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Carbon fibers derived from sustainable precursors

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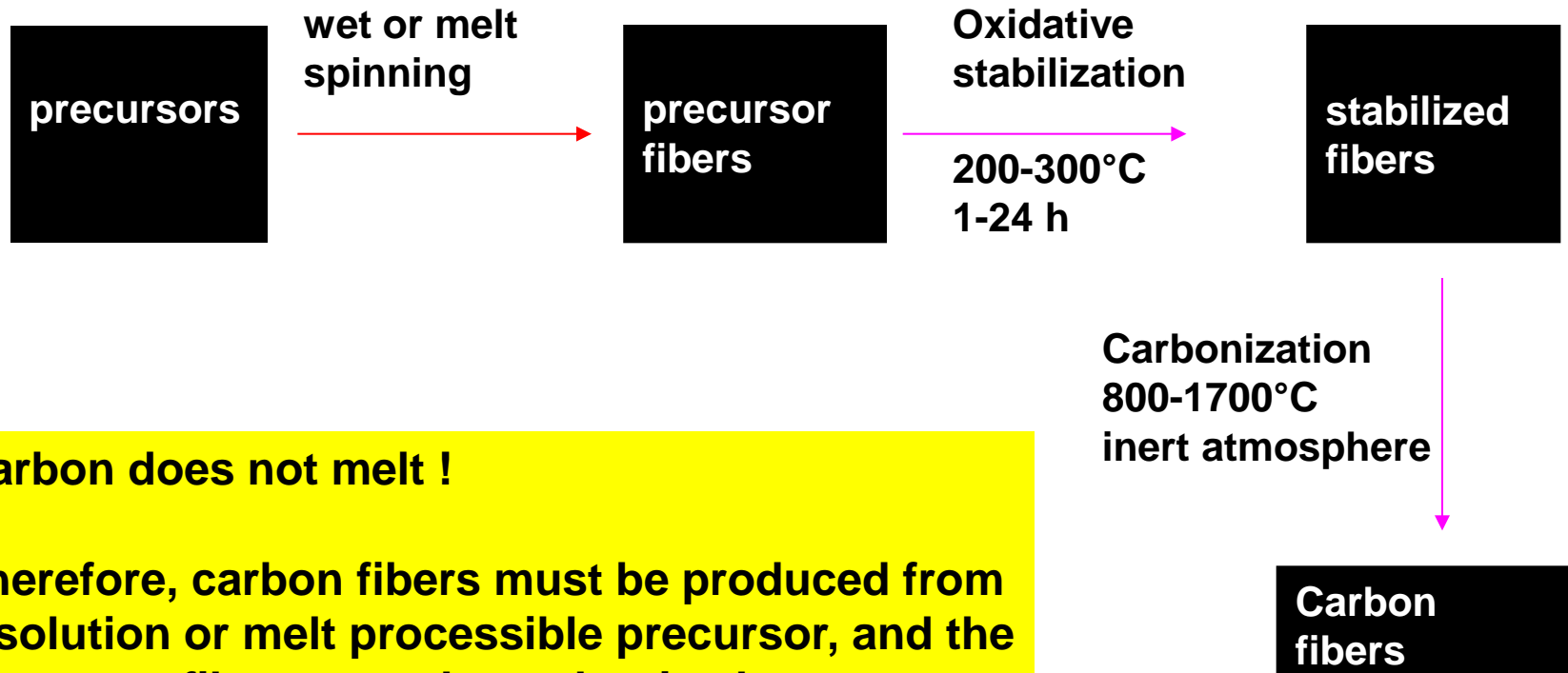
Outline

- **Literature Review: carbon fiber precursors**
PAN, mesophase pitch, rayon, and lignin
- **Motivation and Objectives**
- **Experimental**
 - **Melt spinning with ECN organosolv lignin**
 - **Solution spinning with acetylated Indulin AT**
- **Results and Discussion**
- **Conclusions**
- **Future Work**

Carbon Fiber Characteristics

- ✓ Excellent Strength and Stiffness = high performance
 - ✓ Light-weight = fuel-efficient
 - ✓ Outstanding Electrical and thermal conductivity
 - ✓ Fire-retardant
-
- ❑ Not Cost-Competitive
 - ❑ Current precursors are not bio-based and fibers are not produced by environmentally-friendly processes

Production of Carbon Fibers: Background



Carbon does not melt !

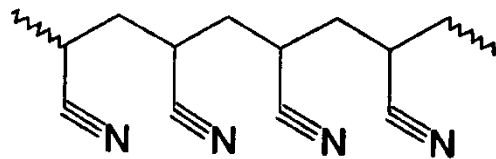
Therefore, carbon fibers must be produced from a solution or melt processible precursor, and the precursor fibers must be carbonized

Carbon Fiber Precursors

- Polyacrylonitrile (PAN)
- Mesophase pitch
- Rayon
- Lignin (current research)

PAN Precursors

The precursor for PAN-based carbon fibers is actually a copolymer.



Polyacrylonitrile

I

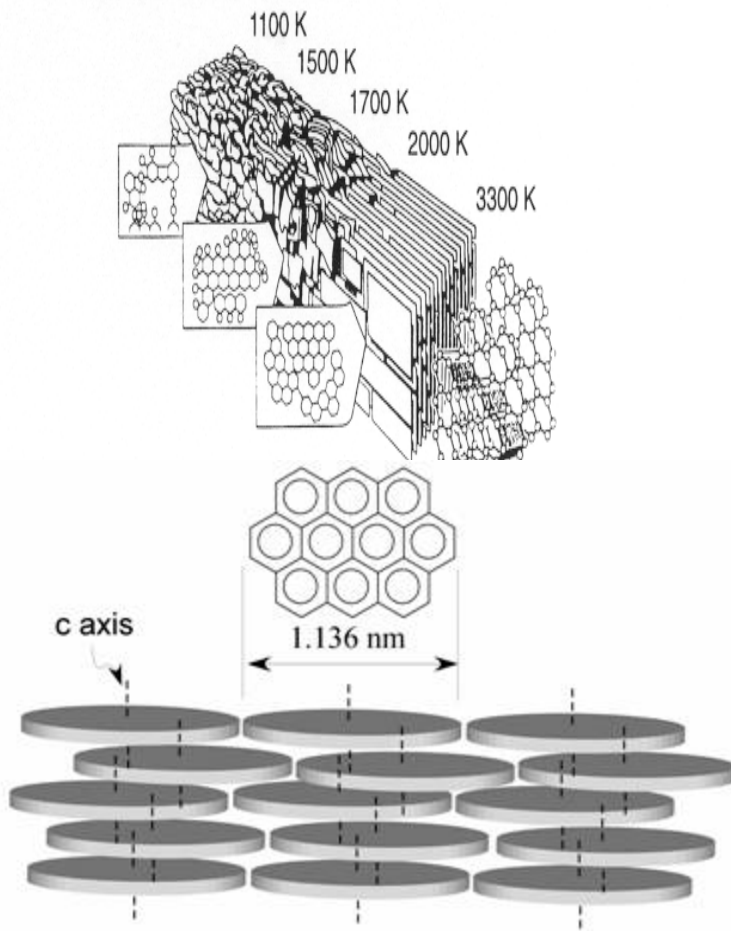
Evolution of HCN and other toxic gases during stabilization and carbonization

