

Assessment of the effect of Cellets' particle size on the flow in a Wurster fluid-bed coater via powder rheology

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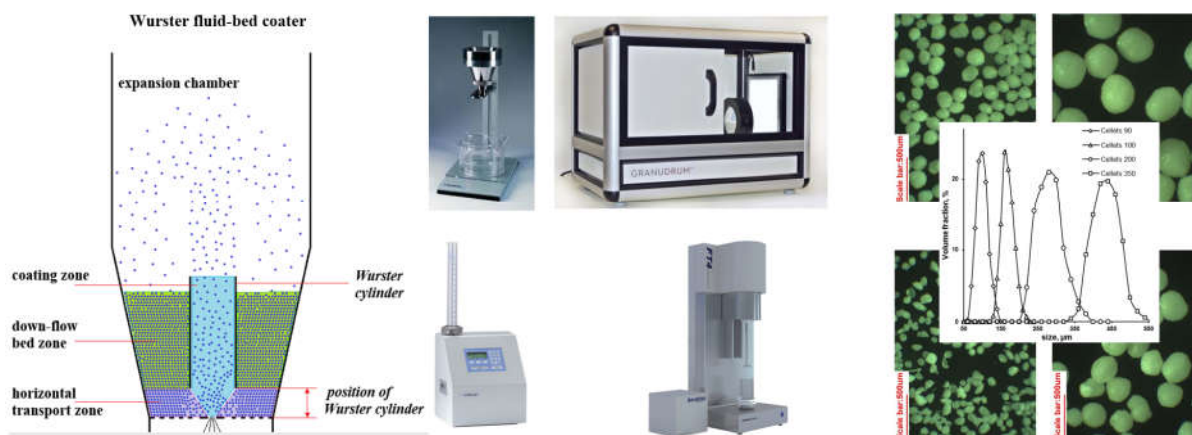
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Graphical abstract

Processability = function of bulk powder behaviour = function of Cellets' size



19

20

21 **Abstract**

22 The main objective of this study was to investigate the effect of microcrystalline cellulose
23 (MCC) spheres (Cellets) size effect on the powder properties to assess the possible impact of
24 the Cellets' size on the powder behaviour in the down-flow bed and horizontal transport zone
25 when a Wurster fluid-bed coater is used.

26 The particle size distribution of Cellets was determined using optical digital microscopy.
27 Standard pharmacopoeia methods as bulk/tapped density and flow rate measured with
28 gravitational funnel method as well as investigation of dynamic angle of repose and dynamic
29 cohesivity index with a rotating drum tester (GranuDrum) and conditioned bulk density, basic
30 flowability energy, specific energy, pressure drop, permeability and compressibility by powder
31 rheometer (FT4 Powder Rheometer[®]), were employed to characterise the powder's properties
32 of Cellets 90, 100, 200 and 350 (D_{50} -size from 94 to 424 μm) alone or premixed with 0.5%
33 w/w magnesium stearate.

34 Specific powder rheology methods were proposed for characterisation of Cellets' behaviour in
35 down-flow bed, the horizontal transport and coating zone.

36 The level of Cellets' processability decreasing in the Wurster fluid-bed coater with decreasing
37 of Cellets size (D_{50}) from 425 to 94 μm was established with different powder rheology
38 methods.

39

40 **Key words:** Cellets; pellets; powder rheology; flowability; fluid-bed; coating; processability

41

42 **1. Introduction**

43 Microcrystalline cellulose (MCC) is one of the most widely used excipients in the formulation
44 of solid dosage forms. Spherical MCC particles, such as Cellets are becoming increasingly
45 popular as the inert core in the fluid-bed coating process [1, 2]. Coated pellets' particle size can
46 influence the solid dosage form's sedimentation rate [3], uniformity [4] and mouthfeel [5].
47 Depending on the specific final product, Cellets with different particle sizes could be used as
48 inert cores in the multi-layered pellets [1].

49 A fluid-bed coater with a Wurster cylinder is one of the most commonly used operations for
50 pellet coating and is a circulating process, but it doesn't contain a fluid-bed in the traditional
51 sense [6]. A Wurster cylinder-containing (two-compartment) design of processing column was
52 intended to reduce inter-particle weight gain variability (compared to a one-compartment
53 design) [7].

54 A few different zones are identified in fluid-bed coater with a Wurster cylinder: the coating
55 zone, the expansion chamber, the down-flow bed zone, and the horizontal transport zone
56 (Figure 1). Cores come into the coating zone via a high velocity air flow, rising and then
57 additionally achieving atomisation air flow and droplets are sprayed from a nozzle. After
58 passing through the Wurster cylinder, cores settle in the expansion chamber on the surface of
59 the down-flow bed and gradually move downwards. Between the bottom part of down-flow
60 bed zone and coating zone is the horizontal transport zone. The adjustable position of the
61 Wurster cylinder (the distance between the air distribution disk and the bottom edge of the
62 Wurster cylinder) controls powder flow rate per unit area. The horizontal transport of powder
63 could be considered similar to pneumatic conveying. The fluidised-bed region is only present
64 in the horizontal transport zone and reduces the friction between particles, which helps to
65 convey them to the coating zone. [6, 8]

66 The particle size, density, surface area, smoothness of the surface and particle-to-particle
67 interaction are amongst the most important powder properties which affects the process
68 performance in different zones. The ideal process settings should provide enough powder flow
69 in the down-flow bed zone and horizontal transport zone to ensure that within the coating zone
70 all droplets will coat the powder particles. Additionally, air flow and atomisation air flow
71 should provide a reasonable throughput of particles through the Wurster cylinder [9].

72 Recently, a new technological platform has been developed overcoming the particle size
73 limitations during sustained release microparticle coating with an aqueous polymer dispersion

74 using Wurster fluid bed coater. Periodically adding a small quantity of dry powder glidant
75 during coating overcame this issue and substantially improved product yield due to the
76 improvement of the powder's flow [10].

77 The purpose of current work is to investigate the effect of Cellets' particle size on powder
78 properties and their possible effect on the processing in a fluid-bed coater with a Wurster
79 cylinder. The behaviour in the down-flow bed and horizontal transport zone is a point of
80 interest. As an additional task, to assess the effect of magnesium stearate at a concentration
81 level of 0.5 % w/w on the powder properties and to compare Cellets with Avicel PH-102 (MCC
82 powder).

83 **2. Materials and methods**

84 **2.1. Materials**

85 Inert spherical particles of MCC (Cellets 90, 100, 200 and 350; manufactured by IPC, Process-
86 Center GmbH & Co KG, Germany) were kindly provided by HARKE Pharma GmbH
87 (Germany). The MCC powder (Avicel PH-102) was supplied as a gift by IMCD UK Ltd.
88 (United Kingdom). Magnesium stearate (MgSt) was gifted by excipient manufacturer Sudeep
89 Pharma Pvt. Ltd. (India).

90 **2.2. Methods**

91 *Moisture content*

92 The moisture content of materials was determined via loss on drying (LOD) of approximately
93 1 g of pellets samples after equilibration at 105°C drying temperature (moisture analyser
94 MB45, Ohaus Corp., Switzerland).

95 *Particle size distribution (PSD)*

96 The PSD of Avicel PH-102 and MgSt was measured using laser diffraction (LD; ASPIROS
97 dosing, RODOS dispersing at 2 bar, and HELOS / KF LD-detector; Sympatec GmbH,
98 Germany). The R5 lens (4.5-875 µm detection range) was used. The average particle diameter
99 was calculated using largest and smallest 2D-dimension of every Cellet (more than 1000
100 particles in total) measured using a digital microscope VHX-600 Series (Keyence Corp.,
101 Osaka, Japan). The D_{10} , D_{50} and D_{90} were extracted from the cumulative volume fraction which
102 was generated from the raw data. The Span was used as an indicator of particle size distribution
103 and calculated using the following equation:

104
$$Span = \frac{D_{90} - D_{10}}{D_{50}} \quad \text{Equation 1}$$

105 ***Determination of particles density***

106 The density of microparticles was determined using the measurement of particle diameter
107 (n=30) with a light microscope (and following calculation of volume), the direct measurement
108 of particle sedimentation in water at room temperature and calculation of the particle density
109 in accordance with the modified Stokes' Law equation:

110
$$\rho = \frac{V_t \times 18\eta}{D^2 \times g} + \rho_0; \quad \text{Equation 2}$$

111 where: ρ and ρ_0 – density of spherical particle and water; η – viscosity of water; V_t –
112 sedimentation velocity; D – diameter of spherical particle; g – gravitational acceleration.

