

1 **Municipal waste liquor treatment via bioelectrochemical and fermentation ($H_2 +$**
2 **CH_4) processes: Assessment of various technological sequences**

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26

27 **Abstract**

28

29 In this paper, the anaerobic treatment of a high organic-strength wastewater-
30 type feedstock, referred as the liquid fraction of pressed municipal solid waste
31 (LPW) was studied for energy recovery and organic matter removal. The processes
32 investigated were (i) dark fermentation to produce biohydrogen, (ii) anaerobic
33 digestion for biogas formation and (iii) microbial fuel cells for electrical energy
34 generation. To find a feasible alternative for LPW treatment (meeting the two-fold
35 aims given above), various one- as well as multi-stage processes were tested. The
36 applications were evaluated based on their (i) COD removal efficiencies and (ii)
37 specific energy gain. As a result, considering the former aspect, the single-stage
38 processes could be ranked as: microbial fuel cell (92.4 %)> anaerobic digestion (50.2
39 %)> hydrogen fermentation (8.8 %). From the latter standpoint, an order of hydrogen
40 fermentation ($2277 \text{ J g}^{-1} \text{ COD}_{\text{removed}} \text{ d}^{-1}$)> anaerobic digestion ($205 \text{ J g}^{-1} \text{ COD}_{\text{removed}} \text{ d}^{-1}$)> microbial fuel cell ($0.43 \text{ J g}^{-1} \text{ COD}_{\text{removed}} \text{ d}^{-1}$) was attained. The assessment showed
41 that combined, multi-step treatment was necessary to simultaneously achieve efficient
42 organic matter removal and energy recovery from LPW. Therefore, a three-stage
43 system (hydrogen fermentation-biomethanation-bioelectrochemical cell in sequence)
44 was suggested. The different approaches were characterized via the estimation of COD
45 balance, as well.

46

47
48 **Keywords:** OFMSW, wastewater, biohydrogen, biogas, microbial fuel cell, multi-
49 stage process

50 **1. Introduction**

51

52 The management of municipal solid waste (MSW) is a crucial issue from an
53 environmental point of view (Eriksson et al., 2005) and has therefore been widely
54 addressed in the literature. As a result, it turned out that this type of problematic
55 material fits well to the waste-to-energy concept (Cheng et al., 2007) and can thus be
56 considered as a potential renewable energy feedstock (Cheng and Hu, 2010). Hence,
57 technologies that not only help to meet environmental protection goals but at the same
58 time, support energy gain from this particular waste stream have been in the focus
59 point worldwide. In accordance, researchers demonstrated the feasibility of MSW
60 (having high organic matter content) for certain bioprocesses, including anaerobic
61 digestion to generate methane (Hilkiah Igoni et al., 2008; Khalid et al., 2011) and dark
62 fermentation to produce biohydrogen (Alzate-Gaviria et al., 2007; Fountoulakis and
63 Manios, 2009; Kobayashi et al., 2012).

64 Besides the possibility of energy recuperation from solid organic matter relying
65 on the techniques mentioned, the utilization of liquid fractions derived from MSW via
66 alternative methods i.e. bioelectrochemical systems (BES) has been a hot topic too.
67 BESs rely on electrochemically-active bacteria (Zhen et al., 2016ab), which are
68 reportedly able to deal with wastewater-like feedstock (Pant et al., 2012) and
69 accordingly, MSW derivatives such as leachate (generated at landfill sites in
70 remarkable quantities) was found as a good candidate for BES feeding and subsequent
71 energy recovery (Iskander et al., 2016).

72 In addition to leachate, another type of waste liquor (obtained from MSW by
73 mechanical pressing and referred as the “liquid fraction of pressed municipal solid
74 waste”, abbreviated as LPW) appeared to be treatable in BES, in particular microbial
75 fuel cells (MFC) (Koók et al., 2016) and microbial electrohydrogenesis cells (MEC)
76 (Zhen et al., 2016c). Moreover, the organic-matter rich LPW was proven to be
77 convertible to gaseous energy carriers (hydrogen and methane) via fermentation, as
78 well (Rózsenszki et al., 2015).

79 Although LPW was found applicable in bioelectrochemical and fermentation
80 processes to generate energy (Koók et al., 2016; Rózsenszki et al., 2015), this
81 landfill-originated fraction can be a challenging starting material for single-stage
82 processes due to its somewhat recalcitrant nature (Zhen et al., 2016c). Therefore, it can
83 be expected that multi-step (two- or three-stage) processes in sequence could lead to
84 performance enhancement and better organic matter degradation. Supportive
85 conclusions on this concept were lately drawn by Mahmoud et al. (2014), who showed
86 that anaerobic pre-fermentation of leachate could improve the stability of subsequent
87 BES operation. Thus, in the frame of a multi-step processes, a pretreatment step can be
88 a proposed to facilitate consecutive energy recovery. Another reason to pay attention
89 to multi-step technologies is that they could make the valorization of effluent coming
90 from a single process possible, which (due to the many times occurring incomplete
91 degradation of poorly biodegradable compounds) may still contain significant amount
92 of organic matter to be removed. In accordance with the study by Wang et al. (2011),
93 integrated systems (coupling BES after dark fermentation step) can have a potential to
94 increase the utilization of substrates inherently resisting the microbiological

95 conversion. Other studies i.e. by [Luongo Malave' et al. \(2015\)](#) also verified the benefit
96 of multi-step anaerobic processes to augment the energy production from simple as
97 well as complex raw materials.

98 Overall, driven by the considerations and recent findings mentioned above, the
99 aim of the current investigation was to assess the utilization of LPW substrate in multi-
100 stage approach, involving MFC, dark fermentation and biogas formation (i) first alone
101 and thereafter (ii) in various, multi(two and three)-step design to find the arrangement
102 that leads to better process performance, both in terms of energy recovery and organic
103 matter degradation. To our knowledge, this concept has not been evaluated in the
104 literature applying the LPW and thus, the experimental results presented in this work
105 can contribute to the new body of knowledge in the field.

106

107 **2. Materials and methods**

108

109 **2.1. Inoculum and substrate**

110

111 Mesophilic anaerobic sludge (MAS) from a biogas plant (Pálhalma, Hungary)
112 was used as a basic source of inoculum in all the single-stage processes (microbial fuel
113 cell, anaerobic digestion and biohydrogen fermentation) involved in this work, which
114 was already proven as a good source of fermentative, methanogenic and eletroactive
115 strains ([Koók et al., 2016](#); [Rózsensberszki et al., 2015](#)). The main initial parameters of
116 MAS were 17 g L^{-1} COD and $\text{pH} = 7.5$. As mentioned above, a particular wastewater
117 fraction, called LPW was taken from a domestic municipal landfill (Királyszentistván,

118 Hungary) and used thoroughly in this investigation as the starting material to feed the
119 first-stage of the tested technological approaches. The methods to obtain the LPW
120 were already described in our earlier papers (Koók et al., 2016; Rózsenszki et al.,
121 2015) and its main characteristics are summarized in Fig. 1.

122

123 **2.2. Process descriptions**

124

125 The evaluation of each single- and multi-stage processes was carried out using
126 the results of triplicates (in all cases the arithmetical mean values are demonstrated
127 with standard deviations lower than 5%).

128

129 **2.2.1. Single-stage biogas and hydrogen fermentations (S-BGF, S-HF)**

130

131 Biogas and hydrogen fermentations were performed in WTW OXITOP 100
132 manometric vessels with 500 mL total capacity (Rózsenszki et al., 2015).
133 Technically, this method is similar to that referred as the Owen method (Logan et al.,
134 2002) since both are based on the intermittent release of pressure created by the gas
135 being formed.

136 Regarding the test conditions: in the single-step biogas (S-BGF) experiments,
137 MAS (as such, without pretreatment) and LPW were mixed in 1:1 ratio to get 50 mL
138 working volume, resulting in 450 mL headspace. Suitable anaerobic circumstances
139 were established by purging the reactor volume with high-purity (>99.99 vol.%) N₂.

140 37 °C temperature and 220 rpm stirring speed were ensured until the gas evolution has
141 stopped (Rózsenszki et al., 2015).

