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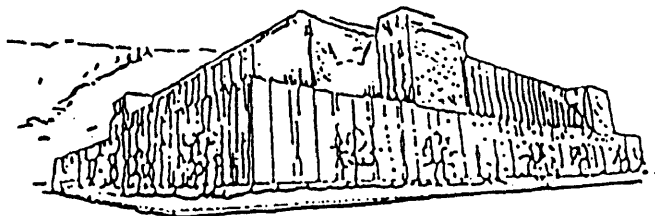
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**Impacts by acidic, metals-rich groundwater on the
hyporheic zone of an intermontane stream**

by

Sonia A. Nagorski

B.A., Amherst College, 1994

Presented in partial fulfillment of the requirements

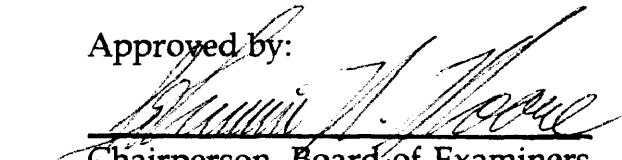
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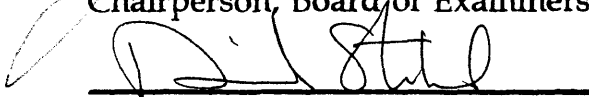
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Impacts by acidic, metals-rich groundwater on the hyporheic zone of an intermontane stream.

Director: Johnnie N. Moore



The hyporheic zone has been gaining ecological recognition as an important site of nutrient cycling and habitat, but very little is known about metal behavior within the zone. This understanding is important for assessing the environmental impacts of mining and for designing remediation strategies for riparian systems. Metal behavior in the hyporheic zone was studied at Silver Bow Creek, MT, where anoxic and acidic (pH=3-5) groundwater with high dissolved metal concentrations travels through a floodplain heavily contaminated with mining wastes and comes into contact with neutral (pH=7-8), oxic, and relatively low metal concentration surface water. A shallow hyporheic zone underlies the streambed where physical mixing and chemical transformation of these waters was found to occur.

Sampling the dissolved (<0.45 μm) metal and As concentrations in surface water, hyporheic zone water (<30 cm below and lateral to the streambed), and adjacent groundwater was conducted along a 1 Km stretch of the creek at three sites with variable surface water to groundwater flow direction relationships. Results of water analyses indicated that water in the shallow subsurface had a mean pH of 6, and mean concentrations of most metals were generally in between mean surface and groundwater concentrations. The highest levels of dissolved As at the site were found in the shallow hyporheic zone, indicating that the hyporheic zone is a distinct geochemical environment. Conservative elements (Ca and Mg) allowed for the calculation of physical mixing ratios, which indicated that 50% of the hyporheic zone samples contained >20% groundwater. All other metals were found to be acting non-conservatively in the hyporheic zone.

The solid phase was examined by setting into the streambed a series of slotted plastic columns filled with 2mm aluminosilicate beads which collected metal precipitates over the course of 52 days. Upon removal, dense bands of iron oxide precipitates were found on many of the bead columns at the surface water- substrate boundary. These precipitation zones, commonly only 1-5 cm thick, are interpreted to be products of metals in the groundwater coming out of solution upon mixing with higher pH and more oxic surface water. The thickness of the mixing zone appeared to be controlled by the the relationship of general groundwater and surface water flow directions, as well as by small scale variability in the permeability of streambed and floodplain sediments. The implication of these processes is that metals transported in solution by the groundwater precipitate onto the hyporheic zone and streambed sediments, thereby contaminating the hyporheic zone and contributing to the surface water metal load.

Acknowledgements

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Introduction

The hyporheic zone has been defined broadly in the literature as the saturated subsurface area connected to a stream channel which shares with it some biological, chemical, or physical characteristics [*Williams and Hynes, 1974; Triska et al., 1989; Valett et al., 1990; Hendricks and White, 1991; Valett et al., 1993*] (Figure 1). A more detailed and universal definition does not exist, because researchers have not used interdisciplinary criteria when proposing definitions of the hyporheic zone [see *White, 1993*].

Nonetheless, this loosely-defined zone has rapidly been gaining recognition as both a key ecological zone crucial to the health of stream biota, as well as a major site of exchange, metabolism, and storage of particulates and solutes [*Grimm and Fisher, 1984; Bencala, 1984; Stanford and Ward, 1988; Triska et al., 1989; Valett, 1993*]. Much progress has been made in the last couple of decades on characterizing many of the biological processes in this zone, yet relatively little has been made in the understanding of the physical and chemical dynamics of hyporheic zones where surface waters and adjacent groundwaters mix [*Bencala, 1993*.] In particular, there is a marked lack of understanding about contaminant storage and exchange through hyporheic zones.

Most studies on the physical and chemical dynamics of the hyporheic zone have concentrated on its connectivity with surface water, and they have generally concluded that the two interact extensively and that hyporheic zones play a major role in storage of stream solutes [*Bencala et al., 1984; Munn and Meyer, 1988; Triska et al., 1989; Valett et al., 1990; Castro and Hornberger, 1991*]. The exchange of nitrogen, oxygen, and organic material between surface water and hyporheic zones has been documented, and many authors contend that hyporheic zones are important sites for nutrient cycling [*Grimm and Fisher, 1984; Jacobs et al., 1988; Hendricks and White, 1991; Findlay et al., 1993*;

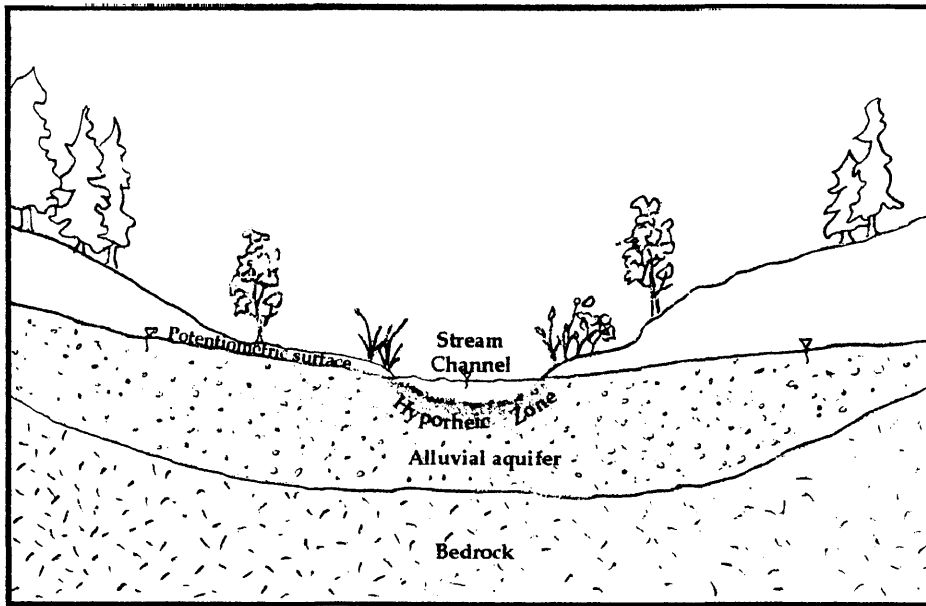


Figure 1: Schematic illustration of the location of the hyporheic zone

Holmes et al., 1994; *Findlay and Sobczak*, 1996; *Pusch*, 1996]. In addition, studies of biologic communities in the hyporheic zones have shown they are important habitats for many macroinvertebrates [*Williams and Hynes*, 1974; *Coleman and Hynes*, 1980; *Stanford and Ward*, 1988; *Williams*, 1989] and are sought by fish for spawning [*Wickett*, 1954; *Hansen*, 1975; *Johnson*, 1980]. A widely-accepted notion that has emerged from this research is that hyporheic animals, organic matter, and solutes are capable of extensive lateral and vertical movement and transport constrained by specific physical and chemical characteristics of each hyporheic environment.

Less is known about the interaction between groundwater and hyporheic zones. Some studies have documented the release of nitrogen and organic matter from groundwater into hyporheic and surface waters [*Coats et al.*, 1976; *Wallis et al.*, 1981; *Rutherford and Hynes*, 1987; *Valett et al.*, 1990; *Triska et al.*, 1993; *Wondzell and Swanson*, 1996]. Other studies have illustrated the volumetric importance of groundwater contributions to streamflow generation during storms [*Freeze*, 1972; *Sklash and Farvolden*, 1979; *Gillham*, 1984; *Blowes and Gillham*, 1988; *Novakowski and Gillham*, 1988; *Squillace*, 1996]. They found that groundwater contributes a much larger component of storm runoff than commonly acknowledged, suggesting that groundwater, surface water, and hyporheic zone interactions are dynamic and transient.

Many rivers and adjacent aquifers are contaminated with heavy metals from mining wastes, and the transport of these contaminants through hyporheic zones needs to be explored and understood. This understanding is crucial for accurately defining routes of contaminant transport targeted in riparian remediation designs. To date, there has been no published research on the geochemistry of metals and arsenic in groundwater entering the hyporheic zone and surface water from groundwater. Instead, most studies on metal contamination have concentrated on transport of metals either through groundwater, surface water, or stream sediments, and not on interrelating these components in the

hyporheic zone. Benner et al. [1995] discussed the behavior of metals in the hyporheic zone at one transect across Silver Bow Creek, MT, the site of this study (Figure 2). Yet their study focussed only on surface water infiltration into the hyporheic zone and was spatially and temporally limited.

The goal of this research was to determine how metals-contaminated groundwater affects the geochemistry of the hyporheic zone and surface water in an intermontane stream. In this paper, the hyporheic zone is defined as the shallow area surrounding the streambed (typically <30 cm below the streambed surface) where the surface water and groundwater physically interact to form a chemically defined transition zone. The examination of both the physical controls on mixing together with the resultant chemical reactions was conducted using geochemical analyses of dissolved and solid phases along a representative reach of the stream during variably sized flow events. This research proposes that a dynamic and spatially complex hyporheic zone exists where dissolved metals from adjacent groundwater are transferred into the solid phase, continually contributing to the streambed's metal load and contaminating the hyporheic zone upon which many aquatic organisms depend in healthy streams.

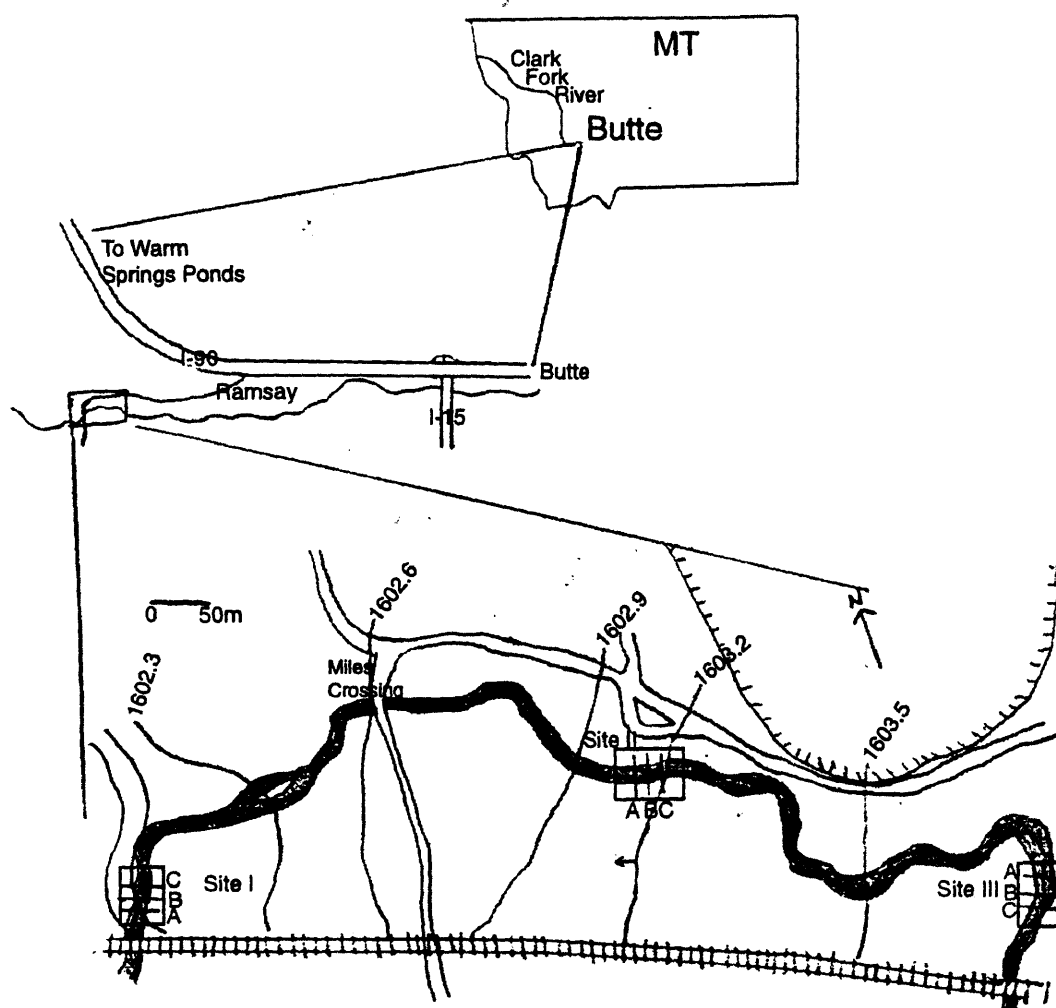


Figure 2: Location map of study area. Map of study area shows the three research sites, with the locations of transects A,B, and C at each site. Potentiometric lines on study area map are from Shay [1997].

Site description:

Silver Bow Creek, at the headwaters of the Clark Fork River in western Montana, was chosen to study metal transport through the hyporheic zone due to the large chemical differences between the acidic, anoxic, and metal-rich groundwater and the relatively dilute, oxic, and neutral-pH surface water [Benner *et al.*, 1995]. The abrupt boundaries between these waters in many portions of this system provide optimal conditions for detecting the small-scale processes occurring on either side of the surface water-groundwater interface. Also, the stream exhibits strong morphological variations, producing a variety of physical relationships between surface water and groundwater flow directions, which are thought to influence the extent of mixing. In addition, the large extent of metal contamination in Silver Bow Creek and its floodplain makes it a useful site for comparison to other systems heavily impacted by mining wastes.

The source of contamination at Silver Bow Creek is over a century of large scale mining in Butte, MT, located 18 km upstream from the study site. The mining resulted in the release of over 100 million metric tons of mining wastes into the creek [Andrews, 1987], much of which was carried downstream by major floods at the turn of the century and became deposited in wide stretches of the floodplain [CH2M Hill Inc., 1989; Nimick and Moore, 1991]. Along the study site portion of the creek, an up to 2 m thick and several hundred meter wide sequence of metal-rich mine tailings intermixed with sediments make up the top portion of the river's floodplain [Benner *et al.*, 1995; Lucy, 1996; Shay, 1997]. These tailings have highly elevated levels of arsenic, cadmium, copper, iron, lead, manganese, and zinc, which have been mobilized through time and have contaminated the underlying pre-mining floodplain sediments as well [Shay, 1997].

The groundwater is acidic and contaminated with dissolved metals due to oxidation of metal-sulfide minerals in the floodplain aquifer and vadose zone [Nordstrom, 1982; Moore and Luoma, 1990; Lucy, 1996]. Benner *et al.* [1995] found the groundwater to

have on average a pH of 4.2-4.9, an alkalinity of 0 meq/l, dissolved oxygen levels less than 1 mg/l, sulfate levels of 1500 mg/l, and metal concentrations of about 30 mg/l Al, 140 mg/l Ca, 20 mg/l, 0.55 mg/l Cd, Cu, 370 mg/l Fe, 35 mg/l Mg, and 54 mg/l Zn. Shay [1997] described in more detail the geochemistry of the aquifer and the relationship of groundwater contamination to the stratigraphy of the floodplain sediments. He found the groundwater chemistry to be extremely spatially heterogeneous, both laterally and with depth, and that zones of highly contaminated groundwater generally correlated with areas within the floodplain where contaminated sediments are continually or seasonally saturated by the groundwater or capillary fringe.

The water table has been found to lie between 0 to 1.5 m below land surface, depending on location and temporal variability, and is hydrologically connected with the creek [Shay, 1997]. Water level measurements taken across the site indicate that general groundwater movement is from east to west [Smart, 1995; Shay, 1997]. Although the surface water is significantly less contaminated than the groundwater, it is devoid of aquatic life with the exception of certain microorganisms [Wielinga *et al.*, 1994], algae, and an extremely depauperate aquatic insect population. During major precipitation events, it becomes particularly contaminated due to input from the floodplain [Lucy, 1996].

Methods

Three sites along the 1 km study area were selected following preliminary measurements of temperature, pH, and specific conductance used to approximate locations of groundwater inflow versus surface water infiltration [Sillman *et al.*, 1995; White *et al.*, 1987; Jones *et al.*, 1994] (Figure 2). Two of the sites were located where groundwater flowed approximately perpendicular to the surface water, thereby presumably forming a flow-through system like that characterized by Benner *et al.* [1995]. The third site was located in an area where the surface water and groundwater flow directions were approximately parallel. Each site was divided into three transects, each about 5 m apart. Detailed streambed topographic maps were constructed at each site. These maps were used to correlate the chemistry of the subsurface water with streambed topography and grain size.

Two shallow piezometers were installed several meters away from the creek banks. Water levels were measured in wells during every sampling session using an electric tape. Concurrent collection of floodplain piezometric data by Shay [1997] further aided in the making of potentiometric maps of the area. These surveyed instruments enabled the measurement and comparison of water levels on both sides of the stream relative to each other and to surface water stage.

Water sampling:

Each transect was instrumented with 5 subsurface water access tubes (1 cm OD polyethelene tubing), which were used for sampling the pore water below the streambed surface and in the adjacent banks and floodplain (Figure 3). Small diameter, flexible tubing was used instead of open PVC wells so that sample exposure to air was minimized. The lower 10 cm of the tubes were perforated and covered with a fine nylon screen. Two of the tubes were placed into the floodplain approximately 1 m away from the stream bank,

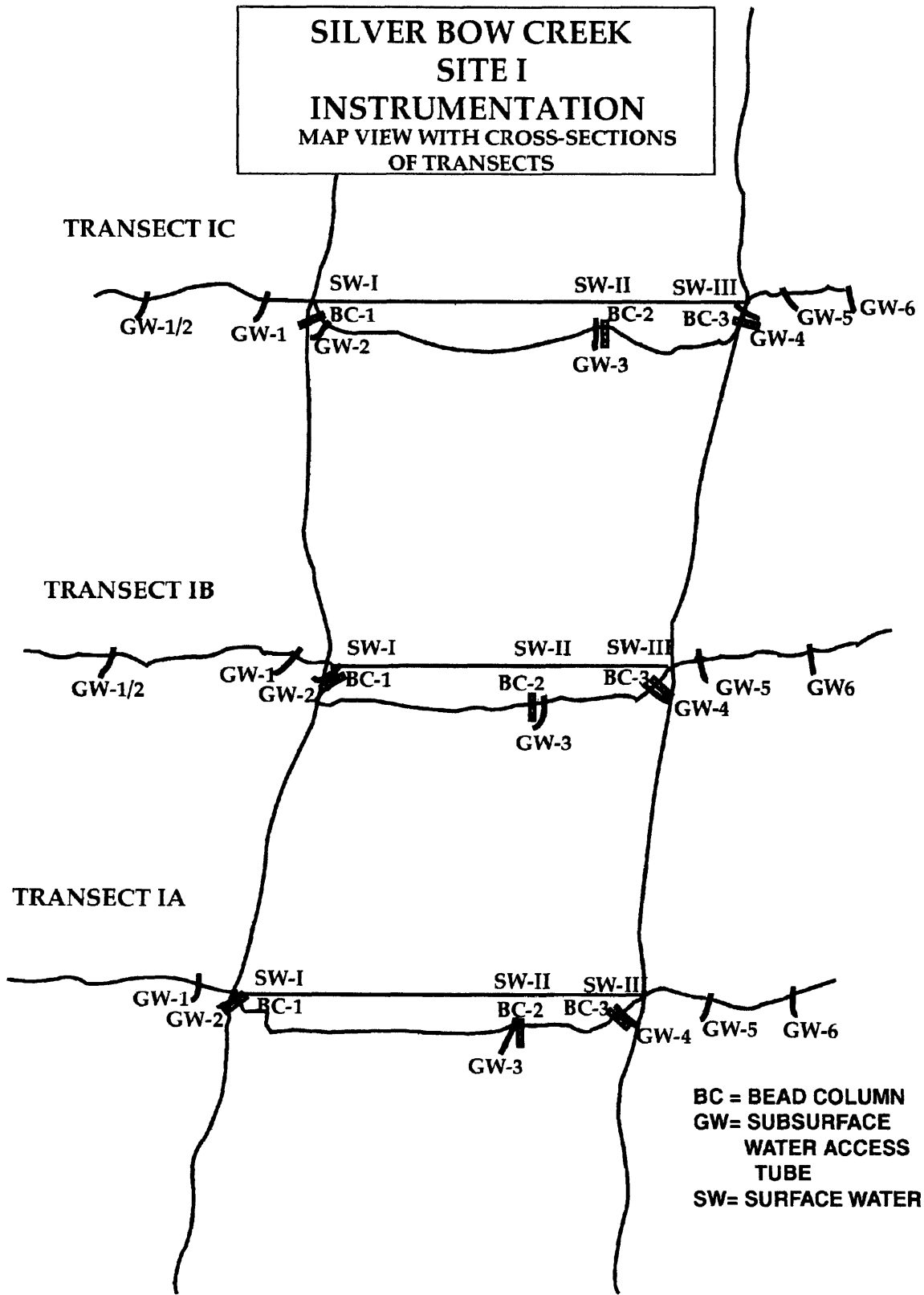


Figure 3: Instrumentation map

another two were submerged and set into the banks of the stream, and one tube was placed into the streambed at the center of the transects. Transects at Site I had two additional tubes, which were installed for purposes of sampling water away from the creek bed during high flow, when all five tubes became submerged. The tubes were installed by driving a steel rod into the sediment and upon its removal immediately inserting the plastic tubing. Wooden stakes attached to the tubes were inserted to stabilize the shallow tubes. The tubes penetrated between 15-25 cm into the streambed.

Each subsurface water sample was taken using an acid washed 60 cc syringe after purging at least one tube and one syringe volume. Small sample volumes were used in order to avoid integrating water from areas far from the tubes. The subsurface water access tubes were filled with water and tightly sealed after each sampling event. Three depth-integrated surface water samples were taken across each transect, one near each bank and one in the center.

Syringe collected samples were immediately pushed through a 0.45 μm filter set in an acid washed filter housing and into an acid washed 30 ml bottle. Approximately 10 ml of sample was used to purge the filter and bottle before taking the sample. Each cation sample was acidified with three drops of concentrated trace-metal grade HNO_3 immediately following collection. Anion samples were unacidified. All samples were immediately placed on ice in the field and kept chilled until analysis. Field blanks and duplicates were collected during every sampling session. In the laboratory, the acidified samples were analyzed using a Thermo Jarrel-Ash Inductively Coupled Argon Plasma Emission Spectrometer (ICAPES) for As, Ca, Cd, Co, Cu, Fe, Mg, Mn, Mo, Na, Ni, Pb, Si, Sr, Ti, and Zn. Unacidified samples were run on a Dionex Ion Chromatograph (IC) for chloride, nitrate, and sulfate analysis within 48 hours of sampling.

Dissolved oxygen, pH, and specific conductance were measured in the field on small sub-samples. Alkalinity was measured in the field by titrating HCl in the sub-sample

until the pH dropped to 4.5. Temperature was measured for all the sample sites at one point in time in order to minimize temperature fluctuations which occur during the course of the day of sampling. A thermometer probe was inserted about 10 cm into the sediment as close as possible to the subsurface water access tubes, and direct readings were taken.

Solid phase sampling:

Using artificial substrates to sample the solid phase by-passed the problems encountered when trying to core coarse-grained stream sediments and delineating coating history on the variably sized sediments [Benner *et al.*, 1995]. Twenty-seven bead columns were constructed using 40 cm long polycarbonate tubing (1 cm OD., 0.6 cm ID) which were slotted (1 mm width) with a band saw on two sides in 3 mm intervals.

Aluminosilicate beads (2 mm average diameter) were inserted into the columns, with plastic dividers separating every 10 cm of tube. The dividers were used to minimize vertical integration of the water flowing through the columns. The assembled columns were soaked in 20% reagent-grade HCl for 2 hours and rinsed repeatedly with deionized water until the ambient pH of deionized water was reached. In the field, each tube was carefully inserted into the substrate so that 30 cm of the column was below the channel bed surface and 10 cm was exposed in the surface water. Three bead columns were placed in each of the 9 transects, with each column placed as close as possible to the three subsurface water sampling tubes in each transect (Figure 3). The bead columns were identified by the transect in which they were placed (e.g. "IA" = Site I, Transect A) followed by 1, 2, or 3, depending on whether they were in the left bank, center, or right bank of the creek (looking upstream), respectively.

After 52 days, the bead columns were removed. With the removal of each column, the location of the water-substrate boundary was marked with a rubber band, and each column was quickly rinsed in the stream to wash off excess sediment and algal growths.

Each column was labeled, photographed, wrapped in plastic wrap, and stored. In the laboratory, the columns were oven-dried at 70°C and then carefully sectioned into 4-7 segments, depending on the amount of visible coating variation on the columns. The beads within the surface water portions of the columns were carefully chosen by using only those with no visible algal coatings. (Only two bead sections (Bead IIC-3, section 0-8.5 cm, and Bead IIIA-3, section 0-7.5 cm) were analyzed using beads with some algal coatings that could not be avoided). This was done because analysis of beads with algal coatings demonstrated that the algae themselves contained highly elevated metal concentrations (200-300% the concentration of beads from the same section without algal coatings), thereby obscuring concentrations of precipitates on the beads. Approximately one gram of bead from each section was measured out. The weighed beads were placed into an acid-washed centrifuge tube, to which 10 mL of 40% trace metal grade HCl was added. The columns were then shaken for 1 hour and centrifuged for 10 minutes. The solutions were analyzed for major metals using the ICAPES.

Sediment sampling

Three sediment samples were taken from along each bank of each of the three study sites. The samples were taken from the top 1-2 cm of stream sediment and wet sieved through a 63 μm mesh, using ambient stream water. The samples were stored on ice for transport to the laboratory, where they were immediately centrifuged and dried in a 70°C oven. The dried sediment then underwent a microwave aqua-regia digest, and metal concentrations were determined with use of the ICAPES.

Quality Assurance/ Quality Control

Accuracy and precision were measured separately for the water samples and for the bead digest samples. Field duplicates, lab duplicates, lab standards (internal and external), and blanks were used to find the % error (or variability) associated with each element in each water and bead sample.

Accuracy was measured by comparison to USGS standards T107 and T117, which were measured during every use of the ICAPES for cation analysis (Table 1). Almost all of the mean measured values were less than 10% different from the reported mean values. For those measured during water analysis, exceptions for USGS T107 were Cd (19.6%), Cu (12%), Mn (12.9%), and Si (12.7%) and for USGS T117 were Ca (12.7%), Mn (10.9%), Si (13.7%), and Zn (13.3%). For the standards measured during bead analysis, the accuracy was not as good, likely due to carry-over from the high concentrations of metals in the bead digests into the relatively dilute standards. All elements were no more than 15% different from the reported values. Exceptions for USGS T107 were Ca (17.1%), Cd (28.4%), Fe (22.4%), Mn (20.4%), Si (20.8%), and Zn (26%), and the exceptions for USGS T117 were Si (15.7%) and Zn (18.8%). Accuracy was less important on the bead digests than was precision, due to the use of the beads for establishing relative concentrations only.

USGS T107					
	Reported values mean (std.dev)	Meas. values (run with water samples) (n=29)	% Diff. from Reported	Meas. values (run with bead samples) (n=20)	% Diff. from Reported
Al	0.22 (0.045)	0.24 (0.022)	9.94	0.24 (0.020)	7.57
Ca	11.7 (0.7)	12.8 (0.4)	8.98	13.9 (0.7)	17.1
Cd	0.0143 (0.002)	0.0174 (0.001)	19.6	0.0190 (0.001)	28.4
Cu	0.030 (0.0023)	0.027 (0.012)	12.0	0.03 (0.002)	7.34
Fe	0.052 (0.007)	0.056 (0.015)	7.23	0.065 (0.0007)	22.4
Mg	2.1 (0.13)	2.2 (0.091)	4.65	2.36 (0.10)	11.5
Mn	0.045 (0.006)	0.051 (0.002)	12.9	0.055 (0.002)	20.4
Mo	0.015 (0.002)	0.016 (0.002)	7.69	0.017 (0.0024)	14
Na	20.7 (1.1)	22.8 (0.44)	9.66	23.6 (1.1)	12.9
Si	3.6 (2.3)	4.1 (0.14)	12.70	4.4 (0.22)	20.8
Sr	0.061 (0.004)	0.063 (0.002)	3.23	0.067 (0.003)	22.9
Zn	0.076 (0.01)	0.08 (0.03)	5.39	0.098 (0.006)	26

USGS T117					
	Reported values mean (std.dev)	Meas. values (run with water samples) (n=12)	% Diff. from Reported	Meas. values (run with bead samples) (n=11)	% Diff. from Reported
Al	0.079 (0.019)	0.084 (0.022)	5.84	0.090 (0.017)	13.3
Ca	20.9 (1.2)	23.7 (0.4)	12.7	24.2 (0.9)	14.6
Fe	0.47 (0.018)	0.51 (0.015)	7.87	0.52 (0.019)	9.78
Mg	10.05 (0.44)	10.99 (0.091)	8.92	11.03 (0.32)	9.27
Mn	0.22 (0.003)	0.25 (0.002)	10.9	0.25 (0.009)	12.8
Mo	0.012 (0.002)	0.013 (0.002)	11.1	0.013 (0.002)	8.70
Na	20.0 (1.26)	22.0 (0.436)	9.67	22.3 (0.89)	10.9
Si	5.54 (0.3)	6.35 (0.1)	13.7	6.48 (0.22)	15.7
Sr	0.265 (0.011)	0.277 (0.002)	4.42	0.277 (0.007)	4.48
Zn	0.176 (0.009)	0.201 (0.025)	13.3	0.212 (0.010)	18.8

Table 1: Results of analyses of Standards USGS T107 and T117 water standards run with water and bead digest samples.

Eighteen field blanks were collected during the water sampling, and all concentrations were below detection, with the exception of Zn, whose average concentration was 0.018 mg/l. Twenty three lab blanks were also run during the course of the water analysis, and again all concentrations were below detection. Lab blanks run during bead digest analyses showed all elements to fall below the detection limit. However, bead blanks, which were digests run on the aluminosilicate beads not installed in the creek bed, showed significant levels of Al and Si, likely resulting from the composition

of the beads themselves. For these reasons, Al and Si values had to be discarded for the bead concentration results. Bead blanks also resulted in mean concentrations of Ca with $5.6 \mu\text{g/g}$ bead; Mg with $1.3 \mu\text{g/g}$ bead; Na with $0.96 \mu\text{g/g}$ bead; and Ti with $0.09 \mu\text{g/g}$ bead. The highest levels found in the bead blanks during each analysis were used to establish the boundary of minimally significant bead concentration values in the samples.

Precision was measured through analysis of both lab and field duplicates. Field precision of the water samples was measured by analysis of eleven duplicate samples collected in the field (about one from each sampling event). Precision of the bead samples was measured by taking 17 duplicate samples of beads from randomly selected sections of the tubes. Lab precision was measured by running the same samples (whether water or bead digest) at least once and calculating the variability in the readings by the ICAPES and IC, whichever pertained. Twenty-two lab duplicates were run during the water sample analysis, and twenty were run during the bead digest analysis.

Each sample was compared with its field and/or lab duplicate and the percent difference between the duplicates was calculated for each element. For purposes of calculations, a value of one-half the detection limit was applied to each metal whose concentration fell below the detection limit. The 95% confidence interval of the mean of all the compiled percent difference values was calculated (Table 2). The variability in the lab duplicates was significantly smaller than that of the field duplicates (for the water samples) and bead coating variability (for the beads); therefore, the latter were used for construction of error/variability bars. Thus, the concentration of each element in each sample had an error bar applied to it which represents the 95% confidence interval of the mean field variability for the specific element. The appendix contains all of the field and lab duplicate samples with the individual percent changes as well as the compilation of percent changes from which the 95% confidence intervals were calculated.

Parameter	Water samples	Bead samples
pH	1.3 %	NA
Diss. oxygen	23 %	NA
Spec. Cond.	3.4 %	NA
Alkalinity	19 %	NA
Cl	4.2 %	NA
NO ₃ -N	22 %	NA
SO ₄ ²⁻	9.7 %	NA
Al	28 %	37 %
As	22 %	18 %
Ca	1.3 %	15 %
Cd	21 %	NA
Cu	21 %	14 %
Fe	8.4 % ^(a) 31% ^(b)	11 %
Mg	2.5 %	24 %
Mn	2.3 %	10 %
Mo	8.3 %	13 %
Na	1.6 %	18 %
Pb	39 %	9.2 %
Si	1.2 %	18 %
Sr	1.7 %	13 %
Ti	NA	20 %
Zn	16 %	12 %

(a): For concentrations > 20 mg/l

(b): For concentrations <1 mg/l

Table 2: Summary of 95 % confidence intervals applied to water and bead samples for construction of error/variability bars. the 95% confidence intervals for water reflect field variability, and those for the beads reflect the coating concentration variability.

Results and Discussion

Physical setting

According to the potentiometric map of the study area, surface water and groundwater had variable flow direction relationships (Figure 2). At Site I, surface water flows approximately perpendicular to the groundwater. There, head measured on the east bank was an average of 9.5 +/- 1.6 cm higher than the head measured in the west bank during all sampling periods. Water levels in the stream fell inbetween the measurements taken on either bank. Furthermore, temperature, assumed to be conservative, indicates that water in the west bank (11.7°C on 6/20/95) was of intermediate composition between east bank water (10.5°C) and surface water (14.8°C). This suggests that at Site I, the stream functioned as a flow-through system, with groundwater moving into it from the east bank and surface water recharging the west bank.

At Site II, the potentiometric maps indicate that groundwater and surface water flow approximately parallel to each other. Water level differences in the north and south banks and in the surface water were small enough to fall within the measurement and surveying error of 3.0 cm. Temperature measurements of the shallow subsurface produced a highly variable pattern of temperature zones which appeared to be controlled by grain size. Generally, there was no spatial regularity to the patterns.

Although the flow relationships at Site III appear to be analogous to those at Site I according to the potentiometric maps, the only discernable head difference was that the east bank generally showed slightly higher (3.9 +/- 0.30 cm) water levels than in the surface water. The difference between the surface water level and head in the west bank was within the measurement and surveying error. Thus, the creek was found to gain water from the east bank, while its potentiometric differences with the west bank are small

enough that flow directions could not be determined. Temperature measurements from across the site show that the subsurface water, both in the banks and below the creek bed were commonly 2-4°C colder than the surface water.

Streambed sediments:

All sites were generally characterized as having sand and gravel stream bottoms with more fine grained sediments (fine sands and silts) forming the banks. At Site I, the streambed ranged in size from coarse sands to medium gravels. The sediments along its banks were mostly fine to medium silts and sands. At Site II, an unstable gravel bar covered most of the stream bottom during the course of the sampling period. Along the banks, the sediments were mostly silty, with only some fine to coarse grained sands. At Site III, the streambed sediments ranged in size from sands to large cobbles, although the sediments along the banks were silts so fine grained that it was usually difficult and sometimes impossible to extract water samples from them.

Flow rate in the creek:

Water sampling took place over the course of four months, during which the flow rate in the creek varied from about 800 to 6100 liters per second (l/s) (USGS, 1996) (Figure 4). Water was sampled from all the transects in two general episodes of flow rate in the creek. One sample set was collected in late June to mid July, when the flow in the creek ranged from 2800 L/s to 6100 l/s. The second set was taken in late August to early September, when the flow rate was considerably lower, ranging from 800 to 1200 l/s. During the lower flow period the bead columns were installed in the creek bed.

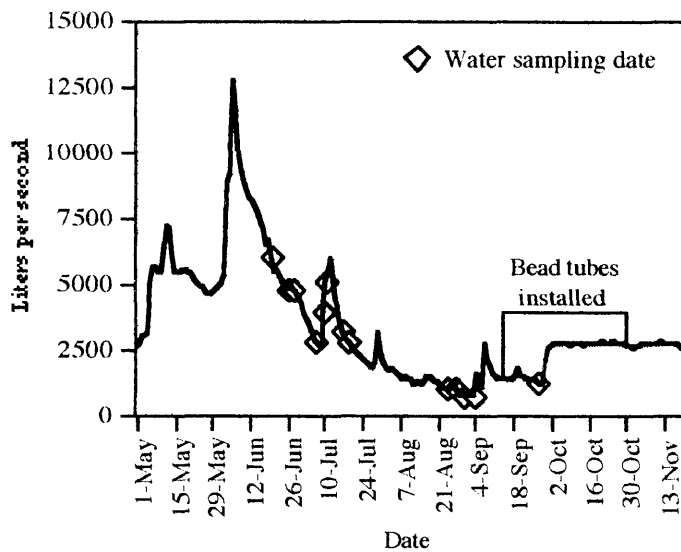


Figure 4: Flow volume rate (l/s) in Silver Bow Creek at the time of water sampling and of bead tube installation in the creek bed. Data from USGS [1996].

The head in the groundwater fluctuated about 0.8 m in the aquifer over the course of the sampling period (Shay, 1997).

Hyporheic Zone Chemistry: Dissolved (<0.45 μm) fraction:

Water highly elevated in metal concentrations compared to the surface water was collected in the shallow hyporheic zone at depths less than 30 cm beneath the creek bed during both low and high flow episodes at all the sites. The chemistry of the hyporheic zone was highly variable both spatially and temporally. Every water sample had a unique chemical composition. Table 3 shows an example of hyporheic zone water chemistry found in two samples from each site. (The appendix shows the entire dataset.)

The pH of water collected from the hyporheic zone commonly measured about 6, which is an intermediate value between typical groundwater ($\approx 3 - 5$) and surface water ($\approx 7 - 8$) at the site. Dissolved As in the hyporheic zone samples were found above the detection limit of 0.07 mg/l in 65% of the samples, with an average of 0.28 mg/l. None

Sample	Site I	IA-GW2 8/24/95	Site II	IIA-GW2 8/31/95	Site III	IIIA-GW2 8/28/95
	IB-GW1 6/26/95		IIA-GW5 7/6/95		IIIB-GW4 7/11/95	
pH	6.8	6.7	6.8	7.0	6.3	6.3
d.O ₂	0.5	1.5	1.3	0.6	1.7	1.7
Cond.	1.20	0.851	2.27	0.755	1.15	1.58
Alkal.	7.6E+02	3.7E+02	1.7E+03	3.8E+02	8.4E+01	9.6E+01
Cl	30	15	26	18	27	38
NO ₃ -N	BDL	0.13	BDL	BDL	BDL	BDL
SO ₄ ²⁻	15.3	36.4	188	17.8	7.5E+02	6.9E+02
Al	0.39	BDL	0.64	BDL	0.14	0.09
As	1.6	0.11	2.6	0.21	0.11	BDL
Ca	110	88	330	77	150	166
Cd	BDL	BDL	BDL	BDL	BDL	BDL
Cu	0.175	BDL	0.954	0.015	BDL	BDL
Fe	71	51	2.1E+02	22	73	60
Mg	21	16	54	11	36	38
Mn	20.0	8.61	45.5	7.79	12.9	10.0
Na	31	32	65	27	78	103
Ni	BDL	BDL	0.023	BDL	BDL	BDL
Pb	BDL	BDL	0.14	BDL	BDL	BDL
Si	21	17	17	20	23	20
Sr	0.618	0.673	1.12	0.450	1.34	1.63
Ti	0.016	BDL	0.023	BDL	BDL	BDL
Zn	0.384	0.042	2.04	0.166	6.4	12.7

Table 3: The chemical characteristics of six selected hyporheic zone samples, two from each site. Concentrations are in mg/l. Error/variability bars are listed in Table 2.

of the 24 groundwater samples collected by Shay [1997] at the same study site had any measurable As, and it was below detection in the surface water. Aluminum, Cd, Co, Cu, Ni, Pb, and Ti were below detection in some samples and above in others. The rest of the measured metals were always at detectable levels, although the concentration ranges for most of them were variable within one to three orders of magnitude. Iron and Zn concentrations had the highest variability, with sample concentrations ranging over three orders of magnitude; dissolved Fe concentrations in the hyporheic zone varied from 0.15 to 350 mg/l, and dissolved Zn concentrations ranged from 0.042 mg/l to 26 mg/l. Dissolved Ca, Cl, Mg, Mn, Na, and Si concentrations had much smaller ranges of variability in the hyporheic zone, generally within one order of magnitude.

Despite the large chemical variation, distinct patterns were found in the distribution of dissolved chemical constituents in the hyporheic zone samples of each of the three sites.

Site I:

At Site I, the water samples taken from the upgradient east bank, interpreted from the piezometric data and from research by Benner et al. [1995] to be the local groundwater, typically contained higher concentrations of metals and SO_4^{2-} and had lower pH levels than did the samples taken from the downgradient west bank, interpreted to be the hyporheic zone. The pH levels averaged 4.3 (standard deviation = 1.2) in the east bank, in contrast with samples from the west bank, where pH levels averaged 6.5 (standard deviation = 0.4). In addition, mean concentrations of SO_4^{2-} , Ca, Mg, Cu, Fe, Mn, and Zn were lower in the west bank than in the east bank, although the standard deviation for these concentrations overlapped due to the temporal and small scale spatial variability.

The general trends observed at the Site is exemplified in the Figure 5, which depicts the results of sampling transect B at Site I on one particular day (6/20/95). The concentration patterns of Fe and Mg show highest concentrations in the east bank, lowest were below the creek bed, and intermediate concentrations were in the west bank. The concentration patterns of Ca, Na, and specific conductance were the same as those of Fe and Mg. Sulfate, Al, Cd, and Zn had the same spatial trends as did Cu (shown in the figure), with high concentrations in the east bank samples, and concentrations close to or below detection in all other subsurface water samples. Two of the three sampling sites in the center of the creek bed usually contained water which chemically resembled the surface water. The third of these, sample site GW-3 at Transect A, was found to contain metal concentrations similar to those found in the east bank all four times it was sampled. (The appendix contains the results of the chemical profiles of the water samples taken along each transect during all the sampling events at Site I.)

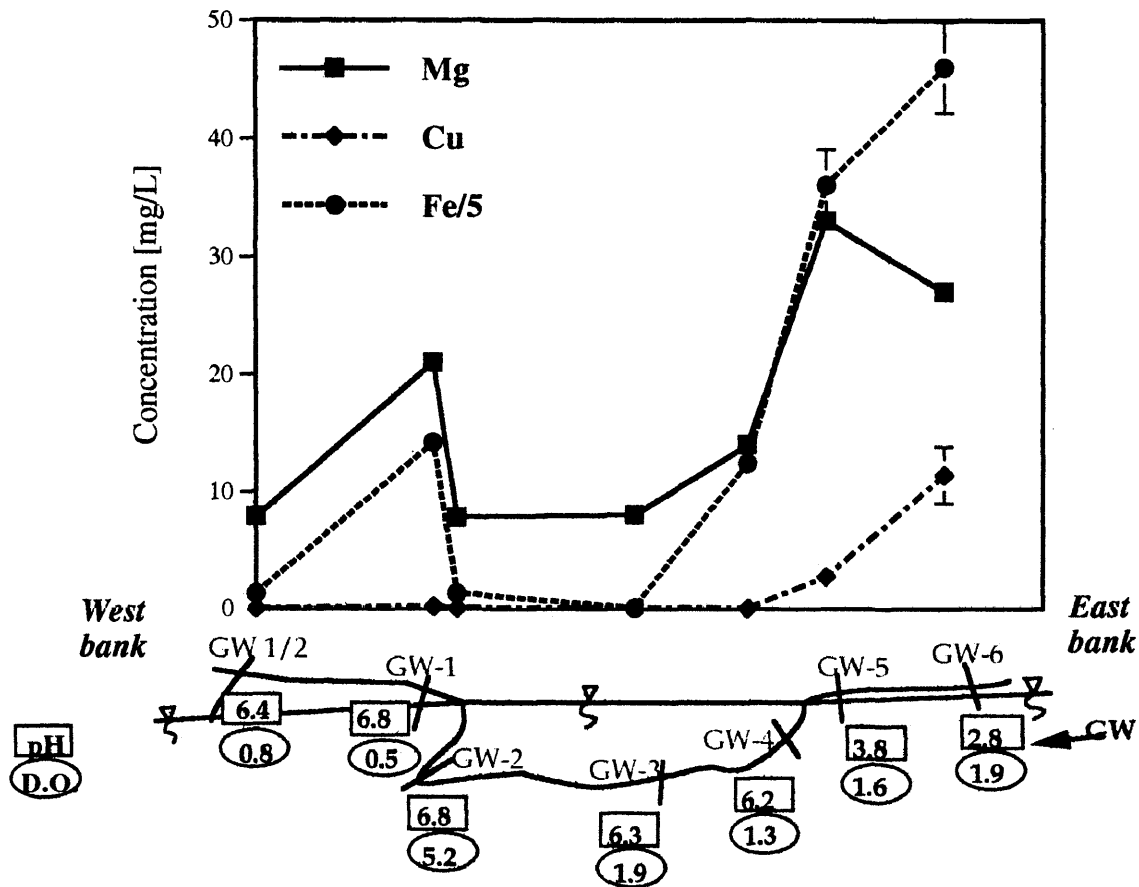


Figure 5: Example of spatial trends of metal concentrations and pH and dissolved oxygen (D.O.) levels found across one transect at Site I. Points on the graph correlate with locations along the transect, shown below the graph.

Site II:

The chemical differences across the transects at Site II were not nearly as clear as those at Site I. Figure 6 exemplifies the general spatial trends found across the site. The trends did not show a relationship of metal concentration and acidity favoring either bank. Although average concentrations of Ca, Fe, Mg, Mn, and Na, and specific conductance were commonly slightly higher on the south bank, average Cu and Zn tended to be slightly higher on the north bank. Chloride, SO_4^{2-} , and Si concentrations varied as well, but their concentrations did not were not different in either stream bank. Samples taken from beneath the gravelly center of the streambed (wells IIA-GW3, IIB-GW3, and IIC-GW3) very closely resembled the chemistry of the surface water. In addition, the mean metal concentrations at the Site were low compared to those found in water samples from Sites I and III. It thus appeared that most of the samples were more heavily dominated by the surface water than at Site I. (The appendix illustrates the results of the chemical profiles of the water collected along transects at Site II.)

Site III:

At Site III, yet another spatial profile was identifiable across the transects. The east and west banks of the creek exhibited almost identical subsurface water chemical compositions (Table 4). The samples were characterized by almost neutral pH and relatively high concentrations of dissolved metals.

The chemistry of the water collected from the subsurface of the central portion of the creek had higher dissolved O_2 and NO_3^- -N, and lower specific conductance, alkalinity, and concentrations of all the constituents measured than did the water collected from within the banks. Still, most of these concentrations were significantly higher than in the surface water.