113 ***Theoretical calculation of one spherical particle mass and apparent specific surface area***
114 ***(SSA_{APP})***

115 The theoretical calculation of the mass of a single particle and SSA_{APP} of Cellets were done
116 using the assumption that particles have an ideal spherical shape and non-hollow, non-porous
117 structure. For every particle size, the volume (V) and mass (m) was calculated using the
118 following equations:

119
$$V = \frac{4}{3}\pi\left(\frac{D_{50}}{2}\right)^3; \quad \text{Equation 3}$$

120
$$m = \rho V; \quad \text{Equation 4}$$

121 using the true density (ρ) of Cellets 350 (1.44 g/cm³ with standard deviation ± 0.03 g/cm³; in
122 accordance to Equation 2).

123 The surface area (A) of a single spherical particle was calculated using:

124
$$A = 4\pi\left(\frac{D_{50}}{2}\right)^2; \quad \text{Equation 5}$$

125 while SSA_{APP} was calculated by taking into the account the number of spherical particles in
126 1 g (N) using:

127
$$SSA_{APP} = A N; \quad \text{Equation 6}$$

128 ***Specific surface area determination (SSA_{BET})***

129 SSA_{BET} analysis was conducted using an Inverse Gas Chromatography Surface Energy
130 Analyser (iGC-SEA, Surface Measurement Systems Ltd, UK). Approximately 1.8-2 g of each

131 sample was packed into silanised iGC glass columns (internal diameter 4 mm). Prior to any
132 measurements, the columns were conditioned using helium carrier gas at 10 scc/min for 2 h at
133 30°C and 0% RH. Methane gas was injected at the start and the end of the experiments for the
134 dead volume calculation. SSA_{BET} was calculated via Brunauer-Emmett-Teller (BET) theory,
135 based on the n-octane adsorption isotherm data (Peak Max parameter) [11, 12]. Every sample
136 was analysed in triplicate. A correlation coefficient ($R^2 > 0.999$) assured linearity within the
137 BET range.

138 ***Mixing with magnesium stearate***

139 Before mixing, MgSt was passed through a sieve with 0.5 mm mesh. Then, MgSt was added
140 to Cellets or Avicel PH-102 and mixed in a Turbula mixer (Turbula T2F, Willy A.
141 Bachofen AG, Switzerland) for a 5 min.

142 ***Mass flow rate determination with gravitational funnel method***

143 A stainless-steel frustum cone funnel was fixed in a strictly vertical position (Flowability Tester
144 model BEP2; Copley Scientific Ltd., England). 50 g samples were weighed and introduced
145 carefully into the dry funnel with 5 mm diameter orifice. During filling, the bottom opening of
146 the funnel was closed. Once the funnel was opened the time taken for the 50 g of powder within
147 the funnel to flow out was measured (n=6). The mass flow rate was calculated as mass per
148 time, expressed in g/s. [13]

149 ***Bulk and tapped density testing (pharmacopoeia method)***

150 To investigate the bulk and tapped density of microparticles alone and with a glidant, the
151 Tapped Density Tester (Copley Scientific JV1000; Copley Scientific Ltd, England) was used.
152 The volume was visually recorded every 3 taps until 33, then at 66, 100, 1000 and 2000 taps.
153 The bulk and tapped density of samples were investigated using a 50 ml graduated volumetric
154 cylinder with 22 mm internal diameter. All measurements were made in triplicate. Bulk and
155 tapped densities (ρ_{bulk} and ρ_{tapped} , respectively) as well as Hausner ratio (HR) were calculated
156 using the following equations:

$$157 \quad \rho_{bulk} = \frac{\text{Powder mass}}{\text{Initial volume}}; \quad \text{Equation 7}$$

$$158 \quad \rho_{tapped} = \frac{\text{Powder mass}}{\text{Volume after 2000 taps}}; \quad \text{Equation 8}$$

$$159 \quad HR = \frac{\rho_{tapped}}{\rho_{bulk}}; \quad \text{Equation 9}$$

160 ***The dynamic cohesive index and dynamic angle of repose determination***

161 The dynamic cohesive index and dynamic angle of repose were measured with a Rotating
162 Drum tester (GranuDrum, GranuTools sprl, Belgium). Approx. 50-60 ml of microparticles
163 were placed into a stainless-steel cylinder (internal diameter 84 mm, length 20 mm and approx.
164 internal volume 111 ml) with round glass side walls that have an anti-sticking internal surface.
165 The cylinder was installed horizontally, rotated around its axis at an angular velocity (2-
166 50 rpm) starting off the flowing motion of the microparticle sample being tested. The rotated
167 drum was backlit and for every rotation speed 50 images of the drum separated by 0.5 s were
168 recorded with a CCD camera. Obtained images were used as raw data and the position of
169 air/powder interface was determined automatically due to the difference in the light intensity.
170 Then, the average air/powder interface position and the fluctuation around this average position
171 as standard deviation (σ_f) were automatically computed using 50 images at every rotation
172 speed. All measurements were made in triplicate. The angle of the average air/powder interface
173 position regarding the horizontal axis was considered as the dynamic angle of repose. The
174 dynamic cohesive index (σ_f , expressed in %) is the deviation from average steady flow and can
175 be used for quantification of cohesion between the microparticles. The fluctuation of
176 air/powder interface and consequently the dynamic cohesive index increases with increasing
177 of particle cohesivity. [14]

178 ***Conditioned Bulk Density, Basic Flowability Energy and Specific Energy*** measurements
179 were performed using the FT4 Powder Rheometer[®] (Freeman Technology Ltd, England). All
180 samples for dynamic tests were conditioned in the test-vessel using the instrument's
181 "conditioning" methodology: the twisted blade's action gently disturbs the microparticle bed
182 and creates a uniform, lightly packed test sample that can be readily reproduced. After the
183 conditioning cycle, the Conditioned Bulk Density (g/ml) of the conditioned microparticle bed
184 in the 25 ml cylindric split borosilicate test vessel (with internal diameter 25 mm) was
185 recorded. To determine the Basic Flowability Energy (BFE, mJ) the conditioned microparticle
186 bed was consolidated by a bulldozing blade (with external rotational diameter 23.5 mm) action
187 that forces the microparticles downwards towards the bottom of the test vessel (where the base
188 of the vessel is limits the powder's movement) at a constant flow rate (anticlockwise blade tip
189 speed of 100 mm/s). The BFE is a key flowability parameter and is calculated as the required
190 energy for the blade to pass through the sample in this downward movement and provides an
191 indication of the resistance to the blade's movement from the powder.

192 The Specific Energy (SE, mJ/g) was measured and calculated from the work done in moving
193 the same blade through the microparticle bed from the bottom of the vessel to the top (upward
194 clockwise motion of blade). This generates a gentle lifting and low stress flow of the
195 microparticles, measuring the level of mechanical interlocking, and inter-particle friction. All
196 measurements were made in triplicate.

197 ***Aeration test (Aerated Energy measurement)***

198 The presence or absence of air in a powder can greatly affect its flow properties. The addition
199 of air can occur naturally when powder is moved freely, e.g. when discharging powder from a
200 hopper. When a powder is aerated, the twisted blade encounters less resistance than in the non-
201 aerated state as it passes through the powder bed. The Aerated Energy is measured using the
202 same blade as is used during BFE measurements, and the blade follows the same downwards,
203 anticlockwise motion through the microparticle bed.

204 ***Bulk property measurements with powder rheology: Pressure Drop, Permeability,*** 205 ***Compressibility***

206 Bulk measurements were performed with FT4 Powder Rheometer (Freeman Technology Ltd,
207 England). A porous vented piston (with external diameter 24 mm) was used to apply increasing
208 levels of normal stress on 10 ml of microparticles in a cylindrical split borosilicate test vessel
209 (with internal diameter 25 mm). The automatically computed volume changes after applied
210 levels of normal stress were used for Compressibility (%) calculation with the following
211 equation:

$$212 \text{ Compressibility } \% = \frac{\text{Initial volume} - \text{Compressed volume}}{\text{Initial volume}} 100\%; \quad \text{Equation 10}$$

213 The pressure drop (ΔP , mbar) across the microparticle bed was recorded whilst the applied
214 normal stress was varied and the air flow through the pellets' bed was maintained at a constant
215 velocity. From ΔP , using Darcy's Law (Eq. 11), rearranging and dividing by the area the
216 Permeability (k , cm^2) can be calculated (Eq. 12):

$$217 Q = \frac{k A P_a - P_b}{\mu L}; \quad \text{Equation 11}$$

218 where: Q is Air volume per unit time (cm^3/s); A is cross-sectional area of powder bed (cm^2);
219 $(P_a - P_b) = \Delta P$ is pressure drop across powder bed (Pa); μ is air viscosity ($\text{Pa}\cdot\text{s}$); L is length of
220 powder bed (cm).

$$221 k = \frac{q \mu L}{\Delta P}; \quad \text{Equation 12}$$

222 where, q is flux, or air flow rate (cm/s) and μ is 1.74×10^{-7} (mbar·s) for air at sea level.
223 Due to the significant differences in the permeability of MCC samples, it was impossible to
224 obtain comparable permeability measurements at the same conditions. So, Cellets 90 and 200
225 were tested at 10 kPa normal stress and 2 mm/s air velocity while Cellets 100, 200, 350 and
226 Avicel PH-102 were tested at 10 kPa and 20 mm/s. To provide comparable data at 20 mm/s air
227 velocity a conversion calculation was used:

$$228 \quad \Delta P(\text{Cellets 90 @ 20mm/s}) = \frac{\Delta P(\text{Cellets 90 @ 2mm/s})}{\Delta P(\text{Cellets 200 @ 2mm/s})} \times \Delta P(\text{Cellets 200 @ 20mm/s}); \quad \text{Equation 13}$$

229 Note that this calculation assumes a linear relationship between Pressure Drop and Air
230 Velocity. All measurements were made in triplicate.