Buckley & Edie, 1986

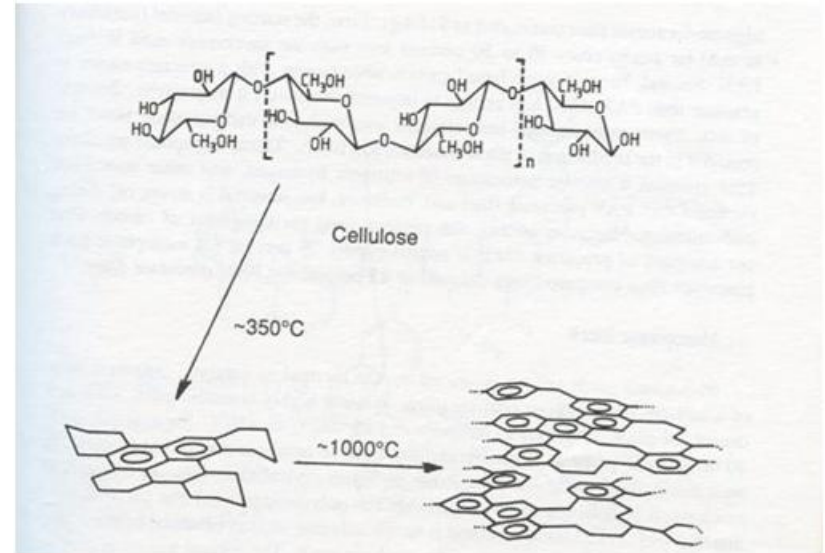
Fitzer & Manocha, 1998

Monomer	Structure
Acrylic Acid	$\begin{array}{c} \text{H} & & \text{H} \\ & \backslash & / \\ & \text{C} = \text{C} \\ & / & \backslash \\ \text{H} & & \text{C} = \text{O} \\ & & \\ & & \text{OH} \end{array}$
Itaconic Acid	$\begin{array}{c} & & \text{OH} \\ & & \\ & & \text{C} = \text{O} \\ & & \\ \text{H} & & \text{C} \\ & \backslash & / \\ & \text{C} = \text{C} \\ & / & \backslash \\ \text{H} & & \text{CH}_2 \\ & & \\ & & \text{C} = \text{O} \\ & & \\ & & \text{OH} \end{array}$
Methacrylic Acid	$\begin{array}{c} & & \text{OH} \\ & & \\ & & \text{C} = \text{O} \\ & & \\ \text{H} & & \text{C} \\ & \backslash & / \\ & \text{C} = \text{C} \\ & / & \backslash \\ \text{H} & & \text{CH}_3 \end{array}$
Methyl Acrylate	$\begin{array}{c} \text{H} & & \text{H} \\ & \backslash & / \\ & \text{C} = \text{C} \\ & / & \backslash \\ \text{H} & & \text{C} = \text{O} \\ & & \\ & & \text{O} \\ & & \\ & & \text{CH}_3 \end{array}$
Vinyl Acetate	$\begin{array}{c} \text{H} & & \text{H} \\ & \backslash & / \\ & \text{C} = \text{C} \\ & / & \backslash \\ \text{H} & & \text{O} \\ & & \\ & & \text{C} = \text{O} \\ & & \\ & & \text{CH}_3 \end{array}$
Acrylonitrile	$\begin{array}{c} \text{H} & & \text{H} \\ & \backslash & / \\ & \text{C} = \text{C} \\ & / & \backslash \\ \text{H} & & \text{CN} \end{array}$

Precursors: Mesophase Pitch and Rayon



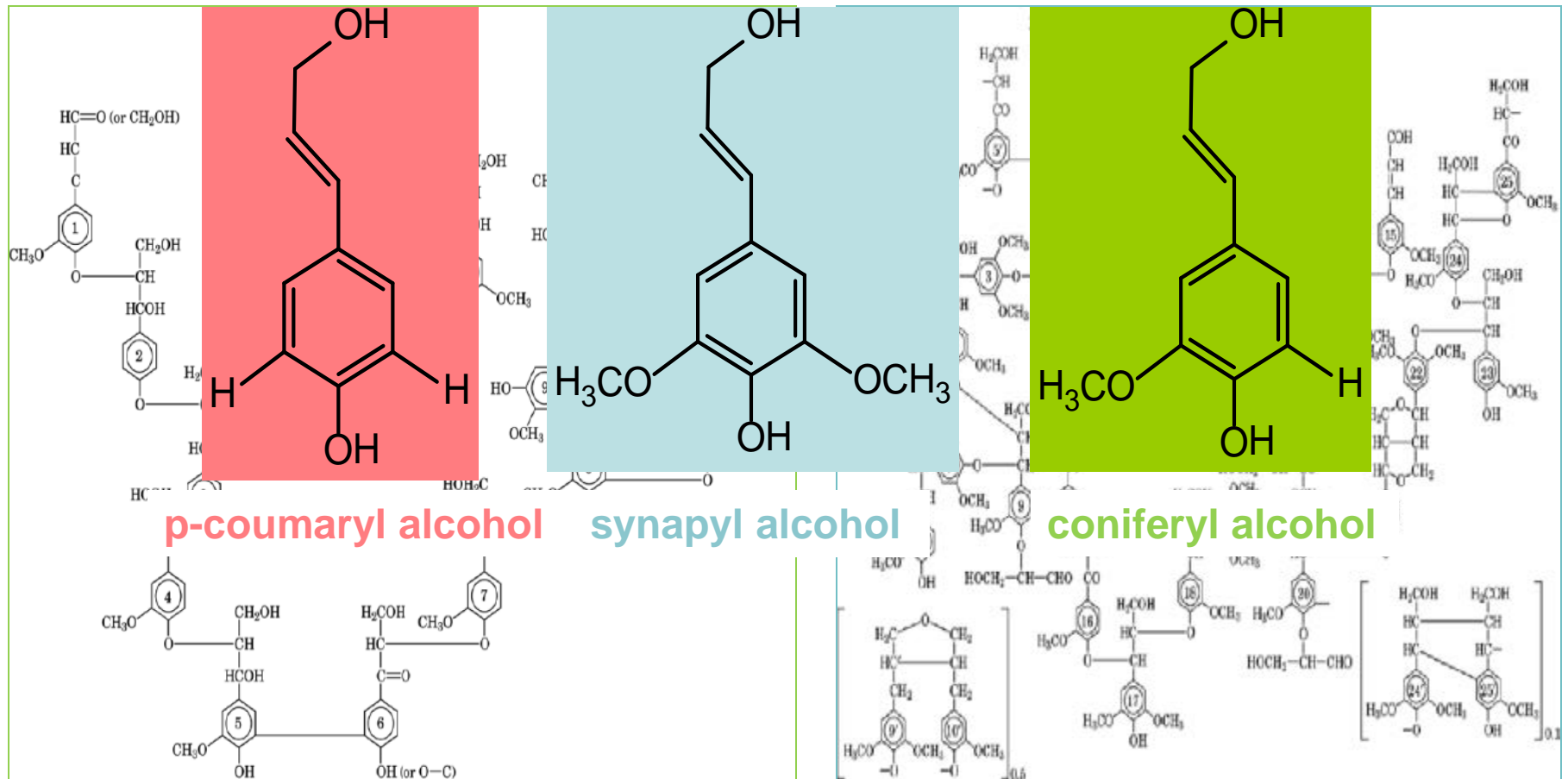
Kundu, ..Ogale, CARBON 2008



Buckley & Edie, 1986

Not for structural applications

Chemical Structure of Lignin



Softwood Lignin

Hardwood Lignin

E. Alder, *Wood Science & Technology*, 11, 169 (1977)
 H. H. Nimz, *Angew. Chem. Int. Ed.*, 13, 313 (1974)

Lignin

Source:

wood, grass, wheat straw, etc

Separation process:

kraft, soda, organosolv pulping, etc

Literature Review

- Different lignin precursor, NaOH solution for dry-spinning / melt spinning. 1969, Otani
- Steam exploded hardwood lignin followed by hydrogenation and several extraction steps, melt-spinning. 1991, K. Sudo *et al*
- Organosolv (acetic acid) hardwood lignin based carbon fiber, melt-spinning 1993, Y. Uraki *et al*; 1995, S. Kubo *et al*
- Organosolv (acetic acid) softwood lignin, melt-spinning, 1998, S. Kubo *et al*
- Hardwood kraft lignin, melt spinning. 2002, J. F. Kadla *et al*
- Acetylated softwood kraft lignin, melt-spinning. 2008, R. C. Eckert
- Softwood kraft lignin using hardwood kraft lignin as plasticizer, melt-spinning. Baker, F. S. EERE, U.S. Dept of Energy
Project ID # Im_03_baker

Mechanical properties of lignin based CF

Precusor Type	Diameter (μm)	Elongation (%)	Modulus (GPa)	Tensile strength (MPa)	Reference
Steam Exploded hardwood	7.6 ± 2.7	1.63 ± 0.19	40.7 ± 6.3	660 ± 230	K. Sudo <i>et al</i> , 1992
Organosolv Hardwood	14-35	0.64-1.12	2.17-39.1	13.3-355	Y. Uraki <i>et al</i> , 1995
Organosolv Softwood	84 ± 15	0.74 ± 0.14	3.59 ± 0.43	26.4 ± 3.1	S. Kubo <i>et al</i> , 1998
Kraft Hardwood	46 ± 8	1.12 ± 0.22	40 ± 11	422 ± 80	J. F. Kadla <i>et al</i> , 2002
Kraft Softwood, acetylated	5-100	N/A	N/A	N/A	Robert C., 2008
Hardwook		2.03	82.7	1070	D. A. Baker, 2013
Rayon based carbon fiber	5-25		100	100-1000	Buckley & Edie; Fitzer & Manocha
PAN based carbon fiber	5-15	2	100-500	3000-7000	Buckley & Edie; Fitzer & Manocha
Mesophase pitch based carbon fiber	5-15	0.6	200-800	1000-3000	Buckley & Edie; Fitzer & Manocha



