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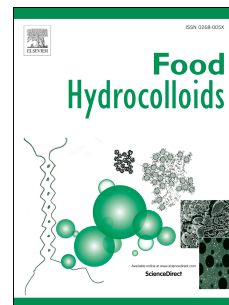
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Improving the *in vitro* digestibility of rice starch by thermomechanically assisted complexation with guar gum

Hai He, Chengdeng Chi, Fengwei Xie, Xiaoxi Li, Yi Liang, Ling Chen



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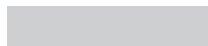
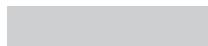
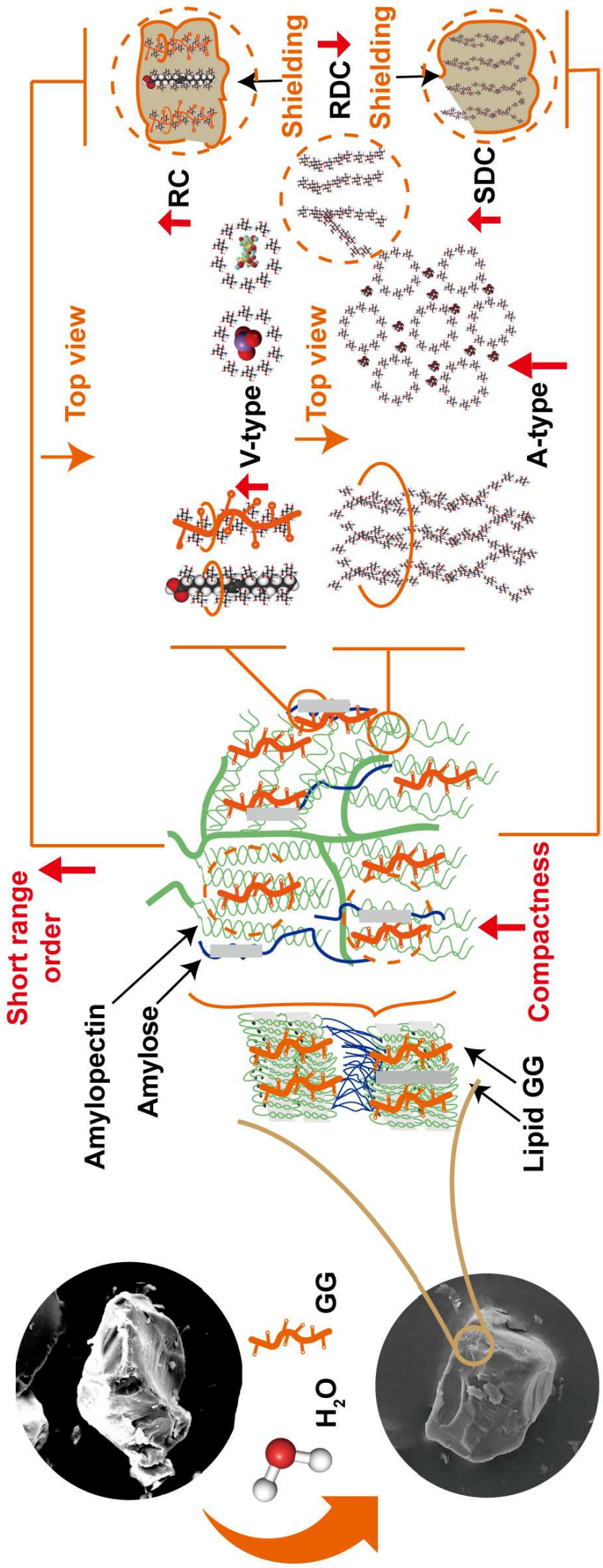
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Journal Pre-proof



1 **Improving the *in vitro* digestibility of rice starch by thermomechanically**
2 **assisted complexation with guar gum**

3 Hai He^a, Chengdeng Chi^a, Fengwei Xie^{b, c, †}, Xiaoxi Li^a, Yi Liang^d, Ling Chen^{a, *}

4 ^a Ministry of Education Engineering Research Center of Starch & Protein Processing, Guangdong
5 Province Key Laboratory for Green Processing of Natural Products and Product Safety, School of
6 Food Science and Engineering, South China University of Technology, Guangzhou, Guangdong,
7 510640, China.

8 ^b International Institute for Nanocomposites Manufacturing (IINM), WMG, University of Warwick,
9 Coventry CV4 7AL, United Kingdom.

10 ^c School of Chemical Engineering, The University of Queensland, Brisbane, Qld 4072, Australia.

11 ^d Guangdong Zhongqing Font Biochemical Science and Technology Co., Ltd., Maoming,
12 Guangdong 525427, China.

13 *Corresponding author. E-mail address: felchen@scut.edu.cn (L. Chen)

14 †Corresponding author. E-mail address: d.xie.2@warwick.ac.uk, f.xie@uq.edu.au (F. Xie)

15

16 **ABSTRACT**

17 The effects of thermomechanical treatment and guar gum (GG) addition on the *in vitro* digestibility
18 of rice starch have been investigated. Rice starch added with GG at concentrations of 0, 0.025, 0.05,
19 0.075, or 0.10 g/100 g (wet basis) was subject to a micro-extrusion process. The *in vitro* digestibility,
20 predicted glycemic index (pGI), and multi-scale structures (granule, lamellar, crystalline, and
21 molecular structures) were examined. Micro-extruded rice starch (MERS) with GG presented
22 reduced digestion rate and pGI, a higher degree of structural ordering, and altered crystalline,
23 single-helical and double-helical structures. Using Pearson correlation analysis, the relationships
24 among extrusion, the molecular interaction and multi-scale structure, and the digestibility were
25 established. The content of resistant components (RC) was positively correlated with crystallinity (r
26 = 0.836, $p < 0.05$), fractal dimension ($r = 0.966$, $p < 0.05$), A-type crystallinity ($r = 0.954$, $p < 0.01$),
27 V-type crystallinity ($r = 0.987$, $p < 0.05$), $R_{1047/1022}$ ($r = 0.987$, $p < 0.05$), single-helix content ($r =$
28 0.897, $p < 0.05$), and double-helix content ($r = 0.991$, $p < 0.01$); and was negatively correlated with
29 pGI ($r = -0.947$, $p < 0.05$). Overall, this study showed that thermomechanical treatment assisted the
30 complexation of GG with starch, which could be an effective means to improve the resistant
31 properties of rice starch.

32 **Keywords:** Rice starch; Extrusion; Guar gum; Molecular interaction; Digestive properties

33

34 **1 Introduction**

35 With the social development and the evolution of the human diet and disease spectrum, there has
36 been an increasing focus on dietary nutrition and health (Crawford et al., 2014; Fang, Li, & Liu,
37 2016). Rice is the main source of carbohydrate for people, especially in Asia. It can be easily
38 digested in the human digestive tract, providing the body with energy to prevent the occurrence of
39 symptoms such as hunger and fatigue (Guerrant, Dutcher, & Brown, 1937). However, rice has a high
40 glycemic index (GI), resulting in a high blood glucose response level after consumption, which leads
41 to unstable blood sugar levels. This can cause insulin resistance and induce type II diabetes mellitus
42 and related complications, thus, worsening human health (Miller, Pang, & Bramall, 1992; Roberts,
43 2000). Starch is an important component of rice, accounting for more than 80% of the total
44 composition of rice. It is crucial to develop rice products with lower postprandial blood glucose
45 response levels (Thompson, Winham, & Hutchins, 2012). Therefore, improving the resistant
46 properties of rice starch, suppressing the postprandial blood glucose response to rice, and regulating
47 the nutritional value of rice products are essential for developing rice-containing healthy foods and
48 adjusting consumers' dietary structure.

