



**Ptaquiloside**  
**an environmental hazard?**

Rasmussen, L. H.

*Publication date:*  
2003

*Document version*  
Publisher's PDF, also known as Version of record

*Citation for published version (APA):*  
Rasmussen, L. H. (2003). *Ptaquiloside: an environmental hazard?* Copenhagen: The Royal Veterinary and Agricultural University, Chemistry Department.

# PTAQUILOSIDE – AN ENVIRONMENTAL HAZARD?



Occurrence and fate of a Bracken (*Pteridium* sp.) toxin in terrestrial environments

**Ph.D.-thesis by Lars Holm Rasmussen**

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July 2003

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in terrestrial environments



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**Examination date:**

October 3, 2003

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## RESUMÉ

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Ørnebregnen (*Pteridium aquilinum* (L.) Kuhn) er en af de hyppigst forekommende planter i Verden. Bregnen findes på alle kontinenter undtagen Antarktis. Ørnebregnen er et almindeligt ukrudt på landbrugsjord, i den primære bush eller i lysbrønde inde i skove. Ørnebregnen indeholder en lang række sekundære planteindholdsstoffer, hvoraf nogle er giftige overfor mennesker og andre levende organismer. Det mest fremtrædende af disse stoffer er det norsesquiterpene glykosid ptaquilosid.

Ptaquilosid forårsager en lang række sygdomme hos dyr strækkende sig fra induceret thiamin-mangel til kræft. Specielt kræft i urinblæren hos køer, kendt som *Bovin Enzootisk Hæmaturi*, forekommer hyppigt i nogle områder af Verden. Ptaquilosid er også mistænkt for at forårsage mave- og spiserørskræft hos mennesker, idet stoffet nemt overføres til mælk hos bregnespisende køer, og idet ptaquilosid muligvis kan udvaskes fra Ørnebregnebevoksninger til grund- og drikkevand. Formålet med denne afhandling har været at undersøge forekomsten af ptaquilosid i terrestriske økosystemer, samt at undersøge ptaquilosids stabilitet og mobilitet i jord for at svare på spørgsmålet: *Udgør ptaquilosid en miljømæssig fare?*

Ptaquilosid findes i koncentrationer op til 15,100  $\mu\text{g g}^{-1}$  i alle bregnedele af alle Ørnebregne-varieteter. Indholdet er normalt størst i de unge bregneblade. Indholdet falder som væksten skrider frem. Indholdet i rhizomer er nogenlunde inverst relateret til indholdet i bladene. Det maksimale indhold på op til 7,050  $\mu\text{g g}^{-1}$  forekommer i rhizomerne når væksten af bladene stopper. Indholdet af ptaquilosid bregne-bevoksninger imellem er meget variabelt. Overordnet set skyldes ptaquilosid-variationen genetiske forhold, mens indvirkning af vækst-forhold kan ses på lokalt niveau (eksempelvist topografi, næringsstofforsyning, lysintensitet o. lign.). Økologisk stress, såsom fysisk påvirkning og skygge, kan forårsage øget indhold af ptaquilosid i blade, mens kvælstof-

mangel kan øge indholdet i rhizomerne. Forskellige studier af variationen i ptaquilosid-indhold har dog leveret modstridende resultater, og yderligere undersøgelser er nødvendige før der kan opstilles en generel model for forekomsten af ptaquilosid i Ørnebregne-bevoksninger.

Nedbør kan udvaske ptaquilosid fra Ørnebregne-bevoksninger til jordbunden. Ptaquilosid er fundet i litter og overjordshorisonter i mængder op til  $8.5 \mu\text{g g}^{-1}$ . Stabiliteten og mobiliteten af ptaquilosid i jord afhænger af jordbundens egenskaber og de klimatiske forhold, idet at: 1) Ptaquilosid er relativt stabilt ved pH 4 til 8; 2) Ptaquilosid kan være beskyttet mod nedbrydning ved svag sorption til uorganiske partikler samt organisk materiale i jorden; og 3) Ptaquilosid nedbrydes langsommere ved lav temperatur i jorden. Under nogle forhold kan ptaquilosid derfor være meget mobilt (som nitrat) og stabilt, hvorved ptaquilosid kan nå grund- og drikkevand. Koncentrationen af ptaquilosid i jordvand blev ikke bestemt direkte i overjordshorisonter, men ligevægtskoncentrationer på op til  $400 \mu\text{g L}^{-1}$  blev estimeret ud fra ptaquilosids sorptionsegenskaber. Disse koncentrationer er i overensstemmelse med koncentrationer på op til  $7 \mu\text{g L}^{-1}$  målt i jordvæske ekstraheret fra 90 cm dybde. Ptaquilosid i koncentrationer op til  $45 \mu\text{g L}^{-1}$  er fundet i drikkevand ved to lejligheder! Sammenlignet med den *tolerable concentration* af ptaquilosid i drikkevand på  $0.002 \mu\text{g L}^{-1}$ , er der tale om meget høje koncentrationer.

Selvom ptaquilosid ser ud til at udgøre en miljømæssig fare, så er den overordnede risiko for mennesker formentlig relativt lille, da ptaquilosidforurening kun vil berøre mennesker der lever og arbejder i områder med mange Ørnebregner. Hvor Ørnebregnen forekommer udbredt, er det dog vigtigt at udvise agtpågivenhed for at minimere risikoen for ptaquilosidforurening af vandressourcer såvel som af mælk.

## ABSTRACT

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Bracken fern (*Pteridium aquilinum* (L.) Kuhn) is one of the most common plant species on Earth. The fern is found on all continents except from Antarctica where it occurs as a common weed on agricultural lands, as part of the primary bush or below canopy openings inside forests. Bracken contains a wide range of secondary metabolites of which some are toxic towards humans and other living organisms. The most prominent of these compounds is the norsesquiterpene glucoside ptaquiloside.

Ptaquiloside causes a wide range of diseases in animals ranging from thiamine deficiency to cancer. Especially bovine urinary bladder cancer known as *Bovine Enzootic Haematuria* occurs widespread in some parts of the world. Ptaquiloside is also suspected of causing human gastric and oesophageal cancer as the carcinogen readily passes into bovine milk of cows browsing on Bracken and may leach from Bracken stands into drinking water supplies. The aim of this thesis has been to monitor the occurrence of ptaquiloside in terrestrial ecosystems and to investigate ptaquiloside stability and mobility in soils, hereby seeking answers to the question: *Is ptaquiloside an environmental hazard?*

Ptaquiloside is found in all fern compartments of all varieties of Bracken in amounts up to 15,100  $\mu\text{g g}^{-1}$ . The content is usually highest in the young fronds. The content decreases as growth proceeds. The content in rhizomes is almost inversely correlated with the content in the fronds. Maximum contents of up to 7,050  $\mu\text{g g}^{-1}$  occur in rhizomes when frond growth ceases. The content of ptaquiloside in Bracken stands is quite variable. On a large-scale, the content seems mainly to be caused by genetic variation, while influence of local growth conditions can be seen on a smaller scale (e.g. altitude, nutrient availability, light intensity etc.). Ecological stress, such as physical disturbance and shade may induce increased contents of ptaquiloside in fronds, while nitrogen-depletion of soils may raise the content found in rhizomes. However, many studies on the effect of growth conditions have been

contradictory and further investigations are needed before a general model for the ptaquiloside occurrence in Bracken stands can be established.

Ptaquiloside can be leached by rainwater from Bracken stands to the soil. Ptaquiloside is found in topsoil's and litter in amounts up to  $8.5 \mu\text{g g}^{-1}$ . The stability and mobility of ptaquiloside in soils depends on the soil properties and climatic conditions, as: 1) ptaquiloside is rather stable when pH is in the range 4 to 8; 2) ptaquiloside may be protected from degradation by weak sorption to inorganic soil particles and soil organic matter; and 3) ptaquiloside degrades slower at low soil temperatures. Hence, under certain soil conditions, ptaquiloside can be highly mobile (comparable to nitrate) and stable. Ptaquiloside may therefore reach ground- and drinking water. Soil solution concentrations were not measured directly in topsoil's, but equilibrium concentrations up to  $400 \mu\text{g L}^{-1}$  was estimated from the sorptive properties of ptaquiloside, which is in accordance with measured amounts of up to  $7 \mu\text{g L}^{-1}$  in soil water extracted from 90 cm depth. On two occasions, ptaquiloside has been found in drinking water in concentrations of up to  $45 \mu\text{g L}^{-1}$ ! Compared to the *tolerable concentration* of ptaquiloside in drinking water of  $0.002 \mu\text{g L}^{-1}$ , these concentrations are very high.

However, even though Bracken may seem like an environmental hazard, the overall risk to humans is likely rather low, as ptaquiloside will only affect people living or working in areas with many Bracken ferns. Where Bracken does occur, care must be taken to minimize the risk of ptaquiloside contamination of water sources as well as of bovine milk!

## PREFACE

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The thesis is composed of two parts: a general part thoroughly reviewing existing literature covering Bracken in general and ptaquiloside in particular. The review includes the work carried out by the author. The second part of the thesis comprises the following papers and submitted manuscripts:

- Rasmussen, L.H., Donnelly, E., Strobel, B.W. Holm, P.E. and Hansen, H.C.B. (2003) Ptaquiloside in Bracken (*Pteridium aquilinum* ssp. *aquilinum*) in Scotland. Contents in fronds, rhizomes, and litter. *Submitted to Journal of Chemical Ecology*.
- Rasmussen, L.H. and Hansen, H.C.B. (2003) Growth of Bracken in Denmark and the content of ptaquiloside in fronds, rhizomes and roots. *In T. Acamovic, C.S. Stewart, and T.W. Pennycott (Eds.). Poisonous Plants and Related Toxins. CAB International, Wallingford. In press.*
- Rasmussen, L.H., Jensen, L.S. and Hansen, H.C.B. (2003) Distribution of the carcinogenic terpene ptaquiloside in Bracken fronds, rhizomes (*Pteridium aquilinum*), and litter in Denmark. *Journal of Chemical Ecology* 29:3:771-778.
- Rasmussen, L.H., Kroghsbo, S., Frisvad, J.C. and Hansen, H.C.B. (2003) Occurrence of the carcinogenic Bracken constituent ptaquiloside in fronds, topsoils and organic soil layers in Denmark. *Chemosphere* 51:117-127.

- Rasmussen, L.H., Lauren, D.R. and Hansen, H.C.B. (2003) Sorption, degradation and mobility of ptaquiloside, a carcinogenic Bracken (*Pteridium* sp.) constituent, in the soil environment. *Submitted to Chemosphere*.
- Rasmussen, L.H., Lauren, D.R., Smith, B.L. and Hansen, H.C.B. (2003) Variation in the ptaquiloside content in Bracken from New Zealand. *Submitted to Journal of Chemical Ecology*.
- Rasmussen, L.H., Schmidt, B., Lauren, D.R., Olsen, C.E., Engelsen, S., and Hansen, H.C.B. (2003) A refined method for isolation of pure ptaquiloside from Bracken (*Pteridium aquilinum* (L.) Kuhn.). *Will be submitted to Journal of Agricultural and Food Chemistry in August 2003*.

Behind these seven papers are many hours of work spent in the laboratory, in the field, and numerous hours behind the computer. The work has been hard, but also inspiring and challenging.

Chemistry Department at the Royal Veterinary and Agricultural University (KVL) has been the base for my work, and I would like to express my sincere gratitude to all the people (faculty, staff and students) affiliated with the department, but especially to my friend and colleague Anne Louise Gimsing.

Almost 6 months passed as visiting scientist at HortResearch in Hamilton, New Zealand. The visit was challenging and invaluable to me, since my New Zealand supervisor Dr. Denis R. Lauren taught me how to prepare ptaquiloside in a much easier way than previously reported. Many other people assisted me in New Zealand, but the list would be too long if I should mention all.

A special thanks goes to my colleague Dr. Carlos Pinto, Servica de Desenvolvimento Agrario de Sao Miguel, Ponta Delgada, the Azores, who turned up to be an excellent guide to the beautiful nature at the Azores as well as a competent and enthusiastic scientist working with the toxicological and veterinary aspects of Bracken and ptaquiloside in dairy and cattle production.

Lots of other people have assisted me in the last four years of Bracken work – ranging from students at the KVL, farmers and foresters in Denmark, Scotland, the Azores, and

New Zealand to scientists at other departments and institutions than the ones mentioned above. All are thanked for their invaluable assistance.

Finally my deepest gratitude and thanks to Solveig Warnecke for her assistance during field work and travels, her patience during my long working hours, and her understanding and tolerance of my love to this challenging fern, the Bracken.

Frederiksberg, July 2003

Lars Holm Rasmussen

This thesis is the 1<sup>st</sup> edition. A 2<sup>nd</sup> edition will be published soon after the examination based on the comments of the examination committee.





# Chapter 1

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## INTRODUCTION

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For many years focus in environmental organic chemistry has been on anthropogenic substances such as pesticides, dioxins, and endocrine disruptors and their occurrences in the environment - especially in ground- and drinking water. However, toxic substances are produced by Nature as well. In fact, some authors estimate that more than 99% of the toxic substances humans are subjected to in total are of natural origin (Ames and Gold, 1990; Ames et al. 1990). These natural toxins are mainly of plant or fungal origin, being a part of their present or past chemical interaction with herbivores as well as other plants or fungi (Ames et al. 1990; Teuscher and Lindequist, 1994; Kaufman et al. 1999).

Humans on all continents have utilized vast numbers of plants and fungi in traditional medicine. This usage can often find its roots in the content of bioactive compounds in the species used, and many species are today screened for their content of bioactives by the pharmaceutical industry. A typical example of this usage is Foxglove (*Digitalis purpurea* L.), which contains a series of glycosides used as heart medicine (Teuscher and Lindequist, 1994). Another widespread usage in North America and Europe of bioactive compounds is aromatherapy based on plant extracts and herbal teas (Anonymous, 2002b). A well-known example of this usage is the Australian *Tea Tree Oil* (made of leaf extracts from the *Melaleuca alternifolia* Cheel) having antiseptic properties. A not so well known example is the extract made from the needles of Western Red Cedar (*Thuja plicata* J. Donn ex D. Don) also having antiseptic properties due to its content of thujone (Teuscher and Lindequist, 1994). The effects of these drugs is caused by their content of bioactive compounds, and are generally considered safe for use due to their natural origin. But the risk of overmedication or intoxication may also be caused by such drugs or the plants/fungi themselves. For example, in Denmark and Britain several people are intoxicated and some are killed every year as a consequence of eating toxic fungi – e.g. Destroying Angel (*Amanita virosa* (Fr.) Bert.) which can be taken for a mushroom when in the young stage

(Cooper and Johnson, 1998). Thus, bioactive compounds may be lethal when consumed in too high amounts.

Due to their potent effect, some bioactive compounds have been examined thoroughly and are monitored regularly by the food safety authorities as some of these compounds occur regularly in human food, traditional/natural medicine, or animal fodder (e.g. Gry et al., 2000). For example, aflatoxins are produced by the fungi *Aspergillus flavus* Link. Aflatoxins are some of the most toxic natural compounds known, and are intensively monitored by the Danish food safety authorities as they occur regularly in certain tropical and subtropical crops, such as groundnuts.

The socio-economic impact of natural toxins can be immense, e.g. in Southern Brazil, cattle intoxication by plants is estimated to make up 17% of all diseases of cattle (resulting in a cattle mortality of 5%). Bracken fern (*Pteridium aquilinum* (Scop.) Gled) is responsible for approx. 40% of these plant intoxications, and in the Santa Catarina State, this sums up to an annual loss of \$2,100,000 (Gava et al. 2002)! The impact is usually socially biased since poor and/or uneducated farmers may not be able to eradicate the fern from pastures and hence improve their living standards. The economic loss is likely higher due to lost productivity of the cattle caused by sub-lethal doses of Bracken toxins (Smith and Beatson, 1970). Cases of Bracken-poisoning is also known from the industrialised part of the world, e.g. Germany and United Kingdom (Caldow and Burns, 2000; Schrader et al. 2001).

How does toxic bioactive compounds behave in Nature seen from an environmental point of view? It is a question yet to be answered. Almost no environmental scientists have addressed this problem so far – most have probably not given it any thoughts, and some scientists generally believe that natural substances are easily biodegraded, only found in small amounts and in general not being as toxic and harmful as anthropogenic toxins. Not much research has therefore been carried out on potential adverse effects of the natural toxins on the environment. Recently, North American and Danish research groups have although looked more carefully at a few toxins, for example: aflatoxins from the fungi *Aspergillus flavus* at Maize (*Zea mays* L.); Bt-toxins from genetically modified Maize – so-called Bt-Maize; and thujone from Western Red Cedar needles. The results show that:

- Aflatoxins from infected Maize may leach through the topsoil and reach drain channels (Madden and Stahr, 1993).

- Bt-toxins are found in the rhizosphere of Bt-Maize. The toxin has also proved to be stable in the soil (Saxena et al. 1999).
- Thujone are found in high concentrations in plant debris (litter) from Western Red Cedar, and in soil layers down to 90 cm depth below such stands (Strobel et al. 2003).

Taken together, these results shows the potential for natural toxins to be transferred from plants/fungi to the soil environment, where they are stable enough to reach deeper soil layers and hence may reach groundwater or other recipients.

Before a natural toxin exhibits a threat to soil or water resources, a series of prerequisites must be fulfilled. The compound must be:

- Produced in vast amounts by plants or fungi.
- Found in commonly occurring plants or fungi.
- Found in plants or fungi with a high biomass-production.
- Relatively stable in soil and soil solutions.
- Highly water-soluble and mobile in the soil environment.
- Highly toxic or carcinogenic.

Ptaquiloside found in Bracken and some other ferns might represent such a substance, since (van der Hoeven et al. 1983; Niwa et al. 1983; Niwa et al. 1983a; Saito et al. 1989; Smith et al. 1994; Taylor, 1999; Rasmussen et al. 2003c):

- ✓ Bracken is one of the most common plants on Earth.
- ✓ Ptaquiloside is found in high amounts in the biomass.
- ✓ Dry matter production of Bracken is high, 1,400 g m<sup>-2</sup> per year or more.
- ✓ The degradation rate of ptaquiloside may be rather slow.
- ✓ Ptaquiloside is readily water soluble and hence theoretically quite mobile in the soil.
- ✓ Ptaquiloside is genotoxic and carcinogenic.

The aim of this thesis is to investigate the potential hazardous nature of ptaquiloside seen from an environmental point of view by seeking answers to the following questions:

1. Can ptaquiloside be found in soil materials?

2. Is ptaquiloside stable in contact with soil materials?
3. What is the mobility of ptaquiloside in soil?

To be able to answer these questions, a few other questions regarding the occurrence of ptaquiloside in different fern components has to be answered:

4. When is ptaquiloside present?
5. What is the natural variation in the ptaquiloside content of Bracken?
6. How much ptaquiloside can be produced under natural conditions?

The research carried out by the author has focused on the Bracken varieties vars. *aquilinum* and *esculentum* (12 varieties exist, see Chapter 2), but reference will also be made to the other Bracken varieties. There is no generally accepted taxonomy of Bracken, as well as many different common names has been used for the different varieties of Bracken. Therefore, Latin names following the taxonomy of Tryon (1941) as referred by Burge and Kirkwood (1992) and Smith et al. (1994) are used instead of common names throughout the text. However, newer taxonomy is used in some of the articles in the second part of the thesis.

## Chapter 2

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### BRACKEN

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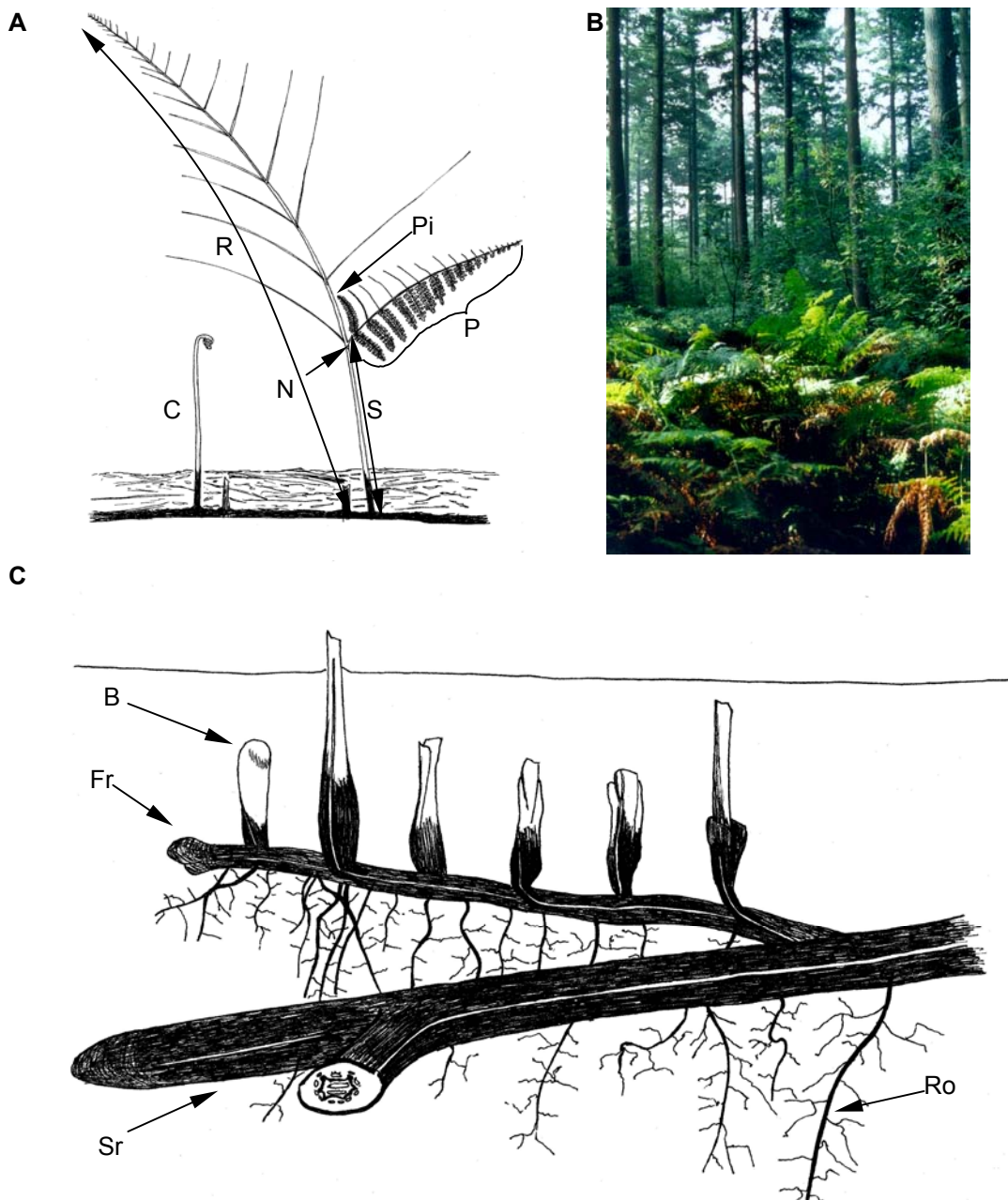
#### 2.1 Morphology, taxonomy and distribution.

Bracken (*Pteridium aquilinum* (Scop.) Gled, Dennstaedtiaceae, Figure 2.1) is a large deciduous rhizomatous perennial fern, having fronds with a maximum height of 2-4 m originating from a well-developed underground system of rhizomes (Øllgaard and Tind, 1993).

*Bracken* is an old English word used for all large ferns, but applied to this species in particular. The name *Bracken* may originate from the German *Brache* or *Brach-feld* which means uncultivated land or land open to tillage – showing old recognition of Bracken as a colonizer of abandoned arable land (Rymer, 1976). Other common names for the different varieties of Bracken include: Brake, Brake Fern, Eagle Fern, Female Fern, Fiddlehead, Hog Brake, Pasture Brake, Western Brackenfern, Devil's Foot (Scotland), Fern of God (Ireland), Grande fougère (France), Fougère d'aigle (France), Warabi (Japan), Örnbräken (Sweden), Einstape (Norway), Ørnebregne (Denmark), Sananjalka (Finland), Adlerfarn (Germany), Kilpjalg (Estonia), Warabi (Japan), Broto de Samambaia (Brazil), and Rahurahu (Maori, New Zealand) (Santos et al. 1987; Brownsey and Smith-Dodsworth, 1989; Rook, 2002; Grieve, 2002). There is some disagreement on the taxonomy of Bracken (see below), and several Latin synonyms have been used: *Allosorus aquilinus*, *Asplenium aquilinum*, *Filix aquilina*, *Filix-foemina aquilina*, *Ornithopteris aquilina*, *Pteris aquilina*, and *Pteris latiuscula* (Rook, 2002).

#### *Morphology and growth.*

The fronds of *Pteridium aquilinum* are soft herbaceous to leathery (depending on light



**FIGURE 2.1.** Bracken (*Pteridium aquilinum* (Scop.) Gled, Dennstaedtiaceae). A: Bracken frond and crozier. B: Mature Bracken stand (var. *aquilinum*), Salten Langsø, Denmark. C: Bracken rhizome system. Abbreviations: B = Bud; C = Crozier; Fr = Frondbearing rhizome; N = Nectaries; P = Pinnae; Pi = Pinnulet; R = Rachis; Ro = Root; S = Stipe; Sr = Storage rhizome. Drawings and photograph by L.H. Rasmussen.

exposition), triangular to triangular-ovate in outline, and bipinnate-pinnatisect to tripinnate. The pinnae are or are nearly opposite (except at apex), the lower ones stalked while the upper are sessile. The pinnae are triangular to triangular-ovate in outline. The outline of the pinnules is linear-lanceolate. The ultimate segments are closely set running together at the base, entire and obtuse, glabrous above and hairy to variable degrees below. The venation

is free, except from fertile segments having continuous marginal, sorus-bearing vein connecting the outer ends of the veinlets (Brownsey and Smith-Dodsworth, 1989; Øllgaard and Tind, 1993). Paired nectaries are found at the base of pinnae. The nectaries are only active in the early development stages of the individual pinnae. The nectaries may be an adaptation to attract ants that will protect the young fronds from herbivores (Øllgaard and Tind, 1993; Rumpf et al. 1994).

The dark rhizomes are subsurface (down to 0.5 m), although some may be found in the litter layer (Watt, 1976). Two types of rhizomes exist: frondbearing rhizomes (blackish short shoots from which the fronds emerge), and storage rhizomes (brownish long shoots forming the main stem of the fern; used for storage of nutrients). A third type of rhizome may be delineated, the transitional shoot, which is transitional between the two other types (Øllgaard and Tind, 1993; Whitehead and Digby, 1995). The frondbearing rhizomes are usually found above the storage rhizomes (Watt, 1976). The thin (1-3 mm thick) brittle blackish roots are attached to both types of rhizomes, and may extend 150 cm into the soil (Watt, 1976; Rook, 2002). The highest density of roots is found at the frondbearing rhizomes (author, personal observation). The roots may penetrate the mineral soil surface and extend into the litter. Up to 7,900 cm of roots per L of soil can be found in this region (Watt, 1976)! Large numbers of buds are found at rhizomes. Between 400 and 600 per m<sup>2</sup> has been reported for var. *aquilinum*, but only approx. 20% was active at the same time. Dormant buds may be up to 10 years old (var. *latiusculum*). Cutting, burning, and application of nitrogen and phosphorous nutrients increase bud activity. The large number of dormant buds is likely partly responsible for the invasive character of Bracken (Figure 2.2) (Shorina, 1990; Tolhurst, 1990; Whitehead and Digby, 1995; Whitehead and Digby, 1997a). The development from bud to sporulating frond takes 3 year (Christiansen and Jahns, 1981). It takes 3 years for the fern to develop from spore to fully mature frond (Figure 2.3) (Bell, 1976).

### ***Bracken reproduction.***

Bracken is genetically diploid and reproduces by spores, but is able to spread vigorously by the use of rhizomes too. The distal part of the rhizomes can reach an age of 35-75 years (Watt, 1940). The expected life span of a clone is uncertain, but estimates of several hundred years or more have been reported (Alonso-Amelot et al. 2001). However, the clone is often killed before reaching this age due to natural succession and ecosystem change (Page, 1976; Rook, 2002). The maximum extent of a single clone has been genetically determined to 390 m (Sheffield et al. 1989).



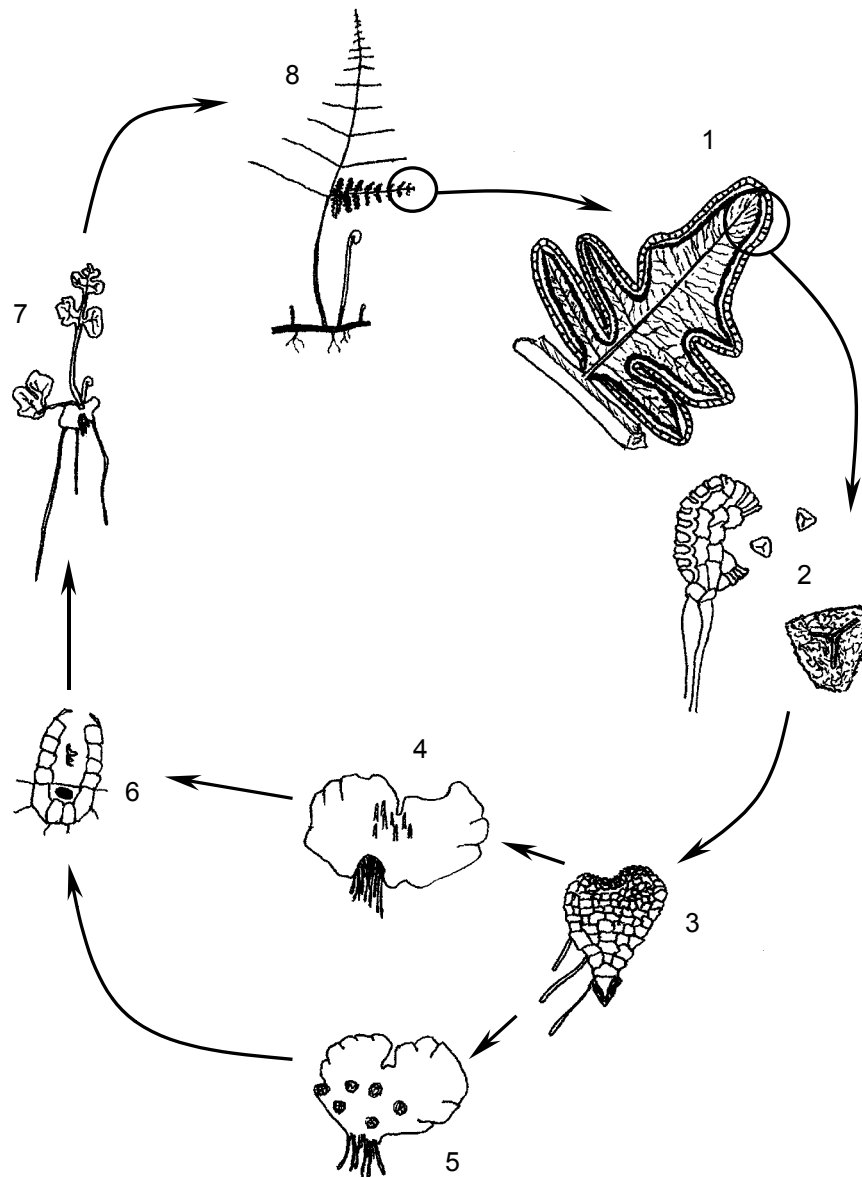
**FIGURE 2.2.** Mature Bracken stand at Præstø Fed, Denmark (var. *aquilinum*). The stand is situated on a former moor dominated by Heather (*Calluna vulgaris* L.). Photograph by L.H. Rasmussen.

Bracken is highly variable at the genetic level, which is interesting since Bracken often reproduces only by extension of the rhizome system. The high genetic variation is likely due to high gene flow rates between populations and to homoeologous pairing and cross over of chromosomes at meiosis resulting in a genetically variable population even when originating from a single spore (Jones, 1983; Wolf et al. 1988; Wolf et al. 1988a; Wolf et al. 1990). Bracken stands may originate from a single spore, but usually several genotypes are present resulting in a heterogeneous stand (Sheffield et al. 1989). At high latitudes, Bracken does not produce spores every year, and gene transfer might therefore be of less significance in such regions. Sporulation is related to the energy balance of the fern at the time of spore-production initiation (midsummer), where it is favoured by increased light and temperature. Sporulation is also affected by genotype. Sporulation takes place in early autumn (Øllgaard and Tind, 1993; Kendall et al. 1995; Simán et al. 1999; Ershova, 2000; Wynn et al. 2000). Vegetative formation seems to be the dominant way of stand propagation inside established stands, while establishment of new stands takes place from spores (Watt, 1976; Øllgaard and Tind, 1993).

### *Bracken taxonomy.*

Bracken is well defined at the generic level, but uncertainties are found regarding the nomenclature and taxonomy within the genus (e.g. Thomson (2000) and Thomson and Alonso-Amelot (2002)). Bracken (*Pteridium aquilinum*) belongs to the family Dennstaedtiaceae, and is divided into two subspecies: ssp. *aquilinum* and ssp. *caudatum* (Figure 2.4 and 2.5, Tryon (1941)).

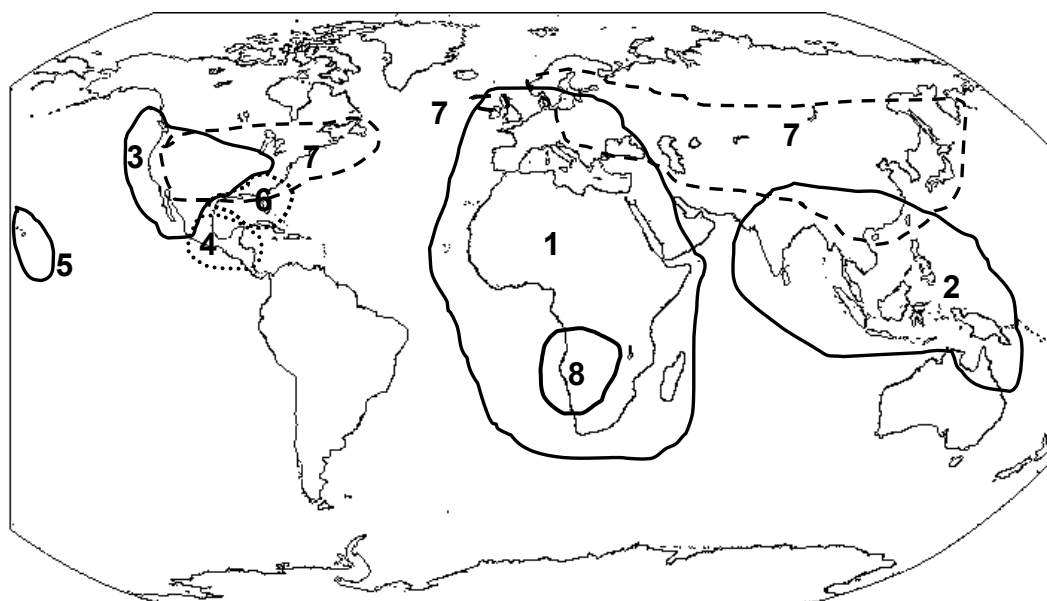




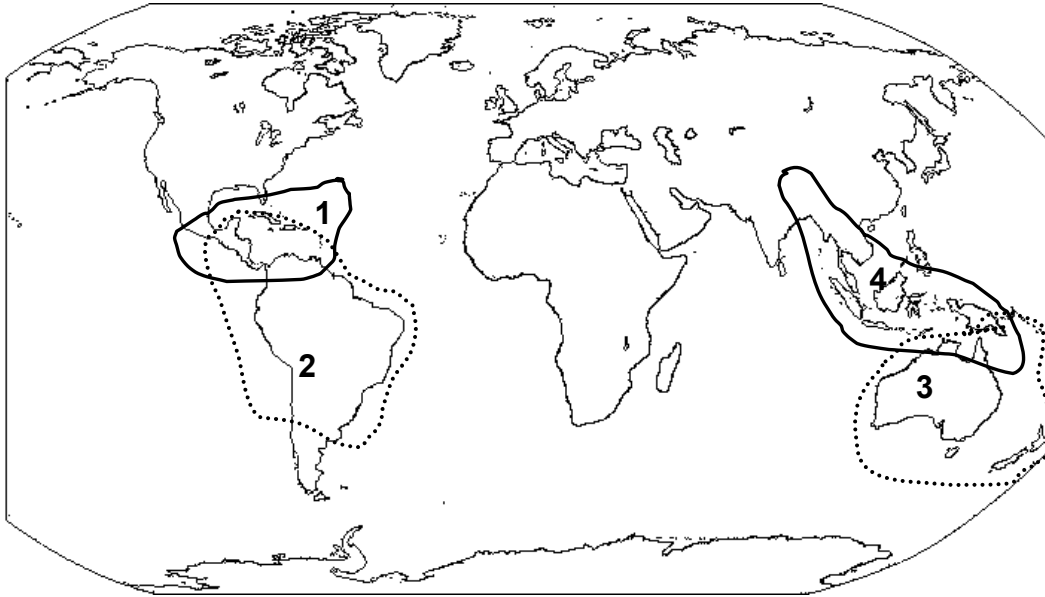
**FIGURE 2.3.** Bracken life cycle. 1: Lower surface of pinnule. Sporangia situated along the marginal sorus. 2: Sporangium and spores. 3: Young prothallus formed from spores. 4: Old prothallus with archegonia. 5: Old prothallus with antheridia. 6: Spermatozoids (from antheridium) migrates to the archegonium where they enter the egg cell to form an gametophyte. 7: Young sporophyte still attached to the gametophyte. 8: Adult sporophyte. Drawing by L.H. Rasmussen. Modified from (Brownsey and Smith-Dodsworth, 1989; Burge and Kirkwood, 1992; Øllgaard and Tind, 1993).

These subspecies are furthermore divided into 8 and 4 geographical varieties, respectively. The 12 geographical varieties may be further divided into a series of chemotypes and/or phenotypes according to their content of different natural compounds/secondary

metabolites and/or morphology. In England, Australia, and New Zealand cyanogenic genotypes have been identified within vars. *aquilinum* and *esculentum* (Cooper-Driver and Swain, 1976; Low and Thomson, 1990), while a ptaquiloside chemotype has been suggested within var. *esculentum* (Rasmussen et al. 2003d). Recently, genetic evidence has proven differences between some varieties at generic level, and the following varieties are now proposed full species status by Thomson (2000) and Thomson & Alonso-Amelot (2002): *africanum*, *aquilinum*, *arachnoideum*, *decompositum*, *esculentum*, *latiusculum*, and *revolutum*. Varieties *pseudocaudatum* and *pubescens* should be treated as varieties within *latiusculum* while var. *yarrabense* is an *esculentum* x *revolutum* hybrid (Thomson, 2000). Var. *aquilinum* should be perceived as a complex and heterogeneous species comprising several phenotypes and hybrids with other bordering Bracken varieties (Cooper-Driver and Swain, 1976; Sheffield et al. 1989; Wolf et al. 1990; Wolf et al. 1995; Thomson, 2000; Speer, 2000; Thomson and Alonso-Amelot, 2002). It must be emphasized that Bracken is an actively evolving species, and further work on its taxonomy is needed before the taxonomy is fully understood (Page, 1990).



**FIGURE 2.4.** Geographical distribution of the Bracken subspecies *aquilinum*. 1: var. *aquilinum*. 2: var. *revolutum* (syn. *wightianum*). 3: var. *pubescens*. 4: var. *feei*. 5: var. *decompositum*. 6: var. *pseudocaudatum*. 7: var. *latiusculum*. 8: var. *africanum*. Taxonomy as defined by Tryon (1941). Map adapted from Burge and Kirkwood (1992), Mosberg et al. (1994) and Page (1989).



**FIGURE 2.5.** Geographical distribution of the Bracken subspecies *caudatum*. 1: var. *caudatum*. 2: var. *arachnoideum*. 3: var. *esculentum*. 4: var. *yarrabense*. Taxonomy as defined by Tryon (1941). Map adapted from Burge and Kirkwood (1992).

### *Bracken distribution and ecology.*

Bracken is found on all continents except Antarctica. It is an invasive and opportunistic species well suited for dominating the forest floor below canopy openings. Hence, Bracken is often found in glades, on clear-cut or newly burned areas, and along tracks in forested areas, on the edge of forests and in open woodland and primary bush. Bracken may also dominate more open landscapes such as moors. Bracken is found in semiarid areas, as well as in tropical and temperate forests, moors and heath lands on both hemispheres (Tryon, 1941; Nicholson and Paterson, 1976; Page, 1976; Fenwick, 1988; Brownsey and Smith-Dodsworth, 1989; Henning and Dickmann, 1996).

There are two ways in which Bracken influences succession in Northern Europe (var. *aquilinum* – see Box 2.1): Through invasion into early-successional plant communities, and through control of invasion of later-successional species (Marrs et al. 2000).

The earliest fossil records of Bracken are from Oligocene (Hungary) and Miocene (England) (Rymer, 1976). Spores have been found in sediments throughout the Quaternary, but it is only during the last 5,000 years Bracken has reached its worldwide distribution and abundance. This increase of abundance is thought related to the advance of Neolithic Man causing widespread forest clearings as agriculture was introduced and domestication

of large herbivores took place (Rymer, 1976). Today, Bracken is one of the five most common plants on Earth. In the industrialized countries, the extensive character of agriculture on marginal land - such as mountains, moors and heather favours the still continuing advance of Bracken (Fenwick, 1988; Taylor, 1999). Overgrazing or grazing by species not willingly eating Bracken such as sheep may also contribute (Smith, 2000).

**Box 2.1.** The five natural reasons for the success of Bracken.

The success of Bracken is due to (Atkinson, 1989; Marrs et al. 2000):

1. Large rhizome system with huge potential for storage of nutrients and carbohydrates.
2. Large number of buds capable of producing new fronds.
3. High productivity resulting in a dense frond canopy that casts deep shade (removes up to 94% of full sunlight (Humphrey and Swaine, 1997)).
4. Large accumulation of litter.
5. A wide range of toxic compounds within living tissues and dead plant material.

Besides from the above mentioned reasons, Bracken also consumes large amounts of water, which adds to the competitiveness of Bracken (Tolhurst and Burgman, 1994; Pakeman and Hay, 2001).

Increased atmospheric deposition of pollutants such as NO<sub>x</sub> and heavy metals favours Bracken in Northern Europe compared to species Bracken usually compete with, e.g. Heather (*Calluna vulgaris* L.) (Smith, 2000). Bracken is not likely directly affected by higher CO<sub>2</sub>-concentrations in the atmosphere, but increased temperature may widen the climatic zones where Bracken can grow (Caporn et al. 1999). Drainage of low-lying soils and peat bog's can also partially explain the Bracken advance (Smith, 2000). In the United Kingdom, Bracken advance has been reported to 1 to 3% per annum, resulting in a total Bracken cover in Scotland and Wales of more than 6% in some areas - in some counties even as much as 40% of the land is covered by Bracken (Rymer, 1976; Fenwick, 1988; Taylor, 1999). Recent counts of Bracken spores in pollen records from the United Kingdom do although indicate that the abundance of Bracken today is less than or only equivalent to maximum historical levels (Pakeman et al. 2000). However, this study is based on few sites only, and includes the questionable assumption that sporulation is constant over time and therefore usable for indication of Bracken cover.

***Bracken soils and elemental composition of Bracken.***

Bracken is usually found on well-drained acid soils (from pH 2.6) such as Spodosols (Figure 2.6), Inceptisols, and Entisols, but may at times be found at nutrientrich and even calcareous soils such as Alfisols and Mollisols (up to pH 7.5) (Watt, 1976; Frankland, 1976; Jarvis and Duncan, 1976; Skeffington, 1983; Ader, 1989; Hetherington and Anderson, 1998; Rasmussen et al. 2003b). This distribution in relation to soil type is a result of the edaphic demands of the fern as well as a result of human clearance of Bracken

from more fertile soils and arable lands (Thompson et al. 1986; Ershova, 2000; Smith, 2000; Wind, 2002).

The elemental composition of Bracken fronds is not related to the nutrient status of the soil (Hunter, 1953), but some variation in the pH of Bracken litter is typically found (Rasmussen et al. 2003b). Bracken growth may result in formation of a thick (up to 86 cm) and dense litter layer composed of frond remnants. In some areas, litter continues to accumulate until breakdown of the stand. The litter is apparently not digested by earthworms, and a strict horizontal layering of the litter can be observed (Watt, 1976; den Ouden and Vogels, 1997). In the first 20-40 cm of the topsoil, large amounts of dead and living rhizomes can be observed, imposing large impact on soil structure and carbon content (Rasmussen et al. 2003). Due to the large production of nutrient rich biomass, high water consumption, increased pH and salt content in frond leachates and internal nutrient recycling within the fern, Bracken can improve the quality of very poor soils such as Spodosols (Table 2.1) (Williams et al. 1987; Smith, 2000). Bracken may also reverse leaching processes at moors compared to vegetation dominated by Heather (Mackney, 1961; Jarvis and Duncan, 1976). On nutrient-rich soils, Bracken (vars. *aquilinum* and *latiusculum*) can have the opposite effect, as the acidity of Bracken litter and Bracken leachates enables buffering of soil solution pH around pH 5-5.5 which is 1-1.3 pH-units lower than in soil solutions from adjacent non-Bracken soils.

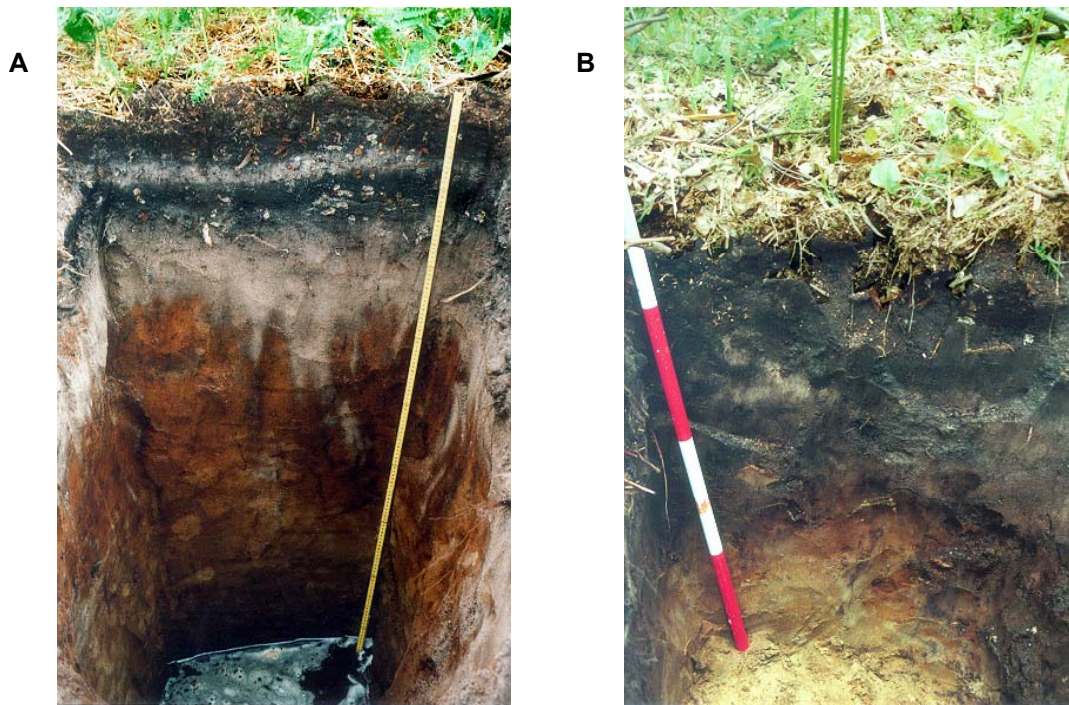
**TABLE 2.1.** Elemental composition of mature var. *aquilinum* fronds (12 sites) and rhizomes (1 site) in Scotland (Hunter, 1953).

Element:	Content in fronds: (% in dry matter)	Content in rhizomes: (% in dry matter)
Ash content	5.20 - 6.45 <sup>†</sup>	5.71
Nitrogen	1.31 - 2.12	1.05
Phosphorous	0.09 - 0.22	0.05
Potassium	1.02 - 2.95	1.24
Calcium	0.15 - 0.37	0.18
Magnesium	0.11 - 0.27	0.18
Sodium	0.07 - 0.24	0.26

<sup>†</sup> Only two sites.

Increased soil solution concentrations of dissolved organic matter (DOM) are also found below Bracken. Up to 4 times the content found in soils with no Brackens have been measured (up to 50 mg L<sup>-1</sup>) (Johnson-Maynard et al. 1998). The acidifying effect and the increased content of DOM in Bracken soils, can change the soil mineralogy by increasing the rate of weathering and the formation of organo-mineral complexes (e.g. aluminium-

iron-humus complexes). The high annual input of organic matter to the Bracken soil can also change the classification of a soil. For example, Haplocryands has been reported to change to Fulvicryands in Bracken dominated areas (Johnson-Maynard et al. 1997; Johnson-Maynard et al. 1998; Pitman and Webber, 1998; Cwen et al. 1999; Soil Survey Staff, 1999). Changes in mineralogical composition may also alter classification (Johnson-Maynard et al. 1997).

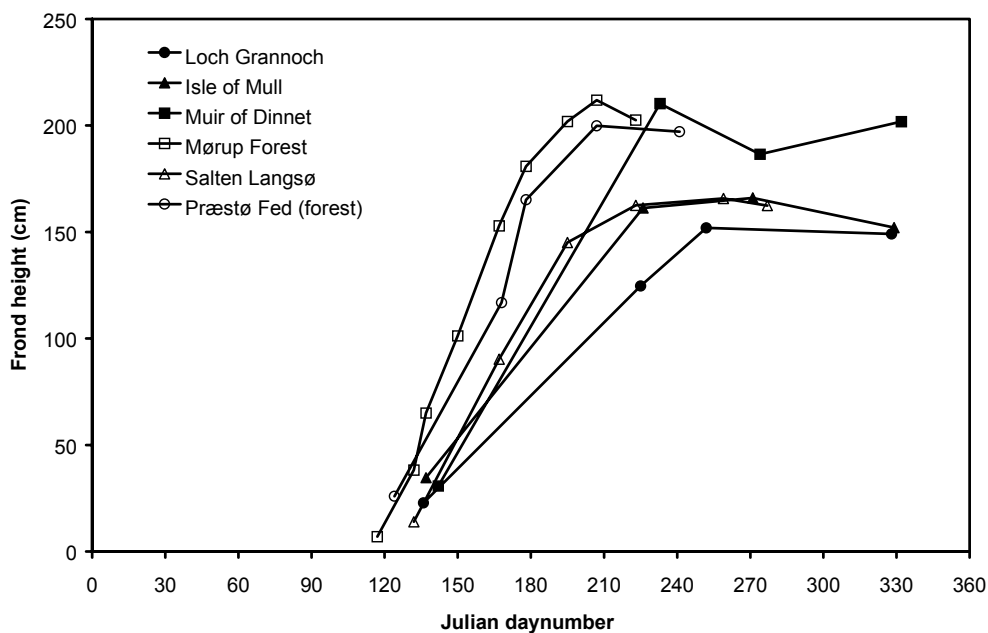


**FIGURE 2.6.** Two soils with Bracken cover in Denmark (*var. aquilinum*). A: Oxyaquic Quartzipsamment at Præstø Fed (June 2000; ptaquiloside load (fronds in September) = 20 mg m<sup>-2</sup>). B: Histic Endoaquod at Sorø (May 2000; ptaquiloside load (fronds in September) = 260 mg m<sup>-2</sup>) (Soil Survey Staff, 1999; Rasmussen et al. 2003b).

## 2.2 Growth of Bracken.

Bracken grows rapidly, and the fronds reach maximum length in a few months (Figure 2.7). The average growth can be several centimetres per day in the initial growth phase (Alonso-Amelot and Rodulfo-Baechler, 1996; Alonso-Amelot et al. 2000; Ershova, 2000; Rasmussen and Hansen, 2003). Frond maturation and spore production in fertile fronds takes place 100-130 days after emergence for *var. aquilinum* in Denmark, while *vars. caudatum* and *arachnoideum* only use 40-70 days in Venezuela (Evans and Galpin, 1990; Alonso-Amelot et al. 1992; Alonso-Amelot et al. 1995; Alonso-Amelot and Rodulfo-Baechler, 1996; Whitehead and Digby, 1997; Alonso-Amelot et al. 2000; Rasmussen and

Hansen, 2003). When fronds mature, translocation of nutrients (carbohydrates and minerals) and secondary metabolites takes place from the fronds to the rhizomes (Hunter, 1953; Watt, 1976; Williams and Foley, 1976; Rasmussen and Hansen, 2003). The precise growth pattern is highly dependent on the physical growth conditions as well as on the variety of Bracken in question. Frond emergence of var. *aquilinum* in Scotland was found to be dependent on soil temperature as bud growth initiates at 5.5°C (Ader, 1989; Birnie et al. 2000). In areas with no frost, individual Bracken fronds may reach an age of more than 12 months. For example, var. *esculentum* fronds in Australia can be up to 36 months old (Tolhurst, 1990).



**FIGURE 2.7.** Growth of var. *aquilinum* in Denmark (white marks) and Scotland (black marks). Data from Rasmussen and Hansen (2003) and Rasmussen et al. (2003).

Five annual growth stages of Bracken can be delineated (partly after Rasmussen et al. (2003b), Tolhurst (1990) and Williams and Foley (1976)):

1. Crosier stage: Fronds has not opened 1<sup>st</sup> pair of pinnules on 1<sup>st</sup> pair of pinnae. Carbohydrates are transferred from rhizomes to fronds.
2. Young stage: Fronds has opened 1<sup>st</sup> pair of pinnules on 1<sup>st</sup> pair of pinnae, but are not fully expanded. Carbohydrates are transferred from rhizomes to fronds. By the end of this stage, carbohydrate replenishment of the rhizomes begins, especially to storage rhizomes.
3. Mature stage: Fronds are fully expanded. Sporulation may take place on fertile fronds. 1<sup>st</sup> and 2<sup>nd</sup> pair of pinnae may exhibit some signs of deterioration in

shaded and dense Bracken stands. Carbohydrate replenishment of both rhizome systems is taking place.

4. Senescent stage: Old fronds show strong signs of deterioration. The colour may change to yellow, orange or brown. Carbohydrate replenishment of both rhizomes systems and translocation of carbohydrates from frondbearing to storage rhizomes.
5. Dormancy: Fronds are dead. No aboveground activity. Slow decrease in reserve carbohydrates and slow increase in the proportion of mobile carbohydrates.

In temperate and subtropical climatic zones, Bracken stands exhibits an annual growth pattern where the crosiers emerge from the soil in the spring and the fronds die back in the autumn (often killed by frost; growth stage 1 to 5) or the following year (growth stage 1 to 4) (Watt, 1976; Øllgaard and Tind, 1993). In the tropics, Bracken stands has a 12-month growing season (growth stage 1 to 4, but only at individual fronds, as new fronds continue to emerge all year round) (Alonso-Amelot et al. 1992).

Bracken (var. *aquilinum*) stand development in forested and open habitats in Denmark, the United Kingdom and New Zealand (var. *esculentum*) can be divided into four separate phases (Watt, 1976; Rasmussen et al. 2003; Rasmussen et al. 2003b; Rasmussen et al. 2003d):

1. Pioneer phase (new community): Low frond density. No Bracken litter. Few rhizomes. No dead rhizomes.
2. Building phase (young community): Medium frond density. Low amounts of Bracken litter. Many rhizomes. No dead rhizomes.
3. Mature phase (mature community): Maximum frond density. Deep layer of Bracken litter. Multiple rhizomes. Dead rhizomes can be found.
4. Degenerate phase (old community): Low-medium frond density. Medium-deep layer of Bracken litter. Multiple rhizomes, high proportion of dead rhizomes. Patches of other herbs (not usually associated with Bracken) can be found inside the stand.

The frond density can be quite high, resulting in an enormous pressure on competing species. Frond (var. *aquilinum*) densities in Denmark and United Kingdom are usually in the range 5-55 fronds per m<sup>2</sup>. The density is affected by natural variations in growth conditions, nutrient availability, and control measures such as mowing (cutting increases the frond density) (Nicholson and Paterson, 1976; Atkinson, 1989; Whitehead and Digby,



1997; Whitehead and Digby, 1997a; Buttenschøn, 2000; Rasmussen et al. 2003; Rasmussen et al. 2003b; Rasmussen and Hansen, 2003). Densities of var. *latiusculum* up to 60 fronds per m<sup>2</sup> has been reported for the former USSR (Shorina, 1990). In Australian forests, var. *esculentum* has densities up to 4 fronds per m<sup>2</sup> (Tolhurst, 1990). In Venezuela, densities between 2 and 67 fronds per m<sup>2</sup> are found for var. *caudatum* and 5-15 fronds per m<sup>2</sup> for *arachnoideum* (Alonso-Amelot et al. 1995; Alonso-Amelot and Rodulfo-Baechler, 1996; Alonso-Amelot et al. 2000). In the United Kingdom, new fronds continue to emerge during most of the growth season. However, in Scotland maximum frond density seems to be reached during the first 30-40 days of the growth season (Ader, 1989; Whitehead and Digby, 1995; Whitehead and Digby, 1997; Rasmussen et al. 2003). This is not the case in Denmark, where almost all fronds have emerged from the ground within 1-2 weeks (author, personal observation).

Several frond biomass production functions exist, e.g. Paterson et al. (1997). Alonso-Amelot et al. (2000), Alonso-Amelot and Rodulfo-Baechler (1996), and Rasmussen and Hansen (2003) have presented simple models for vars. *aquilinum*, *arachnoideum* and *caudatum* based on the length of the frond (Table 2.2). Models like those and direct measurements have been used to quantify the annual frond dry matter production, which reach maximum at frond maturity in late summer/early autumn before the fern begins retrieval of nutrients to rhizomes (Watt, 1976; Williams and Foley, 1976).

**TABLE 2.2.** Bracken biomass production functions.

<b>Bracken:</b>	<b>Equation:</b> Biomass (g frond <sup>-1</sup> )	<b>Country:</b>	<b>Reference:</b>
Unmanaged var. <i>caudatum</i>	$0.29 \times \exp^{(0.044 \times Lf)}$	Venezuela	Alonso-Amelot et al. (2000)
Unmanaged var. <i>caudatum</i>	$-19.33 + 0.512 \times Lf$	Venezuela	Alonso-Amelot and Rodulfo-Baechler (1996)
Unmanaged var. <i>arachnoideum</i>	$-10.80 + 0.458 \times Lf$	Venezuela	Alonso-Amelot and Rodulfo-Baechler (1996)
Unmanaged var. <i>aquilinum</i>	$0.72 \times \exp^{(0.018 \times Lf)}$	Denmark	Partly Rasmussen and Hansen (2003)
Managed var. <i>aquilinum</i>	$0.13 \times \exp^{(0.074 \times Lf)}$	Denmark	Unpublished

Lf: frond length (cm)

Var. *aquilinum* has an annual frond dry matter production up to 1,400 g m<sup>-2</sup> in United Kingdom while the production in Denmark is up to 550 g m<sup>-2</sup> (Watt, 1976; Buttenschøn, 2000; Rasmussen and Hansen, 2003; Rasmussen et al. 2003b). Vars. *caudatum* and *arachnoideum* in Venezuela have a whole year growth cycle resulting in a standing aboveground biomass between 25 and 1,440 g m<sup>-2</sup> with *caudatum* having the highest

production (dry matter) (Watt, 1976; Alonso-Amelot et al. 1995; Alonso-Amelot et al. 2000). The belowground biomass is mainly composed of rhizomes. For var. *aquilinum* this can make up to 660 g dry matter per m<sup>-2</sup>, while vars. *caudatum* and *arachnoideum* have up to 140 and 570 g m<sup>-2</sup> respectively (Galpin and Smith, 1986; Alonso-Amelot et al. 1995; Alonso-Amelot and Rodulfo-Baechler, 1996). Rhizomes may make up around 80% of the total biomass of Bracken populations, with storage rhizomes counting for 60-80% of the rhizome biomass (Whitehead and Digby, 1995; Whitehead and Digby, 1997a; Alonso-Amelot et al. 2001). The dry matter content in the rhizomes is lowest during late spring and early summer while the fronds are growing (var. *aquilinum*). From midsummer till mid autumn, the carbohydrate-content of the rhizomes is replenished before the fern goes into dormancy (Williams and Foley, 1976).

