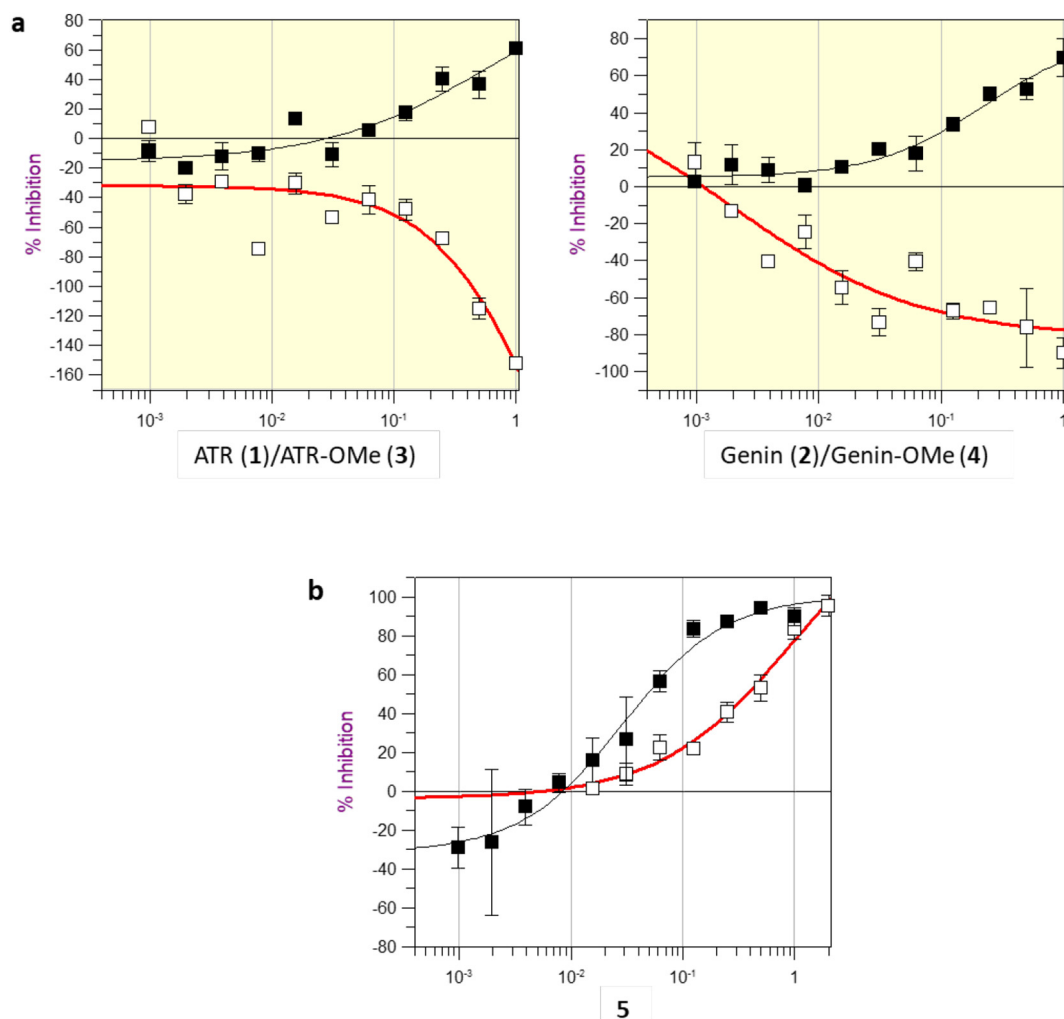


Fig. 3. a-b. Dose response curves. **a**) Effects of ATR (**1**) / ATR-OMe (**3**) and genin (**2**) / genin-OMe (**4**) on *L. donovani* promastigotes growth. ATR **1** and genin **2** are shown as open squares, whereas methylated compounds (**3** and **4**) are shown as filled squares, black line) and amastigotes (open squares, red line). The Y-axis reports the % growth inhibition and X-axis the concentration of the compounds used in this study. Concentrations are in mM and % inhibition was determined using the luciferase assay protocol previously described [29].

