

## Review

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# Property changes in thermo-hydro-mechanical processing

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**Abstract:** Thermo-hydro-mechanical (THM) treatment is a combined action of temperature, moisture, and mechanical force, which leads to modified wood (THMW). Various types of eco-friendly THM processes have been developed to enhance wood properties and generate new materials, such as welding, densification, molding, bending, profiling, artificial aging, panel manufacture, and surface densification. The various transformation processes in the course of THM bring about positive effects in terms of the mechanical and physical properties as well as the biological durability. To the negative effects belong the loss in strength and fracture toughness, and one of the challenges is to minimize these negative aspects. The present paper reviews the chemical transformations processes during THM treatment in a closed processing system and presents the relationship between processing parameters and THMW properties. The discussion includes the problems associated with eliminating the set recovery of densified wood by THM posttreatments and the chemical origin of the relaxation of internal stresses induced by densification.

**Keywords:** chemical-mechanical, chemistry, densification, fixation of compression set, moisture content (MC), temperature, wood

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## Introduction

Thermo-hydro-mechanical (THM) treatment of wood is one of the emerging eco-friendly methods in wood technology, which is based on the combined treatment of wood by elevated temperature, moisture, and application of mechanical forces. These processes can be subdivided into thermo-hydro (TH) and thermo-hydro-mechanical (THM) treatments. TH treatments (THT) change the intrinsic wood properties and dissipate internal stresses, while wood is dried and softened. THT also improves the properties of wood composites. THM treatments (THMT), on the other hand, are employed in molding, shaping, bending, profiling, welding (by friction), and densification of wood. The goals of THT and THMT are different, and for this reason, this paper aims to review and summarize details related solely to THMT and will focus on the chemical changes during the process and their relationship to the properties of THM wood (THMW). The unavoidable side effects of THMT will also be discussed.

Many THM techniques have a historical background. For example, the bending of wood in the longitudinal direction was known in Egypt around 1000 B.C. (Rivers and Umney 2005). In 1850, Michael Thonet (1796–1871) applied steam to soften beech wood and protected the side of the wood submitted to tensile forces during bending by steel strapping. Chair manufacture by this technique is still very successful. Hanemann (1917, 1928) developed a technique to compress plasticized wood along the fiber direction (crushing), which makes wood pliable, and the method found industrial application in the 1990s (Morsing 2000). Compressing softened wood in the transverse direction at high temperature leads to densified wood, which was patented by Stöckhart in 1886 (Vorreiter 1942).

Sandberg et al. (2013) reported the state of the art on THMW up to 1960. It was recognized that densified wood suffers from the “shape memory” or “compression set recovery”. To eliminate this effect, Stamm and Seborg

(1941) developed a multi-layered wood composite (called Impreg) consisting of thin veneers impregnated with phenol formaldehyde resin, which was densified under heat and pressure, while the resin eliminated the compression set recovery. Stamm et al. (1946) and Seborg et al. (1945) observed that a thin plate of densified wood, when treated at 180°C for 30 min, shows reduced compression set recovery following re-wetting and re-heating and called their successful THMW “Staypak” (Kollmann et al. 1975).

Since the 1990s, a renewed interest has arisen in developing THM products. In Japan, Norimoto et al. (1993), Ito et al. (1998a,b), and Tanahashi et al. (2001) adopted the THMT for Staypak production. Reportedly, the THM post-treatment of thin densified wood at 200°C and under conditions of saturated moisture for 4 min eliminated totally the shape memory. Ito et al. (1998a) developed a system to transform small round section wood trunks into trunks with squared cross sections. Another densification process was designed to obtain structural timber of high quality from low-density wood (“compressed lumber processing system”, CLPS). In the first mentioned technique, the stages of moldings and posttreatment follow each other in a continuous process, whereas in CLPS production, the two stages are separated. However, the scaling up of the research results turned out to be difficult. Kyomori et al. (2000) pointed out that more fundamental knowledge is needed for the successful application of THMT in industrial scale.