### **2.3 Ethnobotannical history and recent usage of Bracken.**

Bracken has been widely utilized by Man in all parts of the world. In Europe, Bracken was mainly used as food at times of scarcity – a sort of flour has been made out of the rhizomes. The fronds have also been used as bedding for animals and Man (e.g. by Roman soldiers, Vikings, and peasants in Denmark and Scotland), floor cover, fuel, as ornamental and ritual plant, as bleaching and dyeing agent (tartans and tweed), for dykes, roofing (thatch), baskets, mulch, compost and as source of potash for the glass and soap industry. Bracken fronds have also been used for packaging of fruit and breakable materials (Rymer, 1976; Brøndegaard, 1978; Fenwick, 1988; Taylor and Smith, 1995; Project Runeberg, 1997; Bento et al. 2000; Anonymous, 2002a).

The Forest Authority in United Kingdom recently investigated the potential of using Bracken compost as a peat alternative for making suitable potting media for calcifuge plants (in combination with peat) or alone as a mulching material (Pitman and Webber, 1998). Bracken litter may also be used as a remedy for soil pH modification in the restoration of former moors used as farmland. Here, litter has been showed to reduce soil pH to pH 4-4.5 (Owen et al. 1999). An evaluation of var. *esculentum* for use as mulch in the glasshouse industry showed promising results (Taylor and Thomson, 1998). Bracken may also have a future as fertilizer or mulch in organic farming (Eric Donnelly, University of Aberdeen, personal communication).

In New Zealand, Australia, and North America, the pre-European settlers widely used Bracken rhizomes as everyday food or during travel (Newsome, 1987; Leach, 2000; Turner, 2002). In Japan, Korea and some parts of South America (e.g. Brazil) Bracken is

still used for food (many recipes can be found at the internet: search word *Bracken* or *warabi*). In Tokyo alone, more than 300 tonnes of young fronds were consumed every year in the 1970's, and the annual Bracken import to Japan in the 1980's and 1990's were 6,800-15,500 tonnes (commercial price in Japan (1997): 1.3 US\$ kg<sup>-1</sup>) (Rymer, 1976; Hirono, 1986; Nguyen, 2000). The fern is usually collected from natural habitats, but Bracken cultivation do take place some places in Japan and elsewhere (Fujio Morrow, 2000). Bracken collection does also take place among Japanese and Korean people as a recreational, social and outdoor activity. Immigrants from Japan and Korea tend to continue this activity in a cultural context (Anderson et al. 2000). The crosiers or young fronds are usually steeped or boiled in water sometimes added sodium bicarbonate or ordinary wood ash (to preserve the fresh colour of the crosier and to reduce the toxicity and bitterness of the fern – however, not all toxicity is removed by this treatment) (Hirono et al. 1972; Ortega, 1994; Silva et al. 2000; Fujio Morrow, 2000). However, a study of the ptaquiloside content in some commercial Bracken products from Siberia and Japan showed that the content of ptaquiloside was below the limit of detection (100 µg per g-fresh-weight) (Saito et al. 1989).

From time to time Bracken is advocated as food in Europe and North America (Galpin and Smith, 1986). In Denmark, the crosiers have been used as food too; either cooked or pickled (Brøndegaard, 1978). In Denmark, Bracken is not allowed as human food or as traditional medicine due to its toxic constituents, although the Danish army still advocates Bracken rhizomes as food in time of war (Inspektøren for Hæren, 1986; Gry et al. 2000).

Bracken has and is still some places used for beer brewing (replaces hops), as tea, traditional medicine (e.g. in Denmark and in the United Kingdom for killing intestinal worms, to heal wounds, as an abortifacient and for alleviating constipation), and in modern herbal medicine (e.g. Digestodoron<sup>®</sup>, WELEDA AG, Switzerland - used for regulation of the digestive system) (Rymer, 1976; Øllgaard and Tind, 1993; Ortega, 1994; Burkhalter et al. 1996; Wind, 2002; Anonymous, 2002a).

Besides from local use and cultivation of Bracken in Japan, Brazil and Korea, Bracken is now generally only perceived as a weed in agriculture and forestry. Bracken can be controlled in several ways (Cooper-Driver and Swain, 1976; Martin, 1976; Lowday, 1986; Bassett et al. 1990; Tolhurst, 1990; Taylor, 1990; McElwee et al. 1990; Auld, 1990; Lawton, 1990; Burge and Kirkwood, 1992; Breach, 1999; Backshall, 1999; Anonymous, 1999a; Anonymous, 1999b; Le Duc et al. 2000; Buttenschøn, 2000):

1. Chemical control: RoundUp<sup>®</sup> and Asulam<sup>®</sup> can be used for management. The success is highly dependent on timing of the herbicide application, and use of non-ionic surfactants in the pesticide formulations. Genetic polymorphism in sensitivity towards herbicides may make spraying difficult. Spraying may also damage other species than the fern, hereby making spraying less useful in conservation areas.
2. Biological control: Several mycoherbicides has been tested and some seems promising. Anthropogenic spread of Bracken-feeding insects has also been proposed, but is not likely to take place.
3. Mechanical control: Bracken breakers, ploughing, rotavating, and crushing by rolling are useful tools. The control measures have to be done 2-3 times per year for several years to eradicate the fern.
4. Manual control: Bracken cutting with scythe, fern hooks and similar tools. The control measure has to be done 2-3 times per year for several years to eradicate the fern.
5. Burning: Not advisable, since Bracken seems to become favoured by burning.
6. Grazing: Cows, goats, and sheep may browse Bracken, and tramping by the animals may break and kill young fronds. Generally not advisable due to the content of toxic principles in the fern that may kill or injure the animals (see Chapter 3 and 4). Grazing may be possible in areas with low contents of toxins in the ferns. Cows and goats generally avoid Bracken, and will often only eat the fern when forced to do so. However, calves breed without presence of adult animals and cattle not recognizing Bracken as toxic due to their upbringing in Bracken-free areas may eat the fern more willingly (Carlos Pinto, Servica de Desenvolvimento Agrario de Sao Miguel, and Eric Donnelly, University of Aberdeen, personal communication). In Denmark, poisonous fodder is not allowed, and attempts to use animals for Bracken control is therefore illegal (Anonymous, 2000). Such attempts should also be considered unethical from an animal welfare point-of-view.

## Chapter 3

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### BRACKEN TOXICITY

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#### 3.1 Bracken induced diseases in higher animals and Man.

Bracken causes a wide range of human and animal diseases, including cancer. Sublethal intoxication may also lead to immunosuppression, hence making animals and humans more susceptible for attack from pathogenic bacteria and virus such as the bovine papilloma virus (BPV-4) or MCF-virus (malignant catarrhal fever) (Saveria Campo, 2002; Twomey et al. 2002). Animals do not usually eat large amounts of Bracken, but may do so when: 1) Paddocks are overgrazed; 2) When the young ferns are more palatable than associated forage; 3) When in need of coarse fibres later in the growth season; or 4) When the animals are not familiar with the toxic properties of Bracken (Ralphs (2002); Carlos Pinto, Servica de Desenvolvimento Agrario de Sao Miguel, personal communication). Some animals may even get addicted to Bracken – a phenomena known from many toxic plant species (Figure 3.1) (Watson et al. 1972; Allsup and Griffiths, 1978; Cooper and Johnson, 1998).

In this section, a review will be given of the chemical interaction between Bracken and animals. Emphasis will be on the interaction between Bracken and farm animals, especially on the effect of ptaquiloside. The interaction between Bracken and lower animals, plants and fungi is described in Section 3.2. A selection of the secondary metabolites identified in Bracken will be described and their abundance given in Section 3.3. In the rest of the thesis, all weights are given per unit of air-dry sample, unless otherwise stated.

#### *Thiamine (vitamin B<sub>1</sub>) deficiency.*

Bracken contains thiaminase enzyme (type 1), especially in the rhizomes and in the crosiers. Bracken consumption may therefore lead to thiamine deficiency (*beri beri*) in monogastric (non-ruminant) vertebrates (e.g. horse, mule, and Man) (Nakabayashi, 1955;

Somogyi, 1971; Evans, 1976; Fukuoka, 1982; Evans, 1986; Munro, 1996). The thiaminase enzyme requires a co-substrate to proceed. Caffeic acid, astragalol, isoquercitrin and 5-*o*-caffeoylshikimic acid has been proposed as co-substrates (Nakabayashi, 1955; Somogyi, 1971; Fukuoka, 1982). The enzyme may also act as part of the ferns insect defence (Jones, 1983).



**FIGURE 3.1.** Holstein cow eating Bracken (var. *aquilinum*) growing at stonewall, Sao Miguel Island, Azores. Photograph by L.H. Rasmussen.

### *Potential for cyanogenesis.*

Some Brackens contain prunasin, which is a bitter tasting cyanogenic glycoside deterring herbivores (Cooper-Driver and Swain, 1976). Prunasin could cause cyanogenesis, but no deaths of higher animals has been reported as result of cyanogenesis caused by Bracken (Cooper-Driver and Swain, 1976; Hadfield and Dyer, 1986; Munro, 1996; Smith, 1997; Alonso-Amelot et al. 1998; Smith et al. 2000; Alonso-Amelot et al. 2001). However, insecticidal effects have been observed (Schreiner et al. 1984). Cyanogenic chemotypes have been identified and cyanogenic polymorphism within individual Bracken stands has been found among vars. *aquilinum* and *esculentum*. The content of prunasin in Bracken fronds seems to be correlated with ecosystem type, as higher levels of prunasin are found in shaded woodland habitats compared to open land. The higher content in ferns growing in shaded areas might have been evolved as response to increased grazing pressure in such areas (Cooper-Driver and Swain, 1976; Cooper-Driver et al. 1977; Blair et al. 1983; Hadfield and Dyer, 1986). The HCN release from shaded Brackens seems also correlated with rainfall and temperature (Hadfield and Dyer, 1986).

### ***Haemorrhagic syndromes – internal bleedings in sheep and cattle.***

Bracken intake causes acute internal bleedings in sheep and cattle as well as necrosis of tissue in the gut epithelium. The disease is cumulative, and the animal has to eat approx. 1 kg of Bracken dry matter each day for 2-4 weeks to produce the syndrome (depending on the toxicity of the fern) (Evans, 1986). Internal bleedings in all parts of the body occurs as the bone marrow ceases to produce blood platelets. The production of granulocytes and lymphocytes are also affected (Smith, 1997). Death is often the result of this poisoning known as *the [Acute] Bracken [Fern] Poisoning* (Caldow and Burns, 2000). The haemorrhagic syndromes are likely caused by ptaquiloside or ptaquiloside-like substances (Hirono et al. 1984a; Yoshida and Saito, 1994; Yoshida and Saito, 1994a; Castillo et al. 2000). Other Bracken compounds may be co-responsible for the formation of this disease when acting as immunosuppressants (Evans, 1986).

### ***Progressive retinal degeneration in sheep.***

Degeneration of the retina neuroepithelium in the eyes of sheep is found among animals browsing Bracken for 2-3 years. The syndrome is called *Bright Blindness* due to the increased reflectance of the eyes. The disease is found in the United Kingdom, usually in hill farms. Bright Blindness is caused by ptaquiloside and/or ptaquiloside-like substances (Watson et al. 1972; Allsup and Griffiths, 1978; Hirono et al. 1993; Smith, 1997; Cooper and Johnson, 1998).

### ***Enzootic haematuria – tumours in the urinary bladder in sheep and cattle.***

*Bovine Enzootic Haematuria* (BEH) is found among sheep and cattle feeding on Bracken all over the world, and is characterised by the presence of tumours in the urinary bladder causing blood in the urine (Figures 3.2 and 3.3) (Price and Pamukcu, 1968; Smith and Beatson, 1970). The bladder wall may also become thickened (Hoque et al. 2002).

BEH occurs widespread in some areas, e.g. up to 11.5% of the cattle are affected in some regions of Northern India and Nepal, in the Mérida district of Venezuela 18% of the cattle displayed symptoms of BEH, in Bolivia 11% of the cattle in Bracken affected areas had active BEH while carcinomas were present in 100% of the cattle investigated (206 bovines), 20 cases from 340 farms each year in a district in Wales, and at the Azores, up to 4% of the cows slaughtered at the Ponta Delgada abattoir between 1994 and 1997 had urinary bladder tumours while up to 7% had neoplasias (Hopkins, 1987; Lawson, 1989; Pinto et al. 1997; Dawra et al. 2000; Marrero et al. 2001; Alonso-Amelot and Avendaño, 2001; Dawra et al. 2002).



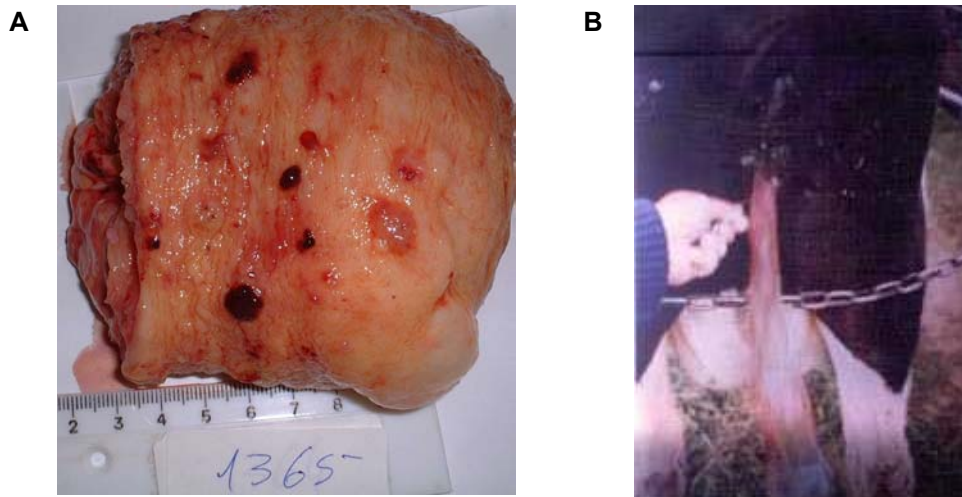
**FIGURE 3.2.** Symptoms of *Bovine Enzootic Haematuria* at anaemic Holstein cow, Sao Miguel Island, Azores. Note the yellowish colour of the udder. Photograph by C. Pinto.

In India, the occurrence of BEH seems somewhat correlated with the content of ptaquiloside in Bracken (Dawra et al. 2002). In China, BEH is common in the mountainous southwestern part and seems to be associated with the occurrence of var. *revolutum*. BEH occurred in 1-16% of the cattle, while the incidence among water buffalos was between 1 and 5% (Xu, 1992). In Costa Rica, Venezuela, and Columbia higher levels of BEH-occurrence is found at higher altitudes in areas of intensive Bracken cover compared to areas of lower altitude. The occurrence of BEH is thought to be caused by higher ptaquiloside contents in Brackens from mountainous areas (Villalobos-Salazar et al. 2000). However, the influence of altitude on the ptaquiloside content is disputed (see Chapter 5). The abundance of BEH may exhibit seasonal variation, e.g. in New Zealand BEH became evident in the early spring and disappeared in late summer (Smith and Beatson, 1970).

Ptaquiloside is likely responsible for the formation of these tumours in the urinary bladder, as administration of pure ptaquiloside to cattle has reproduced the clinical syndromes. Presence of cattle with BEH in Australia is correlated with the occurrence of Brackens having a ptaquiloside content  $>100 \mu\text{g g}^{-1}$  (var. *esculentum*). Similar correlations has been observed in New Zealand and at the Azores too. BEH usually develops after 2-3 years of Bracken exposure, and shorter periods of Bracken ingestion will not cause BEH (Smith et al. 1993; Smith et al. 1994; Smith, 1997; Pinto et al. 1999; Rasmussen et al. 2003d). Ptaquiloside-like substances such as *caudatoside*, *iso-ptaquiloside* and *ptaquiloside Z* may



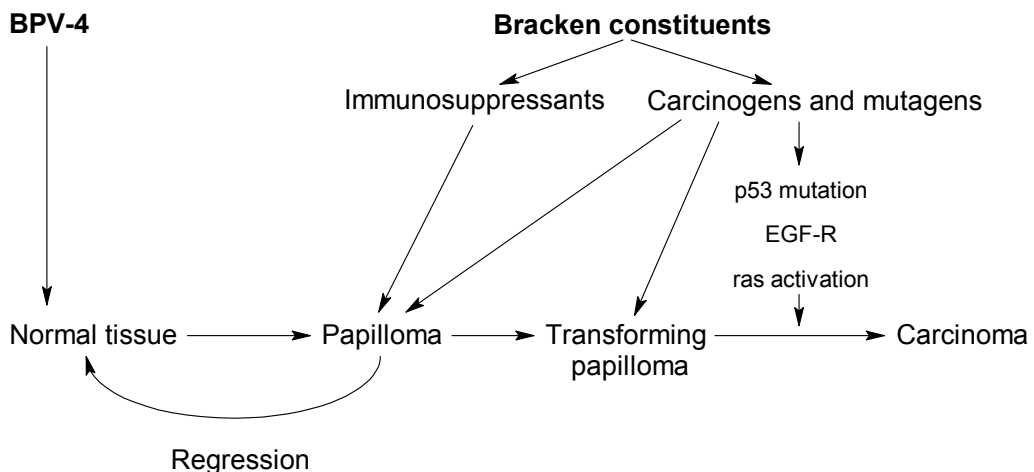
also cause BEH (Castillo et al. 2000). Bovine papilloma virus (BPV-1 and 2) may take part in the formation of bladder cancer (Campo, 1997; Saveria Campo, 2002).



**FIGURE 3.3.** Clinical syndromes of *Bovine Enzootic Haematuria* in cows at the Azores. A: Tumours in urinary bladder. B: Blood in urine. Photographs by NN (photographer could not be traced) (A) and C. Pinto (B).

*Alimentary cancer – tumours in the upper alimentary channel in cattle.*

In United Kingdom, Italy, Kenya, and Brazil epithelial tumours have been found in the upper alimentary channel in cattle from Bracken-infected areas. This disease is likely caused by latent bovine papilloma virus (BPV-4) infections in cattle (Figure 3.4).



**FIGURE 3.4.** Interaction between Bracken constituents (e.g. quercetin or ptaquiloside) and Bovine Papilloma Virus 4 (BPV-4) in the development of upper gastrointestinal tract carcinoma. Modified from Saveria Campo (2002).

In Italy, slaughterhouse monitoring revealed oesophageal lesions caused by papillomas in 13% of the cattle. BPV-4 was identified in more than 60% of these papillomas (Borzacchiello, 2003).

Investigations point at quercetin and/or ptaquiloside as the active mutagens and carcinogens, but the aetiology might be very complex, and the general immunosuppressive effects of Bracken may just be the trigger of this viral cancer (Bjeldanes and Chang, 1977; Moura et al. 1988; Campo et al. 1994; Smith, 1997; Stocco dos Santos et al. 1998; Connolly et al. 1998; Campo et al. 2000; Beniston et al. 2000; Saveria Campo, 2002).

#### *Symptoms found in laboratory animals, bacteria and cell cultures.*

Numerous investigations have been carried out on laboratory rodents to elucidate the toxic effects of Bracken. The animals have been served a diet containing a certain amount of dried Bracken (crosiers, fronds, spores, or rhizomes) or pure ptaquiloside. The result of such feeding experiments has been rapid development of tumours, especially of the ileum, and formation of ptaquiloside DNA-adducts. Tumours are also found in the mammary (mammary gland carcinomas), colon, urinary bladder, and other tissues (Evans and Mason, 1965; Evans and Mason, 1965; Hirono et al. 1972; Hirono et al. 1973; Saito et al. 1975; Yoshihira et al. 1978; El-Mofty et al. 1980; Hirono et al. 1984; Hirono et al. 1984b; Evans, 1986; Hirono et al. 1987; Santos et al. 1987; Smith et al. 1988; Villalobos-Salazar et al. 1995; Shahin et al. 1998; Shahin et al. 1998a; Freitas et al. 2000; Simán et al. 2000a; Silva et al. 2000; Freitas et al. 2001).

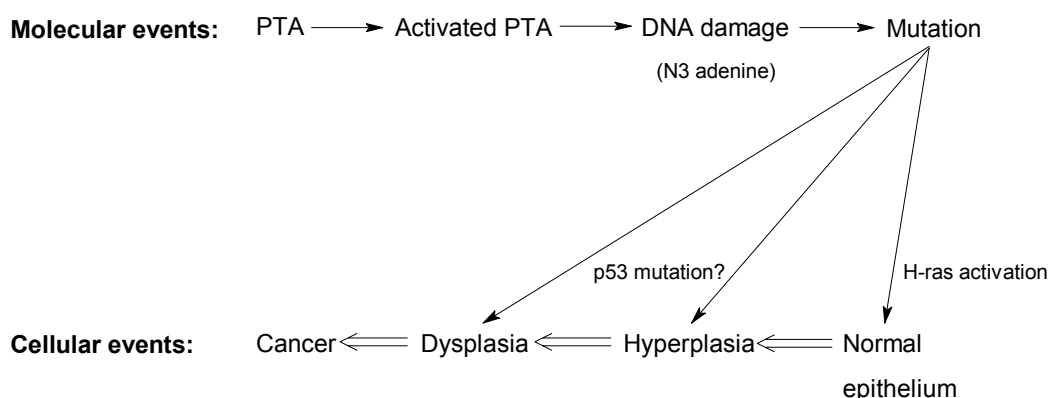
Lymphocytic leukaemia has also been reported in rats given oral or intraperitoneal doses of fresh crosier washings as well as in mice fed with spores (Evans et al. 1984; Evans, 1986; Evans and Galpin, 1990; Povey et al. 1995).

Very often, a large proportion of the laboratory animals dies during the experiments as consequence of fast developing tumours or other Bracken induced diseases (Evans and Mason, 1965; Hirono et al. 1972; Hirono et al. 1984). A single study on the effect of sun dried Bracken crosiers from Brazil (unknown variety) showed no carcinogenic effect on cancer in the urinary bladder (de Oliveira et al. 1995). This may be due to Bracken with no or low content of ptaquiloside/ptaquiloside-like substances (see Chapter 5).

Clinical symptoms have also been found in other animals than rodents used for toxicological testing, such as the Egyptian Toad, Japanese Quail, rabbits and farm animals. In fern fed rabbits, mild to moderate vascular changes were observed (Gounalan et al.

1999). Further examples of syndromes in other types of laboratory animals can be found in several papers, including review papers (El-Mofty et al. 1980; Cooper, 1980; Ushijima et al. 1983; Hirono and Yamada, 1987; Hirono, 1989; Yoshida and Saito, 1994; Yoshida and Saito, 1994a; Shahin et al. 1999).

Ptaquiloside induces chromosomal aberrations in cultured hamster lung cell lines as well as in modified Ames test on *Salmonella typhimurium* strains TA98 and TA100. The clastogenic and mutagenic effect is pH-dependent and effects can be observed after 24 hours at 4.5 mg L<sup>-1</sup> at pH 7.4/8.0 at and at 400 mg L<sup>-1</sup> at pH 5.3 (van der Hoeven et al. 1983; Matoba et al. 1987; Matsuoka et al. 1989; Nagao et al. 1989; Burkhalter et al. 1996). The low concentrations of ptaquiloside causing mutations at high pH compared to the high concentration at low pH are due to activation of ptaquiloside at alkaline pH (see Section 4.1). The cytotoxic effect of ptaquiloside on 3T3-cells (mouse fibroblast) and normal rat kidney cells (NRK) has been measured to 24 g L<sup>-1</sup> (inhibitory concentration, 50% inhibition (IC50), 48 hours). This high concentration indicates no effect of ptaquiloside in acute diseases (e.g. Acute Bracken Poisoning). However, the result may be erroneous due to lack of activation of ptaquiloside in the test experiments (see above and Chapter 4) (Ngomuo and Jones, 1996). Ptaquiloside has a much lower IC50 concentration of 2.9 mg L<sup>-1</sup> towards HL60-cells (McMorris et al. 1992). Acute toxic properties of ptaquiloside has been demonstrated in the Brine Shrimp assay (LC50 (24 hours): 15.3-62.5 mg L<sup>-1</sup>; LC50 (48 hours): 3.9-7.8 mg L<sup>-1</sup>) (Castillo et al. 1998; Castillo et al. 1999). For comparison, strychnine sulphate has a LC50 (48 hours) of 77 mg L<sup>-1</sup> (Meyer et al. 1982). A multistage model for Bracken carcinogenesis has been proposed by Shahin et al. (1999) and is shown in Figure 3.5.



**FIGURE 3.5.** Multistage cancer model for Bracken carcinogenesis. PTA = ptaquiloside. Modified after Shahin et al. (1999).

Aqueous Bracken spore extracts have proved capable of damaging human cell line DNA. However, human liver enzymes are capable of detoxifying the spore extracts (Simán et al. 2000a).

### *Oesophageal and gastric cancer in Man.*

Bracken is used for food in some parts of the world (Section 2.3) (Rymer, 1976; Haenszel et al. 1976; Hirono and Yamada, 1987; Hirono and Yamada, 1987; Fenwick, 1988; Ortega, 1994; Silva et al. 2000). In Japan and Brazil, the habit is correlated with the occurrence of oesophageal and gastric cancer (Hirayama, 1979; Marlière et al. 2000; Marlière et al. 2000a; Marlière et al. 2002). In fact, the occurrence of gastric and oesophageal cancer are between 2.1 and 8.1 times higher among people including Bracken in their daily intake *as well as* for people just living in Bracken infested areas compared when to people not having the risk behavior (Hirayama, 1979; Villalobos-Salazar, 1987; Galpin et al. 1990; Alonso-Amelot and Avendaño, 2001)!

Correlation between Bracken exposure and cancer has been demonstrated in Wales, Costa Rica, and Venezuela – areas where Bracken is not used as food (Galpin and Smith, 1986; Villalobos-Salazar, 1987; Galpin et al. 1990; Villalobos-Salazar et al. 1990; Alonso-Amelot and Avendaño, 2002). It is not finally proved that Bracken - or ptaquiloside - cause this human cancer. Contamination of water sources with Bracken leachates including ptaquiloside has been proposed as the cause. This theory might apply where shallow groundwater or surface water sources are used in the household, as the maximum *tolerable concentration* of ptaquiloside in drinking water is very low (Box 3.1 and Chapter 6) (Galpin and Smith, 1986; Galpin et al. 1990; Rasmussen et al. 2003c). Potentially carcinogenic water soluble compounds have been detected in young fronds and rhizomes (Evans et al. 1984), and recently, Rasmussen et al. (2003c) showed that ptaquiloside can be leached by rain from the fronds of var. *aquilinum*, and is found in soils below Bracken stands (see Chapter 6). Ptaquiloside is generally believed to be responsible for the carcinogenic activity of Bracken towards humans, but ptaquiloside-like substances may also contribute to the carcinogenicity of Bracken towards humans (Castillo et al. 2000).

Besides from leaching from the fern, ptaquiloside may also be transferred to the soil with urine or droppings from animals browsing on Bracken, as the urine from Bracken fed bovines is carcinogenic and therefore may contain ptaquiloside (Pamukcu et al. 1966; Smith, 1997). Another connection between humans and Bracken carcinogens is bovine milk. It has been found that up to 9% of ptaquiloside found in Bracken used to fed cows, will be excreted in their milk (Villalobos-Salazar et al. 1990; Alonso-Amelot et al. 1993;

Alonso-Amelot et al. 1996; Alonso-Amelot, 1997; Alonso-Amelot et al. 1998; Alonso-Amelot et al. 2000a; Alonso-Amelot and Avendaño, 2001): people drinking 0.5 L milk per day from cows eating subtoxic amounts of Bracken will ingest up to 13 mg of ptaquiloside per day – an amount exceeding the tolerable intake of 0.2-3.2 ng per day (Box 3.1)! Pasteurisation or boiling of milk only inactivates approx. 50-75% of the ptaquiloside (Villalobos-Salazar et al. 2000). Milk contaminated with ptaquiloside is thought to be part of the explanation for the abundance of gastric cancer in some parts of South America as high levels of ptaquiloside was found in Brackens from these areas (Villalobos-Salazar et al. 2000). The abundance of bovines with BEH and other Bracken correlated diseases in some areas of South America and Europe may indicate potential for large-scale milk-contamination (e.g. Borzacchiello et al. (2003) - see also Section 3.1). Galpin et al. (1990) demonstrated that consumption of buttermilk by infants was associated with a increased risk of obtaining gastric cancer later in life. Ptaquiloside may also be present in meat from cows browsing on Bracken (Smith, 1997).

**Box 3.1.** Tolerable concentration of ptaquiloside in Danish drinking water.

Following the guidelines from the Danish Environmental Protection Agency, a tolerable concentration of a carcinogenic compound in drinking water for adults can be calculated from results obtained in carcinogenicity tests with laboratory rodents (Miljøstyrelsen, 1992). Using existing laboratory investigations, a tolerable concentration of ptaquiloside in drinking water can be calculated to (see Appendix A):

Animal:	Mode of intake:	Tolerable concentration of ptaquiloside in drinking water: (ng L <sup>-1</sup> )	Reference:
Sprague-Dawley rats	Dried and powdered Bracken frond mixed with fodder	1.4 - 1.6	Smith et al. (1988)
ACI rats	Pure ptaquiloside mixed with fodder	0.1	Hirono et al. (1987)
Sprague-Dawley rats	Pure ptaquiloside fed intragastric	0.1 - 0.3	Hirono et al. (1984)

Burkhalter et al. (1996) proposed a human maximum intake of 700 ng per day (adult) corresponding to a safe drinking water concentration of 350 ng L<sup>-1</sup> based on Hirono et al. (1984). The reason for the discrepancy between the results from Burkhalter et al. (1996) and the tolerable concentrations listed in the table is likely due to different security assessment.

Bracken spores in breathing air have also been considered a potential environmental hazard. Up to 800 spores per L of air can be encountered in sporulating Bracken stands, and up to 10 g of spores can be produced per frond (var. *arachnoideum*). Bracken spores has proven genotoxic towards human cell lines, but only minute amounts of ptaquiloside

have been found in Bracken spores (author; E. Sheffield, the University of Manchester, Manchester, England; and B. Smith, the Royal Veterinary and Agricultural University, Frederiksberg, Denmark unpublished results) (Evans and Galpin, 1990; Simán et al. 1999; Simán et al. 2000; Alonso-Amelot et al. 2001).

The development of oesophageal cancer in Man may be caused by human papilloma virus (HPV-16 and/or 18) interacting with Bracken carcinogens (ptaquiloside and/or quercetin, Figure 3.4) (Chang et al. 1990; Togawa et al. 1994; Suzuk et al. 1996; Campo et al. 2000; Saveria Campo, 2002).

Even though many indices exist for Bracken and/or ptaquiloside induced cancer among humans, the strength of the evidence must be questioned. The results of the animal tests are quite conclusive showing carcinogenicity of Bracken in general and of ptaquiloside in particular. The evidence of human cancer is less conclusive. Epidemiological studies indicate higher abundance of cancer in areas with many Brackens and/or high concentrations of ptaquiloside in the fronds, as well as case-control studies for example indicate higher cancer rates among people living in Bracken infested areas as child compared to those who did not. These retrospective studies should be treated with caution as they include very small populations and are - retrospective (Wilson et al. 1998; Brown et al. 1999). However, the work of Marlière et al. (2002) is very convincing showing a significant higher occurrence of oesophageal cancer among people eating Bracken than people not ingesting Bracken. No raised level of gastric cancer could be observed in this study.

Bracken in itself is classified as possible carcinogenic (Group 2B) by the International Agency for Research on Cancer (IARC - WHO), while ptaquiloside occurs on the Dutch list of cancer promoting substances (Anonymous, 2000a). The British Department of Health evaluated the carcinogenic and epidemiological evidence in 1993, and concluded that the risk to the population was very low (Anonymous, 1998).

### **3.2 Chemical interaction with plants, fungi, and insects.**

Several studies have been performed on the allelopathic potential of Bracken compounds and leachates, but the results have been quite contradictory (e.g. den Ouden (1995) and Dolling (1996)). The reason is likely different methods applied and differences in susceptibility of target species for the allelopathic substances. For example, seedlings of Scots Pine (*Pinus sylvestris* L.) and Norway Spruce (*Picea abies* (L.) Karst.) respond quite differently to

Bracken litter leachates (var. *aquilinum* and/or *latiusculum*) (Dolling, 1996), while var. *esculentum* used as mulch appears to have no phytotoxic effect at all (Taylor and Thomson, 1998). A test of the inhibitory effects of volatile Bracken constituents, fresh fronds and pinnule extracts (vars. *aquilinum* and/or *latiusculum*), showed inhibitory effects on Scots Pine and Aspen (*Populus tremula* L.). The most pronounced effects were observed in spring and autumn (Dolling et al. 1994).

The allelopathic strategy of Bracken is related to the climatic conditions of the fern (Gliessman, 1976; Gliessman and Muller, 1978; den Ouden, 1995):

- Humid areas: Allelopathic substances are released throughout the year, primarily from the fronds.
- Semiarid areas: Allelopathic substances are released from old or dead fronds at the start of the wet season.
- Temperate areas: Allelopathic substances are released from the litter, rhizomes and roots in the spring (after snow melting starts).

As mentioned earlier, the allelopathic substances may also be released from the fronds in semiarid and temperate areas in spring and autumn.

The organisation of Bracken litter - horizontal layering – can constrain root development (den Ouden and Vogels, 1997), and smothering by dying fronds and litter can kill young seedlings (Humphrey and Swaine, 1997).

Bracken is also capable of changing soil properties and soil solution chemistry hereby changing the growth conditions of other plants and of the fern it selves. Aluminium toxicity due to increased levels of aluminium in the soil solution caused by the acidifying effect of Bracken leachates has been proposed as an integral part of the ferns chemical interactions with competing plants (Johnson-Maynard et al. 1997; Johnson-Maynard et al. 1998). However, the aluminium in soil solution below Bracken stands is likely complexed by dissolved organic matter found in the leachates (Johnson-Maynard et al. 1998) (see Section 2.1).

At least 100 species of fungi have been found on dead Bracken fronds and in Bracken derived litter (e.g. *Cladosporium* sp., *Trichoderma* sp., and *Penicillium* sp. – see Frankland (1976) for further information). Only few species are pathogenic to living fronds or prothalli (Figure 2.3) (Hutchinson, 1976; Frankland, 1976).

Few insects are generally observed feeding on living or dead Bracken (e.g. Frankland (1976) and authors own experience). Toxic substances in Bracken or the occurrence of constituents with deterrent activity might be the cause. The deterrent activity of var. *aquilinum* seems to have three annual peaks (Cooper-Driver et al. 1977; Jones and Firm, 1979; Jones, 1983; Schreiner et al. 1984):

1. Early spring: Likely due to prunasin (cyanogenesis), ptaquiloside and pterosin F.
2. Autumn: Likely due to tannins.
3. Winter: Likely due to tannins and lignin.

The highly complex and variable chemical composition of Bracken might be one of the keys explaining the success of Bracken, as it hinders simple adaptation of insects to the chemical composition of the fern (see Section 3.3) (Jones, 1983). However, some authors like Jones (1983) question the general statement of few insects eating Bracken. For example, in Britain 40 species have been found feeding on var. *aquilinum*.

Most studies on Bracken allelopathy have been performed on individual compounds, but it may be necessary to investigate the potential synergetic effects of the different Bracken compounds to fully understand the allelopathic and deterrent activity of Bracken (Jones, 1983).

### **3.3 Toxic and allelopathic compounds in Bracken**

The variation in the chemical composition of Bracken is immense - more than 100 secondary metabolites have been identified in Bracken so far (Fenwick, 1988). The complicated chemical composition may explain the worldwide distribution and success of Brackens in a wide range of ecosystems. A collection of these secondary metabolites is presented below - especially those suspected of having toxic effects towards animals.

#### *Illudanes and related compounds.*

Several compounds having an illudane skeleton are found in Bracken. The most investigated of these is the norsesquiterpene glucoside ptaquiloside (syn.: aquilide A, braxin C - Figure 3.6) (Niwa et al. 1983; Niwa et al. 1983a; van der Hoeven et al. 1983; Hirono et al. 1984). Other so-called *ptaquiloside-like compounds* in Bracken comprises: *iso*-ptaquiloside, caudatoside, and ptaquiloside Z (Figure 3.6) (Saito et al. 1990; Koyama et al. 1991; Potter and Pitman, 1994; Castillo et al. 1997; Castillo et al. 1998; Castillo et al.

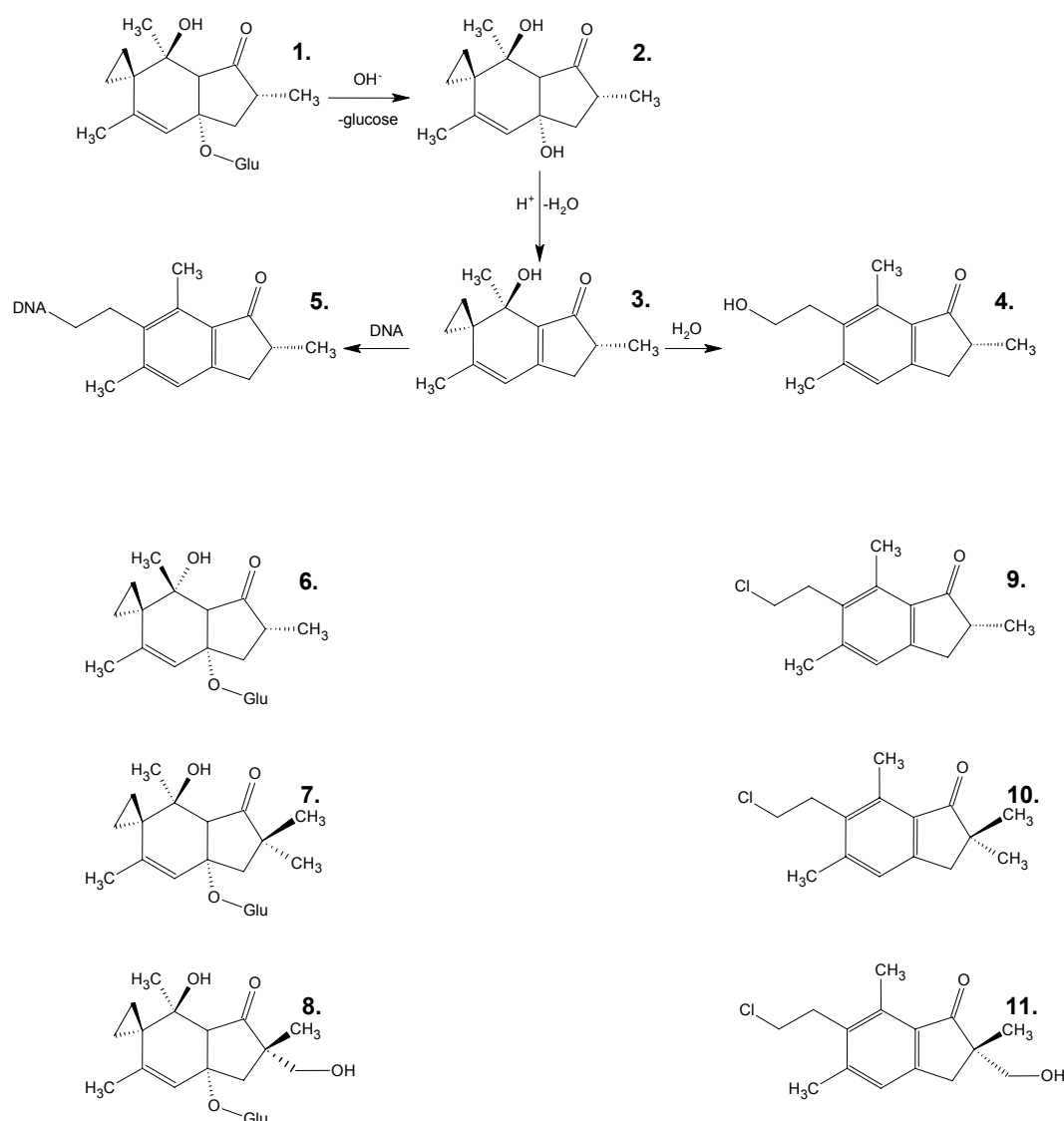


2000; Potter, 2000). Ptaquiloside is the most predominant of these compounds (Alonso-Amelot and Avendaño, 2002). Ptaquiloside-like compounds are also found in other ferns from the Dennstaedtiaceae (e.g. hypacrone and hypolosite A, B and C from *Hypolepis punctata* Mett. and dennstoside A from *Dennstaedtia hirsta* Mett.) (Hayashi et al. 1977; Matsuoka et al. 1989; Saito et al. 1989; Saito et al. 1990; Koyama et al. 1991; Potter, 2000).

Ptaquiloside is acute toxic, clastogenic, mutagenic, carcinogenic and cytotoxic (van der Hoeven et al. 1983; Matsuoka et al. 1989). However, a few authors disagree (see Section 3.1: *Symptoms found in laboratory animals, bacteria and cell cultures*). Ptaquiloside will be described in detail in Chapter 4. Ptaquiloside is unstable in aqueous solution and will transform to give stable indan-1-one derivatives, such as pterosin B, the most common pterosin (at least 29 different pterosins exist in Bracken, Figure 3.6) (Saito et al. 1975; Fukuoka et al. 1978; Jones, 1983; Saito et al. 1989; Castillo et al. 1997). The high number of pterosins indicates the presence of other ptaquiloside-like compounds than those mentioned above, as pterosins are likely to have formed from ptaquiloside-like compounds. The different pterosins are present in fronds and rhizomes, most notably pterosin B. The content ranges between 20 and 2,100  $\mu\text{g g}^{-1}$  in fronds and between 10 and 100  $\mu\text{g g}^{-1}$  in rhizomes (var. *latiusculum*). In fronds from vars. *caudatum* and *arachnoideum* pterosin B contents between 6 and 520  $\mu\text{g g}^{-1}$  and 0.1 and 100  $\mu\text{g g}^{-1}$  are found respectively (Saito et al. 1989).

Several pterosins (especially pterosin F) exhibit cytotoxic effects and antibacterial activity, but these compounds are not suspected of causing cancer (Evans et al. 1983; Finkielstein et al. 1999). Pterosin F has been found in the range 1-27  $\mu\text{g g}^{-1}$  (fresh-weight, var. *aquilinum* frond, highest concentrations found in spring) (Jones and Firn, 1979). Pterosin C has an IC<sub>50</sub> concentration of 2.7  $\mu\text{g mL}^{-1}$  towards HL60-cells (McMorris et al. 1992). Pterosin B does not show any mutagenic activity, but some cytotoxic effects towards HeLa cells and toxicity towards the following soil and plant borne bacteria and fungi has been observed (minimum inhibitory concentration: 100  $\mu\text{g mL}^{-1}$ ): *Bacillus subtilis*, *Alternaria tenuis*, *Ceratocystis fimbriata*, *Endothia parasitica*, *Guignaria laricina*, *Rhizoctonia solani*, and *Trametes sangunea* (Kobayashi et al. 1975; Saito et al. 1975; Yoshihira et al. 1978; van der Hoeven et al. 1983; Niwa et al. 1983; Nagao et al. 1989).

Pteridoside is a stable protoilludane sesquiterpene glycoside found in var. *caudatum*. It is acute toxic (LC<sub>50</sub> (48 hours, Brine Shrimp assay): 63  $\text{mg L}^{-1}$ ) (Castillo et al. 1999; Castillo et al. 2000).



**FIGURE 3.6.** Ptaquiloside, ptaquiloside reactions, some chlorinated pterosins, and ptaquiloside-like compounds from Bracken. 1: Ptaquiloside. 2: Ptaquilosin. 3: Activated ptaquiloside (dienone intermediate). 4: Pterosin B (acid hydrolysis of ptaquiloside yields pterosin B as well). 5: DNA-adduct. 6: *Iso*-ptaquiloside. 7: Ptaquiloside Z (syn: methylptaquiloside). 8: Caudatoside. 9: Pterosin F. 10: Pterosin H. 11: Pterosin K. Reaction scheme partly after Shahin et al. (1999).

Pterosides are glycosides of pterosins. Four compounds have been identified in Bracken (e.g. pteroside B and braxin A2) (Fenwick, 1988; Alonso-Amelot et al. 1992). The content is in the same range as the ptaquiloside-like substances ( $60\text{-}1,000 \mu\text{g g}^{-1}$  in fronds,  $20\text{-}80 \mu\text{g g}^{-1}$  in spores and  $400\text{-}9,500 \mu\text{g g}^{-1}$  in rhizomes) (Saito et al. 1975; Fukuoka et al. 1978; Saito and Mochizuki, 1986).

### *Cyanogenic glycosides and potential for cyanogenesis.*

Prunasin is the only cyanogenic glycoside identified in Bracken (Figure 3.7). Up to 1,600  $\mu\text{g g}^{-1}$  are found in var. *aquilinum* (fronds, fresh weight) (Kofod and Eyjolfsson, 1966; Cooper-Driver and Swain, 1976). Enzymatic hydrolysis of prunasin yields HCN - up to 400  $\mu\text{g g-fresh-weight}^{-1}$  has been observed (pinnae tips, var. *aquilinum*). Hydrolysis of the glucose bond is necessary for the toxic expression of prunasin and the release of HCN (Majak, 2001). Not all Brackens contain the  $\beta$ -glycosidase enzyme necessary for release of HCN, but many rumen bacteria possess this activity making ruminants vulnerable anyway (Cooper-Driver et al. 1977; Majak, 2001). Cyanogenic genotypes have been identified in vars. *aquilinum* and *esculentum* (Cooper-Driver and Swain, 1976; Hadfield and Dyer, 1988; Low and Thomson, 1990). Prunasin may be extruded from Bracken as HCN can be found in organic soil materials inside Bracken stands at contents up to 12.2  $\text{nmol g}^{-1}$ ) (Dartnall and Burns, 1987). Prunasin may act as deterrent towards herbivores due to the potential for cyanogenesis and/or due to its bitter taste. The highest amounts of prunasin and potential for cyanogenesis are found in the spring (var. *aquilinum*) (Cooper-Driver et al. 1977; Schreiner et al. 1984).

### *Flavonol and other glycosides.*

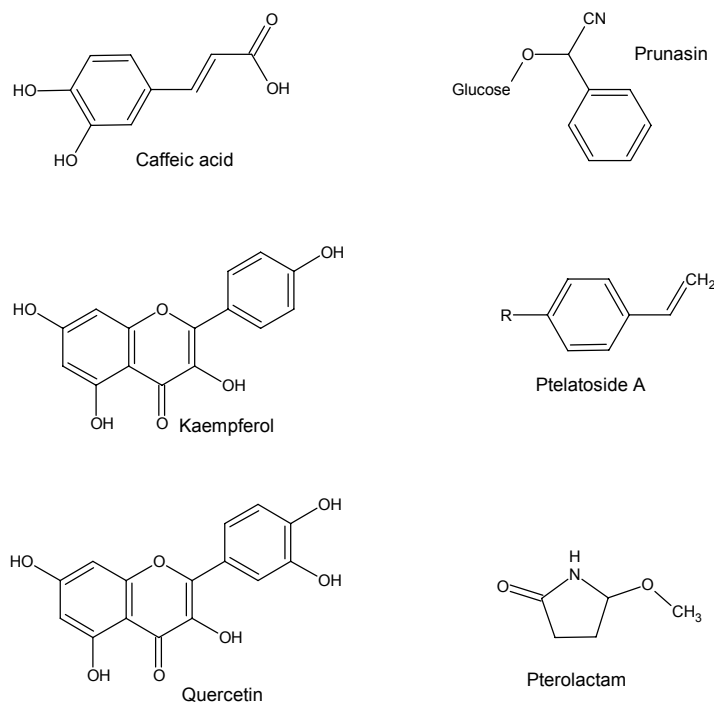
Several glycosides (at least 17 compounds) of quercetin, quercitrin, rhamnetin and kaempferol have been isolated from Bracken (Figure 3.7). Maximum contents are reported in the range 500-1,100  $\mu\text{g g}^{-1}$  (Nakabayashi, 1955; Cooper-Driver and Swain, 1976; Bjeldanes and Chang, 1977; Fukuoka et al. 1978; Fukuoka, 1982; Tanaka et al. 1993; Imperato, 1995; Imperato, 1996; Imperato and Minutiello, 1997; Imperato, 1997; Imperato, 1998; Imperato, 1998a). The role of the flavonols is complex as the individual compounds exhibit different properties towards different insects, ranging from feeding stimulation to deterrence and growth inhibition (Jones, 1983).

Four *p*-hydroxystyrene glycosides have been identified in Bracken: *p*-hydroxystyrene  $\beta$ -vicianoside and ptelatosides A, B and C (Figure 3.7) (Ojika et al. 1985; Ojika et al. 1987a; Tanaka et al. 1993).

The ecdysteroid glycoside ponasteroside A (warabisterone) and (5S,6S,9S,10S)-15-hydroxycadina-3,11-dien-2-one has been identified in var. *latiusculum* (Takemoto et al. 1968; Tanaka et al. 1993).

Braxin A1 is present in rhizomes at concentrations up to 600  $\mu\text{g g}^{-1}$  (Saito and Mochizuki, 1986; Saito et al. 1987; Saito et al. 1990a).

Besides from the glycosides, a range of sugars comprising: glucose, galactose, mannose, glycuronic acid, galacturonic acid, rhamnose, fucose, xylose, arabinose, and sulphoquinovose are present in Bracken (Duncan and Jarvis, 1976).



**FIGURE 3.7.** Overview of some secondary metabolites other than illudanes and illudane-like compounds found in Bracken. R =  $\beta$ -D-xylopyranosyl- $\beta$ -D-glucopyranosyl.

### Flavonoids.

The flavonoids kaempferol, apigenin, and quercetin are pigments producing red and yellow colours (Figure 3.7). Kaempferol and quercetin are aglycons of the corresponding flavonol glycosides. They occur in the range 10,000-25,000  $\mu\text{g g}^{-1}$  – the highest contents observed in spring and early autumn. More flavonoids are produced in open land in contrast to shaded habitats (Cooper-Driver et al. 1977; Smith and Seawright, 1995; Imperato, 1997; Musonda and Chipman, 2000). The flavonoids do not have any deterrent effects towards herbivores, but may have an antimicrobial role (Jones, 1983). Quercetin has been shown to have mutagenic activity towards *Salmonella typhimurium*, carcinogenic activity towards Rats, and might be involved in the papilloma virus carcinogenesis (Figure 3.4) (Cooper-Driver et al. 1977; Pamukcu et al. 1980; Ngomuo and Jones, 1996).

### *Simple phenolic compounds.*

The following cinnamic acid and benzoic acid derivatives has been isolated from Bracken: Benzoic acid, caffeic acid, *o*-coumaric acid, *p*-coumaric acid (*p*-hydroxycinnamic acid), ferulic acid, *p*-hydroxybenzoic acid, protocatechuric acid, vanillic acid, shikimic acid, 5-*o*-caffeoylshikimic acid, chlorogenic acid, dicaffeoyl-tartaric acid, salicylic acid, succinic acid, catecholamines, and *p*-hydroxy benzaldehyde (Figure 3.7) (Bohm and Tryon, 1967; Glass and Bohm, 1969; Somogyi, 1971; Evans, 1976; Whitehead et al. 1982; Fukuoka, 1982; Jones, 1983; Fenwick, 1988; Tanaka et al. 1993; Smith and Seawright, 1995; Alonso-Amelot et al. 1995; Anonymous, 1999; Alonso-Amelot et al. 2001). Some variation in the content of phenolic acids between varieties has been observed (e.g. between vars. *latiusculum* and *caudatum*) (Bohm and Tryon, 1967). The amounts are variable and heritage may influence the abundance of the individual compounds. Contents up to 100  $\mu\text{g g-C}^{-1}$  of *p*-hydroxy benzaldehyde, *p*-coumaric, *p*-hydroxy benzoic, and vanillic acid have been found in aqueous rhizome extracts (Whitehead et al. 1982). Cinnamic acid derivatives seem to have higher deterrent activity than benzoic acid derivatives (Jones, 1983).

### *Other compounds.*

The epicuticular wax of fronds (vars. *aquilinum* and *esculentum*) consists of 92% C<sub>40</sub>-C<sub>50</sub> alkyl esters, 2% C<sub>24</sub>-C<sub>32</sub> n-alkanols, and 2% C<sub>27</sub>-C<sub>31</sub> hydrocarbons (Baker and Gaskin, 1987).

At least four different phytoecdysteroids (polyhydroxy sterols - analogues of insect moulting hormones) are found in Bracken fronds and rhizomes:  $\alpha$ -ecdysone, ecdysterone (20-hydroxyecdysone), ponasterone A, and pterosterone (Kaplanis et al. 1967; Takemoto et al. 1968; Takemoto et al. 1968a; Jones, 1983; El-Mofty et al. 1987; Svatos and Macek, 1994). The frond contents are low and peaks in late autumn (var. *aquilinum*: up to 53  $\mu\text{g kg}^{-1}$  fresh-weight). Frond levels are likely too low to affect insects (Alonso-Amelot et al. 2001). The levels in rhizomes are reported to be much higher, and may occur in sufficiently high concentrations to act as deterrents towards soil-insects or nematodes (Jones, 1983).

Tannins (complex phenolic polymers with quite variable structure) are found in large amounts up to 16% (dry mass)). The total concentration of condensed tannins continues to be build up in the fronds during the growth season and may influence their digestibility. More tannins seems to be produced by Bracken growing in open land compared to shaded stands (Cooper-Driver et al. 1977; Jones, 1983; Alonso-Amelot et al. 2001). Tannins may

act as deterrents towards insects and mammalian herbivores due to their astringent taste. However, interpretations of previous investigations have been problematic (see discussion in Jones (1983)).

Pterolactam has been found in var. *latiusculum* (Figure 3.7) (Takatori et al. 1972). Coumarin occurs in the fronds of var. *caudatum* (Alonso-Amelot et al. 1995), while a range of simple aliphatic carboxylic acids and esters has been identified in frond and rhizome leachates: Propanoic acid, 2-hydroxy-methylester; 2,3-butanediol; 2(3H)-furanone-dihydro; acetic acid, methoxy-ethylester; ethanol, 2,2'-/1,2ethanediylbis(oxy)/bis-3; 3-methylbutan-2-ol; butanoic acid, 4-methoxy-methylester; butanedioic acid, monomethylester; proline, 5-oxo-methylester; cyclohexanol, 2-methyl-, trans; ethanedioic acid; 2,3-butanediol; 3-methylbutan-2-ol; butanedioic acid, monomethylester; proline, 5-oxo-methylester; and methanamine (Evans et al. 1984).

## Chapter 4

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### PTAQUILOSIDE – A CARCINOGENIC TERPENE

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#### 4.1 Chemical and physical properties.

Ptaquiloside is the most intriguing secondary metabolite found in Bracken seen from a toxicological point of view, as ptaquiloside cause more than 50% of the carcinogenic activity of Bracken (van der Hoeven et al. 1983). Besides from Bracken, ptaquiloside is also found in the following ferns: *Cheilanthes farinosa* (Forsk.) Kaulf., *Cheilanthes sieberi* Kunze, *Dryopteris juxtaposita* Christ., *Histiopteris incisa* (Thunb.) J. Sm., *Pteris cretica* L., and *Onychium contiguum* Hope (Smith et al. 1989; Saito et al. 1989; Saito et al. 1990; Agnew and Lauren, 1991; Gounalan et al. 1999; Aswani Kumar et al. 2001). Ptaquiloside and ptaquiloside-like substances has not been found in other organisms.

The chemical name for ptaquiloside is (SciFinder Scholar, 2002):

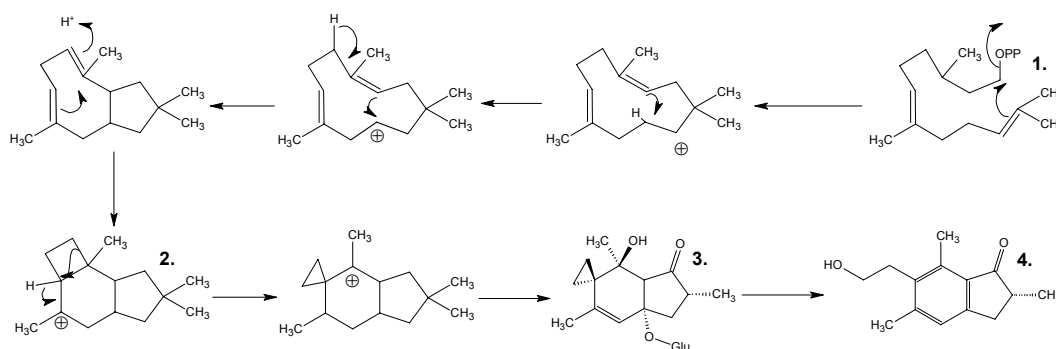
*Spiro[cyclopropane-1,5'-[5H]inden]-3'(2'H)-one,7'a-(β-D-glucopyranosyloxy)-1',3'a,4',7'a-tetrahydro-4'-hydroxy-2',4',6'-trimethyl-, (2'R,3'aR,4'S,7'aR)*

Ptaquiloside is a norsesquiterpene glucoside with an illudane skeleton (Figure 3.6). The biosynthetic pathway for formation of ptaquiloside is not known, but a possible route following humulene cyclization is shown in Figure 4.1 (Castillo et al. 2000). The exact location of ptaquiloside formation inside the fern is not known at present, but biosynthesis of sesquiterpenes is known to take place inside cells following either the acetat/mevalonat path or the glyceraldehyd 3-phosphate/pyruvat path (Chappell, 1995; Newman and Chappell, 1999).

The physical and chemical properties of ptaquiloside are presented in Table 4.1. Ptaquiloside is unstable in aqueous solution, as it decomposes under acid as well as

alkaline conditions due to hydrolysis. Under alkaline conditions, the reaction products are first ptaquilosin and then an unstable dienone formed after liberation of D-(+)-glucose (van der Hoeven et al. 1983; Niwa et al. 1983). Under acid conditions aromatization takes place and the reaction product is mainly pterosin B. Ptaquiloside is rather stable at neutral pH.

The speed of hydrolysis, i.e. disappearance of ptaquiloside from aqueous solutions, is temperature dependent and light sensitive, as the reaction is faster at higher temperature/exposed to light (Figure 3.6 and 4.2) (Ojika et al. 1987; Saito et al. 1989). The reaction is reported to be of second order with respect to ptaquiloside concentration and pH (Saito et al. 1989; Burkhalter et al. 1996). With respect to the stability of ptaquiloside, there is a discrepancy between the former authors (Saito et al. (1989) and Burkhalter et al. (1996) on one side and Ojika et al. (1987) on the other (Table 4.1). Saito et al. (1989) and Burkhalter et al. (1996) finds longer half-life's in their chemical assays than Ojika et al. (1987), although the reaction was running at higher temperature. This may be due to different initial concentrations as the rate of degradation of a second order reaction is dependent on concentration. However, Saito et al. (1989) and Burkhalter et al. (1996) does not report the initial concentrations used in their tests and this discrepancy can therefore not be explained at the moment. The rate of reaction may also depend on how pH-adjustment is accomplished as Saito et al. (1989) used prefabricated commercial buffers while Ojika et al. (1987) used simple sulphuric acid or sodium bicarbonate-buffer.



**FIGURE 4.1.** Proposed biosynthetic pathway for the formation of ptaquiloside in Bracken (var. *caudatum*). 1. Humulene. 2: Protoilludyl cation. 3: Ptaquiloside. 4: Pterosin B (formed after degradation of ptaquiloside). OPP: Pyrophosphate. Modified from Castillo et al. (2000).

A preliminary energy of activation of  $65 \text{ kJ mol}^{-1}$  was estimated by Burkhalter et al. (1996) for the degradation of ptaquiloside in Bracken extracts.



