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1	Stable Isotope Switching (SIS): A New Stable Isotope Probing (SIP) Approach to
2	Determine Carbon Flow in the Soil Food Web and Dynamics in Organic Matter
3	Pools
4	
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14	
15	Running head: ¹³ C Stable Isotope Switching (SIS)
16	Abstract
17	Rationale: Recent advances in stable isotope probing (SIP) have allowed direct linkage of
18	microbial population structure and function. This paper details a new development of SIP,
19	Stable Isotope Switching (SIS), which allows the simultaneous assessment of C uptake,
20	turnover and decay, and the elucidation of soil food webs within complex soils or
21	sedimentary matrices.

Methods: SIS utilises a stable isotope labelling approach whereby the ¹³C-labelled substrate is switched part way through the incubation to a natural abundance substrate. A ¹³CH₄ SIS study of landfill cover soils from Odcombe (Somerset, UK) was conducted. C assimilation and dissimilation processes were monitored through bulk elemental analysis-isotope ratio mass spectrometry and compound specific gas chromatography-combustion-isotope ratio mass spectrometry targeting a wide range of biomolecular components including: lipids, proteins and carbohydrates.

Results: Carbon assimilation by primary consumers (methanotrophs) and sequential transport into secondary (Gram negative and positive bacteria) and tertiary consumers (Eukaryotes) was observed. Up to 45% of bacterial membrane lipid C was determined to be directly derived from CH_4 and at the conclusion of the experiment ca. 50% of bulk soil C derived directly from CH_4 was retained within the soil.

34 **Conclusions:** This is the first estimate of soil organic carbon derived from CH_4 and is 35 comparable to levels observed in lakes that have high levels of benthic methanogenesis. SIS 36 opens the way for a new generation of SIP studies aimed at elucidating total C dynamics 37 (incorporation, turnover and decay) at the molecular level in a wide range of complex 38 environmental and biological matrices.

39

40 Introduction

A range of new techniques, have recently emerged to study environmental microorganisms *in situ* without the need to establish laboratory cultures.^[1] Among these so-called *culture independent* methods' is stable isotope probing (SIP), an approach which involves the incubation of environmental soils or sediments with stable isotope labelled substrates. In

many cases a simple 'pulse chasing' approach is utilised, whereby a short application or 45 single dose of a highly enriched ¹³C-labelled substrate is applied to complex environmental 46 samples and traced to identify the fate of the substrate. Subsequently, the metabolic activities 47 of microorganisms are assessed through determination of label incorporation into 48 49 biochemical components of cells of active members of the microbial population including DNA,^[4] RNA^[5] and phospholipid fatty acids (PLFAs).^[1] Of major importance in this area has 50 51 been the application of highly sensitive detection techniques including gas chromatographycombustion-isotope ratio mass spectrometry (GC-C-IRMS)^[1,2] and more recently liquid 52 chromatography (LC)-IRMS approaches.^[6] 53

54 Initial SIP studies focussed on identification of microorganisms that utilised specific substrates, such as soil dwelling methanotrophic bacteria. Interest in soil methanotrophs 55 56 stems from the fact that such bacteria occur in every soil order and are an important sink for atmospheric CH₄ in well-aerated soils (high affinity methanotrophs)^[7] and a highly efficiency 57 filter that consumes >90% of upward diffusing CH₄ (low affinity methanotrophs) in soils 58 where a sub-surface CH₄ source exists (e.g. landfill cover soils) or *in situ* CH₄ production 59 occurs (e.g. natural wetlands) [8,9,10] Notable successes for SIP in this area include the 60 identification of unculturable high affinity methanotrophic bacteria via PLFA ¹³C-labelling in 61 well-drained non-agricultural soils.^[2] Subsequently, SIP has been extended to quantify 62 methanotroph biomass populations through time series ¹³CH₄ PLFA labelling.^[11] 63 Interestingly, despite their importance as a carbon sink, there has been little study of soil 64 65 methanotrophs in ecological contexts as a potential source of soil organic matter. The quantity of carbon cycled via high and low affinity methanotrophy in soils is globally 66 significant.^[12,13] vet the fate of that carbon remains largely unknown and unquantified. 67 68 Understanding the sources and stability of organic carbon in soils is a prerequisite for

development of realistic global carbon cycle models that contain fully coupled atmosphere-biosphere-geosphere interactions.

