

# Molecular Insight and Mode of Inhibition of $\alpha$ -Glucosidase and $\alpha$ -Amylase by Pahangensin A from *Alpinia pahangensis* Ridley

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The inhibition of carbohydrate-hydrolysing enzymes in human digestive organs is crucial in controlling blood sugar levels, which is important in treating type 2 diabetes. In the current study, Pahangensin A (**1**), a bis-labdanic diterpene previously characterized in the rhizomes of *Alpinia pahangensis* Ridley, was identified as an active dual inhibitor for  $\alpha$ -amylase ( $IC_{50} = 114.80 \mu M$ ) and  $\alpha$ -glucosidase ( $IC_{50} = 153.87 \mu M$ ). This is the first report on the dual  $\alpha$ -amylase and  $\alpha$ -glucosidase inhibitory activities of a bis-labdanic diterpene. The Lineweaver-Burk plots of compound **1** indicate that it is a mixed-type inhibitor with regard to both enzymes. Based on molecular docking studies, compound **1** docked in a non-active site of both enzymes. The dual inhibitory activity of compound **1** makes it a suitable natural alternative in the treatment of type 2 diabetes.

**Keywords:** *Alpinia pahangensis* Ridley; Pahangensin A; bis-labdanic diterpene; dual-inhibitor;  $\alpha$ -glucosidase;  $\alpha$ -amylase

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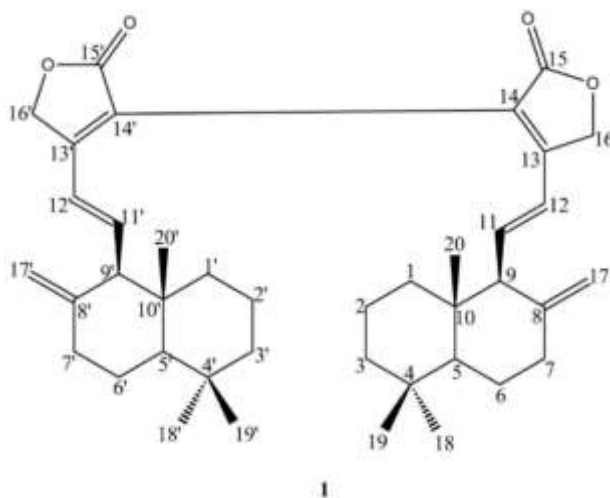
## Introduction

Diabetes mellitus is a metabolic disorder characterized by high plasma glucose levels, classified as either type 1 or 2. Type 1, or insulin-dependent diabetes, is due to failure of the pancreas to secrete insulin, while type 2, or non-insulin-dependent diabetes, is the result of insufficient insulin production. Type 2 diabetes receives more attention than type 1 diabetes because it is considered to be preventable. The former is caused by an imbalance between blood sugar absorption and insulin secretion. Post-prandial hyperglycaemia plays an important role in the development of type 2 diabetes. Controlling plasma glucose levels is essential for delaying or preventing type 2 diabetes.<sup>[1]</sup>

$\alpha$ -Amylase and  $\alpha$ -glucosidase inhibitors belong to a class of anti-diabetic drugs that control the sudden rise in blood sugar levels after meals.  $\alpha$ -Amylase is in a class of enzymes that hydrolyse polysaccharides to oligosaccharides, while  $\alpha$ -glucosidase catalyses the final step in carbohydrate hydrolysis to release the absorbable monosaccharides.<sup>[2]</sup> Both of these carbohydrate-hydrolysing enzymes are secreted in the small intestine, while  $\alpha$ -amylase is also found in the saliva.<sup>[1]</sup> Thus, the inhibition of these enzymes plays an important role in the management of diabetic complications, particularly type 2 diabetes.<sup>[2-3]</sup> Antidiabetic drugs such as acarbose, miglitol and voglibose act by inhibiting the activities of  $\alpha$ -amylase and  $\alpha$ -glucosidase. Although these drugs are effective, continuous use may lead to undesirable side effects, which include liver toxicity and adverse gastrointestinal symptoms.<sup>[4]</sup> Hence, discovering new potential inhibitors from plants may offer reduced side effects during long-term use.