**TABLE 4.1.** Physical and chemical properties of ptaquiloside.

Property:		Reference:
Beilstein registry number	3,632,862	Anonymous (1997)
CAS registry numbers	87,625-62-5; 88,825-03-0	Anonymous (1997)
Molecular formula	C <sub>20</sub> H <sub>30</sub> O <sub>8</sub>	
Molecular weight	398.45 g mol <sup>-1</sup>	
Melting point	85-89°C (hexane, acetone)	Saito et al. (1990)
Optical rotatory power, α (CH <sub>2</sub> OH-solution, at 589 nm)	-188 deg	Saito et al. (1990)
UV absorption maxima, λ nm (Na <sub>2</sub> CO <sub>3</sub> -solution)	212; 325	Saito et al. (1989)
IR absorption maxima, ν cm <sup>-1</sup> (KBr-pellet and ATR-crystal)	3,440/3,400; 3,020; 2,970; 2,932; 2,874; 1,720/1,724; 1,643/1,640; 1,455/1,448; 1,395; 1,383; 1,279; 1,265; 1,196; 1,165/1,160; 1,080/1,074; 1,040; 933; 889	Saito et al. (1990) Ojika et al. (1987) Rasmussen et al. (2003e).
Raman scattering, ν cm <sup>-1</sup> (direct measurement)	3,086; 2,975; 2,932; 2,874; 1,717; 1,642; 1,454; 1,374; 1,275; 601.	Rasmussen et al. (2003e)
Half-life, t <sub>1/2</sub> (25°C)	2.9 hours (pH 2.0, H <sub>2</sub> SO <sub>4</sub> ) 40 min (pH 9.0) 6.3 min (pH 10.0) 69 sec (pH 11.0)	Ojika et al. (1987) All alkaline measurements were performed in a Na <sub>2</sub> CO <sub>3</sub> - buffer.
Half-life, t <sub>1/2</sub> (25°C)	40-50 days (near neutral pH)	Burkhalter et al. (1996)
Half-life, t <sub>1/2</sub> (37°C)	Approx. 12 min. (pH 11.5, buffer) Approx. 25 min. (pH 10.0, buffer) Approx. 8 hours (pH 8.5, buffer) Months or years (pH 7, buffer) Approx. 10 days (pH 5.5, buffer) Approx. 5 days (pH 4.0, buffer)	Saito et al. (1989)
Half-life, t <sub>1/2</sub> (in sunlight)	10 days (solid phase)	Saito et al. (1989)
Half-life, t <sub>1/2</sub> (at 60°C)	7 days (solid phase)	Saito et al. (1989)
Half-life, t <sub>1/2</sub> (75:25 water:methanol)	70-90 days (near neutral pH)	Burkhalter et al. (1996)
Rate constant, k (degradation, aqueous frond extract)	1.56 10 <sup>-2</sup> s <sup>-1</sup> M <sup>-1</sup>	Burkhalter et al. (1996)
Activation energy (ptaquiloside degradation)	65 kJ mol <sup>-1</sup>	Burkhalter et al. (1996)
Partition coefficient, logK <sub>ow</sub> (octanol-water)	-0.63	Rasmussen et al. (2003c)

<sup>1</sup>H and <sup>13</sup>C NMR spectral data can be obtained from several sources (Niwa et al. 1983; Ojika et al. 1987; Oelrichs et al. 1995; Castillo et al. 1997). Note that the half-life's mentioned in Table 4.1 are half-life's of the initial concentrations (half-life of second order reactions are concentration dependent).

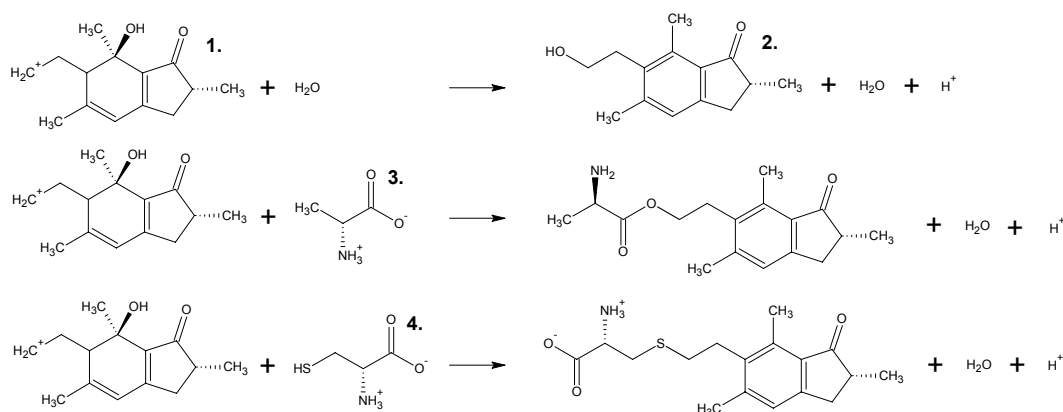
The partition coefficient for the octanol-water system for ptaquiloside is -0.63 (*log*-value), indicating high solubility and strong affinity for the aqueous phase (Rasmussen et al. 2003c).

Pterosin B is stable under both acid and alkaline conditions, while ptaquilosin and the dienone only are stable under alkaline conditions (van der Hoeven et al. 1983; Niwa et al. 1983; Kigoshi et al. 1989). The partition coefficient (*log*K<sub>ow</sub>) for pterosin B is 3.33 while for ptaquilosin it is 1.94 (*log*K<sub>ow</sub>) (Rasmussen et al. 2003c).

## 4.2 Reactivity of ptaquiloside and derivatives.

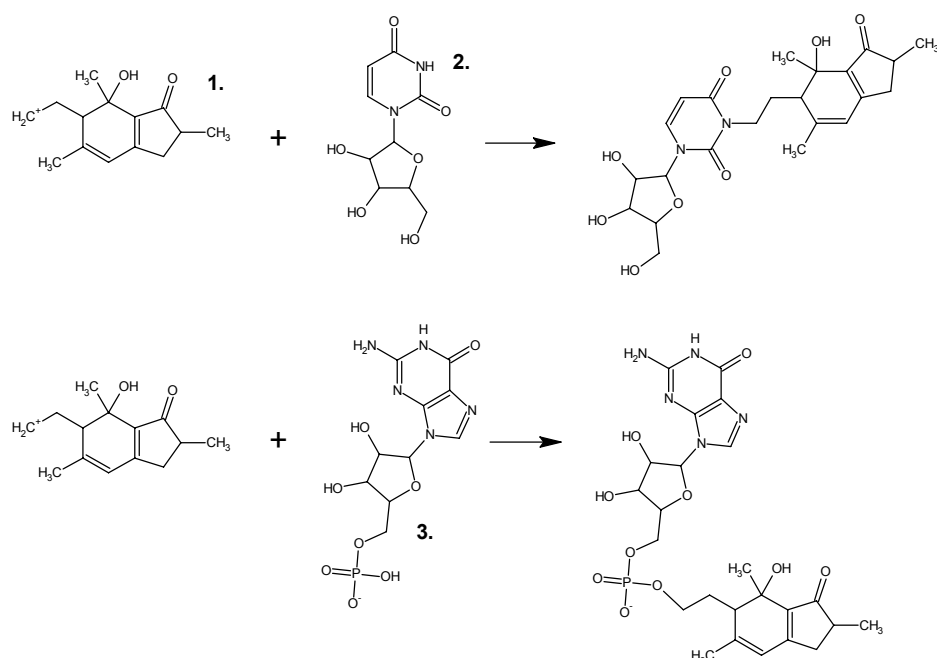
Ptaquiloside is not toxic *per se*, as it has to be activated (i.e. transformed to the unstable dienone, which is believed to be the ultimate carcinogen). Activation takes place at alkaline conditions, which help explain the localization of tumours in bovines (pH(saliva): 8.1-8.2; pH(urine): 7.5-8.5) (Fenwick, 1988).

It is the electrophilic properties of the cyclopropane ring system that cause the alkylating properties of activated ptaquiloside (the dienone) (Prakash et al. 1996). The dienone reacts willingly with nucleophiles such as water, alcohols, amines, thiol and sulphide groups in amino acids, DNA etc. Some reactions are shown in Figure 4.2-4.4 (Ojika et al. 1987; Kushida et al. 1994).



**FIGURE 4.2.** Examples of reactions between activated ptaquiloside (1) water, and the amino acids alanine (3) and cysteine (4). The reaction product in aqueous solutions is mainly pterosin B (2) (Ojika et al. 1987).

The dienone is capable of cleaving DNA by forming adducts through N-3 in adenine and N-7 in guanine Figure 4.4) (Ojika et al. 1987; Ojika et al. 1989; Kushida et al. 1994; Prakash et al. 1996). The DNA-cleaving/alkylating potential combined with potential high biological uptake and mobility due to the glucose moiety are likely responsible for the carcinogenicity and toxicity of ptaquiloside (Prakash et al. 1996). Since water is the usual solvent in chemical analysis and in Nature, pterosin B is likely a major product of all dienone reactions (Ojika et al. 1987; Ojika et al. 1989). Ptaquilosin is reported to have the same reactivity as the dienone and ptaquilosin mutagenicity has been demonstrated (van der Hoeven et al. 1983; Kigoshi et al. 1989; Nagao et al. 1989).

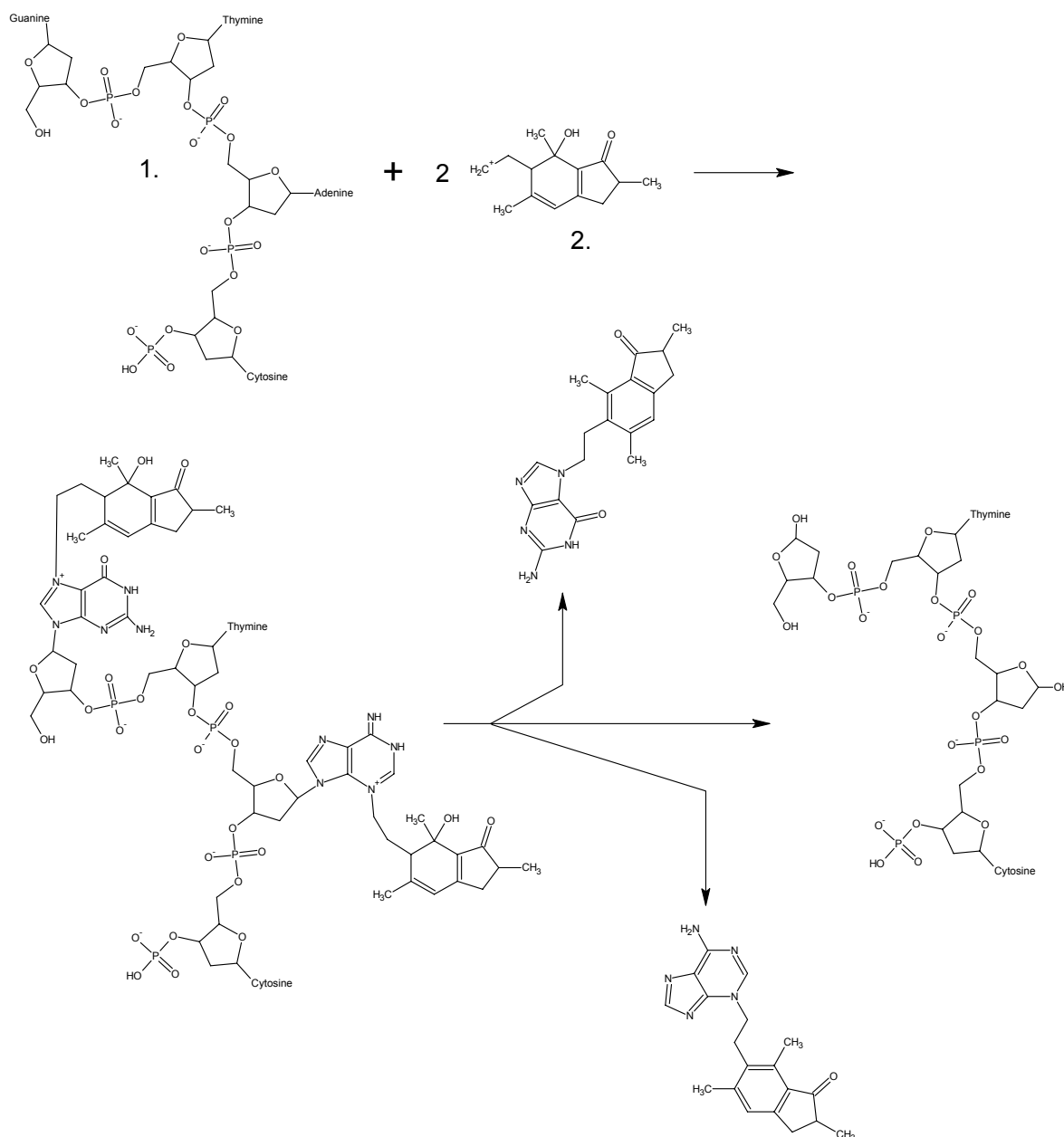


**FIGURE 4.3.** Selected reactions between activated ptaquiloside (1), the nucleoside uridine (2) and the nucleotide 5'-guanylic acid (Ojika et al. 1987).

### 4.3 Purification and determination of ptaquiloside.

No method for the synthesis of ptaquiloside has been published yet, although synthesis of ptaquilosin has been accomplished (Padwa et al. 1994; Cossy et al. 1995; Finkielstein et al. 1999). Existing methods for purification of ptaquiloside from Bracken fern have all been quite laborious, resulting in low yields and have included use of noxious organic solvents such as chloroform (Table 4.2) (van der Hoeven et al. 1983; Niwa et al. 1983; Oelrichs et al. 1995; Burkhalter et al. 1996). However, recently Rasmussen et al. (2003e) developed a fast method with less use of organic solvents. Purification has so far been made using low-pressure chromatography on different resins, partitioning between hydrophilic-hydrophobic solvents and most often a final step of preparative HPLC (high-performance liquid chromatography, sometimes repeated 2 or 3 times with different combinations of eluents; Table 4.2).

The yield and performance of the different methods described has been quite variable, most likely as result of the varying content of ptaquiloside and other secondary metabolites in Bracken (Cooper-Driver et al. 1977; Alonso-Amelot et al. 1995; Rasmussen and Hansen, 2003; Rasmussen et al. 2003). Some other methods than those reported in Table 4.2 exist, but are mainly modifications of the methods reported here.



**FIGURE 4.4.** Reactions between deoxytetranucleotide d(GTAC) (1) and activated ptaquiloside (2). After initial alkylation by activated ptaquiloside, cleavage of the nucleotide chain takes place (Kushida et al. 1994).

Three main methods exist for extraction and determination of ptaquiloside. The methods have been developed for determination of ptaquiloside extracted from Bracken but have since been modified for determination of ptaquiloside in other matrices such as milk and soil (Table 4.3).

**TABLE 4.2.** Some examples of methods for purification of ptaquiloside from Bracken fern<sup>¶</sup>.

Reference:		Method:
van der Hoeven et al. (1983)	Sample pretreatment: Extraction from fern: Purification: Concentration: Purification:	Freeze-drying. Soxhlet (light petroleum and methanol). Polyamide 6 S resin (eluted with water). RP18 Seppak minicolumn (released with methanol). Preperative HPLC (methanol-water eluent).
Niwa et al. (1983)	Sample pretreatment: Extraction from fern: Recovery from water: Concentration: Purification: Purification:	Drying and powdering. Boiling water. XAD-2 resin (released with methanol). Partitioning with butanol-water. Silica gel (eluted with chloroform-methanol). Preperative HPLC (unknown eluent).
Ojika et al. (1987)	Sample pretreatment: Extraction from fern: Recovery from water: Concentration: Purification: Purification: Purification: Purification:	Drying and powdering. Water (30°C). XAD-2 resin (released with methanol-water). Freeze-drying. Partitioning with butanol-water. Toyopearl HW-40 resin (eluted with water). Silica gel (eluted with chloroform-methanol). Develosil ODS 30/60 (eluted with methanol-water).
Saito et al. (1989)	Sample pretreatment: Extraction from fern: Concentration: Purification: Purification:	Drying and powdering. Cold water. Evaporation. Silica gel (eluted with chloroform-methanol). Preperative HPLC (3 times: ethyl acetate-methanol, methanol-water, and dichloro methan-methanol).
Oelrichs et al. (1995)	Sample pretreatment: Extraction from fern: Concentration: Purification: Purification: Recovery from water: Purification: Purification: Purification:	Freeze-drying and powdering. Soxhlet (chloroform and methanol). Evaporation. Partitioning with diethyl ether-water. Polyamide 6 S resin (eluted with water). XAD-2 resin (released with methanol). Silica gel (eluted with ethyl acetate-methanol). LH-20 (eluted with methanol). Preperative HPLC (dichloro methan-methanol).
Burkhalter et al. (1996)	Sample pretreatment: Extraction from fern: Concentration: Purification: Concentration: Purification:	Frozen ferns (liquid N <sub>2</sub> ) were grinded. Cold water. Evaporation. Bakerbond SPE C18 (eluted with water-methanol). Evaporation. Preperative HPLC (water-methanol).
Rasmussen et al. (2003)	Sample pretreatment: Extraction from fern: Purification: Recovery from water: Concentration: Purification: Recovery from water:  Concentration: Purification:	Drying and powdering. Cold water. Polyamide 6 resin (eluted with water). XAD-2 resin (released with methanol). Evaporation. Polyamide 6 resin (eluted with water). Liquid-liquid partitioning (aqueous phase sorbed onto prepacked ChemElut diatomaceous soil columns, ptaquiloside retrieved with ethyl acetate). Evaporation. Preperative HPLC (water-methanol).

<sup>¶</sup> Full description of materials (manufacturers etc.) can be found in the original articles.

**TABLE 4.3.** Some examples of methods for purification of ptaquiloside from Bracken fern<sup>¶</sup>.

Method:	Description:	References:
Organic solvent	Water extraction of fresh fronds for 2 hours. Removal of interfering substances with acetone-dichloromethane. Prepurification on a silica gel microcolumn. Quantification of ptaquiloside after conversion to pterisin B by HPLC.  The method has been modified to allow determination of ptaquiloside in milk.  Recovery of ptaquiloside (standards): 94%.	Alonso-Amelot et al. (1992) Alonso-Amelot et al. (1992a) Alonso-Amelot et al. (1993)
Polyamide resin	Water extraction of dried fronds for 1 hour. Removal of interfering substances with polyamide 6 S resin. Quantification of ptaquiloside by HPLC directly or after conversion to pterisin B.  The method has been modified to allow determination of ptaquiloside in soil and in fresh fronds.  Recovery of ptaquiloside (standards): 95%.	Agnew and Lauren (1991) Rasmussen et al. (2003d) Rasmussen et al. (2003e)
TLC	Water extraction of freeze-dried fronds for 1.5 hour. Lyophilization of aqueous extract. Quantification of ptaquiloside by two-dimensional thin layer chromatography.  Recovery of ptaquiloside: ?.	Matoba et al. (1987) Saito et al. (1989) Oelrichs et al. (1995)

<sup>¶</sup> Full description of materials (manufacturers etc.) can be found in the original articles.

Determination of ptaquiloside in Bracken fern involves three main steps:

1. Extraction of ptaquiloside from the fern. Due to the high hydrophilicity of ptaquiloside, water-extraction is most commonly used. The fern material can be fresh or dried. Care must be taken when drying the ferns as ptaquiloside may decompose at high temperatures (see below).
2. Removal of interfering substances from the extract. Lots of other compounds than ptaquiloside is usually found in Bracken extracts. To secure full resolution of ptaquiloside in the final step of quantification, these compounds must be removed. Especially presence of pterisin B in the aqueous extract can be problematic, as ptaquiloside under some analytical conditions is converted to pterisin B before quantification. Removal of pterisin B and other compounds can be performed by partitioning the aqueous extract with a hydrophobic organic solvent as in the methods of Alonso-Amelot et al. As an alternative, pterisin B can be removed with an organic polymer resin like Polyamide 6 (Fluka, Steinheim, Switzerland).

3. Quantification of ptaquiloside. Quantification is most commonly carried out on HPLC with use of external standards. Under some circumstances, ptaquiloside is converted to pterosin B (Figure 3.6) before analysis, as pterosin B is more stable than ptaquiloside and has stronger absorption. Alternatively 2-dimensional thin layer chromatography can be applied.

No comparative study has been carried out between the different methods. When looking at the recovery of standards, there seems not to be any differences between the resin and organic solvent methods. However, extraction time and sample pre-treatment may be essential for the performance of the methods. Agnew and Lauren (1991) and Burkhalter et al. (1996) found that the amount of ptaquiloside extracted decreased as time of extraction increased. Hence, extraction time should not exceed 1 hour. Drying may also decrease the amount of ptaquiloside extracted, as the compound is heat-sensitive (Saito et al. 1989).

#### **4.4 The role of ptaquiloside in the fern.**

The role of ptaquiloside in Bracken is not quite clear, but ptaquiloside might act as deterrent towards recent or past herbivores (Alonso-Amelot et al. 1992). This hypothesis is supported by the observation that only few insects feed on Bracken, and that no herbivores eat large quantities of Bracken when enough food is available. Bracken crosiers are also very delicate, fragile and sensitive to damage and attacks, and the high ptaquiloside content often observed in young fronds might be an adaptation to prevent attacks from herbivores in this sensitive stage (see Section 5.1) (Alonso-Amelot et al. 1992). However, the *protection* of the fern could just as well be carried out by other secondary metabolites than ptaquiloside. The timespan from ingestion of ptaquiloside by larger herbivores until they become sick is also rather long (days to years). As ptaquiloside does not show any immediate effect, this compound alone will probably not deter such animals. Prunasin or tannins are likely much more efficient in deterring larger herbivores (Cooper-Driver et al. 1977). Ptaquiloside and ptaquiloside-like compounds might instead serve as mobile precursors for the less hydrophilic compounds: pterosin B and F, hereby enabling the fern to allocate these compounds to points of attack. These pterosins are both reported of having some deterrent and/or antimicrobial effect (see Section 3.3) (Alonso-Amelot et al. 1992; Alonso-Amelot et al. 1995).

Another explanation for the occurrence of ptaquiloside that has been proposed states, that it is not Bracken itself that produces ptaquiloside, but a symbiotic endomycorrhizal fungus instead. Due to the close association between fungi and fern in mycorrhizal networks, the

fungi could introduce ptaquiloside into the vascular system of the fern (Schoental, 1984). This hypothesis is partly supported by the observation that some compounds very similar in structure to ptaquiloside as well as some pterosins have been identified in basidiomycetes (e.g. illudin M and S from *Omphalotus illudens/olearis* and pterosins in *Fomes annosus* and *Cyathus bulleri*) (McCloud et al. 1996; Potter, 2000; Rasmussen et al. 2003d). However, no solid evidence for the fungal infection theory exists at present. Hence, the origin and exact role of ptaquiloside in Bracken remains still to be explained.



## Chapter 5

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### PTAQUILOSIDE IN BRACKEN

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#### 5.1 Ptaquiloside content in the frond

The content of ptaquiloside in the fronds of different varieties of Bracken ranges between 0 and 15,100  $\mu\text{g g}^{-1}$  (Table 5.1). There seems to be no significant differences between the varieties (Smith et al. 1994). However, genetic heritage may influence the ptaquiloside content as different proveniences still have different ptaquiloside contents when grown under the same conditions (Smith et al. 1988; Smith et al. 1992; Smith et al. 1994; Rasmussen et al. 2003d). For example, Smith et al. (1984) found that var. *esculentum* from the southern part of Australia had higher ptaquiloside contents than ferns from the northern part, even when grown under uniform conditions. Significant differences were also observed between var. *esculentum* from Australia and New Zealand in that study. Some proveniences from vars. *esculentum* and *revolutum* do not contain ptaquiloside at all, indicating the existence of ptaquiloside chemotypes (Smith et al. 1994; Rasmussen et al. 2003d). The existence of ptaquiloside chemotypes in other Bracken varieties is yet to be unveiled. This variation is discussed further in Sections 5.3-6.

The ptaquiloside content in Bracken fronds from proveniences with ptaquiloside is quite variable. The content in fronds with ptaquiloside is usually at maximum just after the crosiers emerge from the ground. The content then decreases gradually during the growth season - but never reach zero. For example, vars. *caudatum* and *arachnoideum* both have maximum ptaquiloside contents in the crosier-stage, whereafter the content decreases rapidly to 10% of the maximum content 12-20 days after crosier emergence. The distribution of ptaquiloside seems to be constant between different pinnae in these varieties when the fronds are 42-45 days old (85% mature) (Alonso-Amelot et al. 1992; Alonso-Amelot et al. 1995). The annual variation in vars. *aquilinum* and *esculentum* from New Zealand, Scotland and Denmark do also follow this pattern, although not as marked

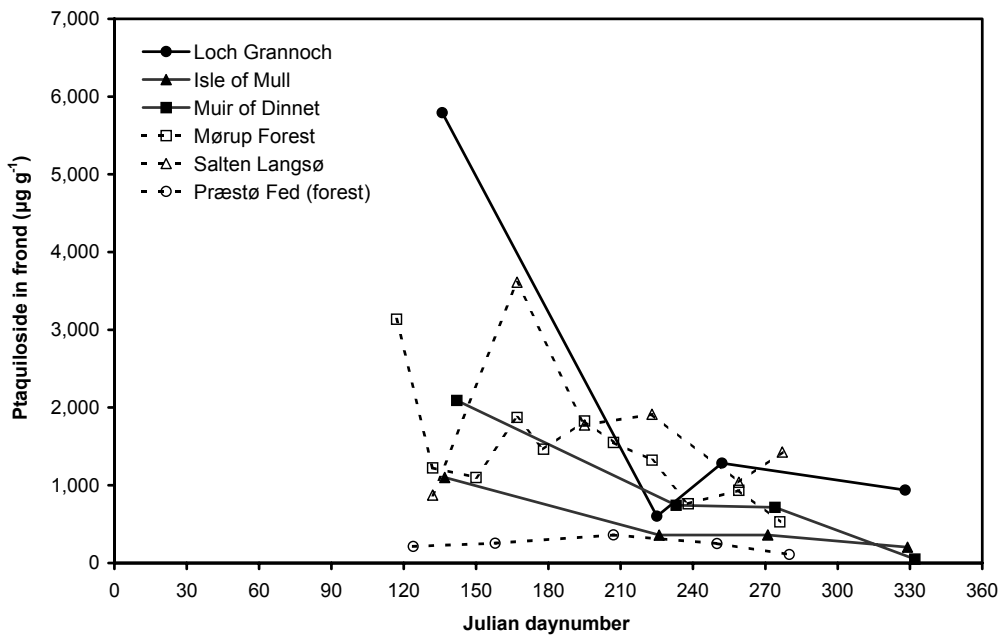
**TABLE 5.1.** Ptaquiloside contents in different Bracken varieties.

Bracken variety:		Ptaquiloside content: ( $\mu\text{g g-drymatter}^{-1}$ )		Reference:	
<i>Pteridium aquilinum</i> ssp. <i>aquilinum</i>	var. <i>aquilinum</i>	210-3,140	C <sup>¶</sup>	Burkhalter et al. (1996)	
		60-9,780	Y	Ghergariu et al. (2000)	
		110-4,450	M	Pinto et al. (1999) Rasmussen and Hansen (2003) Rasmussen et al. 2003) Rasmussen et al. 2003a) Rasmussen et al. 2003b) Smith et al. (1994)	
	var. <i>decompositum</i>	960	Y	Smith et al. (1994)	
	var. <i>latiusculum</i>	40-1,830	C/Y	Gounalan et al. (1999)	
		5-1,200	M	Saito et al. (1989) Smith et al. (1994)	
	var. <i>pseudocaudatum</i>	50	Y	Smith et al. (1994)	
	var. <i>revolutum</i>	0-1,340	Y	Smith et al. (1994)	
	<i>Pteridium aquilinum</i> ssp. <i>caudatum</i>	var. <i>arachnoideum</i>	30-15,100	C	Alonso-Amelot et al. (1995)
			0 <sup>§</sup> -2,130	Y	Smith et al. (1994)
0 <sup>§</sup> -590			M	Villalobos-Salazar et al. (2000)	
var. <i>caudatum</i>		1,880-6,820	C	Alonso-Amelot et al. (1992)	
		10-1,850	Y	Alonso-Amelot et al. (1992a)	
		110-1,960	M	Alonso-Amelot et al. (1995) Alonso-Amelot et al. (2000) Smith et al. (1994) Villalobos-Salazar et al. (2000)	
var. <i>esculentum</i>		700-7,000	C	Smith et al. (1988)	
		0-13,260	Y	Smith et al. (1990)	
		0-8,580	M	Smith et al. (1993) Smith et al. (1994) Rasmussen et al. (2003d)	
var. <i>yarrabense</i>		50-1,330	Y	Smith et al. (1994)	

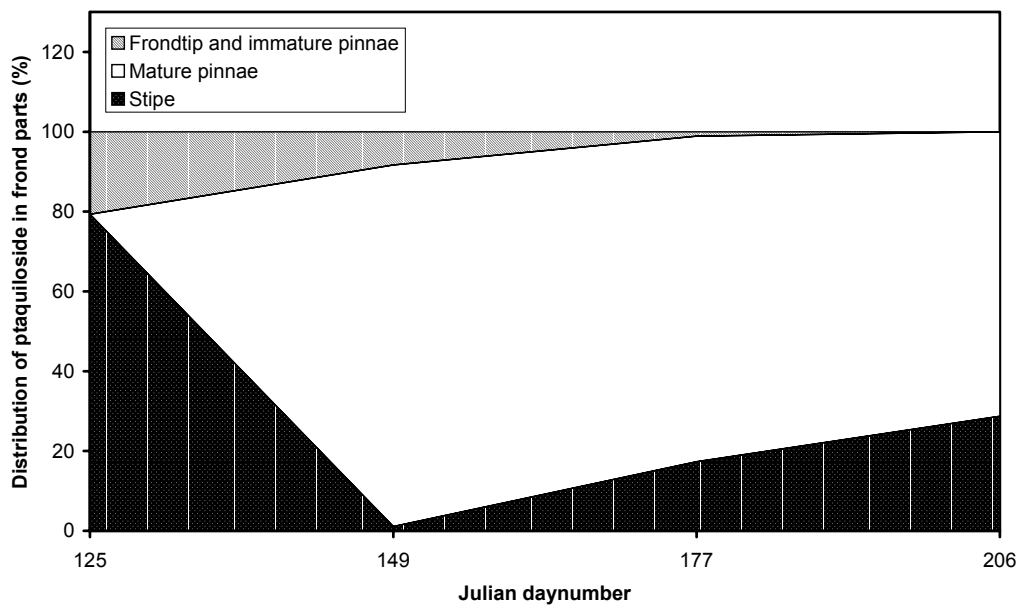
<sup>¶</sup> C: Crosiers; Y: Young developing fronds; M: Mature fronds. <sup>§</sup> Trace amounts (Alonso-Amelot et al. 1995).

(Figure 5.1 and 5.2) (Smith et al. 1990; Rasmussen and Hansen, 2003; Rasmussen et al. 2003). A Japanese survey of var. *latiusculum* fronds showed no marked annual variation, except from a single stand with a shorter summer season compared to the other sites (Saito et al. 1989).

Ptaquiloside is quite stable inside dead fronds exposed to sunlight, but protected from rain. An initial half-life of 3 weeks was observed for var. *latiusculum* fronds kept that way, but the degradation rate seemed to decline with time (Saito et al. 1989).



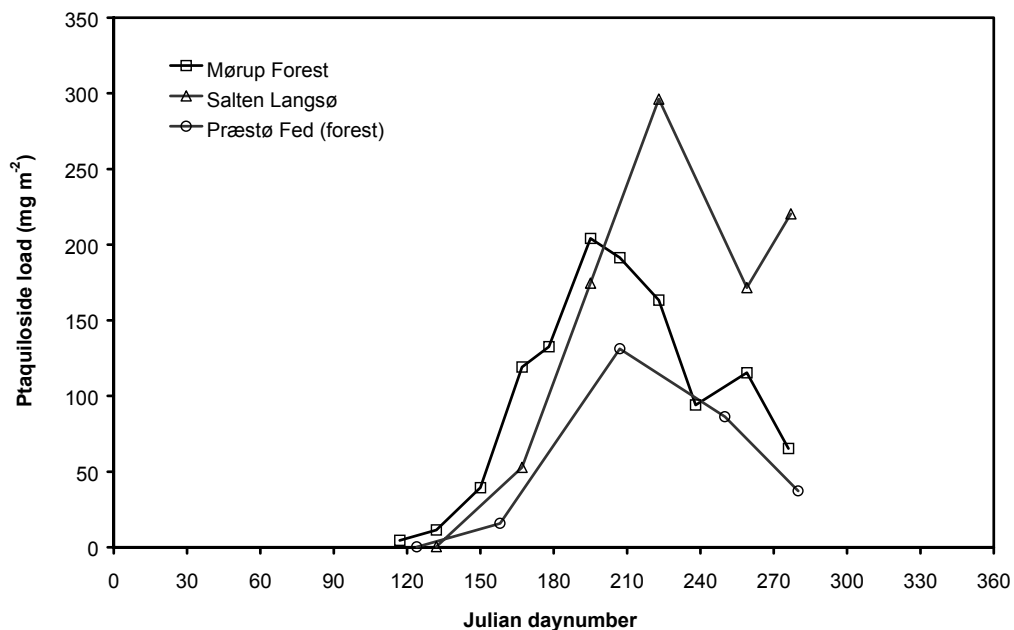
**FIGURE 5.1.** Ptaquiloside contents in *var. aquilinum* fronds in Denmark (open symbols) and Scotland (filled symbols). Data from Rasmussen and Hansen (2003) and Rasmussen et al. (2003).



**FIGURE 5.2.** Ptaquiloside contents in different frond parts of *var. aquilinum* in Denmark. The different frond parts from Denmark were collected at Mørup Forest and Salten Langsø in 2000. The material was analysed as described in Rasmussen and Hansen (2003). Author, unpublished results.

### *Ptaquiloside load in the standing biomass.*

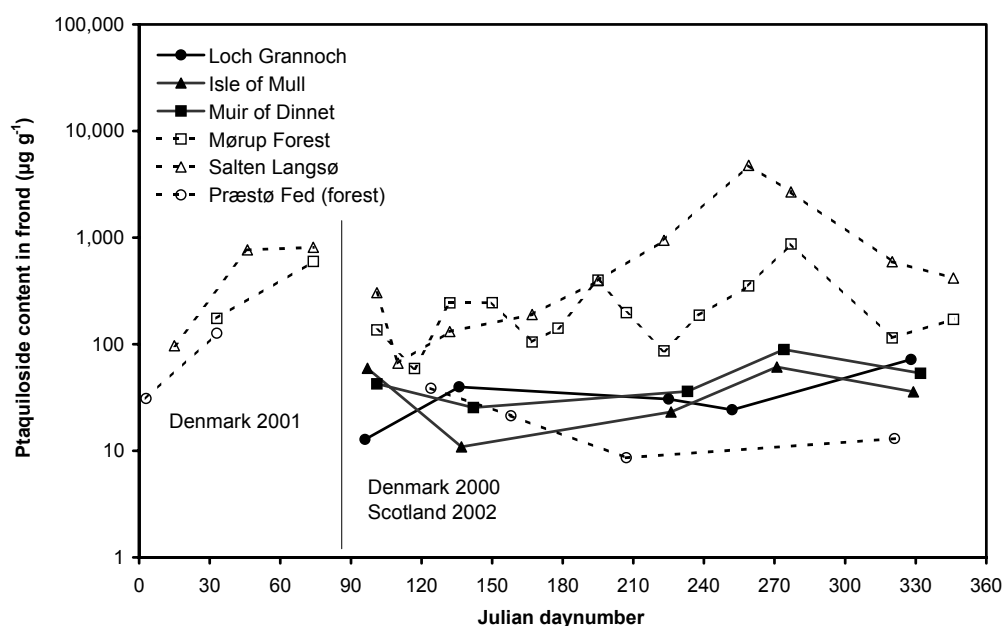
The ptaquiloside load is the total amount found in the fronds per  $\text{m}^2$ . This is the maximum amount of ptaquiloside that can be transferred to the litter-layer when the fronds die naturally or after mowing. It is also the maximum amount that can be ingested by grazing animals or become leached from fronds by rainwater. The annual variation in the ptaquiloside load is shown in Figure 5.3 for 3 Danish sites. The maximum ptaquiloside load is found when the fronds mature in late summer. The load then declines gradually – maybe as a function of rhizome replenishment with carbohydrates and nutrients – an increase in the content of ptaquiloside in the rhizomes has been observed in the autumn (Rasmussen and Hansen, 2003). When compared to Figure 5.1, the differences in ptaquiloside contents between the Danish sites have diminished and the time of maximum content has changed somewhat from spring to late summer. This indicates more constant amounts of ptaquiloside found inside mature and senescent Bracken stands in contrast to what could be believed based on measurements of the ptaquiloside contents alone. This observation indicates rather constant ptaquiloside levels per  $\text{m}^2$  and that the amount of ptaquiloside is *diluted* into the biomass. A survey of 20 Danish Bracken sites in autumn showed that the ptaquiloside load ranged from 10 to  $300 \text{ mg m}^{-2}$  (Figure 5.3 and Rasmussen et al. 2003b).



**FIGURE 5.3.** Ptaquiloside load in the standing biomass (var. *aquilinum*, Denmark). The figure is drawn based on data from Rasmussen and Hansen (2003) and Rasmussen et al. (2003b).

## 5.2 Ptaquiloside content in rhizomes, roots and spores.

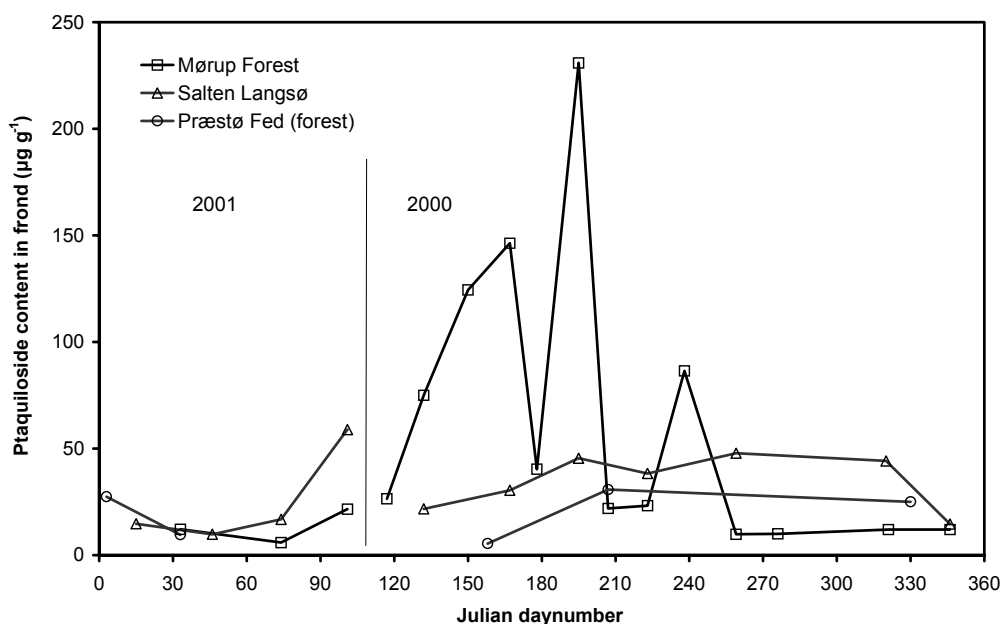
Until recently, rhizome studies have only been carried out during the growth season on vars. *aquilinum*, *latiusculum*, *caudatum* and *arachnoideum*. The results indicated relatively low contents in the rhizomes ( $<1,200 \mu\text{g g}^{-1}$ ) with the highest concentrations in vars. *aquilinum* and *latiusculum* (Saito et al. 1989; Alonso-Amelot et al. 1992; Alonso-Amelot et al. 1992a; Ghergariu et al. 2000; Rasmussen et al. 2003)). Recently, Rasmussen and Hansen (2003) and Ramussen et al. (2003) found drastic annual variations in the ptaquiloside content in the storage and the frond bearing rhizomes of var. *aquilinum* (2-7,050  $\mu\text{g g}^{-1}$ ). The lowest contents were measured during the frond growth season, and the highest concentrations just after fronds had reached maturity (Figure 5.4). Raised levels were also found in late winter/early spring. Storage rhizomes had in general a higher content of ptaquiloside than frond bearing rhizomes. The annual variation in the rhizome content were correlated with the phase of carbohydrate regeneration of the rhizome-system in autumn, and the time of rhizome- and belowground shoot-elongation as well as bud-growth in early spring (Williams and Foley, 1976; Ershova, 2000; Rasmussen and Hansen, 2003).



**FIGURE 5.4.** Ptaquiloside content in rhizomes, var. *aquilinum*, in Denmark (open symbols) and in Scotland (filled symbols). The figure is made from data in Rasmussen and Hansen (2003) and Rasmussen et al. (2003). Note logarithmic scale on abscissa.

The content of ptaquiloside in var. *aquilinum* rhizomes collected in the autumn in Scotland has been found to correlate with longitude, topography and the content of soil nitrogen. Especially nitrogen had strong negative influence on the ptaquiloside content in the rhizomes, indicating higher levels of ptaquiloside in rhizomes from Bracken growing nitrogen-depleted soils (Rasmussen et al. 2003). For further discussion of ptaquiloside variation, see Section 5.6.

Only a single study of the ptaquiloside content in roots has been carried out (var. *aquilinum*) (Rasmussen and Hansen, 2003). The content was very low compared to other parts of the fern, and was found to be highest in the growth season (5-230  $\mu\text{g g}^{-1}$ , Figure 5.5).

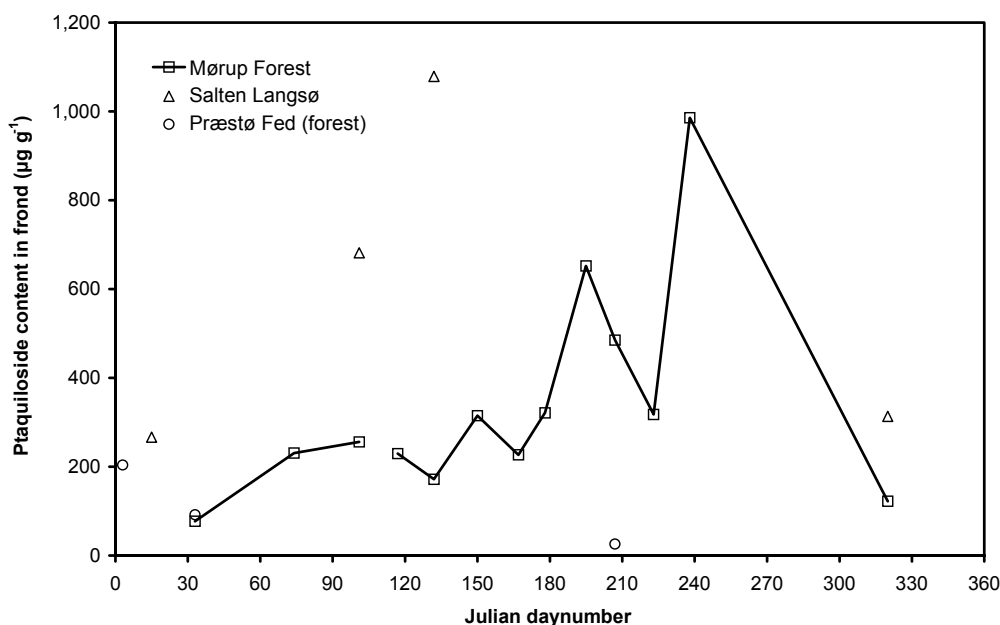


**FIGURE 5.5.** Ptaquiloside content in roots, var. *aquilinum* (Denmark). The content was only measured in duplicate in 48% of the samples due to limited sample amount. Median coefficient of variation = 10.5%. Analytical method from Rasmussen and Hansen (2003). Author, unpublished results.

The annual variation of the ptaquiloside content in rhizome buds is shown in Figure 5.6. The content was in the range 25 to 1,080  $\mu\text{g g}^{-1}$ . The content seems to increase during the growth season. Rhizome growth takes place in late summer and autumn, and the increase in the ptaquiloside content might be a mirror of that.

Large amounts of ptaquiloside are not found in Bracken spores in contrast to what has been expected due to their genotoxic/carcinogenic effects (Matoba et al. 1987; Saito et al. 1989; Evans and Galpin, 1990; Freitas et al. 2001). Recent analyses of ptaquiloside in spores (var.

*aquilinum*) showed that the content in spores was less than 20  $\mu\text{g g}^{-1}$  (author; E. Sheffield, the University of Manchester, Manchester, England; and B. Smith, the Royal Veterinary and Agricultural University, Frederiksberg, Denmark unpublished results).

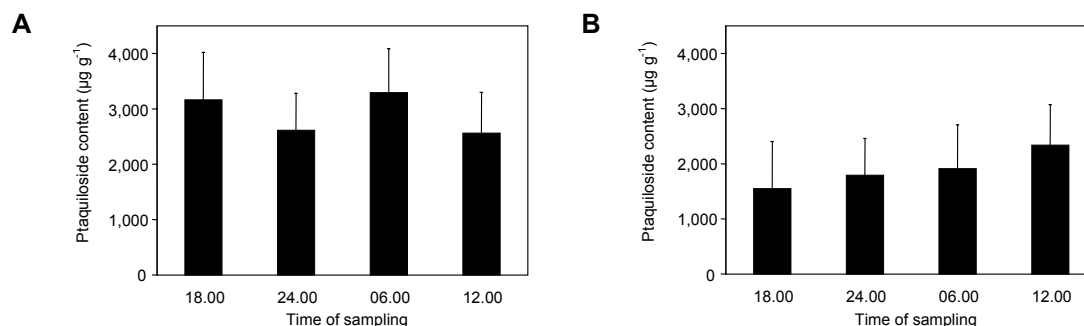


**FIGURE 5.6.** Ptaquiloside content in rhizome buds, var. *aquilinum* (Denmark). The content was only measured in duplicate in 50% of the samples due to limited sample amount. Median coefficient of variation = 9.0%. Analytical method from Rasmussen and Hansen (2003). Author, unpublished results.

### 5.3 Daily variation in the ptaquiloside content in fronds.

The content of secondary metabolites is known to be closely linked to the metabolic activity of plants (Kaufman et al. 1999). The daily variation of the ptaquiloside content in Bracken fronds was therefore investigated by Johansen (2003) (Figure 5.7). No significant differences were found between the different times of sampling. At the first time of sampling (Figure 5.7A, average ptaquiloside content:  $2,900 \pm 800 \mu\text{g g}^{-1}$ ), no increasing or decreasing trend could be observed in the ptaquiloside content during 24 hours, while the ptaquiloside content increased during the second sampling period (Figure 5.7B, average ptaquiloside content:  $1,900 \pm 730 \mu\text{g g}^{-1}$ ). Weather conditions were similar at the two times of sampling (night temperature: 10-12°C; day temperature: 20-22°C; no precipitation). This indicates no influence of the metabolic activity of the fern on the ptaquiloside content in the fronds. A decrease in the ptaquiloside content was found between the two sampling

times in accordance with previously observed ptaquiloside variation in Danish Bracken fronds with respect to age of the fronds (Rasmussen and Hansen, 2003).



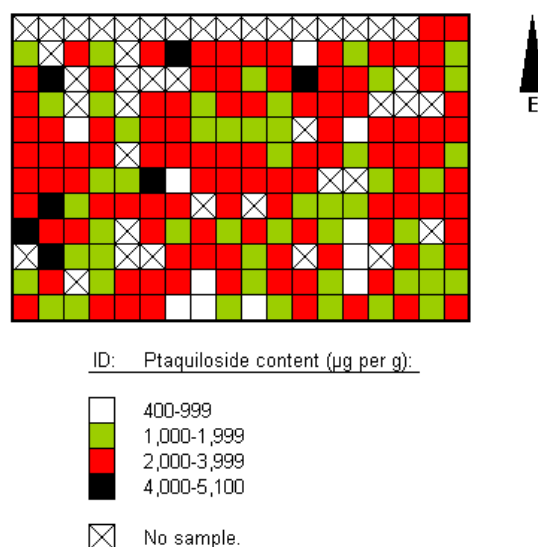
**FIGURE 5.7.** Daily variation in the ptaquiloside content of almost mature Bracken fronds (var. *aquilinum*). A: 5-6 July 2002. B: 20-21 July 2002. 20-25 randomly sampled replicates at each time of sampling, standard deviations shown. The stand was situated at Præstø Fed, Denmark below a stand of Oak (*Quercus robur* L.). Analytical conditions as in Rasmussen et al. (2003). Unpublished data by Johansen (2003).

#### 5.4 Variation in the ptaquiloside content between fronds inside a stand.

In Figure 5.8, the measurements presented in Figure 5.7 are mapped in the grid used for sampling. No correction has been made for the lower level of ptaquiloside in the fronds collected the second time of sampling. The content ranges from 440 to 5,080  $\mu\text{g g}^{-1}$  with most fronds having contents between 2,000 and 3,999  $\mu\text{g g}^{-1}$ . A few fronds had very high or low ptaquiloside contents (<1,000 and >3,999  $\mu\text{g g}^{-1}$ ). These extreme contents were apparently not linked to easily recognizable factors such as height, health status or growth stage. According to Figure 5.8 there seems not to be any pattern in the distribution of ptaquiloside in fronds inside the investigated Bracken stand, i.e. the content of ptaquiloside inside a stand seems to be more or less randomly distributed within a certain concentration range. However, some grouping of samples with extreme ptaquiloside contents could be observed.

The observed pattern of variation in ptaquiloside contents is rather similar to previous observations of cyanogenic polymorphism in Bracken stands (var. *aquilinum*) (Cooper-Driver and Swain, 1976). This polymorphism may be caused by either the possible formation of genetically different sporophytes from the same spore or that several genotypes are present in the stand originating from different spores (Jones, 1983; Wolf et al. 1988; Wolf et al. 1988a; Sheffield et al. 1989; Wolf et al. 1990).

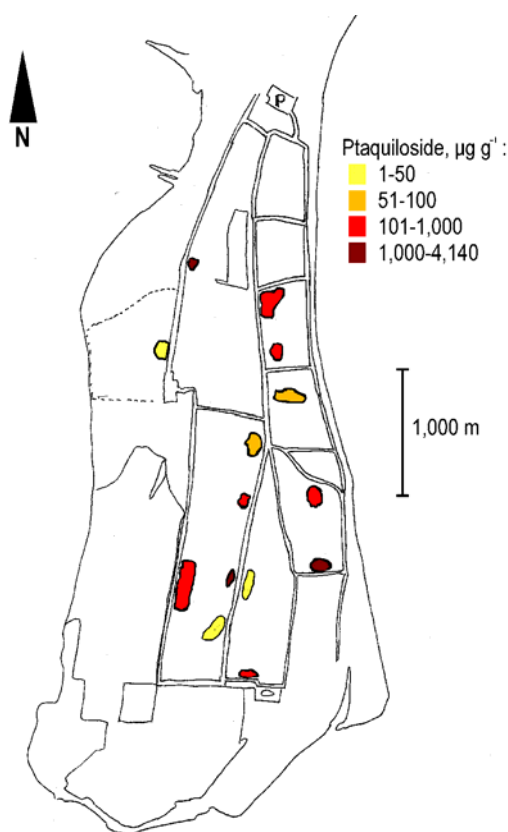




**FIGURE 5.8.** Variation in the ptaquiloside content between individual Bracken fronds (5-21 July 2003, var. *aquilinum*). One frond was sampled from the central part of each square (2x2 m). Details as in Figure 5.7 (Johansen, 2003).

### 5.5 Small-scale geographical variation in the ptaquiloside content of Bracken.

Several investigations of the local variation have been carried out on vars. *aquilinum*, *esculentum*, *caudatum*, and *arachnoideum* (Alonso-Amelot et al. 1992; Alonso-Amelot et al. 1995; Rasmussen et al. 2003a; Rasmussen et al. 2003d). The content seems to be quite variable. In a survey of mature var. *aquilinum* fronds, Rasmussen et al. (2003a) found contents between 210 and 2,150  $\mu\text{g g}^{-1}$  in 12 stands located at 4 geographically distinct locations. Significant differences were found between the locations. A detailed survey of the ptaquiloside content in the fronds of young var. *aquilinum* at Præstø Fed in Denmark did also exhibit quite large variation (Figure 5.8). The same kind of variation was also observed between young var. *esculentum* fronds at a farm in New Zealand (Rasmussen et al. 2003d). In the latter study, stands without any ptaquiloside at all were also found! The variation in the ptaquiloside content in the fronds with ptaquiloside (var. *esculentum*) was positively correlated with the frond development stage (number of pinnae) and the altitude of the stand. However, the model could explain only 76% of the data. The geographical variation of the ptaquiloside content in fronds will be discussed further in Section 5.6) (Cooper-Driver and Swain, 1976; Cooper-Driver et al. 1977; Jones, 1983).



**FIGURE 5.9.** Local variation in the ptaquiloside content in Bracken fronds (var. *aquilinum*), Præstø Fed, Denmark. Analytical method as in Rasmussen et al. (2003). Author, unpublished results.

### 5.6 Large-scale geographical variation in the ptaquiloside content in Bracken fronds.

Large-scale variation in the concentration of terpenoid constituents of higher plants is generally believed to occur in response to variations in genetic, biochemical and ecological conditions. For example, genetic mutations can affect biosynthesis of secondary metabolites; mixtures of terpenes may result of the ability of some plant enzymes to produce different compounds; and the complex composition of the mixture of secondary metabolites may have evolved in response to herbivore pressure (Langenheim, 1994). Besides from these biotic causes of natural variation in the content of secondary metabolites, abiotic factors may influence the levels found as well. According to Gershenzon and Croteau (1991, cited from Langenheim (1994)), the content of terpenes generally increases with increased light intensity and decreases with increased abundance of nitrogen, phosphate and potassium. However, the abundance of secondary metabolites is

quite complex due to the interaction between abiotic and biotic factors (Langenheim, 1995; Kaufman et al. 1999).

Several studies have been carried out to monitor the large-scale geographical variation in the ptaquiloside content of Bracken fronds. In New Zealand and Australia, large geographical surveys comprising hundreds of var. *esculentum* stands have been carried out while geographically less extensive studies have been performed in Costa Rica and Venezuela (vars. *arachnoideum* and *caudatum*) (Smith et al. 1994; Alonso-Amelot et al. 1995; Alonso-Amelot et al. 2000; Villalobos-Salazar et al. 2000; Rasmussen et al. 2003d).

The ptaquiloside content seems to be correlated with altitude for stands in South America (vars. *arachnoideum* and *caudatum*) (Alonso-Amelot et al. 1995; Alonso-Amelot et al. 2000; Villalobos-Salazar et al. 2000; Alonso-Amelot et al. 2001). However, the correlation is complex since Villalobos-Salazar et al. (2000) found a positive correlation between altitude and ptaquiloside content for vars. *arachnoideum* and *caudatum*, while Alonso-Amelot et al. (1995) found a negative correlation for var. *caudatum*. The South American studies did only include few Bracken stands and the observed variation may therefore not be statistically significant. Similar confusing results were obtained in New Zealand (var. *esculentum*), where the ptaquiloside content was positively correlated with altitude in a confined region. However, negative effect of altitude was found in a larger study comprising Bracken stands from all over New Zealand (Rasmussen et al. 2003d). Alonso-Amelot et al. (2000) suggested that the observed decrease in ptaquiloside content with height may be due to lower temperatures at higher altitudes imposing restrictions in the metabolic rate and production of secondary metabolites. As evidence for this hypothesis, Alonso-Amelot et al. (2001) found increased levels of ptaquiloside in mountain-rhizomes moved from high altitudes to lower altitudes. The ptaquiloside production by these rhizomes rose to levels comparable to surrounding low-altitude Brackens. Other edaphic variables such as soil drainage, soil composition and precipitation may also explain this variation, as they are known to vary as result of altitude (Jenny, 1980). Further studies are needed to fully explain the effect of altitude on the ptaquiloside content in fronds.

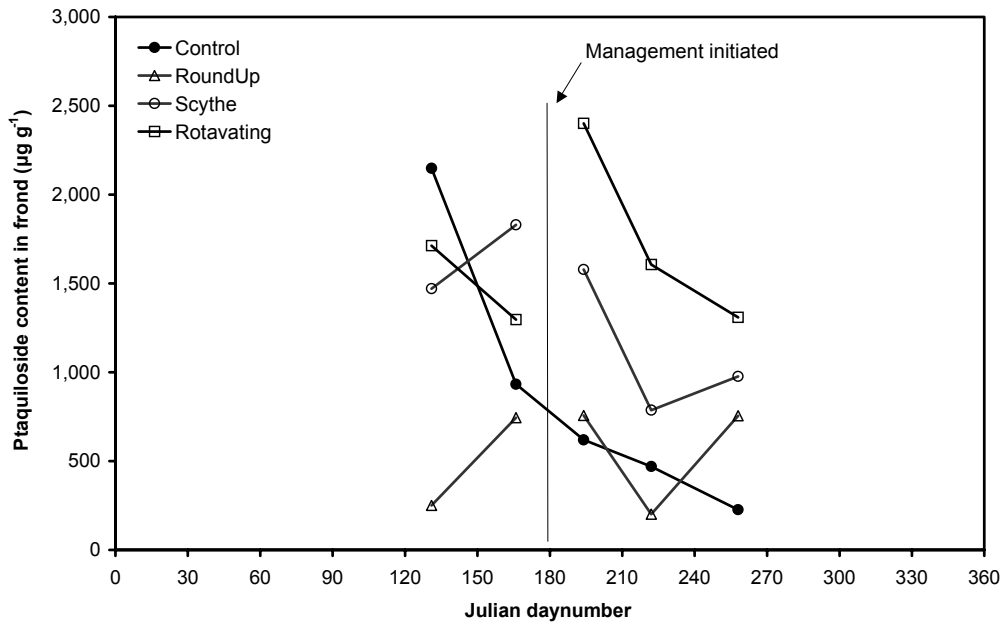
In Australia and in New Zealand, var. *esculentum* from southern latitudes had higher ptaquiloside contents than ferns originating from northern latitudes for Brackens having more or less the same growth stage (Smith et al. 1994; Rasmussen et al. 2003d). The effect of latitude may be due to differences in climate on a North-South gradient, but latitude may also be a proxy variable for some other hidden variable, e.g. parent material and hence soil properties such as content of available nutrients. A similar effect of longitude was also

observed in Scotland (var. *aquilinum*) (Rasmussen et al. 2003). In this study, correlation with aspect was also observed, showing higher ptaquiloside contents in stands with a southwestern aspect.

Looking more into the direct acting edaphic conditions, then the following factors have been found to correlate positively with the ptaquiloside content in var. *aquilinum* fronds from Denmark: Shade, lack of water holding capacity of the soil and low amounts of available nutrients in the soil (Smith et al. 1994; Villalobos-Salazar et al. 2000; Rasmussen et al. 2003b). All three variables indicate that edaphic stress may directly trigger ptaquiloside production. However, Alonso-Amelot et al. (2001) found a 100-fold increase in the ptaquiloside content of var. *caudatum* fronds following a fire that destroyed the surface vegetation - a situation usually associated with increased availability of nutrients. The raised ptaquiloside-level could although also be caused by higher ptaquiloside contents in second growth fronds (see below). A more subtle indirect effect of the edaphic conditions cannot be ruled out, as climatic stress is known to increase toxin concentrations in many plants (Ralphs, 2002). The effect of shading has also been observed in var. *esculentum* growing under controlled circumstances (Smith et al. 1990). Smith et al (1990) investigated the effect of soil substrate, and found that Brackens grown artificially in soil originating from Bracken stands had lower ptaquiloside contents than Brackens grown in sterilized potting soil. The authors did not conclude on this, but the stress hypothesis might be used to explain the observations, as the sterilized soil might lack the soil micro flora necessary for Bracken vigour and establishment of mycorrhizal systems. Hence, Brackens grown in sterilized soil may not receive the necessary nutrients resulting in increased ptaquiloside contents in the fronds.

