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SEDIMENT BASED TURBIDITY ANALYSES FOR REPRESENTATIVE SOUTH CAROLINA SOILS

A Thesis Presented to the Graduate School of Clemson University

In Partial Fulfillment of the Requirements for the Degree Master of Science Biosystems Engineering

> by Katherine E. Resler August 2011

Accepted by: Dr. Calvin B. Sawyer, Committee Chair Dr. John C. Hayes Dr. Charles V. Privette, III

ABSTRACT

Construction activities have been recognized to have significant impacts on the environment. Excess sediment from construction sites is frequently deposited into nearby surface waters, negatively altering the chemical, physical and biological properties of the water body. This environmental concern has led to strict laws concerning erosion and sediment control, such as imposing permit conditions that limit the concentration of suspended solids that can be present in effluent water from construction sites. However, sediment concentration measurements are not routinely used to detect and correct shortterm problems or permit violations because laboratory analysis of sediment concentrations is time-consuming and costly. Nevertheless, timely, accurate field estimation of sediment loading could be facilitated through the development of empirical relationships between suspended solids and turbidity.

Previous research indicates that turbidity measurements may be a more practical method of estimating sediment loads by indirectly relating sediment concentration to turbidity. In addition, recognition of turbidity as an indicator of pollution in surface runoff from disturbed areas has resulted in efforts by the U.S Environmental Protection Agency (EPA) to implement turbidity effluent limitation guidelines to control the discharge of pollutants from construction sites. Therefore, given the importance of a proposed turbidity limit, focus of this research is to determine relationships between representative soils and corresponding turbidity as a function of suspended sediment concentration and sediment settling. Turbidity is not only a function of suspended sediment concentration, but also of particle size, shape, and composition; so this research was needed to analyze turbidity responses based on sediment characteristics of representative South Carolina soils.

First, accuracy and precision of commercially available nephelometers needed to be quantified for use in subsequent sediment/ surface water analysis and potential regulatory compliance. Analysis of accuracy and precision for instruments showed that even though meters may be very precise, they could also be inaccurate. However, three of the four meters that performed well provided statistically accurate and precise results. It was also found that formazin calibration standards may be a better standard than AMCO EPA standards for surface water analysis.

Utilizing representative South Carolina soils, both relationships of turbidity to sediment concentration and turbidity to settling time were used to form mathematical correlations. Turbidity versus suspended sediment concentration and turbidity versus settling time correlated well when top soil and subsoils were classified based on their predominant South Carolina region and their measured clay content. Derived trends for suspended sediment concentration to turbidity correlated well with either a linear or log relationship (R^2 values ranging from 0.7945 to 0.9846) as opposed to previous research utilizing a power function or the assumption of a one-to-one relationship. For the correlation of turbidity and sediment settling time, trends were well correlated with a power function (R^2 values ranging from 0.7674 to 0.9347). This relationship suggests Stoke's Law was followed; where smaller particles remain in suspension longer and contribute more to turbidity compared to soils with less clay content.

Altogether, results of this research provide a step in determining potential sitespecific equations relating sediment concentration to turbidity and sediment settling time to turbidity. With this knowledge, results could ultimately aid in the design of future sediment basins of South Carolina and provide information for potential regulatory compliance.

DEDICATION

This thesis is dedicated, first and foremost, to my parents, Dave and Susan Resler, who passed on to me an insatiable thirst for knowledge and a desire to explore and discover. Due to their unshakable faith and support, I have learned the courage and tenacity to reach my goals and to always give my all. My work is also dedicated to my siblings, Chris, Jay, Nick, and Steph. Despite their insistence of having no involvement in my endeavor, countless phone calls providing support attest otherwise. Always ready to pick up my spirits and to offer me much needed encouragement, to my siblings I owe a debt of gratitude. And finally, I dedicate this work to my grandparents. Grandma Jane and Pop-Pop provided an irreplaceable support network close to Clemson for five years, not to mention endless hospitality, and I cannot thank them enough for the multitude of ways they have seen me through this journey.

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Finally, I would like to recognize several individuals that helped with the completion of this research by either assisting with soil collection or laboratory procedures. This includes Lance Beecher, Chris Bellamy, Stephen Pohlman Jr., Laurens County NRCS, SCDNR, and my fellow graduate students Clark Chewning, Zach Smoot and Elizabeth Tempel.

TABLE OF CONTENTS

TITLE PAGE i
ABSTRACTii
DEDICATIONv
ACKNOWLEDGEMENTS vi
LIST OF TABLES ix
LIST OF FIGURESx
LIST OF ACRONYMNS xii
CHAPTER
1. INTRODUCTION1
2. LITERATURE REVIEW5A. Suspended Solids5B. Construction Sites9C. Sediment Control Structures10D. TSS Measurements15E. Turbidity17F. Turbidity Measurements17G. Future Turbidity-Based Regulations23H. Relationships to Total Suspended Solids24I. Particle Size Distributions & Particle Settling Velocities26
3. ACCURACY AND PRECISION OF PORTABLE TURBIDITY METERS. 28 A. Abstract. 29 B. Introduction. 31 C. Procedures. 35 D. Results and Discussion 44 E. Conclusions. 51

4.	TURBIDITY ANALYSES BASED ON SEDIMENT	
	CHARACTERISTICS OF REPRESENTATIVE SOUTH	
	CAROLINA SOILS	53
	A. Abstract	54
	B. Introduction	56
	C. Procedures	60
	D. Results and Discussion	73
	E. Conclusions	91
5.	SUMMARY CONCLUSIONS	94
APPENDI	CES	99
A.	Tabular data of Experiments 1-4 to determine meters accuracy	
	and precision	100
В.	Tabular particle size distribution data for selected South	
	Carolina soils	107
C.	Tabular data for selected soils' turbidity as a function of	
	suspended sediment concentration	114
D.	Tabular data for selected soils' turbidity as a function of	
	sediment settling time	129
E.	Tabular data for selected soils' turbidity as a function of	
	particle diameter	142

LITERATURE CITED

LIST OF TABLES

Table	Pa	age
3.1	Manufacturer's provided specifications for selected meters	.36
3.2	Experimental procedures summary	.36
3.3	Meter Accuracy: averaged percent difference about the true value for each meter	.46
3.4	Meter Precision: averaged percent difference about measured value for each meter.	.46
3.5	Meters' standard deviation for each field sample	.48
4.1	Representative South Carolina soils utilized for this research	.61
4.2	Pipette analysis schedule	.65
4.3	Tabular data of a soil sample's turbidity and concentration response to particle size	.72

LIST OF FIGURES

Figure	Page
2.1	SCDHEC sediment basin specifications
2.2	Basic turbidimeter design using nephelometric measuring technique19
2.3	Scanning electron microscopy (SEM) images
2.4	Graph compares the particle sizes of formazin and SDVB2
3.1	Selected nephelometers
3.2	Meter's turbidity responses to known Cecil concentrations
3.3	Turbidity measurements from each meter using field samples
4.1	South Carolina physiographic regions61
4.2	Wet sieve and pipette analysis summary photographs65
4.3	Piedmont: Cecil's aggregate size distribution (ASD)66
4.4	Central: Lakeland's aggregate size distribution (ASD)66
4.5	Coastal: Goldsboro's aggregate size distribution (ASD)67
4.6	Size distribution comparison for Cecil subsoil
4.7	Hydrometer analysis photographs69
4.8	All twenty five selected top soil and subsoil's relationship of turbidity and suspended sediment concentration
4.9	Example of a subsoil that aggregated and settled quickly compared to its top soil
4.10	Piedmont top soils relationship of turbidity with respect to concentration77
4.11	Piedmont subsoils relationship of turbidity with respect to concentration77
4.12	Central top soils relationship of turbidity with respect to concentration

4.13	Central subsoils relationship of turbidity with respect to concentration
4.14	Coastal top soils relationship of turbidity with respect to concentration79
4.15	Coastal subsoils relationship of turbidity with respect to concentration79
4.16	The twenty five selected South Carolina soil's top soil and subsoil's turbidity relationship to settling time
4.17	Turbidity as a function of settling time for Piedmont top soils
4.18	Turbidity as a function of settling time for Piedmont subsoils
4.19	Turbidity as a function of settling time for Central top soils
4.20	Turbidity as a function of settling time for Central subsoils
4.21	Turbidity as a function of settling time for Coastal top soils
4.22	Turbidity as a function of settling time for Coastal subsoils
4.23	The ratio of turbidity to concentration versus particle diameter for all soils86
4.24	Piedmont top soil's relationship of particle diameter to the ratio of turbidity to concentration
4.25	Piedmont subsoil's relationship of particle diameter to the ratio of turbidity to concentration
4.26	Central top soil's relationship of particle diameter to the ratio of turbidity to concentration
4.27	Central subsoil's relationship of particle diameter to the ratio of turbidity to concentration
4.28	Coastal top soil's relationship of particle diameter to the ratio of turbidity to concentration
4.29	Coastal subsoil's relationship of particle diameter to the ratio of turbidity to concentration90

LIST OF ACRONYMS

AASHTO: American Association of State Highway and Transportation Officials **APHA:** American Public Health Association **ASD**: Aggregate Size Distribution **ASTM:** American Society for Testing and Materials **BMP**: Best Management Practice **CWA**: Clean Water Act **ELG**: Effluent Limitation Guideline **EPA**: U.S. Environmental Protection Agency **EPSC:** Erosion Prevention and Sediment Control JTU: Jackson Turbidity Unit MCP: Measured Clay Percentage **NPDES**: National Pollutant Discharge Elimination System **NPS**: Nonpoint Source **NTU**: Nephelometric Turbidity Unit PCAL: Primary Calibration Standard **PSD**: Particle Size Distribution SCAL: Secondary Calibration Standard **SCDHEC**: South Carolina Department Health and Environmental Control **SDVB**: Styrene Divinylbenzene SS: Suspended Solids SSC: Suspended Solids Concentration SSS: Stock Standard Suspension **TSS**: Total Suspended Solids USDA: U.S. Department of Agriculture

CHAPTER 1

INTRODUCTON

Increased erosion caused by anthropogenic activities has been recognized as a priority environmental concern. Suspended solids from runoff, such as introduction through construction activity, are a potential water pollutant that can cause significant environmental impacts. Such impacts include adsorption of heavy metals, toxic substances, and biological pollutants to soil particles that are then transported to nearby waters. These chemical and biological pollutants can harm water quality and lead to fish kills and other ecological problems. This environmental concern has led to strict laws concerning erosion and sediment control. The outcome of the sediment and erosion control laws has resulted in construction sites using sediment basins as the conventional method of controlling sediment-laden runoff from these sites. Sediment basins are designed to slow the velocity of runoff and allow sediment particles to settle from the water column before discharge to surface waters offsite (Millen et al., 1997). The efficiency of these ponds is critical to controlling the amount of runoff from these sites.

Several measures can be used to determine the effectiveness of a sedimentation basin including trapping efficiency, average effluent concentration, peak effluent concentration and peak effluent settleable solids (Hoechst, 1997). Trapping efficiencies are the most common performance standard and are generally calculated with the use of sediment concentration (Mitchell, 2000). Water samples from influent and effluent of the basin must be taken to the laboratory and analyzed for sediment content. Methods to obtain the amount of solids contained in stormwater samples includes filtering water, then drying and weighing residue that has remained on a filter (Guo, 2006). Sediment concentrations from the influent and effluent waters are compared to determine the trapping efficiency of the pond using a mass balance approach (Mitchell, 2000).

Total suspended solids (TSS), resulting from erosion, are held in the water column by turbulence and encompass both inorganic solids, such as sand, silt, clay and organic solids, such as algae and detritus (Thackston and Palermo, 2000). Suspended solids measurements are not routinely used to detect and correct short-term problems or permit violations because sediment concentrations cannot be determined easily or quickly in the field, and transportation to a laboratory for analysis is time-consuming and can be costly (Thackston and Palermo, 2000). Timely, accurate field estimation of sediment loading could be facilitated through the development of precise relationships between suspended solids and turbidity. This approach has potential for monitoring any water quality constituent whose concentration is better correlated with an easily measured (in situ) parameter, such as turbidity.

Turbidity is an expression of the optical properties of a liquid that causes light rays to be scattered and absorbed rather than transmitted in straight lines through a sample (Anderson, 2005). Suspended solids have an optical impact on water quality by reducing light transmission through water; this is referred to as light attenuation (Davies-Colley and Smith, 2001). Cloudiness of water results from intense scattering of light by fine particles. Hence, waters with high concentrations of fine suspended sediment are described as turbid. Turbidity measurements are gaining increased usage as an indicator of sediment pollution in surface runoff from disturbed areas such as active construction sites. Recent efforts by the U.S. Environmental Protection Agency (EPA) have resulted in turbidity effluent limitation guidelines and new source performance standards to control the discharge of pollutants from construction sites (EPA, 2009). Numeric turbidity limits for construction site discharge are expected to be required in the near future. Such requirements will likely include subjecting construction site stormwater discharges to a maximum allowable numeric turbidity effluent limit in nephelometric turbidity units (NTU) for sites disturbing 10 acres or more. Where turbidity output readings from nephelometers are given in nephleometric turbidity units (NTU) when scattered light is measured perpendicular to meter's incident beam.

Turbidity is not an inherent property of water, such as temperature or pH. However, the recognition of turbidity as an indicator of the environmental health of water bodies has increased, resulting in a growing demand for high-quality and objective turbidity measurements (Anderson, 2005). Therefore, given the importance of a proposed turbidity limit, the focus of this research is to better understand the relationship between turbidity and suspended sediments. To fulfill this goal, three objectives were established. These objectives are described below.

- 1. Compare selected turbidity meters to quantify accuracy and precision of each instrument for use in subsequent soil/water analysis objectives.
- 2. Establish empirically derived relationships between suspended solids concentration and turbidity for representative South Carolina soils.

3

 Determine the correlation of turbidity with respect to settling time of the selected South Carolina soils.

Before relationships between turbidity and suspended sediments could be established, questions remain as to how samples should be obtained and processed to get reliable readings for future analysis. Therefore, this research will first compare the performance of various commercially available nephelometers to statistically quantify accuracy and precision of selected instruments for use in subsequent analysis and potential regulatory compliance.

A standard practice in relating suspended sediment to turbidity is to take the association as a one-to-one relationship (Hayes et al., 2001). Prior research, however, indicates this assumption is not accurate unless site-specific sampling is utilized to establish unique turbidity-suspended sediment relationships because turbidity is not only a function of TSS concentrations, but also of particle size, shape, and composition (Gippel, 1995: Hayes et al., 2001). As a result, this research is needed to analyze turbidity responses based on sediment characteristics for representative South Carolina soils.

Also, in order to separate particles from the runoff water in best management practices (BMPs) such as sediment basins, it is imperative to know sediment settling time and settling properties required to remove the particles from the water column. Accordingly, this research will also determine empirical relationships of sediment settling time to turbidity.

CHAPTER 2

LITERATURE REVIEW

Suspended Solids

Increased nonpoint source (NPS) pollution in the United States has been recognized as a priority environmental concern. NPS pollution is generated from diffuse land use activities rather than originating from a single discrete point source, such as a pipe (USGAO, 1998). It is conveyed to waterways through natural processes, such as rainfall, storm runoff, or groundwater seepage that generally are associated with land management, construction, and urban runoff (USGAO, 1998). Recent studies and surveys by EPA and state water quality agencies suggest the majority of remaining water quality impairments result from nonpoint source, urban stormwater discharges, and combined sewer overflows compared to point source pollution. Decreases in water quality from NPS pollution are generally reflected by increases of particulate matter (Packman et al., 1999) known as total suspended solids (TSS). According to the National Water Quality Inventory in 1998, suspended solids and sediment are the leading cause of water quality impairment of rivers and lakes (Swietlik, 2002).

TSS refers to the mass or concentration of both inorganic and organic solids that are held in the water column by turbulence (Thackston and Palermo, 2000). Inorganic solids can include sand, silt and clay particles, and organic solids may encompass algae, bacteria and detritus (Thackston and Palermo, 2000). Introduction of suspended solids (SS) in the nation's bodies of water may be caused by many sources, including (LSS, 2009):

- Soil erosion associated with agricultural practices and construction site runoff
- Domestic and industrial wastewater discharge
- Urban runoff from impervious surfaces
- Flooding and chronically increased flow rates
- Algae growth from nutrient enrichment (eutrophication)
- Dredging and channelization, and
- Removal of riparian vegetation and other stream bank disturbances

All streams carry SS under natural flow conditions. However, if concentrations are elevated from the above disturbances, this can lead to alterations to the physical, chemical and biological properties of a water body (Bilotta and Brazier, 2008).

Physical alterations caused by SS pose risks to water quality and aquatic organisms. Increased SS may result in a reduction of light penetration available for aquatic vegetation to grow by photosynthesis (LSS, 2009). A reduction in plant matter results in less energy, oxygen, and habitat for aquatic organisms. In addition, increased SS can negatively affect fish by limiting their ability to find food, increasing susceptibility to predators, and increasing gill abrasion (Packman et al., 1999). Water temperature alterations are another harmful physical stressor caused by increased TSS. Waters usually become warmer because of the greater heat absorbency of particulate matter and their ability to darken water and absorb more heat from sunlight (LSS, 2009). This thermal change can lead to negative effects on cold-water adaptive species (LSS,

2009). Lastly, upon deposition, increased SS loads can cause infilling of channels and reservoirs (Bilotta and Brazier, 2008).

Case studies involving the Chesapeake Bay, the Everglades, Milltown Reservoir, and Lake Hartwell have shown that suspended solids can be another pathway for biological and chemical contaminants to pollute the environment. Chemical contaminants are conveyed from adsorption sites on surfaces of sediments. Constituents of concern can include organic compounds, heavy metals, and some nutrients. Organic contaminants may encompass PCBs, PAHs present in fossil fuels, and pesticides from agricultural practices. Heavy metals include mercury, cadmium, lead, zinc, and chromium that could result from domestic and industrial wastes. Such contaminants are harmful because they can 1) bioaccumulate through the food chain (i.e. mercury); 2) they can settle with the sediments, where bottom-dwelling organisms are exposed to bioavailable contaminants, or 3) pollute groundwater via leaching from settled sediments (LSS, 2009). Sediments can also be a major source of the plant nutrients nitrogen, phosphorous and iron (LSS, 2009). One phenomenon, called eutrophication, is the result of waters becoming nutrientrich, which can lead to increased biological productivity. The most severe consequence of eutrophication is the depletion of oxygen by the decomposition of organic matter (Boesch et al., 2001). Organic matter produced in surface waters sink to the bottom where it decomposes, consuming oxygen inventories that are not replenished by photosynthesis or mixing with oxygen-rich surface waters (Boesch et al., 2001). Lastly, there is a longestablished link between sediment and biological pollutants, such as bacteria, in lentic systems (Sawyer, 2009). Clearly, SS are an important pollutant in surface waters; thus, quantifying and monitoring changes in suspended solids is critical for the nation's bodies of water.

TSS monitoring has become an integral part of programs to reduce nonpoint source pollution, such as those enacted by the Clean Water Act (CWA) (Dahlgren et al., 2004). The goal of the CWA is "to restore and maintain the chemical, physical and biological integrity of the Nation's waters" (EPA, 1972). The Clean Water Act is a comprehensive set of programs and requirements designed to address the complex problems caused by a wide variety of pollution sources (EPA, 1972). The primary focus of the CWA and subsequent 1977 amendments was the prevention of pollution discharges from point sources. In 1987, the act was again amended, this time to focus on non-point sources of pollution. One of the cornerstones of the Act is the National Pollutant Discharge Elimination System (NPDES), which regulates discharge of pollutants into waters of the U.S. Under the CWA, NPDES permits are issued to industrial, municipal, and other point source dischargers by either EPA or a delegated state agency (EPA, 1987).

Construction sites of a designated size or larger require NPDES permit coverage because erosion and sediment laden runoff can have a significant impact on water quality (EPA, 1987). The NPDES stormwater program requires construction site operators engaged in clearing, grading, and excavating activities that disturb one acre or more to obtain coverage under an NPDES general permit for their stormwater discharges. Most states, including South Carolina, are delegated to implement stormwater NPDES permitting programs. Permits require construction site operators to install and maintain erosion and sediment control measures to minimize stormwater from washing soil, nutrients, chemicals and other harmful pollutants into local water bodies.

Construction Sites

According to United States General Accounting Office in 1997, without proper controls at construction sites, sediment loads can reach 35 to 45 tons per acre per year (USGAO, 1998). The two factors accounting for large amounts of sediment coming from construction sites include high erosion rates and high delivery rates.

Erosion is the wearing away of top soil and subsoil by means of running water, wind, ice or other geologic agents (EPA, 1992). The process of erosion features detachment of soil particles from a soil mass, transport of the detached sediment (i.e. via runoff) and deposition of the sediments, known as sedimentation (Johns, 1998). The two types of erosion include geologic erosion and accelerated erosion. Geologic erosion, also known as natural erosion, is caused by geological processes acting over long periods of time without human disturbance (Johns, 1998), whereas accelerated erosion is a more rapid erosion process influenced mostly by human activities (Johns, 1998). Accelerated erosion is most often caused by an alteration of the landscape, resulting from floods, earthquakes, or anthropogenic activities (Morrow et al., 2007). Hence, land disturbance from construction activity exposes large areas of bare soil to water and wind erosion, increases soil erosion rates 2,000 to 40,000 times undisturbed rates (Harbor, 1999: Johns, 1998), and results in approximately 80 million tons per year of sediment supplied to US lakes, rivers and waterways (Harbor, 1999).

More importantly, construction sites can create very high delivery rates compared to pre-construction conditions (NCDENR, 2009). During the initial phase of construction, vegetation is cleared, land is graded, and ditches or storm sewers are installed to provide good drainage. Such alterations increase runoff volume and change the timing, frequency and rate of discharge. Practices are now required to compensate for increased postconstruction peak flows. However, these practices only control the rate of runoff volume leaving sites, but still allow increases in untreated runoff volume. Therefore, this drainage arrangement provides an efficient delivery system for pollutants to reach local waters.

Sediment Control Structures

States implement specific stormwater management and sediment reduction regulations for land disturbing activities that stem from the Clean Water Act. For example, according to South Carolina's Department of Health and Environmental Control (SCDHEC) regulations, construction site activities disturbing one acre or more require the development of erosion prevention and sediment control (EPSC) plans to achieve an 80 percent (minimum) design removal efficiency goal (SCDHEC, 2003). There are many practices that can be implemented on construction sites to reduce erosion. Consequently, even with very aggressive erosion prevention practices, soil erosion will still take place (Harbor, 1999). Therefore, to reduce eroded soil leaving sites, and meet an 80 percent removal efficiency goal, a variety of sediment trapping measures have traditionally been used. Gravitational settling from detention is the main process used to remove sediment from construction site runoff. Commonly used sediment trapping practices on sites include silt fences and sediment basins.

Silt fences are used as temporary perimeter controls that allow sediment to settle out of runoff. They are composed of a geotextile fabric with varying mesh sizes that are stretched across and attached to anchored wooden or metal stakes. The stakes are spaced at regular intervals along the site perimeter (Harbor, 1999). The bottom of the fence is buried in a trench to create a ponding area that allows time for sediment from runoff to settle before water passes through the geotextile (Harbor, 1999). If installed and maintained correctly, filtering efficiencies of silt fences can vary from 75 to 85 percent (EPA, 1992). Unfortunately, the vast majority of silt fences are installed incorrectly, resulting in ineffective sediment controls (Harbor, 1999).

Sediment basins are another common technique to capture sediment from stormwater runoff before it leaves the site. Sediment basins are engineered impoundment structures designed to temporarily detain surface runoff long enough to allow for sediments to settle out of water under the influence of gravity (Millen et al., 1997). They are installed prior to full-scale grading and remain in place until the disturbed portions of the drainage area are fully stabilized (EPA, 1993a). Sediment basins are usually used for drainage areas of five to 100 acres (EPA, 1993a) and are applicable in drainage areas where it is anticipated that controls, such as silt fences, will not be sufficient to prevent off-site transport of sediment (EPA, 1992). Sediment basins are popular with developers because they require less maintenance than other erosion and sediment control techniques and often can be converted into permanent urban runoff management ponds (Harbor, 1999). If sediment basins are converted into permanent stormwater management ponds, they must meet all regulatory requirements for wet or dry ponds (SCDHEC, 2005).

Several measures can be used to determine the effectiveness of a sedimentation basin including trapping efficiency, average effluent concentration, peak effluent concentration and peak effluent settleable solids (Hoechst, 1997). Trapping efficiencies are the most common performance standard and are generally calculated based on certain design requirements (Hoechst, 1997). For example, Figure 2.1 shows sediment basin details according to South Carolina's Regulation 61-9. The general design requirements are as follows (SCDHEC, 2005):

- 1. Minimum drainage area of five acres and maximum drainage area of 150 acres,
- 2. 80 percent design removal efficiency goal for TSS,
- Basin Shape: The effective flow length is at least twice the effective flow width (L=2W minimum),
- 4. Outlet riser has discharge capacity for a 10-year, 24-hour storm event, and
- 5. Effluent 2-year and 10-year, 24-hour storm disturbed flow rates are less than or equal to pre-disturbance peak flow rates.

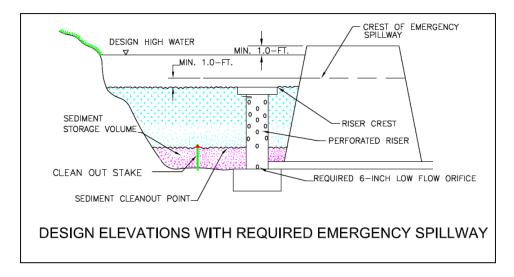


Figure 2.1. SCDHEC sediment basin specifications (SCDHEC, 2005).

