

1 Performance assessment of reinforced concrete after long-term exposure to a
2 marine environment

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25 **ABSTRACT**

26 The performance of CEM I, CEM III/A and CEM II/B-V reinforced concrete slabs which
27 have been exposed to a marine environment for 18 years is presented. Performance is
28 assessed in terms of porosity, degree of saturation, chloride concentration and electrical
29 resistivity gradients evaluated through the surface ~150mm; the condition of the embedded
30 steel was evaluated using electrochemical techniques. Based on the parameters related to
31 chloride transport, two new environmental factors were introduced into the ClinConc model
32 to predict chloride ingress into unsaturated concrete. A monitoring system is also presented
33 enabling remote interrogation of those slabs located at the marine exposure site.

34 **Keywords:** Field concrete; Chloride-induced corrosion; Electrical resistivity;
35 Electrochemical measurements; ClinConc model; Long-term behaviour

36

37 **1. Introduction**

38 Deterioration of reinforced concrete structures caused by corrosion of the embedded steel is a
39 major problem faced by the construction industry, with chloride-induced corrosion being the
40 most commonly reported mechanism affecting concrete durability [1]. Chloride ions come
41 from deicing salt used on roads during the winter months (XD1, XD2 and XD3 environments
42 according to BS EN 206 [2]) or from the marine environment where, for example, bridges
43 span tidal estuaries (XS1, XS2 and XS3 environments according to BS EN 206). As it is the
44 concrete cover-zone (i.e. the surface 50mm or so) which protects the reinforcing steel from
45 the external environment, it has a major influence on the durability of reinforced concrete by
46 providing the only barrier to chloride ingress. The protective qualities of cover-zone concrete
47 are thus of considerable interest to engineers as the service-life of a structure depends, to a
48 large extent, on the permeation properties of the cover-zone concrete which, in turn, will be
49 influenced by the cementitious binder.

50 To ensure the protective quality of the cover-zone, the majority of the current design codes
51 deal with this aspect on the basis of a prescriptive approach by specifying limiting parameters
52 such as minimum strength grade, minimum binder content and maximum water-binder ratio
53 (w/b) for a series of well-defined environmental classes. With regards to the marine
54 environment and chloride induced corrosion, there are, generally, three exposure classes with
55 the tidal/splash zone considered to be the most vulnerable. Table. 1 presents typical values
56 from a number of design codes for structures subjected to this particular exposure condition.
57 The maximum w/b ratio is ~0.4, while the required minimum binder content, compressive
58 strength and maximum chloride content vary from one specification to another, most likely
59 driven by differences in local materials, practices and exposure conditions. As these factors
60 are highly variable and strongly dependent on local conditions, it would be difficult to

61 recommend limiting values which would ensure adequate durability performance throughout
 62 the life of a structure. Satisfactory guidelines for ensuring adequate reinforced concrete
 63 durability can only be developed by evaluating concrete performance under a range of field
 64 exposure conditions over an extended period of time. Only then can there be a move from
 65 prescriptive durability specifications to performance-based methods [3].

Design Code	Maximum w/b	Minimum binder content (kg/m ³)	Minimum f_c (MPa)	Maximum chloride content (% weight of cement)	Country	Reference
EN 206:2013 +A1: 2016	0.45	340	35	0.4	EU	[2]
BS 8500-1:2015 + A1:2016*	0.35-0.55	320-380	25-40	0.3	UK	[4]
ACI 318-14	0.4	+	35	0.15**	USA	[5]
CSA A23.1-09	0.4	+	35	+	Canada	[6]
JGC15/16 SSMC-2007	0.45	330 [#] 300 ^{##}	+	1.2kg/m ^{3###}	Japan	[7,8]

66 **Note:** f_c compressive strength; * Intended working life of at least 100 years; ** water soluble chloride
 67 content; # maximum aggregate size 20 or 25mm; ## maximum aggregate size 40mm; ### total chloride
 68 content per unit volume of concrete.

69 **Table 1** Limiting values of mix design in the codes for marine structures subjected to tidal
 70 and splash zone.

71 Regarding durability design, there is a recent growing interest to move away from the
 72 prescriptive method towards the performance-based approach. For example, whilst BS EN
 73 206 [2] still stipulates prescriptive-based approach for durability, Section 5.3.3 of this code
 74 allows for performance-related methods and defines concrete on the basis of an *equivalent*

75 *durability procedure* (EDP); however, in order to fully implement a performance-based
76 strategy requires:

- 77 (1) long-term experience of local materials and practices, and a detailed knowledge of the
78 local environment;
- 79 (2) test methods based on approved and proven tests that are representative of actual
80 conditions and have approved performance criteria; and,
- 81 (3) analytical models that have been calibrated against test-data representative of actual
82 conditions in practice.

83 Supplementary cementitious materials (SCMs) such as fly ash (FA) and ground granulated
84 blast-furnace slag (GGBS) have been extensively used to replace Portland cement (PC) in
85 concreting operations. It is well-known that these materials are beneficial in resisting chloride
86 transport through their reduced permeability and high chloride binding capacity [9–12]
87 although the chloride threshold level (CTL) regarding these materials is still controversial [13-
88 16]. Moreover, according to BS 8500-1 [4], the use of SCMs is recommended in chloride
89 environments although specification of such concretes still relies on prescriptive methods rather
90 than a performance-based approach. For example, a reinforced concrete element with an
91 intended working life of at least 100 years exposed to XS3 environment, the use of CEM I
92 (Portland cement clinker) concrete is now not recommended [4], whereas CEM III/A (blast
93 furnace cement) concrete or CEM II/B-V (fly ash cement) concrete require a minimum cover to
94 steel of 60mm, plus an allowance in design for deviation. Additionally, maximum w/b and
95 minimum compressive strength are also defined in the specification although these parameters
96 cannot fully ensure the performance of concrete in chloride-rich environments. In comparison
97 to laboratory testing, there is a dearth of long-term, field data for chloride-exposed concrete
98 (see requirements in (1) - (3) above) which has resulted in a slow uptake of a performance-
99 based approach for concrete specification.

100 In the current study, a comprehensive assessment of the performance of 18 year old
101 reinforced concrete slabs subjected to XS3 environment is presented (i.e. concrete in the tidal
102 and splash zone). The testing program included evaluating the chloride concentration,
103 porosity, degree of saturation and electrical resistivity at discrete points within the surface
104 ~150mm. Regarding the condition of the steel surface, half-cell potential, polarisation
105 resistance and *B*-value were measured therefore providing long-term data on the performance
106 of steel in concrete. Furthermore, the ClinConc model [17] was combined with two new
107 environmental factors to predict the time to corrosion initiation for the field concrete slabs.

108 **2. Field durability studies**

109 To date, a number of marine exposure sites have been installed in various countries around
110 the world in an attempt to assess the performance of concrete over an extended period of
111 time. For example, Tang and Utgenannt [18] exposed more than 40 concrete samples to a
112 marine environment on the west coast of Sweden; after 13-years of exposure, it was
113 suggested that oxygen availability plays an important role in corrosion initiation and that a
114 total chloride content of 1% by weight of binder can be regarded as the critical value for
115 significant corrosion to occur. Using the Building Research Establishment (BRE) exposure
116 site on the Thames Estuary in South-East England, Thomas et al. [19] reported that chloride
117 threshold levels for Portland cement concrete containing fly-ash decreases with increasing
118 fly-ash replacement. The chloride concentration was measured using an X-ray fluorescence
119 technique and the condition of the embedded steel bars was evaluated using a weight loss
120 method. After 4-years exposure, it was found that the total chloride threshold levels were
121 0.7%, 0.65%, 0.5% and 0.2% by weight of binder, corresponding to concrete with fly-ash
122 replacement levels of 0%, 15%, 30% and 50%, respectively. It was also found that the
123 diffusion coefficient decreases with increasing fly-ash replacement and this exerts a greater
124 influence on corrosion rate than the effect of decreasing chloride threshold level with

125 increasing fly-ash content. Otieno et al. [20] carried out parallel investigations on both
126 uncracked and pre-cracked concretes (containing SCM's) exposed to accelerated laboratory
127 testing and a natural marine environment at Table Bay in Cape Town (exposure time was
128 2.25 years). Using coulometric and half-cell potential techniques, they found that the corrosion
129 rate increased with crack width and decreased with concrete quality and cover-depth. The use
130 of SCMs was found to result in significant reductions in corrosion rate regardless the
131 exposure condition studied. The corrosion process under the accelerated condition was found
132 to be different to that under the natural environment. Thomas et al. [21] demonstrated that
133 concrete with high levels of GGBS replacement displays a satisfactory performance even
134 though being exposed to a highly aggressive marine environment for 25 years at Treat Island
135 in the USA. However, it was recommended that the w/b ratio must be kept low, ideally ≤ 0.4 .
136 Overall, these studies highlight the importance of field studies to understand the durability
137 performance of different concretes under a particular exposure condition.

138 Considering the complexity of natural exposure, it would be desirable to express the
139 durability of concrete exposed to marine environments by a single factor with electrical
140 resistivity being increasingly attractive for *indexing* durability. Regarding the latter, this
141 testing methodology has been utilized by McCarter et al. [22] in a sensor system to study
142 both the short- and long-term response of the cover-zone concrete subjected to a range of
143 marine exposure conditions. It was shown that the properties of concrete change with time
144 and depth due to further hydration and are affected by ionic ingress. Based on chloride
145 profiling data and electrochemical measurements on concrete exposed to marine
146 environment, Morris et al. [23] proposed a relation between electrical resistivity of concrete
147 and chloride threshold level and highlighted that the resistivity of concrete can be regarded as
148 an effective parameter to evaluate the risk of chloride-induced corrosion. To establish the

149 relationship between performance of the field concrete and electrical resistivity, additional
150 studies are clearly still required.