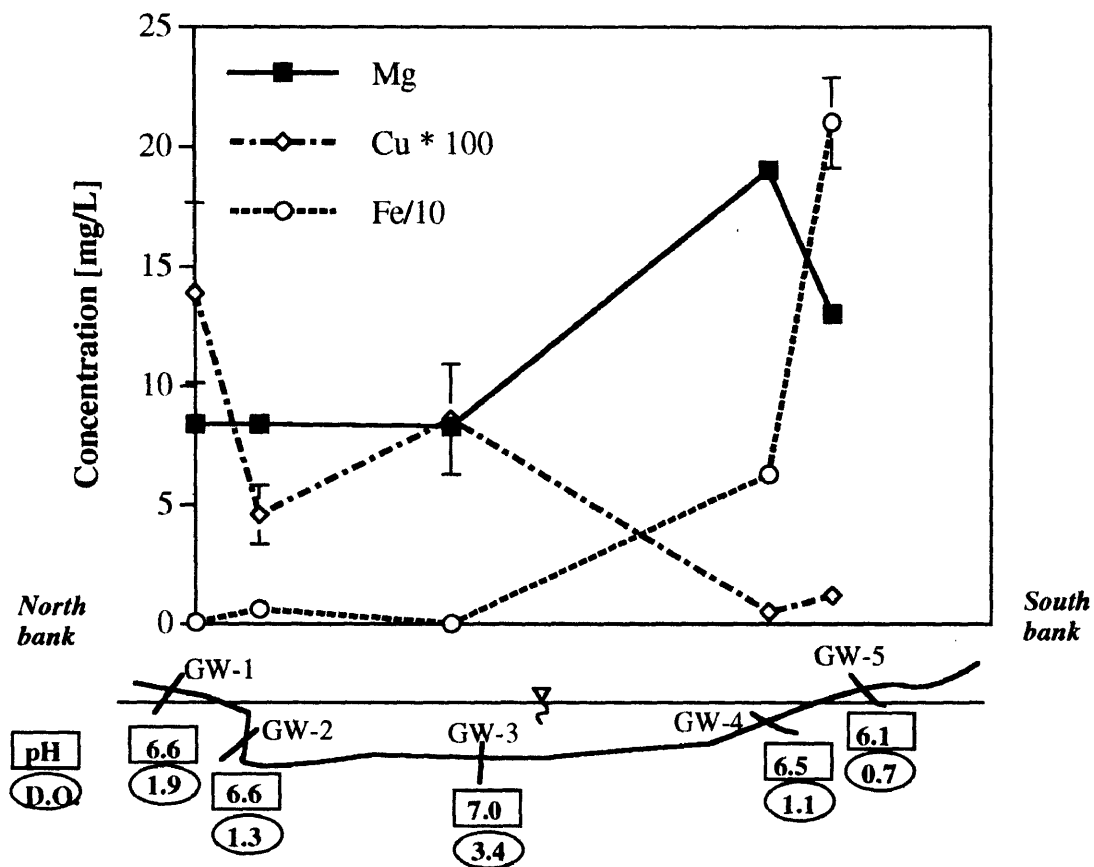


Figure 6: Example of spatial trends of metal concentrations and pH and dissolved oxygen (D.O.) levels found across one transect at Site II. Points on the graph correlate with locations along the transect shown below the graph.

Site III	West Bank Mean (Std. dev.) (n=9)	Below creek center Mean (Std.dev.) (n=5)	East Bank Mean (Std. dev.) (n=10)
pH	6.6 (0.24)	6.6 (0.11)	6.6 (0.37)
D.O.	1.4 (0.9)	3.5 (0.9)	1.8 (0.7)
Cond.	1.27 (0.47)	0.650 (0.23)	1.80 (0.30)
Alkal.	458 (371)	122 (45)	306 (210)
Cl	25.3 (6.3)	15.8 (4.9)	58.1 (13)
NO ₃ ⁻ -N	(<0.15)	0.75 (0.38)	(<0.15)
SO ₄ ²⁻	364 (291)	236 (127)	553 (340)
Al	0.12 (0.10)	0.07 (0.04)	0.11 (0.05)
As	0.25 (0.12)	(<0.07)	0.22 (0.13)
Ca	133 (39)	63 (24)	172 (46)
Cu	0.03 (0.04)	0.08 (0.07)	0.03 (0.02)
Fe	155 (139)	10.3 (6.9)	57.8 (40)
Mg	33.2 (12)	14.4 (5.8)	41.7 (12)
Mn	19.0 (8.9)	3.38 (2.2)	9.30 (4.8)
Mo	0.23 (0.20)	0.019 (0.01)	0.10 (0.05)
Na	72.7 (23.5)	33.2 (13.6)	141 (41.2)
Si	21.9 (2.8)	14.4 (0.55)	20.2 (4.1)
Sr	0.929 (0.42)	0.473 (0.22)	1.51 (0.45)
Zn	1.85 (2.6)	4.73 (5.1)	3.16 (5.7)

Table 4: Mean and standard deviations of concentrations of metals and other measured constituents at Site III. Mean Cd, Ni, Pb, and Ti are not included because their means were below detection.

Figure 7, which depicts the concentrations of some metals found at Transect IIIB on 7/17/95, exemplifies these trends found at Site III. The profiles of Mg, Fe, and Cu show the pattern of increasing concentrations of metals with distance away from the creek center, both in the east and west banks.

Low sulfate samples:

Some samples from the hyporheic zone were found to contain SO₄²⁻ concentrations that were significantly lower than in the groundwater and in the surface water (Table 5). These samples were usually depleted or almost depleted in NO₃⁻-N and dissolved O₂, and they contained relatively high alkalinity concentrations (Figure 8). In addition, they came from locations where the surface water is thought to be infiltrating the hyporheic zone based on potentiometric data, such as at Site II and in many of the west bank (downgradient) samples at Site I. Only one sample (IIC-GW4 (8/31/95)) from

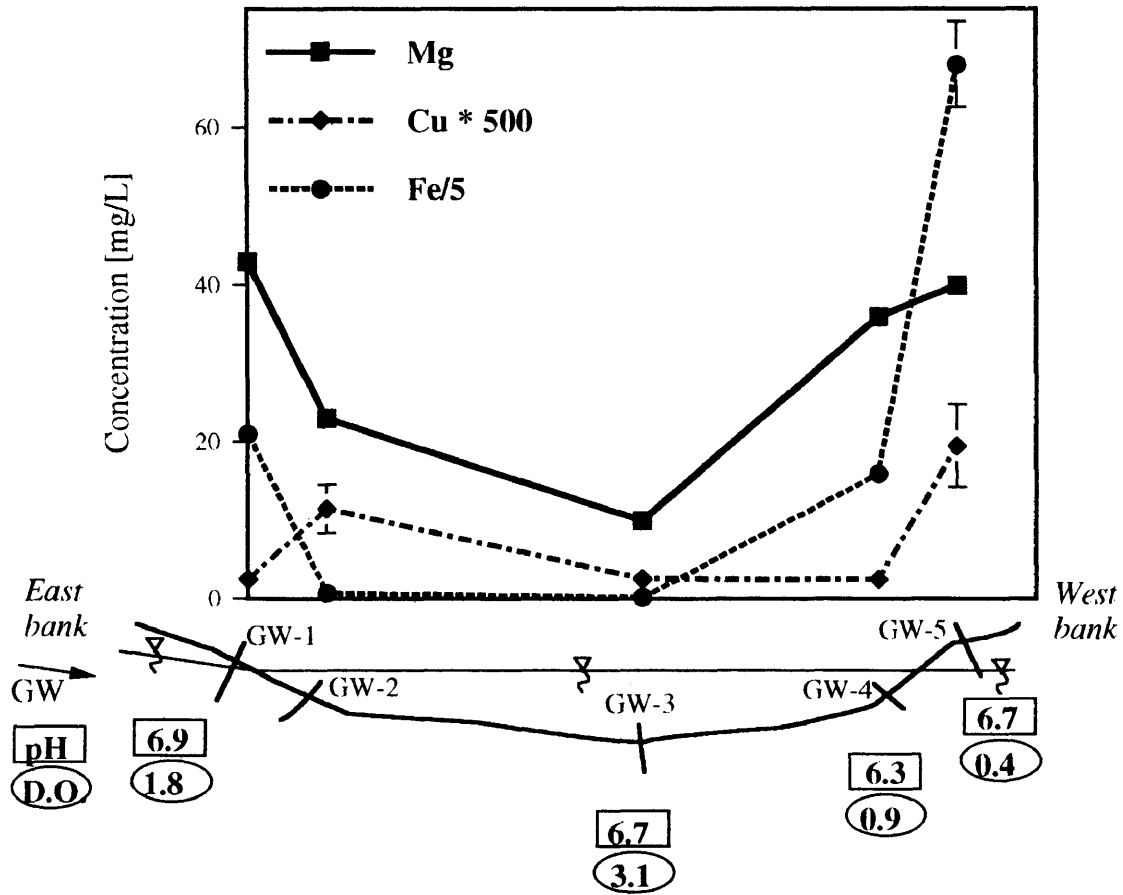


Figure 7: Example of spatial trends of metal concentrations and pH and dissolved oxygen (D.O.) levels found across one transect at Site III. Points on the graph correlate with locations along the transect, shown below the graph.

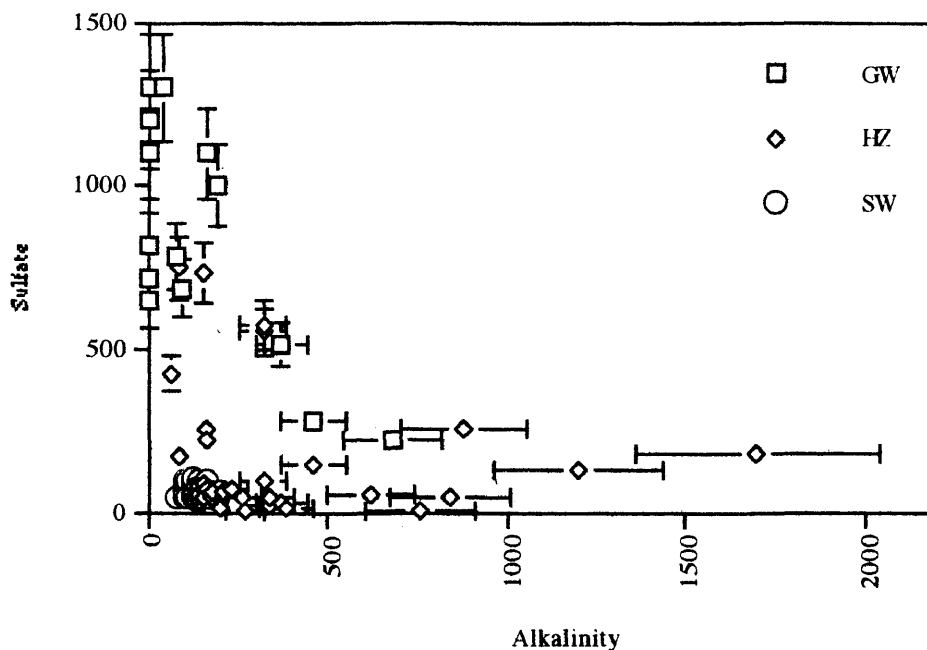


Figure 8: Sulfate vs. Alkalinity concentrations in the three water groups.

Site III (where piezometric data indicate the stream to be gaining from at least one side) had a sulfate concentration lower than that in the surface water.

Hyporheic zone (H.Z.) Sample	H.Z. Diss. O ₂	H.Z. NO ₃ -N	H.Z. SO ₄ ²⁻	H.Z. Alk.	Surface water Diss. O ₂	Surface water NO ₃ -N	Surface water SO ₄ ²⁻	Surface water Alk.
<i>Site I</i>								
IB-GW1 (6/26/95)	0.5	BDL	15.3	7.6E+02	6	0.9-1.0	54-55	1.0E+02
IA-GW2 (7/19/95)	1.2	BDL	28.9	5.7E+02	6	1.1	73	1.7E+02
IA-GW2 (8/24/95)	1.5	BDL	36.4	3.7E+02	7-8	1.6-1.7	90-96	1.3E+02
IA-GW2 (9/27/95)	1.6	0.14	51.4	3.4E+02	7	1.9	108	1.3E+02
<i>Site II</i>								
IIA-GW4 (7/6/95)	1.3	BDL	25.8	3.4E+02	6	0.7-0.9	60-63	1.3E+02
IIB-GW4 (7/6/95)	1.1	BDL	9.25	2.7E+02	6-7	1.2	62-67	1.4E+02
IIA-GW2 (8/31/95)	0.6	BDL	17.8	3.8E+02	8	1.6-1.7	102-104	1.2E+02
IIA-GW4 (8/31/95)	0.9	BDL	55.9	2.6E+02	8	1.6-1.7	102-104	1.2E+02
<i>Site III</i>								
IIIC-GW4(8/31/95)	2.1	0.73	79.1	2.3E+02	8-9	1.5-1.6	97-102	1.2E+02

Table 5: Dissolved oxygen, nitrate-N, sulfate, and alkalinity concentrations in some hyporheic zone samples with most concentrations lower (and alkalinity higher) than surface water and groundwater samples. The surface water chemistry for each given sampling date is listed alongside the hyporheic zone samples for purposes of comparison.

Surface water: physical and chemical results

The surface water chemistry was far less spatially and temporally variable than that of the hyporheic zone water or the groundwater. Generally, it was near neutral in pH, had moderate to high dissolved oxygen levels, and was relatively low in dissolved metal concentrations (Table 6).

	Mean (Std. dev.)	High (6100 L/s)	Low (800 L/s)
pH	7.7 (0.35)	8.2	6.6
D.O.	7.2 (1.3)	11	3.9
Cond.	0.424 (0.09)	0.545	0.304
Alk.	1.3E+02 (25)	2.0E+02	80
Cl	14 (4.2)	28	7.4
NO ₃ ⁻ -N	1.30 (0.53)	1.94	0.304
SO ₄ ²⁻	78.4 (23.7)	109	46.8
Al	(<0.07)	0.34	(<0.07)
As	(<0.07)	(<0.07)	(<0.07)
Ca	47 (8.8)	58	35
Cd	(<0.01)	(<0.01)	(<0.01)
Cu	0.136 (0.04)	0.262	0.06
Fe	0.22 (0.11)	0.46	0.05
Mg	10.4 (2.17)	13.2	7.54
Mn	0.90 (0.29)	1.33	0.44
Mo	(<0.01)	(<0.01)	(<0.01)
Na	23 (4.7)	29	16
Ni	(<0.02)	(<0.02)	(<0.02)
Pb	(<0.1)	(<0.1)	(<0.1)
Si	13 (0.77)	16	12
Sr	0.265 (0.05)	0.330	0.198
Ti	0.003 (0.002)	0.010	0.003
Zn	0.645 (0.255)	1.31	0.323

Table 6: Summary of surface water chemistry (concentrations in mg/l). The mean data are from all surface water samples collected, thus including those collected over a range of flow rates in the creek, spanning from 800 to 6100 L/s.

Surface water chemistry varied at all sites with flow volume. Generally, concentrations of cations and anions decreased with increasing flow rate. Such relationships were particularly clear for Ca, Mg, Mn, Na, Cl, NO₃⁻-N, SO₄²⁻ and specific conductance (Figure 9). Relationships of Cu, Zn, Fe, Si, and pH with flow rate were not as clear (Figure 10).

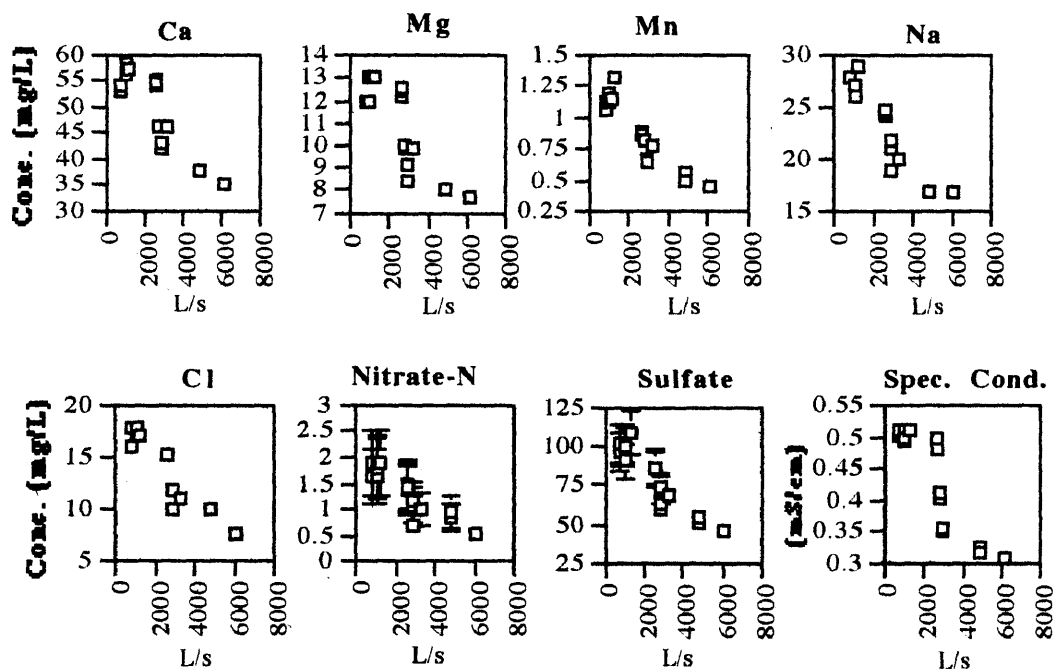


Figure 9: Concentrations of Ca, Mg, Mn, Na, Cl, NO_3^- , SO_4^{2-} , and specific conductance vs. flow rate (L/s) in Silver Bow Creek.

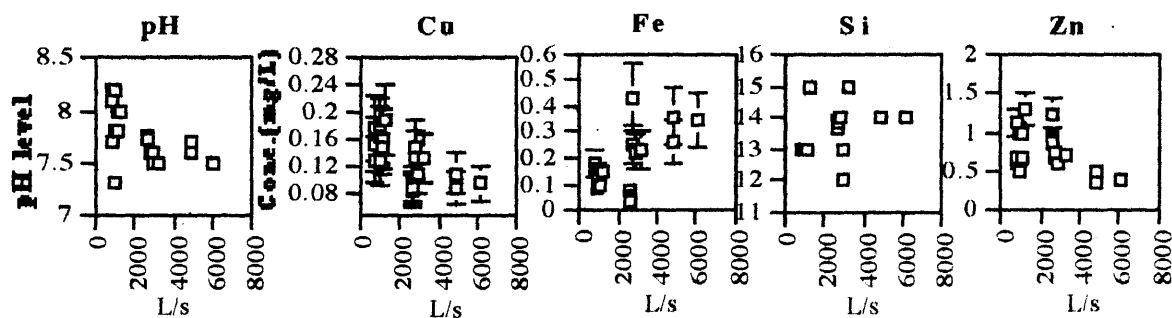


Figure 10: Concentrations of pH, Cu, Fe, Si, and Zn vs. flow rate (L/s) in Silver Bow Creek.

Although the flow rate varied over a 7.5-fold difference over the course of the study period (from 800 to 6100 L/s), the concentrations of the elements did not vary accordingly (Table 7).

Parameter	Conc. at 800 L/s	Conc. at 6100 L/s	Conc. at 800 L/s/ Conc. at 6100 L/s
Mn	1.1 mg/l	0.45 mg/l	2.4
Cl	17 mg/l	7.5 mg/l	2.3
SO ₄ ²⁻	100 mg/l	45 mg/l	2.2
Ca	53 mg/l	35 mg/l	1.5
Mg	13 mg/l	7.6 mg/l	1.6
Na	28 mg/l	17 mg/l	1.6
spec. cond.	0.50	0.31	1.6
Zn	0.5-1.1 mg/l	0.38 mg/l	1.3 - 2.9
Cu	0.12- 0.18 mg/l	0.10 mg/l	1.2 - 1.8

Table 7: Comparison of concentrations of selected parameters during low and high flows in the surface water

Solid phases

The bead columns provided a finer resolution scale of the chemical characteristics of the hyporheic zone. When the three bead columns in each of the nine transects were removed from the streambed, each exhibited two to three distinct color zones. The top section (about 10 cm in the length) that projected out into the surface water was green due to algal coatings on all of the bead columns. Most subsurface portions (the remaining approximately 30 cm in length) of the columns remained white in color. Yet 22 of the 27 bead columns exhibited some amount of precipitation by reddish iron oxides along their lengths, usually right at the surface water - substrate boundary. On some bead columns, this coloration was a thick, dark red, and other columns had much thinner and lighter red coatings. Bead Column IIIA-2 is an example of a bead column that exhibits the three main color zonations (Figure 11).

The red precipitation zones generally spanned no more than about 5 cm of the bead column lengths, although considerable variability existed. Some bead columns exhibited much thicker zones of precipitation (up to 30 cm, in the case of IA-2 (Figure 12) and others showed two separate zones of precipitation (IC-1, IIC-3 (Figure 11), IIB-1, and IIIC-2).

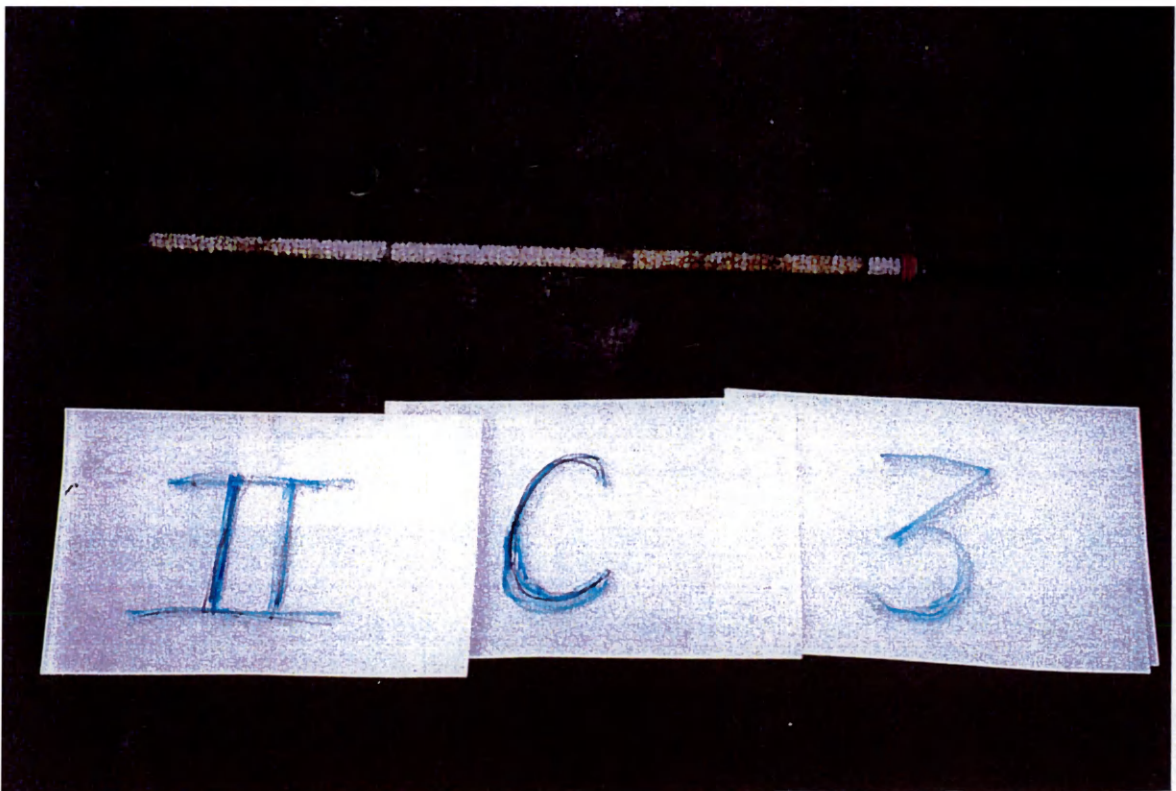


Figure 11: Photographs of bead columns IIIA-2 and IIC-3. Rubber bands mark the surface water - streambed boundary.

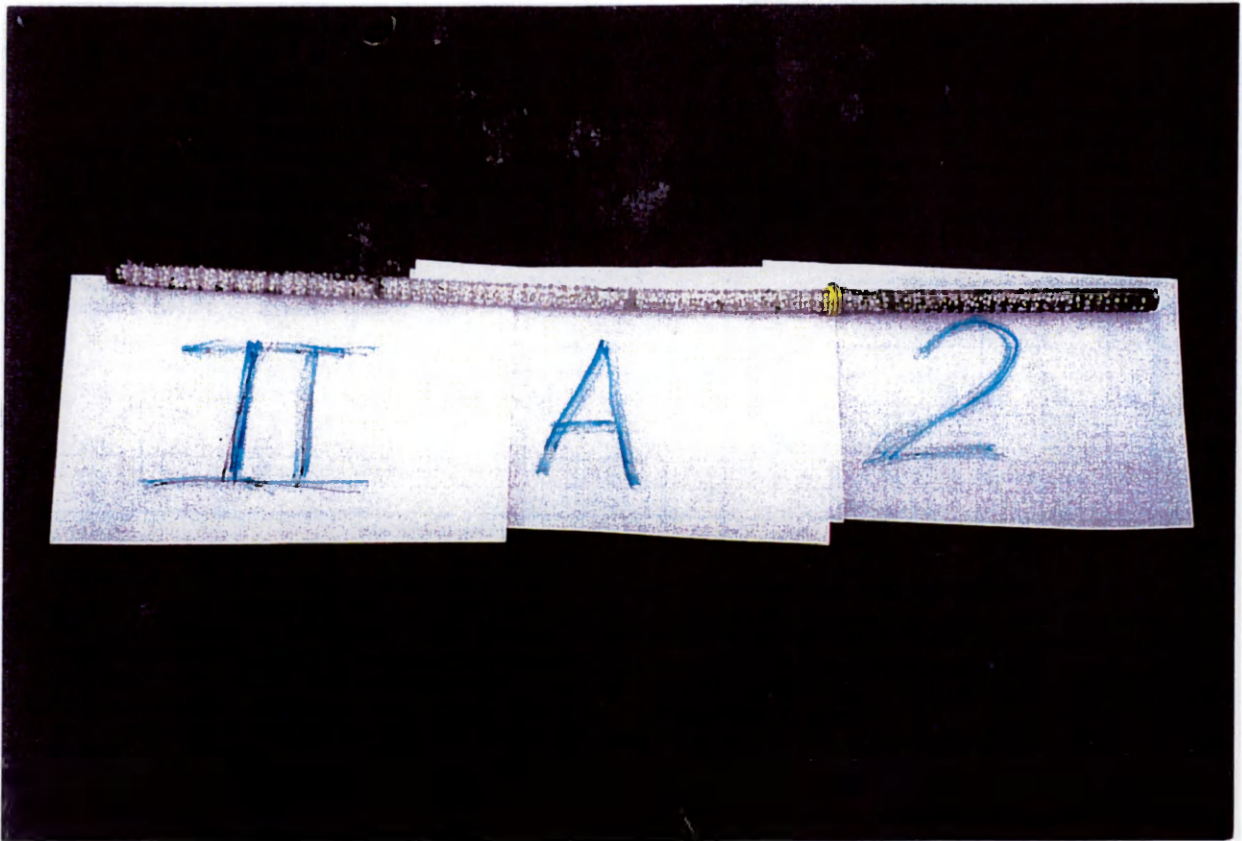


Figure 12: Photographs of bead columns IIA-2 and IA-2. Rubber bands mark the surface water - streambed boundary.

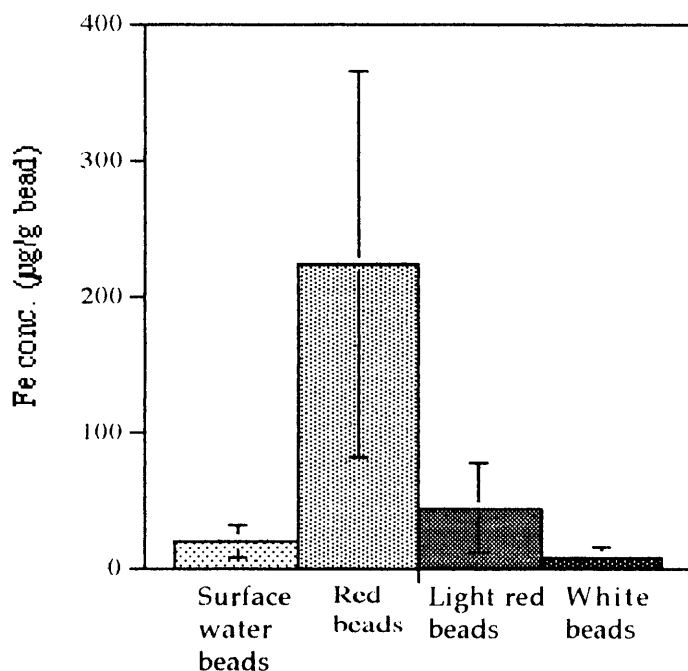
The five columns with no precipitation zones are IIA-2 (Figure 12), IIB-2, IIC-2, all of which were located in the center of the creek at Site II, IB-2, in the center of the creek at Site I, and IIA-1, located along the north bank of Site II. The concentrations of metals and As along the lengths of the columns mentioned above are listed in the Table 8.

CONCENTRATIONS ON BEADS ($\mu\text{g}/\text{g}$ BEAD)																	
SW/ (0cm=top)	Subs.	Colors on	As	Cu	Cd	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Sr	Tl	Zn
Sample Name	bound.	bead tube	As	Cu	Cd	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Sr	Tl	Zn
IA-2 0-12cm	13 cm	0-13 green	0.44	13	0.05	17.8	39	6.4	7.05	0.05	7.6	BDL	9.0	0.97	0.128	0.304	13.3
IA-2 12-15cm		13-15 light red	1.2	11	0.08	31.2	140	3.0	3.36	0.20	4.1	BDL	1.0	1.2	0.108	0.142	15.8
IA-2 15-23cm		15-22 red	0.42	9.6	0.05	18.2	210	3.6	3.77	0.31	4.8	BDL	6.4	3.0	0.085	0.161	12.9
IA-2 23-33.5cm			0.21	22	BDL	10.9	140	3.6	3.25	0.21	5.4	BDL	3.6	1.6	0.117	0.164	6.38
IA-2 33.5-42cm			0.22	9.0	BDL	10.7	190	3.8	3.41	0.30	5.6	BDL	3.1	1.6	0.071	0.164	6.13
IC-1 0-8cm	8 cm	0-8 green w/ red streak	0.48	9.1	BDL	5.55	18	2.6	1.98	BDL	2.0	0.25	3.6	0.71	0.075	0.238	6.79
IC-1 8-11 cm			0.58	11	BDL	2.65	30	4.2	4.81	BDL	3.7	0.73	5.0	1.1	0.094	0.174	8.04
IC-1 11-30 cm		8-11 v.l. red	BDL	8.0	BDL	4.49	7.6	4.1	3.52	BDL	4.9	0.34	2.3	0.48	0.060	0.137	9.84
IC-1 30-42cm		11-30 white 30-42 med. red	2.3	13	0.12	18.0	300	3.8	6.46	0.43	4.4	1.6	4.0	1.2	0.268	0.179	21.9
IIA-1 0-10.5cm	15.5 cm	0-10 drk green	BDL	BDL	BDL	4.22	10	BDL	1.30	BDL	BDL	BDL	3.0	BDL	BDL	BDL	3.61
IIA-1 10.5-15.5cm		10-15.5 green	BDL	BDL	BDL	2.91	9.2	BDL	0.880	BDL	BDL	BDL	2.9	BDL	BDL	BDL	2.12
IIA-1 15.5-22cm		15.5-20.5	BDL	BDL	BDL	3.11	6.3	BDL	0.612	BDL	BDL	BDL	2.4	BDL	BDL	BDL	1.79
IIA-1 22.5-33cm		l. green	BDL	BDL	BDL	2.98	5.5	BDL	0.616	BDL	BDL	BDL	2.5	BDL	BDL	BDL	4.46
IIA-1 33-42cm		20.5-42 white	BDL	BDL	BDL	1.12	3.4	BDL	0.410	BDL	BDL	BDL	2.3	BDL	BDL	BDL	2.17
IIA-2 0-10.5cm	13.5 cm	0-13.5 green	BDL	BDL	BDL	4.75	17	BDL	1.82	BDL	BDL	BDL	5.0	BDL	BDL	BDL	4.12
IIA-2 11-13.5cm		13.5-42 white	BDL	BDL	BDL	2.35	9.9	BDL	0.780	BDL	BDL	BDL	3.4	BDL	BDL	BDL	2.54
IIA-2 13.5-21.5 cm			BDL	BDL	BDL	1.82	7.6	BDL	0.648	BDL	BDL	BDL	2.8	BDL	BDL	BDL	1.98
IIA-2 22-32 cm			BDL	BDL	BDL	0.85	3.7	BDL	0.365	BDL	BDL	BDL	2.1	BDL	BDL	BDL	1.08
IIA-2 32-42cm			BDL	BDL	BDL	0.77	4.4	BDL	0.340	BDL	BDL	BDL	2.0	BDL	BDL	BDL	0.96
IIB-1 0-8cm	8cm	0-8 green	BDL	10	BDL	4.85	16	1.9	1.62	BDL	1.6	0.19	4.1	0.66	0.083	0.192	5.13
IIB-1 8-14cm		8-14 v. light red	0.75	9.4	BDL	4.01	28	1.68	1.27	BDL	1.37	BDL	4.9	1.62	0.098	0.253	7.49
IIB-1 14-26cm		14-26 white	0.6	7.8	BDL	2.65	11	1.71	0.58	BDL	1.51	BDL	1.9	0.85	0.067	0.203	5.79
IIB-1 26-35cm		26-41 light red	1.41	10	BDL	4.10	36	2.14	1.19	BDL	1.56	BDL	5.2	1.87	0.105	0.296	4.65
IIB-1 35-41cm			1.77	12	BDL	2.98	31	2.30	1.17	BDL	2.0	BDL	4.6	1.24	0.117	0.296	5.03
IIB-2 0-7cm	7 cm	0-6 green	BDL	BDL	BDL	5.23	16	BDL	1.75	BDL	BDL	BDL	4.4	BDL	BDL	BDL	4.79
IIB-2 7-12cm		6-8 light green	BDL	BDL	BDL	5.50	21	BDL	1.83	BDL	BDL	0.68	4.6	BDL	BDL	BDL	4.51
IIB-2 12-19.5cm		8-40.5 white	BDL	BDL	BDL	3.85	21	BDL	1.34	BDL	BDL	1.36	3.7	BDL	BDL	BDL	3.53
IIB-2 20-30cm			BDL	BDL	BDL	2.13	16	BDL	0.923	BDL	BDL	1.35	2.7	BDL	BDL	BDL	2.29
IIB-2 30-40.5cm			BDL	BDL	BDL	0.92	3.1	BDL	0.334	BDL	BDL	BDL	1.6	BDL	BDL	BDL	1.26
IIC-2 0-4.5cm			0.32	12	BDL	5.80	18	2.2	2.10	BDL	1.8	BDL	3.9	0.52	0.093	0.162	5.49
IIC-2 4.5-8.5cm			0.24	12	BDL	4.74	15	2.2	1.81	BDL	2.4	BDL	2.8	0.46	0.088	0.172	4.06
IIC-2 8.5-11cm	8.5 cm	0-8.5 green	BDL	11	BDL	3.55	11	1.7	1.50	BDL	2.6	BDL	2.0	BDL	0.070	0.126	3.38
IIC-2 11-21cm		8.5-40 white	BDL	9.9	BDL	2.46	7.7	1.5	0.843	BDL	2.4	BDL	1.5	BDL	0.062	0.122	2.41
IIC-2 21-30cm			BDL	6.9	BDL	1.29	4.0	0.9	0.390	BDL	BDL	BDL	BDL	BDL	BDL	BDL	1.49
IIC-2 30-40cm			BDL	6.5	BDL	1.01	3.1	1.1	0.37	BDL	1.4	BDL	BDL	BDL	0.14	0.090	1.3

Table 8: Concentrations of metals and As ($\mu\text{g}/\text{g}$ bead) along the lengths of 7 of the 27 bead columns.

(A full listing of the concentrations on all the bead columns are in the appendix.)

Analysis of the bead column sections revealed that the white sections generally contained the lowest metal concentrations, the red sections contained the highest concentrations, and the green zones contained concentrations in between the two, although closer to the low concentrations on the white beads. The element for which this was most clearly apparent was Fe (Figure 13).



Error bars are the standard deviation of the mean concentrations of the samples.

Figure 13: Mean Fe concentrations (µg/g bead) on bead color zones (all samples, compiled.)

The standard deviations of the means for Fe concentrations shown in the figure are so large because the figure uses data from all the columns, which come from different portions of the creek, which overly a very chemically heterogeneous groundwater system.

Examination of columns on an individual or site specific scale reveal significantly smaller variability bars among the different color zones. The concentrations of metals on the surface water beads were far less variable than those found on white or red beads, likely due to the relatively homogeneous chemical nature of the surface water across the site.

The comparison of ratios of dissolved metals in the surface water to ratios of metals on bead columns (surface water sections) shows the following sequence of preferential

precipitation in the surface water: $Fe > Cu > Zn > Mn > Na > Mg > Ca$. (i.e., of these metals, Fe is most unstable in the dissolved phase of the surface water, and Ca is most stable).

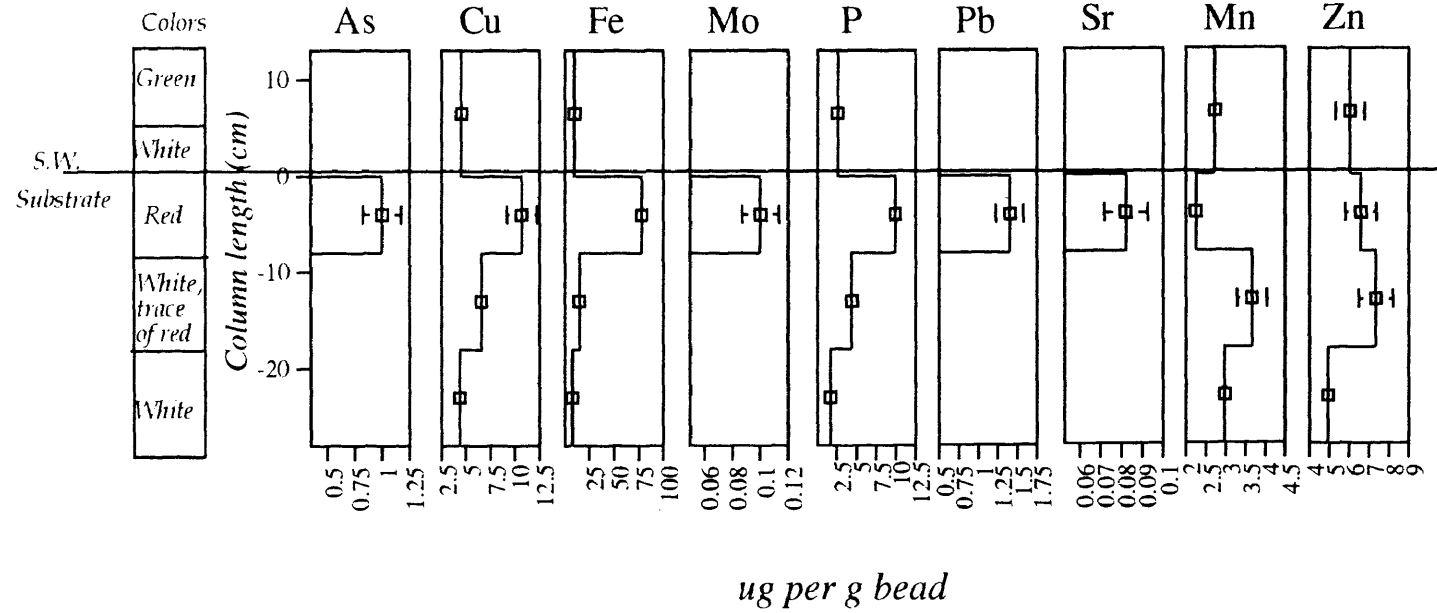
Figures 14-16 illustrate the concentrations of various metals and As along the lengths of a few of the bead columns, one from each site. Note the correlation of concentration with coloration on the columns, and that many metals peak in concentration at the same places along the columns. Metals which typically do not exhibit clear trends along the lengths of column are Ca, Mg, Na, and Ti. The appendix contains the concentrations of the metals and As for each section along all the bead columns.

Site-specific trends: Bead columns

Eight of the nine bead columns from Site I had red precipitation zones on them. The six bead columns within the banks (one in each bank for each transect) had the precipitation zones located right along the interface between the surface water and the streambank sediments, with concentrations of $Fe \approx 100-300 \mu\text{g/g}$ bead; $Mn \approx 3-5 \mu\text{g/g}$ bead; $Zn \approx 6-20 \mu\text{g/g}$ bead. Two of the three bead columns set into the center of the creek bed revealed precipitation zones. That of Transect A had the longest accumulation of precipitates found on any of the columns; the precipitation spanned the entire 30 cm length of its subsurface portion. The bead column in the center of Transect B was the only column at the site without any detectable precipitation on it.

At Site II, none of the three bead columns in the center of each transect had any detectable precipitation zones, yet those along the banks (except for IIA-1) did. However, the precipitation zones on four of the six were very light red in color and had metal concentrations significantly lower ($Fe \approx 30-40 \mu\text{g/g}$ bead; $Mn \approx 1.0-1.5 \mu\text{g/g}$ bead; $Zn \approx 3-7 \mu\text{g/g}$ bead) than on those in the darker red precipitation zones common to bead columns at the other sites.

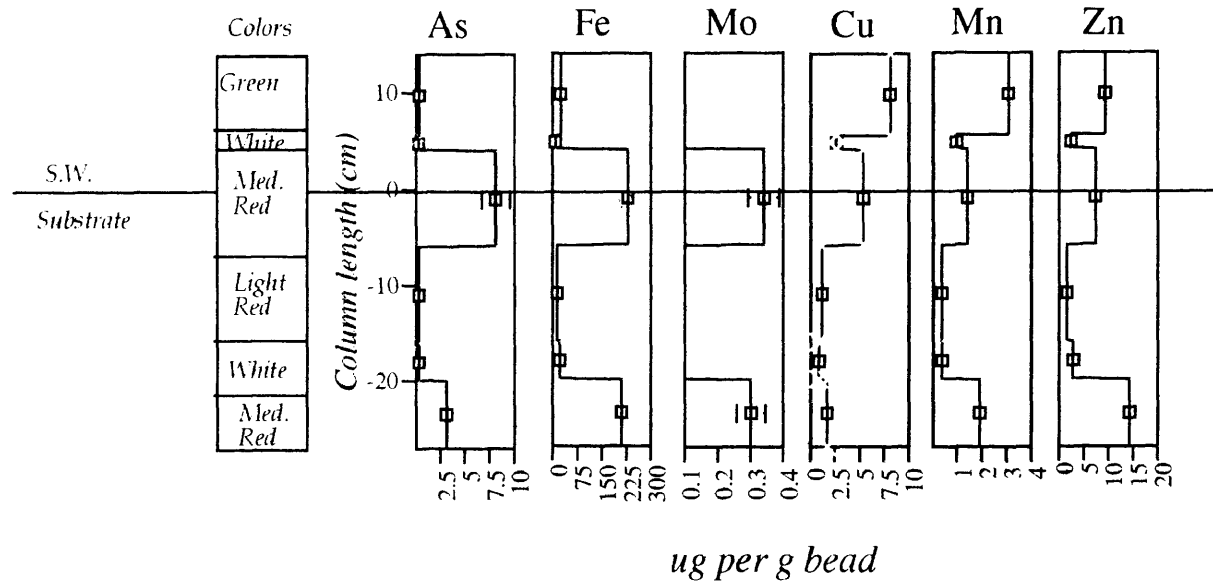
Bead Column IA-1



*BDL/ No significant trends:
Al, Ca, Cd, Mg, Na, Si, Ti*

Figure 14: Bead column IA-1 concentration profile

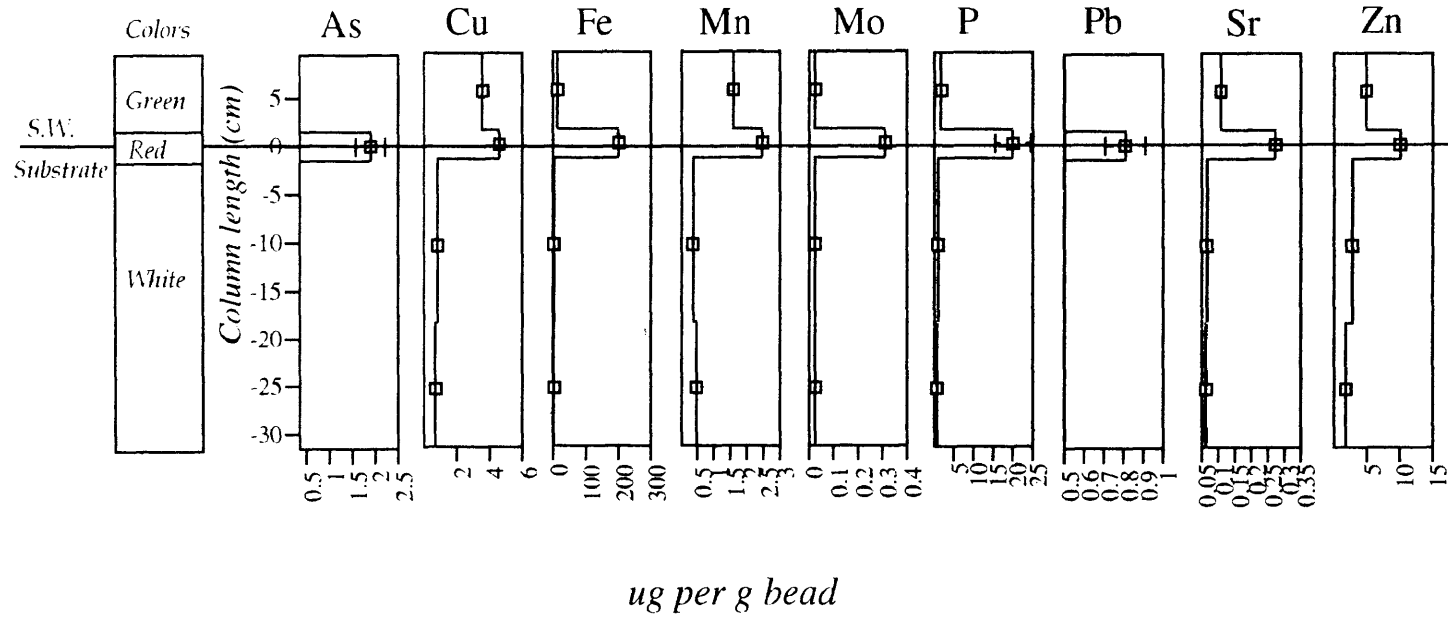
Bead Column IIC-3



*BDL/ No significant trend:
Al, Ca, Cd, Na, P, Pb, Si, Sr, Ti*

Figure 15: Bead column IIC-3 concentration profile

Bead Column IIC-1



*BDL/ No significant trends:
Al, Ca, Cd, Mg, Na, Si, Ti*

Figure 16: Bead column IIC-1 concentration profile

At Site III, every bead column, including the three in the center of the creek bed, had a zone of precipitation on it. The precipitation zones were all relatively dense, with high metal concentrations ($\text{Fe} \approx 100\text{-}700 \mu\text{g/g}$ bead; $\text{Mn} \approx 0.5\text{-}3 \mu\text{g/g}$ bead; $\text{Zn} \approx 5\text{-}18 \mu\text{g/g}$ bead), and occurred right at the surface water - substrate boundary. A few (e.g. IB-1, IIIA-1 and IIIB-3) had metal concentrations on their surface water portions which were significantly elevated (about 2-5 times more concentrated) than the more typical concentrations found on other surface water beads.

Physical mixing in the hyporheic zone:

The results of the examination of the physical relationships of groundwater and surface water flow directions, as well as the temperature patterns found beneath the creek bed at the three sites indicated that the waters were physically mixing. The solute chemical composition of the hyporheic zone samples being distinct from that of both the surface water and local groundwater is more evidence that physical mixing of the waters was likely taking place. The component of physical mixing in the hyporheic zone can be identified through the use of conservative ions, or natural tracers that do not make a transition between the solid and aqueous phases across the groundwater- surface water interface. In addition to not reacting, conservative metals concentrations should form a linear relationship when plotted against each other, indicating that they may be a product of mixing of chemically different waters [Faure, 1991]. The mixing ratios calculated from concentrations of conservative elements in the hyporheic zone indicate the proportion of groundwater to surface water present in each subsurface water sample.

