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Fabrication of Stacked MoS₂ Bilayer with Weak Interlayer Coupling by Reduced Graphene Oxide Spacer

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We fabricated the stacked bilayer molybdenum disulfide (MoS₂) by using reduced graphene oxide (rGO) as a spacer for increasing the optoelectronic properties of MoS₂. The rGO can decrease the interlayer coupling between the stacked bilayer MoS₂ and retain the direct band gap property of MoS₂. We observed a twofold enhancement of the photoluminescence intensity of the stacked MoS₂ bilayer. In the Raman scattering, we observed that the E_{2g}¹ and A_{1g} modes of the stacked bilayer MoS₂ with rGO were further shifted compared to monolayer MoS₂, which is due to the van der Waals (vdW) interaction and the strain effect between the MoS₂ and rGO layers. The findings of this study will expand the applicability of monolayer MoS₂ for high-performance optoelectronic devices by enhancing the optical properties using a vdW spacer.

The recent discovery of a new class of two-dimensional (2D) materials, transition metal dichalcogenides (TMDs), such as molybdenum disulfide (MoS₂) and tungsten disulfide (WS₂), has attracted attention because of their unique layer-dependent electrical and optical properties^{1–3}. For example, MoS₂ possesses an indirect band gap of ~1.29 eV in bulk, but it becomes a direct optical band gap of ~1.90 eV in the monolayer and can affect the electronic and optical properties^{4,5}. These interesting features in MoS₂ have opened up new possibilities for optoelectronic applications⁶. Previous reports have mainly focused on the fabrication of optoelectronic devices, such as phototransistors, photodetectors, light-emitting diodes, and solar cells with monolayer MoS₂^{7–10}. However, the absorbance of monolayer MoS₂ is not strong enough to realize efficient optoelectronic devices⁴. Compared to the monolayer, bilayer or few-layer MoS₂ shows improved absorbance and carrier mobility¹¹. However, few-layer MoS₂ has an indirect band gap that limits its applicability for high-efficiency optoelectronic devices. Therefore, many researchers have worked to improve the mobility and intensity of the photoluminescence (PL) of MoS₂ with doping, strain, and defect engineering^{12–14}. More recently, it was reported that artificially stacked TMD with a spacer between the individual layers can improve the optical properties while maintaining the intrinsic properties of TMD^{15–18}. Among them, Piljae Joo *et al.* demonstrated an enhancement of the photoluminescence (PL) of stacked few-layer MoS₂ with a polymer spacing layer. However, this method has been realized with few-layer or multilayer MoS₂¹⁵. For the stacking of monolayer TMD, hexagonal boron nitride (h-BN) was introduced as the spacer for the TMD hetero-bilayer or homo-bilayer^{16–18}. These results suggest that stacked TMD with the h-BN layer can retain the direct band gap feature of the monolayer TMD. However, the procedure for transferring h-BN onto TMDs is somewhat complicated; thus, it is difficult to apply it to large areas. Therefore, it is essential to devise a simple technique that can be implemented easily into the device-fabrication process to obtain higher optoelectronic properties.

Here, we demonstrate the fabrication of stacked MoS₂ with a reduced graphene oxide (rGO) spacer. rGO has remarkable properties, such as high thermal conductivity, electrical conductivity, and mechanical stability¹⁹. rGO can decrease the interlayer coupling of stacked MoS₂, such that rGO is an ideal sheet to decouple the top and bottom MoS₂ and retain the direct band gap property of MoS₂. We observed an enhancement of the PL intensity of stacked MoS₂ with an rGO spacer compared to that of monolayer MoS₂. We also systematically investigated the origin of the structural, chemical, and optical properties of the stacked MoS₂ with/without spacer layer.