In Denmark, a machine was developed for the pre-compression of wood in the longitudinal direction based on the idea of Morsing (2000). This type of wood element can be bent without the application of steam or jigs. In Switzerland, Navi et al. (1997, 2000), Heger (2004), Navi and Heger (2005), and Popescu et al. (2013) investigated the set recovery of densified wood fabricated in a closed system. In Germany, a system for posttreatment of THM densified wood was developed by means of heating oil (rapeseed oil at 200°C, OHT). Haller and Wehsener (2004) and Welzbacher et al. (2008) investigated the suitability of OHT for large-scale industrial densification by focusing on the elimination of compression set recovery and found that the compression set recovery of large densified spruce panels was completely eliminated. The product has improved resistance to biodegradation, but the mechanical strength is considerably reduced. In Canada, Fang et al. (2011) have also investigated the effects of OHT on densified wood veneer. In the USA, Kamke (2004) and Kutnar et al. (2008) densified fast-grown, low-density wood to composite panels. A semi-open THM reactor was developed for densifying thin layers of hybrid poplar specimens by Kamke and Sizemore (2005).

Militz and Lande (2009) summarized that only a few TH techniques could be industrialized, and the same conclusions were drawn by Kamke (2013) and Sandberg et al. (2013) in terms of THMT. As pointed out previously, more basic knowledge is needed to change this situation. In view of the complexity of the chemical and physical changes of wood in the course of THMT, it can be stated that THM processing is a highly research-oriented technology. Large wood elements behave differently to the small-sized wood elements tested in the laboratory. This is especially true for the detrimental compression set recovery of densified wood. Moreover, the shape memory effects of products manufactured in open systems cannot be compared with the results of materials manufactured in a closed system. Only the closed system provides the possibility of controlling all processing parameters, namely, temperature, MC, and force. Moreover, in a closed system, the moisture content (MC) can be varied from zero to saturated conditions in the course of the whole THM processing including the posttreatment. This is the reason why in this paper, only the chemical changes and subsequent mechanical and physical properties of THMW prepared in a closed system will be reviewed and discussed.

The main aim of this paper is to contribute to a better understanding of the chemical changes that the polymeric constituents of wood undergo during THMT. In other words, the focus is on the chemical origin of the elimination of compression set recovery and the relationship between THMT parameters, and the physical and mechanical performance of the products obtained in a closed system should be illuminated.

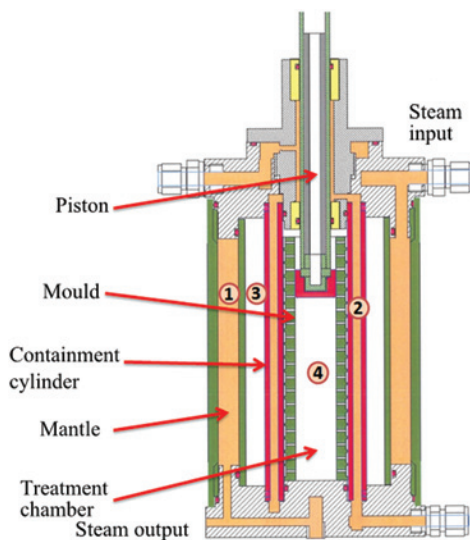
## Methods for THM processing

Densification can be carried out in open, semi-open, or closed systems. In an open system, the process runs at atmospheric pressure, 130°C, and the MC is limited to 13% (Welzbacher et al. 2008). Under this condition, the shape memory effect is pronounced. In a closed system, there is a pressurized gas environment, and thus, the MC can attain saturation conditions, and the temperature can be increased close to pyrolysis conditions. In a semi-open system, called viscoelastic thermal compression (VTC), some stages are controlled by pressurized gas and the others by atmospheric pressure (Kamke and Sizemore 2005). Navi and Sandberg (2012) described examples of the THM processing systems and pointed out the advantages of closed systems that allows for

exact controlling the parameters also in terms of shape memory.

The closed system equipment for densification or molding of wood consists of a reactor, a press, and a high-temperature steam generator (Figure 1). This multi-chamber device is suitable for densification of small wood elements with dimensions  $15 \times 4 \times 4 \text{ cm}^3$  (r, t, l), followed by a posttreatment in saturated or nonsaturated steam at temperatures up to  $200^\circ\text{C}$ .

Treatments often comprise four stages: plasticization, densification, posttreatment, and cooling. A schematic overview of THM processing, including posttreatment is given in Figure 2.