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152 **3. Experimental**

153 *3.1. Field conditions and slab details*

154 The concrete slabs used in this study had been exposed to three different marine
155 environments specified in BS EN 206 [2] and BS 8500-1 [4]

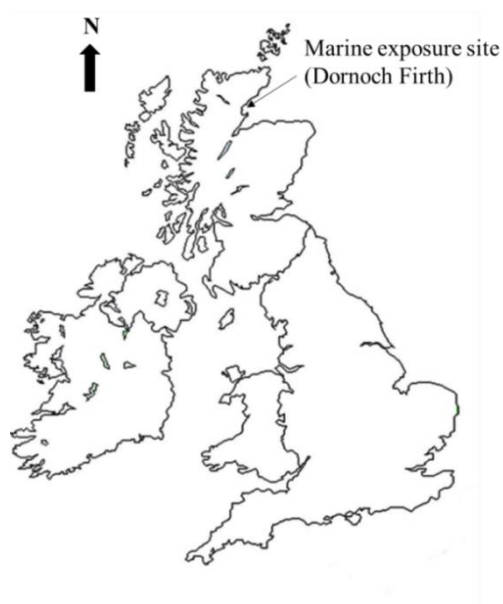
156 (1) above the high-water level in the atmospheric spray zone (XS1);

157 (2) the tidal and splash zone (XS3); and,

158 (3) below mid-tide level (XS2 environment).

159

160 The slabs were positioned at a marine exposure site on the Dornoch Firth in north-east
161 Scotland (see Fig. 1) in 1998.



162

163

Fig. 1. Location of marine exposure site.

164 The concrete mixes used in the marine exposure program are presented in Table 2 [24]. The
 165 slabs were 300×300×200(thick)mm cast in plywood formwork which had been given a coat
 166 of proprietary release agent prior to casting. After casting, the slabs were compacted into two
 167 layers on a vibrating table and demoulded after 24-h. On demoulding, the slabs were then
 168 wrapped in damp hessian and covered with polyethylene sheeting under a sheltered condition
 169 for 7-days and all sides and the top face of the slabs were then coated with several coats of a
 170 high-build epoxy-based paint to ensure one-dimensional movement of water/moisture (see
 171 Figs. 2(a)-(c)). The slabs contained two Ø16mm×200mm ribbed rebar sections which had
 172 been cleaned and degreased prior to installation; they were positioned to have a 50mm cover
 173 from the working face and sides. A Ø6mm×200mm stainless steel (s/s) bar was positioned
 174 centrally between these two rebars (see Fig. 2(b)). All bars had electrical connections and the
 175 cut ends of the rebars were coated with epoxy glue. The rebar and s/s rod formed a working-
 176 electrode/counter-electrode system and, when used in conjunction with a surface mounted
 177 reference electrode, facilitated rebar corrosion measurements viz. half-cell and linear
 178 polarisation.

Mix designation	CEM I (kg/m ³)	GGBS (kg/m ³)	FA (kg/m ³)	Coarse aggregate		Fine aggregate (kg/m ³)	WR* (l/m ³)	w/b	f ₂₈ ** (MPa)
				20 mm (kg/m ³)	10 mm (kg/m ³)				
CEM I (PC)	460	+	+	700	350	700	1.84	0.4	70 (64.9)
CEM III/A (GGBS/40)	270	180	+	700	375	745	3.60	0.44	53 (53.9)
CEM II/B-V (FA/30)	370	+	160	695	345	635	2.65	0.39	58 (49.0)

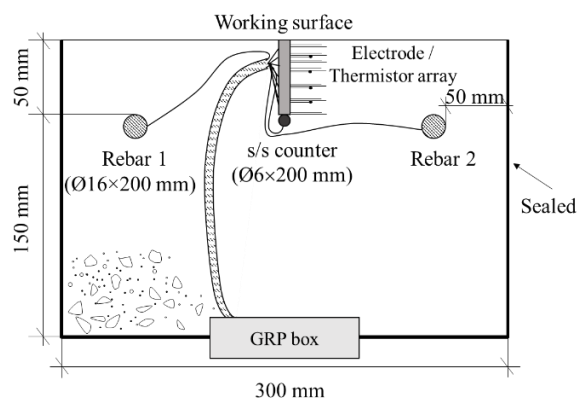
179 *WR water reducer; ** f₂₈ 28-day compressive strength. Figure in brackets represents compressive
 180 strength for migration test samples.

181 **Table 2** Mix design used in field slabs [24].

182 For monitoring the electrical resistance within the cover zone concrete, seven pairs of
 183 stainless-steel pin-electrodes were mounted on a PVC former to form an electrode array [25]

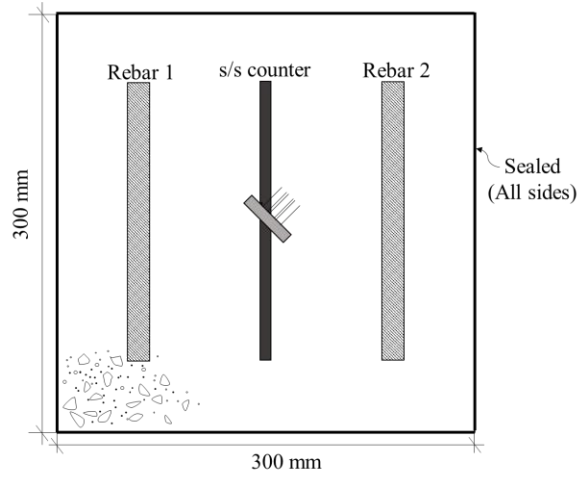
184 which was, subsequently, embedded within each slab at the time of casting (see Fig. 2(a)).
185 This allowed resistance measurements to be obtained at discrete depths from the exposed
186 working surface i.e. 5, 10, 15, 20, 30, 40 and 50 mm. For each pair of electrodes, the spacing
187 between the individual pins was 10mm. Four thermistors were also mounted on the PVC
188 former enabling temperature measurements to be obtained at 10, 20, 30 and 40 mm from the
189 working surface. Colour-coded cables were connected to the electrodes, thermistor, s/s bar
190 and rebars and then were taken into a watertight glass-reinforced plastic (GRP) enclosure
191 placed in the face opposite to the working face (see Fig. 2(c)); a 37-pin, multi-pole female D
192 connector was used to terminate all wires.

193 Within approximately 5-weeks after demoulding, the slabs were transported to the marine site
194 and were placed to the site. The 300×300mm face cast against the plywood formwork was
195 used as the *working face* and positioned vertically facing the sea; six slabs per mix were
196 secured within galvanised steel frames and positioned at each exposure condition (see Fig. 3).
197 In XS2 and XS3, three slabs (one for each mix) were hard-wired (see Fig. 4(a)) to a control-
198 box facility to interrogate/monitor the slabs remotely (see Fig. 4(b) and (c)).



199
200

(a)



(b)

201
202



(c)

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205
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Fig. 2. Concrete slab used in the experimental program showing (a) sectional view, (b) plan view and (c) GRP box enclosing a 37-pin D connector and lead.



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Fig. 3. Frames containing slabs installed at Dornoch field site in XS1, XS2 and XS3 positions.

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Water-proofed hard-wired connection to remote interrogation system



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(a)



Watertight enclosure for monitoring equipment

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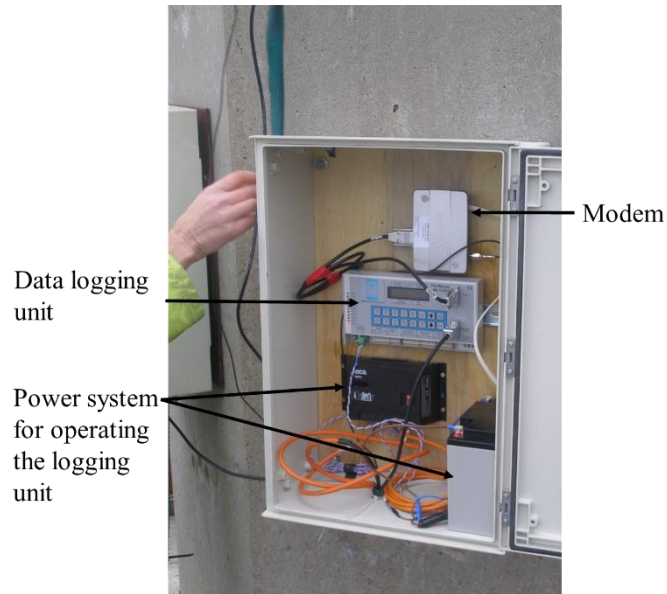
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(b)



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(c)

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Fig. 4. (a) Concrete slabs hard-wired for remote interrogation in XS3 environment (b) watertight enclosure for monitoring equipment and (c) modem, data logging unit and power source for the monitoring system.

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In this study, only those slabs exposed to an XS3 environment were returned to the laboratory as it is in the tidal and splash zones where reinforced concrete is highly susceptible to chloride-induced corrosion [4]. After 18 years exposure, three slabs, one for each mix (see Fig. 5), were retrieved and double-sealed with polythene *film* to preserve the moisture condition of samples. A series of experiments and sampling was then conducted in a laboratory environment when the slabs had equilibrated at laboratory temperature ($20\pm 1^\circ\text{C}$).