231 3. Results

232 The particle size of Cellets was measured with an optical digital microscope to avoid
233 speculation arising from traditional light scattering methods on the measured values. The
234 particle size (D_{50}) values of the Cellets 90, 100, 200 and 350 are 94, 163, 270 and 424 μm ,
235 correspondently (Figure 2, Table 1). Particle size distribution was characterised by narrow span
236 values suggesting a relatively uniform size distribution. Cellets' sphericity increased with
237 increasing particle size (Figure 3).

238 The increase in Cellets' size was accompanied with a decrease in the calculated SSA_{APP} (Table
239 1, Figure 4). Thus, an increase in mechanical interlocking with decrease in the size of the
240 Cellets was expected. The calculated SSA_{APP} values were much lower than the experimentally
241 determined SSA_{BET} (Table 1, Figure 4), because SSA_{BET} method is sensitive to the particle's
242 surface roughness. SSA_{APP} and SSA_{BET} of Cellets 90 and Cellets 100 both show a decrease in
243 surface area with increasing particle size. The difference between SSA_{BET} of Cellets 100 and
244 Cellets 200 was relatively low and can be attributed to the detection limits of the iGC [12].
245 Despite the relatively similar particle size of Cellets 90 and Avicel PH-102, the Avicel's span
246 and SSA_{BET} are approximately 4 and 3 times (respectively) larger than for Cellets 90 (Table
247 1).

248 The bulk and tapped density (Table 1, Figure 5) of the Cellets are approximately 2 times higher
249 than for Avicel PH-102. So, Avicel PH-102 powder had higher SSA_{BET} and was less efficiently
250 packed than Cellets. The bulk and tapped density of Cellets increased with increasing particle
251 size: Cellets 90 < Cellets 100 < Cellets 200 < Cellets 350. The densification kinetics of the
252 different grades of Cellets was approximately the same (Figure 5) with the fastest densification
253 kinetics observed during first 3-12 taps. From a practical point of view, this suggests that even

254 a few periodical vibrations/oscillations (such as impacts with a rubber hammer) during the
255 coating process could significantly change the density of Cellets in the down-flow bed.

256 Mixing Cellets with MgSt improved the packing of the particles and decreased the necessary
257 volume for the same number of particles. Thereby increasing the tapped density of Cellets
258 (Figure 5) without significantly impacting the densification kinetics. The tapped density
259 difference between Cellets 90, Cellets 100 and Cellets 200 mixed with MgSt was very low
260 compared to the significant difference without MgSt. The increase in the Cellets' tapped
261 density after adding of MgSt is likely to be due to a decrease in interparticle friction because
262 of the MgSt's lubrication properties.

263 The additional densification of Cellets was investigated by applying a normal force. At the
264 same applied force, the compressibility of Cellets decreases as particle size increases (*Figure*
265 *6*, Table 2). Avicel PH-102 also demonstrated a decrease in compressibility upon the addition
266 of MgSt. Interestingly, the compressibility of Cellets with MgSt was very similar irrespective
267 of the particle size. This is most likely due to a decrease in frictional forces arising from the
268 lubrication by MgSt.

269 The decrease in the pressure drop from Cellets 90 to Cellets 350 (*Figure 7 A*, Table 2) and
270 corresponding increase in permeability (Figure 7 B, Table 2), alongside the decrease in
271 compressibility indicates that there is a worsening particle packing efficiency (increase in voids
272 between particles) as the particle size increases from Cellets 90 to Cellets 350. Mixing the
273 Cellets with 0.5% MgSt didn't change the trends in pressure drop and permeability but
274 decreased the compressibility, indicating that there was a tighter, more efficient particle
275 packing (Figure 7 A and B, Table 2).

276 The gravitational funnel method suggests an absence of correlation between the Cellets flow
277 properties and their size (Figure 8 A) or SSA_{APP} (Figure 8 B). Notably, in contrast to the
278 similarly sized Cellets 90, the Avicel PH-102 did not flow through the funnel with 5 mm-
279 diameter-opening.

280 The dynamic angle of repose measured with a rotating drum tester gives information regarding
281 the dynamic ability of powders to flow. This could be useful for predicting, to some extent the
282 powders ability to be conveyed or gravitationally discharged. The measurement of dynamic
283 angle of response showed differences between the relatively poor flowing Avicel PH-102 and
284 better flowing Cellets. However, the difference between different grades of Cellets wasn't clear
285 (Figure 9 A).

286 SE measurements, using the FT4 powder rheometer, gives information regarding the level of
287 interlocking and friction between powder particles, which will also inform on how a powder
288 may flow under gravity, i.e. its resistance to flow in an unconstrained environment. SE values
289 of Cellets without MgSt increased with respect to decreasing particle size (Figure 9 B, Table
290 3). The increasing of Cellets sphericity (Figure 3) and decreasing of SSA (Figure 4) as well as
291 decreasing of Cellets' number (Cellets/g) with increasing particle size are suggestive of a
292 decrease in the level of mechanical locking and surface of interparticle friction. So, it can
293 explain the SE decreasing with particle size increase. The addition of MgSt decreased the SE
294 for all Cellets' sizes and decreased the difference between them. This information is in line
295 with the previously shown densification and compressibility measurements, where it was
296 concluded that the addition of MgSt resulted in an increase in the packing efficiency. The
297 greatest SE value was observed for Avicel PH-102, indicating high interparticle friction due to
298 high surface area and roughness, irregular shape and mechanical interlocking, which also
299 decreased with the addition of MgSt (Figure 9 B, Table 3).

300 The BFE is a measure of the resistance required by a twisted blade to displace a powder during
301 non-gravitational, forced flow, i.e. its resistance to flow in a constrained environment, for
302 example, feeding via flowing in a conveyer [15] or mixing [16]. Cellets without MgSt
303 demonstrated an almost linear increase in BFE with decreasing size (Figure 10 A). While
304 Cellets mixed with MgSt showed a decrease in BFE for every Cellets' size, the addition of
305 MgSt almost neutralised the BFE difference between the various particle sizes. As with the
306 Cellets, the BFE of Avicel PH-102 with MgSt was lower than for Avicel PH-102 without
307 MgSt. The BFE of Avicel PH-102 with and without MgSt was lower than the lowest value of
308 all tested Cellets with and without MgSt (Figure 10 A), that is apparently this is due to the
309 relatively low density of Avicel PH-102.

310 The Aeration test measures changes in the flow properties due to the introduction of air into
311 the sample. With increasing levels of air velocity (Figure 10 B) there is a decrease in the aerated
312 energy for all the samples, indicating that the addition of air has some influence during
313 processing on all the samples, with and without MgSt. Cellets 90 and Avicel PH-102 both with
314 and without MgSt had similar responses to being aerated and were completely fluidised at
315 8 mm/s air velocity. Cellets 100 with and without MgSt fluidised at 16 mm/s air velocity.
316 Cellets 200 with MgSt were completely fluidised at 40 mm/s while Cellets 200 without MgSt
317 weren't. The addition of MgSt on the Aerated Energy has had a similar impact on both the
318 Cellets 200 and 350, where the Aerated Energy at 40 mm/s air velocity has decreased by 26.5
319 and 23.5 mJ respectively (Figure 10 C). The Aeration test can determine the powder's Flow

320 Energy with increasing air velocity, and is a measure of the powder's cohesivity, as well as the
321 minimum fluidisation velocity. The Aeration test clearly showed that the minimum fluidisation
322 velocity of Cellets' sizes increased with increasing particle size, and the cohesivity of Cellets
323 and Avicel PH-102 decreased with the addition of MgSt.

324 For the characterisation of cohesivity in another way, Cellets' dynamic cohesive index was
325 determined with the rotating drum tester. The measurement of dynamic cohesive index showed
326 a decrease in cohesivity in the following sequence: Avicel PH-102 > Cellets 90 > Cellets 100
327 > Cellets 200 \approx Cellets 350 (Figure 11). Comparing to the Aeration test (Figure 10 B), the
328 rotating drum tester has a limited extent to which it can determine the cohesivity differences
329 between Cellets 200 and Cellets 350.

