

Full title: Ocean acidification compromises a planktic calcifier with implications for global carbon cycling: Supplemental Material

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Experiment Run	Surface Conditions at Collection			Conditions during Incubation for Oxygen Consumption			Average Conditions for Entire Experiment			
	pH _T	[CO ₃ ²⁻] (μmol kg ⁻¹)	Ω _{Ca}	pH _T	[CO ₃ ²⁻] (μmol kg ⁻¹)	Ω _{Ca}	pH _T	[CO ₃ ²⁻] (μmol kg ⁻¹)	Ω _{Ca}	Salinity (psu)
C1	8.08	132.55	3.47	7.59	57.09	1.31	7.56	53.8	1.39	32.5
C2	8.08	132.55	3.47	8.15	179.65	4.40	8.16	180.97	4.36	32.5
D1	8.18	174.23	4.34	7.75	79.71	1.82	7.72	75.08	1.93	32.98
D2	8.18	174.23	4.34	8.27	221.89	5.30	8.27	219.18	5.36	32.98
E1	8.14	165.08	4.01	7.53	49.76	1.21	7.59	56.78	1.38	32.54
E2	8.14	165.08	4.01	7.75	80.06	1.68	7.69	69.24	1.94	32.54
F1	8.05	139.28	3.38	7.97	117.70	2.90	7.96	119.6	2.86	32.46
F2	8.05	139.28	3.38	8.28	224.22	5.40	8.28	222.58	5.44	32.46
G1	8.05	141.73	3.42	7.72	70.95	1.56	7.66	64.03	1.72	32.19
G2	8.05	141.73	3.42	8.00	129.18	3.04	7.98	125.19	3.14	32.19
H1	8.04	137.87	3.30	7.59	55.69	1.18	7.52	48.47	1.35	32.54

Table S1. Carbonate chemistry parameters for each experimental run, including sea surface parameters at the time of collection, measurements of treatment water at the time of incubation for respirometry, and average conditions (from water samples taken every other day) over the entire experimental run. Each letter represents a single collection and letter and number pair an experimental run.

Experimental Group	pH Before Filling	pH After Filling	[CO ₃ ²⁻] (μmol/kgSW) Before Filling	[CO ₃ ²⁻] (μmol/kgSW) After Filling
C1	7.59	7.59	56.70	57.48
C2	8.15	8.16	178.25	181.04
D1	7.73	7.76	77.22	82.20
D2	8.28	8.27	224.53	219.26
E1	7.57	7.61	54.61	58.95
E2	7.74	7.76	78.39	81.72
F1	7.98	7.96	119.28	116.13
F2	8.28	8.28	223.34	225.10
G1	7.69	7.74	67.17	74.73
G2	8.00	8.00	129.08	129.27
H1	7.58	7.60	54.90	56.49

Table S2. During respirometry incubations, individual foraminifera were maintained in treatment water without a fluorescent calcein label. Treatment water was filtered from 0.6μm to 0.2μm for this purpose and to account for any change in chemistry as a result of the filtration, pH (total) was measured before and after the vials were filled.

