

618 Supplementary Information for

619 Resource limitation modulates the fate of dissimilated nitrogen in a dual-pathway

620 Actinobacterium

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637 **SI Introduction**

638 Dual-pathway respiratory nitrite reducers, where respiratory ammonification and denitrification

639 modules are encoded in the same genome, provide a unique system to investigate the molecular

640 mechanisms of C:NO₃⁻ control on pathway selection as they blend the presumptively ancient

641 (NrfA) and modern (NirK) N-reducing modules. While many dual-pathway denitrifiers and

642 respiratory ammonifiers have been identified (many of which use LP ETCs), most still lack

643 phenotypic characterization[21–23]. Only *Shewanella loihica* PV-4, a Gram-negative, γ -

644 proteobacterium that uses UQ and dimethylmenaquinone (DMK)-based bioenergetics has been

645 thoroughly characterized[3, 24, 43]. Nonetheless, these dual-pathway nitrite reducers are capable

646 of differential energy conservation through pathway bifurcation of nitrite. Yet no mechanistic or

647 metabolic models exist to explain how these dual-pathway organisms partition electron flow and

648 proton translocation between pathways to maximize resource-use efficiency. It is not known if

649 these organisms differentially distribute electron flow through alternative respiratory chains in

650 response to shifting resource availability or control each pathway independently in response to
651 resource thresholds. Is there an evolutionary and biochemical basis for pathway selection?

652 It has long been postulated that C:NO₃⁻ ratio modulates the activity of respiratory
653 ammonification versus denitrification[2], with numerous studies in recent years differing in the
654 methods used to calculate C:NO₃⁻ ratio, carbon sources used, range of C:N ratios tested,
655 magnitudes of C and N concentrations, sample types (enrichment cultures vs isolates), and
656 culturing mode (chemostat/batch culture) (Table S4)[3–6, 34]. The hypothesis for C:NO₃⁻
657 control states that high C:NO₃⁻ ratios (stoichiometric limitation of nitrate relative to C) drive
658 respiratory ammonification while low C:NO₃⁻ ratios (stoichiometric limitation of C relative to
659 nitrate) drive denitrification. This hypothesis is based on the rationale that denitrification
660 theoretically yields more free energy per electron, but respiratory ammonification yields more
661 free energy per nitrate due to the stoichiometry of each reaction: one nitrate is needed to make
662 ammonium while two are needed to make N₂O/N₂[2, 39]. Thus, when C:NO₃⁻ ratio is low, cells
663 select denitrification to maximize the free energy advantage per electron. When C:NO₃⁻ is high,
664 cells select respiratory ammonification to maximize the free energy advantage per nitrate.
665 However, the theoretical basis for this prediction is inconsistent with the observation that growth
666 yields of pure denitrifiers are significantly lower than expected and lower than growth yields of
667 respiratory ammonification[39]. This observation suggests that despite a lower free energy yield
668 (ΔG) compared to denitrification, respiratory ammonification conserves more energy during
669 catabolism, through the generation of a proton motive force (Δp) in the ETC, to build more
670 biomass during anabolism. Thus, there is a need to better understand the effects of different
671 nutrient limitations on growth and pathway selection (i.e., allocation of C and N to dissimilatory
672 and assimilatory processes), which are often confounded by C:NO₃⁻. For example, reportedly
673 “low” C:NO₃⁻ ratios rarely fall below 1.5 have rarely been tested (Table S4), conditions in which
674 C would be a growth-limiting resource. Instead, pathway selection should be better predicted by
675 1) Liebig’s law of the minimum (LM) and 2) the maximum power principle (MPP)[37]. Under
676 the MPP, which states that biological systems are designed to maximize power intake and energy
677 transformation, the cell’s aim is therefore to maximize power (i.e., realized in the form of growth
678 rate and yield) given the constraints of a growth-limiting nutrient (i.e., LM). Therefore, limitation
679 of a growth-limiting nutrient, whether it be C or NO₃⁻, should dictate the selection of the most
680 efficient respiratory module to maximize power (i.e., MPP).

681 **SI Materials and Methods**

682 Sample collection: *I. calvum* was isolated from a groundwater well (GW247) collected on
683 2/18/2013 at Oak Ridge National Laboratory (Lat: 35.97990, Long: 84.27059) with a
684 groundwater temperature of 15.57 °C, conductivity of 521.9 µS/cm, and pH of 7.71. *I. calvum*
685 was isolated and obtained from Dr. R. Chakraborty (Lawrence Berkeley National Laboratory).
686 We began our experimental process by first picking single colonies from LB agar plates and
687 transferring the clonal isolates to LB broth.

688 Media preparation: Media preparation was conducted in a 2L Widdel Flask. After autoclaving,
689 the media was immediately put under an anoxic headspace (N₂/CO₂ 80:20 mix) and sterile
690 filtered (0.2µm) trace elements, trace vitamins, and reducing agent were added. The media was
691 cooled under an anoxic headspace and buffered with bicarbonate to maintain a pH of 7.2.
692 Hungate technique was used to dispense media into culture tubes (20 mL) and serum vials (100
693 mL) pre-flushed with a sterile stream of ultra-high purity (UHP) N₂ and sealed with blue 1” butyl
694 rubber stoppers. End-point cultures were grown in Balch tubes (18x150-mm glass tube) sealed
695 with butyl rubber stoppers. Cultures for time-course sampling were grown in 160ml serum vials.
696 All end-point experiments were terminated after 100 hours unless otherwise noted.