330 **4. Discussion**

331 Usually, cores with particle size larger than 350 μm (e.g. Cellets 350 with D_{50} 424 μm) don't
332 display any process issues in the Wurster fluid-bed coater. So, considering their properties as
333 appropriate or non-problematic the effect of Cellets' size and addition of MgSt on the powder
334 rheology and possible effect on the coating process can be assessed. Avicel PH-102 has been
335 added to the experimental plan as it widely used in industry and has been used in several
336 investigations and can be used by readers to make connections with other studies. From a range
337 of powder flow measurement techniques, some methods are better explaining Cellets'
338 behaviour in a specific zone of fluid-bed coater's processing chamber while some of them are
339 applicable for several zones.

340 **Down-flow bed zone.** The difference in powder behaviour in the down-flow bed zone could
341 be explained using the basic flowability energy (BFE), dynamic cohesive index, densification
342 kinetics, compressibility, pressure drop and permeability measurements.

343 The BFE measurement can be applicable for the understanding of powder flow in a constricted
344 volume. In this situation, it can be used as indicator of the level of particle mobility and powder
345 rearrangement in the down flow bed zone. An almost linear dependence of the BFE decreasing
346 was shown for Cellets' size (D_{50}) increasing from 94 to 424 μm . The BFE is an indicator of
347 sum of different forces, the determination of dynamic cohesive index is elucidating and in some
348 extend representing mostly cohesivity. The trend of cohesive index decreasing with Cellets'
349 size increasing was the same as for BFE but was not able to differentiate between Cellets 200
350 and 350.

351 The results of densification kinetics illustrate the comparison of the interparticle friction and
352 are interesting especially since vibration devices or periodic tapping are used as powder flow
353 facilitation approach during Wurster fluid-bed processing. Compressibility testing, to some
354 extent, could be considered as a continuation of the densification kinetics test with additional
355 vertical axial normal stress. The results of densification kinetics and compressibility
356 measurement were in the agreement with BFE and dynamic cohesive index with an almost
357 linear dependence of a decreasing compressibility with increasing Cellets' size (D_{50}) from 94
358 to 424 μm .

359 In the addition to the other methods to characterise powder behaviour in the down-flow bed
360 zone, the pressure drop across the microparticle bed and permeability provide the information
361 regarding the resistance experienced by the Cellets' bed to passing through air and can explain
362 the difference between drying processes in the down-flow bed for Cellets with different particle
363 sizes. The pressure drop decreases, and permeability correspondingly increases as the Cellets'
364 size increases from 94 to 425 μm . This is suggestive of the drying ability of Cellets in the
365 down-flow bed is improves as Cellets' size increases.

366 **The horizontal transport zone.** To understand powder behaviour as it moves through the
367 horizontal transport zone, the mass flow rate through the gravitational funnel, dynamic angle
368 of repose, specific energy and dynamic cohesive index could be used as they are likely to
369 correlate with powder flow under gravity. In addition, the aeration test which is measures the
370 effect of air flow on the resistance of the FT4 blade's motion though the powder particles and
371 could relate to facilitation of particles rearranging.

372 Gravitational flow through a funnel did not give any clear correlations between the Cellets
373 particle size and the mass flow rate. While, the SE (which is usually used for characterisation
374 of powder flow in an unconfined process) decreased as the Cellets' size increased however
375 there was a minimal difference between Cellets 200 and 350 (as for dynamic cohesive index
376 mentioned above).

377 The aeration test provides information on the changing interparticle interactions with
378 increasing air velocity and the minimum fluidisation velocity for different powder samples.
379 The data from the Aeration test could be used to compare and possibly predict the efficiency
380 of powder passing from the horizontal transport zone to the coating zone. The aerated energy
381 (the energy required for the twisted blade to pass through the aerated Cellets') as expected
382 decrease as the air velocity increased for all of the powders tested, and at 40mm/s air velocity,

383 the aerated energy increased as the Cellets' size increased. In other words, as the particle size
384 decreases, or the air velocity increases, the Cellets' ability to be conveyed to coating zone is
385 likely to improve.

386 **The coating zone.** Within the coating zone, it is essential that particles are kept separated from
387 each other, to avoid coating-induced agglomeration. The dynamic cohesive index and
388 minimum fluidisation velocity could be correlated to a powder particle's ability to become
389 separated from each other in the coating zone. Measurements of the dynamic cohesive index
390 and minimum fluidisation velocity both assess the particle's cohesivity and in both cases, the
391 particle mass and corresponding gravitational force influence the result. For the dynamic
392 cohesive index measurement, the gravitational force is facilitating the flow and in the case of
393 minimum fluidisation velocity determination, all else being equal, heavier particles are likely
394 to require a higher air velocity.

395 Initially, both the cohesive index and minimum fluidisation velocity could be directly applied
396 to the powder's behaviour explanation in the coating zone. However, considering the specific
397 details of each measurement, the aeration test which passes air through the powder bed, is most
398 suitable once the effect of gravity on the particles has been counteracted and building a
399 correlation between the atomisation air shear force in the coating zone and determined
400 cohesivity.

401 **Effect of MgSt.** The addition of MgSt to Cellets of any particle size and Avicel PH-102 had a
402 lubricating effect, as would be expected from a glidant. The addition of MgSt at the 0.5% w/w
403 level increased the tapped density (i.e. increased packing efficiency), decreased BFE, SE, and
404 aerated energy/ minimum fluidisation velocity. The effect of MgSt on the gravitational flow
405 through a funnel did not give clear answer, but the applicability of this method to characterise
406 of Wurster fluid-bed process is questionable too (at least with 5 mm orifice diameter).
407 Generally, almost all flow characterisation methods suggested powder flow improvement in
408 the down-flow bed and horizontal transport zone after adding MgSt. A decrease in the pressure
409 drop and increase permeability after adding MgSt was observed, suggesting that the drying
410 process of lubricated Cellets in the down-flow bed is likely to be slower compared to
411 unlubricated Cellets. This could be correlated to a decrease in the interparticle friction, and
412 generally higher density layer with lower interparticle space in the lubricated Cellets' bed.

413

414 **5. Conclusions**

415 The level of Cellets' flowability decreasing in the down-flow bed and horizontal transport
416 zones of Wurster fluid-bed coater with decreasing of Cellets size (D_{50}) from 424 to 94 μm was
417 established with different powder rheology methods. In the horizontal transport zone, along
418 with decreasing of Cellets' size, the decreasing of Cellets flowability can be to some extent
419 compensated with decrease in the minimum fluidisation velocity and consequent increase in
420 Cellets conveying to the coating zone. While the decrease in Cellets' size decreases the powder
421 flowability in the down-flow and predetermining the powder supply to the horizontal powder
422 zone.

423 Specific powder rheology methods were proposed for characterisation of Cellets' behaviour in
424 down-flow bed, the horizontal transport and coating zone. The effect of a decrease in the
425 Cellets' particle size is likely to worsen Cellets' flowability and consequently processability in
426 a Wurster fluid-bed coater. Whereas the addition of MgSt (only added at 0.5% w/w) is likely
427 to improve the powder flow in a Wurster fluid-bed coater.