Table 2

Anti-leishmanial activity of *A. gummifera* metabolite ATR (1), genin 2, methylated derivatives ATR-OMe (3) and atractyligenin-OMe (4), and 2-furyl(phenyl)methanol (5).

Compound	Cytotoxicity IC ₅₀ (mM) ^a				
	Promastigotes	Amastigotes	Macrophages ^b	SI _{promastigote} ^c	SI _{amastigote} ^d
ATR (1)	N/A ^e	N/A	N/A	–	–
Genin (2)	N/A	N/A	N/A	–	–
ATR-OMe (3)	5.18 ± 9.77	N/A	N/A	–	–
Genin-OMe (4)	0.23 ± 0.16	N/A	N/A	–	–
Crude CHCl ₃ extracts	0.059 ± 0.015	0.88 ± 0.34	> 10	> 169	> 11
5	0.029 ± 0.006	1.00 ± 1.73	> 10	> 345	> 10

^a IC₅₀ is defined as the dose that kills 50% of cells.

^b Bone marrow-derived macrophages from Balb/C mice.

^c Selectivity Index_{promastigote} = IC₅₀ macrophages/ IC₅₀ promastigotes.

^d Selectivity Index_{amastigote} = IC₅₀ macrophages/IC₅₀ amastigotes.

^e N/A – No activity at 10 mM. The experiments were carried out in triplicate.

was identified as the *R* enantiomer, whereas the second peak with a retention time of 30.4 min corresponded to the *S* enantiomer.

The 2-furyl(phenyl)methanol (5) exhibited moderate activity against *L. donovani* amastigote but was 30-fold more active against the promastigote form of the parasite (29 μM). The phenyl-furan scaffold is particularly rare in natural product frameworks and there are only few examples reported in the literature of secondary metabolites from plants and fungi containing this bis-aromatic unit. These include, but are not limited to, peplidiforone C (6) [25] from *Hypericum peplidifolium*, 3-hydroxy-1,4,7-trimethoxydibenzofuran (7) [26] from *Hypericum revolutum* ssp. and myrothecol (8) [27], a polyketide isolated from the fungus *Myrothecium* sp. GS-17 (Fig. 4). The latter (8), which contains a 1,3-dihydroisobenzofuran unit linked to a 2-furyl moiety, is closely related to the structure of the 2-furyl(phenyl)methanol (5) isolated from the chloroform fraction of *A. gummifera*. As a result, we postulate that 5 might be of polyketide biosynthetic origin with a pathway involving prenylation of a benzene substrate followed oxidation and synthase-mediated cyclisation via Michael addition mechanisms.

On the other hand, the phenyl-furan moiety is a privileged chemical scaffold present in the structure of a number of synthetic medicinal compound. For example, furamide (DB75) [28] is a pentamide analogue bearing a core 2,5-disubstituted furan ring, which was evaluated in phase III clinical trials for Human African trypanosomiasis (Fig. 1). As a result, further chemical modifications can be carried to the easily accessible 2-furyl(phenyl)methanol in order to furnish attractive anti-leishmanial chemical probes.

3. Materials and methods

3.1. Experimental section

3.1.1. General experimental information

¹H- and ¹³C-NMR spectra were acquired using Bruker Avance 400 MHz and Avance III 600 MHz NMR spectrometers. Chemical shifts are reported in parts per million (ppm) with the solvent resonance as

the internal standard and coupling constants (*J*) are quoted in Hertz (Hz). Spin multiplicities are described as s (singlet), bs (broad singlet), d (doublet), dd (doublet of doublets), t (triplet), q (quadruplet), and m (multiplet). Optical rotations were determined on a Bellingham and Stanley digital polarimeter ADP 410.

Infrared spectra were acquired on a Perkin Elmer FT-IR Spectrometer (Spectrum 1000) and absorbance frequencies are reported in reciprocal centimeters (cm⁻¹).