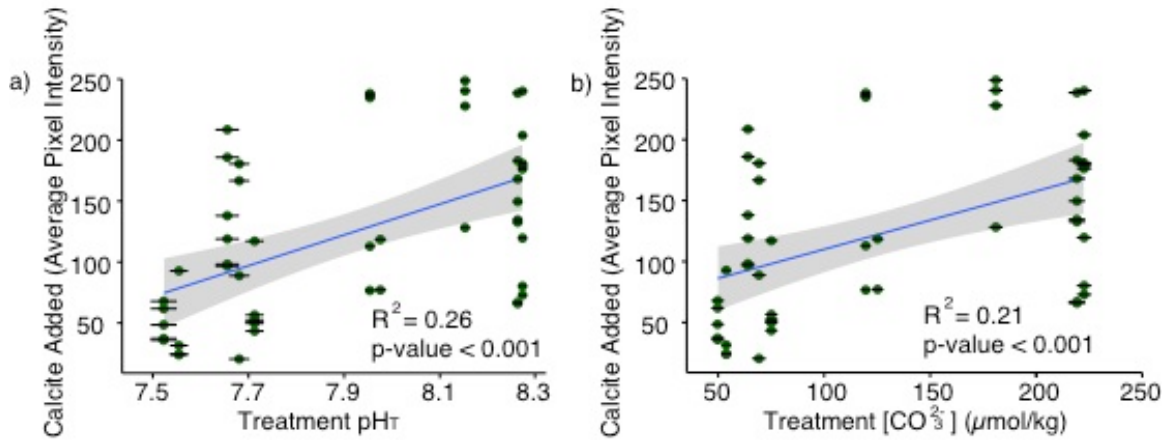


Fig S1. Individual foraminifera pixel intensity shows calcite added for single shells. Reduced calcification under low pH or $[\text{CO}_3^{2-}]$ has previously been described for foraminifera in culture (1-3), in plankton tows (4), and in the fossil record (5, 6), and is generally assessed by a ratio of weight to size. The application of this methodology to foraminifera is novel, allowing for a semi-quantitative documentation of only the amount of calcite grown under known (culture) conditions. This is a useful approach in assessing calcification of individuals that did not add a new chamber under culture conditions.

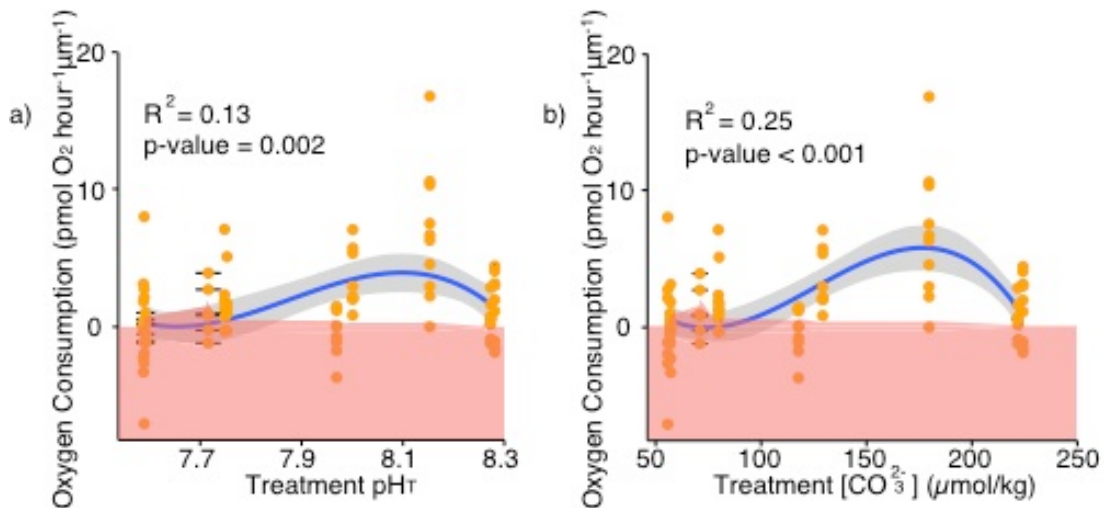


Fig. S2 Aerobic metabolism was assessed as the rate of oxygen consumption relative to background respiration in filtered seawater and normalized to shell length ($\text{pmol O}_2 \text{ foram}^{-1} \text{ hr}^{-1} \mu\text{m}^{-1}$). Individual foraminifera showed a large degree of inter-individual variability in respiration rates. The cause of this variability is indicative of differing physiologic states across individual shells, not captured by quantitative or qualitative metrics of health or life history. Such internal processes must play an important and as of yet undefined role in foraminiferal oxygen consumption. The pink shaded region represents results indistinguishable from background.

