

Probing Multiphased Transition in Bulk MoS₂ by Direct Electron Injection

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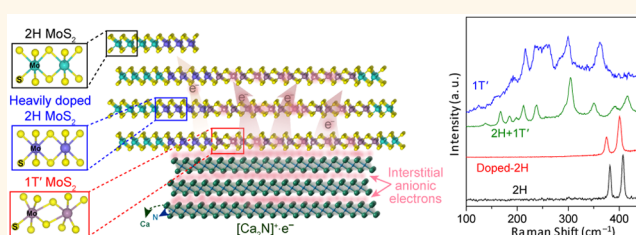
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Supporting Information

ABSTRACT: Structural phase transitions in layered two-dimensional (2D) materials are of significant interest owing to their ability to exist in multiple metastable states with distinctive properties. However, phase transition in bulk MoS₂ by nondestructive electron infusion has not yet been realized. In this study, we report the 2H to 1T' phase transition and in-between intermediates in bulk MoS₂ using MoS₂/[Ca₂N]⁺e⁻ heterostructures, in which kinetic free electrons were directly injected into MoS₂. We observed various phases in MoS₂ ranging from heavily doped 2H to a distorted lattice state and then on to a complete 1T' state. Snapshots of the multiphase transition were captured by extraordinary Raman shift and bandgap reduction and were further elucidated by theoretical calculations. We also observed a weakening in interlayer coupling in the vicinity of the metallic regime, which led to an unusually strong photoluminescence emission, suggesting light-efficient bulk MoS₂. Our results thus suggest the optoelectronic applications that can fully utilize the multiphase transition of bulk 2D materials.

KEYWORDS: MoS₂, [Ca₂N]⁺e⁻, electron injection, multiphase transition, interlayer weakening



Two-dimensional (2D) semiconducting subset of transition-metal dichalcogenide (TMD) crystals, such as molybdenum disulfide (MoS₂), exist in a layered crystal structure in which the S atoms of each monolayer exist in the trigonal prismatic coordination pattern around Mo atoms with a hexagonal (2H) phase atomic stacking sequence.^{1–8} MoS₂ exhibits an indirect bandgap (1.1 eV) in the bulk state but becomes a direct bandgap (1.9 eV) semiconductor in the monolayer state with a signature of strong photoluminescence (PL). Electronic structures of TMDs with interlayer coupling and inversion symmetry that varies with the number of layers enable diverse electronic and optoelectronic applications for MoS₂.^{3–10} Recently, a method to switch between the metallic and insulating phases of semiconductors has been proposed for group VI-TMDs, such as MoS₂ and WS₂ crystals. Their metallic (1T') phases exhibit a low contact resistance for field-effect transistors,¹⁰ Weyl semimetals,¹¹ a quantum spin Hall effect,¹² and strong optical Kerr nonlinearity.¹³ Among Mo- and W-based S, Se, and Te TMDs, metal sulfides (MS₂) exhibit the largest phase-transition energy (0.55 eV for MoS₂) between the 2H and 1T' phases, while MTe₂ exhibits an order of low phase-transition energy (0.06 eV),⁹ which indicates that it is a challenge to produce a metallic MS₂. Previously, several methods had been established to obtain the polymorphs of 2D-MoS₂ using alkali metal intercalation,^{13–16} electron-beam

irradiation,¹⁷ straining,¹⁸ high-pressure flux,^{19,20} and electron doping.^{21–24} Most of the known phase-transition methods for MoS₂ are limited to a single layer,^{13–21} and among them, only the alkali-metal intercalation method could induce complete phase transition from 2H to 1T' MoS₂.^{13–16} However, the resulting 1T' phase due to alkali-metal intercalation method can undergo oxidation and experiences impurity-induced strain in the host lattice; in addition, the phase transition was spatially incomplete, mostly due to a random intercalation of alkali atoms. The other processes mentioned above result in coexisting metallic and semiconductor phases, even for single-layer MoS₂.^{17–21} Therefore, an efficient and direct way for the phase engineering of bulk 2H MoS₂ is necessary.

Recently, bulk 2H MoTe₂ has been converted into the 1T' phase just by making the heterostructures with a 2D electrode material, dicalcium nitride [Ca₂N]⁺e⁻, to realize the solid-state doping of 2D-TMDs with deeply penetrating high-density electrons.²⁴ Because [Ca₂N]⁺e⁻ is a layered material with a very low work function of 2.6 eV and contains layers of freely mobile electrons sandwiched between [Ca₂N]⁺ layers at room temperature, it can transfer a high concentration of electrons (~10¹⁴ cm⁻²) to the material it is in contact with.^{24–26} Here

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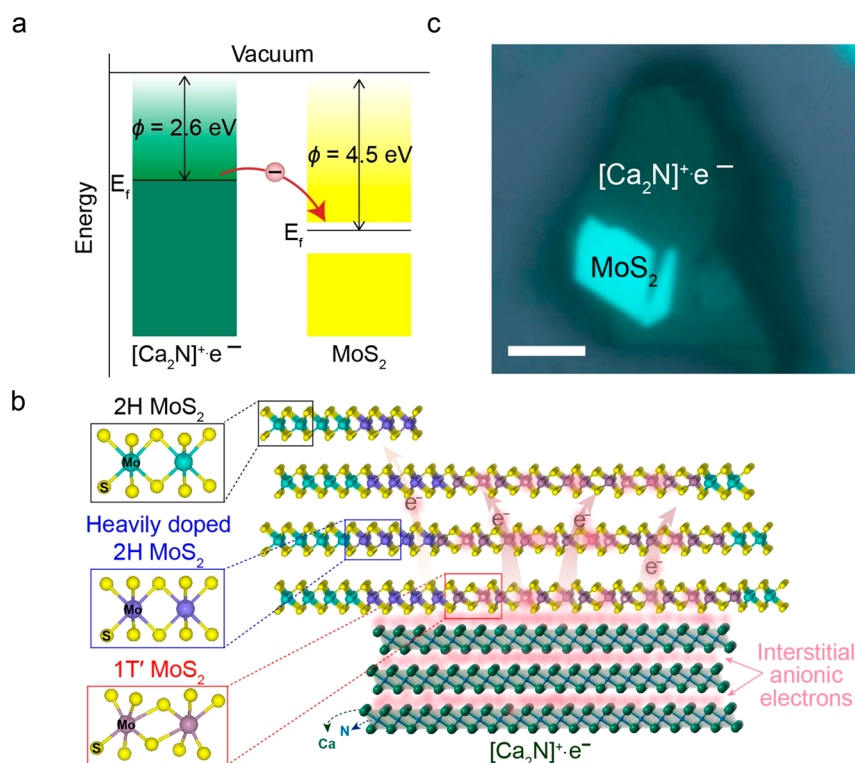