Significantly, it has become apparent that long-term time series ${}^{13}CH_4$ labelling has 71 the potential to yield a wide range of additional information, including: (i) kinetics of ${}^{13}C$ 72 73 uptake, (ii) mechanisms of C incorporation, and (iii) C flow pathways and turnover in soil. For example, in a time series ¹³CH₄-incubation study of methanotrophic bacteria in volcanic 74 soils from Tenerife, Spain high concentrations of ¹³C-label were incorporated into 75 methanotrophic PLFAs.^[14] Due to the high levels of ¹³C- incorporation of ¹³C, at later stages 76 of the ¹³CH₄ incubation, ¹³C-label was detected in non-methanotrophic fungal biomarkers 77 (e.g. C_{18:2}) providing a clear indication of how this approach could be used to investigate 78 79 pathways of C flow through soil microbial communities.

80 In this study we have combined two previous SIP methods i.e. short pulse-chase experiments and long-term continuous labeling approaches to study biosynthesis and C 81 uptake, to enable a comprehensive study of methanotroph C uptake and CH₄ derived C 82 transport and sequestration through the soil food web. A new long-term ¹³C-labelling 83 approach has been applied whereby ¹³CH₄ was switched to CH₄ containing natural abundance 84 levels of ${}^{13}C$ and ${}^{12}C$ when full labelling was achieved (as indicated by a maximum in the ${}^{13}C$ 85 86 label incorporation curve; Fig. 1). The incubation was then continued to monitor C turnover of the incorporated ¹³C-label (hence the term 'stable isotope switching' (SIS)). This approach 87 88 allows short, medium and long-term processes involved in the uptake and turnover of C to be investigated in detail. This continuous labelling method utilises flow-through incubation 89 system and differs from pulse labelling experiments,^[6] which typically are conducted over 90 91 much shorter timescales. The ability to supply a stable flow of isotopically labelled gas at a 92 concentration similar to natural environmental conditions limits disruption to the soil ecosystem and problems associated with selective fertilisation or competitive inhibition of thesoil microorganisms.

To demonstrate the potential of this technique we conducted a detailed SIS investigation on soils from the Odcombe landfill site (Somerset, UK) which had a previous study^[15] has shown to contain a significant population of methanotrophic bacteria. In this methodological paper we document the details of the SIS approach and summarise the range of data produced by SIS to demonstrate its many potential applications. Full datasets, detailed statistical analyses and consideration of all compound classes investigated will be reported in a subsequent communication.

102

103 Experimental

104 Site - Multiple soil cores (5 cm diameter, 10 cm depth) were collected from two sites at the 105 Odcombe landfill (SW England, 50°56'45"N -2°42'19"W) in March 2007. The landfill consists of a terraced area that was formed by the stacking of waste, which was later capped 106 107 with sand and clay. The soil cap was sampled from an area of high CH₄ emissions adjacent to a vent previously identified by Crossman et al.^[15] In this earlier study, conventional PLFA-108 109 SIP identified significant populations of low affinity methane oxidising bacteria similar to 110 known Type II methanotrophs, making the cap soil ideally suited for this trial SIS study 111 because of the potential for incorporation of significant amounts of CH₄-derived C into the 112 soil food web. Additionally, landfill cap material is a unique aerobic mineral soil where 113 methanotrophs are a major component of soil microbial biomass, facilitating both rapid and comprehensive labelling of the methanotrophic population and enabling sensitive and 114 selective tracking of the fate of the ¹³C-signal. 115