Another stress factor known to influence the amount of ptaquiloside found in Bracken stands is physiological stress caused by mowing or grazing. Higher contents have been found in fronds of vars. *esculentum* and *aquilinum* from stands subjected to mowing or grazing compared to undamaged stands (Figure 5.10) (Smith et al. 1994; Rasmussen et al. 2003). This indicates that higher ptaquiloside contents occur in second growth fronds compared to first growth and/or that higher contents are present in damaged fronds. The high contents in ferns collected from stands subjected to browsing or mowing may also be due to the lower growth stage of the actually collected Brackens representing regrowth of new fronds after frond damage (Rasmussen et al. 2003). The physiological response to damage inside a frond needs further investigation.

Genetic variation may also contribute significantly to the observed variations in the ptaquiloside content. Smith et al. (1990 and 1994) found that the primary reason for the differing content of ptaquiloside in fronds was *origin*, i.e. provenience. Bracken from different locations all over the world were grown under controlled environments, and the observed variation in the ptaquiloside content could be tracked to the origin of the Bracken material. However, the study of Smith et al. (1990) was problematic since marked morphological differences were observed between the different proveniences.



**FIGURE 5.10.** Ptaquiloside contents in managed Bracken fronds (*var. aquilinum*). Management trial in a Christmas tree plantation (Noble Fir, *Abies nordmanniana* (Steven) Spach) at Hvalsø, Denmark. The experimental set-up comprised a weed management trial having three treatments and a control: 1) Spraying with RoundUp®; 2) Manual management with scythe; and 3) Mechanical management by rotavating. Analytical details as in Rasmussen and Hansen (2003). Author, unpublished results.

The ptaquiloside content in fronds might also be affected by precipitation – especially heavy showers. Rasmussen et al. (2003a) found a decreased content in Danish *var. aquilinum* stands sampled just after heavy showers. The same pattern was observed in Venezuela with *var. caudatum* by Alonso-Amelot et al. (1995). One plausible reason for this is that ptaquiloside is leached from the fronds by the rain. Rasmussen et al. (2003a) found that ptaquiloside can leach from *var. aquilinum* fronds, although only in minute amounts. Crosier and rhizome leaching tests (*var. aquilinum*) have also previously demonstrated presence of many compounds in leachates (Evans et al. 1984). Leaching of ptaquiloside from fronds was tested by Alonso-Amelot et al. (1995) that submerged a *var. caudatum* frond in water. However, this research group observed no ptaquiloside in the water used for the test. This might have been due to too much water used or that no

leaching actually occurs from these fronds. Physiological and morphological differences between the two Bracken varieties exist, but the effect of these differences on ptaquiloside leaching needs further investigation.

## **5.7 Summary.**

Ptaquiloside can be found in all varieties of Bracken, and in all fern compartments. The frond content is usually highest in young fronds and decreases during the growth season. The content in rhizomes is almost inversely correlated with the frond content. The recorded distribution is likely caused by differences in biological activity of the different compartments during the growth season and/or transfer of ptaquiloside between the compartments. The content of ptaquiloside in different Bracken stands is quite variable. On a large scale, the content of ptaquiloside in Bracken fronds seems mainly to be caused by genetic variation, while influence of local growth conditions can be seen on a smaller scale. However, recorded effects of local growth conditions are conflicting between different studies. There are some indications that ecological stress, such as physical disturbance, shade, and climate, may induce increased ptaquiloside contents in fronds, while nitrogen-depletion of soils may raise the level of ptaquiloside in rhizomes. These indications need further investigation before a general model for the ptaquiloside content in Bracken fronds can be established. Apparently, there is no variation in the frond content during the day, but the content varies significantly between individual fronds inside a single Bracken stand indicating ptaquiloside polymorphism at local scale. Non-ptaquiloside Brackens are found in two varieties of Bracken, indicating the existence of ptaquiloside chemotypes.

## Chapter 6

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### PTAQUILOSIDE IN SOIL

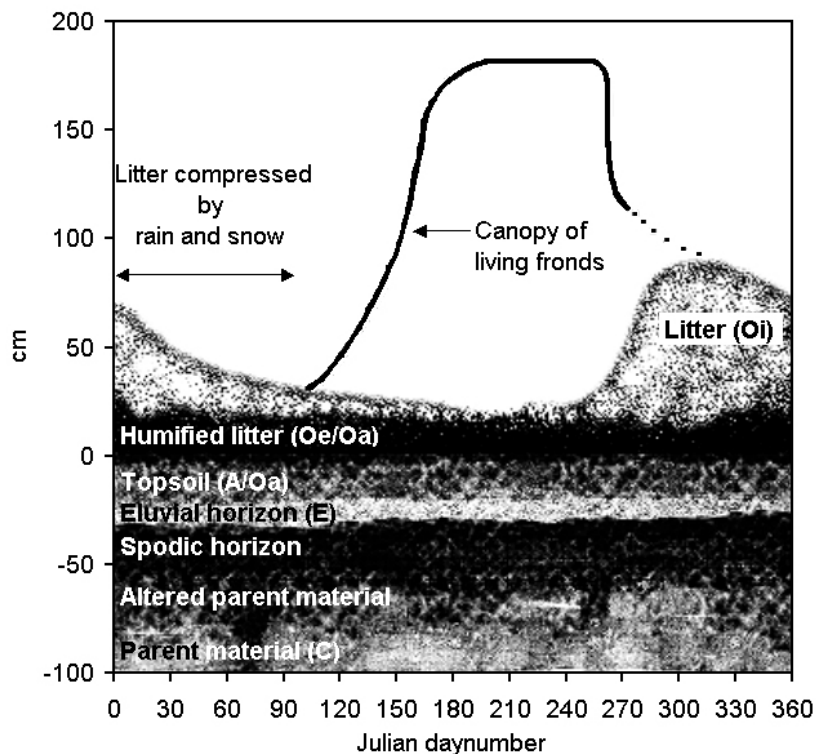
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#### 6.1 Ptaquiloside in soil materials.

The most obvious way for ptaquiloside to enter the soil environment is by leaching from fern material, either as dead or living biomass. Williams et al. (1987) found significantly increased pH (from 4.2 to 5.5) and salt contents in frond leachates compared to plain rain water (through fall and stem flow of var. *aquilinum*). The highest pH values and salt contents were found in the leachates from senescent Bracken stands (Williams et al. 1987), hereby indicating the highest potential of leaching from old Bracken fronds.

Ptaquiloside may be transferred from the fern to the soil environment in the following ways (Figure 6.1):

- Leaching of living fronds by rain (Rasmussen et al. 2003a).
- Physical transfer of ptaquiloside inside fronds to the litter layer when fronds die at the end of the growth season (or when fronds have been managed).
- Leaching of dead fronds in the litter layer by rain or melting snow.
- Leaching of living and dead roots and rhizomes by the soil solution.
- Transfer of ptaquiloside inside dead roots and rhizomes to soil.
- Excretion of ptaquiloside from all fern compartments (fronds, rhizomes and roots).
- Transfer to soil of ptaquiloside dissolved in urine or in droppings of animals eating Bracken.



**FIGURE 6.1.** Annual variation in the physical appearance of a Bracken stand (*var. aquilinum*) growing on a Spodosol. Bracken rhizomes are usually situated in the topsoil (A/Oa-horizons) and in the lower part of the humified litter layer (Oe/Oa-horizons). The litter layer can be very thick, but with very low density. Modified after Nicholson and Paterson (1976).

The content of ptaquiloside ranges between 0.09 and 8.49  $\mu\text{g g}^{-1}$  in litter below stands of *var. aquilinum* in Denmark and Scotland (Oi and/or Oe horizons), while Oa or A-horizons (topsoil) had ptaquiloside contents between 0.01 and 0.09  $\mu\text{g g}^{-1}$  (Denmark only) (Rasmussen et al. 2003; Rasmussen et al. 2003a; Rasmussen et al. 2003b). The variation in the ptaquiloside content of Oi-horizons inside individual stands seems rather high (Rasmussen et al. 2003a). At first sight, the soil contents are quite low compared to frond and rhizome contents. But due to the high density of soil compared to fern materials, contents between 0.3 and 160  $\text{mg m}^{-2}$  are found in Oi and/or Oe-horizons while contents between 0.9 and 57  $\text{mg m}^{-2}$  are found inside Oa or A-horizons. The corresponding frond contents (the ptaquiloside load) were 15-500  $\text{mg m}^{-2}$  (see Section 5.1) (Rasmussen et al. 2003b)!

The fact that ptaquiloside can be leached from fronds and that ptaquiloside occurs in soil strongly indicates that ptaquiloside can be transferred to the soil by rainwater. However, ptaquiloside in Oi- and Oe-horizons may also be relict ptaquiloside found inside remnants of Bracken (Frankland, 1976; Kawai et al. 1981; Saito et al. 1989). The content of



ptaquiloside in Bracken litter from var. *aquilinum* stands in Scotland is correlated with the content of ptaquiloside in the fronds (Rasmussen et al. 2003). Dead Bracken fronds are rather persistent although large amounts of easily soluble compounds are leached within a few months after frond death. Initial half-life's of 3-6 years has been reported for the turnover of var. *aquilinum* litter in Britain (Frankland, 1976). No correlation has been found between the content of ptaquiloside in fronds and soil materials, but a strong correlation has been found between ptaquiloside in Oa or A-horizons (topsoil's) and precipitation showing low ptaquiloside contents in areas with high precipitation in contrast to high ptaquiloside contents in soils from areas with low amounts of precipitation (Rasmussen et al. 2003b). This finding is supported by Rasmussen et al. (2003a) finding higher concentrations of ptaquiloside in soil materials below stands just exposed to heavy showers, reflecting the high solubility and mobility of ptaquiloside in the soil and that ptaquiloside may be added to the soil as discrete additions following the precipitation pattern. Continuous precipitation might instead leach ptaquiloside out of the upper soil layers. No straightforward explanation has been found for the ptaquiloside content in Oi- and Oe-horizons (Rasmussen et al. 2003b).

An example of the annual variation in the ptaquiloside concentration in soil solutions below a Bracken stand is shown in Box 6.1. The concentration is generally low, but peaks in November. The peak is situated in time approx. 1 month after the dieback of the Bracken stand, but also 1 month after the maximum concentrations in the rhizome system. This may indicate leaching of ptaquiloside from dying and degrading fronds or from the rhizome system. This is in line with the findings of Williams et al. (1987) for pH and salts.

**Box 6.1.** Ptaquiloside concentrations in soil solutions.

Ptaquiloside was measured after cleanup and conversion to pterasin B in soil solutions sampled by ceramic suction cells at 90 cm depth below four sub sites inside a Bracken stand (var. *aquilinum*) in a Christmas tree plantation (Noble Fir, *Abies nordmanniana* (Steven) Spach) at Hvalsø, Denmark (see Figure 5.10). Management took place in June. The experimental set-up was a weed management trial having three treatments and a control: 1) Spraying with RoundUp®; 2) Rotavating; and 3) Scythe. Analytical method as in Rasmussen et al. (2003b). No replicate measurements.

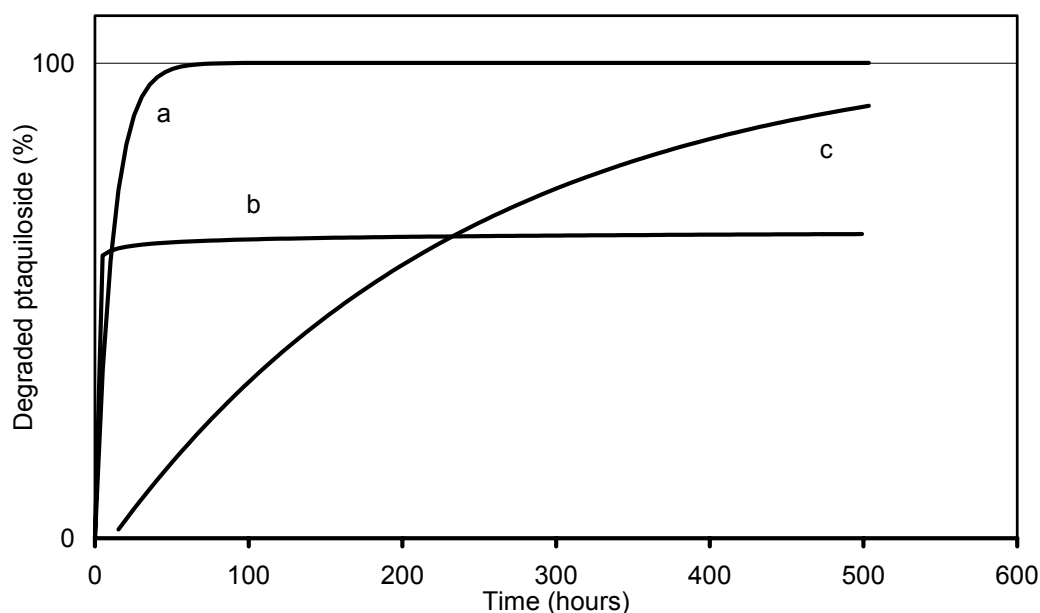
Subsite:	June (ng L <sup>-1</sup> )	July (ng L <sup>-1</sup> )	August (ng L <sup>-1</sup> )	September (ng L <sup>-1</sup> )	October (ng L <sup>-1</sup> )	November (ng L <sup>-1</sup> )	December (ng L <sup>-1</sup> )
Control:	ND	ND	ND	ND	0.0	5.6	0.3
RoundUp®:	6.5	4.8	3.8	3.2	ND	ND	ND
Rotavating:	3.2	0.0	1.9	3.6	ND	1.6	ND
Scythe:	1.3	4.7	1.5	1.3	0.9	7.2	2.4

ND: Not determined.

Ptaquiloside has been found two times in drinking water from small wells using shallow groundwater and situated below Bracken stands (var. *aquilinum*) in Denmark ( $4\text{-}6 \mu\text{g L}^{-1}$ ) and southern part of Sweden ( $45 \mu\text{g L}^{-1}$ ) (author, unpublished results). Both soil water and well water concentrations are very high compared to the tolerable concentration of ptaquiloside in drinkingwater of  $0.002 \mu\text{g L}^{-1}$  (Box 3.1).

## 6.2 Degradation of ptaquiloside in soil.

Only one study has been carried out to investigate the stability of ptaquiloside in contact with soil materials (Rasmussen et al. 2003c). This study shows that ptaquiloside may be stable in soils for relative long time (Figure 6.2).



**FIGURE 6.2.** Stability of ptaquiloside in contact with soil at 25°C. Three degradation patterns exist: a) Fast degradation (within 240 hours, most degraded in 140 hours); b) Fast initial degradation (40-60% within 24 hours) followed by slow degradation; and c) Slow degradation.  $[\text{ptaquiloside}]_{\text{START}} = 22\text{-}30 \mu\text{g mL}^{-1}$ , soil:solution ratio = 1:1 (weight:volume). The figure is modified after Rasmussen et al. (2003c).

Based on Rasmussen et al. (2003c), three degradation patterns can be delineated for Danish soils at 25°C (Figure 6.2):

- a. Fast and full degradation described as a first order degradation reaction



pattern have low clay contents (0-8%), low to high carbon-contents (1-40%), and low pH (less than 4).

- b. Fast initial degradation followed by slow degradation. The overall reaction can be fitted as the sum of a first order degradation reaction and a chemical equilibrium between ptaquiloside in solution and in the solid phase. The soils causing this pattern are weak acid to neutral (pH 4.8-7.9), with low carbon-contents (0.06-1.7%), and have high clay contents (14-23%).
- c. Slow degradation. The reaction can be described as for group-a. The soils resulting in this pattern are weakly acidic (pH 4.0-4.6), with low carbon-contents (0.2-0.8%), and with low clay contents (1-2%).

In this context, *degradation* covers any process resulting in reduction in the water-extractable amount of ptaquiloside. The pattern for group-a is likely due to acid hydrolysis combined with irreversible reactions between ptaquiloside and humic substances/clay. Microbial degradation may also take part in the degradation. The group-b pattern might be due to fast irreversible sorption to clay minerals followed by a period where ptaquiloside is somewhat protected from degradation, e.g. due to weak interlayer sorption in clay minerals. Such sorption has been observed previously with other glycosides and sugars (Greenland, 1956; Greenland, 1956a). The degradation pattern in group-c is likely due to hydrolysis, just at a lower pace than in group-a (due to higher pH than in group-a) (Rasmussen and Hansen, 2003).

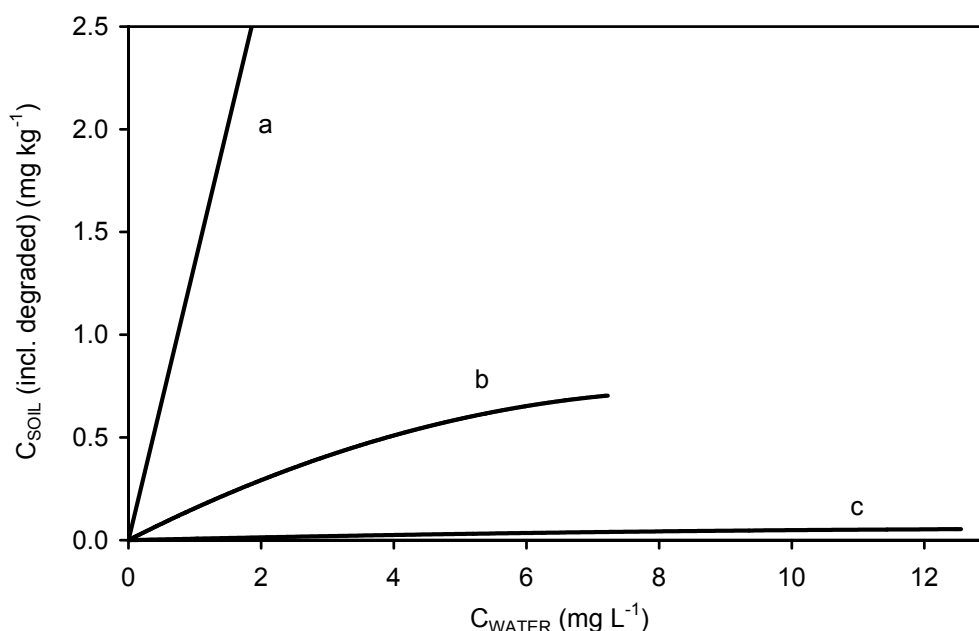
Rasmussen et al. (2003c) repeated their studies of ptaquiloside degradation at 4°C over a period of 1,728 hours. The overall pattern found was the same as at 25°C, but with two notable exceptions:

- 1) Fast group-a like degradation did not take place.
- 2) Slow group-c like degradation ceased after the first days.

In two of the four soils investigated degradation ceased within 144 hours, while in the two other soils degradation continued at a lower pace. A sudden unexplained increase in degradation was although observed by Rasmussen et al. (2003c) in one soil after 350 hours. Interestingly, ptaquiloside was still found in all the soil suspensions when the investigation ceased after 1,728 hours. The effect of the soil microbiology, chemistry and mineralogy needs further investigation before a thorough understanding of ptaquiloside degradation and reactions between ptaquiloside and soil can be achieved.

### 6.3 Soil sorption studies.

Rasmussen et al. (2003c) did also investigate the sorption of ptaquiloside to soil materials. Due to degradation of ptaquiloside caused by hydrolysis and irreversible sorption it was not possible for the authors to determine the sorbed amount of ptaquiloside *sensu strictu*. A further problem faced by the authors was the high solubility of ptaquiloside reflected in the low  $\log K_{ow}$ -value for ptaquiloside (Table 4.1) resulting in a very low – almost non-detectable – amount of sorbed ptaquiloside. Hence, the authors calculated distribution coefficients based on sorption including degradation and irreversible sorption for 24 hours (Equation 6.1, Figure 6.3).



**FIGURE 6.3.** Freundlich sorption isotherms for ptaquiloside in contact with Danish soils. A range of isotherms were observed: a) Organic soil (Oa-horizon),  $K_{\text{FREUNDLICH}} = 1.23 \text{ L kg}^{-1}$ ,  $n = 1.00$ ); b) Mineral soil (clayey A-horizon),  $K_{\text{FREUNDLICH}} = 0.07 \text{ L kg}^{-1}$ ,  $n = 0.75$ ); and c) Mineral soil (sandy C-horizon),  $K_{\text{FREUNDLICH}} = 0.01 \text{ L kg}^{-1}$ ,  $n = 0.73$ ). Partly after Rasmussen et al. (2003b).  $[\text{ptaquiloside}]_{\text{START}} = 0\text{-}14 \text{ mg L}^{-1}$ .  $C_w$ : Concentration in water.  $C_s$ : Content on soil.

The sorption of ptaquiloside could be fitted with Freundlich isotherms (Eq. 6.1):

$$C_s = K_{\text{FREUNDLICH}} \cdot C_w^n \quad \text{Eq. 6.1}$$

where  $C_s$  = equilibrium concentration of ptaquiloside on soil ( $\text{mg kg}^{-1}$ ),  $K_{\text{FREUNDLICH}}$  = Freundlich-type distribution coefficient (the Freundlich constant),  $C_w$  = equilibrium concentration of ptaquiloside in water ( $\text{mg L}^{-1}$ ), and  $n$  = measure of the non-linearity

involved in adsorption. The Freundlich constants ranged from 0.01 to 0.22 with a single extreme constant of 1.23 found in very acid soil materials (probably caused by acid hydrolysis). The Freundlich constant was positively correlated with the carbon-content and the clay content of the soils indicating that sorption and degradation of ptaquiloside is correlated with the clay- and carbon-contents of the soils and that only little sorption of ptaquiloside to soil organic matter takes place ( $P$  (level of significance) = 0.03; Eq. 6.2) (Rasmussen et al. 2003c). However, the correlation coefficient between the Freundlich constant and the carbon/clay content of the soils investigated was not very good indicating that ptaquiloside bonding in soil depends on other soil parameters than the clay and carbon-content of the soil ( $R^2 = 0.75$ ).

$$K_{\text{FREUNDLICH}} = 0.000 \pm 0.036 + 0.082 \pm 0.026 \cdot C(\%) + 0.006 \pm 0.002 \cdot \text{clay}(\%) \quad \text{Eq. 6.2}$$

The non-linearity parameter  $n$  ranged between 0.64 and 1.00, and was correlated with the carbon content and pH ( $P = 0.03$ ;  $R^2 = 0.83$ ; Eq. 6.3).

$$n = 1.074 \pm 0.108 - 0.057 \pm 0.018 \cdot \text{pH} + 0.162 \pm 0.046 \cdot C(\%) \quad \text{Eq. 6.3}$$

Equations 6.2 and 6.3 could only predict the sorption parameters on 8 respectively 7 out of 10 possible sets of data, hereby-indicating influence of other parameters than carbon, pH and clay on the sorption of ptaquiloside to soil materials. However, the models do indicate that sorption of ptaquiloside could be considered as a function of: 1) the acidity, governing the hydrolysis of ptaquiloside; 2) irreversible sorption of ptaquiloside to soil organic matter and clay minerals; and 3) a weak reversible sorption to soil organic matter and clay minerals (Rasmussen et al. 2003c).

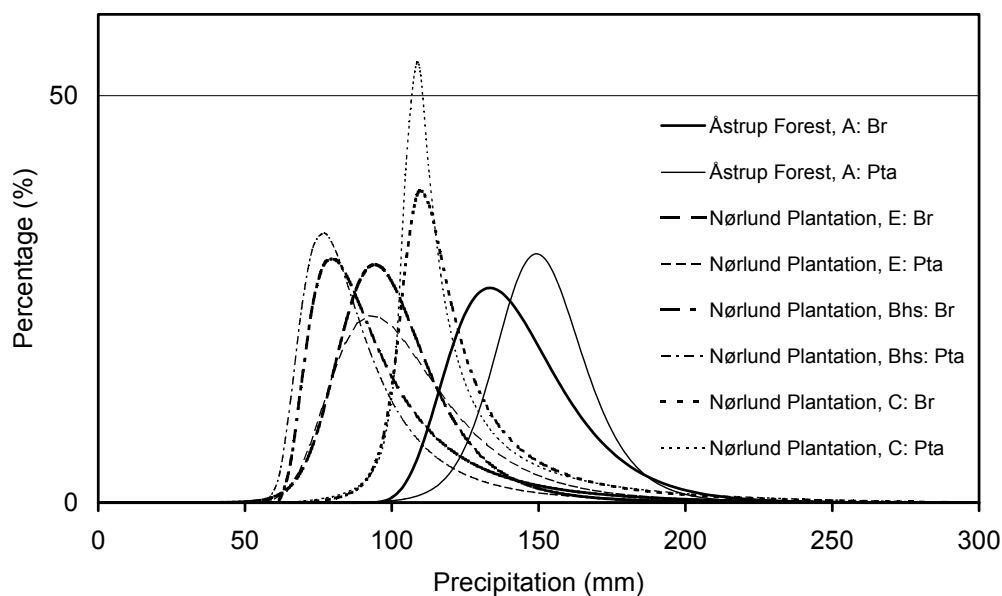
The equilibrium concentrations for ptaquiloside in soil solutions are estimated for 6 common Danish topsoil's from the Bracken stands investigated by Rasmussen et al. (2003b) in Table 6.1. The ptaquiloside concentrations are based on the  $K_{\text{FREUNDLICH}}$  equations (Eq. 6.2-3). These estimated concentrations are in some range with the measured ptaquiloside concentrations in the soil water from Hvalsø (Box 6.1). The concentrations are also well above the maximum tolerable concentration for ptaquiloside in drinking water of 0.1-1.6 ng L<sup>-1</sup> (Box 3.1).

**TABLE 6.1.** Estimated equilibrium concentrations in soil solutions based on the full porosity of the soil horizon (Oa/A-horizons, 0-20 cm), and 100 viz. 30% saturation. Data from Rasmussen et al. (2003b). Details in Appendix B.

Location:	Ptaquiloside content:	Soil carbon	Clay	pH	Equilibrium concentration	Percentage of total in soil water
(saturation in %)	( $\mu\text{g kg}^{-1}$ )	(%)	(%)		( $\mu\text{g L}^{-1}$ )	(%)
<u>Organic soils:</u>						
Hundshage (100)	30	32.5	0.0	2.8	20	30
Præstø I (100)	20	25.7	0.0	3.0	20	30
Præstø II (100)	230	31.3	0.0	3.0	170	30
Ulfshale (100)	710	43.6	0.0	2.8	410	20
<u>Mineral soils:</u>						
Løvenholm (100)	20	2.5	1.8	4.1	30	60
Østbirk (100)	10	2.5	4.3	3.7	20	60
<u>Low saturation:</u>						
Præstø II (30)	230	31.3	0.0	3.0	200	10
Løvenholm (30)	20	2.5	1.8	4.1	60	40

#### 6.4 Soil column studies.

Rasmussen et al. (2003c) investigated the mobility of ptaquiloside in soil materials by leaching experiments in packed soil columns (Spodosol materials).



**FIGURE 6.4.** Breakthrough curves for ptaquiloside and bromide in soil columns comprising soil materials from Danish Spodosols. 80% ptaquiloside recovery from Åstrup Forest, A; 100% ptaquiloside recovery from Nørlund Plantation, E, Bhs, and C (Rasmussen et al. 2003c).

The authors found very weak sorption of ptaquiloside to the soil material and very high mobility (Figure 6.4). Distribution coefficients and retardation factors were calculated from the retardation of ptaquiloside compared to bromide and ranged from 0.00 to 0.05 L kg<sup>-1</sup>, which is in the same range as the Freundlich distribution coefficients (Rasmussen et al. 2003c).

## 6.5 Summary.

Ptaquiloside can be leached from Bracken fronds to the soil by rain, but may also be transferred to the soil from dead Bracken material (roots, rhizomes, and fronds). Whatever reason, ptaquiloside is found in topsoil's and litter in amounts up to 8.5 µg g<sup>-1</sup>. Soil solution concentrations were not measured directly in topsoil's, but equilibrium concentrations up to 400 µg L<sup>-1</sup> in the soil solution can be estimated from the sorptive properties of ptaquiloside. This range is in accordance with measured ptaquiloside concentrations of up to 7 µg L<sup>-1</sup> in soil solutions extracted 90 cm below the soil surface. These concentrations are much higher than the tolerable ptaquiloside concentration in drinking water with respect to the carcinogenicity of ptaquiloside.

Ptaquiloside is highly mobile in soils due to weak sorption and may therefore reach groundwater or other recipients. The stability of ptaquiloside in soil depends on the geochemical properties of the soils:

- 1) Ptaquiloside degrades rapidly when pH is below 4 or higher than 8.
- 2) Ptaquiloside may sorb irreversibly to clay minerals and soil organic matter.
- 3) Ptaquiloside may be sorbed reversibly to clay minerals and soil organic matter
- 4) Ptaquiloside degradation is temperature dependent (faster at higher temperatures).

Besides from the geochemical properties of the soil, soil microbiology may also influence the stability of ptaquiloside in soil.





## Chapter 7

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### PTAQUILOSIDE – AN ENVIRONMENTAL HAZARD?

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#### 7.1 Conclusion.

In Chapter 1, six fundamental questions were raised that this thesis should seek answers to:

*1. Can ptaquiloside be found in soil materials?*

Ptaquiloside can indeed be found in soil materials! Contents ranging between 0.09 and 8.49  $\mu\text{g g}^{-1}$  were found in litter while topsoil material just below the litter had contents between 0.01 and 0.09  $\mu\text{g g}^{-1}$ . The contents seem low, but correspond to 0.3-160  $\text{mg m}^{-2}$  in the litter and 0.9-57  $\text{mg m}^{-2}$  for the topsoil's.

*2. Is ptaquiloside stable in contact with soil materials?*

The stability of ptaquiloside in contact with soil materials depends on the physical composition and chemical properties of the soil. Ptaquiloside degrades rapidly due to hydrolysis when pH is less than 4, especially in acid sandy soils where half-lives in the range of 8 - 30 hours were observed at 25°C. Degradation is probably quite rapid too in soils having pH higher than 8. Ptaquiloside degrades more slowly in slightly acid to neutral soils due to decreased rate of hydrolysis. In clayey soils, the degradation rate is further reduced probably due to a concurrent solid-phase partitioning reaction. The results indicate, that with longer reaction times, clay silicates may sorb ptaquiloside into the interlayer space reducing the amount of ptaquiloside available for degradation in solution.

Ptaquiloside were found to have very low affinity for sorption to soil organic matter. However, irreversible bonding might take place between the alkylating group of ptaquiloside and soil organic matter in line with what has been observed with DNA. Irreversible bonding may also take place between ptaquiloside and soil minerals. However, this topic needs further investigation before final conclusions can be made.

Ptaquiloside degradation in soil is strongly temperature dependent, and at 4°C overall degradation is much reduced compared with 25°C. This raises the question if laboratory studies performed at 20-25°C are capable of mimicking near natural conditions with respect to the fate of ptaquiloside in the terrestrial ecosystem in much of the temperate zone. An unresolved question is also the effect of microbial activity causing degradation of ptaquiloside.

### **3. What is the mobility of ptaquiloside in soil?**

Soil sorption studies revealed very low distribution coefficients between ptaquiloside found in the soil solution and ptaquiloside bound to soil particles, hence demonstrating a high mobility of ptaquiloside in the soil system. This is especially due to the very hydrophilic character of the glucose moiety of ptaquiloside. Even though sorption to soil organic matter is low, ptaquiloside sorption to soil materials seems correlated with the amounts of carbon and clay in the soils as well as to soil pH. However, the correlation is not very good indicating influence of other parameters than those mentioned.

Studies of ptaquiloside mobility in packed soil columns confirmed the results of the soil sorption studies and revealed mobility of ptaquiloside comparable to an inert tracer, e.g. nitrate. The mobility of ptaquiloside was to some degree confirmed by finding of ptaquiloside in soil solutions extracted from 90 cm below the soil surface.

### **4. When is ptaquiloside present in Bracken?**

Ptaquiloside is found in all varieties of Bracken. However, some proveniences of *vars. esculentum* and *revolutum* does not contain ptaquiloside (see below). Ptaquiloside seems to be present in all fern compartments (fronds, rhizomes, roots and spores) at all times, but in very variable amount. The frond content (maximum content approx. 13,000  $\mu\text{g g}^{-1}$ ) is usually highest when the fronds are at the crosier stage and then gradually declines during the growth season. The content in rhizomes (maximum content approx. 7,000  $\mu\text{g g}^{-1}$ ) seems to be highest by the end of the growing season. Only few studies have investigated the content in spores and roots. The content in these parts of the fern seems to be rather low, showing maximum contents of approx. 200 and 20  $\mu\text{g g}^{-1}$  respectively. The highest ptaquiloside contents are apparently found in fern compartments with high biological/metabolic activity. Further studies are needed to fully understand the dynamics and biological role of ptaquiloside in the Bracken fern.

### ***5. What is the natural variation in the ptaquiloside content of Bracken?***

Apparently no significant differences exist between Bracken varieties regarding their content of ptaquiloside. However, genetic heritage is likely to have some impact on the content of ptaquiloside as Bracken proveniences from the same variety have different ptaquiloside contents when grown under the same conditions. Ptaquiloside chemotypes may also exist as some Bracken varieties include proveniences not containing ptaquiloside.

The variation of the ptaquiloside content in fronds is quite large. As described above, annual variation takes place. Besides from that, large variation in the ptaquiloside content is found within Bracken stands; frond contents varying between 400 and 5,000  $\mu\text{g g}^{-1}$  were found in an area of 24m x 36m. However, significantly different frond contents exist between different Bracken stands. The level of ptaquiloside in fronds of different Bracken stands is likely affected by external growth factors such as climate, ecosystem competition and soil properties. The exact cause of the variation is still to be established.

The level of ptaquiloside in fronds is affected by physiological stress as raised levels of ptaquiloside are found in ferns subjected to mowing or browsing by animals. Physiological or edaphical stress may also be the explanation for the effects of the external growth factors on the ptaquiloside content in fronds. However, this stress hypothesis needs further investigation to be proved.

### ***6. How much ptaquiloside can be produced under natural conditions?***

To make a full risk assessment of Bracken stands it is essential to know the amount of ptaquiloside that can be produced in a Bracken stand. Due to the large annual variation of the ptaquiloside content in the different parts of Bracken, this is a very complicated task. Until now, a detailed model has only been achieved for the content of ptaquiloside in fronds. This model indicated that Bracken fronds could produce at least 500 mg ptaquiloside per  $\text{m}^2$ . This model is only a first approximation of the ptaquiloside production potential of Bracken stands as it only takes aboveground production into account and does not consider ptaquiloside flux between plant compartments during the growth season or leaching of ptaquiloside from the fern.

## **7.2 General discussion.**

It is still an open question whether ptaquiloside contamination of milk and/or drinking water or the presence of Bracken spores in breathing air/surface waters forms the link between human cancer and the Bracken toxins (Evans, 1979; Galpin and Smith, 1986;

Galpin et al. 1990; Alonso-Amelot and Avendaño, 2002). In areas where milk consumption is dependent on cows browsing on Bracken, then milk is likely the main vector (Alonso-Amelot and Avendaño, 2001). In areas where only few cows browse on the fern, then water contamination might be important as well, especially in rural areas depending on local water sources like surface wells or shallow aquifers. The findings presented in this review may therefore be a part of the explanation of the high cancer rates observed in these areas. However, actual findings of ptaquiloside in drinking water from these areas are still to be made.

The large variations of ptaquiloside stability in contact with soil materials are due to differences in soil composition. Such variations coupled with the natural variation in the ptaquiloside content in Bracken may explain the lack of correlation between Bracken cover, cancer and water supplies as reported by Galpin and Smith (1986) in the Gwynedd-study (Rasmussen et al. 2003; Rasmussen et al. 2003b; Rasmussen et al. 2003c).

The amount of ptaquiloside found in Bracken stands capable of being transferred to the soil, seems quite high when compared to the amounts of pesticides and other potentially harmful compounds added to soils by humans. The question must therefore be raised if Bracken management in forestry and agriculture should be targeted from an environmental point of view besides from a traditional crop science point of view.

*When considering the possible harmful properties of Bracken, distinguishing between the concepts of hazard and risk is useful. A hazard is a set of circumstances that may have harmful consequences; risk is the probability of harmful consequences occurring from a hazard* (Wilson et al. 1998). Therefore, even though Bracken represents an environmental hazard, it must be remembered that the fern most likely only affects people living and working in areas with large densities of Bracken (Smith, 2000). For people only being in sporadic contact with Bracken and/or ptaquiloside the risk of being affected is low, e.g. people having summer residence homes in Bracken infested areas or scouts tramping through Bracken stands. Further studies are needed to unveil the full environmental impact of Bracken.

### **7.3 Perspectives and further studies.**

Seen from a world-perspective, Bracken has generally changed its role from being a useful or used part of Nature to being perceived as a pest in agriculture and forestry. Most research on Bracken is therefore now conducted on the problematic effects of the fern, in

particular how to eradicate or control it. Much progress in the understanding of Bracken ecology and biochemistry has taken place in this light during the last 10-15 years. However, much is still to be done, such as:

- Development of integrated control systems for organic farmers in industrialized countries.
- Development of low-cost control systems/pesticides for use in the developing world.

A thorough understanding of the connection between the fern and its toxins and their implications on human health and the environment is also strongly needed. This thesis has sought answers to some central questions regarding the presence of ptaquiloside in terrestrial ecosystems. However, several new questions rose such as:

- Can ptaquiloside be leached from roots and rhizomes by the soil water?
- What is the effect of ptaquiloside on the soil microflora?
- Can soil microorganisms mineralise ptaquiloside and how fast?
- Which interactions exist between ptaquiloside and soil minerals?
- To which degree is ptaquiloside transferred from a Bracken stand to the soil?
- What is the total production of ptaquiloside in a Bracken stand (comprising all parts of the fern)?

Until now, most research has been carried out in highly specialized research groups, although some general Bracken conferences have taken place. To secure future advances in this field of science, international cooperation between scientists from different research areas must take place.



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## **APPENDICES**

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- A Calculation of tolerable concentration of ptaquiloside in drinking water.**
- B Calculation of equilibrium concentrations in soil solutions.**





## Appendix A

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### CALCULATION OF TOLERABLE CONCENTRATION OF PTAQUILOSIDE IN DRINKING WATER

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The Danish Environmental Protection Agency recommends the *One-hit* model for estimation of a *tolerable daily intake* of genotoxic compounds through drinking water (Miljøstyrelsen, 1992):

$$\text{One hit model: } C = \frac{W_h \times I_t \times \left(\frac{L_e}{L}\right)^3 \times \frac{D \times I_e}{L_e}}{-\ln\left(\frac{1-P_t}{1-P_c}\right) \times \sqrt[3]{\frac{W_h}{W_a}} \times V_h}$$

where C (tolerable concentration of a compound in drinking water (mg L<sup>-1</sup>)); W<sub>h</sub> (human body mass in kg (70 kg)); I<sub>t</sub> (tolerable life-time risk (10<sup>-6</sup>)); L<sub>e</sub> (actual age of laboratory test animals (number of days)); L (theoretical average life-time of laboratory test animals (days)); D (daily exposure of the carcinogen to the laboratory test animals (mg (kg<sub>BODYWEIGHT</sub> day)<sup>-1</sup>)); I<sub>e</sub> (exposure time (days)); P<sub>t</sub> (tumour incidents among exposed laboratory test animals (0-1)); P<sub>c</sub> (spontaneous tumour incidents among control laboratory test animals (0-1)); W<sub>a</sub> (average weight of laboratory test animals (kg)); and V<sub>h</sub> (daily intake of drinking water (adults: 2 L)).

The estimates of a tolerable concentration of ptaquiloside in Box 3.1 are based on the toxicological studies on rats by Hirono et al. (1984), Hirono et al. (1987) and Smith et al. (1988). These studies are made for estimation of Bracken/ptaquiloside carcinogenicity on laboratory animals, and it must be emphasized that the experiments are not optimised for estimation of a tolerable concentration of ptaquiloside in drinking water. For example, the dose of ptaquiloside is likely somewhat too large resulting in too high mortality of the test animals. The experiments are also somewhat short compared to what usually is performed

in toxicology testing. The estimates were checked by Karl-Heinz Cohr (M.Sc. Biochem) from the Danish Toxicology Centre (Kogle Alle 2, DK-2970 Hørsholm, Denmark).

Example: Smith et al. (1988).

Data obtained from Rat exposure to Taumarunui ferns and information's obtained from the authors.

Calculation of daily dose:	Period 1:	Period 2:	Average:
Average intake of food ( $\text{g day}^{-1}$ ):	25	30	
Ptaquiloside in food ( $\text{mg g}^{-1}$ ):	0.135	0.355	
Weight of laboratory animal (kg):	0.26	0.3	
Daily dose ( $\text{mg (kg}_{\text{BODYWEIGHT}} \cdot \text{day)}^{-1}$ ):	13.0	35.5	<u>27.2</u>

#### Variables

Wh	Human body mass (kg):	70
It	Tolerable life-time risk:	0.000001
Le	Actual age of laboratory test animals (days):	400
L	Theoretical average life-time of laboratory test animals (days):	1095
D	Daily exposure of the carcinogen ( $\text{mg (kg}_{\text{BODYWEIGHT}} \cdot \text{day)}^{-1}$ ):	27.2
le	Exposure time (day):	162
Pt	Tumour incidents among exposed laboratory test animals (0-1):	0.89
Pe	Tumour incidents among control laboratory test animals (0-1):	0.06
Wa	Average weight of laboratory test animals (kg):	0.3
Vh	Daily intake of drinking water (L):	2
C	Tolerable concentration of ptaquiloside in drinking water (ng/L):	<u>1.4</u>

## Appendix B

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### CALCULATION OF EQUILIBRIUM CONCENTRATIONS IN SOIL SOLUTIONS.

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The equilibrium concentrations in Table 6.1 are based on the measured concentrations reported in Rasmussen et al. (2003a) and:

1. Mineral soils: The correlation between  $K_{\text{FREUNDLICH}}$ , soil carbon and clay content (Eq. 6.2), and  $n = 1$ .
2. Organic soils: The  $K_{\text{FREUNDLICH}}$  determined for Præstø soils normalized for carbon content ( $C_{\text{SOIL}} = 0.032 \cdot f_{\text{OC}} \cdot C_{\text{WATER}}$ ) (Rasmussen et al. 2003).

#### Example 1: Præstø II, 100% saturation.

Equilibrium:  $C_{\text{SOIL}} = f_{\text{OC}} \cdot K_{\text{OC}} \cdot C_{\text{WATER}}^n$ ,  $f_{\text{OC}} = 31.3\%$ ,  $K_{\text{OC}} = 0.032$ , due to low concentration in soil water,  $n = 1$ .

$$C_{\text{SOIL}} = 31.3 \cdot 0.032 \cdot C_{\text{WATER}}^1 \Leftrightarrow C_{\text{WATER}} = \frac{1}{1.002} \cdot C_{\text{SOIL}} \Leftrightarrow C_{\text{WATER}} = \frac{1}{1.002} \cdot C_{\text{SOIL}}$$

Porosity:  $1 - \frac{\text{soil density}}{\text{density of soil particles}} = 1 - \frac{0.8 \text{ kg dm}^{-3}}{1.1 \text{ kg dm}^{-3}} = 27\%$ , corresponding to 800 g soil per  $\text{dm}^3$  and 270 mL soil water per  $\text{dm}^3$ .

Total amount of ptaquiloside per  $\text{dm}^3$ : Amount of soil per  $\text{dm}^3 \cdot C_{\text{TOTALSOIL}} = 800 \text{ g} \cdot 225 \text{ ng g}^{-1} = \underline{180,000 \text{ ng}}$ .

$$180,000 \text{ ng} = C_{\text{SOIL}} \cdot 800 \text{ g} + C_{\text{WATER}} \cdot 270 \text{ mL} \Leftrightarrow C_{\text{SOIL}} = 225 \text{ ng} - \frac{270 \text{ mL}}{800 \text{ g}} \cdot C_{\text{WATER}} \Leftrightarrow C_{\text{SOIL}} = 225 \text{ ng} - 0.3375 \cdot C_{\text{WATER}}$$

$$C_{\text{WATER}} = \frac{1}{1.002} \cdot C_{\text{SOIL}} = \frac{1}{1.002} \cdot (225 \text{ ng} - 0.3375 \cdot C_{\text{WATER}}) = \underline{168 \mu\text{g L}^{-1}}$$

$$\text{Percentage in soil water: } \frac{C_{\text{WATER}} \cdot \text{Volume of soil water}}{\text{Total amount of ptaquiloside}} = \frac{168 \text{ ng mL}^{-1} \cdot 270 \text{ mL}}{180,000 \text{ ng}} = \underline{25\%}$$

## Example 2: Løvenholm, 30% saturation,

$$\text{Equilibrium: } C_{\text{SOIL}} = K_{\text{FREUNDLICH}} \cdot C_{\text{WATER}}^n$$

$$K_{\text{FREUNDLICH}} = 0.0819 \cdot C(\%) + 0.0056 \cdot \text{clay}(\%) \Leftrightarrow K_{\text{FREUNDLICH}} = 0.0819 \cdot 2.5 + 0.0056 \cdot 1.8 = 0.2148$$

$$\text{Due to the low concentrations, } n = 1 \Rightarrow C_{\text{WATER}}^1 = \frac{1}{0.2148} \cdot C_{\text{SOIL}} \Leftrightarrow C_{\text{WATER}} = 4.6555 \cdot C_{\text{SOIL}}$$

$$\text{Porosity: } 1 - \frac{\text{soil density}}{\text{density of soil particles}} = 1 - \frac{1.3 \text{ kg dm}^{-3}}{2.65 \text{ kg dm}^{-3}} = 50\%$$

$\Rightarrow$  corresponding to 1,300 g soil per  $\text{dm}^3$  and  $30\% \cdot 500 \text{ mL} = 150 \text{ mL}$  soil water per  $\text{dm}^3$ .

$$\text{Total amount of ptaquiloside per } \text{dm}^3 : \text{Amount of soil per } \text{dm}^3 \cdot C_{\text{TOTAL SOIL}} = 1,300 \text{ g} \cdot 20 \text{ ng g}^{-1} = \underline{26,000 \text{ ng}}$$

$$26,000 \text{ ng} = C_{\text{SOIL}} \cdot 1,300 \text{ g} + C_{\text{WATER}} \cdot 150 \text{ mL} \Leftrightarrow C_{\text{SOIL}} = 20 \text{ ng} - \frac{150 \text{ mL}}{1,300 \text{ g}} \cdot C_{\text{WATER}} \Leftrightarrow C_{\text{SOIL}} = 20 \text{ ng} - 0.1154 \cdot C_{\text{WATER}}$$

$$C_{\text{WATER}} = 4.6555 \cdot C_{\text{SOIL}} = 4.6555 \cdot (20 \text{ ng} - 0.1154 \cdot C_{\text{WATER}}) = \underline{\underline{61 \mu\text{g L}^{-1}}}$$

$$\text{Percentage in soil water: } \frac{C_{\text{WATER}} \cdot \text{Volume of soil water}}{\text{Total amount of ptaquiloside}} = \frac{61 \text{ ng mL}^{-1} \cdot 150 \text{ mL}}{26,000 \text{ ng}} = \underline{\underline{35\%}}$$

## **ARTICLES AND MANUSCRIPTS**

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Due to restrictions from the publishers of the journals in which two of the papers in this thesis have been published, these papers are not present in this PDF. The papers can be found in:

Rasmussen, L. H., Jensen, L. S., & Hansen, H. C. B. (2003). Distribution of the Carcinogenic Terpene Ptaquiloside in Bracken Fronds, Rhizomes (*Pteridium aquilinum*), and Litter in Denmark. *Journal of Chemical Ecology*, 29, 771-778.  
DOI: 10.1023/A:1022885006742

Rasmussen, L. H., Kroghsbo, S., Frisvad, J. C., & Hansen, H. C. (2003). Occurrence of the carcinogenic Bracken constituent ptaquiloside in fronds, topsoils and organic soil layers in Denmark. *Chemosphere*, 51, 117-127.  
DOI: 10.1016/S0045-6535(02)00694-X



1 **Ptaquiloside in Bracken (*Pteridium***  
2 ***aquilinum* ssp. *aquilinum*) in Scotland:**  
3 **Contents in fronds, rhizomes, and litter.**

4

5

6

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1 **Abstract**—Compounds found in bracken (*Pteridium aquilinum* (L.) Kuhn) are suspected of  
2 causing cancer in humans and animals. Ingestion of bracken, or food and drinking water  
3 contaminated with bracken carcinogens may cause the cancer. The main carcinogenic  
4 principle in bracken is the water-soluble indanone-type compound ptaquiloside. The aim of  
5 the study was to investigate the annual and regional variation in the content of ptaquiloside  
6 in Scottish bracken and associated litter, hereby trying to identify environmental conditions  
7 influencing these contents. The ptaquiloside content in fronds, rhizomes and litter (Oi-  
8 horizons) was monitored during the growth season at 3 sites, and at 17 additional sites all  
9 over Scotland by the end of the growth season. Ptaquiloside contents in the fronds ranged  
10 between 50 and 5,800 µg/g, with highest concentrations recorded in May and June.  
11 Ptaquiloside contents in the rhizomes ranged between 10 and 660 µg/g, with the highest  
12 contents recorded by the end of the growth season. Ptaquiloside contents in bracken litter  
13 ranged between 0.1 and 5.8 µg/g, which tended to increase in the autumn. This is the first  
14 report of the annual variation in ptaquiloside concentrations in bracken litter. The latitude  
15 and altitude of the stand, the frond development stage and content correlated with the  
16 amount of ptaquiloside found in the litter, while the content in rhizomes were correlated  
17 with the longitude, topography, and nitrogen content in the soil. This is the first report  
18 correlating the content of ptaquiloside in rhizomes with external growth factors. The  
19 content in the fronds were partly correlated with the longitude and aspect of the stand, the  
20 frond-height and the degree of damage to the fronds. Ptaquiloside contents increased in  
21 fronds that emerged following harvest compared to uncut fronds of the same development  
22 stage. Bracken management should therefore be undertaken with great care if animals are to  
23 graze on paddocks having managed brackens.

24  
25 **Keywords**—Ptaquiloside, bracken, *Pteridium aquilinum*, Scotland, frond, rhizome, litter.

26

## INTRODUCTION

In Scotland, bracken (*Pteridium aquilinum* (L.) Kuhn) is found as an invading weed all over the country, mainly on acid soils in forests, in paddocks, on moors and on less-intensively managed hilly areas (Frankland, 1976; Page, 1997). Bracken is well defined at the generic level, but uncertainties are found regarding the nomenclature and taxonomy within the genus, especially in recent literature (Thomson, 2000; Thomson and Alonso-Amelot, 2002). To allow comparison with older literature, the taxonomy introduced by Tryon (1941) is adapted in this paper: Bracken belongs to the family Dennstaedtiaceae, and the species *P. aquilinum* is divided into two subspecies: ssp. *caudatum* and *aquilinum*, of which only ssp. *aquilinum* is found in Europe. Two varieties of ssp. *aquilinum* grow in Europe: Common Bracken (var. *aquilinum* - Western and Southern Europe) and Pinewood Bracken (var. *latiusculum* - Northern and Eastern Europe). Common and Pinewood Bracken are both found in Scotland as separate species and as hybrids. Common Bracken is although the most widespread variety in Scotland while Pinewood Bracken is mainly found at the westcoast (Page, 1989).

Ptaquiloside is a nor-sesquiterpene glycoside found in all bracken varieties (Figure 1). Ptaquiloside is a compound of environmental and veterinary concern due to its carcinogenic effects to animals and Man. Bracken-fern consumption by browsing animals, especially by cattle and sheep, may lead to severe diseases such as the *Acute Bracken Fern Poisoning* and *Bovine Enzootic Haematuria* (BEH) (Smith, 1997), while bracken consumption by Man may lead to gastric or oesophageal cancer (Haenszel et al., 1976; Marlière et al., 2000). Ptaquiloside has proved to be carcinogenic, mutagenic and genotoxic (van der Hoeven et al., 1983; Ojika et al., 1987; Ojika et al., 1989; Kushida et al., 1994; Shahin et al., 1998). Recently, Marlière et al. (2002) published a study demonstrating a 8.1-fold greater chance of developing oesophageal cancer among people including bracken in their daily intake. Human intake of ptaquiloside with milk contaminated with ptaquiloside may occur in some

1 areas, as up to 9% of ptaquiloside in bracken digested by cows is transferred to the milk.  
2 Water contamination may also occur as at least 0.2% of the total ptaquiloside present in  
3 pinnae can be leached by a single rain-event resulting in a frond-drip concentration of up to  
4 730µg/liter (Galpin and Smith, 1986; Alonso-Amelot et al., 1998; Rasmussen et al., 2003a;  
5 Alonso-Amelot et al., 2000b)

6  
7 Ptaquiloside is stable at neutral pH but degrades rapidly at alkaline conditions or when pH  
8 less than 4. In the pH-range of 4-6.5 ptaquiloside can exist for months - even in contact  
9 with soil. Ptaquiloside is highly water-soluble and is very poorly sorbed to soil organic  
10 matter. However, irreversible sorption to soil components may take place under specific  
11 conditions (Rasmussen et al., 2003c).

12  
13 In the temperate parts of Northern Europe, bracken fronds emerge above the soil surface  
14 from rhizomes in mid-May in the form of tightly curled crosiers, with expansion triggered  
15 by increasing soil temperatures (Ader, 1990; Page, 1997). The fronds reach maturity from  
16 late July to early September, which is followed by senescence. In the United Kingdom  
17 frond heights from 60 to 110 cm are typical in open ecosystems, while the fronds are higher  
18 in forests (150 cm or more) (Watt, 1976; Whitehead and Digby, 1997). Typically, the frond  
19 density is between 16 and 60 fronds m<sup>-2</sup> in the United Kingdom (Nicholson and Paterson,  
20 1976; Watt, 1976; Callaghan et al., 1984; Whitehead and Digby, 1997).

21  
22 The ptaquiloside content in fronds decreases gradually during the growth season but never  
23 reaches zero. Maximum is found in the newly emerged crosiers, e.g. vars. *caudatum* and  
24 *arachnoideum* in Venezuela both show maximum ptaquiloside contents when in the  
25 crosier-stage, decreasing rapidly to 10% of maximum after a few weeks. In 7-8 weeks old  
26 and older fronds, the ptaquiloside contents remains consistently low (Saito et al., 1989;  
27 Alonso-Amelot et al., 1992; Alonso-Amelot et al., 1995). The annual variation in

1 ptaquiloside content in var. *aquilinum* fronds from Denmark follows this pattern, although  
2 not as marked (Rasmussen and Hansen, 2003). The ptaquiloside contents in bracken fronds  
3 of all varieties ranges between 0-15,000 µg/g, from 100-9,800 µg/g in var. *aquilinum* and  
4 40-1,800 µg/g in var. *latiusculum* (Smith et al., 1994). There are no significant differences  
5 between varieties, but genetic heritage may influence the content and explain the  
6 occurrence of fronds free of ptaquiloside (Smith et al., 1994). Smith et al. (1994) reported  
7 that fronds from var. *esculentum* sampled in South Australia contained higher ptaquiloside  
8 contents compared with those from the North Australia. The pattern was consistent when  
9 the ferns were grown at similar conditions for several years.

10

11 Rasmussen and Hansen (2003) investigated the annual variation in the ptaquiloside content  
12 in the rhizomes of var. *aquilinum* growing in Denmark, and reported contents between 2  
13 and 7,046 µg/g. Generally, rhizome contents fell from those in spring, increasing to  
14 maximum after fronds had reached maturity.

15

16 Ptaquiloside contents ranging from 0.09 to 8.49 µg/g have been recorded in bracken litter in  
17 Denmark. In associated topsoil, levels between 0.01 and 0.09 µg/g were found (Rasmussen  
18 et al. 2003a; Rasmussen et al. 2003b).

19

20 Various studies have investigated the environmental/edaphic causes for variations in  
21 ptaquiloside contents in fronds. The content in fronds was correlated with altitude in South  
22 America (vars. *arachnoidum* and *caudatum*; Alonso-Amelot et al., 1995; Alonso-Amelot et  
23 al., 2000a; Villalobos-Salazar et al., 2000). However, the correlation is complex, as  
24 Villalobos-Salazar *et al.* (2000) found positive correlation between altitude and  
25 ptaquiloside content for vars. *arachnoideum* and *caudatum*, while Alonso-Amelot *et al.*  
26 (1995) found negative correlation for var. *caudatum*. Similar confusing results were  
27 obtained in New Zealand for var. *esculentum* (Rasmussen et al. 2003d). Alonso-Amelot et

1 al. (2000a) suggests that the decrease in ptaquiloside content with altitude may be due to  
2 lower temperatures at higher altitudes imposing restrictions in the metabolic rate and  
3 production of secondary metabolites. Climatic stress (drought, temperature and frost) is  
4 known to increase toxin concentrations in many plants (Ralphs, 2002).

5

6 An investigation of frond ptaquiloside contents in var. *aquilinum* in Denmark indicated that  
7 ptaquiloside might be produced in response to environmental stress as growth conditions  
8 characterized by shading from trees, low amounts of plant available water in the soil and  
9 scarcity of nutrients increased the level of ptaquiloside in fronds (Rasmussen et al., 2003b).  
10 The effect of shading has also been demonstrated for var. *esculentum* growing under  
11 controlled circumstances (Smith et al., 1990). The stress-hypothesis is supported by Smith  
12 et al. (1994) finding higher levels of ptaquiloside in fronds of var. *esculentum* subjected to  
13 mowing or grazing compared to non-disturbed fronds. However, the higher contents in  
14 damaged ferns may have been due to the fronds being at an earlier growth stage (regrowth  
15 after damage) than the undamaged fronds included in the study.

16

17 Bracken has been reported to respond to edaphic growth conditions in its production of  
18 other secondary metabolites than ptaquiloside. For example higher prurasin contents are  
19 found in fronds from shaded woodland habitats compared to open land and the release of  
20 HCN from brackens seems to be correlated with rainfall and temperature (Cooper-Driver  
21 and Swain, 1976; Cooper-Driver et al., 1977; Hadfield and Dyer, 1986).

22

23 A general need exists for understanding the dynamics in ptaquiloside variation inside  
24 bracken stands as well as of the geographical variation in the ptaquiloside content of  
25 different bracken stands. The aims of this study were therefore to: 1) Monitor the annual  
26 and regional variation in the level of ptaquiloside encountered in fronds, rhizomes and litter  
27 from Scottish bracken; 2) Unveil any edaphic influence on the observed ptaquiloside

- 1 contents, and 3) Examine the effect of bracken cutting on the ptaquiloside level in fronds,
- 2 rhizomes and litter.

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## MATERIALS AND METHODS

### Field sites, study of annual variation:

Three locations in Scotland with bracken-dominated ecosystems were chosen for the investigation (Loch Grannoch, Muir of Dinnet, Isle of Mull; Figure 2). The locations were situated on open land except from Muir of Dinnet, which was situated in a glade. The stands were growing on different kinds of soil materials (Table 1-4). Sampling was performed on all sites in April, May, August, September, and November 2002.

### Field sites, study of regional variation:

Twenty bracken stands situated across Scotland were chosen for the investigation (Figure 2). The stands were situated in forests or on moors. On all sites the bracken coverage was high or dominant (Table 1-4). Sampling was performed one time at every site in the period September 16 to October 2, 2002, i.e. by the end of the growth season.

### Field site, cutting trial:

A single location (Craibstone) was chosen for the cutting trial (Figure 2, Table 1-4). The site was located on open land. Sampling was performed monthly in the first week of June, July, August, September, October, and November 2002. Fronds, which emerged in response to sampling in June and July plots, were resampled 30 days later.