238 Fig. 5. Concrete slabs retrieved from frame in XS3 environment.

239 *3.2 Tests relating to chloride transport*

240 In relation to the chloride transport behaviour of concrete, the following key performance
 241 indicators were obtained from the slabs, (1) degree of saturation, (2) porosity, and (3)
 242 chloride concentration. With reference to Fig. 6(a), for degree of saturation and porosity
 243 measurements the slab was cut vertically on sections A-A' and B-B' to give a
 244 300×200×50(thick)mm 'slice' of concrete (Fig. 6(b)); seven prismatic samples
 245 50×50×20(thick)mm were then cut from the central portion of this slice as depicted in Figs.
 246 6(b) and (c). To minimise any changes in the moisture state of the concrete, cutting was
 247 carried out using a diamond-concrete saw without water cooling. After cutting, the mass of
 248 each prism was immediately measured. To minimise any potential damage in microstructure
 249 due to micro-cracking at elevated temperatures, a low drying temperature was chosen with
 250 the samples dried to constant mass in an oven at 50±2°C. The samples were then vacuum
 251 saturated with distilled water under a pressure of 30mbar which was maintained for 4 hours.
 252 During the saturation stage, the mass of each sample was periodically measured using a scale
 253 of ± 0.01g accuracy until equilibrium was achieved at which time the following criterion was
 254 satisfied:

255
$$\left| \frac{m(t_i) - m(t_{i+1})}{m(t_{i+1})} \right| \times 100 \leq 0.1(\%) \quad (1)$$

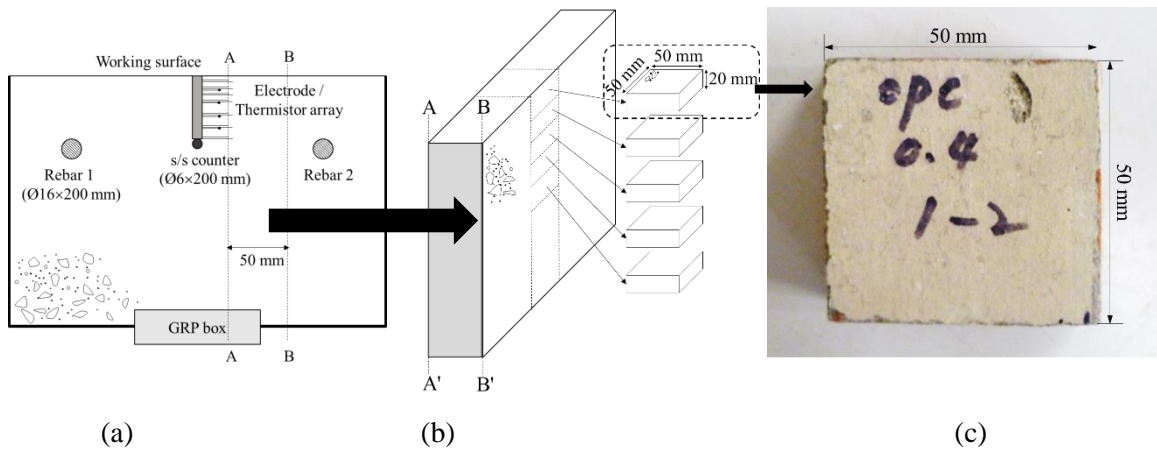
256 where $m(t_i)$ is the mass measured at time t_i and $m(t_{i+1})$ is the mass measured after time
 257 t_{i+1} . The degree of saturation and porosity were then calculated as follows:

258
$$S_r (\%) = \frac{(m_o - m_d)}{(m_v - m_d)} \times 100 \quad (2)$$

259
$$\phi (\%) = \frac{(m_v - m_d)}{(V_s \times \gamma_w)} \times 100 \quad (3)$$

260 where S_r is degree of saturation (%), m_o is the original sample mass (g), m_d is the dried
 261 sample mass (g), m_v is the vacuum saturated sample mass (g), ϕ is porosity (%), γ_w is the
 262 density of distilled water ($=1 \text{ g/cm}^3$), and V_s is the sample volume (cm^3).

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265 Fig. 6. (a) Schematic of concrete slab showing cutting on sections A-A' and B-B', (b) cutting
 266 procedure for concrete 'slice', and (c) cut prisms used for evaluating porosity, degree of saturation and
 267 electrical resistivity.

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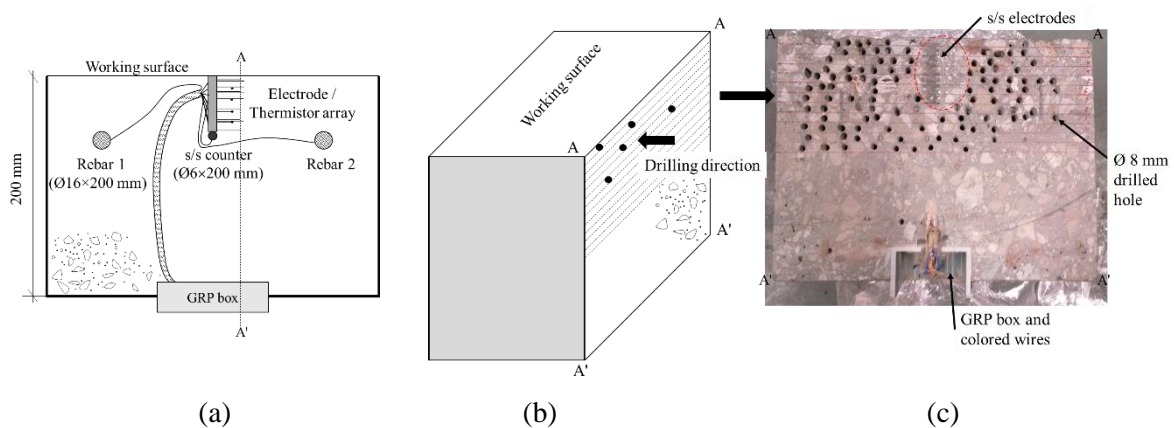
269 The 300mm×200mm exposed surface (A-A/A'-A' on Figs. 7(a) and (b)) of each slab was used
 270 to obtain dust samples for chloride profiling. Dust samples were obtained using a hammer-
 271 action drill with an 8-mm diameter drill bit. The depth of investigation for all concretes
 272 covered the surface 0- ~100 mm using 8-mm depth increments (see Fig. 7(b)) between drill

273 holes. Several sampling locations (6–8 at each depth, see Fig. 7(c)) were drilled to a depth of
 274 30mm depth and powder taken to be representative of bulk concrete. The powder was then
 275 sieved using a 125µm sieve with at least 10g of powder collected at every depth increment.
 276 The total chloride (acid soluble) concentration in ppm at each depth was measured using
 277 Chloride QuanTab® strips (manufactured by HACH). The concentration was then converted
 278 to chloride concentration in % weight of binder. From the chloride profile, the apparent
 279 diffusion coefficient and surface chloride concentration were calculated by fitting the error
 280 function solution to Fick's 2nd law. The pre-existing chloride concentration in the concrete
 281 mixes (C_o) was negligible, hence,

$$282 \quad C(x, t) = C_s \left(1 - \operatorname{erf} \left(\frac{x}{2\sqrt{D_a t}} \right) \right) \quad (4)$$

283 where $C(x, t)$ is the total chloride concentration (%) at depth of x (m) from the surface at
 284 exposure time, t (sec), C_s is the surface chloride concentration (%), D_a is the apparent
 285 diffusion coefficient (m^2/s).

286



287 (a) 288 Fig. 7. Sampling procedure for measuring chloride concentration, (a) cut surface A-A/A'-A', (b)
 289 drilling direction parallel to exposed surface, and (c) at a particular depth, powder was collected from
 290 6-8 drill-holes.

291 The migration coefficient is widely used as a performance indicator to evaluate the resistance
292 of chloride transport, particularly in European countries [26]; in addition, the 6-month
293 migration coefficient is required in the implementation of the ClinConc model. Samples for
294 the migration test were cast at a later stage using a similar mix specification to that used in
295 the 18 year old concrete slabs (see Table 2) thereby conforming to the reference concrete
296 defined in the equivalent durability procedure in Section 6.2 of PD CEN/TR 16563:2013 [26];
297 28-day compressive strength tests were also carried out using 100 mm cubes. For migration
298 testing, Ø100mm×300mm cylinders were cast in PVC moulds; 24-h after casting, the
299 cylinders were placed in curing tank. After 6-months curing, three concrete disks were
300 extracted from the middle of the cylinders using a diamond saw and the migration coefficient
301 was determined following the test procedure specified in NT Build 492 [27]. The average
302 compressive strength and migration coefficient of the reference concrete are summarised in
303 Tables 2 and 6, respectively.

304 *3.3. Tests relating to corrosion of steel*

305 Prior to the tests detailed above, the condition of the embedded steel bars was investigated
306 using the half-cell potential technique, galvanostatic pulse technique and the Tafel
307 extrapolation technique. A three-electrode configuration was used with the two mild steel
308 rebars as individual working electrodes, the stainless-steel rod as a counter electrode and a
309 surface-mounted copper-copper sulphate reference electrode (CSE). A Solartron 1287
310 Electrochemical Interface was used and controlled by CorrWare® software. The corrosion
311 potential was determined when the potential changed by <10 mV/s. The exposed area of each
312 working electrode was 100.53 cm². Between galvanostatic pulse measurement and Tafel
313 extrapolation measurement, a depolarisation time of 10 minutes was maintained to recover
314 the corrosion potential. For the galvanostatic pulse technique, a constant current of 25µA

315 corresponding to an average current density of $0.25\mu\text{A}/\text{cm}^2$ was applied to the working
 316 electrode for 10s. From the result, the ohmic resistance and polarisation resistance were
 317 estimated using a modified Randles circuit combined with an exponent as shown in Eq. (5)
 318 [28],

$$319 \quad V_t(t) = I_{app}R_{\Omega} + I_{app}R_p \left[1 - e^{-\left(\frac{t}{\tau}\right)^{\beta}} \right] \quad (5)$$

320 where $V_t(t)$ is the potential of the steel bar with measuring time t ; I_{app} is the applied
 321 current; R_p is the polarisation resistance; R_{Ω} is the ohmic resistance; τ is the time constant
 322 ($=R_pC_{dl}$); β is the non-ideality exponent ($0 < \beta \leq 1$); and C_{dl} is the double-layer
 323 capacitance.

324 To evaluate the B -value, a Tafel extrapolation test was also carried out. The sweep range was
 325 from -300mV to $+600\text{mV}$ versus the free corrosion potential and the scan rate was
 326 $10\text{mV}/\text{min}$. The curve corresponding to $\pm 200\text{mV}$ around the corrosion potential was fitted
 327 using a commercial program (CView Version 3.5a). The B -value was calculated with the
 328 Tafel constants obtained from the fitting as follows:

$$329 \quad B = \frac{\beta_a\beta_c}{2.3(\beta_a+\beta_c)} \quad (6)$$

330 where β_a and β_c are Tafel constants from the anodic branch and the cathodic branch,
 331 respectively (mV/dec).