For sediment basins to achieve certain sediment removal efficiencies, the basin size must be derived from calculating the rate at which sediments settle. Sediment basin settling theory is generally governed by Stokes' Law (Equation 1) to determine the settling velocity of particles (Haan et al., 1994). According to this law, a particle will settle in the vertical direction and accelerate to a constant velocity under the assumption that all particles have equal densities, are all spherical in shape, and all soil particles fall independent of each other under laminar flow conditions (Scott, 2000). The law states that particles larger in diameter sink farther and quicker than smaller particles when suspended in a liquid (Scott, 2000). Particles with greater surface area per unit of weight have greater frictional resistance and hence settle at slower velocities (Scott, 2000).

$$v_s=\frac{g\bigl(\rho_s-\rho\bigr)d^2}{18\mu}$$

1	1)
J	T)

- v_s = settling velocity (m/s)
- g= gravitational acceleration (m/s^2)
- ρ_s = particle density (kg/m³)
- ρ = fluid density (kg/m³)
- d= particle diameter (m)
- μ = dynamic viscosity (Pa-s)

Along with calculating appropriate basin size, basin geometry is an important design parameter in maintaining higher sediment retention efficiencies (Millen et al., 1997). Poor efficiencies are often found in ponds that have areas of dead storage caused by short circuiting. Bypassing of these dead spaces renders these areas ineffective in the settling process (Mitchell, 2000). As such, South Carolina's R 61-9 requires the flow length to be at least twice the flow width to increase the flow path length to minimize dead spaces (SCDHEC, 2005).

Sediment basins must be maintained to work effectively. Such maintenance may include inspecting basins regularly for sediment deposition and removal of sediment once the basin reaches 50 percent of sediment storage volume (SCDHEC, 2005). Sediment basin effectiveness is a function of eroded particle size distribution of inflow sediment (Mitchell, 2000). Trapping efficiencies will be adversely affected by fine sediments that remain in suspension.

TSS Measurements

Sediment trapping efficiency of a pond may be found by measuring the sediment concentration and flow rate of the runoff entering the structure and the sediment concentration and flow rate of the water leaving the detention structure. Total suspended solids are determined from laboratory analysis by the dry weight of suspended solids per unit volume of water, and are typically reported in milligrams of solids per liter of water (mg/L). There are three different laboratory methods to quantify the amount of solids contained in stormwater samples taken from the field (Guo, 2006). The three methods include EPA's TSS Method (USEPA 1999), the American Public Health Association's (APHA 1995) TSS Method (also referred to as Standard TSS Method), and American Society for Testing and Materials' (ASTM) Suspended Sediment Concentration (SSC) Method (ASTM 1997). All three methods obtain the amount of solids contained in stormwater samples through filtering water, then drying and weighing residue that has remained on the filter (Guo, 2006). EPA's TSS Method stirs and collects the sub-sample by pouring from the whole sample container, the Standard TSS Method stirs and collects the sub-sample using a pipette to draw from the whole sample container, and the ASTM's SSC Method uses the whole sample (Guo, 2006).

Consequently, TSS measurements are not routinely used to detect and correct short-term problems or permit violations because sediment concentrations cannot be determined easily or quickly in the field, and transportation to a laboratory for analysis is time-consuming and can be costly (Thackston and Palermo, 2000). As a result, these laboratory methods are increasingly being replaced in favor of continuously-collected surrogate data for quantification of SSC that may be safer and (or) less expensive to obtain. Other common methods to evaluate stream-water suspended solids concentrations include transparency and turbidity measures. For example, the relation of turbidity to TSS can be used to estimate suspended loads as opposed to estimations based on water discharge from construction sites. This approach has potential for monitoring any water quality constituent whose concentration is better correlated with an easily measured parameter, such as turbidity.

<u>Turbidity</u>

Turbidity is a measurement used to quantify water clarity (Davies-Colley and Smith, 2001). It is an expression of optical properties that cause light rays to be scattered and absorbed rather than transmitted in straight lines through a water sample (Anderson, 2005). Suspended solids have an optical impact on water quality by reducing light transmission through water; this is referred to as light attenuation (Davies-Colley and Smith, 2001). Materials contributing to suspended solids include soil particles (silts and clays), finely divided organic and inorganic matter, soluble organic compounds, plankton, and microscopic organisms. Typically, suspended soil particles exert the dominant influence on light attenuation in natural waters (Davies-Colley and Smith, 2001). Cloudiness of water results from intense scattering of light by fine particles typically with diameters smaller than 0.050 mm (Mitchell, 2000). Hence, waters with high concentrations of fine suspended sediment are described as turbid.

Turbidity Measurements

From the early 20th century, turbidity was measured in Jackson turbidity units (JTU) using a Jackson Candle Turbidimeter (Borok, 2010). This method incorporated a visual method of looking at a black object at certain depths in water to determine turbidity of water (Mitchell, 2000). The turbidimeter consisted of a special candle and a flat-bottom glass tube. Measurements were made by slowly pouring a turbid sample in the tube until the image of the candle flame diffused to a uniform glow (Borok, 2010). Depth of the sample in the tube is read against the ppm-silica scale, and turbidity was

measured in Jackson turbidity units (JTU). Jackson turbidimeters cannot measure turbidity lower than 25 JTU, are cumbersome, and depend on human judgment to determine the extinction point (Borok, 2010). Thus, photoelectric detectors were developed and became the accepted method to measure turbidity (Borok, 2010). Turbidity meters can be used in the field with portable meters, in the lab with benchtop meters, or as probes that can be installed in a stream or lake for continuous measurements. Turbidity is now commonly measured in nephelometric turbidity units (NTU) using nephelometric turbidimeters (Mitchell, 2000).

Nephelometric turbidimeters direct a beam of light into the side of a test sample, measure the amount of light that is reflected at a restricted range of angles (typically 90 degrees) by any particles present, and compare it to the light scattered by standard reference suspensions (Mitchell, 2000 and Borok, 2010). Figure 2.2 displays the basic design for nephelometric turbidimeters where scattered light is captured by a photodiode, which produces an electronic signal that is converted to a turbidity value reported in NTUs (ISO, 1999). Two of the most commonly used nephelometric turbidimeters include nephelometers and near infrared turbidimeters.

18

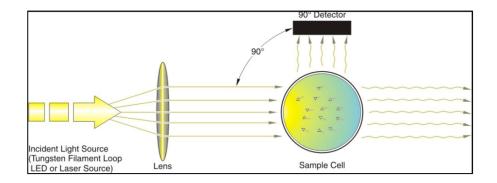


Figure 2.2. Basic turbidimeter design using nephelometric measuring technique (Sadar, 2004).

Nephelometers measure light that has been scattered at a specific angle of 90 degrees from the main light path (EPA, 1993b). Generally, commercially available nephelometers are compliant with EPA's Method 180.1 for determining turbidity by nephelometery (Borok, 2010). Along with the specified angle of 90 degrees, nephelometers under EPA's Method 180.1 require the light source to be a tungsten lamp operating at a color temperature between 2200-3000 K. The light source is a light-emitting diode where the light path is designed to minimize stray light falling on the detector (Thackston and Palermo, 2000). Therefore, a zero reading means no light scattered at 90 degrees (±30 degrees) from the main light path and implies no turbidity (Thackston and Palermo, 2000). Nephelometers must be standardized against either a primary formazin suspension with a value of 4000 NTUs (which can be diluted to desired NTU values) or a commercially available polymer standard identified as AMCO-AEPA-1 (EPA, 1993b).

Formazin was established as the first calibration standard for turbidity meters in the 1950's. Since then, machine performance and EPA approval for meters is structured around formazin as the primary calibration standard (AMCOClear, 2010). Formazin suspensions exhibit a wide range of suspended particle size and shape. Typically, formazin particles are irregular in shape and range from 0.1 to 10.0 μ m in size, closely matching the range of particulates, 0.2 to 50 μ m, found in real world samples (Hach, 2011). The highly predictable light-scattering properties of formazin are the basis of algorithm design for nephelometric instruments. The primary formazin suspension is a stock standard suspension (SSS) that can be prepared following standard methods (APHA, 2005). While SSS can be prepared directly in the laboratory and used to create primary calibration standards (PCALs), the process is labor intensive, time consuming, and requires precise laboratory technique (Hach, 2011). Besides being the only primary standard, formazin is also the least expensive, premixed, commercially available turbidity standard (Downing, 2005). However, routine use of formazin SSS and PCALs have several notable disadvantages that include: 1) formazin's hydrazine sulfate is a carcinogen; 2) turbidity can vary by 2% from batch-to-batch; 3) the size, shape, and aggregation of formazin particles change with temperature, time, and concentration; 4) it settles in storage and must be mixed immediately prior to use; and 5) diluted formazin standards have a storage life as short as one hour (Downing, 2005). As a result, alternative standards, known as secondary calibration standards (SCALs), are created and typically supplied with purchase of turbidity meters. SCALs have been certified by the manufacturer to provide calibration results equivalent to those obtained when the instrument was calibrated with a primary standard (APHA, 2005). Available secondary standards include commercial stock suspensions of 4000 NTU formazin (i.e. StablCal), commercial suspension of microspheres of styrene divinylbenzene (SDVB) microspheres, and items supplied by instrument manufacturers (APHA, 2005). According to Method 180.1, SCALs are acceptable as a daily calibration check, but must be monitored on a routine basis for deterioration and subsequently replaced as required. All secondary standards change with time and must be replaced when their age exceeds shelf life. Deterioration can be detected by measuring the turbidity of the standard after calibrating the instrument with a fresh formazin or microsphere suspension (APHA, 2005).

Along with formazin calibration standards, EPA approved polymer suspensions in 1984 as a secondary calibration standard for turbidity meters under the name AMCO EPA (AMCOClear, 2010). It is made from SDVB microspheres and, unlike formazin, SDVB microspheres have uniform size, shape and particle size distribution. An example is displayed in Figure 3.3 (Downing, 2005). SDVB microspheres have a particle size distribution of 0.02 to 0.2 μ m with a mean size of 0.121 μ m (AMCOClear, 2010). Size distribution of SDVB spheres are adjusted to produce a formazin-equivalent response from a particular turbidity meter (Downing, 2005). SDVB standards are formulated for a specific make and model of turbidimeter, and therefore, cannot be used with a different manufacturer or model even though it conforms to the same standard method (Downing, 2005). Key benefits of SDVB standards are: 1) batch-to-batch variation in turbidity is less than 1%; 2) optical properties are constant from 10 to 30°C; 3) one-year stability is guaranteed; 4) mixing and dilution are not required, and; 5) are not toxic (Downing, 2005).

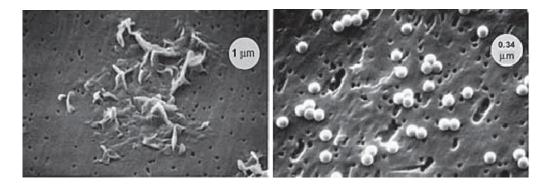


Figure 2.3. Scanning electron microscopy (SEM) images. On the left, formazin particles have many different shapes, whereas SDVB particles on the right are dimensionally uniform (Downing, 2005).

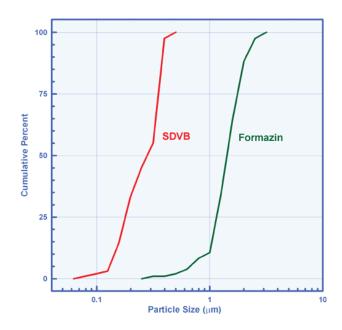


Figure 2.4. Graph compares the particle sizes of formazin and SDVB. The SDVB particles are about 1/5 that of formazin particles (Downing, 2005).

Other regularly accepted nephelometric turbidimeters are near infrared (IR) turbidimeters. These types of nephelometric turbidimeters are compliant with the International Standards Organization (ISO) standard 7027, which is commonly used in Europe (Borok, 2010). This standard requires that particle light scattering should be measured in near infrared (wavelength 860 ± 60 nm) with a light-emitting diode (LED) as the instrument's light source (ISO, 1999). Also, the detector angle must not exceed ± 2.5 degrees from the 90-degree incident path (Borok, 2010).

Turbidity meters vary in design, such as optical design of spectral power of the light source, spectral sensitivity of the detector, angular scattering range, and optical geometry (Pfankuche and Schmidt, 2003). As a result, meters are highly instrument-specific turbidity measurements in spite of identical calibration to formazin (Pfankuche and Schmidt, 2003). For example, infrared turbidimeter beams are unaffected by light absorbance of particles (usually dissolved organic compounds), whereas visible light turbidimeters are more sensitive to scattering from fines (Packman et al., 1999).

Future Turbidity-Based Regulations

Turbidity has been recognized as an indicator of pollution in surface runoff from disturbed areas. In December 2009, EPA released turbidity effluent limitations and new source performance standards to control discharge of pollutants from construction sites (EPA, 2009). Regarding turbidity, the final rule subjected discharges from construction sites disturbing 20 or more acres of land at one time to comply with a numeric effluent limit of 280 NTU, starting in August of 2011 (EPA, 2009). By February 2014 the

limitation would apply to all construction sites disturbing 10 or more acres (EPA, 2009). Interestingly, the final rule does not prescribe specific requirements, such as frequency or location of monitoring related to construction activity (EPA, 2009). Subsequent to the proposed final rule, multiple petitions for reconsideration pointed out potential error in calculating the numeric limit of 280 NTU (EPA, 2010). As a result, EPA concluded that it improperly interpreted the data and is currently reevaluating the numeric effluent limitation of 280 NTU. EPA intends to publish the corrected final rule by May 30, 2011 so that the revised limitation can be effective by June 29, 2011 for its Construction General Permit (EPA, 2010).

Relationships to Total Suspended Solids

Optical gauges are widely used devices to monitor TSS concentrations indirectly since they are generally sensitive to a wide concentration range of TSS and are relatively inexpensive (Foster et al., 1992). Although most are suited to the concentration range of 0-1000 mg/L, higher concentrations can be determined, but these yield a decrease in sensitivity (Foster et al., 1992). Therefore, nephelometric turbidity measurements are gaining recognition because NTU units have been shown to relate to TSS concentrations in many water bodies in regions around the world (Packman et al., 1999). This is because the scattering coefficient measured at an angle of 90 degrees behaves almost proportionally to suspended particle concentration when sensors are calibrated to give a linear response to standards (Pfankuche and Schmidt, 2003). Thus, response to varying TSS concentrations should be linear if the physical properties of suspended particles are

constant (Lewis, 1996). For instance, Lewis (1996) was able to estimate event sediment loads by predicting SSC from linear regressions on turbidity from five storm events on a creek.

Several other studies have illustrated adequate relationships from linear and other regression models of TSS concentration and nephelometric turbidity units in varying water bodies (Ellison et al., 2010; Grayson et al., 1995; Hayes et al., 2001; Packman et al., 1999; Pavanelli and Pagliarani, 2002; Pfankuche and Schmidt, 2003; Mitchell, 2002). However, previous research states turbidity should not be used as a substitute for sediment concentration without a careful study of the relationship between turbidity and suspended load for any proposed watershed monitoring (Pavanelli and Pagliarani, 2002). Hence, previous research stresses the importance of site-specific sampling in order to establish unique turbidity-suspended sediment relationships, (Hayes et al., 2001) because turbidity is not only a function of TSS concentrations, but also of particle size, shape and composition (Foster et al., 1992: Gippel, 1995). As a result of such site-specific properties, there are no universal relationships between turbidity and suspended solids (Borok, 2010). Variable optical properties have resulted in correlations with low coefficients of determination (Pfankuche and Schmidt, 2003). Such variability was generally attributed to changes in particle properties and particularly particle size (Pfankuche and Schmidt, 2003). Finer sediments have more reflective surfaces per unit mass, so, for constant SSC, sensor output increases as suspended sediment becomes finer (Schoellhamer and Wright, 2003), whereas a higher concentration of larger particles produces lower turbidity levels (Hayes et al., 2001). Since turbidity is often an indication of the quantity of clay particles in suspension, it would be beneficial to obtain a relationship between turbidity and particle size, as well as sediment concentration. Also, MacDonald (1991) and Lewis (1996) found that when a watershed displayed similar characteristics, such as soil type, variations either are generally not large or related to SSC, thus the relation between turbidity and SSC may be quite stable and precise (Hayes et al., 2001).

Particle Size Distributions & Particle Settling Velocities

In order to classify a soil for engineering purposes, one needs to know the distribution of particle sizes in a given soil mass. In order to separate particles from the runoff water in best management practices (BMPs) such as sediment basins, it is imperative to know sediment settling time and settling properties required to remove the particles from the water column (Haan et al., 1994: Tempel, 2011). As addressed earlier, particle diameter plays an irmportant role in calculating trapping efficiency of a sediment basin because particle sizes directly relate to settling velocity of the particles (Stoke's Law). Again, these velocities have an effect on required detention time and corresponding area of the pond (Mitchell, 2000).

The scope of this project is on behavior of particles smaller than 0.063 mm (silts and clays) because larger particles settle from surface water flow relatively quickly, whereas small particles remain in suspension longer, thus contributing as sources of turbidity. Therefore, along with determining the relationship between turbidity and particle size, this research will evaluate each soil's relationship of turbidity with respect to time.

A particle size distribution (PSD) curve is created to describe the range of various particle sizes in a given soil. A typical way to express particle size distribution is with a percent finer curve. Such a curve illustrates particle size weight classes versus the entire sample weight. According to the American Association of State Highway and Transportation Officials (AASHTO), particle sizes include four categories of gravel, sand, silt and clay. Their respective particle diameter sizes are listed below following USDA's particle size classification (Das, 2006).

- 1. Gravel: greater than 2 mm.
- 2. Sand: 2 mm to 0.05 mm.
- 3. Silt: 0.05 mm to 0.002 mm
- 4. Clay: less than 0.002 mm

CHAPTER 3

ACCURACY AND PRECISION OF PORTABLE TURBIDITY METERS

ABSTRACT

EPA has published effluent limitation guidelines (ELGs) to control discharge of pollutants from construction sites. Numeric turbidity limits for construction site discharge are expected to be required in the near future. Such requirements will likely include subjecting construction site stormwater discharges to a maximum allowable turbidity numeric effluent limit in nephelometric turbidity units (NTUs) for sites disturbing 10 acres or more.

Turbidity is an expression of the optical properties of a liquid that causes light rays to be scattered and absorbed as measured by a nephelometer. Though constituents such as organic matter can impact water clarity, typically the inorganic fraction derived from particulate matter such as sediment dominates turbidity levels in surface waters. Recognition that water clarity is an important indicator of environmental health has increased, resulting in growing demand for high-quality and objective measurement. However, questions remain as to how samples should be obtained and processed to get reliable readings.

Given the importance of pending numeric effluent limitations related to turbidity, the focus of this research is to compare various nephelometers to quantify accuracy and precision of selected portable instruments for use in routine sediment/surface water analysis and potential regulatory compliance. Instruments were either provided with commercial stock suspensions of formazin (StablCal) or commercial suspension of microspheres of styrene divinylbenzene (SDVB) (AMCO EPA). In order to evaluate

29

meter accuracy and precision, experiments were conducted to compare their responses to true NTU values. This analysis examined performance of each meter when calibrated to manufacturer-supplied secondary calibration standards and when calibrated to primary calibration standards (PCALs).

Results indicated three of the four meters (Hach 1, Hach 2, and GW) were accurate and precise for both calibration studies, but did not fall within the claimed $\pm 2\%$ accuracy range as provided by manufacturer's specifications. The LaMotte meter did not perform as well, and even though the overall accuracies improved when calibrated to the formazin PCALs, the meter was inaccurate for higher turbidity readings. As for the meters' responses to varying field surface water samples, readings produced by the Hach 2100Q meters ('Hach 1' & 'Hach 2') had very similar results and did not change much for either calibration study, but were consistently higher than the LaMotte and GW meters. However, when meters were calibrated to PCALs, the range of readings per sample narrowed. Additionally, the LaMotte and GW meters may provide inaccurate results for higher turbidity readings that may typically contain a wider range of particle sizes. Variability among the Hach meters to the LaMotte and GW is most likely due to differences in formazin standards and SDVB standards. Results may have significant bearing on the construction and development industry as it prepares for proposed monitoring requirements associated with recently promulgated numeric effluent standards for turbidity.

INTRODUCTION

Turbidity measurements are gaining increased usage as an indicator of pollution in surface runoff from disturbed areas such as active construction sites. Recent efforts by the EPA have resulted in proposed turbidity effluent limitations for discharge from construction sites. Turbidity is an expression of optical properties that cause light rays to be scattered and absorbed rather than transmitted in straight lines through a water sample (Anderson, 2005). Though constituents such as organic matter can impact water clarity, typically the inorganic fraction derived from particulate matter, such as sediment, dominates turbidity levels in surface waters (Davies-Colley and Smith, 2001). Cloudiness of water results from intense scattering of light by fine particles typically with diameters smaller than 0.050 mm (Davies-Colley and Smith, 2001). Hence, waters with high concentrations of fine suspended sediment are frequently described as turbid. Turbidity is a vivid visual indicator of pollution associated with sediment-laden runoff.

Turbidity is now commonly measured in nephelometric turbidity units (NTU) using nephelometric turbidity meters (Borok, 2010). Such meters direct a beam of light into the side of a test sample, measure the amount of light that is reflected at a restricted range of angles (typically 90 degrees) by any particles present, and compare it to light scattered by standard reference suspensions (Borok, 2010). Two of the most commonly used nephelometric turbidity meters include nephelometers and near infrared turbidity meters.

Nephelometers measure light that has been scattered at a specific angle of 90 degrees from the main light path (EPA, 1993b). Most commercially available meters comply with EPA's Method 180.1 for determining turbidity by nephelometry (Borok, 2010 and EPA, 1993b). Such instruments must be standardized against either a primary formazin suspension with a value of 4000 NTU (which can be diluted to desired NTU values) or a commercially available polymer standard identified as AMCO-AEPA-1 (EPA, 1993b).

Formazin was established as the first calibration standard for turbidity meters in the 1950's and since then, machine performance and EPA approval for meters is structured around formazin as the primary calibration standard (AMCOClear, 2010). Formazin suspensions are characterized by a wide range of suspended particle size and shape. Typically, formazin particles are irregular in shape and range from 0.1 to 10.0 µm in size, closely matching the range of particulates found in field samples (Hach, 2011). The highly predictable light-scattering properties of formazin are the basis of algorithm design for nephelometric instruments. The primary formazin suspension is a stock standard suspension (SSS) that can be prepared following standard methods (APHA, 2005). While SSS can be prepared directly in the laboratory and used to create primary calibration standards (PCALs), the process is labor intensive, time consuming, and requires precise laboratory technique (Hach, 2011). Besides being the only primary standard, formazin is also the least expensive premixed, commercially available turbidity standard (Downing, 2005). However, routine use of formazin SSS and PCALs have several notable disadvantages: 1) formazin's hydrazine sulfate is a carcinogen; 2) turbidity can vary by 2% from batch-to-batch; 3) the size, shape, and aggregation of formazin particles change with temperature, time, and concentration; 4) it settles in storage and must be mixed immediately prior to use; and 5) diluted formazin standards have a storage life as short as one hour (Downing, 2005).

As a result, alternative standards known as secondary calibration standards (SCALs) are created and typically supplied with purchase of turbidity meters. SCALs have been certified by the manufacturer to provide calibration results equivalent to those obtained when the instrument was calibrated with a primary standard (APHA, 2005). Available secondary standards include commercial stock suspensions of 4000 NTU formazin (i.e. StablCal), commercial suspension of microspheres of styrene divinylbenzene (SDVB) microspheres, and items supplied by instrument manufacturers (APHA, 2005). According to Method 180.1, SCALs are acceptable as a daily calibration check, but must be monitored on a routine basis for deterioration and subsequently replaced as required. All secondary standards change with time and must be replaced when their age exceeds shelf life. Deterioration can be detected by measuring the turbidity of the standard after calibrating the instrument with a fresh formazin or microsphere suspension (APHA, 2005).

Along with formazin calibration standards, EPA approved use of polymer suspensions in 1984 as a secondary calibration standard for turbidity meters under the name AMCO EPA (AMCOClear, 2010). It is made from SDVB microspheres and unlike formazin, the SDVB microspheres have uniform size, shape and particle size distribution (Downing, 2005). The SDVB microspheres have a particle size distribution of 0.02 to 0.2 μm with a mean size of 0.121 μm (AMCOClear, 2010). The size distribution of SDVB spheres is adjusted to produce a formazin-equivalent response from a particular turbidity meter (Downing, 2005). SDVB standards are formulated for a specific make and model of turbidimeter, therefore, cannot be used with one from a different manufacturer or model even though it conforms to the same standard method (Downing, 2005). The key benefits of SDVB standards are: 1) batch-to-batch variation in turbidity is less than 1%; 2) optical properties are constant from 10 to 30°C; 3) one-year stability is guaranteed; 4) mixing and dilution are not required, and; 5) are nontoxic (Downing, 2005).

Turbidity is not an inherent property of water, such as temperature or pH, yet recognition of turbidity as an indicator of the environmental health of water bodies has increased, resulting in a growing demand for high-quality and objective turbidity measurements (Anderson, 2005). Questions remain as to how samples should be obtained and processed to get reliable readings. Therefore, given the importance of proposed turbidity limits, focus of this research is to compare various commercially available portable nephelometers to quantify accuracy and precision of the selected instruments for use in routine sediment/ surface water analysis and potential regulatory compliance.

34

PROCEDURES

To evaluate instrument performance, accuracy is defined as the nearness of a measurement to the accepted or true value. Accuracy can be expressed as a range, about the true value, in which a measurement occurs. Instrument precision is defined as the tightness of measurements for one sample. Precision can be expressed as a range about the averaged reading (LaMotte, 2010).

Four commonly utilized and commercially available turbidity meters were selected for evaluation in this study (see Figure 3.1). Nephelometers chosen included two Hach 2100Q, one LaMotte 2020e and Global Water's 'GW' Turb430T¹. Note that meters Hach 1 and 2 are the same model, and were selected to evaluate any statistical differences between two meters from the same manufacturer. Each meter specified compliance with EPA's Method 180.1, and meter specifications are provided in Table 3.1. Meters Hach 1 and 2 are both supplied with SCALs known as StablCal (a stable formazin secondary standard) to calibrate the meters, whereas the LaMotte and GW meters are supplied with AMCO EPA calibration standards. Experimental procedures are summarized and outlined in Table 3.2.

¹ Disclaimer: Mention of a trade name does not imply endorsement of the product by Clemson University to the exclusion of others that might be available. Users are encouraged to fully evaluate the suitability of any equipment for their intended application.



Figure 3.1. Selected nephelometers. From left to right GW, Hach 1, and LaMotte.