142 The single-stage hydrogen fermentations (S-HF) were conducted in a similar
143 way to biogas experiments, except two things: Firstly, thermal pretreatment (75 °C, 45
144 min) was applied to MAS to suppress CH₄-forming activity and at the same time,
145 promote H₂-fermenting microorganisms (Bakonyi et al., 2014). Secondly, initial pH
146 was adjusted to 5 using 10 (m/m)% H₂SO₄ solution to facilitate H₂-formation
147 (Rózsenszki et al., 2015). Volumetric gas evolution (mL) in both cases (S-BGF, S-
148 HF) was derived from the pressure data acquired by manometric caps attached to the
149 OXITOP 100 devices – as detailed earlier (Rózsenszki et al., 2015) – and converted
150 to STP conditions. Control experiments were also conducted to check the indigenous
151 gas formation potential of heat-pretreated as well as untreated, raw MAS alone and
152 was found negligible compared to those observed from fresh LPW.

153

154 **2.2.2. Single-stage microbial fuel cells (S-MFC)**

155

156 Small bench-scale, two-chambered microbial fuel cells (MFC) were fabricated
157 from plexiglass material. The anode and cathode cells were designed with equal, 60
158 mL total volumes. To separate the chambers, Nafion[®] N 115 proton-selective
159 membrane (purchased from Sigma-Aldrich) with 7 cm² effective area was used after
160 an activation procedure (Kim et al., 2007). In the anode half-cell, carbon cloth having
161 25 cm² apparent surface area was employed to be colonized by the exo-electrogenic
162 strains contained in raw MAS. To connect the carbon cloth anode to the external

163 circuit (made of copper wiring and including a 100 Ω external resistor), graphite rod
164 with 1.5 mm diameter was applied. In the cathode side of the MFC, 25 cm² carbon
165 cloth (connected to the external circuit similar to the anode) was used as cathode,
166 immersed in 60 mL continuously aerated catholyte. The scheme of the MFC can be
167 found in our previous publication (Koók et al., 2016). The MFCs in this work were
168 inoculated with 55 cm³ MAS (as such, without pretreatment) and started-up with Na-
169 acetate (10 g/L stock solution) to select the electrochemically-active biofilm and help
170 its initial stabilization (Liu et al., 2005) (no any other external nutrients were added),
171 making the working volume up to 60 cm³. More information on start-up methods and
172 related operation can be read in our recent communications (Koók et al., 2016;
173 Rózsenszki et al., 2015). The anodic biofilm formation process was considered
174 done once reproducible batch cycles (Carmona-Martínez et al., 2015) (here in terms of
175 the voltage time profile) could be observed. The detectable potential difference
176 between the anode and cathode was monitored on-line and saved by a DAQ (National
177 Instruments USB-6008) and LabView installed on PC. The MFCs were operated at 37
178 °C. The anode and cathode chambers were not mechanically stirred. The initial pH in
179 the anode chamber was set to 7, which is within the interval (6-9) that was found
180 sufficient for electrochemically-active bacteria to carry out the transfer of electrons to
181 the anode (Patil et al., 2011).

182 Once the biofilm development on the anode surface was finished, the single-
183 stage MFC (S-MFC) was fed with LPW rather than Na-acetate. Noteworthy, LPW in
184 all S-MFCs was injected after drawing the equivalent amount of spent media
185 remaining in the MFC from the previous cycle. At first, when the MFC start-up was

186 completed, only a small dose of LPW (3 mL) was added to the anode chamber to test
187 the reaction of the anodic biofilm to this substrate. Afterwards, in the consecutive
188 cycle, the volume of fresh LPW substrate was increased to 25 mL.

189

190 **2.2.3. Two-stage process I – Coupling anaerobic digestion to hydrogen** 191 **fermentation (HF-BGF)**

192

193 The fermentation effluent of S-HF was investigated further on as the input
194 material for sequential biogas production. In essence, 25 mL residue from S-HF was
195 mixed with 25 mL sludge coming from S-BGF. Afterwards, the experiments were
196 conducted under the conditions as specified for S-BGF (Section 2.2.1.), except
197 certainly that no fresh LPW was supplemented into the OXITOP bottles.

198

199 **2.2.4. Two-stage process II – Coupling microbial fuel cell to hydrogen** 200 **fermentation (HF-MFC)**

201

202 The fermentation effluent of S-HF was subjected to microbial fuel cell for its
203 further utilization. In particular, 25 mL from the anode chamber of stabilized MFC
204 was removed and subsequently replaced by 25 mL residue from S-HF. Thereafter, the
205 measurements were conducted under the conditions as specified for S-MFC (Section
206 2.2.2.), except certainly that no fresh LPW was fed into the system again.

207

208 **2.2.5. Three-stage process – hydrogen fermentation, anaerobic digestion and**
209 **microbial fuel cell attached in sequence (HF-BGF-MFC)**

210

211 Once the two-stage HF-BGF experiments were finished, the fermentation
212 residue of HF-BGF process was transferred to microbial fuel cell. As a matter of fact,
213 25 mL from the anode chamber of stabilized MFC was drawn and replaced by 25 mL
214 residue from HF-BGF. Further on, the tests were carried out under the conditions
215 detailed for S-MFC (Section 2.2.2.), except certainly that no fresh LPW was
216 introduced to the MFC again.

217

218 **2.3. Analytical methods**

219

220 To determine the composition of headspace gases obtained in the S-HF, S-BGF
221 and HF-BGF (in terms of H₂, CH₄ and CO₂), samples were taken (when the particular
222 process was terminated) and analyzed by gas chromatography as described elsewhere
223 ([Rózsenszki et al., 2015](#)). To follow the biodegradation of LPW in the S-HF, S-
224 BGF and S-MFC, samples from the liquid phase at the end of the experiments and
225 their chemical oxygen demand (COD) was measured by the standard methods ([APHA,](#)
226 [1995](#)). In the multi-stage processes (Section 2.2.3.-2.2.5.), the COD of the effluent
227 after each step was determined, as well. The initial COD values of the MAS and fresh
228 LPW were considered according to Section 2.1 (**Fig. 1**), respectively.

229

230

231 **2.4. Calculations**

232

233 To complete COD balance calculations (**Table 1**), the following COD
234 equivalents for gaseous products, in particular H₂ and CH₄ gases were employed: 8 g
235 COD/g H₂ and 4 g COD/g CH₄. To determine the COD of biocurrent generation, the
236 calculation took into account the next steps:

237 First, the Coulombic efficiency (C_e) was calculated, in accordance with Eq. 1
238 ([Oh and Logan, 2005](#)):

239

240
$$C_e = \frac{C_{ex}}{C_{th}} \times 100 \quad (1)$$

241

242 where C_{ex} is the amount of coulombs actually recovered in the course of the MFC
243 operation time (t) (measured in the form of biocurrent), according to Eq. 2:

244

245
$$C_{ex} = \int_0^t I dt \quad (2)$$

246

247 *I* is the (bio)current, computed from Ohm's law using the voltage measured between
248 the anode and cathode through the external resistance ([Logan et al., 2006](#)). *t* is the
249 MFC operating time (during which the current was measured).

250

251

252 C_{th} is the theoretical coulombs that can be derived from the total amount of COD
253 removed (Eq. 3).

254

$$255 \quad C_{th} = \frac{F b \Delta COD V}{M} \quad (3)$$

256

257 where ΔCOD is the difference of initial and final/residual liquid phase COD
258 concentrations in the anode chamber (g/L). F is the Faraday's constant, b is 4 (the
259 number of electrons exchanged per mole of O_2), V is the anode working volume (L).
260 M is the molar mass of O_2 (32 g/mole) (Logan et al., 2006; Pasupuleti et al., 2016).

261 Afterwards, the portion of COD (ΔCOD_C , in the unit of g) accounted for C_{ex} is
262 expressed by Eq. 4:

263

$$264 \quad \Delta COD_C = \frac{C_e \Delta COD V}{100} \quad (4)$$

265

266 The specific energy yield (E^*) of a given process was calculated according to Eq. 5:

267

$$268 \quad E^* (\text{J g}^{-1} \text{COD}_{\text{removed}} \text{d}^{-1}) = E_t [(\text{COD}_{\text{initial}} - \text{COD}_{\text{final}}) V_L]^{-1} \text{t}^{-1} \quad (5)$$

269

270 where E_t is the energy cumulated in the course of reactor operation (t) either in the
271 form of gaseous energy carriers (H_2 or CH_4) or bioelectricity (see Section 3.).

272 COD_{initial} and COD_{final} are the initial and final/residual liquid phase COD
273 concentrations in the liquid phase (working volume), respectively, while V_L is the
274 working volume in the particular bioreactor.