332 *3.4 Electrical resistivity of concrete*

333 The electrical resistivity of the concrete was evaluated for two test conditions (a) the field
 334 slabs which were returned to the laboratory, and (b) their counterparts positioned at the XS3
 335 environment at the marine exposure site. For (a), the electrical resistance of the vacuum-

336 saturated 50×50×20mm prisms used in the porosity and saturation tests was measured using a
 337 LCR meter (Agilent 4263B) at a frequency of 1kHz and signal amplitude of 350mV. A two-
 338 electrode technique was employed with intimate contact between the electrodes and prism
 339 obtained using synthetic sponges which had been saturated with Ca(OH)₂ solution. The
 340 resistance was then converted to resistivity by Eq. (7),

$$341 \quad \rho = \frac{A}{L} R_c \quad (7)$$

342 where ρ is the resistivity of the concrete ($\Omega \cdot m$), A is the cross-sectional area of the sample (= $0.025m^2$), L is the sample thickness (= 0.02m) and R_c is the measured concrete resistance (Ω).

344 For condition (b), the electrode and thermistor array described above was used to obtain
 345 discretized resistance and temperature measurements using the remote interrogation system.
 346 Data were uploaded through a dial-up modem and then recorded in a CSV file using a
 347 software installed in the office-based computer. A measurement cycle was carried out every
 348 6-hours for approximately 1 month (September/October). The resistance measured by the
 349 thermistor was converted to *degree Celsius* using the Steinhart-Hart equation:

$$350 \quad T = [A + B \ln R + C (\ln R)^3]^{-1} - 273.15 \quad (8)$$

351 where R is the measured thermistor resistance (Ω); T is temperature ($^{\circ}C$); and A , B , and C
 352 are coefficients depending on the type of thermistor which were, 1.29×10^{-3} , 2.36×10^{-4} and
 353 $9.51 \times 10^{-8} K^{-1}$, respectively [29].

354 Prior to embedding the electrode array in the slabs, the electrode arrangement was calibrated
 355 in solutions of known resistivity thereby enabling the measured resistance of concrete,
 356 between the electrodes, R_c (Ω), to be converted to resistivity, ρ ($\Omega \cdot m$) through Eq. (9):

$$357 \quad \rho = k R_c \quad (9)$$

358 where k is the calibration factor for the electrode geometry which was obtained as $0.0125\text{m} \pm$
359 5%. The resistivity of the calibrating solutions covered the anticipated range of concrete
360 resistivity (i.e. $10\text{--}100\Omega\text{m}$).

361 **4. Results and discussion**

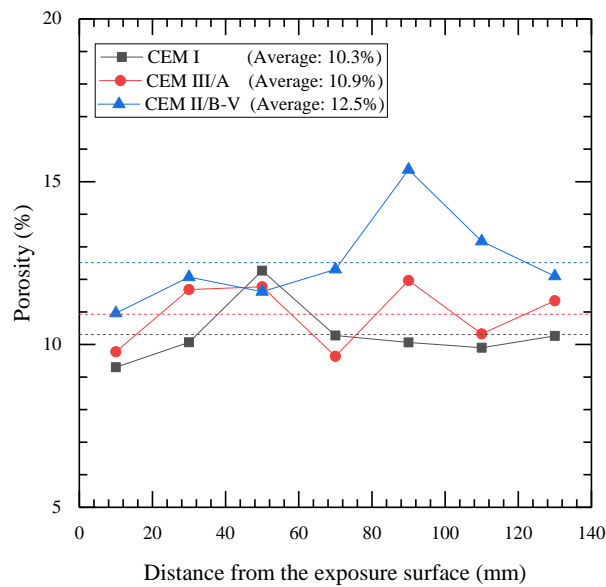
362 *4.1 Performance Indicators for Chloride Transport*

363 *4.1.1 Porosity Gradient*

364 Dissolved chlorides move through the connected pore network in concrete and, when the
365 chloride concentration at the steel surface exceeds a certain critical value (the chloride
366 threshold level CTL), depassivation will occur and corrosion will be initiated. Before
367 depassivation, however, the capillary pore structure within the cementitious binder is a key
368 factor in determining chloride transport behaviour in concrete and porosity is an important
369 performance parameter in this respect. Fig. 8 presents the porosity gradient for each concrete
370 slab through the surface 140mm with the average porosity given in the Figure legend.
371 Interestingly, the porosity increases with type of binder in the order: CEM I (10.3%) < CEM
372 III/A (10.9%) < CEM II/B-V (12.5%). From the results, it is evident that concretes containing
373 SCMs have a slightly higher porosity than the plain Portland cement (CEM I) concrete;
374 however, it is well-established that concretes containing SCMs have a higher resistance to
375 chloride transport than plain Portland cement concretes [10]. In attempting to explain this
376 anomaly, the various types of pores that exist in concrete are shown schematically in Fig. 9
377 [30]. It could be deduced that whilst the overall porosity of the concretes containing SCMs is
378 higher than plain CEM I concrete, the capillary pore network within these concretes will be
379 more tortuous and disconnected in comparison to the CEM I concrete. This would be in
380 agreement with previous studies [31, 32]. The result of this would imply that the rate of

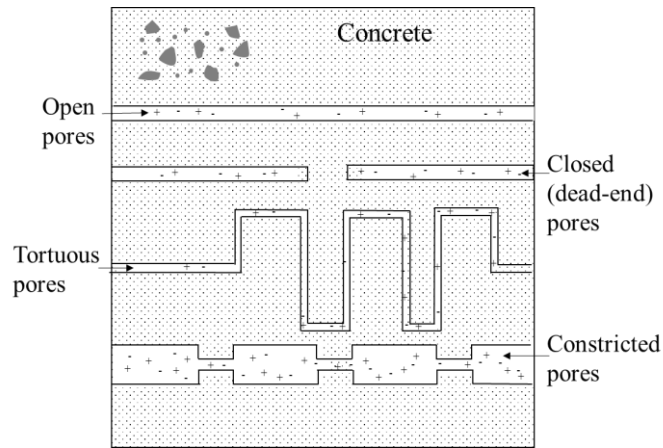
381 chloride transport would be slower in these concretes and that porosity, itself, cannot be used
382 solely as an indicator of concrete performance without due consideration of pore type as
383 depicted in Fig. 9.

384 Although porosity varied with depth due to heterogeneity of the concrete (see Fig. 8), the
385 variation is considered small. It is noteworthy that the porosity of the surface 0-20mm was
386 generally lower than those at other depths. This could be explained, in part, by chloride
387 binding effects which can constrict capillary pores and reduce porosity.



388
389 Fig. 8. Porosity gradient of 18-year old concrete through the surface 140mm (XS3 environment).

390



391

392

Fig. 9. Schematic representation of different types of capillary pore in concrete.

393 *4.1.2 Degree of Saturation*

394

As (free) chlorides dissolved in the pore-water are only mobile through the continuous water-

395

filled capillary pores, chloride transport will be significantly influenced by the moisture state

396

of the concrete. At low moisture-contents corresponding to a value which is higher than the

397

critical moisture content below which capillary absorption becomes dominant, less chloride

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will be transferred through the pore solution assuming that there are no external forces for

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chloride transport. Until now, it is noted that there is no evident boundary value between

400

capillary suction and diffusion with respect to chloride transport in concrete. From Fig. 10,

401

the degree of saturation was shown to be in the order: CEM I (83%) > CEM III/A (73%) >

402

CEM II/B-V (67%) and is the reverse of the order in porosity. This could be explained as the

403

CEM I concrete will have a more connected porosity hence allow a more rapid ingress of

404

water. At 10mm depth, the degree of saturation was, generally, higher than at other depths as

405

it is this region which is more susceptible to the wet/dry cyclic regime and represents the

406

convective zone. It could be inferred that the degree of saturation of the concrete at depths

407

>10mm remains relatively stable at this exposure environment. In other words, chloride

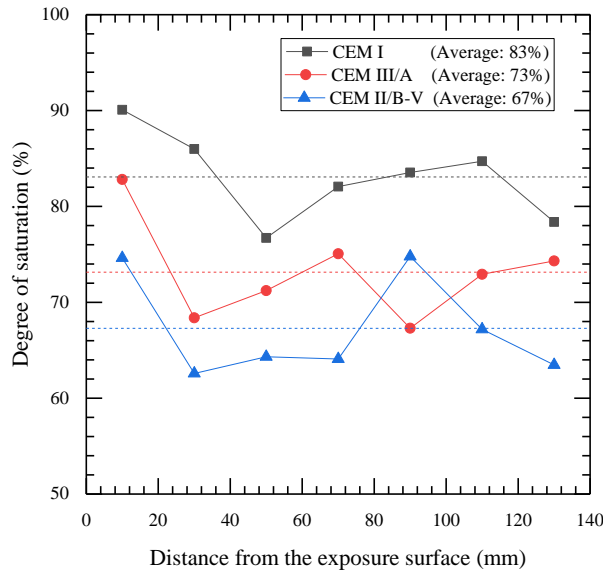
408

transport rate in the tidal zone is lower than that in the submerged zone as the convective

409

zone in tidal zone is negligible (~5mm) [33] and the concrete is not continuously exposed to

410 chlorides. This result is in agreement with a previous study [34]. The degree of saturation
 411 must be considered when the chloride transport rate is determined and represents an
 412 important performance indicator in this respect.