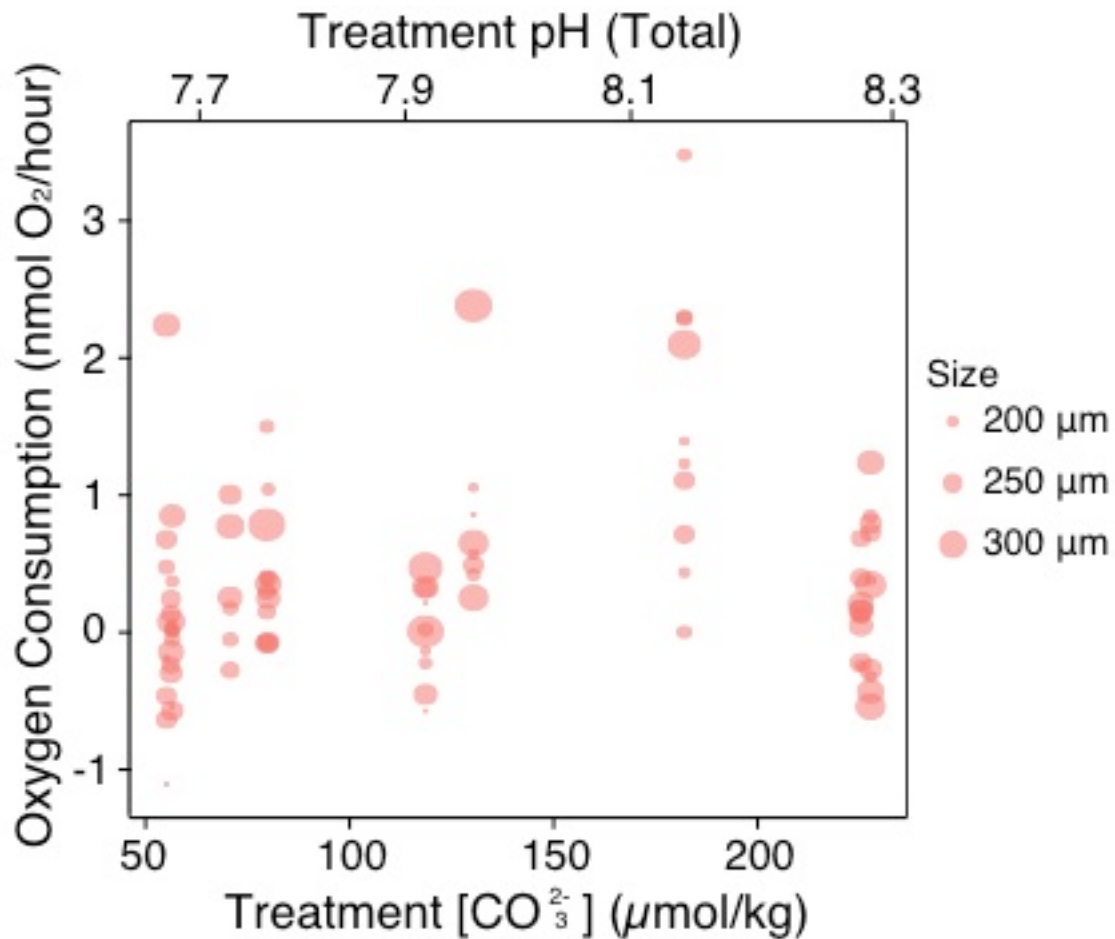


Fig S3. Oxygen consumption rates ($\text{nmol O}_2 \text{ foram}^{-1} \text{ hr}^{-1}$) across pH treatments, sized according to the longest shell dimension. Shell length has been shown to relate to biomass in both planktic (7) and benthic foraminifera (8, 9) and with protein biomass in planktic foraminifera (10), and as such, maximum shell length from post-respirometry observations was used to normalize oxygen consumption rates to account for variation associated with approximate biomass ($\text{pmol O}_2 \text{ foram}^{-1} \text{ hr}^{-1} \mu\text{m}^{-1}$). Here, un-normalized oxygen consumption rates ($\text{nmol O}_2 \text{ foram}^{-1} \text{ hr}^{-1} \mu\text{m}^{-1}$) are shown, sized according to the longest shell dimension.