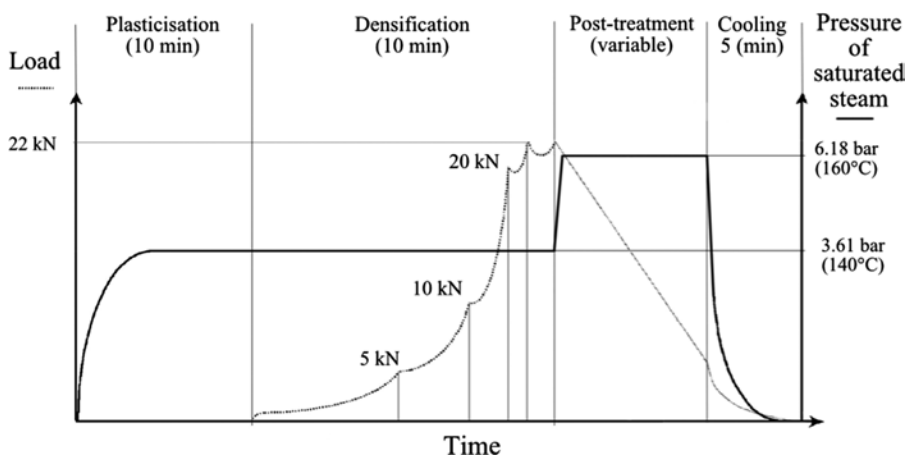
**Figure 1:** Multichamber THM closed reactor. Steam injection into the containment cylinder (3 and 4) and the mantle (yellow circuit 1 and 2) are shown (designed by RINO SärI, Switzerland).

## Compression set recovery as a function of temperature and time

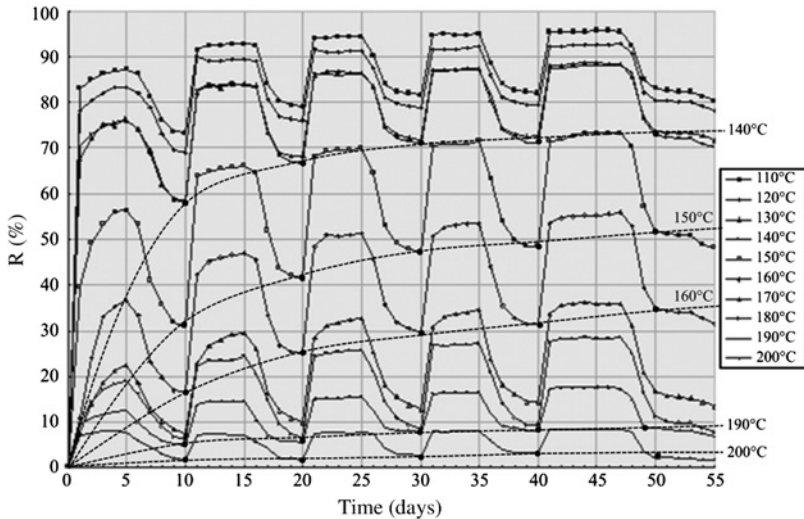
Densified wood is submitted to soaking-drying cycles, and the compression set, short-recovery  $R$  observed provides information about the dimensional stability as a function of the parameters (Norimoto et al. 1993; Takahashi et al. 1998; Ito et al. 1998a). Heger (2004) observed the influence of temperature and time separately in two experimental series. In the first one, small wood specimens were densified and posttreated between  $110^\circ\text{C}$  and  $200^\circ\text{C}$  under saturated steam conditions. The postprocessing time was 9 min, and compression set was at ca. 60%. The recovery test consisted of five cycles, which involved immersing the compressed samples in water at  $60^\circ\text{C}$  for 5 days and then drying at  $30^\circ\text{C}$  for 5 days. Finally, the samples were totally dried over a period of another 5 days. The recovery  $R$  was determined based on the following relationship:

$$R = \frac{X'_c - X_c}{X_o - X_c} \times 100 \quad (1)$$

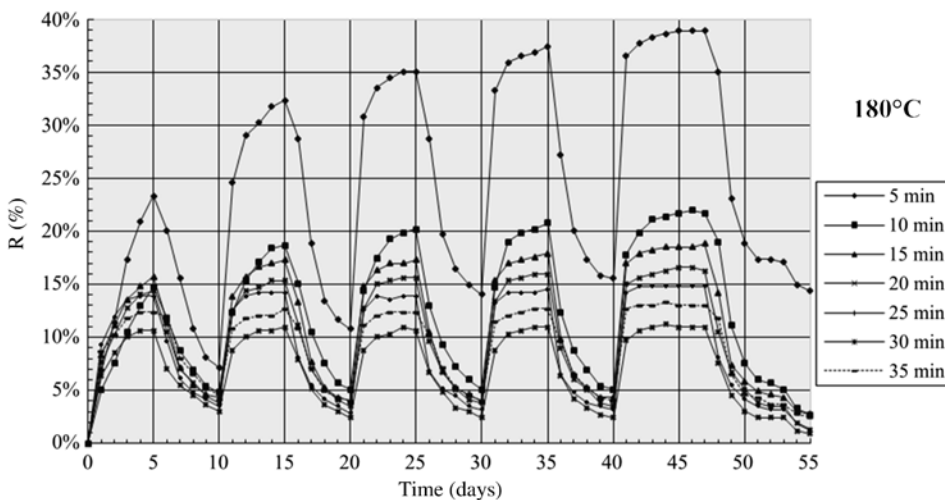
where,  $X_o$  is the initial length of specimen before densification,  $X_c$  is the length after densification, and  $X'_c$  is the length of the specimen after recovery test. The variation of  $R$  (Figure 3) indicates that each treatment modifies the wood with new moisture-related properties that are different from native wood and to each other. It is important to note that the value of the  $(X'_c - X_c)$  represents the sum of two deformations accounted for moisture-related swelling and compression set recovery. To separate these two deformations, curves have been constructed for each temperature (shown as the dashed-line) by connecting the origin of Figure 3 to the points at the end of each cycle after



**Figure 2:** Diagram of THM processing of small elements: THM densification and posttreatment under saturated steam.



**Figure 3:** Set recovery  $R$  measured during soaking-drying cycles of densified and posttreated wood at saturated vapor under different temperatures for similar processing time and compression rate of  $10 \text{ mm min}^{-1}$ . Curves with dashed line show the variation of compression set recovery for the indicated temperature.



**Figure 4:** Set recovery  $R$  during soaking-drying cycles;  $R$  curves of densified samples treated at saturated vapor at  $180^\circ\text{C}$  under different processing times.

10, 20, 30, 40, and 50 days. These dashed line curves indicate the variation of  $R$  (%) of the densified wood, which are smaller if the compressed wood is treated at higher temperatures. Figure 3 also shows that the moisture-related behavior becomes entirely fixed only when the specimen is densified at  $200^\circ\text{C}$ . When the  $R$  is about 2%, the recovery is considered to be completely eliminated.

Another experimental series of Heger (2004) under constant temperature at different postprocessing times between 5 and 35 min are presented in Figure 4 showing  $R$  after five cycles. Accordingly, longer posttreatment times entail smaller  $R$ . The recovery is entirely eliminated at  $180^\circ\text{C}$  with 10 min posttreatment time.

## Chemical degradation as a function of processing parameters

Chemical changes in heat-treated wood were reported, among others, by Windeisen et al. (2009). No doubt, chemical analysis contributes a lot to estimate the optimal THT (Hill 2006; Militz and Lande 2009). Wood degrades faster when heated by steam or water (Millet and Gerhards 1972; Hillis 1975). Heat treatments are mainly conducted in a dry environment in an inert gas or in a moist environment with steam at temperatures up to  $240^\circ\text{C}$ . Under these conditions, the hemicelluloses are hydrolyzed, and



**Table 1:** Percentage of neutral monosaccharide in treated and native wood from the GPC spectra (Heger 2004).

Specimen	Sugars by hydrolysis (%)				
	Man	Xyl	Gal	Ara	Total
Ref. (natural spruce)	13.6	5.6	2.8	1.2	23.2
Only densified	9.7	4.7	1.2	0.7	16.3
Densified+posttreated (180°C, 30 min)	5.6	2.0	0.5	elim. <sup>a</sup>	8.1
Densified+posttreated (200°C, 5 min)	5.4	2.8	0.4	elim. <sup>a</sup>	8.6

<sup>a</sup>Eliminated.

the crystallinity index of cellulose increases, but lignin is only slightly affected (Pelaez-Samaniego et al. 2013). The pyrolysis of hemicelluloses begins at about 270°C followed closely by cellulose (Rowell et al. 2009). The THM processing diagram (Figure 2) illustrates that the period of plasticization and densification for small elements is short even at lower temperatures. However, the posttreatment temperature must be higher, and the treatment time depends on the MC. It was demonstrated that at higher temperature and MC with long processing time, the wood constituents can be seriously degraded, and as a result, the mechanical and fracture strength of THM wood is lowered.