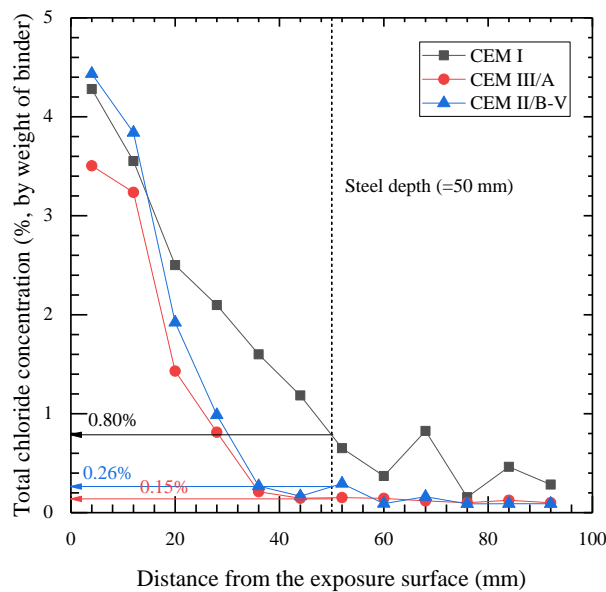


413 Fig. 10. The variation in the degree of saturation of 18-year old concrete through the surface 140mm
 414 (XS3 environment).
 415

416 4.1.3 Chloride Ingress

417 The chloride profiles for each concrete (one slab per mix) after 18 years exposure are
 418 presented in Fig. 11. As expected, the CEM I concrete has a higher concentration of chlorides
 419 at all depths in comparison to the other concretes. Curve-fitting using Eq. (4), obtained
 420 diffusion coefficients, D_a , of $1.39 \times 10^{-12} \text{ m}^2/\text{s}$, $4.21 \times 10^{-13} \text{ m}^2/\text{s}$, and $4.21 \times 10^{-13} \text{ m}^2/\text{s}$ for CEM I,
 421 CEM III/A and CEM II/B-V concrete, respectively, and surface chloride concentrations, C_s ,
 422 of 4.54%, 4.46% and 5.56% (by weight of binder), respectively. Direct comparison with
 423 previous studies is difficult as the concretes had different mix compositions and were
 424 exposed to different curing and exposure conditions [10, 35]. However, it is clear that SCM
 425 concretes were observed to be beneficial in resisting chloride transport, particularly at depths
 426 $>20\text{mm}$.

427 Regarding chloride-induced corrosion, the chloride concentration at the surface of the steel is
 428 a critical factor in corrosion initiation and values of 0.4% and 0.6% (by weight of binder) has
 429 been presented in BS EN 206 [2] and fib [36], respectively as an acceptable chloride
 430 threshold level (CTL). From the chloride profiles in Fig. 11, the chloride concentration at
 431 steel depth for both SCM concretes was 0.15% and 0.26% for CEM III/A and CEM II/B-V,
 432 respectively, which are lower than the tolerance value in the specifications. For the CEM I
 433 concrete, the chloride concentration was 0.8%; however, there were no visible signs of
 434 corrosion (e.g. rust staining, cracking). When compared to the data obtained from the field,
 435 the CTL specified in the code would appear to significantly underestimate the chloride
 436 concentration for corrosion initiation.

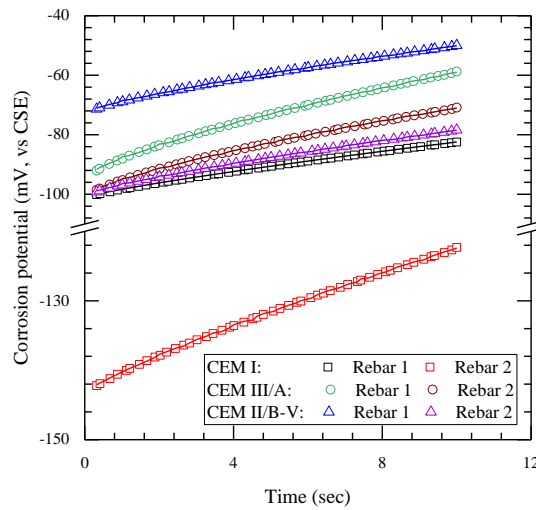


437 Fig. 11 Chloride profiles through slabs after 18 years exposure (XS3 environment)

438 *4.2 Electrochemical Measurements on Embedded Steel*

439 The products of corrosion result in cracking and spalling of the concrete cover which further
 440 accelerates the deterioration process. Although chloride ingress is a determinant in the
 441 corrosion of steel, it is still difficult to predict time to corrosion initiation as the CTL is not a

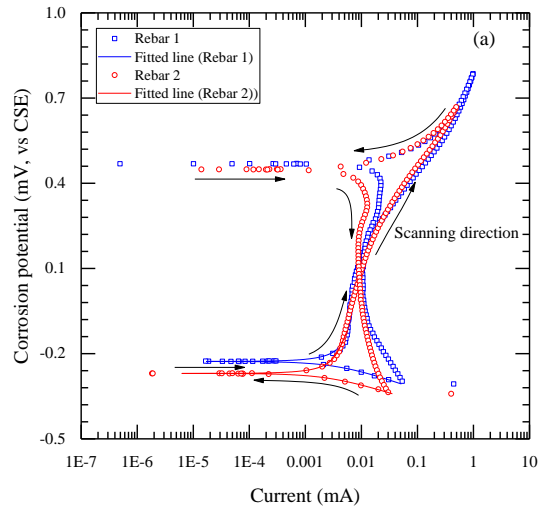
442 single value. To manage reinforced concrete structures subjected to chloride environments
443 regular monitoring of the condition of steel is required. In this study, corrosion parameters
444 including corrosion potential, ohmic resistance, polarisation resistance and B value were
445 investigated together with half-cell potential, galvanostatic pulse technique and Tafel
446 extrapolation technique. Figs. 12 and 13 show measured and fitting data in galvanostatic
447 pulse and Tafel extrapolation measurements.



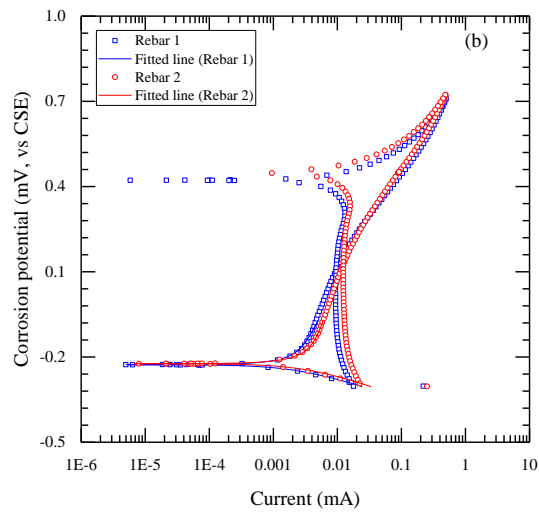
448 Fig. 12 Analysis of corrosion parameters in galvanostatic pulse technique (Solid line = Fitted line)
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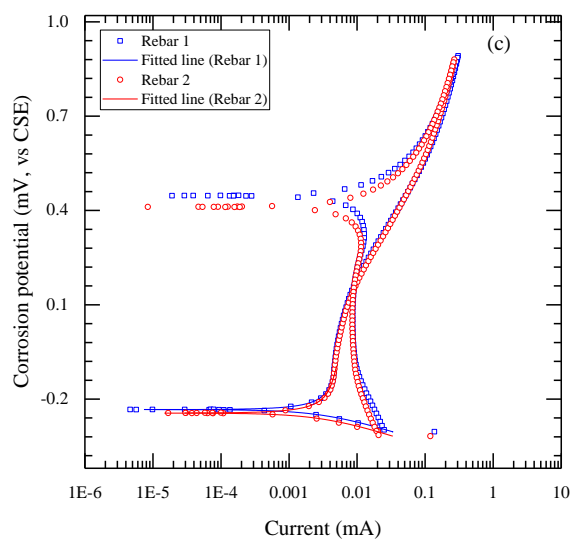
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Fig .13. Measurements used for analysis of corrosion parameter (B value) in Tafel extrapolation technique for (a) CEM I, (b) CEM III/A and (c) CEM II/B-V.

457 Table 3 presents the corrosion parameters for steel embedded in the slabs. Corrosion potential
458 is widely used and according to ASTM C876 [37] the probability of corrosion is specified in
459 Table. 4. Although the method is qualitative, all corrosion potentials were $> -140\text{mV}$ (vs
460 CSE), corresponding to a low probability of corrosion ($<10\%$) in the specification. In order to
461 quantify the corrosion rate, polarisation resistance and B values were evaluated. Although the
462 rate is significantly lower in the passive state, the corrosion rates ranged from $2.2\mu\text{A}/\text{m}^2$ to
463 $6.3\mu\text{A}/\text{m}^2$. Assuming that the corrosion rate ranges from $1 - 2 \text{ mA}/\text{m}^2$ at the onset of
464 corrosion in concrete [38–41], the values presented were ~ 3 orders lower than the value
465 required for the onset of corrosion. Hence, in the passive state, the corrosion rate is negligible.

466 The B -values obtained from the Tafel extrapolation method are similar to the value of 26mV
467 which is normally recommended [42]. In the passive state, the B -value and polarisation
468 resistance are minor factors in quantifying the corrosion rate; however, these values become
469 critical after depassivation. Generally, the B value is set as 26mV in the active state, but can
470 be widely distributed ($2.2\text{--}86\text{mV}$ in the active state) [43–45]. The error caused by the B -value
471 can be considerable. On the other hand, from the Tafel slopes, the corrosion activity was
472 similar in a cathodic environment, irrespective of type of binder, as the samples were exposed
473 to the same environment, i.e. XS3. The anodic reaction rate, however, varied with type of
474 binder and steel rebar. Anodic slopes in the CEM II/B-V concrete were higher than those in
475 the other concretes; hence, steel rebars embedded in CEM II/B-V concrete can have a better
476 inhibition against corrosion. For CEM III/A concrete, the chloride concentration at steel
477 depth (i.e. 50 mm) was lower than the CEM I concrete, but anodic slopes in both concretes
478 ranged from 423.0 to $708.4\text{mV}/\text{dec}$. It could be inferred that the CTL of CEM III/A concrete
479 is lower than of CEM I concrete due to low buffering capacity in CEM III/A by a lower pH in
480 pore solution.

481 In reinforced concrete, the ohmic resistance has an important influence on the corrosion rate
 482 and should be considered: CEM II/B-V concrete had the highest resistance followed by CEM
 483 III/A and CEM I concretes. Based on these measurements, CEM II/B-V concrete is the best
 484 binder in resisting chloride ingress and, once initiated, in slowing down the corrosion process.