Table 3.1. Manufacturer's	provided specifications	s for selected meters.
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Meter	Accuracy	Precision	Price	Meter Range	
Hach 1&2	$\pm 2\%$ of true values from 0 to 1000 NTU	$\pm 1\%$ of the measured value	\$930	0-1000 NTU	
LaMotte	±2% of true values from 0 to 100 NTU ±3% of true values above 100 NTU	0.02 NTU of the measured value	\$900	0-4000 NTU	
GW	$\pm 2\%$ of true values from 0 to 500 NTU $\pm 3\%$ of true values 500 to 1100 NTU	$\pm 1\%$ of the measured value	\$2100	0.01-1100 NTU	

Table 3.2. Experimental procedures summary.

	Exp. 1	Exp. 2	Exp. 3a	Exp. 3b	Exp. 4a	Exp. 4b
Calibrated with	Supplied SCALs	Supplied SCALs	Supplied SCALs	Created PCALs	Supplied SCALs	Created PCALs
Tested Solution	Cecil TSS concentrations	All Supplied SCALs	Created PCALs	Created PCALs	Field Samples	Field Samples

Experiment 1: Concentration vs. Turbidity

Initially, known total suspended solids (TSS) concentrations of a Cecil soil series were tested in the Hach 1, LaMotte and GW meters when calibrated to their provided calibration standards. Cecil soil, a predominant sandy loam commonly found in Piedmont uplands of the southeastern U.S., was used to create several concentrations of TSS for analysis. First, a dry sieve analysis was performed where soils were oven dried at 105°C and passed through a 230 sieve with a mesh opening of 0.063mm to obtain material for use in experimentation. Resulting soil material was weighed on an analytical balance to the nearest 0.0001 g to achieve concentrations of 3000, 2000, 1000, 500, 100, 50 and 25 mg/L. Each concentration was created individually and was mixed using a magnetic stir plate.

Each meter was calibrated as specified to the manufacturer's operating instructions and then the known concentrations were placed in corresponding sample vials for analysis. Each sample was read five times by each meter and then averaged (see Appendix A). Between each reading, the sample vial was gently agitated by inverting the vial roughly ten times to adequately suspend and disperse all sediment particles before replacing it into the turbidity meter (Hayes et al., 2001). Using standard pipette protocol, dilutions were made from the sample vial if concentrations were out of instrument turbidity range (see Table 3.1).

Meters did not produce the same turbidity readings for each known concentration and results are displayed in Figure 3.2. After performing an ANOVA two factor analysis without replication it was concluded that the meters were different (p-value less than 0.05), initially suggesting they may differ in the manufacturer's calibration standards. Therefore, the following experiment was conducted to test the standards provided with each meter.

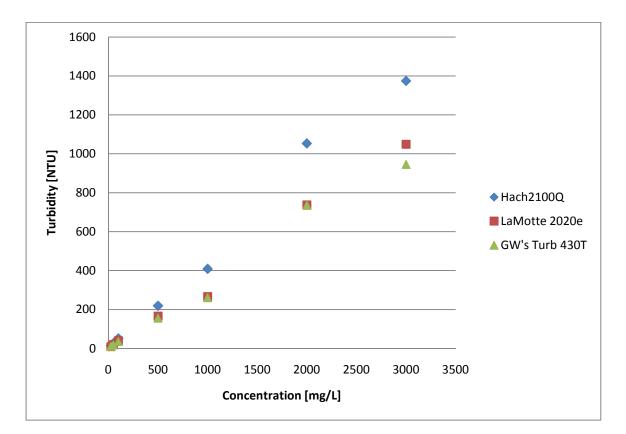


Figure 3.2. Meters' turbidity responses to known Cecil concentrations.

Experiment 2: Verification of Manufacturer Supplied Standards

For this experiment, meters Hach 1, LaMotte and GW were available to evaluate meter responses to all supplied calibration standards. Again, each meter was calibrated with its provided standards as specified in the manufacturer's operating instructions.

First, standards from Hach 1 were placed in each meter (including Hach 1) to distinguish if subsequent readings gave the same result as the standard. Five readings were taken for each of Hach 1's standards, recorded and then averaged. The procedure was repeated using meters LaMotte and GW's calibration standards. Resulting tabular data are included in Appendix A.

Results indicated each meter was calibrated to its own standards. This also verifies that meters calibrated with SDVB standards (AMCO EPA) are formulated exclusively for the specified meter and cannot be used with different meters (Downing, 2005). Therefore to find true readings of each meter, the primary formazin stock standard suspension (SSS) with a value of 4000 NTUs was used for the following experiments.

Experiment 3: Meter Accuracy and Precision

To evaluate meter accuracy and precision, the third experiment compared responses to true NTU values. This analysis examined meters' performance when calibrated to manufacturer-supplied secondary calibration standards and when calibrated to primary calibration standards (PCALs).

3a: Meters calibrated to supplied SCALs

Preparation of NTU standards used the 4000 NTU primary formazin SSS. Following Standard Methods, dilutions with high-quality dilution water was used to create PCALs with values of 1, 10, 50, 100, 280, 500, 750 and 1000 NTU. First, one liter of the 1000 NTU standard was created from the 4000 NTU formazin primary standard and placed in an amber bottle. 100 milliliters of each standard was prepared from the 1000 NTU solution and placed in opaque plastic bottles to eliminate light penetration. Because hydrazine sulfate (constituent in formazin) is a carcinogen, all dilutions were executed under laboratory hood to avoid inhalation, ingestion and skin contact (APHA, 2005).

Meters were calibrated as specified in the manufacturer's operating instructions with provided calibration standards. Each created NTU standard was placed in the meter's specified sample vial, positioned in the turbidity meter five times, and the results were recorded. Between each reading, sample vials were carefully inverted ten times before being placed back in the meter. A replicated experiment was performed for statistical purposes. Resulting data tables are in Appendix A.

To determine meter accuracy for each sample, percent difference from the true value (value derived from formazin SSS) was calculated for each reading and then averaged among percent differences (Equation 2). Therefore meter accuracy, displayed in Table 3.3, is represented as an averaged percentage about the true value. To determine meter precision for each sample, the same five readings were utilized and averaged. Then, each reading's percent difference from the averaged measured value was calculated and these five percent differences were averaged (Equation 3). Results are shown in Table 3.4 and are represented as an averaged percentage about the measured value.

Accuracy [%] =
$$\sum \overline{\left(\frac{|MR - TV|}{TV} * 100\right)}$$
 (2)

MR= measured turbidity reading [NTU],

TV= true turbidity value [NTU]

$$Precision = \sum \overline{\left(\frac{MR - \overline{MR}}{\overline{MR}} * 100\right)}$$
(3)

MR= measured turbidity reading [NTU]

3b: Meters calibrated to PCALs

The fourth experiment was conducted to compare each meter's performance when calibrated to formazin primary calibration standards (PCALs). Therefore, values of the meters provided calibration standards created following the Standard Methods procedure above, and meters were calibrated as specified to the manufacturer's operating instructions using these PCALs.

The experiment was performed, as described above, to test each meter's readings of the created standards of 1, 10, 50, 100, 280, 500, 750 and 1000 NTU. Again, a replicated experiment for the meters calibrated to PCALs was executed for statistical purposes. To determine each meter's accuracy and precision, calculations described previously were performed with results shown in Tables 3.3 and 3.4.

Experiment 4: Turbidity in Natural Water Samples

The final experiment was to evaluate meter response to surface water samples collected in the field from both disturbed and undisturbed sites. Since future use of these instruments will be to monitor field conditions, it was important to examine meter behavior with real surface water samples of unknown turbidity collected in the field.

Samples used for this experiment included three composites collected at the discharge point from an active construction site at varying time intervals after a rain event began, one sample from a golf course creek, and another sample from a botanical garden's pond. It was desired to get a variety of samples with varying TSS concentrations to evaluate meter range. Samples were mixed using a magnetic stir plate in the laboratory. Laboratory analysis was chosen over field practice because some of the samples needed to be diluted using laboratory techniques. This would eliminate field dilution errors when calculating actual turbidity readings.

Meters were calibrated to their provided calibration standards. Each field sample was drawn using a 10ml pipette, placed in the specified vial and positioned in the corresponding turbidity meter. Each sample was read five times and averaged. Sample vials were inverted ten times between each reading to adequately suspend and disperse all sediment particles before placing it into the turbidimeters. A total of five replications for each sample were analyzed. Dilutions were made from the main sample if the concentration was out of an instruments' turbidity range (see Table 3.1).

Finally, the same procedure was repeated except meters were calibrated to PCALs instead of the meter's calibration standards. Resulting data tables are shown in Appendix A.

RESULTS AND DISCUSSION

The goal of this paper was to compare accuracy and precision for selected portable turbidity meters that comply with Method 180.1. Computed results are presented in Tables 3.3-3.4 and also account for experimental error associated with the volumetric flasks used to create primary formazin standards.

In Table 3.3, for each meter, overall averaged percent error ("AVG") from the true value was calculated. However, statistical calculations were computed without the meters' averaged percent differences from the 1 NTU value because the low end produced a large discrepancy; and focus of this research is surface water samples that have typical values above 10 NTU. Overall accuracy improved somewhat for all meters when calibrated to the formazin PCALs by 0.34%, 0.52%, 8.57% and 1.28% for Hach 1, Hach 2, LaMotte and GW, respectively (Table 3.3). However, meters Hach 1, Hach 2, and GW for both calibration studies are slightly above the specifications' claimed $\pm 2\%$. Even though LaMotte meter improved when calibrated to the PCALs, the instrument still provided statistically inaccurate results, especially for readings above 500 NTUs. The LaMotte meter provided significantly higher percent differences at higher turbidity readings.

As for the meters' repeatability of a sample, Hach 1, Hach 2 and GW, for both calibration studies, fell within the claimed $\pm 1\%$ of the measured value (Table 3.4).

LaMotte was slightly above $\pm 1\%$ of the measured value, but improved when calibrated to PCALs.

	Hach 1		Hach 2		LaMotte		GW	
PCAL values ³ [NTU]	Calibrated to provided SCALs	Calibrated to PCALs						
1	55.6	44.0	53.0	54.0	61.1	23.6	6.20	31.2
10	6.70	6.20	5.70	5.00	14.9	6.13	3.32	0.50
50	1.58	2.50	1.52	1.00	16.4	5.14	3.70	3.12
100	3.96	0.99	1.63	0.56	10.1	7.24	5.00	2.15
280	4.68	3.18	3.21	2.50	3.14	7.64	3.39	2.11
500	2.20	2.72	2.56	2.46	22.1	6.54	4.48	3.10
750	1.53	2.27	2.15	2.20	78.1	47.9	2.72	1.03
1000	1.12	1.54	1.24	0.62	21.3	24.2	1.11	2.79
AVG	3.11	2.77	2.57	2.05	23.7	14.9	3.39	2.11

Table 3.3. Meter Accuracy: averaged percent difference about the true value for each $meter^2$.

Table 3.4. Meter Precision: averaged percent difference about measured value for each

meter⁴.

	Hach 1		Hach 2		LaMotte		GW	
PCAL values ⁵ [NTU]	Calibrated to provided SCALs	Calibrated to PCALs						
1	6.99	2.24	5.00	2.69	9.93	9.81	7.50	18.92
10	1.69	1.13	0.75	0.57	1.34	4.35	0.50	0.93
50	0.37	0.72	0.61	0.40	0.34	0.40	0.73	0.38
100	0.69	0.54	0.49	0.71	0.47	0.49	0.93	0.60
280	1.05	0.59	0.41	0.22	0.10	0.45	0.67	0.58
500	0.82	1.15	1.29	0.84	2.10	1.03	0.54	0.31
750	1.19	0.70	0.70	0.68	4.83	1.33	0.88	1.22
1000	1.21	0.77	0.65	0.46	1.07	0.42	0.89	0.99
AVG	1.00	0.80	0.70	0.56	1.46	1.21	0.73	0.72

^{2,4} Overall averaged percent error ('AVG') from the true value was calculated without the meters' averaged percent differences from the 1 NTU value. ^{3,5} Known NTU values ('PCAL values') were derived by diluting 4000 NTU primary formazin SSS with

distilled (DI) water.

Finally, meters were tested with field samples of unknown turbidity values in order to understand their behavior for future analysis. Meter responses are provided in Figure 3.3. Overall, Hach 1 and 2 meters produced very similar results, as expected, and did not change much for either calibration study. Also, it was observed from Figure 3.3 that Hach 1 and 2 consistently produced higher results compared to meters LaMotte and GW. This relationship was also shown in Experiment 1, Figure 3.2. However, when meters were calibrated to PCALs, the range of readings per sample narrowed (see Table 3.5). This may be due to formazin resembling sediment found in surface waters. Again, AMCO EPA solutions have uniform microspheres ranging from 0.02 to 0.2 μ m; whereas formazin particles are irregular in shape and range from particles less than 0.2 to 50 μ m. Therefore, formazin may more closely match the range of particulates found in collected field samples.

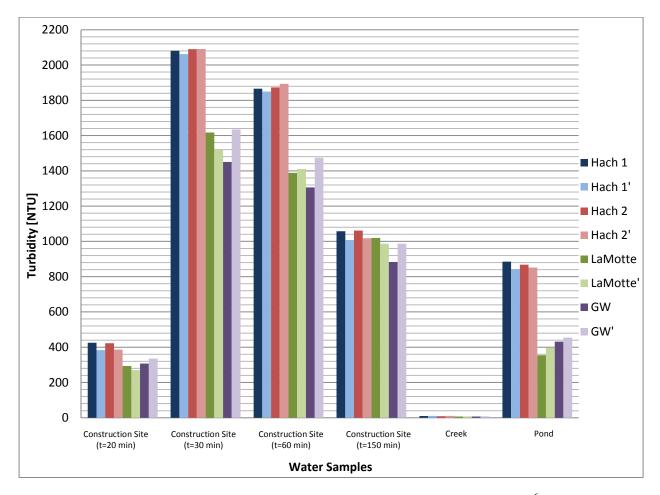


Figure 3.3. Turbidity measurements from each meter using field samples⁶.

	Const. Site (t=20 min)	Const. Site (t=30 min)	Const. Site (t=60 min)	Const. Site (t=150 min)	Creek	Pond
Calibrated with SCALs	71.5	326	304	83.5	1.43	281
Calibrated with PCALs	54.2	292	250	15.4	1.40	244

⁶ Hach 1, Hach 2, LaMotte and GW refer to the meters calibrated to their provided SCALs. Hach 1', Hach 2', LaMotte' and GW' refer the meters calibrated to the created PCALs.

⁷ Derived calculations are shown in Appendix A; Table A.12 and Table A.14.

Furthermore, two sediment basin composite samples collected 30 minutes and 60 minutes after a rainfall event were relatively turbid, due to samples' initial readings exceeding Hach's instrument range, and had to be diluted by a factor of 4 and 5, respectively. According to Method 180.1, turbidity of the original sample is then computed from turbidity of the diluted sample and dilution factor (EPA, 1993b). For example, if 5 volumes of turbidity-free water were added to 1 volume of sample and the diluted sample showed a turbidity of 30 NTU, then the turbidity of the original sample was 180 NTU (EPA, 1993b). Therefore, differences between Hach meters to LaMotte and GW meters were pronounced due to the multiplication factor of the actual turbidity readings and possible errors introduced with dilution techniques.

In addition, dilution procedures did not follow Method 180.1's protocol to dilute samples until turbidity readings fell below 40 NTU mainly because of the difficulty to get field samples below this value. Turbid samples that were diluted by a factor of 4 and 5 produced turbidity readings in a range of 200 to 500 NTU before multiplication. To dilute these samples even farther to obtain values 40 or below would potentially introduce many errors. From a practical standpoint issues related to dilution and the potential for compounded error for turbid field samples need to be avoided.

However, for meters designed with the AMCO EPA standards, before and after dilution turbidity readings did not display similar proportionality. For example, as shown in Figure 3.2, pond sample turbidity for meters Hach 1, Hach 2, LaMotte and GW read 886, 868, 354, and 432 NTU, respectively. Then, samples were diluted by a factor of two. After incorporating the multiplication of two with the diluted readings, the computed

actual turbidity readings were 813, 816, 534, and 619 NTU for meters Hach 1, Hach 2, LaMotte and GW. Therefore, according to EPA's Method 180.1 proportionality, meters Hach 1, Hach 2, LaMotte and GW are 8.3, 5.9, 50.7, and 43.2 percent different from the undiluted reading. Therefore, LaMotte and GW may be altogether inaccurate for higher turbidity readings of surface water samples, further supporting the design of AMCO EPA standards to specifically resemble remaining particulates in finished treated drinking water that are primarily submicron in size following filtration (AMCOClear, 2010).

CONCLUSIONS

Given potential impact of proposed numeric effluent limitations related to turbidity on construction activities, the goal of this research was to quantify accuracy and precision of selected instruments in routine surface water analysis and potential regulatory compliance. Disclaimer: Mention of a trade name does not imply endorsement of the product by Clemson University to the exclusion of others that might be available. Users are encouraged to fully evaluate the suitability of any equipment for their intended application.

When subjected to both calibration experimental procedures, meters Hach 1, Hach 2 and GW provided accurate results for their overall averaged percent differences about true NTU values, but were not within their claimed $\pm 2\%$. The LaMotte meter was not as statistically accurate by comparison. Though overall averages improved when calibrated to the PCALs, LaMotte remained inaccurate for higher turbidity readings. Possible inaccuracy results for the LaMotte may result from its provided SCALs. The highest SCAL provided was 100 NTU, even though range of the instrument is 0 to 2000 NTU.

As for meter precision, Hach 1, Hach 2, and GW were precise under both calibration studies and were overall within the claimed $\pm 1\%$ of measured values; whereas the LaMotte was slightly above $\pm 1\%$ of the measured values for both calibration studies. Analysis showed that even though meters may be very precise, the meters could be inaccurate. If inaccuracy is compounded by precision, misinterpretation of results is likely.

51

With an understanding of each meter's accuracies and precisions to formazin NTU values, instruments were tested with field water samples of unknown turbidity or concentration values to observe their behavior for future analysis. Readings produced by the Hach meters had very similar results for either calibration study, but were consistently higher than the LaMotte and GW meters. However, when meters were calibrated to PCALs, the range of readings per sample narrowed, suggesting that formazin is a better calibration standard. Also, the LaMotte and GW meters may provide inaccurate results for higher turbidity readings that contain a wider range of particle sizes. This again suggests that AMCO EPA standards more adequately resemble treated drinking water samples. In addition, the collected field samples were not diluted until turbidity readings fell below 40 NTU mainly because of the difficulty to get field samples below this value. Issues related to dilution and the potential for compounded error for turbid field samples need to be avoided. Therefore, it is recommended that Method 180.1 be reevaluated in order to clarify standards used for either drinking water or surface water evaluations.

CHAPTER 4

TURBIDITY ANALYSES BASED ON SEDIMENT CHARACTERISTICS OF REPRESENTATIVE SOUTH CAROLINA SOILS

ABSTRACT

EPA has published effluent limitation guidelines (ELGs) to control discharge of pollutants from construction sites. Numeric turbidity limits for construction site discharge are expected to be required in the near future Such requirements will likely include subjecting construction site stormwater discharges to a maximum allowable turbidity numeric effluent limit in nephelometric turbidity units (NTUs) for sites disturbing 10 acres or more.

Turbidity is a measurement used to quantify water clarity. Turbidity is an expression of the optical properties of a liquid that causes light rays to be scattered and absorbed as measured by a nephelometer. Though constituents such as organic matter can impact water clarity, typically the inorganic fraction derived from particulate matter such as sediment dominates turbidity levels in surface waters. Recognition that water clarity is an important indicator of environmental health has increased, resulting in growing demand for high-quality objective measurement.

Given the importance of pending numeric effluent limitations related to turbidity, focus of this research is to determine relationships between representative South Carolina soils and corresponding turbidity as a function of suspended sediment concentration and settling time.

The relationship between turbidity and suspended sediment concentration is complex. Experimental results indicate this relationship is well correlated when top soil and subsoil trends were based on predominant South Carolina region and measured clay content. Derived trends either correlated well with a linear or a log relationship (\mathbb{R}^2 values ranging from 0.7945 to 0.9846) as opposed to a power function from previous research. Therefore, for each region, research confirmed that as concentration of fines increased, turbidity increased; and soils with higher clay content produced higher turbidity values compared to soils with less clay.

As for the correlation of turbidity and settling time, top soil and subsoil results were also separated by South Carolina region and sorted based on the same measured clay content ranges formulated above. All trends correlated well with a power function (R^2 values ranging from 0.7674 to 0.9347). This relationship therefore followed Stoke's Law, where smaller particles remain in suspension longer and contribute more to turbidity as opposed to soils with less clay content. Such results may have significant bearing on the construction and development industry as it prepares for proposed monitoring requirements associated with recently promulgated numeric effluent standards for turbidity.

INTRODUCTION

Accelerated erosion due to construction can potentially cause much damage to the surrounding ecosystem (Haan et al., 1994). For example, introduction of excessive suspended solids (SS) from runoff, such as construction activity, are a potential water pollutant that can cause significant environmental impacts. All streams carry some SS under natural conditions, but if concentrations are elevated from human disturbance, it can lead to alterations to the physical, chemical and biological properties of a water body (Bilotta and Brazier, 2008). Such impacts can include the adsorption of heavy metals, toxic substances, and biological pollutants to soil particles that are then transported downstream. Chemical and biological pollutants can harm water quality and other ecological problems. Clearly, SS are an important pollutant in surface waters; thus, quantifying and monitoring changes in suspended solids is critical for the nation's bodies of water. As a result, water quality regulatory agencies often impose permit conditions that limit the concentration of suspended solids that can be present in effluent waters (Thackston and Palermo, 2000).

Total suspended solids (TSS) resulting from erosion encompass both inorganic solids and organic solids. Inorganic solids may include sand, silt, clay sediment particles, and organic solids can consist of algae and detritus. TSS is computed from laboratory analysis by the dry weight of suspended solids per unit volume of water, and is reported in milligrams of solids per liter of water (mg/L). However, TSS measurements are not routinely used to detect and correct short-term problems or permit violations because

sediment concentrations cannot be determined easily or quickly in the field, and transportation to a laboratory for analysis is time-consuming and can be costly (Thackston and Palermo, 2000). As a result, these traditional methods are increasingly being replaced in favor of accurate, continuously-collected surrogate data for quantification of SSC that may be safer and (or) less expensive to obtain, such as turbidity measurements.

Turbidity measurements are gaining increased usage as an indicator of pollution in surface runoff from disturbed areas such as active construction sites. For example, timely, accurate field estimation of sediment loading could be facilitated through the development of precise relationships between suspended solids and turbidity. This approach has potential for monitoring any water quality constituent whose concentration is better correlated with an easily measured (in situ) parameter, such as turbidity.

Turbidity is an expression of the optical properties of a liquid that causes light rays to be scattered and absorbed rather than transmitted in straight lines through a water sample (Anderson, 2005). Though constituents such as organic matter can impact water clarity, typically the inorganic fraction derived from particulate matter such as sediment dominates turbidity levels in surface waters (Davies-Colley and Smith, 2001). Cloudiness of water results from intense scattering of light by fine particles typically with diameters smaller than 0.050 mm (Davies-Colley and Smith, 2001). Hence, waters with high concentrations of fine suspended sediment are frequently described as turbid. Turbidity is a vivid visual indicator of pollution associated with sediment-laden runoff. Turbidity is now commonly measured in nephelometric turbidity units (NTU) using nephelometric turbidimeters (Mitchell, 2000), currently two of the most frequently used nephelometric turbidimeters include nephelometers and near infrared turbidimeters. For this research, nephelometers are of interest because of their common use in the U.S. as opposed to near infrared turbidimeters that are most common overseas.

Nephelometers measure light that has been scattered at a specific angle of 90 degrees from the main light path (EPA, 1993b). Generally, nephelometers are compliant with EPA Method 180.1 for determining turbidity by nephelometry (Borok, 2010). Along with the specified angle of 90 degrees, nephelometers under EPA's Method 180.1 require the light source to be a tungsten lamp operating at a color temperature between 2200-3000 K. The light source is a light-emitting diode where the light path is designed to minimize stray light falling on the detector (Thackston and Palermo, 2000). Therefore, a zero reading means no light scattered at 90 degrees (\pm 30 degrees) from the main light path and implies no turbidity (Thackston and Palermo, 2000). Nephelometers must be standardized against either a primary formazin suspension with a value of 4000 NTUs (which can be diluted to desired NTU values) or a commercially available polymer standard identified as AMCO-AEPA-1 (EPA, 1993b).

Turbidity is not an inherent property of water, such as temperature or pH. However, the recognition of turbidity as an indicator of the environmental health of water bodies has increased, resulting in a growing demand for high-quality and objective turbidity measurements (Anderson, 2005). Therefore, given the importance of a proposed turbidity limit, the focus of this research is to determine relationships between representative sediments and corresponding turbidity as a function of suspended sediment concentration. Also, in order to separate particles from the runoff water in best management practices (BMPs) such as sediment basins, it is imperative to know sediment settling time and settling properties required to remove the particles from the water column (Haan et al., 1994: Tempel, 2011). Thus, this research will also determine if any relationship exists between turbidity and settling time of suspended sediments.

A standard practice in relating suspended sediment to turbidity is to take the association as a one-to-one relationship (Hayes et al, 2001). Prior research, however, indicates this assumption is not accurate unless site-specific sampling is utilized to establish unique turbidity-suspended sediment relationships because turbidity is not only a function of TSS concentrations, but also of particle size, shape, and composition (Foster et al., 1992: Hayes et al., 2001). Hence, this research is needed to analyze turbidity responses based on sediment characteristics for representative South Carolina soils.

PROCEDURES

Soil Collection

The following section describes the materials and methods used to complete the objectives of relating turbidity to suspended sediment concentration and settling time of representative South Carolina soils. First, the most representative soils for South Carolina were found using U.S. Department of Agriculture's (USDA) Natural Resources Conservation Service (NRCS) soils data for each county in South Carolina. Soils are listed by county with corresponding area provided in acres. Soil area by county was aggregated and ranked based on percent coverage for the entire state. The most predominant twenty five soils represent fifty percent of South Carolina's total area (see Table 4.1). Table 4.1 also provides the soil's predominant region, consisting of the upstate Piedmont Region, the mid-state Central region and the Coastal Plains region (see Figure 4.1) as defined by Russ (2009). Soils were located around the state in order for data collection to proceed. Most were found on Clemson University experiment stations located in various geographic regions across the state. These stations included the Simpson Experiment Station (Anderson, SC), the Pee Dee Research and Education Center (REC) (Florence, SC), and the Edisto REC (Blackville, SC). Other samples were collected in Aiken, Edgefield and Laurens counties. Each representative soil was identified at the experiment stations and other parts of the state using the Soil Conservation Service county soil surveys or NRCS's Web Soil Survey online service.