Benner et al. [1995] used Cl and Mn as conservative elements in their study at Silver Bow Creek. Chloride is typically thought to be one of the least reactive solutes in aquatic systems and has been used in many tracer studies [Feth, 1981; Legrand-Marcq and Laudelot, 1985; Triska et al., 1989; Stollenwerk, 1994], and Mn has been used in other studies as a conservative tracer as well [Bencala et al., 1987]. However, other than in the surface water, Cl and Mn do not form a linear relationship according to the data collected for this study. Na has been found to act conservatively in aquatic systems together with Cl [Theobald et al., 1963; Chapman, 1982]. However, these two do not form a neatly linear relationship either, likely indicating an unidentifiable measurement problem with Cl. A small range of Cl concentrations (16-22 mg/l) are connected with a wider range of Na concentrations (approximately 20-120 mg/l). Most importantly, the very small differences

between the surface water Cl concentrations and the groundwater Cl concentrations (all but the ones collected near Site III) made it a poor choice for calculations of mixing ratios.

Ca and Mg appear to be acting conservatively at the site, and their concentrations in the surface water and the groundwater are relatively large. Ca and Mg follow linear conservative mixing relationships [Faure, 1991] (Figure 17).

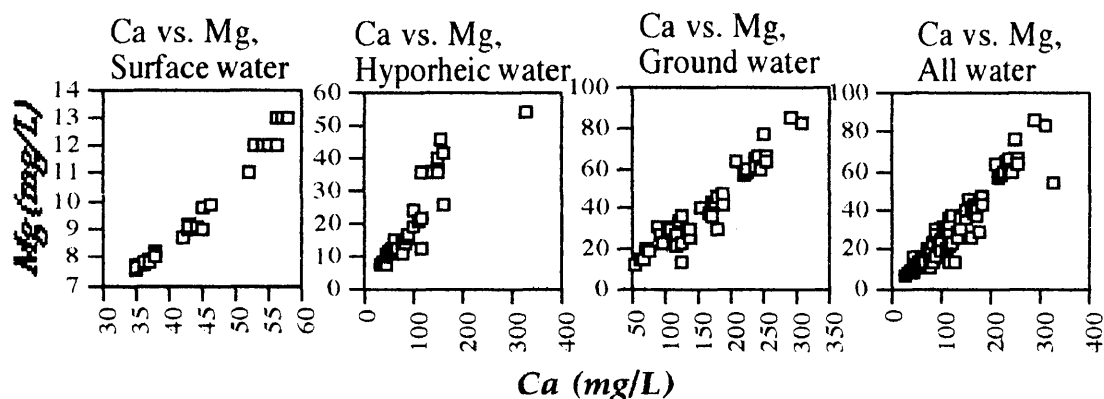


Figure 17: Dissolved Ca vs. Mg concentrations [mg/l] in water samples.

In addition, the solid phase sampling results support the conservative behavior of Ca and Mg. On 20 of the 27 bead columns, Ca and Mg did not exhibit any statistically significant trends along the length of the bead columns, suggesting that they generally undergo very small to insignificant chemical changes across the surface water - groundwater interface as reflected in the solid phase. The comparison of dissolved vs. solid phase ratios in the surface water further confirm the observed conservative nature of Mg and Ca, since they were shown to be most stable in the surface water's dissolved phase compared to the other metals. Other researchers have found one or both of these elements to behave conservatively in aqueous systems as well [Stauffer, 1985; McKnight and Bencala, 1990; Wetherbee and Kimball, 1991]. Thus, Ca and Mg were used to calculate mixing ratios in the hyporheic zone.

By using the mixing equation:

$$y = 100 * ([HZ]-[SW])/([GW]-[SW]) \quad [Benner \textit{et al.}, 1995; \textit{modified} \\ \textit{from Triska \textit{et al.}, 1989}]$$

where

y= the percent groundwater in the hyporheic zone water
 [HZ]= the concentration of the element in the hyporheic zone
 [SW]= the concentration of the element in the surface water
 [GW]= the concentration of the element in the groundwater

the percent of groundwater in the hyporheic zone water samples could be calculated using Ca and Mg concentrations (Figures 18-20). Because each sample was unique, a separate mixing ratio was calculated for each.

Calcium and Mg (and Na at Site II) yielded nearly identical values of mixing ratios for most of the subsurface water samples, usually within +/- 10%. Variabilities exist due to possible sampling errors, analysis errors, and inaccurate concentrations used to represent the local groundwater chemistry.

Finding the endmember groundwater concentration ([GW]) proved difficult. Due to the variability of the metals concentrations in the groundwater generally existing on scales smaller than the size of any of the three sites used in this study [Shay, 1997], it was not feasible to make a generalization of the groundwater chemistry across the study area for calculating mixing equations. Instead, values for the groundwater concentrations were assigned according to the chemistry of the closest high-metal, upgradient sample at each transect. Although some of these samples may have been influenced by the surface water due to their close proximity (within 3 meters) to the creek, they likely represent the local groundwater interacting with the creek better than would samples taken from piezometers further away on the floodplain which commonly contain water of greatly different chemical composition. If these samples have some surface water component within them, the mixing ratios may be an overestimation of the amount of groundwater calculated per sample. However, there is no way to draw the line between "pure" groundwater and

SITE I
 Percent groundwater in subsurface water samples
 at various flow rates
Map view with cross-sections of transects

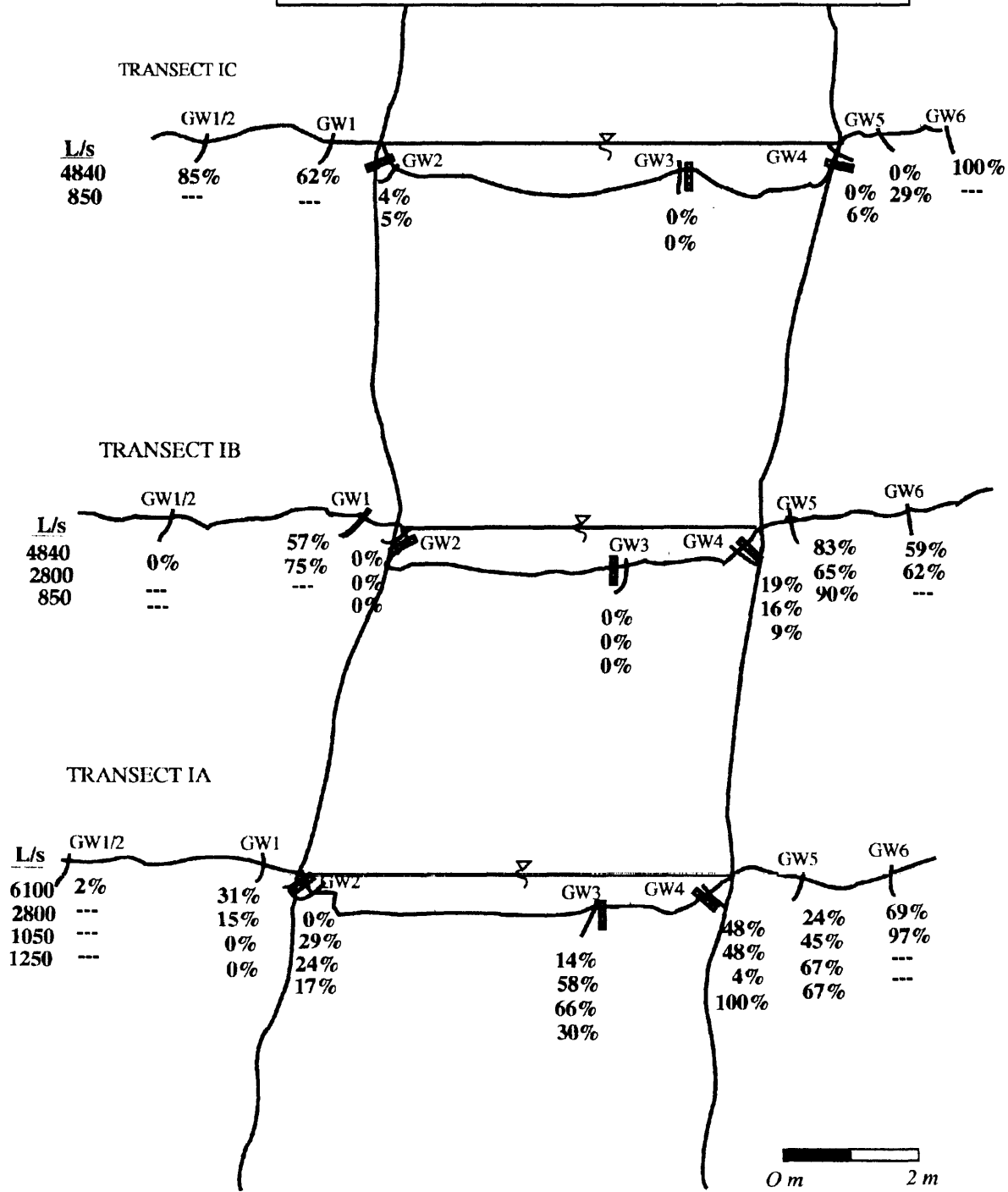


Figure 18

SITE II
Percent groundwater in subsurface water samples
at various flow rates
Map view with cross-sections of transects

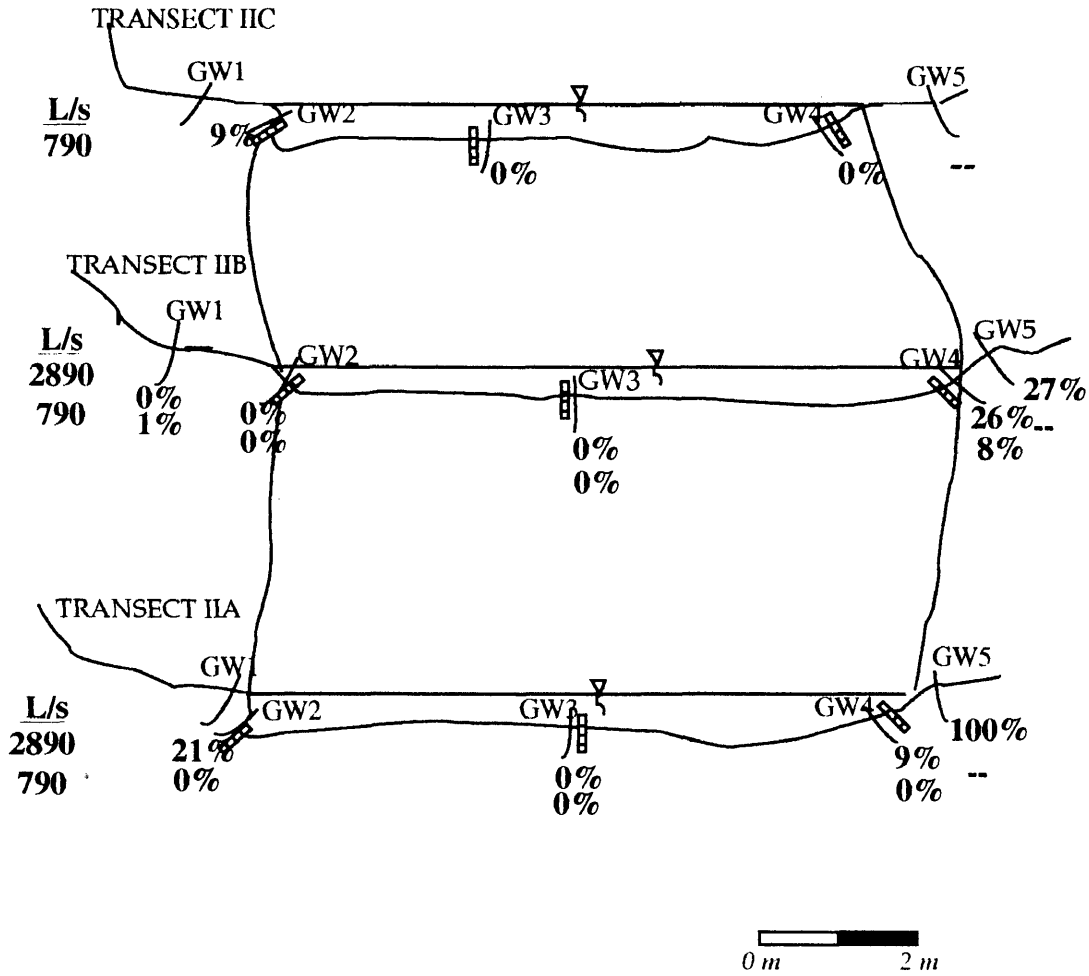


Figure 19

SITE III
Percent groundwater in subsurface water samples
at various flow volumes
Map view with cross-sections of transects

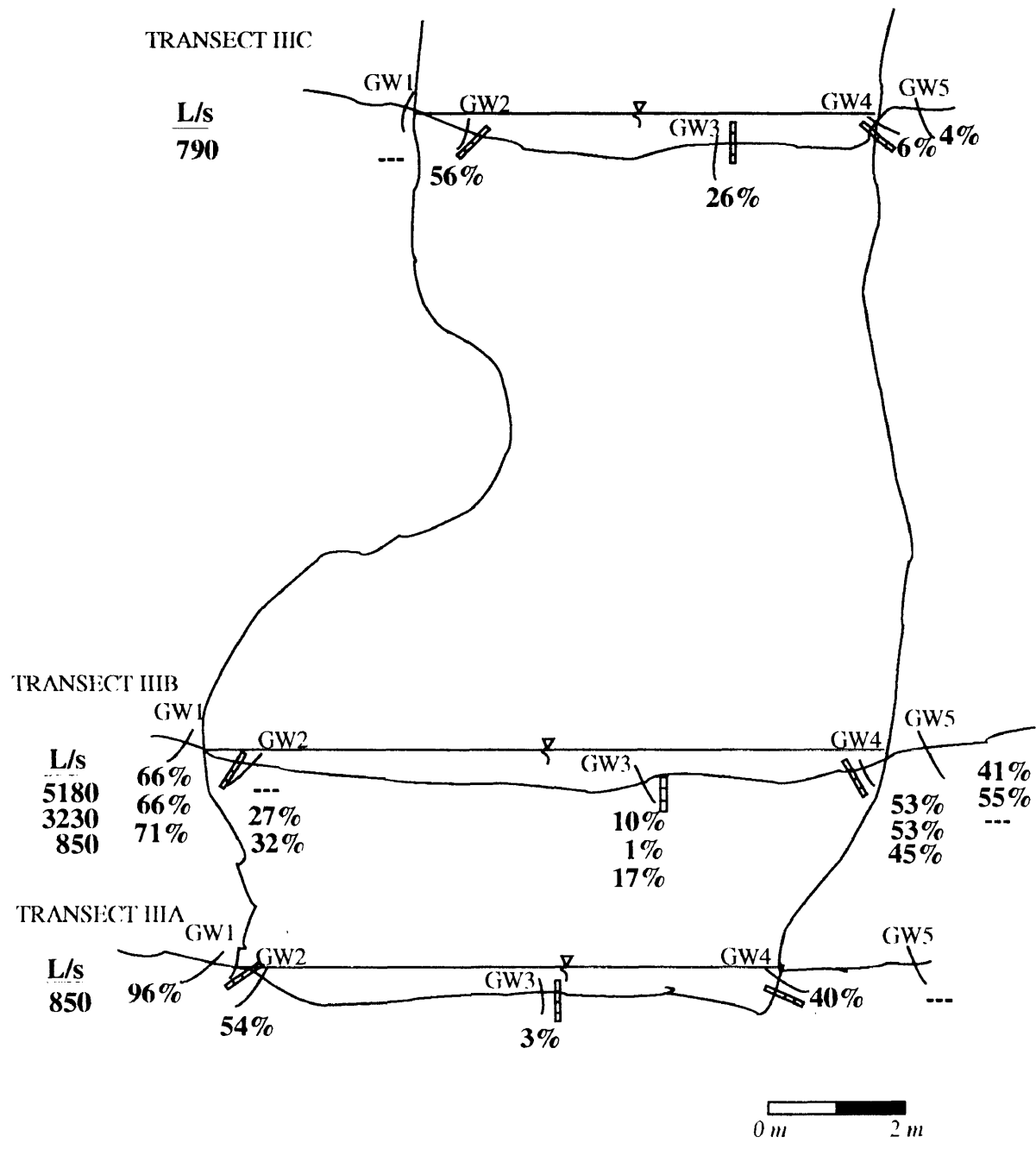


Figure 20

groundwater partially diluted by inflowing surface water, given the heterogeneity of the groundwater system itself.

At Site I, the groundwater Ca and Mg concentrations used were taken from east bank (upgradient) samples with the highest concentrations of these elements. For transect IA, water sample IA-GW4 (9/27/95) was used as the groundwater end member; for Transect IB, sample IB-GW5 (8/27/95) was used; and for Transect C, sample IC-GW6 (6/28/95) was used. At Site II, the values used were derived from IIA-GW5 (7/6/95), a sample collected from the stream bank about 0.5 m south of the creek at the site. It contained significantly higher concentrations of Ca, Mg, and Na than did the two closest (10 and 20 m away) floodplain piezometers. The mixing ratios for Site II indicate that Na is acting conservatively together with Ca and Mg, and thus was included in calculating the average mixing ratios at this site. At Site III, the values were taken from the subsurface water access tube GW-1 in transect IIIA. The sampling tube was chosen because it was upgradient of the creek, and its concentrations of conservative elements were highest.

The surface water concentrations [SW] varied significantly with flow volume, and thus the end member surface water values varied with each sampling date. This variability was accounted for in the calculations; surface water values used were adjusted for the specific concentration found on the same day as the hyporheic zone water sample was collected.

Site-specific trends with mixing ratio results

As seen in Figures 18-20, the mixing ratio results indicate that significant portions of the shallow subsurface water samples are groundwater in a the samples (> 20% groundwater in 48% of the samples). The mixing ratios also illustrate the large degree of heterogeneity seen in the samples, both in terms of spatial distribution and changes with

flow rate in the creek. Nonetheless, on a site by site analysis, general trends were discernable.

At Site I's east bank, where the highest metal concentrations were observed in the water samples, the mixing ratios further indicate that groundwater comprises a relatively large portion of the samples (mean 48%). The west bank had lower mean percentages of groundwater composition (17%), and the samples from the below the center of the creek at Transects B and C were calculated as having 0% groundwater. The samples from below the center of the creek at Transect A, which contained relatively high metal concentrations, show groundwater comprising proportions of the samples similar to those in the east bank. Thus, the west bank samples are again interpreted as representing the local hyporheic zone, where infiltrating surface water mixes with the groundwater in the west bank. The upgradient east bank samples more closely resemble the local groundwater as characterized by Shay [1997], and are interpreted to represent the groundwater moving into the creek. This general relationship was also described by Benner et al. [1995] at the same site (specifically, at this study's transect IC).

At Site II, the mixing ratios indicate that both banks contained some proportion of groundwater, although not more than about 27% at any location (other than IIA-GW5, which was used as the groundwater end member). All samples taken from beneath the center of the streambed were calculated to contain 0% groundwater. The mixing ratios suggest that this site is heavily controlled by the surface water, as compared with other sites, and groundwater is not a major component of the shallow hyporheic zone (Figure 19).

The mixing ratios at Transects A and B at Site III also complement the general trends seen in the metal concentrations across the transects. Generally, both east and west banks contained similarly high proportions of groundwater according to the mixing ratios. Each transect exhibited increasing groundwater concentrations further from the center of the

creek. Yet, even those samples below the center of the creek bed contained from 1 to 27% groundwater, depending on sample time and location. Thus, the east banks samples are interpreted to represent the local inflowing groundwater, due to the high metal concentrations found in the samples and because of the higher head in the east bank piezometer. Because of the undetectable head differences between the west bank and the surface water, there does not appear to be a strong gradient moving surface water into the groundwater or vice versa. This, in conjunction with the high concentrations of metals and the calculated proportion of groundwater in the west bank samples and in the center of the creek, suggest that the creek may be gaining water from all sides at this site.

Chemical transition in the hyporheic zone

No metals other than Ca and Mg (and in places Na and Cl) form linear plots of their concentration relationships, and their mixing ratios do not match those produced using the conservative elements. This indicates that chemical reactions are taking place in the hyporheic zone in conjunction with the physical mixing of the waters.

Examination of the bead columns with precipitation bands further indicates that physical mixing is not the only factor controlling the chemistry of the hyporheic zone. The highest concentrations of metal precipitates typically occurred right at the surface water - substrate boundary. These precipitation zones are interpreted to represent the portion of the hyporheic zone where acidic, reduced, and high-dissolved metal concentration water comes into contact with large enough amounts of the neutral, oxic, and more dilute surface water to induce the precipitation of metal oxides [Benner *et al.*, 1995]. Very minimal precipitation of metals is found on the white portions of the bead tubes below the red precipitation zones. These white portions are thus interpreted to have resided in more acidic and less oxic environments which are favorable to retaining metals in solution [Stumm and Morgan, 1970]. Such environments are found where larger proportions of

groundwater are present. The concentrations of precipitates observed in the surface water portions of the bead columns are also relatively low. Therefore, conservative physical mixing between the two low solid-phase concentration zones cannot explain the thick bands of high metal concentration precipitates that separate the two. Instead, these zones of precipitation are products of chemical reactions which occur where the chemically distinct surface and groundwaters mix to the extent that metals in the groundwater become unstable in the dissolved phase due to the dilution by oxic and neutral surface water.

The precipitation of metals within the mixing zone agrees with chemical theories and field and laboratory observations of pH and Eh controlling the partitioning of metals into their dissolved and solid phases [Stumm and Morgan, 1970; Chapman *et al.*, 1983; Nordstrom and Ball, 1986; Stollenwerk, 1994]. In numerous studies on streams affected by acid mine drainage, researchers have illustrated the close correlation of dissolution/precipitation and sorption/desorption reactions with changing pH and redox conditions [Theobald *et al.*, 1963; McKnight and Bencala, 1990; Davis *et al.*, 1991; Smith *et al.*, 1992]. Generally, metals become increasingly less soluble in less acidic and more oxidizing conditions, and microbial action greatly accelerates the rate of redox reactions [Nordstrom, 1982]. The oxidation rate of Fe (II) to Fe (III) is pH dependent [Nordstrom, 1982], typically occurring between a pH of about 4.5 and 5.0, and commonly results in the formation of Fe oxides or oxyhydroxides [McKnight and Bencala, 1990]. At higher pH levels, metals such as Al, Cd, Cu, Mn, Pb, and Zn come out of solution and can sorb with the Fe oxides [Johnson, 1986; Filipek *et al.*, 1987; Rampe and Runnells, 1989; Callender *et al.*, 1991].

Most of the hyporheic zone water samples had near neutral pH levels, yet they commonly contained high levels of metal concentrations. This suggests that many of the metals remain in the dissolved phase (or in colloids $<0.45 \mu\text{m}$ that can pass through the filter) at least until a near-neutral pH is reached. The relationship between pH levels and

metal concentrations is seen in Figure 21. The relatively steady concentrations of metals such as Fe, Co, Mn, Mo, and Zn between pH levels of 2-6 contrast with the range of concentrations present between pH units 6 and 8, where the levels of dissolved metals concentrations drop drastically, presumably due to their precipitation or adsorption onto Fe oxides and mixing with the more dilute and oxic surface water. Constituents without clear pH-dependent dissolved phase concentrations are Ca, Mg, Na, Sr, Ti, Cl, and NO_3^- -N, at least for the pH ranges encountered in this study.

Interestingly, the behavior of dissolved As differs from that of the metals. At near neutral pH levels, As is observed to occur in solution (Figure 21), as reflected in the appearance of As in about half of the hyporheic zone water samples and in a few of the near-stream groundwater samples. As previously mentioned, dissolved As was not found above the detection limit of 0.07 mg/l in any of the floodplain piezometers sampled by Shay [1997] nor in the surface water. Of the hyporheic zone samples with detectable As concentrations, 88% had a pH between 6 and 7. What may be occurring is that the dissolved As captured in the samples had been in a chemical environment in which As is neither stable as associated with sulfides (as it is in reduced environments) nor with iron oxides (as it is in oxidized environments). The source for the As is likely the streambed sediments in the hyporheic zone, which were found to have average concentrations of 433 ppm (stdev= 75) in the <63 μm size fraction. Still, the bead columns show a very close correlation of solid As and Fe in precipitation zones, which is what was expected based on reports by other researchers [DeCarlo and Thomas, 1985; Rampe and Runnells, 1989; Smith et al., 1992; Moore, 1994]. On all of the bead columns with measurable levels of As, the concentration profiles along the columns exactly matched those of Fe. This indicates that the As sorbs with iron oxyhydroxides in the hyporheic zone, where conditions are oxic and neutral enough to foster the precipitation of iron oxyhydroxides and

As. Thus, the water and bead samples captured two different phases of As existing in the hyporheic zone.

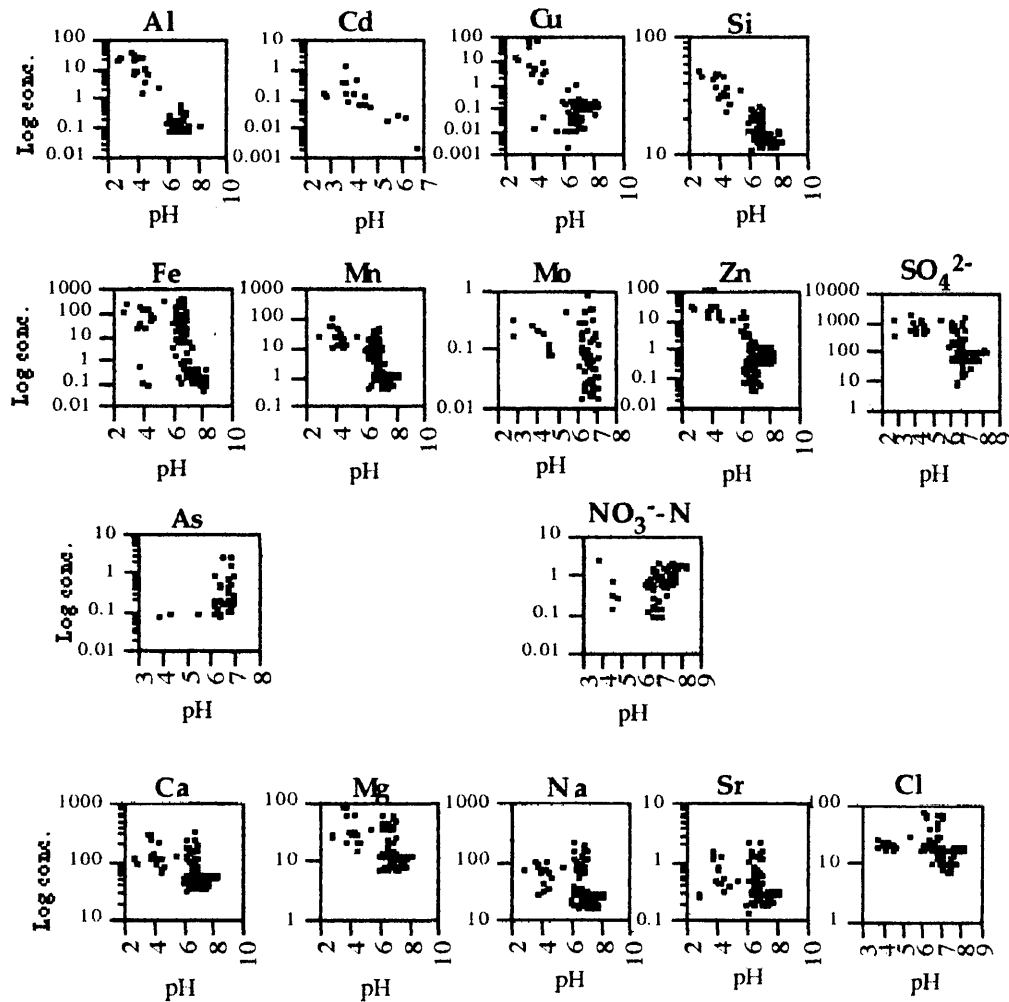


Figure 21: Concentrations [mg/l] of elements versus pH levels. Note log scale. The graphs include data from all the surface, ground, and hyporheic zone water samples, in addition to 12 groundwater samples collected by Shay (1997).

Surface water infiltration versus groundwater upwelling:

The low dissolved SO_4^{2-} , NO_3^- -N, and O_2 concentrations and the high alkalinity levels in many of the hyporheic zone samples (Table 5) can be explained by microbial reduction. Sulfate reduction occurs in less oxic environments than does NO_3^- reduction [Brock et al., 1994], meaning that SO_4^{2-} reduction will not occur before NO_3^- reduction in an increasingly reducing environment. This may suggest that SO_4^{2-} reducing bacteria and

NO_3^- reducing bacteria inhabit the zones where surface water infiltrates the hyporheic zone and supplies SO_4^{2-} , NO_3^- , and O_2 . The bacteria reduce the NO_3^- and subsequently the SO_4^{2-} as well, where the conditions become increasingly more reducing with depth. In their study of the hyporheic zone of Little Lost Man Creek in California, Triska et al. [1989] also attributed the non-conservative behavior on NO_3^- -N either to microbial uptake or dissimilatory reduction. The relatively high alkalinity levels associated with most of these samples can be attributed to the products of microbial respiration [Brock et al., 1994]. These patterns are helpful in distinguishing between zones of surface water infiltration versus groundwater upwelling. These samples with the low SO_4^{2-} , NO_3^- -N, and O_2 concentrations are interpreted as having been taken from zones of surface water infiltration.

Precipitation patterns on some of the bead columns also aided in making the distinction between areas where surface water was infiltrating into the hyporheic zone versus where groundwater was upwelling. For example, bead columns with no precipitation zones (Columns IIA-1, IIA-2, IIB-2, and IIC-2 (Figure 22) exhibited a steady

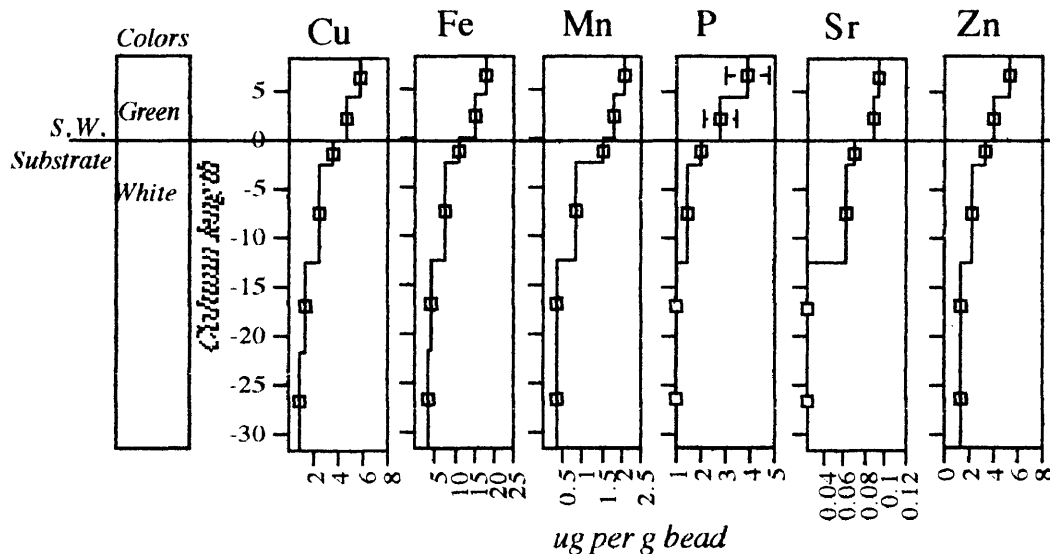


Figure 22: Concentration of metals coatings on bead column IIC-2, an example of surface water infiltration along the length of the column. Metals whose concentrations are not shown were either below detection or did not exhibit any significant differences in concentrations along the length of the column.

decreasing trend of metals concentrations with depth and were interpreted as having been in zones of surface water infiltration. The concentrations of metals along the subsurface portions of these bead columns decrease with depth presumably due to redox gradients and dropping pH levels in the substrate, which will increasingly drive metals to go into solution. The relatively small but detectable amounts of accumulation on the surface water portions of the bead columns are thought to be products of redox reactions precipitating metals in the oxic, neutral pH surface water. They also may be due to elevated metal algal coatings on the beads.

Metal concentration trends along bead columns IA-2, IB-1 (Figure 23), IC-3, IIIA-1, IIIA-2, IIIA-3, and IIIC-3 are interpreted as being indicators of groundwater infiltration into the stream. On these columns, Fe concentrations come to a maximum lower down (i.e. deeper in the substrate) on the bead column than do metals such as Cu, Mn, and Zn. These other metals were found to precipitate out higher on the bead column, closer to the surface water. As discussed earlier, this sequence follows redox patterns found by other researchers in which Fe precipitates out more readily in conditions less oxic and alkalinity than those required for the precipitation of metals such as Mn, Cu, and Zn. Such conditions can be met with a greater proportion of neutral pH and more oxic surface water, and it is thought that the proportion of surface water in the hyporheic zone grades upward towards the surface water boundary. Thus, this sequence of metal concentrations can be interpreted as having been formed as a result of groundwater recharging the surface water. (It is possible that more bead columns exhibited this sequence of metals precipitation, but the separation of beads columns into no less than 2 cm sections for purposes of analysis was not on a small enough scale to detect the possible trends.)

On some of the bead columns (IA-2, IB-1, IB-3, IC-3, IIIA-1, IIIB-3, and IIIC-3), Cu, Mn, and Zn have their maximum concentrations on the surface water portions of the bead columns. These concentrations are significantly higher than the typical concentrations

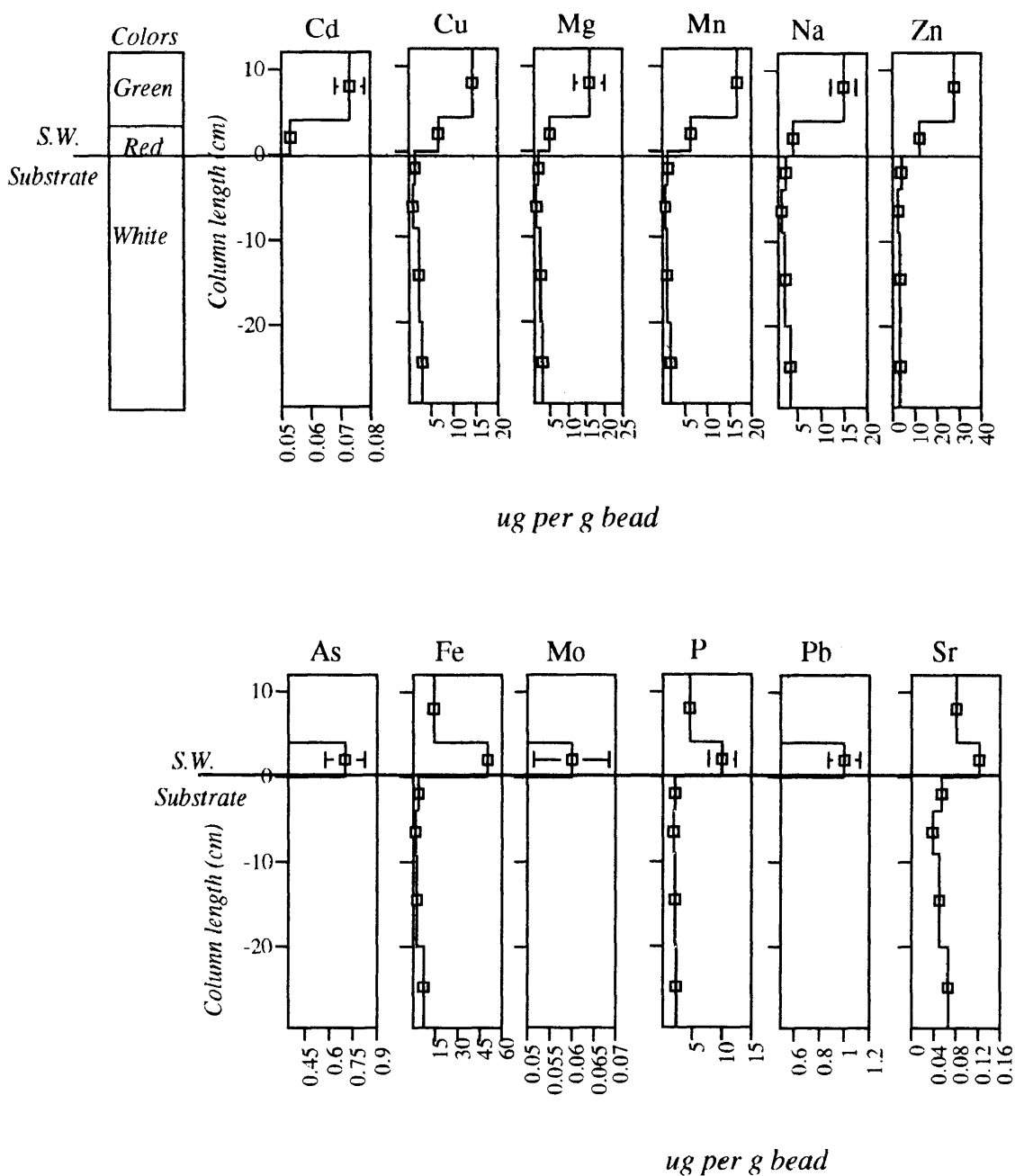


Figure 23: Concentrations of metals along the length of bead column IB-1, an example of groundwater upwelling along the length of the column. Metals Al, Ca, not shown (Al, Ca, Ti) had no significant trends.

of these metals found on most of the other surface water beads. This implies that the source for these metals must be in the upwelling groundwater and not in the surface water. It shows that these metals may stay in solution throughout their migration through the hyporheic zone and precipitate only once in the surface water.

The implication of these processes is that groundwater is upwelling into the creek in many areas, supplying dissolved As and metals to the stream sediments and water. Many of the metals appear to precipitate right at the surface water - substrate boundary, suggesting that the bed sediments are a sink for metals loading from the groundwater. Blowes et al. [1991] reported the presence of 1-5 cm thick "hardpans" at the depth of active oxidation of sulfide-rich tailings, and attributed their formation to the precipitation of iron hydroxide and oxyhydroxide minerals upon contact with porewater of increased pH. The thickness of these precipitation zones and processes of their formation are analogous to those found at the mixing zone at Silver Bow Creek. It is likely that hardpans do not form at Silver Bow Creek because the precipitation occurs on unstable creek sediments, which are continuously being transported downstream.

Spatial and temporal controls on mixing:

As illustrated by the mixing ratios, there appear to exist highly variable distributions of the degree of mixing on both spatial and temporal scales. The small scale heterogeneity in the groundwater chemistry across the site is likely an important chemical control on the variability of the hyporheic zone chemistry. A major physical control in the extent of mixing is thought to be the general groundwater to surface water flow direction relationships at the sites. This is seen at Site II, where surface water and groundwater flow approximately parallel to each other, and the stream subsurface was more dominated by surface water than at either of the other two sites. Sites I and III, where perpendicular

relationships between groundwater and surface water exist, showed much stronger groundwater signatures in the chemistry of the water collected from their banks.

More complex spatial variability is seen in that the concentration of groundwater in each hyporheic zone water sample usually did not decrease with increasing proximity to the creek at Sites I and II. This is in contrast with the findings of Triska et al. [1989], who concluded that all well locations within 3.5 meters of the wetted channel at Little Lost Man Creek contained at least 80% stream water. An example of this variability is at sample site (IA-GW 3) below the center of the creek at Site I, which was found to consistently contain proportions of groundwater similar to or higher than those found in the east bank samples. The bead column which was set into the creek bottom a few centimeters away from the IA-3 water tube was found to have a large mixing zone as suggested by the thick and long (at least 30 cm) band of precipitation visible on the column. The solid and dissolved chemistry collected at this location show that zones of high groundwater concentrations are found 0-30 cm under the creek channel and not exclusively in the banks or at deeper levels below the creek bed. The water sampling tubes in the center of the creek at the other 2 transects (IB- and IC- GW3) at the site were set in slightly coarser-grained sediments and had concentrations nearly identical to those in the surface water, and the mixing equation showed no groundwater component in these samples. In these places, the grain size is largest and the sediments least consolidated, allowing for a less obstructed infiltration of surface water. This exemplifies the amount of spatial variability found at the sites, as well as the importance that small scale physical heterogeneities in the streambed sediments can have in controlling the degree and depth of surface water - groundwater mixing. Such physical controls by streambed topography and sediment size have been credited by other researchers as controlling the extent of surface water - groundwater interaction as well [Bencala, 1984; Savant et al., 1987; Thiobodeaux and Boyle, 1987; Valett et al., 1990;

Harvey and Bencala, 1993; White, 1993; Pusch, 1996; Henry et al., 1984; Vervier et al., 1992].

Physical factors may also complicate mixing within the banks. The sediments within the floodplain of Silver Bow Creek contain complex layers of variably sized grains due to the history of meandering and flooding of the creek. Thus, the mixing between surface water and groundwater within the banks of the creek is thought to take place in a series of complex settings which provide for variably sized and conductive flowpaths through which the waters can travel and mix. Evidence for this is seen in that a few of the bead columns exhibited two separate bands of precipitation (Figure 10) which are thought to be a product of the small scale interfingering of the chemically distinct surface and groundwaters. The importance of floodplain stratigraphy and permeability in controlling the spatial distribution of the hyporheic zone has been noted by other researchers as well [*Stanford and Ward, 1993*]. For instance, *Ward et al. [1994]* contend that the abundance of invertebrates residing within the Flathead River floodplain is largely determined by site-specific geomorphic and hydrogeologic features, as opposed to mere distance from the river channel. Similarly, *Triska et al. [1993]* reported that distance from the channel accounted for only 40% of the variance in nominal travel time of a chloride tracer injected into Little Lost Man Creek in California. They credited the rest of the variance to the complex flowpaths caused by heterogeneous structure, size, and hydraulic conductivity of the floodplain sediments.

In some places, such small scale flowpaths appear to have a stronger control on the mixing zone than do the general groundwater and surface water flow direction relationships. This is seen in the downgradient west bank of Site I, where all the bead columns (IA-1, IB-1, and IC-1) emerged with zones of precipitation on them at the surface water - substrate boundary, despite physical and chemical evidence from the water suggesting that the creek is recharging this area. A reddish streak on the surface water

portion of the bead column IC-1 was noted, and the precipitation sequence on IB-1 implies that groundwater-rich water is infiltrating the surface water. Thus, groundwater flow may be strongly controlled by small scale flowpaths that defy the larger-scale flow directions and water with large groundwater compositions may be entering the creek along some portions of the downgradient banks. This means that the surface water is not uniformly dominating the chemistry in the west bank hyporheic zone, which was found by Benner et al. [1995] to be comprised almost entirely of inflowing surface water up to about 1 m in depth. Instead, the mixing zone appears to vary in width on the scale of centimeters. In some areas it may be a lot smaller and shallower than previously thought, with groundwater located much closer to the creek than indicated in the previous study.

Temporal variability in the Silver Bow Creek hyporheic zone is irregular as well. The amount of groundwater in the shallow hyporheic zone generally did not decrease with respect to increasing flow volumes in the creek. For example, transect IA was sampled four times, during which the flow rate measured 1050, 1250, 2800 and 6100 l/s in the creek. The amount of groundwater in the hyporheic zone was different at each location during each flow event, (according to the mixing ratios) and the concentrations of non-conservative elements changed as well. However, these changes did not form any particular patterns with the flow volume in the creek. The percent of groundwater calculated to be present at a rate of 6100 L/s was about 14%; at a rate of 2800 L/s, there was 58%; during a rate of 1250, there was 66%, and finally, during rate of 1050 there was 30% (Figure 18). This indicates that during high flow events, the hyporheic zone is not flushed out by the surface water, and mixing with groundwater continues. This is similar to the findings of Harvey et al. [1996], who found that hyporheic exchange occurred during both low and high base flow in St. Kevin Gulch, and in contrast to those of Legrand-Marcq and Laudelot [1985], who concluded that the influence of transient storage mechanisms are greatest during low flow. The hyporheic zone in Silver Bow Creek

appears to continue to play a significant and sometimes larger role in solute transport into the creek during high flows, at least along certain portions of the creek. This process can be explained by the close hydrologic connection between the groundwater and the creek. Shay [1997] reported that rises in the water table of the adjacent aquifer brings larger volumes of groundwater into contact with the tailings, and this process causes the upper 1 m of groundwater to become significantly more contaminated with metals. This more highly contaminated and higher elevation groundwater is thought to continue to interact with the creek during the high flow events, causing the continued-- and in places, more extensive-- contamination of hyporheic zone.

Groundwater impacts on the surface water:

If dilution were the only control on the change in surface water chemistry with flow rate, the expected concentration change of conservative elements would be the same as the change in flow rate (a 7.5-fold change over the course of the study period). If the same sources and sinks were working on the system during all flow rate episodes in the creek, then no change in concentration would be observed. Transition metals (e.g. Fe, Cu, and Zn) generally do not have a well defined relationship with discharge, due to the various chemical, physical, and biological factors in the stream that may influence their concentrations at a given time (Forstner and Wittmann, 1979; Wetherbee and Kimball, 1991.) However, conservative elements should not be impacted by chemical changes in the stream. Yet, the concentrations of conservative elements increased by 150% and 170% for Ca and Mg, respectively, when flow rate was only 13% of the highest flow measured (Table 7). This suggests that dilution is not the only control on the surface water chemistry during high flow events and that there exists a source which contributes to the changing chemistry with flow rate.

Thus, this portion of Silver Bow Creek is not a losing reach, rather it was gaining dissolved Ca and Mg from some source, assumed to be the groundwater. Surface runoff was not observed on any surface water sampling dates, and thus it is assumed that direct floodplain runoff during the sampling period makes up only a small to insignificant portion of the source. Calculations of baseflow estimates, based on Ca and Mg concentrations, were made using the assumption that Ca and Mg behave conservatively and are of consistent concentrations in the groundwater within the entire site area and the upstream portions of the Silver Bow Creek system.

Estimates of the groundwater component (baseflow) in the surface water were made using the following mixing equation:

$$\%GW = \frac{[LF] - [HF]}{[GW] - [HF]} * 100$$

where

$\% GW$ = percent groundwater (baseflow) during low flow

$[LF]$ = concentration of conservative element in surface water during low flow, when highest concentration was found

$[HF]$ = concentration of conservative element in surface water during high flow, when lowest concentration was found

$[GW]$ = average concentration of conservative element in groundwater (Ca = 175mg/l, Mg = 43 mg/l; Na = 65 mg/l; from Shay [1997])

As a conservative lower range estimate, the amount of Ca in the surface water high flow (35 mg/l) is assumed to not be of groundwater origin at all; thus, the equation yields 16% as a low-end estimate for the percent of groundwater in the surface water during low flow. Using Mg, the same percentage is found. For the high-end estimate, it is assumed that all the Ca and Mg come from the groundwater alone, and thus $[HF]=0$. In this case, the estimated percent groundwater in the surface water is calculated to be 33% according to Ca, and 30% according to Mg. Na concentrations show slightly higher percentages of baseflow, with 23% for the low end estimate and 43% for the high end estimate.

Due to the extent of contamination of the groundwater, this amount of loading into the surface water likely accounts for a significant amount of the contamination in the water and sediment of Silver Bow Creek.