116 SIS CH₄ incubation

117 Soils were incubated in a flow through incubation chamber that has been previously described in Maxfield *et al.*^[11] All soils were sieved (<2 mm) and incubated in petri dishes 118 119 (ca. 20 g) over a time course of 154 days. Synthetic air premixed with CH_4 (1.33%) was 120 flowed continuously through the chamber to maintain a steady CH₄ mixing ratio similar to 121 that measured at the landfill site. For the initial 50 days of the incubation 1% of the CH₄ was ¹³CH₄ (mixed from > 99% ¹³C; CK Gas Products Ltd, Hook, UK; equivalent to 133 ppmv 122 ¹³CH₄). Following 50 days of incubation the input of ¹³CH₄-bearing gas was discontinued and 123 switched to a pre-mixed supply of CH₄ (13300 ppmv) containing natural abundance levels of 124 ¹³C and ¹²C. The gas flow rate through the chamber was maintained at 44 mL min⁻¹ 125 throughout the entire incubation period, which flushed the entire headspace (63 L) every 24 126 h. Soil samples were removed in triplicate and from random positions in the incubator at 127 128 regular intervals during the incubation period at times of 0, 3, 6, 9, 12, 18, 21, 27, 38, 50, 53, 56, 65, 85, 117, 154 days. The initial moisture content of the soil was maintained through the 129 regular addition of double-distilled water (DDW) to each sample (determined 130 gravimetrically). All samples removed from the incubator were stored at -20° C until analysed. 131 132 Soil samples were freeze-dried and ground prior to extraction and analysis for PLFAs, total amino acids (AAs), total monosaccharides glycolipids (GLFAs), free fatty acids (FFAs), 133 134 hopanoids, sterols, *n*-alkanes and *n*-alkanols.

135

136 Lipid extraction and fractionation

All soils were extracted using a modified Bligh Dyer extraction methodology as described
 previously.^[11] Lipid fractionation was obtained using a modified silicic acid fractionation
 protocol^[16] to yield three fractions: simple lipids, glycolipids and phospholipids. Simple

140 lipids were further fractioned using silicic acid columns into four fractions; hydrocarbons 141 (eluted with hexane, 4 mL), ketones and wax esters (dichloromethane (DCM), 6 mL), 142 alcohols (DCM/methanol (MeOH), 1:1 v/v, 4 mL), and polar poly-functionalised compounds 143 (MeOH, 4 mL). Alcohols and hydrocarbons were further separated by urea adduction to 144 separate cyclic from acyclic components.^[17]

145

146 Lipid derivitization

147 The PLFA, GLFA and FFA fractions were methylated with $BF_3/MeOH (14\% w/v)$ by heating 148 at 70°C for 1 h. Fatty acid methyl esters (FAMES) were dissolved in *n*-hexane for analysis by 149 GC, GC/mass spectrometry (MS) and GC/C/IRMS. All neutral polar lipids were derivatized 150 with *N*,*O*-bis(trimethylsilyl)trifluoroacetamide (BSTFA) prior to analysis by GC, GC/MS and 151 GC/C/IRMS.

152

153 Extraction and derivatization of carbohydrates

The method of Blakeney *et al.*^[18] modified by Docherty *et al.*^[19] was employed to prepare 154 alditol acetate derivatives of total monosaccharides. Soils were hydrolysed with H₂SO₄ and 155 156 the released monosaccharides reduced with NaBH₄. Excess NaBH₄ was destroyed by the 157 addition of glacial acetic acid to reduce the liberated monosaccharides to their corresponding alditols. The alditols were acetylated by reaction with N-methylimidazole and acetic 158 anhydride. A standard mix of rhamnose, fucose, galactose, mannose, xylose, arabinose 159 glucose, inositol and pentaerythritol was prepared according to Docherty et al.^[19] The DDW 160 161 (400 µl) was added to the dry monosaccharide standards which were derivatized as above. A

162 20 μg μl⁻¹ internal standard (I.S.) was prepared by dissolving 0.04 g pentaerythritol in 2 ml 1
163 M NH₃ solution.