### Field work and sample pre-treatment:

Before sampling, the stand was divided into 5 homogeneous sub-sites, from which one square metre (the sampling-area) was delineated in the central part of each sub-site. At each sampling-area the frond-density was counted, and 2 whole fronds were harvested for analysis, while approx. 0.5 m of rhizomes (both frondbearing and storage rhizomes) were collected from a small trench dug at the southern part of the sampling-area. The diameter

1 and abundance of rhizomes was measured and counted in the trench. All fronds were  
2 separated from the rhizome at the soil surface, their frond- and stipe length was measured,  
3 and the number of unfurling pinnae counted, enabling determination of the bracken growth  
4 stage. Each trench was dug to the depth of the R-horizon, and this depth recorded. Detailed  
5 site-descriptions were performed according to the FAO-Unesco system (FAO, 1990). The  
6 FAO-Unesco site-descriptions were modified to include: 1) Size of bracken stand (<50m<sup>2</sup>,  
7 50-200m<sup>2</sup>, 200-500m<sup>2</sup>, and >500m<sup>2</sup>); 2) Bracken ground cover (low [15%], moderate [15-  
8 40%], high [40-80%], and dominant [80-100%]); 3) Light conditions (shaded [below light  
9 tree species], glade [stand encircled by trees {>3m tall}], shrub [fronds growing between  
10 small trees {<3m tall}, open land]); 4) Bracken status (suppressed [other species dominate  
11 the herbal part of the ecosystem], subdominant [bracken is outnumbered by another herb],  
12 dominant [bracken is the most prominent herb]); 5) Bracken community growth stage (new  
13 [only new plants {max. 1 year}], young [no old fronds present, no litter], mature [old  
14 fronds and litter present], and old [bracken community is degenerating] (Watt, 1976)); and  
15 6) Bracken disturbance (none, bitten [by animals], managed [cut, sprayed etc.]). All  
16 collected samples from the 5 sub-sites were pooled to give one sample from each site.  
17 Growth, density, soil- and stand descriptions were reported as averages of the 5 sub-sites.  
18 All samples were brought back to the laboratory as soon as possible and dried at 45°C for 5  
19 days before milling to a particle size less than 2 mm and storage at -20°C.

20

#### 21 Ptaquiloside determination:

22 The ptaquiloside-content was determined in aqueous bracken extracts after cleaning of the  
23 sample with Polyamide 6 resin (Fluka, Steinheim, Switzerland) (Agnew and Lauren, 1991).  
24 Ptaquiloside from dried samples of fronds and rhizomes were extracted with deionized  
25 water for 60 min. in the ratio 1:50 (weight:volume). Ptaquiloside from litter were extracted  
26 the same way, but in the ratio 1:7.5 (weight:volume). The suspension was centrifuged at  
27 11,200 g for 5 min. before an aliquot was passed through a 1.0 x 10 cm I.D. glass Econo-



1 Column<sup>®</sup> (BIO-RAD, New York, USA) drypacked with Polyamide resin in the ratio 8:1  
2 (volume:weight). Ptaquiloside was converted to pterosin B by base-acid treatment  
3 following the method of Agnew and Lauren (1991), and pterosin B determined by UV-  
4 absorption at 214 nm at 35°C using a Merck Hitachi HPLC system (L-4200 UV-VIS  
5 Detector; 655A-40 Auto-Sampler; D-6000A Interface; L-6200 Intelligent Pump) equipped  
6 with Merck LiChroCART<sup>®</sup> column (125 x 4 mm) packed with LiChrospher<sup>®</sup> 100 RP-8, 4  
7 µm. The separations were performed at isocratic mode with a water-acetonitril eluent  
8 (74:26 volume:volume) using a flow of 1.0 ml/min. The column was flushed after each  
9 analysis, and the total time of analysis was 30 min, with a retention time for pterosin of  
10 approx. 9 min. Sample injection-volumes of 20-100 µl were used for analysis, and all  
11 samples were measured in duplicate. Pterosin B was quantified by using external standards.  
12 The limits of detection were 0.1 0.1 µg/g for litter and 0.7 µg/g for fronds and rhizomes.  
13 Standards of pterosin B was made from pure ptaquiloside obtained from Professor Ojika  
14 (Nagoya University, Japan).

15

16 Auxiliary analysis:

17 Soil pH was measured in a 1:2.5 or 1:10 soil:0.01 M CaCl<sub>2</sub> suspension in mineral soil or  
18 organic soil samples respectively (Rowell, 1994). The content of organic carbon was  
19 determined in all samples (soil, litter, rhizomes and fronds) by dry combustion using an  
20 Eltra CS 500 Carbon Sulphur Determinator (Rowell, 1994). Soil, litter, frond and rhizome  
21 nitrogen was determined by the Kjeldahl method (Rowell, 1994) in air-dried samples. All  
22 analyses were made as dup- or triplicates. The results are given as weight percentages on  
23 air-dry basis.

24

25 Statistical analysis:

26 Multiple linear regression (MLR) were carried out on all ptaquiloside- site- and stand data  
27 shown in Tables 1 and 2 to unveil correlations between environmental/edaphic parameters

1 and the ptaquiloside-content in fronds, rhizomes and litter. The significance of individual  
2 variables and of the resulting model was tested in analysis of variance. All statistical  
3 calculations were performed in the software package *The Unscrambler*<sup>®</sup> ver. 7.6 SR1  
4 (Camo ASA, Oslo, Norway). Geographical coordinates in latitude and longitude were  
5 transformed to 10-digit numbers before modelling.

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## RESULTS AND DISCUSSION

### Bracken growth at Loch Grannoch, Muir of Dinnet, and Isle of Mull:

Aboveground frond growth began in the first part of May, and the fronds matured in August (Figure 3). The maximum frond lengths were between 150 and 200 cm, which are in accordance with previous investigations of var. *aquilinum* in the United Kingdom. The frond densities were in the range 24 to 42 fronds m<sup>-2</sup>, which is relative high compared to investigations of var. *aquilinum* in Denmark, but in accordance with previously reports from the United Kingdom ( Nicholson and Paterson, 1976; Whitehead and Digby, 1997; Rasmussen and Hansen, 2003; Rasmussen et al. 2003a; Rasmussen et al. 2003b). The longest fronds were found at Muir of Dinnet, the site with the lowest density, while the smallest fronds were encountered at Loch Grannoch, which had the highest density of the three sites. However, this inverse relationship between frond height and density does not exist at regional scale (Table 3). Frond density increased from May to August, in accordance with earlier reports from the United Kingdom, but in contrast with the growth pattern of var. *aquilinum* observed in forested ecosystems in Denmark (Rasmussen et al. 2003b).

### Annual variation in the ptaquiloside content in fronds, rhizomes and litter at Loch Grannoch, Muir of Dinnet, and Isle of Mull:

Ptaquiloside was found in all frond, rhizome and litter samples analysed (Figure 4A-C). Ptaquiloside contents in the fronds ranged between 50 and 5,790 µg/g, with maximum contents at all sites recorded in May. Concentrations decreased as the fronds matured, similar to previous observations of ptaquiloside variations in var. *aquilinum* fronds. However, the ptaquiloside content was higher when compared to var. *latiusculum*. Ptaquiloside contents in the rhizomes ranged between 11 and 90 µg/g. Contents at Isle of Mull and Muir of Dinnet were both low between June and August, with maximum contents

1 around 1<sup>st</sup> of October. Rasmussen and Hansen (2003) noted a similar trend in rhizome  
2 contents in Denmark, although with considerably higher contents in the autumn of up to  
3 7,000 µg/g. Ptaquiloside contents in the rhizomes at Loch Grannoch were lowest in May  
4 and August/September, with maximum contents recorded in November. The difference  
5 between Loch Grannoch and the other sites is likely due to fronds reaching maturity later at  
6 this site due to higher altitude (Atkinson, 1989). Ptaquiloside contents in bracken litter  
7 ranged between 0.1 and 3.5 µg/g, similar to that observed for var. *aquilinum* in Denmark  
8 (Rasmussen et al., 2003a; Rasmussen et al., 2003b). At Loch Grannoch and Muir of Dinnet,  
9 the contents increased towards autumn. Ptaquiloside contents at Isle of Mull displayed no  
10 marked variation. Higher contents in the litter were expected in the autumn due to higher  
11 amounts of precipitation in this season that might leach ptaquiloside from living, dead or  
12 dying bracken fronds. Also, the litter at this time would contain fronds that had died after  
13 being shaded out, releasing ptaquiloside. Why this was not the case at Isle of Mull cannot  
14 be explained from the observations performed in this study, although it might have been  
15 due to the higher rate of precipitation in West Scotland compared to the eastern part.

16

17 Regional variation in ptaquiloside contents in Scotland:

18 Ptaquiloside was found in all samples analysed (Table 4). Contents in the fronds ranged  
19 between 87 and 2,450 µg/g with a median content of 787 µg/g. This is in accordance with  
20 earlier findings for vars. *aquilinum* and *latiusculum*. MLR analysis resulted in a rather poor  
21 cross-validated model for the impact of the environmental/edaphic parameters on the  
22 ptaquiloside content in fronds, although showing significant impact of longitude (western  
23 longitude), orientation (aspect), disturbance (mowing or grazing damage), and frond height  
24 (Figure 5A; Equation 1, standard deviations shown; units as in Table 1-4).

25

$$\log Pta = 3.085(\pm 0.310) - 0.103(\pm 0.046) \cdot \text{Longitude}$$

Eq. 1

$$+ 0.004(\pm 0.001) \cdot \text{Orientation} - 0.591(\pm 0.196) \cdot \text{Disturbance} \\ - 0.004(\pm 0.002) \cdot \text{Fronnd height}$$

1

2 However, the calibration correlation coefficient was acceptable indicating high leverage of  
 3 individual sites on the cross-validated model. The model indicates that fronds with high  
 4 ptaquiloside contents are found in the eastern part of Scotland, in stands pointing to the  
 5 south and west, which are not being disturbed and which are small. The model might reflect  
 6 genetic variation across Scotland like previously observed in Australia for var. *esculentum*  
 7 or it merely reflects the distribution of the two Scottish bracken varieties and hybrids. The  
 8 negative effect of longitude might also be due to the more senescent stage of brackens  
 9 sampled in the western part of Scotland compared to the eastern part. A climatic gradient  
 10 with respect to precipitation is also found going east-west in Scotland, with lower amounts  
 11 of precipitation found in the eastern part. However, it is warmer in the western part of  
 12 Scotland, due to the Gulf Stream, leading to a higher number of frost-free days. The  
 13 negative effect of disturbance is surprising, but might be due to the fact that only two stands  
 14 showed signs of physical disturbance and the effect could therefore be random. The  
 15 negative effect of frond height is in accordance with the findings of Rasmussen et al.  
 16 (2003b).

17

18 Ptaquiloside contents in rhizomes ranged between 17 and 657  $\mu\text{g/g}$ , with a median content  
 19 of 118  $\mu\text{g/g}$  (Table 4). MLR-analysis showed correlation between ptaquiloside in the  
 20 rhizomes and longitude (western longitude), topography (slope), and content of soil  
 21 nitrogen (Figure 5B; Equation 2, standard deviations shown; units as in Table 1-4).

22

$$\log \text{Pta} = 2.532(\pm 0.286) + 0.124(\pm 0.057) \cdot \text{Longitude} \quad \text{Eq. 2} \\ - 0.040(\pm 0.009) \cdot \text{Topography} - 1.844(\pm 0.636) \cdot \text{Soil N}$$

23

1 The model indicates that high levels of ptaquiloside are found in rhizomes originating from  
2 stands growing at flat terrain in the western part of Scotland on soils with low amounts of  
3 nitrogen. The effect of longitude might be due to the same reasons as indicated for the  
4 fronds. The effect of topography is not easily explainable, and might just be a proxy  
5 variable for some other cause. This is the first attempt to correlate the ptaquiloside content  
6 in the rhizome-system with external growth factors.

7

8 Ptaquiloside contents in bracken litter ranged between 0.13 and 5.77 µg/g with a median  
9 content of 0.62 µg/g (Table 4). The content was well correlated with the latitude (northern  
10 latitude), altitude, frond development stage (number of pinnae) and the content of  
11 ptaquiloside in the frond (Figure 5C; Equation 3, standard deviations shown; units as in  
12 Table 1-4).

13

$$\begin{aligned} \log(\text{Pta}+1) = & -6.202(\pm 1.548) + 0.095(\pm 0.027) \cdot \text{N} - 0.001(\pm 0.000) \cdot \text{Altitude} & \text{Eq. 3} \\ & + 0.047(\pm 0.010) \cdot \text{Development stage} + 0.185(\pm 0.079) \cdot \log \text{Pta}(\text{frond}) \end{aligned}$$

14

15 High levels of ptaquiloside in litter was found in the northern part of Scotland, inside stands  
16 situated at low altitudes, below well-developed fronds having high ptaquiloside contents.  
17 The effect of altitude and latitude is not readily explainable. It is colder in the northern part  
18 of Scotland and as ptaquiloside degradation is strongly temperature dependent, colder  
19 climate should preserve ptaquiloside longer (Rasmussen et al., 2003c). The same effect  
20 should be seen with respect to altitude, as temperature tends to decrease with this  
21 parameter. Instead, the effect of altitude is negative. However, the negative effect could be  
22 caused by increased precipitation at altitude. The effect of altitude needs further  
23 investigation. The positive effect of frond content, may have three reasons: 1) Ptaquiloside is  
24 present in the litter from dead fronds originating from earlier growth seasons; 2)  
25 Ptaquiloside is leaching from the fronds to the soil, or 3) Ptaquiloside originates from parts

1 of damaged fronds from the actual growth season (broken pinnae etc.). The effect of frond  
2 development stage is unclear, since it was expected that all fronds would have reached the  
3 same and maximum number of pinnae and pinnules by the end of September. However,  
4 stands at altitude have shorter growing season, so they would not necessarily reach the  
5 same stage before growth ceased. Compared to the explanation of the ptaquiloside content  
6 in litter given by Rasmussen et al. (2003b), the result obtained in the present investigation  
7 is more straightforward, but still indicates the complexity of the origin of ptaquiloside in  
8 litter.

9

10 When compared to Danish investigations of the regional variation in the ptaquiloside  
11 content, the result is somewhat complex. Broadly speaking, ptaquiloside in bracken may be  
12 a function of the ferns genetic heritage and its growth conditions. The inability to obtain  
13 results similar to the Danish findings may therefore indicate: 1) Relatively large genetic  
14 variation in Scottish bracken compared to Danish bracken partly attributed to the presence  
15 of var. *latiusculum* and *latiusculum-aquilinum* hybrids in Scotland, and/or 2) More complex  
16 or different interactions between environmental/edaphic variables and ptaquiloside contents  
17 than in Denmark; 3) Greater range of habitats sampled.

18

#### 19 Cutting trial:

20 The effect of the harvesting of fronds on the ptaquiloside content in fronds, rhizomes, and  
21 litter is shown in Figure 6A-B. The ptaquiloside content in the fronds and rhizomes follow  
22 a declining pattern as shown for Loch Grannoch, Isle of Mull, and Muir of Dinnet.  
23 However, in October a large increase in the frond content was observed. The content in the  
24 recut fronds originating from rhizomes, which have had their fronds cut 30 days earlier, is  
25 significantly higher than in the uncut fronds. There is no effect on the ptaquiloside content  
26 in rhizomes. The content in the recut fronds is also much higher than in uncut fronds at the  
27 same growth stage (i.e. compare July-Recut with June). It is possible that some ptaquiloside

1 in the re-cut fronds is translocated from storage rhizomes, as frond-bearing rhizome  
2 carbohydrate stores are likely to be depleted due to harvesting. Storage rhizomes contain  
3 higher contents of ptaquiloside (Rasmussen and Hansen, 2003). When comparing the litter  
4 at the cut area with the uncut area, a lower ptaquiloside content in the recut area is seen in  
5 July and August. The lower content is likely due to the more exposed nature of the recut  
6 litter lying on bare ground compared to litter below protecting bracken fronds. Less  
7 ptaquiloside leaching from fronds into the litter may also explain this, as the fronds were  
8 cut.



## CONCLUSION

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Ptaquiloside contents in Scottish bracken fronds, rhizomes and litter are in accordance with previous findings for vars. *aquilinum* and *latiusculum*. Also, the annual variation in contents is consistent with earlier reports for var. *aquilinum* in Denmark. The latitude and altitude of the stand, the frond development stage and the ptaquiloside-content was well correlated with the ptaquiloside content in the litter, while the content in rhizomes were correlated with the longitude, topography and development stage of the stand. The ptaquiloside levels in the fronds were partly correlated with longitude and aspect of the stand, frond-height and presence of frond damage. Fronds that emerged following harvesting, contained higher amounts of ptaquiloside compared to uncut fronds of the same development stage. However, decreased levels of ptaquiloside in the litter were found on harvested areas. Compared to similar studies of the ptaquiloside occurrence in bracken from Denmark, the genetic heritage and other growth factors than the recorded may influence the ptaquiloside content in bracken.

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## FIGURE CAPTIONS

<b>Table no:</b>	<b>Text:</b>
<b>1</b>	TABLE 1 SITE DESCRIPTION I: LOCATION AND SITE CHARACTERS.
<b>2</b>	TABLE 2 SITE DESCRIPTION II: SOIL PROPERTIES. SD <sup>a</sup> SHOWN.
<b>3</b>	TABLE 3 STAND DESCRIPTIONS (SEPTEMBER 2002) I: PHYSICAL DESCRIPTION. SD <sup>a</sup> SHOWN.
<b>4</b>	TABLE 4 STAND DESCRIPTIONS (AUTUMN 2002) II: PTAQUILOSIDE CONTENTS ETC. SD <sup>a</sup> SHOWN.
<b>Figure no:</b>	<b>Text:</b>
<b>1</b>	FIG. 1. Ptaquiloside.
<b>2</b>	FIG. 2. Map of Scotland (minus Shetland and Orkney Islands). Study sites marked. All sites were used for the regional survey. Craibstone (white star) also used for cutting trial. Loch Grannoch, Isle of Mull, and Muir of Dinnet (grey stars) also used for monitoring the annual variation in the ptaquiloside content.
<b>3</b>	FIG. 3. Bracken growth pattern at Loch Grannoch, Isle of Mull, and Muir of Dinnet. Standard deviations marked.
<b>4</b>	FIG. 4. Annual variations in the ptaquiloside content in: A) Fronds; B) Rhizomes; C) Litter. Sites: Loch Grannoch, Isle of Mull, and Muir of Dinnet. Standard deviations shown.
<b>5</b>	FIG. 5. Multiple linear regression models for the ptaquiloside content in: A) Fronds ( $P = 0.0035$ ); B) Rhizomes ( $P = 0.0005$ ); C) Litter ( $P = 0.0001$ ). All variables have $P < 0.05$ . $R^2$ = Calibration correlation coefficient. $Q^2$ = Cross-validated correlation coefficient.



<b>6</b>	FIG. 6. Craibstone cutting trial: A) Ptaquiloside content in fronds (filled bars) and rhizomes (open bars); B) Ptaquiloside content in litter (September is missing due to contamination of sample). Standard deviations shown.
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**TABLE 1 SITE DESCRIPTION I: LOCATION AND SITE CHARACTERS.**

Site	N	W	Altitude (m)	Orientation (deg)	Topography (%)	Landform	Land element	Human influence	Land use	Vegetation	Light conditions
<b>Loch Grannoch</b>	54:59:29	4:16:49	200	276	15	upland	slope	none	conservation	forest-scrub	open land
<b>Isle of Mull</b>	56:28:52	6:07:18	50	210	20	upland	slope	none	semi-extensive sheep	grassland	open land
<b>Muir of Dinnet</b>	57:05:16	2:54:39	170	198	5	upland	slope	none	conservation	forest	glade
<b>Craibstone</b>	57:11:20	2:13:00	90	115	2	hill	depression	none	conservation	grassland-scrub	open land
<b>Balloch Wood</b>	54:54:45	4:33:51	40	210	0.5	valley	floodplain	none	forest	oak, alder, hazel	glade
<b>Ben Nevis</b>	56:46:47	4:59:46	210	175	27	mountain	slope	none	conservation	grassland	open land
<b>Elrick Hill High</b>	57:10:54	2:15:01	190	155	20	hill	slope	none	conservation	grassland-scrub	open land
<b>Elrick Hill Low</b>	57:10:53	2:14:51	150	260	5	valley	valley floor	none	conservation	grassland-forest	glade
<b>Glencoe</b>	56:39:43	4:57:31	280	20	15	mountain	slope	none	conservation	grassland	open land
<b>Grey Mare's Tail</b>	55:25:18	3:16:58	320	185	15	mountain	slope	none	semi-extensive sheep	grassland	open land
<b>Kirkhill</b>	57:11:37	2:14:49	150	100	3	upland	terrace	none	exotic timber production	forest	glade
<b>Loch Trool</b>	55:05:36	4:29:09	150	170	15	upland	slope	burning	conservation	grassland-scrub	open land
<b>Muir of Dinnet cut</b>	57:05:21	2:54:56	170	200	2	upland	slope	clearing	conservation	grassland-forest	glade
<b>Newton Stewart</b>	54:58:46	4:33:51	100	160	5	upland	plateau	none	exotic timber forestry	forest-scrub	glade
<b>Rockcliffe</b>	54:52:14	3:48:25	30	140	2	hill	slope	none	conservation	grassland-forest	open land
<b>Schiehallion</b>	56:40:32	4:02:25	350	75	5	hill	slope	none	semi-extensive sheep	grassland	open land
<b>Skye</b>	57:12:53	5:57:05	30	240	2	upland	valley floor	none	semi-extensive sheep	grassland	open land
<b>Stronord</b>	54:57:26	4:25:11	70	130	2	hill	terrace	none	exotic timber production	forest	glade
<b>Torridon Beach</b>	57:32:23	5:30:39	5	260	2	coastal plain	floodplain	none	conservation	misc.	open land
<b>Torridon Hill</b>	57:32:50	5:30:36	40	160	7	upland	slope	none	conservation	grassland-scrub	open land

**TABLE 2 SITE DESCRIPTION II: SOIL PROPERTIES. SD<sup>a</sup> SHOWN.**

Site	Thickness of litter <sup>b</sup> (cm)	Effective soil depth (cm)	Litter C <sup>bc</sup> (%)	Litter N <sup>bc</sup> (%)	Soil pH <sup>cd</sup>	Soil C <sup>cd</sup> (%)	Soil N <sup>cd</sup> (%)
<b>Loch Grannoch</b>	9±5	36±12	48.7	1.12	3.46	8.90	0.49
<b>Isle of Mull</b>	9±3	56±35	46.8	1.23	4.40	8.94	0.47
<b>Muir of Dinnet</b>	8±2	44±20	47.8	1.06	4.03	5.17	0.32
<b>Craibstone</b>	6±3	54±25	50.1	1.37	3.93	3.76	0.28
<b>Balloch Wood</b>	0±0	79±7	49.2	0.81	3.85	2.43	0.14
<b>Ben Nevis</b>	5±5	32±18	47.9	0.60	4.11	4.01	0.25
<b>Elrick Hill High</b>	13±2	56±4	49.3	1.01	3.31	4.77	0.20
<b>Elrick Hill Low</b>	6±3	66±18	45.3	1.27	3.51	9.08	0.48
<b>Glencoe</b>	3±2	41±20	48.2	0.62	4.09	4.42	0.31
<b>Grey Mare's Tail</b>	4±2	24±6	48.2	1.37	3.72	6.68	0.51
<b>Kirkhill</b>	7±2	43±10	48.4	1.00	3.68	4.95	0.17
<b>Loch Trool</b>	3±1	31±15	48.4	0.73	4.19	5.35	0.36
<b>Muir of Dinnet cut</b>	4±1	44±34	45.4	1.51	4.02	3.48	0.24
<b>Newton Stewart</b>	13±5	44±17	49.7	1.29	3.60	9.69	0.42
<b>Rockliffe</b>	14±3	84±8	49.6	1.26	3.86	4.44	0.32
<b>Schiehallion</b>	1±0	76±21	46.8	0.56	4.34	3.98	0.28
<b>Skye</b>	2±1	71±23	48.3	0.47	4.22	4.24	0.36
<b>Stronord</b>	12±4	43±12	48.9	1.46	3.99	6.42	0.43
<b>Torridon Beach</b>	4±2	71±17	46.3	0.86	4.00	6.15	0.44
<b>Torridon Hill</b>	9±2	44±20	46.7	0.60	3.28	2.86	0.21

<sup>a</sup> Standard deviation.

<sup>b</sup> Litter comprises Oi, Oe and Oa-horizons.

<sup>c</sup> Average coefficient of variation: 3.0% (litter C); 2.1% (litter N); 0.2% (soil pH); 0.7% (soil C); 0.2% (soil N).

<sup>d</sup> Soil refers to the 20 cm below the litter layer (A or A/Oa horizons).

**TABLE 3 STAND DESCRIPTIONS (SEPTEMBER 2002) I: PHYSICAL DESCRIPTION. SD<sup>a</sup> SHOWN.**

Site	Community development stage	Disturbance	Bracken status	Bracken coverage	Stand size (m <sup>2</sup> )	Fron density (#/m <sup>2</sup> )	Fron height (cm)	Bracken development stage (# pinnae)	Mineralization of litter	Rhizomes (#/m)	Rhizomes <sup>b</sup> (FBR:SR)
<b>Loch Grannoch</b>	developing	none	dominant	high	50-200	42.6±7.2	152±35	17.0±2.0	strong	25.0±2.0	69:31
<b>Isle of Mull</b>	mature	none	dominant	dominant	200-500	29.6±4.4	166±14	15.0±0.0	moderate	37.8±11.9	73:27
<b>Muir of Dinnet</b>	mature	none	dominant	dominant	>500	24.2±4.3	187±23	20.6±1.7	strong	23.2±3.6	68:32
<b>Craibstone</b>	mature	none	dominant	dominant	50-200	34.4±7.8	183±18	ND <sup>c</sup>	moderate	32.2±10.8	72:29
<b>Balloch Wood</b>	mature	none	subdominant	high	<50	17.0±4.3	176±19	12.3±5.9	strong	13.6±5.9	59:40
<b>Ben Nevis</b>	mature	none	subdominant	high	200-500	44.8±8.2	79±11	16.0±2.2	strong	19.6±14.4	73:27
<b>Elrick Hill High</b>	mature	none	dominant	dominant	>500	28.4±6.2	105±16	ND	slight	26.0±3.7	62:38
<b>Elrick Hill Low</b>	mature	none	subdominant	dominant	50-200	33.0±8.5	153±24	10.0±2.8	moderate	23.0±5.6	61:39
<b>Glencoe</b>	mature	none	subdominant	high	50-200	36.6±11.1	88±10	16.1±1.6	strong	14.8±6.5	76:24
<b>Grey Mare's Tail</b>	mature	none	subdominant	dominant	>500	40.4±13.0	97±14	14.3±2.9	strong	30.8±7.5	68:32
<b>Kirkhill</b>	mature	none	subdominant	high	50-200	21.0±4.8	102±18	14.4±2.7	moderate	23.4±11.2	67:33
<b>Loch Trool</b>	mature	none	dominant	dominant	>500	32.8±6.5	113±9	13.6±2.3	moderate	37.2±10.1	71:29
<b>Muir of Dinnet cut</b>	mature	managed	subdominant	dominant	50-200	20.4±5.1	129±25	15.3±3.9	strong	24.2±18.3	59:41
<b>Newton Stewart</b>	developing	none	dominant	high	50-200	27.0±6.9	133±17	12.2±3.2	moderate	26.8±8.0	70:30
<b>Rockcliffe</b>	mature	none	dominant	dominant	>500	27.2±9.3	172±15	16.8±5.2	slight	38.8±10.6	65:35
<b>Schiehallion</b>	mature	none	dominant	dominant	200-500	47.4±7.5	97±11	14.6±2.7	strong	36.0±8.3	66:34
<b>Skye</b>	developing	grazing damage	suppressed	high	200-500	50.0±4.0	57±12	12.1±2.7	strong	29.0±4.7	55:45
<b>Stronord</b>	mature	none	dominant	dominant	50-200	38.2±2.4	117±10	12.7±4.8	slight	38.6±12.2	79:21
<b>Torridon Beach</b>	mature	none	dominant	dominant	50-200	48.6±4.0	126±25	ND	moderate	27.4±9.7	75:25
<b>Torridon Hill</b>	mature	none	subdominant	high	>500	47.8±4.4	94±12	13.7±3.5	slight	26.4±3.0	75:25

<sup>a</sup> Standard deviation.

<sup>b</sup> FBR: Frond bearing rhizomes; SR: Storage rhizomes.

<sup>c</sup> ND: Not determined.

**TABLE 4 STAND DESCRIPTIONS (AUTUMN 2002) II: PTAQUILOSIDE CONTENTS ETC. SD<sup>a</sup> SHOWN.**

Site	Fronde C <sup>b</sup> (%)	Fronde N <sup>b</sup> (%)	Rhizome C <sup>b</sup> (%)	Rhizome N <sup>b</sup> (%)	Fronde-PTA <sup>c</sup> (µg/g)	Rhizome-PTA <sup>c</sup> (µg/g)	Litter-PTA <sup>c</sup> (µg/g)
<b>Loch Grannoch</b>	45.9	1.71	ND	ND	1,280±30	24±0	0.13±0.02
<b>Isle of Mull</b>	45.9	1.72	ND	ND	360±15	61±2	0.40±0.03
<b>Muir of Dinnet</b>	46.0	1.25	ND	ND	720±30	89±5	3.52±0.42
<b>Craibstone</b>	46.9	1.32	43.5	1.40	650±20	125±5	ND <sup>d</sup>
<b>Balloch Wood</b>	48.1	1.22	45.1	0.90	510±15	271±1	0.22±0.04
<b>Ben Nevis</b>	44.9	0.72	45.1	0.42	980±20	18±1	1.29±0.09
<b>Elrick Hill High</b>	46.3	0.81	43.7	0.74	1,280±10	50±2	3.85±0.66
<b>Elrick Hill Low</b>	47.1	1.23	44.1	1.18	1,490±50	38±1	0.43±0.15
<b>Glencoe</b>	43.8	1.03	44.2	0.67	90±10	290±3	0.43±0.13
<b>Grey Mare's Tail</b>	46.3	1.72	46.3	1.11	2,450±280	37±5	0.63±0.05
<b>Kirkhill</b>	45.2	1.03	44.4	0.88	490±60	282±1	1.37±0.10
<b>Loch Trool</b>	45.6	1.32	46.1	0.69	1,150±180	42±1	0.44±0.25
<b>Muir of Dinnet cut</b>	46.3	1.10	46.4	1.34	240±30	256±18	0.62±0.05
<b>Newton Stewart</b>	46.3	1.45	46.2	1.20	1,115±40	60±6	0.84±0.73
<b>Rockcliffe</b>	47.6	1.52	45.9	0.91	820±20	510±10	0.96±0.51
<b>Schiehallion</b>	45.3	1.22	42.4	1.11	860±10	657±18	0.41±0.01
<b>Skye</b>	44.6	1.13	44.7	0.43	760±30	155±6	0.35±0.17
<b>Stronord</b>	45.6	1.86	44.7	1.31	260±5	112±7	0.24±0.04
<b>Torridon Beach</b>	45.3	1.24	43.5	1.33	1,780±10	421±2	5.77±0.71
<b>Torridon Hill</b>	44.3	1.08	43.2	0.41	565±5	519±15	2.52±0.39

<sup>a</sup> Standard deviation.

<sup>b</sup> Average coefficient of variation: 1.8% (frond C); 1.3% (frond N); 2.3% (rhizome C); 2.0% (rhizome N).

<sup>c</sup> PTA: Ptaquiloside-equivalents.

<sup>d</sup> ND: Not determined.

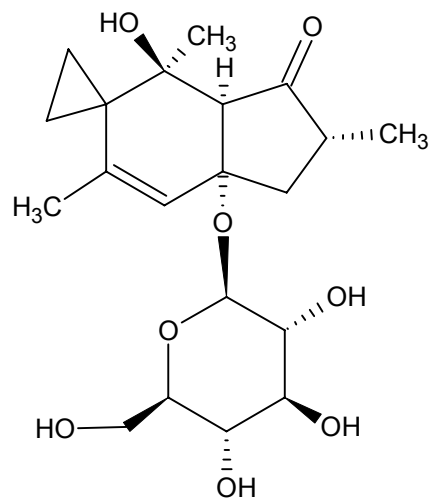


FIG. 1. Ptaquiloside.

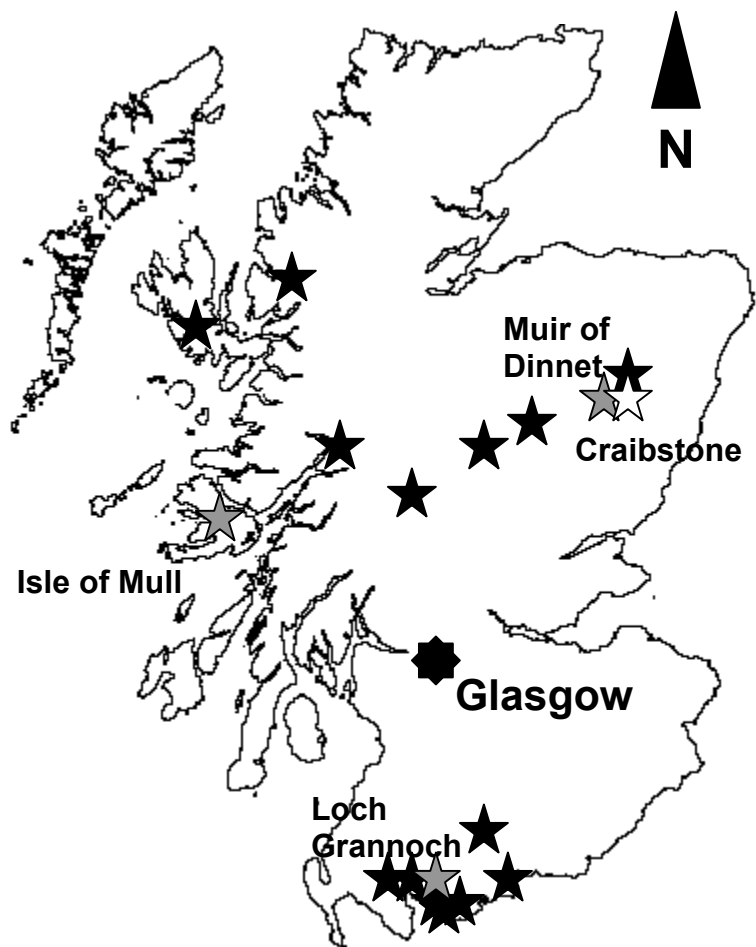


FIG. 2. Map of Scotland (minus Shetland and Orkney Islands). Study sites marked. All sites were used for the regional survey. Craibstone (white star) also used for cutting trial. Loch Grannoch, Isle of Mull, and Muir of Dinnet (grey stars) also used for monitoring the annual variation in the ptaquiloside content.

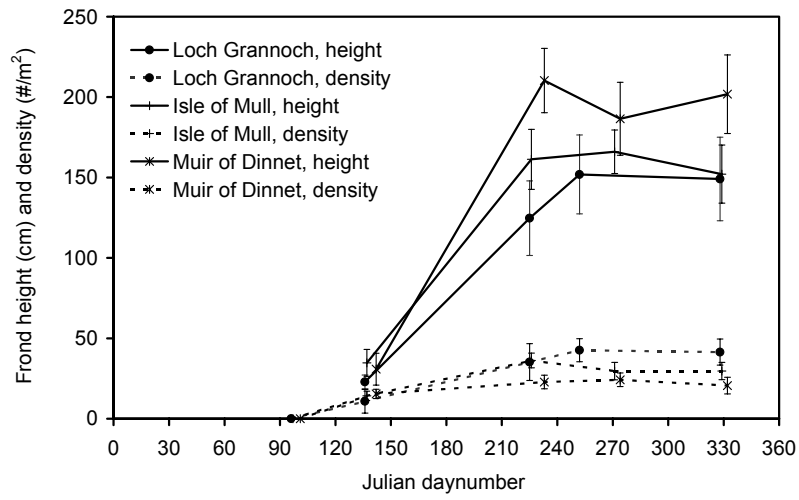


FIG. 3. Bracken growth pattern at Loch Grannoch, Isle of Mull, and Muir of Dinnet. Standard deviations marked.



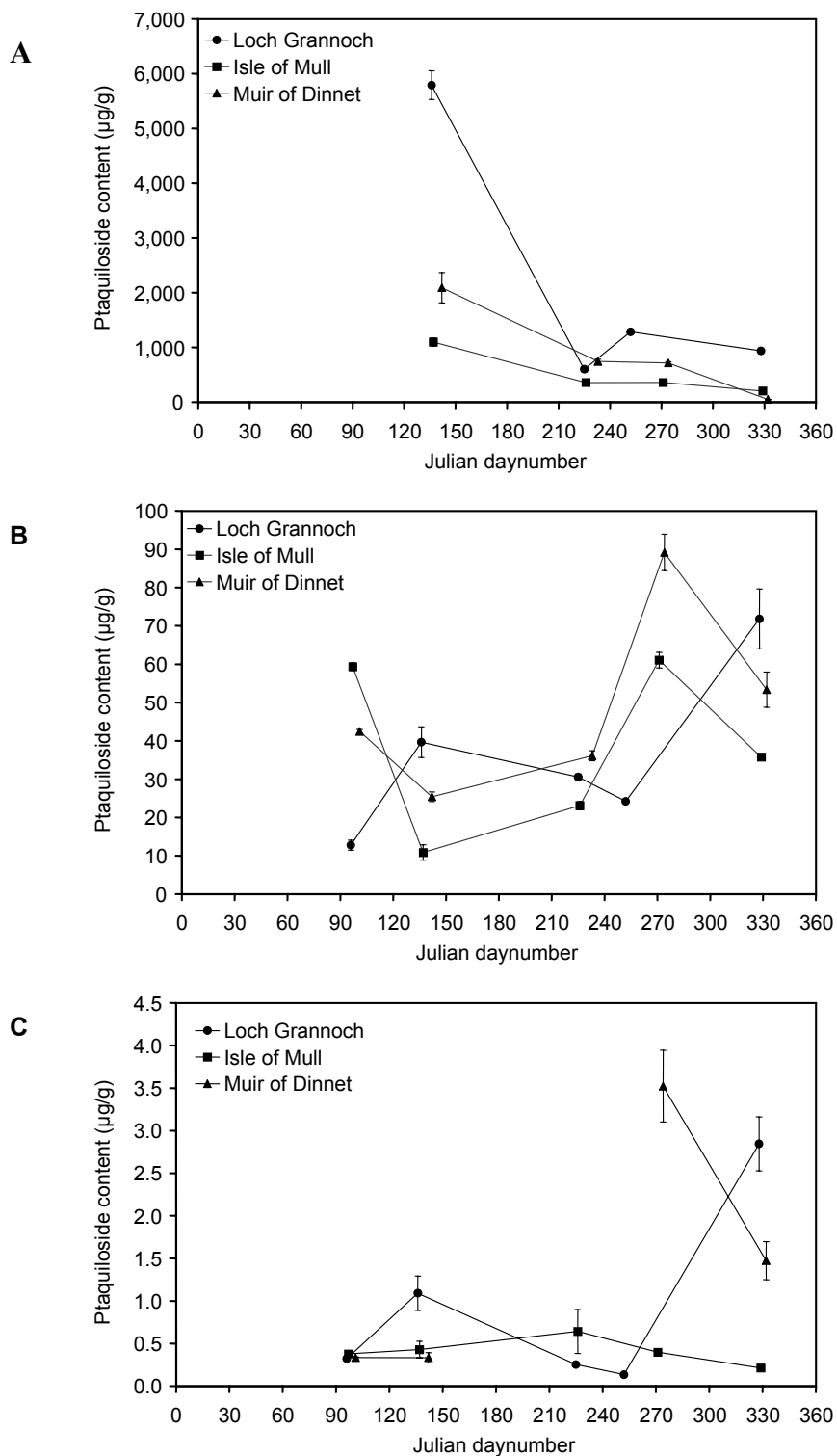


FIG. 4. Annual variations in the ptaquiloside content in: A) Fronds; B) Rhizomes; C) Litter. Sites: Loch Grannoch, Isle of Mull, and Muir of Dinnet. Standard deviations shown.

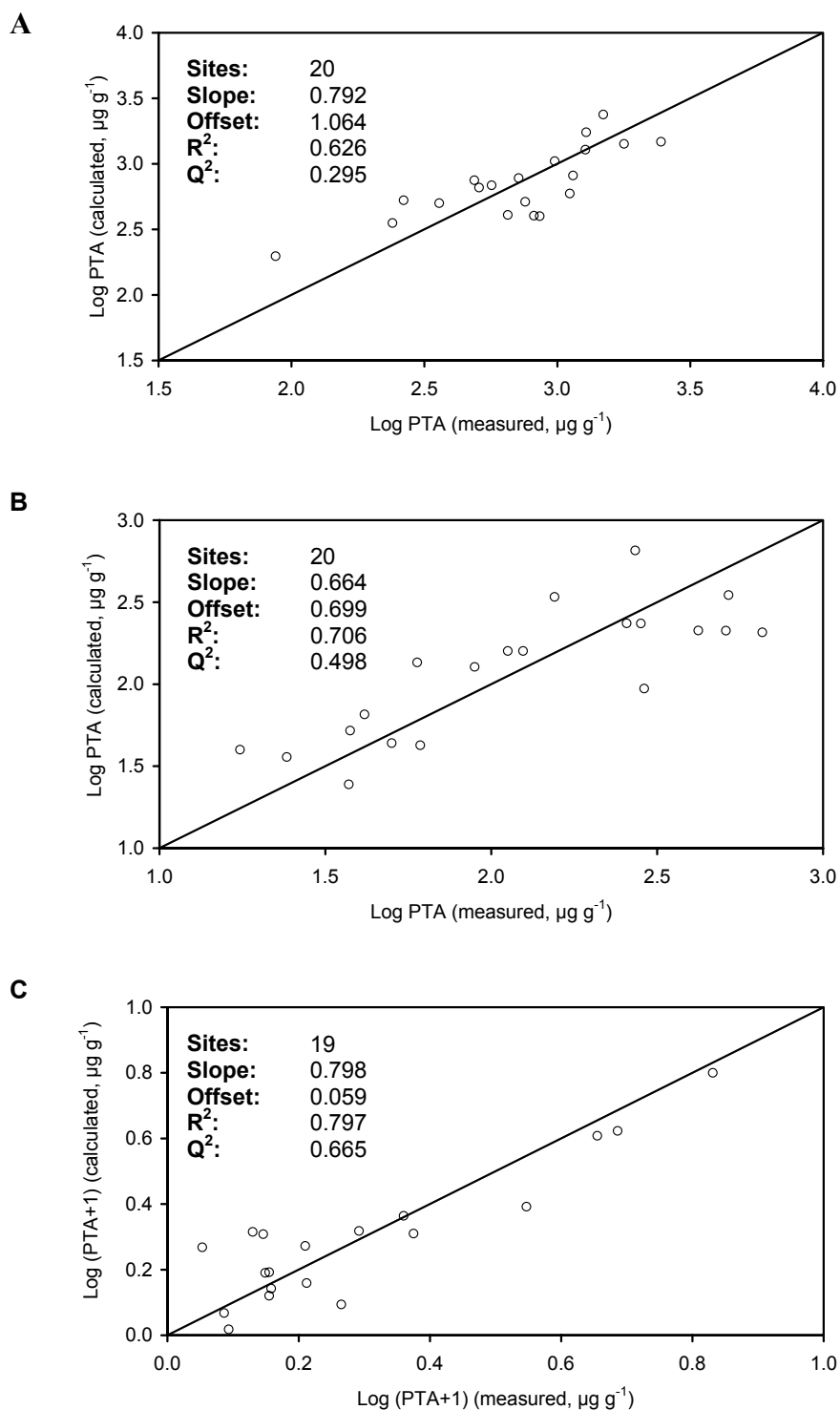


FIG. 5. Multiple linear regression models for the ptaquiloside content in: A) Fronds ( $P = 0.0035$ ); B) Rhizomes ( $P = 0.0005$ ); C) Litter ( $P = 0.0001$ ). All variables have  $P < 0.05$ .  $R^2$  = Calibration correlation coefficient.  $Q^2$  = Cross-validated correlation coefficient.

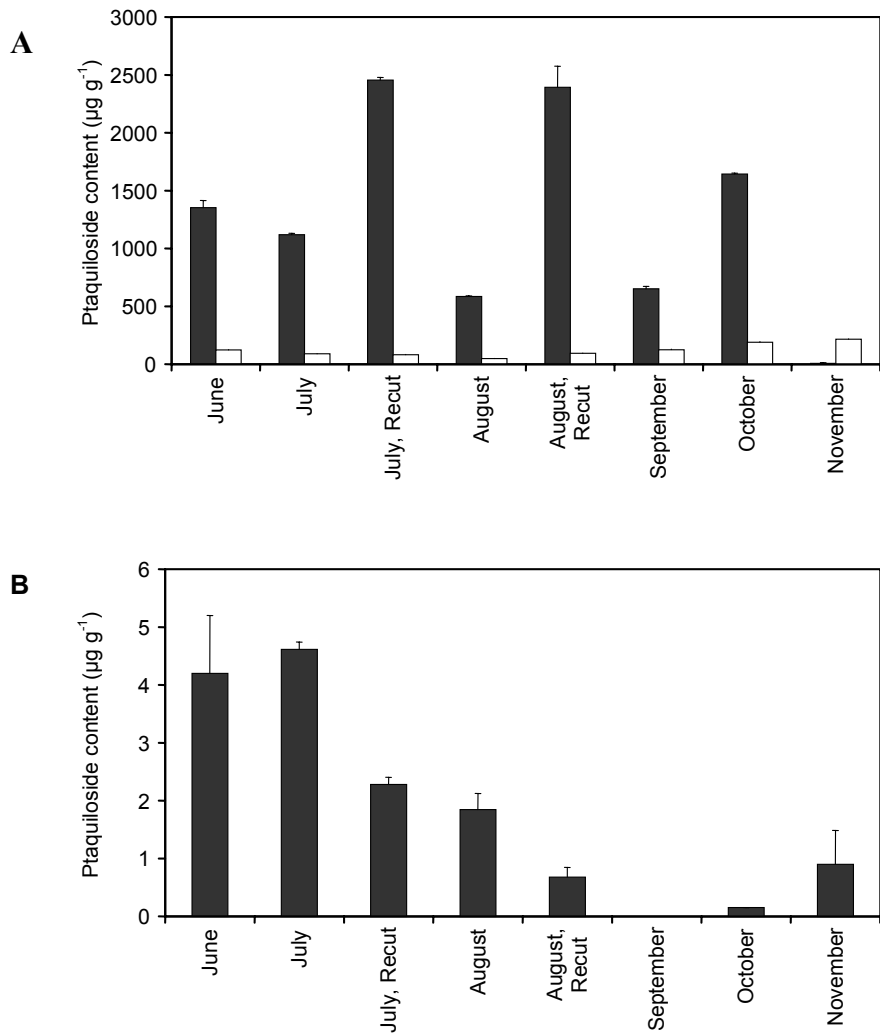


FIG. 6. Craibstone cutting trial: A) Ptaquiloside content in fronds (filled bars) and rhizomes (open bars); B) Ptaquiloside content in litter (September is missing due to contamination of sample). Standard deviations shown.

# Growth of Bracken in Denmark and the Content of Ptaquiloside in Fronds, Rhizomes and Roots

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## Introduction

In Denmark, the Bracken variety Common Bracken (*Pteridium aquilinum* ssp. *aquilinum* var. *aquilinum* (L.) Kuhn) is encountered as a weed on acid soils in plantation forestry and on recreational areas. As a part of an investigation of soil- and groundwater-contamination with the carcinogenic compound ptaquiloside, three typical Common Bracken populations situated in plantations were investigated from April 2000 until April 2001 to quantify the above ground biomass-production as well as in ptaquiloside-production in different plant-compartments (fronds, frondbearing rhizomes, storage rhizomes, and roots).

Frond growth commences around May 1 in Denmark. The fronds reach maturity in late summer (July and August), and finally die back from late September. In case of early autumn night frost, the growth season may stop at an earlier time (Hansen, 1991; Øllgaard and Tind, 1993; Watt, 1976). The fronds reach a height of 100-250 cm. The highest fronds are typically found within forests and plantations, where the ferns are protected against wind and frost. In the United Kingdom where Common Bracken also are found widespread in forests, on moors, and on heath lands, frond heights from 60 to 110 cm are typical on open terrain, while the fronds generally are higher in protecting ecosystems like forests (150 cm or more) (Watt, 1976; Whitehead and Digby, 1997; Nicholson and Paterson, 1976). No investigations have been published regarding the frond density of Common Bracken in Denmark. In the United Kingdom, densities between 16 and 51 fronds m<sup>-2</sup> was measured (Whitehead and Digby, 1997; Nicholson and Paterson, 1976).

Several frond biomass production functions have been established over the years (Paterson *et al.* 1997; Alonso-Amelot *et al.* 2000). Recently Alonso-Amelot *et al.* (2000) presented a simple exponential model for the Neotropical Bracken variety *caudatum* based on rachis length. Models like this and direct measurements have been used to estimate the annual frond dry matter production. The dry matter content in Common Bracken fronds reach a maximum at frond maturity in late summer/early autumn before the fern begins retrieval of nutrients to rhizomes (Watt, 1976; Williams and Foley, 1976). Common Bracken has been reported to have an annual frond dry matter production up to 1400 g m<sup>-2</sup>, while Neotropical Brackens (var. *caudatum* and

*arachnoidum*) with a whole year growth cycle have a standing biomass in the fronds between 25 and 1439 g m<sup>-2</sup> (Alonso-Amelot *et al.* 1995; Watt, 1976; Alonso-Amelot *et al.* 2000). A few investigations have been performed regarding the belowground biomass found in the rhizomes. The dry matter content of Common Bracken rhizomes is lowest during late spring and early summer while the fronds are growing. From July to October the Common Brackens in England replenish the carbohydrate-content of the rhizomes before going into dormancy by the end of October (Williams and Foley, 1976). The belowground biomass for Common Bracken can be very large (660 g-dry-matter m<sup>-2</sup>, 8630 g-fresh-matter m<sup>-2</sup>), and it is estimated that the rhizomes makes up around 80% of the total biomass of Bracken populations. Of the rhizomes, the storage rhizomes make up around 60% of the rhizome biomass (Whitehead and Digby, 1997; Baker *et al.* 1997).

The ptaquiloside-content in fronds is usually highest just after the crosiers emerge from the ground. The content typically decreases during the growing season. In Neotropical Brackens less than 5% of the maximum content is found in mature fronds (Alonso-Amelot *et al.* 1995; Alonso-Amelot *et al.* 1992; Saito *et al.* 1989). In Common Bracken contents between 60 and 9800 µg g<sup>-1</sup> were encountered in a world-wide collection held in Sydney (Smith *et al.* 1994).

The ptaquiloside-content in rhizomes lies between 5 and 1200 µg g<sup>-1</sup>. Only a few studies have been performed on rhizomes, and they were performed in the growth season on Common Bracken and Neotropical Bracken. Highest contents were found in Common Bracken rhizomes (Alonso-Amelot *et al.* 1992; Saito *et al.* 1989; Rasmussen *et al.* 2001).

## Materials and Methods

Three Bracken-dominated ecosystems situated all-over Denmark on different kinds of soil materials were chosen for the investigation (Figure 1). Description of the sites can be found in Table 1. Frond-heights (30 fronds per site) and dry matter content of fronds (3 fronds per site) were measured once (Præstø Fed and Salten Langsø) or twice (Mørup Skov) a month from the time crosiers emerged in April 2000 until the fronds reached maturity in September 2000. The dry matter content was measured after drying for 5 days at 110°C. In addition frond-densities were measured at the end of the growing season (3 times per site (1 m<sup>2</sup>). Several fronds, rhizomes, and roots were sampled from each site for ptaquiloside analysis. A distinction was made between frondbearing (short shots) and storage rhizomes (long shots). Roots from both kinds of rhizomes were pooled to obtain a suitable sample size. Plant material were brought to the laboratory, cut into small pieces, and dried at 50°C for 5 days, milled (resulting in diameter less than 2 mm), and finally stored at 4°C until further analysis. Before drying of rhizomes and roots, they were gently washed free from soil. The ptaquiloside-content was measured in duplicate once or twice a month during the period of investigation following the water-extraction method of Agnew and Lauren (1991). Ptaquiloside were measured as ptaquiloside-



**Fig. 1.** Map of Denmark. Study sites shown on map.

equivalents after conversion to pterisin B at 214 nm on a Perkin Elmer Series 10 liquid chromatographer, equipped with a Shimadzu SPD-10A UV Detector, a Perkin Elmer LCI-100 Laboratory Computing Integrator, and a Merck LiChroCART 250-4, Lichrospher 100 RP-8 as analytical column. Water-methanol (45:55 vol-%) was used as eluent with a flow rate of 1.50 ml min<sup>-1</sup>. A 50 µL sample loop was used for all samples. Standards in the range of 0.0112-22.5000 µg ml<sup>-1</sup> ptaquiloside-equivalents were used for quantification, resulting in measurement of both ptaquiloside and *iso*-ptaquiloside (Agnew and Lauren, 1991; Castillo *et al.*

1997). A standard of ptaquiloside was obtained from Professor Ojika, Nagoya University.

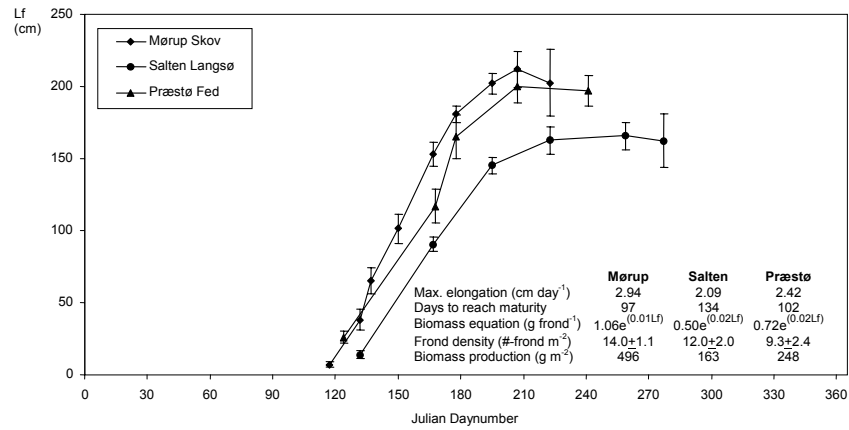
**Table 1.** Site description according to FAO (1990) and Soil Survey Staff (1999).

Site:	Parent material	Soil order	Drainage class	Land use	Herbal dominants
Mørup Skov	Glaciofluvial deposits	Inceptisol	Well drained	Pedunculate Oak ( <i>Quercus robur</i> L.)	B,C,D,E,F G,H
Salten Langsø	Fine sandy till	Inceptisol	Well drained	Douglas Fir ( <i>Pseudotsuga menziesii</i> Franko.)	B,D,I,J,C K
Præstø Fed	Beach ridges	Spodosol	Somewhat excessively drained	Silver Birch ( <i>Betula pendula</i> Roth)	B,I

B=Bracken (*Pteridium aquilinum* (L.) Kuhn), C=Shield fern (*Dryopteris dilatata* (Hoffm.) A.Gray), D=Honeysuckle (*Lonicera periclymenum* L.), E=Wood anemone (*Anemona nemorosa* L.), F=Dutch rush (*Equisetum hyemale* L.), G=Raspberry (*Rubus idaeus* L.), H=Nettle (*Urtica dioica* L.), I=Wavy Hair-grass (*Deschampsia flexuosa* (L.) Trin.), J=Blackberry (*Rubus fruticosus* L.), K= Wood sorrel (*Oxalis acetosella* L.).

## Results and Discussion

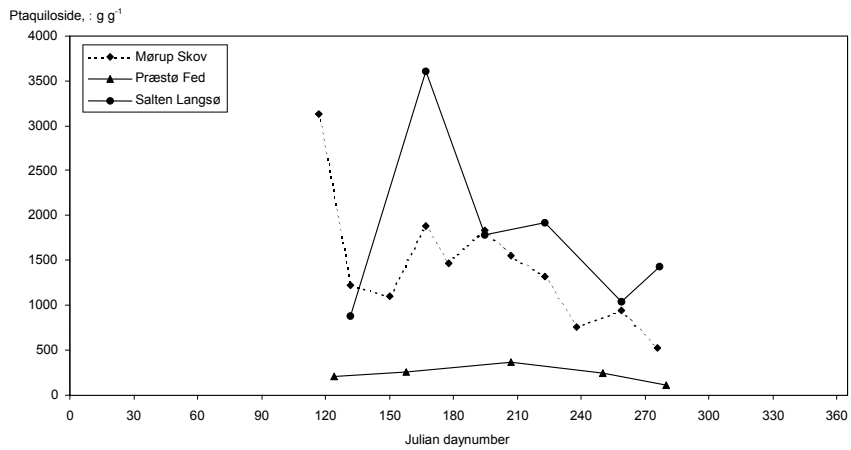
The fronds reached maturity 97-134 days after emergence from the ground (Fig. 2). The maximum frond lengths were 166-212 cm, which is in accordance with previous investigations of Common Bracken in Denmark. The frond densities were 9-14 fronds m<sup>-2</sup>, which is rather low compared to investigations of



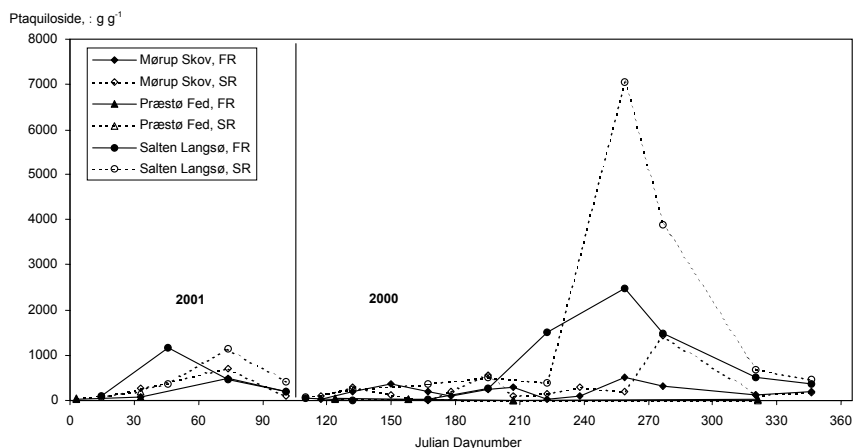
**Fig. 2.** Bracken growth. Lf = Length of frond (cm). The population at Mørup Skov tilted after heavy rain at day number 210.

Common Bracken in the United Kingdom (Whitehead and Digby, 1997; Nicholson and Paterson, 1976). Exponential biomass equations based on frond length and dry matter content were established and yielded correlation coefficients between 0.9209 and 0.9277 (Fig. 2). Using these functions and the frond density, a total aboveground biomass production between 163 and 496 g dry-matter m<sup>-2</sup> were estimated at the end of the growth season, which is somewhat low compared to other investigations (Watt, 1976).

Ptaquiloside were found in all plant-compartments all year round (Fig. 3-4). The fronds had ptaquiloside-contents between 360 and 3612 µg g<sup>-1</sup>. Salten Langsø and Præstø Fed showed maximum ptaquiloside-content in June and July, while Mørup Skov showed maximum in April.



**Fig. 3.** Ptaquiloside-content in Bracken fronds. SDev<10%.



**Fig. 4.** Ptaquiloside in rhizomes. FR=Frondbearing rhizome. SR=Storage Rhizome. SDev<10%.

Both Mørup Skov and Salten Langsø exhibited a tendency for the ptaquiloside-content to decrease during the growth-season in accordance with other findings, while on Præstø Fed no marked difference in the ptaquiloside-content was observed. The different kinds of rhizomes had ptaquiloside-contents between 2 and 7046  $\mu\text{g g}^{-1}$ , with the lowest contents just after the fronds emerged from ground and during the frond growth season (Fig. 4). The largest contents were encountered after fronds had reached maturity as fully enrolled fronds, in late summer and autumn (after day number 210). Raised ptaquiloside-content was also measured in late winter/early spring before the crosiers emerged from the ground. The storage rhizomes had in general a higher content of ptaquiloside than the frondbearing rhizomes, especially in the autumn. The content seem to be rather high compared to other investigations of rhizomes (Alonso-Amelot *et al.* 1992; Saito *et al.* 1989). The rhizome-contents tend to be positively correlated with carbohydrate regeneration of rhizomes in the autumn, rhizome- and belowground shoot-elongation, and negatively correlated with frondgrowth during summer (Williams and Foley, 1976).

The ptaquiloside-content in roots was generally low compared to the other fern-compartments (5-230  $\mu\text{g g}^{-1}$ ). This is the first time ptaquiloside is measured in Bracken roots. The content seems not to be correlated with the ptaquiloside-content in fronds and rhizomes.