Figure 1. Charge injection from 2D- $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ to bulk MoS_2 to induce phase transition. (a) Band alignment between $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ and MoS_2 with work functions of 2.6 and 4.5 eV. (b) A schematic describing that electron injection from $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ induces a 2H to 1T' phase change in bulk MoS_2 ; MoS_2 is heavily doped in the thicker sections. (c) Optical view of a typical heterostructure of MoS_2 on 2D- $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$. The scale bar is 5 μm .

we note that the difference in work function between $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ and MoS_2 is higher (>2 eV) than that between $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ and MoTe_2 (1.3 eV),^{24,27} which suggests that the kinetic energy of the doped surface electrons is larger in MoS_2 ; furthermore, a high density of electrons can be injected into bulk MoS_2 by heterostructuring with $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$. While previous surface electron-doping methods such as chemical functionalization or electrical gating were applied mostly for monolayer TMDs, they resulted in only a moderate (30 to 50 meV) bandgap reduction,^{2–7,28,29} and complete metallization of bulk MoS_2 could not be achieved.

In this study, we report a series of lattice symmetry changes from the hexagonal phase to a distorted octahedral phase in bulk MoS_2 induced by the injection of a very high density of electrons by making the solid-state heterostructures with 2D- $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ crystals. The resultant degenerate electron doping, at an estimated sheet density of $\sim 10^{14}$ cm^{-2} , penetrates tens of nanometers into bulk MoS_2 from the contact interface, leading to the gradual evolution of lattice symmetry and band-structure renormalization in bulk MoS_2 . These changes were analyzed by PL spectroscopy, Raman spectroscopy, and density functional theory (DFT) calculations. We could observe several distinct phases of MoS_2 , including heavily doped 2H, metallic phase (1T'), and possibly a combination of both hexagonal and 1T'-phases, which exhibited strong PL emission due to weak interlayer interactions. Our results illustrate the exceptionally controlled lattice symmetry and electronic band structure of MoS_2 forming the heterostructures with 2D- $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$.

RESULTS AND DISCUSSION

The schematic in Figure 1a indicates a large work function difference (~ 2 eV) between $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ and MoS_2 owing to

the electrons in $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ can be easily injected into MoS_2 across the contact area with a high kinetic energy. These electrons penetrate deep into the bulk structure,²⁴ resulting in a structural change in MoS_2 from 2H to a metallic state with a distorted lattice (1T') as shown in the schematic Figure 1b. Figure 1c shows optical views of the typical heterostructures made by preparing MoS_2 crystals on $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$. We prepared such heterostructures in the argon atmosphere to prevent the oxidation of $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ that can occur at the ambient environment. If $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ is oxidized, the Raman spectrum is greatly modified and its electron-doping capability is compromised (see Supporting Information, Figure S1 for the Raman spectra of $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ before and after oxidation). For confocal Raman and PL spectroscopy, we air-sealed the heterostructures by capping them with transparent cover glass in an argon atmosphere.

We first show the correlated Raman and PL spectra of pristine bulk MoS_2 and $\text{MoS}_2/[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ heterostructures. The representative optical images are shown in Figure 2a, and the corresponding Raman and PL spectra obtained from these regions are displayed in Figure 2b,c, respectively. The Raman spectrum of pristine MoS_2 (Figure 2b, black curve) includes characteristic E_{2g}^1 and A_{1g} phonon modes at 382 and 408 cm^{-1} respectively, confirming the bulk thickness 2H MoS_2 crystals^{1–5} (see Supporting Information, Figure S2 for thickness verification). Based on the complete absence of a $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ Raman peak, it is deduced that the thickness of MoS_2 in Figure 2 is greater than ~ 30 nm (equal to the laser-penetration depth of MoS_2 at 514.5 nm).³⁰ The Raman spectra of bulk MoS_2 on $[\text{Ca}_2\text{N}]^+\cdot\text{e}^-$ exhibited softening in the A_{1g} and E_{2g}^1 modes to different extents (Figure 2b). Maximum red-shift values of 10 and 8 cm^{-1} were observed for the A_{1g} and E_{2g}^1