275 **3. Results and discussion**

276

277 **3.1. Assessment of single-stage processes for LPW treatment**

278

279 **3.1.1. Single-stage hydrogen and biogas fermentations (S-HF, S-BGF)**

280

281 A typical progress curve for S-HF process is illustrated in **Fig. 2**, which has a
282 shape quite similar to that obtained by [Logan et al. \(2002\)](#). After 12 hours of lag time,
283 intense gas formation could be observed, leading to the end of fermentation within 2
284 days. It is to note that methane could not be detected in the headspace, which verifies
285 the suppression of methanogens by the applied sludge pretreatment ([Bakonyi et al.,](#)
286 [2014](#)). As a result, it turned out that a total volume of 194 (STP) cm³ gas (mixture of
287 hydrogen and carbon dioxide) could be obtained. This, together with the gas
288 composition analysis (47.1 vol.% H₂) reveals that 91 (STP) cm³ hydrogen was formed.

289 As for the S-BGF, the time profile of the biogas production can be seen in **Fig.**
290 **3**. In fact, the biogas formation began shortly after the inoculation of LPW with MAS.
291 In the course of the S-BGF process, 527 cm³ (STP) biogas was produced with 56.7
292 vol.% methane content, meaning a gain of 299 cm³ (STP) CH₄. Taking into account
293 the gravimetric energy densities of hydrogen and methane gases (142 kJ/g and 55.5
294 kJ/g, respectively), the theoretically recoverable amount of energy in the S-HF and S-
295 BGF from the organic matter contained in 25 mL LPW was 1.14 kJ and 11.7 kJ,
296 respectively.

297 In the light of the current results about the S-HF and S-BGF it can be inferred
298 that LPW was suitable for the generation of bioenergy carriers, in particular H₂ and
299 CH₄ via fermentation, which is in agreement with our previous findings
300 (Rózsensberszki et al., 2015). However, in the case of S-HF, the short gas evolution
301 phase was accompanied by only a limited organic matter removal (8.8 %) measured as
302 COD (Table 2), which assumes the quick depletion of components the
303 microorganisms could metabolize into H₂ and CO₂. As for the S-BGF, representing a
304 technology with markedly longer time-demand, COD elimination efficiencies as high
305 as 50.2 % were realized (Table 2). Consequently, it seems that the LPW is composed
306 of an (i) easily i.e. the marginal reducing sugar content (Fig. 1) and a (ii) hardly
307 biodegradable fraction of organic substances (Zhen et al., 2016c). It is a reasonable
308 assumption (based on the brief fermentation time and relatively poor COD removal)
309 that the S-HF was able to deal only with the former, simply convertible part. On the
310 other hand, the S-BGF had better capability to process the refractory COD content,
311 presumably attributed to the more sufficient adaption of hydrolytic bacteria contained
312 in the MAS as inoculum. This step, the hydrolysis is a crucial to achieve the efficient
313 solubilization and biotransformation during anaerobic digestion, especially when
314 complex organic matters are supplied (Zheng et al., 2014), such as in the case of LPW.
315 Besides, in both the S-HF and S-BGF, the rapid initial gas generation may have
316 occurred (at least in part) because of the naturally-occurring microbes hosted by the
317 LPW, contributing to the biodegradation via so-called self-fermentation (Marone et al.,
318 2012). Though the BOD₅/COD ratio as high as 0.6-0.7 predicts good biodegradability
319 (Buitrón et al., 2014), it would appear that LPW decomposition (despite a BOD₅/COD

320 = 0.73, **Fig. 1**) apparently faces more challenges under anaerobic fermentation
321 conditions, especially during the short S-HF.

322

323 **3.1.2. Single-stage microbial fuel cells (S-MFC)**

324

325 As mentioned in Section 2.2.2., two methods were used to test the conversion of
326 LPW in microbial fuel cells. Firstly, the injection of 3 mL LPW resulted in an almost
327 immediate response, detected in the form of a gradually growing electric potential
328 between the anode and cathode (**Fig. 4a**). Similar progress curves were reported by
329 [Rabaey et al. \(2003\)](#), presenting the response of MFC to substrate injections. **Fig. 4a**
330 was an important feedback to confirm the degradability of LPW by the exo-
331 electrogenic strains located on the anode surface. Afterwards, in the subsequent cycle,
332 the MFC was loaded with 25 mL LPW (same amount as used in the S-HF and S-BGF)
333 to perform the S-MFC experiments. The results are plotted in **Fig. 4b**. The maximum
334 potential values reached the vicinity of 50 mV (measured through 100 Ω external
335 resistor) approximately for 6-7 days (between Day 10 and 16), before noting a
336 gradually declining tendency (**Fig. 4b**). It is clear from the comparison of **Fig. 4a** and
337 **Fig. 4b** that the higher LPW dosage induced the extension of process (biodegradation)
338 time. From the data collected during the 30 days operation (**Fig. 4b**) of S-MFC, it was
339 computed – in accordance with papers by [Koók et al. \(2016\)](#) and [Rabaey et al. \(2003\)](#)
340 – that 31 J cumulated energy (as electricity) could be extracted from 25 mL LPW. On
341 one hand, this value is significantly lower in comparison with those attained with the
342 S-HF and S-BGF. Nonetheless, on the other hand, the COD in the residue at the end of

343 the S-MFC measurements was decreased dramatically, meaning that only 7.5 % of the
344 initial COD was left behind (**Table 2**).

345 The maximal current density of the S-MFC – taken into account the (i)
346 maximum voltage measured under stationary conditions (Day 10-16, **Fig. 4b**), (ii)
347 external resistor and (iii) anode surface data, according to [Koók et al. \(2016\)](#) – was
348 calculated as 216 mA m⁻². This is close to the value (253 mA m⁻²) reported by
349 [Cercado-Quezada et al. \(2010\)](#) for microbial fuel cells inoculated with garden compost
350 leachate (somewhat similar to LPW) for the valorization of food industry waste.
351 Current densities in the same order of magnitude (114 mA m⁻²) were achieved by
352 [Ganesh and Jambeck \(2013\)](#) in microbial fuel cells designed for the treatment of
353 landfill leachate, a problematic substrate with more or less similar origin than LPW.

354 Although salient COD removal performance was experienced with the S-MFC,
355 the small amount of electrical energy obtained and the low, ~3 % Coulombic
356 efficiency – helping to reveal what portion of the substrate consumed is transformed
357 into electricity ([Logan et al. 2006](#); [Ren et al., 2014](#)) – together indicate that the major
358 part of COD was converted via side-reactions that did not result in registered bio-
359 current generation.

360 Among the reasons behind, it is supposed that COD was partly consumed via
361 alternative metabolism of bacteria – living either (i) in the anode-surface biofilm or (ii)
362 in the bulk phase (planktonic strains), etc. – as a response to the appearance of O₂ in
363 the anode chamber. This may have occurred, at least in part, due to the diffusion of O₂
364 through the Nafion[®] membrane placed in between the anaerobic anode and aerated
365 cathode compartments. This polymeric, proton exchange-type membrane (PEM) plays

366 a key-role in two-chamber MFCs to sustain its function, however, it is not indeed
367 impermeable to the gases present at either side of the bioelectrochemical system ([Chae
368 et al., 2008a](#)). As a matter of fact, noticeable transport of O₂ across the Nafion[®]
369 membrane can occur in MFCs, as found by [Chae et al. \(2008b\)](#). This phenomena will
370 potentially make the metabolic pathways of the (electroactive) strains shift – attributed
371 to the absence of fully anaerobic conditions – and aerobic respiration of heterotrophs
372 can take over, simultaneously causing the loss of precious substrate and the depression
373 of Coulombic efficiency ([Logan, 2012](#)). For example, [Liu and Logan \(2004\)](#) reported
374 Coulombic efficiency values below 10 %, ascribed to substantial penetration of
375 oxygen to the MFC anode chamber. Under some circumstances in two-chambered
376 MFCs (separated by PEM) fed with glucose, [Rabaey et al. \(2003\)](#) experienced
377 Coulombic efficiencies in the range of 8-12 % (same order of magnitude with those
378 reported in this current research, **Table 3**). Furthermore, [Hernández-Fernández et al.
379 \(2015\)](#) obtained 4.4 % Coulombic efficiency in a two-chamber microbial fuel cell with
380 Nafion[®] membrane utilizing waste water feedstock, which represents a comparable
381 level of system performance as demonstrated in this work with the S-MFC process.