Steel Bar Number.	CEM I		CEM III/A		CEM II/B-V	
	1	2	1	2	1	2
Ohmic resistance* (Ωm^2)	2.3	2.6	3.5	3.5	12.0	13.0
Polarisation resistance* (Ωm^2)	105.7	89.1	83.8	48.7	112.7	134.7
Corrosion potential (mV)	-99.8	-141.0	-94.8	-102.9	-96.2	-126.5
Time constant* (s)	212.1	142.7	84.7	48.9	226.8	293.7
Anodic slope** (mV/dec)	708.4	475.7	423.0	616.4	930.3	1082.9
Cathodic slope** (mV/dec)	70.2	62.9	71.0	79.1	72.2	70.7
<i>B</i> value (mV)	27.8	24.1	26.4	30.5	29.1	28.9
Corrosion current density ($\mu\text{A}/\text{m}^2$)	2.6	2.7	3.2	6.3	2.6	2.2

* Data obtained from Galvanostatic pulse technique

** Data obtained from Tafel slope

485 Table 3 Corrosion parameters of steel for samples at 18 years.

486

Potential (mV vs CSE)	Potential (mV vs Calomel electrode)	Probability of corrosion
>-200	>-126	Low (<10 %)
-200 - -350	-126 - -276	Intermediate
<-350	<-276	High (>90 %)

487 Table 4 Probability of corrosion from potential measurements [37].

488 4.3 Monitoring the Electrical Resistivity and Temperature of Field Concrete

489 The electrical resistivity is closely related to permeability/diffusivity of concrete according to
 490 the Nernst-Einstein equation [46]. The electrical resistivity of concrete is, on the other hand,

491 sensitive to both the ambient environment and material properties. Considering the age of
492 concrete used in this study, the influence of hydration on microstructure will be negligible;
493 however, other factors should be considered in evaluating concrete resistivity in the field
494 including temperature and degree of saturation and is developed below.

495 4.3.1. The effect of temperature on the resistivity

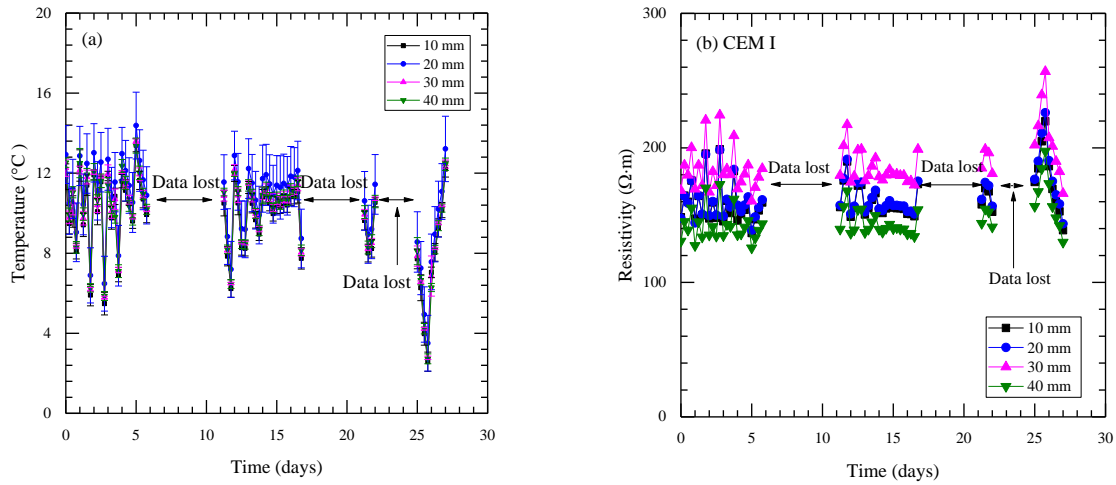
496 As conduction through concrete is, essentially, electrolytic in nature, it will be temperature
497 dependent. The influence of temperature on resistivity is linked through an Arrhenius
498 relationship [24] and, providing that the activation energy is known, the measured resistivity
499 can be *standardised* to an equivalent resistivity at a predefined reference temperature using
500 Eq. (10):

$$501 \quad \rho_R = \rho_M e^{\frac{E_a}{R} \left[\frac{1}{T_{K,R}} - \frac{1}{T_{K,M}} \right]} \quad (10)$$

502 where ρ_R is the resistivity (Ωm) at the reference temperature; $T_{K,R}$ is the reference
503 temperature (=293.15K in this study) (K); ρ_M is the measured resistivity (Ωm) at
504 temperature, $T_{K,M}$ (K); R is the gas constant (8.314 J/mol·K); and E_a is the activation
505 energy for conduction processes in concrete (kJ/mol).

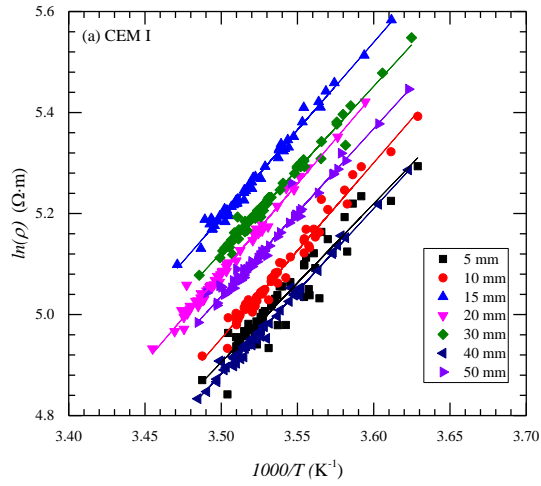
506 Fig 14(a) presents averaged ambient temperatures at discrete depths from the exposed surface
507 of the concrete slabs (error bars represent ± 1 standard deviation); over the time scale
508 presented, the temperature varied from 2.5°C to 13.9°C. From Fig. 14(b), it is evident that the
509 resistivity fluctuated in sympathy with the temperature – as the temperature increased, the
510 resistivity decreased and vice versa (note: in this Figure, for clarity, only the response from
511 10, 20, 30 and 40mm is presented).

512

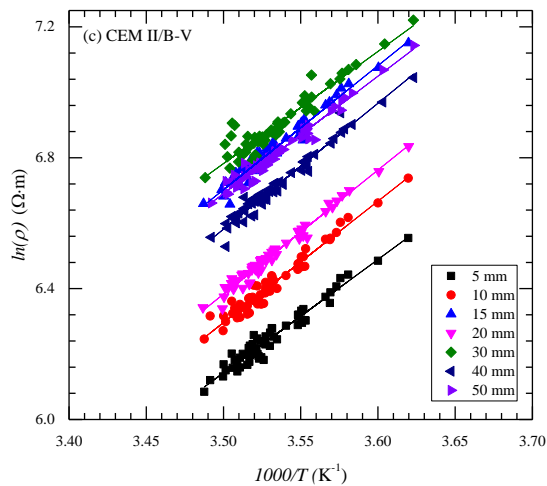
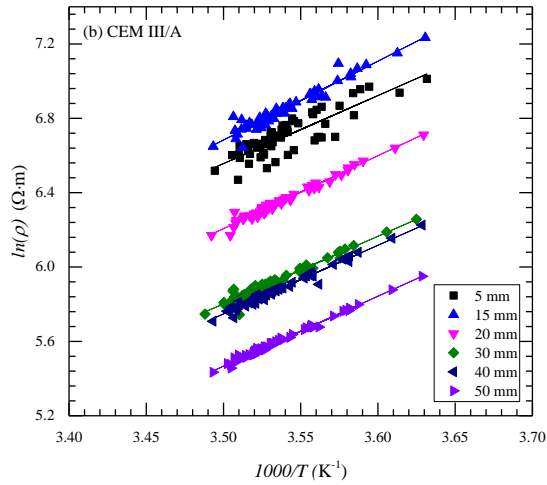


514 Fig. 14 Measurements obtained from remote monitoring of field concretes (XS3) showing the
 515 variation in (a) temperature and (b) electrical resistivity (CEM I concrete).
 516

517 Fig. 15(a)-(c) presents the temperature/resistivity measurements in an Arrhenius format with
 518 the activation energy obtained from the slope of the straight line through the data; Fig. 16
 519 presents the variation of activation energy with depth for each concrete type with the
 520 averaged values presented in the figure legend (27.94 kJ/mol for CEM I, 31.75 kJ/mol for
 521 CEM III/A and 30.46 kJ/mol for CEM II/B-V). Considering conduction occurs via capillary
 522 pore water, the low activation energy in CEM I concrete indicates a less tortuous and
 523 disconnected pore network compared to CEM III/A and CEM II/B-V concrete.



524



525

Fig 15. Field data plotted in Arrhenius format for (a) CEM I, (b) CEM III/A and (c) CEM II/B-V.

527

Note: data at 10mm on (b) was lost.