Figure 4.1. South Carolina physiographic regions (Russ, 2009).

	Soil Series	Percent of SC	Predominant SC Region
1	Cecil	8.940	Piedmont
2	Lakeland	2.936	Central
3	Goldsboro	2.540	Coastal
4	Pacolet	2.507	Piedmont
5	Madison	2.443	Piedmont
6	Rains	2.327	Coastal
7	Lynchburg	2.164	Coastal
8	Wilkes	2.086	Piedmont
9	Coxville	1.929	Coastal
10	Johnston	1.674	Central
11	Georgeville	1.665	Piedmont
12	Troup	1.656	Central
13	Norfolk	1.617	Coastal
14	Hiwassee	1.441	Piedmont
15	Cataula	1.378	Piedmont
16	Blanton	1.368	Central
17	Fuquay	1.334	Central
18	Dothan	1.307	Central
19	Ailey	1.284	Central
20	Vaucluse	1.273	Central
21	Appling	1.205	Piedmont
22		1.156	Central
23	Noboco	1.149	Coastal
24	Herndon	1.110	Piedmont
25	Wagram	1.002	Central

Table 4.1. Representative South Carolina soils utilized for this research in order of area.

When a desired soil series was identified at the sampling location, cover vegetation was removed, and a shovel was used to determine soil profile. Using a shovel, top soil was collected in a nineteen liter bucket, and subsoil was separately placed in another nineteen liter bucket. Soil collection was carefully executed to avoid mixing top soil and subsoil. Lids were tightened on each bucket, and samples were brought back to the lab for analysis.

Data Analysis

First, a wet sieve and pipette analysis were conducted in order to obtain an aggregate size distribution (ASD) for each soil. For a primary particle size distribution, procedures require sodium hexametaphosphate to disperse aggregates formed in order to obtain all particle sizes (Das, 2006). However, samples collected were not dispersed for a primary particle size distribution. As a result, for this research, ASD analysis was conducted using representative soils as they would be found on sites across South Carolina.

Wet Sieve Analysis:

Sieve analysis is applicable to separate granular material from soil mass (Das, 2006). First, 100 grams of each soil sample was measured and mixed with 600 ml of water. The mixture was washed through a series of six sieves with mesh openings of 2.0, 1.0, 0.425, 0.15, 0.075, and 0.063 mm. The procedure consisted of first passing the soil mixture through the 2.0, 1.0 and 0.425 mm mesh openings that were stacked in a five

gallon bucket. Next, the sieves were placed in a second nineteen liter bucket. The first bucket was stirred and its mixture was poured over the sieves in the second bucket. The soil mixture went through these sieves three times. During this process, an additional 400 ml of water was used for the soil mixture to thoroughly pass through the sieves. After running the mixture through these three sieves three times, the sieves were removed from the bucket and the remaining three sieves, 0.15, 0.075, and 0.063 mm, were placed in the bucket. The remaining soil mixture was passed through the smaller sieves three times. The soil-water mixture left in the bucket was measured and saved for pipette analysis to determine sizes less than 0.063 mm. Approximately 1.0 L was used for each pipette analysis.

Once the mixture was passed through all the sieves, the soil remaining on each sieve was collected in a pre-weighed tin and placed in the drying oven at 105 °C for a minimum 24 hours (Bolton, 1979). Once dried, the tins were weighed on a balance to the nearest 0.001 gram and recorded.

Pipette Analysis:

As stated before, sieve analysis was used to separate gravel (particles coarser than 2 mm) from particles less than 2 mm in diameter. Percent sand was isolated by wet sieving (Scott, 2000). However, particles less than 0.063 mm were of particular interest for this work because larger particles will settle from surface water flow relatively quickly leaving smaller particles that contribute as sources of turbidity (Hayes et al., 2001). Therefore to determine the proportion of silt and clay in the sample, a pipette

procedure was conducted. Silt and clay can be separated by settling in water, where the rate of settling is governed by Stoke's Law (Scott, 2000). For the pipette procedure, once the soil-water mixture had been thoroughly mixed, all particles of one size will have fallen below a certain level in the suspension after a certain amount of time has elapsed (Scott, 2000).

Therefore, the remaining 1 L soil-water mixture from the wet sieve analysis was completely mixed, collected and used for pipette analysis. Pipette withdrawals followed the schedule shown in Table 4.2 to obtain the size distribution range between 0.063 mm and 0.002 mm. Samples were assumed to be collected at a temperature of 21 degrees Celsius since this temperature was the approximate room temperature. The particular sizes displayed in Table 4.2 represent the largest particle size collected at the elapsed time (Johns, 1998). Depth is the distance the 25 ml pipette tip was lowered from the actual (declining) water surface at each sample time.

Table 4.2. Pipette analysis schedule at 21°C (USDA, 1979)⁸.

Sediment Size (mm)	0.062	0.031	0.016	0.008	0.004	0.002
Depth (mm)	150	150	150	150	100	50
Time (hr:min:sec)	0:00:42	0:02:48	0:11:14	0:29:58	1:00:00	4:02:00

Once the 25 ml sample was drawn, it was collected in a pre-weighed tin and placed in a drying oven at 105 °C for a minimum 24 hours to evaporate any water present (Bolton, 1979). After each sample was dried, tins were weighed on a balance to the

⁸ All tests withdrew 25 ml from the soil-water mixture

nearest 0.0001 gram and results were recorded. The remaining soil-water mixture in the 1L graduated cylinder was saved for subsequent analysis. After the completion of the wet sieve and pipette analysis for each soil, results were computed to create an ASD. ASD examples for each region's dominant soil series are displayed in Figures 4.3 to 4.5. Figure 4.6 illustrates differences among soil's ASD and eroded PSD. Tabular data for all soil's ASDs are found in Appendix B.

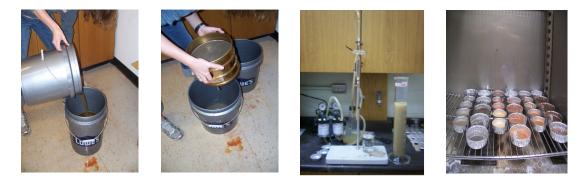


Figure 4.2 Wet sieve and pipette analysis summary photographs.

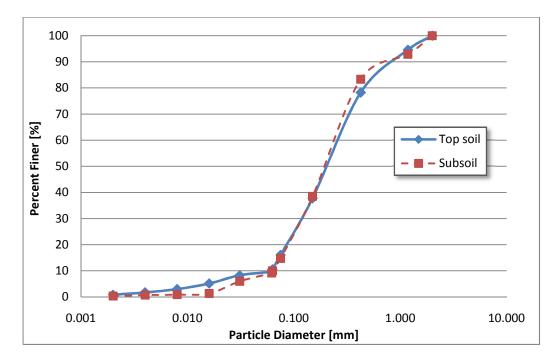


Figure 4.3. Piedmont: Cecil's aggregate size distribution (ASD).

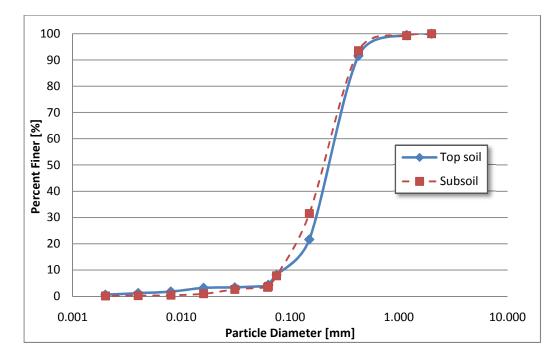


Figure 4.4. Central: Lakeland's aggregate size distribution (ASD).

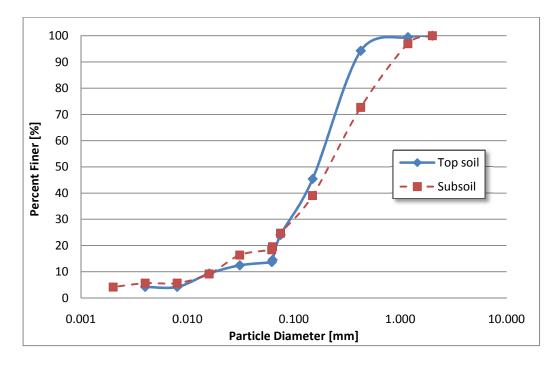


Figure 4.5. Coastal: Goldsboro's aggregate size distribution (ASD).

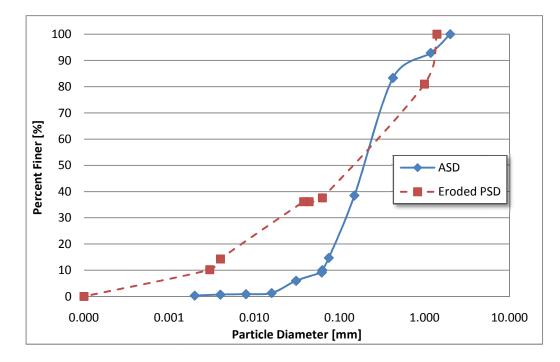


Figure 4.6. Size distribution comparison for Cecil subsoil.

Hydrometer Analysis

As stated previously, the particle size distribution obtained from the wet sieve and pipette analysis was not a primary particle size distribution, but rather a modified eroded particle size distribution that would resemble sediment found in surface waters during active construction phases. To gather additional information on characteristics of each soil series, primary particle sizes of selected soils were established through performance of a hydrometer analysis (Scott, 2000). Hydrometer analysis uses a dispersed sample of soil that has been thoroughly mixed with water in a tall glass cylinder and allowed to settle. Soil particles in a dispersed state in water will settle individually according to Stoke's Law (Das, 2006). Therefore, after specified settling times, density (g/L) of each suspension is measured with a hydrometer to determine the mass of particles remaining in suspension.

For the hydrometer method, 50 g of soil for each series was used to determine soil particle analysis using an ASTM 152-H hydrometer. To achieve separation among soil particles, 40 g/L of sodium hexametaphosphate was utilized. Procedures followed ASTM's Standard Test Method for Particle-Size Analysis of Soils (ASTM D 422-63, 2004), and hydrometer readings were recorded 40 seconds and 2 hours after settling. At 40 seconds, sand-sized particles will have settled out of the suspension leaving only silt and clay sized particles. After 2 hours of settling, the silt sized particles have settled leaving the hydrometer to read the density of clays remaining in suspension. Due to hydrometers being calibrated to 20 degrees Celsius and because density and viscosity of water change with temperature, temperatures were recorded for each reading to correct

the hydrometer readings (hydrometer paper). Readings were corrected for temperatures above 20 degrees Celsius using the following equation (Equation 4). The primary particle size distribution results from the hydrometer analysis are in Appendix B.

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Corrected hydrometer reading(g/L)
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= mesaured reading (g/L)+ [(measured temperature - 20°C) * 0.36 g/L]

(4)



Figure 4.7. Hydrometer analysis photographs.

Turbidity versus Settling Time

To derive empirical relationships between turbidity and settling time of selected South Carolina soils, the procedure utilized the remaining soil-water mixtures from pipette analysis. Again, both top soil and subsoil were examined for this experiment. To obtain turbidity readings the Hach 2100Q was chosen based upon determining meter's reliable accuracies and precisions. It was also chosen because its secondary calibration standard of a stable formazin was found to be better suited to mimic sediment found in surface water samples. The Hach 2100Q was calibrated daily with its provided calibration standards (StabCal) as specified in the manufacturer's operating instructions.

Once the pipette method was complete, the graduated cylinder containing the soilwater mixture was filled back up to 1 liter with distilled (DI) water and repeatedly inverted until solution was completely mixed. Fifteen milliliters of the soil-water mixture was drawn at specified times from two inches below the surface of the sample and placed in the Hach 2100Q's cuvette. Specified times spanned two weeks and included readings taken at 0 min, 5 min, 30 min, 1 hour, 2 hour, 4 hour, 24 hours, 48 hours, 4 days, 7 days, and 14 days. A depth of two inches below the water surface level was chosen to minimize surface and edge effects from the glass cylinders. This depth would also mimic surface withdrawal from sediment basins through such practices as skimmers and flashboard risers.

The cuvette containing each sample was inverted ten times between readings to adequately resuspend and disperse all sediment particles before placement in turbidimeters (Hayes et al., 2001). The Hach 2100Q cannot measure turbidities higher than 1000 NTU. Therefore, if a reading exceeded this range, dilutions were made from the cuvette to follow EPA's Method 180.1 proportionality example to obtain sample's actual turbidity reading. For instance, if the soil-water solution was diluted by half, then the diluted sample's turbidity reading was multiplied by two. However, for all turbidity readings, dilution procedures did not follow EPA's Method 180.1 protocol to dilute samples until turbidity readings fell below 40 NTU. This was mainly because of the difficulty to get soil-water samples below this value. From a practical standpoint, issues related to dilution and the potential for compounded error for turbid samples need to be avoided.

Five readings were recorded for each sample and averaged. For each soil's top soil and subsoil, results were plotted on a graph. Each sample was saved for utilization in subsequent TSS analysis.

Turbidity versus Total Suspend Solids Concentration

To develop unique relationships for concentration of South Carolina suspended sediments versus turbidity, analysis was conducted in sequence with the turbidity versus settling time procedure. After the five turbidity readings were recorded, 10 ml of each sample was drawn using a 5 ml Eppendorf Pipette, placed in a pre-weighed dish and dried in the oven at 105 degrees Celsius for a minimum of 24 hours (Bolton, 1979). After the sample was dried, tins were removed from the oven using tongs to avoid additional errors. Samples were weighed on a balance to the nearest 0.0001 gram to obtain concentrations in mg/L. Resulting concentrations were plotted on a graph versus its corresponding turbidity from the procedure above.

Turbidity versus Particle Size

To better understand sediment effects on turbidity, it was important to analyze turbidity response to particle size. Particle diameters analyzed for this project were derived using the parameters selected in the turbidity versus settling time procedure. Knowing samples were drawn from a depth of two inches below the water surface level, Stoke's Law was applied for the specified time intervals of 0 min, 5 min, 30 min, 1 hour, 2 hours, and 4 hours to calculate the particle diameters of 0.063, 0.014, 0.006, 0.004, 0.003 and 0.002 mm, respectively. Altogether, at the specified times, particle sizes were compared to their corresponding turbidities and concentrations that were determined from the above procedures. However, actual turbidity and concentration values for the particle sizes needed to be determined. This is because when samples were drawn at the given times; the particle size calculated included that size, along with any sizes smaller that also remained in suspension. Therefore, a subtraction method was adopted to determine the particle sizes true turbidity and concentration values ('Actual Conc.' and 'Actual Turb.' in Table 4.3). If the subtraction method produced a negative value, values were assigned zero, indicating that particle diameter did not contribute to turbidity. Because all soils vary in turbidity and concentrations for a given time, in order to compare all soils, each particle diameter was plotted versus the ratio of turbidity to concentration.

Actual Actual Actual Turb/ Conc. Turbidity D [mm] Conc. Turbidity [NTU] [mg/L] **Actual Conc** [mg/L][NTU] 0.063 6510 2240 6818 292 0.130 4270 1.151 0.014 1830 6526 2106 0.006 2440 4420 1.217 720 876 1.148 0.004 1720 210 3544 241 0.003 993 2.207 1510 450 3303 0.002 1060 1060 2310 2310 2.179

Table 4.3. Tabular data of a soil sample's turbidity and concentration response to particle

size.

RESULTS AND DISCUSSION

Turbidity with Respect to Total Suspended Solids Concentration

Selected soils were first divided by top soil and subsoil and plotted to determine if any relationships existed between suspended solids concentration and turbidity (Figure 4.4). This approach was taken to evaluate if the representative South Carolina soils collectively could be classified by certain properties. Factors taken into consideration encompassed the experimental particle size distribution values, primary particle size distribution values from the hydrometer analysis, South Carolina region, soil classification (soil order, suborder, great group, subgroup) and soil family (particle size and mineralogy).

Measured turbidity averages and calculated sediment concentrations were plotted and used in a regression analysis to model turbidity as a function of sediment concentration. These values can be viewed in Appendix C. First, trends were arbitrarily chosen based on similar turbidity responses to suspended sediment concentration. However, trends could not be explained because there were no common denominators to classify the soils' relationship of turbidity with suspend solids concentration. As a result, top soils and subsoils were examined depending on South Carolina regions.

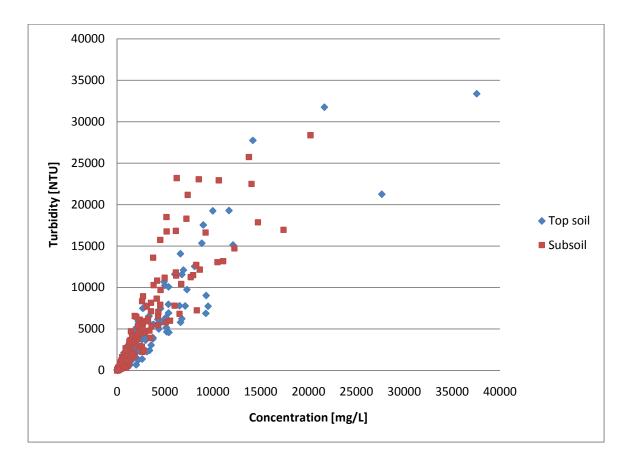


Figure 4.8. All twenty five selected top soil and subsoil's relationship of turbidity and suspended sediment concentration.

Once soils were divided by South Carolina regions, top soil and subsoils were sorted based on 1) measured clay content, 2) measured fine content, and 3) the ratio of measured clay to fines to establish any trends from each scenario. Measured values were derived from ASD. Each scenario was evaluated and it was found that for each region, trends were best formulated from soils' measured clay content compared to the other two scenarios. Fines content would be expected to explain the relationships of turbidity to concentration because both silt and clay particle sizes contribute to turbidity. However, this scenario did not rationalize the behavior of certain subsoils that aggregated and quickly settled from suspension. Aggregates formed in these subsoils contributed to a higher percentage of silt in the upper range of 0.016 to 0.05 mm, but settled rapidly, leaving very small amounts of clay in suspension. As a result, these soils would contribute to high turbidity and high concentration readings, but values would then substantially drop after the first minutes. Hence, when subsoils were sorted according to the amount of measured fines, poor correlations resulted because aggregated soils were incorrectly grouped with soils that had both higher amounts of silt and clay. Figure 4.9 depicts an example of a subsoil that aggregated and quickly settled compared to its top soil.

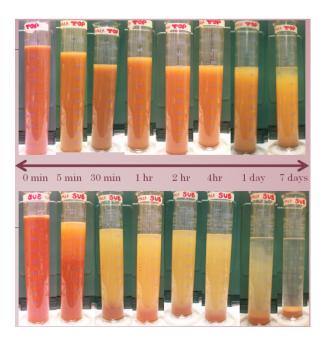


Figure 4.9. Example of a subsoil (image on bottom) that aggregated and settled quickly compared to its top soil (image on top).

Based upon correlation coefficients of the predicted equations, trends correlated well with either a linear or a log relationship. These relationships are plotted in Figures 4.10-4.15⁹. From the figures, not all results followed a one to one relationship, as previous research had assumed (Kundell, 1995). Yet, soils that did conform to a linear relationship were not far off from a one to one relationship. A power function in the form of $T=a*SS^b$ has previously been found to correlate sediment concentration and turbidity (Hayes, et al. 2001 and Mitchell 2000). However, this relationship proved inconsequential and was not utilized because, even though it produced high correlation coefficients, it did not accurately model the behavior of soils with higher clay content. For the most part, these soils in their higher turbidity and concentration range behaved asymptotically rather than turbidity continuing to increase with increasing concentration.

⁹ For Figures 4.10-4.15, n refers to the number of soil samples. Each soil sample contains 11 data points.

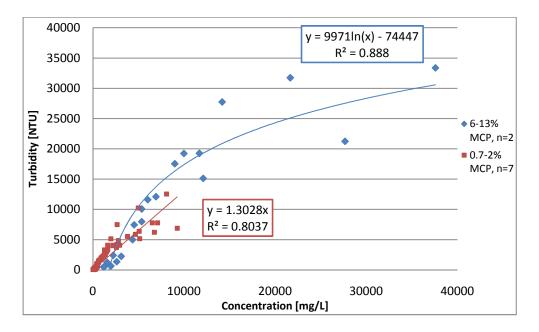
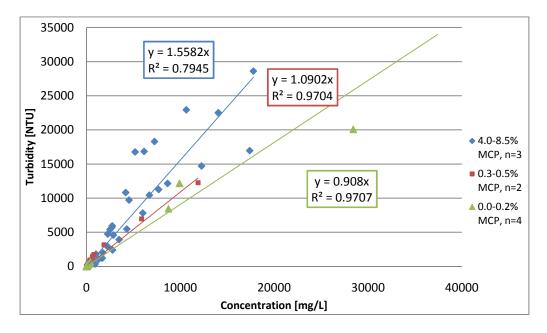
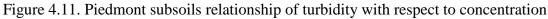


Figure 4.10. Piedmont top soils relationship of turbidity with respect to concentration

(n=number of soils). Trends were formulated based on MCP.





(n=number of soils). Trends were formulated based on MCP.

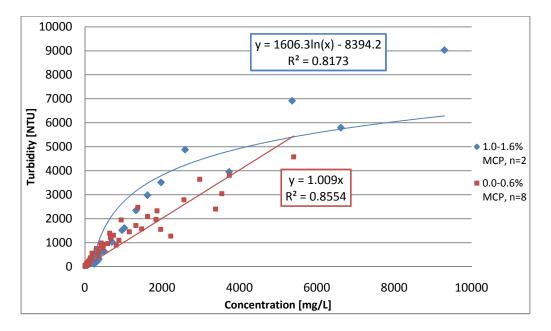
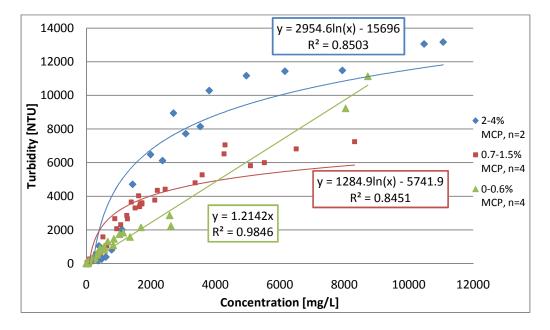
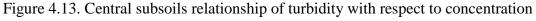


Figure 4.12. Central top soils relationship of turbidity with respect to concentration

(n=number of soils). Trends were formulated based on MCP.





(n=number of soils). Trends were formulated based on MCP.

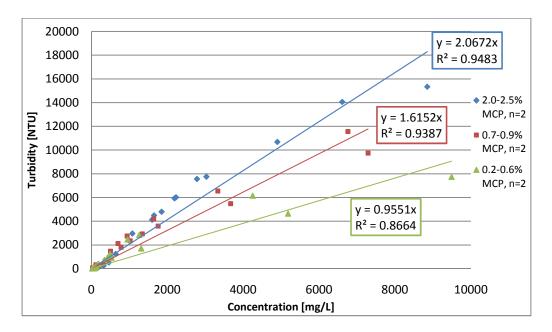
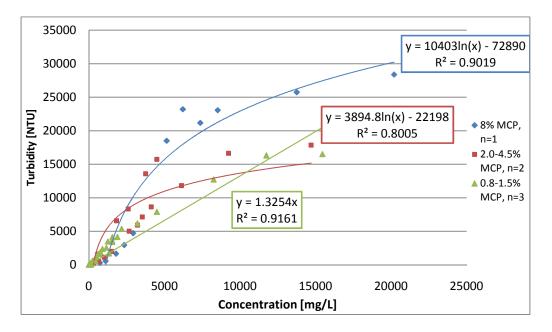
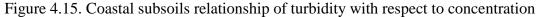


Figure 4.14. Coastal top soils relationship of turbidity with respect to concentration

(n=number of soils). Trends were formulated based on MCP.





(n=number of soils). Trends were formulated based on MCP.

Turbidity with Respect to Settling Time

To determine if an empirically derived relationship between settling time and turbidity existed for the selected soils, measured turbidity averages from above and their corresponding time values were evaluated. These values are plotted in Figure 4.16 and raw data may be viewed in Appendix D. In addition, Figure 4.16 shows that majority of the soils were not below a value of 280 NTU until 24 hours, or later, of settling. In other words, it would take longer than a day for the soils to be below EPA's effluent limitation guideline of 280 NTU.

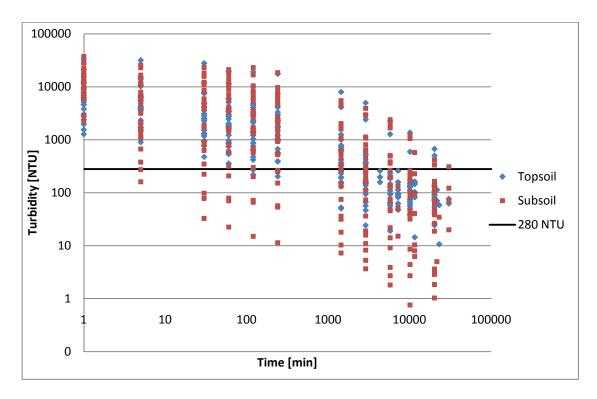


Figure 4.16. The twenty five selected South Carolina soil's top soil and subsoil's turbidity relationship to settling time¹⁰.

¹⁰ For Figure 4.16, 0 minute values were plotted as 1 to be shown on logarithmic scale

To remain consistent with classifying the soils, results were divided by South Carolina regions and the top soil and subsoils were sorted based on the same measured clay content ranges formulated previously. Resembling the relationships established above from turbidity with respect to suspend sediment concentration, it was found that for each region, trends were best formulated from soils' measured clay content as well.

Based upon correlation coefficients of the predicted equations, all trends correlated well with a power function. This suggests the relationship is governed by Stoke's Law; where smaller particles remain in suspension longer and contribute more to turbidity compared to soils with less clay content. These relationships are displayed in Figures 4.17-4.22¹¹ on log-log plots. For Figures 4.17-4.22, 0 minute values were assigned a value of 1 in order to be able to be shown on a logarithmic scale.