49 Guar gum (GG) is another important food component. The molecular structure of GG consists of
50 a D-mannose unit connected by a β -1,4-glycosidic bond to form its main chain, with D-galactose as
51 side chains connected to the main chain by α -1,6 glycoside bond. It has a galactose/mannose ratio of
52 approximately 1:2, with an average relative molecular mass of 200,000–300,000 (Mudgil, Barak, &
53 Khatkar, 2014). GG can affect the functionality and nutritional characteristics of starch-based food
54 products by non-covalent actions. Moreover, GG can be used as a dietary fiber to promote human

55 health (Chung, Liu, & Lim, 2007). As a dietary supplement, GG swells and then intertwines with
56 other food components to form a network structure. This slows down the process of gastric emptying
57 and subsides hunger and eating desire, which is particularly suitable for those who are obese or on
58 diet (Butt, Shahzadi, Sharif, & Nasir, 2007). In the human colon, GG can be fermented by resident
59 bacteria, producing short-chain fatty acids that protect the digestive tract and prevent bowel cancer
60 (Shahzadi, Butt, Sharif, & Nasir, 2007). Bordoloi et al. (Bordoloi, Singh, & Kaur, 2012) reported that
61 when potato starch was added with pectin, cellulose or GG, the digestion rate of the starch was
62 significantly reduced owing to the physical embedding of GG. Jang et al. (Jang, Bae, & Lee, 2015)
63 suggested that GG could induce the structural recombination of starch and significantly reduce the
64 sugar index of noodles made from wheat flour or whole wheat flour.

65 The extrusion of starch can be considered as a 'green' processing technology where starch is
66 plasticized with water under high temperature, shear, and pressure. During the extrusion process,
67 starch is re-linked with a strengthened intermolecular hydrogen-bonded network, finally resulting in
68 kinetic and thermodynamic stabilities (Gomez & Aguilera, 2010). Extrusion can modify the starch
69 granule morphology, semicrystalline lamellar thickness, crystalline structure, molecular mass and its
70 distribution, and linear/branched-chain starch contents (Lai & Kokini, 1991; Ye, Hu, Luo, Wei, &
71 Liu, 2017). Besides, starch chain aggregates on different scales can be formed and the starch
72 digestibility can be regulated (Fan et al., 2018).

73 The combination of extrusion treatment and GG addition is likely to produce rice starch with
74 increased enzymatic resistance, which is conducive to human health. However, limited research can
75 be found regarding the effect of extrusion on the GG complexation with starch. GG molecules may

76 induce chain rearrangement and aggregation through non-covalent interactions, such as hydrogen
77 bonding and van der Waals forces, thus, changing the susceptibility of starch molecules to enzymes
78 (Zheng et al., 2019). This current study aimed at understanding the combined effects of GG addition
79 and extrusion treatment, for which the digestibility of micro-extruded rice starch (MERS) complexed
80 with GG was analyzed using scanning electron microscopy (SEM), Fourier-transform infrared
81 spectroscopy (FTIR), X-ray diffraction (XRD), small-angle X-ray scattering (SAXS), and ^{13}C
82 nuclear magnetic resonance (NMR). Based on these analyses, we explored how the altered
83 multi-scale structure of rice starch influence its digestibility. A relevant model was established and
84 the mechanism for these effects was elucidated.

85 **2 Materials and methods**

86 **2.1 Materials**

87 Rice starch (GABIOSTA-F) was supplied by Jiangxi Jinnong Biotechnology Co., Ltd. (Wuxi,
88 China), which contains 11.55% (d.b) moisture, 0.20% (d.b) lipid, 0.65% (d.b) protein, and 0.38%
89 (d.b) ash. Food-grade guar gum (GG) was provided by Guangzhou Feibo Biotechnology Co., Ltd.
90 (China). Pancreatic α -amylase (A3306) was purchased from Sigma (USA). A glucose oxidation kit
91 (GOPOD) was bought from Megazyme (Ireland). All other chemical reagents used in this study were
92 of analytical grade.

93 **2.2 Preparation of micro-extruded rice starch (MERS) with guar gum (GG)**

94 Rice starch (20 g, dry weight) was mixed with GG (2.5%, 5%, 7.5%, 10%, w/w, based on rice
95 starch) in a flask. The mixture of rice starch and GG was adjusted to 40% moisture content, which

96 was then extruded in a Haake MiniLab II micro-compounder (Thermo Fisher Scientific, USA) at 150
97 rpm and 85 °C. The material was directly extruded without circulation in the micro-compounder. The
98 residence time was about 5 min. The extrudates were dried at 40 °C for 24 h, and ground with a
99 laboratory-scale grinder to pass through a 100-mesh sieve. According to the different amounts of GG
100 added, micro-extruded rice starch (MERS) samples with GG were named in the form of
101 “MERS/GG-X”, in which “GG” represented guar gum, and “X” represented the content of added
102 GG (%). The MERS sample without GG was named MERS/GG-0. These samples were compared
103 with native rice starch (NRS).

104 **2.3 *In vitro* starch digestibility**

105 *In vitro* digestibility was measured based on the Englyst method (Englyst & Cummings, 1985)
106 with slight modifications (Wang, Wang, Li, Chen, & Zhang, 2017). Considering the chemical
107 structure of GG, non-digestive components may include not only resistant starch but also GG.
108 Therefore, the digestive properties of MERS samples with GG were measured by the proportions of
109 rapidly digestible components (RDC), which were hydrolyzed by incubation within 20 min, of
110 resistant components (RC), which were remain unhydrolyzed after 120 min of incubation, and of
111 slowly digestible components (SDC), which were calculated as the difference between RDC and RC.

112 **2.4 Predicted glycemic index (pGI)**

113 The hydrolysis index (HI) was calculated on the basis of the starch hydrolysis curve (0–180 min)
114 as the percentage of total glucose released over 180 min in comparison with that released from white
115 bread over the same duration (Goñi, Garciaalonso, & Sauracalixto, 1997; Jenkins et al., 1981). The

116 glycemic index (GI) was then estimated by using the equation used by Goni et al. (Goñi et al., 1997)
117 with slight modifications ($GI = 44.78 + 0.3797 \times HI$).

118 2.5 First-order kinetic analysis of *in vitro* digestibility

119 The logarithm of the slope (LoS) graphical method was used to study the *in vitro* digestibility of
120 MERS samples and to predict their physiological response (Butterworth, Frederick, Terri, Hamung,
121 & Peter, 2012; Dhital, Warren, Butterworth, Ellis, & Gidley, 2017; Xu, Kuang, Wang, Zhou, & Wang,
122 2017). The digestibility was fitted to the first-order kinetic as shown in Eq. (1):

$$123 C_t = C_\infty (1 - e^{-kt}) \quad (1)$$

124 where C_t is the digested components ratio of the samples, C_∞ is the digested components ratio at the
125 end of the reaction, and k is the first-order rate constant.

126 By expressing the first derivative of the first-order equation in a logarithmic form, we obtained a
127 logarithmic slope (LoS) plot:

$$128 \ln\left(\frac{dC}{dt}\right) = -kt + \ln(C_\infty k) \quad (2)$$

129 where $\ln(dC/dt)$ represents the logarithm of the slope, k and C_∞ were used to construct a
130 model-fitted starch digestion curve according to Eq. (1).

131 2.6 GPC-MALS analysis

132 The weight-average molecular mass (M_w), mean square radius of gyration, and molecular mass
133 (M) distribution of the samples were analysed using a gel permeation chromatography (GPC) system
134 coupled with a multi-angle light scattering detector and a refractive index (RI) detector, as described
135 in our previous studies (Liu, Chen, Xu, Liang, & Zheng, 2019).

136 **2.7 Determination of amylose content**

137 The amylose content of starch was determined using the iodine reagent method (Lim et al.,
138 2015). 100 mg of starch was accurately weighed and dispersed in a 1M NaOH solution, which was
139 then diluted with distilled water to form a 1 mg/mL solution. This solution was then mixed with
140 iodine-KI reagent, and measured at 620 nm using an Evolution 201 UV-Visible Spectrophotometer
141 (Thermo Scientific Inc., Waltham, USA). The amylose content was determined using a standard
142 curve established by a mixed solution of amylose and amylopectin.

143 **2.8 Scanning electron microscopy (SEM)**

144 MERS samples were spread onto circular metal stubs covered with double-sided adhesive
145 carbon tape and then sputter-coated with gold. Images of starch granules were obtained using an
146 EM-30 Plus scanning electron microscope (SEM, COXEM, South Korea) under an accelerating
147 voltage of 20 kV at 500× magnification.