164

165 Extraction and derivatization of amino acids

An internal standard of nor-leucine (0.2 mg ml⁻¹) was prepared for quantification of AAs. 166 167 Each AA produced a different GC-FID response which required correction by the application 168 of response factor. The FID response factors were determined from a standard solution of AAs of known concentration according to the method of Corr *et al.*^[20] Extracted soil samples 169 were hydrolysed with HCl and purified by ion exchange chromatography^[21] using Dowex 170 171 50WX8-400 Ion Exchange Resin (Acros Organics, Geel, Belgium). The purified amino acids 172 were derivatized to N-acetyl methyl esters using acetone, triethylamine and acetic anhydride $(5:2:1, v/v)^{[20]}$. 173

174

175 Instrumental analyses

GC analysis were performed using a Hewlett-Packard Series 5890 Series II gas 176 177 chromatograph (Agilent Technologies UK Ltd., Edinburgh, UK) equipped with a flame 178 ionisation detectors (FID) using H₂ carrier gas (10 psi). Non-polar fractions were analysed 179 using a Chrompack CPSil-5CB (50 m x 0.32 mm i.d. x 0.12 µm film thickness). The temperature conditions were 50°C to 200°C at 10° C min⁻¹, to 300°C at 3°C min⁻¹ (held for 20 180 181 min). Polar compounds were analysed using a Varian VF23ms (Varian BV, Middelburg, The 182 Netherlands) 50% cyanopropyl equivalent fused-silica column (60 m x 0.32 mm i.d. x 0.25 183 µm film thickness). The temperature conditions for fatty acid derivatives were 50°C (2 min) to 100°C at 15°C min⁻¹, to 240°C at 4°C min⁻¹ (held for 20 min). The temperature programme 184

for monosaccharides was 50°C (1 min) to 200°C at 20°C min⁻¹, to 230°C at 4°C min⁻¹, (held for 22 min). The temperature programme for amino acids was 40°C (1 min) to 120°C at 15°C min⁻¹, to 190°C at 3°C min⁻¹, to 250°C at 5°C min⁻¹ (held for 20 min).

GC-MS analyses were performed using a Thermo Finnigan Trace GC-MS (Thermo Fisher Scientific, Hemel Hempstead, UK). All the GC conditions were the same with the exception of helium being used as carrier gas. The interface was held at the maximum oven temperature, the ion source was held at 200°C and the quadropole mass analyser operated in EI mode scanning over the range m/z 50-650 at 1.7 scans s⁻¹. The emission current was maintained at 300 µA and electron energy was 70 eV. The data were acquired and analysed using the Excalibur software Version 1.2 (Thermo Fisher Scientific, Hemel Hempstead, UK).

195 GC-C-IRMS analysis were carried out using a Varian 3500 GC (Varian BV, Middelburg, 196 The Netherlands) coupled to a Finnigan MAT DELTA-S isotope ratio mass spectrometer 197 (Thermo Fisher Scientific, Hemel Hempstead, UK). Analytes in He were combusted to CO₂ 198 using a modified Type I Finnigan MAT combustion interface with a CuO/Pt combustion 199 reactor set at a temperature of 850°C prior to entry into the mass spectrometer source via an 200 open split. The ionisation source electron energy was 100 eV with an electron current of 1 201 mA. Detection was via 3 Faraday cup collectors set at m/z 44, 45 and 46. The 202 chromatographic conditions were the same as those described previously for GC analyses. All samples were analysed in duplicate to verify reliability of δ^{13} C values. Samples were 203 204 calibrated against reference CO₂ of known isotopic composition which was introduced 205 directly into the source three times at the beginning and end of every run. Compound specific IRMS performance was determined using a suite of externally calibrated reference fatty acid 206 207 methyl esters. Analytical precision was $<0.5 \$ ($\pm 1 \$ standard deviation) based upon replicate analysis of reference standards (n=5). 208

EA-IRMS analysis of freeze-dried and ground soil samples was conducted using a Eurovector C-N (Eurovector, Milan, Italy) elemental analyser interfaced to an Isoprime (Isoprime Ltd., Manchester, UK) stable isotope ratio mass spectrometer (IRMS). Samples were weighed (1–2 mg), placed into tin capsules and combusted. δ^{13} C values of the resultant CO₂ from combustion were determined at the NERC Stable Isotope Facility at CEH Lancaster with an analytical precision of <0.15 ‰ (± 1 standard deviation, n=16).