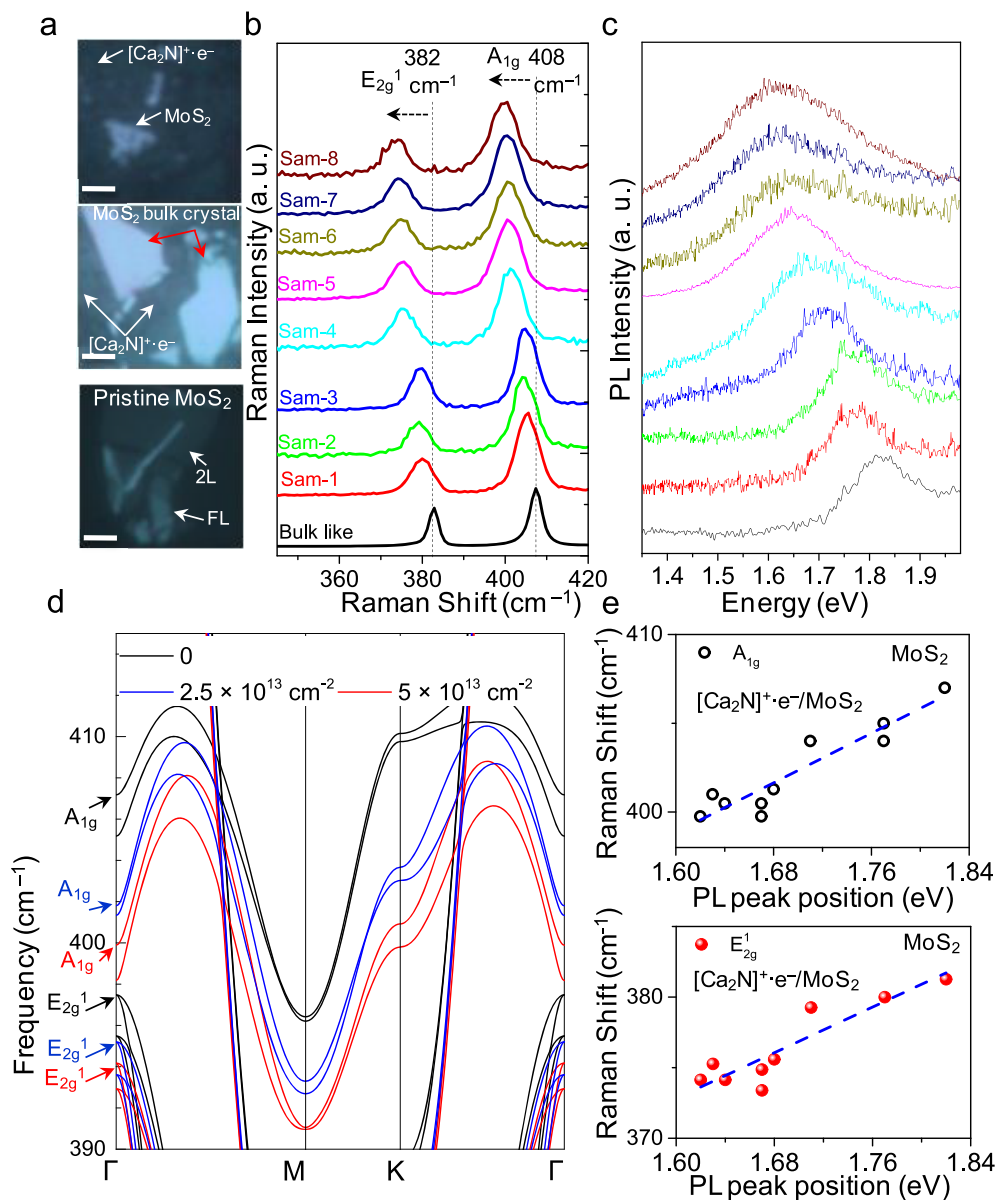


Figure 2. Renormalization of the optical bandgap and phonon energies of MoS₂ by electron injection. (a) Optical views of the MoS₂/[Ca₂N]⁺e⁻ heterostructure (scale bar = 5 μm). (b) Raman and (c) PL spectra obtained from eight different MoS₂/[Ca₂N]⁺e⁻ heterostructures. The dotted line act as a visual guide. Color-coded Raman and PL spectra were obtained from the same locations. (d) Phonon dispersion curves of undoped and doped (electron density (n) = 2.5×10^{13} and 5×10^{13} cm⁻²) 2L-MoS₂. A_{1g} and E_{2g}¹ Raman frequencies are indicated by the arrow. (e) Correlation plots between E_{2g}¹ and A_{1g} phonon frequencies and the exciton peaks obtained from eight samples.

phonon modes, respectively. The red-shift of A_{1g} peak in the heterostructure as compared to the pristine value (408 cm⁻¹ for bulk) could be attributed to electron transfer from 2D-[Ca₂N]⁺e⁻ to MoS₂. According to previous studies, observed softening in the A_{1g} mode (10 cm⁻¹) corresponds to an electron density of $\sim 5 \times 10^{13}$ cm⁻²,²⁻⁷ demonstrating that direct injection of degenerate electrons in MoS₂ was achieved using 2D-[Ca₂N]⁺e⁻. We believe that the A_{1g} frequency shift (ranging 0–10 cm⁻¹) originates from the varying local thickness of MoS₂ on [Ca₂N]⁺e⁻. The thinner the MoS₂ flake is, the greater is the expected doping effect. This is because a finite amount of doping electrons are injected from 2D-[Ca₂N]⁺e⁻; such dependence of doping electron density

on semiconductor thickness was previously observed with MoS₂⁷ and MoTe₂.²⁴

We also noticed a significant red-shift in the E_{2g}¹ mode of MoS₂ placed on (Ca₂N)⁺e⁻. Generally the change of the E_{2g}¹ mode energy was interpreted as the presence of a strain.^{30,31} It is to be noted that that the in-plane lattice mismatch between 2D-[Ca₂N]⁺e⁻ and MoS₂ cannot cause such a large strain, because previously lattice mismatch with various substrates or hetero van der Waals layers were reported to cause less than 2 cm⁻¹ of peak shift of E_{2g}¹ mode in thin (1–2 nm) MoS₂.^{31–35} Therefore, the large softening amounting to ~ 8 cm⁻¹ for the E_{2g}¹ mode indicates a significant strain and structural changes due to electron doping by 2D-[Ca₂N]⁺e⁻. Previously, doping with degenerate electrons was predicted to cause structural