148 **2.9 Small-angle X-ray scattering (SAXS)**

149 SAXS measurements were performed on a SAXSess small-angle X-ray scattering (SAXS)
150 system (Anton-Paar, Austria) according to our previous methods (Zhang, Chen, Li, Li, & Zhang,
151 2015; Zhang, Chen, Zhao, & Li, 2013; Zhang, Li, Liu, Xie, & Chen, 2013). Samples were measured
152 at 50 mA and 40 kV using a PW3830 X-ray generator (PANalytical) equipped with an X-ray source
153 of Cu K α radiation ($\lambda = 0.1542$ nm). Each sample with a moisture content of about 60% was
154 prepared and equilibrated at room temperature for 24 h, before being placed into a paste cell and
155 measured for 5 min under X-ray. The data, recorded in an image plate, were collected using the IP
156 Reader software using a PerkinElmer Storage Phosphor System. All collected data were normalized,

157 and the background intensity and smeared intensity were subtracted by using SAXSquant 2D
158 software and SAXSquant 3.0 software, respectively.

159 **2.10 X-ray diffraction (XRD)**

160 XRD analysis was performed using an Xpert PRO diffractometer (PANalytical B.V.,
161 Netherlands), operated at 40 mA and 40 kV with an X-ray source of Cu-K α radiation ($\lambda = 0.15424$
162 nm). The range of the diffraction angle (2θ) was from 5° to 40° with a scanning speed of $10^\circ/\text{min}$
163 and a scanning step of 0.033° . The characteristic diffraction peaks of the A-type crystal are mainly
164 9.9° , 11.2° , 15° , 17° , 18° , and 23.5° , and those of the V-type crystal were mainly 7.4° , 13.1° , and
165 20.1° (Singh, Dartois, & Kaur, 2010).

166 The MDI Jade 6.0 software was used to compute the crystallinity between 4° and $30^\circ 2\theta$,
167 according to our previous studies (Xu, Tan, Chen, Li, & Xie, 2019). The integrated areas of all
168 crystalline peaks halo on the X-ray diffractogram was the crystallization area (A_c), and the integrated
169 areas of the amorphous halo on the X-ray diffractogram was the amorphous region (A_a). The
170 equations used for calculating A-type or V-type crystallinity are: X_A (%) or X_V (%) = $100 \times A_c / (A_c +$
171 $A_a)$; and the total crystallinity is calculated using X_{Total} (%) = X_A (%) + X_V (%).

172 **2.11 Fourier-transform infrared (FTIR)**

173 FTIR spectra with a range of 4000 to 400 cm^{-1} were obtained on a Tensor 37 spectrometer
174 (Bruker, Germany) with a DTGS (deuterated triglycine sulfate) detector using an attenuated total
175 reflectance accessory. For each spectrum, 64 scans with air as the background were obtained at a
176 resolution of 4 cm^{-1} .

177 **2.12 ¹³C Cross-polarized magic angle spinning nuclear magnetic resonance (CP/MAS NMR)**
178 **spectroscopy**

179 Solid-state ¹³C CP/MAS NMR analysis was performed on a Bruker AVANCE III HD 400
180 spectrometer (Bruker, Germany) equipped with a 4-mm broadband double-resonance MAS probe.
181 Approximately 500 mg of the sample was placed into the spinner and inserted into the center of the
182 magnetic field. The NMR spectrum with CP and MAS was recorded at 100.613 MHz at a
183 temperature of 295 K. Over 6000 scans were accumulated for each spectrum with a recycle delay of
184 2 s (Mihhalevski et al., 2012; Tan, Flanagan, Halley, Whittaker, & Gidley, 2007).

185 **2.13 Statistical analysis**

186 The data were statistically analyzed using the SPSS 22.0 statistical package and presented as the
187 mean ± standard deviation (SD). The differences between the groups were estimated using an
188 analysis of variance (ANOVA), and $p < 0.05$ was considered to indicate a statistically significant
189 difference between the two groups. Pearson correlation analysis was also conducted to determine the
190 relationships between GG content, the structures and digestibility.

191 **3 Results**

192 **3.1 *In vitro* digestibility**

193 Table 1 shows the digestibility of NRS and MERS samples. Compared with those of NRS, the
194 SDC and RC contents of MERS samples remarkably increased. The migration of water molecules to
195 the interior of the starch granules driven by the thermomechanical energy during extrusion could
196 destroy the hydrogen bonding between the starch molecules and induce the reorganization of starch

197 molecules to form a more thermodynamically stable structure, leading to a significantly RDC content
 198 (Wang et al., 2018). The addition of GG apparently reduced the digestibility of MERS. With the
 199 addition of GG, the RDC content of MERS samples was decreased by 26–41.8% and the RC content
 200 was increased by 27.1–41.5% gradually. The effect of extrusion on the SDC content was less
 201 significant, with an increase from 9.2% to 20.6%.

202

203

Table 1. RDC, SDC and RC contents and pGI of NRS and MERS samples *

Samples	RDC content (%)	SDC content (%)	RC content (%)	pGI
NRS	95.0±0.2 ^e	3.4±1.1 ^a	1.6±0.9 ^a	–
MERS/GG-0	75.2±0.6 ^d	10.3±0.6 ^b	14.5±0.5 ^a	82.3±1.5 ^d
MERS/GG-2.5%	49.2±1.0 ^c	9.2±0.9 ^a	41.6±1.2 ^b	64.3±2.2 ^c
MERS/GG-5%	42.8±0.7 ^c	11.2±0.5 ^c	46.0±0.2 ^c	63.2±0.9 ^b
MERS/GG-7.5%	33.2±1.5 ^b	16.4±1.4 ^d	50.4±1.7 ^d	63.0±1.3 ^b
MERS/GG-10%	23.4±1.1 ^a	20.6±0.2 ^e	56.0±0.9 ^e	61.8±1.2 ^a

204 * Mean value ± standard deviation of duplicate analysis is given. Values with different letters within
 205 the same column differ significantly ($p < 0.05$).

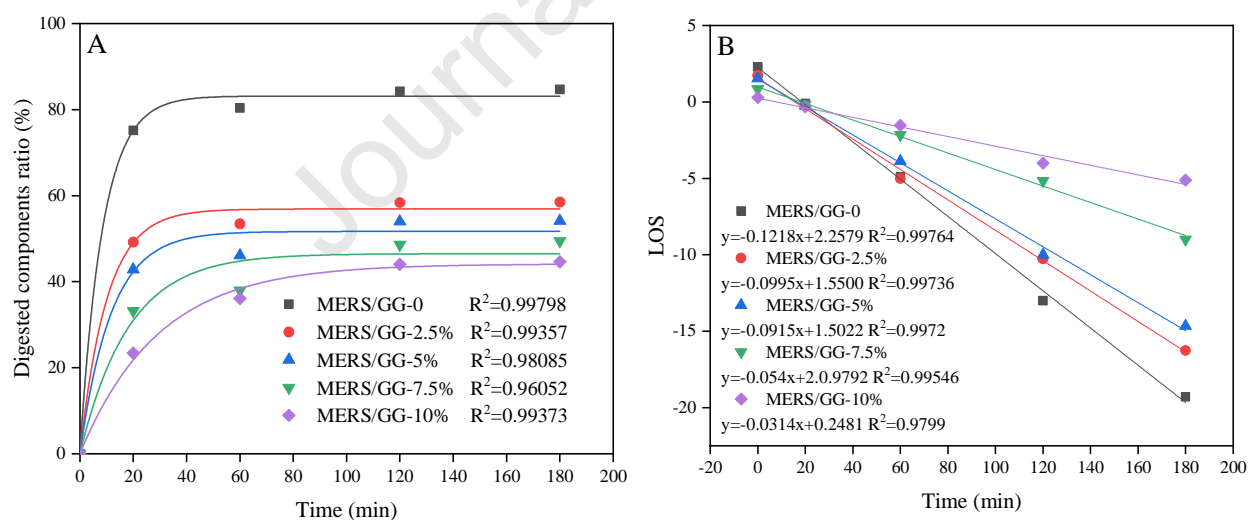
206

207 3.2 Predicted glyceic index (pGI)

208 Table 1 also shows the pGI values of NRS and MERS samples. Specifically, the interaction with
 209 GG significantly reduced the pGI of MERS. With increasing amounts of GG added, the pGI of
 210 MERS decreased gradually. Along with the results about RC and SDC contents, it could be proposed
 211 that the formation of SDC and RC reduced the pGI of MERS.