382 Besides O₂ leakage, obstacles related to biofouling of the membrane due to the
383 growth of bacteria on its anode side-faced surface should be taken into account ([Sun et
384 al., 2016](#)). According to [Chae et al. \(2008b\)](#), biofilm layer can develop over time on
385 Nafion[®], which was identified as a factor affecting MFC performance ([Xu et al., 2012](#))
386 and to certain extent, it can be made responsible for deteriorated Coulombic
387 efficiencies ([Choi et al., 2011](#)).

388 Another probable explanation for COD consumption without decent Coulombic
389 efficiency and electrical energy recovery may be associated with the formation of
390 gases e.g. methane and hydrogen. Since the MFCs were inoculated with untreated
391 mesophilic anaerobic sludge in this work, fermentative microorganisms and H₂-
392 scavenging, methanogenic archaea (in addition to exo-electrogens) were inherently
393 added to anode chamber. In such cases, when a mixed microbial community is applied
394 as seed source, the gaseous compounds mentioned may be generated in certain
395 amounts (Kim et al., 2005), however, in this particular work, it was not quantified.

396 Overall, though several assumptions concerning the behavior of the S-MFC
397 were given above, further research and technical improvements will have to be made
398 to tackle the issues raised, leading to better competitiveness from an energy yield
399 standpoint.

400

401 **3.1.3. Comparison of single-stage processes for LPW treatment and** 402 **considerations to design the multi-stage processes**

403

404 To comparatively assess the single-stage, either bioelectrochemical or
405 fermentative technologies for LPW management, two indicators, in particular the (i)
406 COD removal efficiency and the (ii) specific energy yield were used.

407 From an energetic point of view, taking into account the specific energy yields
408 (expressing the amount of energy recovered from the COD removed on daily average
409 basis) the following ranking could be established (**Table 2**): S-HF > S-BGF > S-MFC.

410 From the aspect of COD removal, the following, contrary order was found (**Table 2**):
411 S-MFC > S-BGF > S-HF.

412 The outcomes indicate that the best energy recovery from LPW could be done
413 via hydrogen fermentation, however, this application does not seem to be able to
414 reduce the organic matter content. In other words, S-HF alone suffers from the
415 bottleneck of insufficient COD conversion efficacy. Thus, if the effluent coming from
416 the S-HF was discharged without further treatment, it would cause environmental
417 concerns and on the top of that, the major part of chemical energy bound in the LPW
418 would remain unexploited.

419 Overall, based on the considerations so far made, it could be recommended that
420 the S-HF, as first-step is integrated to a multi-stage process. It was already
421 demonstrated by researchers that the residue of hydrogen fermentation can be
422 subjected to anaerobic digestion to get extra methane and in that way, enhance the
423 energy productivity (Buitrón et al., 2014; Intanoo et al., 2016; Nualsri et al., 2016).
424 Besides that, it was reported that the effluent of dark fermentation step can be a viable
425 starting material in bioelectrochemical cells, as an alternative solution to valorize the
426 unutilized organic matter (Kumar et al., 2016; Marone et al., 2016; Rivera et al.,
427 2015). For instance, Oh and Logan (2005) studied the concept of an attached,
428 hydrogen fermentation–microbial fuel cell system in order that the high amount of
429 organic matter remaining in the effluent of the former application is subsequently
430 utilized for bioelectricity generation in the latter one.

431 Therefore, two-stage processes including a (i) HF-BGF as well as a (ii) HF-
432 MFC were tested to evaluate the fate of COD in the S-HF effluent and at the same

433 time, to see how much energy could be further gained. The results will be presented in
434 the next sections.

435

436 **3.2. Assessment of multi-stage processes to treat the effluent of applications fed** 437 **with raw LPW substrate**

438

439 To utilize the effluent containing the recalcitrant COD fraction – which was
440 undegradable by microbes present in the S-HF – it was fed to complementary (i)
441 anaerobic digestion and (ii) microbial fuel cell under conditions described in Section
442 2.2.3 and 2.2.4, respectively.

443 In the HF-BGF, as it can be seen in **Table 4**, the BGF as a second-step was able
444 to work with a COD removal efficiency of 55.7 %, cutting the initial 34.5 g COD/L to
445 15.3 g COD/L at the end of the test. In the course of the fermentation, as a result of
446 organic matter decomposition, 157 mL (STP) CH₄ was generated, which has a total
447 energy content of 6130 J. The COD removal efficiency was comparable with the S-
448 BGF process, as reflected by the 55.7 % and 50.2 % values, respectively.
449 Nevertheless, the specific energy yield achieved was enhanced by 20 %, according to
450 the 246 J g⁻¹ COD_{removed} d⁻¹ instead of 205 J g⁻¹ COD_{removed} d⁻¹ (as listed in **Table 2** for
451 S-BGF). In fact, it can be stated that biogas could be generated from the effluent of S-
452 HF process treating fresh LPW along with additional COD elimination. Successful
453 attempts with two-stage systems (attaching anaerobic digestion after dark fermentative
454 hydrogen production) were communicated in recent research articles, as well ([Intanoo](#)

455 [et al., 2016; Nualsri et al., 2016](#)). For instance, [Buitrón et al. \(2014\)](#) obtained COD
456 removal efficiencies in the range of 56-67 % (depending on organic matter loadings)
457 during the methanation of effluent coming from preliminary H₂ fermentation, which
458 matches well with our relevant result (55.7 %).

459 By evaluating the results obtained during the HF-MFC process, it can be
460 pointed out that the COD removal from the residue of S-HF could exceed 90 %
461 efficiency (**Table 4**), similar to that observed for the S-MFC from raw LPW. However,
462 the specific energy yield, compared to other applications investigated herewith, still
463 remained rather poor, and the loss of performance was probably caused by the reasons
464 considered and detailed in Section 3.1.2. for the S-MFC. The Coulombic efficiency
465 characteristic is given in **Table 3**.

466 Thus, as proven for the second time in the course of this study, the MFC could
467 be used efficiently to decrease the organic matter content, which is desired from an
468 environmental protection point of view. However, it was unable to noticeably
469 participate in the total energy production, which was more the responsibility of the
470 fermentation processes, generating energy carriers (H₂ + CH₄) in decent quantities.
471 These gaseous biofuels formed can be potentially utilized for sustainable energy
472 production, in case an appropriate purification (downstream) technique is employed.
473 To accomplish this goal, membrane separation can be considered as a solution,
474 attributed to their attractiveness for the concentration of such energy carriers ([Bakonyi
475 et al., 2013; Basu et al., 2010; Ryckebosch et al., 2011](#)).

476 Though the coupled, HF-BGF was found favorable to harvest energy in the
477 form of hydrogen and methane, organic matter removal after auxiliary biogas

478 formation was still incomplete, reflected by the 15.3 g/L COD concentration in the
479 residue, as mentioned above. Thus, in the last part of the measurements, its removal
480 was attempted. For this purpose, MFC was chosen, creating a HF-BGF-MFC three-
481 stage, sequential approach. Certainly, in this MFC stage, further energy gain was not
482 really expected based on the experiences described. Nevertheless, keeping its already
483 demonstrated capability to remove COD in mind (over 90% in all cases tested so far),
484 it was considered that it was able to serve as a final, polishing step. This set of
485 experiments (carried out in accordance with Section 2.2.5.) fairly confirmed our
486 previous expectation since as it can be seen in **Table 4**, initial COD content was
487 drastically reduced, thank to the 89 % removal efficiency (without realizing notable,
488 additional energy recovery). As a result, the effluent remained at the end of the
489 operation had more or less negligible, 0.9 g/L COD concentration. For related
490 Coulombic efficiency, **Table 3** should be consulted.

491 Consequently, in agreement with the evaluation made so far, the proposed route
492 for LPW management should consist of three, consecutive steps, involving, hydrogen
493 fermentation, followed by methanogenesis and microbial fuel cell, as depicted in **Fig.**
494 **5**.