## Chemical degradation of hemicelluloses

The pH of softwood varies between 4 and 6 (Sandermann and Rothkam 1959). For example, the pH of spruce (*Picea abies* Linn) is 5.3 at ambient temperature. At elevated temperatures, the dissociation constant of water is higher (Sweeton et al. 1974; Verma 2003). Thus, during THM and particularly during posttreatment, the pH of the water inside the wood at 200°C is around 5.8, based on data from the International Critical Tables. The degradation reactions liberate acetic acid, the concentration of which is almost constant during processing. According to Hsu et al. (1988), the pH of pine decreases rapidly in the beginning of the heat treatment and stabilizes at about 3.5. The major reason for pH decrement is deacetylation of hemicelluloses, which is the best established reaction in wood chemistry.

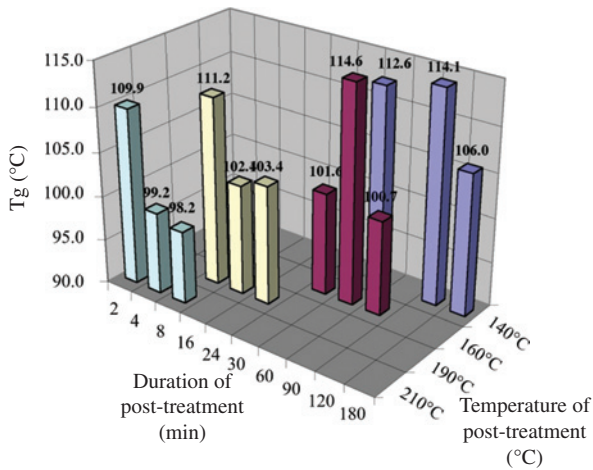
Under these acidic conditions, the hemicelluloses are easily hydrolyzed, their DP is lowering, and mainly xylan fragments down to monomeric sugars can be solubilized by water (Fengel and Wegener 1984). Typical yield of neutral monosaccharides (glucose, galactose, mannose, xylose, and arabinose) of native spruce such as densified spruce, and densified posttreated spruce are

presented in Table 1 (Heger 2004). Accordingly, the mass loss (ML) is 6.9% after densification and washing of the specimens. The ML increases significantly when the specimen undergoes posttreatment, i.e., ML is 15.1% and 14.6% after posttreatment at 180°C for 30 min and at 200°C for 5 min, respectively. The 175-ppm NMR signal characteristic of acetyl -COOH groups of hemicelluloses decreases markedly upon THMT, which is a proof that the increasing acidity is due to hemicellulose degradation. This effect is also seen by the decrease of the 20- to 22-ppm signal characteristic of the -CH<sub>3</sub> group of acetic acid and by the corresponding FTIR bands.

## Chemical degradation of lignin

The degree of cross-linking and glass transition temperature ( $T_g$ ) of lignin might be affected by THMT. According to Heger (2004), the  $T_g$  of lignin between 140°C and 210°C decreases with increasing treatment time (Figure 5). This might be interpreted as a partial depolymerization of lignin in the initial phase and degradation during posttreatment. Microcracking occurs between cells in the lignin-rich intercellular region during posttreatment at 200°C and above. During wood welding, similar observations were made.

Solid phase CP MAS <sup>13</sup>C NMR spectra reveal the cleavage of linkages at C $\alpha$  and 4-O, and demethylation and demethoxylation reactions are also typical. NMR analysis confirms that initially, hydrolysis occurs only on the positions C3 and C5 but not on the C1 and C4 of the ring (Delmotte et al. 2008). The appearance of phenolic hydroxyl groups, as a result of lignin splitting, was also shown during welding, which is as a special type of THMT and which also leads to polycondensation of the aromatic nuclei accompanied by formation of Ar-Ar and Ar-CH<sub>2</sub>-Ar bridges (Delmotte et al. 2008). The  $T_g$  of lignin may be affected in two ways, while during welding cross-linking reactions are typical. Under HTM conditions in the heating phase, however, the



**Figure 5:** Tg of the lignin of densified and posttreated specimens in saturated steam, function of the temperature, and posttreatment time.

situation is more complex. The lignin rearrangement leads to (1) a decrease in the Tg of lignin due to its partial depolymerization, and (2) to a change in Tg due to the internal rearrangements without cross-linking that would occur during dry heating. A notable increase in the NMR signals between 190 and 220 ppm indicates the formation of aldehydes. The predominance of the NMR bands at 191–192 ppm and particularly at 196 ppm are for formaldehyde, furfural, or methyl furfural. The considerable rise of the 148-ppm and 152-ppm NMR signals during wood welding confirms the formation of considerable proportions of furfural (Delmotte et al. 2008). The formation of aldehyde groups implies extensive depolymerization of both lignin and carbohydrates and can also be interpreted as sign of lignin rearrangement, which is obvious by Tg change.