212 3.3 First-order kinetics analysis

213 Fig. 1 shows the digestibility curves and LoS plots for NRS and MERS samples. Table 2
 214 shows the k and C_{∞} values obtained from the LoS plots derived from the first-order kinetics of the
 215 digestion of MERS samples. In the first 1 h, MERS samples were digested rapidly and nearly
 216 reached a plateau at 80–120 min (Fig. 1A). In addition, for all MERS samples, the LoS plot exhibits
 217 a linear relationship with a constant k value, suggesting that the digestion of MERS was a
 218 single-phase process. The k value for MERS/GG-0 was significantly higher than that for MERS
 219 samples with GG (Table 2). C_{∞} represents the theoretical percentage of starch digested at the end of
 220 the reaction. The addition of GG obviously reduced the C_{∞} values for MERS samples. These results
 221 were consistent with the digestibility data as shown in Table 1.



223
 224 Fig. 1. Digestibility curves (A) and associated LoS plots (B) for MERS samples.

225

226 Table 2. Digestibility parameters obtained from the LoS plots for MERS samples*

Samples	k (min^{-1})	C_{∞} (%)
---------	---------------------------	---------------------

MERS/GG-0	0.122±0.003 ^d	83.169±1.104 ^c
MERS/GG-2.5%	0.099±0.003 ^c	56.898±1.353 ^b
MERS/GG-5%	0.092±0.002 ^c	51.682±2.148 ^b
MERS/GG-7.5%	0.054±0.002 ^b	46.504±2.956 ^a
MERS/GG-10%	0.031±0.002 ^a	44.124±1.283 ^a

227 *Mean value ± standard deviation of duplicate analysis is given. Values with different letters within
 228 the same column differ significantly ($p < 0.05$).

229

230 3.4 Molecular mass distribution and amylose content

231 The weight-average molecular mass (M_w), mean square radius of gyration, and molecular
 232 mass distribution of NRS and MERS/GG-0 were investigated by GPC-MALS. The parameters of the
 233 different samples are shown in Table 3. The M_w and R_g of NRS were 2.250×10^7 g/mol and 112.9 nm,
 234 respectively. After extrusion, the M_w of MERS/GG-0 decreased to 1.048×10^7 g/mol. Along with that,
 235 the R_g of MERS/GG-0 also decreased. These results show that the thermomechanical treatment could
 236 break starch chains, reducing the molecular weight and the size of chain aggregates. Table 3 also
 237 shows the cumulative weight fractions at different molecular mass distribution ranges for MERS
 238 samples. For NRS, the fractions higher than 1×10^7 g/mol accounted for 100%. After extrusion, the
 239 molecular mass distribution of MERS/GG-0 moved towards lower values.

240

241

Table 3 GPC-MALS parameters of NRS and MERS/GG-0.

Samples	Amylose content	M_w (g/mol)	R_g	Molecular mass distribution (%)			
				$<10^{6b}$	$1-5 \times 10^{6b}$	$0.5-1 \times 10^{7b}$	$>10^{7b}$
NRS	22.90%	2.250×10^7	112.9 (0.5%)	0	0	0	100%

		(1%) ^a					
MERS/GG-0	25.30%	1.048×10 ⁷	94.0 (0.5%)	0	23.6%	30.7%	45.8%
		(0.9%)					

242 a. Fitting precision. b. g/mol.

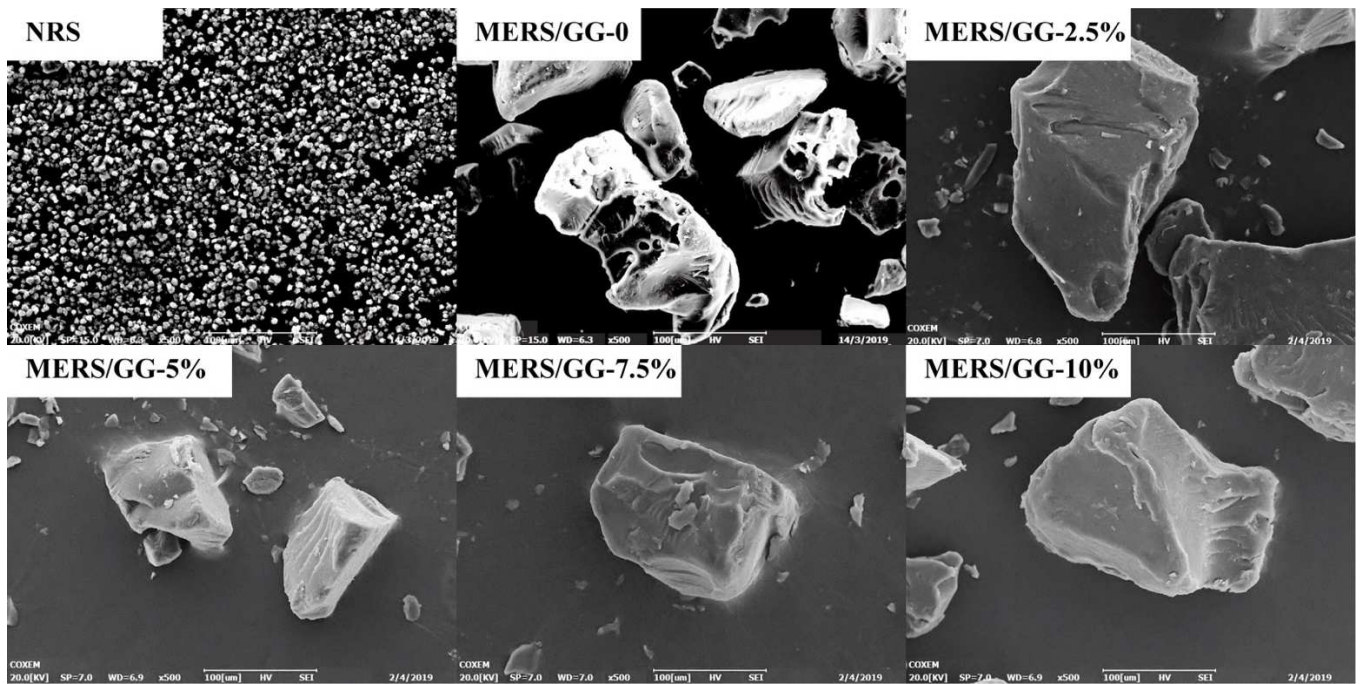
243

244 Table 3 also shows the amylose contents of NRS and MERS/GG-0. Along with the molecular
 245 mass distribution results measured by GPC-MALS, it could be found that the thermomechanical
 246 treatment caused the breakage of α -1,4 glycosidic and α -1,6 glycosidic bonds, with increasing amylose
 247 content.

248 3.5 Granule microscopy

249 Fig. 2 shows the granule morphology of NRS and MERS samples. The granules of MERS
 250 presented irregular shapes and a rough surface, with pores on the surface. This indicates that the
 251 thermomechanical treatment significantly destroyed the granule structure of NRS. In comparison, the
 252 size of MERS became more homogeneous, ranging from 80 to 150 μ m. MERS samples with GG
 253 were in the form of polygonal chunks with a smooth surface without apparent porosity, and the size
 254 of MERS with GG was approximately 120 μ m. This indicates that the addition of GG significantly
 255 destroyed the granule morphology of MERS.