697 Growth Curve/Cell counts/Yield Measurements: Growth curves were measured from scratch-
698 free Balch-tubes grown cultures using an automated optical density reader at OD₆₀₀ nm
699 (Lumenautix LLC, Reno, NV). End-point cultures were monitored until all replicates reached
700 stationary phase (65-100 hours depending on C:NO₃⁻ treatment) (Figure S6).

701 Cell counts were performed by fixing cells in 4% paraformaldehyde (final concentration) for 20
702 minutes, filtered onto 0.2µm pore-sized black polycarbonate filters, and washed three times with
703 phosphate buffered saline (PBS, pH 7.2). Filtered cells captured on the black polycarbonate
704 filters were stained with SYBR[®] Gold nucleic acid stain (10-minute incubation) (ThermoFisher
705 Scientific) and counted manually with a fluorescence microscope (Olympus BX60, Tokyo,
706 Japan). We collected cells from during lag phase, exponential phase, and stationary phase in
707 order to create a standard curve of cell counts versus optical density (OD₆₀₀). We fit a linear
708 model to cell count versus OD₆₀₀ (R²=0.99) and used the resulting linear equation for cell count
709 enumeration for growth curves during our various treatment conditions.

710 Biomass concentrations were measured by filtration and drying as per standard protocol [26] for
711 8mM lactate/12mM nitrate and 0.8mM lactate/1.2mM nitrate treatments and conducted in
712 parallel with growth curve/cell counts as described above. Analysis from triplicate cultures
713 yielded (0.064 ± 0.003) and (0.016 ± 0.001) mg of biomass (dry weight) ml^{-1} for 8mM and 0.8
714 mM lactate cultures, respectively. Cell counts from stationary phase cultures were $(1.5 \pm 0.05) \times$
715 10^7 and $(1.16 \pm 0.09) \times 10^6$ for 8mM and 0.8 mM lactate cultures, respectively. From these
716 values the dry weight of a single *I. calvum* cell was estimated to be 1.09×10^{-10} g. Growth yield
717 (*Y*) (Table S3) was calculated by dividing biomass (g) by lactate mass (g) and moles consumed,
718 as described by [56]. Lactate measurements are described below.

719 Thermodynamic calculations for anaerobic lactate oxidation with nitrate and nitrite were carried
720 out using standard Gibbs free-energy values defined by Thauer *et al.*, [57].

721 Ion and Gas Chromatography Measurements: New glass IC vials were used for every sample in
722 order to ensure no cross contamination of analytes. Ammonium production via respiratory nitrite
723 ammonification was measured as described by [24]. Briefly, because the bacterium
724 simultaneously produces (via dissimilation) and consumes (via assimilation) ammonium,
725 ammonium consumption was first measured with O_2 and lactate by calculating the difference
726 between starting and ending ammonium concentrations. These ammonium consumption values
727 were then normalized to lactate consumed ($0.31 \mu\text{mols NH}_4^+/\text{lactate}$) ($7.07 \times 10^{-7} \mu\text{mols NH}_4^+/\text{cell}$
728 calculated from average cell number of stationary phase biomass; Figure S6). Ammonium
729 production during nitrate reducing conditions was then calculated using the mass balance
730 approach from [58] for Total Belowground Carbon Allocation (TBCA) but adapted for nitrogen
731 flux instead of carbon flux:

$$732 \quad \Delta\text{NH}_4^+ = (\Delta\text{lactate}_{\text{start-end}} \times 0.31 \mu\text{mols NH}_4^+/\text{lactate}) + \Delta\text{NH}_4^+_{\text{end-start}} \quad (1)$$

733 Here, the $\Delta\text{lactate}_{\text{start-end}}$ variable (μmols) is multiplied by the ammonium consumed per lactate
734 consumed constant. This value is added to the $\Delta\text{NH}_4^+_{\text{end-start}}$ variable (μmols), denoted as ending
735 minus starting concentration, which defines whether the change in ammonium is positive (more
736 ammonium produced than consumed) or negative (more ammonium consumed than produced).

737 Headspace gas from Balch tubes and serum vials was sampled with volume appropriate gastight
738 syringes (Hamilton Company, Reno, NV) pre-flushed with UHP N₂. For high and low nutrient
739 treatments, 10µl and 100µl of headspace were sampled and diluted into 12ml exetainers (Labco,
740 Lampter, Wales, UK) over-pressurized with 15ml UHP N₂, respectively. Similar dilutions were
741 performed for nitrite as e-acceptor experiments, ammonium-deplete experiments, and time-series
742 experiments. For time-series experiments, an equal volume of headspace gas that was removed at
743 each time-point was replaced with sterile UHP N₂. N₂O and NO were measured by gas
744 chromatography (Shimadzu Greenhouse Gas Analyzer GC-2014) using a 500µl injection
745 volume. The rubber septa on the injection port of the GC was replaced after 100 injections in
746 order to prevent leakage of the sample after the injection needle was lifted out from the injection
747 port. Aqueous concentrations of N₂O were calculated using a Henry's constant of 1.751 (mM
748 (g)/mM (aq)) corrected for the medium's ionic strength and temperature. A total of 8-11
749 replicates per treatment were analyzed for all experiments discussed in this work (Table S5).