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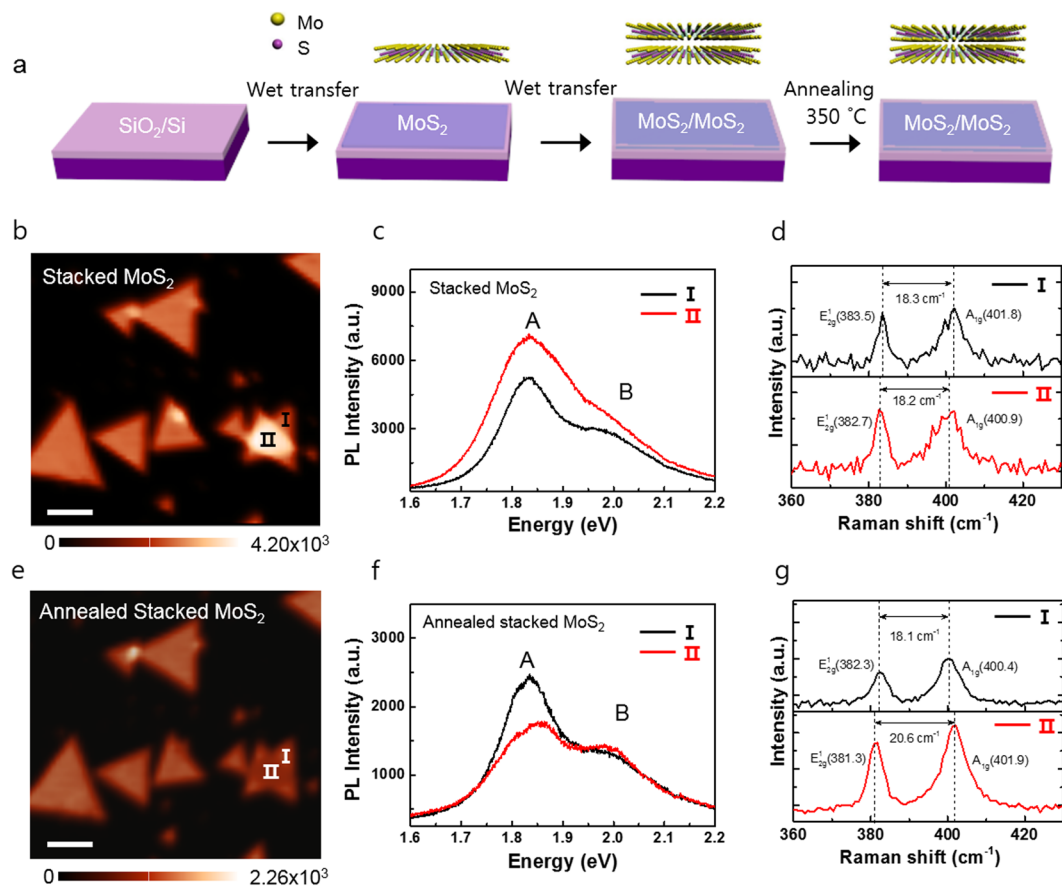


Figure 1. (a) schematic of sample preparation of stacked MoS₂ with/without thermal annealing. PL intensity map and PL and Raman spectra of I and II region in the stacked MoS₂ (b–d) before and (e–g) after annealing.

Results and Discussion

Figure 1a illustrates the preparation process of stacked MoS₂. The chemical vapor deposition (CVD)-grown monolayer MoS₂ films were transferred onto the 300-nm-thick of SiO₂/Si substrate using the conventional wet transfer method. (See Methods section for details) To prepare stacked bilayer MoS₂, another CVD-grown monolayer MoS₂ film was transferred onto the monolayer-MoS₂/SiO₂/Si template. First, we confirmed the basic optical properties of the monolayer MoS₂ by using the PL and Raman analysis (Supporting Information (SI), Fig. S1). Figure 1b shows the PL intensity map of monolayer MoS₂ flakes and stacked MoS₂ flakes. Normally, the PL intensity of bilayer MoS₂ is much lower than that of monolayer MoS₂ because it has an indirect band gap^{4,5}.

Interestingly, we observed that the integrated PL intensity of stacked MoS₂ flake (II) is slightly higher than that of the monolayer MoS₂ flake (I). It is implied that stacked bilayer MoS₂ flake has different optical features that are irrelevant to the layer-dependent optical properties of MoS₂.