215

216 Statistical analyses

Two approaches were used to analyse the ¹³C-label incorporation curves. Linear and non-217 linear regressions were performed using GraphPad Prism version 5.02 for Windows 218 219 (GraphPad Software, San Diego, CA, USA), R 2.8.1 (R foundation), and Microsoft Excel 220 (Microsoft, Redmond, WA, USA). Least squares linear regression analysis was applied to %-221 incorporation data from the first 9 days of the incubation to obtain initial zero-order rate 222 constants, whereas separate non-linear regressions were applied to incorporation data from 223 the first 50 d of the incubation before the switchover to unlabelled CH₄, and the C turnover data from 50 d to 150 d. The equations of the fitted lines were of the form: 224

- 225 Incorporation: $F = F^0 + (P Y^0) * (1 \exp(-K * t))$ (1)
- 226 Decay: $F = (F^0 P)^* \exp(-K^* t) + P$ (2)

Where F is the fraction of excess 13 C incorporated into organic material, t is the incubation time in days, R⁰ is the initial R value at initial t, *P* is the plateau at maximum 13 C incorporation (incorporation) or maximum 13 C loss (decay) and K is the first order rate constant expressed in inverse days.

232 **Results and Discussion**

The SIS approach differs from pulse-chase methods in two key respects: (i) substrate delivery 233 234 is maintained throughout the experiment at a constant concentration in order to establish 235 equilibrium conditions between substrate and the primary substrate consumer population thus 236 allowing evaluation of substrate cycling purely based on changes in isotopic labelling patterns in different endogenous chemical species, and (ii) the only change in substrate 237 delivery is in its stable isotopic composition, i.e. ¹³C-enriched to natural abundance, which is 238 239 switched when isotopic equilibrium with the primary consumers has been established. As a result it is possible to study both assimilation and decay of the 13 C-label in a wide range of C 240 pools and molecular species (including biomarker compounds) within a complex ecosystem, 241 242 in order to provide insights into assimilation pathways, kinetics of turnover and quantitative estimates of pool sizes. 243

244

245 Methanotrophic bacteria as primary consumers

246 Soil PLFA profiles provide general information about soil microbial community structure, 247 diversity and size. PLFA nomenclature indicates the length of the C chain (first number), the number of double bonds (number after the colon), the position of the double bond 248 249 $(\omega, \text{ counted from the defunctionalised end of the molecule})$ and geometry (for full details see Zelles).^[22] The extremely high abundance of 18:1007c observed in the Odcombe vent PLFA 250 profile (Fig. 2) indicates the likely dominance of α -proteobacterial (type II) methanotrophs 251 linked to the high CH₄ flux at the site. This predominance was confirmed by GC-C-IRMS 252 analysis of the PLFA fraction following the ¹³CH₄-enriched incubation which also showed 253 with the largest proportion of 13 C was incorporated into PLFA 18:1 ω 7c. The prevalence of 254 methanotrophic bacteria resulted in highly 13 C-labelled soils following the 13 CH₄/ 12 CH₄ 255

incubation enabling methanotroph C derived from ${}^{13}CH_4$ to be traced through the complex soil food web.

258

259 Stable Isotope Switching

The extracts from the ¹³C-labelled soils were analysed by GC-C-IRMS to determine $\delta^{13}C$ 260 261 values to quantify and monitor the fate of metabolised CH₄ across a wide range of compound classes representative of a range of soil biota. Whilst PLFAs are the most commonly studied 262 biomarkers in methanotrophic bacteria SIP studies we and others have shown previously the 263 potential for linking hopanoids with methanotrophic bacteria through ¹³CH₄-labelling 264 studies.^[23,24,25] However, the purpose of SIS is to move beyond functional taxonomic 265 266 profiling to explore more fully C cycling and soil microbiological function. Thus, we have conducted a comprehensive survey of methanotroph-derived biochemicals tracing ¹³C-label 267 268 into amino acids, carbohydrates, glycolipids, free fatty acids, alcohols, alkanes, hopanoids, 269 sterols and resorcinols (Fig. 3). Different compound classes represent soil C pools of 270 differing stability and recalcitrance. Further, the magnitude of methanotroph C turnover by the soil microbial community suggests that those biochemicals which display little or no 13 C-271 272 labelling are not closely linked with soil C turnover by soil microorganisms, and thus 273 represent either stable soil C pools receiving little fresh C input, or are soil C pools with a conserved source of C. In addition the total incorporation of ¹³C into the bulk soil also was 274 assessed. Comparison with this bulk $\delta^{13}C$ value indicates the recalcitrance of individual C 275 276 pools relative to the total pool of soil organic C (Fig. 3).