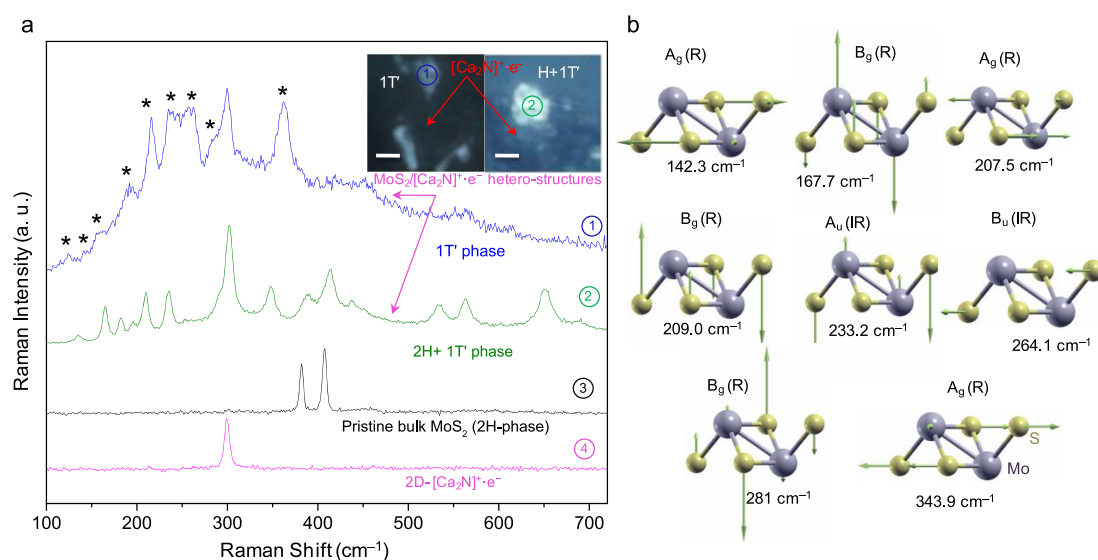


Figure 3. MoS₂ phase transition from 2H to 1T'. (a) Raman spectra of two different MoS₂/[Ca₂N]⁺e⁻ heterostructures (spectra 1 and 2), pristine bulk MoS₂ (black curve, spectrum 3), and [Ca₂N]⁺e⁻ (spectrum 4). Optical views of MoS₂/[Ca₂N]⁺e⁻ heterostructures representing spectra 1 and 2 are shown in the inset (scale bar = 5 μm). (b) Assignment of the Raman vibrational modes of 1T' MoS₂, obtained from theoretical calculations.

changes and plasma screening of coulomb interactions, thus causing phonon softening in 1L-MoS₂.^{22,36,37} In Figure 2d, we show the phonon dispersion relationship of bilayer MoS₂ calculated by DFT at an electron density of $5 \times 10^{13} \text{ cm}^{-2}$; it predicts red-shifts of 7 and 4 cm⁻¹ for the Raman A_{1g} and E_{2g}¹ modes, respectively, thus affirming that the observed giant softening of Raman modes may be attributed to the high density of electrons injected from 2D-[Ca₂N]⁺e⁻ (see the calculated phonon dispersion curves at other electron densities in Supporting Information, Figures S3 and S4).

Interestingly, we observed a direct correlation between the red-shift of Raman peaks and PL peak positions. In Figure 2c, we display PL spectra obtained simultaneously with Raman spectra (Raman and PL spectra obtained from the same sample locations are color-coded). We found that the sample with a large red-shift for the Raman peak displayed a large PL peak shift. Correlated optical bandgap vs E_{2g}¹ or A_{1g} Raman frequency plots displayed a linear correlation as shown in Figure 2e. An optical bandgap reduction of ~200 meV was observed for the sample with the maximum phonon softening of ~10 and ~8 cm⁻¹ for the A_{1g} and E_{2g}¹ modes, respectively, indicating that electron injection by heterostructuring with 2D[Ca₂N]⁺e⁻ caused giant renormalization in the bandgap and phonon modes of bulk MoS₂.

In Figure 3a, we show the Raman spectra of relatively thinner MoS₂ flakes on [Ca₂N]⁺e⁻ (spectra 1 and 2), which are vastly different from that of 2H phase MoS₂ (spectrum 3). We believe that the layer thickness of MoS₂ shown in Figure 3a is <30 nm (penetration depth of the excitation laser) and the thickness of the samples shown in Figure 2, because a peak at ~300 cm⁻¹ corresponding to [Ca₂N]⁺e⁻ could still be observed, unlike in the case of the Raman spectra in Figure 2b (see the non-normalized Raman spectra of MoS₂/[Ca₂N]⁺e⁻ heterostructures and bare [Ca₂N]⁺e⁻ in Supporting Information, Figure S5). The Raman spectrum of bare [Ca₂N]⁺e⁻ is shown for reference (spectrum 4). We should emphasize here that spectra 1 and 2 were obtained still from the bulk thickness of MoS₂; the optical contrast of the topmost

MoS₂ layer in the heterostructure (shown in the inset of Figure 3a) is greater than that of pristine bulk MoS₂ (see the Raman spectra of pristine bulk MoS₂ in Supporting Information, Figure S5; its optical contrast was compared to that of the sample in Figure 3a). In thinner bulk MoS₂, we expect that the electron doping density is higher than $5 \times 10^{13} \text{ cm}^{-2}$, which is the value estimated for the samples shown in Figure 2 and may have reached $\sim 10^{14} \text{ cm}^{-2}$, a value high enough to cause transition to 1T' MoS₂.^{4,13–15,21} First, we noticed in spectra 1 and 2 that the E_{2g}¹ and A_{1g} Raman modes, which are characteristic of 2H MoS₂, disappeared completely and instead, multiple Raman peaks newly emerged. In spectrum 2, we identified 17 Raman modes as corresponding to the 1T' phase (B_g: 167 (J₁) and 236 (J₂) cm⁻¹ and A_g: 138, 212, and 350 (J₃) cm⁻¹ modes), infrared (IR) active phonon modes at 290 (E_u + E_{1g}), 393 (E_{1u}), and 420 cm⁻¹ (B_{2g}² + E_{1u}²), and 2H Raman modes (B_{2g}²: 415 cm⁻¹, E_{2g}²: 440 cm⁻¹, 2LA (M): 454 cm⁻¹, E_g (M) + TA (M): 535 cm⁻¹, E_{1g}: 564 cm⁻¹, A_g (M) + LA (M): 653 cm⁻¹, and B_{3g}: 693 cm⁻¹), thus suggesting a metaphase different from 2H or 1T' MoS₂ crystals.^{17–22,36–48} Details of Raman-peak assignment are provided in Table S1. Indeed, the observation of a large number of Raman peaks even with off-resonance laser excitation (2.41 eV) indicates that this particular thickness of MoS₂ on [Ca₂N]⁺e⁻ possesses low crystal symmetry, similar to other 2D semiconductors such as rhenium disulfide and rhenium diselenide, which normally exist in the 1T' phase.^{49,50} Another possible reason for the emergence of multiple Raman modes is the formation of a twisted multilayer MoS₂ structure due to a heavy electron doping density, which weakens interlayer attractions (will be discussed later). Because supercell structures are formed in twisted multilayer MoS₂, the zone-folding effect generates multivibration modes at the γ point, activating multiple Raman modes, as observed in twisted bilayer MoS₂ or graphene.^{40,51}