To confirm the PL peak position of the sample, we extracted the PL spectrum from the stacked MoS₂ flakes marked I and II in Fig. 1b. The A and B exciton peak positions at ~1.83 and ~1.98 eV of I and II were almost the same (Fig. 1c). Figure 1d shows the Raman spectra of I and II. We observed the two Raman modes at approximately 383.5 and 401.8 cm⁻¹ of monolayer region I in the stacked MoS₂, corresponding to the E_{2g}¹ and A_{1g} modes²⁰. The distance between the E_{2g}¹ and A_{1g} modes was 18.3 cm⁻¹. The A_{1g} and E_{2g}¹ modes of the region II were shifted compared to those of region I, but the distance between the two Raman modes is around 18.2 cm⁻¹, which is related to the monolayer properties of MoS₂²⁰. We confirmed the same trend from other stacked MoS₂ flakes, regardless of stacking angle (see SI, Fig. S2). Figure 1e shows the PL intensity map of the same stacked MoS₂ sample after thermal annealing at ~300 °C for 1 h under nitrogen gas. After thermal annealing, the integrated PL intensity was reduced for all of the stacked MoS₂ flakes. The PL intensity of region II is decreased compared to that of region I. In region II of stacked MoS₂, the intensity of exciton peak A decreases significantly and is slightly shifted (Fig. 1f). These PL results show a similar characteristic to the natural bilayer^{21,22}. In addition, the distance between the E_{2g}¹ and A_{1g} modes increased from ~18.2 to ~20.6 cm⁻¹, which is somewhat similar to the bilayer properties (Fig. 1g)²⁰. From the PL and Raman results, we found that annealed stacked MoS₂ regions have bilayer properties.

To investigate the phenomena of the different optical properties of the stacked MoS₂ before and after thermal annealing, we carried out photothermal induced infrared resonance (PTIR) spectroscopy, which simultaneously provides topographical information and infrared absorbance (see SI, Fig. S3)²³. Fig. 2a,b show the AFM topography images and height profiles of stacked MoS₂ before and after thermal annealing, respectively. The

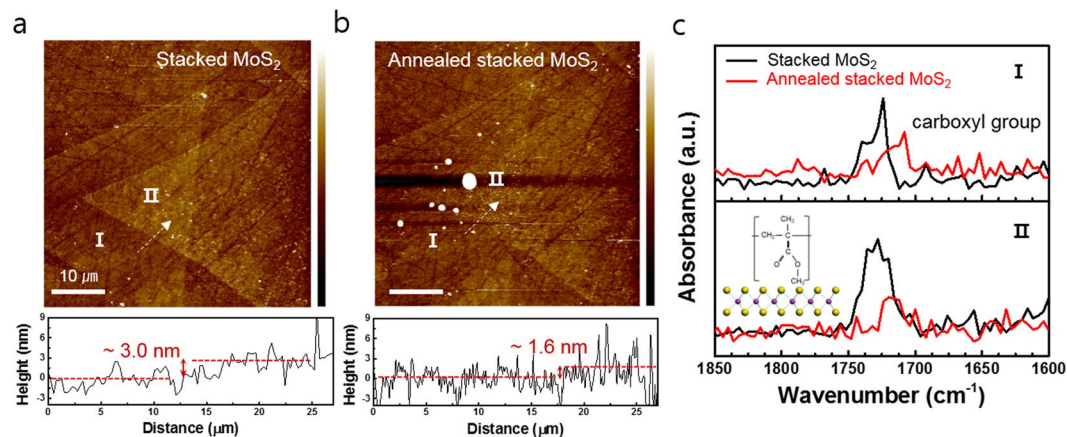


Figure 2. (a,b) is AFM height image of the stacked bilayer MoS₂ before and after annealing. (c) Absorbance spectra of each position of stacked bilayer MoS₂ before and after annealing treatment.