750 Phylogenetic, Genomic, and Transcriptomic Analysis: A set of 34 NrfA amino acid sequences,
751 representing 33 complete genome sequences and 1 octaheme nitrite reductase (ONR) from
752 known respiratory ammonification organisms were downloaded from GenBank (Table S1). A
753 multiple sequence alignment (MSA) was generated from the sequences annotated as cytochrome
754 c nitrite reductase and ONR using MUSCLE [59]. The resulting alignment was visualized within
755 MEGA5 [60] where the alignment was manually screened for the presence of conserved amino
756 acid residues consistent with those found in NrfA (i.e., heme motifs). A maximum likelihood tree
757 was created from the alignment using RAxML [61] with 500 bootstrap iterations. The presence
758 of NapA, NarG, NirK, and Nor modules were manually queried from each NCBI genome in our
759 set and confirmed by MSA, as described above. Metabolic pathway for pool quinone type was
760 queried on BioCyc Pathway/Genome Database (biocyc.org) for each organism in our set. The
761 structure of *I. calvum*'s NirK protein was predicted using the protein structure predicting
762 algorithm Phyre2 [62]. Protein atomic composition for C and N was calculated from amino acid
763 sequences as input files, as described by [14, 15], using custom python scripts for each element
764 separately (github.com/dvuono/Cost_minimization).

765 Due to the high similarity of C5 to 7KIP, reads were aligned to the *Intrasporangium calvum*
766 genomic reference sequence and gtf file (Acc: NC_014830.1) using the STAR RNA-seq

767 aligner[63], with the --limitBAMsortRam parameter set to the recommended value by STAR.
768 Sequence reads were mapped to genomic features to obtain count data using featureCounts [64].
769 Systematic changes across experimental conditions were performed on normalized read counts in
770 DESeq2 [65]. The RNA-seq data reported in this study are available within the NCBI BioProject
771 number PRJNA475609.

772 **SI Results**

773 **Genomic analysis of *I. calvum* C5.** *I. calvum* C5 was isolated from a nitrate contaminated well
774 (>200mM nitrate) at the Oak Ridge National Laboratory Field Research Station in Oak Ridge,
775 TN. The strain was selected for further analysis due to its nitrate reducing phenotype in minimal
776 media. The genome consists of a 4,025,044-base pair chromosome and encodes for 3,722
777 predicted genes, 2,665 protein coding genes, 57 RNA genes, and two rRNA operons.

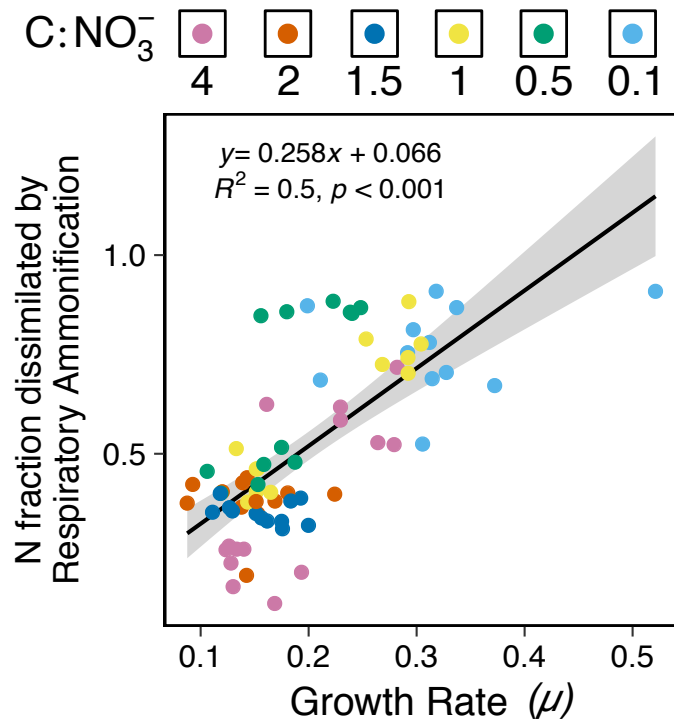
778 ***I. calvum* encodes for a functional NrfAH complex and assimilates NH₄⁺ via respiratory**
779 **nitrite ammonification.** The potential routes for N-assimilation were screened using aerobic
780 minimal media with defined C-source/e-donor and e-acceptor, and with nitrate or ammonium as
781 assimilatory N-sources. Based on genomic information, the bacterium possesses no known
782 assimilatory nitrate reductase, but encodes for an ammonium transporter (Intca_RS11655) and
783 GS/GOGAT pathways (Intca_RS13810; Intca_RS11930; Intca_RS08335, 08340), suggesting
784 that ammonium is its sole assimilatory N source. Indeed, we observed no aerobic growth with
785 8mM lactate, O₂, and nitrate as N-source. However, when grown on 8mM lactate, O₂, and
786 1.5mM ammonium as N-source, *I. calvum* displayed a typical growth curve with a specific
787 growth rate of 0.4±0.02 μ (1.7±0.1 doublings/hour) (Figure S7).