A large number of traditional medicinal plants and plant-derived constituents, such as terpenes, alkaloids, curcuminoids, anthocyanins, flavonoids, quinones, phenols, phenylpropanoids, acylphenols and dimeric acylphenols, are known to exhibit  $\alpha$ -amylase and  $\alpha$ -glucosidase inhibitory activities.<sup>[2] [5-8]</sup> However, to date, bis-labdanic diterpenes have never been investigated for their potential to inhibit these enzymes. Therefore, herein, we report the dual inhibiting potentials of Pahangensin A (**1**) (*Figure 1*), a bis-labdanic diterpene previously

isolated and characterized from the rhizomes of *Alpinia pahangensis* Ridley, against  $\alpha$ -amylase and  $\alpha$ -glucosidase.<sup>[9]</sup> Kinetic studies were subsequently carried out on compound **1** to determine its mode of inhibition against each enzyme. Additionally, molecular docking was performed to provide insights into the binding interactions between compound **1** and the enzymes.



**Figure 1:** Structure of compound **1**

## Results and Discussion

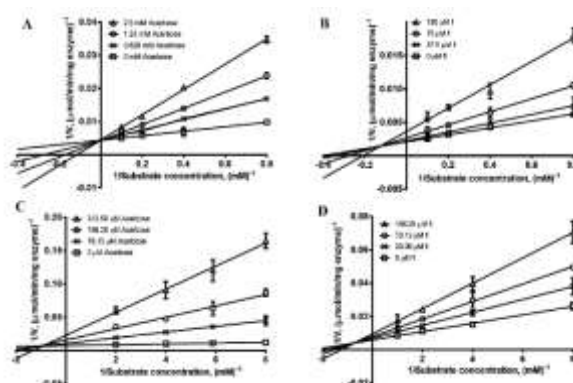
The  $IC_{50}$  values of compound **1** and the positive control, acarbose are given in Table 1. The inhibition potential of compound **1** against the carbohydrate-hydrolysing enzymes was higher compared to acarbose, thus making compound **1** a more effective inhibitor than the positive control.

**Table 1:** Carbohydrate hydrolyzing enzymes inhibition of compound **1** and acarbose.

Compounds	$IC_{50}$ ( $\mu M$ ) <sup>a</sup>	
	$\alpha$ -Amylase	$\alpha$ -Glucosidase
<b>1</b>	114.80 $\pm$ 8.49	153.87 $\pm$ 9.28
<b>Acarbose</b>	280.47 $\pm$ 10.19	1449.67 $\pm$ 46.52

<sup>a</sup> Data presented as Mean  $\pm$  SD (n = 3).

As illustrated by Lineweaver-Burk plot analyses (*Figure 2*), compound **1** displayed mixed-mode inhibition against  $\alpha$ -amylase and  $\alpha$ -glucosidase, as indicated by the intersect in the second quadrant of each respective plot. This finding suggested that compound **1** will be able to bind to the free enzymes and enzyme-substrate complexes.<sup>[5]</sup> [10-11]



**Figure 2:** Lineweaver-Burk plots for (A and B)  $\alpha$ -glucosidase and (C and D)  $\alpha$ -amylase in the presence of acarbose and compound **1**, respectively. Mode of inhibition was determined by analysing the Lineweaver–Burk plots, which was obtained from the Michaelis–Menten kinetics (*Supp 1*).