428

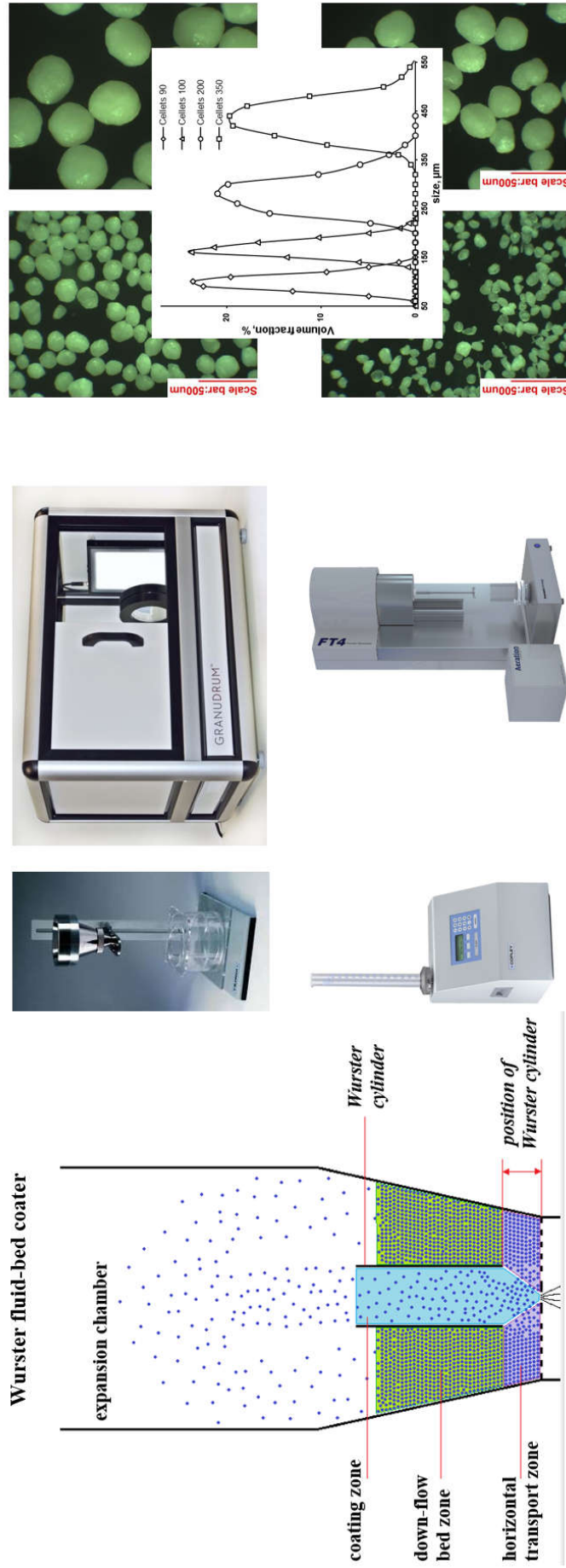
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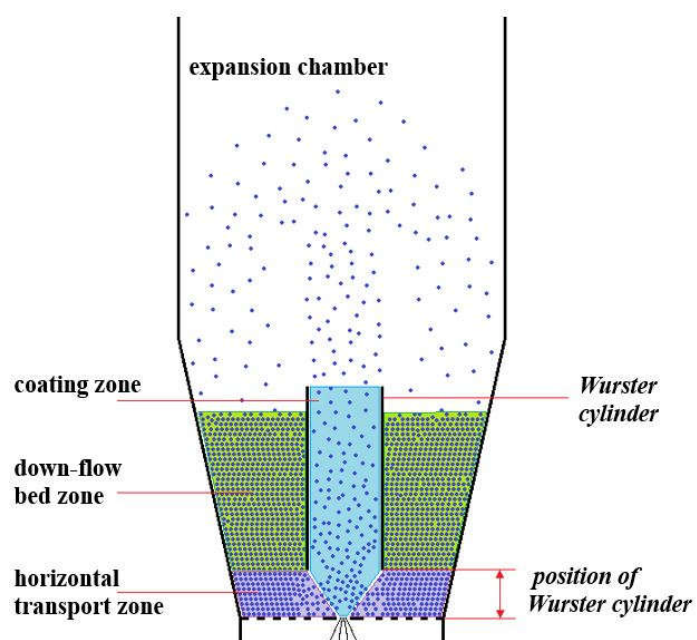
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476

Processability = function of bulk powder behaviour = function of Cellets' size



482



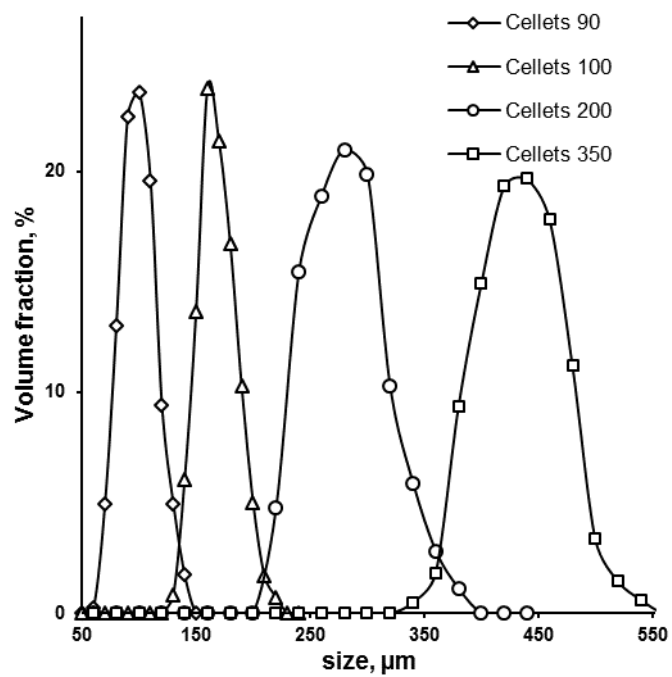
483

484 Figure 1. Schematic presentation of the working chamber of the Wurster fluid-bed coater.

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489 Figure 2. Particle size distribution of the Cellets used in this study.

490

491

492 Table 1. Particle size distribution (PSD), moisture content (LOD method), bulk and tapped density, Hausner ratio and mass flow rate.

Excipients	PSD ¹		SSA, cm ² /g		LOD, % (w/w)		Density ³ , g/ml				Hausner ratio		Mass flow rate ⁵ , g/s			
	D ₅₀ , μm	Span	SSA _{APP}	SSA _{BET}		Av	SD	bulk		tapped (2000 taps)		True ⁴	with-out MgSt	with 0.5% MgSt	Av	SD
				Av	SD			Av	SD	Av	SD					
1 MgSt	10.5 ²	2.93	-	-	-	3.4	0.2	0.275	0.007	0.387	0.005	1.092	1.41	-	no flow	-
2 Avicel PH-102	115 ²	1.85	-	11033	72	5.2	0.2	0.351	0.002	0.412	0.001		1.21	1.08	no flow	-
3 Cellets 90	94	0.44	433	3274	23	4.9	0.1	0.750	0.004	0.796	0.003		1.09	1.07	1.76	0.09
4 Cellets 100	163	0.27	249	906	12	4.9	0.1	0.811	0.005	0.855	0.004	1.460	1.09	1.07	2.06	0.01
5 Cellets 200	270	0.34	152	1133	2	4.1	0.1	0.809	0.003	0.875	0.003	-1.668	1.11	1.06	1.89	0.01
6 Cellets 350	424	0.22	97	-	-	4.0	0.2	0.833	0.005	0.906	0.004		1.10	1.09	1.83	0.01

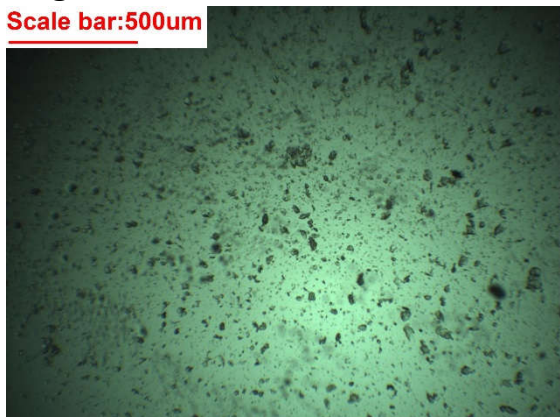
493 ¹ – digital microscopy; ² – measured by laser diffraction method; ³ – tapped density test; ⁴ – True density [17]; ⁵ – gravitational funnel method;