788 **SI Discussion**

789 *S. loihica* PV-4 (a Gram-negative dual-pathway γ-proteobacterium) and *I. calvum* use different
790 types of nitrate reducing modules. *S. loihica* PV-4 utilizes NapA whereas *I. calvum* utilizes NarG
791 (Figure 1). The latter translocates two H⁺ per nitrate reduced while NapA consumes two H⁺ in
792 the periplasm [17]. Both reductases would generate a Δ*p* via NADH dehydrogenase H⁺
793 translocation, but the NapA module would result in a net loss of two H⁺, which may impact the
794 selection of downstream respiratory modules. For example, given the lower-than-expected
795 observed growth yields of denitrifiers compared to respiratory ammonifiers [39], if a NapA

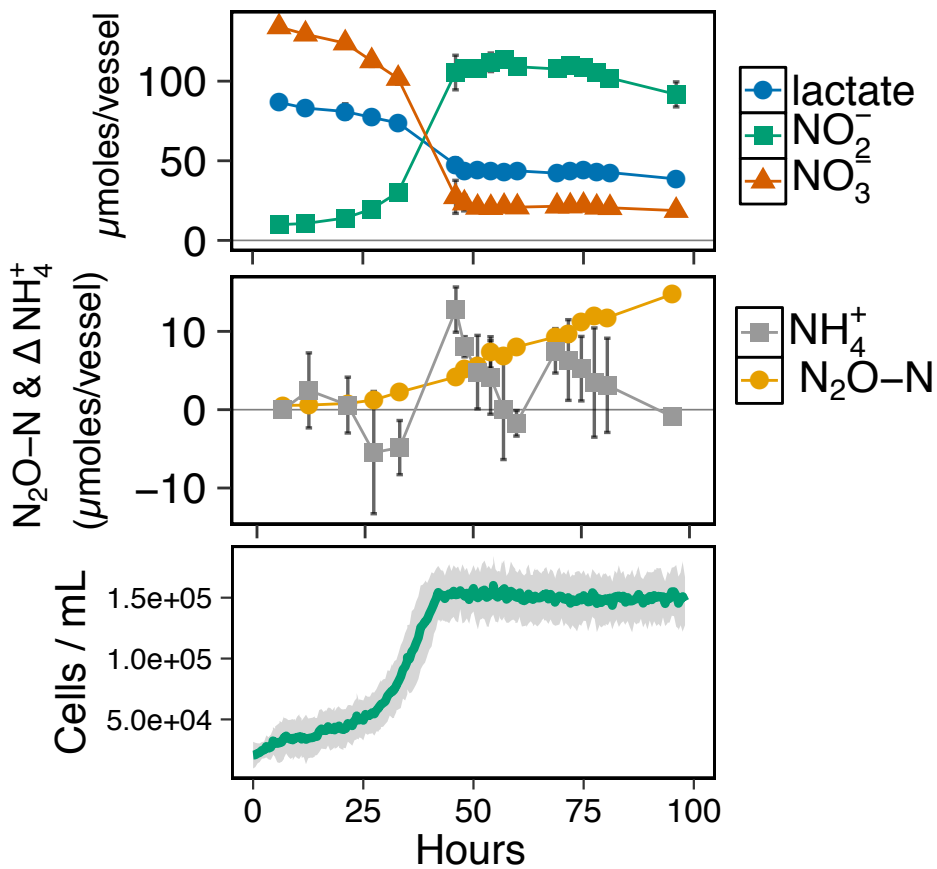
796 module is used, it would make sense for respiratory ammonification to be selected under high
797 C:N ratios because the cell would need to compensate for less energy conservation during nitrate
798 reduction. Nitric oxide reductase composition may also impact pathway selection. For example,
799 qNor does not translocate H⁺, while sNor, eNor, and gNor are predicted to conserve energy
800 through H⁺ translocation [41]. Thus, the modularity of dissimilatory N reduction processes, and
801 whether those modules conserve energy or not, may impose certain constraints on pathway
802 selection in different organisms and should be further investigated.

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Figure S1. Relationship between growth rate and the fraction of N dissimilated by respiratory ammonification for high and low nutrient concentrations. Treatments under C and NO_3^- scarcity, even with low $C:NO_3^-$ ratios, disproportionately produce more ammonium and have higher growth rates.