LC-MS analysis was conducted on an Agilent 6100 series Quadrupole LC-MS system coupled with a G4220A 1290 binary pump/DAD, using an Agilent Zorbax SB-C19 reverse-phase column (2.1 × 50 mm) with a flow rate of 1 mL min⁻¹. Solution B was kept at 5% for 0.8 min, then a gradient was applied up to 100% B at 6.6 min and solution B was held at 100% for 1.6 min, then B was reverted back to 5% until 10 min. Eluent A: H₂O/0.1% formic acid; eluent B: CH₃CN/0.1% formic acid.

HR-MS data were recorded on Thermo LTQ Orbitrap coupled to an HPLC system (PDA detector, PDA autosampler, and pump). The following conditions were used: capillary voltage of 45 V, capillary temperature of 260 °C, auxiliary gas flow rate of 10–20 arbitrary units, sheath gas flow rate of 40–50 arbitrary units, spray voltage of 4.5 kV, and mass range of 100–2000 amu (maximal resolution of 30,000). For LC/MS, a Sunfire C18 analytical HPLC column (5 μm, 4.6 mm × 150 mm) was used with a mobile phase of 0 to 100% MeOH over 30 min at a flow rate of 1 mL min⁻¹.

Chiral HPLC analysis was performed using a Daicel Chiralcel OD analytical column (250 × 4.6 mm, 10 μm) and UV detection was monitored at 230 nm.

TLC and preparative TLC were performed on Merck Silicagel 60 F₂₅₄ aluminum sheets. TLC system 1: Si gel, EtOAc:acetone:H₂O:HOAc / 5:3:1:1, sprayed with a vanillin-H₂SO₄ solution prepared by dissolving 15 g of vanillin in 250 mL of EtOH with 0.01% conc H₂SO₄. TLC system 2: Si gel, hexane: EtOAc / 9:1. TLC system 3: Si gel, CHCl₃:MeOH:HOAc / 8:2:0.5.

Purifications of the intermediates and final products were

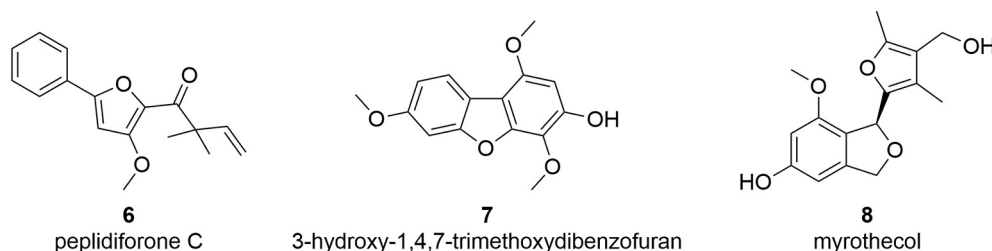


Fig. 4. Chemical structures of natural products bearing phenyl-furan moieties, including peplidiforone C (6), 3-hydroxy-1,4,7-trimethoxydibenzofuran (7) and myrothecol (8) isolated from *Hypericum peplidifolium*, *Hypericum revolutum* ssp. and the fungus *Myrothecium* sp. GS-17, respectively.

conducted using an Interchim puriflash 4100 automatic purification machine. The crude compounds were injected into the system at a rate of 250 mL/min at 100 bar, using a 15 μ g silica, high-capacity columns.

All chemicals were purchased from Fisher Scientific, Sigma Aldrich and Merck Chemicals, and used without further purification.

3.1.2. Plant material

The *A. gummifera* rhizomes were identified through morphological analysis and collected on 19/03/2014 by Professor Francesco Maria Raimondo in Piana degli Albanesi, Loc. Maganoce, Palermo, Italy. Fresh rhizomes were shipped to UK and stored at -20°C .

3.1.3. Extraction and isolation

The *A. gummifera* rhizomes (439.64 g) were cut and blended to obtain a coarse powder, which was initially extracted (1 L/100 g) with acetone at 40°C for 2 h. The powdered roots were further extracted three times with an acetone:water (9:1) mixture (0.5 L/100 g) for 1 h each time. The four fractions were combined and evaporated under reduced pressure at a temperature below 40°C and condensed to 350 mL to give a dark-green oily residue, which was extracted with CHCl_3 (3×500 mL and 1×150 mL). The organic fractions were combined, filtered and evaporated at a reduced pressure to yield a brown oil (1.052 g), which was analysed to identify novel secondary metabolites.