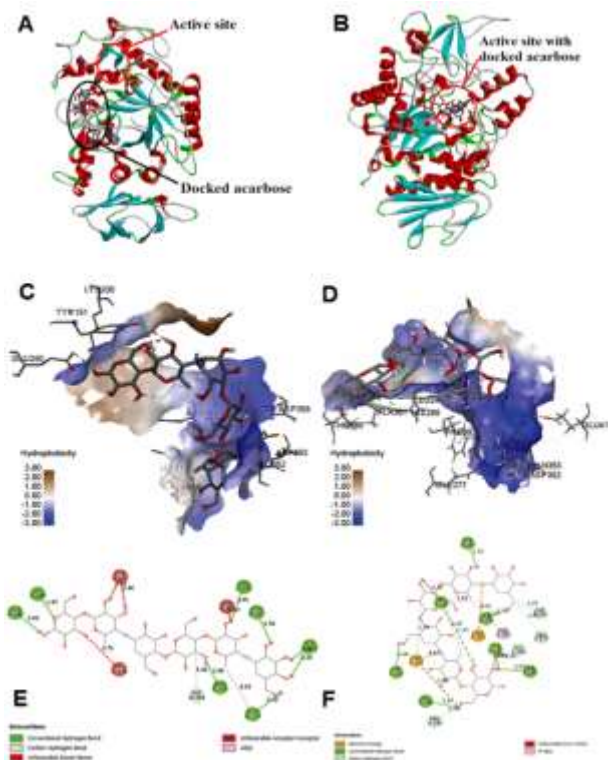
The free enzyme inhibition of compound **1**, which was smaller than its enzyme-substrate complex inhibition constants (*Table 2*), enabled us to postulate that compound **1** has a higher affinity towards the free enzymes rather than the enzyme-substrate complexes, which is typical for mixed-mode inhibitors.<sup>[11]</sup> In the current study, acarbose was identified as a competitive inhibitor of  $\alpha$ -glucosidase and a mixed-mode inhibitor of  $\alpha$ -amylase, which is consistent with previous reports.<sup>[5] [12]</sup>

**Table 2:** Mode of inhibition and inhibitory constants of compound **1** and acarbose against  $\alpha$ -amylase and  $\alpha$ -glucosidase.

Compound	$\alpha$ -Amylase			$\alpha$ -Glucosidase		
	Mode of Inhibition	Ki <sub>1</sub> ( $\mu$ M)	Ki <sub>2</sub> ( $\mu$ M)	Mode of Inhibition	*Ki <sub>1</sub> ( $\mu$ M)	*Ki <sub>2</sub> ( $\mu$ M)
<b>1</b>	Mixed-mode	47.66 $\pm$ 8.24	68.78 $\pm$ 11.75	Mixed-mode	124.77 $\pm$ 12.16	140.93 $\pm$ 20.49
<b>Acarbose</b>	Mixed-mode	26.04 $\pm$ 2.67	61.18 $\pm$ 10.45	Competitive	568.47 $\pm$ 107.04	-

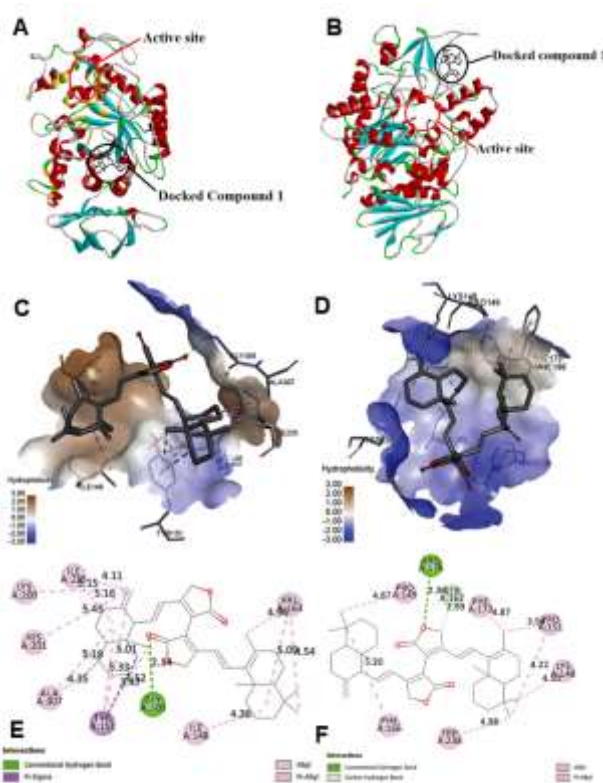
\*Ki<sub>1</sub> and Ki<sub>2</sub> indicate the affinity of the inhibitors to the free enzyme and the enzyme-substrate complex, respectively.