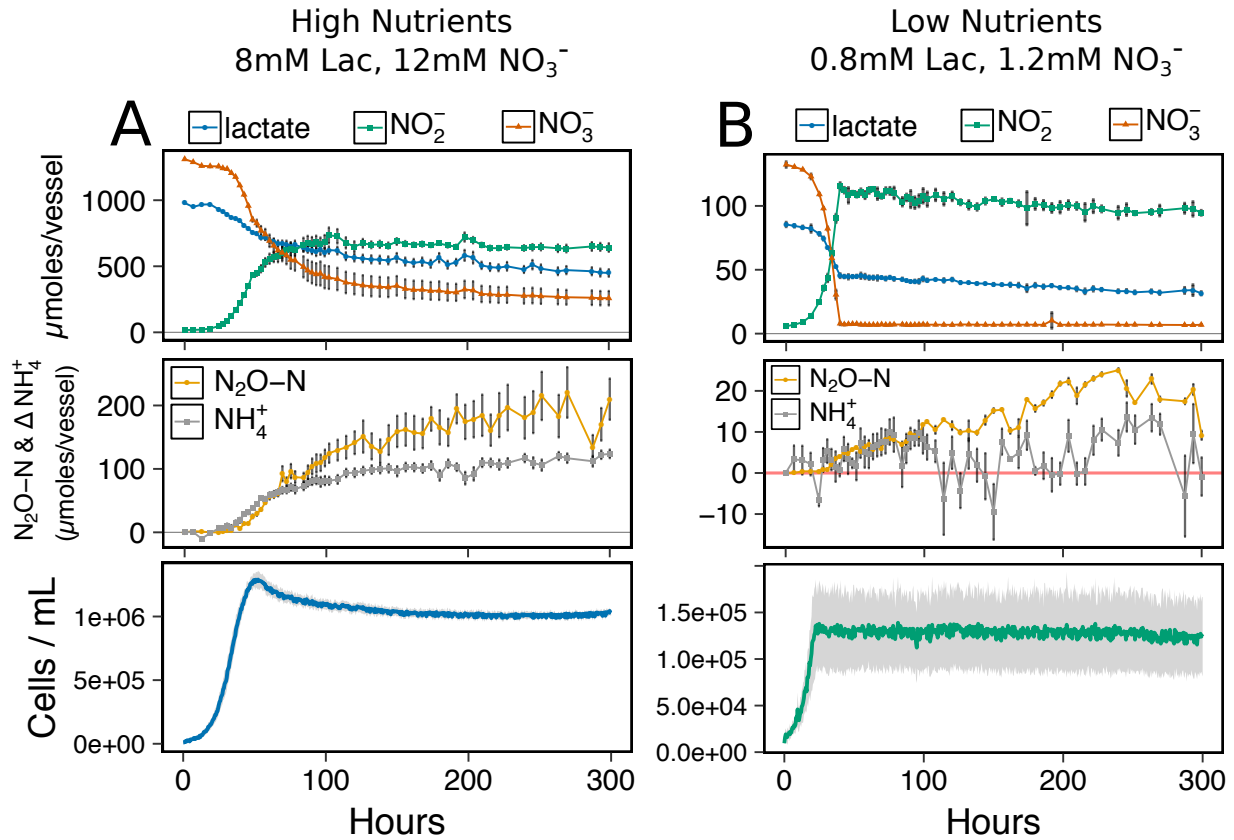
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817 Figure S2. Time-series metabolite profiles of a 96-hour incubation for lactate, nitrate, and nitrite

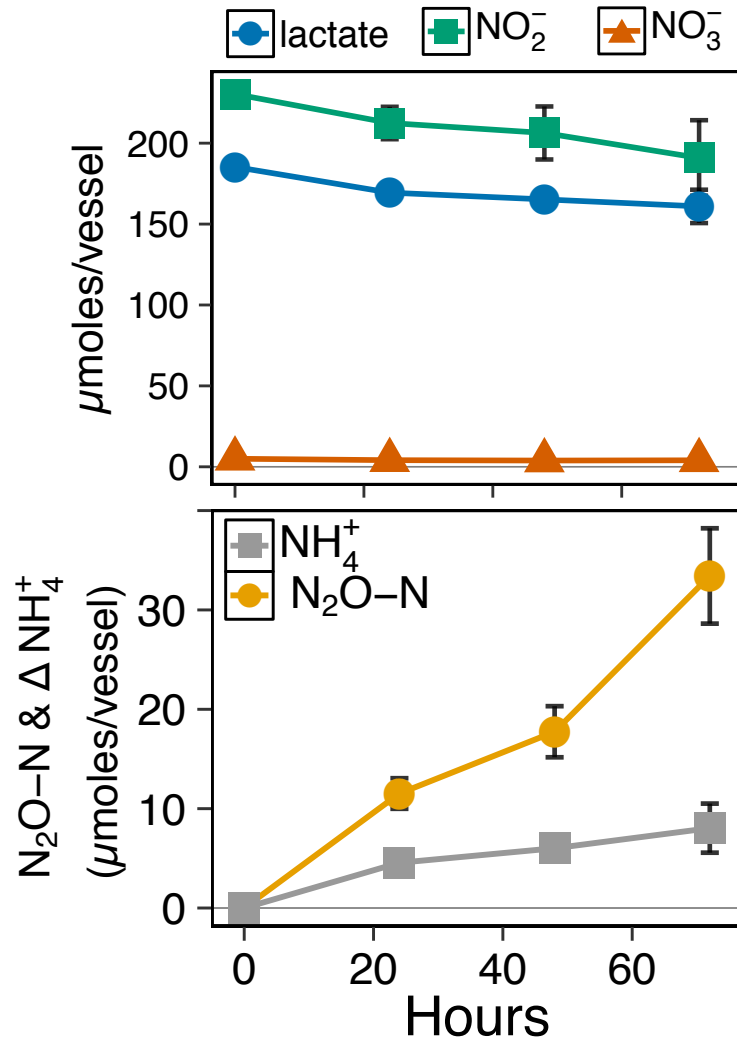
818 ($\text{N}_2\text{O-N}$ and ΔNH_4^+ production (middle pane), and corresponding growth curve of *I. calvum* cells grown