Goal:

Lignin-based carbon fibers with higher performance properties

Specific objectives:

Chemical modification of separated lignin

Preparation of lignin based carbon fiber

Spinning

Thermostabilization

Carbonization

Microstructure and Properties

Tensile

Nanotexture and Graphitic Crystallinity



Experimental

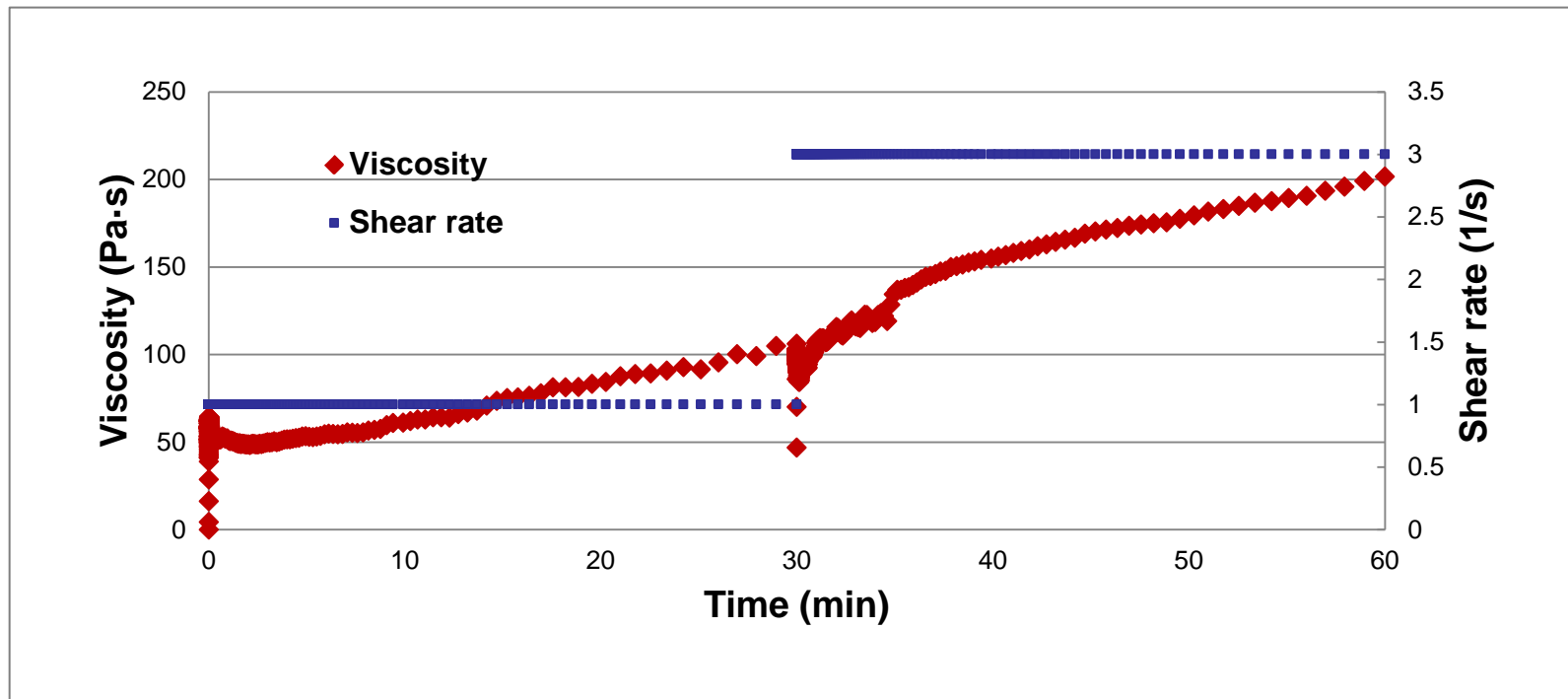
Materials

- ECN lignin (Organosolv lignin, Energy Research Centre of the Netherlands)
- SKL Softwood Kraft lignin (Indulin AT, MeadWestvaco, Charleston, SC)

Melt spinning of ECN organosolv lignin

- Source: Poplar wood lignin from ethanol/H₂O pulping
- Softening point: 155°C
- Decomposition temperature: ~280°C from TGA result in N₂ purge

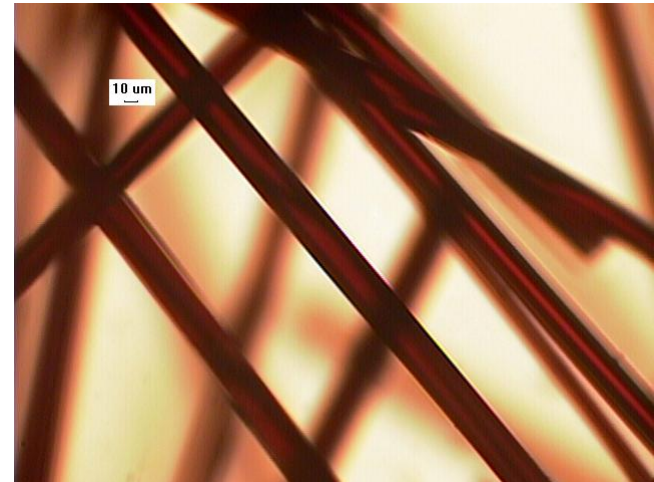
Transient shear viscosity of ECN lignin (@ 160°C)



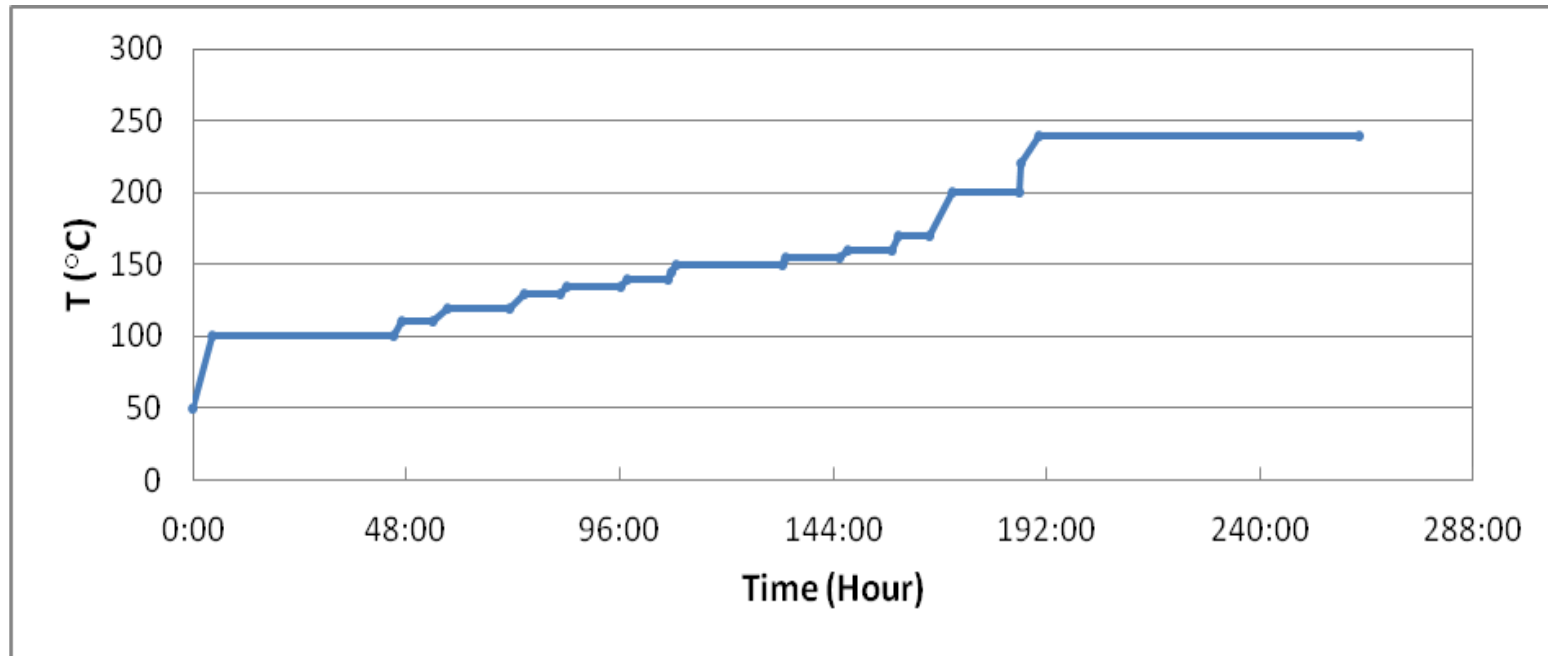
Melt spinning of ECN lignin



- Instron Capillary Rheometer
- Temperature: 160°C
- Winding rate: 190 m/min
- Capillary diameter: 254 μm
- Fiber diameter: $29 \pm 1 \mu\text{m}$