The crystal structures of  $\alpha$ -amylase (PDB ID: 1OSE)<sup>[13]</sup> and  $\alpha$ -glucosidase (PDB ID: 3A4A)<sup>[14]</sup> have identified active sites and based on the molecular docking studies (*Figures 3 and 4*), acarbose and compound **1** showed favourable interactions with residues of the enzyme's binding pockets. Acarbose docked in the active site of  $\alpha$ -glucosidase, but only interacted with a non-active site in  $\alpha$ -amylase. This finding in turn provided supportive data of our experimental results that acarbose inhibited  $\alpha$ -glucosidase and  $\alpha$ -amylase in competitive and mixed-mode manners, respectively. As shown in Figure 3, the docked conformation of acarbose revealed the formation of two hydrogen bonds with the important active site residues of  $\alpha$ -glucosidase, GLU277 and ASP352. On the other hand, compound **1** docked into the non-active site of both  $\alpha$ -glucosidase and  $\alpha$ -amylase (*Figure 4*), which is in agreement with our experimental data suggested that compound **1** was a mixed-mode inhibitor. In contrast to acarbose, compound **1** has only one hydrogen bond with  $\alpha$ -amylase (GLY306). The majority of the interactions involved alkyl-alkyl and pi-alkyl interactions with ILE235, LYS 200, HIS201, ALA307, ILE148 and VAL163. Similar types of interactions were also observed with  $\alpha$ -glucosidase, with only one hydrogen bond to ARG176 and the majority being alkyl-alkyl and pi-alkyl interactions (PRO149, PHE166, TRP238, PHE173, PRO151 and LYS148). The results also revealed better binding energies for compound **1** (-9.8 and -11.0 kcal/mol) compared to acarbose (-8.8 and -8.4 kcal/mol) for  $\alpha$ -amylase and  $\alpha$ -glucosidase, respectively.



**Figure 3:** The docked acarbose in (A, C and E)  $\alpha$ -amylase (PDB: 1OSE)<sup>[13]</sup> and (B, D and F)  $\alpha$ -glucosidase (PDB: 3A4A)<sup>[14]</sup> is shown in 3-D and 2-D. The results show acarbose is docked into the  $\alpha$ -glucosidase's active

site with 2 important active site residues, Glu277 and Asp352 via hydrogen bonding. However, acarbose only docked into the non-active site of the  $\alpha$ -amylase enzyme.



**Figure 4:** The docked compound 1 in (A, C and E)  $\alpha$ -amylase (PDB: 1OSE)<sup>[13]</sup> and (B, D and F)  $\alpha$ -glucosidase (PDB: 3A4A)<sup>[14]</sup> is shown in 3-D and 2-D. The results show compound 1 is docked into the non-active site of the  $\alpha$ -amylase and  $\alpha$ -glucosidase enzymes.

## Conclusions

Diabetes mellitus is a metabolic disorder that leads to comorbidities such as neuropathy, nephropathy, retinopathy and cardiovascular diseases.<sup>[15]</sup> Control of postprandial glucose levels is one way to reduce postprandial spikes.<sup>[16]</sup> Hence, controlling the activities of  $\alpha$ -amylase and  $\alpha$ -glucosidase seems to be useful in preventing a postprandial spike.<sup>[16]</sup> In the present study, compound 1 exhibited stronger inhibitory effects towards  $\alpha$ -amylase and  $\alpha$ -glucosidase in comparison to acarbose. Molecular docking revealed that the interactions of compound 1 with both enzymes involved mainly hydrophobic interactions (alkyl-alkyl and pi-alkyl). The weak inhibition of  $\alpha$ -glucosidase by acarbose is known to cause adverse effects, such as flatulence from the breakdown of undigested carbohydrates by intestinal bacteria.<sup>[15]</sup> Therefore, a compound with dual inhibition may elicit its pharmacological effects while minimizing its side effects.<sup>[18]</sup> The results presented here indicate that compound 1 is a promising  $\alpha$ -amylase and  $\alpha$ -glucosidase inhibitor for further development as a potential alternative to acarbose. Since, the enzymes used are non-human types, compound 1 can be a lead compound for medicinal chemists to further develop analogues, which may potentially inhibit human  $\alpha$ -amylase and  $\alpha$ -glucosidase.