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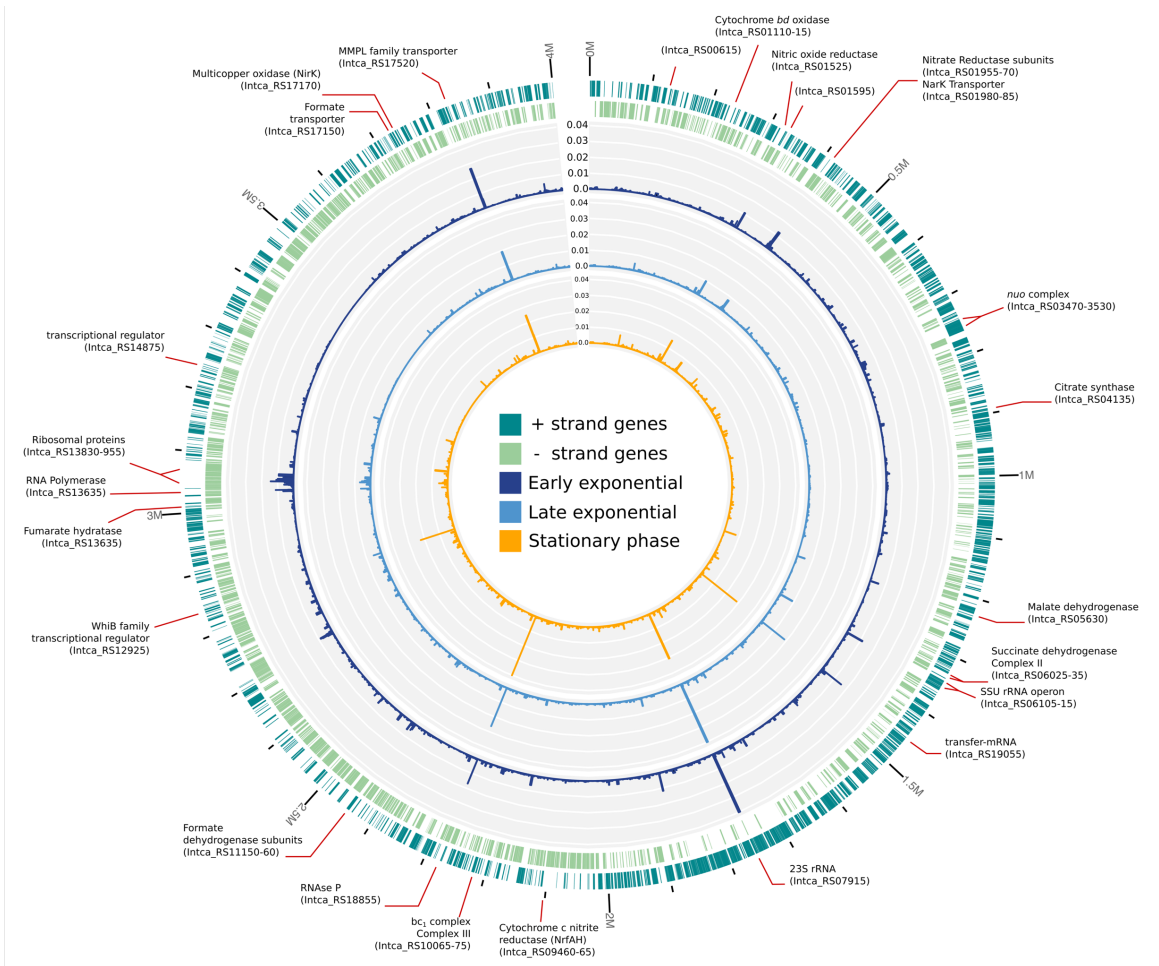
821 under 0.8mM lactate 1.2mM nitrate (C: NO_3^- ratio = 2) (bottom pane).



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 824 Figure S3. Time-series metabolite profiles of a 300-hour incubation for (A) high nutrient and (B)
 825 low nutrient concentrations. Shown are the profiles of lactate, nitrate, and nitrite (top pane),
 826 production of dissimilated end-products as $\text{N}_2\text{O-N}$ and net change in NH_4^+ ammonium
 827 production (middle pane), and corresponding growth curves of *I. calvum* cells ($\text{C}:\text{NO}_3^-$ ratio = 2)
 828 (bottom pane).



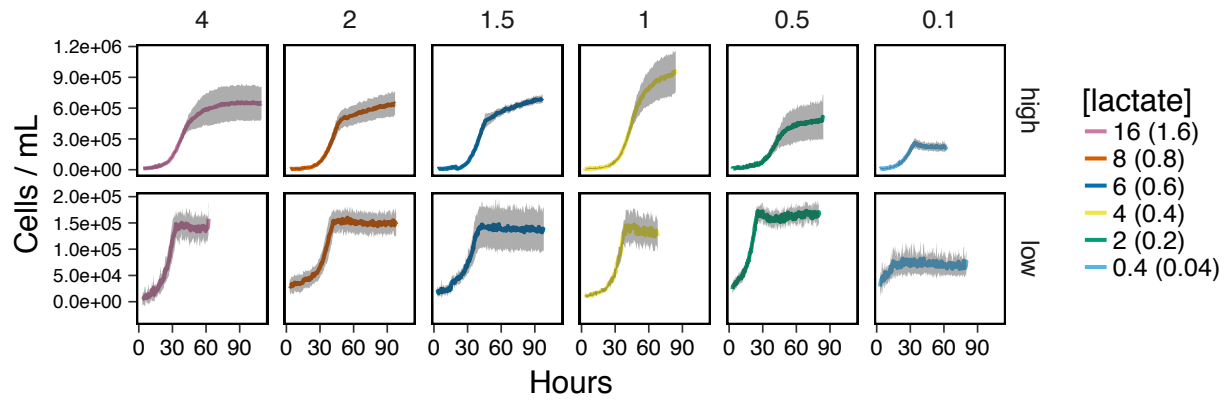
830
 831 Figure S4. Time-series metabolite profiles of a 72-hour incubation conducted in balch-tubes
 832 grown under 8mM lactate 12mM nitrite (C:NO₂⁻ ratio = 2). Profiles for lactate and nitrite (top
 833 pane) and production of dissimilated end-products as N₂O-N and net change in NH₄⁺ ammonium
 834 production (bottom pane).
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 837 Figure S5. The genome-wide transcriptional changes of early exponential, late exponential, and
 838 stationary phase *I. calvum* cells. The first and second outermost rings (dark and light green
 839 indicate the open reading frames (ORFs) on the positive and negative strands. The third, fourth,
 840 and fifth rings are the relative abundance of transcripts mapped onto the *I. calvum* genome based
 841 on the transcript read counts from early exponential phase, late exponential phase and stationary
 842 phase, respectively. The position and locus IDs are marked for the most highly expressed genes
 843 and genes involved in the ETC.