256



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Fig. 2 SEM images of NRS and MERS samples (scale bar: 100 μm ; magnification: $\times 500$)

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260 3.6 Fractal structure

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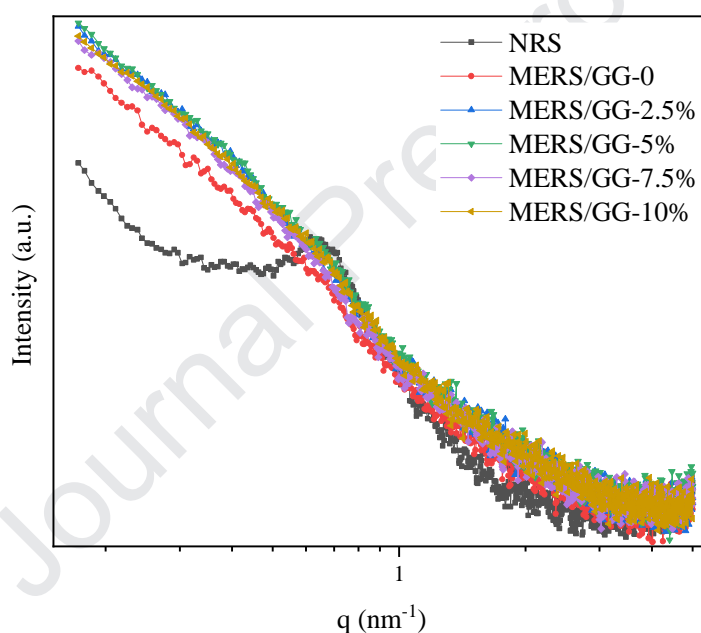
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Fig. 3 shows double-logarithmic SAXS curves for NRS and MERS samples. While NRS showed a characteristic scattering peak at q of 0.614 nm^{-1} representing its lamellar structure, this peak disappeared for all MERS samples either without or with GG, indicating the lamellar structure was completely destroyed by thermomechanical treatment. The α value (Table 4) was calculated by fitting the SAXS scattering signal to the power-law equation, $I(q) \sim q^{-\alpha}$, where I is the SAXS scattering intensity and α , obtained from the slope of the regression line of the double logarithmic SAXS curves, can be used to analyze the D characteristics of the surface/mass fractal structure. The scattering objects displayed a surface fractal structure with a fractal dimension $D_s = 6 - \alpha$ in the case of $3 < \alpha < 4$, whereas a mass fractal structure was described with a fractal dimension $D_m = \alpha$ in the case of $1 < \alpha < 3$ (Cameron & Donald, 1993a, b). The fractal dimension of rice starch after extrusion

271 increased significantly. This suggests that the thermomechanical treatment caused the migration of
 272 water molecules to the interior of the starch granules, which destroyed the hydrogen bonding
 273 between starch chains and induced the reorganization of starch chains to form new aggregates.
 274 Compared with that of MERS/GG-0, the fractal dimension of MERS with GG was significantly
 275 larger. With increasing amounts of GG added, the fractal dimension of MERS and the density of
 276 nano-aggregates increased moderately.



278 Fig. 3 Double-logarithmic SAXS patterns of NRS and MERS samples.

279
280

281 Table 4. α , total crystallinity (X_{Total}), A-type crystallinity (X_A), and V-type crystallinity (X_V) of NRS
 282 and MERS samples*

Samples	α	X_{Total} (%)	X_A (%)	X_V (%)
NRS	1.66 ± 0.02^a	32.4 ± 0.5^f	32.0 ± 0.4^e	0.4 ± 0.1^a
MERS/GG-0	2.10 ± 0.02^b	17.8 ± 0.3^a	16.0 ± 0.1^a	1.8 ± 0.2^b
MERS/GG-2.5%	2.27 ± 0.01^c	19.7 ± 0.3^b	16.5 ± 0.2^b	3.2 ± 0.1^c

MERS/GG-5%	2.28±0.02 ^d	21.5±0.2 ^c	17.6±0.1 ^c	3.9±0.1 ^c
MERS/GG-7.5%	2.31±0.03 ^e	22.5±0.3 ^d	18.3±0.2 ^d	4.2±0.1 ^d
MERS/GG-10%	2.33±0.01 ^f	23.3±0.4 ^e	18.7±0.2 ^d	4.6±0.2 ^d

283 *Mean value ± standard deviation of duplicate analysis was given. Values with different letters
 284 within the same column differ significantly ($p < 0.05$).

285

286 3.7 Crystalline structure

287 Fig. 4 shows the XRD spectra of MERS samples. After extrusion, the rice starch maintained its
 288 A-type crystal structure, but the intensity of characteristic diffraction peaks at 15°, 17°, 18° and 23.3°
 289 (2θ) was significantly reduced, indicating reduced A-type crystallinity. The V-type crystalline
 290 scattering peak increased significantly after extrusion. Regarding this, the thermomechanical
 291 treatment may have caused the unwinding of double helices into single helices and the complexation
 292 between amylose and the starch granule endogenous lipid (Wang et al., 2018). MERS/GG-0 shows
 293 an A+V hybrid crystalline type. For MERS samples with GG, while the starch crystalline type
 294 remained the same, the diffraction peak intensity increased. As shown in Table 4, the addition of GG
 295 apparently increased the crystallinity of MERS, which was more so with a higher content of GG.

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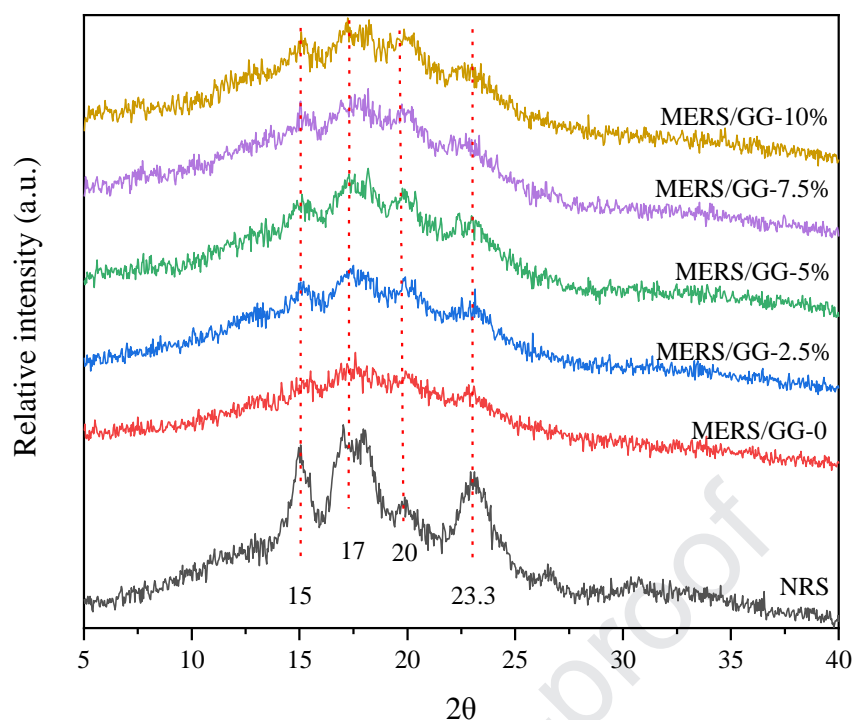


Fig. 4 X-ray diffraction patterns for NRS and MERS samples.

3.8 FTIR analysis

Fig. 5 shows the FTIR spectra for GG and MERS samples. The bands at 1047 cm^{-1} , 1081 cm^{-1} and 1022 cm^{-1} can be attributed to the C—O stretching vibration of the anhydroglucose unit (AGU); and additional strong signals at 3320 cm^{-1} and 2932 cm^{-1} could be assigned to O—H and C—H stretching vibrations (Fang, Fowler, Tomkinson, & Hill, 2002; Zhou et al., 2014). The infrared spectrum of GG was similar to that of MERS samples without or with GG. No new reflections were detected for the complexation between rice starch and GG. With increasing amounts of GG added, this absorption peak at around $3,300\text{ cm}^{-1}$ red-shifted gradually with increasing intensity, indicating hydrogen-bonding interactions between GG and starch chains.