## Conclusion

Compared to other investigations, Danish populations of Common Bracken seem to have relatively low growth rates and frond densities resulting in a low biomass production. The content of ptaquiloside in the fronds is in accordance with other observations but do not follow the marked decline after frond



emergence as reported elsewhere. The ptaquiloside contents found in roots and rhizomes are higher than previously reported. The high ptaquiloside-contents found in the rhizomes in the autumn are probably caused by replenishment of the rhizomes with carbohydrates.

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**Sorption, degradation and mobility of ptaquiloside, a  
carcinogenic Bracken (*Pteridium* sp.) constituent, in the  
soil environment**

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**Keywords:** *Natural toxin, cancer, soil contamination, soil sorption,  $K_{ow}$ .*

## Abstract

1  
2  
3 Ptaquiloside (PTA) is a carcinogenic norsesquiterpene glucoside produced by Bracken in  
4 amounts up to at least 500 mg m<sup>-2</sup>. The toxin is transferred from Bracken to the underlying  
5 soil from where it may leach to surface and groundwater's impairing the quality of drinking  
6 water. The objectives of the present study were to characterize the solubility, degradation and  
7 retention of PTA in soils in order to evaluate the risk for groundwater contamination. PTA  
8 was isolated from Bracken. The logarithmic octanol-water and ethyl acetate-water partitioning  
9 coefficients for PTA were -0.63 and -0.88, respectively, in agreement with the high water  
10 solubility of the compound. PTA hydrolysed rapidly in aqueous solution at pH 4 or lower, but  
11 was stable above pH 4. Incubation of PTA with 10 different soils at 25°C showed three  
12 different first-order degradation patterns; i) rapid degradation observed for acid sandy soils  
13 with half life's ranging between 8 and 30 hours decreasing with the soil content of organic  
14 matter, ii) slow degradation in less acid sandy soils with half lives of several days, and iii) fast  
15 initial degradation with a concurrent solid phase-water partitioning reaction observed for non-  
16 acid, mostly clayey soils. The presence of clay silicates appears to retard the degradation of  
17 PTA, possibly through sorption. Degradation at 4°C was generally of type (iii) and  
18 degradation rates were up to 800 times lower than at 25°C. Sorption isotherms for the same  
19 set of soils were almost linear and generally showed very low sorption affinity with  
20 distribution coefficients in the range 0.01 to 0.22 L kg<sup>-1</sup> at a solution concentration of 1 mg L<sup>-1</sup>  
21 except for the most acid soil; Freundlich affinity coefficients increased linearly with clay  
22 and organic matter contents. Negligible sorption was also observed in column studies where  
23 PTA and a non-sorbing tracer showed almost coincident break-through. Based on these  
24 results leaching of PTA to the aqueous environment will be most extensive on sandy soils, pH  
25 > 4 and poor in organic matter exposed to high precipitation rates during cold seasons.

## 1. Introduction

Plants and fungi produce vast amounts of natural substances as part of their chemical interaction with other species. Just like pesticides of human origin, such substances can be quite toxic towards intended or accidental target organisms (Kaufman et al., 1999; Teuscher and Lindequist, 1994). Such a toxic natural compound is the norsesquiterpene glycoside ptaquiloside (PTA, Table 1) encountered in several ferns, but most notably in the Bracken ferns, which are classified as carcinogenic by the International Agency of Research on Cancer (IARC, 1998).

Bracken fern (*Pteridium aquilinum* (L.) Kuhn., Dennstediaceae) is a globally widespread species comprising 2 subspecies and 12 geographic varieties. It is found in a wide range of ecosystems, mainly on acid sandy soils (Mosberg et al., 1994; Page, 1976). PTA cause acute poisoning, blindness and cancer in animals browsing on Bracken (Fenwick, 1988; Smith, 1997; Hirono, 1989). PTA can be transferred to Man when Bracken is utilized as food. In Japan and Brazil occurrence of oesophageal cancer is observed among people eating Bracken (Hirono et al., 1972; Haenszel et al., 1976; Marlière et al., 2000). Correlation between Bracken exposure and cancer has also been demonstrated in rural parts of South America where Bracken is not used for food. Milk from cows eating Bracken are thought to be the vector for the transfer of PTA to humans in such areas, as approx. 9% of the PTA content in fronds eaten by cows are excreted into the milk (Alonso-Amelot et al., 1998; Alonso-Amelot, 1997; Alonso-Amelot et al., 1996; Alonso-Amelot, 1999). Contamination of drinking water with Bracken leachates including PTA has also been proposed. No direct or indirect evidence has although been found for contamination of drinking water (Galpin and Smith, 1986; Galpin et al., 1990). Recently, Danish investigations have demonstrated the occurrence of PTA in

1 soil materials, and the drinking water hypothesis was repostulated (Rasmussen et al., 2003a;  
2 Rasmussen et al., 2003b).

3 PTA is found in all varieties of Bracken, but in rather variable amounts - between 0 and  
4 13,000  $\mu\text{g g}^{-1}$  in the fronds (Smith et al., 1994; Alonso-Amelot et al., 1995; Rasmussen and  
5 Hansen, 2003; Rasmussen et al., 2003a; Rasmussen et al., 2003b). The content is usually  
6 highest at the beginning of the growth season, and is determined by the genetic heritage of the  
7 ferns as well as external growth factors like climate and soil properties (Smith et al., 1994;  
8 Smith et al., 1988; Alonso-Amelot et al., 1992; Rasmussen and Hansen, 2003; Rasmussen et  
9 al., 2003b). The content in the belowground fern rhizomes lies between 5 and 7,050  $\mu\text{g g}^{-1}$ ,  
10 exhibiting peak concentrations in the autumn, while the PTA-content in roots is in the range  
11 5-230  $\mu\text{g g}^{-1}$  (Alonso-Amelot et al., 1992; Rasmussen and Hansen, 2003; Saito et al., 1989).

12 PTA might be transferred to the soil when dead or living Bracken materials are leached by  
13 rain (Ojika et al., 1987; Rasmussen et al., 2003a). PTA-contents in the range of 0.09 to 8.49  
14  $\mu\text{g g}^{-1}$  has been found in Oi/Oe-horizons from Bracken stands in Denmark, while  
15 corresponding Oa and A-horizons had contents between 0.01 and 0.09  $\mu\text{g g}^{-1}$  (Rasmussen et  
16 al., 2003a; Rasmussen et al., 2003b). These levels equal 0.3-160  $\text{mg m}^{-2}$  (Oi/Oe-horizons) and  
17 0.9-57  $\text{mg m}^{-2}$  (Oa or A-horizons). The corresponding content in the fronds were 15-500  $\text{mg}$   
18  $\text{m}^{-2}$  (Rasmussen et al., 2003b). Saito et al. (1989) found that more than 25 % of the initial  
19 PTA-content were still present after 6 weeks in dead fronds exposed to sunlight in open air  
20 protected from rain. However, it is until now an unresolved question whether PTA is stable  
21 enough and has sufficient mobility to reach streams, lakes or aquifers.

22 Another question is whether the potential concentrations in drinking water could be high  
23 enough to cause cancer. Based on carcinogenicity-tests with rats (Smith et al., 1988; Hirono et  
24 al., 1987; Hirono et al., 1984), it is possible to estimate a maximum tolerable concentration of  
25 PTA in drinking water by using the *One-hit model* and a tolerable increase in mortality of 1

1 per million (Danish standards and methodology (Miljøstyrelsen, 1992)). This gives a  
2 maximum tolerable concentration of PTA in drinking water of 0.5-16 ng L<sup>-1</sup>. Rasmussen et al.  
3 (2003a) found PTA-concentrations in artificial frond run-offs of 100,000-730,000 ng L<sup>-1</sup>,  
4 hereby indicating a large potential for dangerous soil and water contamination.

5 The degradation of PTA by hydrolysis (Fig. 1) is a second order reaction in aqueous  
6 systems, with approx. 50% degradation after 7 days at pH 4.0 (37°C), while at pH 5.5 and 7.0  
7 approx. 60 and 90% of the initial concentration were still to be found after 7 days (Saito et al.,  
8 1989). The estimated activation energy of hydrolysis of 65 kJ mol<sup>-1</sup> is high, showing that the  
9 reaction is strongly temperature dependent (Burkhalter et al., 1996). Topsoil's with Bracken-  
10 cover usually exhibits pH-values between 3 and 5 (Nagao et al., 1989; Johnson-Maynard et  
11 al., 1998; Johnson-Maynard et al., 1997; Hetherington and Anderson, 1998; Mackney, 1961;  
12 Rasmussen et al., 2003a; Rasmussen et al., 2003b). Such pH-values should result in a relative  
13 fast degradation of PTA due to hydrolysis. As well as simple hydrolysis, PTA, or more  
14 correctly, the unstable dienone (Fig. 1: [3]) is known to be involved in a series of reactions  
15 due to the electrophile nature of the cyclopropane ring system. The cyclopropane ring has a  
16 high affinity for thiol and sulphide groups in amino acids, and is capable of cleaving DNA by  
17 forming DNA-adducts (Kushida et al., 1994; Ojika et al., 1989; Shahin et al., 1998). An  
18 unresolved question is whether microbial or enzymatic degradation of PTA occurs in soil.

19 The overall aim of this study is to characterize the stability and retention of PTA in soil. The  
20 specific aims comprise: 1) Determination of the octanol-water distribution coefficient for  
21 PTA and its reaction products; 2) Investigation of the stability of PTA in different soils and  
22 fern extracts at near natural temperatures; 3) Determination of the distribution coefficient for  
23 PTA in soil materials, and 4) Determination of the distribution coefficient for PTA in  
24 different soil horizons in a typical Bracken soil as well as the leaching pattern for PTA. As an

- 1 integrated part of the study, methods for PTA analysis in soil samples were developed as well
- 2 as a new fast and simple method for PTA recovery and purification from ferns.

## 2. Materials and methods

Figure 2 provides an overview of the different experiments performed. All studies were performed with non-sterile soil materials hereby mimicking near-natural conditions. The development of the soil sorption and stability studies was performed according to the guidelines in OECD-method 106 (OECD, 2000). Sterilization of soil samples by HgCl<sub>2</sub>-addition was tested, but was found unusable due to changes in the soil sorption properties caused by this treatment. Sterilization with sodium azide was not possible in the experimental set-up due to acid-treatment of the soils involved in PTA-determination (see below), as this would cause liberation of toxic HN<sub>3</sub>-gasses and nucleophiles which could interfere with quantitative conversion of PTA to pterosin B for analysis (see Section 2.1). All laboratory equipment was washed thoroughly with ordinary cleaning detergent, thoroughly rinsed with deionised water and left to dry after being washed with methanol or ethanol (technical grade). Due to its high hydrophilicity, PTA did not bind to laboratory glassware or plastic used in the experiments.

**2.1. PTA Analysis.** The PTA-content in all samples was determined after clean-up of the sample with Polyamide 6 resin (Fluka, Steinheim, Switzerland) (Agnew and Lauren, 1991): 4 mL aqueous extract was passed through 0.50 g resin (dry packed) on a 1.0 x 10 cm I.D. glass Econo-Column<sup>®</sup> (BIO-RAD, New York, USA) fitted with a 28 µm polymer filter. In all soil extracts, the PTA was converted to pterosin B following the method of Agnew and Lauren (1991). Pterosin B was determined at 214 nm at 35°C (Merck Hitachi L-4200 UV-VIS Detector, Merck Hitachi oven) using a Merck Hitachi 655A-40 Auto-Sampler, Merck Hitachi D-6000A Interface and a Merck Hitachi L-6200 Intelligent Pump equipped with a Merck LiChroCART<sup>®</sup> column (125 x 4 mm) packed with LiChrospher<sup>®</sup> 100 RP-8, 4 µm. The



1 measurements were performed with a double deionised-H<sub>2</sub>O:acetonitrile (HPLC grade) eluent  
2 (83:17 V:V) at a flow rate of 1.00 mL min<sup>-1</sup>. 20-100 µl sample injection volumes were used  
3 for analysis resulting in a detection limit of 0.1 µg mL<sup>-1</sup>. Standards of pterosisin B was made  
4 from analytically pure PTA obtained from Professor Ojika (Nagoya University, Japan). PTA  
5 for extraction studies was determined at 220 nm (Shimadzu SPD-6AV UV-VIS  
6 Spectrophotometric Detector) on a Shimadzu LC-6A Liquid Chromatograph (Shimadzu SCL-  
7 6A System Controller) equipped with similar columns as used for pterosisin B. The  
8 measurements were performed at 35°C (Shimadzu CTO-6A Column Oven) with an MillQ-  
9 H<sub>2</sub>O:acetonitrile (HPLC grade) eluent (92:8 V:V) at a flow rate of 1.00 mL min<sup>-1</sup>. A 20 µl  
10 sample injection-volume was used for PTA analysis resulting in a detection limit of 5 µg mL<sup>-1</sup>.  
11 Standards of PTA were made from pure PTA obtained from Professor Ojika (Nagoya  
12 University, Japan). The stability of PTA as well as of pterosisin B standards was evaluated each  
13 day of analysis. PTA standards were generally only stable for a single working day while  
14 standards of pterosisin B were stable for at least 5 days (likely well over that). New working  
15 standards were prepared each week.

16

17 **2.2. Extraction and Purification of PTA.** Semi-pure PTA for the soil sorption,  
18 degradation, and partitioning studies, was obtained by extraction of PTA from freeze-dried  
19 and milled Bracken with MillQ-water (40 g Bracken in 400 mL, shaken for 60 min). 100 mL  
20 fractions of the crude extract were passed through dry packed Polyamide 6 resin (5.00 g in a  
21 1.0 x 20 cm I.D. glass Econo-Column<sup>®</sup>) under gravity. The first 50 mL of eluate was used  
22 directly in the further clean-up, while the last 40 mL eluate was cleaned a second time with  
23 Polyamide 6 (2.00 g in a 1.0 x 20 cm I.D. glass Econo-Column<sup>®</sup>). The combined eluates  
24 (approx. 200 mL) were then added to a XAD2-column and passed through at a rate of approx.  
25 2 drops per second (20 mL Amberlite<sup>®</sup> XAD<sup>®</sup> 2 resin (Supelco, Bellefonte, PA, USA) in a 2.5

1 x 20 cm I.D. glass Econo-Column<sup>®</sup>, wetpacked). The column was washed with 3 x 15 mL of  
2 MillQ-water. All eluates to this point were discarded. The PTA was extracted with 3 x 15 mL  
3 methanol (HPLC grade) into a round-bottomed flask. This extract was evaporated to dryness  
4 under reduced pressure at 45°C using a rotary-evaporator. The residue was dissolved in 4 mL  
5 MillQ-water and the solution added to a Polyamide 6 column (1.00 g Polyamide, 1.0 x 10 cm  
6 I.D. glass Econo-Column<sup>®</sup>, drypacked). The first 0.8 mL eluate was discarded, then a Chem  
7 Elut 1020 column (Varian, USA) was then mounted below the Polyamide column and the rest  
8 of the eluate passed on to the Chem Elut column. The round-bottomed flask was washed with  
9 4 mL of MillQ-water, and was also added to the Polyamide column. The wash was repeated  
10 twice with 2 mL of MillQ-water. Once all the aqueous eluates had adsorbed into the Chem  
11 Elut column, it was flushed with 10 x 20 mL ethyl acetate (HPLC grade), and the eluates  
12 collected in a round-bottomed flask and evaporated until dry at low pressure (rotary-  
13 evaporator, 45°C). The resulting amorphous yellowish gum of raw-PTA was stored in a  
14 desiccator in vials under N<sub>2</sub>-atmosphere at -20°C. The PTA yield through this process was 90-  
15 100%, and the resulting raw-PTA had a purity of 20-80% (Fig. 3). The purity of the raw-PTA  
16 was determined against a standard of pure PTA. The resulting purity depends on the PTA-  
17 content in the ferns used for extraction as well as the content of interfering substances. Only  
18 raw-PTA with a purity of 70-80% was used in subsequent work.

19

20 **2.3. Determination of Octanol-Water Partitioning Coefficient ( $K_{ow}$ ) for PTA.** Analytical  
21 pure PTA was dissolved in MillQ-water resulting in a stock solution with [PTA] = 88  $\mu\text{g mL}^{-1}$ .  
22 The PTA was partitioned between water and n-octanol (analytical grade) by careful shaking  
23 1 part of the PTA stock solution with 6 parts of n-octanol for 5 min in a 10 mL glass  
24 centrifuge tube at 25°C before separation of the phases by centrifugation at 3,600 g (Hansch

1 and Leo, 1995). The PTA-content in the aqueous fraction was measured directly. The  
2 experiment was repeated 8 times.

3

#### 4 **2.4. Determination of Octanol-Water Partitioning Coefficient ( $K_{ow}$ ) for Pterosin B.**

5 Pterosin B was made from analytical pure PTA following the method of Agnew and Lauren  
6 (1991) and dissolved in MillQ-water resulting in a stock solution with [pterosin B] = 31  $\mu\text{g}$   
7  $\text{mL}^{-1}$ . The pterosin B was partitioned between water and n-octanol (analytical grade) by  
8 careful shaking 500 parts of pterosin B stock solution with 1 part n-octanol for 5 min in a 10  
9 mL glass centrifuge tube at 25°C before separation of the phases by centrifugation at 3,600 g  
10 (Hansch and Leo, 1995). The pterosin B-content in the aqueous fraction was measured  
11 directly. The experiment was repeated 10 times.

12

#### 13 **2.5. Determination of Ethyl Acetate-Water Partitioning Coefficient ( $K_{EtOAc/H_2O}$ ) for**

14 **PTA.** A stock solution of analytical pure PTA was made in MillQ-water with [PTA] = 72  $\mu\text{g}$   
15  $\text{mL}^{-1}$ . The PTA was partitioned between water and ethyl acetate (HPLC grade) by careful  
16 shaking 1 part PTA stock solution with 2 parts of ethyl acetate for 5 min in a 10 mL glass  
17 centrifuge tube at 25°C before separation of the phases by centrifugation at 3,600 G (Hansch  
18 and Leo, 1995). The PTA-content in the aqueous fraction was measured directly. The  
19 experiment was repeated 9 times.

20

#### 21 **2.6. Estimation of Octanol-Water Partitioning Coefficient for Ptaquilosin and the**

22 **Unstable Dienone.** The distribution coefficients were estimated by using the method of  
23 structural groups contribution by the ACD log*P*-algorithm (ACD, 1995).

24

1     **2.7. Stability of PTA in Aqueous Solutions.** Raw-PTA was dissolved in double deionised-  
2 water or in 0.01 M CaCl<sub>2</sub> with preset pH in the range 2-7 resulting in [PTA]<sub>START</sub> = 13 µg mL<sup>-1</sup>.  
3 0.001 M acetic acid – acetate buffer systems were used as well as pH adjustment with 0.001  
4 M HCl or 0.001 M NaOH. The stability was investigated during 24 hours at 25°C to enable  
5 comparison with soil sorption studies. PTA-concentrations were measured after Polyamide 6  
6 treatment and conversion to pteroin B.

7  
8     **2.8. Stability of PTA in Aqueous Bracken Extracts.** 16.00 g Bracken (dried and milled)  
9 was extracted with 800 mL MillQ-water for 60 min. The suspension was centrifuged and  
10 filtered (Whatman No. 3). Two aliquots were stored in 500 mL Schott®-bottles at 25°C  
11 ([PTA]<sub>START</sub> = 113 µg mL<sup>-1</sup>), while two similar samples were kept protected from light at 4°C  
12 in a refrigerator ([PTA]<sub>START</sub> = 105 µg mL<sup>-1</sup>). The PTA-contents were measured after clean-  
13 up (Polyamide 6) and conversion to pteroin B daily for the first 5 days, and at day 7, 10 and  
14 25. In addition, pH was measured in the extracts. The extracts were aerated for 5-10 min.  
15 when samples were collected until day 10, whereafter aeration was carried out three times a  
16 week.

17  
18     **2.9. Soil Materials and Extraction of PTA from Soil.** 10 soil types representing typical  
19 Danish natural and cultivated soils with properties similar to soils where Bracken could be  
20 found were chosen for the sorption and degradation studies. The criteria given in OECD  
21 method 106 were used as a guideline to ensure a wide range in soil properties with respect to  
22 content of organic carbon, pH, and texture (Table 2, OECD, 2000).

23     In the sorption and degradation studies, the content of PTA in contact with soil was  
24 determined after extraction of 10 g soil with 10 mL 0.01 M CaCl<sub>2</sub> for 60 min (flatbed shaker,  
25 approx. 40 rpm.). The soil was then separated from the solution by centrifuging for 5 min at

1 11,200 G and 4 mL of the clear extract added to a Polyamide 6 column (0.50 g Polyamide, 1.0  
2 x 10 cm I.D. glass Econo-Column<sup>®</sup>, drypacked). For each sample, the extraction was repeated  
3 4 times with 4 mL 0.01 M CaCl<sub>2</sub> added to the soil and shaking by hand for 5 min before  
4 centrifuging. The PTA-content was determined in the different fractions after conversion to  
5 pterosin B. Extracts estimated of having PTA-content well above the detection limit were  
6 pooled before measurement. No further PTA could be extracted after 5 extractions. The  
7 recovery for spiked samples was generally below 100% since some PTA reacts irreversibly  
8 with the soil materials (see: 3. *Results and Discussion*).

9  
10 **2.10. Stability of PTA in Contact with Soil Materials.** 10.00 g of dry soil was equilibrated  
11 with 9 mL 0.01 M CaCl<sub>2</sub> over night. 1 mL raw-PTA dissolved in 0.01 M CaCl<sub>2</sub> was then  
12 added to the suspension resulting in a total of 10 mL aqueous solution in contact with the soil  
13 material. The [PTA]<sub>START</sub> in the aqueous phase ranged between 22 and 30 µg mL<sup>-1</sup>. The  
14 suspensions were gently shaken on a flatbed shaker (approx. 40 rpm) protected from light in a  
15 black box kept at 25°C for 1 to 21 days. The samples were aerated 3 times a week. PTA was  
16 extracted from the soil according to the method described above. All analyses were made in  
17 duplicate. The experiment was repeated at 4°C for 65 days without shaking with 4 soils using  
18 a soil:solution ratio of 10:3 (W:V), [PTA]<sub>START</sub> in the aqueous phase was 94 µg mL<sup>-1</sup>.

19  
20 **2.11. Sorption of PTA to Soil Materials.** For each soil material 10.00 g of dry soil was  
21 equilibrated with 9 mL 0.01 M CaCl<sub>2</sub> over night. 0-1 mL raw-PTA dissolved in 0.01 M CaCl<sub>2</sub>  
22 and 1-0 mL 0.01 M CaCl<sub>2</sub> was then added to the suspension resulting in a total of 11 samples  
23 containing 10 mL aqueous solution with [PTA]<sub>START</sub> ranging between 0 and 14 µg mL<sup>-1</sup>. The  
24 suspensions were shaken on an orbital shaker (30-40 rpm) protected from light at 25°C for 24  
25 hours (wrapped in aluminium foil). The aqueous phase was then separated from the soil solids

1 by centrifugation for 5 min at 11,200 G, and the PTA-content analysed after conversion to  
2 pterisin B and clean-up on Polyamide 6.

3

4 **2.12. Soil Column Studies.** Soil columns (30 x 5 cm I.D.) with densities of 1.1-1.6 g cm<sup>-3</sup>  
5 were prepared in acid proof stainless steel columns according to Danish standard (Kjølholt,  
6 1998). The columns were equilibrated with 0.01 M CaCl<sub>2</sub> at approx. 65 mm hour<sup>-1</sup>. The  
7 influent was added with a peristaltic pump (Alitea C-6Y, Ventur Alitea AB, Sweden). Raw-  
8 PTA was dissolved in 0.01 M CaCl<sub>2</sub>, and 1 mL of the PTA-solution was then mixed with 4  
9 mL of 0.01 M KBr (used as inert tracer). Initial PTA-concentrations were in the range 177-  
10 680 µg mL<sup>-1</sup>. The PTA/bromide solution was added to the column by pipette and the  
11 breakthrough curve was obtained by analysing 23 mL fractions each equalling 11.7 mm of  
12 precipitation/influent added. PTA in the fractions was measured after conversion to pterisin  
13 B, while bromide was measured directly with a Br-sensitive crystal-membrane electrode  
14 (Metrohm, Denmark). The pH of the eluates was measured after 35, 130, 220, and 310 mm of  
15 influent had passed the column.

16

17 **2.13. Soil Chemical and Physical Analysis.** Soil pH was measured in a 1:2.5 or 1:10  
18 soil:0.01 M CaCl<sub>2</sub> suspension in mineral soil or organic soil samples, respectively. Organic  
19 carbon was determined by dry combustion using an Eltra CS 500 Carbon Sulfur Determinator  
20 (Nelson and Sommers, 1982). The content of calcite was measured gas-volumetrically  
21 (Collins, 1906). The soil samples were fractionated into the following particle-size classes  
22 after treatment with 0.1 M sodium pyrophosphate: Sand (20-2,000 µm), silt (2-20 µm), and  
23 clay (<2 µm). All soil chemical and physical analysis except from soil textures were  
24 performed as dup- or triplicates.

25

1     **2.14. Statistical and computational methods.** Statistical analyses were performed in the  
2     *Unscrambler ver. 7.6 SR-1* software package (Camo ASA software, Oslo, Norway) and in  
3     *Microsoft® Excel 2000* (Microsoft Corporation, USA). Multiple Linear Regression (MLR)  
4     analysis was conducted on soil properties (pH, texture, and carbon-content) and soil sorption  
5     data obtained for PTA in the soil sorption and degradation studies to unveil correlations  
6     between soil properties and PTA-behaviour in soil. The MLR-models were validated and  
7     evaluated by full cross validation and one-way analysis of variance. Distribution coefficients  
8     for PTA in soil materials were determined by fitting data from soil sorption studies with  
9     Freundlich-isotherms in the software package *TableCurve™ 2D* (Jandel Scientific, AISN  
10    Software, USA).

### 3. Results and discussion

**3.1. Partitioning Coefficients in Non-soil Systems.** The logarithm of the partitioning coefficient for PTA between water and ethyl acetate was -0.88 (Table 3), which is in accordance with previous findings (Ojika et al., 1987). The experimental logarithmic partitioning coefficients between octanol and water for PTA, pterosin B, and the estimated coefficients for ptaquilosin and the dienone intermediate were -0.63, 3.33, 1.03 and 1.94 respectively (Fig. 1, Table 3). The low value for PTA indicate a low sorption and hence a high mobility of the compound in the soil environment due to weak sorption to soil organic matter, all other things being equal (Schwarzenbach et al., 1993). However, formation of covalent bonds by reaction of the cyclopropane moiety of ptaquiloside as well as of ptaquilosin and the unstable dienone with functional groups of humic substances or soil minerals may reduce mobility. Due to the high reactivity of the cyclopropane moiety of ptaquilosin and the unstable dienone, formation of covalent bonds is likely the main reaction of these compounds.

**3.2. Stability of PTA in Aqueous Solutions.** PTA is stable in the range pH 4-7 for at least 24 hours, but hydrolyses rapidly at pH-values below 4 (Fig. 4), which is in accordance with other findings (Burkhalter et al., 1996; Saito et al., 1989).

**3.3. Stability of PTA in Bracken Extracts.** The time course of PTA degradation and the resulting variation in pH of aqueous Bracken extracts stored at 4°C and 25°C are shown in Figure 5. The incubation temperature was the only parameter which was varied between the two experiments. Initial pH is 6.1 in all extracts for the first 24 hours. Hereafter pH drops to 5 followed by an increase to 7.5 in the extracts stored at 25°C. The pH-variation is likely due to fermentation, as production of different carboxylic acids will lower the pH followed by an



1 increase in pH resulting from the liberation of proteins from the decaying Bracken  
2 compounds. Extracts stored at 4°C do not exhibit the same variations in pH. At ambient  
3 temperature (25°C), PTA seems to be stable for 192 hours after which degradation of PTA  
4 starts. Thereafter, the concentration of PTA decreases with approx. 5.25 % per day. This  
5 pattern is likely due to growth of bacteria capable of utilizing PTA as energy source,  
6 increased concentration of  $\beta$ -glucosidase-enzyme in the suspension combined with increased  
7 hydrolysis of PTA when pH increases after 192 hours (Saito et al. 1989). For the experiment  
8 run at 4°C, the ptaquiloside degradation is much slower with no degradation taking place  
9 during the first 260 hours. These investigations clearly demonstrate that PTA can exist outside  
10 the fern in aqueous extracts for prolonged periods of time, especially at low temperatures.

11

12 **3.4. Stability of PTA in Contact with Soil Materials.** The stability of PTA in contact with  
13 different soil materials at 4 and 25°C is plotted against time in Figure 6. At 25°C (Fig. 6A),  
14 three degradation patterns can be distinguished:

- 15 1) Fast degradation (full degradation within 240 hours, half life between 8 and 30  
16 hours, most degraded within 140 hours). The reaction can be described as a first  
17 order degradation reaction:  $PTA_{\text{SOLUTION}} \xrightarrow{k_1} PTA_{\text{DEGRADED}}$ . The rate constants  
18 ( $k_1$ ) are listed in Table 4. This group (group 1) comprises: Jyndevad Farm A;  
19 Åstrup Forest A; Nørlund Plantation E, and Præstø Bracken Stand Oa.
- 20 2) Fast initial degradation (degradation of 40-60% within 24 hours) followed by slow  
21 or almost no degradation. The overall reaction can be fitted as the sum of a first  
22 order reaction degradation reaction (rate constant  $k_{21}$ , Table 4) and an equilibrium  
23 step characterized by the forward and reverse constants  $k_{22}$  and  $k_{23}$  (Fig. 7, Table  
24 4). This group (group 2) comprises: Tåstrup Organic Farm A; Christianssæde Ap;  
25 Flakkebjerg Farm Btg2, and Flakkebjerg Farm 2Cg1.

1           3) Slow degradation throughout the whole timespan (half life between 150 and 180  
2           hours). The reaction can be described as degradation pattern 1, and first order rate  
3           constants are found in Table 4. This group (group 3) consists of Nørlund  
4           Plantation Bhs and Nørlund Plantation C.

5           In this context, *degraded* is defined as the amount of PTA, which cannot be extracted by  
6           water, comprising irreversible bound PTA, hydrolysed PTA, and PTA otherwise degraded.

7           The soils from group 1 are characterized by having low clay contents (0-8%), low to high  
8           carbon-contents (1-40%), and low pH (pH < 4 - with Jynde vad Farm A as an notable  
9           exception). The soils making up group 2 are slightly acid to neutral (pH 4.8-7.9), with low  
10          carbon-contents (0.06-1.7%), and higher clay contents (14-23%). The two soils in group 3 are  
11          slightly acid (pH 4.0-4.6), having low carbon-contents (0.17-0.78%), and low clay contents  
12          (1-2%). It is interesting to note that PTA in several of the investigated soils is still present in  
13          the soil suspensions after incubation for more than 500 hours at 25°C. The results are in  
14          accordance with the recent detection of PTA in the soil environment, but in contrast with the  
15          general perception of PTA as being highly unstable (Castillo et al., 1998; Rasmussen et al.,  
16          2003a; Rasmussen et al., 2003b).

17          The degradation pattern observed in group 1 is likely due to a combination of hydrolysis  
18          (Burkhalter et al., 1996; Saito et al., 1989), irreversible reactions between PTA and humic  
19          substances/clay and possibly microbial degradation in the case of the non-acid Jynde vad Farm  
20          A soil. The hypothesis was supported by MLR performed on the reaction constant and soil  
21          pH, carbon and clay content showing strong correlation between the first order rate constant  
22          and the carbon content (Eq. 1; 95% confidence limits shown,  $r^2$  (calibration) = 0.87,  $q^2$  (cross  
23          validation) = 0.80)).

$$k_1 = 0.0150 \pm 0.0058 + 0.0019 \pm 0.0004 \cdot C(\%) \quad \text{Eq. 1}$$

1        However, it should be emphasized that the distribution of data with respect to content of  
2 organic matter was somewhat skewed due to Præstø Bracken Stand Oa being an organic soil  
3 horizon, in contrast to the other horizons being mineral soils (Table 2).

4        The pattern for group 2 is likely due to fast irreversible sorption of PTA to organic matter  
5 and clay minerals combined with hydrolysis followed by a period where PTA is somewhat  
6 protected from degradation by reversible sorption to clay minerals (Fig. 7). The protection of  
7 PTA by sorption to clay minerals might be due to weak interlayer sorption in clay minerals, a  
8 phenomena previously observed for other glycosides (Greenland, 1956a; Greenland, 1956b).  
9 The same sorption reaction has been reproduced by using pure smectites, and is affected  
10 negatively by application of surfactants (phthalates) to the soil solution (Authors, unpublished  
11 results). One exception from the general group 2 pattern is Taastrup Organic Farm A, where a  
12 sudden degradation of PTA takes place from 140 to 240 hours, which could be due to  
13 microbial degradation. The degradation-pattern observed for group 3 is similar to group 1, but  
14 with degradation reactions running at a lower rate. No correlations were found between the  
15 reaction constants for group 2 and soil pH, carbon- or clay-content.

16        The degradation of PTA was repeated at 4°C without shaking for 1,728 hours (Fig. 6B). The  
17 overall pattern found at 25°C was observed at 4°C too, but with three exceptions:

- 18            1) Fast group 1-like degradation did not take place (see Åstrup Forest A).
- 19            2) Slow group 3-like degradation ceased after the first days (see Nørlund Plantation  
20            C).
- 21            3) All degradation patterns could be modelled as the sum of a first order degradation  
22            reaction and an equilibrium step governed by the constants  $k_{22}$  and  $k_{23}$  (Table 4),  
23            hereby indicating temperature influence on the degradation pattern and speed of  
24            hydrolysis.

1 At Nørlund Plantation C, and Flakkebjerg Farm Btg2, the degradation ceased after 24-144  
2 hours, while at Åstrup Forest A, and Taastrup Organic Farm A, it continued, but at a lower  
3 rate than in the first few days. A sudden increase in degradation took place at the Flakkebjerg  
4 Farm Btg2-soil after 350 hours. These degradation patterns clearly demonstrate the ability for  
5 PTA to persist in contact with soil for prolonged time. Since the experimental conditions for  
6 the degradation experiment performed at 4°C are more similar to natural conditions in  
7 temperate climates than the experiments performed at 25°C, this study indicates that PTA can  
8 exist in soil environments for prolonged periods, and supports the findings of Rasmussen et  
9 al. (2003a,b).

10

11 **3.5. Sorption of PTA to Soil Materials.** Freundlich-modelling of the sorption data were  
12 chosen due to the lack of obvious adsorption maximum in the isotherms:

$$C_{\text{SOIL}} = K_{\text{FREUNDLICH}} \cdot C_{\text{WATER}}^n \quad \text{Eq. 2}$$

13 where  $C_{\text{SOIL}}$  = Content of PTA sorbed on soil (: g kg<sup>-1</sup>),  $K_{\text{FREUNDLICH}}$  = Freundlich  
14 coefficient,  $C_{\text{WATER}}$  = Concentration of PTA in water (: g L<sup>-1</sup>), and  $n$  = measure of non-  
15 linearity. The amount of PTA sorbed to the soils ( $C_{\text{SOIL}}$ ) was calculated as the difference of  
16 the amount added and the amount remaining in the soil solution after 24 hours ( $C_{\text{WATER}}$ ). Due  
17 to degradation of PTA caused by hydrolysis in some of the soils (mainly group 1) and/or as a  
18 result of irreversible sorption (mainly group 2)  $C_{\text{SOIL}}$  is the sum of sorbed PTA and PTA  
19 degraded during 24 hours. It was attempted to extract the actual amount of PTA sorbed to the  
20 soil, but due to extremely low sorption to some soils, this was not experimentally possible  
21 (Authors, unpublished results). Three soils had pH less than 4.0: Præstø Bracken stand Oa  
22 (pH 2.7); Åstrup Forest A (pH 3.3); and Nørlund Plantation E (pH 3.3). According to the rates  
23 of hydrolysis shown in Figure 4, this would result in degradation of approx. 55, 30, and 30%  
24 of the added PTA within 24 hours when in pure aqueous solution. In the less acid soils, the

1 amount of PTA which had degraded due to hydrolysis during the 24 hours of sorption is likely  
2 close to zero. The results of the soil sorption studies are plotted in Figure 8. Two types of  
3 isotherms exist:

- 4 1) A linear type with  $n = 1$  (Præstø Bracken Stand Oa; Nørlund Plantation E and  
5 Bhs).
- 6 2) A convex type with  $n < 1$  (all other soils).

7 The sorption parameters and the regression coefficients of the sorption isotherms are listed  
8 in Table 5. The Freundlich-constants ( $K_{\text{FREUNDLICH}}$ ) were found to be in the range 0.0070 to  
9 0.2165 with a single extreme constant of 1.2311 (Præstø Bracken Stand, Oa), clearly  
10 demonstrating very low sorption of PTA to soil materials. The  $n$ -values were between 0.65  
11 and 1.00. The sites with  $n = 1.00$  did actually have  $n$ -values slightly above 1 (1.04 – 1.09), but  
12 could be fitted with linear-isotherms as indicated by the  $r^2$ -values (Table 5). MLR was  
13 performed using the soil characteristics from Table 2 to estimate any correlation to  
14  $K_{\text{FREUNDLICH}}$  and  $n$  (Fig. 9). Præstø Bracken Stand Oa and Åstrup Forest A, were removed  
15 from the  $K_{\text{FREUNDLICH}}$  analysis due to high leverage on the model, while Christianssæde Forest  
16 A in addition was removed from the  $n$  analysis. The leverage of Præstø Bracken Stand Oa and  
17 Åstrup Forest A might be due to a large proportion of degraded PTA caused by hydrolysis.  
18  $K_{\text{FREUNDLICH}}$  was positively correlated with the carbon-content and the clay-content (Eq. 3;  
19 95%-confidence limits shown,  $r^2$  (calibration) = 0.75,  $q^2$  (cross validation) = 0.42).

$$K_{\text{FREUNDLICH}} = 0.0819 \pm 0.0264 \cdot C(\%) + 0.0056 \pm 0.0021 \cdot \text{clay}(\%) \quad \text{Eq. 3}$$

20 The positive regression coefficients clearly indicate that *sorption* of PTA is correlated with  
21 the clay- and carbon-contents of the soils.

22 The non-linearity parameter  $n$  was correlated with the carbon-content and pH (Eq. 4, 95%-  
23 confidence limits shown,  $r^2$  (calibration) = 0.83,  $q^2$  (cross validation) = 0.57).

$$n = 1.074 \pm 0.108 - 0.0567 \pm 0.0181 \cdot \text{pH} + 0.162 \pm 0.046 \cdot C(\%) \quad \text{Eq. 4}$$

1 The positive correlation with the carbon-content is likely a result of PTA-sorption to organic  
2 matter, while the negative correlation with pH could indicate that the sorption capacity of the  
3 soil is pH-dependent, i.e. high pH results in a convex adsorption isotherm likely as a result of  
4 surface charge changes of the organic matter.

5 The calibration correlations obtained for  $K_{\text{FREUNDLICH}}$  and  $n$  are rather good when looking at  
6 the calibration correlation coefficients only. It must although be emphasized that samples  
7 were removed from the analysis, and the cross-validated correlation coefficients are rather  
8 low, indicating high leverage of individual soils.

9

10 **3.6. Soil Column Studies.** Breakthrough curves for PTA and bromide are presented in  
11 Figure 10. The pH-values measured in the eluates from Nørlund Plantation E and C differed  
12 slightly from the reported soil pH values in Table 2, as they were 0.6 and 0.5 units lower  
13 respectively. No decreasing or increasing trend in pH development was observed in eluates  
14 during the experiment. All added PTA was recovered in the eluates from Nørlund Plantation  
15 soils, while 80% was recovered from Åstrup Forest A. Satisfactory modelling of all the  
16 resulting breakthrough curves for PTA and bromide could not be performed by the use of  
17 convection-dispersion equation due to tailing likely caused by preferential flow (Hansen,  
18 1997). Instead, curves were fitted as Pearson peaks allowing more tailing (Fig. 10).  
19 Calculation of distribution coefficients ( $K_D$ ) based on the retardation of PTA compared to  
20 bromide at maximum breakthrough was only possible for Åstrup Forest A, since PTA moved  
21 slightly faster than bromide in Nørlund Plantation Bhs and Nørlund Plantation C, and no  
22 significant difference between PTA and bromide was observed in Nørlund Plantation E  
23 (Table 6). The retardation factor ranged from 0.96 to 1.12. The retardation factors less than 1  
24 is likely due to experimental conditions resulting in insufficient accuracy of the breakthrough  
25 curve, but indicates retardation very close to 1 resulting in PTA distribution coefficients close

1 to 0. The corresponding distribution coefficient therefore ranges from 0 to 0.05 hence  
2 showing PTA mobility comparable to that of bromide and demonstrating that PTA does not  
3 sorb significantly to soil materials, which is in agreement with the results of the sorption  
4 studies, where the  $K_{\text{FREUNDLICH}}$ -coefficients for ranged from 0.0089 to 0.0661. Better  
5 resolution of the peaks is likely obtained if longer soil columns are used.

6

7 **3.7. General Discussion.** This investigation shows that PTA is quite mobile in many soil  
8 types and may persist for long time when present in sandy, non-acid and cold subsoils - such  
9 as many aquifers - or in clayey soil horizons. Based on the results reported above the  
10 following statements can be made:

- 11 1) PTA degrades rapidly due to hydrolysis when pH is below 4 (i.e. soils in group 1 and 3,  
12 Fig. 6) or higher than 8 (Burkhalter et al., 1996; Saito et al., 1989). Consequently, half-  
13 lives of PTA in acid sandy soils are short, in the range of 8 - 30 hours at 25 °C (Fig. 6,  
14 Table 4). Half-lives in calcareous soil materials have not been quantified, but probably  
15 are quite short due to  $\text{pH} > 8$ .
- 16 2) PTA degrades more slowly in slightly acid to neutral soils where observed half-lives are  
17 usually in the range of days. For clayey soils, the degradation rate at longer reaction  
18 times is further reduced probably due to a concurrent solid-phase partitioning reaction  
19 (i.e. group 2 soils, Fig. 6, Table 4). The results indicate, that with longer reaction times,  
20 clay silicates may sorb PTA in the interlayer space reducing the amount of PTA that is  
21 available for degradation in solution.
- 22 3) PTA shows low affinity for sorption to organic matter in agreement with its low  
23 octanol-water partitioning coefficient (Fig. 8, Table 3). For group 1 soils degradation  
24 rates increases with increasing soil organic matter contents which may be due to

1 irreversible bonding between the alkylating group of PTA and organic matter, or due to  
2 increasing microbial activity with increasing content of organic matter (Eq. 1).

- 3 4) PTA degradation is strongly temperature-dependent; at 4 °C overall degradation is much  
4 reduced compared with 25 °C, either due to a decrease in the rate of the first-order  
5 degradation reaction, strong retardation due to the solid phase-partitioning process, or  
6 both (i.e. Fig. 6A vs. 6B).

7 These sorption and degradation properties of PTA might help explain the high occurrences  
8 of cancer in some Bracken dominated areas, as PTA can leach from Bracken stands and enter  
9 aquifers or other recipients such as streams and lakes. The apparent lack of correlation  
10 between Bracken cover and human cancer on a small scale in some investigations might be  
11 partly due to variations in soil properties and/or microbial activity as mirrored in the  
12 degradation patterns observed for PTA. It is although still an open question whether PTA-  
13 contaminated milk, water or Bracken spores form the links between humans and Bracken. In  
14 areas where milk-consumption is dependent on cows browsing on Bracken, then milk is likely  
15 the main vector. In areas with only few cows browsing on Bracken or where people obtains  
16 milk from areas without Bracken-browsing, then water-contamination might be important as  
17 well. This may be the case in areas with many surface wells or shallow aquifers situated  
18 below slightly acid clay soils. However, the PTA content in drinking waters still needs to be  
19 measured. A preliminary study carried out by the authors did show PTA-contents between 0  
20 and 7  $\mu\text{g L}^{-1}$  in 15 samples of soil solutions sampled at 90 cm depth in a Bracken stand, while  
21 two water samples from surface wells encircled by Bracken yielded PTA-contents of 30 and  
22 45  $\mu\text{g L}^{-1}$ . The transfer of PTA from the fern to the soil environment is not yet fully  
23 understood; especially leaching from living and dead Brackens needs further examination.  
24 PTA can be leached from living fronds by rain (Rasmussen et al., 2003a). This, combined  
25 with the fact that PTA can be quite mobile in the soil, indicates that PTA may occur in the soil



1 water in very variable concentrations correlated with the precipitation pattern. Leaching  
2 studies must therefore be very detailed with respect to sampling of soil water. Currently,  
3 further studies are being carried out to unveil the incidence of PTA in drinking water.

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#### 4. Conclusion

PTA is capable of existing in the soil environment for prolonged time as a highly mobile compound. The octanol-water partitioning coefficient for PTA was extremely low ( $\log K_{ow} = -0.63$ ), while the *log* values of the coefficients for ptaquilosin, the dienone and pterosin B were between 1.03 and 3.33. The stability of PTA in contact with soil or fern extracts was highly temperature dependent as ptaquiloside degraded faster at high temperatures. The observed patterns for degradation of PTA in contact with soil could be grouped into three patterns: 1) Fast degradation (half life between 8 and 30 hours, all degraded within 240 hours); 2) Fast initial degradation (40-60% degraded within 24 hours) followed by poor or almost no degradation; and 3) Slow degradation (half life between 150 and 180 hours). In several soils, PTA was still present after 72 days. The degradation patterns observed was due to differences in acidity, clay content, carbon-content, and probably microbial activity. Sorption isotherms could be fitted by the Freundlich equation with  $K_{FREUNDLICH}$ -constants between 0.01 and 0.22 except from one soil with a constant of 1.23. *n* values were between 0.64 and 1.00. The shape of the isotherms was correlated with pH, carbon- and clay-content. Soil column studies performed in a sandy media resulted in breakthrough curves very similar to the inert tracer bromide, hence demonstrating high mobility of PTA in all soils examined.

#### 5. Safety

Use respiration protection-gear while working with powdered Bracken.

#### Acknowledgement

The authors are grateful to Professor Ojika (Nagoya University, Japan), for the gift of PTA, which made this study possible.

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## FIGURE CAPTIONS

<b>Table no:</b>	<b>Text:</b>
<b>1</b>	Table 1. Properties of PTA.
<b>2</b>	Table 2. Properties of soils used in the investigation.
<b>3</b>	Table 3. Distribution coefficients for PTA and derivatives in n-octanol-water and ethyl acetate-water systems (95% confidence limits shown). The coefficients for ptaquilosin and the dienone are estimated by the <i>ACDlogP</i> -algorithm.
<b>4</b>	Table 4. First order rate constants for PTA degradation in the soil-water systems (see Figures 6 and 7).
<b>5</b>	Table 5. Fitted Freundlich parameters for sorption of ptaquiloside to the soils investigated.
<b>6</b>	Table 6. Retardation factors and distribution coefficients for the soil column studies.
<b>Figure no:</b>	<b>Text:</b>
<b>1</b>	Fig. 1. PTA [1] transformation reactions. During alkaline conditions, first ptaquilosin [2] and then the unstable dienone [3] are formed. Both [1] and [3] yields pterosin B [4] under acid conditions. The unstable dienone may alkylate DNA [5] and is likely to react with numerous other substances including humic substances.
<b>2</b>	Fig. 2. Overview of main analyses.
<b>3</b>	Fig. 3. HPLC chromatograms of raw-ptaquiloside. A) Raw-ptaquiloside (purity: 80%). Substances causing peaks eluting later than 15 min was removed from the raw-ptaquiloside by Polyamide treatment before use; B)

	The same sample as A, but after conversion to pterosin B (late eluting peaks not removed).
<b>4</b>	Fig. 4. PTA stability in aqueous solutions at 25°C after shaking for 24 hours: ○) pH adjusted with HCl or NaOH; ◇) acetic acid-acetate buffer. $[PTA]_{START} = 13 \mu\text{g mL}^{-1}$ . Dotted line represents the range where PTA is unstable (pH < 4), while the solid line is the stable range (pH 4-7).
<b>5</b>	Fig. 5. PTA stability in aqueous Bracken extracts (25 days): A) PTA stability; B) Variation of pH; ○) At 4°C ( $[PTA]_{START} = 105 \mu\text{g mL}^{-1}$ ); □) At 25°C ( $[PTA]_{START} = 113 \mu\text{g mL}^{-1}$ ).
<b>6</b>	Fig. 6. PTA stability in soil suspensions: A) At 25°C, $[PTA]_{START}$ ranges between 22 and 30 $\mu\text{g mL}^{-1}$ , thick lines represent first order degradation reactions, dotted lines represent the sum of a first order degradation reactions and a concurrent solid phase-water equilibrium; B) At 4°C, $[PTA]_{START} = 94 \mu\text{g mL}^{-1}$ .
<b>7</b>	Fig. 7. Proposed PTA degradation reaction scheme in clayey soils comprising a first order degradation reaction (rate constant $k_{21}$ ) and a concurrent solid phase-water partitioning equilibrium characterized by the forward ( $k_{22}$ ) and reverse ( $k_{23}$ ) rate constants.
<b>8</b>	Fig. 8. Sorption isotherms for PTA bonding to soil materials (0.01 M CaCl <sub>2</sub> , 24 hours, $[PTA]_{START}$ in the range 0 to 14 $\mu\text{g mL}^{-1}$ ). Solid lines represent Freundlich-fits. Fitting parameters are listed in Table 5. Only the first 5 data points are shown for Præstø Bracken Stand, Oa.
<b>9</b>	Fig. 9. Multiple linear regressions of sorption parameters in Freundlich models versus soil properties: A) $K_{FREUNDLICH}$ ; B) n. Two soils removed from both regressions: Præstø Bracken Stand Oa and Åstrup Forest A.

	Christiansæde Forest A removed from B. ○ = Calibration. ● = Validation .
<b>10</b>	<p>Fig. 10. Breakthrough curves for ptaquiloside and bromide in packed soil columns composed of soil materials from Danish Spodosols: A) Nørlund Plantation, E-curves fitted with the convection-dispersion equation (dispersion coefficients: <math>2.1 \text{ cm}^2 \text{ hour}^{-1}</math> (bromide) and <math>17.0 \text{ cm}^2 \text{ hour}^{-1}</math> (ptaquiloside)); B) Curves fitted with Pearson curves, <math>R^2</math> (calibration curves, bromide/ptaquiloside): Åstrup Forest, A = 0.8300/0.9283; Nørlund Plantation, E = 0.9934/0.9582; Nørlund Plantation, Bhs = 0.9982/0.9995; Nørlund Plantation, C = 0.9841/0.9868. 80% ptaquiloside recovery from Åstrup Forest, A; 100% ptaquiloside recovery from Nørlund Plantation, E, Bhs, and C.</p>

Table 1  
Properties of PTA

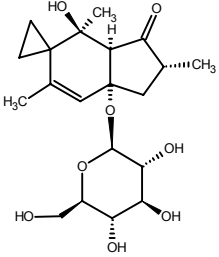
CAS numbers	87625-62-5 88825-03-0
Chemical name	[2'R-(2'α,3'α,4'β,7'α)]-7'a-(β-D-glucopyranosyloxy)-1',3',a,4',7'a-tetrahydro-4'-hydroxy-2',4',6'-trimethyl-spiro[cyclopropane-1-5'-[5H]inden]-3'(2'H)-one
Chemical structure	
Molecular formula	C <sub>20</sub> H <sub>30</sub> O <sub>8</sub>
Molecular weight	398.45 g mol <sup>-1</sup>

Table 2  
Properties of soils used in the investigation

Site and soil horizon	pH (0.01 M CaCl <sub>2</sub> )	Organic C (%)	Clay content (%)
Præstø Bracken stand, Oa	2.7	38.3	ND <sup>¶</sup>
Jyndevad Farm, A	6.1	1.8	4
Åstrup Forest, A	3.3	5.2	8
Tåstrup Organic Farm, A	7.2	1.3	14
Christanssæde Forest, Ap	4.8	1.7	17
Nørlund Plantation, E	3.3	0.29	1
Flakkebjerg Farm, Btg2	6.6	0.1	23
Nørlund Plantation, Bhs	4.0	0.78	1
Nørlund Plantation, C	4.6	0.17	2
Flakkebjerg Farm, 2Cg1	7.9	0.06	19

<sup>¶</sup> Not determined, but probably close to 0.

Table 3  
 Distribution coefficients for PTA  
 and derivatives in n-octanol-water  
 and ethyl acetate-water systems  
 (95% confidence limits shown). The  
 coefficients for ptaquilosin and the  
 dienone are estimated by the  
 ACDlogP-algorithm

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<u>PTA</u>	
$\log K_{\text{EtOAc/H}_2\text{O}}$	-0.88±0.06
$\log K_{\text{ow}}$	-0.63±0.01
<u>Pterisin B</u>	
$\log K_{\text{ow}}$	3.33±0.01
<u>Ptaquilosin</u>	
$\log K_{\text{ow}}$	1.03±0.48
<u>Dienone</u>	
$\log K_{\text{ow}}$	1.94±0.41

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Table 4  
First order rate constants for PTA degradation in the soil-water systems (see Figures 6 and 7)

Temperature	Site and soil horizon	$k_1^{\dagger}$	$r_1^2$	$k_{21}$	$k_{22}$	$k_{23}$	$r_2^2$
		hours <sup>-1</sup>		hours <sup>-1</sup>	hours <sup>-1</sup>	hours <sup>-1</sup>	
25°C	Præstø Bracken stand, Oa	0.0864	0.9998	¥			
	Jyndeved Farm, A	0.0239	0.9994				
	Åstrup Forest, A	0.0246	0.9959				
	Tåstrup Organic Farm, A			0.0379	0.0575	0.0943	0.9859
	Christanssæde Forest, Ap			0.0024	0.0038	0.1082	0.9948
	Nørlund Plantation, E	0.0331	0.9941				
	Flakkebjerg Farm, Btg2			0.0009	0.0012	0.0766	0.9791
	Nørlund Plantation, Bhs	0.0045	0.9794				
4°C	Nørlund Plantation, C	0.0043	0.9814				
	Flakkebjerg Farm, 2Cg1			0.0007	0.0008	0.2907	0.9924
	Åstrup Forest, A			$5.40 \cdot 10^{-5}$	0.0011	0.0174	0.9880
	Tåstrup Organic Farm, A			$4.81 \cdot 10^{-5}$	0.0005	0.0189	0.9829
	Flakkebjerg Farm, Btg2			0.0029	0.0033	0.0546	0.9128
	Nørlund Plantation, C			0.0043	0.0053	0.3525	0.7268

<sup>†</sup> k-values: Rate constants,  $k_1$  = 1. order rate constant;  $k_{21}$  = 1. order degradation rate constant;  $k_{22}$  = forward rate equilibrium constant (sorption);  $k_{23}$  = reverse rate equilibrium constant (desorption), ¥ Empty cells: Not appropriate.

Table 5  
Fitted Freundlich parameters for sorption of ptaquiloside to the soils investigated

Site and soil horizon	$K_{\text{FREUNDLICH}}^{\ddagger}$	n	$r^2$
Præstø Bracken stand, Oa	1.23	1.00	0.99
Jynde vad Farm, A	0.22	0.96	0.97
Åstrup Forest, A	0.07	0.75	0.99
Tåstrup Organic Farm, A	0.19	0.94	0.99
Christanssæde Forest, Ap	0.20	0.78	0.99
Nørlund Plantation, E	0.08	1.00	0.99
Flakkebjerg Farm, Btg2	0.18	0.72	0.97
Nørlund Plantation, Bhs	0.01	1.00	0.98
Nørlund Plantation, C	0.01	0.73	0.80
Flakkebjerg Farm, 2Cg1	0.08	0.64	0.87

$\ddagger$   $C_{\text{SOIL}}$  measured in  $\text{mg kg}^{-1}$ ,  $C_{\text{WATER}}$  measured in  $\text{mg L}^{-1}$ .



Table 6  
Retardation factors and distribution coefficients for the soil column studies

Site and soil horizon	Soil density (kg L <sup>-1</sup> )	Porosity (%)	Effluent at max. bromide concentration (mm)	Effluent at max. PTA concentration (mm)	Retardation <sup>¶</sup>	K <sub>D</sub> (L kg <sup>-1</sup> )	K <sub>OC</sub> (L f <sub>OC</sub> <sup>-1</sup> kg <sup>-1</sup> ) <sup>§</sup>
Åstrup Forest, A	1.3	51	133.4±4.8	149.3±2.1	1.12	0.05	0.91
Nørlund Plantation, E	1.4	47	94.2±0.6	93.3±2.0	0.99	NA <sup>‡</sup>	NA
Nørlund Plantation, Bhs	1.5	43	79.4±1.7	76.4±0.3	0.96	NA	NA
Nørlund Plantation, C	1.6	40	110.0±0.7	108.0±0.3	0.98	NA	NA

<sup>¶</sup>: Retardation of PTA max. concentration compared to bromide max. concentration. <sup>§</sup>: f<sub>OC</sub>: fraction of organic matter. <sup>‡</sup> NA: Not available.

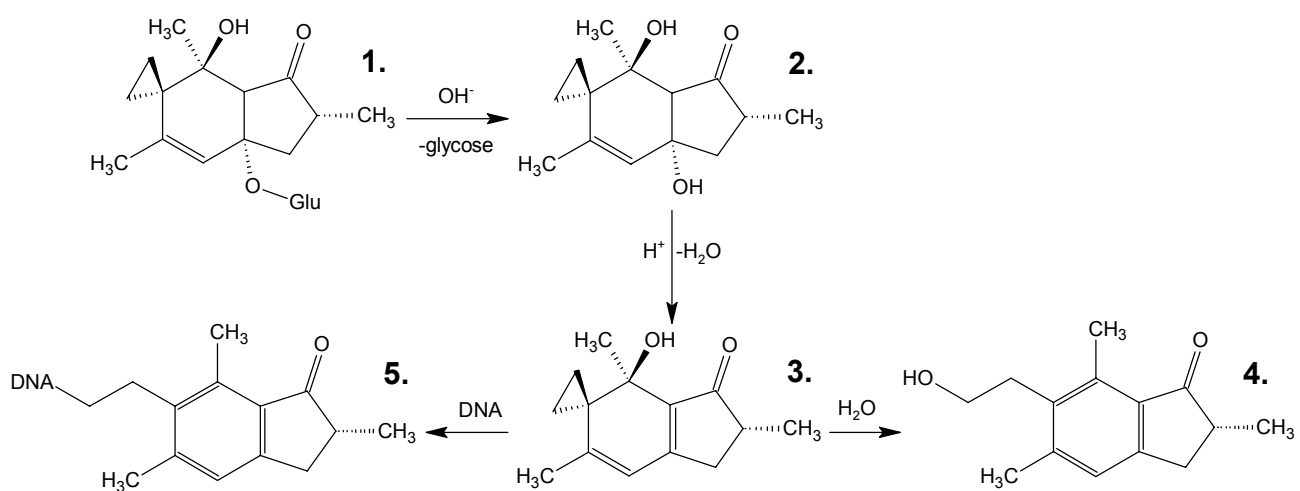


Fig. 1. PTA [1] transformation reactions. During alkaline conditions, first ptaquilosin [2] and then the unstable dienone [3] are formed. Both [1] and [3] yields pterisin B [4] under acid conditions. The unstable dienone may alkylate DNA [5] and is likely to react with numerous other substances including humic substances.