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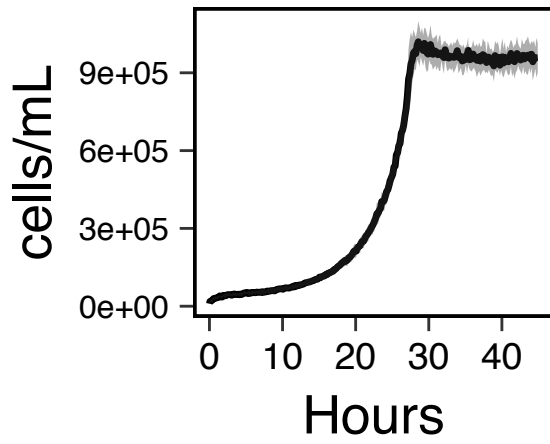
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Figure S6. Mean cell concentrations for *I. calvum* cultures grown over a range of C:NO₃⁻ ratios (columns) at high nutrient (top row) and low nutrient (bottom row) concentrations of the same ratio. Each growth curve consists of n=6 replicates.

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Figure S7. Growth curve of *I. calvum* in a sealed Balch-tube with lactate and O₂ as electron donor/acceptor pair and with ammonium as sole nitrogen source.

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Table S1. Organism accession numbers for NrfA and NirK modules.

Organisms	Accession #
	NrfA
<i>Escherichia coli</i> K-12	NC_000913.3
<i>Salmonella enterica</i> CT18	NC_003198.1
<i>Yersinia kristensenii</i>	NZ_CP009997.1
<i>Yersinia frederiksenii</i>	NZ_CP009364.1
<i>Vibrio fischeri</i> ES114	NC_006840.2
<i>S. loihica</i> -PV-4	NC_009092.1
<i>Shewanella oneidensis</i> MR-1	NC_004347.2
<i>Desulfotalea psychrophila</i> LSV54	NC_006138.1
<i>Sulfurospirillum deleyianum</i>	NC_013512.1
<i>Wolinella succinogenes</i>	NC_005090.1
<i>Flexibacter tractuosus</i>	NC_014759.1
<i>Porphyromonas gingivalis</i> W83	NC_010729.1
<i>Symbiobacterium thermophilum</i>	NC_006177.1
<i>Carboxydotherrmus hydrogenoformans</i>	NC_007503.1
<i>Desulfovibrio vulgaris</i> Hildenborough	NC_002937.3
<i>Bacillus vireti</i>	NZ_LDNB01000003.1
<i>Bacillus bataviensis</i>	NZ_AJLS01000002.1
<i>Bacillus azotoformans</i>	NZ_AJLR01000001.1
<i>Bacillus selenitireducens</i> MLS10	NC_014219.1
<i>Campylobacter jejuni</i>	NC_002163.1
<i>Opitutus terrae</i>	NC_010571.1
<i>Anaeromyxobacter dehalogenans</i> 2_CP-1	NC_011891.1
<i>Rhodopirellula baltica</i>	NC_005027.1
<i>Intrasporangium calvum</i> 7KIP	NC_014830.1
<i>Intrasporangium calvum</i> C5	This study
<i>Bdellovibrio bacteriovorus</i>	NC_005363.1
<i>Gimesia maris</i>	NZ_ABCE01000001.1
<i>Candidatus Nitrospira inopinata</i>	NZ_LN885086.1
<i>Myxococcus xanthus</i>	NC_008095.1
<i>Geobacter metallireducens</i> GS_15	NC_007517.1
<i>Geobacter sulfurreducens</i> PCA	NC_002939.5
<i>Thioalkalivibrio nitratireducens</i>	NC_019902.2
<i>Thermodesulfovibrio yellowstonii</i> THEYE_A0193	NC_011296.1
	NirK
multicopper oxidase [Intrasporangium calvum]	WP_013494195.1
nitrite reductase, copper-containing [Shewanella loihica]	WP_011867131.1
nitrite reductase [Candidatus Nitrospira inopinata]	WP_062488124.1
nitrite reductase, copper-containing [Marivirga tractuosa]	WP_013454821.1
nitrite reductase, copper-containing [Symbiobacterium thermophilum]	WP_070105442.1
nitrite reductase [Opitutus terrae]	WP_012373845.1