CONCLUSIONS

The quality of water in the hyporheic zone is impacted by the groundwater to a significant extent in many portions of the study area at Silver Bow Creek. High concentrations of metals were present during both low and high flows in this shallow subsurface periphery of the streambed. According to conservative element mixing calculations, half of the hyporheic zone samples (<30 cm below the creek bed) were composed of at least 20% groundwater. Many of the metals remained in their dissolved phases throughout the hyporheic zone and precipitated only once in the surface water or just a few centimeters below. The precipitation reactions appeared to be most strongly controlled by changes in pH. In some cases, Fe and associated elements were found to precipitate out deeper in the substrate than other more mobile metals such as Cu, Mn, and Zn, suggesting that groundwater is moving into the more neutral and oxic creek.

The heterogeneity of the subsurface water chemistry and the differences in precipitation on the beads illustrate the need for very detailed sampling in order to capture the small-scale flowpaths which appear to regulate the nature and magnitude of surface water and groundwater mixing. Sampling of a small area during a limited time cannot be used to adequately represent the entire system. Variations in local groundwater metal concentrations need to be considered as well as the physical controls on the extent of mixing between the surface water and the groundwater. In some locations, groundwater appears to infiltrate the creek even from downgradient locations. This indicates that small scale flowpaths on the scale of centimeters may have stronger controls on groundwater interaction with the hyporheic zone and surface water despite the larger scale flow paths in the adjacent alluvial aquifer.

Thus, the mixing of the groundwater with the surface water takes place in a series of complex physical settings which create a chemically heterogeneous hyporheic zone

underlying and lateral to the creek bed. The product of the interaction is a transfer of metals into various depths of the hyporheic zone and into the stream channel, primarily into the solid phase. This precipitation of groundwater-borne metals onto stream and hyporheic zone sediments appears to be a constant source of pollution to the creek and must be recognized in remediation designs. In addition, the geochemical environment in the hyporheic zone creates a area in which As was found to be present in solution, although it is not found in solution in either the surface water or groundwater. Therefore, the hyporheic zone is a spatially and temporally heterogeneous, geochemically distinct environment in which metals and arsenic undergo chemical transformations which significantly control contaminant cycling between surface water and groundwater in the Silver Bow Creek hydrologic system.

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Water levels at each site (Piezometers and staff gauges)

SITE I											
	(measurements in meters above sea level)										
<u>DATE</u>	<u>Top of casing</u>	<u>5/9/95</u>	<u>5/15/95</u>	<u>5/22/95</u>	<u>5/30/95</u>	<u>6/12/95</u>	<u>6/20/95</u>	<u>6/23/95</u>	<u>6/28/95</u>	<u>7/6/95</u>	<u>7/12/95</u>
P-44 (E bank)	1602.952	1602.038	1602.075	1601.986	1601.895	1602.151	1602.038	1602.007	1601.953		1602.050
Staff Gauge 1	1603.083										
P-38 (W bank)	1602.587	1601.916	1601.998	1601.895	1601.806	1602.081	1601.947	1601.916	1601.855	1601.767	
Lls in creek		5437	5465	5295	4701	8297	6088	5295	4842	2888	5465
							P38inSBC				
<u>DATE</u>	<u>Top of casing</u>	<u>7/19/95</u>	<u>8/15/95</u>	<u>8/24/95</u>	<u>8/27/95</u>	<u>8/31/95</u>	<u>9/27/95</u>	<u>10/1/95</u>	<u>10/20/95</u>	<u>11/3/95</u>	<u>11/4/95</u>
P-44	1601.910				1601.669			1601.709			1601.697
Staff Gauge 1	1601.901	1601.663	1601.663	1601.657	1601.642	1601.666	1601.706	1601.706	1601.706	1601.703	1601.383
P-38	1601.843	1601.617	1601.593	1601.581	1601.569			1601.596	1601.611	1601.587	1601.581
Lls in creek	2803	1218	1048			793	1246	2577	2747	2605	2662
SITE II											
<u>DATE</u>	<u>Top of casing</u>	<u>7/6/95</u>	<u>7/12/95</u>	<u>8/15/95</u>	<u>8/24/95</u>	<u>8/31/95</u>	<u>9/4/95</u>	<u>10/20/95</u>	<u>11/3/95</u>	<u>11/4/95</u>	
D-18 (S bank)	1603.278	1602.815	1602.980	1602.657	1602.654		1602.635	1602.687	1602.678	1602.675	
Staff Gauge 8	1603.617			1602.675	1602.660	1602.648	1602.648	1602.699	1602.687	1602.702	
D-19 (N bank)	1603.778	1602.882					1602.666	1602.718	1602.693	1602.693	
Lls in creek	2888	5465	1218	1048	793	793	793	2747	2605	2662	
SITE III											
<u>DATE</u>	<u>Top of casing</u>	<u>7/11/95</u>	<u>7/12/95</u>	<u>8/15/95</u>	<u>8/24/95</u>	<u>8/28/95</u>	<u>8/31/95</u>	<u>10/20/95</u>	<u>11/3/95</u>	<u>11/4/95</u>	
D-30 (W bank)	1604.269	1603.714	1603.708	1603.397	1603.388	1603.379	1603.379	1603.440	1603.419	1603.422	
Staff gauge 10	1604.391			1603.385	1603.361	1603.355	1603.355	1603.413	1603.397	1603.391	
D-31 (E bank)	1604.357	1603.742			1603.400	1603.397	1603.388	1603.452	1603.437	1603.431	
Lls in creek	5182	5465	1218	1048			793	2747	2605	2662	

ICAPES data from water sample analysis on 7/5/95																							
Sample No	Sample Name	Sample Date	Sample Analytic Date	As	Cu	Cd	Co	Cr	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Ph	SI	Sr	Tl	Zn		
1	STD1 Blank	7/5/95	9:21	0.04887	0.0045	0.0014	0.0047	0.0009	0.0095	0.0025	0	3.90757	0.0042	0.0009	0.07376	0.00228	0.10319	0.07338	0.00495	0.03171	0.01985	0.0018	
2	STD2	7/5/95	9:24	1.92266	1.53071	2.30695	2.85233	1.29247	1.60704	1.29247	1.60704	2.41338	2.13333	2.41338	1.98399	0.9028	8.75271	12.0843	17.217	2.66623	0.029	3.0829	
3	STD3	7/5/95	9:24																				
4	STD4	7/5/95	9:27																				
5	STD5	7/5/95	9:31																				
6	BLANK	7/5/95	9:34	0.006	0.001	0.002	0.001	0.008	0.014	0.015	0.004	0.0004	0.0035	0.019	0.002	0.07	0.002	0.007	0.0009	0.0009	0.002	0.0026	
7	USGS T107	7/5/95	9:38	0.25	0.014	0.018	0.049	0.043	0.057	0.042	2.34	0.0042	0.018	252	0.015	0.15	0.026	0.441	0.066	0.0036	0.0006	0.0006	
8	USGS T117	7/5/95	9:42	0.073	0.003	0.003	0.006	0.005	0.008	0.009	0.002	0.003	0.003	0.013	0.002	0.15	0.003	0.657	0.277	0.0035	0.0038	0.0008	
9	BLANK	7/5/95	9:48	0.004	0.012	0.009	0.008	0.019	0.017	0.012	0.031	0.0017	0.008	0.381	0.004	0.1	0.003	0.02	0.0115	0.0011	0.0034	0.0011	
10	BLANK	7/5/95	9:52	0.005	0.001	0.006	0.003	0.019	0.015	0.012	0.013	0.0007	0.0025	0.136	0.002	0.08	0	0.017	0.0011	0.0007	0.0010	0.0010	
11	IA SW1	7/5/95	9:55	0.046	0.012	0.018	0.005	0.024	0.066	0.031	7.88	0.4445	0.0076	16.5	0.002	0.25	0	1.43	0.2007	0.0022	0.3831	0.0022	
12	IB SW1	7/5/95	9:58	0.051	0.026	0.027	0.007	0.023	0.106	0.279	8.24	0.5197	0.0085	17.6	0.006	0.28	0.013	1.37	0.2161	0.0007	0.3959	0.0007	
13	IC SW1	7/5/95	10:02	0.043	0.03	0.024	0.008	0.029	0.1106	0.357	8.14	0.5555	0.0087	17.1	0.005	0.25	0.012	1.37	0.2105	0.0011	0.4996	0.0011	
14	IA SW1	7/5/95	10:05	0.063	0.013	0.024	0.0015	0.0236	0.0962	0.338	7.71	0.4436	0.0074	16.5	0.006	0.26	0.019	1.42	0.2021	0.0018	0.3932	0.0018	
15	IB SW1	7/5/95	10:08	0.043	0.001	0.025	0.007	0.0205	0.09	0.261	8.16	0.4952	0.0085	17.5	0.006	0.26	0.022	1.36	0.2155	0.0004	0.3567	0.0004	
16	IC SW1	7/5/95	10:11	0.048	0.02	0.027	0.0018	0.0263	0.1121	0.333	8.09	0.5546	0.0078	17.2	0.007	0.3	0.011	1.36	0.2101	0.0011	0.5289	0.0011	
17	IA SW1	7/5/95	10:14	0.048	0.005	0.027	0.004	0.022	0.0925	0.352	7.54	0.4709	0.0076	16.3	0.002	0.24	-0.01	1.39	0.1981	0.0011	0.3708	0.0011	
18	IB SW1	7/5/95	10:17	0.041	0.021	0.027	0.0012	0.0217	0.0811	0.249	8.1	0.4873	0.0041	17.3	0.004	0.24	0.001	1.35	0.2143	0	0.3344	0	
19	USGS T107	7/5/95	10:21	0.238	0.015	0.0185	0.0132	0.0355	0.0317	0.039	2.23	0.0513	0.0154	22.5	0.037	0.1	0.036	0.417	0.0638	0.0011	0.0862	0.0011	
20	BLANK	7/5/95	10:24	0.02	0.002	0.006	0.0005	0.016	0.096	0.01	0.014	0	0.0012	0.027	0.003	0.05	-0.009	0.001	0.0006	-0.0043	0.0019	0.0019	
21	IC SW1	7/5/95	10:28	0.043	0.028	0.023	0.0013	0.027	0.0889	0.283	8.14	0.5836	0.0039	17.4	0.01	0.27	-0.012	1.37	0.2099	0.0022	0.4419	0.0022	
22	IA SW1	7/5/95	10:31	0.1	0.012	0.024	0.0018	0.0276	0.1213	0.463	7.62	0.5238	0.0046	16.6	0.001	0.28	0.002	1.41	0.1979	0.0029	0.4725	0.0029	
23	IB SW1	7/5/95	10:34	0.043	0.01	0.026	0.0007	0.026	0.0859	0.275	8.01	0.4889	0.0067	17.4	0.003	0.26	0.009	1.34	0.2123	0.0007	0.3483	0.0007	
24	IC SW1	7/5/95	10:37	0.05	0.012	0.028	0.0005	0.0222	0.1066	0.332	7.97	0.5497	0.0076	17.1	0.004	0.4	-0.02	1.34	0.2081	0.0004	0.5091	0.0004	
25	IB SW1	7/5/95	10:40	0.051	0.016	0.028	0.0017	0.0268	0.0852	0.198	8.1	0.4914	0.0062	17.3	0.004	0.28	-0.016	1.35	0.2143	0.0016	0.3726	0.0016	
26	IC SW1	7/5/95	10:43	0.052	0.016	0.024	0.0028	0.023	0.0892	0.284	8	0.6568	0.0056	17.3	0.003	0.29	0.003	1.34	0.2077	0.0007	0.4344	0.0007	
27	IA SW1	7/5/95	10:46	0.055	0.02	0.024	0.0029	0.0267	0.0917	0.15	11.9	0.4425	0.0039	21.9	0.015	0.25	-0.004	1.63	0.2768	0.0001	0.4143	0.0001	
28	IB SW1	7/5/95	10:49	0.054	0.003	0.026	0.0028	0.0281	0.0856	0.186	8.04	0.5459	0.005	17.6	0.006	0.22	0.001	1.37	0.205	0.0014	0.4415	0.0014	
29	IC SW1	7/5/95	10:51	0.024	0.039	0.026	0.0025	0.024	0.0961	0.18	7.66	1.42	0.135	16.8	0.005	0.25	0.012	1.51	0.1992	0	0.5782	0	
30	IA SW1	7/5/95	10:55	0.079	0.008	0.024	0.0008	0.0216	0.1128	0.395	7.5	0.4474	0.0081	16.4	0	0.28	-0.004	1.4	0.1976	0.0014	0.3955	0.0014	
31	IB SW1	7/5/95	10:58	0.04	0.169	0.018	0.0035	0.031	0.0225	0.636	7.81	1.886	0.017	18	0.006	0.26	-0.001	1.35	0.2311	0.0018	0.2904	0.0018	
32	IB SW1	7/5/95	11:01	0.025	0.11	0.008	0.0001	0.0244	0.1084	2.32	7.64	1.38	0.0114	17.8	0.005	0.21	-0.006	1.31	0.2303	0.0011	0.2285	0.0011	
33	IC SW1	7/5/95	11:04	0.038	0.13	0.022	0.0025	0.0375	0.0099	2.9	8.8	4.206	0.0472	18.6	0.003	0.79	0.009	1.87	0.273	0.0014	0.0381	0.0014	
34	IA SW1	7/5/95	11:07	0.026	0.108	0.018	0.0012	0.0384	0.0015	25.4	8.09	3.966	0.042	18	0.003	0.56	0.003	1.79	0.2449	0.0018	0.0415	0.0018	
35	USGS T107	7/5/95	11:10	0.246	0.13	0.127	0.0172	0.0142	0.0354	0.0306	0.081	2.2	0.0567	0.0179	22.5	0.034	0.17	0.015	0.413	0.0636	0.0013	0.0855	0.0013
36	BLANK	7/5/95	11:13	0.016	0.015	0.005	0.0005	0.0017	0.0075	0.0085	-0.002	0.0009	0.004	0.088	0	0.13	-0.006	0.001	0.0009	-0.0029	0.0022	0.0022	
37	IA SW1	7/5/95	11:17	0.103	0.12	0.06	0.0025	0.0015	0.017	0.136	0.455	7.67	0.4776	0.007	16.8	0.006	0.73	0.003	1.43	0.2017	0.0047	0.4362	0.0047
38	USGS T117	7/5/95	11:20	0.099	0.16	0.094	0.0057	0.0374	0.0274	0.467	10.7	0.2402	0.028	21.8	0.014	0.41	0.006	0.628	0.2733	0.0039	0.1977	0.0039	
39	IA SW1	7/5/95	11:23	0.042	0.66	0.028	0.0028	0.0438	0.017	3.96	22.5	35.36	0.86	63.8	0.01	0.21	0.642	2.09	0.4461	0.0004	1.337	0.0004	
40	IB SW1	7/5/95	11:31	0.035	0.07	0.007	0.0075	0.0259	0.0022	61.6	13.7	7.332	0.1899	36.5	0.001	0.28	0.006	1.64	0.3071	0.0007	0.1966	0.0007	
41	IC SW1	7/5/95	11:33	0.062	0.058	0.018	0.0018	0.0008	0.0267	0.0354	7.82	7.28	0.8041	0.0569	17.5	0.001	0.4	-0.012	1.25	0.1837	0.0029	0.1099	0.0029
42	IA SW1	7/5/95	11:36	0.055	0.36	0.012	0.0085	0.0262	0.0258	88.6	14.3	15.37	0.1879	24.1	0.002	0.63	0.006	1.76	0.342	0.0025	0.1514	0.0025	
43	IC SW1/2	7/5/95	11:39	0.24	0.021	0.022	0.0035	0.026	0.1822	26	25.5	29.72	0.0467	40.5	0.064	1.1	0.033	5.05	0.3078	0.0144	0.3639	0.0144	
44	IA SW1	7/5/95	11:42	0.016	0.107	0.173	0.06	0.1328	0.167	11.4	28.8	35.67	0.1673	74.9	0.063	0.79	0.011	5.05	0.2565	0.0153	32.42	0.0153	
45	IB SW1	7/5/95	11:45	0.023	0.005	0.024	0.067	0.144	0.1135	22.7	26.5	22.78	0.3225	74.1	0.061	0.61	0.025	4.82	0.2927	0.0061	24.17	0.0061	
46	IC SW1	7/5/95	11:48	0.18	0.183	0.117	0.0888	0.147	0.2037	29.1	28.9	34.96	0.0488	43.1	0.073	1.2	-0.024	5.49	0.3474	0.0173	29.27	0.0173	
47	BLANK	7/5/95	11:51	0.087	0.008	0.005	0.0005	0.0025	0.0428	0.099	0.864	0.0605	0.0157	0.03	0	0.13	-0.014	0.009	0.0014	0.0022	0.0022	0.0022	
48	BLANK	7/5/95	11:54	0.03	0.019	0.014	0.0003	0.0023	0.0147	0.018	0.058	0.0127	0.0077	0.035	0.001	0.14	0.018	0	0.0005	-0.002	0.011	-0.002	
49	USGS T117	7/5/95	11:57	0.245	-0.01	0.039	0.0038	0.0432	0.0088	0.015	11	0.248	0.0194	22.5	0.016	0.43	0.012	0.647	0.2807	0.004	0.3076	0.004	
50	USGS T107	7/5/95	12:00	0.245	0.009	0.13	0.0179	0.0123	0.0389	0.034	2.21	0.0516	0.0245	23.1	0.002	0.2	0.016	0.417	0.0646	0.0018	0.0669	0.0018	
51	IA SW1	7/5/95	12:04	0.393	1.62	1.09	0.0033	0.0179	0.0876	0.1748	71.3	20.7	19.9	15.08	31.1	0.011	0.35	0.842	2.05	0.6179	0.0164	0.3843	0.0164
52	IC SW1	7/5/95	12:08	0.189	0.149	0.124	0.0007	0.0033	0.0223	0.0413	17.5	21.5	13.1	25.33	41	0.048	0.21	0.64	2.39	1.453	0.0122	0.1391	0.0122
53	IA SW1	7/5/95	12:11	0.372	0.06	0.24	0.001	0.0209	0.0639	0.0454	54.2	14.8	10.63	0.086	35	0.039	0.21	-0.004	2.33	0.3054	0.0014	13.46	0.0014
54																							

ICAPTS data from water sample analyses on 7/23/95 and 7/24/95																								
#	Sample Name	Sample Date	Analysis Date	Analysis Time	Al	As	Cu	Cd	Co	Cr	Cs	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Si	Sr	Tl	Zn	
213	IA-SW11(2)	7/19/95	7/23/95	16:46	0.051	0.222	46.4	0.0334	0.0009	0.0062	0.262	0.143	10.1	8.257	0.0036	21.6	0.006	0.09	0.17	0.012	1.35	0.2489	0.0026	0.7043
219	IA-SW11	7/19/95	7/24/95	14:30	0.059	0.011	45	0.0026	0.0001	0.01	0.129	0.165	9.66	8.158	0.0049	20.7	0.001	0.09	0.12	0.012	1.35	0.2489	0.0026	0.7043
289	IA-SW11	7/19/95	7/24/95	15:19	0.063	0.016	45.9	0.0028	0.0001	0.0136	0.131	0.173	9.66	8.352	0.0064	21	0.002	0.21	0.015	1.37	0.2528	0.002	0.7358	
304	IA-SW11	7/19/95	7/24/95	16:09	0.073	0.014	46.4	0.0024	0.0004	0.0102	0.142	0.172	9.68	8.498	0.0072	21.5	0.002	0.19	0.008	1.38	0.2558	0.001	0.7418	
319	IA-SW11	7/19/95	7/24/95	16:53	0.079	0.007	46.6	0.0027	0.0001	0.012	0.1363	0.172	9.6	8.5	0.0061	21.5	0.004	0.21	0.013	1.38	0.2551	0.0013	0.7403	
334	IA-SW11	7/19/95	7/24/95	17:40	0.084	0.011	45.2	0.003	0.0004	0.014	0.1306	0.177	9.22	8.225	0.0057	20.9	0.001	0.26	0.002	1.33	0.2456	0.0016	0.7161	
349	IA-SW11	7/19/95	7/24/95	18:24	0.084	0.016	45.2	0.003	0.0007	0.0141	0.1313	0.177	9.12	8.231	0.0059	21	0.004	0.26	0.002	1.33	0.2456	0.0016	0.7145	
363	IA-SW11	7/19/95	7/24/95	19:05	0.091	0.011	45.7	0.0026	0.0001	0.0136	0.1335	0.186	9.19	8.346	0.0056	21.3	0.003	0.27	0.004	1.34	0.2481	0.0013	0.7214	
175	IB-SW11	7/19/95	7/24/95	19:40	0.088	0.007	45.4	0.0027	0.0003	0.0118	0.1313	0.187	9.07	8.308	0.0034	21.1	0.003	0.31	0.004	1.33	0.2453	0.002	0.7144	
206	BLANK C	7/19/95	7/24/95	16:02	0.018	0.009	0.006	0.0001	0.0005	0.0062	0.0303	0.001	0.026	0.0012	0	0.006	0.001	0.03	0.016	0.003	0.0003	0.0018	0.0087	
207	BLANK C	7/19/95	7/24/95	16:06	0.009	0.011	0.019	0.001	0.0005	0.0005	0.0007	0.014	0.014	0.0012	0.0009	0.057	0.004	0.01	0.001	0.005	0.0001	0.0002	0.0352	
208	BLANK C	7/19/95	7/24/95	16:09	0.023	0.017	0.11	0.001	0.0026	0.0031	0.063	0.018	0.034	0.0662	0.0014	0.044	0.001	0.012	0.011	0.002	0.0001	0.0002	0.0427	
141	IA-GW1	7/19/95	7/24/95	18:01	0.069	0.26	68.1	0.0008	0.0007	0.0437	0.0007	77.4	11.7	8.373	0.1176	29.8	0.003	0.98	0.022	1.76	0.652	0.002	0.0713	
113	IA-GW2	7/19/95	7/24/95	16:35	0.06	0.355	86.4	0.0014	0.0037	0.0479	0	73.1	14.6	9.63	0.1143	29.8	0.003	0.98	0.022	1.76	0.652	0.002	0.0713	
119	IA-GW3	7/19/95	7/24/95	16:27	1.43	0.009	105	0.1108	0.0662	0.0644	1.806	55.8	25.5	16.29	0.0823	60.1	0.042	0.18	0.042	2.32	0.5375	0.0013	13.28	
125	IA-GW4	7/19/95	7/24/95	17:14	0.072	0.765	63.8	0.0025	0.0177	0.149	0.0242	41.8	20	28.57	0.6077	80.9	0.001	0.02	0.07	2.16	0.5454	0.001	0.5524	
318	IA-GW5	7/19/95	7/24/95	17:52	20.6	0.112	85.3	0.0076	0.0548	0.0919	0.0042	125	23.1	21.07	0.1764	65.1	0.062	0.18	0.055	4.4	0.3891	0.0012	26.9	
318	IA-GW5	7/19/95	7/24/95	17:52	20.6	0.112	85.3	0.0076	0.0548	0.0919	0.0042	125	23.1	21.07	0.1764	65.1	0.062	0.18	0.055	4.4	0.3891	0.0012	26.9	
374	IA-GW6	7/19/95	7/24/95	19:37	19.9	0.097	84.4	0.0264	0.0548	0.0937	0.0042	124	22.3	20.89	0.1743	64.5	0.062	0.2	0.035	4.32	0.3823	0.0016	26.34	
332	IA-GW6	7/19/95	7/24/95	18:33	21.6	0.041	140	0.2664	0.0708	0.1484	0.484	15.42	112	37.3	37.01	0.1887	100	0.079	1.2	0.024	5.37	0.469	0.0153	43.7
366	IA-SW1	7/19/95	7/24/95	19:14	0.081	0.006	46.4	0.0037	0	0.0154	0.1261	0.201	9.31	8.237	0.0053	21.4	0.003	0.29	0.004	1.36	0.2518	0.002	0.7128	
227	IA-SW1	7/19/95	7/24/95	17:10	0.025	0.02	46.7	0.0032	0.0008	0.001	0.1322	0.193	10.1	8.324	0.0038	21.4	0.004	0.14	0.004	1.43	0.259	0.001	0.7409	
298	IB-GW1	7/19/95	7/24/95	15:47	0.723	1.88	139	0.0056	0.0108	0.082	0.3108	139	23.9	23.54	0.2592	39.9	0.005	0.36	0.123	2.23	0.9619	0.0036	0.3675	
312	IB-GW2	7/19/95	7/24/95	16:33	0.087	0.098	42.9	0.0009	0.0016	0.0183	0.0214	10.7	8.47	3.32	0.0218	21.4	0	0.21	0.001	1.39	0.2575	0.0016	0.2702	
312	IB-GW3	7/19/95	7/24/95	17:05	0.085	0.02	42.8	0.0029	0.0025	0.0122	0.0913	1.2	8.8	7.739	0.0086	21.1	0.001	0.31	0.004	1.33	0.2316	0.0016	0.5112	
323	IB-GW4	7/19/95	7/24/95	17:08	0.067	0.045	60.8	0.0011	0.0099	0.0436	0.0126	63.5	14.3	7.26	0.0442	36.2	0.007	0.27	0.027	1.68	0.3109	0.0013	0.4897	
262	IB-SW1	7/19/95	7/24/95	13:58	0.054	0.024	45.7	0.0025	0.0006	0.0097	0.1229	0.215	9.92	7.843	0.0075	20.6	0.001	0.13	0.019	1.38	0.2553	0.001	0.6293	
271	IA-SW11	7/19/95	7/24/95	14:54	0.073	0.019	45.3	0.0039	0.0003	0.0098	0.1447	0.298	9.74	8.669	0.0086	21.1	0.003	0.17	0.019	1.37	0.251	0.001	0.8568	
221	IA-SW11(2)	7/19/95	7/24/95	16:52	0.056	0.022	45.3	0.003	0.0008	0.0024	0.1382	0.258	9.77	8.751	0.0048	21.1	0.006	0.14	0.012	1.38	0.2512	0.001	0.7492	
261	IA-SW11(3)	7/19/95	7/24/95	13:55	0.058	0.015	45.7	0.003	0.0006	0.0091	0.1352	0.245	9.88	8.755	0.0058	20.8	0	0.14	0.003	1.38	0.253	0.002	0.7476	
210	IA-SW1	7/19/95	7/23/95	16:17	0.042	0.016	46.3	0.0036	0.0009	0.0006	0.1309	0.195	10.1	8.184	0.0067	21.1	0.001	0.11	0.012	1.41	0.2554	0.0008	0.7346	
188	BLANK	7/19/95	7/23/95	15:02	0.03	0.008	0.001	0.0003	0.0001	0.0034	0.048	0.017	0.014	0.001	0.0007	0.005	0	0.14	0.006	0.001	0.0009	0.0032	0.0078	
186	BLANK	7/19/95	7/23/95	14:55	0.011	0.005	0.032	0.0008	0	0.0209	0.0501	0.001	0.014	0.0002	0.0023	0.007	0	0.14	0.012	0.002	0.0011	0.0035	0.0212	
189	BLANK	7/19/95	7/23/95	13:05	0.024	0.009	0.029	0.0015	0.0009	0.0195	0.0476	0.008	0.002	0.0004	0.0005	0.041	0.004	0.16	0.007	0.001	0.001	0.0035	0.0138	
190	BLANK	7/19/95	7/23/95	15:08	0.021	0.003	0.006	0.0002	0.0003	0.0198	0.049	0	0.024	0.0013	0.0009	0.031	0.001	0.13	0.005	0.002	0.0009	0.0032	0.0081	
295	BLANK	7/19/95	7/24/95	15:37	0.21	0.152	245	0.0008	0.0113	0.05	0.006	35.1	59.3	5.04	0.0538	21.0	0.016	0.22	0.008	2	2.176	0.0036	2.584	
316	IB-GW1	7/19/95	7/24/95	16:44	0.083	0.191	170	0.001	0.0062	0.0614	0.0025	104	42.9	9.615	0.1548	159	0.003	1.1	0.023	2.22	1.26	0.0095	0.1914	
369	IB-GW1	7/19/95	7/24/95	19:33	0.13	0.228	174	0.0009	0.0052	0.0636	0.0049	112	44.1	9.334	0.1672	155	0.004	1.1	0.018	2.25	1.271	0.0097	0.1942	
292	IB-GW1	7/19/95	7/24/95	15:27	0.107	0.229	173	0.0009	0.0052	0.0636	0.0049	112	44.1	9.334	0.1672	155	0.004	1.1	0.018	2.25	1.271	0.0097	0.1942	
318	IB-GW2	7/19/95	7/24/95	16:50	0.143	0.09	173	0.0009	0.0057	0.0636	0.0049	112	44.1	9.334	0.1672	155	0.004	1.1	0.018	2.25	1.271	0.0097	0.1942	
302	IB-GW2	7/19/95	7/24/95	16:02	0.055	0.012	96.7	0.0008	0.0012	0.0197	0.0232	3.55	23	2.64	0.0075	79.2	0.001	0.21	0.005	1.08	0.8377	0.0016	0.4111	
314	IA-GW2(2)	7/19/95	7/24/95	16:39	0.129	0.097	170	0.0009	0.0054	0.0644	0.0077	59.8	37	10.33	0.0923	111	0.023	0.13	0.024	1.98	1.653	0.0015	14.5	
303	IA-GW3	7/19/95	7/24/95	16:06	0.059	0.11	46.3	0.0023	0.0084	0.0197	0.0471	6.83	9.88	1.797	0.0167	24.2	0.002	0.18	0.013	1.35	0.3225	0.002	2.869	
309	IB-GW3	7/19/95	7/24/95	16:24	0.088	0.026	56.2	0.0016	0.0117	0.0296	0.0443	17.4	13.4	4.401	0.0278	29.7	0.006	0.14	0.016	1.54	0.426	0.003	2.782	
308	IB-GW3	7/19/95	7/24/95	16:21	0.07	0.027	46.4	0.0016	0.0036	0.0144	0.0513	0.903	10.2	1.601	0.0087	22	0.004	0.16	0.006	1.43	0.2887	0.002	0.8178	
373	IA-GW4	7/19/95	7/24/95	19:34	0.327	0.36	154	0.001	0.0045	0.0779	0.1377	117	36.3	15.04	0.1742	81.5	0.007	0.45	0.064	2.36	1.91	0.0156	0.2315	
327	IB-GW4	7/19/95	7/24/95	17:20	0.276	0.259	158	0.001	0.0033	0.0796	0.1356	121	37.9	15.39	0.1823	82.7	0.004	0.44	0.074	2.41	1.411	0.0125	0.2315	
311	IB-GW4	7/19/95	7/24/95	16:30	0.138	0.111	147	0.0001	0.0044	0.065	0.0039	72.6	35.5	12.91	0.1068	78.1	0.016	1.3	0.018	2.34	1.338	0.0021	6.39	
329	IB-GW4	7/19/95	7/24/95	17:25	0.156	0.117	152	0.0003	0.0041	0.0676	0.1055	79.4	36.2	13.11	0.1149	80.6	0.021	1.4	0.017	2.36	1.391	0.002	6.347	

ICAPES data from water sample analysis on 9/25/95																							
Sample Name	Sample Date	Analysis Date	Analysis Time	Al	As	Ca	Cd	Co	Cr	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Si	Sr	Ti	Zn	
USGS T107		9/25/95	13:39	0.25	-0.003	12.4	0.0168	0.012	0.0052	0.0321	0.07	2.13	0.0506	0.0137	23.8	0.03	0.02	0.025	4.04	0.0619	-0.002	0.087	
BLANK		9/25/95	13:44	0.03	-0.018	0.005	0.0006	-0.001	-0.01	-0.0024	0.008	-0.014	-0.0002	-0.001	-0.023	-0.003	0.02	0	0	-4.00E-04	-0.003	0.0024	
USGS T107		9/25/95	13:47	0.25	-0.004	12.3	0.0164	0.0127	0.0068	0.0303	0.06	2.11	0.0492	0.0131	23.3	0.033	0.04	0.022	3.98	0.0607	-0.002	0.0832	
FIELD BLANK 1	8/31/95	9/25/95	13:56	0.03	-0.022	0.013	0.0008	-0.0002	-0.0139	-0.0017	0.044	-0.033	0.0009	-0.0039	0.031	-0.001	0.04	-0.003	0.008	-4.00E-04	-0.003	0.0281	
8-IIA-SW1	8/31/95	9/25/95	13:59	0.11	-0.006	54.4	0.0026	0.0009	-0.0022	0.1669	0.409	12	1.104	0.0012	28.3	-0.001	0.15	-0.015	12.9	0.306	0.003	0.7471	
8-IIA-SWII	8/31/95	9/25/95	14:02	0.07	-0.008	54.1	0.003	-0.001	-0.0013	0.1526	0.163	11.9	1.073	0.0023	28.1	0.001	0.13	-0.006	12.7	0.3037	-0.002	0.6191	
8-IIA-SWIII	8/31/95	9/25/95	14:05	0.07	0	53.4	0.0021	0.0007	-0.0022	0.139	0.168	11.8	1.069	0.0017	28.1	0.005	0.12	-0.009	12.6	0.3025	-0.002	0.5414	
8-IIB-SW1	9/4/95	9/25/95	14:08	0.05	0.004	54.1	0.0029	-0.0012	-0.002	0.1516	0.072	12	1.041	0.0016	28.3	-0.001	0.14	-0.027	12.8	0.3114	-0.002	0.455	
8-IIB-SWII	9/4/95	9/25/95	14:10	0.06	-0.009	53.6	0.0028	-0.0002	0.0001	0.1313	0.15	11.8	1.067	0.0023	27.8	0.002	0.17	-0.003	12.8	0.3064	-0.001	0.681	
8-IIB-SWIII	9/4/95	9/25/95	14:13	0.05	0.005	53.7	0.0027	0.0003	-0.0015	0.1195	0.164	11.9	1.063	0.003	27.8	-0.001	0.14	-0.009	12.8	0.3098	-0.002	0.5443	
8-IIA-SWIII(2)	8/31/95	9/25/95	14:16	0.06	-0.006	53.9	0.0015	0	-0.0019	0.1	0.114	11.9	1.077	-0.0007	28.5	-0.002	0.14	-0.013	12.6	0.3058	-0.002	0.3782	
8-IIIC-SW1	8/31/95	9/25/95	14:19	0.06	-0.008	52	0.0041	0.0003	0.0002	0.1732	0.19	11.4	1.108	0.0026	27.4	0	0.17	-0.017	12.3	0.2898	-0.002	1.134	
8-IIIC-SWII	8/31/95	9/25/95	14:24	0.05	-0.009	52.9	0.0036	-0.0003	-0.0024	0.1773	0.183	11.6	1.126	0.0022	28.2	0	0.12	-0.002	12.6	0.2965	-0.003	1.109	
8-IIIC-SWIII(CHK)		9/25/95	14:27	0.07	0.015	53.1	0.0039	0.0009	0.0007	0.177	0.194	11.6	1.132	0.0003	28	-0.002	0.16	-0.017	12.5	0.2956	-0.002	1.156	
8-IIB-SW1 DUP		9/25/95	14:30	0.05	0.004	53.6	0.0027	-0.0002	-0.002	0.1523	0.071	11.8	1.033	0.0037	28.2	0.001	0.13	0	12.8	0.3089	-0.002	0.4519	
USGS T107		9/25/95	14:32	0.24	-0.003	12	0.0162	0.0113	0.0063	0.0272	0.052	2.03	0.0486	0.0116	22.9	0.032	0.06	0.008	3.9	0.0594	-0.002	0.0817	
IIIC-SWIII	8/31/95	9/25/95	14:36	0.05	0.004	52.1	0.0033	0	0.001	0.1662	0.149	11.3	1.101	0.0011	27.3	0	0.18	-0.007	12.3	0.2887	-0.002	1.044	
IIIC-SW1 DUP	8/31/95	9/25/95	14:39	0.06	-0.013	52	0.003	0.0002	-0.0009	0.1746	0.189	11.4	1.112	0.0028	27.7	0.002	0.17	0.008	12.4	0.2905	-0.002	1.138	
IIIC-SWII DUP	8/31/95	9/25/95	14:42	0.07	-0.005	53.3	0.0042	0.0007	0.0011	0.1777	0.184	11.5	1.132	0.0034	27.9	0.003	0.18	0.005	12.6	0.2952	-0.002	1.114	
IIA-GW2	8/31/95	9/25/95	14:45	0.06	0.214	76.6	0.0004	0.0015	0.0204	0.0153	21.5	10.5	7.788	0.0356	26.8	0.002	0.28	0.004	20.3	0.4504	-0.001	0.1661	
8-IIA-GW3	8/31/95	9/25/95	14:47	0.05	0.015	48.4	0.0003	-0.0012	-0.0061	0.0439	0.999	10.7	0.9367	0.0046	27.5	0.001	0.16	-0.001	13	0.2841	-0.002	0.3736	
8-IIA-GW4	8/31/95	9/25/95	14:50	0.06	0.136	61.2	-4E-04	0.0019	0.0214	0.007	46.3	11.3	6.033	0.067	27	0.002	1.2	0.003	18.5	0.4374	#####	0.1156	
8-IIB-GW3	9/4/95	9/25/95	14:53	0.06	0	52.9	0.0021	0.0003	-0.0018	0.1136	0.15	11.7	0.7139	0.0017	28	0.001	0.17	-0.004	12.7	0.3004	-0.002	0.567	
8-IIB-GW1	9/4/95	9/25/95	14:56	0.05	0.014	54.6	0.003	-0.0002	-0.0031	0.1171	0.17	12	1.057	0.0019	28.7	0.001	0.16	-0.014	12.8	0.3142	-0.002	1.078	
8-IIB-GW2	9/4/95	9/25/95	14:59	0.09	0.724	54.5	-3E-04	0.0019	0.0099	0.039	15.3	11.7	3.954	0.0297	27.7	0.002	0.16	0.001	12.4	0.3628	0.002	0.4714	
8-IIIC-GW4	9/4/95	9/25/95	15:02	0.05	0.06	68.6	-1E-04	0.0005	0.0197	0.0063	37.9	13.8	5.888	0.0546	33.2	0.003	0.31	0.001	16.2	0.4894	#####	0.243	
8-IIA-GW3	8/31/95	9/25/95	15:04	0.06	0.019	48.6	0.0001	0	-0.0042	0.0446	1	10.7	0.9408	0.0025	27.6	-0.002	0.17	0.017	13.1	0.2841	-0.002	0.3754	
8-IIIC-GW3	9/4/95	9/25/95	15:07	0.18	0.005	54.3	0.0042	0.0015	0.0021	0.2212	0.565	11.9	0.944	0.0029	28.4	0.001	0.25	-0.004	13.3	0.3102	0.004	1.235	
8-IIIC-GW4	9/4/95	9/25/95	15:10	0.05	0.061	53	-5E-04	0.0003	0.0047	0.0125	3.63	11.7	3.157	0.009	29	0.001	0.98	-0.009	13.9	0.3034	-0.002	0.0642	
8-IIIC-GW3	8/31/95	9/25/95	15:13	0.12	-0.012	104	0.0089	0.0355	0.0214	0.1996	16.6	24	6.738	0.0248	56.1	0.022	0.15	0.001	15.3	0.8486	-0.002	13.74	
8-IIA-SW1 DUP	8/31/95	9/25/95	15:16	0.12	-0.015	54	0.0035	0	0.003	0.1658	0.401	11.7	1.099	0.0025	27.9	0	0.18	-0.013	12.8	0.3005	0.004	0.7452	
8-IIA-SW1 DUP	8/31/95	9/25/95	15:18	0.11	0.009	53.2	0.0037	0.0007	0.0023	0.1641	0.395	11.5	1.084	0.0012	27.6	0	0.18	-0.019	12.6	0.2968	0.004	0.7352	
8-IIIC-GW4	8/31/95	9/25/95	15:21	0.08	0.242	64.8	0.001	0.0103	0.0268	0.0425	22.8	14.5	10.15	0.0463	31.5	0.004	0.26	0	14.7	0.4184	#####	0.4874	
8-IIIC-GW5	8/31/95	9/25/95	15:24	0.05	0.392	63.6	-4E-04	0.023	0.0545	0.0118	38.1	12.1	18.8	0.0634	32.5	0.002	0.12	-0.014	20.6	0.38	#####	1.75	
8-IIIC-GW2	8/31/95	9/25/95	15:27	0.07	0.459	154	0.0006	0.0046	0.0764	0.0111	89.9	40.4	20.14	0.1353	130	0	1.9	0.021	26.2	1.217	5E-04	0.0929	
8-IIIC-GW2	9/4/95	9/25/95	15:30	2.6	0.32	76	0.0029	0.0033	0.0366	1.035	68.4	14.3	9.626	0.096	32.5	0.002	0.74	0.462	19.4	0.6133	0.119	0.8309	
8-IIA-SW1 DUP	8/31/95	9/25/95	15:37	0.12	0.008	53.5	0.0037	0.0009	-0.0019	0.1665	0.406	11.7	1.092	0.0021	28.2	0.001	0.16	-0.003	12.7	0.301	0.004	0.742	
USGS T107		9/25/95	15:40	0.25	-0.002	12.2	0.0166	0.0118	0.0057	0.0272	0.056	2.05	0.0497	0.0118	23.4	0.033	0.12	0.005	3.98	0.0601	-0.002	0.0833	
USGS T117		9/25/95	15:43	0.12	-0.002	22.1	0.0025	0.0039	0.0079	0.0021	0.506	9.96	0.2342	0.0085	22.3	0.008	0.31	0.002	6.02	0.2612	#####	0.1956	

June/July 1995 Laboratory duplicates (water samples); Calculations worksheet

Sample Name	Lab Dupl.			Lab Dupl.			Lab Dupl.			Lab Dupl.			Lab Dupl.			Lab Dupl.					
	IIA-SWI	IIA-SWII	%CIIG	IIA-SWII	IIA-SWII	%CIIG	IIIA-GW4	IIIA-GW4	%CHG	IB-SWIII	IB-SWIII	%CHG	IA-GW5	IA-GW5	%CHG	IA-SWII	IA-SWII	%CIIG	IA-SWII	IA-SWII	%CIIG
Sample Date	7/6/95	7/6/95		7/6/95	7/6/95																
Analysis Date	7/23/95	7/23/95		7/24/95	7/24/95		7/24/95	7/24/95		7/24/95	7/24/95		7/24/95	7/24/95		7/24/95	7/24/95		7/24/95	7/24/95	
Analysis Time	16.14	16.23			19.17		19.34	17.20		14.21	14.26		17.52	19.37		14.30	15.19				
Al	BDL	BDL	BDL	0.065	0.091	BDL	0.327	0.276	16.915	0.08	0.069	14.765	20.6	19.9	3.457	0.059	0.063	BDL	0.073	0.079	BDL
As	BDL	BDL	BDL	0.015	0.009	BDL	0.26	0.259	0.385	0.028	0.013	BDL	0.112	0.097	14.354	0.011	0.016	BDL	0.014	0.007	BDL
Ca	42.9	42.5	0.937	41.9	41.8	0.239	154	158	2.564	45.1	45.4	0.663	85.3	84.4	1.061	45	45.9	1.980	46.4	46.6	0.430
Cd	BDL	BDL	BDL	0.0019	0.0026	BDL	0.001	0.001	BDL	0.0025	0.0026	BDL	0.0076	0.0076	BDL	0.0026	0.0028	BDL	0.0024	0.0027	BDL
Co	BDL	BDL	BDL	0.0007	0	BDL	0.0045	0.0053	BDL	0.0021	0.0006	BDL	0.0548	0.0548	0	0.0001	0.001	BDL	0.0004	0.001	BDL
Cu	0.0905	0.0898	0.776	0.1124	0.1117	0.625	0.1377	0.1356	1.537	0.1243	0.1433	14.200	0.0042	0.0042	BDL	0.1299	0.1313	1.072	0.1342	0.1363	1.553
Fe	0.26	0.255	1.942	0.223	0.234	4.814	117	121	3.361	0.276	0.299	8	125	124	0.803	0.163	0.173	5.952	0.172	0.172	0
Mg	9.04	9.01	0.332	8.71	8.12	7.011	36.3	37.9	4.313	9.66	9.72	0.619	23.1	22.3	3.524	9.66	9.66	0	9.68	9.6	0.830
Mn	0.6746	0.6696	0.744	0.6547	0.6566	0.290	15.04	15.39	2.300	0.8058	0.8665	7.259	21.07	20.89	0.858	0.8158	0.8352	2.350	0.8438	0.85	0.732
Mo	BDL	BDL	BDL	0.007	0.0042	BDL	0.1742	0.1823	4.544	0.007	0.0074	BDL	0.1764	0.1743	1.198	0.0049	0.0064	BDL	0.0072	0.0061	BDL
Na	19.1	19	0.525	19	19.2	1.047	81.5	82.7	1.462	20.3	20.9	2.913	65.1	64.5	0.926	20.7	21	1.439	21.5	21.5	0
Ni	BDL	BDL	BDL	0	-0.003	BDL	0.007	0.004	BDL	0	0.003	BDL	0.062	0.062	0	0.001	0.002	BDL	0.002	0.004	BDL
Pb	BDL	BDL	BDL	0.002	0.005	BDL	0.064	0.074	BDL	0.014	0.017	BDL	0.055	0.035	BDL	0.012	0.015	BDL	0.008	0.013	BDL
Si	1.27	1.26	0.791	1.24	1.21	2.449	2.36	2.41	2.096	1.35	1.37	1.471	4.4	4.32	1.835	1.35	1.37	1.471	1.38	1.38	0
Sr	0.229	0.2284	0.262	0.2264	0.2211	2.369	1.381	1.431	3.556	0.2493	0.2498	0.200	0.3891	0.3823	1.763	0.2489	0.2528	1.555	0.2558	0.2551	0.274
TI	BDL	BDL	BDL	0.001	0.0013	BDL	0.0156	0.0125	22.064	0.002	0.002	BDL	0.0012	0.0016	BDL	0.0013	0.002	BDL	0.001	0.0013	BDL
Zn	0.4606	0.4587	0.413	0.62	0.6042	2.581	0.228	0.2315	1.523	0.6629	0.8374	23.262	26.9	26.34	2.104	0.7249	0.7358	1.492	0.7418	0.7403	0.202
Sample Name	Lab Dupl.			Lab Dupl.			Lab Dupl.			Lab Dupl.			Lab Dupl.			Lab Dupl.					
	IA-SWII	IA-SWII	%CIIG	IA-SWII	IA-SWII	%CIIG	IA-SWII	IA-SWII	%CIIG	IA-SW1	IA-SW1	%CHG	%CHG	IIIB-GW1	IIIB-GW1	%CIIG	IIIB-SWI	IIIB-SWI	%CIIG		
Sample Date	7/24/95	7/24/95		7/24/95	7/24/95		7/24/95	7/24/95		7/24/95	7/23/95	7/23/95		7/24/95	7/24/95		7/24/95	7/23/95			
Analysis Time	16.09	16.53		17.40	18.24		19.05	19.40		19.14	17.10	16.17		19.23	15.27		19.29	16.55			
Al	0.073	0.079	BDL	0.084	0.084	0	0.091	0.088	3.352	0.081	0.052	0.042	BDL	BDL	0.13	0.107	19.409	0.106	0.057	BDL	
As	0.014	0.007	BDL	0.011	0.016	BDL	0.011	0.007	BDL	0.006	0.02	0.016	BDL	BDL	0.228	0.229	0.438	0.018	0.008	BDL	
Ca	46.4	46.6	0.430	45.1	45.2	0.221	45.7	45.4	0.659	46.4	46.7	46.3	0.860	0.216	174	173	0.576	45	45.5	1.105	
Cd	0.0024	0.0027	BDL	0.003	0.003	BDL	0.0026	0.0027	BDL	0.0027	0.0032	0.0036	BDL	BDL	0.0002	0.0009	BDL	0.0029	0.0031	BDL	
Co	0.0004	0.001	BDL	0.0004	0.0007	BDL	0.0001	0.0013	BDL	0	0.0008	0.0009	BDL	BDL	0.0061	0.0052	BDL	0	0.0003	BDL	
Cu	0.1342	0.1363	1.553	0.1306	0.1313	0.535	0.1335	0.1313	1.662	0.1261	0.1322	0.1309	0.988	3.735	0.006	0.0049	BDL	0.1177	0.1233	4.647	
Fe	0.172	0.172	0	0.169	0.177	4.624	0.186	0.187	0.536	0.201	0.193	0.195	1.031	3.030	115	112	2.643	0.312	0.3	3.922	
Mg	9.68	9.6	0.830	9.22	9.12	1.091	9.19	9.07	1.314	9.31	10.1	10.1	0	8.140	42.1	44.1	4.640	9.04	9.88	8.879	
Mn	0.8438	0.85	0.732	0.8221	0.8235	0.170	0.8346	0.8308	0.456	0.8237	0.8324	0.8184	1.696	0.646	9.415	9.334	0.864	0.8348	0.8428	0.954	
Mo	0.0072	0.0061	BDL	0.0057	0.0059	BDL	0.0056	0.0034	BDL	0.0053	0.0058	0.0067	BDL	BDL	0.1713	0.1672	2.422	0.0059	0.0079	BDL	
Na	21.5	21.5	0	20.9	21	0.477	21.3	21.1	0.943	21.4	21.4	21.1	1.412	1.412	1.59	1.55	2.548	20.1	20.4	1.481	
Ni	0.002	0.004	BDL	0.001	0.004	BDL	0.003	0.003	BDL	0.003	0.004	-0.001	BDL	BDL	0.003	0.004	BDL	0.001	0.007	BDL	
Pb	0.008	0.013	BDL	0.003	0.002	BDL	-0.004	0.004	BDL	-0.004	-0.004	-0.012	BDL	BDL	0.023	0.018	BDL	-0.007	-0.009	BDL	
Si	1.38	1.38	0	1.33	1.33	0	1.34	1.33	0.749	1.36	1.43	1.41	1.408	3.610	2.22	2.25	1.342	1.36	1.43	5.018	
Sr	0.2558	0.2551	0.274	0.2457	0.2456	0.041	0.2481	0.2453	1.135	0.2518	0.259	0.2554	1.400	1.420	1.26	1.271	0.869	0.2467	0.2564	3.856	
TI	0.001	0.0013	BDL	0.0016	0.0016	BDL	0.0013	0.002	BDL	0.002	-0.001	-0.0008	BDL	BDL	0.0095	0.0097	2.083	0.0044	0.0008	BDL	
Zn	0.7418	0.7403	0.202	0.7161	0.7145	0.224	0.7214	0.7144	0.975	0.7128	0.7409	0.7346	0.854	3.012	0.1914	0.1942	1.452	0.6469	0.6757	4.36	