Figure 4 shows an overview of ¹³C-label assimilation and dissimilation profiles for a range of selected soil compounds from several of the main compound classes: $16:1\omega7c$ PLFA, C₃₁ homohopenol, glucose, valine and the bulk soil. Whilst there are clear differences in the extent of ¹³C-labelling of these compound classes the δ^{13} C curves exhibit the general shape predicted from the experimental design (Fig. 1) with C turnover following a first order rate dissociation curve (Fig. 4, right hand side). However, it is important to note both the differences in the relative uptake of ¹³C into the different soil compound classes and the wide range of turnover rates which reflects the refractiveness of different biochemicals analysed in this study.

286

287 Soil Food Web

As with previous conventional time series labelling studies ¹³C-label uptake monitored by 288 GC-C-IRMS and expressed by δ^{13} C values represents the magnitude of 13 C-label uptake into 289 a specific compound as a proportion of the total concentration of that compound. Thus, $\delta^{13}C$ 290 291 values do not indicate the absolute amount of ¹³C-label present in a specific pool but rather the proportion of that C pool that is derived from the ¹³C-labelled C source. Hence SIS 292 293 provides a potential new tool for use in soil food web studies which can be employed to 294 determine the rate of C flow through a soil microbial network and the rate of in situ 295 biosynthesis of specific compounds within that network. Figure 4 indicates that a significant 296 proportion of PLFAs within the Odcombe soil were synthesised by methanotrophs directly from CH₄. All PLFAs displayed uptake of the ¹³C-label indicating both the high magnitude of 297 initial ¹³CH₄ incorporation and the extensive redistribution of assimilated C within the soil 298 system. The rates and magnitudes of ¹³C uptake vary widely indicating that the PLFA 299 producing organisms differed in terms of their usage of CH₄ derived C in biosynthesis, and 300 the biochemical proximity of different PLFAs to the ¹³C source (¹³CH₄). To more easily 301 visualize PLFA ¹³C incorporation profiles those data are shown separately (in Fig. 5) 302 highlighting the differences in ¹³C uptake rate between different groups of PLFA sources. 303

304 Despite their lack of taxonomic specificity PLFAs have been widely used to broadly characterize microbial populations^[22] and as such PLFAs are an extremely effective tool for 305 identifying active bacterial groups utilizing ¹³C-labelled tracer compounds. For example, 306 307 PLFAs commonly linked to methanotrophic bacteria ($18:1\omega7c$, $16:1\omega7$, $16:1\omega5$) most rapidly incorporate the ¹³C-label (Fig 5, panel a). The primary methanotroph PLFA in this soil, 308 18:1007c also reaches a plateau before the ¹³CH₄ supply was discontinued following 50 days 309 incubation, indicating maximum ¹³C-labeling of the methanotroph population and 310 311 establishment of isotopic equilibrium within the primary consumer population. Primary 312 methanotrophic PLFAs can be separated from PLFAs that incorporated a lower proportion of 13 C at a slower rate, the latter likely being indicative of secondary (Fig 5, panel b) and tertiary 313 consumers (Fig 5, panel c). Secondary and tertiary consumers incorporate ¹³C-label at much 314 315 reduced rates compared to primary consumers. For example following 18 days incubation under ${}^{13}CH_4$ primary consumer $\delta^{13}C$ values increased by +100 to +300 ‰ (Fig. 5a), 316 secondary consumer δ^{13} C values increased by +30 to +60 % (Fig. 5b), and tertiary consumer 317 δ^{13} C values increased by +10 to +20 ‰ (Fig. 5c). Because microorganisms linked with non-318 primary methanotroph PLFAs are not obtaining ¹³C label direct from ¹³CH₄ the likely 319 alternative ¹³C-label redistribution mechanisms are metabolite release and uptake, necromass 320 321 grazing and direct predation.