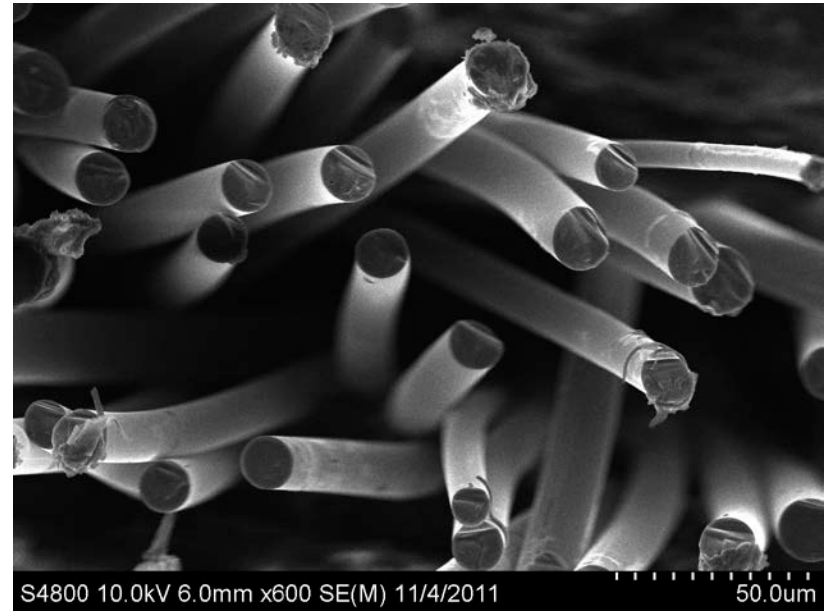
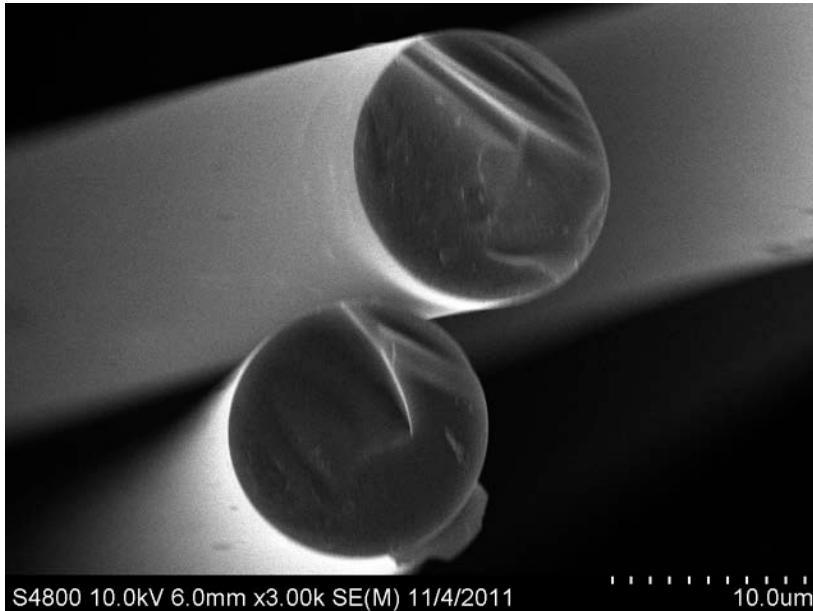


Thermostabilization of ECN fibers



It takes more than 10 days to stabilize to prevent fibers from being tacky





ECN carbon fibers had a smooth surface and circular cross section

Mechanical properties of ECN carbon fibers

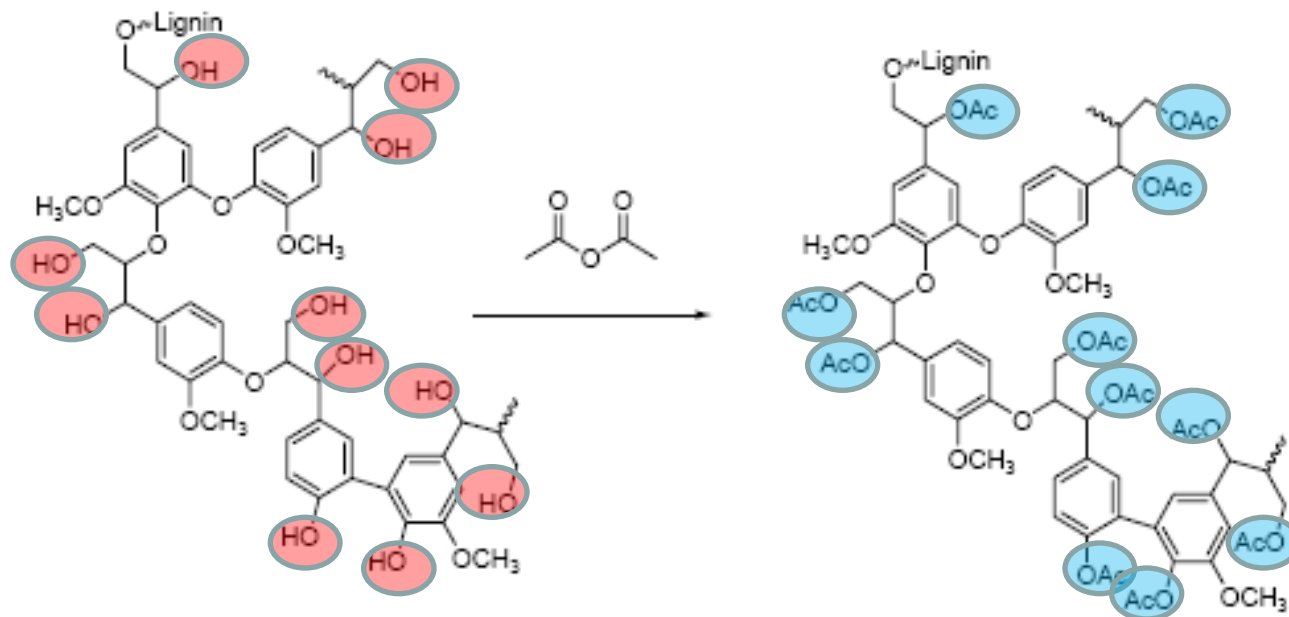
Diameter (μm)	Elongation (%)	Modulus(GPa)	Tensile strength (MPa)
14 ± 1	1.4 ± 0.4	34 ± 4	450 ± 130

Indulin AT Lignin

- Indulin AT (Softwood Kraft lignin, MeadWestvaco, Charleston, SC)
- No Softening Point, charring occurred due to high molecular weight fraction and dehydration reaction