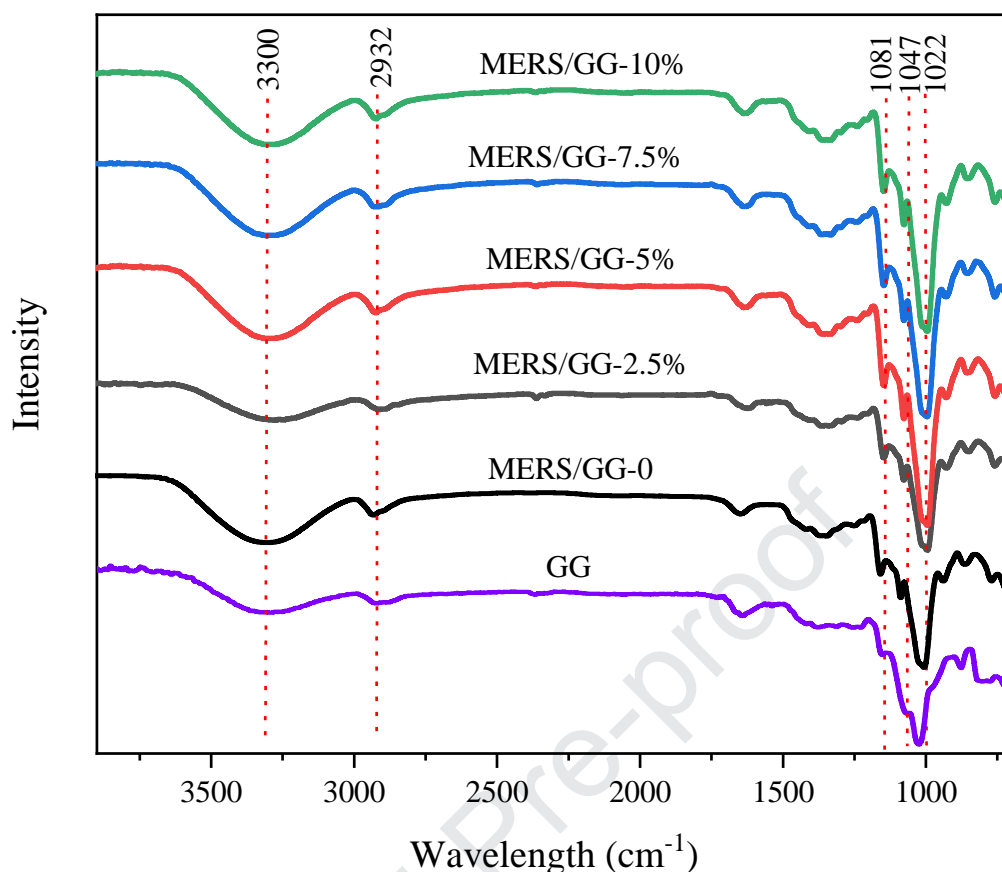


Fig. 5 FT-IR spectra for GG and MERS samples.

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313 The complexation between GG and rice starch could change the starch molecular conformation

314 and chain interactions. Fig. 6 shows the second-order derivative infrared spectra for MERS samples.

315 The peak at 991 cm⁻¹ represents the vibration absorption of C—O—H, which is related to hydrogen

316 bonding associated with the C₆ hydroxyl group of the AGU of starch (Soest, Tournois, Wit, &

317 Vliegthart, 1995). Compared with MERS/GG-0, the addition of GG red-shifted the absorption

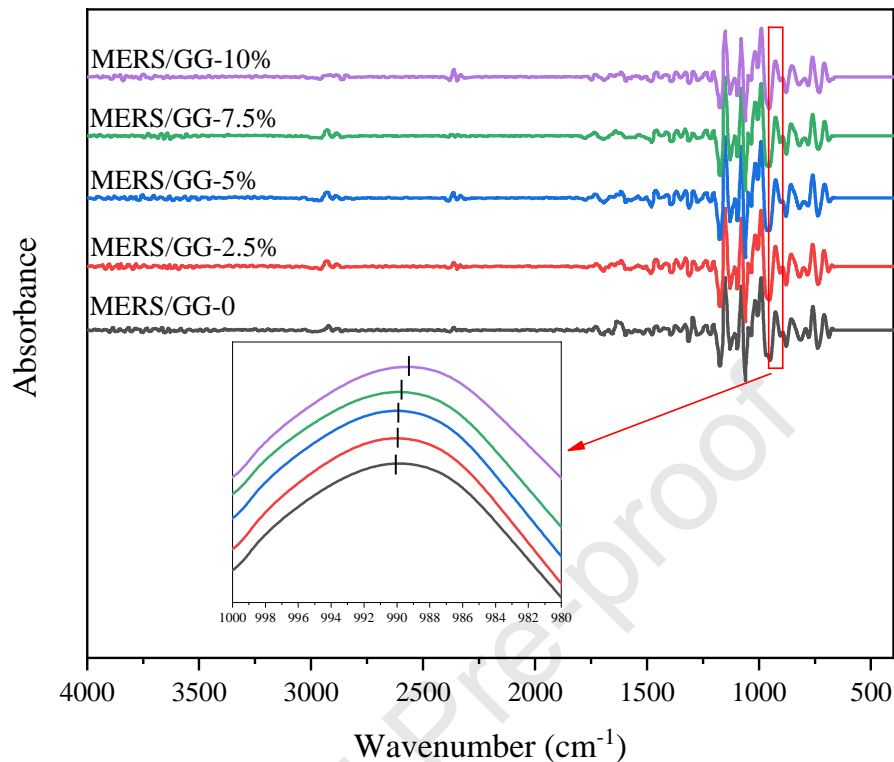
318 peak of C—O—H at 991 cm⁻¹, and higher amounts of GG led to a greater shift. This result suggests

319 that GG molecules could destroy the original hydrogen bonding in starch molecules by forming new

320 hydrogen bonds with the starch AGU, which is in agreement with the infrared spectroscopy data in

321 Fig. 5.

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Fig. 6 Second-derivative infrared spectra for MERS samples.

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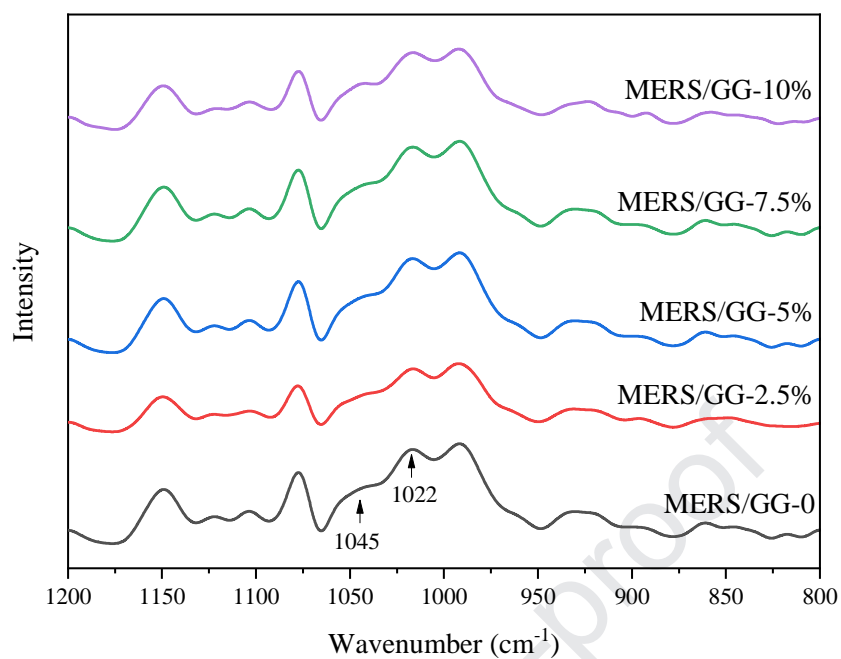
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A mechanistic analysis of the change in starch digestibility was performed by studying the change in the short-range ordered structure of starch. Fig. 7 shows an infrared deconvolution map for MERS samples, which could be used to calculate the degree of short-range ordering of the starch granule surface structure based on the ratio of intensity at $1,045/1,022\text{ cm}^{-1}$. As shown in Table 5, during extrusion, the semi-crystalline structure was completely destroyed, and the degree of starch ordering was reduced; however, the hydro-thermo-mechanical treatment could lead to the reassembly of starch chains and thus a higher degree of aggregation (Wang, Li, Copeland, Niu, & Wang, 2015; Wang, Wang, Wang, & Wang, 2017). Complexation with GG improved the ordered structure of the starch granule surface, which was more apparent with higher amounts of GG added.

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Fig. 7 Infrared deconvolution map for MERS samples.

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Table 5. $R_{1045/1022}$ and proportions of amorphous starch and helical structures of MERS samples*

Samples	$R_{1045/1022}$	Amorphous starch	Single helix	Double helix
MERS/GG-0	0.815 ± 0.002^a	79.2 ± 0.5^e	2.5 ± 0.3^a	18.3 ± 0.8^a
MERS/GG-2.5%	0.833 ± 0.014^b	77.7 ± 0.1^d	3.1 ± 0.3^b	19.2 ± 0.2^b
MERS/GG-5%	0.837 ± 0.002^c	74.4 ± 0.1^c	5.9 ± 0.0^c	19.7 ± 0.1^b
MERS/GG-7.5%	0.845 ± 0.013^d	72.6 ± 0.5^b	6.8 ± 0.3^d	20.6 ± 0.8^c
MERS/GG-10%	0.870 ± 0.006^e	70.4 ± 0.1^a	7.9 ± 0.0^e	21.7 ± 0.0^d

340

*Mean value \pm standard deviation of duplicate analysis is given. Values with different letters within

341

the same column differ significantly ($p < 0.05$).