494

495

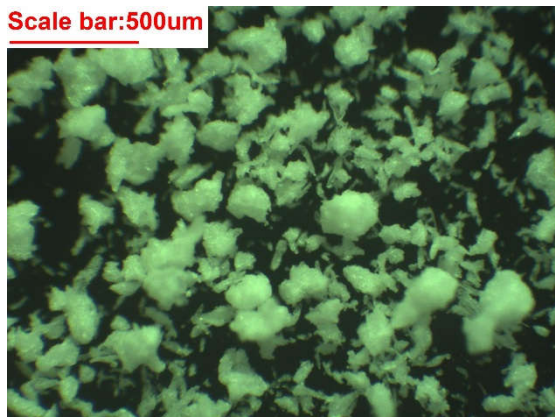
A) Magnesium stearate

Scale bar:500um



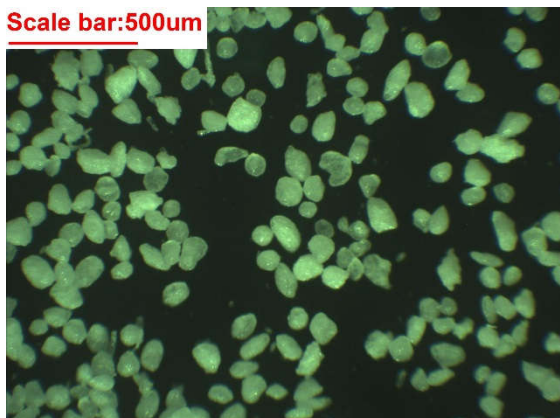
B) Avicel PH-102

Scale bar:500um



C) Cellets 90

Scale bar:500um



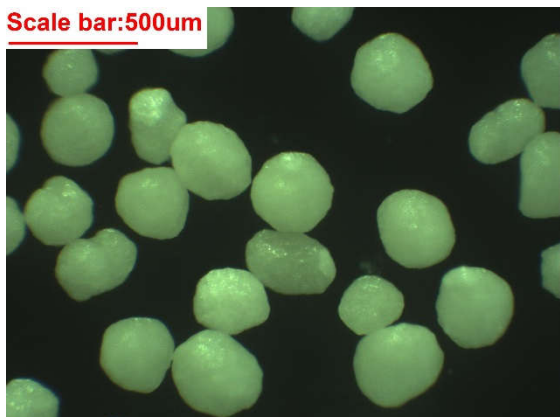
D) Cellets 100

Scale bar:500um



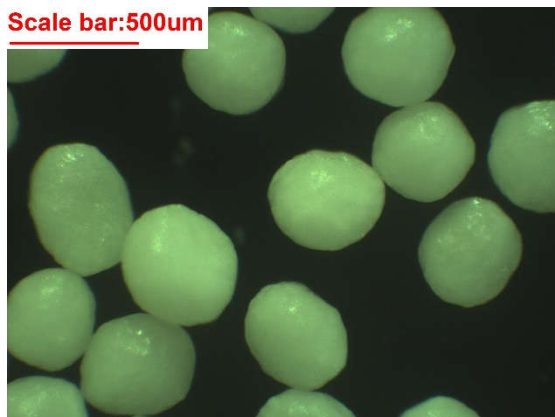
E) Cellets 200

Scale bar:500um



F) Cellets 350

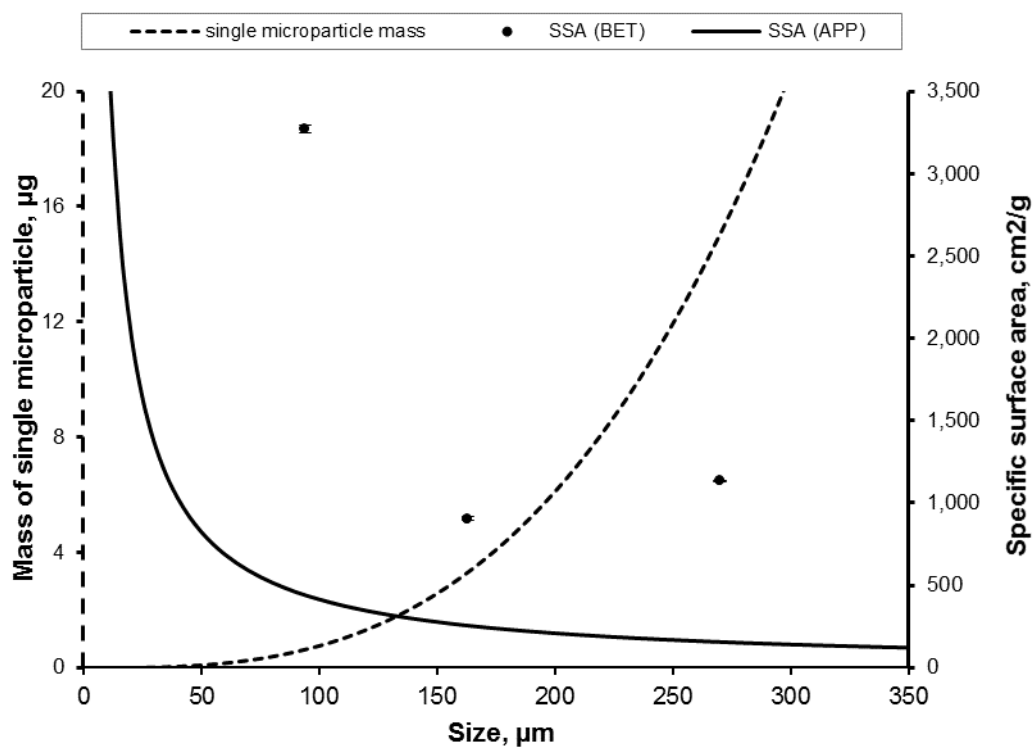
Scale bar:500um



496 Figure 3. Optical microscopy images of excipients: **A)** magnesium stearate (in liquid paraffin);
497 **B)** Avicel PH-102; **C-F)** Cellets 90, 100, 200 and 350, respectively.

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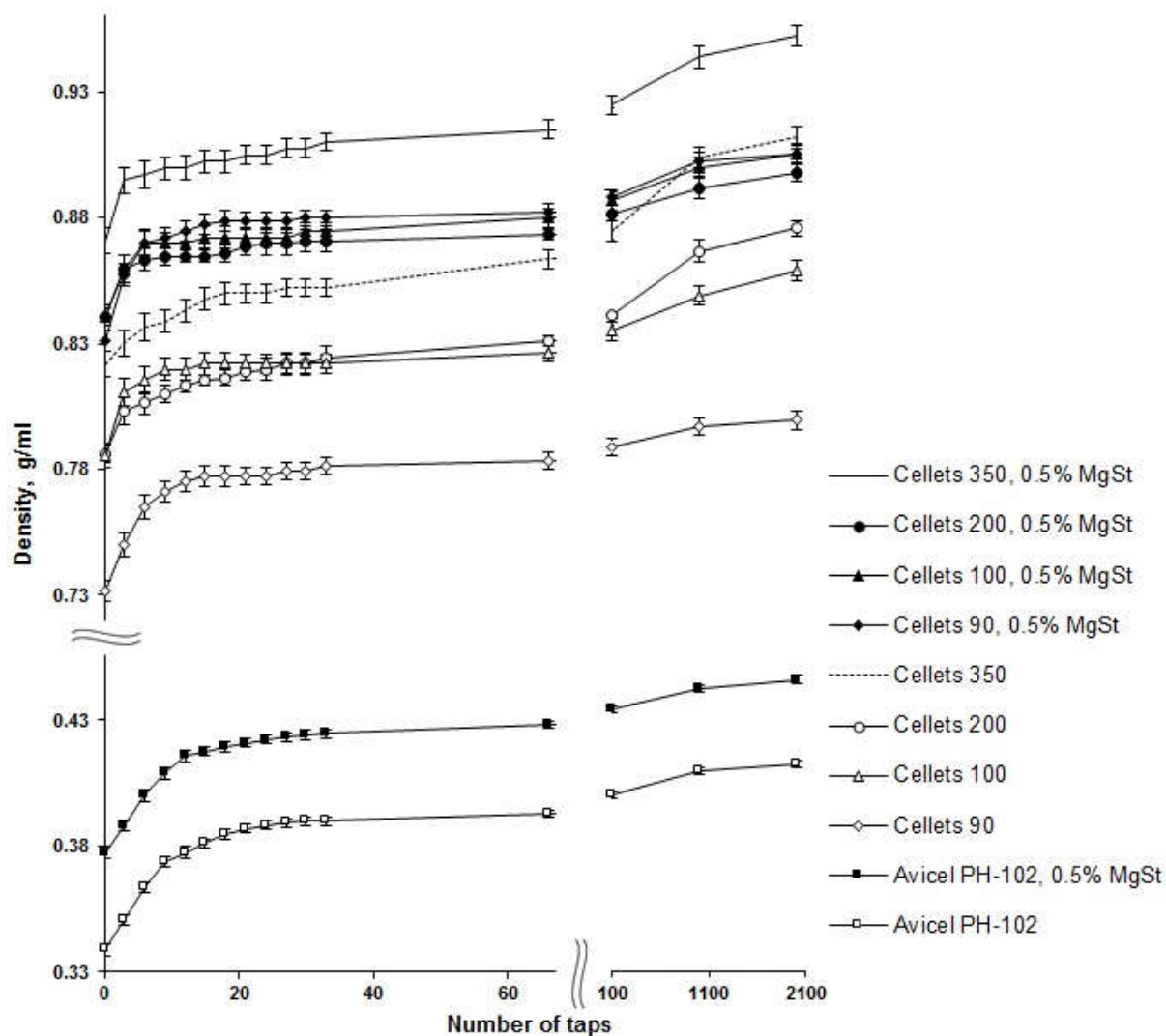


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501 Figure 4. The effect of particle size (D_{50}) on the calculated single microparticle mass and
502 apparent specific surface area (SSA_{APP}) in comparison to the experimentally measured
503 SSA_{BET}

504

505

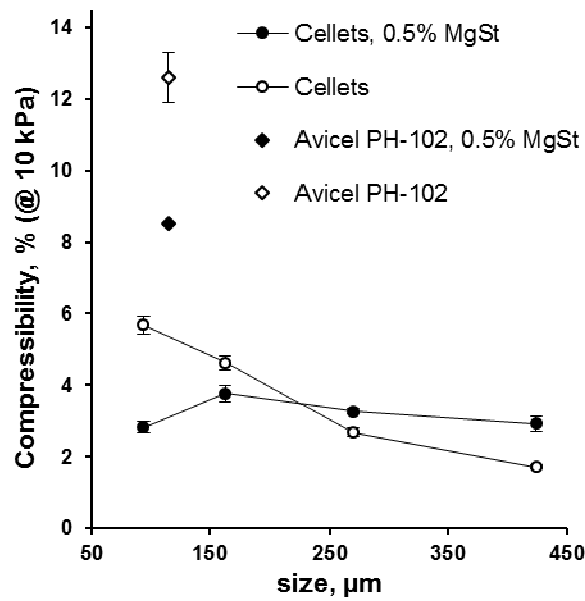


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507 Figure 5. Densification kinetics of Cellets and Avicel PH-102 without and with 0.5% MgSt.