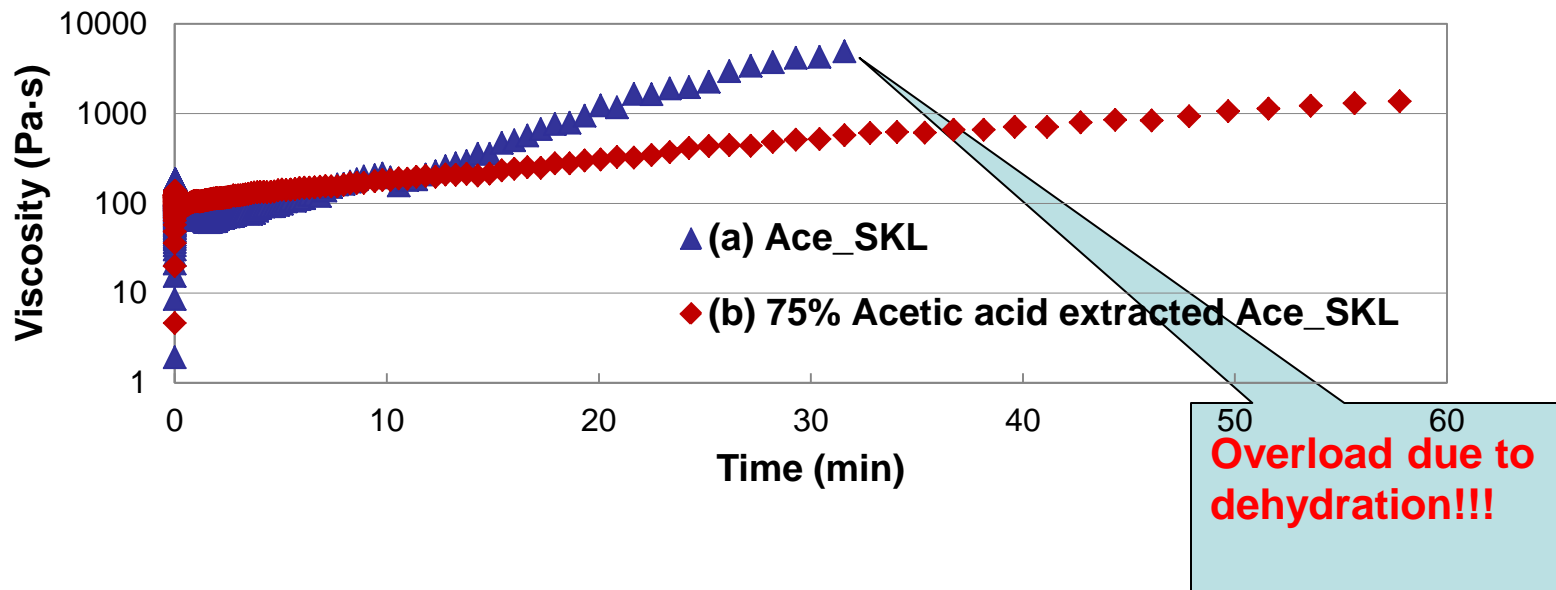
Previous modification of Indulin AT-Acetylation with high extent of substitution on -OH group and fractionation



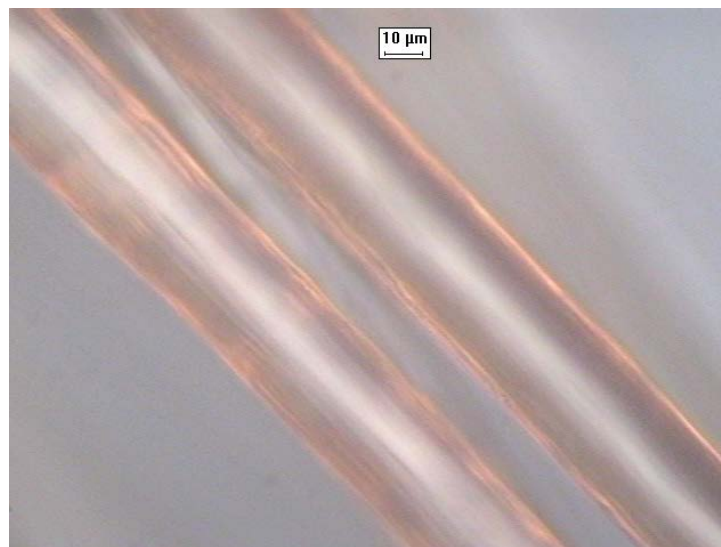
- 1 g lignin + 15 ml acetic anhydride, 85°C, 2 hour
- Acetylated Indulin AT (**Ace-SKL**) had a softening point between 156 and 167°C



- Ace-SKL had a softening point, but unstable melt viscosity.
- Ace-SKL was extracted with 75% acetic acid aqueous solution. Resulted material (**75%AA-Ace-SKL**) had a softening point of 136-145°C.
- 75% acetic acid extracted Ace-SKL had relatively stable melt viscosity.



- 75%AA-Ace-SKL was melt spun into fibers
- 75%AA-Ace-SKL fibers became tacky during oxidative stabilization



As-spun Ace-SKL fibers

Thermostabilization

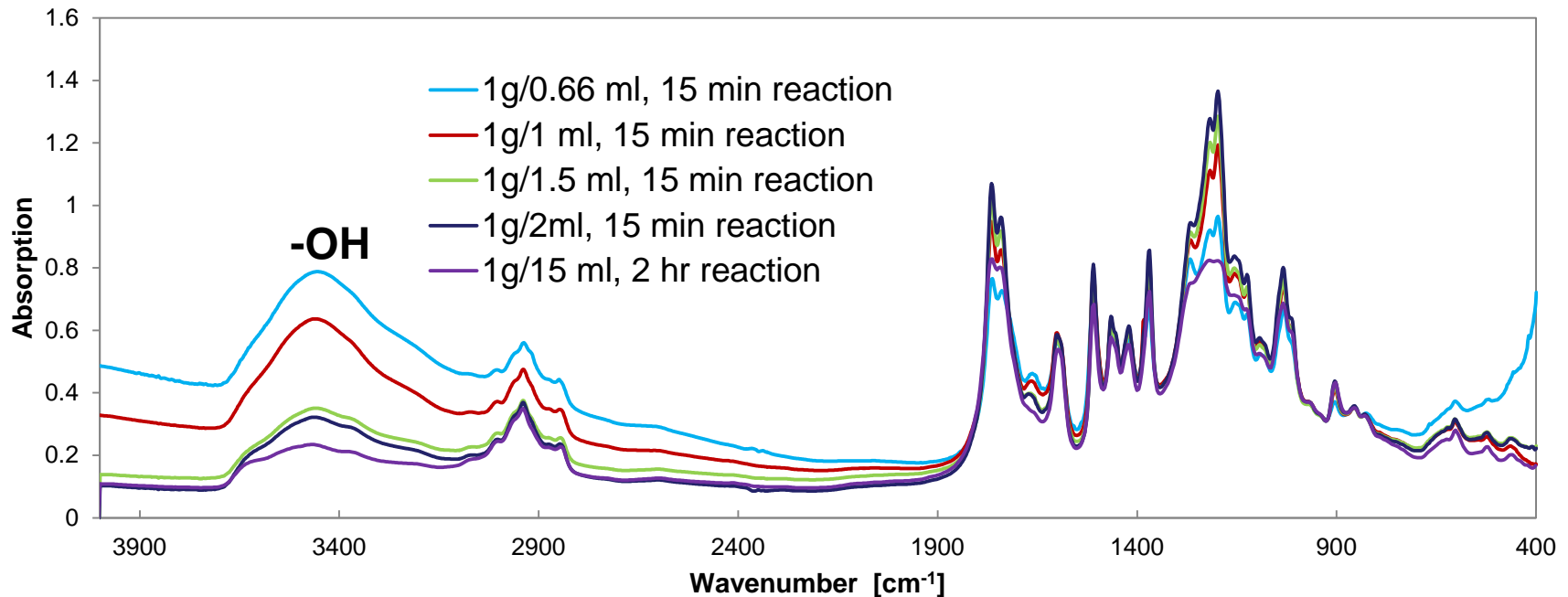


Tacky Ace-SKL fibers

Ace-SKL fibers obtained from high extent of acetylation (15 ml AA/g SKL) could not be stabilized due to the presence of a significant extent of substitution of hydroxyl groups by thermally stable acetyl groups

Alternative way:

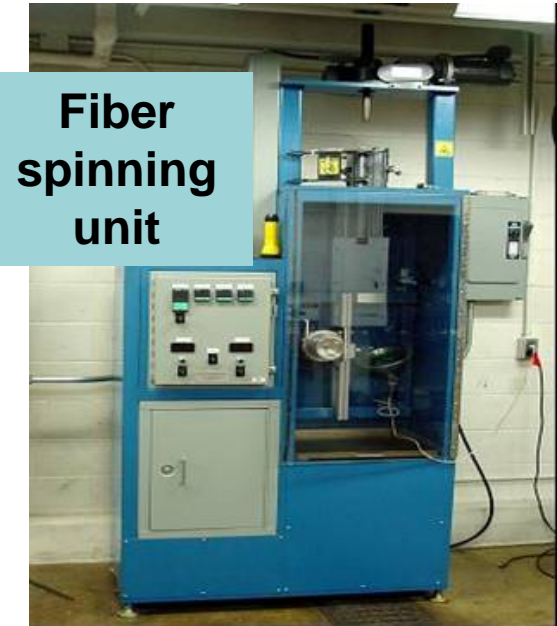
- **Ace-SKL lower extent of -OH group substitution, which is favorable for thermostabilization**
- **Solution spinning instead of melt spinning**