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343 3.9 ¹³C CP/MAS NMR Analysis

344 The change in the helical structure of MERS samples was studied by ¹³C CP/MAS NMR and the
345 results are shown in Fig. 8 and Table 5. For MERS samples with GG, the C₁ peak was located
346 between 96 and 106 ppm, the C₄ peak between 80 and 85 ppm, the C_{2,3,5} peak between 68 and 78
347 ppm, and the C₆ peak between 58 and 65 ppm. Peaks at 99–102 ppm in C₁ region indicate V-type
348 single helices (eight glucose cycles per turn), and 103.2 ppm is also related to the amorphous starch
349 content associated with the junction points of amylopectin double helices. In addition, the peaks
350 centered on 101.5, 100.5, and 99.4 ppm in the C₁ region are characteristic of the double-helical
351 structure of A-type starch (Gidley & Bociek, 2002; Morrison, Tester, Gidley, & Karkalas, 1993). The
352 ratio between the area of these characteristic peaks and the area of the C₁ peak can be used to
353 calculate the proportions of single/double helices and amorphous structure (Fan et al., 2013). The
354 thermomechanical treatment can easily induce the breakage of starch α -1,6 glycosidic bonds, further
355 increasing the amylose content and promoting the formation of single helices (Liu et al., 2017).
356 Increasing amounts of GG led to higher amounts of single helices, a lower proportion of double
357 helices, and less amorphous content.

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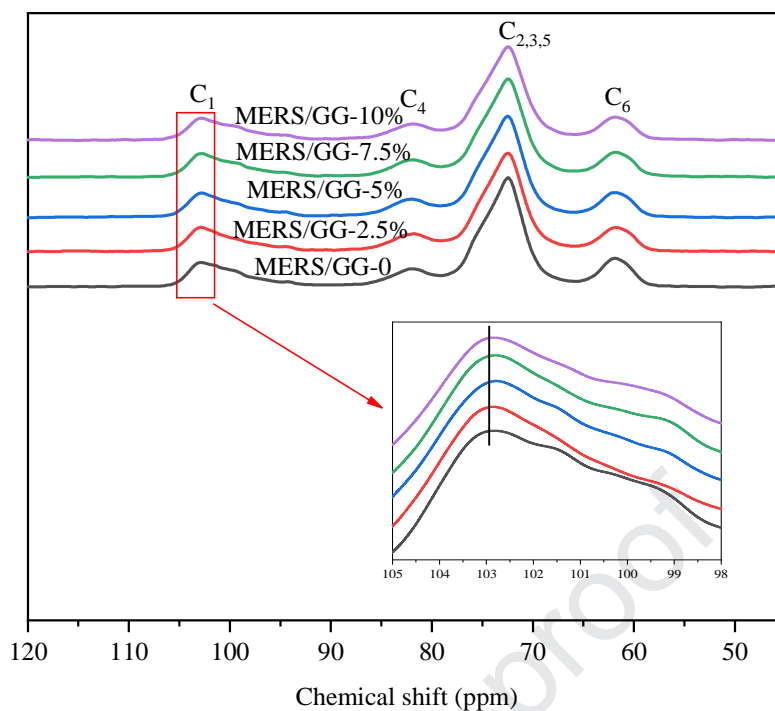


Fig. 8 ^{13}C CP/MAS NMR spectra for MERS samples

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362 4 Discussion

363 The thermomechanical treatment by extrusion facilitates the interaction of starch chains with
 364 water, causes changes in the starch multiscale structures (granule morphology, semi-crystalline
 365 structure, crystallites, and helical structures) and the structural ordering, and leads to the formation of
 366 new starch aggregates. These structural changes vary the digestibility and the blood glucose response
 367 level of rice starch.

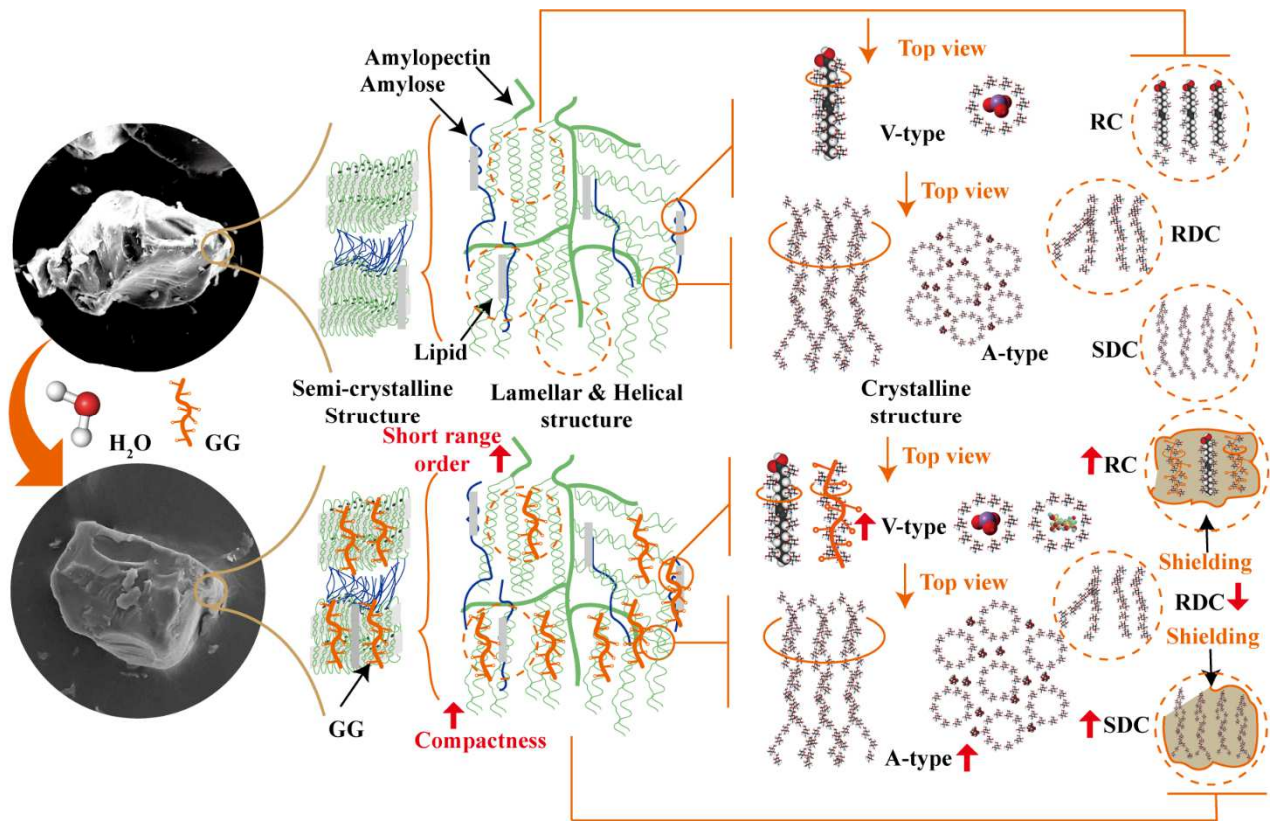
368 GG can be incorporated into starch-based food products through forming hydrogen bonding, van
 369 der Waals forces, and other non-covalent forces with starch chains. These interactions induce the
 370 rearrangement of starch chains to form more-ordered aggregates, which are more resistant to
 371 amylase (Zheng et al., 2019). Besides, the aqueous solution formed by GG has a high viscosity,

372 which inhibits the migration of amylase in solution and thus restricts the access of amylase to starch
373 chains (Sasaki & Kohyama, 2012). Moreover, Slaughter et al. (Slaughter, Ellis, Jackson, &
374 Butterworth, 2002) have suggested that some parts of an enzyme molecule may have a certain
375 affinity with GG, making GG a non-competitive inhibitor for the α -amylase-catalyzed hydrolysis of
376 starch molecules.