495

496 **3.3. Estimation of organic matter (COD) balance**

497

498 Analyzing the fate of COD is a useful approach to track how the organic matter
499 added to a particular process was converted ([Zhen et al., 2016c](#)). To estimate the COD
500 balance, as to be observed in **Table 1**, (i) initial and (ii) final COD contents in the

501 liquid phase of the respective system and furthermore, (iii) product CODs (either H₂,
502 CH₄ or biocurrent) were considered. COD recovery was established according to Eq.
503 6.

$$505 \text{ COD recovery (\%)} = 100[(\text{COD}_{\text{residual}} + \text{COD}_{\text{product}}) \text{COD}^{-1}_{\text{initial}}] \quad (6)$$

506

507 It is evident from **Table 1** that there was a remarkable variation of COD
508 recoveries, dependent on the particular treatment process. The best COD recovery was
509 attained for the hydrogen fermentation, with only a 6-7 % deviation of the organic
510 matter balance. As for methane production, accomplished either in the first- or second-
511 stage, COD recoveries in the range of 70-80 % could be estimated, which is still
512 acceptable. Similar COD recovery values were reported by [Yu and Fang \(2003\)](#) for an
513 anaerobic reactor fed with wastewater-type feedstock.

514 In the case of MFC process, regardless of the conditions, the low COD
515 recoveries (10-20 %, **Table 1**) present an unclear fate for the major part of COD
516 converted (removed). This problem is reflected by the low Coulombic efficiencies
517 (specified e.g. in Section 3.1.2. and given in **Table 3**) and means that the electrons
518 liberated from organic matter decomposition in MFC were not captured by the anode
519 and registered in the form of biocurrent. Although several distinct mechanisms were
520 proposed to explain it (seen in Section 3.1.2.), further research will be required to
521 better understand the underlying phenomena and can be a subject for the continuation
522 of our work.

523

524

525 **4. Conclusions**

526

527 The assessment of various, single- and multi-stage anaerobic process for
528 municipal waste liquor (LPW) valorization was carried out in this study. It was
529 demonstrated that one-step technologies (in particular dark fermentative H₂
530 production, biogas fermentation and microbial fuel cells) alone were not able to
531 simultaneously achieve efficient organic matter removal and energy recovery from
532 LPW substrate. Nevertheless, a sequential, three-stage technology was found
533 promising to meet these aims. Organic matter balances (according to COD recoveries)
534 were quite satisfactory for the hydrogen and methane productions, however, further
535 investigation on the microbial fuel cell process will be needed to get a better
536 comprehension about fate of COD added in this particular application.

537

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549

550 **References**

551

- 552 1. Alzate-Gaviria, L.M., Sebastian, P.J., Pérez-Hernández, A., Eapen, D., 2007.
553 Comparison of two anaerobic systems for hydrogen production from the
554 organic fraction of municipal solid waste and synthetic wastewater. *Int. J.*
555 *Hydrogen Energy* 32, 3141-3146.
- 556 2. APHA. Standard methods for the examination of water and wastewater. 19th
557 ed. New York, USA: American Public Health Association, 1995.
- 558 3. Bakonyi, P., Borza, B., Orlovits, K., Simon, V., Nemestóthy, N., Bélafi-Bakó,
559 K., 2014. Fermentative hydrogen production by conventionally and
560 unconventionally heat pretreated seed cultures: a comparative assessment. *Int.*
561 *J. Hydrogen Energy* 39, 5589-5596.
- 562 4. Bakonyi, P., Nemestóthy, N., Bélafi-Bakó, K., 2013. Biohydrogen purification
563 by membranes: An overview on the operational conditions affecting the
564 performance of non-porous, polymeric and ionic liquid based gas separation
565 membranes. *Int. J. Hydrogen Energy* 38, 9673-9687.
- 566 5. Basu, S., Khan, A.L., Cano-Odena, A., Liu, C., Vankelecom, I.F.J., 2010.
567 Membrane-based technologies for biogas separations. *Chem. Soc. Rev.* 39, 750-
568 768.

- 569 6. Buitrón, G., Kumar, G., Martínez-Arce, A., Moreno, G., 2014. Hydrogen and
570 methane production via a two-stage processes (H₂-SBR + CH₄-UASB) using
571 tequila vinasses. *Int. J. Hydrogen Energy* 39, 19249-19255.
- 572 7. Carmona-Martínez, A.A., Trably, E., Milferstedt, K., Lacroix, R., Etcheverry,
573 L., Bernet, N., 2015. Long-term continuous production of H₂ in microbial
574 electrolysis cell (MEC) treating saline wastewater. *Water Res.* 81, 149-156.
- 575 8. Cercado-Quezada, B., Delia, M.L., Bergel, A., 2010. Testing various food-
576 industry wastes for electricity production in microbial fuel cell. *Bioresour.*
577 *Technol.* 101, 2748-2754.
- 578 9. Chae, K.J., Choi, M., Ajayi, F.F., Park, W., Chang, I.S., Kim, I.S., 2008b. Mass
579 transport through a proton exchange membrane (Nafion) in microbial fuel cells.
580 *Energy Fuels* 22, 169-176.
- 581 10. Chae, K.J., Choi, M.J., Lee, J., Ajayi, F.F., Kim, I.S., 2008a. Biohydrogen
582 production via biocatalyzed electrolysis in acetate-fed bioelectrochemical cells
583 and microbial community analysis. *Int. J. Hydrogen Energy* 33, 5184-5192.
- 584 11. Cheng, H., Hu, Y., 2010. Municipal solid waste (MSW) as a renewable source
585 of energy: Current and future practices in China. *Bioresour. Technol.* 101,
586 3816-3824.
- 587 12. Cheng, H., Zhang, Y., Meng, A., Li, Q., 2007. Municipal solid waste fueled
588 power generation in China: A case study of waste-to-energy in Changchun city.
589 *Environ. Sci. Technol.* 41, 7509-7515.
- 590 13. Choi, M.J., Chae, K.J., Ajayi, F.F., Kim, K.Y., Yu, H.W., Kim, C.W., et al.,
591 2011. Effects of biofouling on ion transport through cation exchange

- 592 membranes and microbial fuel cell performance. *Bioresour. Technol.* 102, 298-
593 303.
- 594 14. Eriksson, O., Carlsson Reich, M., Frostell, B., Björklund, A., Assefa, A.,
595 Sundqvist, J.O., et al., 2005. Municipal solid waste management from a systems
596 perspective. *J. Clean. Prod.* 13, 241-252.
- 597 15. Fountoulakis, M.S., Manios, T., 2009. Enhanced methane and hydrogen
598 production from municipal solid waste and agro-industrial by-products co-
599 digested with crude glycerol. *Bioresour. Technol.* 100, 3043-3047.
- 600 16. Ganesh, K., Jambeck, J.R., 2013. Treatment of landfill leachate using microbial
601 fuel cells: alternative anodes and semi-continuous operation. *Bioresour.*
602 *Technol.* 139, 383-387.
- 603 17. Hernández-Fernández, F.J., de los Ríos, A.P., Mateo-Ramírez, F., Godínez, C.,
604 Lozano-Blanco, L.J., Moreno, J.I., et al., 2015. New application of supported
605 ionic liquids membranes as proton exchange membranes in microbial fuel cell
606 for waste water treatment. *Chem. Eng. J.* 279, 115-119.
- 607 18. Hilkieh Igoni, A., Ayotamuno, M.J., Eze, C.J., Ogaji, S.O.T., Probert, S.D.,
608 2008. Designs of anaerobic digesters for producing biogas from municipal
609 solid-waste. *Appl. Energy* 85, 430-438.
- 610 19. Intanoo, P., Chaimongkol, P., Chavadej, S., 2016. Hydrogen and methane
611 production from cassava wastewater using two-stage upflow anaerobic sludge
612 blanket reactors (UASB) with an emphasis on maximum hydrogen production.
613 *Int. J. Hydrogen Energy* 2016;41:6107-6114.

- 614 20. Iskander, S.M., Brazil, B., Novak, J.T., He, Z., 2016. Resource recovery from
615 landfill leachate using bioelectrochemical systems: Opportunities, challenges,
616 and perspectives. *Bioresour Technol* 2016;201:347-54
- 617 21. Khalid, A., Arshad, M., Anjum, M., Mahmood, T., Dawson, L., 2011. The
618 anaerobic digestion of solid organic waste. *Waste Manage.* 31, 1737-1744.
- 619 22. Kim, J.R., Cheng, S., Oh, S.E., Logan, B.E., 2007. Power generation using
620 different cation, anion, and ultrafiltration membranes in microbial fuel cells.
621 *Environ. Sci. Technol.* 41, 1004-1009.
- 622 23. Kim, J.R., Min, B., Logan, B.E., 2005. Evaluation of procedures to acclimate a
623 microbial fuel cell for electricity production. *Appl. Microbiol. Biotechnol.* 68,
624 23-30.
- 625 24. Kobayashi, T., Xu, K.Q., Li, Y.Y., Inamori, Y., 2012. Evaluation of hydrogen
626 and methane production from municipal solid wastes with different
627 compositions of fat, protein, cellulosic material and the other carbohydrates. *Int.*
628 *J. Hydrogen Energy* 37, 15711-15718.
- 629 25. Koók, L., Rózsenszki, T., Nemestóthy, N., Bélafi-Bakó, K., Bakonyi, P.,
630 2016. Bioelectrochemical treatment of municipal waste liquor in microbial fuel
631 cells for energy valorization. *J. Clean. Prod.* 112, 4406-4412.
- 632 26. Kumar, G., Bakonyi, P., Kobayashi, T., Xu, K.Q., Sivagurunathan, P., Kim,
633 S.H., et al., 2016. Enhancement of biofuel production via microbial
634 augmentation: The case of dark fermentative hydrogen. *Renew. Sustain. Energy*
635 *Rev.* 57, 879-891.