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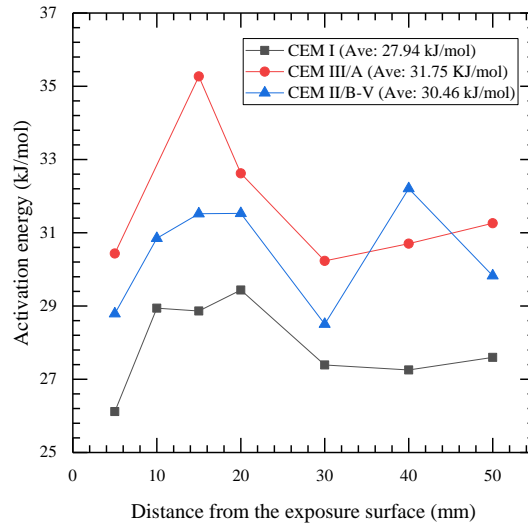
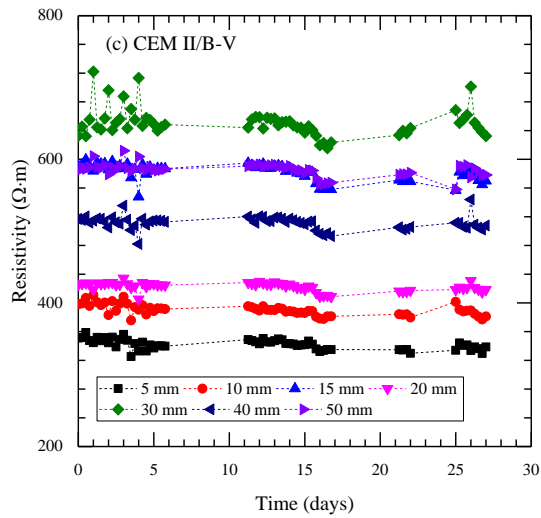
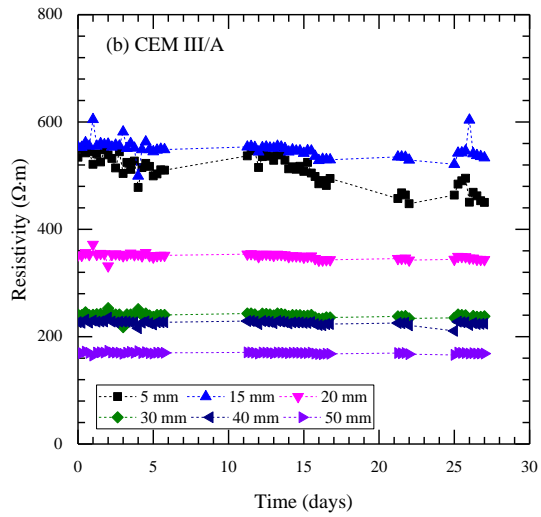
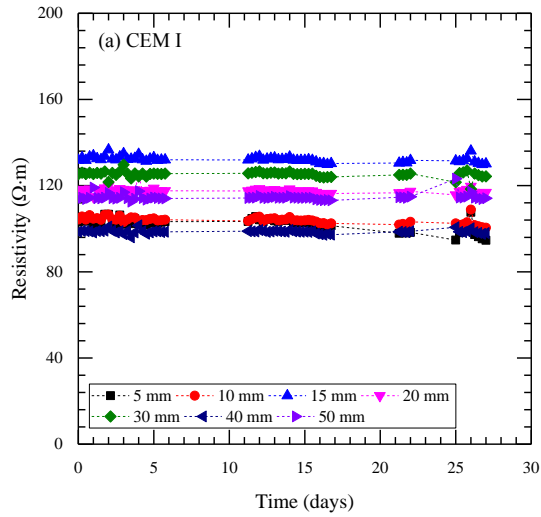


Fig. 16. Variation of activation energy with depth obtained from Fig. 15.

531
532

533 Fig. 17 shows the electrical resistivity of concrete after the correction using Eq. (10) and the
 534 average activation energy given above; the reference temperature was 20°C. During
 535 monitoring, some of the data were lost due to technical problems which remain unclear. It
 536 can be observed that the fluctuations in electrical resistivity observed in the uncorrected (i.e.
 537 raw) data become stable after the correction. As expected, CEM I concrete had the lowest
 538 resistivity; however, as the measured resistivity was $> 100\Omega\text{m}$, irrespective of type of binder
 539 and depth, Table. 5 would indicate that all concretes could be regarded as offering good
 540 protection to the steel reinforcement [47].

541



542
543
544

Fig. 17. Temperature corrected resistivity for (a) CEM I, (b) CEM III/A and (c) CEM II/B-V concretes. Note: data at 10mm on (b) were lost.

545

Resistivity (Ωm)	Probable Corrosion Rate
< 50	Very High
50 – 100	High
100 – 200	Moderate / low
> 200	Low

546 Table. 5 Empirical resistivity thresholds for protection of embedded steel reinforcement [47].

547

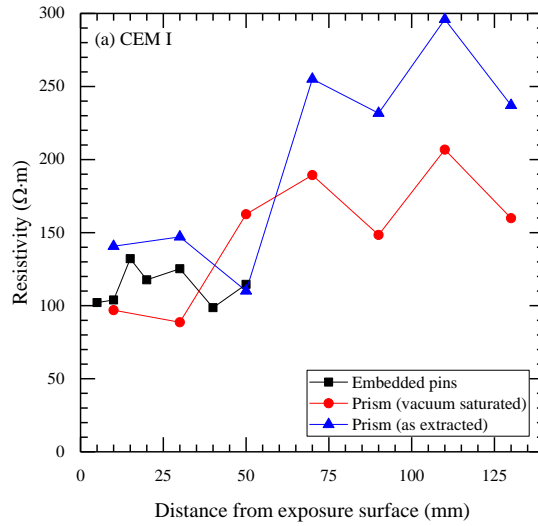
548 4.3.2 Prism and Field Resistivity Measurements

549 In this study, the resistivity of the prisms extracted from the slabs (i.e. the prisms used to
550 measure degree of saturation in Section 3.1.2) was compared to the resistivity obtained from
551 the slabs positioned at the marine exposure site using the remote interrogation facility. The
552 results are presented in Fig. 18. It was observed that the resistivity of prisms in the vacuum-
553 saturated condition is lower than the resistivity obtained from the embedded pins whilst the
554 resistivity of the as-extracted prisms is similar to the resistivity of embedded pins. This result
555 coincides with the result of degree of saturation. From the results, it could be inferred that the
556 field concretes in the XS3 environment are in an unsaturated state under natural exposure.
557 Again, considering that moisture movement is marginal, an increase in moisture content leads
558 to an increase in chloride transport. It is interesting to note that the difference in the resistivity
559 between the as-extracted prisms and vacuum-saturated prisms for the CEM II/B-V concrete is
560 considerably greater than the other two concretes perhaps reflecting the lower degree of
561 saturation of the as-extracted concrete prisms (see Fig. 10).

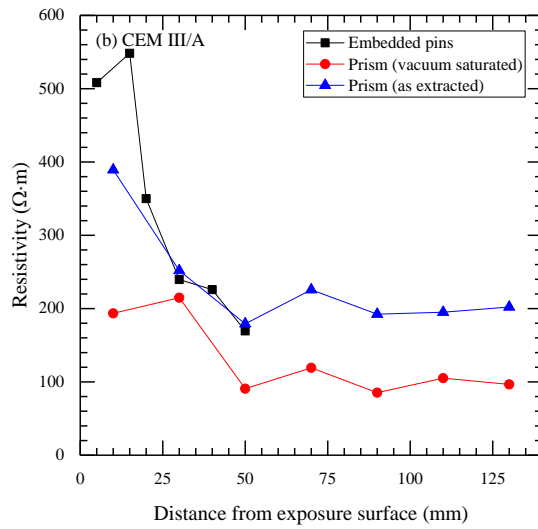
562 Finally, the electrical resistivity measurements using embedded pins, coupled with remote
563 interrogation, is an attractive method for real-time monitoring of the long-term performance

564 of concrete in the field; however, further work is required to improve data acquisition in this
565 respect. Considering that the chloride concentration in SCM concretes at 50 mm depth is low,
566 the increase in resistivity may be caused by a continuous pozzolanic reaction. This is
567 beneficial to the performance of concrete and may also affect the resistivity of pore solution
568 in concrete which is one of the factors which determine the resistivity of concrete.

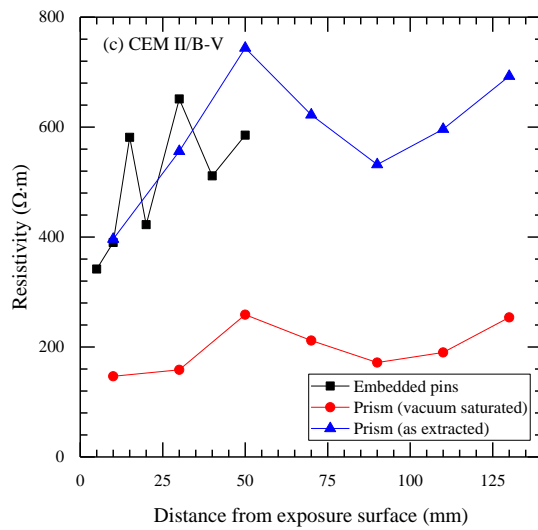
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Fig. 18. Comparison of electrical resistivity of prisms (as-extracted and vacuum saturated conditions) and embedded pins.

576 *4.4 Time to corrosion initiation considering environmental factors*

577 Prediction of the onset of corrosion in concrete was undertaken using the ClinConc model
578 [17]. This model describes chloride transport within concrete based on the principle of Fick's
579 law. The basic governing equation consists of the mass balance equation and a non-linear
580 chloride binding isotherm, as given by:

$$\frac{\partial q_{cl}}{\partial x} = -\frac{\partial}{\partial x} \left(D_0 \frac{\partial C_f}{\partial x} \right)$$

$$\frac{\partial C_t}{\partial t} = \frac{\partial C_f}{\partial t} + \frac{\partial C_b}{\partial t}$$

581 where, C_t , C_f and C_b are, respectively, the total, free and bound chloride concentrations,
582 q_{cl} is the net flux of free chloride per unit area and D_0 is the intrinsic diffusion coefficient.

583 The model input parameters include concrete mix proportions, binder type(s), curing and
584 environmental temperatures, the concentration of chloride in the solution to which a concrete
585 is exposed, and a 6-month chloride migration coefficient which needs to be obtained
586 experimentally from the non-steady migration test. Given that the ClinConc model is based
587 on diffusion, it is able to predict the chloride profile in concrete which, essentially, remains
588 fully saturated (i.e. completely submerged to seawater or below the mid-tide level). However,
589 it would be difficult to accurately determine an irregular boundary condition for concrete
590 exposed to tidal or splash zones where the concrete would be partially saturated and chloride
591 ingress due to direct contact with seawater tends to be random.