¹¹ For Figures 4.17-4.22, n refers to the number of soil samples. Each soil sample contains 11 data points.

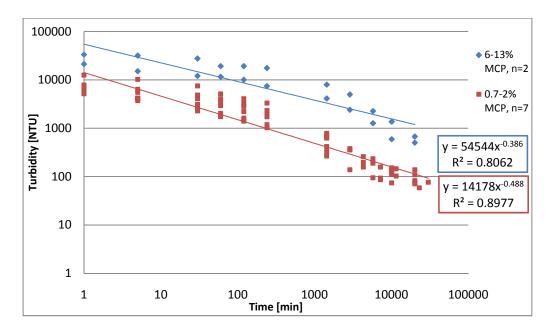


Figure 4.17. Turbidity as a function of settling time for Piedmont top soils (n= number of soils). Trends were formulated based on measured clay percentages (MCP).

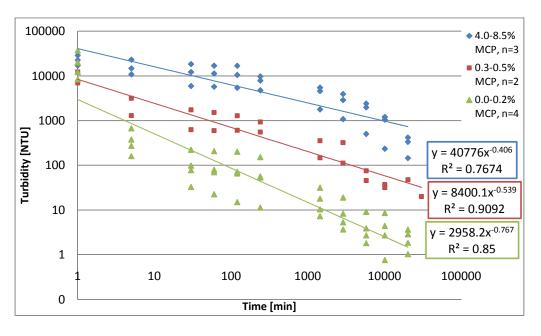


Figure 4.18. Turbidity as a function of settling time for Piedmont subsoils (n= number of soils). Trends were formulated based on measured clay percentages (MCP).

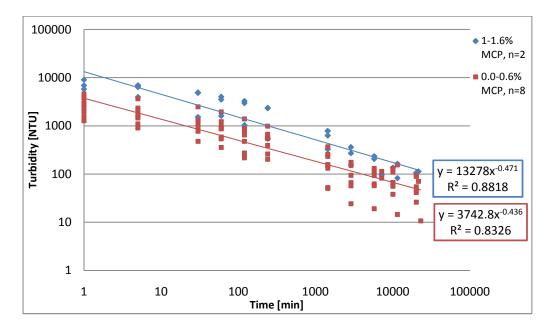


Figure 4.19. Turbidity as a function of settling time for Central top soils (n= number of soils). Trends were formulated based on measured clay percentages (MCP).

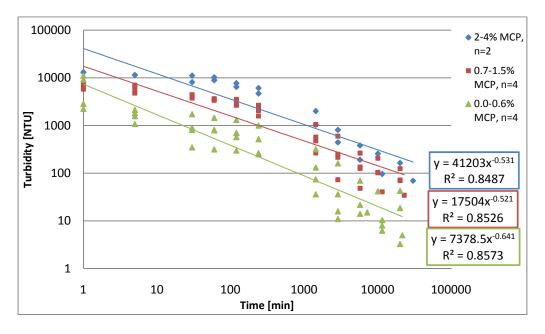


Figure 4.20. Turbidity as a function of settling time for Central subsoils (n= number of soils). Trends were formulated based on measured clay percentages (MCP).

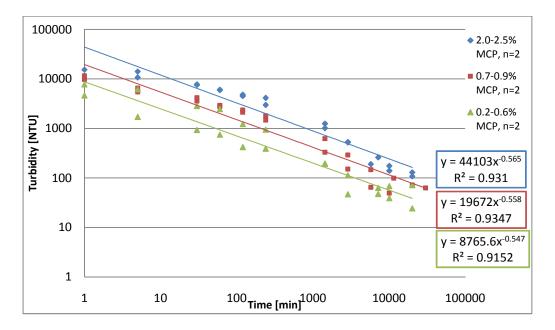


Figure 4.21. Turbidity as a function of settling time for Coastal top (n= number of soils). Trends were formulated based on measured clay percentages (MCP).

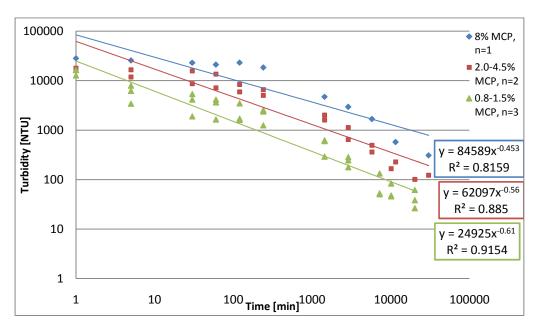


Figure 4.22. Turbidity as a function of settling time for Coastal subsoils (n= number of soils). Trends were formulated based on measured clay percentages (MCP).

Figures 4.17-4.22 illustrate noticeable differences among the smallest percentages of measured clay content. However, for both relationships of turbidity versus suspended sediment concentration and turbidity versus settling time, the trends formulated from measured clay content have gaps between the ranges. To possibly explain gaps between measured clay content or why certain soils aggregated, both soils' primary particle size distribution and eroded particle size distribution values were examined. If either of these values could describe the ranges that were formulated, then lab analysis could possibly be avoided in order to predict the behavior of all soils in the respective South Carolina Nevertheless, primary particle size distribution and eroded particle size regions. distribution values could not relate to the ranges obtained for this project (tabular data shown in Appendix B). An aggregate size distribution would have to be executed in order to use the empirically derived correlations established in this project. If this procedure is performed and the soil's ASD clay content does not follow within the empirical ranges, interpolation between trendlines could be conducted to predict the soil's behavior with respect to turbidity. However, in Figures 4.13 and 4.15, trendlines cross due to high initial values from the certain subsoils that aggregated and quickly settled. Therefore, if interpolation was needed where lines cross, still use the trendlines that the measured clay content lies between.

Turbidity with Respect to Particle Size

After the fulfillment of analyzing turbidity with respect to concentration and settling time, a potential predictive model of turbidity's response to particle size was evaluated to possibly support the previous findings. Prediction models, such as SEDIMOT II, utilize particle diameters from an eroded particle size distribution to estimate effluent sediment loads from best management practices. Therefore, it would be beneficial if a correlation existed between particle diameter and turbidity for use in prediction models. As stated earlier, to be able to compare all selected soils, the ratio of turbidity to concentration was plotted versus particle diameter (see Figure 4.23).

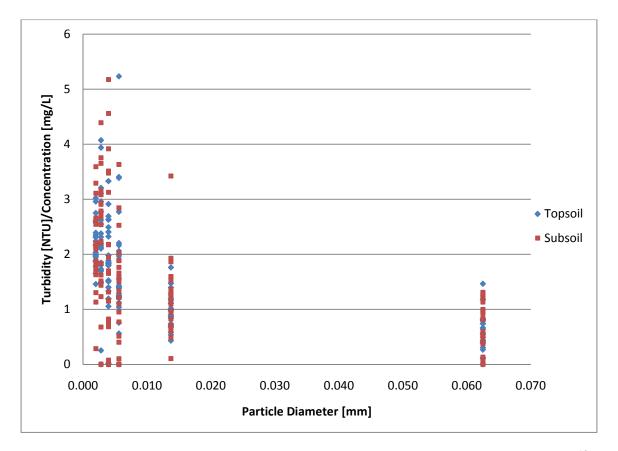


Figure 4.23. The ratio of turbidity to concentration versus particle diameter for all soils¹².

¹² Two values above a ratio of 6 were allocated as outliers and were not plotted on Figure 4.19.

First, top soils and subsoils were evaluated collectively, but no correlations were determined. As a result, soils were once again evaluated depending on South Carolina regions, and the top soil and subsoils were sorted based on the same measured clay content ranges formulated above. Unlike trends found in the above analyses, correlation coefficients of the predicted equations did not correlate as well with a classification based on measured clay content. These relationships are displayed in Figures 4.24-4.29. Because a ratio of zero indicated a particle diameter not contributing to turbidity, these ratios were not plotted. Appendix E contains the modified and raw data used to create Figures 4.25-4.29. For the most part, the figures seem to follow an increasing trend from larger particles to smaller particles. As particle size decreases, the ratio of turbidity to concentration increases. Soils were chosen to fit a power function as opposed to a logarithmic regression because the power function equations will not produce negative ratios if the equation was carried out for larger particle sizes. However, soils viewed individually (Appendix E) do not follow this, or any apparent, trend mainly due to the subtraction method that was computed. Errors were most likely introduced with the lab procedure of pulling a sample at a certain depth and time to calculate its corresponding particle diameter because the true amount of a particle was still unknown even though a subtraction method was performed. Hence, correlations were inconclusive and future analysis is encouraged to refine this relationship of turbidity/concentration to particle size for it to be used in prediction models.

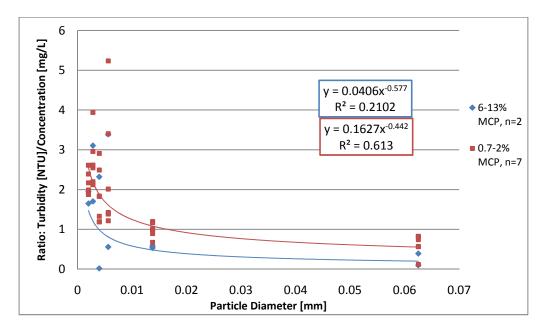


Figure 4.24. Piedmont top soil's relationship of particle diameter to y, where y is the ratio of turbidity to concentration (n= number of soils). Trends were evaluated based on MCP.

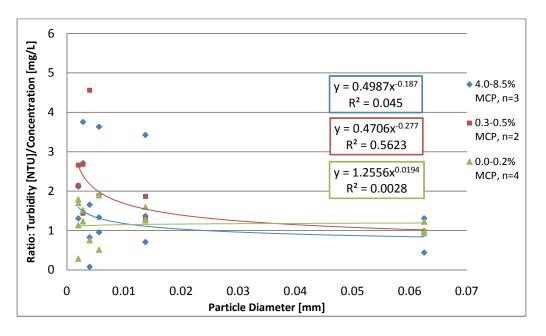


Figure 4.25. Piedmont subsoil's relationship of particle diameter to y, where y the ratio of turbidity to concentration (n= number of soils). Trends were evaluated based on MCP

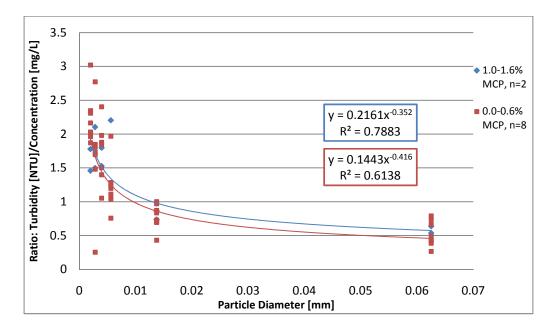


Figure 4.26. Central top soil's relationship of particle diameter to y, where y the ratio of turbidity to concentration (n= number of soils). Trends were evaluated based on MCP.

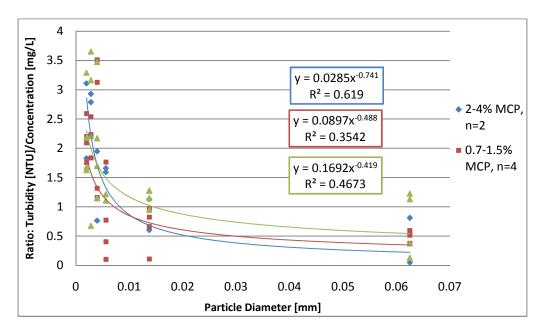


Figure 4.27. Central subsoil's relationship of particle diameter to y, where y the ratio of turbidity to concentration (n= number of soils). Trends were evaluated based on MCP.

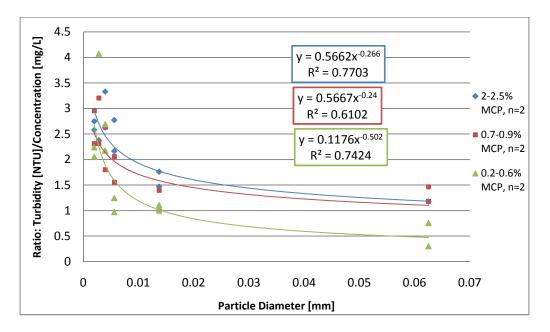


Figure 4.28. Coastal top soil's relationship of particle diameter to y, where y the ratio of turbidity to concentration (n= number of soils). Trends were evaluated based on MCP.

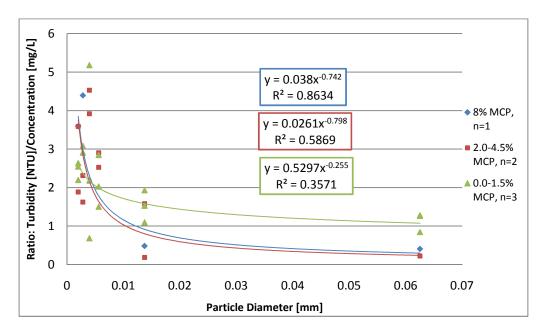


Figure 4.29. Coastal subsoil's relationship of particle diameter to y, where y the ratio of turbidity to concentration (n= number of soils). Trends were evaluated based on MCP.

CONCLUSIONS

The goal of this research was to determine if relationships could be established to relate sediment concentration and sediment settling time to turbidity based on sediment characteristics of representative South Carolina soils.

For the relationship of suspended sediment concentration to turbidity, soils were first divided by top soils and subsoils to determine if the behavior of the selected soils could be collectively classified by soil characteristics. However, relationships found among top soils and subsoils could not be explained or classified because common physical properties could not be established. Further analysis found correlations among top soil and subsoil based on predominant South Carolina region and measured clay content. Such results are supported by previous research that relationships must be derived from site specific characteristics. Based on correlation coefficients of the predicted equations, trends correlated well with either a linear or a log relationship (\mathbb{R}^2) values ranging from 0.7945 to 0.9846). Therefore, for each region, research confirmed that as concentration of fines increased, turbidity increased; and soils with higher clay content produced higher turbidity values compared to soils with less clay. As opposed to measured fines content (silt and clay), empirically derived measured clay content best modeled the behavior of soils' turbidity with respect to concentration. This was because fines content could not rationalize the behavior of soils that aggregated and quickly settled from suspension.

As for the correlation of turbidity and settling time, to remain consistent with classifying the soils, they were once again evaluated depending on the assortment determined from the above analyses. Therefore, results were separated by South Carolina region where the top soil and subsoils were sorted based on the same measured clay content ranges formulated above. Based on the correlation coefficients of the predicted equations, all trends correlated well with a power function (R^2 values ranging from 0.7674 to 0.9347). This relationship therefore followed Stoke's Law, where smaller particles remain in suspension longer and contribute more to turbidity as opposed to soils with less clay content.

It was also determined that majority of the selected soils did not below a value of 280 NTU until 24 hours, or later, of settling. In other words, it would take longer than a day for the soils to be below EPA's effluent limitation guideline of 280 NTU.

For both relationships of turbidity versus suspended sediment concentration and turbidity versus settling time, trends formulated from measured clay content have gaps between the ranges and could not be explained by either soil's primary or eroded particle size distributions. Therefore, if relationships formulated in this project are to be used for future predictive purposes, a modified eroded particle size distribution would be required. Also, if this procedure is performed and the soil's ASD clay content does not fall within the empirical ranges, to account for the gaps in clay content ranges, interpolation between trendlines could be conducted to predict the soil's behavior with respect to turbidity, concentration, and settling time.

92

Lastly, the relationship of turbidity to particle size was evaluated to potentially support findings of turbidity to sediment concentration and settling time. Once again, top soils and subsoils were examined based on South Carolina region and measured clay content. It was found that the ratio of turbidity to concentration versus particle diameter did not correlate well (R^2 values ranging from 0.0028 to 0.8634 from power functions), resulting in inconclusive relationships. Due to potential procedural errors, it is encouraged to refine this relationship of turbidity/concentration to particle size in order for it to be used in future prediction models. Better separation of particle sizes, as opposed to conducting a subtraction method, should yield in better results for determining particle size's effect on turbidity.

CHAPTER 5

SUMMARY CONCLUSIONS

The overall goal of this research was to determine relationships between turbidity and sediment concentration and sediment settling time, based on properties of representative South Carolina soils. Before this could be achieved, accuracy and precision of commercially available nephelometers were evaluated for use in subsequent analysis and potential regulatory compliance. The overall goal was met, and the following conclusions can be made from the from instrument analysis results.

- 1. Analysis of instruments' accuracy and precision showed that even though meters may be precise, there is potential for inaccuracy. If inaccuracy is compounded by precision, misinterpretation of results is likely.
- 2. Meters that performed well provided accurate and precise results for both calibration studies. As a result, secondary calibration standards (SCALs) did provide calibration results equivalent to those obtained when the instrument was calibrated with the primary calibration standard (PCAL) of formazin.
- 3. When subjected to varying field surface water samples, readings produced by meters calibrated to formazin standards produced higher, but consistent, NTU values compared to meters calibrated to styrene divinylbenzene (SDVB) standards. However, when all meters were calibrated to formazin PCALs, range of readings per sample narrowed. These results suggested that differences from meters are most likely due to differences in formazin standards and SDVB

standards. It was also concluded that formazin standards may be a better standard for surface water analysis.

After meter performance conclusions were determined, the following conclusions can be made from the soil-water analysis results.

- 4. It was found that the relationship of turbidity versus sediment concentration of selected South Carolina soils were well correlated when top soil and subsoils were classified by South Carolina physiographic region and measured clay content. As a result, for each region, research suggests that as concentration of fines increase, turbidity increases; and soils with higher clay content produce higher turbidity values compared to soils with less clay.
- 5. Based on correlation coefficients of the predicted equations, trends for suspended sediment concentration to turbidity correlated well with either a linear or a log relationship (R^2 values ranging from 0.7945 to 0.9846) as opposed to previous research utilizing a power function or the assumption of a one-to-one relationship.
- 6. When TSS Cecil concentrations from a dry sieve analysis were created and tested with the four commercially available turbidity meters, results were not equivalent to turbidity and concentration values that were obtained from soil-water remains of a wet sieve analysis. For example, Cecil subsoil that was dry sieved, at higher concentration values, turbidity was roughly half of its concentration value. For Cecil soil that was wet sieved, at higher concentration values, turbidity was roughly half of its concentration value.

likely due to the difference in sieve procedures. The wet sieve analysis would have disassociated more clay particles to pass through sieve openings as opposed to the dry sieve analysis.

- 7. Trends that correlated well with a linear relationship for suspended sediment concentration and turbidity can follow the dilution protocol in EPA's Method 180.1. Where, if a sample is diluted by half, then its turbidity reading is multiplied by two. Thus, following a linear proportionality. However, this would not be the case for trends that correlated well with a log relationship.
- 8. For correlations of turbidity and sediment settling time, trends also correlated well when top soil and subsoils were classified based on their predominant South Carolina physiographic region and measured clay content. From the correlation coefficients of the predicted equations (R² values ranging from 0.7674 to 0.9347), all trends correlated well with a power function. This suggests that the relationship was governed by Stoke's Law; where smaller particles remain in suspension longer. As a result, the smaller particles contributed more to turbidity compared to soils with less clay content.
- 9. Majority of the soils were not below a value of 280 NTU until 24 hours, or later, of settling. In other words, it would take longer than a day for the soils to be below EPA's proposed effluent limitation guideline of 280 NTU.
- 10. The empirical relationships found for turbidity versus suspended sediment concentration and turbidity to sediment settling time are expected to work well for predicting the behavior of all South Carolina soils. Correlations can be utilized if

soil's measured clay content is determined from an aggregate size distribution (ASD). Also, if soil's measured clay content does not fall within the empirical ranges, to account for gaps in clay content ranges, interpolation between trendlines can be conducted to predict soil's behavior with respect to turbidity, concentration and sediment settling time.

Altogether, results of this research will provide a step in determining 1) potential site-specific equations relating sediment concentration to turbidity and sediment settling time to turbidity, 2) aid in the design of future best management practices on construction sites, and 3) provide information for potential regulatory compliance. However, future analysis is encouraged to refine turbidity relationships and suggestions are listed below.

- EPA's Method 180.1 is focused on turbidity analyses for drinking water. If future limitation guidelines are proposed for construction site activities, then revisions to EPA Method 180.1 are needed. For example, it was found that formazin may be a better standard for surface water samples as opposed to the other accepted standard, AMCO EPA. Also, it was impractical and difficult to dilute samples below a turbidity value of 40 NTU. Lastly, not all soils' relationship of turbidity and suspended sediment concentration was linear, therefore, cannot follow the dilution proportionality outlined in 180.1.
- 2. The research found well-correlated relationships of turbidity to sediment concentration and turbidity to sediment settling time, but these trends were formulated from specific measured clay content ranges. Therefore, further research is needed to see if the correlations found could be used to accurately

97

predict the responses of other South Carolina soils. It would also be of interest to evaluate if interpolation between trendlines could be executed in order to calculate relationships of other South Carolina soils that don't fall within the empirically derived clay content ranges.

3. Again, from this research it was determined that future laboratory analysis, namely conducting an aggregate size distribution, cannot be avoided if data are going to be used. However, the potential for particle diameter relating to the ratio of turbidity to concentration could be a link to explaining turbidity relationships for prediction models. It is suggested this relationship be further refined with a different lab procedure.

APPENDICES

APPENDIX A.

Tabular data of Experiments 1-4 to determine meters accuracy and precision^{13,14}

Appendix A contains the tabular data calculated in order to evaluate instrument performance of four commonly utilized and commercially available nephelometers. Determining meters' accuracy and precision may be found in the tables for Experiments 3a and 3b. Tabular data for meters' response to surface water samples may be found in the tables for Experiments 4a and 4b.

¹³ Meters used:

Hach 1&2= Hach 2100Q LaMotte= LaMotte 2020e GW= Global Water's Turb 430T

¹⁴ Experiment 1= Meters' turbidity responses to TSS concentrations of a Cecil soil (calibrated to provided SCALs).

Experiment 2= Verification of manufacturer supplied standards (calibrated to provided SCALs)

Experiment 3a= Meter accuracy and precision (calibrated to provided SCALs)

Experiment 3b= Meter accuracy and precision (calibrated to created PCALs)

Experiment 4a= Meters' turbidity responses to natural water samples (calibrated to provided SCALs)

Experiment 4b= Meters' turbidity responses to natural water samples (calibrated to created PCALs)

			Hach 1			
3000 mg/L	2000 mg/L	1000 mg/L	500 mg/L	100 mg/L	50 mg/L	25 mg/L
1360	1070	414	216	51.6	26.9	15.6
1385	1030	407	219	53.1	27	14.7
1375	1065	401	220	52.0	25.7	14.3
1370	1030	411	219	49.5	26.9	14.7
1380	1070	410	220	50.1	26.5	14.5
1374	1053	408.6	219	51.3	26.6	14.8

Table A.1. Experiment 1: Hach 1, LaMotte and GW's turbidity [NTU] responses to Cecil TSS [mg/L].

			LaMotte			
3000 mg/L	2000 mg/L	1000 mg/L	500 mg/L	100 mg/L	50 mg/L	25 mg/L
1060	724	263	175	36.9	18	9.4
1032	696	270	163	40.1	20.1	10.3
1036	756	263	162	40.1	19.8	8.1
1036	820	264	160	39.7	20.7	8.9
1080	692	271	172	40.9	21.2	9.8
1049	738	266	166	39.5	19.9	9.30

			GW			
3000 mg/L	2000 mg/L	1000 mg/L	500 mg/L	100 mg/L	50 mg/L	25 mg/L
970	750	266	158	35.5	18.4	11.0
945	740	261	163	35.5	18.2	9.5
955	740	263	146	34.6	16.9	11.4
940	710	252	156	38	18.6	9.9
915	735	262	154	30.8	15.8	11.0
945	735	261	155	34.9	17.6	10.6

	Hae	ch 1			LaMott	e				GW	
10 NTU	20 NTU	100 NTU	800 NTU	10 NTU	20 NTU	100 NTU	800 NTU	10 NTU	20 NTU	100 NTU	800 NTU
9.79	20.7	103	797	9.83	22.1	110	675	9.43	20.3	107	1048
9.77	20.7	102	801	9.83	22.1	110	677	7.49	20.5	105	1074
9.72	20.7	103	796	9.83	22.1	111	677	7.48	19.3	107	1026
9.71	20.8	103	801	9.84	22	111	677	8.92	20.5	104	1024
9.76	20.8	102	805	9.83	22	111	677	9.22	20.1	107	1030
9.75	20.7	103	800	9.83	22.1	111	677	8.51	20.1	106	1040

Table A.2. Experiment 2: meter's turbidity [NTU] responses to Hach 2100Q's provided SCALs.

Table A.3. Experiment 2: meter's turbidity [NTU] responses to LaMotte 2020e's provided SCALs.

	Ha	ch 1			LaN	Iotte			G	W	
0 NTU	1 NTU	10 NTU	100 NTU	0 NTU	1 NTU	10 NTU	100 NTU	0 NTU	1 NTU	10 NTU	100 NTU
0.58	1.37	10.2	68.1	0.08	1.15	10.0	104	0.23	1.18	10.5	86.9
0.60	1.26	10.4	67.9	0.00	1.14	10.6	99.0	0.22	1.10	10.1	86.5
0.62	1.32	10.3	68.1	0.00	1.15	9.9	99.0	0.22	1.06	10.2	87.1
0.56	1.42	10.2	67.8	0.11	0.95	10.9	98.7	0.20	1.11	10.0	86.9
0.71	1.32	10.1	67.7	0.00	0.95	10.0	98.7	0.19	1.08	10.1	87.3
0.61	1.34	10.2	67.9	0.04	1.07	10.3	99.9	0.21	1.11	10.2	86.9

Table A.4. Experiment 2: meter's turbidity [NTU] responses to Global Water's provided SCALs.