topography images provide distinguishable contrast between the monolayer (I) and stacked (II) MoS₂ region. We measured the thickness of the stacked MoS₂ from region I to II. The thickness of the dashed line at the stacked MoS₂ before thermal annealing was ~3.0 nm. This value is much larger than the expected thickness of the stacked bilayer MoS₂²². However, after thermal annealing, the thickness of the stacked MoS₂ decreased to ~1.6 nm, which is almost identical to the thickness of the bilayer MoS₂²². According to the previous results, it is assumed that defects or organic molecules exist in between the stacked flakes²¹. Fig. 2c presents the FTIR absorbance spectra of regions I and II of the stacked MoS₂ sample before and after thermal annealing treatment. We observed a peak at approximately 1728 cm⁻¹, corresponding to the C=O mode of the poly(methyl methacrylate) (PMMA) residues on the overall area of the stacked MoS₂ sample before thermal annealing²⁴. Normally, PMMA is used as a supporting material during the wet-transfer process. However, polymer residues usually remain on the surface of the TMD thin film after the transfer process, and it is difficult to completely remove them from the TMD thin film^{21,25}. Therefore, the stacked bilayer MoS₂ are thicker than the original bilayer MoS₂ because polymer residues are present on MoS₂. The polymer residue suppresses the interlayer interaction between stacked MoS₂ monolayers. Thus, the PL and Raman results of stacked MoS₂ before annealing showed monolayer characteristics (Fig. 1b–d). In contrast, after thermal annealing, the intensity of the C=O mode decreased. This indicates that the concentration of the PMMA is reduced (Fig. 2c). The thermal annealing removed the polymer residues and caused interlayer interaction in stacked MoS₂. Thus, the thickness of stacked MoS₂ decreased after annealing, and it exhibited bilayer characteristics. From these results, the polymer residues between stacked MoS₂ flakes can act as a spacer, which can retain the intrinsic properties of monolayer MoS₂ prior to thermal annealing. However, during the device-fabrication process, the thermal annealing process inevitably decreases the resistivity of the interface for better device performance²⁶. Therefore, it is necessary to find the optimum spacer in the stacked MoS₂ that blocks the intercoupling between flakes without damage in the thermal annealing process. Figure 3a shows an illustration of the sample-preparation process of the stacked MoS₂ with spacer layer. The CVD-grown monolayer MoS₂ films were transferred onto the 300-nm SiO₂/Si substrate using the wet-transfer method. The prepared GO solution was spin-coated at 500 rpm for 5 s, followed by 1500 rpm for 60 s on top of the monolayer MoS₂ on the SiO₂/Si substrate. The average thickness of the coated rGO on the MoS₂ sheet was approximately 5 ± 2 nm (see SI, Fig. S4). To prepare hybrid stacked MoS₂, another CVD-grown monolayer MoS₂ film was transferred onto the GO-coated monolayer MoS₂/SiO₂/Si template. By thermal annealing treatment at 350 °C for 3 h, we fabricated the stacked MoS₂ with an rGO spacer. To investigate the reduction of GO, we carried out Raman spectroscopy measurement of the GO before and after thermal annealing (see SI, Fig. S5). Figure 3b is the AFM topography image of the stacked MoS₂ with rGO spacer. According to the AFM topography image, the monolayer and stacked MoS₂ region can be distinguished. Additionally, we observed wrinkles and some bubbles on the stacked MoS₂ with rGO sample. These seem to be formed during the transfer process (see SI, Fig. S6). The thickness of the monolayer MoS₂ with rGO (green dot) is approximately 1.1 nm and that of the stacked bilayer MoS₂ with rGO region (orange dot) is approximately 2.7 nm (see SI, Fig. S7). For the spatially resolved optical characterization, confocal PL and Raman measurements were performed for stacked MoS₂ with rGO spacer. Figure 3c shows the PL spectra obtained from each position of the sample. We observed the A and B excitons (~1.86 and 2.0 eV) of MoS₂ and the Raman G and D bands (~2.13 and 2.16 eV) of the rGO at PL spectra¹². As shown in Fig. 3c, the PL intensity of stacked bilayer MoS₂ with rGO (M/rGO/M) region is twice that of rGO on monolayer MoS₂ (rGO/M) or monolayer MoS₂ on rGO (M/rGO) regions. In previous reports, PL quenching was observed in MoS₂ over mechanically exfoliated graphene (MEG) because MEG has semi-metallic properties that result in a metal-semiconductor interface^{27,28}. On the other hand, rGO show semiconductor behavior because of residual oxygen functional groups after thermal annealing^{29,30}. Therefore, PL quenching is not observed in the interface between rGO and MoS₂ stacked structure. Figure 3d represents the PL intensity map for the energy range of 1.67–2.11 eV of stacked bilayer MoS₂ with rGO spacer. The PL intensity map could also be used to distinguish the stacked M/rGO/M, M/rGO, and rGO/M regions because these PL intensity values are different from

