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CLEANING ABILITY OF CHLORHEXIDINE GELAND SODIUM HYPOCHLORITE ASSOCIATED OR NOT WITH **EDTA AS ROOT CANAL IRRIGANTS: A SCANNING ELECTRON MICROSCOPY STUDY**

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ABSTRACT

The aim of this study was to evaluate the cleaning efficacy of 2% chlorhexidine gluconate gel (CHX) compared to 2.5% sodium hypochlorite solution (NaOCl) associated or not with 17% EDTA used as irrigants during the biomechanical preparation. Fifty freshly extracted single-rooted human teeth with complete apex formation were randomly divided into five groups: G1 sterile saline, G2 - 2.5% NaOCl, G3 - 2% CHX, G4 - 2.5% NaOCl + EDTA and G5 - 2% CHX + EDTA. The specimens of G1 were subdivided into two control groups. The teeth were decoronated and the coronal and middle root thirds were prepared with Gates-Glidden burs, and the apical third was reserved to manual instrumentation. All procedures were performed by a single operator. In all groups, 2 mL of irrigant was delivered between each file change. The teeth were sectioned and prepared for analyses under scanning electron microscopy (SEM). SEM micrographs were graded according to a score scale by two examiners. Data were analyzed statistically by Kruskal-Wallis and Dunn tests at 1% significance level. The best results were obtained in the groups in which the irrigant was used followed by the chelating agent. No statistically significant difference was observed among G4, G5 and the positive control group (p<0.01). The groups G2 and G3 were significantly different from the others, presenting the worst cleaning capacity. In conclusion, the use of the chelating agent is necessary to obtain clean canal walls, with open tubules and no heavy debris. The use of 2% chlorhexidine gluconate gel alone is not able to remove the smear layer.

Uniterms: Root canal preparation; Root canal irrigation; Root canal cleaning.

INTRODUCTION

The presence of bacteria and their byproducts in the development of pulp and periapical lesions is well established¹². Therefore, one of the objectives of endodontic treatment is the elimination of microorganisms from the root canal system. However, no instrumentation technique or irrigating solution is able to totally eliminate these microorganisms or their toxins^{20,22}.

After biomechanical preparation, residual pulp tissue, bacteria and dentin chips may persist in the irregularities of the root canal system, covering the canal walls or lodged in the dentinal tubules even after careful instrumentation^{1,22,23}.

The presence of smear layer and debris decreases the penetration ability of intracanal dressings and also prevents complete adaptation of obturation materials^{5,25,28}. An irrigant that could completely remove the smear layer would contribute to reduce the microbiota and associated endotoxins and decrease the potential of bacteria to survive and reproduce¹³.

Selection of irrigants is very important because they can act as lubricants during instrumentation, remove the debris and necrotic pulp tissue and help eliminating or neutralizing microorganisms and their byproducts7,28,31.

The most popular endodontic irrigant is sodium hypochlorite (NaOCl)^{3,32}. The large use of NaOCl is related

to its great physicochemical, antibacterial and tissue dissolving properties. Nevertheless, it is know to be highly irritating to the periapical tissues when used at high concentrations^{6,17}. It has undesirable characteristics such as toxicity, risk of emphysema, allergic potential and unpleasant smell and taste^{6,11,31}.

Chlorhexidine gluconate has been extensively used in dentistry, showing good results in caries control and as an aid in periodontal therapy³¹. It has been suggested as an endodontic irrigant, in liquid or gel form, as an alternative to NaOCl, because it has a broad-spectrum antimicrobial effect, presents residual action and biocompatibility, and has great physicochemical properties^{4,9,15,27,29}.

The capacity of chlorhexidine gluconate gel to remove the smear layer is not yet clear and the literature brings ambiguous results of the cleaning ability of this substance as an endodontic irrigant^{7,14,20,31}. Thus, the purpose of this study was to evaluate the efficacy of cleaning and flushing of debris of 2% chlorhexidine gluconate gel and 2.5% NaOCl associated or not to 17% EDTA.

MATERIAL AND METHODS

Fifty freshly extracted, single-rooted human teeth with complete apex formation were used. The study was approved by the institutional Ethics Committee. The crowns were sectioned with a bur in a low-speed handpiece producing 17-mm root specimens.

Roots were randomly divided into five groups, with ten roots each: group 1 (G1) sterile saline (Farmence, Barbalha, CE, Brazil); group 2 (G2) 2.5% NaOCl (Biodinâmica, Ibiporã, PR, Brazil); group 3 (G3) 2% chlorhexidine gluconate gel (Essencial Farma, Itapetininga, SP, Brazil); group 4 (G4) 2.5% NaOCl followed by 17% EDTA (Biodinâmica, Ibiporã, PR, Brazil); and group 5 (G5) 2% chlorhexidine gluconate gel followed by 17% EDTA (Biodinâmica, Ibiporã, PR, Brazil); and group 5 (G5) 2% chlorhexidine gluconate gel followed by 17% EDTA (Biodinâmica, Ibiporã, PR, Brazil). The teeth in G1 were subdivided into two control groups (n=5): a negative control group, irrigated only with sterile saline; and a positive control group irrigated with sterile saline and submitted, after preparation, to a 5-miniute ultrasonic bath in 2.5% NaOCl, followed by an additional 1minute bath in 17% EDTA.