Furthermore, it was observed following prolonged incubation under 13 CH₄ that a significant proportion of PLFA C was derived from CH₄. Accounting for the fact that the 13 Cenriched CH₄ used in this study was only 1% enriched in 13 CH₄, and extrapolating the results to take this dilution into account, following 38 days of the incubation 45% of the C present in the Odcombe soil PLFAs was derived from CH₄. This proportion includes all PLFAs extracted from the soil, which suggests that >45% of total bacterial PLFA C consists of C derived from CH₄. Even allowing for the high abundance of methanotrophs in the Odcombe 329 soil this is an unexpectedly high proportion of bacterial C after a relatively short incubation 330 periods, suggesting that microbial communities in soil and sedimentary environments where a 331 significant source of CH₄ is present must utilise CH₄ as a major source of C and energy for 332 growth. This high proportion of C is consistent with the appreciable amount of CH₄-derived 333 C that is incorporated into certain lake food webs where CH₄ production is prevalent in anoxic sediments.^[26] Whilst there is a wide range of estimates between different lakes, 334 335 chironomid larvae have been observed as a primary conduit for the trophic transfer of biogenic CH₄ gaining >60% of their C from CH₄^[27] and zooplankton in small boreal lakes 336 ~50% through grazing on methanotrophs.^[28] In these lake studies trophic C transfer was 337 338 determined via physical separation of the organisms of interest prior to bulk isotopic 339 measurement by EA-IRMS. The work reported herein is the first assessment of the proportion 340 of CH₄ derived C that is transmitted through microbial food webs, which has been made 341 possible through the application of SIS in combination with GC-C-IRMS analysis of a wide 342 range of soil biochemical components.

343

344 C sequestration

In addition to C uptake kinetics, the SIS method also enables the study of rates of C turnover, 345 346 redistribution and sequestration. Figure 4 (right hand side) shows C turnover of PLFAs following the change from incubation under ¹³CH₄ to natural abundance CH₄ and the fitted 347 348 first order exponential dissociation curves used to determine C turnover rate constants. Box 349 plots of C turnover for each group of compounds analyzed by GC-C-IRMS indicate no 350 observable relationship between C assimilation and turnover (Fig. 3). There is a high degree of variability in both C turnover rates and loss both within and between various compound 351 352 classes. Although CH₄ is a significant source of C in the Odcombe soils, a large amount of 353 the C utilized in cellular biosynthesis is rapidly lost from the system and not sequestered in 354 long-term C pools. The C pools where C retention is poor include PLFAs, GLFAs, FFAs, carbohydrates, resorcinols and *n*-alkanols, which all lose >60% of their CH₄-derived C after 355 100 days of incubation under natural abundance CH₄. These losses suggest that despite 356 significant initial incorporation of CH₄-derived C, little of the C will be retained long-term 357 358 within the Odcombe landfill cover soils. The most recalcitrant forms of C include proteinderived amino acids, steroids and hopanoids (Fig. 3). Notably, a wide range of turnover rates 359 360 were observed for the hopanoids, and the diversity is likely due to functionalised hopanoids 361 being converted to their de-functionalised more stable analogues, indicating long residence 362 times for the pentacylcic hopanoid core structures.

First order dissociation curves were fitted to 13 C decay curves (Fig. 4) and associated rate constants were used to calculate CH₄-derived C residence times in the Odcombe soils. Total bulk soil 13 C had a half-life of 68 days and 47% of the total C derived from CH₄ was retained within the soil after 100 days. Extrapolation of these values indicates that 1 year after SIS just 19% of the CH₄-derived C will remain in the Odcombe soil, with 81% being released primarily as CO₂ via aerobic soil respiration.