- **Spectra normalized with peaks at 856 cm⁻¹ (C-H bending on benzene rings).**
- **The hydroxyl peak decreased as the amount of acetic anhydride per gram of SKL increased.**
- **Higher content of hydroxyl group is favorable for thermostabilization.**

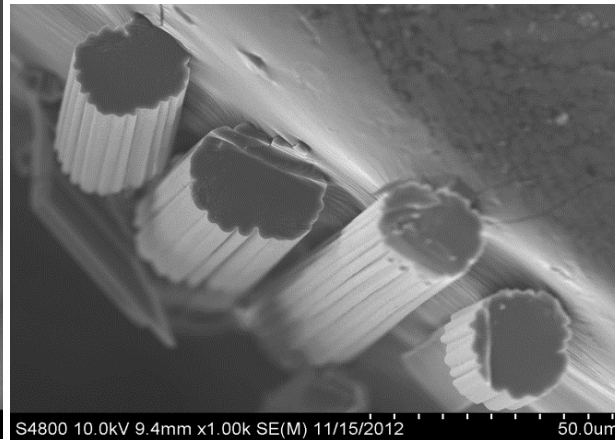
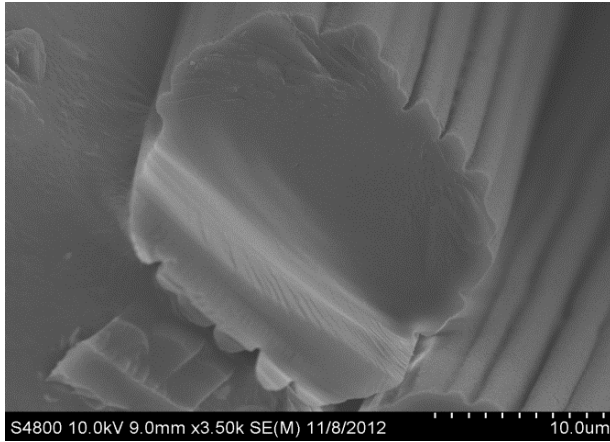
Solution spinning with Ace-SKL

- Ace-SKL acetone solution concentrated
- Take up speed: 50 m/min
- Spinneret diameter: 75-150 μm
- Fiber diameter: $27 \pm 3 \mu\text{m}$

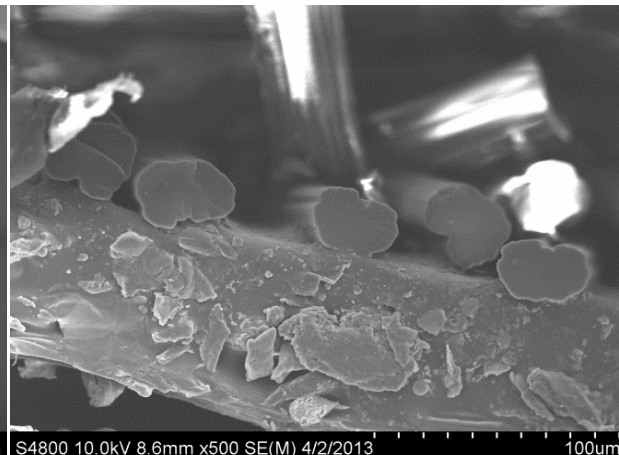
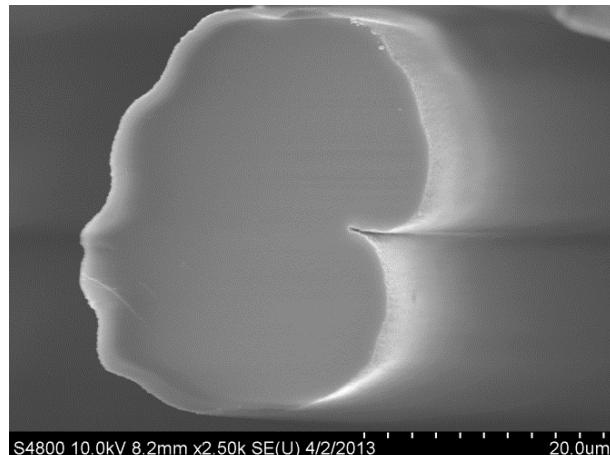


Solution spinning with Ace-SKL

2.1 g Ace-SKL / ml acetone, 45-45°C spinning



2.1 g/ml acetone room temperature spinning



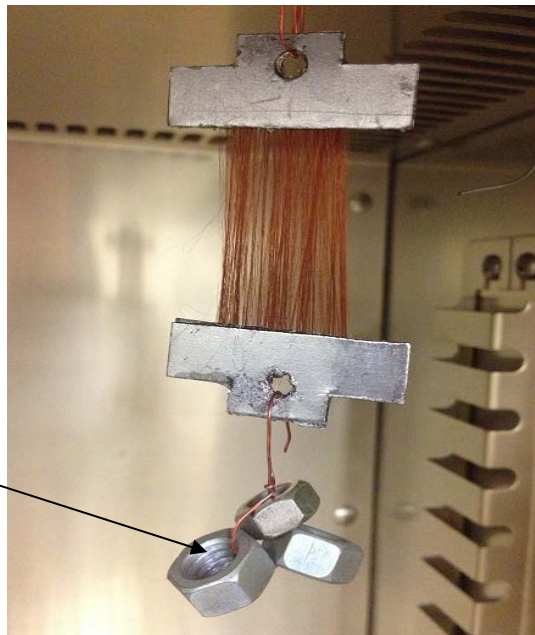
Thermostabilization of Ace-SKL fibers under tension

Stabilization with constant load

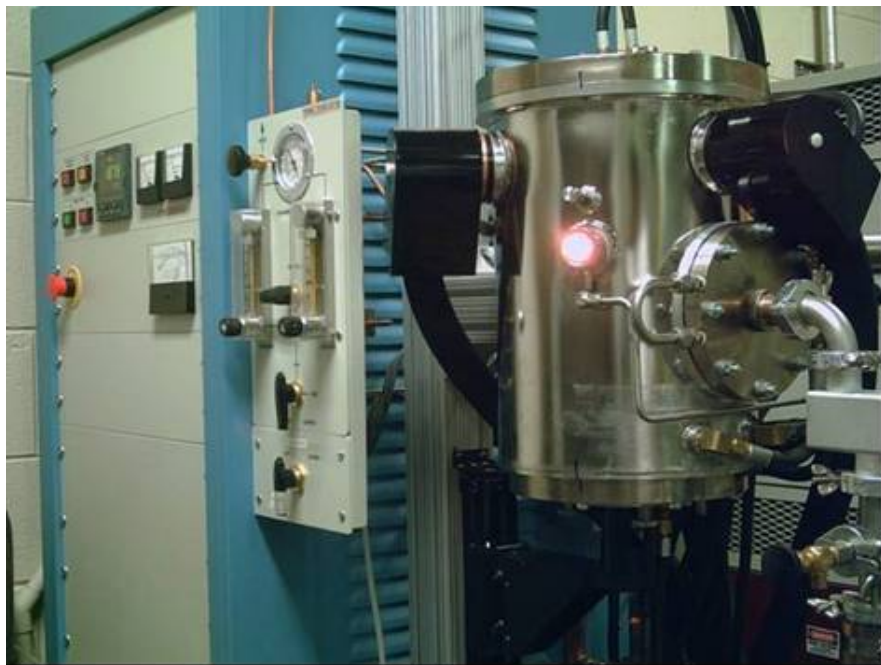
As-spun fiber was glued on both ends with hook and hanging in the oxidation oven with weight loaded.

Fibers can be stabilized and extended up to 800% of original length during stabilization.