In spectrum 1, the Raman modes of the metallic phase (1T') were dominant, while the Raman modes of the 2H phase could not be observed; this is similar to the spectrum of 1T' MoS₂

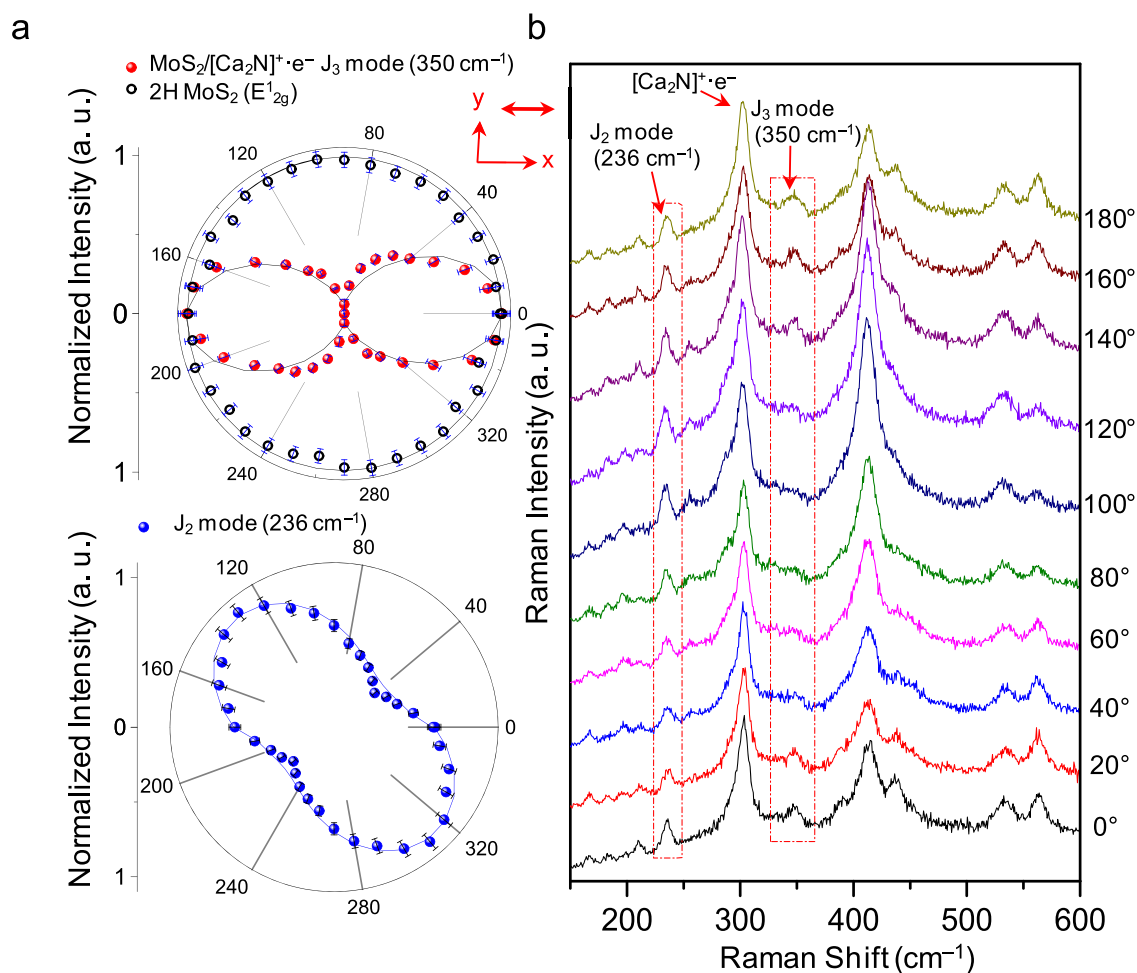


Figure 4. Anisotropic Raman response of the distorted lattice of bulk MoS₂ due to electron doping from [Ca₂N]⁺e⁻. (a) Polar plots of the polarization-dependent Raman intensities of 2H MoS₂ (E_{2g}¹ mode), J₃ mode at 350 cm⁻¹, and J₂ mode at 236 cm⁻¹ of bulk MoS₂ on [Ca₂N]⁺e⁻ (see the main text for the measurement scheme). An angle of 0° represents a horizontal placement of the sample stage. The polar plot of the E_{2g}¹ mode is shown for comparison. Solid lines are fitted calculations using the Raman tensors of E_{2g}¹ mode of 2H MoS₂ and A_g mode of 1T' MoS₂ and the experimental polarization configuration (see the main text). (b) Raman spectra at several selected rotation angles are shown.

obtained by Li-intercalation,¹³ indicating that MoS₂ on [Ca₂N]⁺e⁻ was converted completely into the 1T' phase. The phonon modes of 1T' MoS₂ were theoretically calculated. Considering the doped condition of MoS₂, we assumed an electron density of 2.2×10^{14} cm⁻² for calculating the vibration modes of 1T'-MoS₂. We found that the observed Raman spectrum of 1T'-MoS₂ (spectrum 1 in Figure 3a) matched well with our calculated 1T' MoS₂ in terms of the position and number of Raman modes, signifying that a very high level of electron injection ($\sim 10^{14}$ cm⁻²) at the MoS₂/[Ca₂N]⁺e⁻ heterostructure to yield a 1T' MoS₂ lattice structure. Several 1T' MoS₂ modes, identified in our calculation, are indicated by (*) in Figure 3a, and their vibration modes are displayed in Figure 3b (refer to Supporting Information, Table S2 for the assignment of all other Raman modes corresponding to 1T' MoS₂).