Mechanical preparation was performed by a single operator according to a crown-down technique. The coronal and middle root thirds were prepared with Gates-Glidden burs (Dentsply/Maillefer, Ballaigues, Switzerland) used in decreasing order: GG#6, GG#5, GG#4, GG#3⁸. Manual instrumentation was performed on the apical third with sizes #60, #55, #50 and #45 K-files (Dentsply/Maillefer, Ballaigues, Switzerland), using the balanced force technique. In all experimental groups, 2 mL of freshly prepared irrigant were applied between each change of instrument or bur, and a final flush was performed with 3 mL of sterile saline. In groups G4 and G5, the chelating agent was used during 3 minutes under agitation performed with a #30 K-file (Dentsply/Maillefer, Ballaigues, Switzerland), followed by another flush with 3 mL of sterile saline³¹. Irrigation was performed using

10-mL disposable syringes with 20x5.5 disposable needles without bevel (Becton Dickinsin, Juiz de Fora, MG, Brazil).

The canals were dried with paper points and the roots were split along the long axis in a buccolingual direction. The root halves were mounted with conductive adhesive onto metal stubs, sputter-coated with gold and examined on a scanning electron microscope (Zeiss DSM 940A, Jena, Germany). The entire surface of the apical third of each half was examined at x12 to x1500 magnifications and then, micrographs were taken at x1000 magnification.

The evaluation was performed by classification of the root canal surfaces in different levels of amount of smear layer and debris by two examiners in a double-blind design. The root canal walls were then graded according to Ahlquist, et al. 2 (2001), as follows: for the presence of smear layer, 1 = no smear layer, open dentinal tubules; 2 = little smear layer, most dentinal tubules were open; 3 = homogeneous smear layer covering most part of the surface, a few tubules open; 4 = homogeneous smear layer covering the surface, no dentinal tubules open; 5 = thick non-homogeneous smear layer covering the surface; for the presence of debris, 1 = clean root canal wall, very little amount of debris; 2 = little amount of debris; 3 = moderate amount of debris, less than 50% of the specimen surface covered; 4 = substantial amount of debris, more than 50% of the specimen surface covered; 5 = the root specimen surface was completely or almost completely covered with debris.

Data were statistically analyzed using the Kruskal-Wallis and Dunn tests, applied at a significance level of 1% (p<0.01) as data were not normally distributed. In addition, the degrees of inter and intra-examiner reliability was assessed by calculating Kappa values.

RESULTS

Statistical analysis confirmed the reliability of the method, as the Kappa values were higher than 0.9 for both intra and inter-examiner, indicating high observer agreement. Table 1 shows the values of median and medium post for each group, besides the statistical differences. The analysis of the specimens in the positive control group showed that all surfaces were totally clean and free from smear layer and debris (Figure 1A), without any signs of covered dentinal tubules. In the specimens of the negative control group, the canal walls were totally covered by smear layer and debris were formed over the inner surface of dentinal walls (Figure 1F).

The analysis of groups G4 and G5, respectively treated with 2.5% NaOCl and 2% chlorhexidine gluconate gel followed by 17% EDTA (Figures. 1B and 1C), showed no statistically significant difference between them and the positive control group, presenting the best results (p<0.01). Generally, there was typical thin smear layer and light debris formed by mechanical preparation. Most dentinal tubules of these specimens were open.

The specimens in G2 and G3, respectively irrigated only with 2.5% NaOCl and 2% chlorhexidine gluconate gel

(Figures. 1D and 1E), showed the worst results, with no significant difference (p<0.01) between them and the negative control group. In general, these specimens showed heavy smear layer and debris covering the openings of dentinal tubules; at some areas, the debris and smear layer produced by biomechanical preparation partially covered the dentinal walls. Some dentinal tubules were open and some were covered.

DISCUSSION

A laboratory test is only the first step in the investigation of efficacy of irrigants⁴. The SEM analysis is probably the best tool to identify organic and inorganic debris and smear layer on the root canal walls after endodontic preparation, producing high resolution and magnification images⁸. However, other methods can be found in the literature, as light microscopic evaluation, so, currently there is no consensus on the standardization of methods for measurement of intracanal debris and smear layer^{4,7,8}. Although SEM analysis represents the most popular

TABLE 1- Values of median and mean post for the presence of smear layer and debris for each group

Group	Smear layer		Debris	
	Median	Mean Post	Median	Mean Post
C-	4	188.54	3	202.78
G2	4	204.59	3	181.81
G3	4	185.23	3	177.31
G4	2	74.95 _b	2	101.82 _b
G5	2	90.26 _b	2	88.69 _b
C+	2	53.59 _b	1	41.22

Different letters indicate statistically significant difference (p<0.01; Kruskall Wallis test).