- 636 27. Liu, H., Logan, B.E., 2004. Electricity generation using an air-cathode single
637 chamber microbial fuel cell in the presence and absence of a proton exchange
638 membrane. *Environ. Sci. Technol.* 38, 4040-4046.
- 639 28. Liu, H., Cheng, S., Logan, B.E., 2005. Production of electricity from acetate or
640 butyrate using a single-chamber microbial fuel cell. *Environ. Sci. Technol.* 39,
641 658-662.
- 642 29. Logan, B.E., Hamelers, B., Rozendal, R., Schröder, U., Keller, J., Freguia, S., et
643 al., 2006. Microbial fuel cells: Methodology and technology. *Environ. Sci.*
644 *Technol.* 40, 5181-5192.
- 645 30. Logan, B.E., Oh, S.E., Kim, I.S., Van Ginkel, S., 2002. Biological hydrogen
646 production measured in batch anaerobic respirometers. *Environ. Sci. Technol.*
647 36, 2530-2535.
- 648 31. Logan, B.E., 2012. Essential data and techniques for conducting microbial fuel
649 cell and other types of bioelectrochemical system experiments. *ChemSusChem*
650 5, 988–994.
- 651 32. Luongo Malave', A.C., Bernardi, M., Fino, D., Ruggeri, B., 2015. Multistep
652 anaerobic digestion (MAD) as a tool to increase energy production via H₂ +
653 CH₄. *Int. J. Hydrogen Energy* 40, 5050-5061.
- 654 33. Mahmoud, M., Parameswaran, P., Torres, C.I., Rittmann, B.E., 2014.
655 Fermentation pre-treatment of landfill leachate for enhanced electron recovery
656 in a microbial electrolysis cell. *Bioresour. Technol.* 151, 151-158.
- 657 34. Marone, A., Ayala-Campos, O.R., Trably, E., Carmona-Martinez, A.A.,
658 Moscoviz, R., Latrille, E., et al., 2016. Coupling dark fermentation and

659 microbial electrolysis to enhance bio-hydrogen production from agro-industrial
660 wastewaters and by-products in a bio-refinery framework. *Int. J. Hydrogen*
661 *Energy*, <http://dx.doi.org/10.1016/j.ijhydene.2016.09.166>

662 35. Marone, A., Massini, G., Patriarca, C., Signorini, A., Varrone, C., Izzo, G.,
663 2012. Hydrogen production from vegetable waste by bioaugmentation of
664 indigenous fermentative communities. *Int. J. Hydrogen Energy* 37, 5612-5622.

665 36. Nualsri, C., Kongjan, P., Reungsang, A., 2016. Direct integration of CSTR-
666 UASB reactors for two-stage hydrogen and methane production from sugarcane
667 syrup. *Int. J. Hydrogen Energy* 41, 17884-17895.

668 37. Oh, S., Logan, B.E., 2005. Hydrogen and electricity production from a food
669 processing wastewater using fermentation and microbial fuel cell
670 technologies. *Water Res.* 39, 4673-4682.

671 38. Pant, D., Singh, A., Van Bogaert, G., Irving Olsen, S., Singh Nigam, P., Diels,
672 L., et al., 2012. Bioelectrochemical systems (BES) for sustainable energy
673 production and product recovery from organic wastes and industrial
674 wastewaters. *RSC Adv.* 2, 1248-1263.

675 39. Pasupuleti, S.B., Srikanth, S., Dominguez-Benetton, X., Venkata Mohan, S.,
676 Pant, D., 2016. Dual gas diffusion cathode design for microbial fuel cell
677 (MFC): optimizing the suitable mode of operation in terms of
678 bioelectrochemical and bioelectro-kinetic evaluation. *J. Chem. Technol.*
679 *Biotechnol.* 91, 624-639.

680 40. Patil, S.A., Harnisch, F., Koch, C., Hübschmann, T., Fetzer, I., Carmona-
681 Martínez, A.A., et al., 2011. Electroactive mixed culture derived biofilms in

- 682 microbial bioelectrochemical systems: The role of pH on biofilm formation,
683 performance and composition. *Bioresour. Technol.* 102, 9683-9690.
- 684 41. Rabaey, K., Lissens, G., Siciliano, S.D., Verstraete, W., 2003. A microbial fuel
685 cell capable of converting glucose to electricity at high rate and
686 efficiency. *Biotechnol. Lett.* 25, 1531-1535.
- 687 42. Ren, L., Zhang, X., He, W., Logan, B.E., 2014. High current densities enable
688 exoelectrogens to outcompete aerobic heterotrophs for substrate. *Biotechnol.*
689 *Bioeng.* 111, 2163-2169.
- 690 43. Rivera, I., Buitrón, G., Bakonyi, P., Nemestóthy, N., Bélafi-Bakó, K., 2015.
691 Hydrogen production in a microbial electrolysis cell fed with a dark
692 fermentation effluent. *J. Appl. Electrochem.* 45, 1223-1229.
- 693 44. Rózsenberszki, T., Koók, L., Hutvágner, D., Nemestóthy, N., Bélafi-Bakó, K.,
694 Bakonyi, P., et al., 2015. Comparison of anaerobic degradation processes for
695 bioenergy generation from liquid fraction of pressed solid waste. *Waste*
696 *Biomass Valor.* 6, 465-473.
- 697 45. Ryckebosch, E., Drouillon, M., Vervaeren, H., 2011. Techniques for
698 transformation of biogas to biomethane. *Biomass Bioenergy* 35, 1633-1645.
- 699 46. Sun, M., Zhai, L.F., Li, W.W., Yu, H.Q., 2016. Harvest and utilization of
700 chemical energy in wastes by microbial fuel cells. *Chem. Soc. Rev.* 45, 2847-
701 2870.
- 702 47. Wang, A., Sun, D., Cao, G., Wang, H., Ren, N., Wu, W.M., et al., 2011.
703 Integrated hydrogen production process from cellulose by combining dark

704 fermentation, microbial fuel cells, and a microbial electrolysis cell. *Bioresour.*
705 *Technol.* 102, 4137-4143.

706 48. Xu, J., Sheng, G.P., Luo, H.W., Li, W.W., Wang, L.F., Yu, H.Q., 2012. Fouling
707 of proton exchange membrane (PEM) deteriorates the performance of microbial
708 fuel cell. *Water Res.* 46, 1817-1824.

709 49. Yu, H.Q., Fang, H.H.P., 2003. Acidogenesis of gelatin-rich wastewater in an
710 upflow anaerobic reactor: influence of pH and temperature. *Water Res.* 37, 55-
711 66.

712 50. Zhen, G., Kobayashi, T., Lu, X., Kumar, G., Xu, K., 2016a. Biomethane
713 recovery from *Egeria densa* in a microbial electrolysis cell-assisted anaerobic
714 system: Performance and stability assessment. *Chemosphere* 149, 121-129.

715 51. Zhen, G., Lu, X., Kobayashi, T., Kumar, G., Xu, K., 2016b. Promoted
716 electromethanosynthesis in a two-chamber microbial electrolysis cells (MECs)
717 containing a hybrid biocathode covered with graphite felt (GF). *Chem. Eng. J.*
718 284, 1146-1155.

719 52. Zhen, G., Kobayashi, T., Lu, X., Kumar, G., Hu, Y., Bakonyi, P., et al., 2016c.
720 Recovery of biohydrogen in a single-chamber microbial electrohydrogenesis
721 cell using liquid fraction of pressed municipal solid waste (LPW) as substrate.
722 *Int. J. Hydrogen Energy* 41, 17896-17906.