Aug/Sept 1995 Laboratory duplicates (water samples): Calculations worksheet																											
Sample Name	Lab Dcpt. IA-SW1			Lab Dcpt. IA-SW1			Lab Dcpt. IIB-SW1			Lab Dcpt. IIC-SW1			Lab Dcpt. IID-SW1			Lab Dcpt. IIA-GW3			Lab Dcpt. IIA-SW1			Lab Dcpt. IIA-SW1			Lab Dcpt. IA-SW1		
	IA-SW1	IA-SW1	%CHG	IA-SW1	IA-SW1	%CHG	IIB-SW1	IIB-SW1	%CHG	IIC-SW1	IIC-SW1	%CHG	IID-SW1	IID-SW1	%CHG	IIA-GW3	IIA-GW3	%CHG	IIA-SW1	IIA-SW1	%CHG	IIA-SW1	IIA-SW1	%CHG	IA-SW1	IA-SW1	%CHG
Sample Date	9/27/95	9/27/95		9/27/95	9/27/95		9/4/95	9/4/95		8/31/95	8/31/95		8/31/95	8/31/95		9/25/95	9/25/95		8/31/95	8/31/95		9/25/95	9/25/95		8/24/95	8/24/95	
Analyte Date																9/25/95	9/25/95		8/31/95	8/31/95		9/25/95	9/25/95		8/24/95	8/24/95	
Analyte Time																14:47	15:04		14:47	15:04		15:15	15:18		15:15	15:18	
Al	0.022	0.014	BDL	0.018	0.021	BDL	0.05	0.051	BDL	0.057	0.07	BDL	0.053	0.065	BDL	0.053	0.061	BDL	0.115	0.11	4.444	0.029	0.027	BDL	0.022	0.019	BDL
Au	0.006	0.008	BDL	0.01	-0.003	BDL	0.004	0.004	BDL	0.008	0.015	BDL	0.009	-0.005	BDL	0.015	0.019	BDL	-0.015	0.009	BDL	0.005	-0.007	BDL	0.006	-0.009	BDL
Ca	57.3	56.9	0.701	56.9	56.5	0.705	54.1	53.6	0.929	52	53.1	2.093	52.9	53.3	0.753	48.4	48.6	0.412	54	53.2	1.493	55.8	57	2.128	55	54.3	1.281
Cd	0.0034	0.0045	BDL	0.0039	0.004	BDL	0.0029	0.0027	BDL	0.0041	0.0039	BDL	0.0036	0.0042	BDL	0.0003	0.0001	BDL	0.0035	0.0037	BDL	0.0024	0.0023	BDL	0.0025	0.0025	BDL
Co	0.0007	-0.0022	BDL	0.0003	-0.0007	BDL	0.0012	-0.0002	BDL	0.0003	0.0009	BDL	0.0003	0.0007	BDL	0.0012	0	BDL	0	0.0007	BDL	0.0002	0	BDL	0.0005	-0.0003	BDL
Cu	0.1788	0.1785	0.168	0.1879	0.1892	0.689	0.1516	0.1523	0.461	0.1732	0.177	2.170	0.1773	0.1777	0.225	0.0439	0.0446	1.582	0.1658	0.1641	1.031	0.1616	0.1657	2.505	0.1606	0.1576	1.886
Fe	0.047	0.051	1.163	0.151	0.153	1.316	0.072	0.071	1.399	0.19	0.194	2.083	0.183	0.184	0.545	0.999	1	0.100	0.401	0.395	1.508	0.1	0.103	2.956	0.11	0.103	6.573
Mg	12.6	12.5	0.797	12.5	12.6	0.797	12	11.8	1.681	11.4	11.6	1.739	11.6	11.5	0.866	10.7	10.7	0	11.7	11.5	1.724	12.5	12.7	1.587	12.4	12.2	1.626
Mn	1.126	1.318	0.605	1.324	1.319	0.378	1.041	1.033	0.771	1.108	1.132	2.143	1.126	1.132	0.531	0.9367	0.9408	0.437	1.099	1.084	1.374	1.141	1.164	1.996	1.118	1.102	1.441
Nb	0.0034	0.0049	BDL	0.0058	0.0063	BDL	0.0016	0.0037	BDL	0.0026	0.0003	BDL	0.0022	0.0034	BDL	0.0046	0.0025	BDL	0.0023	0.0012	BDL	0.0053	0.006	BDL	0.0047	0.0056	BDL
Na	29.1	29.1	0	28.9	29.4	1.715	28.9	28.2	0.354	27.4	28	2.166	28.2	27.9	1.070	27.5	27.6	0.363	27.9	27.6	1.081	26	26.5	1.905	25.7	25.5	0.781
Ni	0.002	0.002	BDL	0	0.003	BDL	0.001	0.001	BDL	0	-0.002	BDL	0	0.003	BDL	0.001	-0.002	BDL	0	0	BDL	0.003	0.001	BDL	0.001	0.002	BDL
Pb	0.004	0.016	BDL	0.019	-0.01	BDL	0.027	0	BDL	0.017	0.017	BDL	0.007	0.005	BDL	0.001	0.017	BDL	0.013	-0.019	BDL	0.005	0.018	BDL	0.013	0.016	BDL
Si	15.6	15.4	1.290	15.5	15.4	0.647	12.8	12.8	0	12.3	12.5	1.613	12.6	12.6	0	13	13.1	0.766	12.8	12.6	1.575	1.27	1.29	1.563	1.26	1.24	1.6
Sr	0.3303	0.3294	0.273	0.329	0.3304	0.425	0.3114	0.3089	0.806	0.2898	0.2956	1.982	0.2965	0.2952	0.439	0.2841	0.2841	0	0.3095	0.2968	1.239	0.3032	0.3106	2.411	0.3	0.2962	1.275
TI	0.0039	0.0035	BDL	0.0035	0.0035	BDL	0.002	-0.002	BDL	0.002	-0.002	BDL	0.003	-0.002	BDL	-0.002	-0.002	BDL	0.0043	0.0043	BDL	0.0011	0.0011	BDL	0.0014	0.0018	BDL
Zn	1.203	1.184	1.592	1.308	1.293	1.153	0.455	0.4519	0.684	1.134	1.136	1.921	1.109	1.114	0.450	0.3736	0.3754	0.481	0.7352	0.7352	1.351	0.7957	0.8162	2.544	0.7919	0.786	0.748

Sample Name	Compilation of all laboratory duplicates (water samples)																															
	JUN/JUL/95													AUG/SEPT/95													STATISTICS					
	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	MEAN	STAND. DEV.	MEAN+ ST. DEV	# OF DATA	STD. ERROR	95% CONF. OF MEAN	
Al			16.92	14.77	3.46			7.89	0.00	3.35			19.41						4.44	8.8	7.3	16.0	8	0.9	10.6							
Au			0.39		14.35								0.44							5.1	8.0	13.1	3	2.7	10.3							
Ca	0.94	0.24	2.56	0.66	1.06	1.98	0.43	0.22	0.66	0.86	0.22	0.58	1.10	0.70	0.71	0.93	2.09	0.75	0.41	1.49	2.13	1.28	1.0	0.7	1.7	22	0.0	1.1				
Cd																										0						
Co					0.00																				0.0			1	0.0	0.0		
Cu	0.78	0.62	1.54	14.20			1.07	1.55	0.53	1.66	0.99	3.74		4.65	0.17	0.69	0.46	2.17	0.23	1.58	1.03	2.51	1.89	2.1	3.1	5.2	20	0.2	2.4			
Fe	1.94	4.81	3.36	8.00	0.80	5.95	0.00	4.62	0.54	1.03	3.03	2.64	3.92	8.16	1.32	1.40	2.08	0.54	0.10	1.51	2.96	6.57	3.0	2.5	5.4	22	0.1	3.2				
Mg	0.33	7.01	4.31	0.62	3.52	0.00	0.83	1.09	1.31	0.00	8.14	4.64	8.88	0.80	0.80	1.68	1.74	0.87	0.00	1.72	1.59	1.63	2.3	2.6	5.0	22	0.1	2.6				
Mn	0.74	0.29	2.30	7.26	0.86	2.35	0.73	0.17	0.46	1.70	0.65	0.86	0.95	0.61	0.38	0.77	2.14	0.53	0.44	1.37	2.00	1.44	1.3	1.5	2.8	22	0.1	1.5				
Nb					4.54									2.42											2.7	1.7	4.4	3	0.6	3.8		
Na	0.52	1.05	1.46	2.91	0.93	1.44	0.00	0.48	0.94	1.41	1.41	2.55	1.48	0.00	1.72	0.35	2.17	1.07	0.36	1.08	1.90	0.78	1.2	0.8	2.0	22	0.0	1.3				
Ni					0.00																				0.0			1	0.0	0.0		
Pb																									0							
Si	0.79	2.45	2.10	1.47	1.83	1.47	0.00	0.00	0.75	1.41	3.61	1.34	5.02	1.29	0.65	0.00	1.61	0.00	0.77	1.57	1.56	1.60	1.4	1.2	2.6	22	0.1	1.5				
Sr	0.26	2.37	3.56	0.20	1.76	1.55	0.27	0.04	1.13	1.40	1.42	0.87	3.86	0.27	0.42	0.81	1.98	0.44	0.00	1.24	2.41	1.27	1.3	1.1	2.3	22	0.0	1.3				
TI					22.06									2.08											12.1	14.1	26.2	2	7.1	25.9		
Zn	0.41	2.58	1.52	23.26	2.10	1.49	0.20	0.22	0.98	0.85	3.01	1.45	4.36	1.59	1.15	0.68	1.92	0.45	0.48	1.35	2.54	0.75	2.4	4.8	7.2	22	0.2	2.9				

June/July 1995 Field duplicates (water samples): Calculations worksheet																				
Sample Name	Field Dupl.			Lab Dupl.	Lab Dupl.			AVG.	Field Dupl.			AVG.	Field Dupl.			Field Dupl.				
	IC GW2	IC GW2(2)	% CHG		IA SW1	IA SW1	IA SW1		IA SW1	IA-SW1(2)	% CHG		IA SW1	IA-SW1(2)	% CHG	IIA-GW1	IIA-GW1(2)	% CHG	IIA-GW1	IIA-GW1(2)
Sample Date	6/28/95	6/28/95		7/19/95	7/19/95	7/19/95	7/19/95	7/19/95	7/19/95	7/19/95		7/19/95	7/19/95		7/24/95	7/24/95		7/24/95	7/24/95	
Analysis Date	7/5/95	7/5/95		7/24/95	7/23/95	7/23/95	7/23/95	7/23/95	7/23/95	7/23/95		(of lab dupl.s)	7/24/95		7/24/95	7/24/95		7/24/95	7/24/95	
Analysis Time	11:07	11:10		19:14	17:10	16:17		16:46				13:55		15:41	18:56					
Al	0.038	0.026	37.50	BDL	0.081	0.052	0.042	0.0583	0.051	13.41	BDL	0.0728	0.058	22.63	BDL	0.067	0.063	6.15	DL	
As	0.13	0.108	18.49		0.006	0.02	0.016	0.014	0.022	44.44	BDL	0.0128	0.015	15.83	BDL	0.04	0.039	2.53	BDL	
Ca	44.8	41.8	6.93		46.4	46.7	46.3	46.467	46.4	0.14		45.64	45.7	0.13		42.6	43.1	1.17		
Cd	0.0013	0	200	BDL	0.0027	0.0032	0.0036	0.0032	0.0034	7.11	BDL	0.00281	0.003	6.54	BDL	0.0042	0.0051	19.35	BDL	
Co	0.0025	0.0012	70.27	BDL	0	0.0008	0.0009	0.0006	0.0009	45.45	BDL	0.00067	0.0006	3628.57	BDL	0.0018	0.001	57.14	BDL	
Cu	0.0039	0.0015	119	BDL	0.1261	0.1322	0.1309	0.1297	0.1282	2.76		0.13301	0.1352	1.63		0.1335	0.1454	8.53		
Fe	29	25.4	13.24		0.201	0.195	0.195	0.1963	0.145	30.08		0.1805	0.245	30.32		0.741	0.663	11.11		
Mg	8.8	8.09	8.41		9.31	10.1	10.1	9.8367	10.1	2.64		9.487	9.88	4.06		8.48	8.24	2.87		
Mn	4.206	3.966	5.87		0.8237	0.8324	0.8184	0.8248	0.8257	0.11		0.83076	0.8235	0.88		0.9934	0.9379	5.75		
Mo	0.0472	0.042	11.66		0.0053	0.0058	0.0067	0.0059	0.0056	5.78	BDL	0.00577	0.0058	0.52	BDL	0.0072	0.004	57.14	BDL	
Ni	18.6	18	3.28		21.4	21.4	21.1	21.3	21.6	1.40		21.08	20.8	1.34		19.1	19.4	1.56		
Nb	0.003	0.003	0.00	BDL	0.003	0.004	-0.001	0.002	0.006	100.00	BDL	0.0027	0	200.00	BDL	0.002	-0.001	600.00	BDL	
Pb	0.009	0.003	100	BDL	-0.004	-0.004	0.012	-0.007	-0.01	-40.00	BDL	0.0031	-0.003	12200.00	BDL	0.017	0.003	140.00	BDL	
Si	1.87	1.79	4.37		1.36	1.43	1.41	1.4	1.41	0.71		1.357	1.38	1.68		1.26	1.24	1.60		
Sr	0.273	0.2449	10.85		0.2518	0.259	0.2554	0.2554	0.2598	1.71		0.25804	0.253	1.18		0.2233	0.2234	0.04		
Ti	0.0014	0.0018	25.00	BDL	0.002	-0.001	-0.0008	0.000	-0.0026	210.53	BDL	0.00132	0.002	40.96	BDL	0.002	0.0012	50.00	BDL	
Zn	0.0381	0.0415	8.54				0.7346	0.7346	0.7043	4.21		0.72945	0.7476	2.46		0.8734	1.069	20.14		
D.G.	1.4	2.8	66.67									7.7	6	24.82						
pH	6.73	6.82	1.33									7.58	7.58	0.00						
Cond.	0.425	0.443	4.15										0.402							
Alk.	164	228	32.65									144	176	20.00						
Cl												11.689	12.264	4.80						
NO3 N	0.15	0.335	76.29	DL								1.098	1.136	3.40						
PO4 P	0	0.295	200.00	BDL								72.702	72.688	0.02						
SO4	19.179	30.548	45.73																	

Sample Name	Field Dupl.			Field Dupl.			IIA-GW1	IIA-GW1(2)			IIIB-GW3	IIIB-GW3(2)		
	IIIA-GW2	IIIA-GW2(2)	% CHG	IIA-SW1	IIA-SW1(2)	% CHG		IIA-GW1	IIA-GW1(2)	% CHG		IIIB-GW3	IIIB-GW3(2)	% CHG
Sample Date	7/10/95	7/10/95		7/6/95	7/6/95		7/10/95	7/10/95		7/11/95	7/11/95			
Analysis Date	7/24/95	7/24/95		7/24/95	7/24/95		7/24/95	7/24/95		7/24/95	7/24/95			
Analysis Time	16:50	0.69375		14:40	14:47		15:37	15:37						
Al	0.143	0.129	10.29		0.057	0.055	3.57	BDL						
As	0.09	0.097	7.49		0.012	0.012	0.00	BDL						
Ca	173	170	1.75		41.8	41.7	0.24							
Cd	0.0009	0.0009	0.00	BDL	0.0021	0.0019	10.00	BDL						
Co	0.0557	0.0534	4.22		0.0004	0.0012	100.00	BDL						
Cu	0.007	-0.0077	-9.52	BDL	0.1061	0.0913	14.99							
Fe	57.4	59.8	4.10		0.302	0.176	52.72							
Mg	37.4	37	1.08		8.73	8.68	0.57							
Mn	10.39	10.33	0.58		0.6948	0.6948	0.00							
Mo	0.0862	0.0923	6.83		0.0064	0.0056	13.33	BDL						
Ni	112	111	0.90		19	18.9	0.53							
Nb	0.026	0.023	12.24		0.002	0.001	66.67	BDL						
Pb	0.003	0.024	155.36	BDL	0.009	0.015	50.00	BDL						
Si	1.98	1.98	0.00		1.22	1.21	0.82							
Sr	1.667	1.653	0.84		0.2244	0.2239	0.22							
Ti	0.0021	0.0015	33.33	BDL	0.001	0.001	0.00	BDL						
Zn	14.89	14.5	2.65		0.5747	0.5249	9.06							
D.O.														
pH														
Cond.														
Alk.														
Cl														
NO3 N							72.96	78.76	7.65	13.89	11.95	15.02		
PO4 P							BDL	BDL		0.262	0.405	42.88		
SO4							BDL	BDL						
							1136.25	1100	3.24	233.34	176.7	27.63		

Aug/Sept 1995 Field duplicates (water samples): Calculations worksheet																		
Sample Name Sample Date	Field Dupl.			Field Dupl.			Field Dupl.			Field Dupl.			Field Dupl.			M. Xing		
	IB-GW6 8/27/95 8/30/95	IB-GW6(2) 8/27/95 8/30/95	%CHG	IA-SWIII 8/24/95 8/30/95	IA-SWIII(2) 8/24/95 8/30/95	%CHG	IA-GW3 8/24/95 8/30/95	IA-GW3(2) 8/24/95 8/30/95	%CHG	IA-GW6 8/24/95 8/30/95	IA-GW6(2) 8/24/95 8/30/95	%CHG	IIA-SWIII 8/31/95	IIA-SWIII(2) 8/31/95	%CHG	M. Xing 11/3/95	M. Xing (2) 11/3/95	%CHG
Al	2.5	1.54	47.525	0.028	0.023	BDL	10.4	9.63	7.688	23.8			0.07	0.062	BDL	0.006	0.015	BDL
Au	0.088	0.112	24	0	0	BDL	0.005	0.014	BDL	0.069			0	-0.006	BDL	0.001	0.016	BDL
Cu	124	123	0.810	55.3	55.2	0.181	111	111	0	108			53.4	53.9	0.932	54.1	54.5	0.737
Cd	0.0186	0.0176	5.525	0.0032	0.0031	BDL	0.1165	0.1406	18.748	BDL			0.0021	0.0015	BDL	0.0023	0.0027	BDL
Co	0.05	0.0529	5.637	0.0006	0.0005	BDL	0.0452	0.0432	4.525	0.0526			0.0007	0	BDL	0.0002	0.0024	BDL
Cm	0.0101	0.0057	BDL	0.1455	0.1798	21.088	2.833	6.706	81.203	0.0141			0.139	0.1	32.636	0.0874	0.0867	0.804
Fe	319	312	2.219	0.087	0.096	9.836	93.2	89.1	4.498	146			0.168	0.114	38.298	0.051	0.044	14.737
Mg	37	37.6	1.609	12.4	12.3	0.810	29.9	29.9	0	30.8			11.8	11.9	0.844	12.4	12.2	1.626
Mn	26.76	27.21	1.668	1.117	1.163	4.035	20.52	20.23	1.423	25.37			1.069	1.077	0.746	0.863	0.873	1.152
Mo	0.4447	0.4358	2.022	0.007	0.0068	BDL	0.1296	0.1238	4.578	0.2036			0.0017	-7.00E-04	BDL	0.0059	0.0061	BDL
Na	81.8	84.7	3.483	25.6	26	1.550	65.7	65.8	0.152	71.7			28.1	28.5	1.413	34.6	24.3	1.227
NI	0.052	0.044	16.667	0.004	0.001	BDL	0.044	0.045	2.247	0.058			0.005	-0.002	BDL	0.001	0.001	BDL
Pb	0.057	0.041	32.653	0.006	0.011	BDL	0.367	0.37	0.814	BDL			0.009	-0.013	BDL	0.01	0.009	BDL
SI	3.62	3.66	1.099	1.26	1.26	0	3.3	3.25	1.527	4.9			12.6	12.6	0	13.8	13.7	0.727
Sr	0.4846	0.4837	0.186	0.3015	0.3	0.499	0.5427	0.543	0.055	0.463			0.3025	0.3058	1.085	0.2815	0.284	0.176
TI	0.0016	0.0019	BDL	0.0008	0.0011	BDL	0.0013	0.0054	BDL	BDL			0.002	-0.002	BDL	0.0035	0.0035	BDL
Zn	9.933	10.71	7.528	0.6584	1	41.196	16.65	17.12	2.784	27.17			0.5414	0.3782	35.494	0.9635	0.8558	11.840
pH	5.35	5.38	0.559	6.84	6.64	2.967	4.45	4.39	1.357	3.98	4.09	2.726	8.16	8.19	0.367	7.72	7.72	0
D.O.	2.7	4	38.806	7.3	7	4.196	1.7	2.1	21.053	4.6	4.5	2.198		8.3		7.5	7.5	0
Cond.	1.95	1.94	0.514	0.489	0.492	0.612	1.36	1.35	0.738	1.79	1.63	9.357	0.504	0.511	1.379	0.501	0.482	3.866
Alk.	36	36	0.000	192	148	25.882	NA	NA		NA	NA		124	128	3.175	172	152	12.346
Cl	29.033	28.816	0.750	16.908	17.124	1.269	19.501	19.697	1.000	22.17			17.5	17.8	1.700	15.22	15.28	0.393
NO3-N	BDL	BDL	BDL	1.673	1.615	3.528	BDL	BDL	BDL	BDL			1.68	1.74	3.509	1.401	1.469	4.739
PO4-P	0.5	0.55	9.524	BDL	BDL	BDL	BDL	BDL	BDL	BDL								
SO4	1257.03	1276.41	1.530	97.2	94.2	3.135	820.1	828.1	0.971	723.02			102.7	104.6	1.833	85.86	86.93	1.238

Compilation of all field duplicates (water samples)												STATISTICS			
	JUNE/JULY 1995						AUGUST/SEPTEMBER 1995						STD	# OF DATA	95% CONF. OF MEAN
	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	%CHG	MEAN	DEV			
pH	1.3						0.559	1.357	2.726	0.367	1.2	1.2	8	1.3	
D.O.	66.667						38.806	21.053	2.198	BDL	18.7	21.5	7	22.9	
Cond.	4.147						0.514	0.738	9.357	1.379	2.8	2.9	7	3.4	
Alk.	32.653						0.000			3.175	16.9	10.9	6	19.3	
Cl						7.646	15.015	0.750	1.000	1.700	3.5	4.3	9	4.2	
NO3-N	76.289					BDL	42.879	BDL	BDL	3.509	16.2	26.0	6	21.9	
PO4-P	200					BDL	BDL	9.524	BDL	BDL	104.8	134.7	2	236.8	
SO4	45.726					3.242	27.627	1.530	0.971	1.833	7.7	13.6	10	9.7	
Al	BDL	BDL	6.154	10.294	BDL		47.525	7.688		BDL	17.9	19.8	4	27.6	
Au	18.487	BDL	BDL	7.487	BDL		24.000	BDL		BDL	16.7	8.4	3	22.2	
Cu	6.928	0.144	1.167	1.749	0.2		0.810	0.000		0.932	1.0	1.8	11	1.3	
Cd	BDL	BDL	BDL	BDL	BDL		5.525	18.748		BDL	12.1	9.3	2	21.3	
Co	BDL	BDL	BDL	4.22	BDL		5.637	4.525		BDL	4.8	0.7	3	6.3	
Cm	BDL	2.761	8.534	BDL	14.995		BDL	81.203		32.636	17.0	23.8	8	21.3	
Fe	13.235	30.078	1.111	4.096	52.72		2.219	4.498		38.298	19.0	14.0	11	21.1	
Mg	8.4	2.642	2.87	1.08	0.574		1.609	0.000		0.844	2.2	2.2	11	2.8	
Mn	5.874	0.1	5.7	0.6	0.00		1.668	1.423		0.746	2.0	2.0	11	2.3	
Mo	11.659	BDL	BDL	6.83	BDL		2.022	4.578		BDL	6.3	4.1	4	8.3	
Na	3.3	1.40	1.558	0.90	0.528		3.483	0.152		1.413	1.5	0.9	11	1.6	
NI	BDL	BDL	BDL	12.245	BDL		16.667	2.247		BDL	10.4	7.4	3	16.2	
Pb	BDL	BDL	BDL	BDL	BDL		32.653	0.814		BDL	16.7	22.5	2	38.8	
SI	4.372	0.712	1.6	0	0.8		1.099	1.527		0.000	1.1	1.1	11	1.2	
Sr	10.852	1.708	0.045	0.843	0.223		0.186	0.055		1.085	1.3	2.8	11	1.7	
TI	BDL	BDL	BDL	BDL	BDL		BDL	BDL		BDL	#DIV/0!	#DIV/0!	0	#DIV/0!	
Zn	8.54	4.2	20.14	2.654	9.058		7.528	2.784		35.494	14.4	14.4	11	16.4	

Summary of mean (+/- standard deviation) blank concentrations for water and bead analyses .

Element	Water field blanks (mg/L) n=11	Milli-Q lab blanks run with water samples (mg/L) n = 23	Bead Digest Blanks (ug/ g bead) n=13	Bead Digest Blanks without beads (mg/L) n=11	Milli-Q lab blanks run with bead samples (mg/L) n=32
Al	(<0.07)	(<0.07)	32.13 +/- 29.4	(<0.07)	(<0.07)
As	(<0.07)	(<0.07)	(<0.07)	(<0.07)	(<0.07)
Ca	(<0.1)	(<0.1)	5.55 +/- 4.11	(<0.1)	(<0.1)
Cd	(<0.01)	(<0.01)	(<0.01)	(<0.01)	(<0.01)
Co	(<0.01)	(<0.01)	(<0.01)	(<0.01)	(<0.01)
Cu	(<0.01)	(<0.01)	(<0.01)	(<0.01)	(<0.01)
Fe	(<0.03)	(<0.03)	0.24 +/- 0.25	(<0.03)	(<0.03)
Mg	(<0.1)	(<0.1)	1.32 +/- 0.94	(<0.1)	(<0.1)
Mn	(<0.005)	(<0.005)	0.06 +/- 0.04	(<0.005)	(<0.005)
Mo	(<0.01)	(<0.01)	(<0.01)	(<0.01)	(<0.01)
Na	(<0.1)	(<0.1)	0.96 +/- 0.74	(<0.1)	(<0.1)
Ni	(<0.02)	(<0.02)	(<0.02)	(<0.02)	(<0.02)
P	(<0.2)	(<0.2)	(<0.2)	(<0.2)	(<0.2)
Pb	(<0.1)	(<0.1)	(<0.1)	(<0.1)	(<0.1)
Si	(<0.1)	(<0.1)	6.47 +/- 3.93	(<0.1)	(<0.1)
Sr	(<0.005)	(<0.005)	0.03 +/- 0.02	(<0.005)	(<0.005)
Ti	(<0.005)	(<0.005)	0.09 +/- 0.06	(<0.005)	(<0.005)
Zn	0.018 +/- 0.017	(<0.005)	(<0.005)	(<0.005)	(<0.005)

n= number of samples; numbers in parentheses are the detection limits of the ICAPES.

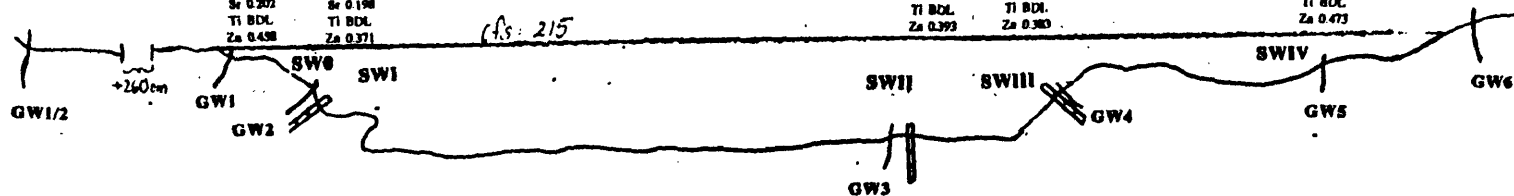
Fe QA/QC: low vs. high concentrations			
Conc. range	% change btw. duplicates	Conc. range	% change btw. duplicates
319-312	2.2	0.087-.096	9.8
93.2-89.1	4.5	0.168-0.114	38.3
29-25.4	4.1	0.051-0.044	14.7
57.4-59.8	13.2	0.196-0.145	30.1
		0.181-0.245	30.3
		0.741-0.663	11.11
		0.302-0.176	52
avg	6.0	avg	26.6
std dev.	4.9	stdev	14.5
# of data	4	# of data	7
std.error	1.2	std.error	2.1
95% conf. mean	8.4	95% conf. mean	30.7

SILVER BOW CREEK
TRANSECT IA
6/20/95

1A- SW0	1A- SW1
pH 7.3	pH 7.3
D.O. 11	D.O. 10
Cond. 0.324	Cond. 0.304
Alk.	Alk.
Temp. 14.8	Temp. 14.9
Cl 10.13	Cl 7.4
NO3-N 0.55	NO3-N 0.33
SO4 46.8	SO4 46.8
Al BDL	Al BDL
As BDL	As BDL
Ca 36	Ca 35
Cd BDL	Cd BDL
Co BDL	Co BDL
Cu 0.114	Cu 0.090
Fe 0.46	Fe 0.35
Mg 7.7	Mg 7.5
Mn 0.478	Mn 0.471
Mo BDL	Mo BDL
Na 17	Na 16
Ni BDL	Ni BDL
Pb BDL	Pb BDL
Si 14	Si 14
Sr 0.202	Sr 0.198
Tl BDL	Tl BDL
Zn 0.458	Zn 0.371

1A- SWII	1A- SWIII
pH 7.5	pH 7.4
D.O. 9.2	D.O. 9.1
Cond. 0.307	Cond. 0.321
Alk.	Alk.
Temp. 14.8	Temp. 14.9
Cl 7.6	Cl 7.5
NO3-N 0.37	NO3-N 0.37
SO4 46.8	SO4 46.9
Al BDL	Al BDL
As BDL	As BDL
Ca 35	Ca 35
Cd BDL	Cd BDL
Co BDL	Co BDL
Cu 0.096	Cu 0.097
Fe 0.34	Fe 0.33
Mg 7.7	Mg 7.7
Mn 0.444	Mn 0.445
Mo BDL	Mo BDL
Na 17	Na 17
Ni BDL	Ni BDL
Pb BDL	Pb BDL
Si 14	Si 14
Sr 0.202	Sr 0.201
Tl BDL	Tl BDL
Zn 0.393	Zn 0.380

1A- SWIV
pH 7.1
D.O. 8.1
Cond. 0.308
Alk.
Temp. 14.8
Cl 7.6
NO3-N 0.37
SO4 48.5
Al 0.10
As BDL
Ca 35
Cd BDL
Co BDL
Cu 0.121
Fe 0.46
Mg 7.6
Mn 0.524
Mo BDL
Na 17
Ni BDL
Pb BDL
Si 14
Sr 0.198
Tl BDL
Zn 0.473



1A- GW1/2	1A- GW1	1A- GW2
pH 5.9	pH 6.9	pH 7.0
D.O. 8.0	D.O. 2.4	D.O. 10
Cond. 0.474	Cond. 1.040	Cond. 0.262
Alk.	Alk.	Alk.
Temp. 13.4	Temp. 11.7	Temp. 11.7
Cl 7.5	Cl 36	Cl 7.5
NO3-N BDL	NO3-N BDL	NO3-N 0.55
SO4 144	SO4 1.98E+02	SO4 47.1
Al 0.14	Al BDL	Al 0.08
As BDL	As 0.80	As BDL
Ca 41	Ca 79	Ca 34
Cd BDL	Cd BDL	Cd BDL
Co BDL	Co BDL	Co BDL
Cu 0.180	Cu 0.026	Cu 0.113
Fe 3.6	Fe 89	Fe 0.40
Mg 7.3	Mg 14	Mg 7.5
Mn 8.15	Mn 15.4	Mn 0.447
Mo BDL	Mo 0.19	Mo BDL
Na 22	Na 34	Na 16
Ni BDL	Ni BDL	Ni BDL
Pb BDL	Pb BDL	Pb BDL
Si 16	Si 18	Si 14
Sr 0.189	Sr 0.34	Sr 0.198
Tl BDL	Tl BDL	Tl BDL
Zn 13	Zn 0.151	Zn 0.398

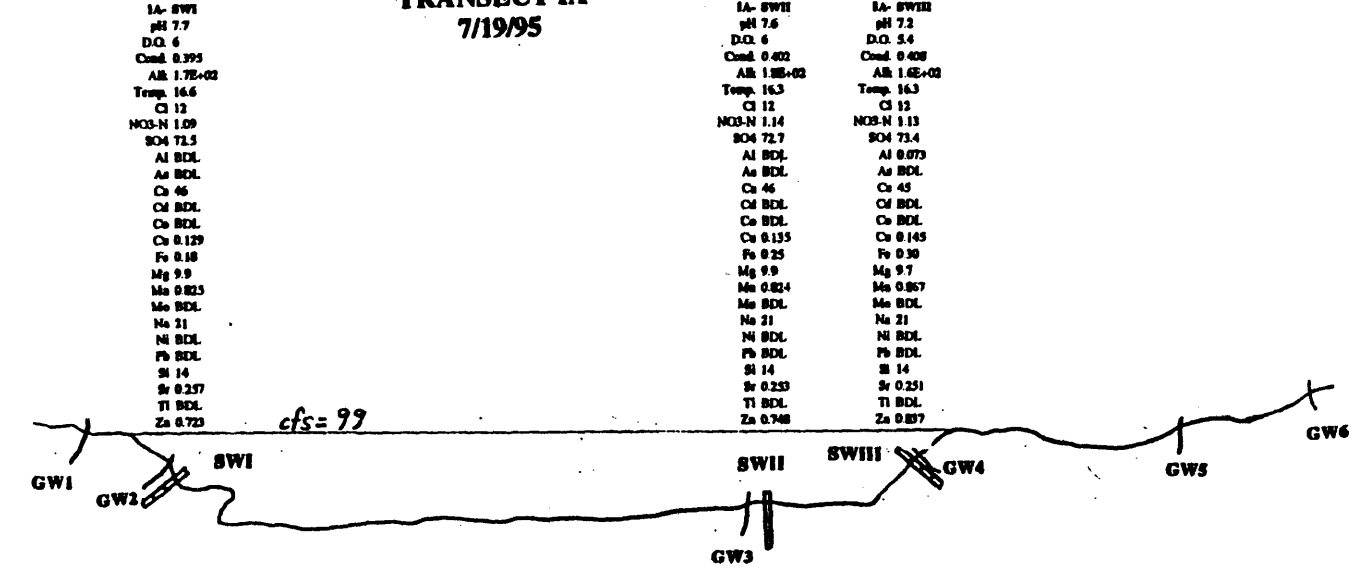
1A- GW3	1A- GW4
pH 6.2	pH 6.3
D.O. 5.9	D.O. 4.3
Cond. 0.490	Cond. 1.41
Alk.	Alk.
Temp. 11.0	Temp. 10.5
Cl 10	Cl 10
NO3-N 0.34	NO3-N BDL
SO4 172	SO4 93.0
Al 0.17	Al BDL
As BDL	As 0.7
Ca 51	Ca 85
Cd 0.023	Cd BDL
Co BDL	Co 0.044
Cu 0.602	Cu BDL
Fe 7.2	Fe 4.0E+02
Mg 12	Mg 23
Mn 4.43	Mn 35.4
Mo 0.014	Mo 0.86
Na 28	Na 64
Ni BDL	Ni BDL
Pb BDL	Pb BDL
Si 16	Si 21
Sr 0.277	Sr 0.446
Tl BDL	Tl BDL
Zn 4.14	Zn 134

1A- GW5
pH 4.5
D.O. 3.8
Cond. 0.967
Alk.
Temp. 10.5
Cl 9.8
NO3-N 0.32
SO4 4.2E+02
Al 3.7
As BDL
Ca 63
Cd BDL
Co 0.081
Cu 0.045
Fe 54
Mg 15
Mn 10.6
Mo 0.079
Na 35
Ni 0.009
Pb BDL
Si 23
Sr 0.305
Tl BDL
Zn 13.5

1A- GW6
pH 2.7
D.O. 5.4
Cond. 2.32
Alk.
Temp. 10.8
Cl 21.12
NO3-N BDL
SO4 3.8E+02
Al 20
As BDL
Ca 107
Cd 0.173
Co 0.06
Cu 13.7
Fe 1.1E+02
Mg 29
Mn 25.7
Mo 0.17
Na 75
Ni 0.06
Pb BDL
Si 51
Sr 0.257
Tl 0.015
Zn 32.4

1 Meter

SILVER BOW CREEK TRANSECT IA 7/19/95



1A- SWI

pH	7.7
D.O.	6
Cond.	0.395
Alk	1.7E+02
Temp.	16.6
Cl	12
NO3-N	1.09
SO4	72.5
Al	BDL
As	BDL
Cd	46
Cf	BDL
Co	BDL
Cu	0.129
Fo	0.18
Mg	9.9
Mn	0.825
Mo	BDL
Na	21
Ni	BDL
Pb	BDL
Si	14
Sr	0.257
Tl	BDL
Zn	0.723

1A- SWII

pH	7.6
D.O.	6
Cond.	0.402
Alk	1.8E+02
Temp.	16.3
Cl	12
NO3-N	1.14
SO4	72.7
Al	BDL
As	BDL
Cd	46
Cf	BDL
Co	BDL
Cu	0.135
Fo	0.25
Mg	9.9
Mn	0.824
Mo	BDL
Na	21
Ni	BDL
Pb	BDL
Si	14
Sr	0.253
Tl	BDL
Zn	0.748

1A- SWIII

pH	7.2
D.O.	5.4
Cond.	0.408
Alk	1.6E+02
Temp.	16.3
Cl	13
NO3-N	1.13
SO4	73.4
Al	0.073
As	BDL
Cd	45
Cf	BDL
Co	BDL
Cu	0.145
Fo	0.30
Mg	9.7
Mn	0.867
Mo	BDL
Na	21
Ni	BDL
Pb	BDL
Si	14
Sr	0.251
Tl	BDL
Zn	0.837

1A- GW1

pH	6.7
D.O.	1.3
Cond.	0.768
Alk	3.4E+02
Temp.	15.3
Cl	10
NO3-N	0.1
SO4	37.8
Al	BDL
As	0.26
Cd	68
Cf	BDL
Co	BDL
Cu	BDL
Fo	77
Mg	12
Mn	0.57
Mo	0.13
Na	29
Ni	BDL
Pb	BDL
Si	16
Sr	0.496
Tl	BDL
Zn	0.664

1A- GW2

pH	6.8
D.O.	1.2
Cond.	0.910
Alk	5.7E+02
Temp.	14.7
Cl	9.0
NO3-N	BDL
SO4	28.9
Al	BDL
As	0.36
Cd	86
Cf	BDL
Co	BDL
Cu	BDL
Fo	73
Mg	15
Mn	0.96
Mo	0.11
Na	30
Ni	BDL
Pb	BDL
Si	18
Sr	0.63
Tl	BDL
Zn	0.872

1A- GW3

pH	5.8
D.O.	2.7
Cond.	1.12
Alk	1.1E+02
Temp.	13.8
Cl	19
NO3-N	0.31
SO4	7.3E+02
Al	1.4
As	BDL
Cd	105
Cf	0.111
Co	0.036
Cu	1.81
Fo	36
Mg	26
Mn	16.3
Mo	0.082
Na	60
Ni	0.042
Pb	BDL
Si	23
Sr	0.538
Tl	BDL
Zn	13.3

1A- GW4

pH	6.4
D.O.	1.6
Cond.	1.81
Alk	7.3E+02
Temp.	13.7
Cl	25
NO3-N	BDL
SO4	4.1E+02
Al	0.072
As	0.77
Cd	64
Cf	BDL
Co	BDL
Cu	BDL
Fo	4.2E+02
Mg	30
Mn	28.6
Mo	0.61
Na	81
Ni	BDL
Pb	BDL
Si	22
Sr	0.343
Tl	BDL
Zn	0.592

1A- GW5

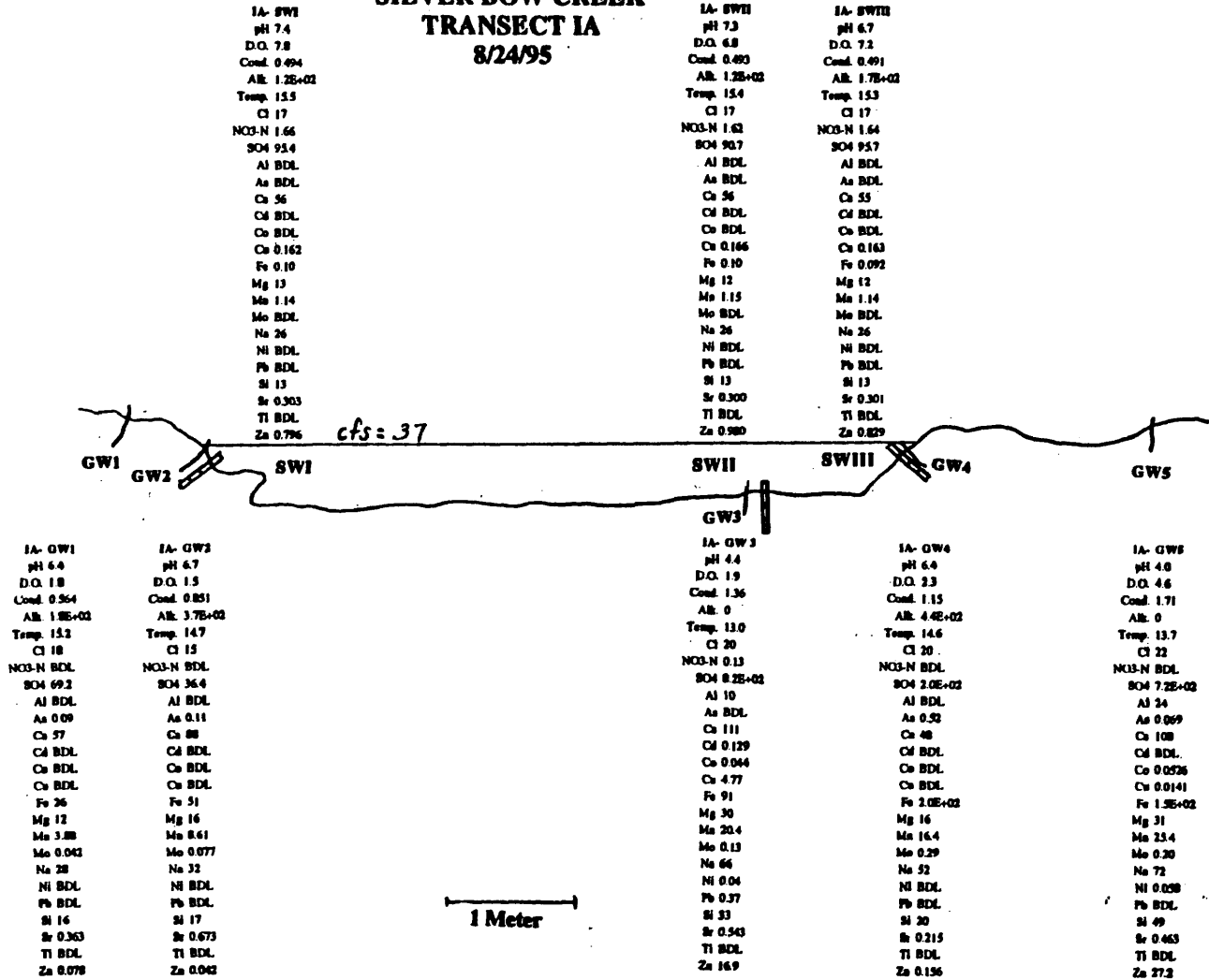
pH	4.8
D.O.	1.8
Cond.	1.53
Alk	0
Temp.	14.3
Cl	14
NO3-N	BDL
SO4	1.2E+03
Al	21
As	0.11
Cd	83
Cf	BDL
Co	0.055
Cu	BDL
Fo	1.3E+02
Mg	23
Mn	31.1
Mo	0.18
Na	63
Ni	0.06
Pb	BDL
Si	44
Sr	0.589
Tl	BDL
Zn	26.9