Even thinner MoS₂, of which thickness we believe is only a few nanometers based on the nonattenuation of Raman peak at ~ 300 cm⁻¹ of underneath [Ca₂N]⁺e⁻ and the appeared optical contrast, displayed only two Raman peaks of A_g mode of 1T' lattice at 146 cm⁻¹ and the IR-active mode at 255 cm⁻¹ with all other 1T' specific Raman modes totally quenched (See the Supporting Information, Figure S6). Such distinct

quenching of 1T' lattice Raman modes with strong electron doping was not observed in electrolyte-doped^{4,52} or Li-intercalated TMDs¹³ and suggests that this particular MoS₂ on [Ca₂N]⁺e⁻ may have reached a highly conducting phase or a lattice disordered phase due to excessive electron density.⁵³

Though the exact origin of Raman spectra containing both 1T' and 2H specific modes (spectrum 2 in Figure 3a) is not clearly known, we found that these modes are highly polarization-dependent. We obtained a series of Raman spectra while rotating the sample between a fixed polarizer and analyzer in the same direction as the horizontal *x*-axis of the sample stage of our Raman microscope. The polar plots of the Raman intensity of J₃ and J₂ modes are shown in Figure 4a, and Raman spectra at several selected rotation angles are displayed in Figure 4b, in which the angles represent sample rotation angles from the horizontal *x*-axis of the sample stage. The polarization dependence of the E_{2g}¹ mode of pristine bulk 2H MoS₂ is displayed for comparison in Figure 4a; it exhibited an isotropic response under the polarization-resolved Raman measurement as shown in Figure 4a (black circles). Raman spectra at all the measured angles for pristine 2H MoS₂ are included in Supporting Information, Figure S7. The isotropic response of the E_{2g}¹ mode of pristine 2H MoS₂ is expected

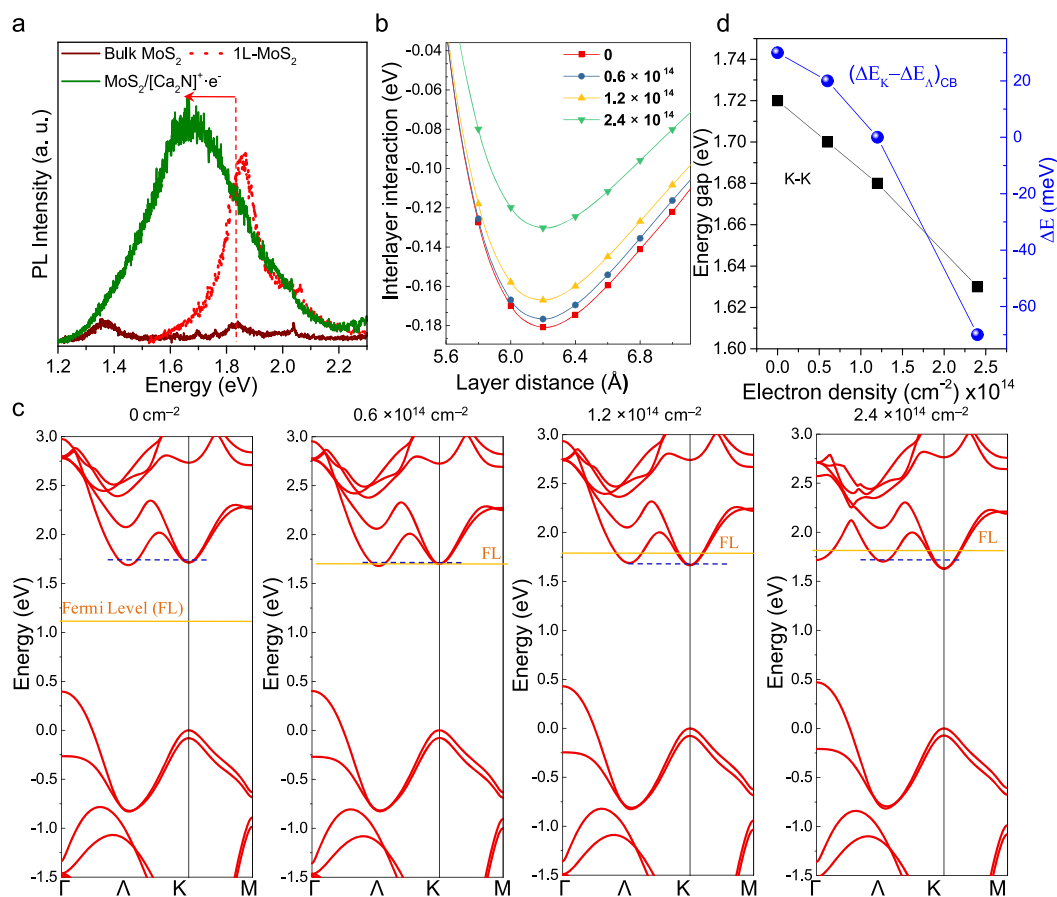


Figure 5. MoS₂ band-structure modulation due to the degenerate electron doping by [Ca₂N]⁺e⁻. (a) PL spectra of pristine 1L-MoS₂ (red line), pristine bulk MoS₂ (brown line), and MoS₂/[Ca₂N]⁺e⁻. The MoS₂/[Ca₂N]⁺e⁻ structure displayed a broad red-shifted (by ~200 meV) PL intensity greater than that of pristine 1L-MoS₂. (b) Interlayer coupling energies calculated for 2L-MoS₂ at three different doping densities. Interlayer interaction of 2L-MoS₂ decreases with increasing doping density. (c) Electronic band structures of 2L-MoS₂ calculated at various doping densities. Note the gradual lowering (raising) of the CBV at the K (Λ) point with an increase in electron density. (d) Plot of the doping density-dependent bandgap reduction at the K point (black squares) and energy difference in the CBV at the K and Λ points ($\Delta E_K - \Delta E_\Lambda$) (blue balls).