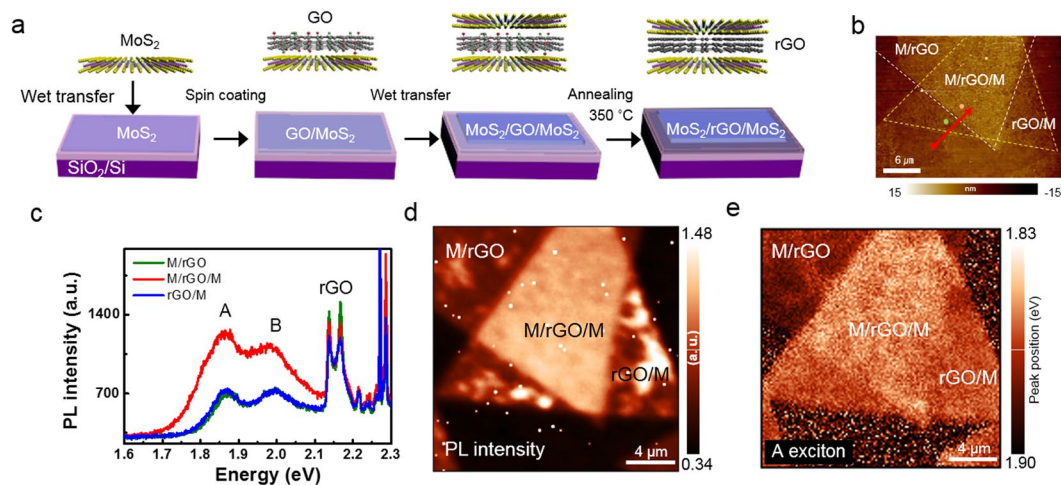


Figure 3. (a) Illustration for sample preparation of stacked MoS₂ with rGO. (b) AFM topography image of the stacked MoS₂ with rGO. (c) PL spectra of the various position of the stacked MoS₂ with rGO. (d) PL intensity map image of stacked MoS₂ with rGO for 1.67~2.11 eV corresponding to the MoS₂. (e) PL peak position map image of the stacked MoS₂ with rGO.

each other. As observed in Fig. 3d, some position of rGO/M region has a higher PL intensity rather than the M/rGO/M region. To compare the intensity, we extracted the PL spectra of bright spot of rGO/M region and M/rGO/M region (see SI, Fig. S8). According to previous report, GO induce the p-type doping of monolayer MoS₂ because of the functional groups of GO¹². We believe that the cause of high PL intensity in the rGO/M region is p-type doping by residue of functional group after thermal annealing. A slight red shift in the A exciton peak is observed in the M/rGO/M region. In addition, the PL peak position of the stacked MoS₂ with rGO sample is different (Fig. 3e). We confirmed that the PL results at stacked MoS₂ with rGO spacer showed different characteristics from those of the stacked MoS₂ without rGO spacer.

To clarify the cause for the considerably changed optical and structural properties of stacked MoS₂ with rGO spacer, we have performed confocal Raman mapping. Figure 4a shows the Raman intensity map of the stacked MoS₂ with rGO. The Raman intensity map provides distinguishable contrast between the stacked and monolayer MoS₂ regions. Figure 4b shows the local Raman spectra of each position. The peak positions of the two Raman modes of stacked MoS₂ with rGO spacer are different from those of the stacked MoS₂ without spacer. The A_{1g} and E_{2g} modes of stacked MoS₂ with rGO were further shifted compared to those of stacked MoS₂ without spacer. In order to understand these results, we focus on the E_{2g}¹ and A_{1g} modes of MoS₂, depending on the region. Figure 4c presents the Raman intensity of the E_{2g}¹ and A_{1g} modes of the MoS₂ according to each position. The higher Raman intensity of the M/rGO/M region originates from the increased scattering cross section by stacked MoS₂ with rGO. Figure 4d shows the peak position of the two Raman modes according to the position of the stacked MoS₂ with rGO spacer. The E_{2g}¹ and A_{1g} modes of the stacked MoS₂ with rGO spacer were observed to shift to the opposite direction compared to that of pristine monolayer MoS₂ (M). Interestingly, we observed that the red shift of E_{2g}¹ on M/rGO/M is large compared to M/rGO and rGO/M, which is shown in the Raman peak position map in Fig. 4e. Furthermore, the full width at half maximum (FWHM) of the E_{2g}¹ peak of M/rGO/M broader than that of other regions (Fig. 4f). This indicates that interfacial strain is generated more on the M/rGO/M than on the M/rGO and rGO/M^{31,32}. In contrast, the blue shift of A_{1g} on M/rGO and rGO/M is larger than the shift observed in M/rGO/M, which is confirmed in the Raman map shown in Fig. 4g. The FWHM of the A_{1g} peak of M/rGO/M slightly broader than that of other regions (Fig. 4h). This means that the van der Waals (vdW) interlayer interaction between the MoS₂ and rGO on M/rGO and rGO/M is stronger than that of M/rGO/M³¹. According to the previous results, the blue shift of A_{1g} of MoS₂ can originate from the p-doping effect by the functional group of GO^{12,33}. The A_{1g} peak appeared to be more strongly shifted as the amount of GO functional groups increased¹². In this study, rGO was prepared with GO by thermal annealing. Thus, most functional groups of GO were removed and only a small amount of the functional group remained on the surface²⁵. In addition, we confirmed that the A_{1g} mode of stacked MoS₂ with rGO was further shifted compared to stacked MoS₂ with GO (see SI, Fig. S9). This means that p-doping of MoS₂ by rGO has a negligible effect on the shift of the A_{1g} peak observed in our stacked MoS₂ with rGO. Therefore, it is concluded that the stacked MoS₂ with rGO spacer is in close contact with vdW interaction between MoS₂ and rGO. We suggest that the rGO can serve as a spacer to decouple the top and bottom MoS₂ and retain the direct band gap of MoS₂.