1A- GW6

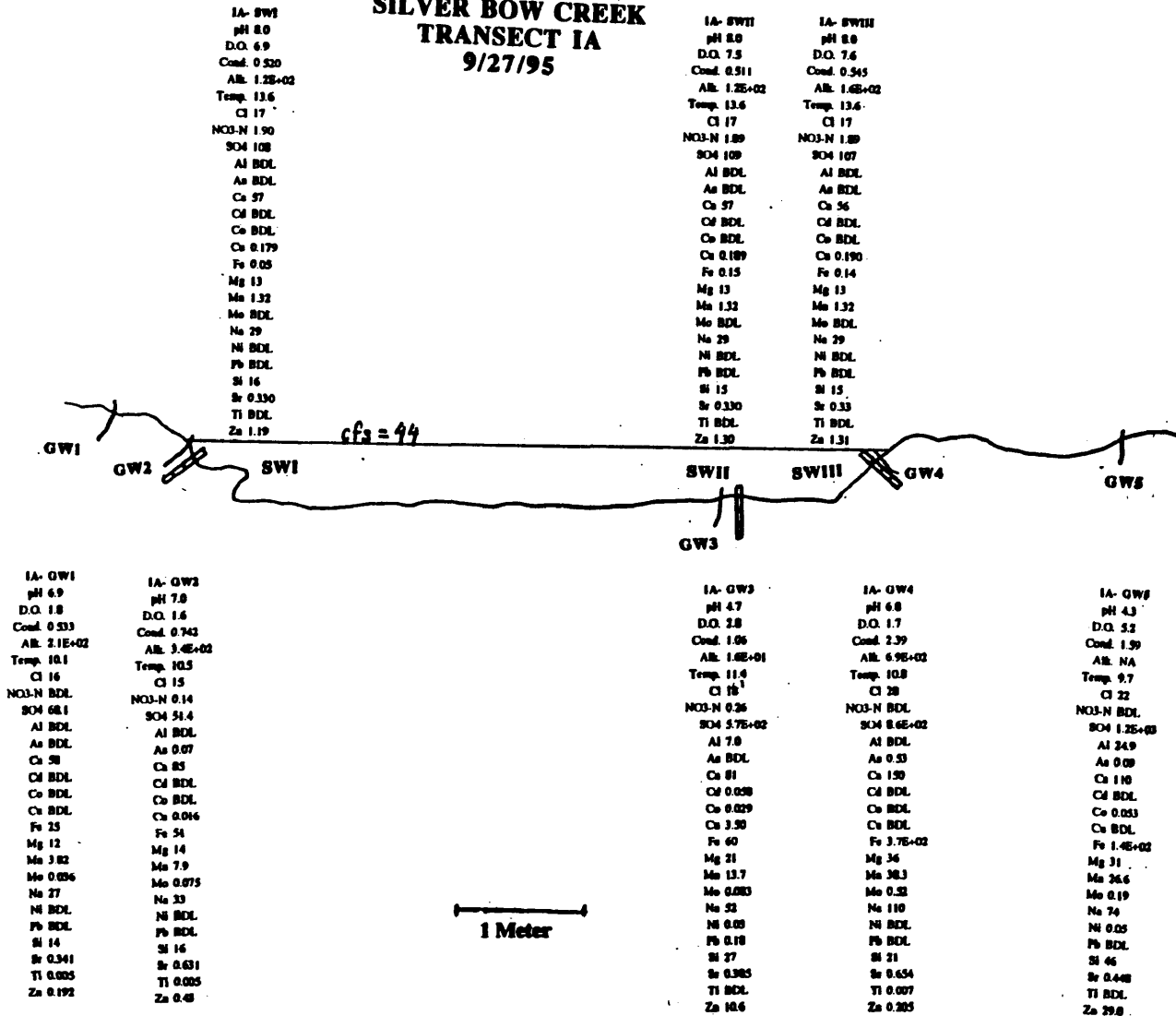
pH	2.6
D.O.	5.2
Cond.	3.15
Alk	0
Temp.	14.8
Cl	28
NO3-N	BDL
SO4	1.2E+03
Al	22
As	BDL
Cd	140
Cf	0.269
Co	0.071
Cu	15.4
Fo	1.1E+02
Mg	37
Mn	37.0
Mo	0.16
Na	100
Ni	0.08
Pb	BDL
Si	54
Sr	0.469
Tl	0.015
Zn	44

1 Meter

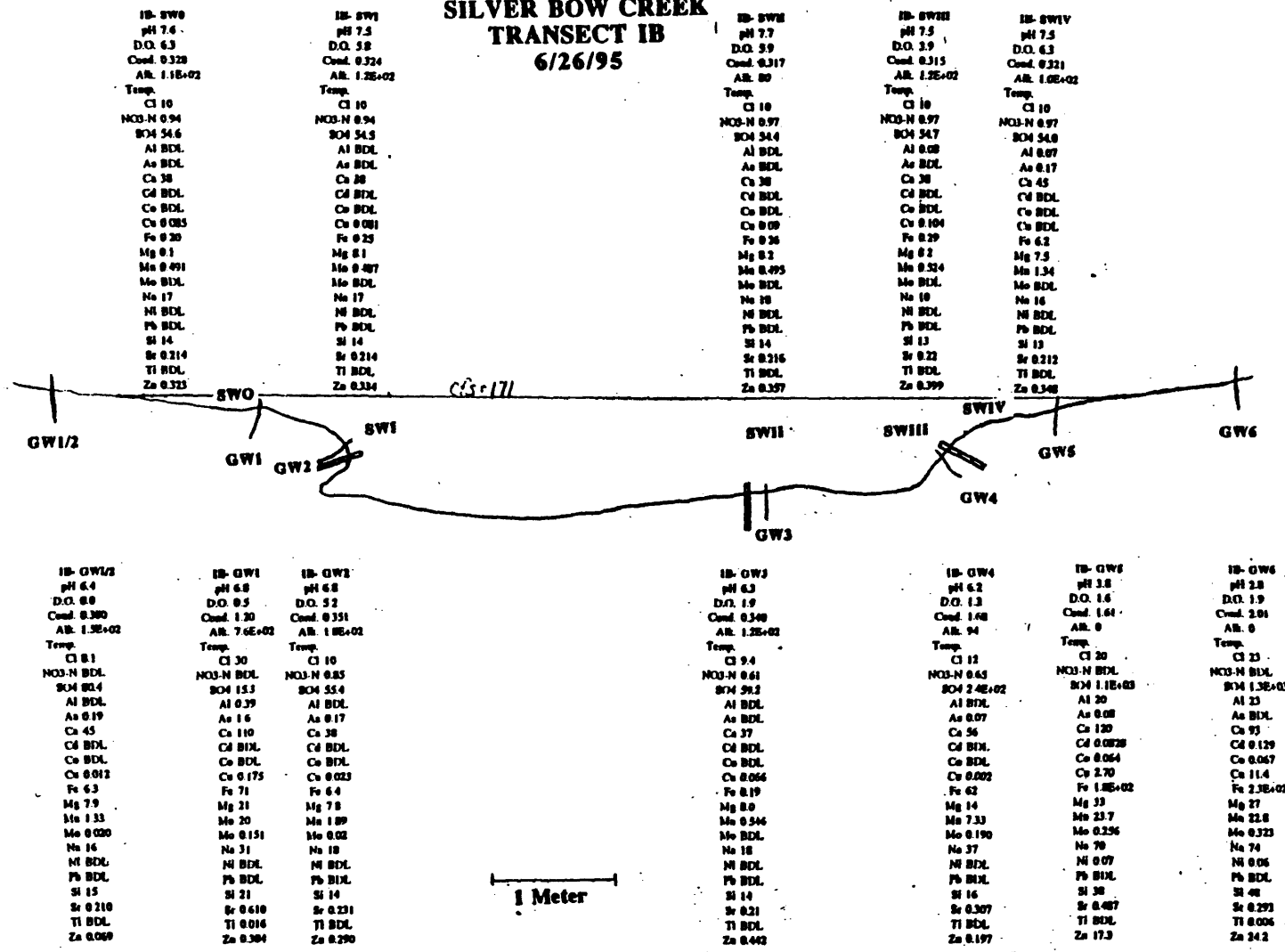
**SILVER BOW CREEK
TRANSECT IA
8/24/95**



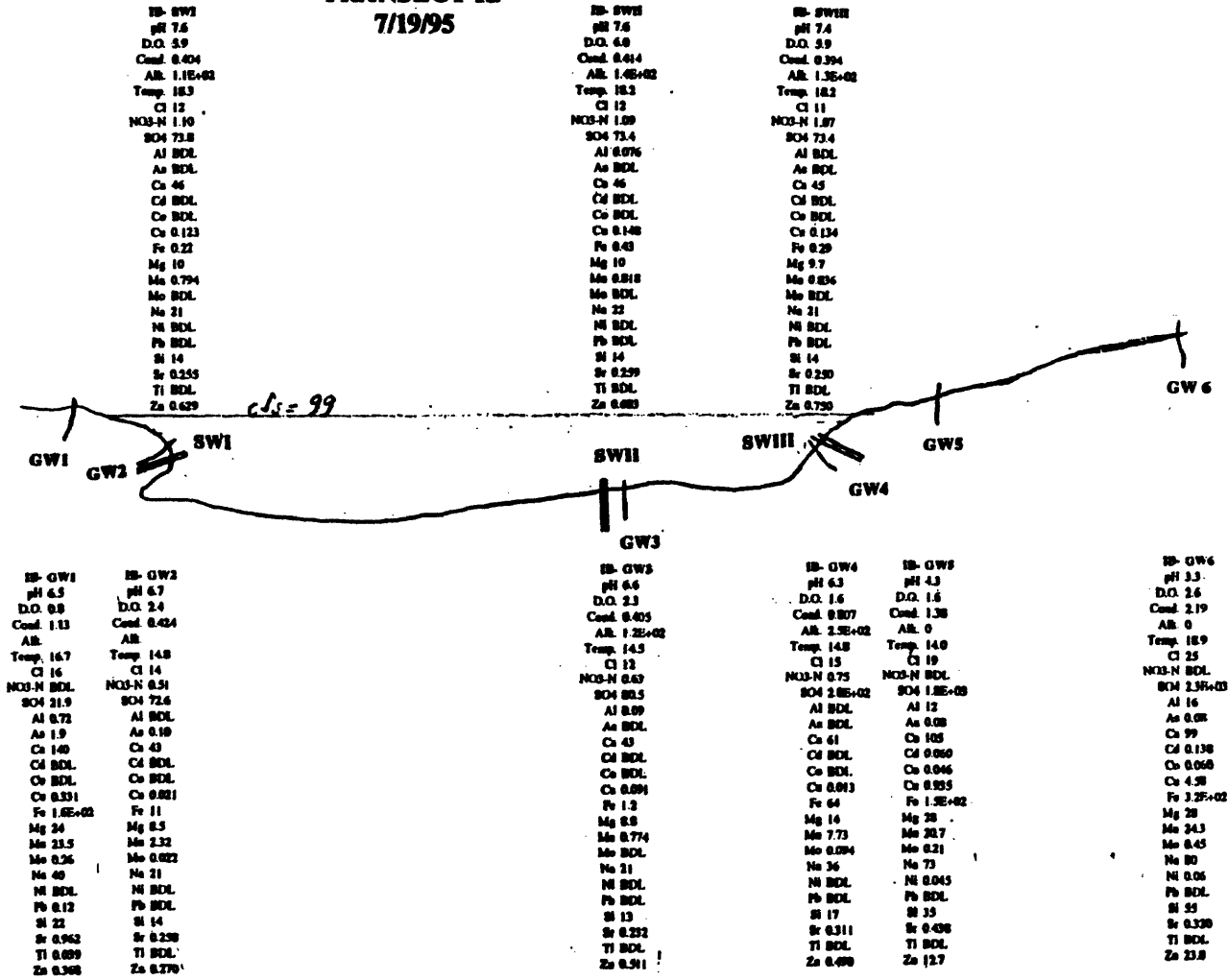
SILVER BOW CREEK TRANSECT IA 9/27/95

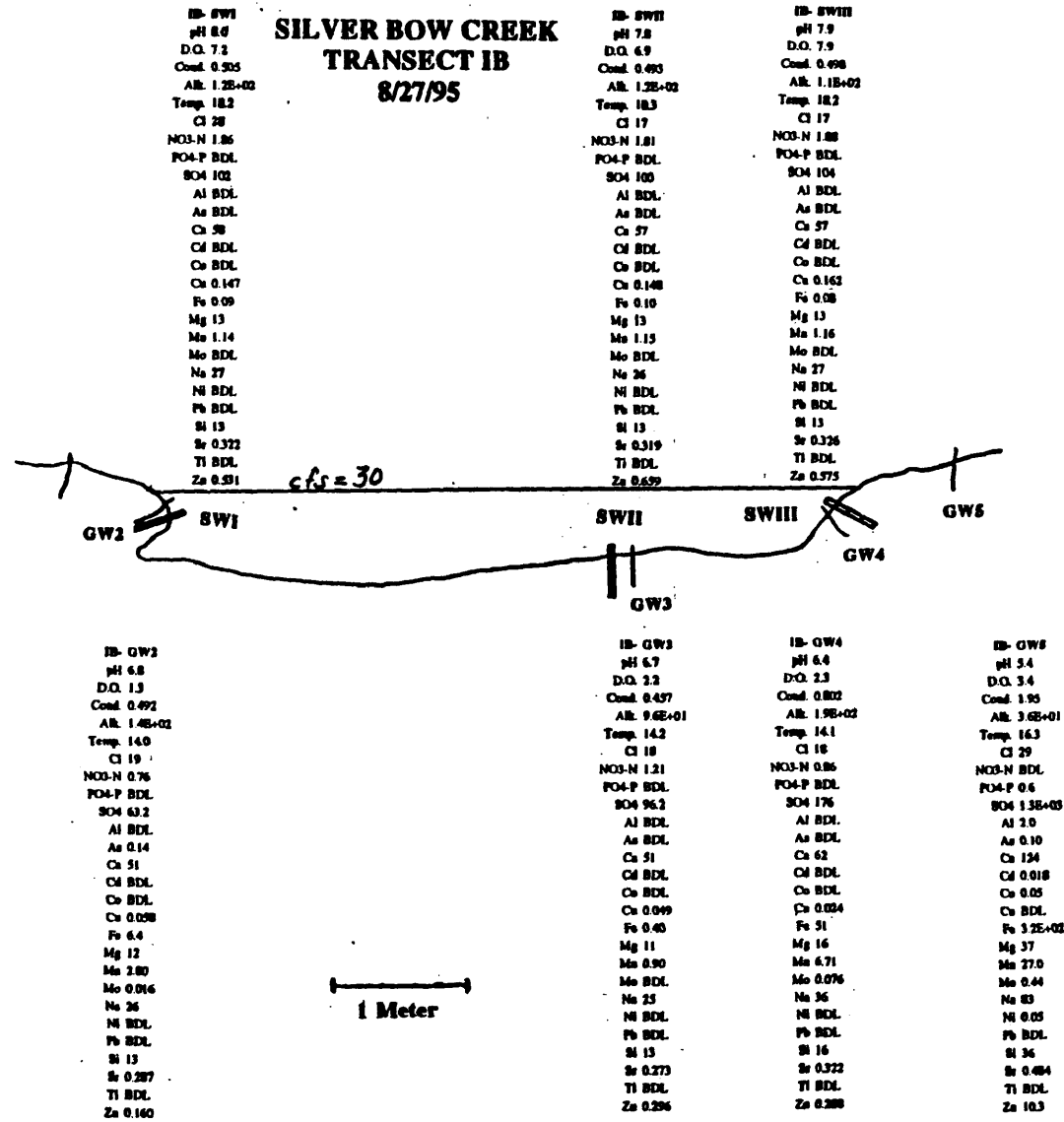


**SILVER BOW CREEK
TRANSECT IB
6/26/95**

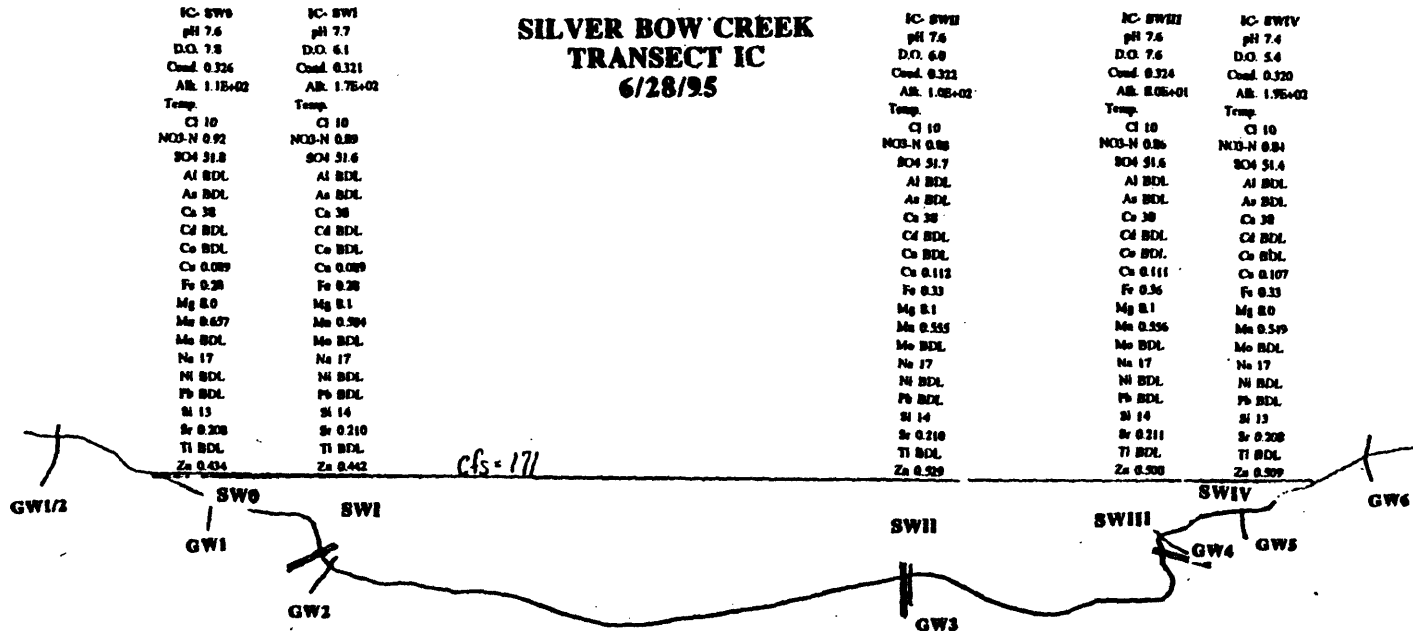


**SILVER BOW CREEK
TRANSECT IB
7/19/95**





SILVER BOW CREEK
TRANSECT IC
6/28/95



IC-SW0	IC-SW1
pH 7.6	pH 7.7
D.O. 7.8	D.O. 6.1
Cond. 0.326	Cond. 0.321
AR. 1.18+02	AR. 1.78+02
Temp.	Temp.
Cl 10	Cl 10
NO3-N 0.92	NO3-N 0.89
SO4 31.8	SO4 31.6
Al BDL	Al BDL
As BDL	As BDL
Cd 38	Cd 38
Cd BDL	Cd BDL
Co BDL	Co BDL
Cu 0.089	Cu 0.089
Fe 0.28	Fe 0.28
Mg 8.0	Mg 8.1
Mn 0.637	Mn 0.390
Mo BDL	Mo BDL
Ni 17	Ni 17
Ni BDL	Ni BDL
Pb BDL	Pb BDL
Si 13	Si 14
Sr 0.208	Sr 0.210
Tl BDL	Tl BDL
Zn 0.434	Zn 0.442

IC-SW11
pH 7.6
D.O. 6.0
Cond. 0.322
AR. 1.08+02
Temp.
Cl 10
NO3-N 0.88
SO4 31.7
Al BDL
As BDL
Cd 38
Cd BDL
Co BDL
Cu 0.112
Fe 0.33
Mg 8.1
Mn 0.585
Mo BDL
Ni 17
Ni BDL
Pb BDL
Si 14
Sr 0.210
Tl BDL
Zn 0.329

IC-SW12
pH 7.6
D.O. 7.6
Cond. 0.324
AR. 0.05+01
Temp.
Cl 10
NO3-N 0.86
SO4 31.6
Al BDL
As BDL
Cd 38
Cd BDL
Co BDL
Cu 0.111
Fe 0.36
Mg 8.1
Mn 0.526
Mo BDL
Ni 17
Ni BDL
Pb BDL
Si 14
Sr 0.211
Tl BDL
Zn 0.500

IC-SW13
pH 7.4
D.O. 5.4
Cond. 0.320
AR. 1.95+02
Temp.
Cl 10
NO3-N 0.84
SO4 31.4
Al BDL
As BDL
Cd 38
Cd BDL
Co BDL
Cu 0.107
Fe 0.33
Mg 8.0
Mn 0.549
Mo BDL
Ni 17
Ni BDL
Pb BDL
Si 13
Sr 0.208
Tl BDL
Zn 0.309

IC-GW12	IC-GW1	IC-GW2
pH 6.8	pH 6.8	pH 6.8
D.O. 1.9	D.O. 1.9	D.O. 2.1
Cond. 0.434	Cond. 0.434	Cond. 0.434
AR. 2.05+02	AR. 2.05+02	AR. 2.05+02
Temp.	Temp.	Temp.
Cl 22	Cl 22	Cl 9.9
NO3-N BDL	NO3-N BDL	NO3-N BDL
SO4 1.88	SO4 1.88	SO4 24.9
Al 0.19	Al 0.19	Al BDL
As 0.15	As 0.15	As 0.12
Cd 120	Cd 120	Cd 43
Cd BDL	Cd BDL	Cd BDL
Co BDL	Co BDL	Co BDL
Cu 0.0413	Cu 0.0413	Cu BDL
Fe 1.8E+02	Fe 1.8E+02	Fe 27
Mg 22	Mg 22	Mg 8.4
Mn 13.1	Mn 13.1	Mn 4.09
Mo 0.253	Mo 0.253	Mo 0.045
Ni 40	Ni 40	Ni 18
Ni 0.064	Ni 0.05	Ni BDL
Pb BDL	Pb BDL	Pb BDL
Si 24	Si 24	Si 10
Sr 1.49	Sr 1.49	Sr 0.259
Tl 0.012	Tl 0.012	Tl BDL
Zn 0.159	Zn 0.159	Zn 0.048

IC-GW3
pH 7.1
D.O. 3.3
Cond. 0.312
AR. 1.2E+02
Temp.
Cl 13
NO3-N 0.44
SO4 48.3
Al BDL
As BDL
Cd 37
Cd BDL
Co BDL
Cu 0.096
Fe 5.2
Mg 7.7
Mn 1.42
Mo 0.014
Ni 17
Ni BDL
Pb BDL
Si 15
Sr 0.199
Tl BDL
Zn 0.570

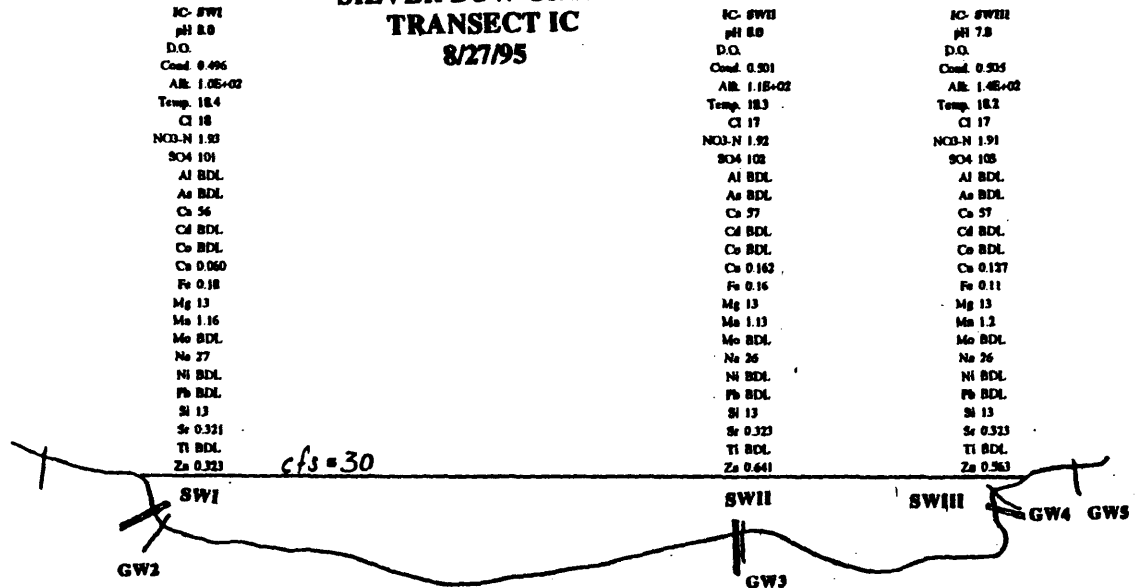
IC-GW4
pH 6.5
D.O. 2.6
Cond. 0.296
AR. 1.2E+02
Temp.
Cl 16
NO3-N 0.16
SO4 53.3
Al BDL
As BDL
Cd 35
Cd BDL
Co BDL
Cu 0.093
Fe 7.8
Mg 7.3
Mn 0.834
Mo 0.057
Ni 18
Ni BDL
Pb BDL
Si 13
Sr 0.184
Tl BDL
Zn 0.110

IC-GW5
pH 6.1
D.O. 2.4
Cond. 0.307
AR. 31
Temp.
Cl 10
NO3-N 0.63
SO4 64.3
Al 0.26
As BDL
Cd 31
Cd BDL
Co BDL
Cu 0.143
Fe 1.6
Mg 7.1
Mn 0.428
Mo BDL
Ni 18
Ni BDL
Pb BDL
Si 11
Sr 0.144
Tl 0.006
Zn 0.226

IC-GW6
pH 2.7
D.O. 4.5
Cond. 2.13
AR. 0
Temp.
Cl 7
NO3-N
SO4
Al 31
As BDL
Cd 180
Cd 0.118
Cu 0.099
Fe 29
Mg 29
Mn 55.0
Mo 0.049
Ni 43
Ni 0.073
Pb BDL
Si 35
Sr 0.347
Tl 0.017
Zn 29.3

1 Meter

**SILVER BOW CREEK
TRANSECT IC
8/27/95**



IC- SWI
pH 8.0
D.O.
Cond. 0.496
AR. 1.0E+02
Temp. 18.4
Cl 18
NO3-N 1.93
SO4 101
Al BDL
As BDL
Ca 56
Cd BDL
Co BDL
Cu 0.060
Fe 0.18
Mg 13
Mn 1.16
Mo BDL
Na 27
Ni BDL
Pb BDL
Si 13
Sr 0.321
Ti BDL
Zn 0.323

IC- SWII
pH 8.0
D.O.
Cond. 0.501
AR. 1.1E+02
Temp. 18.3
Cl 17
NO3-N 1.92
SO4 108
Al BDL
As BDL
Ca 57
Cd BDL
Co BDL
Cu 0.162
Fe 0.16
Mg 13
Mn 1.13
Mo BDL
Na 26
Ni BDL
Pb BDL
Si 13
Sr 0.323
Ti BDL
Zn 0.641

IC- SWIII
pH 7.8
D.O.
Cond. 0.505
AR. 1.4E+02
Temp. 18.2
Cl 17
NO3-N 1.91
SO4 108
Al BDL
As BDL
Ca 57
Cd BDL
Co BDL
Cu 0.127
Fe 0.11
Mg 13
Mn 1.3
Mo BDL
Na 26
Ni BDL
Pb BDL
Si 13
Sr 0.323
Ti BDL
Zn 0.563

IC- GW2
pH 6.7
D.O.
Cond. 0.761
AR. 3.4E+02
Temp. 14.2
Cl 16
NO3-N 0.14
SO4 31.9
Al BDL
As 0.30
Ca 68
Cd BDL
Co BDL
Cu BDL
Fe 55
Mg 13
Mn 7.30
Mo 0.082
Na 27
Ni BDL
Pb BDL
Si 19
Sr 0.473
Ti BDL
Zn 0.063

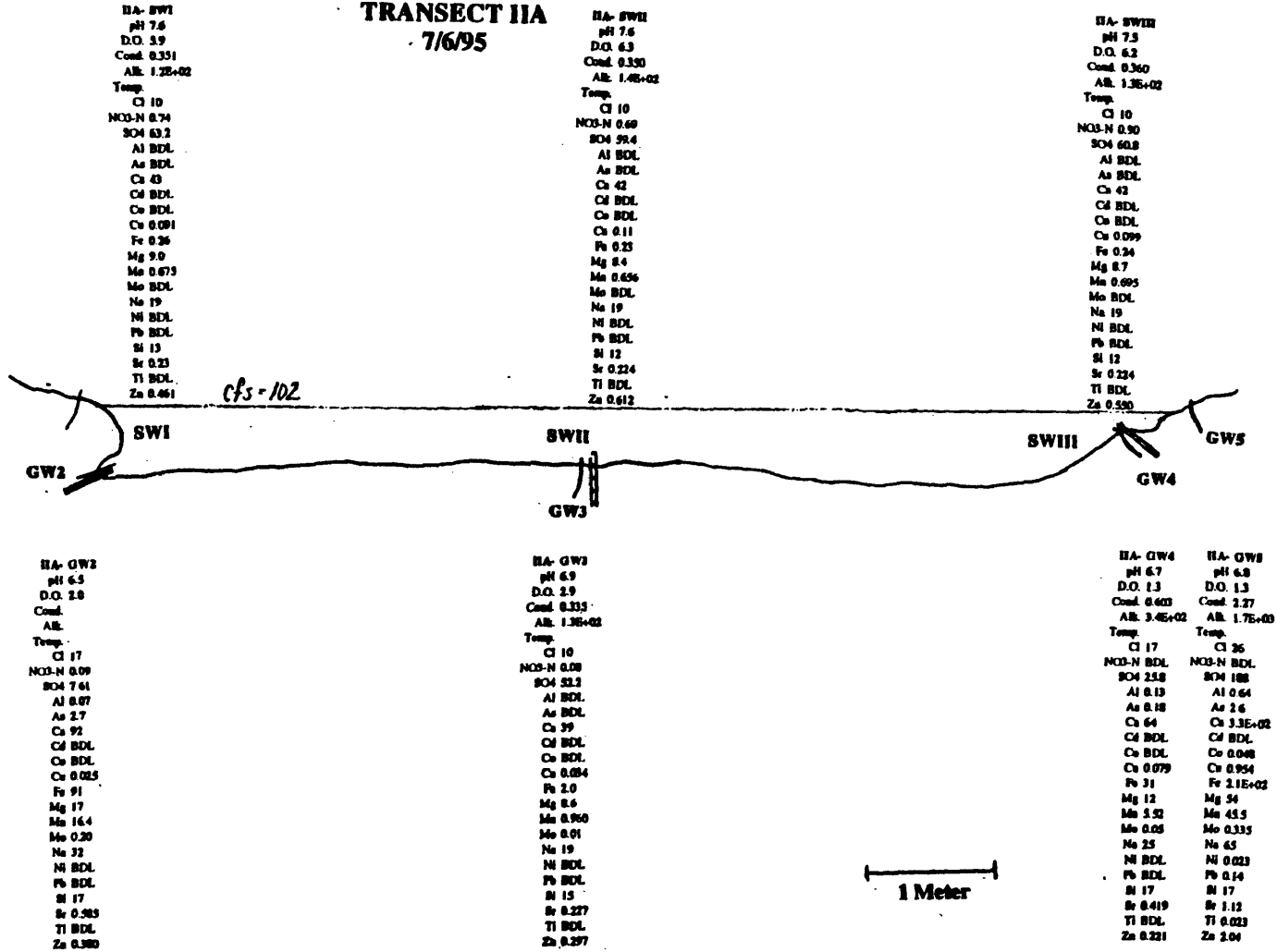
IC- GW3
pH 6.8
D.O.
Cond. 0.515
AR. 1.4E+02
Temp. 14.0
Cl 17
NO3-N 0.08
SO4 62.1
Al BDL
As 0.10
Ca 52
Cd BDL
Co BDL
Cu 0.092
Fe 10
Mg 12
Mn 3.79
Mo 0.021
Na 26
Ni BDL
Pb BDL
Si 15
Sr 0.298
Ti BDL
Zn 0.164

IC- GW4
pH 6.6
D.O.
Cond. 0.364
AR. 1.4E+02
Temp. 14.4
Cl 17
NO3-N 0.14
SO4 143
Al BDL
As BDL
Ca 64
Cd BDL
Co BDL
Cu 0.053
Fe 9.4
Mg 14
Mn 3.24
Mo 0.019
Na 28
Ni BDL
Pb BDL
Si 16
Sr 0.357
Ti 0.0051
Zn 0.360

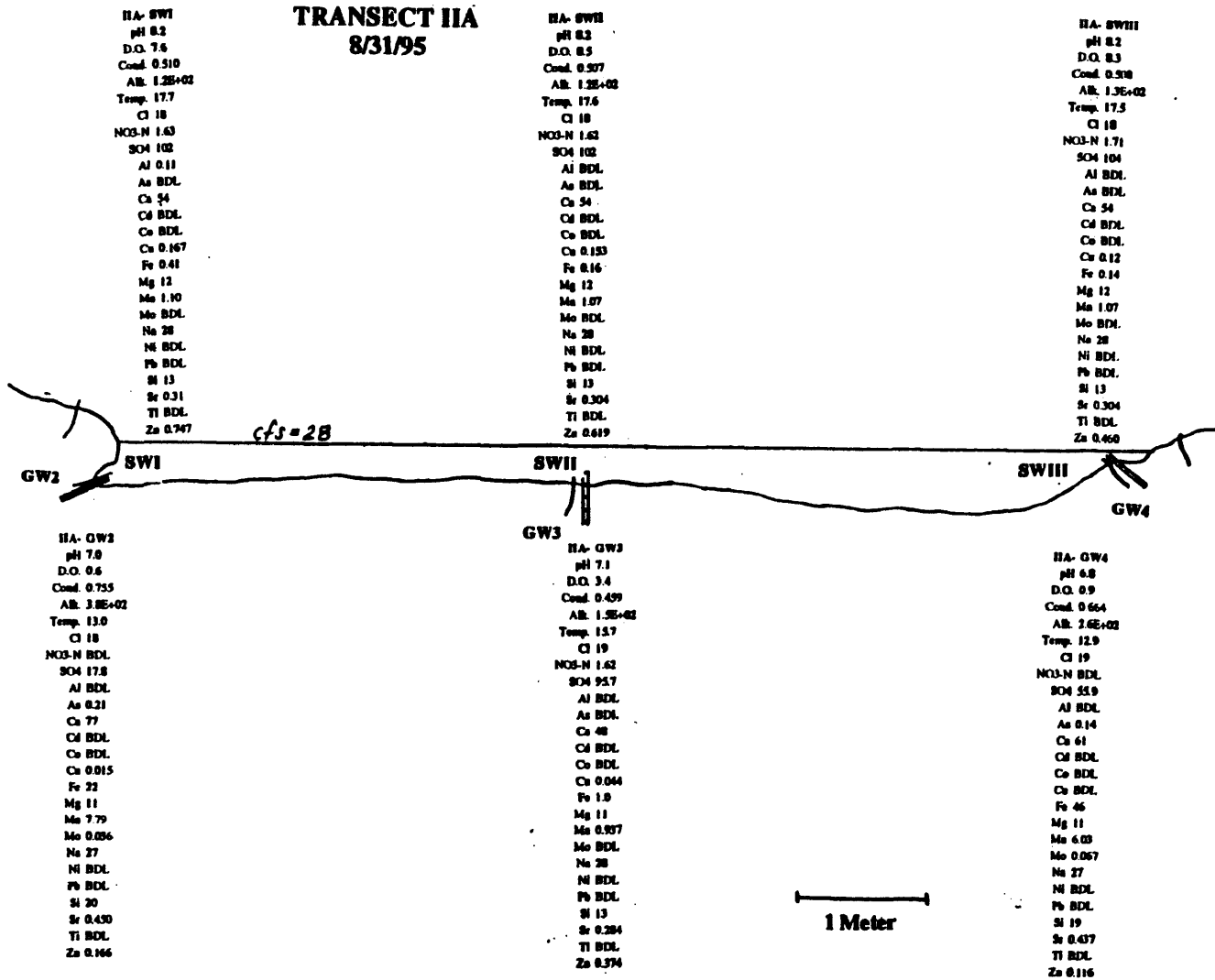
IC- GW5
pH 4.5
D.O.
Cond. 1.01
AR. 0
Temp. 16.0
Cl 17
NO3-N 0.68
SO4 6.9E+02
Al 9.78
As BDL
Ca 79
Cd 0.073
Co 0.020
Cu 8.38
Fe 70
Mg 20
Mn 16.0
Mo 0.098
Na 33
Ni 0.046
Pb BDL
Si 38
Sr 0.32
Ti BDL
Zn 187

1 Meter

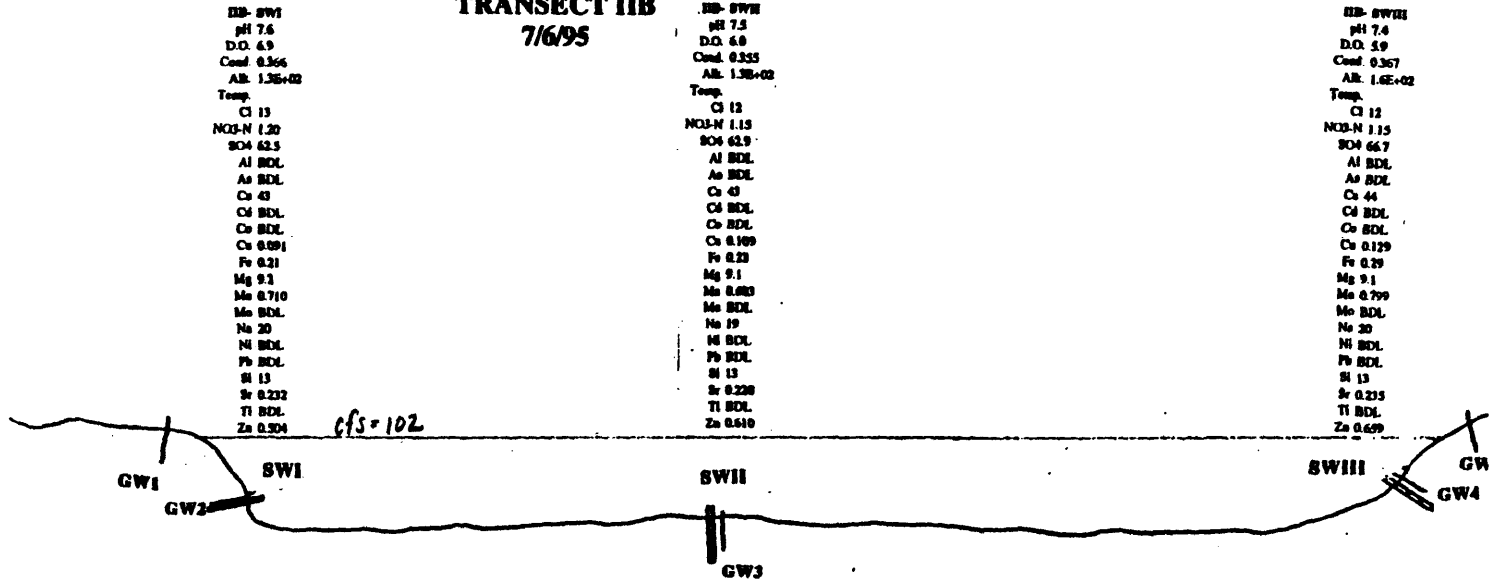
**SILVER BOW CREEK
TRANSECT IIA
7/6/95**



**SILVER BOW CREEK
TRANSECT IIA
8/31/95**



**SILVER BOW CREEK
TRANSECT IIB
7/6/95**



IBB- SWI
pH 7.6
D.O. 6.9
Cond. 0.366
Alk. 1.35E+02
Temp.
Cl 13
NO3-N 1.20
SO4 62.5
Al BDL
As BDL
Cr 43
Cd BDL
Co BDL
Cu 0.091
Fe 0.21
Mg 9.1
Mn 0.710
Mo BDL
Na 20
Ni BDL
Pb BDL
Si 13
Sr 0.232
Ti BDL
Zn 0.304

IBB- SWII
pH 7.3
D.O. 6.8
Cond. 0.325
Alk. 1.38E+02
Temp.
Cl 12
NO3-N 1.15
SO4 62.9
Al BDL
As BDL
Cr 43
Cd BDL
Co BDL
Cu 0.109
Fe 0.23
Mg 9.1
Mn 0.683
Mo BDL
Na 19
Ni BDL
Pb BDL
Si 13
Sr 0.228
Ti BDL
Zn 0.610

IBB- SWIII
pH 7.4
D.O. 5.9
Cond. 0.367
Alk. 1.68E+02
Temp.
Cl 12
NO3-N 1.15
SO4 66.7
Al BDL
As BDL
Cr 44
Cd BDL
Co BDL
Cu 0.129
Fe 0.29
Mg 9.1
Mn 0.799
Mo BDL
Na 20
Ni BDL
Pb BDL
Si 13
Sr 0.233
Ti BDL
Zn 0.699

IBB- GW1
pH 6.6
D.O. 1.9
Cond. 0.336
Alk. 1.48E+02
Temp.
Cl 23
NO3-N 0.46
SO4 63.8
Al BDL
As BDL
Cr 43
Cd BDL
Co BDL
Cu 0.139
Fe 0.70
Mg 8.4
Mn 0.966
Mo BDL
Na 19
Ni BDL
Pb BDL
Si 13
Sr 0.223
Ti BDL
Zn 0.971

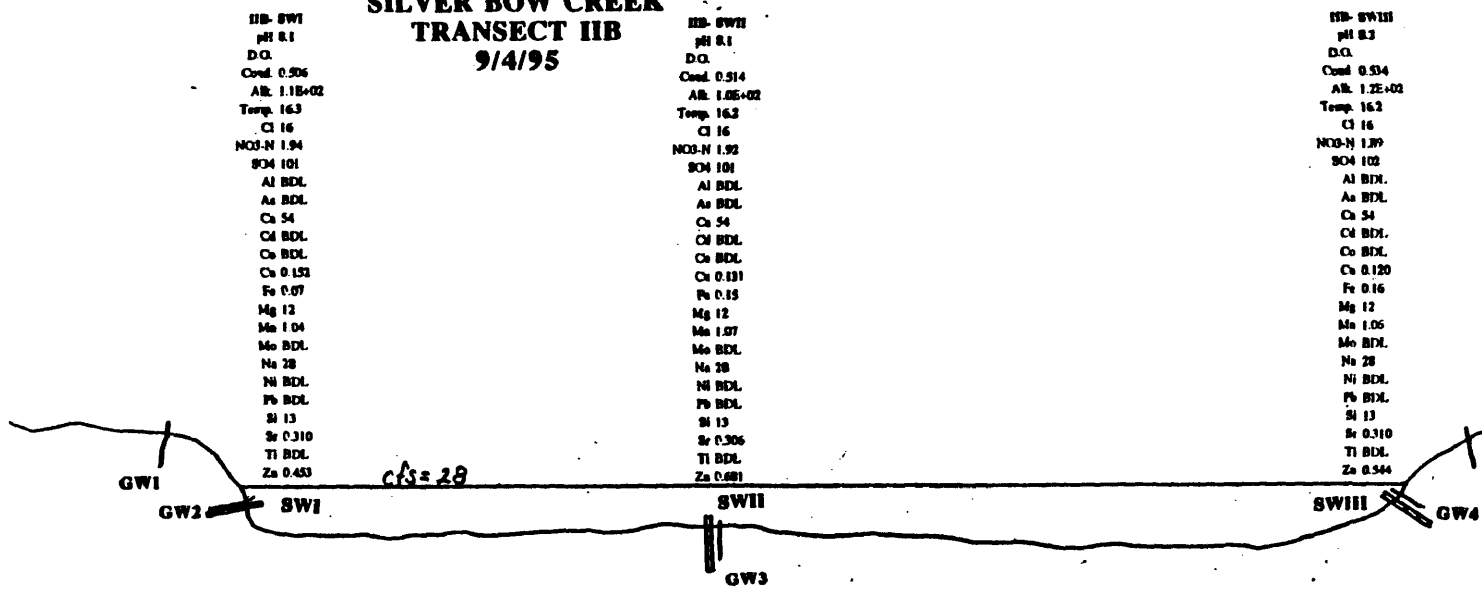
IBB- GW2
pH 6.6
D.O. 1.3
Cond. 0.355
Alk. 1.5E+02
Temp.
Cl 16
NO3-N 0.60
SO4 56.4
Al 0.11
As 0.16
Cr 40
Cd BDL
Co BDL
Cu 0.046
Fe 6.2
Mg 8.4
Mn 0.216
Mo 0.018
Na 20
Ni BDL
Pb BDL
Si 13
Sr 0.233
Ti BDL
Zn 0.214

IBB- GW3
pH 7.0
D.O. 3.4
Cond. 0.343
Alk. 1.58E+02
Temp.
Cl 12
NO3-N 1.01
SO4 58.1
Al BDL
As BDL
Cr 41
Cd BDL
Co BDL
Cu 0.086
Fe 0.18
Mg 8.3
Mn 0.473
Mo BDL
Na 19
Ni BDL
Pb BDL
Si 12
Sr 0.228
Ti BDL
Zn 0.523

IBB- GW4 pH 6.3 D.O. 1.1 Cond. 0.607 Alk. 2.75E+02 Temp. Cl 16 NO3-N BDL SO4 9.25 Al BDL As 0.16 Cr 100 Cd BDL Co BDL Cu BDL Fe 63 Mg 19 Mn 0.33 Mo 0.095 Na 36 Ni BDL Pb BDL Si 21 Sr 0.233 Ti BDL Zn 0.072	IBB- GW5 pH 6.1 D.O. 0.7 Cond. 1.495 Alk. 9.48E+02 Temp. Cl 21 NO3-N BDL SO4 51.5 Al 0.16 As 0.90 Cr 118 Cd BDL Co BDL Cu 0.012 Fe 2.1E+02 Mg 13 Mn 20.0 Mo 0.304 Na 40 Ni BDL Pb BDL Si 23 Sr 0.207 Ti 0.009 Zn 0.069
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1 Meter

**SILVER BOW CREEK
TRANSECT IIB
9/4/95**



IIB- GW1
 pH 8.1
 D.O.
 Cond. 0.506
 AR. 1.18E+02
 Temp. 16.3
 Cl 16
 NO3-N 1.94
 SO4 101
 Al BDL
 As BDL
 Ca 54
 Cd BDL
 Co BDL
 Cr 0.151
 Fe 0.07
 Mg 12
 Mn 1.04
 Mo BDL
 Na 28
 Ni BDL
 Pb BDL
 Si 13
 Sr 0.310
 Tl BDL
 Zn 0.453

IIB- GW2
 pH 8.1
 D.O.
 Cond. 0.514
 AR. 1.05E+02
 Temp. 16.3
 Cl 16
 NO3-N 1.92
 SO4 101
 Al BDL
 As BDL
 Ca 54
 Cd BDL
 Co BDL
 Cr 0.131
 Fe 0.15
 Mg 12
 Mn 1.07
 Mo BDL
 Na 28
 Ni BDL
 Pb BDL
 Si 13
 Sr 0.306
 Tl BDL
 Zn 0.481

IIB- SWIII
 pH 8.3
 D.O.
 Cond. 0.534
 AR. 1.2E+02
 Temp. 16.2
 Cl 16
 NO3-N 1.99
 SO4 102
 Al BDL
 As BDL
 Ca 54
 Cd BDL
 Co BDL
 Cr 0.120
 Fe 0.16
 Mg 12
 Mn 1.05
 Mo BDL
 Na 28
 Ni BDL
 Pb BDL
 Si 13
 Sr 0.310
 Tl BDL
 Zn 0.544

IIB- GW1
 pH 7.3
 D.O.
 Cond. 0.515
 AR. 1.25E+02
 Temp. 15.0
 Cl 17
 NO3-N 1.38
 SO4 87.1
 Al BDL
 As BDL
 Ca 55
 Cd BDL
 Co BDL
 Cr 0.117
 Fe 0.17
 Mg 12
 Mn 1.06
 Mo BDL
 Na 29
 Ni BDL
 Pb BDL
 Si 13
 Sr 0.314
 Tl BDL
 Zn 1.08