508

509



510

511 Figure 6. Compressibility at 10 kPa normal stress on Cellets with varying particle size and
 512 Avicel PH-102, with and without 0.5% MgSt.

513

514

515 Table 2. Bulk properties of Cellets measured with FT4 powder rheometer at 10 kPa normal
 516 stress

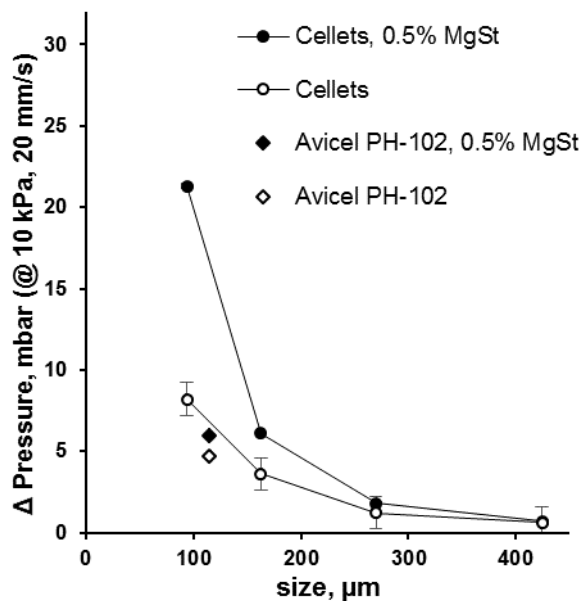
	Pressure Drop		Permeability		Compressibility	
	Av	SD	Av	SD	Av	SD
	mbar	mbar	$\times 10^{-9} \text{ cm}^2$	$\times 10^{-9} \text{ cm}^2$	%	%
Cellets 350	0.6	0.0	1014.0	16.2	1.7	0.0
Cellets 200	1.2	0.0	467.0	25.3	2.7	0.1
Cellets 100	3.6	0.1	178.0	3.0	4.6	0.2
Cellets 90*	8.2	0.1	69.9	8.1	5.7	0.3
Avicel PH102	4.7	0.1	123.0	3.3	12.6	0.7
Cellets 350, MgSt (0.5%)	0.7	0.0	960.0	9.8	2.9	0.2
Cellets 200, MgSt (0.5%)	1.8	0.0	365.0	8.4	3.3	0.1
Cellets 100, MgSt (0.5%)	6.1	0.0	106.0	1.1	3.8	0.2
Cellets 90, MgSt (0.5%) *	21.3	0.1	27.2	0.1	2.8	0.2
Avicel PH102, MgSt (0.5%)	6.0	0.7	100.0	0.7	8.5	0.0

517 * note that the Pressure Drop of Cellets 90 was calculated using the equation given in the
 518 method description

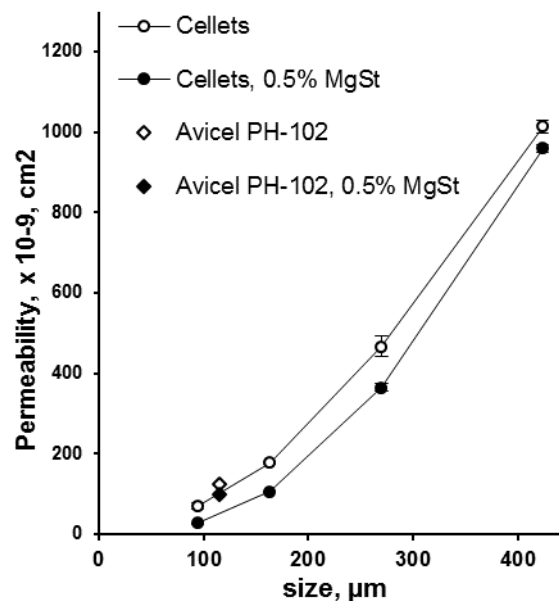
519

520

A



B

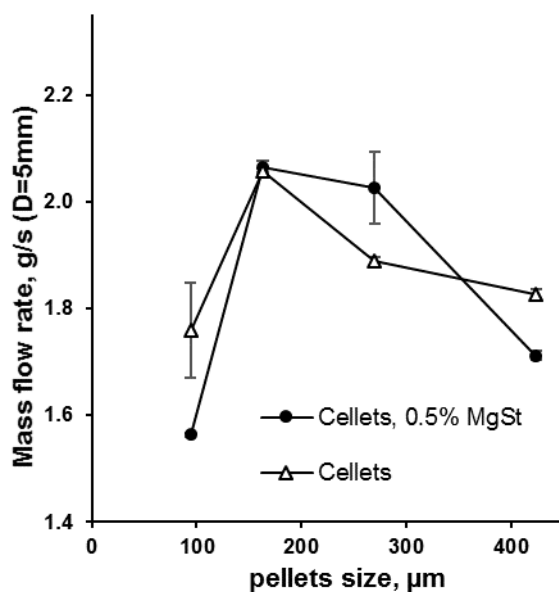


521 Figure 7. Pressure drop (ΔP) across the microparticle bed (A) and Permeability (B) with
522 increasing Cellets particle size and Avicel PH-102 without and with 0.5% MgSt.

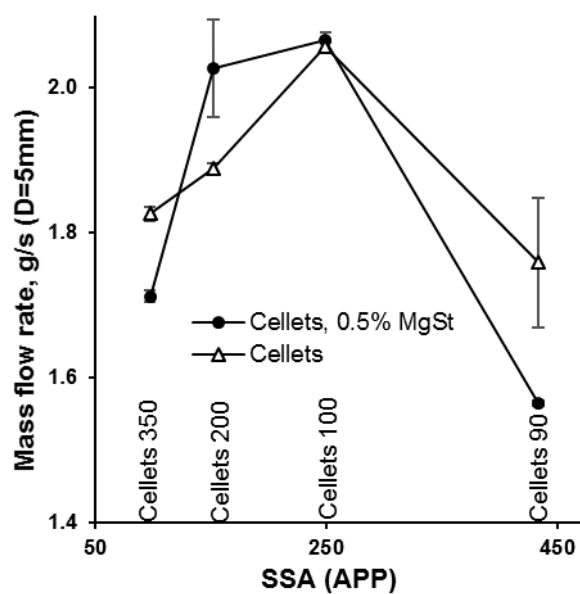
523

524

A



B

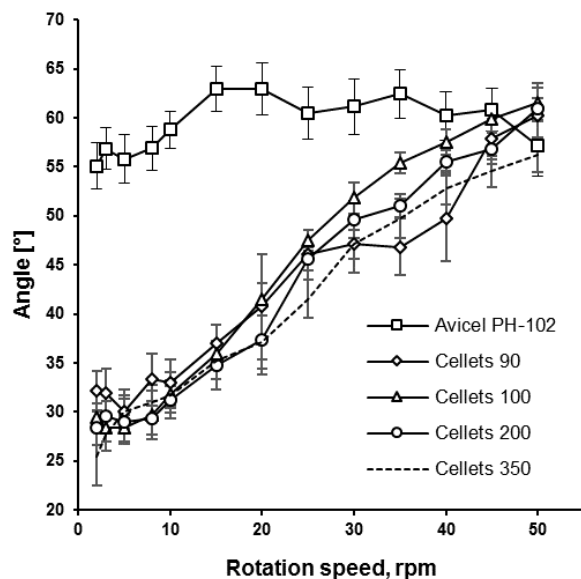


525 Figure 8. Effect of Cellets' size (A) and apparent specific surface area (SSA_{APP}) of Cellets
526 (B) on the mass flow rate without and with 0.5% MgSt.

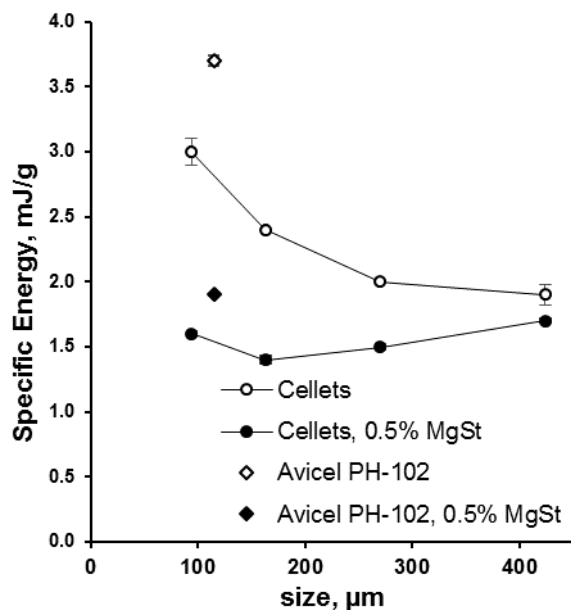
527

528

A



B



529 Figure 9. Dynamic angle of repose at different rotation speed (A) and Specific Energy (B) for
 530 different Cellets' particle size and Avicel PH-102 without and with MgSt.