369

370 Conclusions

This study demonstrates that SIS has considerable potential as a new method for determining the kinetics of soil C uptake, turnover, release and sequestration at the molecular level in complex soil and sedimentary matrices. Our findings indicate that in the Odcombe landfill cover soil 45% of bacterial membrane lipid C was directly derived from CH_4 and at the end of the experiment 47% of bulk soil C derived from CH_4 was retained within the soil. In this communication we have demonstrated the potential power of this technique to resolve 377 discreet C cycling processes within a complex environmental sample. Whilst we have 378 initially focussed on the soil food web in a landfill cover soil, SIS has considerable potential 379 for identifying and elucidating hitherto elusive aspects of nutrient cycling in many different 380 environments using a range of gaseous or liquid substrates, and will lead to new molecular 381 level estimates of ecosystem nutrient dynamics for use in process-based models of element cycling. Previous applications of SIP have utilised a range of ¹³C-labelled substrates 382 383 including acetate, glucose, methanol, CO₂ (both directly and indirectly via individual plants; for full overview see Maxfield and Evershed and references therein).^[29] The main drawback 384 of the approach is the requirement to incubate soils removed from their natural habitat for 385 386 long periods, which has the potential to cause changes in the composition of the wider soil 387 microbial community. This issue could be overcome by conducting SIS incubations in situ (i.e., in field setting, using an appropriately designed isotope delivery system).^[30] The SIS 388 technique could be readily applied to other light rare isotopes including ²H and ¹⁵N and we 389 390 recently developed an aqueous-based re-circulating isotope delivery system that can be 391 employed to conduct SIS experiments using a wide range of different substrates.

392

393

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495 **Figure Captions**

Figure 1: Schematic representation of stable isotope switching (SIS) experiment based upon previously observed (light grey) and theoretical (dark grey) curves indicating how 13 C-label from 13 CH₄ is incorporated by methanotrophs and turned-over within soil ecosystems. The timescale to achieve full labelling of the target population will vary depending upon the isotopically labelled substrate composition, concentration, delivery method and the nature of the environmental sample to be studied.

Figure 2: Partial gas chromatogram of the Odcombe landfill soil PLFA fraction (T_0 d). Where 1, C_{19} alkane; 2, *i*14:0; 3, *a*14:0; 4, *i*15:0; 5, *a*15:0; 6, 15:0; 7, *i*16:0; 8, 16:0; 9, 16:1 ω 7, 16:1 ω 5; 10, *i*17:0; 11, *a*17:0; 12, 17:0; 13, 17:1 ω 8; 14, 18:0; 15, 18:1 ω 7; 16, 18:2; 17, 20:0; 18, 18:3; 19, 11-CH₃O-17:0; 20, 22:0; 21, br23:0; 22, 23:0; 23, 24:0. See text for explanation of PLFA nomenclature.^[26]

Figure 3: Top panel - ¹³C-label uptake rate (zero order) from ¹³CH₄ into a wide range of soil biochemicals; Bottom panel - ¹³C-label turnover (first order) following SIS to natural abundance. Whiskers represent the lowest data point still within 1.5 of the interquartile range of the lower quartile, and the highest data point still within 1.5 of the interquartile range of the upper quartile.

512 **Figure 4:** Mean δ^{13} C values for bulk C and selected components extracted from Odcombe

513 landfill soil following incubation under 1.3% CH₄. Error bars represent ±1 standard deviation

514 (n = 3). Left hand side = 1% enriched in ${}^{13}CH_4$. Right hand side = CH₄ with a natural

abundance of ${}^{13}C$ and ${}^{12}C$. C decay curves were fit to a first order dissociation rate constant

516 (Equation 2)

517	Figure 5: ¹³ C-label incorporation by selected PLFAs. (a) primary consumers, (b) secondary
518	consumers, and (c) tertiary consumers. Note the different y-axis scales for plots a, b and c.
519	Error bars represent ± 1 standard deviation from the mean (n = 3).
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Carbon flow to non-methanotrophic pools (2°, 3° uptake)



⁵³⁷ Figure 1: Maxfield *et al*.





557 Figure 3: Maxfield *et al.*



Figure 4: Maxfield et al.



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563 Figure 5: Maxfield *et al.*