Conclusions

In conclusion, we successfully fabricated stacked bilayer MoS₂ with rGO spacer. We observed that the decreased PL intensity in the stacked MoS₂ layers without spacer because of bilayer properties. Meanwhile, when the rGO is used as a spacer in between MoS₂ sheets, PL intensity shows sum up of PL from two individual MoS₂ sheets. It clearly indicates that rGO is a suitable material to decouple the top and bottom MoS₂ and retain the direct band gap of MoS₂. In the Raman results, we observed the vdW interaction and the strain effect between the MoS₂ and

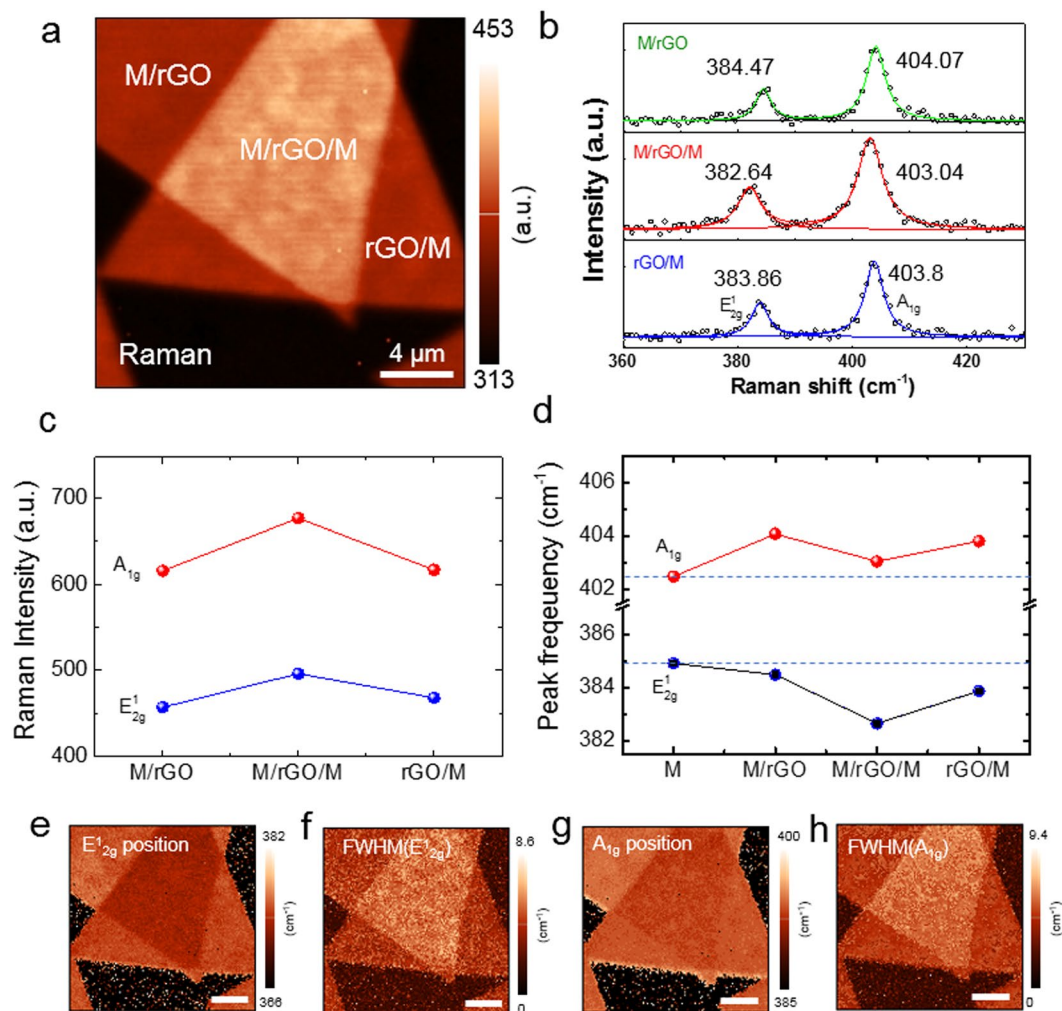


Figure 4. (a) Raman intensity map of the stacked MoS₂ with rGO. (b) Local Raman spectra of various position of the stacked MoS₂ with rGO. (c) Raman intensity and (d) peak position of the E_{12g} and A_{1g} of the various position of stacked MoS₂ with rGO. (e) Peak position and FWHM map of E_{12g} (f) peak position and FWHM map of the A_{1g} of the sample.

rGO layers. We expect that these results will provide a fundamental understanding of the interlayer interaction of stacked 2D materials and enable further development of stacked MoS₂-based devices.

Methods

Conventional wet transfer method. Poly(methyl methacrylate) (PMMA) (Micro Chem, 4 wt% in chlorobenzene) was coated onto monolayer MoS₂ grown on a SiO₂/Si substrate to serve as a supporting layer for the transfer process. After being coated with PMMA, the sample was floated on a 1 M potassium hydroxide (KOH) solution at 80 °C to remove the SiO₂ layer. Subsequently, only monolayer MoS₂ with PMMA remained on the KOH solution. The remaining PMMA/MoS₂ was washed with DI water to remove any residual KOH etchant. The washed PMMA/MoS₂ was transferred to the SiO₂/Si substrate. After the drying process was complete, the PMMA was removed using acetone.