3.1.3.1. Aqueous phase. The aqueous phase was concentrated to 500 mL at a reduced pressure at 40°C . A 20% aqueous solution of KCl (100 mL) was then added and a white precipitate, containing ATR, was filtered off and washed with cold water. The mother solution was kept a $0-4^{\circ}\text{C}$ overnight and more precipitate was formed. A total of 3.750 g of ATR 1 (0.85% yield) was recovered as a pale yellow solid, which was characterised by TLC ($R_f = 0.45$, purple, TLC system 1), LC-MS and $^1\text{H-NMR}$ spectroscopy. NMR spectral data were identical to previously analysed pure sample of ATR [17].

3.1.3.2. Organic solvent phase. The brown oil (1.052 g) was subjected to column chromatography over silica gel column with 0.5% MeOH in CHCl_3 to yield four fractions Fr. 1–4. Further purification of Fr 3 (46 mg) by preparative TLC (hexane:ethyl acetate - 9.5:0.5) allowed the isolation of pure 5 ($R_f = 0.29$, TLC system 2) as a pale yellow solid (3.5 mg).

3.1.4. Synthetic procedures

The physical and spectral data of compounds 1 were identical to those of pure samples previously isolated [17]. Compounds 2, 3 and 4 were prepared according to published procedures and their physical and spectroscopic data agreed with those reported in the literature [22].

Hydrolysis of ATR 1 to give atractyligenin 2. A solution of 500 mg of ATR (1) in 20% KOH in water (5 mL) was heated under reflux for 8 h. The solution was acidified with 20% HCl to $\text{pH} = 4$ at 0°C and the solvent evaporated under reduced pressure. The residue was purified by column flash chromatography over silica gel to give 187 mg (94% yield) of pure genin 2 ($R_f = 0.17$, TLC system 3) as a white solid. $[\alpha]_{20}^{\text{D}}-145$ (c 0.2, EtOH); ESIMS m/z (negative mode) 319 $[\text{M} - \text{H}]^-$.

General procedure for the synthesis of 3 and 4. To a solution of the desired compound (1 and 2) in a mixture of dry THF:MeOH / 5:3 (8 mL) was added TMSCHN_2 (1.5 equiv., 2 M in Et_2O) dissolved in dry MeOH (0.5 mL) at -40°C over 5 min. The reaction was allowed to reach room temperature and react for 4 h. The reaction mixture was filtered off through a silica pad and eluted with MeOH (15 mL) to give the desired product (3 and 4) that were judged pure by analysis of TLC, LC-MS, NMR, and IR spectra.

Atractyliside methyl ester (ATR-OMe) 3. Starting from 220 mg (0.303 mmol) of ATR (1), 197 mg (0.266) of 3 were obtained as a pale yellow solid (88% yield).

Atractyligenin methyl ester (genin-OMe) 4. Starting from 130 mg (0.406 mmol) of atractyligenin 2, 110 mg (0.329 mmol) of 4 were obtained as a white amorphous solid (81% yield). $R_f = 0.32$, purple, (TLC = 20% MeOH in CHCl_3 , stained with vanillin- H_2SO_4 solution). $[\alpha]_{20}^{\text{D}}-141$ (c 0.2, EtOH); ESIMS m/z (negative mode) 333 $[\text{M} - \text{H}]^-$. $^1\text{H-NMR}$ (600 MHz, MeOD) δ_{H} 5.18 (bs, 1H, H-17 α), 5.08 (bs, 1H, H-17 α), 3.97 (bs, 1H, H-2), 3.77 (bs, 1H, H-15), 3.65 (s, 3H, OCH_3), 2.71 (bs, 1H, H-13), 2.65–2.49 (m, 1H, H-4).