	Hach	1		LaMot	te		GW	r
0.02 NTU	10 NTU	1000 NTU	0.02 NTU	10 NTU	1000 NTU	0.02 NTU	10 NTU	1000 NTU
0.33	10.1	429	0.00	10.7	1744	0.00	9.95	1000
0.34	10.1	430	0.12	10.7	1744	0.00	9.97	1001
0.32	10.1	432	0.03	10.8	1726	0.00	9.91	1001
0.26	10.1	429	0.03	10.8	1746	0.01	9.95	1000
0.30	10.1	431	0.08	10.8	1717	0.00	9.96	996
0.31	10.1	430	0.05	10.8	1735	0.00	9.95	1000

		Hach 1			Hach 2			LaMotte			GW	
PCAL NTU	Round A	Round B	AVG	Round A	Round B	AVG	Round A	Round B	AVG	Round A	Round B	AVG
1	1.71	1.40	1.56	1.62	1.44	1.53	1.90	1.32	1.61	1.00	1.07	1.03
10	10.4	10.9	10.7	10.1	11.0	10.6	12.1	10.9	11.5	9.37	9.96	9.67
50	50.7	50.9	50.8	49.7	51.2	50.5	62.2	54.2	58.2	48.5	47.8	48.2
100	95.8	96.3	96.0	98.0	98.7	98.4	117	103	110	97.0	92.9	95.0
280	264	270	267	269	273	271	291	287	289	273	268	271
500	486	492	489	483	492	487	600	621	611	479	477	478
750	731	752	741	724	744	734	1411	1261	1336	729	730	730
1000	1008	989	999	991	984	988	724	851	787	1004	984	994

Table A.5. Experiment 3a, meters' averaged turbidity readings [NTU].

Table A.6. Experiment 3a, meter accuracy: averaged percent difference about the true value.

		Hach 1			Hach 2			LaMotte			GW	
PCAL NTU	Round A	Round B	AVG	Round A	Round B	AVG	Round A	Round B	AVG	Round A	Round B	AVG
1	71.00	40.20	55.60	62.20	43.80	53.00	89.80	32.40	61.10	5.60	6.80	6.20
10	4.00	9.40	6.70	1.40	10.00	5.70	21.00	8.88	14.94	6.28	0.36	3.32
50	1.32	1.84	1.58	0.60	2.44	1.52	24.44	8.40	16.42	3.00	4.40	3.70
100	4.20	3.72	3.96	2.00	1.26	1.63	17.00	3.22	10.11	2.98	7.02	5.00
280	5.64	3.71	4.68	3.79	2.64	3.21	3.79	2.50	3.14	2.57	4.21	3.39
500	2.76	1.64	2.20	3.48	1.64	2.56	19.96	24.24	22.10	4.28	4.68	4.48
750	2.59	0.48	1.53	3.52	0.77	2.14	88.11	68.13	78.12	2.75	2.69	2.72
1000	1.16	1.08	1.12	0.92	1.56	1.24	27.64	14.94	21.29	0.64	1.58	1.11

Table A.7. Experiment 3a, meter precision: averaged percent difference about measured value.

		Hach 1			Hach 2			LaMotte			GW	
PCAL NTU	Round A	Round B	AVG									
1	11.70	2.28	6.99	6.66	3.34	5.00	9.59	10.27	9.93	12.00	3.00	7.50
10	1.92	1.46	1.69	0.59	0.91	0.75	1.65	1.03	1.34	0.77	0.24	0.50
50	0.51	0.24	0.37	0.40	0.82	0.61	0.68	0.00	0.34	0.62	0.84	0.73
100	0.52	0.85	0.69	0.31	0.67	0.49	0.00	0.95	0.47	0.70	1.16	0.93
280	1.21	0.89	1.05	0.59	0.22	0.41	0.21	0.00	0.10	0.66	0.67	0.67
500	0.99	0.65	0.82	1.12	1.46	1.29	0.37	3.83	2.10	0.54	0.55	0.54
750	1.18	1.20	1.19	0.75	0.64	0.70	8.00	1.67	4.83	1.04	0.71	0.88
1000	1.94	0.49	1.21	0.93	0.37	0.65	1.71	0.42	1.07	1.00	0.79	0.89

	Ha	ch 1			Hach 2			LaMotte			GW	
PCAL NTU	Round A	Round B	AVG									
1	1.33	1.55	1.44	1.41	1.67	1.54	0.79	1.26	1.03	1.02	0.41	0.71
10	10.6	10.7	10.6	10.9	10.1	10.5	10.7	9.49	10.1	10.0	10.0	10.0
50	48.9	51.4	50.2	49.4	49.6	49.5	49.3	45.6	47.4	48.4	51.5	49.9
100	99.4	101	100	99.6	99.7	99.6	99.2	86.4	92.8	97.1	101	99.3
280	267	275	271	270	276	273	274	243	259	272	284	278
500	482	491	486	482	494	488	56	504	533	478	491	485
750	725	741	733	725	742	734	1175	1044	1110	723	735	729
1000	980	989	985	993	994	994	810	706	758	983	961	972

Table A.8. Exp. 3b, meters' averaged turbidity readings [NTU].

Table A.9. Experiment 3b, meter accuracy: averaged percent difference about the true value.

		Hach 1			Hach 2			LaMotte	•		GW	
PCAL NTU	Round A	Round B	AVG									
1	32.8	55.2	44.0	40.6	67.4	54.0	21.0	26.2	23.6	3.40	59.0	31.2
10	5.80	6.60	6.20	8.60	1.40	5.00	6.64	5.62	6.13	0.38	0.62	0.50
50	2.12	2.88	2.50	1.20	0.80	1.00	1.40	8.88	5.14	3.24	3.00	3.12
100	0.58	1.40	0.99	0.40	0.72	0.56	0.84	13.64	7.24	2.90	1.40	2.15
280	4.57	1.79	3.18	3.43	1.57	2.50	2.14	13.14	7.64	2.79	1.43	2.10
500	3.64	1.80	2.72	3.64	1.28	2.46	12.20	0.88	6.54	4.44	1.76	3.10
750	3.39	1.15	2.26	3.33	1.07	2.20	56.67	39.25	47.9	0.04	2.03	1.03
1000	2.00	1.08	1.54	0.68	0.56	0.62	19.00	29.40	24.2	1.68	3.90	2.79

Table A.10. Experiment 3b, meter precision: averaged percent difference about measured value.

		Hach 1			Hach 1			LaMotte			GW	
PCAL NTU	Round A	Round B	AVG									
1	2.41	2.06	2.24	4.55	0.84	2.69	18.9	0.63	9.81	11.0	26.8	18.9
10	1.70	0.56	1.13	0.55	0.59	0.57	2.01	6.68	4.35	0.98	0.88	0.93
50	0.74	0.70	0.72	0.40	0.40	0.40	0.00	0.79	0.40	0.37	0.39	0.38
100	0.48	0.59	0.54	0.10	1.32	0.71	0.91	0.07	0.49	0.62	0.59	0.60
280	0.45	0.73	0.59	0.22	0.22	0.22	0.00	0.90	0.45	0.81	0.35	0.58
500	1.49	0.81	1.15	1.00	0.69	0.84	1.07	0.99	1.03	0.38	0.24	0.31
750	0.63	0.76	0.70	0.69	0.67	0.68	1.45	1.21	1.33	1.61	0.84	1.22
1000	1.22	0.32	0.77	0.28	0.64	0.46	0.00	0.85	0.42	1.14	0.83	0.99

			ction Site) min			Construct t=30				Constructi t=60 n		
	Hach 1	Hach 2	Lamotte	GW	Hach 1	Hach 2	Lamotte	GW	Hach 1	Hach 2	Lamotte	GW
Rep 1	408	406	290	297	2030	2059	1621	1430	1864	1854	1418	1289
Rep 2	430	429	291	307	2022	2025	1643	1400	1850	1853	1366	1296
Rep 3	427	423	298	307	2070	2074	1560	1436	1831	1834	1349	1307
Rep 4	434	431	299	313	2071	2089	1607	1474	1852	1886	1377	1298
Rep 5	5 426 426 292 312			312	2211	2201	1655	1511	1933	1942	1432	1340
AVG	425	423	294	307	2081	2090	1617	1450	1866	1874	1388	1306

Table A.11. Experiment 4a, meters' averaged turbidity readings. Each repetition is an average of five readings.

		Construct t=150			Hunnington Creek				BG's Pond			
	Hach 1	Hach 2	Lamotte	GW	Hach 1	Hach 2	Lamotte	GW	Hach 1	Hach 2	Lamotte	GW
Rep 1	1001	1010	966	846	9.6	9.0	6.7	7.0	884	863	353	417
Rep 2	1002	1006	974	849	9.6	9.4	6.8	6.7	879	859	352	417
Rep 3	1043	1055	1052	866	9.0	8.6	6.5	6.2	898	874	356	433
Rep 4	1079	1086	1053	898	9.2	9.2	7.1	6.4	879	871	352	448
Rep 5	1161	1147	1053	955	8.6	8.9	6.4	6.6	887	875	359	444
AVG	1057	1061	1020	883	9.2	9.0	6.7	6.6	886	868	354	432

Table A.12. Experiment 4a, meter's standard deviation [NTU] for each field sample¹⁵.

	Construction Site t=20 min	Construction Site t=30 min	Construction Site t=60 min	Construction Site t=150 min	Hunnington Creek	BG's Pond
	Avg Rep [NTU]	Avg Rep [NTU]	Avg Rep [NTU]	Avg Rep [NTU]	Avg Rep [NTU]	Avg Rep [NTU]
Hach 1	425	2081	1866	1057	9.2	886
Hach 2	423	2090	1874	1061	9.0	868
LaMotte	294	1617	1388	1020	6.7	354
GW	307	1450	1306	883	6.6	432
Average	362	1809	1608	1005	7.9	635
St. Dev	71.5	326	304	83.5	1.43	281

¹⁵ Averaged Rep NTU values obtained from averages calculated in Table A.11.

			ction Site) min			Construct t=30	ction Site min		Construction Site t=60 min			
	Hach 1	Hach 2	Lamotte	GW	Hach 1	Hach 2	Lamotte	GW	Hach 1	Hach 2	Lamotte	GW
Rep 1	380	382	266	327	2049	2077	1397	1651	1798	1854	1397	1434
Rep 2	387	390	274	344	2047	2072	1317	1604	1820	1859	1415	1485
Rep 3					2055	2086	1649	1620	1838	1889	1367	143
Rep 4					2027	2061	1537	1618	1836	1870	1372	1440
Rep 5					2130	2157	1696	1678	1958	1995	1502	156
AVG	383	386	270	336	2062	2091	1519	1634	1850	1893	1411	147

Table A.13. Experiment 4b, meters' averaged turbidity readings. Each replication, 'Rep', is an average of five readings 16 .

		Construct t=150]	Hunningto	on Creek		BG's Pond			
	Hach 1	Hach 2	Lamotte	GW	Hach 1	Hach 2	Lamotte	GW	Hach 1	Hach 2	Lamotte	GW
Rep 1	1014	1013	984	992	12.1	12.1	10.7	9.3	828	845	390	443
Rep 2	1002	1022	1004	995	11.1	10.8	8.2	7.9	861	870	392	449
Rep 3	1014	1016	963	996	9.9	9.9	7.3	7.5	862	873	401	456
Rep 4	1007	1010	988	974	9.2	9.6	7.0	7.5	871	871	411	460
Rep 5	1003	1026	997	975	9.9	9.7	7.4	7.3	790	799	397	463
AVG	1008	1017	987	986	10.4	10.4	8.1	7.9	842	852	398	454

Table A.14. Experiment 4b, meter's standard deviation [NTU] for each field sample ¹⁷.

	Construction Site t=20 min	Construction Site t=30 min	Construction Site t=60 min	Construction Site t=150 min	Hunnington Creek	BG's Pond
	Avg Rep [NTU]	Avg Rep [NTU]	Avg Rep [NTU]	Avg Rep [NTU]	Avg Rep [NTU]	Avg Rep [NTU]
Hach 1	383	2062	1850	1008	10.4	842
Hach 2	386	2091	1893	1017	10.4	852
LaMotte	270	1519	1411	987	8.1	398
GW	336	1634	1474	986	7.9	454
Average	344	1826	1657	1000	9	637
St. Dev	54.2	292	250	15.4	1.40	244

¹⁶ Not enough sample was collected from 'Construction Site at t=20 min' to carry out five replications. ¹⁷ Averaged Rep NTU values obtained from averages calculated in Table A.13.

APPENDIX B.

Soil properties of the selected South Carolina soils

Appendix B encompasses size distributions for each of the selected soil series based on soil's physiographic South Carolina region. Regions include the upstate Piedmont region, the mid-state Central region and the Coastal Plains region. The first table for each region compares the size distributions obtained from the aggregate size distribution (ASD) analysis and the primary particle size distributions (PSD) from the hydrometer analysis. Following this includes the tabular data from the ASD analysis.

		Me	asured			Hyd	rometer	
Soil Series	Sand ¹⁸	Silt ¹⁹	Clay ²⁰	Silt+Clay	Sand	Silt	Clay	Silt+Clay
Georgeville Top	51.00	36.58	12.42	49.00	59.62	27.72	12.66	40.38
Herndon Top	65.00	28.58	6.42	35.00	56.54	31.58	11.88	43.46
Pacolet Top	91.00	7.48	1.52	9.00	65.98	18.22	15.81	34.02
Wilkes Top	88.50	10.06	1.44	11.50	72.90	14.50	12.59	27.10
Cataula Top	88.00	10.58	1.42	12.00	58.98	26.22	14.81	41.02
Appling Top	87.00	11.98	1.02	13.00	74.76	12.50	12.74	25.24
Cecil Top	91.00	8.13	0.87	9.00	69.40	15.72	14.88	30.60
Madison Top	91.00	8.23	0.77	9.00	68.62	18.43	12.95	31.38
Hiwassee Top	93.50	5.75	0.75	6.50	64.83	20.29	14.88	35.17
Wilkes Sub	76.50	15.19	8.31	23.50	74.98	12.29	12.74	25.02
Hiwassee Sub	80.50	13.76	5.74	19.50	73.26	12.00	14.74	26.74
Madison Sub	73.00	23.04	3.96	27.00	71.26	13.93	14.81	28.74
Cataula Sub	82.00	17.52	0.48	18.00	71.26	14.86	13.88	28.74
Cecil Sub	92.00	7.66	0.34	8.00	78.05	9.22	12.74	21.95
Appling Sub	88.50	11.35	0.15	11.50	75.83	11.36	12.81	24.17
Herndon Sub	67.00	32.91	0.09	33.00	63.19	23.86	12.95	36.81
Pacolet Sub	90.00	9.93	0.07	10.00	75.19	12.07	12.74	24.81
Georgeville Sub	46.00	53.97	0.03	54.00	54.98	27.22	17.81	45.02

Table B.1. Piedmont soils' ASD ('Measured') and primary PSD ('Hydrometer'). Values are percent finer [%]

 ¹⁸ For ASD analysis, Sand refers to aggregate and particle sizes ranging from 2 to 0.05 mm
¹⁹ For ASD analysis, Silt refers to aggregate and particle sizes ranging from 0.05 to 0.002 mm
²⁰ For ASD analysis, Clay refers to aggregate and particle sizes less than 0.002 mm

	Georgeville	Herndon	Pacolet	Wilkes	Cataula	Appling	Cecil	Madison	Hiwassee
Particle Size [mm]				Perc	ent finer [%	6]			
2.000	100	100	100	100	100	100	100	100	100
1.180	96.4	89.2	98.8	92.2	97.6	97.9	94.6	96.9	98.9
0.425	91.8	83.3	68.1	83.8	85.9	84.1	78.3	75.3	66.5
0.150	86.4	75.7	34.2	53.3	37.8	39.8	37.8	40.6	25.1
0.075	74.9	57.0	16.8	24.2	20.8	20.6	16.1	18.6	12.3
0.063	56.4	42.7	10.2	13.7	13.3	14.4	10.6	10.9	7.79
0.062	52.7	38.7	9.49	12.6	12.1	13.8	9.68	9.46	7.02
0.031	43.8	29.1	8.40	10.0	10.6	11.0	8.33	8.04	5.77
0.016	43.2	23.9	7.10	6.59	4.72	4.09	5.17	5.43	4.33
0.008	22.7	11.82	3.82	2.64	4.21	2.69	3.02	2.45	2.12
0.004	16.7	8.85	2.68	2.64	2.50	1.92	1.75	1.49	1.30
0.002	12.4	6.42	1.52	1.44	1.42	1.02	0.87	0.77	0.75

Table B.2. Piedmont top soils' ASD tabular data.

Table B.3. Piedmont subsoils' ASD tabular data.

	Georgeville	Herndon	Pacolet	Wilkes	Cataula	Appling	Cecil	Madison	Hiwassee
Particle Size [mm]				Perc	ent finer [%	6]			
2.000	100	100	100	100	100	100	100	100	100
1.180	99.9	82.1	99.5	97.7	99.3	99.0	92.9	99.2	97.4
0.425	97.5	76.3	89.2	90.2	86.2	87.1	83.3	84.6	90.7
0.150	90.0	61.9	51.6	60.9	43.4	38.8	38.5	57.3	51.1
0.075	74.4	46.4	22.1	32.1	28.7	21.4	14.7	35.6	28.3
0.063	59.3	36.9	14.8	25.4	19.8	14.5	10.0	28.3	21.0
0.062	55.0	34.5	13.8	23.9	18.6	13.2	9.1	27.2	20.0
0.031	50.5	27.3	3.94	22.6	16.2	7.22	5.97	25.4	18.4
0.016	0.43	0.51	0.26	16.9	1.07	0.38	1.31	9.63	20.3
0.008	0.09	0.16	0.08	10.9	0.92	0.24	0.87	4.95	12.7
0.004	0.05	0.12	0.09	10.9	0.75	0.22	0.70	4.71	8.03
0.002	0.03	0.09	0.07	8.31	0.48	0.15	0.34	3.96	5.74

		Me	asured			Hyd	romete	<u>r</u>
Soil Series	Sand ²¹	Silt ²²	Clay ²³	Silt+Clay	Sand	Silt	Clay	Silt+Clay
Ailey Top	88.00	10.38	1.62	12.00	71.05	14.36	14.59	28.95
Johnston Top	89.00	9.84	1.16	11.00	58.05	28.29	13.66	41.95
Lakeland Top	96.00	3.39	0.61	4.00	82.76	4.50	12.74	17.24
Dothan Top	93.00	6.50	0.50	7.00	75.47	10.58	13.95	24.53
Fuquay Top	98.00	1.56	0.44	2.00	85.05	2.22	12.74	14.95
Troup Top	96.00	3.68	0.32	4.00	81.40	6.72	11.88	18.60
Bonneau Top	92.50	7.24	0.26	7.50	62.83	24.22	12.95	37.17
Blanton Top	97.50	2.28	0.22	2.50	83.33	4.79	11.88	16.67
Vaucluse Top	97.00	2.85	0.15	3.00	81.40	5.72	12.88	18.60
Wagram Top	94.00	5.99	0.01	6.00	75.54	10.50	13.95	24.46
Dothan Sub	84.50	12.21	3.29	15.50	79.34	8.78	11.88	20.66
Bonneau Sub	80.50	16.94	2.56	19.50	63.26	23.86	12.88	36.74
Blanton Sub	94.00	4.49	1.51	6.00	85.05	2.22	12.74	14.95
Troup Sub	92.00	6.66	1.34	8.00	80.12	7.14	12.74	19.88
Vaucluse Sub	94.00	4.79	1.21	6.00	81.12	6.22	12.66	18.88
Wagram Sub	94.00	5.22	0.78	6.00	74.12	13.14	12.74	25.88
Fuquay Sub	97.00	2.40	0.60	3.00	86.05	3.22	10.74	13.95
Johnston Sub	89.50	10.27	0.23	10.50	81.12	6.14	12.74	18.88
Ailey Sub	88.00	11.80	0.20	12.00	79.05	8.22	12.74	20.95
Lakeland Sub	97.00	2.86	0.14	3.00	84.19	4.14	11.66	15.81

Table B.4. Central soils' ASD ('Measured') and primary PSD ('Hydrometer'). Values are percent finer [%]

 ²¹ For ASD analysis, Sand refers to aggregate and particle sizes ranging from 2 to 0.05 mm
²² For ASD analysis, Silt refers to aggregate and particle sizes ranging from 0.05 to 0.002 mm
²³ For ASD analysis, Clay refers to aggregate and particle sizes less than 0.002 mm

	Ailey	Blanton	Bonneau	Dothan	Fuquay	Johnston	Lakeland	Troup	Vaucluse	Wagram
Particle Size[mm]					Perce	nt finer [%]				
2.000	100	100	100	100	100	100	100	100	100	100
1.180	99.6	99.8	99.8	99.8	99.9	99.5	99.5	99.5	99.6	99.9
0.425	93.9	97.3	78.4	94.7	98.5	86.7	91.6	97.0	95.4	98.8
0.150	45.7	34.5	35.5	51.7	37.7	38.1	21.6	51.1	55.0	45.8
0.075	19.9	9.32	19.8	13.7	6.20	22.0	8.17	13.4	12.8	17.2
0.063	14.5	3.37	9.01	8.17	2.66	12.8	4.35	5.21	3.72	7.76
0.062	13.2	2.71	8.16	7.39	2.40	11.7	3.94	4.45	3.04	6.55
0.031	10.8	1.96	5.38	6.14	2.22	9.98	3.40	3.54	2.25	4.58
0.016	8.91	1.45	2.11	4.77	2.15	7.82	3.19	2.52	1.43	2.21
0.008	4.38	0.69	1.18	2.58	1.12	3.74	1.79	1.12	0.55	1.05
0.004	2.73	0.44	0.93	1.35	0.76	2.34	1.18	0.57	0.28	0.36
0.002	1.62	0.22	0.26	0.50	0.44	1.16	0.61	0.32	0.15	0.01

Table B.5. Central top soils' ASD tabular data.

Table B.6. Central subsoils' ASD tabular data.

	Ailey	Blanton	Bonneau	Dothan	Fuquay	Johnston	Lakeland	Troup	Vaucluse	Wagram
Particle Size[mm]					Perce	ent finer [%]				
2.000	100	100	100	100	100	100	100	100	100	100
1.180	97.5	99.9	99.8	97.3	99.9	98.9	99.3	99.3	99.1	99.93
0.425	89.7	97.6	98.6	92.6	98.4	89.9	93.5	97.7	95.7	99.20
0.150	26.7	44.5	62.4	40.9	56.3	30.4	31.5	66.1	55.7	71.34
0.075	16.6	12.0	29.3	21.9	7.13	15.4	7.78	18.2	17.5	26.60
0.063	12.6	7.04	21.7	16.6	3.45	11.6	3.78	9.04	7.31	16.25
0.062	11.9	6.41	20.4	15.7	3.23	11.1	3.39	8.18	6.68	15.01
0.031	10.5	5.68	17.68	14.6	2.91	8.71	2.61	7.13	5.80	11.45
0.016	0.75	6.22	12.95	14.8	2.64	1.02	0.97	6.99	5.62	8.71
0.008	0.33	3.19	8.29	8.1	1.49	0.45	0.41	3.86	3.47	4.37
0.004	0.26	2.34	4.93	5.2	1.08	0.36	0.25	2.24	1.83	2.35
0.002	0.20	1.51	2.56	3.29	0.60	0.23	0.14	1.34	1.21	0.78

		Me	asured			Hyd	romete	er
Soil Series	Sand ²⁴	Silt ²⁵	Clay ²⁶	Silt+Clay	Sand	Silt	Clay	Silt+Clay
Goldsboro Top	86.50	11.12	2.38	13.50	73.34	10.07	16.59	26.66
Lynchburg Top	77.50	20.51	1.99	22.50	64.76	21.36	13.88	35.24
Coxville Top	92.00	7.16	0.84	8.00	60.95	20.65	18.40	39.05
Norfolk Top	89.00	10.31	0.69	11.00	75.40	12.72	11.88	24.60
Rains Top	87.50	11.90	0.60	12.50	62.40	24.79	12.81	37.60
Noboco Top	93.00	6.79	0.21	7.00	66.62	20.50	12.88	33.38
Coxville Sub	64.50	27.83	7.67	35.50	71.34	14.00	14.66	28.66
Goldsboro Sub	82.00	13.85	4.15	18.00	65.26	24.86	9.88	34.74
Noboco Sub	76.50	21.53	1.97	23.50	67.19	18.93	13.88	32.81
Rains Sub	86.00	12.70	1.30	14.00	73.12	13.22	13.66	26.88
Lynchburg Sub	69.00	30.01	0.99	31.00	71.26	16.93	11.81	28.74
Norfolk Sub	83.00	16.12	0.88	17.00	79.05	10.22	10.74	20.95

Table B.7. Coastal soils' ASD ('Measured') and primary PSD ('Hydrometer'). Values are percent finer [%]

 ²⁴ For ASD analysis, Sand refers to aggregate and particle sizes ranging from 2 to 0.05 mm
²⁵ For ASD analysis, Silt refers to aggregate and particle sizes ranging from 0.05 to 0.002 mm
²⁶ For ASD analysis, Clay refers to aggregate and particle sizes less than 0.002 mm

	Coxville	Goldsboro	Lynchburg	Norfolk	Rains
Particle Size [mm]		Р	ercent finer [%]		
2.000	100	100	100	100	100
1.180	99.3	99.4	99.8	99.5	99.8
0.425	69.3	94.3	92.7	97.0	91.8
0.150	25.9	45.4	59.3	53.1	46.9
0.075	14.1	24.1	41.5	23.6	25.7
0.063	8.79	14.5	27.9	14.1	14.6
0.062	8.18	13.7	24.9	12.0	13.6
0.031	6.90	12.4	18.6	8.81	10.6
0.016	4.79	9.32	13.4	6.06	6.67
0.008	2.67	4.21	6.39	3.17	2.73
0.004	1.62	4.21	3.88	1.49	1.23
0.002	0.84	2.38	1.99	0.69	0.60

Table B.8. Coastal top soils' ASD tabular data.

Table B.9. Coastal subsoils' ASD tabular data.

	Coxville	Goldsboro	Lynchburg	Norfolk	Rains
Particle Size [mm]		Р	ercent finer [%]		
2.000	100	100	100	100	100
1.180	99.9	97.0	99.9	99.4	99.7
0.425	94.7	72.7	95.1	91.5	97.2
0.150	59.5	39.1	65.6	50.4	46.6
0.075	47.6	24.7	45.3	28.5	26.6
0.063	38.8	19.5	33.4	19.5	16.7
0.062	36.9	18.4	30.5	17.9	15.1
0.031	32.1	16.4	29.0	14.1	12.0
0.016	24.6	9.16	3.48	2.61	9.43
0.008	17.1	5.66	1.76	1.42	4.32
0.004	12.4	5.66	1.61	1.20	2.59
0.002	7.67	4.15	0.99	0.88	1.30

APPENDIX C.