Weight



Carbonization of stabilized Ace-SKL fibers

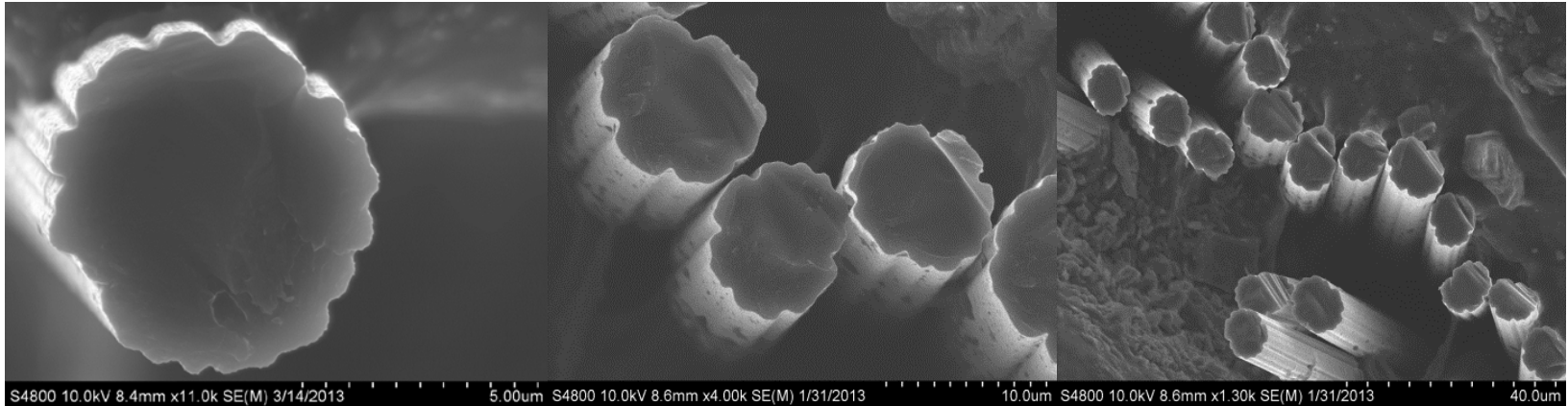


**Carbonization and Graphitization
furnaces: 1000-2700°C**



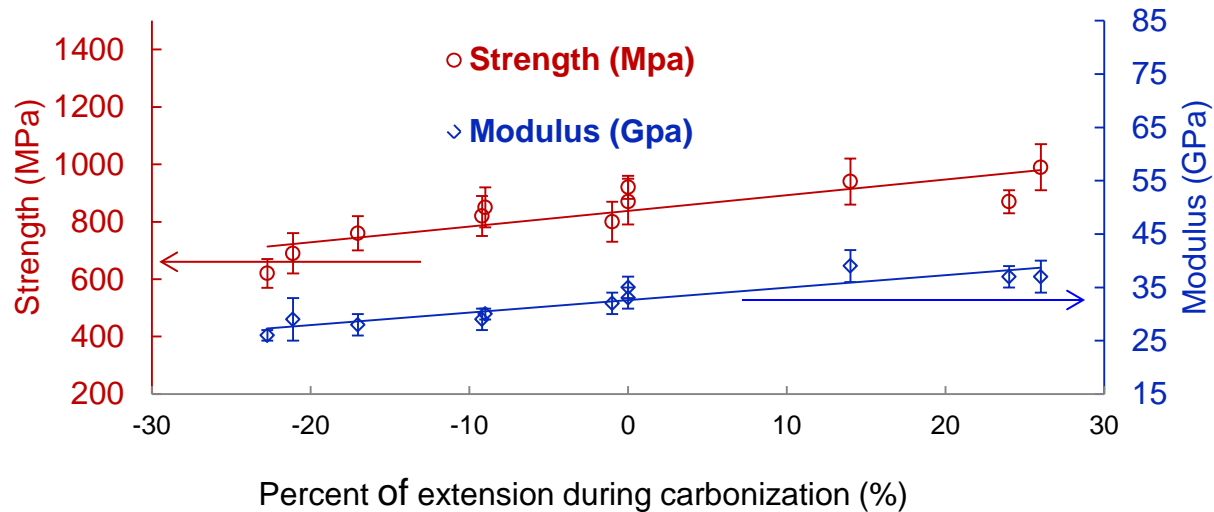
Carbonization under tension

1000°C carbonized



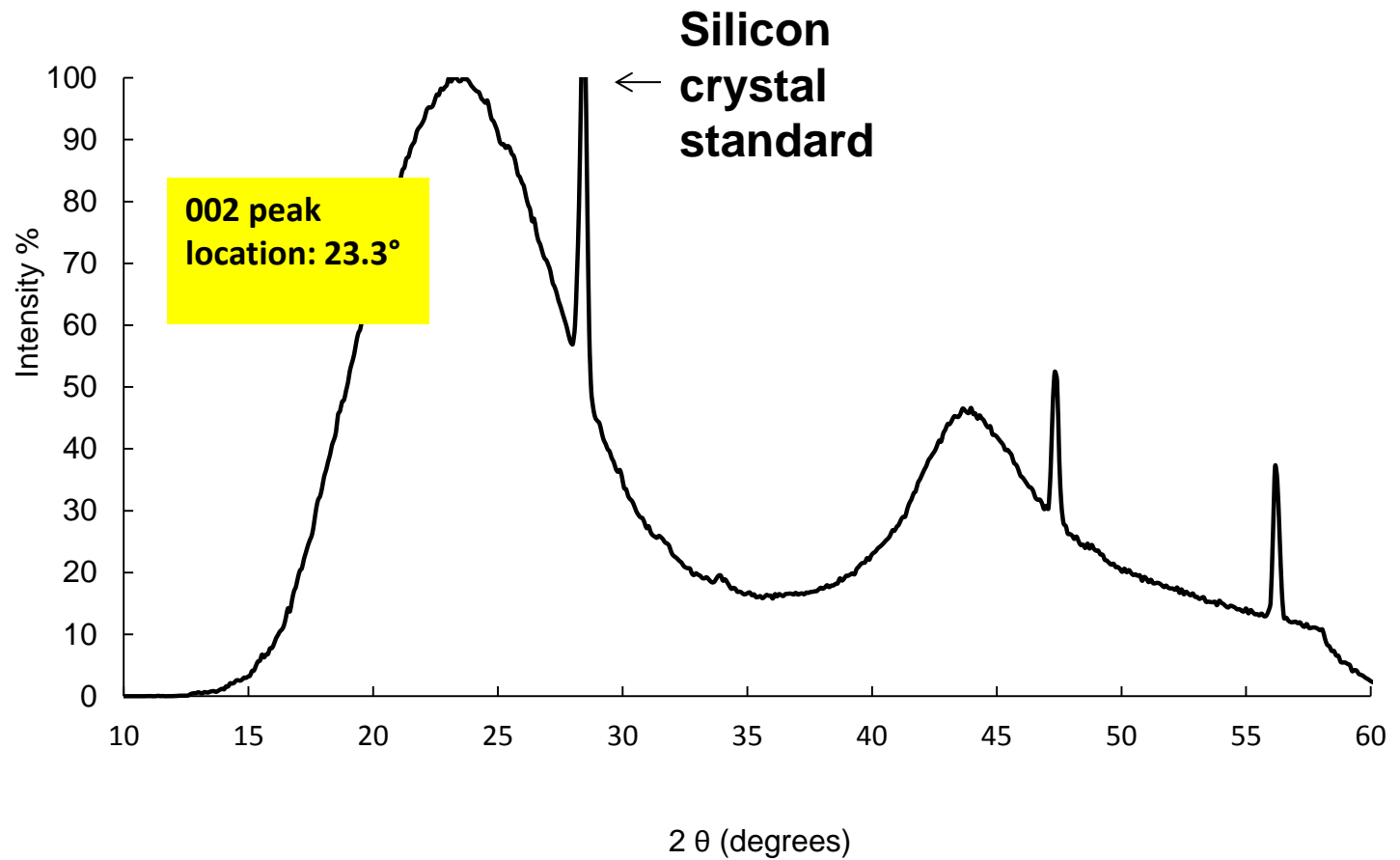
- **Crenulated CF have 35% larger surface area as compared with equivalent circular fibers**
- **This could lead to higher fiber-matrix interfacial bond strength, and ultimately better realizability of carbon fiber properties in the composites**