377 Fig. 9 shows a schematic representation of the non-covalent interaction of GG with rice starch
378 during extrusion and the associated changes in the multi-scale structures and digestibility of rice
379 starch. The thermomechanical treatment could facilitate hydrogen-bonding interactions between GG
380 and starch chains and the formation of new aggregates (i.e., single- or double-helices), which
381 effectively shielded the action sites on starch chains for amylase and thus reduced the hydrolysis by
382 amylase.

383 At a low content of GG addition (up to 10%), the complexation between GG and starch chains
384 led to a significantly higher RC content (Table 1). Regarding this, GG increased the ordering of the
385 aggregated structure (Fig. 3 and Table 4), the contents of single- and double-helices (Fig. 8 and Table
386 5), X_A and X_V (Fig. 4 and Table 4), thereby improving the digestion resistance.

387



388

389 Fig. 9 Schematic representation for the changes of multi-scale structure and digestibility of MERS
 390 samples.

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392 As the GG content increases, GG begins to adhere and wrap the starch granule fragments and
 393 the structure of the rice starch-GG complex becomes denser (Zhang, Gu, Hong, & Cai, 2012). In
 394 other words, with more GG added, GG may provide a spatial resistance (Bordoloi et al., 2012;
 395 Dartois, Singh, Kaur, & Singh, 2010). Increasing the amount of GG enhances the interaction
 396 between GG and starch chains and promotes the formation of an ordered, complexed structure
 397 (Zheng et al., 2019). While the ordered structure formed by starch chains highly depends on strong
 398 hydrogen bonding interactions and chain arrangements, there are also many weak hydrogen bonds
 399 between starch chains. The co-existence of these hydrogen bonds with different strengths results in a
 400 wider frequency range of the characteristic peak of hydroxyls (Bemiller & Huber, 2015). As amylose

401 readily forms complexes with organic molecules such as GG especially during thermomechanical
402 processing (Chaisawang & Suphantharika, 2005), the complexation can lead to increased V-type
403 crystallinity.

404 In ^{13}C CP/MAS NMR spectra, the position of C_1 is related to the helical structure, as the spiral
405 arrangement of glucose molecules in V_6 , V_7 , and V_8 structures is different. In other words, as the
406 bonding position of hydrogen bonds is different, the chemical displacement position of the C_1 peak
407 for the three types of helical structure varies. The C_1 chemical displacement signals of the V_6 , V_7 ,
408 and V_8 helical structures are 102.2–102.7 ppm, 103.3–103.4 ppm, and 103.9–104.3 ppm,
409 respectively (Bail, Rondeau, & Buleon, 2005). A higher chemical displacement value of general C_1
410 indicates a greater amount of helical structure in starch (Tian et al., 2018). The chemical
411 displacement of C_1 for MERS with GG was close to the V_6 value, indicating that the single-helical
412 structure in the complex was of two main types, namely V_6 and V_7 .

413 Table 6 shows the results of Pearson correlation analysis between the GG content and the
414 structure and digestive properties of MERS with GG. There was a significant correlation between the
415 RDC, SDC, and RC contents. The RDC content was significantly negatively correlated with the SDC
416 content (-0.997 , $p < 0.01$), RC content (-0.998 , $p < 0.01$), α (-0.994 , $p < 0.01$), X_{Total} (-0.959 , $p <$
417 0.05), X_{A} (-0.955 , $p < 0.05$), X_{V} (-0.962 , $p < 0.05$), $R_{1047/1022}$ (-0.950 , $p < 0.05$), single-helix content
418 (-0.952 , $p < 0.05$), and double-helix content (-0.997 , $p < 0.01$), and was significantly positively
419 correlated with amorphous starch content (0.975 , $p < 0.05$). In addition, the RC content was
420 significantly positively correlated with α (0.985 , $p < 0.05$), X_{Total} (0.970 , $p < 0.05$), X_{A} (0.964 , $p <$
421 0.05), X_{V} (0.976 , $p < 0.05$), $R_{1047/1022}$ (0.987 , $p < 0.05$), single-helix content (0.950 , $p < 0.05$), and

422 double-helix content (0.994, $p < 0.01$), and was significantly negatively correlated with amorphous
 423 starch content (-0.986 , $p < 0.05$). The addition of GG was significantly positively correlated with α
 424 (0.984, $p < 0.05$), X_{Total} (0.981, $p < 0.05$), X_A (0.978, $p < 0.05$), X_V (0.983, $p < 0.05$), single-helix
 425 content (0.962, $p < 0.05$), and double-helix content (0.987, $p < 0.05$). In addition, GG content was
 426 strongly negatively correlated with pGI (-0.970 , $p < 0.05$), which means the level of postprandial
 427 blood glucose response to MERS could be reduced by increasing the amount of GG added. RDC and
 428 pGI were positively correlated with each other (0.960, $p < 0.05$), while the RDC and RC contents
 429 were negatively correlated (-0.998 , $p < 0.01$). This indicates that GG reduced the RDC content and
 430 increased the RC content by changing the RC content, X_V , X_A , and the amounts of single- and
 431 double-helices.

432

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Table 6. Pearson correlation coefficients of GG content with starch structures and digestibility

	GG content	RDC content	SDC content	RC content
GG content	1	–	–	–
RDC content	-0.996^{**}	1	–	–
SDC content	0.987^*	-0.997^{**}	1	–
RC content	0.998^{**}	-0.998^{**}	0.989^*	1
α	0.984^*	-0.994^{**}	0.999^{**}	0.985^*
X_{Total}	0.981^*	-0.959^*	0.940	0.970^*
X_A	0.978^*	-0.955^*	0.938	0.964^*
X_V	0.983^*	-0.962^*	0.941	0.976^*
$R_{1047/1022}$	0.925	-0.950^*	0.948	0.987^*
Amorphous starch content	-0.992^{**}	0.975^*	-0.957^*	-0.986^*
Single helix content	0.962^*	-0.952^*	0.906	0.950^*

Double helix content	0.987*	-0.997**	0.996	0.994**
pGI	-0.970*	0.960*	-0.936	-0.976*

434 Mean value \pm standard deviation of duplicate analysis is given; * $p < 0.05$, ** $p < 0.01$.

435

436 5 Conclusion

437 For MERS, the addition of GG significantly increased the RC content and reduced pGI.

438 Moreover, MERS with GG showed increases in the degrees of the ordering of nano-aggregates and

439 the granule surface structure, higher crystallinity (X_{Total} , X_{A} , and X_{V}), and greater amounts of single-

440 and double- helices. Pearson correlation analysis shows that X_{Total} , X_{A} , X_{V} , and single- and

441 double-helix contents were correlated with the SDC and RC contents. Thus, this study demonstrates

442 that the starch structure and digestibility could be well regulated by the addition of GG combined

443 with thermomechanical treatment.

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453 **Conflicts of interest**

454 There are no conflicts of interest to declare.

455

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Journal Pre-proof

Highlights

- ✓ Micro-extruded rice starch (MERS) was prepared with guar gum (GG)
- ✓ GG addition enhanced slowly-digestible (SDC) and resistant (RC) components in MERS
- ✓ MERS with GG presented reduced digestion rate and predicted glycemic index (pGI)
- ✓ Crystallinity and helical structures contributed to SDC and RC in MERS with GG

– Declaration of Interest –

**Improving the *in vitro* digestibility of rice starch by
thermomechanically assisted complexation with guar gum**

Hai He^a, Chengdeng Chi^a, Fengwei Xie^{b, c, *}, Xiaoxi Li^a, Yi Liang^d, Ling Chen^{a, *}

^a Ministry of Education Engineering Research Center of Starch & Protein Processing, Guangdong Province Key Laboratory for Green Processing of Natural Products and Product Safety, School of Food Science and Engineering, South China University of Technology, Guangzhou, Guangdong, 510640, China.

^b International Institute for Nanocomposites Manufacturing (IINM), WMG, University of Warwick, Coventry CV4 7AL, United Kingdom.

^c School of Chemical Engineering, The University of Queensland, Brisbane, Qld 4072, Australia.

^d Guangdong Zhongqing Font Biochemical Science and Technology Co., Ltd., Maoming, Guangdong 525427, China.

***The authors declare that there is no conflict of interest regarding the
publication of this article.***