## Experimental Section

### Chemicals and reagents

Analytical and preparative TLC were carried out on Merck 60 F<sub>254</sub> silica gel plates (absorbent thickness: 0.25 and 0.50 mm). Column chromatography was performed using silica gel (Merck 230-400 mesh, ASTM). The IR spectrum was recorded using a Perkin-Elmer Spectrum 400 FT-IR Spectrometer. NMR spectra were recorded in CDCl<sub>3</sub> (Merck, Germany) with tetramethylsilane as an internal standard using a JEOL ECA 400 MHz NMR spectrometer. The LCMS-IT-TOF spectrum was recorded on a UFLC Shimadzu Liquid Chromatograph with an SPD-M20A diode array detector coupled to an IT-TOF mass spectrometer. The UV spectrum was recorded using

a Shimadzu 1650 PC UV-Vis Spectrophotometer. All solvents were of analytical grade and were distilled prior to use.<sup>[9]</sup>

The  $\alpha$ -glucosidase enzyme (EC 3.2.1.20) was obtained from the yeast *Saccharomyces cerevisiae*, while the  $\alpha$ -amylase enzyme (EC 3.2.1.1) was obtained from the porcine pancreas. All reagents, chemicals and enzymes were purchased from Sigma-Aldrich Co. LLC. (St. Louis, MO, USA).

#### *Plant material*

*A. pahangensis* was collected from Pahang in 2011. The plant was identified by one of our authors, Professor Halijah Ibrahim, and a voucher specimen (KU001) was deposited with the University of Malaya herbarium.<sup>[9]</sup>

#### *Extraction, isolation and characterization of compound 1*

Dried powdered rhizomes (1.0 kg) of *A. pahangensis* were extracted with CH<sub>2</sub>Cl<sub>2</sub> (5.0 L, 2 $\times$ ) followed by MeOH (5.0 L, 2 $\times$ ) at room temperature, giving 15.55 g and 19.04 g of extracts, respectively. The CH<sub>2</sub>Cl<sub>2</sub> extract was fractionated over a silica gel column eluting with mixtures of hexane:CH<sub>2</sub>Cl<sub>2</sub> in proportions of 50:50 (v/v, 450 mL), 25:75 (v/v, 550 mL), and 0:100 (v/v, 1950 mL), followed by mixtures of CH<sub>2</sub>Cl<sub>2</sub>:MeOH in proportions of 95:5 (v/v, 400 mL) and 85:15 (v/v, 400 mL) to yield 10 main fractions (AP1-AP10). Repeated preparative TLC of fraction AP 4 (1.00 g; eluted with hexane:CH<sub>2</sub>Cl<sub>2</sub> [25:75 v/v]) with hexane:EtOAc (95:5 v/v) led to the isolation of **1** (20.0 mg).<sup>[9]</sup> The purity of Pahangensin A is 98.2%, determined from the peak area of the LCMS-DAD-IT-TOF analysis (Supp 2).