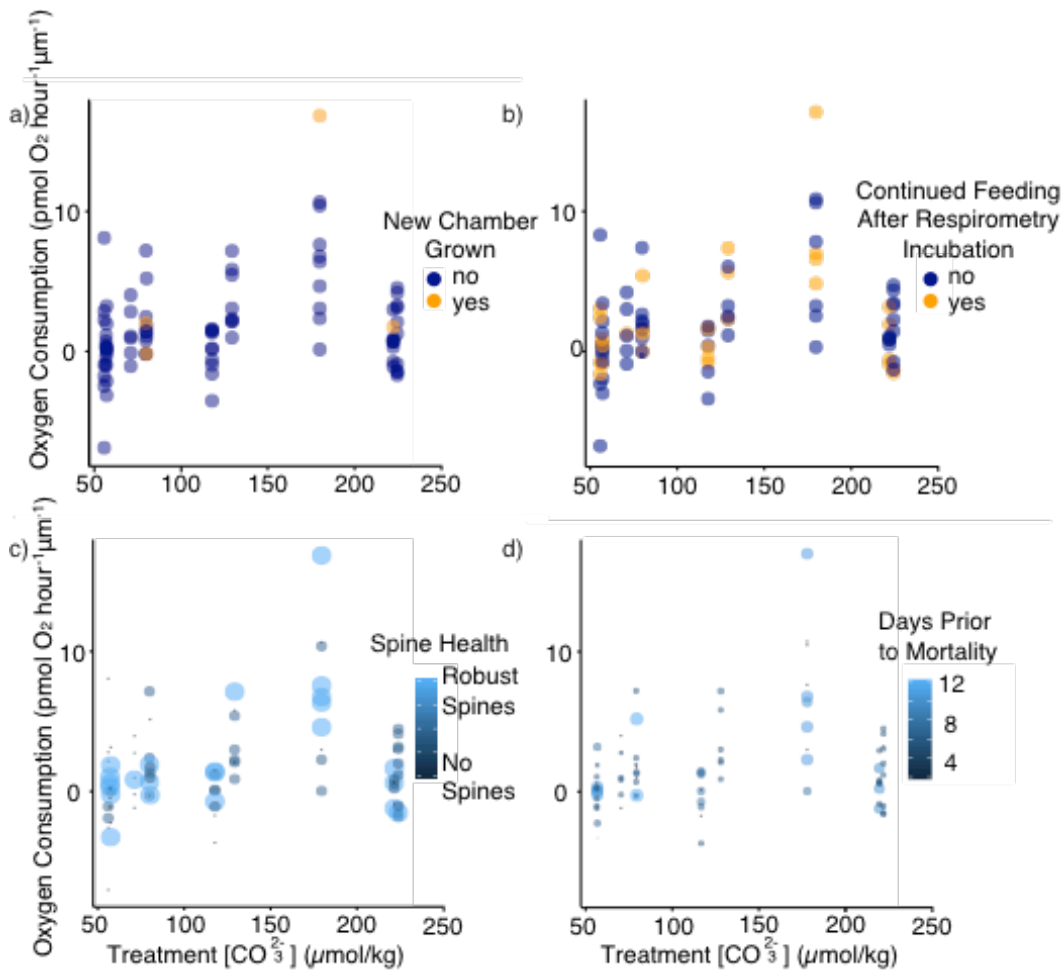


Fig. S4 The relationship between oxygen consumption and other observed traits in *G. bulloides* relative to [CO₃²⁻]. (A) Individuals that added a new chamber during their respirometry incubation are highlighted in orange. (B) Individuals that were able to feed the day following respirometry incubation (thus both healthy and not pre-gametogenic) are highlighted in orange. (C) Individuals with an extensive spine array at the time of incubation are highlighted in lighter blue. (D) The number of total days in culture before death or gametogenesis, with individuals having longer culture lifespans in lighter blue.

Water Chemistry

Water for each treatment condition was prepared ahead of collection. Individual *G. bulloides* were immediately isolated from plankton tows and randomly assigned to a target pH treatment (pH ~ 7.5, 7.7, 8.0, or 8.3) prior to the initiation of experiments. Limited abundance in tow material prevented all four treatments from being run

simultaneously. Consequently, the experimental conditions were replicated in non-repeating pairs until all four targets had been run at least twice. At the start of each experiment, foraminifera were placed directly into 25 mL vials, sealed without headspace, and filled with treatment water filtered to 0.6 μm containing fluorescent calcein dye. A regression of measured conditions was used in most analyses, through, to assess such categorical traits (i.e. spine regrowth, chamber addition), the proportion of individuals with a trait are reported relative to the measured average condition within each of four pH groups, averaging pH_T 7.54 (SE \pm 0.02), 7.69 (SE \pm 0.01), 8.02 (SE \pm 0.02) and 8.27 (SE \pm 0.01), respectively.