Tabular data for selected soils' turbidity as a function of suspended sediment concentration

Appendix C contains raw data used to establish relationships between turbidity and suspended sediment concentration. First, turbidity versus suspended sediment concentration is plotted for all top soils and subsoils. Next, soils are evaluated based on South Carolina physiographic region. Tables include all concentration and corresponding turbidity values for each region's top soil and subsoils. This tabular data was used for subsequent figures that plot region's top soil turbidity versus suspended sediment concentration and region's subsoil turbidity versus suspended sediment concentration.

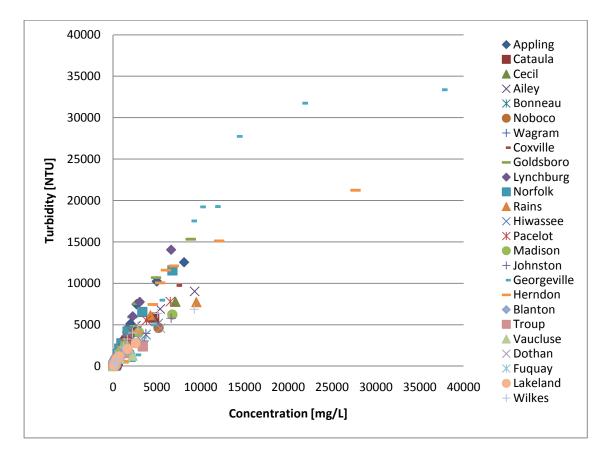


Figure C.1. Relationship of turbidity versus suspended sediment concentration for all top

soils.

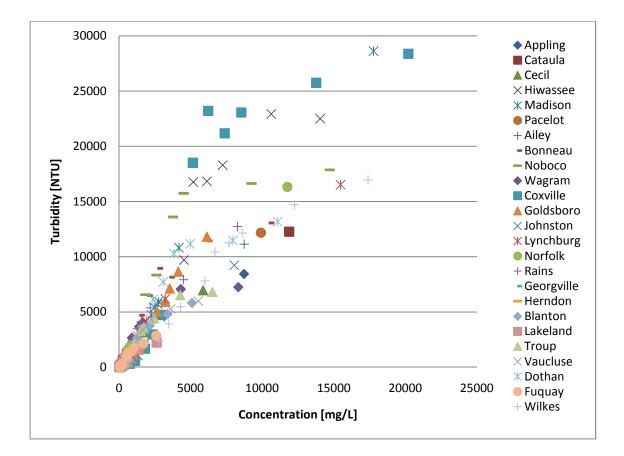


Figure C.2. Relationship of turbidity versus suspended sediment concentration for all

subsoils.

Ар	pling	Cataula		C	Cecil		rgeville	Hiwassee	
Conc. [mg/L]	Turbidity [NTU]								
8090	12548	4670	5864	7100	7780	37580	33370	5130	5158
4980	10244	2630	4214	2680	4178	21650	31740	2590	3718
2670	7480	1500	3094	1280	2542	14180	27740	1170	2278
1980	5132	1470	2937	910	2030	11680	19270	900	1735
1620	4084	1080	2224	600	1461	9990	19236	650	1403
1270	3315	910	1721	430	1028	8990	17532	480	1042
380	657	520	628	160	267	5350	7972	190	332
250	378	400	357	90	139	4350	4996	160	158
150	190	330	192	90	94.9	3110	2261	140	86.5
150	103	280	145	110	74.2	2610	1366	150	111
20	58.6	260	76.8	100	70.9	1980	673	130	82.4

Table C.1. Piedmont top soils ²⁷

Ma	dison	Pa	colet	He	rndon	W	ilkes
Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]
6750	6228	6520	7788	27660	21240	9260	16956
2940	4078	3800	5536	12100	15126	5080	14712
1360	2500	2180	4086	6920	12100	2810	12156
880	1826	1640	3430	6020	11596	2220	11252
710	1624	1400	2832	5360	10064	1610	10436
600	1191	1120	2099	4520	7456	1170	7808
270	419	450	725	2730	4118	490	5466
140	198	250	259	2230	2409	320	3917
110	93.4	170	158	1560	1272	220	2400
130	127	150	152	1290	594	160	1204
100	138	120	109	1170	503	70	417

²⁷ Turbidity values are an average of five readings

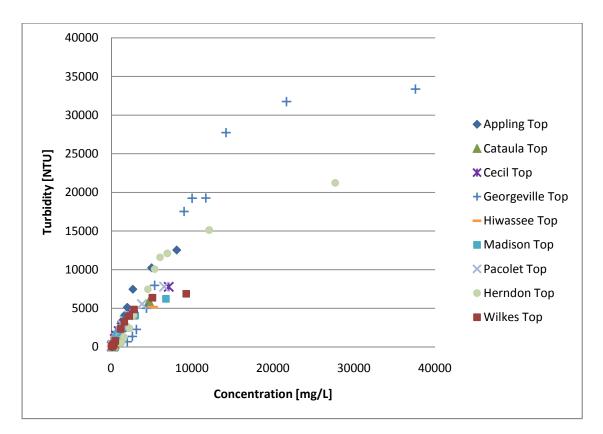


Figure C.3. Piedmont top soils' relationship of turbidity versus suspended sediment

Ар	pling	Cataula		C	ecil	Geor	rgeville	Hiwassee	
Conc. [mg/L]	Turbidity [NTU]								
8720	8440	11870	12264	5860	6960	37400	37340	14040	22500
400	671	640	1301	1850	3156	150	1601	10620	22926
120	223	270	629	770	1744	40	32.8	7230	18294
90	208	270	597	650	1518	50	22.6	6130	16830
90	202	280	609	600	1290	40	15.1	5170	16758
90	153	260	555	350	931	40	11.4	4530	9704
0.001	31.6	80	147	130	354	10	7.27	2860	4570
0.001	19.0	60	112	160	321	0	5.31	2250	2882
20	9.12	60	45.6	30	75.5	0	2.71	1750	1974
0.001	4.41	50	31.8	10	37.1	0	0.76	1300	1051
0.001	1.85	50	20.0	30	47.5	0	2.88	910	334

0.001	1.05	50	20.0	50	4 7.5	0	2.00	
Madison		Pa	colet	Her	rndon	Wilkes		
Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	
17770	28608	9900	12174	28410	20070	17380	16956	
4170	10824	190	274	280	378	12250	14712	
2740	5928	50	98.4	80	77.9	8630	12156	
2680	5710	40	79.5	40	70.5	7680	11252	
2490	5396	40	66.0	60	71.8	6690	10436	
2250	4746	30	53.7	50	56.6	5990	7808	
990	1781	0	10.36	20	17.9	4290	5466	
600	1077	0	3.66	10	8.29	3460	3917	
320	503.4	0	1.82	0	3.91	2770	2400	
160	234.8	0	2.71	10	8.60	1670	1204	
90	145.2	0	1.04	0	3.64	900	417	

Table C.2. Piedmont subsoils²⁸.

²⁸ Turbidity values are an average of five readings

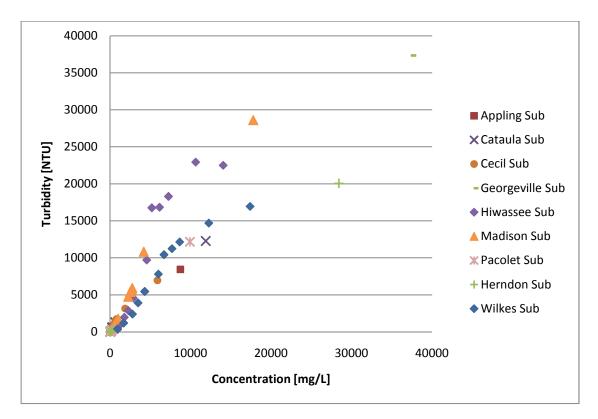


Figure C.4. Piedmont subsoils' relationship of turbidity versus suspended sediment concentration.

A	Ailey		neau	Wa	gram	Bla	nton	Lak	Lakeland	
Conc. [mg/L]	Turbidity [NTU]									
9300	9032	3736	3810	3540	3042	1960	1555	2560	2788	
5360	6916	1864	2330	1150	1454	880	1100	1620	2094	
2590	4878	676	1297	470	774	490	933	680	1183	
1970	3512	415	972	300	585	290	539	420	888	
1620	2978	300	756	170	272	230	476	450	872	
1320	2347	186	563	120	260	200	393	270	547	
500	632	86	132	20	52	50	153	140	232	
340	271	79	175	10	24	50	94	90	153	
310	207	50	82	10	19	60	93	100	113	
250	163	40	67	10	14	60	109	120	155	
240	113	50	41	20	11	50	54	80	71	

Table C.3. Central top soils	²⁹ .
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Tr	Troup		cluse	Dot	than	Fu	quay	Johnston	
Conc. [mg/L]	Turbidity [NTU]								
3380	2400	2220	1277	5400	4580	1840	1970	6620	5794
1470	1574	810	903	2970	3642	1320	1714	3730	3952
590	964	300	477	1370	2476	740	1306	960	1519
460	866	200	358	940	1946	660	1223	1020	1608
350	648	100	218	640	1392	420	862	700	1033
300	561	100	203	420	986	290	670	370	539
70	158	10	51	140	367	130	256	360	328
20	66	10	57	80	163	90	151	140	153
20	62	20	58	70	109	80	131	80	96
20	61	10	38	80	130	20	56	70	82
20	47	10	26	80	90	60	93	90	106

²⁹ Turbidity values are an average of five readings

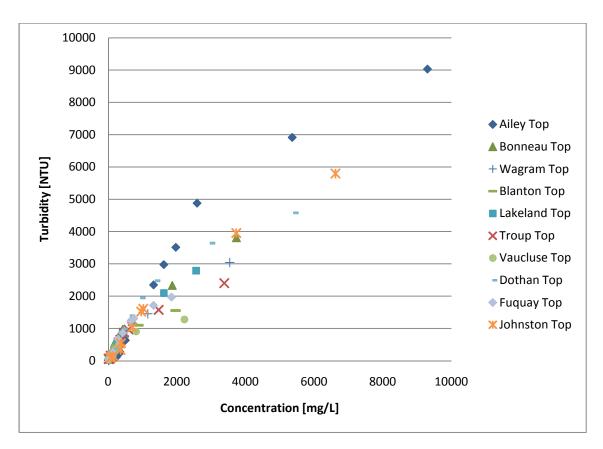


Figure C.5. Central top soils' relationship of turbidity versus suspended sediment

A	Ailey		Bonneau		gram	Bla	anton	Lakeland	
Conc. [mg/L]	Turbidity [NTU]								
8730	11136	10480	13060	8320	7248	5090	5822	2620	2230
1140	1830	6160	11436	4300	7058	3370	4800	1350	1576
420	911	3530	8148	1620	4028	2120	3770	560	811
420	812	2700	8946	1390	3662	1720	3609	390	794
370	704	1990	6478	880	2669	1670	3551	320	575
320	521	1430	4706	510	1585	1270	2657	150	263
20	36	390	1038	100	266	630	1051	60	132
20	11	260	444	40	73	440	601	60	36
20	22	180	192	20	48	300	265	0	14
10	10	50	96	30	41	230	206	0	8
10	5	80	70	20	35	170	125	0	19

Т	Troup		Vaucluse		Dothan		Fuquay		Johnston	
Conc. [mg/L]	Turbidity [NTU]									
6510	6818	5520	5996	11070	13172	2580	2864	8040	9230	
4270	6526	3590	5274	7940	11488	1690	2142	830	1078	
2440	4420	2200	4352	4960	11172	1020	1738	260	349	
1720	3544	1650	3382	3810	10284	850	1456	230	316	
1510	3303	1250	2856	3080	7720	660	1311	220	299	
1060	2310	940	2069	2360	6116	550	1004	150	252	
280	478	290	573	1100	2009	210	320	50	75	
170	216	180	251	780	810	100	161	30	16	
100	135	110	126	600	383	50	70	30	15	
90	102	100	105	470	257	50	42	10	6	
80	70	80	72	360	165	40	43	40	3	

Table C.4. Central subsoils³⁰.

³⁰ Turbidity values are an average of five readings

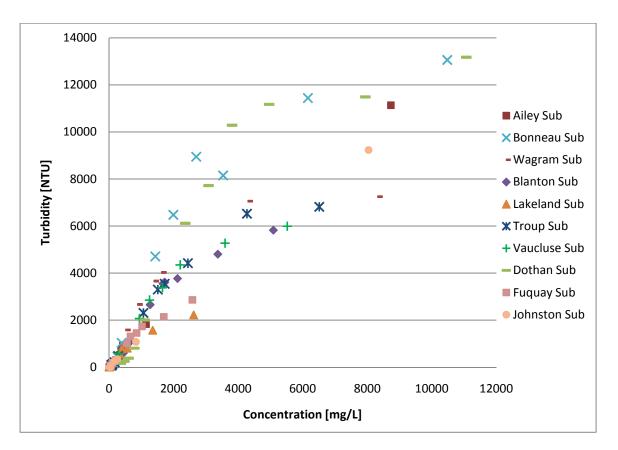


Figure C.6. Central subsoils' relationship of turbidity versus suspended sediment concentration.

Co	xville	Gold	lsboro	Lynchburg		
Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	
7290	9744	8850	15336	14890	11316	
3670	5476	4900	10686	6610	14058	
1760	3586	2780	7568	3030	7754	
1340	2933	2190	5932	2230	6022	
1020	2357	1850	4800	1650	4498	
780	1802	1080	2970	1600	4126	
360	621	510	1012	640	1248	
220	292	360	520	460	529	
140	147	170	189	320	260	
130	99	140	139	200	175	
120	63	150	107	190	129	

No	rfolk	R	ains	Noboco		
Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	
6760	11568	9490	7740	5180	4644	
3330	6548	4250	6156	1310	1712	
1640	4192	1260	2850	530	931	
940	2752	950	2464	340	747	
700	2120	490	1226	180	421	
500	1479	420	941	190	390	
120	331	110	192	130	195	
100	151	110	113	90	47	
50	65	90	63	50	48	
50	50	60	68	60	39	
30	72	40	71	20	24	

³¹ Turbidity values are an average of five readings

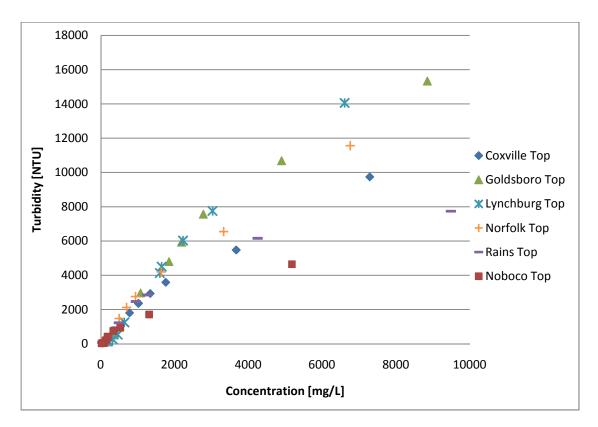


Figure C.7. Coastal top soils' relationship of turbidity versus suspended sediment

Со	xville	Gold	dsboro	Lynchburg		
Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	Conc. [mg/L]	Turbidity [NTU]	
20200	28370	10410	10968	15460	16524	
13760	25740	6130	11816	3220	6192	
8530	23060	4130	8650	1890	4160	
7370	21170	3530	7134	1540	3635	
6220	23200	3220	5920	1330	1702	
5150	18500	2670	5028	1160	2548	
2930	4724	1510	2010	430	600	
2340	2967	1020	1127	230	245	
1800	1681	650	494	60	51	
1110	577	300	228	50	46	
730	309	190	122	10	26	

No	orfolk	R	ains	Noboco		
Conc. [mg/L]			Conc. Turbidity [mg/L] [NTU]		Turbidity [NTU]	
11750	16324	8260	12732	14710	17862	
1550	3416	4500	7920	9240	16632	
760	1893	2170	5370	4500	15744	
670	1637	1560	4134	3760	13596	
610	1596	1260	3480	2600	8344	
490	1247	900	2370	1830	6564	
140	291	290	622	580	1595	
110	177	200	284	360	647	
160	53	100	133	260	361	
10	48	50	83	170	166	
10	38	70	62	150	101	

³² Turbidity values are an average of five readings

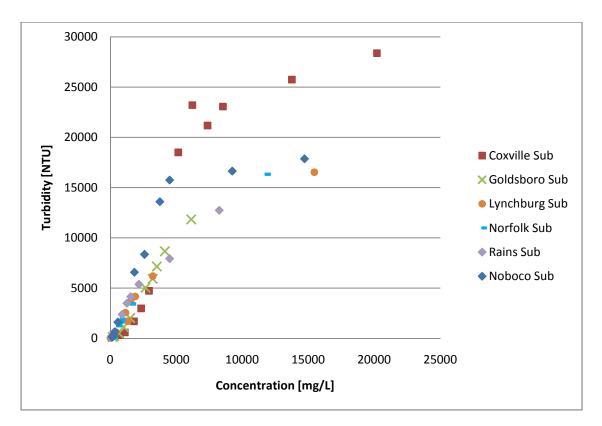


Figure C.8. Coast subsoils' relationship of turbidity versus suspended sediment

APPENDIX D

Tabular data for selected soils' turbidity as a function of sediment settling time

Appendix D contains raw data used to establish relationships between turbidity and sediment settling time. First, turbidity versus settling time is plotted for all top soils and subsoils. Next, soils are evaluated based on South Carolina physiographic region. Tables include all sample times and corresponding turbidity values for each region's top soil and subsoils. This tabular data was used for subsequent figures that plot region's top soil turbidity versus settling time and region's subsoil turbidity versus settling time.

Ар	Appling		Cataula		Cecil		Georgeville		Hiwassee	
Time [min]	Turbidity [NTU]									
0	12548	0	5864	0	7780	0	33370	0	5158	
5	10244	5	4214	5	4178	5	31740	5	3718	
30	7480	30	3094	30	2542	30	27740	30	2278	
60	5132	60	2937	60	2030	60	19270	60	1735	
120	4084	120	2224	120	1461	120	19236	120	1403	
240	3315	240	1721	240	1028	240	17532	240	1042	
1440	657	1440	628	1440	267	1440	7972	1440	332	
2880	378	2880	357	2880	139	2880	4996	4320	158	
5760	190	5760	192	5760	95	5760	2261	7200	87	
11520	103	11520	145	10080	74	10080	1366	10080	111	
23040	59	30240	77	20160	71	20160	673	20160	82	

Ma	Madison		Pacolet		Herndon		lkes
Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]
0	6228	0	7788	0	21240	0	6876
5	4078	5	5536	5	15126	5	6374
30	2500	30	4086	30	12100	30	4846
60	1826	60	3430	60	11596	60	4004
120	1624	120	2832	120	10064	120	3276
240	1191	240	2099	240	7456	240	2326
1440	419	1440	725	1440	4118	1440	784
4320	198	4320	259	2880	2409	2880	361
7200	93	7200	158	5760	1272	5760	235
10080	127	10080	152	10080	594	10080	136
20160	138	20160	109	20160	503	20160	76

Table D.1. Piedmont top soils.

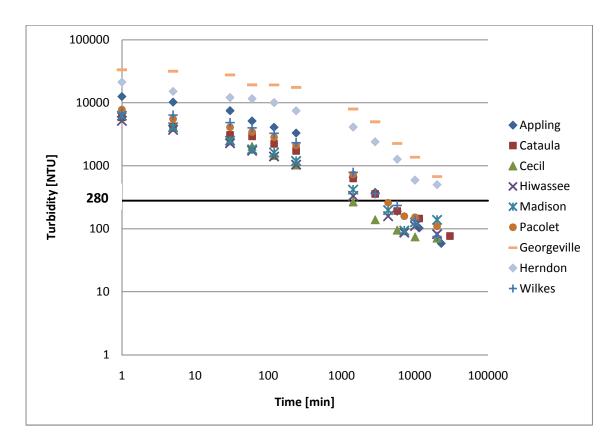


Figure D.1. Piedmont top soils' relationship of turbidity and sediment settling time.

Appling		Cataula		Cecil		Geo	rgeville	Hiwassee	
Time [min]	Turbidity [NTU]								
1	8440	1	12264	1	6960	1	37340	1	22500
5	671	5	1301	5	3156	5	161	5	22926
30	223	30	629	30	1744	30	33	30	18294
60	208	60	597	60	1518	60	23	60	16830
120	202	120	609	120	1290	120	15	120	16758
240	153	240	555	240	931	240	11	240	9704
1440	32	1440	147	1440	354	1440	7	1440	4570
2880	19	2880	112	2880	321	2880	5	2880	2882
5760	9	5760	46	5760	75	5760	3	5760	1974
10080	4	10080	32	10080	37	10080	1	10080	1051
20160	2	30240	20	20160	48	20160	3	20160	334

Ma	adison	Pa	acolet	Не	rndon	W	likes
Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]
1	28608	1	12174	1	20070	1	16956
5	10824	5	274	5	378	5	14712
30	5928	30	98	30	78	30	12156
60	5710	60	80	60	70	60	11252
120	5396	120	66	120	72	120	10436
240	4746	240	54	240	57	240	7808
1440	1781	1440	10	1440	18	1440	5466
2880	1077	2880	4	2880	8	2880	3917
5760	503	5760	2	5760	4	5760	2400
10080	235	10080	3	10080	9	10080	1204
20160	145	20160	1	20160	4	20160	417

Table D.2.	Piedmont	subsoils.
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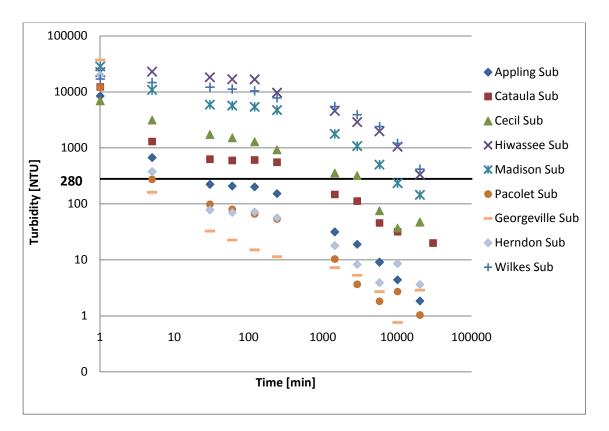


Figure D.2. Piedmont subsoils' relationship of turbidity and sediment settling time.

A	Ailey	Bo	nneau	Wa	agram	Bl	anton	La	keland
Time [min]	Turbidity [NTU]								
1	9032	1	3810	1	3042	1	1555	1	2788
5	6916	5	2330	5	1454	5	1100	5	2094
30	4878	30	1297	30	774	30	933	30	1183
60	3512	60	972	60	585	60	539	60	888
120	2978	120	756	120	272	120	476	120	872
240	2347	240	563	240	260	240	393	240	547
1440	632	1440	132	1440	52	1440	153	1440	232
2880	271	2880	175	2880	24	2880	94	2880	153
5760	207	7200	82	5760	19	5760	93	7200	113
11520	163	10080	67	11520	14	10080	109	11520	155
21600	113	20160	41	23040	11	20160	54	21600	71

Т	roup	Vaucluse		D	othan	Fu	iquay	Johnston	
Time [min]	Turbidity [NTU]								
1	2400	1	1277	1	4580	1	1970	1	5794
5	1574	5	903	5	3642	5	1714	5	3952
30	964	30	477	30	2476	30	1306	30	1519
60	866	60	358	60	1946	60	1223	60	1608
120	648	120	218	120	1392	120	862	120	1033
240	561	240	203	240	986	240	670	240	539
1440	158	1440	51	1440	367	1440	256	1440	328
2880	66	2880	57	2880	163	2880	151	2880	153
5760	62	5760	58	5760	109	5760	131	7200	96
10080	61	10080	38	10080	130	10080	56	11520	82
20160	47	20160	26	20160	90	20160	93	20160	106

Table D.3. Central top soils.

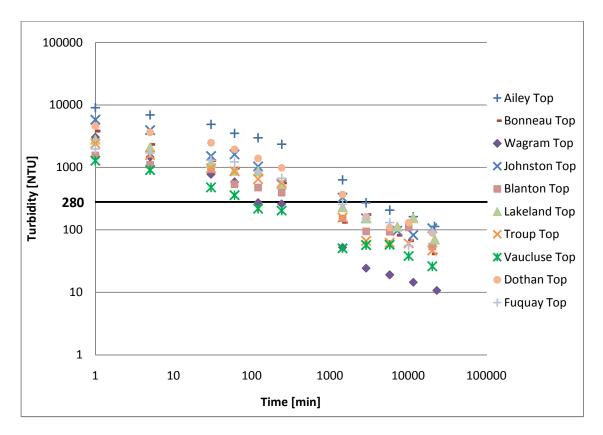


Figure D.3. Central top soils' relationship of turbidity and sediment settling time.

A	Ailey	Bonneau		Wa	agram	Bl	anton	Lakeland	
Time [min]	Turbidity [NTU]								
1	11136	1	13060	1	7248	1	5822	1	2230
5	1830	5	11436	5	7058	5	4800	5	1576
30	911	30	8148	30	4028	30	3770	30	811
60	812	60	8946	60	3662	60	3609	60	794
120	704	120	6478	120	2669	120	3551	120	575
240	521	240	4706	240	1585	240	2657	240	263
1440	36	1440	1038	1440	266	1440	1051	1440	132
2880	11	2880	444	2880	73	2880	601	2880	36
5760	22	5760	192	5760	48	5760	265	5760	14
11520	10	11520	96	11520	41	10080	206	11520	8
21600	5	30240	70	23040	35	20160	125	20160	19

Т	roup	Va	ucluse	D	othan	Fu	iquay	Johnston	
Time [min]	Turbidity [NTU]								
1	6818	1	5996	1	13172	1	2864	1	9230
5	6526	5	5274	5	11488	5	2142	5	1078
30	4420	30	4352	30	11172	30	1738	30	349
60	3544	60	3382	60	10284	60	1456	60	316
120	3303	120	2856	120	7720	120	1311	120	299
240	2310	240	2069	240	6116	240	1004	240	252
1440	478	1440	573	1440	2009	1440	320	1440	75
2880	216	2880	251	2880	810	2880	161	2880	16
5760	135	5760	126	5760	383	5760	70	7200	15
10080	102	10080	105	10080	257	10080	42	11520	6
20160	70	20160	72	20160	165	20160	43	20160	3

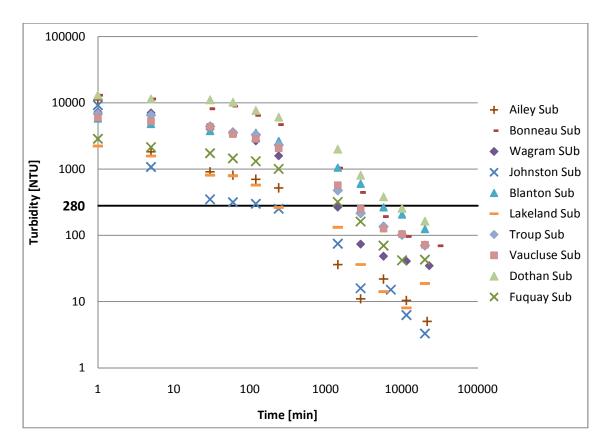


Figure D.4. Central subsoils' relationship of turbidity and sediment settling time.