531

532 Table 3. Dynamic properties of Cellets measured with FT4 powder rheometer

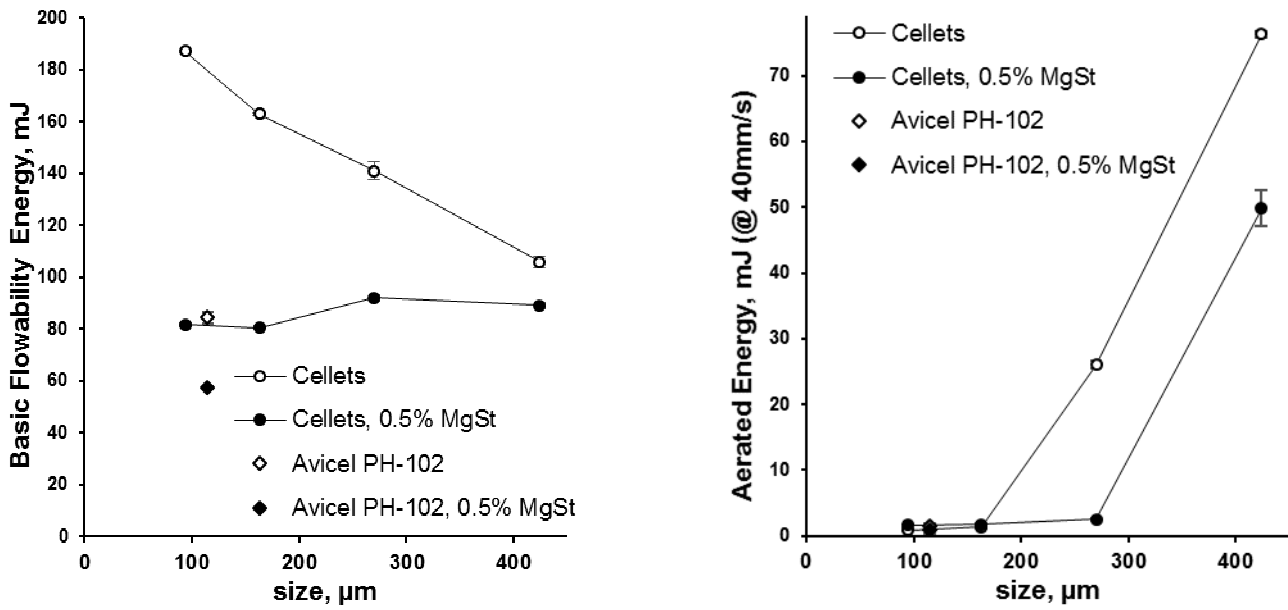
	Basic Flowability Energy		Specific Energy		Conditioned Bulk Density		Aerated Energy (at 40 mm/s)	
	Av	SD	Av	SD	Av	SD	Av	SD
	mJ	mJ	mJ/g	mJ/g	g/ml	g/ml	mJ	mJ
Cellets 350	106.0	2.2	1.9	0.1	0.9	0.002	76.3	0.5
Cellets 200	141.0	3.5	2.0	0.0	0.8	0.003	26.0	0.6
Cellets 100	163.0	0.5	2.4	0.0	0.8	0.007	1.4	0.3
Cellets 90	187.0	0.4	3.0	0.1	0.8	0.006	0.9	0.2
Avicel PH102	84.3	2.4	3.7	0.0	0.3	0.002	1.5	0.4
Cellets 350, MgSt (0.5%)	89.1	1.0	1.7	0.0	0.9	0.002	49.8	2.7
Cellets 200, MgSt (0.5%)	92.0	1.1	1.5	0.0	0.9	0.002	2.5	0.1
Cellets 100, MgSt (0.5%)	80.5	0.6	1.4	0.0	0.9	0.003	1.8	0.1
Cellets 90, MgSt (0.5%)	81.7	0.3	1.6	0.0	0.9	0.003	1.7	0.0
Avicel PH102, MgSt (0.5%)	57.4	0.5	1.9	0.0	0.4	0.004	0.9	0.5

533

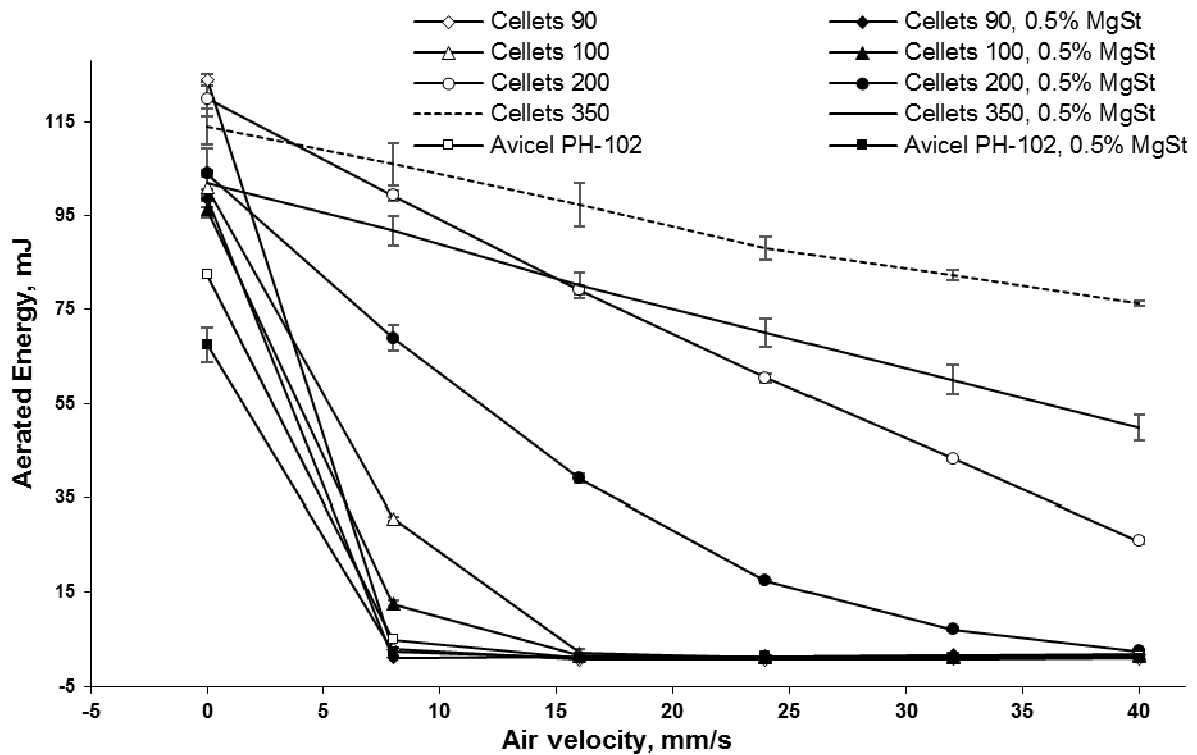
534

A

C



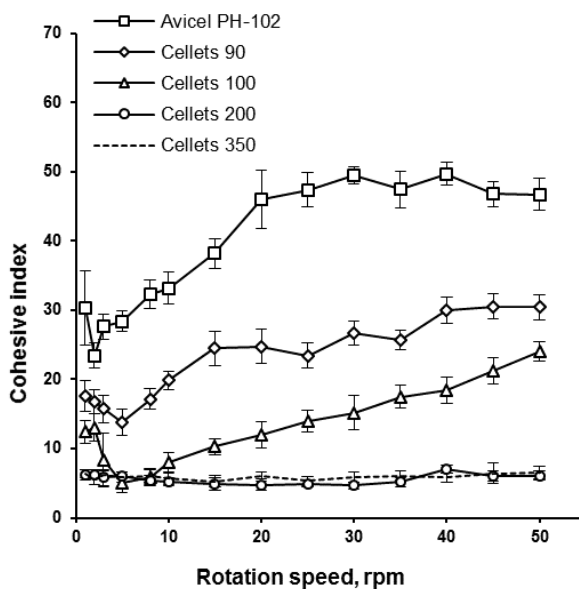
B



535 Figure 10. Flow Energy measurements of Cellets of varying particle size and Avicel PH-102
 536 without and with 0.5% MgSt. Basic Flowability Energy (A), increasing levels of air velocity
 537 on the Aerated Energy (B) and Aerated Energy at 40mm/s air velocity (C).

538

539



540

541 Figure 11. Dynamic cohesive index at different rotation speeds on Cellets with varying
542 particle size and Avicel PH-102.