**Pahangensin A (= bis-labda-8(17),11,13-trien-16,15-olide; 1):** Yellow oil, UV  $\lambda_{\max}$  (MeOH) nm (log  $\epsilon$ ): 205 (7.58), and 281 (11.80); IR  $\lambda_{\max}$  (NaCl) cm<sup>-1</sup>: 3069, 2928, 2863, 2852, 1738, 1646, 1459, 1367, 889; LCMS-IT-TOF m/z : 599.4071 [M + H]<sup>+</sup> (calcd. for C<sub>40</sub>H<sub>55</sub>O<sub>4</sub> 599.4095); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  0.84 (3H, s, H-20/H-20'),  $\delta$  0.85 (3H, s, H-19/H-19'),  $\delta$  0.90 (3H, s, H-18/H-18'),  $\delta$  1.05 (1H, m, H-1 $\alpha$ /H-1' $\alpha$ ),  $\delta$  1.13 (1H, dd, J = 12.4, 2.3 Hz, H-5 $\alpha$ /H-5' $\alpha$ ),  $\delta$  1.20 (1H, m, H-3 $\alpha$ /H-3' $\alpha$ ),  $\delta$  1.37-1.45 (4H, m, H-1 $\beta$ /H-1' $\beta$ , H-2 $\alpha$ /H-2' $\alpha$ , H-3 $\beta$ /H-3' $\beta$ , H-6 $\alpha$ /H-6' $\alpha$ ),  $\delta$  1.52 (1H, m, H-2 $\beta$ /H-2' $\beta$ ),  $\delta$  1.73 (1H, m, H-6 $\beta$ /H-6' $\beta$ ),  $\delta$  2.10 (1H, m, H-7 $\alpha$ /H-7' $\alpha$ ),  $\delta$  2.44 (1H, m, H-7 $\beta$ /H-7' $\beta$ ),  $\delta$  2.48 (1H, brd, J = 11.0 Hz, H-9 $\alpha$ /H-9' $\alpha$ ),  $\delta$  4.42 (1H, d, J = 1.2 Hz, H-17a/H-17'a),  $\delta$  4.78 (1H, d, J = 1.2 Hz, H-17b/H-17'b),  $\delta$  4.88 (2H, brs, H-16/H-16'),  $\delta$  6.00 (1H, dd, J = 16.5, 10.0 Hz, H-11/H-11'),  $\delta$  6.36 (1H, d, J = 16.5 Hz, H-12/H-12'); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  15.3 (C-20/C-20'),  $\delta$  19.3 (C-2/C-2'),  $\delta$  22.2 (C-19/C-19'),  $\delta$  23.5 (C-6/C-6'),  $\delta$  33.8 (C-4/C-4', C-18/C-18'),  $\delta$  36.9 (C-7/C-7'),  $\delta$  39.6 (C-10/C-10'),  $\delta$  41.2 (C-1/C-1'),  $\delta$  42.4 (C-3/C-3'),  $\delta$  54.8 (C-5/C-5'),  $\delta$  62.2 (C-9/C-9'),  $\delta$  68.2 (C-16/C-16'),  $\delta$  108.5 (C-17/C-17'),  $\delta$  120.9 (C-12/C-12'),  $\delta$  127.5 (C-13/C-13'),  $\delta$  135.5 (C-11/C-11'),  $\delta$  135.7 (C-14/C-14'),  $\delta$  149.6 (C-8/C-8'),  $\delta$  171.3 (C-15/C-15').<sup>[9]</sup>

#### *$\alpha$ -Glucosidase inhibitory assay*

Compound **1** was initially dissolved in dimethyl sulfoxide (DMSO) and further diluted with the respective assay buffers to yield a final concentration of 15% DMSO in buffer. Acarbose was directly dissolved in the respective assay buffers. Negative controls with 15% DMSO in the respective assay buffers were included to account for the effect of the solvent. A total volume of 40  $\mu$ L of compound **1** at different concentrations (0.02 – 2.5 mM) was pre-incubated with 80  $\mu$ L of potassium phosphate buffer (pH 6.8) containing 67 mM potassium phosphate and 2.0 units/mL  $\alpha$ -glucosidase in a 96-well plate for 10 minutes at 25 °C. Subsequently, 40  $\mu$ L of 5 mM p-nitrophenyl- $\alpha$ -D-glucopyranoside solution (p-NPG) in potassium phosphate buffer was introduced and incubated for another 10 minutes. After incubation, 60  $\mu$ L of 100 mM sodium carbonate was added to terminate the reaction, and the absorbance was measured at a wavelength of 415 nm using a microplate reader (Infinite 200, Tecan). Acarbose was utilized as a positive control. The concentration of compound **1** resulting in 50 % inhibition of the  $\alpha$ -glucosidase activity (IC<sub>50</sub>) was determined using the GraphPad Prism 5 statistical package (GraphPad Software, USA). All data are expressed as the mean  $\pm$  standard deviation of triplicate determinations.<sup>[5] [19]</sup>