because of its isotropic lattice structure.^{47,48} On the contrary, the J_3 mode at 350 cm⁻¹ and J_2 mode at 236 cm⁻¹ in the Raman spectrum of thinner bulk MoS₂ on [Ca₂N]⁺e⁻ were highly dependent on sample rotation under the polarized Raman measurement as shown in Figure 4a,b. Polarization dependence of A_g modes of 1T' MoTe₂ is given by^{23,54,55}

$$I = [c + (b - c)\cos^2\theta]^2$$

where I represents the Raman intensity, b and c the diagonal elements of the A_g mode Raman tensor, and θ represents the angle between the crystalline axis and the parallel polarization directions of the incident and the scattered light. Previously, bifold anisotropy of polarized Raman scattering was observed in distorted octahedral lattices such as ReS₂ and ReSe₂,^{49,50} and the anisotropy factor (b/c) of 2.3 was obtained for 1T' MoTe₂.²³ Our DFT calculation suggests that the A_g mode of 1T' MoS₂ must have the similar dependence, but the anisotropy factor has not been estimated so far. We fitted the observed polarization dependence of J_3 mode and J_2 modes with the above equation, and the anisotropy factors were estimated to be 2.4 and 1.9, respectively, which confirms that this specific bulk MoS₂ underwent phase transition from a 2H

hexagonal lattice to a distorted octahedral lattice upon electron injection by 2D-[Ca₂N]⁺e⁻.

MoS₂/[Ca₂N]⁺e⁻ that displayed the spectrum 1 in Figure 3a was characterized to be a completely metallic phase and it exhibited no detectable PL intensity, which is consistent with the zero gap metallic structure of 1T' MoS₂.^{13,21,22} In contrast, we observed unusually strong PL from bulk MoS₂ that displayed both 2H and 1T' specific Raman modes (spectrum 2 in Figure 3a); the observed PL intensity is even greater than the PL intensity of pristine monolayer MoS₂, whereas the PL intensity of the pristine bulk MoS₂ is much smaller than that of monolayer MoS₂ as shown in Figure 5a and Supporting Information, Figure S2. We performed the differential reflectance mapping on these samples (see the Supporting Information, Figure S8) and found that absorption and the surface quality were similar to pristine bulk MoS₂, which excludes the roles of higher absorption or scattering to observed strong PL from MoS₂/[Ca₂N]⁺e⁻. The PL peak position of this MoS₂ was located at 1.63 eV, ~200 meV lower than the original peak position corresponding to the direct transition of bulk MoS₂ at the K-point of the Brillouin zone. The brown line in the Figure 5a, shows the PL spectrum of pristine bulk MoS₂ with distinct PL emission at 1.82 and 1.35

eV, which could be attributed to the transitions along the K–K and Λ – Γ lines, respectively.^{1,56} Our observations suggest that long-range electron injection may have resulted in a meta-phase MoS₂ with weak interlayer coupling and strong PL.

To describe the origin of various footprints of lattice evolution due to long-range electron injection, we performed DFT calculations on the electronic band structure of bilayer MoS₂ with increasing electron doping density to simulate electron transfer from [Ca₂N]⁺e[−], and the results are shown in Figure 5b–d. First, in Figure 5b, we show the variation in the interlayer coupling of bilayer MoS₂ with respect to electron density. Interlayer coupling reached its minimum value at ~ 0.62 nm, which is the typical interlayer distance of 2L-MoS₂ at 0 doping density. While this interlayer distance was insensitive to doping concentration, interlayer coupling at this distance gradually decreased from -0.18 to -0.13 eV as the doped electron density increased from 0 to 2.4×10^{14} cm^{−2}, suggesting interlayer decoupling in MoS₂. Such decoupling has a direct effect on the electronic band structure. In Figure 5c, we show the band structures of 2L-MoS₂ calculated at electron densities of 0, 0.6×10^{14} , 1.2×10^{14} , and 2.4×10^{14} cm^{−2}. It is to be noted that the valence band at the K point was taken as the reference for all calculations. We noticed that the conduction band valley (CBV) at the K-point lowered monotonically with an increase in electron density, exhibiting a 90 meV bandgap reduction at the K-point at an electron density of 2.4×10^{14} cm^{−2}. A plot of K–K transition energy vs electron density is shown in Figure 5d. The observed reduction in K–K transition energy with an increase in electron density is consistent with the observation of large red-shift of the PL peak of MoS₂/[Ca₂N]⁺e[−] heterostructures. In contrast, our calculations showed that the CBV at the Λ point increased with an increase in electron density (indicated by a horizontal dashed line in Figure 5c). As a result, at an electron density of 1.2×10^{14} cm^{−2}, direct bandgap energy at the K–K point is equal to the indirect band (Λ – Γ) gap energy. At 2.4×10^{14} cm^{−2}, the difference ($\Delta E_K - \Delta E_\Lambda$) between Λ –K and K–K energy gaps was -70 meV, as shown in Figure 5d. Our DFT results on the inversion between Λ –K and K–K energy gaps with respect to an increase in the electron density are consistent with previously calculated band structures of MoS₂ at high electron concentrations.^{36,52} Meanwhile, the minimum energy of the conduction band shifted from the Λ point to the K point due to electron doping, which enhances the PL signal at the K-valley as suggested by the previously reported local strain-induced band gap modulation in bulk TMD crystal,⁵⁶ although the indirect band gap in TMDs (Γ –K) was maintained, similar to the doping concentration-dependent band structure of 2L-MoS₂ (Figure 5c). Another possible explanation for the strong PL of bulk MoS₂ at high electron densities could be the formation of twisted bulk MoS₂ due to weaker interlayer coupling, as suggested by the multiple phonon modes observed in Raman measurements (Figure 3a).⁵⁷ The band structure of twisted bulk MoS₂ may be different from that calculated for the AB stacking bilayer in Figure 5c; the position of the topmost valence band at the Γ point in the twisted structure is supposed to be down-shifted to a greater extent than that in the case of AB stacking.⁵⁸ We conjecture that the twisted band structure may be responsible for the stronger PL signal of bulk MoS₂ at the K-point than that observed with AB stacking. Our calculations provide a convincing explanation for the structural modulation of bulk MoS₂ by injecting electrons from [Ca₂N]⁺e[−], implying that