723 53. Zheng, Y., Zhao, J., Xu, F., Li, Y., 2014. Pretreatment of lignocellulosic
724 biomass for enhanced biogas production. *Prog. Energy Combust. Sci.* 42, 35-
725 53.

726

Figure legend

727

728

729 **Fig. 1** – The origin of LPW and its main characteristics

730 **Fig. 2** – Gas formation (H_2+CO_2) as a function of time in the S-HF process

731 **Fig. 3** – Biogas generation (CH_4+CO_2) time profile for the S-BGF process

732 **Fig. 4** – Potential changes after injecting (a) 3 mL and (b) 25 mL LPW to the S-MFC
733 system

734 **Fig. 5** – The processes tested in this work. Red arrows indicate the proposed, multi-
735 stage treatment to efficiently accomplish energy recovery and COD removal from
736 LPW substrate

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740

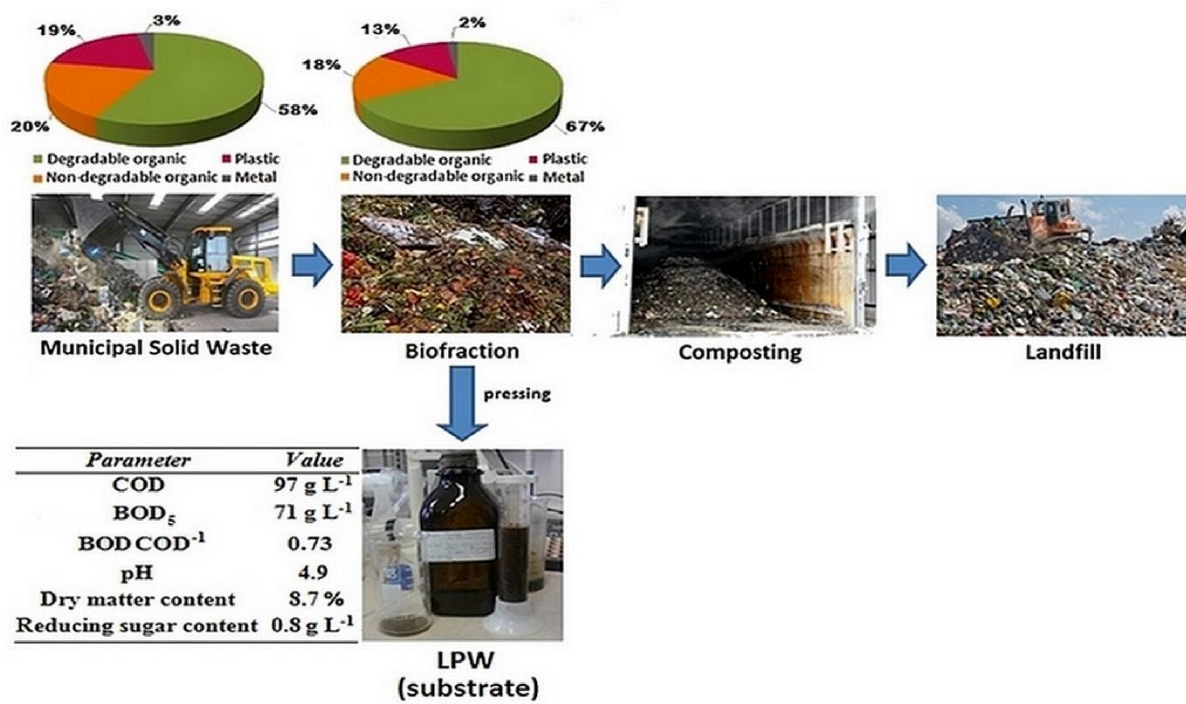
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744 **Fig. 1**

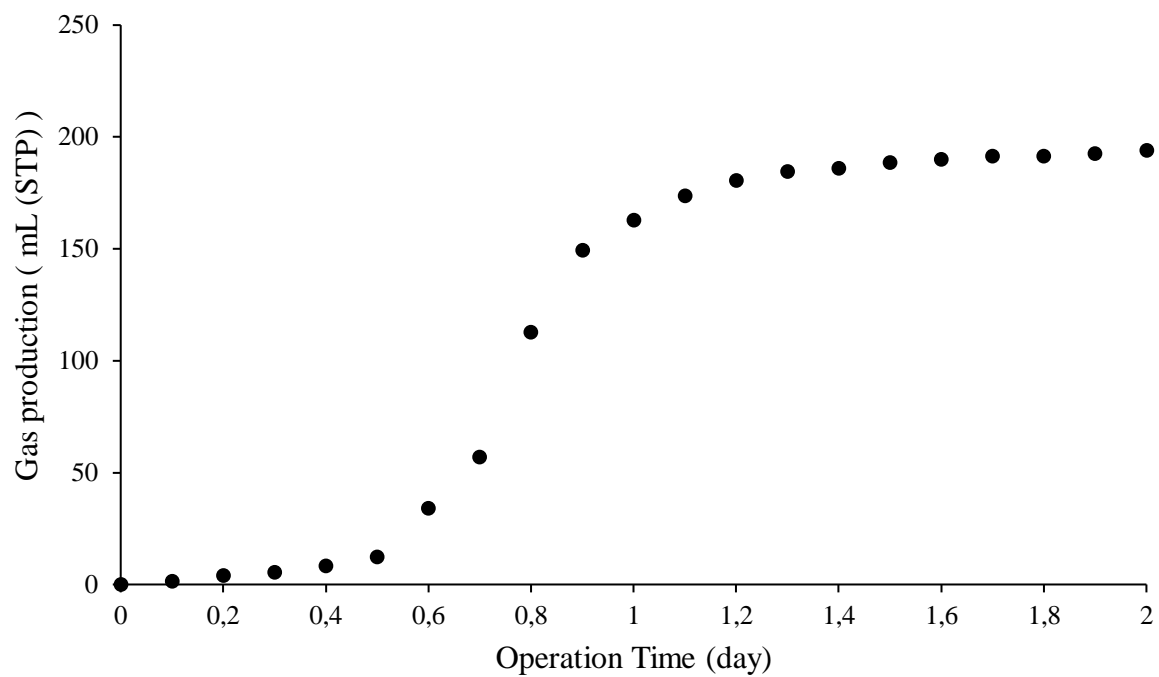
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748 **Fig. 2**