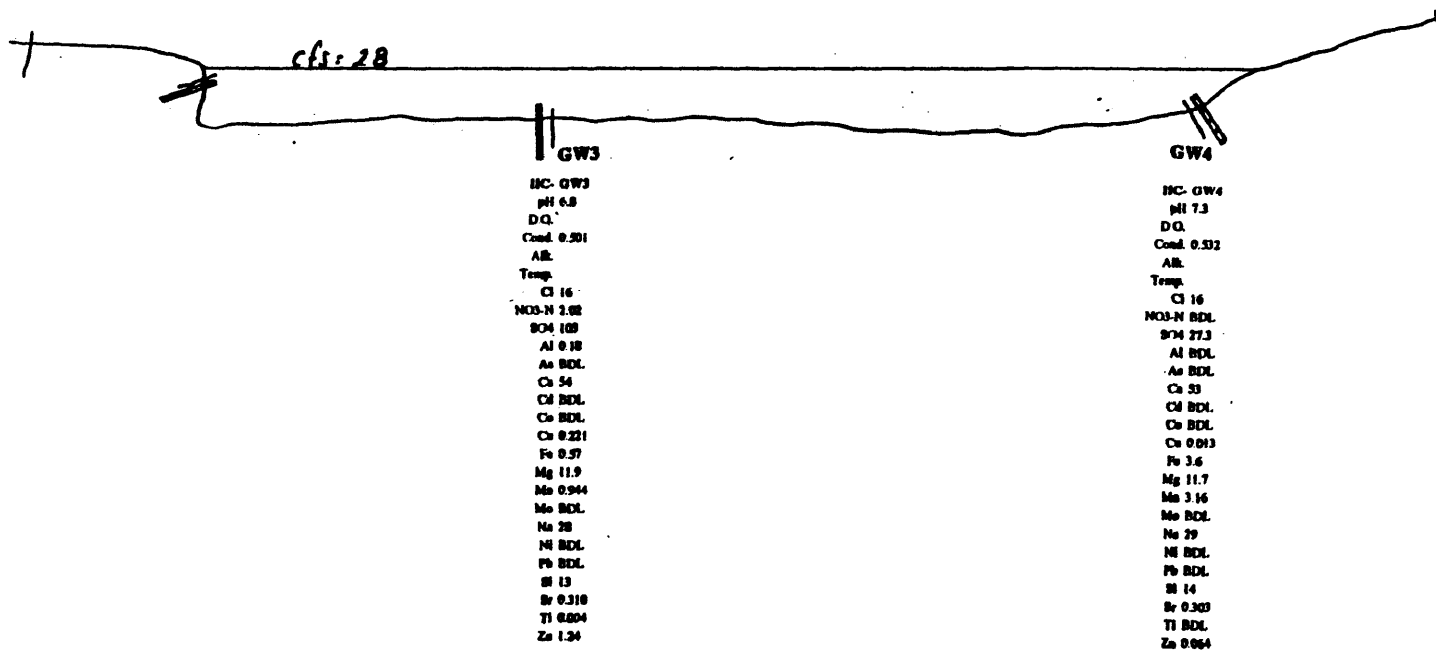
IIB- GW2
 pH 6.7
 D.O.
 Cond. 0.452
 AR. 1.65E+02
 Temp. 14.3
 Cl 17
 NO3-N 1.27
 SO4 83.3
 Al 0.09
 As 0.724
 Ca 55
 Cd BDL
 Co BDL
 Cr 0.089
 Fe 15
 Mg 12
 Mn 3.95
 Mo 0.090
 Na 28
 Ni BDL
 Pb BDL
 Si 12
 Sr 0.363
 Tl BDL
 Zn 0.471

IIB- GW3
 pH 7.5
 D.O.
 Cond. 0.498
 AR. 1.18E+02
 Temp. 15.9
 Cl 16
 NO3-N 1.05
 SO4 100
 Al BDL
 As BDL
 Ca 53
 Cd BDL
 Co BDL
 Cr 0.114
 Fe 0.15
 Mg 12
 Mn 0.714
 Mo BDL
 Na 28
 Ni BDL
 Pb BDL
 Si 13
 Sr 0.300
 Tl BDL
 Zn 0.967

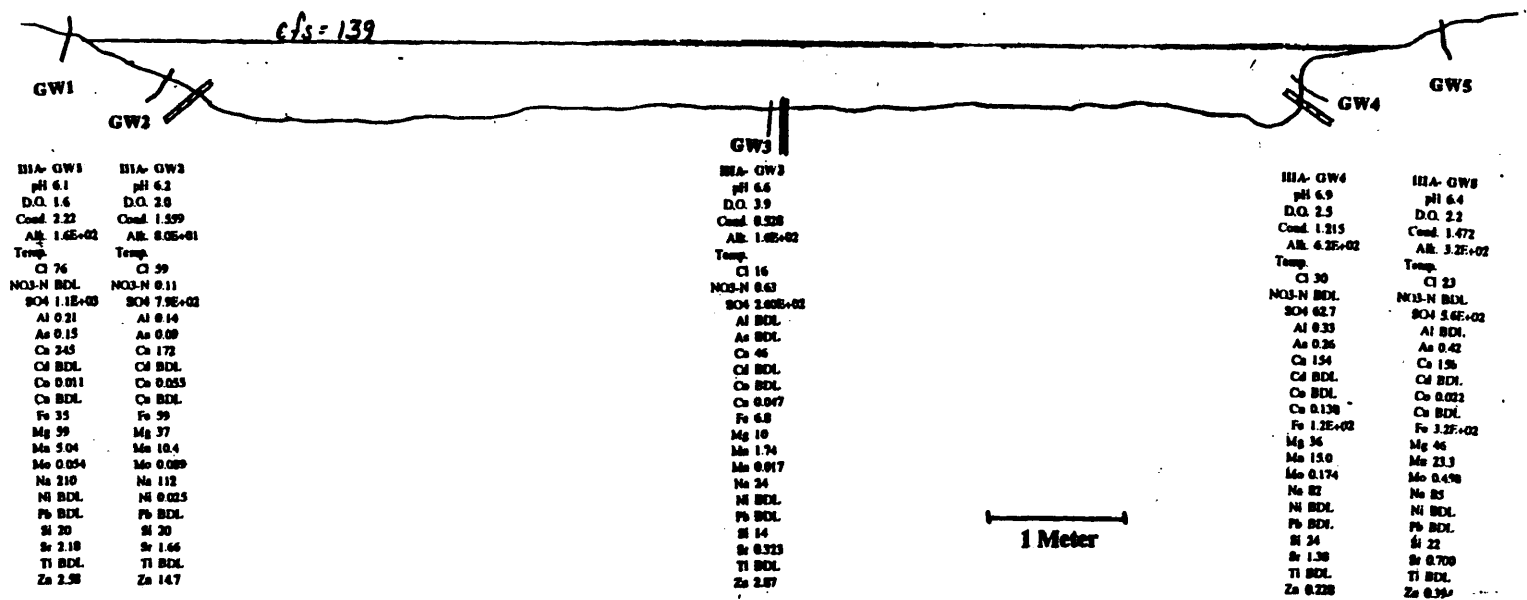
IIB- GW4
 pH 6.7
 D.O.
 Cond. 0.666
 AR. 3.05E+02
 Temp. 13.8
 Cl 16
 NO3-N BDL
 SO4 16.4
 Al BDL
 As BDL
 Ca 69
 Cd BDL
 Co BDL
 Cr BDL
 Fe 38
 Mg 14
 Mn 5.89
 Mo 0.055
 Na 33
 Ni BDL
 Pb BDL
 Si 16
 Sr 0.489
 Tl BDL
 Zn 0.243

1 Meter

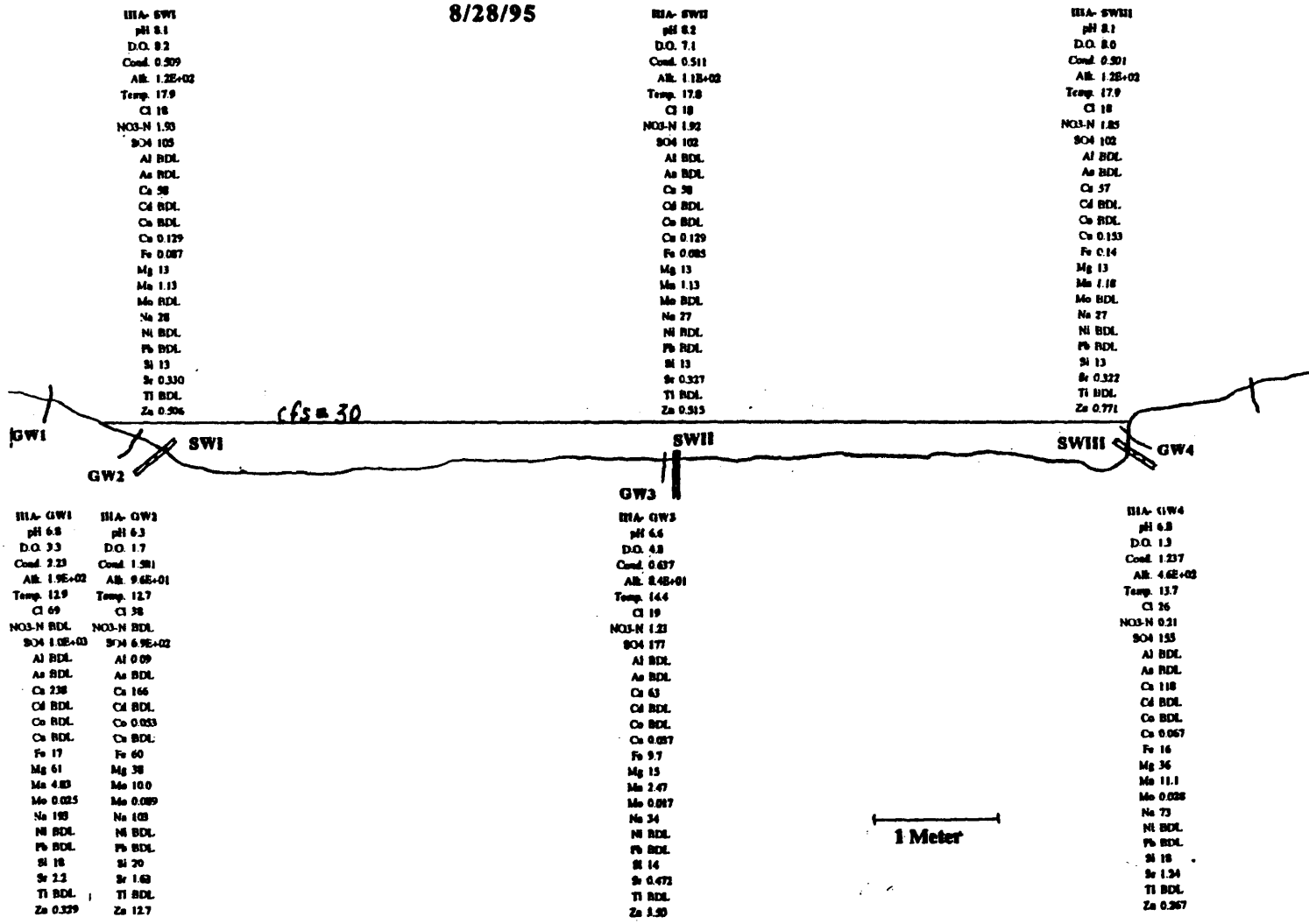
**SILVER BOW CREEK
TRANSECT IIC
9/4/95**



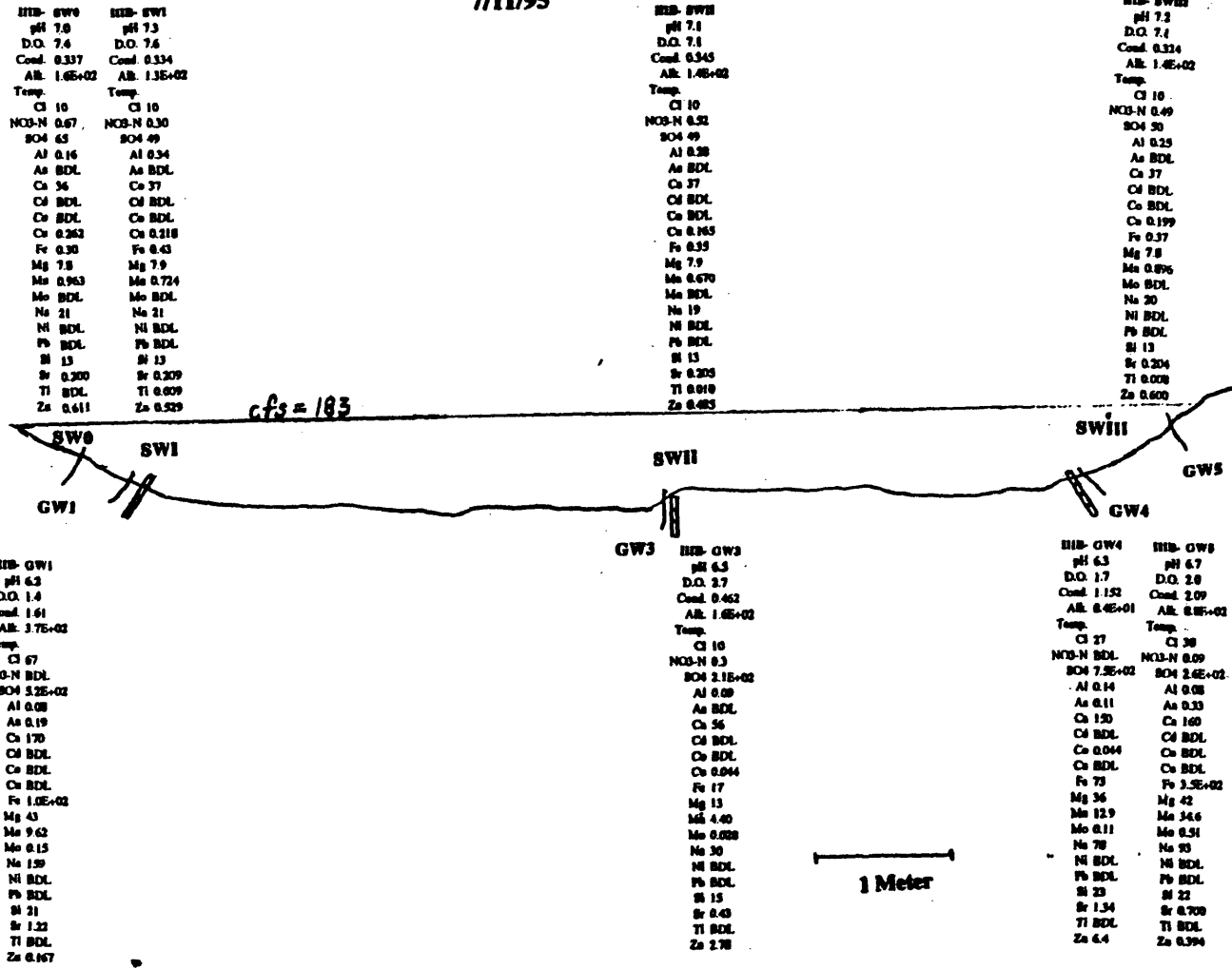
**SILVER BOW CREEK
TRANSECT IIIA
7/10/95**



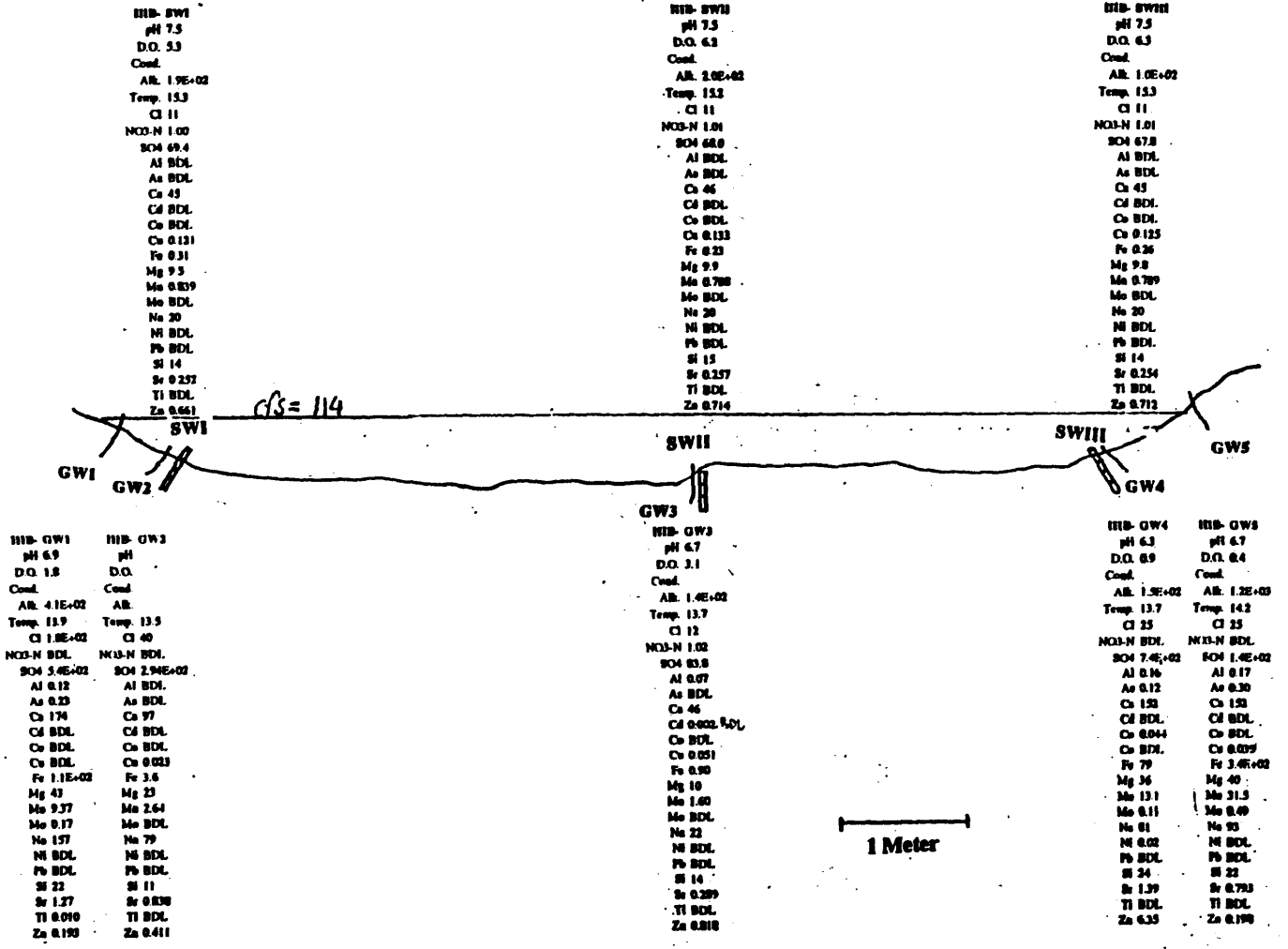
**SILVER BOW CREEK
TRANSECT IIIA
8/28/95**



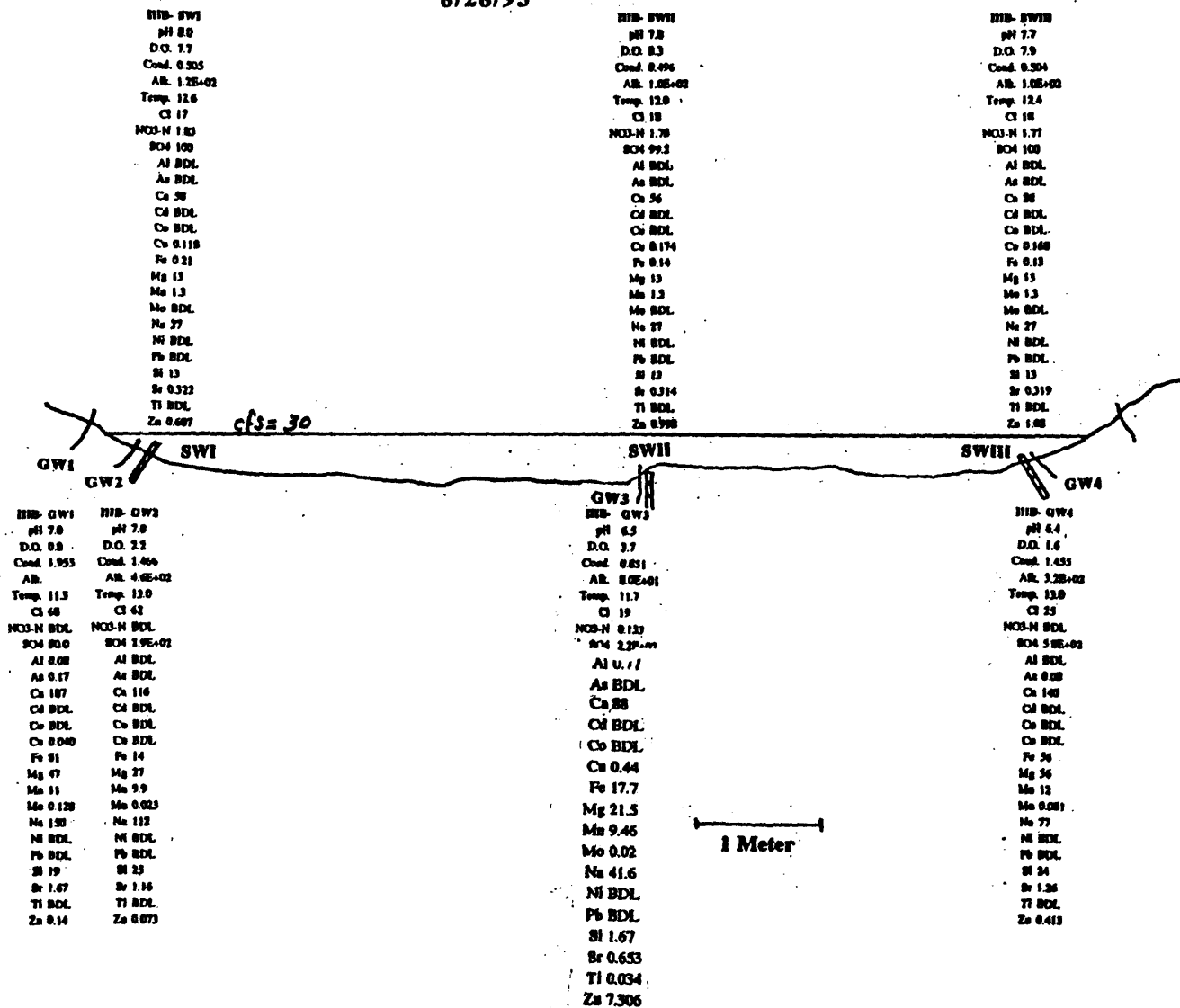
**SILVER BOW CREEK
TRANSECT IIIB
7/11/95**



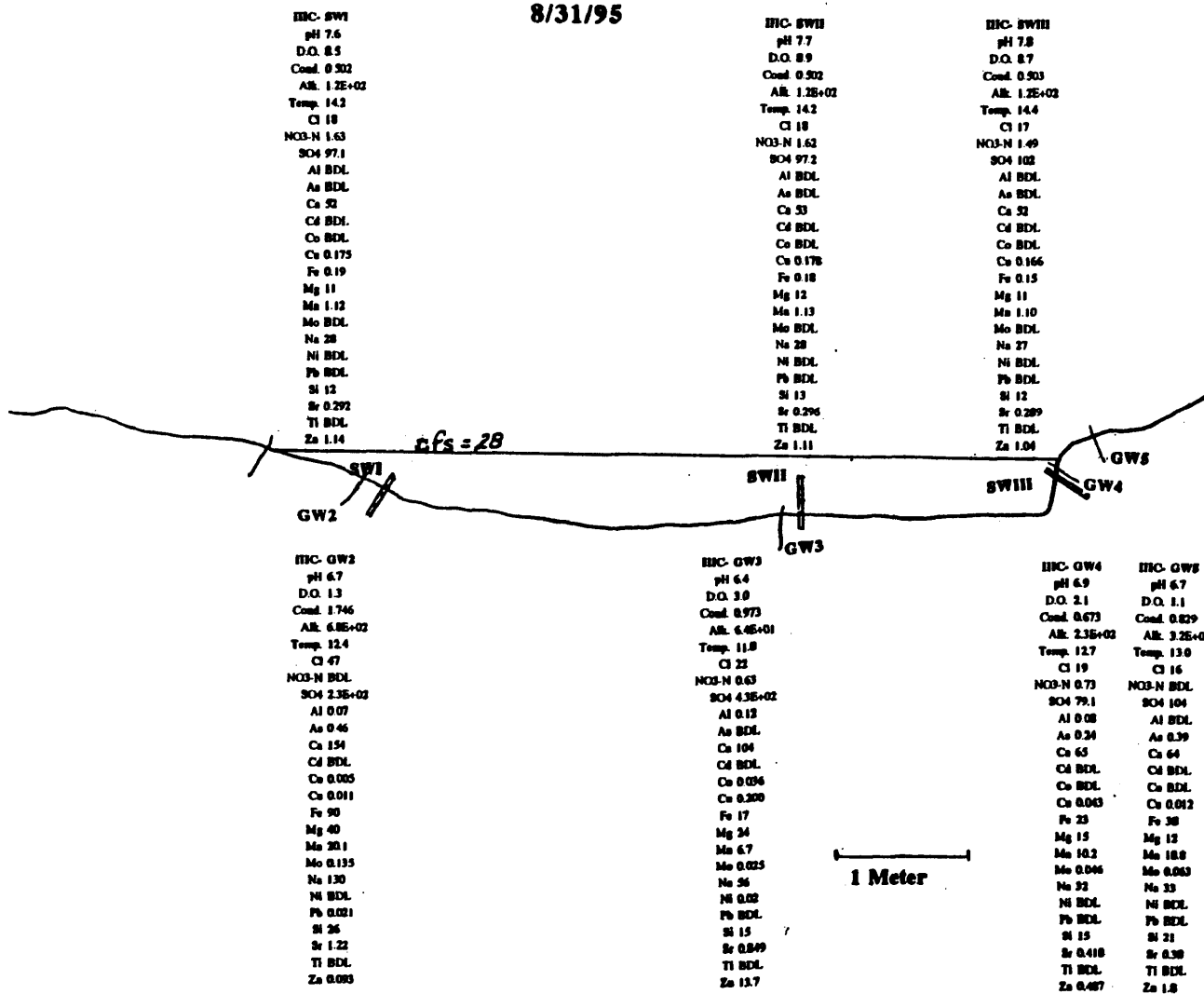
**SILVER BOW CREEK
TRANSECT IIB
7/17/95**



**SILVER BOW CREEK
TRANSECT IIB
8/28/95**



**SILVER BOW CREEK
TRANSECT IIC
8/31/95**

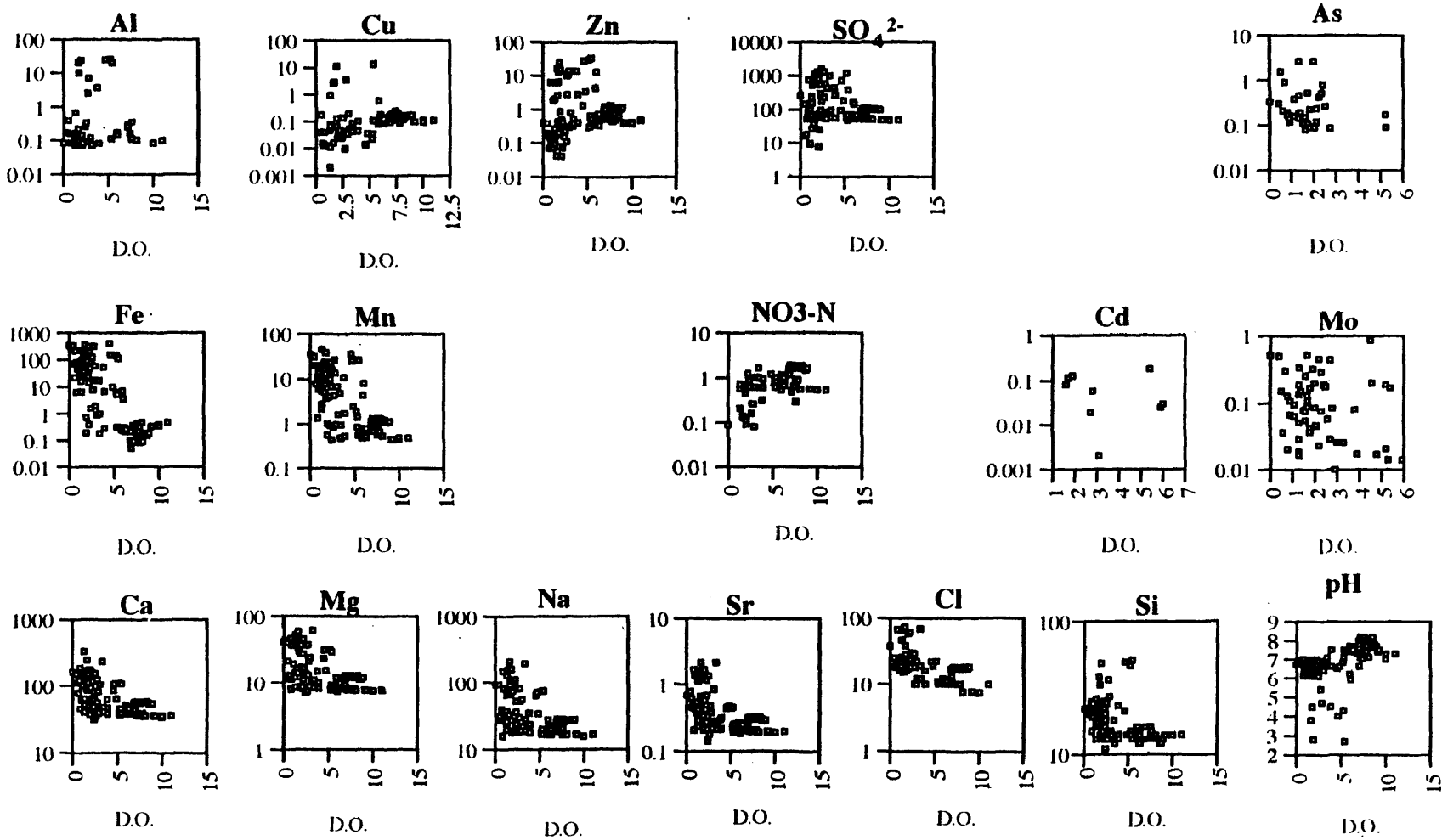


Mixing ratios for transects at Site 1														
Transect	Date	1/2 in SW	1/4 in SW	1/8 in SW	1/16 in SW	1/32 in SW	1/64 in SW	1/128 in SW	1/256 in SW	1/512 in SW	1/1024 in SW	1/2048 in SW	1/4096 in SW	1/8192 in SW
IA	6/20/95	6100	7	28	7.64	36	16.5	110	35.3	150	670/95	IA-GW 1/2	1.4	1.13
IA	7/19/95	2800	11	28	9.64	36	21.1	110	43.8	150	6/20/95	IA-GW 1	23.48	0.48
IA	8/24/95	1050	16	28	12.4	36	26	110	55.5	150	6/20/95	IA-GW 2	2.4	0.78
IA	9/27/95	1230	16	28	13	36	29	110	57	150	6/20/95	IA-GW 3	14	13.43
IB	6/26/95	4840	9	29	8.1	37	17.4	83	38.2	140	6/20/95	IA-GW 4	27.40	43.97
IB	7/19/95	2800	11	29	9.85	37	20.8	83	45.6	140	6/20/95	IA-GW 5	13	23.80
IB	8/27/95	850	17	29	13	37	26.6	83	57.6	140	6/20/95	IA-GW 6	67	62.51
IC	6/28/95	4840	9	28	8.1	39	17.4	43	38.2	180	7/19/95	IA-GW 1	8.8	7.81
IC	8/27/95	850	17.3	28	12.9	39	26.3	43	56.9	180	7/19/95	IA-GW 2	31.8	14.61
											7/19/95	IA-GW 3	47.1	28.89
											7/19/95	IA-GW 4	82.4	58.49
											7/19/95	IA-GW 5	17.6	47.58
											7/19/95	IA-GW 6	100.0	37.91
											7/19/95	IA-GW 6	100.0	97.10
											8/24/96	IA-GW 1	8.3	90.40
											8/24/96	IA-GW 2	8.3	2.14
											8/24/96	IA-GW 3	29.2	1.80
											8/24/96	IA-GW 4	30.8	24.29
											8/24/96	IA-GW 5	30.0	66.44
											8/24/96	IA-GW 6	100.0	5.45
											9/27/95	IA-GW 1	3.3	7.94
											9/27/95	IA-GW 2	6.7	15.56
											9/27/95	IA-GW 3	19.2	1.08
											9/27/95	IA-GW 4	100.0	2.47
											9/27/95	IA-GW 5	53.3	4.94
											9/27/95	IA-GW 6	100.0	17.33
											6/26/95	IB-GW 1/2	4.5	30.39
											6/26/95	IB-GW 1	105.0	100.00
											6/26/95	IB-GW 2	5.0	55.56
											6/26/95	IB-GW 3	2.0	2.47
											6/26/95	IB-GW 4	14.0	1.64
											6/26/95	IB-GW 5	55.0	30.11
											6/26/95	IB-GW 6	70.0	30.39
											7/19/95	IB-GW 1	26.1	100.00
											7/19/95	IB-GW 2	17.8	67.63
											7/19/95	IB-GW 3	6.8	3.19
											7/19/95	IB-GW 4	20.6	7.07
											7/19/95	IB-GW 5	44.4	69.53
											7/19/95	IB-GW 6	77.8	37.09
											8/27/95	IB-GW 1	16.7	0.81
											8/27/95	IB-GW 2	8.3	2.75
											8/27/95	IB-GW 3	8.3	2.75
											8/27/95	IB-GW 4	8.3	8.17
											8/27/95	IB-GW 5	100.0	8.92
											6/28/95	IC-GW 1/2		90.29
											6/28/95	IC-GW 1	66.3	80.58
											6/28/95	IC-GW 2	11.6	85.28
											6/28/95	IC-GW 3	22.6	62.31
											6/28/95	IC-GW 4	36.8	4.00
											6/28/95	IC-GW 5	7.9	4.65
											6/28/95	IC-GW 6	100.0	1.35
											8/27/95	IC-GW 1	12.1	0.56
											8/27/95	IC-GW 2	2.8	2.47
											8/27/95	IC-GW 3	2.8	3.15
											8/27/95	IC-GW 4	2.8	5.08
											8/27/95	IC-GW 5	2.8	100.00
											8/27/95	IC-GW 6	44.10	100.00
											8/27/95	IC-GW 7	4.19	9.02
											8/27/95	IC-GW 8	5.59	4.82
											8/27/95	IC-GW 9	5.77	4.79
											8/27/95	IC-GW 10	5.77	6.30
											8/27/95	IC-GW 11	13.08	28.59

Mixing ratios for transects at Site II																	
Transect	Date	L/A	Cl - In		Mg In		Na - In		Ca - In		Sample Date	Subsurface				Average % GW from Ca, Mg, Na	
			In SW	SW	I/A-GW5	SW	I/A-GW5	SW	I/A-GW5	SW		I/A-GW5	water ID	Cl mix. rat.	Mg mix. rat.		Na mix. rat.
I/A	7/6/95	2890	10.3	26	8.7	54	19	65	41.9	330	7/6/95	I/A-GW2	43.3	17.4	27.4	17.4	20.8
I/A	8/31/95	790	17.5		11.9		28.3		54		7/6/95	I/A-GW3	0.0	-0.2	0.4	-1.1	0.6
											7/6/95	I/A-GW4	60.8	7.1	13.5	7.5	9.4
I/B	7/6/95	2890	12		9.1		19.6		43.3		7/6/95	I/A-GW5	100.0	100.0	100.0	100.0	100.0
I/B	9/4/95	790	15.9		11.9		28		53.8		8/31/95	I/A-GW2	0.0	3.3	4.1	8.2	0.3
											8/31/95	I/A-GW3	14.1	2.9	2.2	-2.0	-2.4
											8/31/95	I/A-GW4	12.9	-1.4	3.5	2.6	-0.8
											7/6/95	I/B-GW1	80.0	1.3	1.1	0.2	0.9
											7/6/95	I/B-GW2	25.7	1.6	0.9	1.1	0.6
											7/6/95	I/B-GW3	-3.6	1.8	-0.4	0.9	-1.0
											7/6/95	I/B-GW4	30.0	22.0	36.3	19.8	26.1
											7/6/95	I/B-GW5	64.3	8.7	44.9	26.1	26.6
											9/4/95	I/B-GW1	9.9	0.2	2.7	0.4	1.1
											9/4/95	I/B-GW2	5.9	0.2	0.0	0.4	0.2
											9/4/95	I/B-GW3	2.0	0.2	0.0	0.3	0.0
											9/4/95	I/B-GW4	2.0	5.0	13.5	5.5	8.0
											9/4/95	I/C-GW2	3.0	5.7	12.2	8.0	8.6
											9/4/95	I/C-GW3	0.0	0.0	1.1	0.2	8.4
											9/4/95	I/C-GW4	4.0	-0.5	2.7	-0.3	0.6

Mixing ratios for transects at Site III																	
Transect	Date	L/A	Cl - In		Mg In		Na - In		Ca - In		Sample Date	Subsurface				Average % GW from Ca and Mg ratios	
			In SW	SW	I/A-GW5	SW	I/A-GW5	SW	I/A-GW5	SW		I/A-GW5	water ID	Cl mix. rat.	Mg mix. ratio		Na mix. rat.
III/A	8/28/95	850	17.7	76	13.1	61	27.4	210	57.7	245	8/28/95	III/A-GW1	87.8	95.8	90.7	96.3	96.0
											8/28/95	III/A-GW2	34.5	50.9	41.4	57.8	54.4
III/B	7/11/95	5180	9		7.8		20.6		36.5		8/28/95	III/A-GW3	2.4	2.9	3.8	2.7	2.8
III/B	7/17/95	3230	10		9.7		20		45		8/28/95	III/A-GW4	13.7	48.2	24.9	32.2	40.2
III/B	8/28/95	850	17.6		12.9		27		57.3		7/11/95	III/B-GW1	86.6	66.0	73.1	64.0	65.0
											7/11/95	III/B-GW3	7.5	10.6	4.8	9.4	10.0
											7/11/95	III/B-GW4	26.9	52.1	30.4	53.0	52.5
											7/11/95	III/B-GW5	43.3	64.3	38.2	58.6	61.5
											7/17/95	III/B-GW1		63.2	73.2	64.5	63.8
											7/17/95	III/B-GW2	45.5	25.9	31.2	25.9	25.9
											7/17/95	III/B-GW3	3.5	1.0	1.1	0.7	0.8
											7/17/95	III/B-GW4	23.0	51.7	31.9	53.5	52.6
											7/17/95	III/B-GW5	22.7	58.5	38.5	53.5	56.0
											8/28/95	III/B-GW1	85.4	70.1	67.2	69.1	69.6
											8/28/95	III/B-GW2	75.5	28.3	46.4	31.3	29.8
											8/28/95	III/B-GW3	3.6	17.9	8.0	16.4	17.1
											8/28/95	III/B-GW4	12.5	48.2	27.2	44.1	44.1
											8/31/95	III/C-GW2	49.6	58.4	56.1	52.7	55.5
											8/31/95	III/C-GW3	7.5	25.3	15.5	26.7	26.0
											8/31/95	III/C-GW4	2.4	6.1	2.0	6.3	6.2
											8/31/95	III/C-GW5	2.2	1.2	2.6	5.7	3.5

Concentrations vs. Dissolved oxygen [mg/L]



Full listing of hyporheic zone samples with dissolved oxygen, nitrogen-N, and sulfate concentrations lower than (and alkalinity levels higher than) surface water and groundwater samples:

Hyporheic zone (H.Z.) Sample	H.Z. Diss. O ₂	H.Z. NO ₃ -N	H.Z. SO ₄ ²⁻	H.Z. Alk.	Surface water Diss. O ₂	Surface water NO ₃ -N	Surface water SO ₄ ²⁻	Surface water Alk.
<i>Site I</i>								
IB-GW1 (6/26/95)	0.5	BDL	15.3	7.6E+02	6	0.9-1.0	54-55	1.0E+02
IC-GW1 (6/28/95)	1.9	BDL	1.88	NA	6-7	0.9	51-52	1.5E+02
IC-GW2 (6/28/95)	2.1	BDL	24.9	2.0E+02	6-7	0.9	51-52	1.5E+02
IA-GW1 (7/19/95)	1.3	0.1	37.8	3.4E+02	6	1.1	73	1.7E+02
IA-GW2 (7/19/95)	1.2	BDL	28.9	5.7E+02	6	1.1	73	1.7E+02
IB-GW1 (7/19/95)	0.8	BDL	21.9	NA	6	1.1	73	1.7E+02
IA-GW1 (8/24/95)	1.8	BDL	69.2	1.8E+02	7-8	1.6-1.7	90-96	1.3E+02
IA-GW2 (8/24/95)	1.5	BDL	36.4	3.7E+02	7-8	1.6-1.7	90-96	1.3E+02
IB-GW2 (8/27/95)	1.3	0.76	63.2	1.4E+02	7-8	1.9	100-104	1.2E+02
IC-GW2 (8/27/95)	NA	0.14	31.9	3.4E+02	NA	1.9	101-103	1.2E+02
IC-GW3 (8/27/95)	NA	0.08	62.1	1.4E+02	NA	1.9	101-103	1.2E+02
IA-GW1 (9/27/95)	1.8	BDL	68.1	2.1E+02	7	1.9	108	1.3E+02
IA-GW2 (9/27/95)	1.6	0.14	51.4	3.4E+02	7	1.9	108	1.3E+02
<i>Site II</i>								
IIA-GW2 (7/6/95)	2.0	0.09	7.61	NA	6	0.7-0.9	60-63	1.3E+02
IIA-GW4 (7/6/95)	1.3	BDL	25.8	3.4E+02	6	0.7-0.9	60-63	1.3E+02
IIB-GW2 (7/6/95)	1.3	0.60	56.4	1.5E+02	6-7	1.2	62-67	1.4E+02
IIB-GW4 (7/6/95)	1.1	BDL	9.25	2.7E+02	6-7	1.2	62-67	1.4E+02
IIA-GW2 (8/31/95)	0.6	BDL	17.8	3.8E+02	8	1.6-1.7	102-104	1.2E+02
IIA-GW4 (8/31/95)	0.9	BDL	55.9	2.6E+02	8	1.6-1.7	102-104	1.2E+02
IIB-GW1 (9/4/95)	NA	1.38	87.1	1.2E+02	NA	1.9	101	1.1E+02
IIB-GW2 (9/4/95)	NA	1.27	83.3	1.6E+02	NA	1.9	101	1.1E+02
IIB-GW4 (9/4/95)	NA	BDL	16.4	3.8E+02	NA	1.9	101	1.1E+02
IIC-GW4 (9/4/95)	NA	BDL	27.3	NA	NA	1.9	101	1.1E+02
<i>Site III</i>								
IIIC-GW4(8/31/95)	2.1	0.73	79.1	2.3E+02	8-9	1.5-1.6	97-102	1.2E+02

Transect-scale trends in surface water chemistry:

A simple and quick measurement of groundwater infiltration into the surface water was measured directly with the specific conductance meter. When the meter was placed directly against (but not into) the stream bank, significantly higher specific conductance was measured where water data indicated that ground water was infiltrating.

Transect	Spec. Cond. (mS/cm) in SW center	Spec. Cond. (mS/cm) in SW, against creek bank	Spec. Cond. (mS/cm) in ground water <1m from creek
IA (2/26/95)	0.220	0.290	0.950
IA (5/15/95)	0.200	0.220	1.20
IB (5/15/95)	0.210	0.290	1.50
IC (5/15/95)	0.210	0.220	0.370
IB (8/27/95)	0.498	0.625	1.95
IIIA (8/28/95)	0.510	0.640 (E bank)	2.23
IIIA (8/28/95)	0.510	0.599 (W bank)	1.24

Table 6: Examples of specific conductance measurements of water along the creek banks, as compared to those of the surface water in the central portion of the creek channel and to nearby groundwater.

This was found primarily on the east bank of Site I, where piezometric data indicate groundwater is flowing into the creek from the east side. When the specific conductance meter was placed in the surface water, right up against the east bank, higher measurements were recorded. The most striking example was measured on 8/27/95, when the specific conductance in the central portion of the stream channel measured at 0.498 mS/cm, whereas along the east bank it was measured 0.625 mS/cm. The suspected source for this elevated level, the groundwater within the east bank, was measured to have a specific conductance of 1.95 mS/cm. The 95% confidence interval for the mean error associated with specific conductance measurements was calculated to be 3.4%, and thus these differences observed in the surface water are significant. Such measurements indicate that it is possible to directly detect the high-conductivity groundwater infiltrating the creek, at least along those portions of the creek where the differences between the surface water and ground water specific conductance values are large. Although the measurements are not useful in indicating the type and concentrations of the specific ions are moving into the surface water system, they are useful in detecting locations of ground water movement into the creek.