#### *$\alpha$ -Amylase inhibitory assay*

Porcine  $\alpha$ -amylase (1.0 unit/mL) in a volume of 100  $\mu$ L was pre-incubated with 10  $\mu$ L of compound **1** at different concentrations (0.02 – 2.5 mM) for 10 minutes at 25 °C. Next, 100  $\mu$ L of 0.5 % (w/v) starch solution containing 0.5 % (w/v) potato starch in 20 mM sodium phosphate buffer (pH 6.9) with 6.7 mM sodium chloride was added to the solution and incubated for 8 minutes at 25 °C. Then, 100  $\mu$ L of DNS colour agent solution containing 96 mM 3,5-dinitrosalicylic acid solution and 5.31 M sodium potassium tartrate in 2 M sodium hydroxide was added into the solution and incubated again for 15 minutes at 85 °C. After incubation, the absorbance was measured at a wavelength of 540 nm using a microplate reader (Infinite 200, Tecan). The experiment was also carried out using the positive control, acarbose. Similarly, the concentration of compound **1** resulting in 50 % inhibition of the  $\alpha$ -amylase activity (IC<sub>50</sub>) was determined using the GraphPad Prism 5 statistical package (GraphPad Software, USA). All data are expressed as the mean  $\pm$  standard deviation of triplicate determinations.<sup>[20]</sup>

## Mode of $\alpha$ -glucosidase and $\alpha$ -amylase inhibition

The inhibition modes of the compound **1** and acarbose samples against  $\alpha$ -glucosidase and  $\alpha$ -amylase were measured at different concentrations of their respective substrates (p-NPG or potato starch) in the presence or absence of samples at various concentrations. The mode of inhibition was obtained by Lineweaver-Burk plot analysis and calculated using Michaelis-Menten kinetics. A Dixon plot and the Y-intercept of the Lineweaver-Burk plot versus [inhibitor] were used to determine the inhibition constants ( $K_i$ ). All data are expressed as the mean  $\pm$  standard deviation of triplicate determinations.<sup>[5] [19] [21]</sup> The pathlength of the microplate reader (Infinite 200, Tecan) is 0.511 cm and extinction coefficient of compound **1** is 2.74 L mol<sup>-1</sup> cm<sup>-1</sup>.

## Molecular docking

Molecular docking was performed to investigate the binding mode between the samples (compound **1** and acarbose) and the enzymes ( $\alpha$ -amylase and  $\alpha$ -glucosidase) using Autodock Vina 1.1.2. The 3-D structures of compound **1** and acarbose were drawn and energy minimized using ChemBio3D Ultra 12.0. The crystal structures of  $\alpha$ -amylase (PDB ID: 1OSE)<sup>[13]</sup> and  $\alpha$ -glucosidase (PDB ID: 3A4A)<sup>[14]</sup> were prepared using AutoDockTools 1.5.6 to remove water molecules and ligands (acarbose,  $\alpha$ -D-glucose and  $\beta$ -D-glucose) and to add missing hydrogens. Although the active sites are identified in the crystal structures, the enzymes were enclosed in a grid box with 1.00 Å spacing, and the search exhaustiveness value was set to 100 to perform blind docking encompassing the whole enzyme structure. The top best-scoring pose from the AutoDock Vina results was analysed using Discovery Studio visualizer 4.5.<sup>[22]</sup>

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## Author Contribution Statement

K.Y. Loo conducted the inhibitory assays, kinetic studies and molecular docking studies and analysed the data. Y. Sivasothy isolated and elucidated the structure of compound **1**. K.H. Leong, K.Y. Loo and Y. Sivasothy wrote the manuscript. K.H. Leong and K. Awang designed the experiments. H. Ibrahim collected and identified the plant material.

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