To monitor carbonate chemistry conditions throughout each run, a new container of treatment water was opened for each water change and measurements of pH_T and total alkalinity were performed both upon opening and after filling vials. pH_T was measured using a spectrophotometric method based on *m*-cresol purple (10) (UV-1800 UV-VIS spectrophotometer, Shimadzu Corp., Kyoto, Japan) (\pm 0.03 pH_T). Total alkalinity was measured using an automated, open-cell potentiometric titration (12) with a Metrohm 809 Titrando titrator and a Metrohm 6.0262.100 pH probe (Metrohm AG, Herisau, Switzerland) (\pm 4.2 $\mu\text{mol/L}$). Titrations were performed using a titrant of 0.025N HCl in a matrix of NaCl (33.5 PSU) and were standardized using certified references materials from A. Dickson laboratory (Scripps Institute of Oceanography). Temperature, salinity, pH_T , and total alkalinity were used to calculate the complete suite of carbonate chemistry parameters for each water change, using CO2CALC (13), with CO_2 constants K_1 , K_2 (14) and pH_T (mol kg-SW^{-1}).

Defining Foraminiferal Life Stage

We used established understandings of foraminiferal life history to guide our observations. This included the important observation that the planktic foraminiferal life cycle ends in sexual reproduction and the release of gametes, with the parent cytoplasm being nearly or entirely consumed or converted in the process (15). Individuals were photographed every other day in culture and observed every day for the presence and robustness of spines, extended rhizopodia, cytoplasm color, and the degree to which cytoplasm filled the youngest chambers. Individual *G. bulloides* generally completed their life cycle within 3-10 days of collection. 1-2 days prior to gametogenesis, *G. bulloides* began to show “pre-gametogenic” characteristics, including change of cytoplasm color from brown/gold to white, rhizopodial shortening and decreased “streaming”, and cessation of feeding. In some cases foraminifera died without completing their life cycle and “death” was defined as the inability to feed during two consecutive feedings accompanied by an absence of rhizopodial activity without pre-gametogenic characteristics. Foraminifera that did not feed but appeared to be pre-gametogenic were subject to continued observations, imaging, and water changes until gamete release.

Incubation in Calcein

The fluorescent compound, calcein (Sigma-Aldrich, Co., St. Louis, MO, USA), was used to label foraminiferal calcite grown under treatment conditions. Calcein binds to free calcium ions and is incorporated in calcite minerals during shell formation from seawater.

Calcein was shown to be non-toxic to foraminifera during short (<5 week exposures) at concentrations <20 mg/L (16, 17) and has since been used in additional experiments e.g. (18, 19). All foraminifera were incubated in treatment seawater containing a final calcein concentration of 10 mg/L. Calcein addition was found to cause a small decrease on final seawater pH (< 0.02), which given the slightness of this change and consistency of calcein addition across treatments was not chemically corrected. When the number of days spent in culture, and therefore in calcein was included in initial statistical models of calcification relative to pH, there was no significant interaction between time and either carbonate chemistry parameter.

Analogous uses of calcein as a quantitative or semi-quantitative measurement of calcite bound calcite has been undertaken by many previous authors, for example as a quantitative metric of coral calcification (20) or semi-quantitatively for *in vitro* analyses (21). Results here are presented as semi-quantitative, as the relationship between calcification and calcein uptake under these conditions in foraminiferal calcite could not be quantified. Our semi-quantitative analysis of bound calcein was carried out using the Metamorph software, by defining the limits of the foraminiferal shell from images taken using an epifluorescent microscope. An upper and lower threshold of brightness consistent across all images was defined, and then the relative brightness or intensity of each pixel could be determined. The measure reported, of Average Pixel Intensity, was the average intensity of each pixel of our image defined as shell.

Predicting Changes in the 'Rain Ratio'

In our estimations of foraminiferal contribution to ‘rain ratio’ changes, we used previously published records of foraminiferal CaCO₃, total CaCO₃, and POC flux below 1000 m (Table 2), along with an approximation that each individual foraminifera contains on average 5 µg CaCO₃ and 1 µg POC. We furthermore assumed that our calcification reduction results are representative of all foraminifera at pH <8.0, and that absence of spine repair is a reasonable measure of pre-reproductive mortality and rapid export. Thus, foraminifera shells would contain on average 38% less CaCO₃. If a spine loss event occurs (we estimate this at 25% likelihood within a lifetime), 70% of foraminifera would fail to recover, resulting in export of POC associated with cytoplasm. We also assume no change in “background” mortality, such as that of thinly calcified sub-adults, for which remobilization of both organic and inorganic carbon likely occurs in the upper water column, and thus would not impact export.

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