Preparation of the GO. GO was synthesized from natural graphite (Alfa Aesar, 99.999% purity, 200 mesh) by modified Hummers' method. First, 5 g of graphite powder and 350 mL of 10 M Sulfuric acid (H₂SO₄) were blended. KMnO₄ (15 g) was slowly added over approximately an 1 h. Stirring was continued for 2 h in a cooled water bath. The mixture was strongly stirred for 3 days at room temperature. Deionized water was added and stirring for 10 min. The mixture was stirred for 2 h at room temperature after the addition of an aqueous solution of H₂O₂ (30 wt%). Aqueous solution of HCl (35 wt%) was then added and stirred for 30 min at room temperature. After the supernatant solution was decanted, deionized water was slowly added and stirred for 30 min. The GO solution 1 g/l in water was sonicated for 1 h to exfoliate the GO sheets. To obtain dispersed GO, centrifugation at 10,000 rpm was performed for 1 h, and the supernatant solution was decanted.

Characterization Methods. PL spectra were obtained using a confocal PL spectrometer equipped with an objective lens with high numerical aperture of 0.7 and a diode-pumped solid-state laser (532 nm). Confocal

Raman spectroscopy was conducted using a commercial multifunctional microscope (NTEGRA, NT-MDT). The atomic force microscopy (AFM) topography image and photothermal induced infrared resonance (PTIR) absorption of the sample was obtained using a commercial Nano-IR system (Anasys Instruments).

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Author Contributions

H.M.O. and M.S.J. designed the experiments. H.M.O. and H.K. fabricated the stacked MoS₂ sample conducted the Nano I.R. experiment. H.M.O. performed P.L. and Raman experiments. H.K. performed growth experiments for graphene. H.M.O. and M.S.J. analyzed the results and wrote the manuscript. All authors discussed the results and commented on the manuscript.

Additional Information

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