2-furyl(phenyl)methanol 5. IR ν_{max} (golden gate) cm^{-1} 3366, 2925, 2853, 1641, 1455, 1210, 1032, 969, 749. $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ (DMSO- d_6) please refer to Table 1; ESIMS m/z (positive mode) 175.01 $[\text{M} + \text{H}]^+$; HRMS calculated for $\text{C}_{11}\text{H}_{10}\text{O}_2$ 174.0681, found 175.0755 $[\text{M} + \text{H}]^+$. Chiral HPLC (Chiralcel OD, 2% isopropanol in hexane, flow rate = 1 mL min^{-1} , $\lambda = 230$ nm): $R_{\text{tR}} = 25.4$ min for enantiomer S and $R_{\text{tR}} = 30.4$ min for enantiomer R. The configuration of the enantiomers was assigned by comparing the HPLC elution order with previously published data [24].

4. Biology

4.1. Parasite lines and cell culture microorganisms and growth conditions

Luciferase-expressing transgenic *Leishmania donovani* LV82 promastigotes and amastigotes, within macrophage derived from bone marrow, were cultured in M199 and RPMI1640 medium (Gibco $\text{\textcircled{C}}$), supplemented with 20% v/v heat-inactivated foetal calf serum (FCS) (Gibco $\text{\textcircled{C}}$), and 1% penicillin streptomycin (Pen/Strep; Gibco $\text{\textcircled{C}}$) at 25°C [29].

Amastigotes of these lines were cultured within macrophage derived from bone marrow. The bone marrow derived macrophage were extracted as non-differentiated monocytes from bone marrow of the femur, cultured in Dulbecco's modified eagle medium (DMEM, Invitrogen) supplemented with 10% v/v FCS, 1% pen/strep, 30% v/v cell supernatant and 1% L-glutamine for 7–10 days, after which they were transferred to RPMI 1640 medium supplemented with 10% FCS (v/v), 1% pen/strep (v/v) and L-glutamine. Viability of macrophage was determined with the Trypan blue assay [29] prior to use for in vitro luminescence assay of intramacrophage amastigotes.

4.2. In vitro Luminescence assay

The previously reported luciferase assay [30] was used to estimate the cytotoxic effect of the compounds used in this study against transgenic *L. donovani* promastigotes and amastigotes. For promastigotes, the assay was carried in a 96 well format and 1×10^6 log phase promastigotes, with or without drugs at a concentration determined by the experimental protocol in a final volume of 100 μL , were incubated for 72 h at 26°C . The cell viability was estimated by adding 0.5 $\mu\text{g/mL}$ of luciferine to the assay plate and the luminescence was measured with the IVIS bio-imaging platform. Luminescence of drug-treated wells relative to their no drug control expressed as a percentage was used as a measure of cell viability.

For amastigotes, 10^5 activated bone derived macrophages were incubated for 24 h in a 96 well plate at 37°C , 5% CO_2 to adhere to the bottom of the well. Subsequently, the macrophages were infected with promastigotes at a ratio promastigotes:macrophage of 20:1, and incubated further for 24 h to allow phagocytosis of the parasites. After the extracellular parasites were washed out, the drugs were added to the wells at concentrations determined by the experimental protocol for 72 h, and the viability was estimated as described for promastigotes. The data presented are derived from independent experiments done in triplicate. Dose response curves were used to estimate the IC_{50} .

5. Macrophage cellular toxicity

The IC_{50} of macrophage for the compounds used in this study was

estimated using the alamar blue assay. Briefly, macrophages (10 [5]) were incubated with different concentrations on drugs determined by the experimental protocol for 72 h. Viability was estimated by adding alamar blue and incubating the macrophages further for 6 h. Absorbance readings at 595 nm and 570 nm were used to estimate cell viability as described for promastigotes. Dose response curves were used to estimate the IC₅₀. The selectivity indices were estimated by expressing the IC₅₀ of macrophages to IC₅₀ on *L. donovani* amastigotes intracellularly within macrophage or IC₅₀ of promastigotes.

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Declaration of Competing Interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Appendix A. Supplementary data

¹H- and ¹³C-NMR spectra. This material is available free of charge via the Internet at <https://doi.org/10.1016/j.fitote.2019.104420>

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