Co	oxville	Gol	dsboro	Lyn	chburg
Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]
1	9744	1	15336	1	11316
5	5476	5	10686	5	14058
30	3586	30	7568	30	7754
60	2933	60	5932	60	6022
120	2357	120	4800	120	4498
240	1802	240	2970	240	4126
1440	621	1440	1012	1440	1248
2880	292	2880	520	2880	529
5760	147	5760	189	7200	260
11520	99	10080	139	10080	175
30240	63	20160	107	20160	129

Table D.5. Coastal top soils.

No	orfolk	R	lains	No	oboco
Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]
1	11568	1	7740	1	4644
5	6548	5	6156	5	1712
30	4192	30	2850	30	931
60	2752	60	2464	60	747
120	2120	120	1226	120	421
240	1479	240	941	240	390
1440	331	1440	192	1440	195
2880	151	2880	113	2880	47
5760	65	7200	63	7200	48
10080	50	10080	68	10080	39
20160	72	20160	71	20160	24

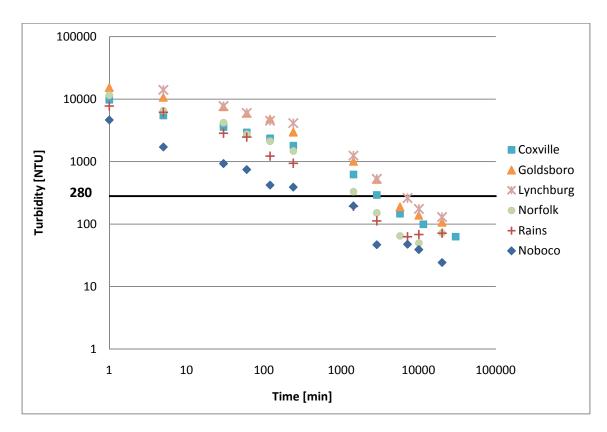


Figure D.5. Coastal top soils' relationship of turbidity and sediment settling time.

Co	xville	Gol	dsboro	Lyn	chburg
Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]
1	28370	1	10968	1	16524
5	25740	5	11816	5	6192
30	23060	30	8650	30	4160
60	21170	60	7134	60	3635
120	23200	120	5920	120	1702
240	18500	240	5028	240	2548
1440	4724	1440	2010	1440	600
2880	2967	2880	1127	2880	245
5760	1681	5760	494	7200	51
11520	577	11520	228	10080	46
30240	309	30240	122	20160	26

Table D.6. Coastal subsoils.

No	orfolk	R	Rains	N	oboco
Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]	Time [min]	Turbidity [NTU]
1	16324	1	12732	1	17862
5	3416	5	7920	5	16632
30	1893	30	5370	30	15744
60	1637	60	4134	60	13596
120	1596	120	3480	120	8344
240	1247	240	2370	240	6564
1440	291	1440	622	1440	1595
2880	177	2880	284	2880	647
7200	53	7200	133	5760	361
10080	48	10080	83	10080	166
20160	38	20160	62	20160	101

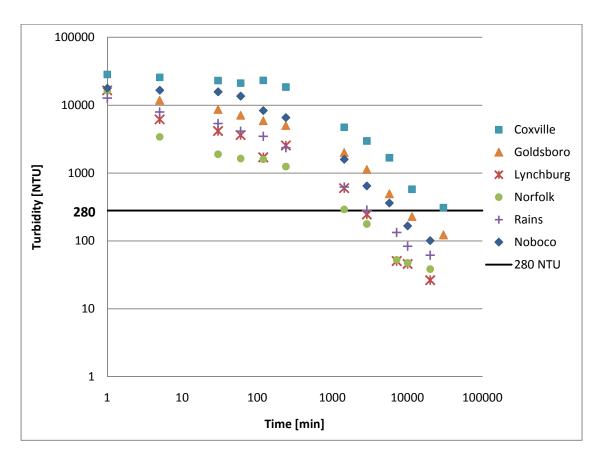


Figure D.6. Coastal subsoils' relationship of turbidity and sediment settling time.

APPENDIX E

Tabular data for selected soils' turbidity as a function of particle diameter

Appendix E contains raw data used to establish relationships between turbidity and aggregate/particle size. From applying Stoke's Law, particle diameters evaluated included 0.063, 0.014, 0.006, 0.004, 0.003, and 0.002 mm based on the previous sample times of 0 min, 5 min, 30 min, 1 hr, 2 hr and 4 hr. Particles sizes were compared to their corresponding turbidities and concentrations that were determined from previous procedures. Actual turbidity and concentration values were calculated using a subtraction method. Next, in order to compare all soils, the ratio of actual turbidity to concentration was calculated. All tabular data are reported in Appendix E tables for each region's top soil and subsoil. Lastly, subsequent figures plot region's top soil versus the ratio of turbidity to concentration and region's subsoil turbidity versus the ratio of turbidity to concentration.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Georgeville	0.063	37580	33370	15930	1630	0.10
	0.014	21650	31740	7470	4000	0.54
	0.006	14180	27740	2500	8470	3.39
	0.004	11680	19270	1690	34	0.02
	0.003	9990	19236	1000	1704	1.70
	0.002	8990	17532	8990	17532	1.95
Herndon	0.063	27660	21240	15560	6114	0.39
	0.014	12100	15126	5180	3026	0.58
	0.006	6920	12100	900	504	0.56
	0.004	6020	11596	660	1532	2.32
	0.003	5360	10064	840	2608	3.10
	0.002	4520	7456	4520	7456	1.65
Appling	0.063	8090	12548	3110	2304	0.74
	0.014	4980	10244	2310	2764	1.20
	0.006	2670	7480	690	2348	3.40
	0.004	1980	5132	360	1048	2.91
	0.003	1620	4084	350	769	2.20
	0.002	1270	3315	1270	3315	2.61
Cataula	0.063	4670	5864	2040	1650	0.81
	0.014	2630	4214	1130	1120	0.99
	0.006	1500	3094	30	157	5.23
	0.004	1470	2937	390	713	1.83
	0.003	1080	2224	170	503	2.96
	0.002	910	1721	910	1721	1.89
Cecil	0.063	7100	7780	4420	3602	0.81
	0.014	2680	4178	1400	1636	1.17
	0.006	1280	2542	370	512	1.38
	0.004	910	2030	310	569	1.84
	0.003	600	1461	170	433	2.55
	0.002	430	1028	430	1028	2.39
Hiwassee	0.063	5130	5158	2540	1440	0.57
	0.014	2590	3718	1420	1440	1.01
	0.006	1170	2278	270	543	2.01
	0.004	900	1735	250	332	1.33
	0.003	650	1403	170	361	2.12
	0.002	480	1042	480	1042	2.17

Table E.1. Piedmont top soils.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Madison	0.063	6750	6228	3810	2150	0.56
	0.014	2940	4078	1580	1578	1.00
	0.006	1360	2500	480	674	1.40
	0.004	880	1826	170	202	1.19
	0.003	710	1624	110	433	3.94
	0.002	600	1191	600	1191	1.99
Pacelot	0.063	6520	7788	2720	2252	0.83
	0.014	3800	5536	1620	1450	0.90
	0.006	2180	4086	540	656	1.21
	0.004	1640	3430	240	598	2.49
	0.003	1400	2832	280	733	2.62
	0.002	1120	2099	1120	2099	1.87
Wilkes	0.063	9260	6876	4180	502	0.12
	0.014	5080	6374	2270	1528	0.67
	0.006	2810	4846	590	842	1.43
	0.004	2220	4004	610	728	1.19
	0.003	1610	3276	440	950	2.16
	0.002	1170	2326	1170	2326	1.99

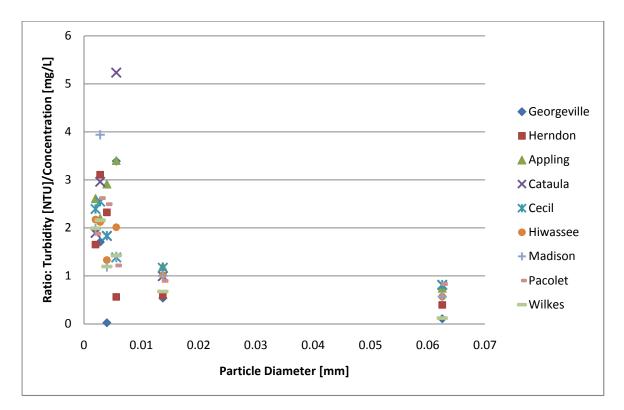


Figure E.1. Piedmont top soils' relationship of particle diameter and the ratio of turbidity

to concentration.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Hiwassee	0.063	14040	22500	3420	0	0.00
	0.014	10620	22926	3390	4632	1.37
	0.006	7230	18294	1100	1464	1.33
	0.004	6130	16830	960	72	0.07
	0.003	5170	16758	640	7054	11.02
	0.002	4530	9704	4530	9704	2.14
Madison	0.063	17770	28608	13600	17784	1.31
	0.014	4170	10824	1430	4896	3.42
	0.006	2740	5928	60	218	3.63
	0.004	2680	5710	190	314	1.65
	0.003	2490	5396	240	650	2.71
	0.002	2250	4746	2250	4746	2.11
Wilkes	0.063	17380	16956	5130	2244	0.44
	0.014	12250	14712	3620	2556	0.71
	0.006	8630	12156	950	904	0.95
	0.004	7680	11252	990	816	0.82
	0.003	6690	10436	700	2628	3.75
	0.002	5990	7808	5990	7808	1.30
Cataula	0.063	11870	12264	11230	10963	0.98
	0.014	640	1301	360	672	1.87
	0.006	270	629	0	20	0.00
	0.004	270	597	0	0	0.00
	0.003	280	609	20	54	2.68
	0.002	260	555	260	555	2.14
Cecil	0.063	5860	6960	4010	3804	0.95
	0.014	1850	3156	1080	1412	1.31
	0.006	770	1744	120	226	1.88
	0.004	650	1518	50	228	4.56
	0.003	600	1290	250	359	1.44
	0.002	350	931	350	931	2.66
Georgeville	0.063	37400	37340	37250	37179	1.00
	0.014	150	161	100	128	1.28
	0.006	40	33	0	10	0.00
	0.004	50	23	10	8	0.75
	0.003	40	15	0	4	0.00
	0.002	40	11	40	11	0.28

Table E.2. Piedmont subsoils.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Herndon	0.063	28410	20070	9710	11900	1.23
	0.014	280	378	140	176	1.26
	0.006	80	78	10	19	1.89
	0.004	40	70	0	14	0.00
	0.003	60	72	10	15	1.52
	0.002	50	57	50	57	1.13
Appling	0.063	8720	8440	8320	7769	0.93
	0.014	400	671	280	447	1.60
	0.006	120	223	30	15	0.51
	0.004	90	208	0	6	0.00
	0.003	90	202	0	49	0.00
	0.002	90	153	90	153	1.70
Pacelot	0.063	9900	12174	9710	11900	1.23
	0.014	190	274	140	176	1.26
	0.006	50	98	10	19	1.89
	0.004	40	80	0	14	0.00
	0.003	40	66	10	12	1.23
	0.002	30	54	30	54	1.79

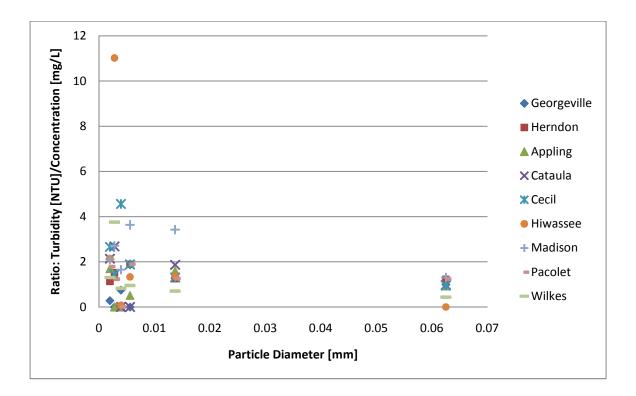


Figure E.2. Piedmont subsoils' relationship of particle diameter and the ratio of turbidity

to concentration.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Ailey	0.063	9300	9032	3940	2116	0.54
	0.014	5360	6916	2770	2038	0.74
	0.006	2590	4878	620	1366	2.20
	0.004	1970	3512	350	534	1.53
	0.003	1620	2978	300	631	2.10
	0.002	1320	2347	1320	2347	1.78
Johnston	0.063	6620	5794	2890	1842	0.64
	0.014	3730	3952	2710	2344	0.86
	0.006	960	1519	0	0	0.00
	0.004	1020	1608	320	575	1.80
	0.003	700	1033	330	494	1.50
	0.002	370	539	370	539	1.46
Bonneau	0.063	3736	3810	1872	1480	0.79
	0.014	1864	2330	1188	1033	0.87
	0.006	676	1297	261	325	1.24
	0.004	415	972	115	217	1.88
	0.003	300	756	114	193	1.70
	0.002	186	563	186	563	3.02
Wagram	0.063	3540	3042	2390	1588	0.66
	0.014	1150	1454	680	680	1.00
	0.006	470	774	170	189	1.11
	0.004	300	585	130	313	2.40
	0.003	170	272	50	13	0.25
	0.002	120	260	120	260	2.16
Blanton	0.063	1960	1555	1080	455	0.42
	0.014	880	1100	390	168	0.43
	0.006	490	933	200	394	1.97
	0.004	290	539	60	63	1.05
	0.003	230	476	30	83	2.77
	0.002	200	393	200	393	1.96
Lakeland	0.063	2560	2788	940	694	0.74
	0.014	1620	2094	940	911	0.97
	0.006	680	1183	230	295	1.28
	0.004	420	888	0	16	0.00
	0.003	450	872	180	325	1.80
	0.002	270	547	270	547	2.03

Table E.3. Central top soils.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Troup	0.063	3380	2400	1910	826	0.43
	0.014	1470	1574	880	610	0.69
	0.006	590	964	130	98	0.76
	0.004	460	866	110	218	1.98
	0.003	350	648	50	87	1.73
	0.002	300	561	300	561	1.87
Vaucluse	0.063	2220	1277	1410	374	0.27
	0.014	810	903	510	426	0.83
	0.006	300	477	100	120	1.20
	0.004	200	358	100	140	1.40
	0.003	100	218	0	15	0.00
	0.002	100	203	100	203	2.03
Dothan	0.063	5400	4580	2430	938	0.39
	0.014	2970	3642	1600	1166	0.73
	0.006	1370	2476	430	530	1.23
	0.004	940	1946	300	554	1.85
	0.003	640	1392	220	406	1.85
	0.002	420	986	420	986	2.35
Fuquay	0.063	1840	1970	520	256	0.49
	0.014	1320	1714	580	408	0.70
	0.006	740	1306	80	83	1.04
	0.004	660	1223	240	361	1.50
	0.003	420	862	130	193	1.48
	0.002	290	670	290	670	2.31

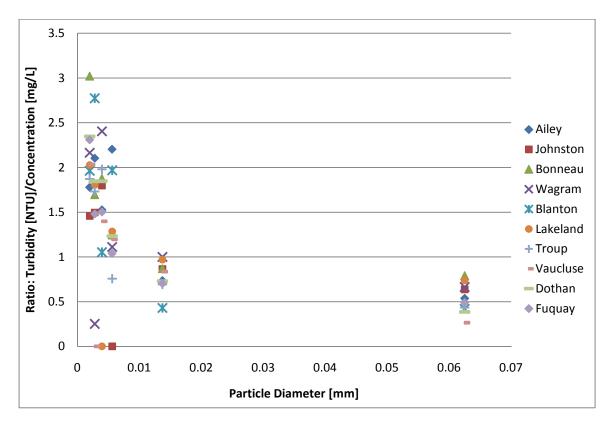


Figure E.3. Cental top soils' relationship of particle diameter and the ratio of turbidity to

concentration.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Bonneau	0.063	8320	7248	4020	190	0.05
	0.014	4300	7058	2680	3030	1.13
	0.006	1620	4028	230	366	1.59
	0.004	1390	3662	510	993	1.95
	0.003	880	2669	370	1084	2.93
	0.002	510	1585	510	1585	3.11
Dothan	0.063	2580	2864	890	722	0.81
	0.014	1690	2142	670	404	0.60
	0.006	1020	1738	170	282	1.66
	0.004	850	1456	190	145	0.76
	0.003	660	1311	110	307	2.79
	0.002	550	1004	550	1004	1.83
Wagram	0.063	5090	5822	1720	1022	0.59
C	0.014	3370	4800	1250	1030	0.82
	0.006	2120	3770	400	161	0.40
	0.004	1720	3609	50	58	1.16
	0.003	1670	3551	400	894	2.24
	0.002	1270	2657	1270	2657	2.09
Blanton	0.063	2620	2230	1270	654	0.51
	0.014	1350	1576	790	765	0.97
	0.006	560	811	170	17	0.10
	0.004	390	794	70	219	3.13
	0.003	320	575	170	312	1.83
	0.002	150	263	150	263	1.76
Troup	0.063	5520	5996	1930	722	0.37
_	0.014	3590	5274	1390	922	0.66
	0.006	2200	4352	550	970	1.76
	0.004	1650	3382	400	526	1.32
	0.003	1250	2856	310	787	2.54
	0.002	940	2069	940	2069	2.20
Vaucluse	0.063	11070	13172	3130	1684	0.54
	0.014	7940	11488	2980	316	0.11
	0.006	4960	11172	1150	888	0.77
	0.004	3810	10284	730	2564	3.51
	0.003	3080	7720	720	1604	2.23
	0.002	2360	6116	2360	6116	2.59

Table E.4. Central subsoils.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Ailey	0.063	8730	11136	7590	9306	1.23
	0.014	1140	1830	720	919	1.28
	0.006	420	911	0	99	0.00
	0.004	420	812	50	109	2.17
	0.003	370	704	50	183	3.65
	0.002	320	521	320	521	1.63
Johnston	0.063	10480	13060	4320	1624	0.38
	0.014	6160	11436	2630	2490	0.95
	0.006	3530	8148	830	0	0.00
	0.004	2700	8946	710	2468	3.48
	0.003	1990	6478	560	1772	3.16
	0.002	1430	4706	1430	4706	3.29
Lakeland	0.063	6510	6818	2240	292	0.13
	0.014	4270	6526	1830	2106	1.15
	0.006	2440	4420	720	876	1.22
	0.004	1720	3544	210	241	1.15
	0.003	1510	3303	450	993	2.21
	0.002	1060	2310	1060	2310	2.18
Fuquay	0.063	8040	9230	7210	8152	1.13
	0.014	830	1078	570	729	1.28
	0.006	260	349	30	33	1.11
	0.004	230	316	10	17	1.70
	0.003	220	299	70	47	0.68
	0.002	150	252	150	252	1.68

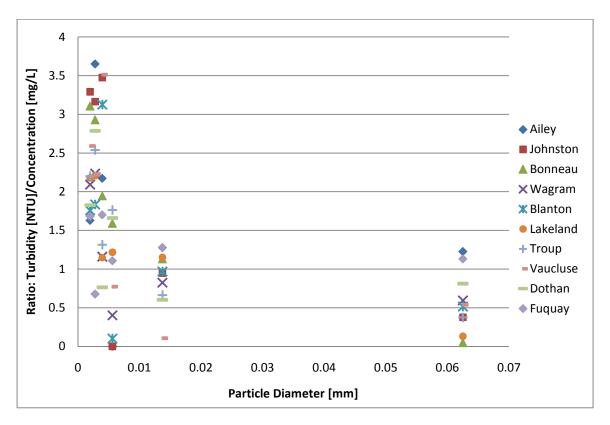


Figure E.4. Central subsoils' relationship of particle diameter and the ratio of turbidity to

concentration.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Goldsboro	0.063	8850	15336	3950	4650	1.18
	0.014	4900	10686	2120	3118	1.47
	0.006	2780	7568	590	1636	2.77
	0.004	2190	5932	340	1132	3.33
	0.003	1850	4800	770	1830	2.38
	0.002	1080	2970	1080	2970	2.75
Lynchburg	0.063	14890	11316	8280	0	0.00
	0.014	6610	14058	3580	6304	1.76
	0.006	3030	7754	800	1732	2.17
	0.004	2230	6022	580	1524	2.63
	0.003	1650	4498	50	372	7.44
	0.002	1600	4126	1600	4126	2.58
Coxville	0.063	7290	9744	3620	4268	1.18
	0.014	3670	5476	1910	1890	0.99
	0.006	1760	3586	420	653	1.55
	0.004	1340	2933	320	576	1.80
	0.003	1020	2357	240	555	2.31
	0.002	780	1802	780	1802	2.31
Norfolk	0.063	6760	11568	3430	5020	1.46
	0.014	3330	6548	1690	2356	1.39
	0.006	1640	4192	700	1440	2.06
	0.004	940	2752	240	632	2.63
	0.003	700	2120	200	641	3.21
	0.002	500	1479	500	1479	2.96
Noboco	0.063	5180	4644	3870	2932	0.76
	0.014	1310	1712	780	781	1.00
	0.006	530	931	190	184	0.97
	0.004	340	747	150	326	2.17
	0.003	180	421	0	31	0.00
	0.002	190	390	190	390	2.05
Rains	0.063	9490	7740	5240	1584	0.30
	0.014	4250	6156	2990	3306	1.11
	0.006	1260	2850	310	386	1.25
	0.004	950	2464	460	1238	2.69
	0.003	490	1226	70	285	4.07
	0.002	420	941	420	941	2.24

Table E.5. Coastal top soils.

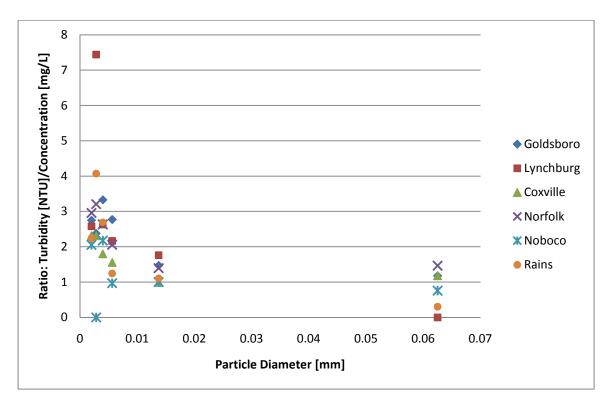


Figure E.5. Coastal top soils' relationship of particle diameter and the ratio of turbidity to

concentration.

Soil Series	D [mm]	Conc. [mg/L]	Turbidity [NTU]	Actual Conc. [mg/L]	Actual Turbidity [NTU]	Turb./Conc.
Coxville	0.063	20200	28370	6440	2630	0.41
	0.014	13760	25740	5230	2540	0.49
	0.006	8530	23060	1160	0	0.00
	0.004	7370	21170	1150	0	0.00
	0.003	6220	23200	1070	4700	4.39
	0.002	5150	18500	5150	18500	3.59
Noboco	0.063	14710	17862	5470	1230	0.22
	0.014	9240	16632	4740	888	0.19
	0.006	4500	15744	740	2148	2.90
	0.004	3760	13596	1160	5252	4.53
	0.003	2600	8344	770	1780	2.31
	0.002	1830	6564	1830	6564	3.59
Goldsboro	0.063	10410	10968	4280	0	0.00
	0.014	6130	11816	2000	3166	1.58
	0.006	4130	8650	600	1516	2.53
	0.004	3530	7134	310	1214	3.92
	0.003	3220	5920	550	892	1.62
	0.002	2670	5028	2670	5028	1.88
Lynchburg	0.063	15460	16524	12240	10332	0.84
	0.014	3220	6192	1330	2032	1.53
	0.006	1890	4160	350	525	1.50
	0.004	1540	3635	210	1087	5.18
	0.003	1330	1702	170	0	0.00
	0.002	1160	2548	1160	2548	2.20
Norfolk	0.063	11750	16324	10200	12908	1.27
	0.014	1550	3416	790	1523	1.93
	0.006	760	1893	90	256	2.84
	0.004	670	1637	60	41	0.68
	0.003	610	1596	120	349	2.91
	0.002	490	1247	490	1247	2.54
Rains	0.063	8260	12732	3760	4812	1.28
	0.014	4500	7920	2330	2550	1.09
	0.006	2170	5370	610	1236	2.03
	0.004	1560	4134	300	654	2.18
	0.003	1260	3480	360	1110	3.08
	0.002	900	2370	900	2370	2.63

Table E.6. Coastal subsoils.

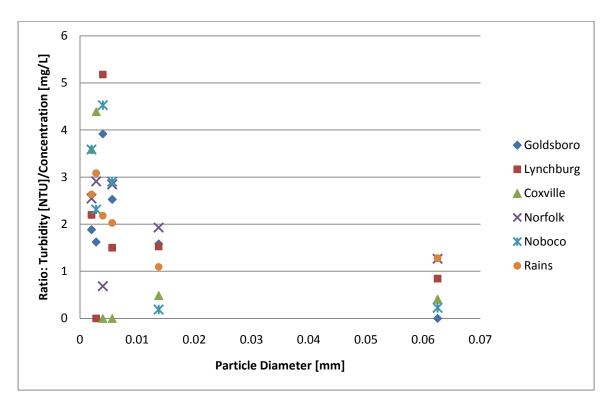


Figure E.6. Coastal subsoils' relationship of particle diameter and the ratio of turbidity to concentration.

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