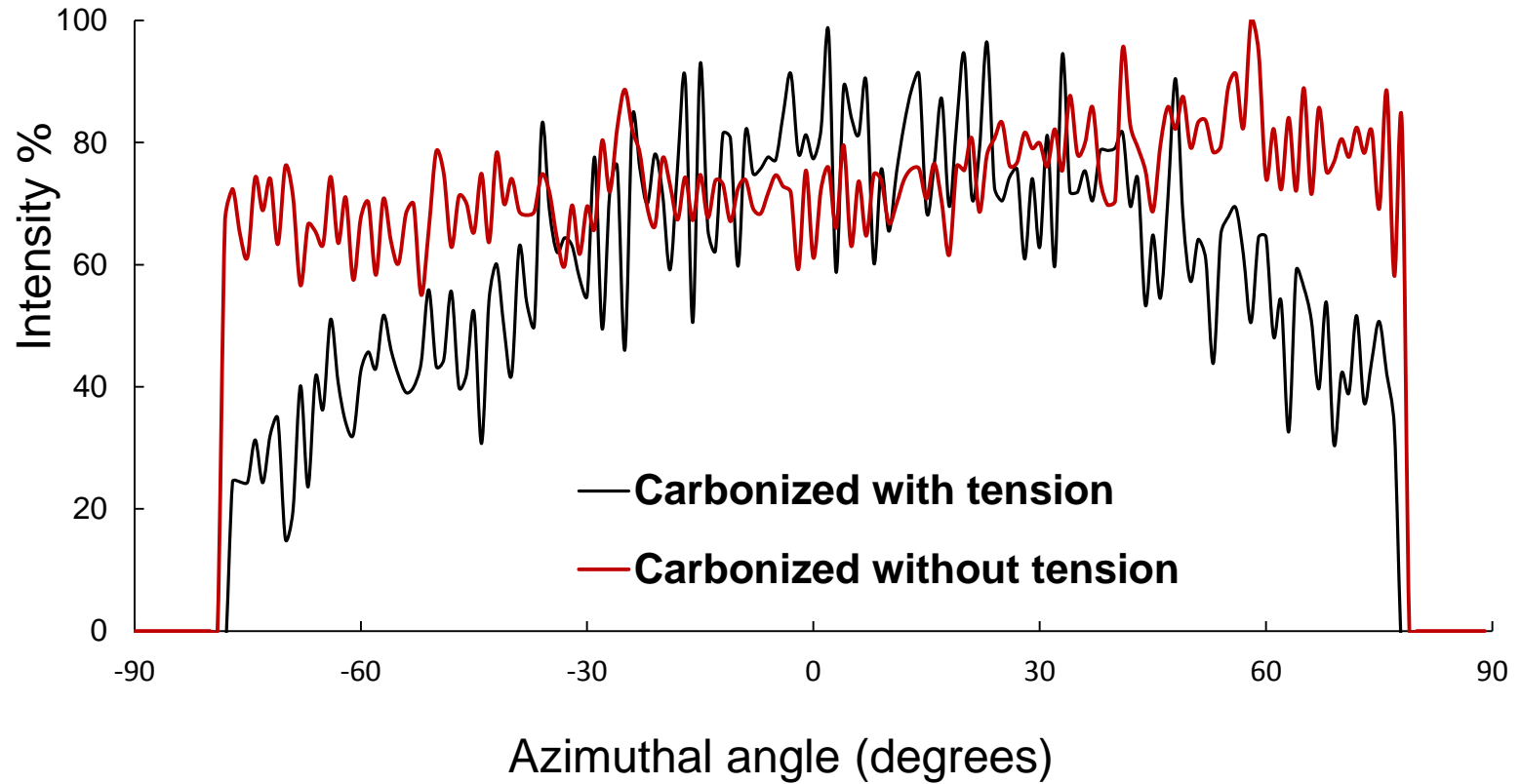
Mechanical properties of Ace-SKL carbon fibers



	Diameter (μm)	Strength (MPa)	Apparent Modulus (GPa)	Apparent strain to failure (%)
Ace-SKL CF (processed without tension)	22.5 ± 0.4	510 ± 50	30 ± 2	1.7 ± 0.1
Ace-SKL CF (processed with tension)	5.9 ± 0.2	1050 ± 70	35 ± 3	3.0 ± 0.2

Ace-SKL CF X-ray Diffraction Spectrum







Conclusions

- A softwood kraft lignin was modified by controlled acetylation and the precursor (Ace-SKL) was solution-spun into fibers, which is capable of thermal-oxidation.
- Mechanical properties of Ace-SKL carbon fibers (CF) can be enhanced by tension. The tensile properties reported here is among the best for lignin-based CF.
- Crenulation on Ace-SKL CF surface lead to larger surface area and potential higher fiber-matrix interfacial strength.

Next steps...

- Rheology of spinning solution is being studied
- Relationship between fiber cross-section shape and mechanical properties will be studied
- UV/thermostabilization to increase stabilization speed

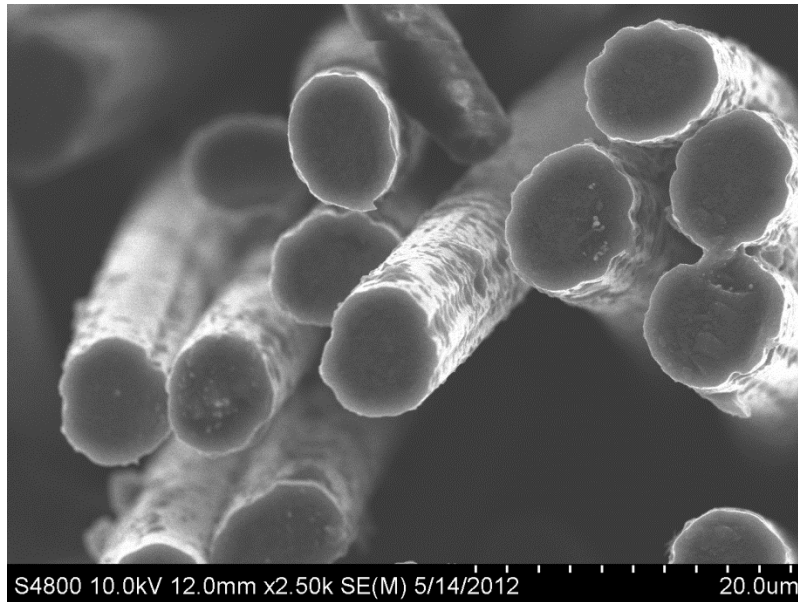


Acknowledgment

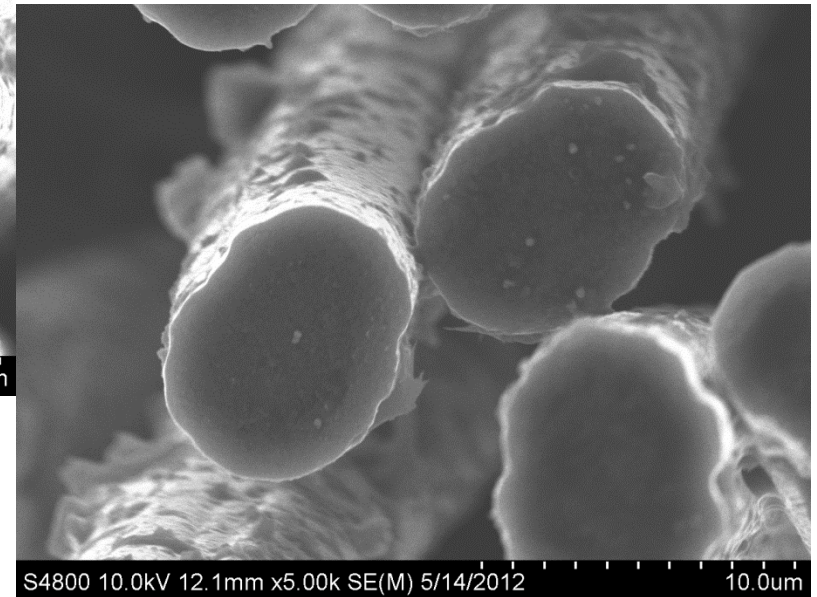
ARL-SERDP Grant WPSO-N-10-03 / W911NF-10-2-0024

ARL/UD/Drexel/Clemson team members including Dr. Marlon Morales and Dr. Young-pyo Jeon

ECN, Netherlands for providing lignin



**2400°C
carbonized**



Crenulated surface are desirable for enhancing fiber-matrix interfacial area