## DP of cellulose

The degree of polymerization of the cellulose decreases during THM posttreatment as measured by capillary viscosimetry (Table 2). The average  $D\bar{P}_v$  value can be determined from its intrinsic viscosity measured in cupriethylene-diamine (CED) based on French norm AFNOR, Heger (2004).

## Fixation of compression set

Salmén and Burgert (2009) and Stevanic and Salmén (2009) reviewed the supramolecular architecture of the cell wall with the special arrangement of cellulose, hemicelluloses,

**Table 2:** Cellulose overall viscosimetric degree of polymerization  $D\bar{P}_v$  of three spruce samples.

Specimen	$D\bar{P}_v$
Only densified	1533
Densified+posttreated (180°C, 20 min)	949
Densified+posttreated (200°C, 4 min)	837

and lignin. In this model, the amorphous hemicelluloses adjacent to the microfibrils (MF) are connected by hydrogen bonds and joined to the lignin mainly by covalent bonding. The cellulose domain consists of crystalline and semi-crystalline regions. With the exception of the crystalline cellulose, nearly all regions of the cell wall are plasticized during the densification process under the combined action of steam and high temperature (over 100°C) (Back and Salmén 1982). When wood is transversally compressed in these conditions, elastic and elastic-viscoplastic deformations occur. The elastic energy stored in the helical semi-crystalline microfibrils and the lignin is rather high; the applied compression stress on the sample at the end of the densification reaches about 17.5 MPa. This stored elastic energy is the main cause of the compression set recovery, R.

Two mechanisms account for the fixation of R: After the hot densified wood is cooled down below the Tg of lignin, the lignin passes from the rubbery to the glassy state, and the deformed cellulose is confined in this rigid matrix. During drying, the formation of hydrogen bonds between the cell wall polymers also contributes to the fixation of the deformed state. The fixation is not permanent (Navi and Heger 2005). After re-humidification/heating, the hydrogen bonds disrupt, and lignin may become rubbery again. The release of the elastic energy leads to total or partial R of the densified wood.

Norimoto et al. (1993) and Morsing (2000) proposed three methods for avoiding R: 1. To prevent wood from being re-softened by changing the hygroscopicity of the cell wall; 2. Promoting formation of covalent crosslinks between wood components in the deformed state; 3. To relax the stresses stored in the microfibrils and matrix polymers during densification. The latter can be considered as the most promising approach in practice.

## Fixation of compression set by THM treatment

To obtain stable densified wood without R or with low R, a THM posttreatment is necessary. Fixation by means of saturated steam was investigated by Van Niekerk and Pizzi

**Table 3:** Posttreatment processing time for complete fixation of the compression-set (R=2) at different temperatures (T) and relative humidities (RH).

T (°C)	Fixation time (min) at RH				
	0%	40%	60%	80%	100%
140	No fix. <sup>a</sup>	6000	No fix.	2020	210
160	24250	No fix.	No fix.	135	70
180	5680	341	167	50	20
200	No fix.	No fix.	43	16	4

<sup>a</sup>No fixation.

(1994), Inoue et al. (1993), Dwianto et al. (1997), Ito et al. (1998a,b), Heger (2004), Navi and Girardet (2000), and Navi and Heger (2005). Heger (2004) and Groux (2004) observed the parameters influencing R in a closed thermo-hydro-mechanical multichamber reactor (Figure 1).

Table 3 gives the posttreatment time necessary for complete fixation of compression-set at different temperatures under saturated and unsaturated conditions. Obviously, a complete fixation of R (R=2%) at any temperature and RH could not be achieved. When the specimens with low MC were kept for more than 50 min at 200°C in the reactor, the specimens started to burn at 0 and 40% RH. Probably, complete fixation was unsuccessful under these conditions because the lignin remained glassy (Salmén 1982).