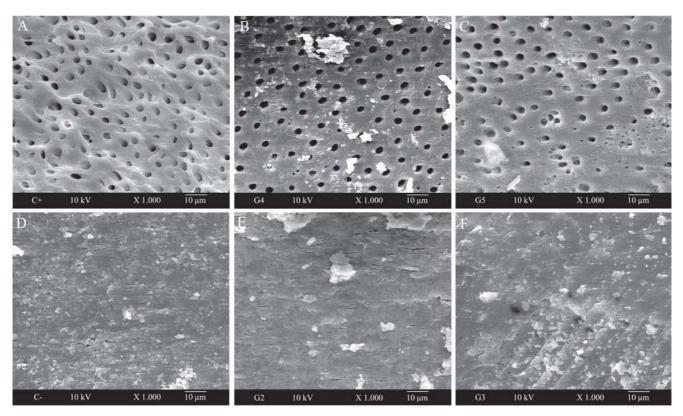


FIGURE 1- SEM micrographs of the groups. (A) Positive control group showing canal walls totally free of debris or smear layer, all tubules opened. (B-C) Typical thin smear layer and little amount of debris, with open dentinal tubules. (D) Negative control group showing canal walls covered by smear layer and debris with no open tubules. (E-F) Heavy smear layer and debris covering the dentinal walls, with a small number of open dentinal tubules

method to identify debris and smear layer, light microscopic evaluation could probably archive more consistent results in the identification of organic particles that represents minor part of them¹⁶.

Several models have been suggested for *ex vivo* investigation of the cleaning efficacy. However, most studies obtained micrographs from the middle root third^{2,28,30,31}. Although this study analyzed only the apical third, this fact was based on the literature, which states that worst cleaning results are observed at the apical third^{2,28,30,31}. This may be related to the decreasing diameter of the root canal and consequent decrease in flow of the irrigant, concomitant with the smallest diameter of tubules in that portion. In order to evaluate the efficacy for cleaning the canals and flushing away debris, this study used a score scale according to Ahlquist, et al.² (2001). This scale was used because it was clear and easy to learn, which was confirmed by the high Kappa values.

Several irrigants have been suggested for use during and after root canal preparation not only as antimicrobial agents, but also to increase the cutting efficiency of root canal instruments, remove smear layer and flush away the debris. NaOCl has been the irrigant of choice for endodontic treatment for several decades because of its excellent properties of organic tissue dissolution and antimicrobial activity^{11,31}. Chlorhexidine gluconate is a cationic bisguanide that causes leakage of intracellular components and, if used as a water-soluble gel, as Natrosol, it is able to aggregate dentin chips, decreasing the production of smear layer and debris, without the outcome of other gel-based solutions⁷. In addition, a previous study showed that this gel did not interfere with the capacity of sealers to fill artificially lateral canals mechanically prepared with it²⁶.

Debris is defined as dentin chips, pulp remnants or other particles loosely found on the root canal walls¹⁰. On the other hand, smear layer is defined as a film of debris intensely attached to the dentin and other surfaces following instrumentation with rotary drills or endodontic files, being composed of dentin particles, remnants of vital or necrotic pulp tissue, bacterial products and retained irrigants¹⁸. Therefore, in an infected root canal, the smear layer and debris should be removed to eliminate bacteria, facilitate the action of intracanal medicaments, and improve the sealing of obturation materials²⁴.

The results of the present study showed that when 2.5% NaOCl and 2% chlorhexidine gluconate gel were used without 17% EDTA, they did not efficiently remove the smear layer and debris, these findings corroborate those of Yamashita, et al.³¹ (2003), Menezes, et al.²⁰ (2003), Medici; Fröner¹⁹ (2006), who advocate the use of a mixture or combination of one of those solutions and chelating substances to reach improve action on the smear layer. These authors attribute to the chelating agent the capacity to remove the major component of the smear layer and debris, the inorganic components, favoring the action of others solutions. On the other hand, Ferraz, et al.⁷ (2001) reported better cleaning results using 2% chlorhexidine gluconate gel compared to 5.25% NaOCl or 2% chlorhexidine gluconate

in liquid form. In their study, Ferraz, et al.⁷ (2001) submitted the specimens to a previous ultrasonic bath, which might probably have opened the tubules that were maintained free of smear and debris during the preparation, whereas, in this present study, the specimens did not receive any type of previous treatment, which can explain the different outcomes.

In the present study, the root canal walls were free of smear layer and debris only when the irrigant was followed by 17% EDTA, independently of the type of irrigating solution employed. The root canal walls were almost always free of residues, and the dentinal tubules were visible. These results are in agreement with those of other authors, who have also reported that the removal inorganic residues is mainly dependent on the action of chelating agents^{24,28,30}. As the smear layer is mostly composed of 70% of inorganic particles²¹, the performance of the irrigating solutions can be explained by the inability of both products to remove this type of residues.

CONCLUSIONS

Under the tested conditions, it may be concluded that the use of 2.5% sodium hypochlorite or 2% chlorhexidine gluconate gel as irrigating solutions without association with a chelating agent was not effective in smear layer or debris removal. Thus, use of a chelating agent can contribute significantly for achievement of clean root canal walls with open tubules.

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