592 To take into account these aspects, two additional environmental factors have been
593 introduced [48]. From Fig. 10, the concretes used in this study were in an unsaturated state
594 (i.e. $S_r < 100\%$) and to reflect this condition, the environmental factors employed are:

- 595 (i) K_s which accounts for the reduction of diffusion coefficient with degree of saturation; and,
 596 (ii) K_{exp} which accounts for the degree of contact of the concrete surface with the external
 597 *chloride* environment (i.e. tidal cycles).

598 The procedure for estimating the factors was as follows;

$$599 \quad K_s = \frac{D_a}{D_0} = S_r^{4.863-3.441\frac{w}{b}} \quad (11)$$

$$600 \quad C_t = K_{exp} \times \Phi(C_f) \quad (12)$$

601 where D_0 is the diffusion coefficient in saturated concrete; and D_a is the diffusion
 602 coefficient in the unsaturated concrete including the degree of saturation, S_r , and w/b is the
 603 water-to-binder ratio, C_t is total chloride content, Φ is the functional relationship defined
 604 within the ClinConc model [17] and C_f is free chloride content.

605 In formulating Eq. (11), the empirical relationship relating diffusion coefficient, degree of
 606 saturation and w/b has been adapted from the work of Kumar [49]. The ClinConc model
 607 requires the 6-month migration coefficient for the concrete as an input parameter obtained
 608 from the NT Build 492 test [27]. All the input parameters used in the ClinConc model are
 609 presented in Table 6. The mix specification was the same as that used in the 18-year old
 610 sample. K_s can be estimated from Eq. (11) and the averaged degree of saturation from Fig. 10
 611 while K_{exp} values were found using the least square method between the measured and the
 612 predicted value. The details for the calculation procedure can be found in a previous study
 613 [48]. Fig. 19 presents the measured values and two predicted curves. One predicted curve is
 614 K_{exp} based on the best fitting for each type of concrete and the other is K_{exp} based on the
 615 averaged value for all three concrete mixes. As shown in the graphs, K_{exp} values are within a
 616 relatively narrow range 0.60-0.82 which is not surprising as the samples are in the same

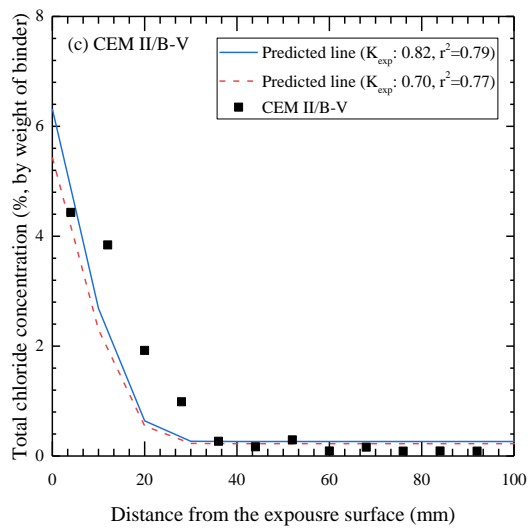
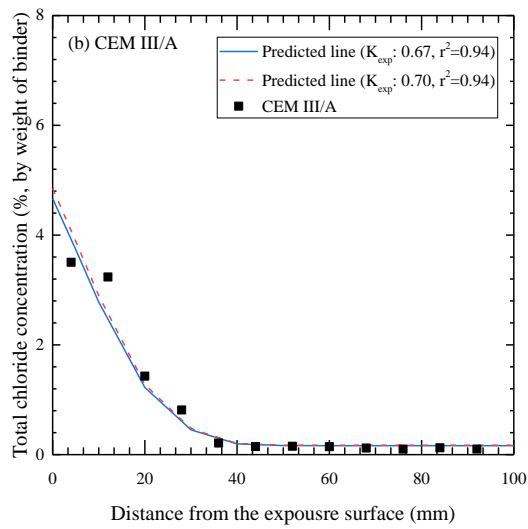
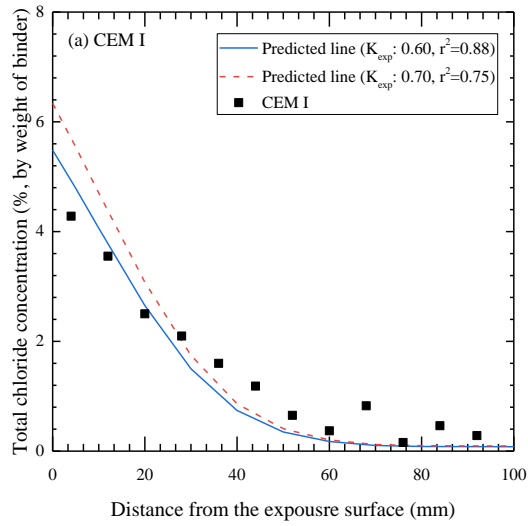
617 exposure condition and have a similar w/b (0.39–0.44). In addition, the difference between
 618 two predicted curves was marginal when the average value for K_{exp} was used.

	CEM I	CEM III/A	CEM II/B-V
Cement content(kg/m ³)	460	270	370
Slag content(kg/m ³) (<i>k</i> -factor*)	+	180 (0.6)	+
Fly ash content (kg/m ³) (<i>k</i> -factor*)	+	+	160 (0.4)
Water content (l/m ³)	184	198	206.7
Average saturation degree	0.83	0.73	0.67
Average Temperature	9		
Concrete age at first exposure (days)	35		
Exposure duration (years)	18		
Chloride concentration applied at the surface (g/l)	19.6		
Migration coefficient at 6 months $\times 10^{-12}$ (m ² /s) (Standard deviation)	8.87 (± 0.2)	3.33 (± 0.5)	2.92 (± 0.3)

* *k*-factor used in calculating equivalent binder content where the term ‘w/b’ is now w/(cement + k×SCM) defined in BS EN 206 [2]

619
 620
 621

Table. 6 Input parameters for the ClinConc model.



622 Fig. 19 Measured profile and predicted chloride profiles from the modified ClinConc model for (a)
 623 CEM I, (b) CEM III/A and (c) CEM II/B-V concrete subjected to XS3 environment.

624 The time to corrosion initiation was calculated with the average value as shown in Table. 7.
 625 CTLs can vary over the wide range, typically from 0.2% to 2.2%; hence, two CTLs were
 626 employed: 0.4% as the lower-bound value and 2.0% as upper-bound value. The time to attain
 627 the CTL at 50mm depth was highest for the CEM II/B-V concrete and increased by a factor
 628 of ~3.2 when the CTL increased from 0.4% to 2.0%. SCM concrete is thus beneficial in
 629 resisting chloride induced corrosion due to its low diffusivity. To increase the service-life of
 630 concrete structures subjected to marine environment, an increase in the acceptable CTL for
 631 the concrete can be an alternative as shown in the result; however, as the CTL is still
 632 controversial and is influenced by numerous factors [13, 14, 40] it is significantly difficult to
 633 control.

CTL*	Predicted time to corrosion initiation (year)		
	CEM I	CEM III/A	CEM II/B-V
0.4	17.8	49.8	114.3
2.0	55.6	179.5	342.5

634 * Chloride threshold level (% by weight of binder).

635 Table 7 The predicted time to attain the chloride threshold level at the steel depth.
 636

637 5. CONCLUSIONS

638 In the current study, a comprehensive suite of tests was undertaken on concretes, with and
 639 without SCMs, subjected to a marine environment (XS3 exposure). As the concrete slabs
 640 were 18 years old, the data are representative of the performance of mature concrete, which is
 641 of importance in the development of performance-based specifications for concrete. The
 642 degree of saturation, porosity and chloride concentration were measured, together with their
 643 variation with depth from the concrete surface. Regarding the electrochemical condition of
 644 the steel, corrosion rate and half-cell potential were also evaluated. Based on chloride profiles,

645 the service life of reinforced concrete was predicted using the ClinConc model combined
646 with two environmental factors to take account for concrete in an unsaturated state and cyclic
647 exposure to chlorides. The following general conclusions can be drawn from the work:

648 (1) Regarding the degree of saturation and diffusion coefficient, CEM I concrete had the
649 highest, but the porosity of CEM I concrete was the lowest. To evaluate the resistance
650 of chloride transport in reinforced concrete, pore structure, i.e. tortuosity and
651 connectivity, should be considered rather than overall porosity. It was confirmed that
652 SCMs concretes were beneficial to resist chloride transport due to disconnected pore
653 structure caused by continuous hydration and high chloride binding.

654 (2) Although the chloride concentration at steel depth was 0.8% by weight of binder for
655 CEM I concrete, corrosion of steel was not detected. Corrosion potentials for all
656 concretes were $> -140\text{mV}$ vs CSE corresponding to $<10\%$ in probability of corrosion
657 as specified in ASTM C876. The B -values ranged from 24.1 to 30.5mV irrespective
658 of type of binder. It is considered that a low B -value resulted from high availability of
659 oxygen as concretes were exposed to the tidal and splash zones.

660 (3) To reflect the effect of temperature on electrical resistivity, the activation energy was
661 evaluated with the field concrete. It was observed that the activation energy varied
662 with depth and type of binder, ranging from 27.94 to 31.75kJ/mol. It was confirmed
663 that the activation energy was also influenced by contamination of chloride and
664 moisture-content. Depending on type of chlorides present in concrete (i.e. free/bound),
665 the electrical resistivity, especially in the surface region, increased or decreased.

666 (4) The electrical resistivity of all concretes was $> 100\ \Omega\text{m}$, indicating that the probability
667 of chloride-induced corrosion of steel in the reinforced concrete is low. From

668 comparison of resistivity between prisms and field samples, all concretes were in an
669 unsaturated state in their natural condition. In addition, it was confirmed that moisture
670 movement is marginal from the stable values in the resistivity, irrespective of depth
671 and binder type although field concretes investigated in this study were exposed to
672 tidal action.

673 Evaluating the long-term performance of concrete in the field yields valuable information as
674 the data reflect both exposure condition and material properties. As the exposure condition
675 cannot be exactly simulated within a laboratory environment, field monitoring assumes even
676 greater significance. Electrical resistivity represents an important candidate testing
677 methodology for non-destructive (and real-time) monitoring of the performance of as-placed
678 concrete and warrants further development.

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