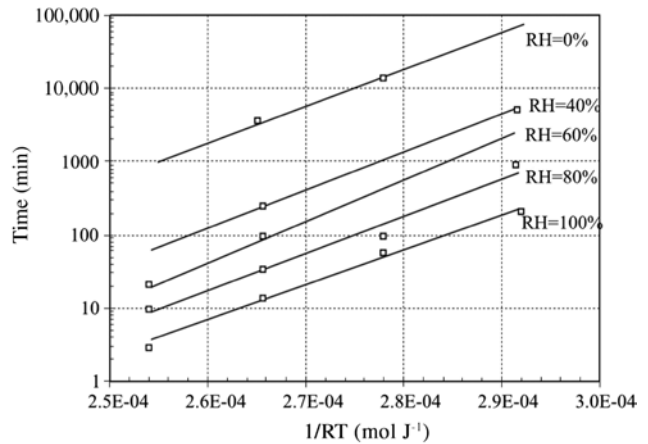
## The role of hemicelluloses hydrolysis on the fixation

The effect of the THM parameters on R (Table 3) is illustrated in Figure 6. As indicated, the posttreatment time (log scale) is a function of temperature and RH. For a given RH, the logarithm of time varies linearly with respect to the inverse of the absolute value of temperature. This regularity suggests a mathematical relationship between the time, temperature, and RH for eliminating shape memory, or relaxation of the strain energy accumulated during densification. The plot in Figure 6 allows the calculation of the activation energy ( $E_A$ ) of the process responsible for eliminating the shape memory effect for each RH:

$$t(T, h) = \alpha e^{E_A / RT} \quad (2)$$

where,  $\alpha$  is a parameter that depends on RH and pH of the water during the posttreatment.

$E_A$  is measured from the slope of the lines at different RH and are given in Table 4. The value is roughly constant at around 98.5–118.4 kJ mol<sup>-1</sup> and independent of RH. These  $E_A$  data are very close to the value  $E_A=118$  kJ mol<sup>-1</sup>



**Figure 6:** Value of posttreatment time of complete fixation of the compression set vs.  $1/(RT)$  and relative humidity (RH) for spruce wood.

**Table 4:** Values of the activation energy  $E_A$  (kJ mol<sup>-1</sup>) at various relative humidities derived from Figure 6.

	Relative humidity (%)				
	100	80	60	40	0
$E_A$ (kJ mol <sup>-1</sup> )	98.5	116.6	118.8	111.5	118.4

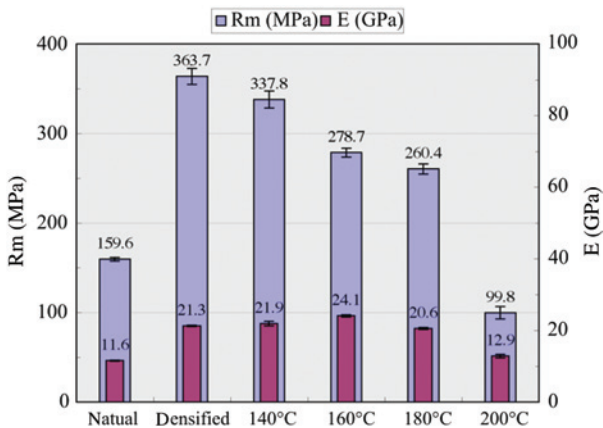
found by Springer (1966) for the hydrolysis of xylan. The low value measured under 100% RH might be due to the easy initiation of hydrolysis in the presence of saturated steam. There are several literature data available with this regard. Experiments were conducted by Mittal et al. (2009), Nabarlantz et al. (2004), Garrote et al. (1999), and Grénman et al. (2011) concerning the extraction of hemicelluloses from various wood species (Table 5). Accordingly, the  $E_A$  varies with the experimental parameters and the wood species. The result obtained by Grénman et al. (2011) on spruce can be compared with the one calculated from the results given in Figure 6 for spruce. The difference between the two results might be attributed to the different extraction methods. Grénman et al. (2011) used a cascade type reactor with solid-liquid aqueous extraction, whereas Heger (2004) used a single reactor and saturated steam.

Obviously, the hydrolysis of hemicelluloses as manifested by ML, the subsequent lowering of hydrophobicity and reactions of furfural leading to cross-linking reactions, increasing of porosity, and mainly microcracks and new surfaces in the hemicellulose matrix, play an essential role for low R values. Table 1 shows about 2/3 ML of hemicelluloses. This finding was also presented by Van Niekerk and Pizzi (1994), who checked more than 29 000 full-scale industrial panels made of eucalyptus wood and

**Table 5:** Activation energies  $E_A$  (kJ mol<sup>-1</sup>) and the experimental parameters (Grénman et al. 2011).

Parameter	Wood species			
	Sugar maple <sup>a</sup>	Corn cob <sup>b</sup>	Eucalyptus <sup>c</sup>	Spruce <sup>d</sup>
Chip thickness (mm)	<0.6	1	<8	1.25-2
Temperature (°C)	145–185	150–190	145–190	150–170
Liquid/solid ratio	20	8	6–10	160
$E_A$ (kJ mol <sup>-1</sup> )	114–117	127–251	82–156	150–170

<sup>a</sup>Mittal et al. (2009); <sup>b</sup>Nabarlatz et al. (2004); <sup>c</sup>Carrote et al. (1999); <sup>d</sup>Grénman et al. (2011).