nitrite_reductase,_copper-containing_[Bdellovibrio_bacteriovorus]	WP_011165004.1
Nitrite_reductase_OS=Bacillus_azotoformans_GN=nirK	ZP_08007035.1
Ochrobactrum_anthropi_ATCC_49188	NC_009668.1
Bradyrhizobium_japonicum_USDA_110	NC_004463.1
Agrobacterium_fabrum_str._C58	NC_003063.2
Sinorhizobium_meliloti_1021	NC_003037.1
Pseudomonas_citronellolis_strain_SJTE-3	NZ_CP015878.1
Rhodanobacter_denitrificans_strain_2APBS1	NC_020541.1
Taylorella_equigenitalis_ATCC_35865	NC_018108.1
Flavobacterium_columnare_ATCC_49512	NC_016510.2
Actinobacillus_suis_ATCC_33415	NZ_CP009159.1
Chromobacterium_violaceum_ATCC_12472	NC_005085.1
Halopiger_xanaduensis_SH-6	NC_015666.1
Halopiger_xanaduensis_SH-6	NC_015666.1
inorhizobium_fredii_HH103	NC_016812.1
Pseudomonas_entomophila_str._L48	NC_008027.1
Pseudomonas_denitrificans_ATCC_13867	NC_020829.1
Flavobacterium_johnsoniae_UW101	NC_009441.1
Rhizobium_etli_CFN_42	NC_007766.1
Ochrobactrum_anthropi_ATCC_49188	NC_009667.1
Caulobacter_segnis_ATCC_21756	NC_014100.1
Rhizobium_giardinii_bv._giardinii_H152	NZ_KB902685.1

Table S2. Concentration and ratio experimental design and production values for NH_4^+ and $\text{N}_2\text{O-N}$.

[C] (mM)	[NO ₃] (mM)	C:NO ₃ - ratio	Ammonia per Cell		Ammonia per Lactate		% Recovery of Dissimilated N
			NH ₄ produced (μmoles)	NH ₄ produced (μmoles)	N ₂ O produced (μmoles)	N ₂ O produced (μmoles)	
16	12	4	1.94 ± 1.31	7.79 ± 3.3	27.4 ± 7.5	91.64 ± 12.9	
8	12	2	4.91 ± 1.07	10.8 ± 4.1	18.1 ± 6.7	72.87 ± 9.3	
6	12	1.5	3.07 ± 4.50	10.2 ± 3.8	19.1 ± 6.2	72.77 ± 4.1	
4	12	1	8.06 ± 2.19	14.5 ± 4.2	18.1 ± 6.8	61.79 ± 5.1	
2	12	0.5	3.82 ± 1.92	8.76 ± 2.9	10.2 ± 3.7	64.21 ± 9.8	
0.4	12	0.1	2.05 ± 0.50	1.55 ± 0.2	0.48 ± 0.1	24.44 ± 7.5	
1.6	1.2	4	1.12 ± 0.99	2.39 ± 0.7	1.77 ± 0.2	70.47 ± 10.4	
0.8	1.2	2	1.50 ± 0.57	2.32 ± 0.5	3.72 ± 0.4	90.71 ± 8.2	
0.6	1.2	1.5	0.90 ± 0.53	2.27 ± 0.5	4.53 ± 0.7	88.31 ± 8.6	
0.4	1.2	1	1.18 ± 1.17	2.45 ± 0.3	0.88 ± 0.3	50.20 ± 9.4	
0.2	1.2	0.5	1.91 ± 0.33	1.13 ± 0.2	0.18 ± 0.0	43.34 ± 20.0	
0.04	1.2	0.1	0.03 ± 0.95	0.28 ± 0.2	0.06 ± 0.0	18.10 ± 10.9	

Table S4. Literature summary of C:N ratio controls on N dissimilation.

Citation	C-source	C:N range	C conc range	NO3 conc range	units	calc method
Kraft <i>et al.</i> 2014 (17)	amino acids	1.5-3	4.4-43.5	0.5-14.4	mmol	mmol-C/mmol- \sum NO _x
Yoon <i>et al.</i> 2015 (3)	lactate	1.5-150	0.1-10	0.2	mM	nC*mM-C/nN*mM-N
Van den Berg <i>et al.</i> 2015 (4)	acetate	1.8-7.7	160-595	82-93	mg/L	mg-COD/mg-N
Schmidt <i>et al.</i> 2011 (5)	Soil organic-C	not specified	2.7-11.4	22.4-79.8	C%,mg-N/kg soil	not specified
Hardison <i>et al.</i> 2015 (6)	complex	not specified	C+ - C-	0.6-5	μg	not specified
Fazzolari <i>et al.</i> 1998 (7)	glucose	2.5-10	250-1000	100	mg/kg dried soil	mg-C/mg-N
This study	lactate	0.1-4	0.004-16	1.2-12	mM	nC*mM-C/nN*mM-N

Table S5. Summary of all experimental conditions and replicate number in the current study (Figure 2 in main text).

NO ₃ ⁻ (mM)	Lactate (mM)	Ratio C:NO ₃ ⁻	experiment type	ammonium-deplete	replicates	Number of samples taken
1.2	0.04	0.1	end-point	n	9	2
1.2	0.2	0.5	end-point	n	9	2
1.2	0.4	1.0	end-point	n	10	2
1.2	0.6	1.5	end-point	n	10	2
1.2	0.8	2.0	end-point	n	9	2
1.2	1.6	4.0	end-point	n	10	2
12	0.4	0.1	end-point	n	10	2
12	2	0.5	end-point	n	8	2
12	4	1.0	end-point	n	8	2
12	6	1.5	end-point	n	10	2
12	8	2.0	end-point	n	10	2
12	16	4.0	end-point	n	8	2
12	8	2.0	time-series	n	3	17
1.2	0.8	2.0	time-series	n	3	17
12	8	2.0	time-series	n	3	59
1.2	0.8	2.0	time-series	n	3	59
12*	8	2.0	time-series	n	11	4
12	8	2.0	time-series	y	10	3

*nitrite is used as the electron acceptor