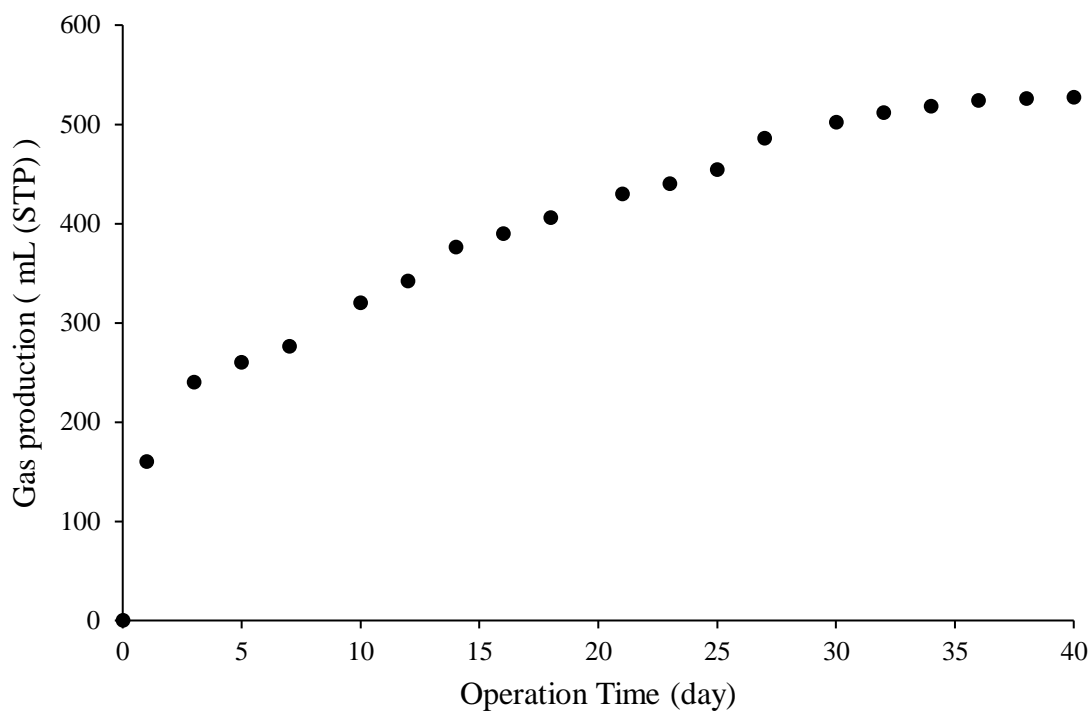


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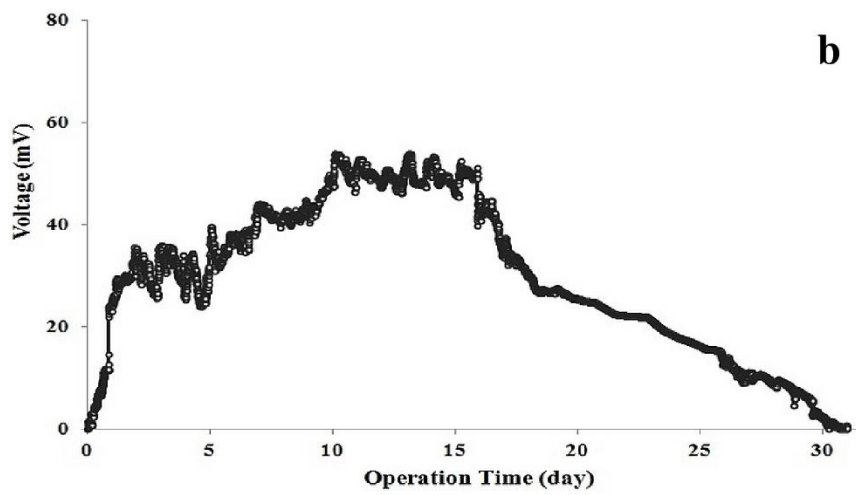
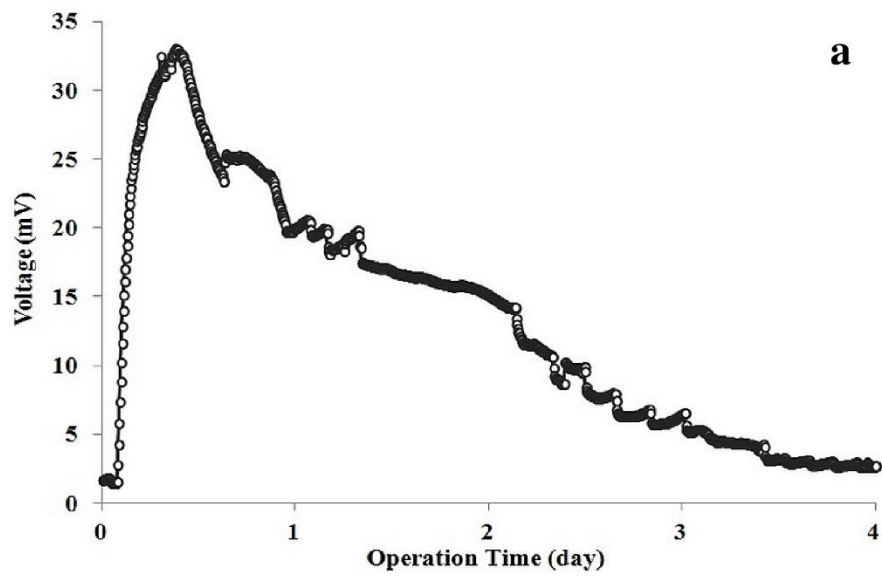
752 **Fig. 3**



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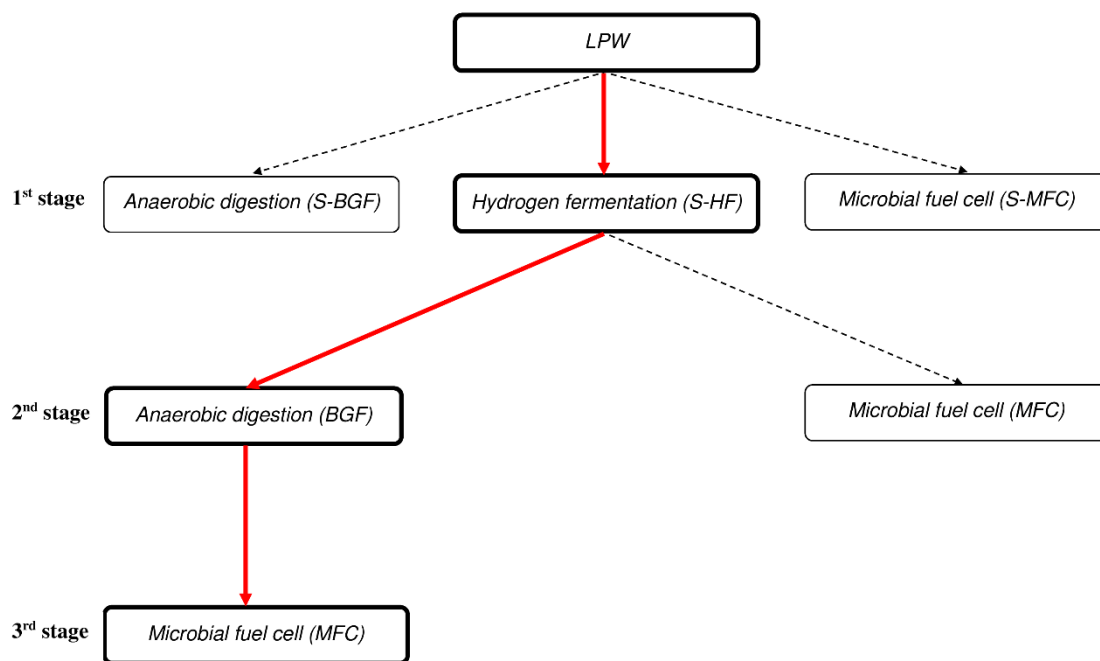
755 **Fig. 4**



756

757

758 **Fig. 5**



759

760

761 **Table 1** – COD balance for the processes studied for LPW utilization

762

Process	Liquid phase total COD (g)		Product COD (g)			COD recovery (%)	
	Initial	Residual	H ₂	CH ₄	Biocurrent ^a		
Single-stage	S-HF	2.85	2.6	0.064		93.5	
	S-BGF	2.85	1.42		0.843	79.4	
	S-MFC	2.52	0.192			0.066	10.2
Two-stage	HF-BGF	HF	2.85	2.6	0.064		93.5
		BGF	1.73	0.77		0.442	70.1
	HF-MFC	HF	2.85	2.6	0.064		93.5
		MFC	1.39	0.084			0.055
Three-stage	HF-BGF-MFC	HF	2.85	2.6	0.064		93.5
		BGF	1.73	0.77		0.442	70.1
		MFC	0.48	0.054			0.042

^a: as ΔCOD_C from **Table 3**.

763

764

765 **Table 2** – Comparison of single-stage processes for LPW treatment

Process	COD (g/L)		COD removal efficiency (%)	Cumulated energy (J)	Operational time (d)	E* (J g ⁻¹ COD _{removed} d ⁻¹)
	Initial	Residual				
S-HF	57	52	8.8	1139	2	2277
S-BGF	57	28.4	50.2	11698	40	205
S-MFC	42	3.2	92.4	31	30	0.43

766

767 **Table 3** – Coulombic efficiencies (C_e) and related data for the MFCs

768

Process	COD _{removed} (g)	C_{th} (C)	C_{ex} (C)	C_e (%)	ΔCOD_C (g)
S-MFC	2.33	28059	801	2.9	0.066
MFC (in the HF-MFC)	1.31	15793	658	4.2	0.055
MFC (in the HF-BGF-MFC)	0.43	5090	492	9.7	0.041

769

Table 4 – Comparison of multi-stage processes for LPW treatment

Process	COD (g/L)		COD removal efficiency (%)	Cumulated energy (J)	Operational time (d)	E* (J g ⁻¹ COD _{removed} d ⁻¹)		
	Initial	Residual						
Two-stage	HF-MFC	HF	57	52	8.8	1139	2	2277
		MFC	23.3	1.4	94.0	25	25	0.8
	HF-BGF	HF	57	52	8.8	1139	2	2277
		BGF	34.5	15.3	55.7	6130	26	246
Three-stage	HF-BGF-MFC	HF	57	52	8.8	1139	2	2277
		BGF	34.5	15.3	55.7	6130	26	246
		MFC	7.9	0.9	88.6	24	15	3.8