**Figure 7:** Short-term strength (Rm) and longitudinal Young's modulus (E) of natural spruce, densified, densified, and posttreated at 140°C during 3 h, 160°C during 1 h, 180°C during 16 min, and 200°C during 4 min under saturated steam.

found evidence that hemicelluloses hydrolysis is the main cause of R fixation. This effect was greater in the case of high MC of the wood furnish. The compression strength of the eucalyptus wood cells is relatively high, and the cell walls are only deformed under standard particleboard producing conditions, which is a source of considerable residual strain in the particles. However, if much higher specific pressure is applied at higher MC of the furnish, hemicelluloses are more degraded, then more micropores arise, and the residual strain is lowered.

## Effect of processing parameters on the wood mechanical properties

The effects of processing parameters on longitudinal strength and Young's modulus of THMW are exemplified in Figure 7 for spruce wood according to Heger (2004). Here, the density of the wood and consequently its longitudinal tensile strength (without posttreatment) were

increased by a factor of 3. As indicated, the tensile strength was 364 MPa, which is by a factor of 3 is less than that of natural spruce. Posttreatment at 140°C leads to a loss of about 7% of the wood strength, but the reductions are greater at 160°C, 180°C, and 200°C, where the corresponding losses are 23%, 28%, and 73%, respectively.

The high stress losses at higher temperatures are partly due to microcracking in the samples, which are prominent around 200°C. In fact, the pressure of saturated steam increases exponentially with temperature. The pressure of the saturated steam at 200°C is about 16 kg cm<sup>-2</sup>, whereas it is only 4 kg cm<sup>-2</sup> at 140°C. The middle lamella becomes also fragile above 190° due to lignin softening (Navi and Sandberg 2012), which leads to fractures and heavy distortions relative to the other cells. Microfissures during processing and drying contribute a lot to strength losses.

Theoretically, the Young's modulus (MOE) of the three times densified spruce should be triplicated, but MOE increased only by a factor of 1.9. Table 1 shows that about 2/3 of hemicelluloses was washed out by water in the densification stage that generated pores. MOE is sensitive to variations in the material porosity. The MOE was increased by 3% at 140°C and 13% at 160°C. Above 160°C, the MOE begins to decrease for the effects indicated above. It can be safely concluded that the increased number of pores in the wood are responsible for MOE losses, though the cell wall crystallinity index,  $C_rI$ , and the diameter of cellulose crystallite are increased (Heger 2004).

## Conclusions

Various types of THM processing technique can be utilized to enhance wood properties and produce eco-friendly new materials. The modification of mechanical properties, dimensional stability, resistance to microorganisms, change in the color, and odor are functions of the broad array of processing parameters and consequences of wood



chemical degradation. The changes show positive as well as negative aspects. Examples of these unwanted features are diminution of strength, increased brittleness, and recovery of compression set.

Altering the processing parameters, in which wood is treated, alters the product performance. THM posttreatment is an effective method of permanently fixing the compression set. The temperature should not, however, exceed 160°C or 170°C as the mechanical properties of products may then greatly deteriorate.

Hydrolysis of hemicelluloses during THM treatment plays an active role in the dissipation of the stresses stored in the microfibrils and lignin. About 2/3 of the mass of hemicellulose hydrolysis leaves the wood during THM treatment and produces new surfaces, and pores helping dissipate the strain energy. To achieve a total elimination of shape memory, the lignin must be in its rubbery state during the THM treatment.

One of the important challenges for expanding the commercialization of THM technology is scaling up the laboratory processes to industrial levels. The time required for densification and posttreatment of a specimen is approximately proportional to the second power of the principal length of the densification. For large elements, the consequence of this time on the chemical degradation of wood would be dramatic, i.e., would be prohibitive for the industrial production. In order to reduce the processing time, investigations with interdisciplinary approach are required under participation of researchers from academia and industry.

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