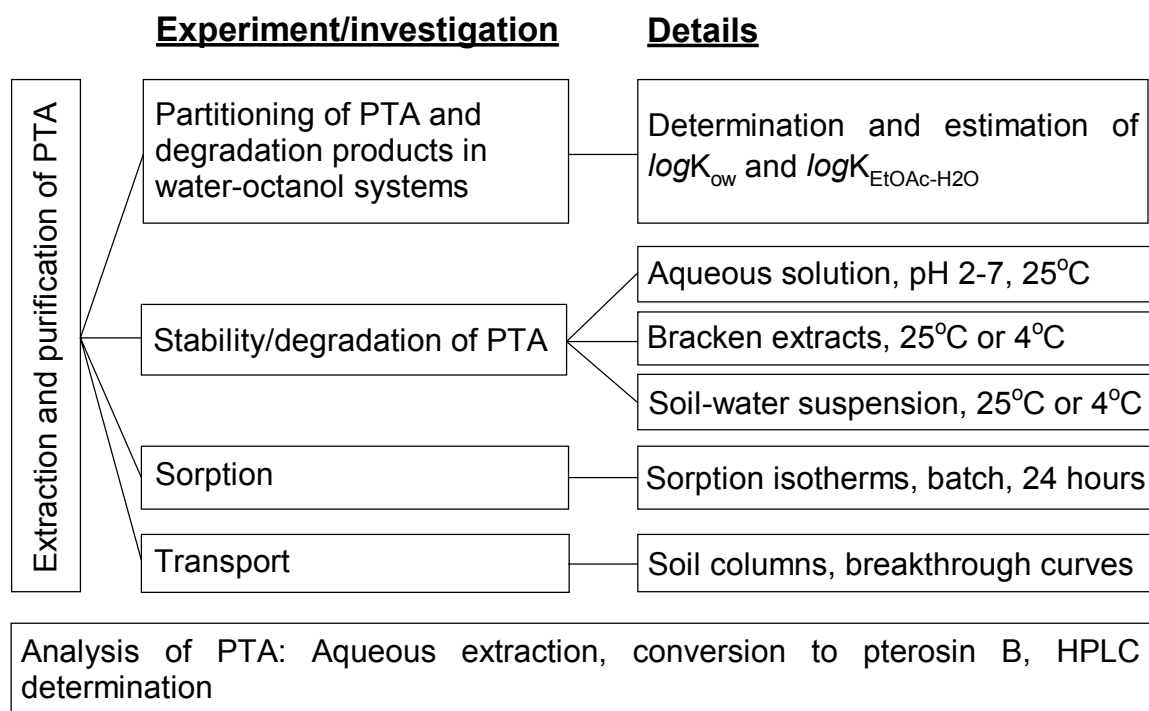


Fig. 2. Overview of main analyses.

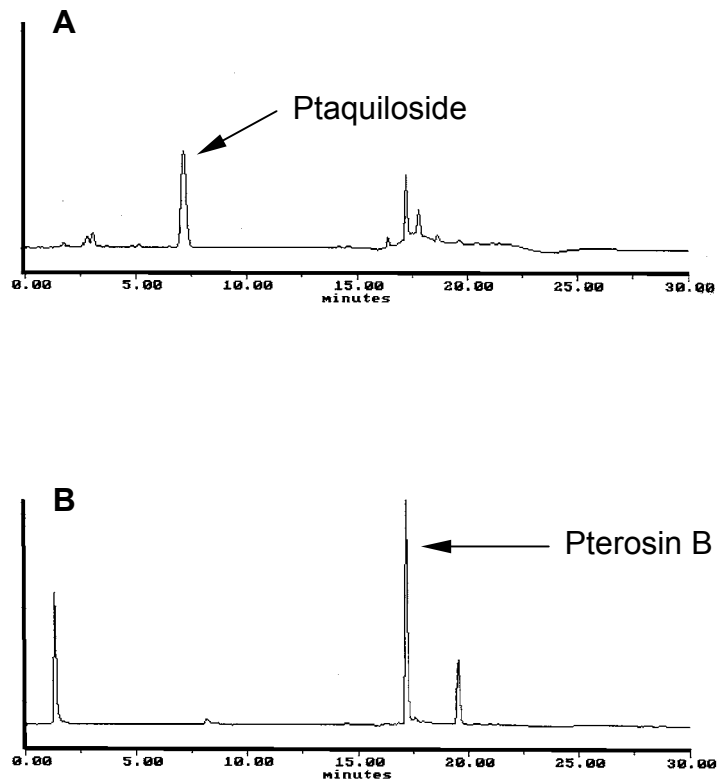


Fig. 3. HPLC chromatograms of raw-ptaquiloside. A) Raw-ptaquiloside (purity: 80%). Substances causing peaks eluting later than 15 min was removed from the raw-ptaquiloside by Polyamide treatment before use; B) The same sample as A, but after conversion to pterisin B (late eluting peaks not removed).

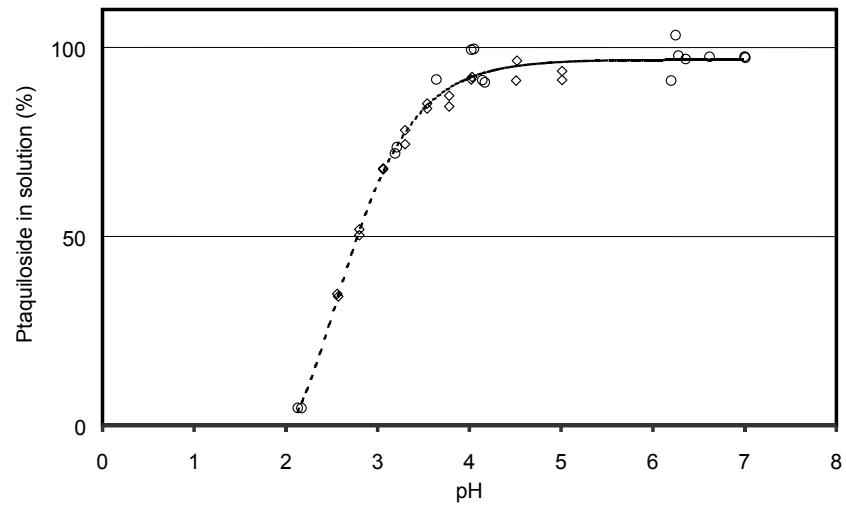


Fig. 4. PTA stability in aqueous solutions at 25°C after shaking for 24 hours: ○) pH adjusted with HCl or NaOH; ◇) acetic acid-acetate buffer.  $[PTA]_{START} = 13 \mu\text{g mL}^{-1}$ . Dotted line represents the range where PTA is unstable (pH < 4), while the solid line is the stable range (pH 4-7).

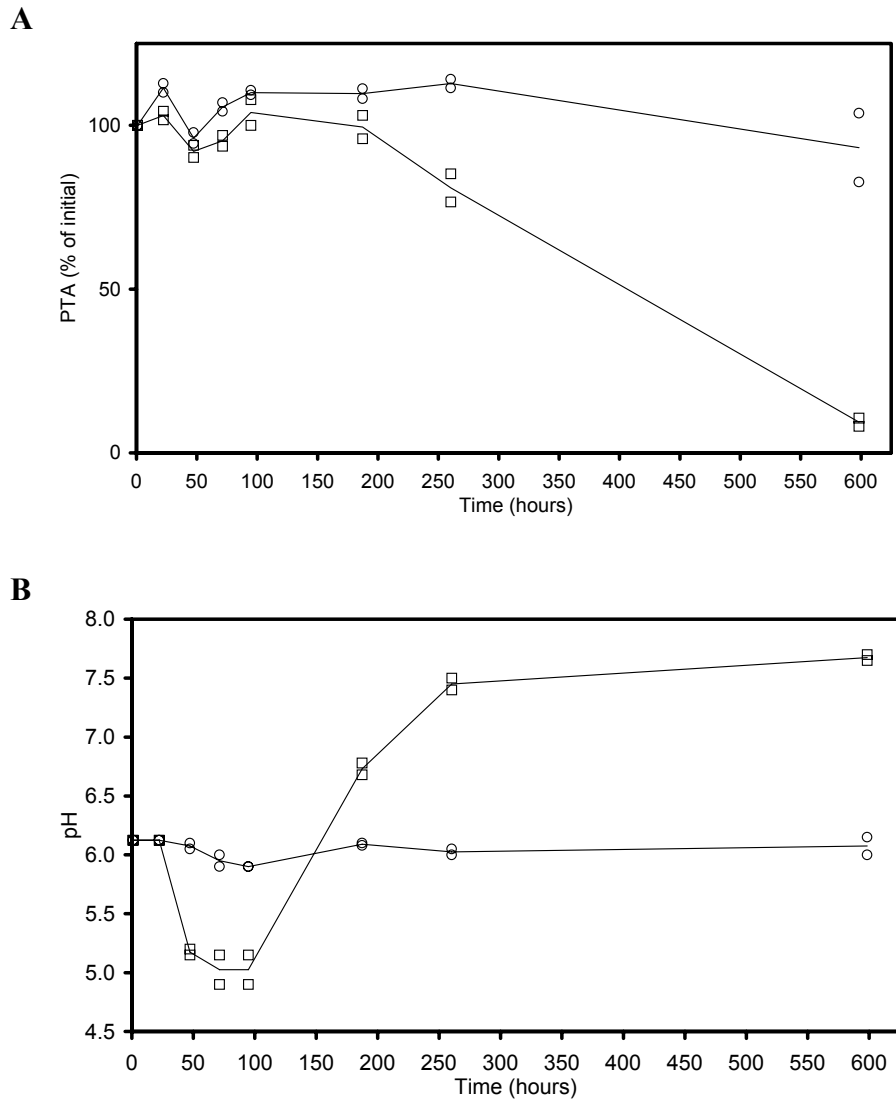


Fig. 5. PTA stability in aqueous Bracken extracts (25 days): A) PTA stability; B) Variation of pH; ○) At 4°C ( $[PTA]_{START} = 105 \mu\text{g mL}^{-1}$ ); □) At 25°C ( $[PTA]_{START} = 113 \mu\text{g mL}^{-1}$ ).

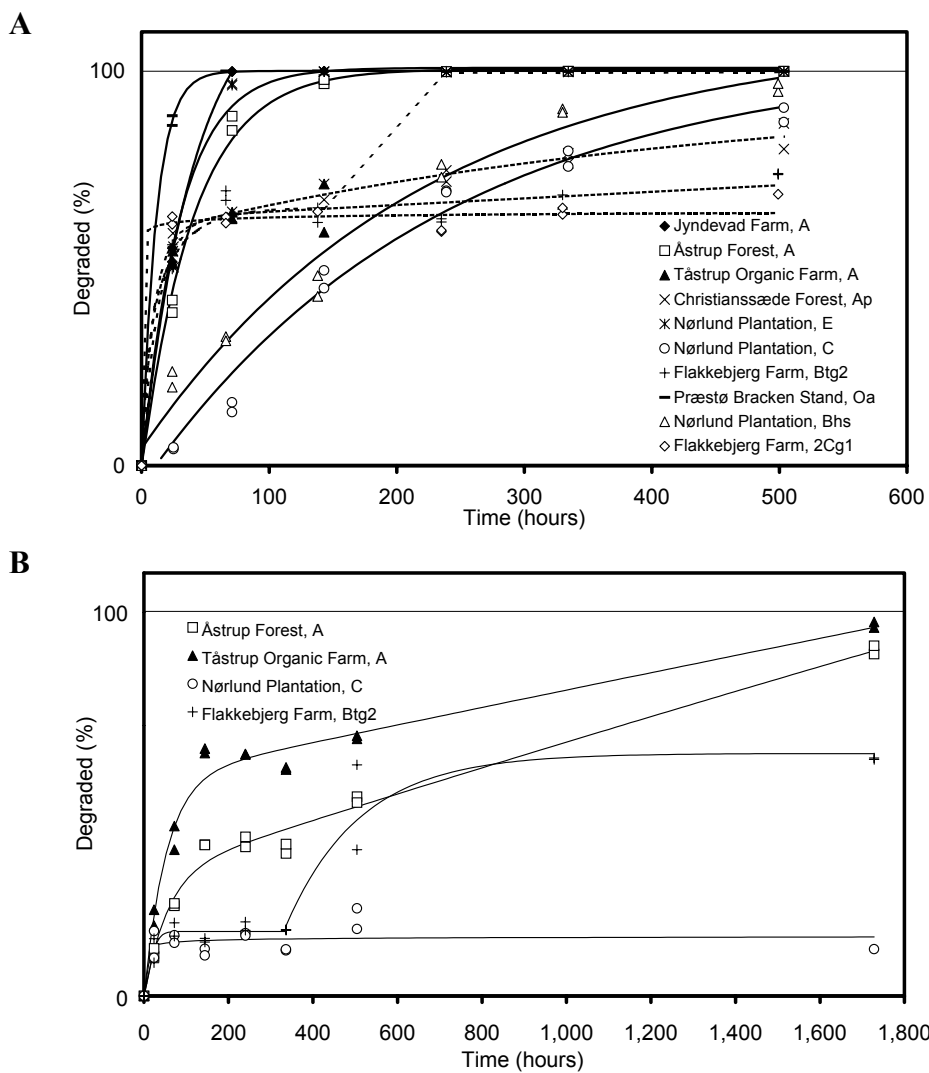


Fig. 6. PTA stability in soil suspensions: A) At 25°C,  $[PTA]_{START}$  ranges between 22 and 30  $\mu\text{g mL}^{-1}$ , thick lines represent first order degradation reactions, thin lines represent the sum of a first order degradation reactions and a concurrent solid phase-water equilibrium; B) At 4°C,  $[PTA]_{START} = 94 \mu\text{g mL}^{-1}$ .

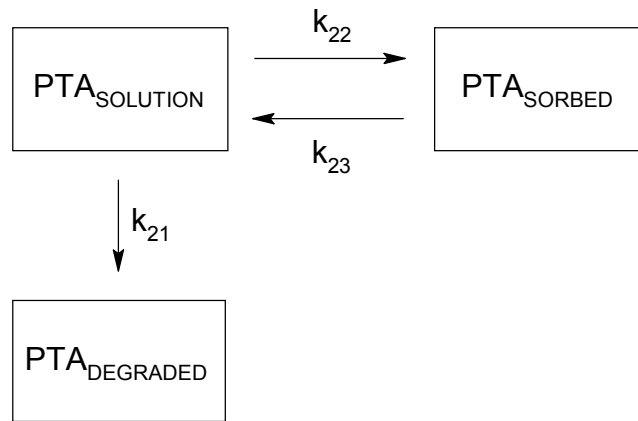


Fig. 7. Proposed PTA degradation reaction scheme in clayey soils comprising a first order degradation reaction (rate constant  $k_{21}$ ) and a concurrent solid phase-water partitioning equilibrium characterized by the forward ( $k_{22}$ ) and reverse ( $k_{23}$ ) rate constants.



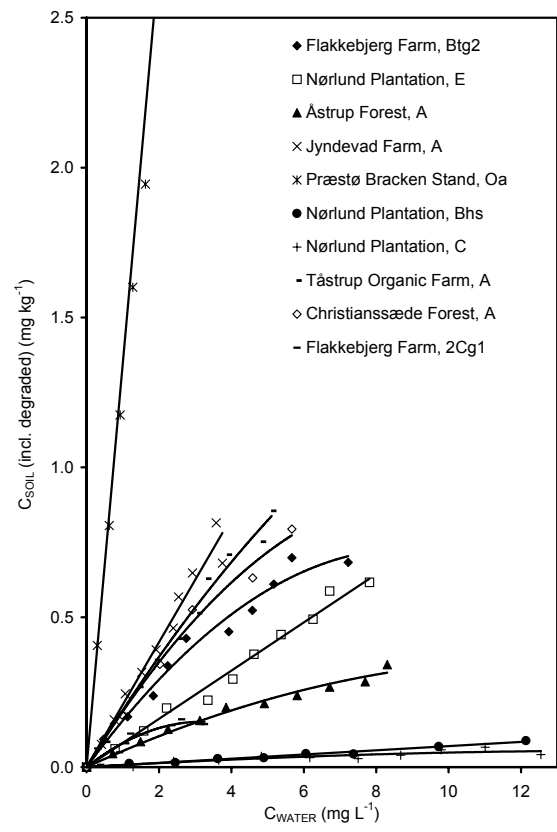


Fig. 8. Sorption isotherms for PTA bonding to soil materials (0.01 M CaCl<sub>2</sub>, 24 hours, [PTA]<sub>START</sub> in the range 0 to 14 μg mL<sup>-1</sup>). Solid lines represent Freundlich-fits. Fitting parameters are listed in Table 5. Only the first 5 data points are shown for Præstø Bracken Stand, Oa.

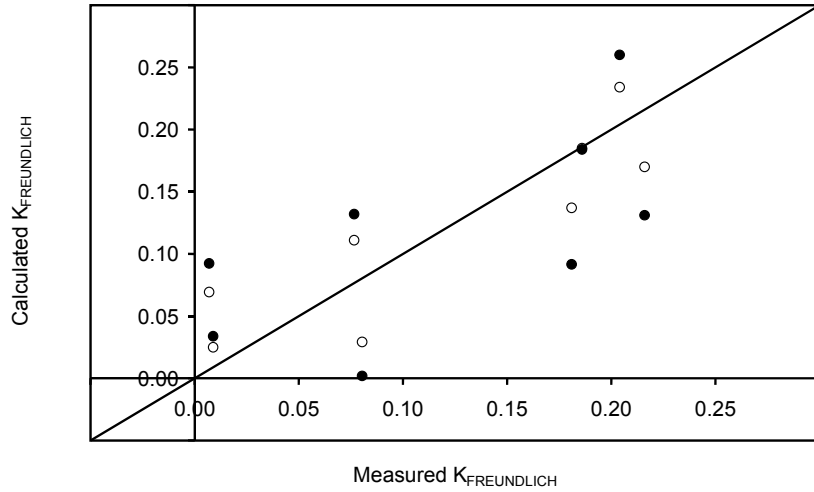
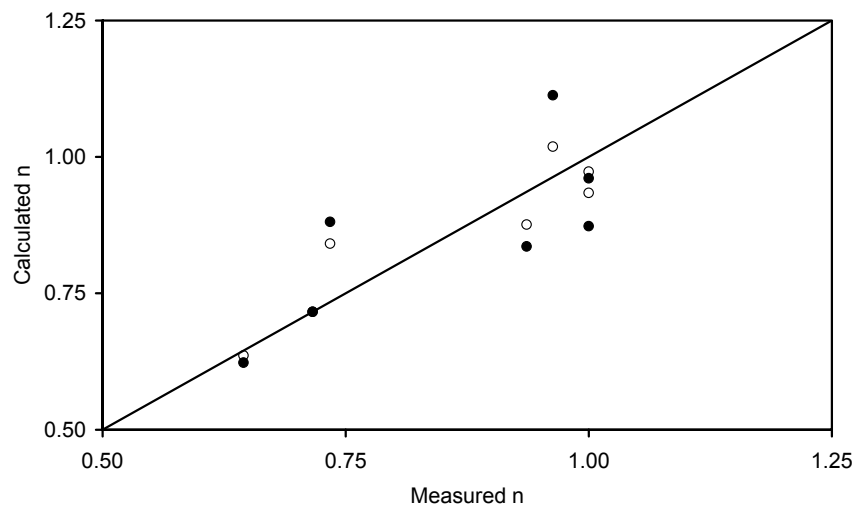
**A****B**

Fig. 9. Multiple linear regressions of sorption parameters in Freundlich models versus soil properties: A)  $K_{\text{FREUNDLICH}}$ ; B)  $n$ . Two soils removed from both regressions: Præstø Bracken Stand Oa and Åstrup Forest A. Christiansæde Forest A removed from B.  $\circ$  = Calibration.  $\bullet$  = Validation .

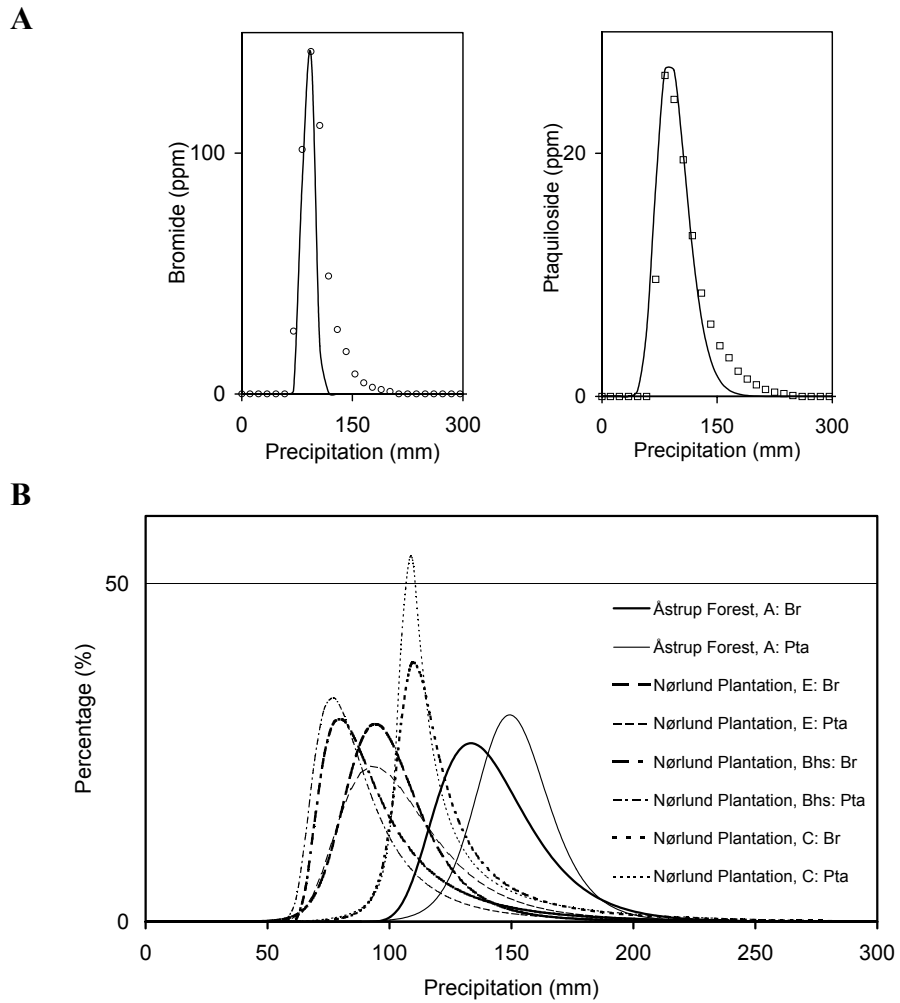


Fig. 10. Breakthrough curves for ptaquiloside and bromide in packed soil columns composed of soil materials from Danish Spodosols: A) Nørlund Plantation, E-curves fitted with the convection-dispersion equation (dispersion coefficients:  $2.1 \text{ cm}^2 \text{ hour}^{-1}$  (bromide) and  $17.0 \text{ cm}^2 \text{ hour}^{-1}$  (ptaquiloside)); B) Curves fitted with Pearson curves,  $R^2$  (calibration curves, bromide/ptaquiloside): Åstrup Forest, A = 0.8300/0.9283; Nørlund Plantation, E = 0.9934/0.9582; Nørlund Plantation, Bhs = 0.9982/0.9995; Nørlund Plantation, C = 0.9841/0.9868. 80% ptaquiloside recovery from Åstrup Forest, A; 100% ptaquiloside recovery from Nørlund Plantation, E, Bhs, and C.



1 **Abstract**—Bracken (*Pteridium esculentum* (Forst. f) Cockayne) is a common fern  
2 encountered all over New Zealand and in many parts of the surrounding pacific islands and  
3 in Australia. Bracken causes tumours in the urinary bladder of cattle browsing on the fern.  
4 This disease, known as *Bovine Enzootic Haematuria* (BEH), is caused by the  
5 norsesquiterpene glycoside ptaquiloside, which can be found in appreciable amounts in the  
6 fern. The purpose of the study was to investigate the variation in the ptaquiloside content in  
7 Bracken from New Zealand, and to compare the ptaquiloside content with the occurrence of  
8 BEH. A further purpose of the study was to identify the correlation between geographic and  
9 edaphic parameters and the ptaquiloside content in the fronds. The frond samples collected  
10 in the survey had ptaquiloside content between 0 and 13,259 µg/g (on dry matter basis). A  
11 significant proportion (87%) of the investigated bracken stands did not contain  
12 ptaquiloside, but brackens in 6 out of 11 areas with reported cases of BEH did contain  
13 ptaquiloside. The correlation between ptaquiloside content and geographic and edaphic  
14 parameters was modelled by partial least squares regression. At national level, the content  
15 in fronds having ptaquiloside correlated positively with latitude and negatively with the  
16 covarying variables altitude and degree of disturbance (signs of mowing or browsing  
17 damage). However, these relationships could not be reproduced at local level. The possible  
18 existence of ptaquiloside chemotypes as cause of the observed ptaquiloside distribution is  
19 discussed.

20

21 **Key Words**—Bracken, New Zealand, ptaquiloside, Bovine Enzootic Haematuria,  
22 *Pteridium*, chemotaxonomy

## INTRODUCTION

Bracken ferns (*Pteridium sp.* (Scop.) Gled, Dennstadiaceae) are some of the most common plants on Earth, found on all continents except Antarctica (Page, 1976). The genus comprises at least 12 species and geographic varieties. The species *Pteridium esculentum* (Forst. f) Cockayne occurs on the Pacific islands, New Zealand and in Australia. In New Zealand, this Bracken is found as a common invasive species throughout the country, in open places, along paddocks and roads, in disturbed habitats, in forest glades and clearings, on the edge of wooded areas and as part of the native bush. It normally reaches a height of 2 m, although taller fronds (3-4 m) appear in warm lowland environments (Brownsey and Smith-Dodsworth, 1989). Bracken is usually considered a weed in New Zealand, but the pre-European settlers, the Maoris, have used the fern rhizome as food, especially during travel (Leach, 2001).

Bracken ferns contain several toxins, which can cause a range of diseases in animals and possible man, ranging from induced thiamine deficiency to cancer, e.g. gastric cancer is found among humans who utilize the fern as food, and *Bovine Enzootic Haematuria* (BEH) occurs in cattle browsing on bracken (Hirayama, 1979; Smith et al., 1988; Ojika et al., 1989; Kushida et al., 1994; Marlière et al., 2000; Potter and Baird, 2000; Alonso-Amelot and Avendaño, 2002; Gava et al., 2002). BEH is a chronic neoplastic disease among cattle older than 3 years characterized by intermittent haematuria and numerous tumours in the urinary bladder (Xu, 1992; Hopkins, 1995; Smith, 1997; Pinto et al., 1999; Caldow and Burns, 2000; Dawra et al., 2000). The norsesquiterpene glycoside ptaquiloside has been shown to be responsible for the formation of these tumours in laboratory animals and administration of pure ptaquiloside to cattle has reproduced similar haematological changes (Hirono et al., 1984). Presence of cattle with BEH is also correlated with the ptaquiloside-content (>100 µg/g) in Australia (Smith et al., 1993; Smith et al., 1994). Ptaquiloside-like substances in bracken such as caudatoside, isoptaquiloside and ptaquiloside Z may also

1 cause BEH (Castillo et al., 1997; Castillo et al., 1998). Bovine papillomavirus (BPV 1 and  
2 2) may also take part in the formation this bladder cancer (Saveria Campo, 2002).

3 Ptaquiloside (Table 1) is regarded as the major and most significant carcinogenic  
4 compound in bracken although other compounds with mutagenic and carcinogenic  
5 properties like have been described (e.g. caudatoside). Ptaquiloside is found in the fronds of  
6 all bracken varieties, although in quite variable amounts; surveys have shown contents  
7 between 0 and 13,000  $\mu\text{g/g}$  (Smith et al., 1994; Alonso-Amelot et al., 1995; Rasmussen et  
8 al., 2003b). The frond content is usually highest at the start of the growing season when the  
9 young crosiers emerge from the ground. The content then declines as growth proceeds, but  
10 never reaches zero (Alonso-Amelot et al., 1992; Alonso-Amelot et al., 1995; Rasmussen  
11 and Hansen, 2003). Spring is the time of year when cattle are most susceptible to poisoning  
12 by bracken, since the young fronds are appealing to the cattle, while at the same time being  
13 in their most toxic state. Surveys of bracken populations in South America (*P. aquilinum*  
14 vars. *arachnoidum* and *caudatum* (Alonso-Amelot et al., 1992; Alonso-Amelot et al.,  
15 1995)) and Australia (*P. esculentum* (Smith et al., 1994)) showed that altitude and latitude  
16 might affect the ptaquiloside content in fronds. In Australia, the brackens growing at  
17 southern latitudes had higher ptaquiloside content than ferns coming from northern  
18 latitudes, while in Venezuela the fern on mountain slopes at higher altitudes had lower  
19 ptaquiloside content than those growing at lower altitudes. However, a Costa Rican study  
20 indicated the opposite pattern (Villalobos-Salazar et al., 2000). In a recent Danish  
21 investigation of the ptaquiloside contents in *P. aquilinum* var. *aquilinum*, it was  
22 demonstrated that the ptaquiloside content in fronds was correlated with the edaphic  
23 conditions of light, water and available nutrients (Rasmussen et al., 2003b). Together, these  
24 investigations indicate possible existence of ptaquiloside variation caused by recognizable  
25 geographic and edaphic parameters. Smith et al. (1994) examined also the ptaquiloside  
26 content in fronds from a collection of brackens from all-over the World grown at the same  
27 conditions in Sydney, Australia, and found large variations in the ptaquiloside content

1 inside different bracken species and varieties. However, no significant differences could be  
2 found between different varieties and species. In addition, some samples did not contain  
3 ptaquiloside at all, indicating existence of ptaquiloside chemotypes within several species  
4 and varieties. These observations, although based on small numbers of samples, suggest  
5 that the level of ptaquiloside encountered in a particular bracken frond is partly due to  
6 genetic heritage as well as edaphic parameters.

7 The aims of the present study were to: 1) Investigate the range of ptaquiloside contents in  
8 young bracken fronds from different sites in New Zealand; 2) Investigate the correlation  
9 between the ptaquiloside content in bracken fronds and the occurrence of BEH in New  
10 Zealand, and 3) Investigate possible correlations between ptaquiloside in bracken fronds  
11 and edaphic parameters.

12



## MATERIALS AND METHODS

*Organization of the Study.* The study were carried out as 3 separate surveys:

- 1) Regional Survey: Investigation of the variation in the ptaquiloside content in bracken fronds on a regional basis. This survey was carried out in the Waikato-region and on the Coromandel Peninsula (North Island) on several sampling-trips in September and October 2001. Both old and young fronds were sampled. A total number of 62 Bracken stands were investigated (Fig. 1A). Bracken samples were taken along roadsides, near fences, and in paddocks, covering a wide geographical range of locations in the Waikato-region and on the Coromandel Peninsula.
- 2) Farm Survey: Investigation of the local variation in the ptaquiloside content in bracken fronds encountered on a sheep and cattle farm situated 30 km west of Taumarunui, North Island. BEH has previously been reported from this farm, and ptaquiloside was found in fronds from this site in the Regional Survey. The survey was carried out November 5-9, 2001 by sampling young fronds from 27 bracken stands within the farm. Samples were taken over a steep hilly section of the farm on grazed paddocks (Fig. 1B).
- 3) National Survey: Investigation of the variation in the ptaquiloside content in bracken fronds on a national level. This survey was carried out November 21-23 and 27-29 (North Island), and December 15-29, 2001 (South Island). In total, 186 bracken stands were investigated over both main islands in the National survey (Fig. 1C). The survey included 9 areas where BEH was known to occur. Bracken stands sampled were located 30-40 km apart (2-3 km in BEH-areas) in different ecosystems, hereby trying to include as many bracken ecosystems as possible

1            (roadsides, fences, forests, bush, and paddocks grazed by deer, sheep,  
2            and cattle).

3        All surveys taken together, 275 different bracken stands all over New Zealand were  
4 investigated as a part of the study.

5        *Fieldwork and Sample Preparation.* At each bracken stand in the Farm and Regional  
6 Surveys, 3-20 whole fronds were harvested for analysis, while a number of 6 young fronds  
7 were used in the National Survey. The fronds were collected among average sized  
8 undamaged fronds from the central part of the bracken stands. In the National Survey only  
9 the unfurling frond tips were collected resulting in sampling of the outermost 20-30 cm of  
10 frond. All fronds were separated from the rhizome at the soil surface, their length was  
11 measured, and the number of unfurling pinnae counted, in order to determine the Bracken  
12 growth stage. In the National Survey, three mature fronds from the previous growth season  
13 were also harvested for measurement of height and maximum development stage. In the  
14 Farm and the National Surveys, detailed site descriptions were performed according to the  
15 FAO-Unesco system (FAO, 1990), although adapted to New Zealand (NZ) conditions  
16 based on the experiences obtained during the Regional Survey, where a simplified FAO-  
17 Unesco site description system was used (FAO, 1990). The FAO-Unesco site descriptions  
18 used in the Farm and National Surveys included; NZ landuse classes, NZ vegetation  
19 classes, size of bracken stand (<50m<sup>2</sup>, 50-200m<sup>2</sup>, 200-500m<sup>2</sup>, and >500m<sup>2</sup>), degree of  
20 bracken ground cover (low [15%], moderate [15-40%], high [40-80%], and dominant [80-  
21 100%]), light conditions (shaded [below light tree species], glade [stand encircled by trees  
22 {>3m tall}], scrub [fronds growing between small trees {<3m tall}, open land]), bracken  
23 status (suppressed [other plant species dominate the ecosystem], subdominant [bracken is  
24 outnumbered by another plant], dominant [bracken is the most prominent plant]), bracken  
25 community growth stage (new [only new plants {max. 1 year}], young [no old fronds  
26 present, no litter], mature [old fronds and litter present], and old [bracken community is

1 degenerating] (Watt, 1976)), and bracken disturbance (none, bitten [by animals], managed  
2 [mowed, sprayed etc.]).

3 All samples from the Farm and Regional Surveys were frozen within 4 hours of  
4 sampling, while samples from the National Survey were dried in an Ezidry FD 1000 Food  
5 Dehydrator at 45°C (until crisp dry - 12 hours) before freezing.

6 *Ptaquiloside Analysis.* Ptaquiloside was extracted from thawed and milled or macerated  
7 fronds using MilliQ-water at room temperature for 60 min in BlueCab<sup>®</sup> bottles at a flat-bed  
8 orbital shaker (350 rpm) (1:10 w/v plant-water ratio, Farm and Regional Survey; 1:25 w/v  
9 plant-water ratio, National Survey). The ptaquiloside content in the fronds was measured in  
10 the aqueous extracts, after treating the extracts with resin (Polyamide 6, Fluka, Steinheim,  
11 Switzerland) to remove interfering substances and conversion to pterosin B following the  
12 method of Agnew and Lauren (1991). The conversion results in the determination of the  
13 sum of ptaquiloside and of any isoptaquiloside present. Both compounds are carcinogenic,  
14 therefore this conversion should be a better measure than determination of ptaquiloside  
15 alone when comparing the occurrence of bracken related diseases with the content of  
16 ptaquiloside in bracken (Castillo et al., 1997). The pterosin B was determined at 220 nm  
17 (Shimadzu SPD-6AV UV-VIS Spectrophotometric Detector) on a Shimadzu LC-6A Liquid  
18 Chromatograph (Shimadzu SCL-6A System Controller) equipped with Merck  
19 LiChroCART<sup>®</sup> column (125×4 mm) packed with LiChrospher<sup>®</sup> 100 RP-8, 5 µm. The  
20 measurements were performed at 35°C (Shimadzu CTO-6A Column Oven) using a water-  
21 acetonitrile (83:17 v/v) eluent and a flow rate of 1.00 ml/min. A 20 µl sample injection-  
22 volume was used for all analyses. The limit of detection of the method was below 5 µg/g  
23 when expressed as ptaquiloside in dried plant (Agnew & Lauren, 1991). The ptaquiloside  
24 content of all samples in the Farm and Regional Surveys was determined in duplicate. The  
25 average coefficient of variance ( $CV_{AVG}$ ) in the samples containing ptaquiloside was 19%.  
26 This relatively high CV was due to a few samples which gave markedly different results for

1 the duplicates, but further analysis was not practical due to the potential degradation of  
2 ptaquiloside in the thawed and chopped samples. The median CV ( $CV_{MED}$ ) was 9%. For the  
3 National Survey 70% of the samples were analysed in duplicate, and the samples with  
4 ptaquiloside gave a  $CV_{AVG}$  of 6%. The water-content in the samples from the Farm and the  
5 Regional Surveys were measured after drying subsamples of the thawed and chopped  
6 samples at 50°C. In the National Survey, the dry weight at 45°C was used for  
7 quantification. All results are expressed on a dry weight basis.

8 *Statistical Methods.* Statistical analyses were performed in the *Unscrambler ver. 7.6 SR-1*  
9 software package (Camo ASA software, Oslo, Norway) and in *Microsoft® Excel 2000*  
10 (Microsoft Corporation, USA). Partial Least Squares Regression (PLSR) was carried out on  
11 data from the Farm and National Surveys, using all qualitative and quantitative data, to  
12 investigate potential correlations between ptaquiloside contents and ecological parameters.  
13 PLSR models were developed using jack-knifing techniques and Martens' uncertainty-test.  
14 Removal of outliers and non-significant variables were permitted. The models were  
15 validated and evaluated by full cross validation (Bro, 1996; Esbensen, 2000).  
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## RESULTS AND DISCUSSION

The investigated Bracken stands are located all-over New Zealand in a multitude of ecosystems ranging from roadsides to undisturbed forests (Fig. 1, Tables 2-6).

*Regional Survey.* The content of ptaquiloside in the young unfurling fronds as well as in the old fronds from the previous growth season sampled in the Regional survey was zero or below the detection limit in all samples except from 4 samples collected at a cattle and sheep farm situated 30 km west of Taumarunui and in 2 samples from bracken stands nearby this farm (Fig. 1A, Table 2). The content in these fronds ranged from 5 to 1,200 µg/g. The highest levels were found in stands from a hill ridge at the farm, and in a small stand growing at the roadside just outside the farm. BEH has previously been reported at the farm. The low number of stands with ptaquiloside and the occurrence of stands with and without ptaquiloside close to each other at the Taumarunui farm within an area of a few square kilometres prompted a more detailed survey at this site.

*Farm Survey.* 27 Bracken stands were described and sampled (Tables 3 and 4). The median development stage was 3.1 pinnae resulting in a median height of 45.3 cm. Most stands were situated on sloping land or on hill ridges. The majority had a southern or western aspect, which was due to the topography of the farm. All areas investigated were used as sheep and cattle paddocks. The vegetation class of the paddocks were mainly mixed as many shrubs, small patches of three-ferns and other native as well as exotic species were scattered all-over the farm growing mixed with bracken. In most stands, no signs of disturbance were observed.

The content of ptaquiloside in the fronds ranged from 0 to 13,300 µg/g with a median content of 2,151 µg/g among 17 out of 27 sites (62%) that had measurable ptaquiloside contents (Fig. 1B, Table 4). The median development stage of all stands was 3.1 pinnae, while the development stage among stands having ptaquiloside was only 2.6 pinnae. Compared to stands containing no ptaquiloside, the ptaquiloside containing stands were in

1 general mature stands placed in grasslands with higher bracken cover. The content was  
2 within the range observed previously in bracken fronds in New Zealand and Australia  
3 (Smith et al., 1988; Smith et al., 1994). The ptaquiloside content in the stands having  
4 ptaquiloside was generally high, and high enough to cause BEH, which is in accordance  
5 with the disease history of the farm (Smith et al., 1994). The observed pattern with nearby  
6 stands having ptaquiloside concentrations ranging from zero to over 10,000  $\mu\text{g/g}$  resembles  
7 previously observed patterns of variation among different stands and even individual fronds  
8 of *P. aquilinum* var. *aquilinum* in England regarding the content of the cyanogenic  
9 glycoside prunasin (cyanogenic polymorphism) (Cooper-Driver and Swain, 1976; Cooper-  
10 Driver et al., 1977).

11 PLSR was carried out to unveil correlations between the ptaquiloside content and the  
12 edaphic parameters (Fig. 2). The analysis was restricted to sites having ptaquiloside in the  
13 fronds. By using a logarithmic model based on the number of pinnae (development stage)  
14 and the altitude of the stand it was possible to predict the ptaquiloside content on 13 out of  
15 17 sites (76%). Both variables were positively correlated with the ptaquiloside content. The  
16 model could not describe 4 sites out of 17, which subsequently were removed from the  
17 modelling. Those were outlying stands having either (Table 4): 1) Extreme ptaquiloside  
18 contents (F15 and F32); 2) Extremely long stipes (F14); and 3) Extreme variation in the  
19 number of pinnae (F19). In addition, two (F14 and F15) of the outliers showed very high  
20 variation among duplicates in the ptaquiloside determination. The PLSR analysis indicate  
21 that the ptaquiloside content in the fronds could be partly caused by ecological stress since  
22 higher contents were found in fronds growing in more stressed environments (hill-ridges;  
23 harsher micro climate and lower soil quality compared to down-slope soils). It should be  
24 noticed that the development stage is positively correlated with the ptaquiloside content,  
25 which is in contradiction to the usual finding that the ptaquiloside content decreases with  
26 the age and hence, the development stage of the frond. This might be due to the time span  
27 of the sampling. The Farm Survey was carried out over a relatively short time, while the

1 ptaquiloside content usually decreases during the entire growth season, i.e. over a period of  
2 several months.

3 *National Survey.* 186 Bracken stands were investigated in the national survey (Table 5  
4 and 6). The median development stage of the young fronds from 2001 was 3.3 pinnae  
5 resulting in a median height of 68.3 cm, while the old fronds from the previous year (2000)  
6 had reached a median height of 138 cm and a development stage of 17.3 pinnae for mature  
7 fronds. 36 % of the stands were situated along roads on strongly disturbed soils, while the  
8 rest of the stands were situated on flat terrain, valley floors or on sloping land on more  
9 natural soils. Most stands had a northern aspect - oriented towards the sun - and were  
10 situated on unused areas or in forests. The vegetation class was therefore mainly grassland  
11 and grassland mixed with scrub. The stands were generally situated in open land or among  
12 scrub, as small dominant mature communities. Approximately 25% of the stands showed  
13 signs of disturbance.

14 Among the 24 sites (13%) that contained detectable ptaquiloside in the fronds, the  
15 concentration ranged from 5 to 2,603  $\mu\text{g/g}$  with a median content of 290  $\mu\text{g/g}$  (Fig. 1C,  
16 Table 6). The median development stage among stands having ptaquiloside was 3.2 pinnae,  
17 close to the development stage of fronds without ptaquiloside. The ptaquiloside contents  
18 were lower than the concentrations found at the Taumarunui Farm, and within the range  
19 observed elsewhere for bracken (Smith et al., 1988; Smith et al., 1994). The contents were  
20 generally moderate, and only a few sites had ptaquiloside contents considered high enough  
21 to cause BEH. There were no marked morphological or edaphic differences between  
22 Bracken stands having ptaquiloside and those without.

23 PLSR was carried out on ptaquiloside containing stands to investigate any correlations  
24 between ptaquiloside content and edaphic parameters (Fig. 3). It was possible to predict the  
25 ptaquiloside content in the fronds on 21 out of 24 sites (88%) using a logarithmic model  
26 based on the altitude of the stand, the latitude (geographical coordinate: Southern latitude),  
27 and the degree of bracken disturbance (signs of animals browsing). The altitude and

1 bracken disturbance were negatively correlated, while latitude was positively correlated  
2 with the content of ptaquiloside. In addition, altitude and bracken disturbance were  
3 covariate. The significance of altitude is subordinate to the significance of latitude and  
4 disturbance. The model was not able to describe three stands, of which two were stands  
5 having extremely low ptaquiloside contents (Table 6: N457 and N467) while stand N228  
6 did not have any measurement of altitude, and hence could not be modelled. Those sites  
7 were therefore removed from the final model. The PLSR analysis indicates that the  
8 ptaquiloside content in the fronds is generally higher in fronds growing in the southern part  
9 of New Zealand, which is in line with the findings from Australia (Smith et al., 1994). The  
10 altitude of the stand is negatively correlated with the ptaquiloside content, which is in  
11 contrast with the results of the Farm Survey. This confusing but intriguing influence of  
12 altitude on ptaquiloside in fronds is in line with previous reports from South America and  
13 needs further investigation.

14 *Correlation between ptaquiloside content and the occurrence of BEH.* Among the 24  
15 ptaquiloside containing stands from the National Survey, 13 (54%) were found in areas  
16 where BEH had been observed (Table 7). Among the 10 BEH areas investigated in the  
17 whole study, 6 areas had bracken containing ptaquiloside. The number of BEH areas with  
18 ptaquiloside was expected to be higher, due to previous field and laboratory-findings  
19 indicating the connection between ptaquiloside, bracken, and BEH (Hirono et al., 1984;  
20 Smith et al., 1988). Furthermore, bracken from some of the BEH-areas did not show any  
21 presence of ptaquiloside. The explanation for these observations may be due to the limited  
22 number of bracken stands investigated inside the BEH areas, i.e. the high degree of  
23 ptaquiloside polymorphism results in high proportion of no-ptaquiloside stands and  
24 therefore a high probability of not finding brackens with ptaquiloside even if they are  
25 present in the investigated area.

26



## GENERAL DISCUSSION

1  
2 Ptaquiloside was found in the range 0-13,259  $\mu\text{g/g}$  in young unfurling fronds of bracken  
3 in the Farm Survey, while the levels were 0-2,603  $\mu\text{g/g}$  in the National Survey, which is  
4 within the range reported earlier for bracken in Australia and in New Zealand. A high  
5 proportion (84%) of the stands did not contain ptaquiloside at all. This predominant lack of  
6 ptaquiloside producing bracken for large areas of the country has not been reported until  
7 now, although consistently large differences between two locations have been reported  
8 earlier in New Zealand (Smith et al., 1988). The finding is similar to the work of Smith et  
9 al., (1994) who found large variations in the ptaquiloside content as well as some ferns  
10 containing no ptaquiloside at all. Bracken chemotypes have been found with respect to the  
11 occurrence of the cyanogenic glycoside prunasin (Cooper-Driver, 1976; Cooper-Driver and  
12 Swain, 1976; Low and Thomson, 1990). The lack of ptaquiloside in some stands in the  
13 present study might therefore be explained by the existence of ptaquiloside chemotypes of  
14 bracken. An attempt was made to model the occurrence of zero-ptaquiloside stands by  
15 multiple linear regression, PLSR and principal component analysis in line with the analysis  
16 carried out on the fronds containing ptaquiloside, but no significant correlations could be  
17 identified. Another plausible explanation of the lack of ptaquiloside in some brackens could  
18 be the existence of more than one bracken variety in New Zealand, or that the fern itself  
19 does not produce ptaquiloside. Instead, a symbiotic endomycorrhizal fungus could produce  
20 the compound. Due to the close association between fungi and fern in mycorrhizal  
21 networks, the fungi might introduce ptaquiloside or a biosynthetic precursor into the ferns  
22 vascular system (Schoental, 1984). Ptaquiloside-like substances and their products of  
23 hydrolysis, the pterosins, have been identified in some basidiomycetes (e.g. illudin M and S  
24 from *Omphalotus illudens* and *O. olearis* and pterosins in *Fomes annosus* and *Cyathus*  
25 *bulleri*) (McCloud et al., 1996, Potter, 2000). Ptaquiloside is, however, found in ferns  
26 grown in sterilized potting soil. Those had even higher contents than similar ferns grown

1 artificially in natural bracken soil (Smith et al., 1990). The origin and exact role of  
2 ptaquiloside in bracken remains still to be explained. If the fungal theory is correct, then the  
3 presence of ptaquiloside might just be an effect of fungal infection, and the observed  
4 pattern for the ptaquiloside content is a mirror of the distribution of a ptaquiloside  
5 producing fungi.

6 Detailed studies of the ptaquiloside content in different species and varieties of bracken  
7 from Australia, South America and Europe have demonstrated correlations between  
8 ptaquiloside content, growth stage, and edaphic/geographic variables, such as altitude,  
9 latitude, light exposure and nutritional status. The present study supports these results to  
10 some degree, although the Farm and National Surveys exhibited contradictory results  
11 regarding the effect of altitude. Based on the present study, it is not possible to explain the  
12 variations in the frond ptaquiloside content. It is likely that the variations may be caused by  
13 unexplained variations in other edaphic parameters, such as plant available nutrients, soil  
14 texture, precipitation, fungal infection, genetic heritage, a combination of those, or by other  
15 factors entirely. Further studies are needed to uncover the complex correlations between  
16 edaphic parameters, the occurrence of ptaquiloside chemotypes and contents before general  
17 applicable conclusions can be made.

18 There is a reasonable association of ptaquiloside containing stands of bracken to areas  
19 where BEH has been observed. However, the proportion of stands containing ptaquiloside  
20 was low. In addition, the ptaquiloside contents were in general low compared to the  
21 contents, which have been found to cause BEH. These findings may simply reflect the very  
22 localised nature of high-ptaquiloside bracken as shown in the Farm Survey. The overall low  
23 ptaquiloside contents observed are in agreement with the declining occurrence of BEH in  
24 New Zealand. The declining BEH-occurrence is probably due to improved farming  
25 practises, namely, the recognition of bracken as a significant weed, the use of high stock  
26 numbers and effective herbicides to remove bracken from grazed areas.

1       Great care must be taken in future investigations of the geographical variation in the  
2       ptaquiloside content in bracken. Such surveys must be very detailed to ensure accurate and  
3       reliable results. Interpretations of studies based on low sample numbers must therefore be  
4       made with great care.  
5

## CONCLUSION

This study shows that ptaquiloside is found in the range 0-13,259  $\mu\text{g/g}$  in young unfurling bracken fronds in New Zealand, although a high proportion of the investigated bracken stands did not contain ptaquiloside at all. The contents of ptaquiloside are within the range previously reported for bracken in Australia and in New Zealand. The content in fronds with ptaquiloside was modelled by partial least squares regression and was found to be correlated with the following edaphic and geographic variables: Altitude, development stage, latitude, and degree of disturbance. At the local level (the Farm Survey), more ptaquiloside was found in older fronds growing at higher altitudes compared to younger fronds at lower altitudes. At national level, more ptaquiloside was found in fronds from southern compared to northern latitudes, while altitude and the degree of disturbance decreased the ptaquiloside content in a covarying complex manner. Models developed at local and national levels were convincing, but due to opposing correlation coefficients between the two separate studies, a general model for the ptaquiloside content could not be obtained. Possible causes for the high variation in ptaquiloside concentrations and occurrence could be the existence of ptaquiloside chemotypes of bracken, the existence of endomycorrhizal fungi, which produces, or triggers the production of ptaquiloside in the fern, or the fern responses to some edaphic growth factors not investigated.

Compared to previous reports on the correlation between ptaquiloside and BEH, the number of stands found to contain ptaquiloside in BEH areas compared to the total number of stands investigated in those areas ptaquiloside was low, as was the concentrations of ptaquiloside found. However, overall, 60% of the stands found nationally to contain ptaquiloside were found in areas with a known record of BEH, supporting the association between BEH and ptaquiloside.

1     *Acknowledgements*—This study was made possible by generous grants from Eides Legat,  
2     Studiefonden for Danmarks Jordbrugsvidenskabelige Ph.D.-forening, Skovrider Mogens  
3     Andersens og Hustrus Legat, Knud Højgaards Fond and OHF og AJ-E Heilmanns Fond.  
4     Dawei Deng, Dwayne Jensen, Heather McBridie, Bruce Willoughby, and Solveig  
5     Warnecke are all thanked for their valuable assistance during the bracken sampling trips.

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## FIGURE CAPTIONS

<b>Table no:</b>	<b>Text:</b>
<b>1</b>	TABLE 1. Properties of ptaquiloside.
<b>2</b>	TABLE 2. Stand descriptions and ptaquiloside contents, Regional survey.
<b>3</b>	TABLE 3. Comparison of ecological variables between ptaquiloside (Pta) and non-ptaquiloside (Non-Pta) containing bracken stands, Farm Survey.
<b>4</b>	TABLE 4. Descriptions of ptaquiloside containing stands, Farm Survey.
<b>5</b>	TABLE 5. Comparison of ecological variables between ptaquiloside (Pta) and non-ptaquiloside (Non-Pta) containing bracken stands, National Survey.
<b>6</b>	TABLE 6. Description of ptaquiloside containing stands, National Survey.
<b>7</b>	TABLE 7. Correlation between the occurrence of BEH and ptaquiloside.
<b>Figure no:</b>	<b>Text:</b>
<b>1</b>	<p>FIG. 1. A: Map of the North Island, Waikato and Coromandel Regions. Investigated sites in the Regional Survey are marked on the map (●) as well as the Taumarunui farm area (●). B: Map of the investigated area at the farm at Taumarunui, North Island, showing the location of the investigated sites (UTM 60H) and the ptaquiloside levels encountered (<math>\mu\text{g/g}</math>): ○ = 0; ✕ = 1-999; + = 1,000-9,999; ● = &gt;10,000. C: Map of New Zealand. Investigated sites in the National Survey are marked on the map (●) as well as BEH-sites included in the survey (●). P indicates bracken stands with ptaquiloside.</p>
<b>2</b>	<p>FIG. 2. PLSR-model of the ptaquiloside content in young fronds, Farm Survey (77% explained calibration Y-variance, 70% explained validation Y-variance, 1 PC (principal component)). ○: Samples used for modelling. ✕: Outlying samples not included in the model. Thick line: <math>\text{LogPTA (measured)} = \text{LogPTA}</math></p>

	<p>(calculated). Thin line: Model trend line (offset and slope shown). <math>R^2</math> = the variance in LogPTA (calculated) according to the model. <math>Q^2</math> = the variance in LogPTA (calculated) according to cross validation. RMSEP = root mean square error of prediction.</p>
<p><b>3</b></p>	<p>FIG. 3. PLSR-model of the ptaquiloside content in young fronds, National Survey (82% explained calibration Y-variance, 78% explained validation Y-variance, 2 PC's). ○ Samples used for modelling. ×: Outlying samples not included in the model. Solid line: LogPTA (measured) = LogPTA (calculated). Thin line: Model trend line. <math>R^2</math> = the variance in LogPTA (calculated) according to the model. <math>Q^2</math> = the variance in LogPTA (calculated) according to cross validation. RMSEP = root mean square error of prediction.</p>

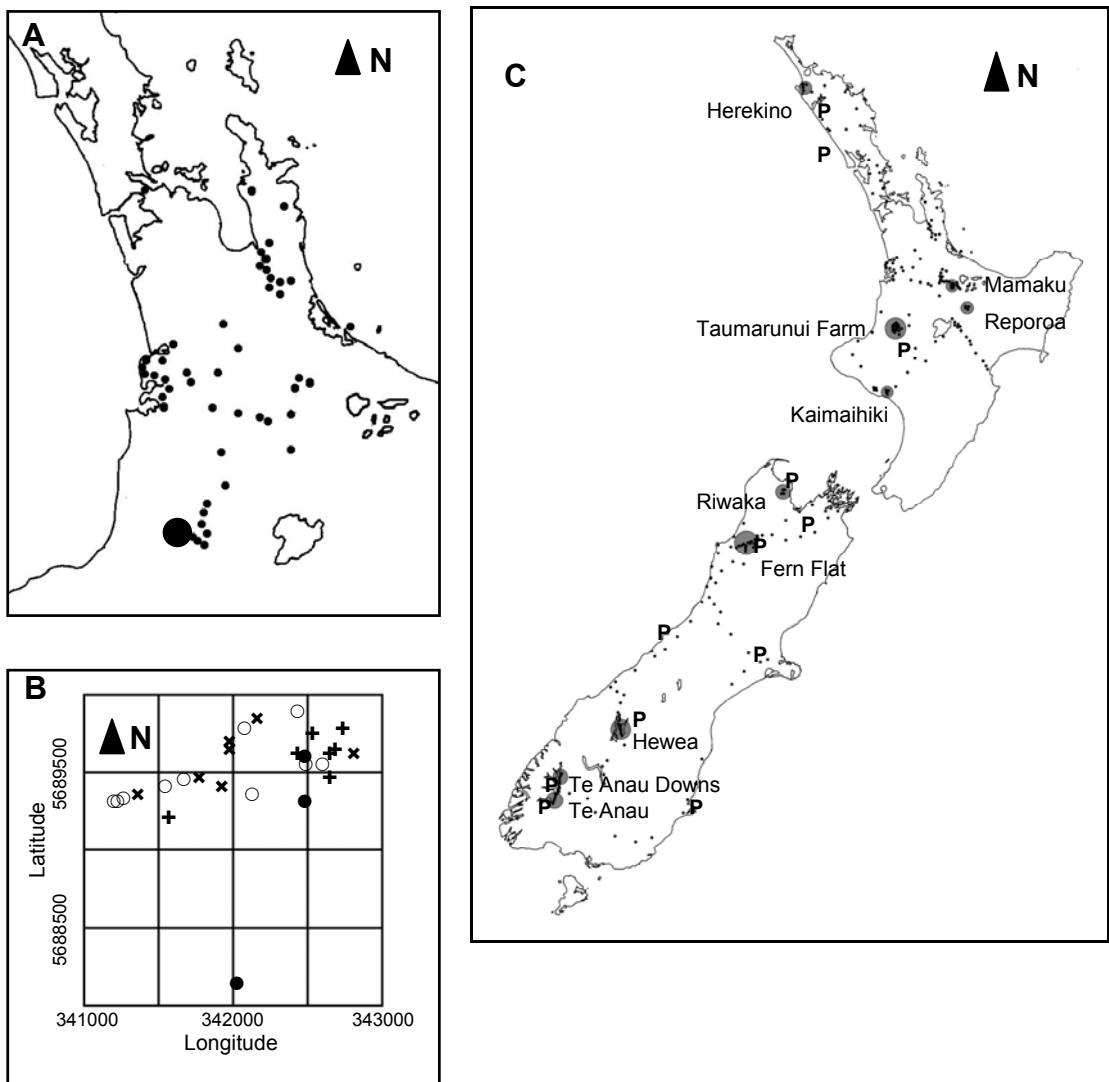


FIG. 1. A: Map of the North Island, Waikato and Coromandel Regions. Investigated sites in the Regional Survey are marked on the map (●) as well as the Taumarunui farm area (●). B: Map of the investigated area at the farm at Taumarunui, North Island, showing the location of the investigated sites (UTM 60H) and the ptaquiloside levels encountered ( $\mu\text{g/g}$ ): ○ = 0; ✕ = 1-999; + = 1,000-9,999; ● = >10,000. C: Map of New Zealand. Investigated sites in the National Survey are marked on the map (●) as well as BEH-sites included in the survey (●). P indicates bracken stands with ptaquiloside.

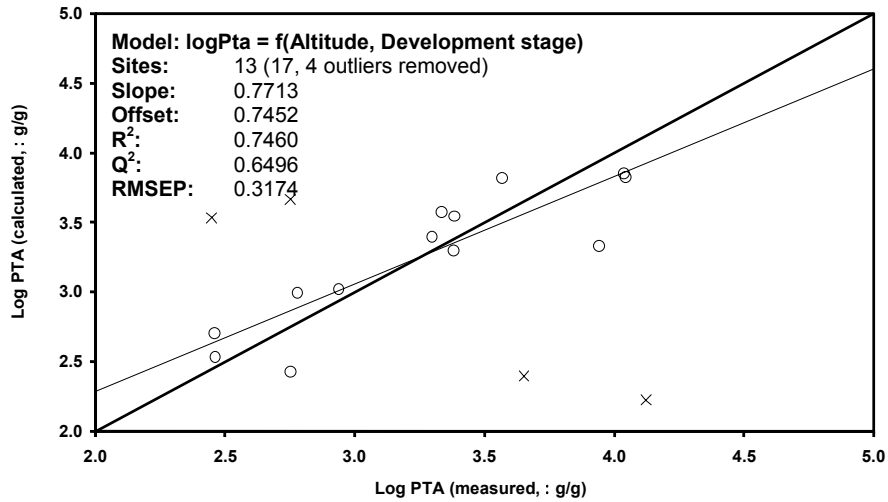


FIG. 2. PLSR-model of the ptaquiloside content in young fronds, Farm Survey (77% explained calibration Y-variance, 70% explained validation Y-variance, 1 PC (principal component)). ○: Samples used for modelling. ×: Outlying samples not included in the model. Thick line:  $\text{LogPTA (measured)} = \text{LogPTA (calculated)}$ . Thin line: Model trend line (offset and slope shown).  $R^2$  = the variance in  $\text{LogPTA (calculated)}$  according to the model.  $Q^2$  = the variance in  $\text{LogPTA (calculated)}$  according to cross validation. RMSEP = root mean square error of prediction.