strong doping can lead to a giant bandgap renormalization. The greatly enhanced PL emission of bulk MoS₂ may be attributed to the formation of twisted multilayer MoS₂ due to the decoupling of interlayer interactions.

CONCLUSIONS

In summary, we demonstrated the complete phase conversion of bulk MoS₂ from 2H to 1T' by preparing heterostructures with 2D-[Ca₂N]⁺e[−]. This conversion was confirmed by the presence of metallic-phase Raman modes and their polarization dependency. We observed a giant band structure renormalization (>200 meV) due to degenerate electron doping by charge injection from 2D-[Ca₂N]⁺e[−]. We also observed a large softening of Raman modes and strong PL intensity in the vicinity of the metallic regime, suggesting weakened interlayer coupling in electron-injected bulk MoS₂, which was further confirmed by DFT calculations. Our demonstration of giant bandgap renormalization, complete metallization, and facile phase engineering of bulk MoS₂ using 2D[Ca₂N]⁺e[−] as the substrate provides clear snapshots of the intriguing phase-transition process of MoS₂ due to direct electron injection from solid-state 2D electrides and helps to develop 2D materials of diverse phases of electrical and optical properties.

EXPERIMENTAL METHODS

Synthesis of Dicalcium Nitride [Ca₂N]⁺e[−]. Two-dimensional crystals of [Ca₂N]⁺e[−] were synthesized using a previous protocol on the solid-state reaction between calcium nitride (Ca₃N₂) powders and calcium metal.^{25,26} A mixture of Ca₃N₂ powders and calcium chips at a molar ratio of 1:1 was pressed into a pellet at a pressure of 20–30 MPa. The pellet was fully covered with molybdenum foil and annealed at 800 °C for 48 h under vacuum ($\sim 10^{-3}$ Pa), after which the sample was quenched with water. The obtained [Ca₂N]⁺e[−] was ground into powder in an agate mortar in a nitrogen-filled glovebox and reannealed at 800 °C for 48 h under vacuum to improve homogeneity.

Preparation of the Heterostructure Using [Ca₂N]⁺e[−] and 2H MoS₂. We first exfoliated bulk [Ca₂N]⁺e[−] crystals using the scotch-tape method onto a glass slide as the substrate in an argon environment and then heated them at 80 °C. Later, we conducted the exfoliation of bulk 2H MoS₂ (2D Semiconductor Supplies Corp.) on a polydimethylsiloxane (PDMS) substrate. Exfoliated MoS₂ on the PDMS substrate was directly transferred onto pre-exfoliated 2D-[Ca₂N]⁺e[−] to obtain the desired heterostructure. The sample was heated at 80 °C to ensure strong contact between MoS₂ crystals and [Ca₂N]⁺e[−]. The entire sample-preparation process was carried out in an argon gas environment to prevent the oxidation of [Ca₂N]⁺e[−]. Finally, the heterostructure samples were air-sealed for in-air optical characterization by capping with a cover glass.

PL and Raman Spectral Measurements. Confocal PL and Raman spectral measurements were performed on a lab-made confocal microscope system.⁷ An argon laser (514.5 nm) was used at a laser power of 500 μ W for obtaining the Raman and PL spectra. An objective lens with a numerical aperture of 0.95 was used to illuminate the samples and collect the scattered light. The collected light was guided to a 50 cm-long monochromator equipped with a cooled charge-coupled device through an optical fiber with a core diameter of 100 μ m; this acted as a confocal detection pinhole. Diffraction gratings with 150 and 1800 grooves mm^{−1} were used for PL and Raman measurements, respectively.

DFT Calculation Methods. The electronic band structure of bilayer MoS₂ was investigated by DFT analysis performed using Quantum Espresso code.⁵⁹ Ultrasoft potentials were used in all calculations with a cutoff energy of 60 Ry. A Monkhorst–Pack *k*-grid (24 \times 24 \times 1) was used to optimize the cell structure at a pressure and force <0.5 kbar and 10^{−3} Ry/Bohr, respectively. We implemented

the VDW-DF-OB86 functional to treat interlayer interactions between MoS₂ layers.^{60–62}

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsnano.9b08037>.

Raman spectra of dicalcium nitride crystals, PL and Raman spectra of MoS₂ films, theoretical evidence of degenerate electron doping induced phonon softening in the MoS₂ crystal, thickness estimation of the MoS₂ crystal placed on a [Ca₂N]⁺e⁻ substrate, assignment of Raman peak positions of MoS₂/[Ca₂N]⁺e⁻ heterostructures, Raman spectrum of the thin MoS₂ on the top of the [Ca₂N]⁺e⁻, polarization-dependent Raman vibrational modes of 2H MoS₂ and differential reflectance measurement (PDF)

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Notes

The authors declare no competing financial interest.

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