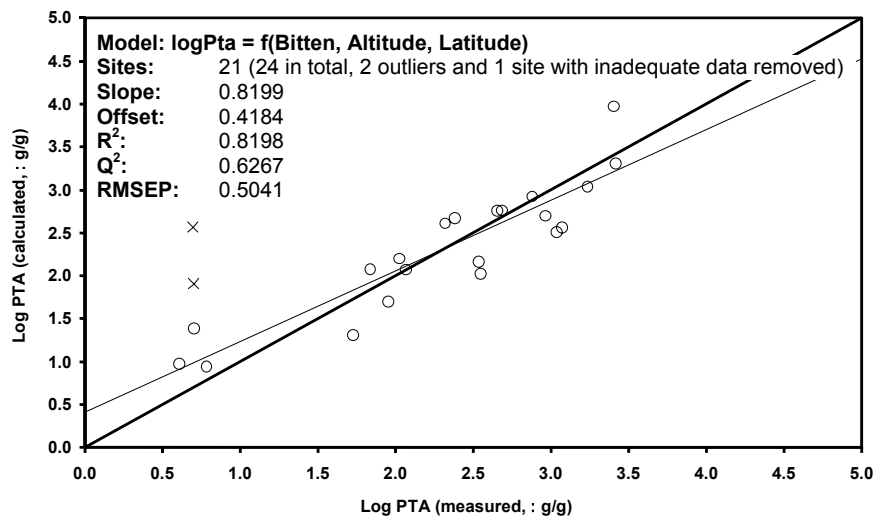
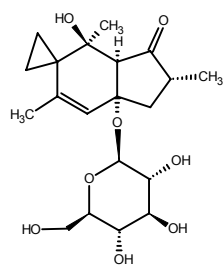


FIG. 3. PLSR-model of the ptaquiloside content in young fronds, National Survey (82% explained calibration Y-variance, 78% explained validation Y-variance, 2 PC's).  $\circ$  Samples used for modelling.  $\times$ : Outlying samples not included in the model. Solid line:  $\text{LogPTA}(\text{measured}) = \text{LogPTA}(\text{calculated})$ . Thin line: Model trend line.  $R^2$  = the variance in  $\text{LogPTA}(\text{calculated})$  according to the model.  $Q^2$  = the variance in  $\text{LogPTA}(\text{calculated})$  according to cross validation. RMSEP = root mean square error of prediction.

TABLE 1. Properties of ptaquiloside.

CAS numbers: 87625-62-5  
88825-03-0  
Chemical name: Ptaquiloside

Chemical structure:



Molecular formula:  $C_{20}H_{30}O_8$   
Molecular weight:  $398.45 \text{ g mol}^{-1}$



TABLE 2. Stand descriptions and ptaquiloside contents, Regional survey.

Stand	Elevation m	Frond length cm	Growth stage pinnae	Ptaquiloside content µg/g	Site description						
					Land element	Aspect	Vegetation	Stand size m <sup>2</sup>	Bracken cover %	Light conditions	Growth stage
R1	N.D.	200	N.D.	0	Road side	N.D.	Grass-Scrub	50 - 200	80 - 100	Open land	Old
R2	N.D.	150	N.D.	0	Road side	N.D.	Grassland	50 - 200	40 - 80	Open land	Mature
R3	N.D.	150	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R4	N.D.	50	N.D.	0	Slope	East	Scrub	< 50	< 15	Shaded	Mature
R5	N.D.	15	N.D.	40±0	Hill ridge	None	Grassland	N.D.	N.D.	Open land	Mature
R6	N.D.	150	N.D.	30±10	Hill ridge	None	Grassland	> 500	80 - 100	Open land	Mature
R7	N.D.	150	N.D.	0	Valley	N.D.	Scrub	> 500	80 - 100	Open land	Mature
R8	N.D.	150	N.D.	0	Valley	N.D.	Scrub	> 500	80 - 100	Open land	Mature
R9	N.D.	200	N.D.	6±1	Valley	South	Scrub	> 500	80 - 100	Open land	Mature
R10	N.D.	200	N.D.	139±36	Valley	N.D.	Scrub	> 500	N.D.	Open land	Mature
R11	N.D.	200	N.D.	0	Valley floor	N.D.	Scrub	< 50	N.D.	Open land	Mature
R12	N.D.	200	N.D.	2±0	Valley floor	N.D.	Forest	< 50	< 15	Shaded	Old
R13	N.D.	200	N.D.	0	Valley floor	East	Forest	< 50	80 - 100	Open land	Mature
R14	N.D.	150	N.D.	1198±57	Road side	South	Scrub	< 50	80 - 100	Open land	Mature
R15	N.D.	150	N.D.	0	Slope	N.D.	Scrub	< 50	< 15	Shaded	Old
R16	N.D.	150	N.D.	0	Slope	N.D.	Grass-Scrub	50 - 200	N.D.	Open land	Mature
R17	N.D.	200	N.D.	0	Road side	N.D.	Grassland	200 - 500	80 - 100	Open land	Mature
R18	N.D.	150	N.D.	0	N.D.	N.D.	N.D.	N.D.	N.D.	N.D.	N.D.
R19	N.D.	150	N.D.	0	Road side	N.D.	Grassland	< 50	40 - 80	Open land	Mature
R20	N.D.	N.D.	N.D.	0	Road side	N.D.	Grassland	< 50	40 - 80	Open land	Mature
R21	N.D.	200	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R22	N.D.	200	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R23	N.D.	200	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R24	N.D.	150	N.D.	0	Road side	N.D.	Grassland	< 50	40 - 80	Open land	Mature
R25	N.D.	150	N.D.	0	Slope	N.D.	Scrub	200 - 500	< 15	Scrubs	Old
R26	N.D.	200	N.D.	0	Road side	N.D.	Grass-Scrub	50 - 200	80 - 100	Open land	Mature
R27	N.D.	200	N.D.	0	Slope	N.D.	Forest	< 50	< 15	Scrubs	Old
R28	N.D.	100	N.D.	0	Ridge	N.D.	Grass-Scrub	200 - 500	< 15	Open land	Old
R29	217	150	N.D.	0	Slope	N.D.	Scrub	> 500	< 15	Scrubs	Old
R30	200	150	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R31	9	125	N.D.	0	Road side	North	Grassland	< 50	80 - 100	Open land	Mature
R32	40	150	N.D.	0	Road side	N.D.	Grassland	< 50	< 15	Open land	Mature
R33	40	125	N.D.	0	Road side	N.D.	Scrub	< 50	< 15	Open land	Mature

R34	42	N.D.	N.D.	0	Slope	N.D.	Scrub	> 500	15 - 40	Scrubs	Mature
R35	44	150	N.D.	0	Road side	None	Grassland	< 50	80 - 100	Open land	Mature
R36	15	100	N.D.	0	Valley	N.D.	Grass-Scrub	> 500	40 - 80	Scrubs	Mature
R37	60	80	N.D.	0	Road side	N.D.	Grassland	N.D.	40 - 80	Open land	Mature
R38	11	110	N.D.	0	Road side	South	Grassland	< 50	40 - 80	Open land	Mature
R39	8	225	N.D.	0	Slope	South	Scrub	> 500	80 - 100	Glade	Mature
R40	35	150	N.D.	0	Slope	None	Scrub	< 50	< 15	Glade	Old
R41	22	125	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R42	20	125	N.D.	0	Road side	South	Grassland	< 50	80 - 100	Open land	Mature
R43	182	175	N.D.	0	Slope	None	Grass-Scrub	> 500	40 - 80	Scrubs	Old
R44	317	175	N.D.	0	Road side	South	Grass-Scrub	< 50	40 - 80	Scrubs	Mature
R45	237	175	N.D.	0	Road side	West	Grassland	< 50	40 - 80	Open land	Mature
R46	N.D.	150	N.D.	0	Road side	N.D.	Grassland	50 - 200	80 - 100	Open land	Mature
R47	125	125	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R48	209	175	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R49	113	125	N.D.	0	Road side	South	Grassland	50 - 200	80 - 100	Open land	Mature
R50	207	225	N.D.	0	Road side	East	Scrub	< 50	80 - 100	Scrubs	Old
R51	195	90	N.D.	0	Road side	North	Grass-Scrub	< 50	80 - 100	Scrubs	Mature
R52	N.D.	200	N.D.	0	Road side	N.D.	Grass-Scrub	< 50	80 - 100	Open land	Mature
R53	N.D.	150	N.D.	0	Road side	N.D.	Grass-Scrub	< 50	40 - 80	Scrubs	Young
R54	N.D.	150	N.D.	0	Road side	N.D.	Grass-Scrub	< 50	< 15	Scrubs	Young
R55	N.D.	100	N.D.	0	Road side	N.D.	Grass-Scrub	< 50	< 15	Scrubs	Young
R56	N.D.	100	N.D.	0	Road side	N.D.	Grass-Scrub	50 - 200	80 - 100	Shaded	Mature
R57	N.D.	150	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R58	N.D.	125	N.D.	0	Road side	N.D.	Grass-Scrub	200 - 500	80 - 100	Shaded	Mature
R59	N.D.	200	N.D.	0	Road side	N.D.	Grassland	< 50	80 - 100	Open land	Mature
R60	N.D.	N.D.	N.D.	0	Road side	N.D.	Forest	< 50	40 - 80	Glade	Mature
R61	N.D.	N.D.	N.D.	0	Road side	N.D.	Grass-Scrub	< 50	80 - 100	Shaded	Mature
R62	N.D.	150	N.D.	0	Road side	N.D.	Grassland	50 - 200	80-100	Open land	Mature

N.D. Not described.

TABLE 3. Comparison of ecological variables between ptaquiloside (Pta) and non-ptaquiloside (Non-Pta) containing bracken stands, Farm Survey.

	Altitude m	2001 growth			2000 growth			Land element					Aspect				
		Fron length cm	Stipe length cm	Develop. stage pinnae	Fron length cm	Stipe length cm	Develop. stage pinnae	%					%				
		Ridge	Slope	Valley	Road	Other	North	East	South	West	None						
Non-Pta	221	45.1	N.D. <sup>¶</sup>	3.7	N.D.	N.D.	N.D.	20	80	0	0	0	30	20	20	30	0
Pta	346	45.3	N.D.	2.6	N.D.	N.D.	N.D.	29	53	12	6	0	12	18	41	24	6
All	257	45.3	N.D.	3.1	N.D.	N.D.	N.D.	26	63	7	4	0	19	19	33	26	4

	Landuse					Vegetation					Bracken cover				
	%					%					%				
	Sheep	Cows	Forestry	Unused	Other	Grassland scrub	Scrub	Grassland forest	Forest scrub	Forest	Grassland	Low	Moderate	High	Dominant
Non-Pta	100	0	0	0	0	80	0	10	10	0	0	70	30	0	0
Pta	100	0	0	0	0	41	6	18	6	0	29	29	41	18	12
All	100	0	0	0	0	55	4	15	7	0	19	45	37	11	7

	Light conditions				Bracken status			Community development stage			Bracken disturbance			Size of Bracken area			
	%				%			%			%			%			
	Open land	Scrubs	Glade	Shaded	Sup- pressed	Sub- dominant	Dominant	Young	Mature	Old	None	Bitten	Managed	< 50 m <sup>2</sup>	50-200 m <sup>2</sup>	200-500 m <sup>2</sup>	> 500 m <sup>2</sup>
Non-Pta	20	60	20	0	0	0	100	80	10	10	100	0	0	20	10	70	0
Pta	35	35	12	18	0	0	100	35	59	6	94	6	0	25	25	50	0
All	30	44	15	11	0	0	100	55	41	4	96	4	0	23	58	19	0

¶ N.D. = Not determined.

TABLE 4. Descriptions of ptaquiloside containing stands, Farm Survey.

Stand	Altitude m	Fronde length cm	Growth stage pinnae	Ptaquiloside content $\mu\text{g/g}^{\ddagger}$
F7	205	37.7±5.5	3.6±1.8	862±240
F11	204	72.1±19.1	2.0±1.1	288±19
F12	216	62.8±25.7	1.5±0.8	562±17
F14	192	113.3±14.6	1.7±0.6	4,491±2,375
F15	164	57.9±33.6	1.5±1.1	13,259±6,918
F19	231	45.3±17.5	5.4±2.9	564±117
F20	235	59.0±11.6	3.1±1.5	598±98
F22	273	45.3±18.4	1.7±1.2	286±9
F24	430	18.1±5.6	2.5±1.2	2,151±1,521
F25	386	32.0±5.8	3.9±1.7	3,684±0
F26	378	33.1±10.8	4.0±1.3	11,052±797
F29	365	49.6±14.5	3.3±1.2	2,416±111
F30	368	61.8±15.3	2.5±0.5	8,721±364
F31	379	35.8±8.5	2.3±1.3	2,395±100
F32	380	29.6±4.8	3.0±2.1	279±38
F33	379	34.1±12.9	2.4±1.0	1,978±47
F34	346	40.8±15.8	4.5±2.4	10,872±1,185

$\ddagger$  CV<sub>AVG</sub> = 19%, CV<sub>MED</sub> = 9%.

TABLE 5. Comparison of ecological variables between ptaquiloside (Pta) and non-ptaquiloside (Non-Pta) containing bracken stands, National Survey.

	Altitude m	2001 growth			2000 growth			Land element					Aspect				
		Fron	Stipe	Develop.	Fron	Stipe	Develop.	%					%				
		length	length	stage	length	length	stage	Ridge	Slope	Valley	Road	Other	North	East	South	West	None
	cm	cm	pinnac	cm	cm	pinnac											
Non-Pta	196	119.0	68.3	3.3	138.0	61.0	17.3	16	28	13	37	6	33	17	4	17	29
Pta	167	121.1	67.4	3.2	135.0	67.3	17.0	14	19	22	36	6	25	11	13	21	30
All	183	119.3	68.3	3.3	138.0	61.5	17.3	16	27	14	36	7	26	12	12	21	30

	Landuse					Vegetation					Bracken cover				
	%					%					%				
	Sheep	Cows	Forestry	Unused	Other	Grassland scrub	Scrub	Grassland forest	Forest scrub	Forest	Grassland	Low	Moderate	High	Dominant
Non-Pta	13	10	19	57	1	25	14	8	13	2	38	19	27	31	23
Pta	4	4	25	63	4	13	17	17	4	0	49	13	29	37	21
All	12	9	20	57	2	24	15	10	12	2	37	18	27	32	23

	Light conditions				Bracken status			Community development stage			Bracken disturbance			Size of Bracken area			
	%				%			%			%			%			
	Open land	Scrubs	Glade	Shaded	Sup-pressed	Sub-dominant	Dominant	Young	Mature	Old	None	Bitten	Managed	< 50 m <sup>2</sup>	50-200 m <sup>2</sup>	200-500 m <sup>2</sup>	> 500 m <sup>2</sup>
Non-Pta	48	39	6	7	8	23	69	11	86	3	73	12	15	48	27	12	13
Pta	60	16	8	16	8	25	67	8	88	4	76	12	12	42	25	29	4
All	50	36	6	8	8	23	69	10	87	3	74	12	14	46	27	15	12

TABLE 6. Description of ptaquiloside containing stands, National Survey.

Stand	Altitude m	Frond length cm	Growth stage pinnae	Ptaquiloside content <sup>¶</sup> µg/g
N227	19	83.8±22.2	4.7±2.0	53±2
N228	N.D.	94.7±43.5	1.5±1.8	455±10
N238	106	97.2±13.1	3.0±1.4	4±0
N401	278	120.0±18.7	6.0±2.1	5 N.D.
N411	241	101.7±26.3	3.4±1.4	6 N.D.
N414	167	214.7±31.2	2.3±1.4	68 N.D.
N416	146	122.2±31.9	3.8±1.9	105 N.D.
N417	153	127.8±18.3	1.0±0.6	341 N.D.
N418	179	181.2±26.3	2.7±1.5	351 N.D.
N419	172	122.5±17.3	2.2±1.0	115±8
N423	86	59.8±15.6	3.9±1.9	239 N.D.
N428	65	106.2±18.5	3.7±1.2	1,709 N.D.
N429	125	140.0±21.9	3.0±1.5	482 N.D.
N442	362	110.2±38.9	1.7±0.8	89±2
N455	221	152.5±19.8	4.2±2.1	450±35
N456	246	155.0±17.7	3.0±0.6	206 N.D.
N457	253	125.3±27.1	0.2±0.4	5 N.D.
N458	248	69.5±9.3	3.0±0.6	1,080 N.D.
N459	239	161.0±14.3	3.5±1.0	1,173 N.D.
N467	241	103.0±4.8	4.0±1.6	5±1
N471	25	56.6±22.9	1.6±1.4	2,603 N.D.
N472	19	106.3±5.7	3.8±1.0	2,530 N.D.
N480	98	148.8±12.7	5.8±1.3	748±51
N505	22	164.8±34.0	5.7±2.4	913 N.D.

N.D. Not determined. <sup>¶</sup> VC<sub>AVG</sub> = 6.

TABLE 7. Correlation between the occurrence of BEH and ptaquiloside.

BEH area	Number of stands	Stands with Pta	Median Pta-content $\mu\text{g/g}^{\text{¶}}$
Taumarunui Farm	27	17	2,151
Herekino	4	0	0
Mamaku	6	0	0
Reporoa	4	0	0
Kaimaihiki	5	0	0
Fern Flat	10	6	115
Hewea	5	1	89
Te Anau	5	3	206
Te Anau Downs	5	2	1,080
Riwaka	4	1	913

<sup>¶</sup> Median values only for ptaquiloside-containing stands.

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3 **A Refined Method for Isolation of pure Ptaquiloside**  
4 **from Bracken (*Pteridium aquilinum* (L.) Kuhn.)**

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1 ABSTRACT

2 A simplified and refined method for isolation of the carcinogen ptaquiloside from  
3 processed Bracken fern was developed. Crude ptaquiloside was obtained by extraction  
4 with water followed by stepwise purification by low-pressure liquid chromatography using  
5 various resins including Fluka Polyamide 6, Amberlite XAD-2 and the Varian Chem Elut  
6 1020 columns. Further purification of the crude ptaquiloside by preparative HPLC afforded  
7 ptaquiloside as a whitish amorphous powder with a purity of close to 100% and the gross  
8 formula  $C_{20}H_{30}O_8 \cdot \frac{2}{3}H_2O$ . The purity was confirmed by  $^1H$ - and  $^{13}C$ -NMR and mass  
9 spectrometry. UV/VIS, FTIR and FT-Raman spectra are presented. This method enables  
10 purification of large amounts of ptaquiloside in a timesaving manner, which can be used  
11 for analytical purposes as well as for toxicological testing.

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18 **Keywords:** *Cancer, HPLC, low-pressure chromatography, Raman.*

## INTRODUCTION

Bracken (*Pteridium aquilinum* (L.) Kuhn., Dennstedtiaceae) is a globally widespread fern encountered on all continents except Antarctica (1,2). Bracken consumption by cattle and sheep can result in acute intoxication, blindness and cancer among the animals, and the fern is consequently classified as carcinogenic by the International Agency of Research on Cancer (1,3-5). Ptaquiloside (Figure 1) is a carcinogenic and mutagenic nor-sesquiterpene glucoside found in high concentrations in Bracken (6-10). Ptaquiloside is likely responsible for these clinical syndromes, as administration of pure ptaquiloside to laboratory rodents, sheep and calves have reproduced the syndromes (11-13). Bracken and ptaquiloside are also suspected of causing cancer among humans as intake of young Bracken fronds in Japan and Brazil is associated with the occurrence of human gastric and oesophageal cancer (14,15). Ptaquiloside may also be consumed by humans via cow milk (16-18), or, as recently postulated, through drinking water (19-21). However, an unresolved question is if ptaquiloside is stable enough to exist in the soil environment for prolonged periods of time enabling penetration of the carcinogen into deeper soil layers from where contamination of ground and drinking water may occur. Toxicological, ecotoxicological and chemical investigations of the fate of ptaquiloside in the soil environment have been limited due to lack of larger amounts of pure, analytical grade ptaquiloside.

Ptaquiloside is relatively stable in neutral solution, however, in alkaline solution ptaquiloside forms first ptaquilosin and then an unstable dienone (Figure 1). Under acid conditions both ptaquiloside and the unstable dienone react to give pterosin B. Other compounds than pterosin B may also be formed from ptaquiloside under acid conditions (7,10).

1 No method for the synthesis of ptaquiloside has been published so far. Existing methods  
2 for purification of ptaquiloside from Bracken fern have all been quite laborious, resulting  
3 in low yields and have included use of toxic organic solvents (e.g. chloroform) (7,9,10,22-  
4 24).

5 Here, we report a simple, refined and time-efficient method for the isolation and  
6 purification of ptaquiloside from Bracken fern based on aqueous extraction, low-pressure  
7 liquid chromatography and preparative HPLC.

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## EXPERIMENTAL PROCEDURES

**General Experimental Procedures.** Figure 2 provides an overview of the different experimental procedures used for ptaquiloside purification. Analytical HPLC was performed as gradient elution with a water:acetonitrile eluent at 35°C to quantify both ptaquiloside and pterisin B (25) by using a Merck Hitachi (Tokyo, Japan) HPLC system (T-6300 Column Thermostat, L-4200 UV-VIS Detector; 655A-40 Auto-Sampler; D-6000A Interface; L-6200 Intelligent Pump) and 125 x 4 mm I.D. Merck LiChroCART<sup>®</sup> (LiChrospher<sup>®</sup> 100 RP-8, 5 µm) columns equipped with 4 x 4 mm I.D. Merck LiChroCART<sup>®</sup> (LiChrospher<sup>®</sup> 100 RP-18e, 5 µm) guard columns. Quantification was by UV at 220 nm (ptaquiloside) or 214 nm (pterisin B). Preparative HPLC was performed in isocratic mode with a water:methanol (65:35 V:V) eluent at 35°C to purify ptaquiloside with a Perkin Elmer Series 10 liquid chromatograph (Connecticut, USA) equipped with a Shimadzu SPD-10A UV-VIS Detector (Kyoto, Japan) and a Perkin Elmer LCI-100 Laboratory Computing Integrator (Connecticut, USA). The preparative column was a 250 x 10 mm I.D. Merck LiChroCART<sup>®</sup> (LiChrospher<sup>®</sup> 100 RP-8e, 10 µm) column equipped with a 10 x 10 mm I.D. Merck LiChroCART<sup>®</sup> (LiChrospher<sup>®</sup> 100 RP-8, 10 µm) precolumn. Detection was by UV at 220 nm. The UV-VIS spectrum was obtained using a Hewlett Packard – 8453 UV-VIS spectrophotometer with a 2 mm QX absorption cell, and methanol as the solvent. NMR spectra were recorded on a Bruker Avance 400 instrument in methanol-*d*<sub>4</sub>. The <sup>1</sup>H and <sup>13</sup>C NMR spectra was obtained at 400,1 and 100,6 MHz, respectively. The chemical shifts are relative to internal TMS. LC-MS with ESI ionisation was performed on a Bruker Esquire 3000+ ion trap mass spectrometer (Bruker Daltonics, Bremen, Germany) using an Xterra C18 column (Waters, Milford, USA) and a NaCl modified acetonitrile:water gradient. FTIR spectra were obtained using a Perkin Elmer

1 System 2000 Fourier Transform infrared spectrometer with a KBr beam splitter. Spectra  
2 were obtained directly using an ATR-crystal. FT-Raman spectra were obtained using a  
3 Perkin Elmer System 2000 Fourier Transform infrared spectrometer equipped with a  
4 Raman module attachment (quartz beam splitter). Excitations of the Raman spectra were  
5 made with an Nd<sup>3+</sup>/YAG laser operating at 1064 nm with a source power of 400 mW.

6 **Reagents, Materials and Standard Solutions.** HPLC grade acetonitrile and methanol  
7 were supplied by Sigma-Aldrich (Seelze, Germany), ethyl acetate from Fischer Scientific  
8 (Loughborough, U.K), and analytical grade sodium hydroxide and hydrochloric acid from  
9 Mallinckrodt Baker B.V. (Deventer, Holland). Polyamide 6 resin was supplied by Fluka  
10 (Steinheim, Switzerland), and Amberlite supplied XAD-2 resin (Supelco, Bellefonte, PA,  
11 USA). Prepacked Chem Elut 1020 columns were obtained from Varian, Inc. (Palo Alto,  
12 CA, USA). A standard of pure ptaquiloside was donated by Professor Ojika (Nagoya  
13 University, Japan), while standards of pterosin B were made from pure ptaquiloside (25).  
14 Glass-columns equipped with 28 µm polymer filters (1.0 and 2.5 cm I.D. Econo-Columns,  
15 length: 10 and 20 cm) were purchased from BIO-RAD (Hercules, CA, USA). Absorption  
16 cells used for obtaining UV-VIS spectrum was obtained from Hellma GmbH & Co KG  
17 (Müllheim, Germany). Methanol-*d*4 used for NMR was supplied by Cambridge Isotope  
18 Laboratories, Inc. (Andover, USA).

19 **Quantification of Ptaquiloside and Pterosin B.** The ptaquiloside content in crude  
20 Bracken extracts was quantified by analytical HPLC after removal of interfering  
21 substances with Polyamide 6 resin following the method of Agnew and Lauren (25). The  
22 extract (4 mL) was passed through 0.50 g Polyamide 6 resin (dry packed in a 1.0 cm I.D.  
23 glass column). Ptaquiloside in the crude samples was then converted to pterosin B by  
24 treatment of a 1 mL aliquot of the purified extract with 0.075 mL 1 M NaOH, heating at  
25 45°C in sealed autosampler vials for 60 min., followed by addition of 0.075 mL 5 M HCl

1 (25). This method ensures full conversion of ptaquiloside to pterosin B. The stability of  
2 ptaquiloside and pterosin B standards was evaluated each working day. Ptaquiloside  
3 standards were stable for a single working day while standards of pterosin B were stable  
4 for at least 5 days.

5 The purity of extracted and purified ptaquiloside (crude ptaquiloside and pure  
6 ptaquiloside) was quantified by analytical HPLC both directly and after conversion to  
7 pterosin B. These quantifications were performed without prior Polyamide treatment.

8 **Harvest and Sample Pretreatment.** Pinnae (approx. 10 kg) were harvested 12 August  
9 2002 from mature Bracken (*Pteridium aquilinum* var. *aquilinum*) growing at Præstø Fed  
10 approx. 60 km south of Copenhagen, Denmark, at 2-4 m above sea level (20). The pinnae  
11 were dried at room temperature for 7 days before being milled into a powder (particle  
12 diameter less than 2 mm). The resulting Bracken powder had a water-extractable  
13 ptaquiloside content of approx. 2,150  $\mu\text{g g}^{-1}$  and was stored at  $-20^{\circ}\text{C}$ .

14 **Extraction of Bracken Materials and Isolation of Crude Ptaquiloside by Low-**  
15 **Pressure Liquid Chromatography.** Aqueous extraction of ptaquiloside from Bracken  
16 powder was carried out by shaking 60 g Bracken powder with 900 mL triple deionised  
17 water for 60 min on a flat bed shaker (approx. 200 shakes per min.). The plant material was  
18 then separated from the aqueous solution by centrifuging at 11.200 g for 15 min. Aqueous  
19 extract retained by the fern materials fern materials was expelled by squeezing the  
20 material. This liquid was combined with the supernatant resulting in a total volume of  
21 approx. 600 mL crude extract. Aliquots of 50 mL crude extract were then passed through  
22 twelve dry packed Polyamide 6 columns (5.00 g resin in 1.0 cm I.D. glass columns). The  
23 combined eluates (approx. 550 mL) were loaded onto 100 mL of pre-conditioned  
24 (methanol washed) and wet packed (3-times deionised water) XAD-2 resin (2.5 cm I.D.  
25 glass column) at a rate of 360 mL hour<sup>-1</sup>. After loading of the column, the XAD-2 resin

1 was washed three times with 50 of mL 3-times deionised water before elution of the  
2 column by addition of three times 50 mL methanol. The column was not allowed to run  
3 dry at any moment. The methanol extract was evaporated until dryness under vacuum  
4 using a rotary evaporator (45°C). The resulting gum was dissolved in 10 mL 3-times  
5 deionised water and passed through a dry packed Polyamide 6 column (2.00 g resin in a  
6 1.0 cm I.D. glass column) directly onto a dry Chem Elut CE1020 column. The polyamide  
7 column was rinsed with a further 2 x 3 mL 3-times deionised water, collecting the total  
8 eluate in the Chem Elut column. The Chem Elut column was finally eluted with 2 x 100  
9 mL of ethyl acetate and this eluate was evaporated to dryness under vacuum with a rotary  
10 evaporator (45°C). The resulting gum was dissolved in small amounts of ethyl acetate and  
11 transferred to glass storage vials with PTFE-caps before being evaporated till dryness by  
12 flushing with N<sub>2</sub>. The resulting crude ptaquiloside (Figure 3A) was stored at -20°C in  
13 sealed vials in a desiccator over silica gel. The entire preparation of crude ptaquiloside  
14 lasted one working day. Up to 120 mg of crude ptaquiloside could be produced in this time  
15 from the Bracken material obtained in this study.

16 **Purification of Ptaquiloside from Crude Ptaquiloside by Preparative HPLC.** One  
17 batch of crude ptaquiloside produced as described above was dissolved in 6 mL of 3-times  
18 deionised water. Preparative HPLC separation was performed with 1 mL injections.  
19 Ptaquiloside containing fractions from each separation run was combined to give approx.  
20 60 mL water-methanolic solution of ptaquiloside (65:35 V:V). Aliquots of 20 mL were  
21 then added to each of three Chem Elut CE1020 columns. Ptaquiloside was recovered from  
22 the Chem Elut columns with two times 100 mL ethyl acetate. The eluates were combined  
23 and evaporated to dryness under vacuum using a rotary evaporator at 45°C. The resulting  
24 gum was dissolved in ethyl acetate, transferred to a glass sample vial and evaporated till  
25 dry with a stream of N<sub>2</sub> giving ptaquiloside as a gum. When ptaquiloside powder was

1 needed, the gum was dissolved in 5 mL 3-times deionised water and frozen immediately  
2 before being freeze-dried. Ptaquiloside gum and powder (Figure 3B) were stored in sealed  
3 glass sample vials with PTFE caps in N<sub>2</sub>-atmosphere at -20°C. The purification of  
4 ptaquiloside from crude ptaquiloside lasted one working day.

5 **Confirmation of Ptaquiloside Structure by NMR.** Pure ptaquiloside was dissolved in  
6 methanol-*d*<sub>4</sub>, and the identity confirmed by <sup>1</sup>H and <sup>13</sup>C NMR spectra (Table 1 and 2). The  
7 spectra proved similar to previously published NMR spectra (7,8,10,22). No signs of  
8 impurities could be detected in the <sup>13</sup>C spectra.

9 **LC-MS of Purified Ptaquiloside.** Pure ptaquiloside was dissolved in 3 times deionized  
10 water and the mass spectrum obtained by electro spray ionization. Mass spectrum (ESI) of  
11 ptaquiloside, *m/z*: 421 [M+Na<sup>+</sup>]. The spectrum confirmed the identity of the compound.

12 **Determination of UV-VIS, Infrared, and Raman Spectra of Purified Ptaquiloside.**

13 Pure ptaquiloside was dissolved in methanol resulting in a concentration of ptaquiloside  
14 of 3.65 mg mL<sup>-1</sup> and the UV-VIS absorption spectrum obtained in the range 190-1100 nm.

15 FTIR-spectra were obtained from pure ptaquiloside directly using an ATR-crystal. 256  
16 scans were obtained for the spectrum at 4 cm<sup>-1</sup> spectral resolution from 4000-550 cm<sup>-1</sup>.

17 FT-Raman spectra were obtained directly from pure ptaquiloside at 4 cm<sup>-1</sup> spectral  
18 resolution from 3600-200 cm<sup>-1</sup> (64 scans in total).

19 **Determination of Stability for Purified Ptaquiloside stored as a Gum.**

20 A stock solution of pure ptaquiloside dissolved in ethyl acetate was produced to give a  
21 ptaquiloside-concentration of 65.9±5.2 mg L<sup>-1</sup>, then 32 autosampler vials, each containing  
22 40 µl of stock solution, were prepared and evaporated until dry by a stream of N<sub>2</sub>. The  
23 vials were kept at -18° C in a desiccator. On each day of analysis (Figure 6), the content of  
24 ptaquiloside was determined by dissolving the content of four vials in 1 mL of 3-times  
25 deionised water. Ptaquiloside was converted to pterosin B in two of the vials, and the two



1 remaining vials left unaltered to give a measurement of the background conversion of  
2 ptaquiloside to pterosin B. All four vials were quantified using analytical HPLC with an  
3 external standard of pterosin B. This procedure was repeated seven times during a time  
4 span of 116 days and the stability of purified ptaquiloside determined.

5

## RESULTS

**Purification Technique.** Crude ptaquiloside in the form of a yellowish transparent gum was prepared successfully by low-pressure liquid chromatography. The yield was 90-100% of the ptaquiloside content in the crude Bracken extract when quantified by HPLC by use of an external standard. The purity was approx. 70% as determined by HPLC and quantified on a weight basis (Figure 3A). Preparation of crude ptaquiloside from other specimens of Bracken than those used in this study revealed that the purity of crude ptaquiloside produced by this method depends on the initial ptaquiloside content in the ferns harvested as well as the content of interfering substances with solubility properties similar to ptaquiloside. Crude ptaquiloside purities between 20 and 80% have been obtained by this method from different batches of Bracken collected at different sites and times of the year.

Pure ptaquiloside was obtained by preparative HPLC in combination with low-pressure liquid chromatography (Figure 3B). The yield was 15-20% of the ptaquiloside content in the dried and milled bracken fern. Ptaquiloside was isolated either as a transparent gum after evaporation of ptaquiloside dissolved in ethyl acetate or as a whitish transparent amorphous powder after freeze-drying an aqueous solution of ptaquiloside. The purity was 100% ( $107 \pm 5$ ). Clarification of the purity and identity of ptaquiloside by elemental analysis, LC-MS,  $^1\text{H}$  and  $^{13}\text{C}$  NMR (Tables 1 and 2) revealed that water was bound to ptaquiloside resulting in a gross formula for ptaquiloside of  $\text{C}_{20}\text{H}_{30}\text{O}_8 \cdot \frac{2}{3}\text{H}_2\text{O}$ .

The UV-VIS spectrum of pure ptaquiloside dissolved in methanol had 3 absorbance maxima,  $\lambda_{\text{max}}$  nm: 222 ( $\epsilon$  1810); 259 ( $\epsilon$  1060); and 303 ( $\epsilon$  250). The spectrum was in accordance with previously published spectra. However, the absorbance at 303 nm was weaker than reported earlier (24).

1 The FTIR spectrum (Figure 4) was characterized by the following absorbance maxima,  
2  $cm^{-1}$ : 3,394 (broad); 2,970; 2,932; 2,874; 1,720; 1,643; 1,448; 1,383; 1,279; 1,196; 1,160;  
3 1,074 (very strong); 1,040 (very strong); 933; 889. The spectrum was rather similar to  
4 previously reported spectra. However, new assignments were made in the range 3,000-  
5 2,800  $cm^{-1}$ , likely originating from  $sp^3$  C-H stretch (7,9,26,27).

6 The FT-Raman spectrum (Figure 5) was characterized by the following transmittance  
7 maxima,  $cm^{-1}$ : 3,086; 2,975; 2,932 (very strong); 2,874 (very strong); 1,717; 1,642; 1,454;  
8 1,374; 1,275; 601. This is the first published FT-Raman spectrum of ptaquiloside.

9 Ptaquiloside as a gum was stored in sealed vials at  $-20^{\circ}C$  in  $N_2$ -atmosphere. Ptaquiloside  
10 was stable for the first 50 days (Figure 6). Signs of degradation were observed after 68  
11 days (2.6%). After 116 days 17.6% of the initial amount had degraded.

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## DISCUSSION

Several methods for recovery of ptaquiloside from Bracken already exist. However, these methods are all rather laborious, include the use of highly toxic chemicals or have low yield. No methods have so far been published for the synthesis of ptaquiloside. The method described in this paper yields high purity ptaquiloside by a combination of low-pressure liquid chromatography and preparative HPLC in a timesaving manner with limited use of organic solvents. If a lower quality of ptaquiloside is acceptable for analytical work or experimental work, the crude ptaquiloside can be used directly. This quality can be produced in large quantities very quickly.

The purity of ptaquiloside produced by this method is very high and comparable to existing methods as demonstrated by LC-MS and NMR-spectroscopy (Figure 4, Tables 1 and 2). The mass spectrum matched previously reported spectra, and the  $^1\text{H}$  and  $^{13}\text{C}$  NMR chemical shifts for ptaquiloside were similar to existing data.

Spectral analysis of ptaquiloside by UV-VIS and FTIR methods confirmed previously published spectra, although new peaks were observed in the range  $3,000\text{-}2,800\text{ cm}^{-1}$ . The FT-Raman spectrum was characterized by distinct peaks between  $3,086$  and  $2,874$ .

## CONCLUSION

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2  
3 In summary, a rapid method of producing a high yield of approx. 70% pure ptaquiloside  
4 has been developed. This has then led to the production of pure production as a gum or as  
5 an amorphous powder using a refined and simplified method. Ptaquiloside can be rapidly  
6 produced in sufficient amounts for use in soil chemical, toxicological and ecotoxicological  
7 studies. Compared to previously published methods for ptaquiloside production, this  
8 refined method facilitates production of large amounts of high-quality ptaquiloside in a  
9 time-efficient manner with less use of harmful chemicals. Full FTIR- and FT-Raman  
10 spectra were recorded. Ptaquiloside produced in this manner was stable for 50 days when  
11 stored at -18°C.

## ABBREVIATIONS USED

<sup>1</sup>H NMR, proton nuclear magnetic resonance; <sup>13</sup>C NMR, carbon nuclear magnetic resonance; ATR, attenuated total reflection; COSY, correlated spectroscopy; DMF, dimethylformamide; TMS, tetra methyl silane; I.D., inner diameter; FT, Fourier transform; FTIR, Fourier transform infrared spectrometry; HPLC, high-performance liquid chromatography; MS, mass spectrometry; LC-MS, liquid chromatography-mass spectrometry; ESI, electrospray ionization; *m/z*, mass-to-charge ratio; PTFE, poly(tetrafluoroethylene); UV-VIS, ultraviolet-visible range; V, volumetric percentage.

*Safety*-Because of the presence of carcinogens in Bracken, respiration protection-gear must be used while working with powdered Bracken. Plant materials and extracts should be handled wearing gloves.

*Acknowledgement*-Professor Ojika (Nagoya University, Japan) is gratefully thanked for his donation of ptaquiloside making this study possible.

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## FIGURE CAPTIONS

<b>Table no:</b>	<b>Text:</b>
<b>1</b>	<b>Table 1.</b> $^1\text{H}$ NMR Chemical Shifts ( $\delta$ in ppm relative to TMS, MeOD) and Coupling Constants (Hz) of Ptaquiloside.
<b>2</b>	<b>Table 2.</b> $^{13}\text{C}$ NMR Chemical Shifts ( $\delta$ in ppm relative to TMS, MeOD) of Ptaquiloside.
<b>Figure no:</b>	<b>Text:</b>
<b>1</b>	<b>Figure 1.</b> Ptaquiloside [1] reactions. During alkaline conditions, first ptaquilosin [2] and then the unstable dienone [3] are formed. Both [1] and [3] yield pterosin B [4] under acid conditions. Other compounds than pterosin B may also be formed from ptaquiloside under acid conditions. The unstable dienone may alkylate DNA [5], and hence be responsible for the mutagenicity and carcinogenicity associated with ptaquiloside.
<b>2</b>	<b>Figure 2.</b> Overview of ptaquiloside purification.
<b>3</b>	<b>Figure 3.</b> HPLC chromatograms of (A) crude ptaquiloside (purity: 40%) and (B) pure ptaquiloside (purity: 105 $\pm$ 11%) using UV 220 nm. Chromatogram A obtained at ambient temperature while B was obtained at 35°C.
<b>4</b>	<b>Figure 4.</b> FTIR spectrum of pure ptaquiloside (purity 105 $\pm$ 11%).
<b>5</b>	<b>Figure 5.</b> Raman spectrum of pure ptaquiloside (purity 105 $\pm$ 11%).
<b>6</b>	<b>Figure 6.</b> Stability of ptaquiloside stored at -18°C (initial concentration: 65.9 $\pm$ 5.2 mg L <sup>-1</sup> ). Standard error bars shown.

**Table 1.** <sup>1</sup>H NMR Chemical Shifts (δ in ppm relative to TMS, MeOD) and Coupling Constants (Hz) of Ptaquiloside.

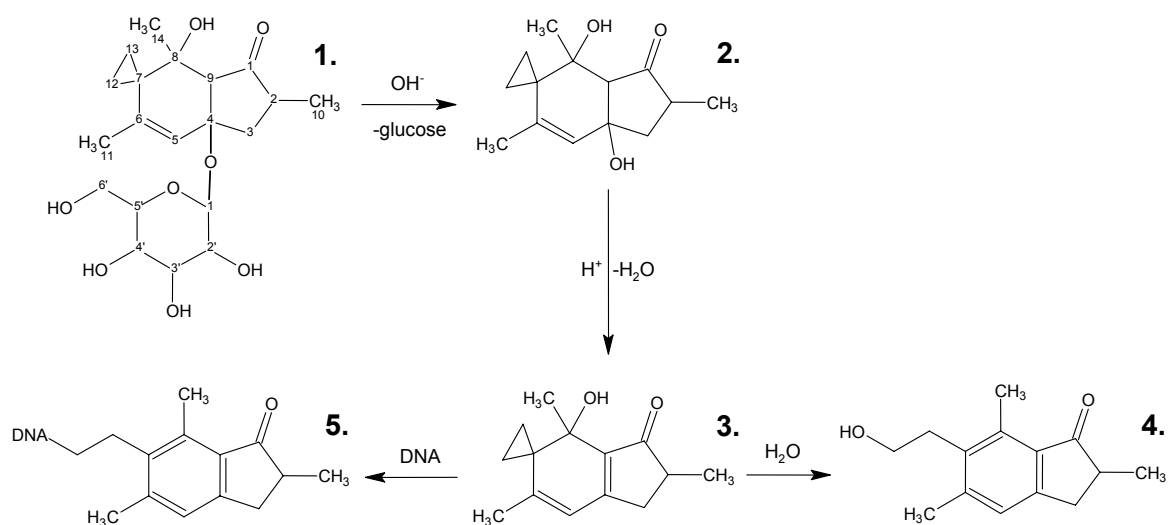
signal	chemical shift <sup>a</sup>	coupling constants
H-2	2.17-2.28 (m)	
H-3a	1.93 (t)	12.7
H-3b	2.50 (dd)	8.2; 12.4
H-5	5.76 (q)	1.3
H-9	2.64 (d)	1.3
H-10	1.07 (d)	6.9
H-11	1.53 (d)	1.2
H-12a -H-13b	0.45-0.90 (m)	
H-14	1.28 (s)	
H-1'	4.60 (d)	7.7
H-2'	3.20 (dd)	7.7; 9.1
H-3' - H-5'	3.26-3.38 (m)	
H-6'a	3.65 (dd)	5.7; 11.8
H-6'b	3.89 (dd)	1.9; 12.0

m: multiplet; s: singlet; d: doublet; t: triplet; q: quintet;  
dd: doublet-doublet

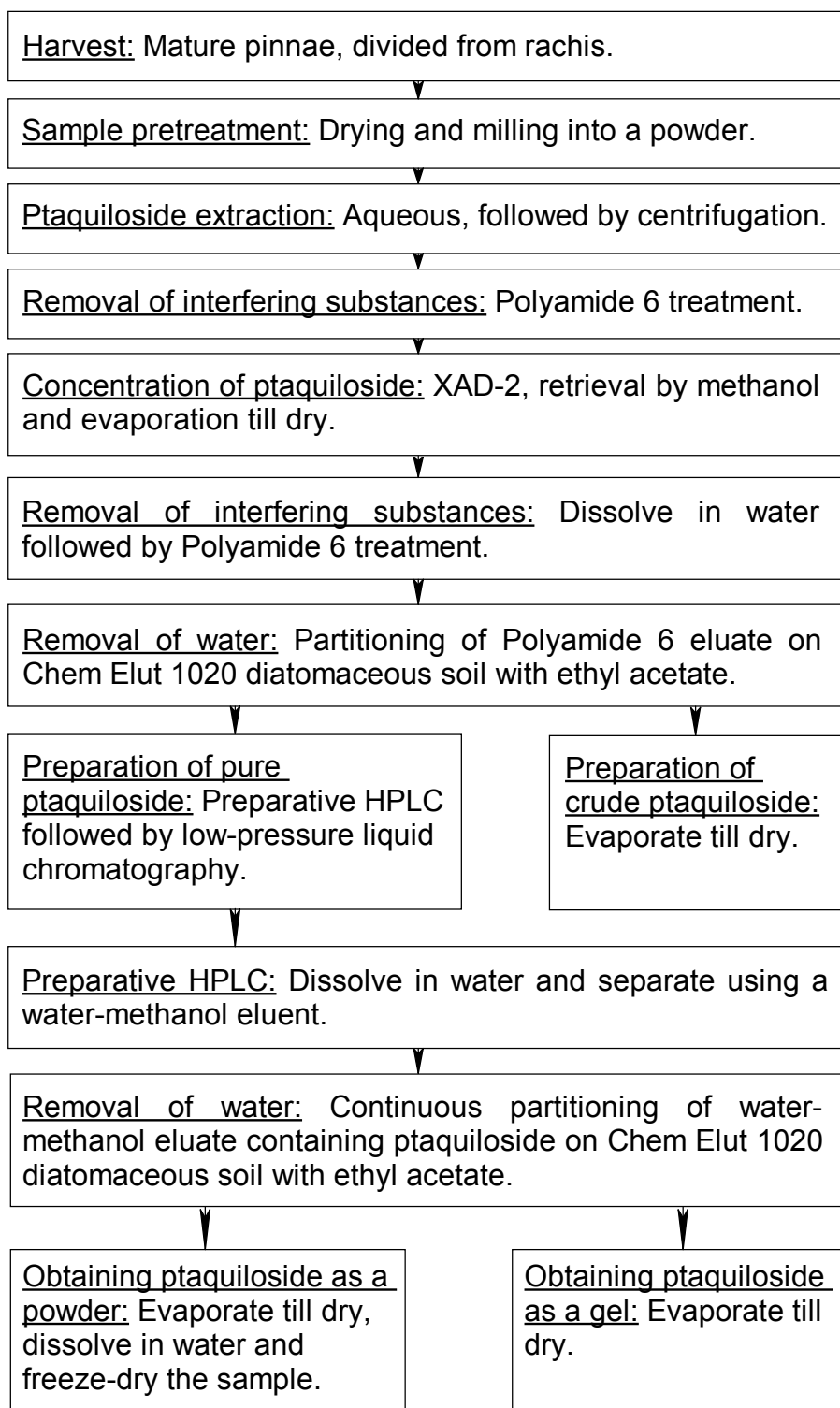
**Table 2.**  $^{13}\text{C}$  NMR Chemical Shifts ( $\delta$  in ppm relative to TMS, MeOD) of Ptaquiloside.

signal	chemical shift <sup>a</sup>
C-1	224.1 (s)
C-2	45.3 (d)
C-3	45.3 (t)
C-4	82.1 (s)
C-5	123.2 (d)
C-6	144.5 (s)
C-7	30.1 (s)
C-8	71.9 (s)
C-9	62.5 (d)
C-10	13.6 (q)
C-11	19.5 (q)
C-12	5.9 (t)
C-13	10.6 (t)
C-14	27.0 (q)
C-1'	99.4 (d)
C-2'	75.3 (d)
C-3'	77.8 (d)
C-4'	71.9 (d)
C-5'	78.2 (d)
C-6'	63.0 (t)

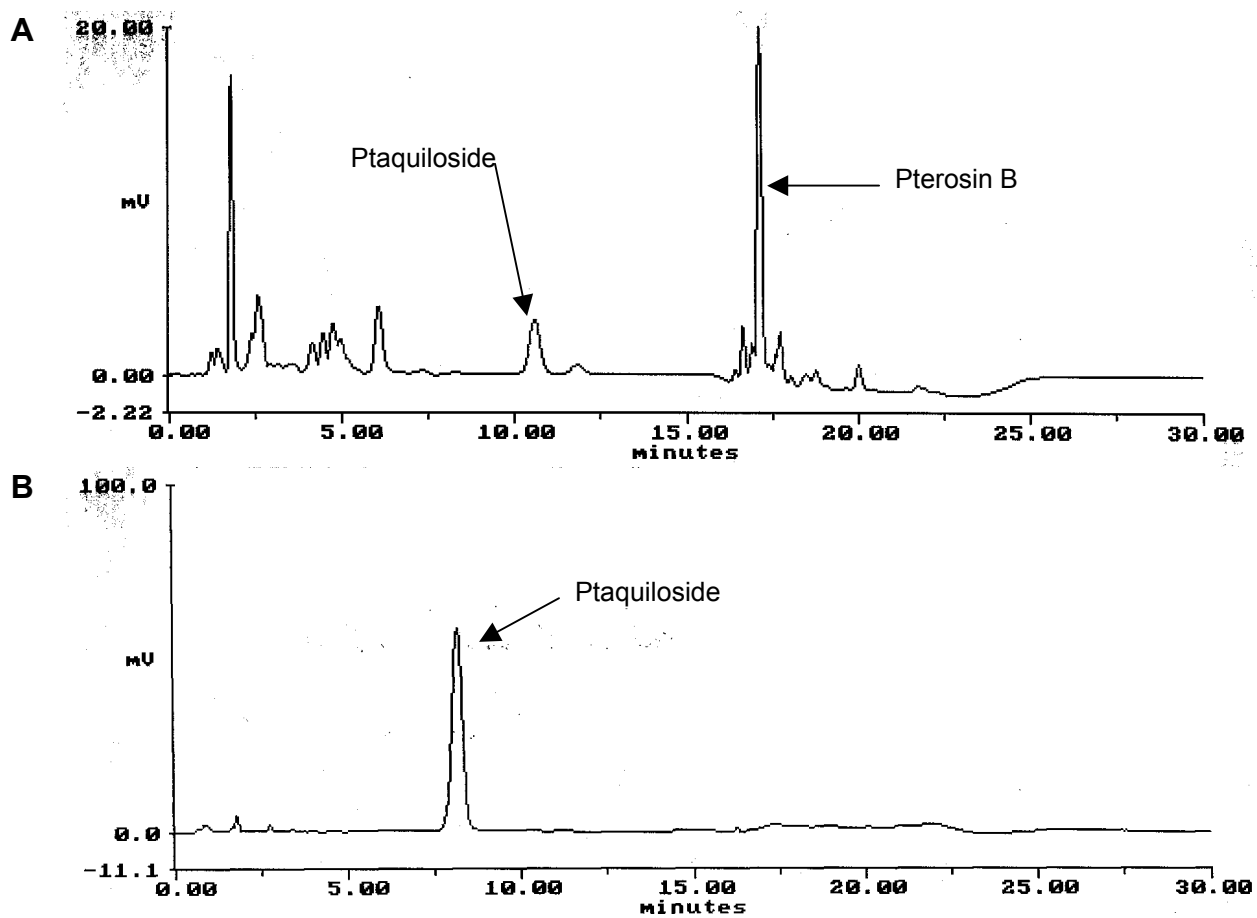
<sup>a</sup> Nomenclature: C (s), CH (d), CH<sub>2</sub> (t), or CH<sub>3</sub> (q) given in brackets.



**Figure 1.** Ptaquiloside [1] reactions. During alkaline conditions, first ptaquilosin [2] and then the unstable dienone [3] are formed. Both [1] and [3] yield pterisin B [4] under acid conditions. Other compounds than pterisin B may also be formed from ptaquiloside under acid conditions. The unstable dienone may alkylate DNA [5], and hence be responsible for the mutagenicity and carcinogenicity associated with ptaquiloside.

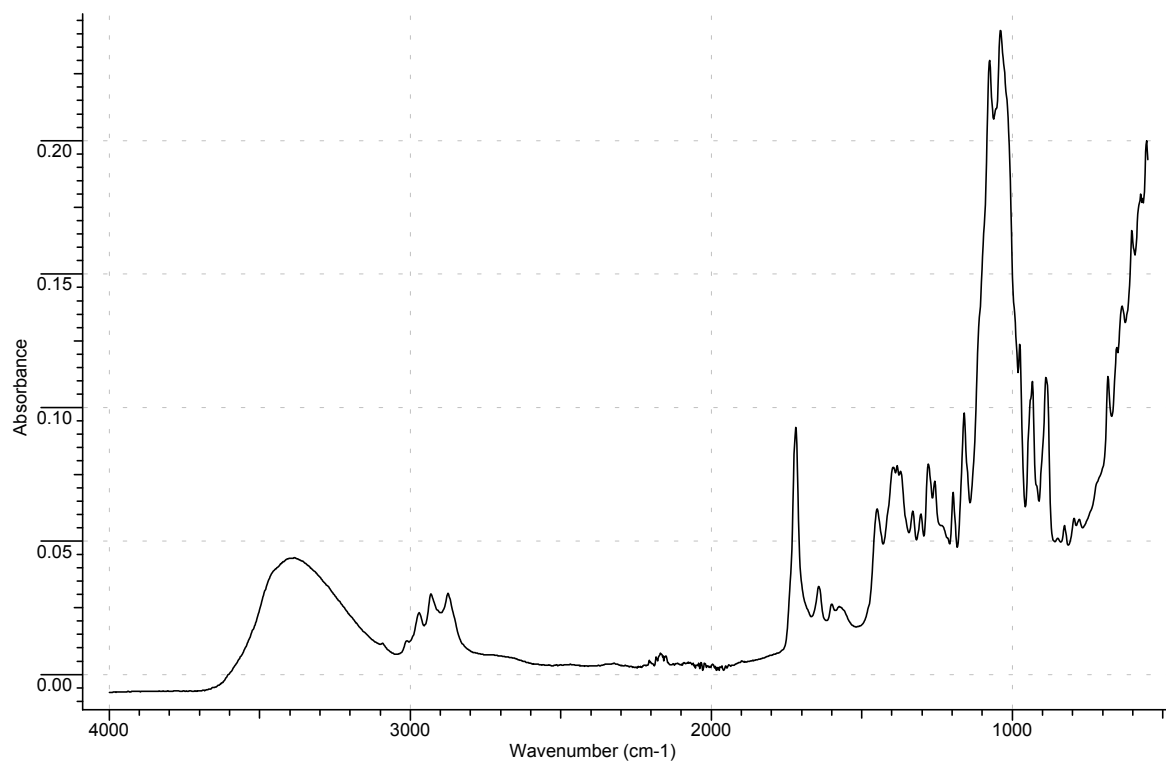


**Figure 2.** Overview of ptaquiloside purification.

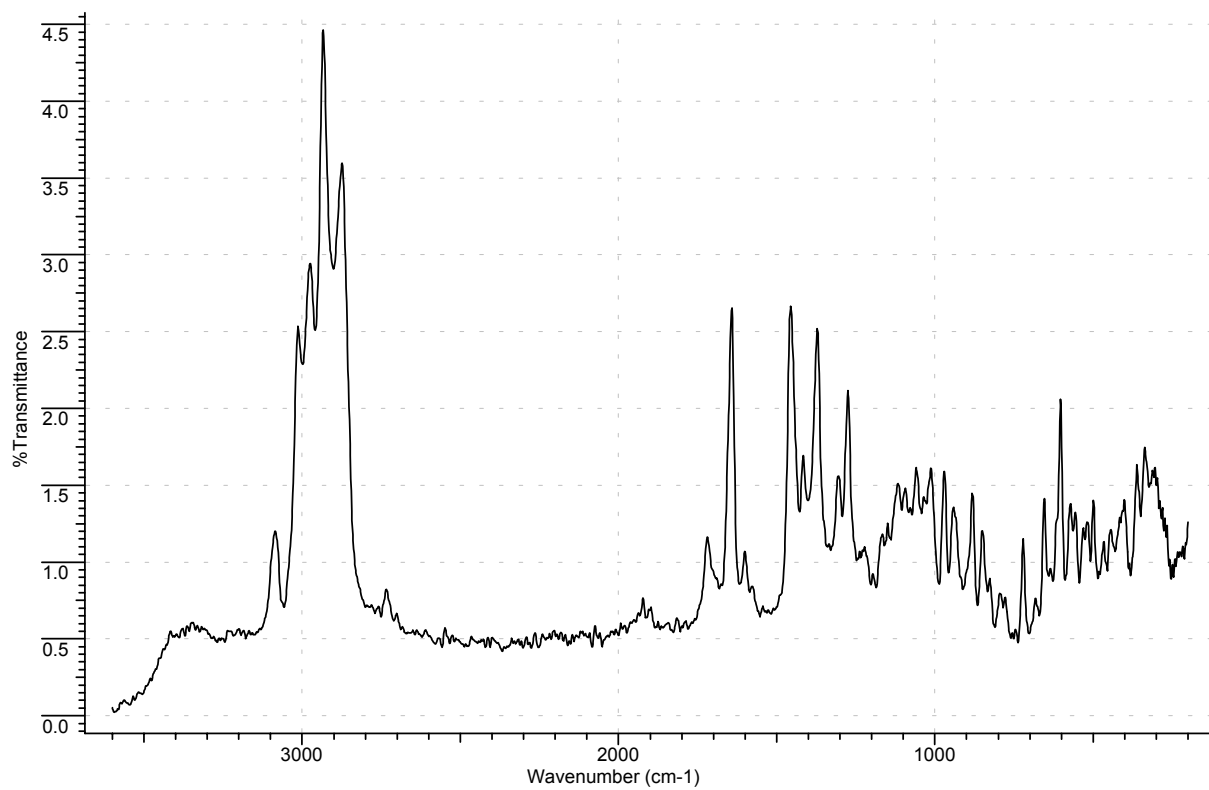


**Figure 3.** HPLC chromatograms of (A) crude ptaquiloside (purity: 40%) and (B) pure ptaquiloside (purity: 105±11%) using UV 220 nm. Chromatogram A obtained at ambient temperature while B was obtained at 35°C.

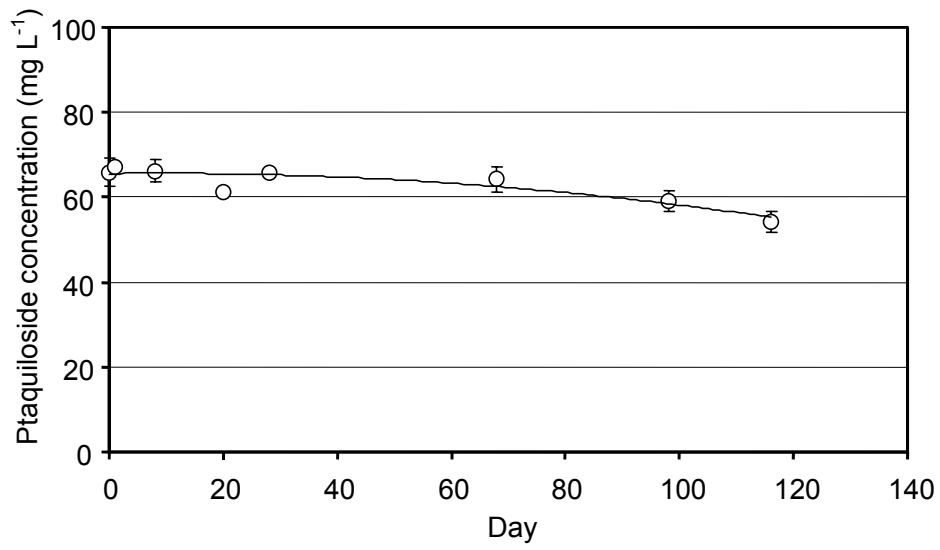




**Figure 4.** FTIR spectrum of pure ptaquiloside (purity 105±11%).



**Figure 5.** Raman spectrum of pure ptaquiloside (purity  $105 \pm 11\%$ ).



**Figure 6.** Stability of ptaquiloside stored at  $-18^{\circ}\text{C}$  (initial concentration:  $65.9 \pm 5.2 \text{ mg L}^{-1}$ ). Standard error bars shown.