ICAPES data from bead digests run on 1/24/96 and 1/25/96

Sample Name	Analysis Date	Time	Al	As	Ca	Cd	Co	Cr	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Si	Sr	Ti	Zn
BLANK	1/24/96	8 45	0.003	-0.021	0.001	0.0011	0.0004	0.0005	0.0023	0.003	0.01	0	-0.0011	0.009	-0.004	0.02	0.014	0.034	0.0003	-0.0002	0.0012
8-STR-BLANK1	1/24/96	8 48	0.023	-0.011	0	0.0006	0.0046	0.0006	0.0027	0.013	0.03	5E-04	0.0034	-0.013	0.005	0.08	-0.005	0.029	0.0002	0	0.0071
8-STR-BLANK2	1/24/96	8 50	0.013	0.018	0	0.0006	0.0016	0.0077	0.0012	0.01	0.02	0	0.0005	-0.009	-0.001	0.08	-0.005	0.028	0	0.0004	0.0051
8-BEADBLANK1.1.0022g	1/24/96	8 54	4.05	0.016	0.211	0.0001	0.0002	0.0067	0.0019	0.024	0.157	0.003	0.0006	0.031	0.001	0.11	0.005	0.417	0.0012	0.0065	0.0049
8-BEADBLANK2.0.9939g	1/24/96	8 57	2.27	0.03	0.249	0.0007	0.0008	0.0147	0.0008	0.017	0.116	0.005	-0.0003	0.043	-0.001	0.09	-0.002	0.377	0.0015	0.0066	0.0022
BLANK	1/24/96	9 09	0.02	0.003	0.006	0	0.001	0.0011	0.0027	0.001	0.014	0	-0.0011	0.006	-0.003	0.09	-0.026	-0.007	0.0003	0	0.0035
USGS T107	1/24/96	9 13	0.25	0	13	0.0183	0.0123	0.0263	0.0329	0.061	2.27	0.052	0.0159	22.9	0.027	0.14	0.033	4.16	0.6649	0.0025	0.0906
USGS T117	1/24/96	9 17	0.106	-0.006	23.3	0.0033	0.0067	0.0234	0.0077	0.52	11	0.243	0.0154	22	0.014	0.37	-0.006	6.21	0.7779	0.0027	0.2034
8-BLANK	1/26/96	9 06	0.004	-0.009	0	0.0001	2E-04	0.0019	-0.001	-0.004	0	0	0.0019	0.01	-0.005	0.02	0.003	0.011	0.0020	0	0.0002
USGS T107	1/26/96	9 08	0.236	-0.011	12.9	0.0173	0.0142	0.0276	0.0301	0.061	2.28	0.052	0.0156	22.7	0.029	0.05	0.017	4.14	0.645	0.0022	0.0887
USGS T117	1/26/96	9 12	0.092	0.01	23.2	0.0032	0.0055	0.0274	0.0053	0.517	11	0.241	0.0132	21.9	0.006	0.27	-0.006	6.24	0.278	0.0025	0.1988
USGS T-91	1/26/96	9 15	0.413	-0.01	27.8	0.0397	0.0131	0.0196	0.997	0.313	11.1	2.551	0.0014	6.23	0.018	0.08	0.009	7.37	0.123	0.0018	6.276
USGS AMW-2	1/26/96	9 18	20.5	-0.028	339	0.1393	0.1458	0.3491	4.939	102	116	92.03	0.1445	25.2	0.247	0.51	0.037	23	1.624	0.0148	47.61
8-BLANK	1/26/96	9 28	0.006	-0.012	0.006	-1E-04	0.0009	0.0357	-0.002	-0.003	0.007	0.002	0.0021	0.07	-0.004	0.14	0.002	-0.004	0.0015	0.006	0.0009
8-STR-BLANK1	1/26/96	9 31	0.017	-0.005	0.053	2E-04	0.0011	0.028	-0.001	0.032	0.06	0.001	-0.0009	0.052	-0.004	0.14	0	0.071	0.0008	0.0022	0.0038
8-STR-BLANK2	1/26/96	9 34	0.065	-0.017	0.084	-1E-04	0.0017	0.036	-0.001	0.041	0.366	0.001	-0.0004	0.125	-0.002	0.17	-0.006	0.206	0.0014	0.0065	0.0022
8-STR-BLANK3	1/26/96	9 37	0.011	-0.013	0.065	0.0003	0.0009	0.0289	-7E-04	0.032	0.057	5E-04	0.0012	0.12	0.017	0.14	-0.01	0.079	0.001	0.004	0.0162
8-BLANK	1/26/96	9 40	-0.003	0.005	0.006	-6E-04	0.0009	0.0389	-7E-04	-0.004	0.01	7E-04	0.0024	0.08	-0.001	0.18	-0.001	0.002	0.0017	0.0076	0.0007
8-BEADBLANK3	1/26/96	9 43	3	-0.015	0.676	0.0008	0.0017	0.0323	-0.001	0.056	0.169	0.009	-0.0011	0.175	-0.006	0.17	0.002	0.777	0.0042	0.0143	0.0028
8-IC-1 0-8cm	1/26/96	9 46	4.1	0.095	1.81	0.0037	0.0019	0.0366	1.107	3.51	0.52	0.395	0.0031	0.394	0.049	0.72	0.141	1.68	0.015	0.0474	1.355
8-IC-1 8-11 cm	1/26/96	9 50	4.23	0.116	2.28	0.0057	0.003	0.0404	0.5275	1.01	0.841	0.958	0.007	0.728	0.146	0.99	0.214	1.47	0.0188	0.0347	1.601
8-IC-1 11-30 cm	1/26/96	9 53	3.65	0.022	1.61	0.0031	0.0015	0.037	0.9004	1.53	0.83	0.706	0.0006	0.991	0.069	0.46	0.097	0.873	0.012	0.0275	1.973
8-IC-3 0-9.5cm	1/26/96	9 56	5.06	0.063	2	0.0057	0.0008	0.0371	1.512	3.93	0.536	0.504	0.0066	0.667	0.103	0.75	0.188	1.87	0.0162	0.0419	1.999
8-IC-3 9.5-21cm	1/26/96	10 00	4.56	0.016	1.42	0.0021	0.0008	0.0403	0.9113	2.81	0.426	0.262	0.003	0.45	0.041	0.45	0.065	1.23	0.0101	0.0358	0.5246
USGS T107	1/26/96	10 03	0.226	0.002	13.6	0.0187	0.0157	0.0776	0.0294	0.063	2.28	0.054	0.0132	22.3	0.034	0.28	0.039	4.34	0.0644	0.0116	0.0969
USGS T117	1/26/96	10 06	0.07	-0.01	23.8	0.0027	0.0047	0.0688	0.0035	0.513	10.7	0.243	0.0115	20.8	0.009	0.49	0	6.32	0.0654	0.0112	0.2086
8-IC-3 21-32cm	1/26/96	10 09	11.3	0.027	1.25	0.0044	0.0011	0.0422	1.108	13.2	0.555	0.207	0.02	0.386	0.059	0.47	0.065	3.01	0.0095	0.0506	0.7684
8-IC-3 32-42cm	1/26/96	10 12	5.96	0.014	1.22	0.0025	0.0004	0.0396	0.951	5.41	0.529	0.327	0.0052	0.433	0.065	0.37	0.063	1.46	0.0084	0.0344	0.6865
8-IC-1 30-42cm	1/26/96	10 15	4.63	0.468	2.65	0.0245	0.0034	0.0556	3.598	6.04	0.752	1.29	0.0861	0.879	0.317	0.87	0.244	3.61	0.0535	0.0358	4.367
8-IC-2 0-14cm	1/26/96	10 18	3.13	0.055	2.45	0.0066	0.0017	0.0332	1.424	3.85	0.575	0.609	0.0041	0.397	0.109	0.79	0.156	1.86	0.0196	0.0456	2.093
8-IC-2 14-18cm	1/26/96	10 21	10.8	0.124	2.68	0.0098	0.0027	0.0439	1.773	5.43	1.03	0.71	0.0067	0.758	0.045	0.86	0.329	2.69	0.0233	0.0617	1.618
8-IC-2 18-32cm	1/26/96	10 24	4.74	0.85	3.35	0.023	0.0036	0.0573	2	53.8	0.966	1.567	0.0749	0.894	0.052	5.9	0.287	4.22	0.0483	0.0391	4.512
8-IC-2 18-32cm (2)	1/26/96	10 28	5.86	0.758	2.8	0.0202	0.0017	0.0522	2	44.6	1.03	1.478	0.0647	0.934	0.03	5.4	0.241	3.95	0.0432	0.0358	3.945
8-IC-2 32-41cm	1/26/96	10 31	2.73	0	1.37	0.0042	0.0009	0.0431	0.552	1.4	0.431	0.34	0.0028	0.57	0.199	0.49	0.047	0.762	0.0097	0.0239	0.6602
8-IB-1 0-8cm	1/26/96	10 38	6.97	0.034	1.76	0.0147	0.0051	0.0476	6.275	2.61	2.95	3.352	0.0037	2.93	0.005	0.87	0.093	1.61	0.0157	0.0331	5.522
8-IB-1 0-8cm (2)	1/26/96	10 41	3.36	0.014	1.96	0.0153	0.0051	0.0527	2.91	3.05	2.81	3.353	0.0041	3.05	0.007	0.98	0.105	1.38	0.0169	0.0297	5.593
8-IB-1 0-4cm DUP	1/26/96	10 43	37.1	0.023	1.95	0.0137	0.0055	0.0288	8.229	2.87	3.79	3.359	0.0053	3.01	0.008	0.9	0.086	4.39	0.0163	0.0803	5.576
8-IB-1 8-12cm	1/26/96	10 47	5.45	0.158	1.9	0.0093	0.0025	0.0437	1.366	10.1	0.821	1.3	0.011	0.798	0.003	2.1	0.201	1.63	0.0236	0.0347	2.476
8-BLANK	1/26/96	10 50	0.012	-0.023	0.01	0.0008	0.0006	0.0562	-0.002	-0.002	0.02	5E-04	-0.0024	0.13	-0.003	0.3	-0.004	0	0.0023	0.0094	0.0025
8-IB-1 12-16cm	1/26/96	10 53	2.61	0.004	1.44	0.0028	0.0015	0.038	0.291	0.757	0.396	0.252	-0.0008	0.504	-0.003	0.43	0.053	0.895	0.0109	0.0249	0.7794
8-BLANK	1/26/96	10 56	0.015	-0.02	0.012	-3E-04	0.0009	0.0553	-7E-04	-0.004	0.004	2E-04	-0.001	0.112	-0.001	0.32	0.006	-0.002	0.0023	0.009	0.0024
USGS T107	1/26/96	10 59	0.226	-0.1	13.6	0.0188	0.0138	0.0847	0.029	0.058	2.22	0.054	0.0137	22.2	0.031	0.36	0.011	4.26	0.0641	0.0128	0.0974
USGS T117	1/26/96	11 02	0.094	-0.007	24	0.0032	0.0051	0.0756	0.0042	0.521	10.6	0.248	0.012	21.3	0.01	0.56	-0.002	6.39	0.2674	0.0119	0.2131
8-IB-1 16-21cm	1/26/96	11 05	3.31	-0.032	1.01	0.0011	0	0.0405	0.1947	0.358	0.275	0.128	0.0003	0.321	-0.003	0.37	0.006	0.819	0.0076	0.0206	0.6695
8-IB-22-32cm	1/26/96	11 08	8.85	0.008	1.51	0.0019	0.0021	0.0427	0.4582	0.517	0.494	0.205	0.0001	0.455	0.002	0.41	0.047	1.33	0.0099	0.034	0.6558
8-IB-1 32-41.5	1/26/96	11 11	5.86	0.004	1.74	0.0022	0.0021	0.0409	0.6104	1.44	0.606	0.364	0.0007	0.687	0.001	0.46	0.069	1.38	0.0132	0.0372	0.6588
8-IB-1 8-12cm DUP	1/26/96	11 14	21.1	0.123	2.24	0.0107	0.0017	0.0489	1.339	1.01	1.24	1.296	0.0138	0.877	0.007	2.2	0.207	3.59	0.0254	0.0692	2.451
8-IA-3 0-10cm	1/26/96	11 17	4.43	0.098	2.18	0.0122	0.0027	0.0458	3.566	6.93	1.33	1.641	0.0066	1.94	0.013	1.5	0.217	2.53	0.0225	0.0474	3.596
8-IA-3 10-20cm	1/26/96	11 20	5.42	0.13	1.7	0.0126	0.0021	0.0432	2.655	1.51	1.35	1.543	0.0178	2.02	0.131	0.87	0.222	3.02	0.0144	0.032	2.692
8-IA-3 20-31cm	1/26/96	11 23	5.7	0.001	1.76	0.0132	0.0025	0.0458	1.483	1.13	2	1.967	0.0005	2.06	0.016	0.6	0.093	1.23	0.0127	0.0278	3.091
8-IA-3 31-42cm	1/26/96	11 26	5.8	0.019	3.44	0.0047	0.0025	0.0444	1.077	1.12	1.32	1.082	0.0017	1.22	0.14	0.44	0.065	1.61	0.0151	0.0376	1.773
8-IA-2 0-12cm	1/26/96	11 30	3.7	0.074	2.47	0.0084	0.0025	0.0455	3.386	7.07	1.08	1.333	0.0092	1.45	0.045	1.5	0.203	2.31	0.0241	0.0481	2.471
8-IA-2 0-12cm (2)	1/26/96	11 33	5.92	0.096	2.67	0.0091	0.0015	0													

Appendix

Sample Name	Date	Time	Al	As	Ca	Cd	Co	Cr	Cu	Fe	Mg	Mn	Mo	Ni	Nb	P	Pb	Si	Sr	Ti	Zn
USGS T107	1/26/96	13:56	0.225	-0.008	12.9	0.0177	0.0129	0.0179	0.0326	0.067	2.28	0.052	0.0182	22.9	0.03	0.02	0.041	4.18	0.0644	0.0007	0.0898
USGS T117	1/26/96	13:59	0.081	-0.001	23.5	0.0038	0.0037	0.0225	0.0078	0.513	11.1	0.244	0.0166	21.9	0.009	0.26	0.001	6.32	0.2788	0.0022	0.2025
USGS T91	1/26/96	14:02	0.408	-0.031	28.5	0.0401	0.0112	0.0193	1.005	0.317	11.1	2.587	0.0041	6.19	0.025	0.1	0.029	7.52	0.1237	0.0015	6.422
8-BLANK	1/26/96	14:05	-0.001	-0.01	0.004	0.0003	0.0006	-0.0048	0.0004	-0.002	0.001	0	0.002	-0.017	0.005	-0.01	-0.007	0.002	-1E-04	-0.0022	0.0012
USGS T-91	1/26/96	14:08	0.399	-0.027	28.4	0.0414	0.0105	0.02	0.9996	0.31	11	2.572	0.0073	6.2	0.022	0.12	0.019	7.48	0.1231	0.0015	6.4
8-III-B-3 30-41cm	1/26/96	14:13	4.43	0.019	1.97	0.0007	0.0007	-0.0095	0.3302	1.11	0.333	0.071	0.0039	0.336	0.002	0.21	0.117	1.54	0.0155	0.0274	0.4533
8-III-B-3 21-30cm	1/26/96	14:16	5.5	0.004	2.15	0.0019	9E-04	-0.0065	0.6094	1.13	0.35	0.053	0.0053	0.337	0.002	0.28	0.118	1.82	0.0157	0.0345	0.6875
8-III-B-3 9-21cm	1/26/96	14:19	3.88	0	2.08	0.0025	0.0004	-0.005	0.3828	1.11	0.272	0.071	0.0033	0.407	0.004	0.3	0.091	1.54	0.0154	0.0255	0.921
8-III-B-3 0-4cm	1/26/96	14:21	5.46	0.283	3.65	0.008	-0.001	-0.0038	1.739	23.9	0.433	0.28	0.0368	0.5	0.006	2.9	0.222	3.37	0.0485	0.0345	1.644
8-III-B-2 0-8cm	1/26/96	14:25	6.7	0.533	4.54	0.0071	0.0004	0.0244	2.818	141	0.369	0.166	0.2101	0.55	0.061	3.5	0.083	12.8	0.0604	0.0304	3.701
8-III-B-2 8-18cm	1/26/96	14:28	5.09	-0.013	2.29	0.0001	-0.002	-0.0073	0.1385	0.971	0.318	0.033	0.0037	0.358	0.055	0.1	0.028	1.79	0.0121	0.0296	0.2256
8-III-B-2 18-29cm	1/26/96	14:31	3.3	0	1.69	0.0008	9E-04	-0.0083	0.1499	0.694	0.236	0.029	0.004	0.342	0.179	0.06	0.018	1.06	0.0094	0.0176	0.2154
8-III-B-2 18-29cm(2)	1/26/96	14:33	4.29	-0.02	1.96	0.0004	7E-04	-0.0031	0.0574	0.612	0.275	0.032	0.0032	0.308	0.046	0.09	0.029	1.52	0.0107	0.0236	0.1605
8-III-B-3 4-9cm	1/26/96	14:36	4.86	-0.008	2.3	0.0022	-0.002	-0.0044	0.6053	1.41	0.334	0.077	0.0039	0.471	-0.004	0.3	0.112	1.8	0.018	0.0334	0.8537
USGS T107	1/26/96	14:39	0.229	0.01	14.1	0.0201	0.0137	0.0428	0.0333	0.072	2.45	0.057	0.0205	24.7	0.035	0.15	0.035	4.54	0.0695	0.0049	0.0989
USGS T107	1/26/96	14:42	0.23	-0.011	13.7	0.0194	0.0129	0.0422	0.0344	0.072	2.36	0.055	0.018	23.6	0.033	0.1	0.036	4.37	0.0672	0.0041	0.0948
USGS T107	1/26/96	14:43	0.219	0.005	13.7	0.0188	0.0133	0.0395	0.0337	0.07	2.34	0.055	0.0209	23.6	0.032	0.09	0.025	4.34	0.067	0.0037	0.0952
8-III-C-1 0-8cm	1/26/96	14:46	6.1	0.026	3.16	0.0043	4E-04	-0.003	0.7179	2.57	0.664	0.319	0.0066	1.44	0.002	0.38	0.094	2.57	0.0216	0.039	0.9871
8-BLANK	1/26/96	14:49	-0.015	-0.025	0.004	0.0014	7E-04	0.0035	0	-0.001	-0.015	0	0.0027	0.02	0.001	0.06	0.009	0.003	0.0002	-0.0004	0
8-III-C-1 8-11cm	1/26/96	14:52	4.89	0.375	3.93	0.0029	4E-04	0.0079	0.9178	39.7	0.418	0.493	0.0614	0.595	0.002	3.6	0.161	4.25	0.0544	0.033	2.016
8-III-C-1 11-28cm	1/26/96	14:55	4.5	-0.004	2.25	0.0011	9E-04	-0.0098	0.1807	0.68	0.326	0.078	0.0051	0.925	0.007	0.22	0.066	1.63	0.0139	0.0234	0.6266
8-III-C-1 11-28cm(2)	1/26/96	14:58	3.7	-0.02	2.03	0.0025	-0.001	-0.0089	0.1684	0.57	0.286	0.072	0.0013	0.93	0.001	0.21	0.064	1.37	0.0124	0.0191	0.4939
8-III-C-1 28-42cm	1/26/96	15:01	3.45	-0.009	2.09	0.0022	7E-04	-0.005	0.1462	0.452	0.294	0.095	0.0029	1.31	0.001	0.17	0.041	1.35	0.0127	0.021	0.3694
8-III-C-3 0-11cm	1/26/96	15:04	6.3	0.048	3.45	0.004	7E-04	-0.0066	1.188	4.62	0.519	0.312	0.0074	0.476	0.003	0.63	0.111	2.86	0.0238	0.0469	1.117
8-III-C-3 11-17cm	1/26/96	15:08	4.33	0.095	2.34	0.003	-0.002	-0.0049	0.6212	6.95	0.296	0.107	0.0122	0.413	0.002	0.53	0.134	2.27	0.0163	0.0289	0.6445
8-III-C-3 17-31cm	1/26/96	15:11	4.4	-0.014	2.02	0.0009	7E-04	-0.0068	0.2118	0.674	0.265	0.058	0.003	0.33	0.036	0.18	0.063	1.67	0.0122	0.0276	0.7046
8-III-C-3 31-42cm	1/26/96	15:14	6	-0.022	2.26	0.001	-0.002	-0.0126	0.1274	0.52	0.345	0.048	0.003	0.413	0.026	0.14	0.056	1.83	0.0137	0.0296	0.5837
8-BLANK	1/26/96	15:16	-0.013	-0.014	0.007	0.001	-0.002	0.0105	0	0.001	-0.013	0	0.002	0.035	0.001	0.12	-0.003	0.003	0.0006	0.0007	0.0015
USGS T107	1/26/96	15:19	0.231	-0.008	14	0.019	0.0118	0.0422	0.0344	0.071	2.42	0.056	0.0182	24.3	0.039	0.14	0.024	4.48	0.0687	0.0045	0.0979
8-III-A-3 0-7.5cm	1/26/96	15:22	3.76	0.038	2.38	0.0052	4E-04	-0.0059	1.121	3.22	0.436	0.304	0.0084	0.309	0.003	0.63	0.121	2.03	0.0175	0.0239	1.042
8-III-A-3 7.5-11cm	1/26/96	15:25	4.09	0.005	2.38	0.0042	-0.002	-0.0023	0.8208	2.16	0.353	0.173	0.0069	0.548	0.721	0.42	0.083	1.91	0.0151	0.0285	0.9478
8-III-A-1 0-5.5cm	1/26/96	15:28	4.78	0.063	2.92	0.006	-0.002	-0.001	1.534	6.88	0.474	0.456	0.0135	0.445	0.005	0.67	0.151	2.84	0.0233	0.0427	1.812
8-III-A-1 5.5-8.5cm	1/26/96	15:31	5.45	0.412	3.62	0.0047	4E-04	0.0092	1.026	65.3	0.354	0.21	0.0996	0.389	0.001	1.3	0.2	6.19	0.0416	0.0345	2.424
8-III-A-1 0-5.5cm DUP	1/26/96	15:34	6.12	0.052	3.06	0.0069	2E-04	0.0016	1.534	6.99	0.517	0.461	0.0141	0.472	0.005	0.7	0.155	3.12	0.024	0.0476	1.828
8-III-A-1 8-14cm	1/26/96	15:37	6.71	0.001	2.25	0.0008	-0.001	-0.0062	0.1973	0.729	0.326	0.042	0.0025	0.462	0.006	0.13	0.059	1.94	0.0143	0.0264	0.4793
8-III-A-1 14-25cm	1/26/96	15:40	4.4	-0.018	2.02	0.0018	-0.002	-0.0071	0.1899	0.702	0.274	0.039	0.0033	0.393	0.001	0.14	0.067	1.71	0.0132	0.0236	0.4772
8-III-A-1 25-42cm	1/26/96	15:43	4.68	-0.019	2.13	0.002	-0.001	-0.0033	0.1377	0.631	0.273	0.042	0	0.41	0.004	0.15	0.057	1.73	0.0136	0.0242	0.427
8-III-B-3 30-41cm DUP	1/26/96	15:45	42.3	-0.007	2.69	0.0022	-0.002	-0.0041	0.3524	1.39	1.36	0.096	0.003	0.502	0.003	0.26	0.108	4.27	0.0202	0.0812	0.4824
8-III-B-2 0-8cm DUP	1/26/96	15:51	15	0.53	5.08	0.0065	0.0007	0.0337	2.807	142	0.62	0.176	0.213	0.64	0.066	3.6	0.118	13.6	0.063	0.0444	3.761
8-BLANK	1/26/96	15:55	-0.003	-0.019	0.008	0.0008	0.0011	0.0096	0.0004	0.018	-0.005	5E-04	0.0023	0.021	0.002	0.11	-0.001	0.005	0.0004	0.0007	0.0017
8-STR.BLANK1	1/26/96	15:58	0.024	-0.017	0.053	0.0006	-0.001	-0.0056	0.0019	0.024	0.046	0	0.0023	0.018	-0.001	0.08	0.008	0.061	5E-04	-0.0037	0.0063
8-STR.BLANK2	1/26/96	16:00	0.072	-0.017	0.078	0.0006	-0.001	-0.0052	0.0011	0.047	0.34	0	0.0039	0.04	-0.001	0.07	0.003	0.194	2E-04	-0.0015	0.0022
8-STR.BLANK3	1/26/96	16:04	0.017	-0.008	0.06	0.0013	-0.002	-0.0077	0.0019	0.038	0.049	8E-04	0.0007	0.029	0.019	0.09	0	0.067	6E-04	-0.0041	0.0163
8-USGST107	1/26/96	16:06	0.243	-0.002	14.6	0.0204	0.0142	0.0576	0.0352	0.077	2.47	0.059	0.0202	24.9	0.035	0.21	0.036	4.62	0.0705	0.0075	0.1044
8-USGST107	1/26/96	16:08	0.236	0.013	14	0.0188	0.0139	0.0478	0.0337	0.073	2.4	0.056	0.0189	23.9	0.038	0.17	0.014	4.43	0.0682	0.006	0.1001
USGS T117	1/26/96	16:11	0.077	0.003	24.9	0.0029	0.0037	0.04	0.0078	0.542	11.5	0.257	0.0151	22.8	0.01	0.4	0.004	6.65	0.2886	0.0056	0.2172
USGS T91	1/26/96	16:14	0.406	-0.036	30.1	0.0431	0.0114	0.0354	1.037	0.329	11.4	2.722	0.0046	6.45	0.03	0.22	0.009	7.87	0.128	0.0045	6.871

ICAPES data of bead digests analysis of 2/6/96																						
Sample Name	Date	Al	As	Ca	Cd	Co	Cr	Cu	Fe	Mg	Mn	Mo	Nb	Ni	P	Pb	Si	Sr	Ti	Zn		
8-STR.BLANK A	2/6/96	0.046	-0.028	0.054	7E-04	-0.002	0.0074	0.0007	0.083	0.02	0.0017	-0.002	-0.037	0.018	0.13	-0.042	0.035	-6E-04	-0.0044	0.0043		
8-STR.BLANK B	2/6/96	-0.068	0.024	0.037	3E-04	0	0.0264	-0.001	0.033	-0.006	0.0007	0.002	0.054	0.004	0.15	-0.006	0.024	0.0014	0.0039	0.0028		
RINSE HCL	2/6/96	0.07	-0.017	0.043	8E-04	-4E-04	0.0176	-7E-04	0.037	0.095	0.0012	-0.002	0.008	-0.002	0.2	-0.018	0.073	0.0004	0.0007	0.2679		
RINSE HCL	2/6/96	0.079	-0.027	0.044	2E-04	0	0.0151	-0.001	0.038	0.106	0.0012	0.001	0.008	-0.002	0.24	-0.027	0.074	0.0004	0.0007	0.2693		
RINSE HCL	2/6/96	0.038	-0.005	0.071	0	-0.002	0.0173	-0.001	0.012	0.006	0.0002	-0.002	0.002	-0.001	0.24	-0.019	0.019	0.0001	-0.0011	0.0039		
RINSE HCL	2/6/96	0.043	-0.027	0.07	0.002	-6E-04	0.0152	-0.001	0.012	0.017	0.0007	-0.002	0.003	0.002	0.2	-0.028	0.016	0.0001	-0.0015	0.0039		
8-STR.BLANK A	2/6/96	0.043	-0.031	0.064	9E-04	-8E-04	0.0153	-0.001	0.095	0.022	0.0027	-0.002	-0.015	0.013	0.18	-0.041	0.035	-1E-04	-0.0026	0.0059		
50% HCL	2/6/96	0.041	-0.016	0.073	0.001	0.0006	0.0179	-7E-04	0.015	0.023	0.0007	-5E-04	0.017	0.001	0.26	-0.023	0.02	0.0002	-0.0004	0.0043		
NEW 50%HCL	2/6/96	0.025	-0.008	0.013	0.001	-0.002	0.0016	-0.0007	0.025	0.026	0.001	-0.004	-0.038	-0.006	0.18	0.043	0.007	-5E-04	-0.0037	0.0583		
NEW 50%HCL	2/6/96	0.028	-0.01	0.013	3E-04	-0.001	0.0101	-0.001	0.035	0.036	0.0007	-0.004	-0.023	-0.005	0.25	-0.04	0.012	-3E-04	-0.0026	0.06		
NEW 50%HCL	2/6/96	0.008	-0.017	0.019	1E-04	-6E-04	0.0043	-0.002	0.017	0.04	-0.0002	-0.001	-0.031	-0.003	0.13	-0.037	0.011	-3E-04	-0.0026	0.0569		
8-STR.BLANK X	2/6/96	0.005	0.002	0.059	-5E-04	-0.001	0.006	-0.004	0.034	0.015	0.0002	-0.004	0.008	-0.005	0.14	-0.039	0.03	0.0002	-0.0015	0.0035		
8-STR.BLANK X	2/6/96	0.006	0.003	0.059	0	-0.001	0.0063	-0.003	0.036	0.029	0.0007	-8E-04	0.003	-0.004	0.16	-0.053	0.027	0.0002	-0.0015	0.0035		
8-STR.BLANK Y	2/6/96	0.008	-0.002	0.061	0	-0.002	0.0124	-0.003	0.037	0.015	0.001	-0.003	0.009	-0.005	0.2	-0.065	0.025	0.0003	-0.0007	0.0031		
8-STR.BLANK Z	2/6/96	0.005	-0.012	0.059	3E-04	-0.003	0.0077	-0.004	0.032	0.018	0.0002	-0.003	0.01	-0.008	0.14	-0.044	0.019	0.0001	-0.0015	0.0022		
8-BEADBLANK4	2/6/96	11.6	-0.002	0.97	0	-0.003	0.0124	-4E-04	0.133	0.477	0.0163	-0.002	0.198	0.007	0.16	-0.05	1.37	0.0049	0.0206	0.007		
8-BEADBLANK5	2/6/96	9	0	0.911	2E-04	-0.002	0.0109	-0.003	0.08	0.324	0.016	-0.003	0.173	-0.005	0.16	-0.038	1.11	0.0047	0.0198	0.005		
8-BEADBLANK6	2/6/96	20.3	-0.025	1.62	2E-04	-0.002	0.0116	0	0.138	0.709	0.0244	-0.001	0.313	-0.001	0.17	-0.03	2.13	0.0086	0.0411	0.0075		
8-BEADBLANK7	2/6/96	5.77	0.003	1.11	-5E-04	-0.002	0.0132	-0.002	0.14	0.25	0.0133	-0.005	0.215	0.011	0.16	-0.054	1.11	0.0057	0.0158	0.0076		
50% HCL	2/6/96	0	-0.004	0.022	-6E-04	-0.005	0.0071	-0.003	0.056	0.04	0.0002	-0.003	-0.012	-0.009	0.12	-0.048	0.016	0	-0.0015	0.0569		
50% HCL	2/6/96	-0.008	-0.007	0.022	1E-04	-0.003	0.0057	-0.003	0.062	0.031	0	-0.004	-0.013	-0.008	0.12	-0.037	0.013	-1E-04	-0.0022	0.0573		
8-IA-1-0-13cm	2/6/96	6.53	0.046	2.2	0.004	-0.002	0.0186	0.9043	2.02	0.726	0.5973	-0.001	0.703	0.008	0.52	0.061	1.71	0.0151	0.0341	1.195		
8-IA-1-0-13cm(2)	2/6/96	6.39	0.045	1.65	0.004	-0.001	0.015	0.9246	2	0.575	0.5022	-9E-04	0.511	0.004	0.54	0.053	1.65	0.012	0.0268	1.233		
8-IA-1-13-21cm	2/6/96	3.18	0.201	1.46	0.007	-0.003	0.0168	2.152	15.6	0.484	4.532	0.0204	0.483	0.006	2	0.28	1.82	0.0165	0.0305	1.324		
8-IA-1-21-31cm	2/6/96	3.68	0.04	1.54	0.004	-0.002	0.0215	1.325	3.06	0.872	0.737	0.001	0.947	0.002	0.89	0.093	1.11	0.0135	0.0264	1.472		
8-IA-1-31-41cm	2/6/96	2.55	0.022	1.19	0.003	-0.001	0.017	0.8738	1.53	0.665	0.5971	0.0009	0.892	0.005	0.34	0.018	0.842	0.01	0.0224	0.9862		
8-IB-2-0-7.5cm	2/6/96	3.75	0.106	2.28	0.005	-0.003	0.0168	2.427	5.82	6.56	0.7476	0.0065	0.57	-0.001	0.98	0.142	2.35	0.0214	0.0397	2.425		
8-IB-2-7.5-12cm	2/6/96	4.22	0.111	2.7	0.006	-0.001	0.0161	2.568	6.52	7.005	0.7673	0.0081	0.596	0.009	1.2	0.187	2.48	0.0244	0.0487	2.14		
8-IB-2-12-16cm	2/6/96	3.69	0.003	1.9	0.002	-0.002	0.0163	0.8785	1.34	0.648	0.6623	-0.002	0.843	0.008	0.39	0.007	1.31	0.0129	0.0246	1.138		
USGS T107	2/6/96	0.252	0.003	14.8	0.019	0.0111	0.0661	0.0323	0.064	2.46	0.0581	0.018	26	0.032	0.34	0	4.7	0.07	0.0092	0.1053		
USGS T107	2/6/96	0.238	0.026	14.2	0.019	0.0113	0.058	0.0316	0.062	2.37	0.0552	0.0131	25.2	0.033	0.29	-0.017	4.52	0.0678	0.0073	0.101		
USGS T117	2/6/96	0.08	-0.005	25.2	0.003	0.0008	0.0541	0.0051	0.529	11.2	0.2596	0.0088	23.7	0.007	0.53	-0.045	6.69	0.2847	0.0077	0.222		
USGS T117	2/6/96	0.094	-0.002	25.1	0.003	0.0019	0.0561	0.0051	0.53	11.1	0.2584	0.0082	23.5	0.006	0.53	-0.04	6.66	0.2831	0.0077	0.2204		
FLUSHING HCL	2/6/96	0.06	0.002	0.055	4E-04	-0.002	0.0182	0.002	0.038	0.1	0.0012	-0.002	0.023	-0.003	0.23	-0.05	0.075	0.0008	0.0018	0.2665		
90%EST.HCL	2/6/96	0.03	-0.01	0.077	5E-04	-0.002	0.0088	-7E-04	0.01	0.222	0.0007	-0.004	-0.004	-0.005	0.19	-0.051	0.024	0.0001	-0.0018	0.0039		
BLANK-STD1	2/6/96	0.009	-0.001	0.025	5E-04	-0.002	0.0242	-7E-04	0.001	0.026	0.0002	-0.003	0.061	-0.007	0.25	-0.058	0	0.0011	0.0029	0.0039		
8-IB-2-16-21.5cm	2/6/96	3.61	-0.007	1.82	0.002	-8E-04	0.0123	0.6719	0.656	0.6	0.6392	-0.003	0.787	0.006	0.33	-0.029	1.21	0.0109	0.0246	0.9286		
8-IB-2-22-32cm	2/6/96	4.73	0.007	2.07	0.002	-2E-04	0.0076	0.3593	0.467	0.573	0.3993	-0.002	0.732	0.008	0.24	-0.047	1.45	0.0118	0.0294	0.7007		
8-IB-2-32-42cm	2/6/96	6.22	0.002	2.19	0.002	-0.001	0.0151	0.5772	1.3	0.696	0.3741	0.0011	0.786	0.004	0.34	-0.011	1.56	0.0144	0.0296	0.6086		
8-IB-3-0-8cm	2/6/96	5.99	0.038	2.2	0.006	-0.001	0.0208	1.589	3.83	1.31	1.462	0.0016	1.64	0.01	0.75	0.033	2.04	0.0152	0.0312	2.478		
8-IB-3-8-11cm	2/6/96	4.24	0.052	1.97	0.007	0.0011	0.0159	1.33	6.7	1.5	1.644	0.0059	2.12	0.011	0.58	0.072	1.72	0.0129	0.0231	2.808		
8-IB-3-11-21cm	2/6/96	4.03	-0.006	2.02	0.011	0.0011	0.0225	1.174	1.05	1.88	2.017	-1E-04	2.45	0.011	0.48	-0.006	1.18	0.0119	0.0206	3.233		
8-IB-3-21-31cm	2/6/96	3.72	0.009	1.52	0.007	-8E-04	0.0217	1.022	0.95	1.19	1.238	-0.001	1.78	0.008	0.41	-0.021	1.16	0.01	0.0195	1.93		
8-IB-3-21-31cm DUP	2/6/96	3.65	-0.001	1.44	0.005	-8E-04	0.0195	1.015	0.939	1.18	1.221	0.0003	1.78	0.008	0.38	-0.023	1.11	0.0095	0.0184	1.906		
8-IB-3-11-21cm DUP	2/6/96	5.1	-0.001	2.32	0.009	-0.001	0.0209	1.179	1.24	2.98	2.04	-0.002	2.53	0.012	0.5	0.034	3.54	0.0136	0.0745	3.203		
8-IB-3-31-41cm	2/6/96	5.03	0.007	1.45	0.006	-0.002	0.0173	0.7979	0.986	1.02	0.9862	-0.001	1.07	0.006	0.43	0.001	1.2	0.0095	0.0217	1.552		
8-IB-3-31-41cm DUP	2/6/96	4.85	-0.009	1.49	0.007	-8E-04	0.018	0.7957	0.981	0.994	0.9759	-0.003	1.09	0.009	0.45	-0.002	1.22	0.0097	0.0217	1.537		
8-IIB-2-0-7cm	2/6/96	5.23	0.036	2.34	0.002	-0.002	0.0138	1.041	3.09	0.428	0.348	0.0022	0.312	-0.001	0.88	0.047	1.98	0.0177	0.0354	0.9546		
8-IIB-2-7-12cm	2/6/96	3.49	0.058	2.16	0.003	-8E-04	0.0627	1.101	4.17	0.733	0.3653	0.0063	0.354	0.137	0.93	0.07	1.77	0.0172	0.0327	0.9012		
BLANK-STD1	2/6/96	0.013	-0.015	0.028	-1E-04	-0.001	0.0241	-0.003	0.003	0.02	0.0005	-0.004	0.074	-0.006	0.26	-0.049	-0.002	0.0012	0.0029	0.0049		
8-IIB-2-12-19.5cm	2/6/96	4.16	0.037	2.4	0.001	-0.001	0.0153	0.7743	4.22	0.398	0.2685	0.0055	0.494	0.273	0.74	0.042	1.95	0.0164	0.0036	0.7093		
8-IIB-2-20-30cm	2/6/96	4.62	0.017	2.01	0.001	-4E-04	0.0281	0.4255	3.13	0.365	0.185	0.0044	0.441	0.271	0.56	0.015	1.68	0.0129	0.0292	0.4571		
8-IIB-2-20-30cm DU	2/6/96	3.87	0.019	2	0.002	0	0.2244	0.4265	3.13	0.352	0.184	0.0058	0.441	0.268	0.51	0.01	1.61	0.0128	0.0279	0.4567		
8-IIB-2-30-40.5cm	2/6/96	4.73	-0.013	2.04	1E-04	-0.003	0.0135	0.1848	0.63	0.328	0.067	-0.002	0.378	0.001	0.32	-0.021	1.6	0.0122	0.0263	0.2521		
USGS T107	2/6/96	0.243	0.001	14.7	0.021	0.0119	0.0769	0.0327	0.061	2.41	0.0579	0.0153	25.9	0.034	0.42	0.008	4.69	0.0696	0.0099	0.1059		
USGS T117	2/6/96	0.094	0.005	25.4																		

Appendix

ICAPES data of head digest analysis of Z/15/96

Sample Name	Analysis Date	Analysis Time	Al	As	Ca	Cd	Co	Cr	Cu	Fe	Mg	Mn	Mu	Na	Ni	P	Pb	Si	Sr	Ti	Zn
STD1-Blank	2/15/96	9:51	0.066	-0.01	0.002	0.001	0.002	0.189	0.001	9E-05	-2E-04	7E-04	-3E-04	0.052	0.0052	0.003	0.045	0.003	0.015	0.015	0.00185
STD2	2/15/96	9:55	2.088							1.557			1.635	11.04	1.7486					2.861	
STD4	2/15/96	9:58		0.975				8.133								0.664		11.89		19.26	2.25
STD8	2/15/96	10:01			18.77	2.05	2.935		1.493		-0.0214	2.044					7.283				
STD1	2/15/96	10:06	0.014	-0.01	0.006	0.002	0.003	0.006	0.002	0.001	0.011	9E-04	4E-04	0.002	0.007	0.11	0.014	0.01	7E-04	-2E-04	0.0015
USGS T107	2/15/96	10:09	0.254	0.087	12.9	0.018	0.013	0.033	0.032	0.056	2.26	0.052	0.016	23	0.038	0.15	0.038	4.23	0.063	0.002	0.0098
USGS T117	2/15/96	10:12	0.102	0.025	23.1	0.004	0.007	0.03	0.006	5.07	11	0.239	0.013	21.9	0.015	0.35	0.008	6.33	0.272	0.003	0.2001
8-STRBLANKL	2/15/96	10:16	0.023	0.017	0.024	8E-04	0.001	0.006	6E-04	0.004	0.013	9E-04	0.002	-0.02	0.008	0.14	0.009	0.052	-0	-0.002	0.0013
8-STRBLANKM	2/15/96	10:19	0.021	-0	0.016	0.001	0.002	0.015	-3E-04	0.002	0.018	9E-04	0.003	-0.03	0.006	0.17	0.021	0.044	0	-3E-04	0.0004
8-STRBLANKN	2/15/96	10:22	0.018	0.006	0.006	0.002	0.003	0.013	-6E-04	0.003	0.002	9E-04	0.001	-0.02	0.001	0.15	0.023	0.038	-0	-0.001	0.0023
8-BEADBLANK8	2/15/96	10:25	1.06	-0	0.242	7E-04	4E-04	0.011	0	0.01	0.05	0.003	0.001	0.024	0.008	0.14	0.006	0.478	0.001	0.002	0.0002
8-BEADBLANK9	2/15/96	10:27	1.06	-0	0.282	3E-04	9E-04	0.016	-6E-04	0.01	0.06	0.004	0	0.038	0.006	0.16	0.011	0.396	0.001	0.003	0.0014
8-BEADBLANK10	2/15/96	10:30	4.94	-0	0.778	0.001	0.001	0.017	-6E-04	0.027	0.246	0.011	0.001	0.117	0.002	0.16	0.008	0.929	0.004	0.016	0.0024
8-IC3 8.5-10cm	2/15/96	10:35	0.937	0.018	0.527	0.001	0.001	0.016	0.218	0.729	0.095	0.082	8E-04	0.084	0.007	0.28	0.05	0.857	0.004	0.006	0.2064
8-IC3 8.0-8.5cm	2/15/96	10:38	22.2	0.083	2.59	0.007	0.002	0.024	1.621	5	1.13	0.614	0.008	0.317	0.003	1	0.198	5.47	0.023	0.072	1.875
8-IC3 20-30cm	2/15/96	10:41	0.934	0.063	0.525	0.002	0.003	0.015	0.113	1.48	0.087	0.037	0.002	0.074	0.006	0.36	0.028	0.446	0.004	0.006	0.1625
8-IC3 30-34cm	2/15/96	10:44	1.09	0.029	0.691	0.002	4E-04	0.015	0.083	2.08	0.03	0.04	0.002	0.088	0.004	0.69	0.05	0.329	0.007	0.007	0.2901
8-IA-3 15-21.5cm	2/15/96	10:47	3.7	0.015	1.07	0.003	0.002	0.027	0.33	0.944	0.236	0.042	6E-04	0.282	0.005	0.45	0.069	0.893	0.009	0.02	0.9985
STD1-BLANK	2/15/96	10:50	0.025	-0.001	0.001	0.001	4E-04	0.021	6E-04	0	0.002	7E-04	4E-04	0.028	0.002	0.23	0.011	0.01	7E-04	0.002	0.0023
8-IC3 30-34cm DUP	2/15/96	10:52	2.28	0.023	0.739	0.001	9E-04	0.012	0.082	2.04	0.132	0.04	0.005	0.09	0.002	0.69	0.063	0.62	0.007	0.008	0.2797
8-IC3 30-34cm DUP	2/15/96	10:54	2.13	0.026	0.757	0.002	0.002	0.017	0.082	2.08	0.122	0.04	0.001	0.095	0.005	0.75	0.068	0.63	0.007	0.009	0.289
USGS T107	2/15/96	10:57	0.276	0.019	15.1	0.021	0.016	0.059	0.033	0.067	2.52	0.059	0.018	24.9	0.043	0.31	0.04	4.89	0.07	0.007	0.1076
USGS T107	2/15/96	10:59	0.276	0.003	14.2	0.021	0.017	0.05	0.033	0.062	2.41	0.056	0.019	24.3	0.045	0.29	0.049	4.62	0.067	0.005	0.1006
8-STRBLANKL DUP	2/15/96	11:02	0.083	-0.01	0.014	8E-04	8E-04	0.013	-6E-04	0.002	0.009	2E-04	0.002	-0.03	0.005	0.22	0.008	0.03	-0	-0.001	0.0015
USGS T107	2/15/96	11:05	0.276	0.003	14.7	0.021	0.015	0.055	0.033	0.065	2.46	0.058	0.017	24.7	0.041	0.28	0.034	4.75	0.069	0.006	0.104
USGS T117	2/15/96	11:10	0.124	0.005	25.1	0.005	0.007	0.045	0.037	0.543	11.4	0.259	0.015	23.4	0.016	0.56	0.028	6.78	0.286	0.005	0.2179
8-IC3 20-30cm DUP	2/15/96	11:16	6.65	0.011	0.923	0.003	0.001	0.009	0.084	2.08	0.266	0.044	0.003	0.111	0.005	0.73	0.045	1.02	0.008	0.014	0.2801
8-IC3 20-30cm DUP	2/15/96	11:19	2.17	0.05	0.607	0.003	6E-04	0.014	0.115	1.44	0.118	0.059	0.001	0.091	0.001	0.44	0.048	0.362	0.005	0.007	0.1624
Std 1 blank	2/15/96	11:39	0.144	-0.008	0.005	0.005	0.007	0.067	0.003	0.053	0.017	0.003	0.002	0.094	0.017	0.59	0.037	0.238	0.002	0.012	0.0106
COMP 1-4	2/15/96	11:44	2.73	2.99	31.6	3.06	3.058	3.001	2.667	2.96	27.1	2.983	2.97	25.6	2.99	3.6	3.09	29.3	2.607	2.796	3.138
COMP 1-4	2/15/96	12:10	2.68	2.96	31	3.012	3	2.949	2.638	2.92	26.6	2.942	2.932	25.6	2.98	3.7	2.98	28.9	2.596	2.762	3.09
COMP 1-4	2/15/96	12:21	2.68	2.99	31.4	3.039	3.027	2.978	2.646	2.92	26.6	2.962	2.956	25.2	3.02	3.7	3.02	29	2.984	2.761	3.12
COMP 1-4	2/15/96	12:35	2.67	2.97	31.2	3.018	3.01	2.96	2.647	2.9	26.4	2.947	2.939	25.7	3.01	3.7	3.01	28.8	2.584	2.747	3.097
COMP 1-4	2/15/96	12:41	2.69	2.99	31.2	3.017	3.004	2.955	2.661	2.91	26.5	2.947	2.932	25.9	3.01	3.7	2.99	28.8	2.595	2.751	3.093
COMP 1-4	2/15/96	12:48	2.66	2.97	31.2	3.016	3.002	2.95	2.632	2.9	26.3	2.938	2.938	25.5	3.01	3.6	2.98	28.6	2.568	2.731	3.091
COMP 1-4	2/15/96	13:00	2.66	3.02	31	3.004	2.987	2.938	2.637	2.88	26.2	2.93	2.921	25.5	3.02	3.7	2.97	28.6	2.571	2.725	3.081
COMP 1-4	2/15/96	13:08	2.65	2.98	31	2.999	2.978	2.93	2.646	2.88	26.2	2.936	2.919	25.8	3	3.7	2.95	28.6	2.58	2.727	3.072
COMP 1-4	2/15/96	13:19	2.64	2.98	31.3	3.026	3.002	2.954	2.631	2.89	26.1	2.945	2.946	25.5	3.04	3.8	2.94	28.6	2.567	2.722	3.096
COMP 1-4	2/15/96	13:27	2.64	2.99	31	3.008	2.978	2.936	2.622	2.87	25.9	2.925	2.926	25.5	3.02	3.8	2.95	28.4	2.557	2.706	3.077
STD1-Blank	2/15/96	13:44	0.091	-0.01	0.003	0.002	0.004	0.344	0.002	0.005	0.001	-2E-04	0.062	0.0089	0.115	0.087	0.021	0.019	0.018	0.00342	
STD2	2/15/96	13:47	2.143							1.741			1.928	10.55	1.9563					3.083	
STD4	2/15/96	13:50		1.179				9.455													
STD2	2/15/96	13:53			22.57	2.321	2.936		1.52		4.2017	2.307						8.451		19.86	2.6319
STD1-BLANK	2/15/96	13:58	-0.01	-0.02	0.001	9E-04		0.001	-0.001	0.001	7E-04	9E-04	0	-0.003				0.01	0.02	3E-04	9E-04
STD1-BLANK	2/15/96	14:00	0	0	0	6E-04		0	-0.001	0.009	-6E-04	7E-04	0.017	-0.003				0.01	0.015	3E-04	4E-04
COMP 1-4	2/15/96	14:08	2.54	2.55	26.3	2.66		2.991	2.6	25.6	2.604	2.599	25.9	2.61				2.63	26.1	2.591	2.574
STD1-BLANK	2/15/96	14:08	-0.01	-0.01	0.001	3E-04		-6E-04	0	0.001	0	0.004	0.014	-0.004				0	0.014	3E-04	4E-04
8-IC3 8.5-10cm	2/15/96	14:09	1.28	0.012	0.699	7E-04		0.21	0.72	0.094	0.078	0.005	0.123	-0.001				0.024	0.317	0.004	0.006
8-IA-3 14.5-21cm	2/15/96	14:13	1.29	0.009	0.683	0.001		0.048	0.465	0.139	0.017	0.001	0.175	-0.001				-0.01	0.42	0.004	0.005
8-IC3 20-30cm	2/15/96	14:16	3.62	0.035	1.02	8E-04		0.113	1.44	0.139	0.041	0.006	0.133	-0.004				0.02	0.912	0.006	0.01
8-IA-3 22-31.5cm	2/15/96	14:19	4.17	0.004	1.66	0.001		0.228	0.392	0.254	0.044	0.002	0.322	0				0.089	1.27	0.011	0.022
COMP 1-4	2/15/96	14:23	2.58	2.61	26.9	2.745		2.635	2.64	26.1	2.635	2.647	26.7	2.69							

ICAPES data for bead samples analyzed on 3/7/96																					
Sample Name	Date	Time	Al	As	Ca	Cd	Co	Cr	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Si	Sr	Tl	Zn
STD1-Blank	3/7/96	13:17	0.06738	-0.00733	0.00128	0.00109	0.00171	0.21623	0.00109	0.00342	-4E-05	0.00057	0.00023	0.06052	0.00409	0.10566	0.04228	0.00704	0.01814	0.01714	0.0018
STD2	3/7/96	13:20			22.308	2.38004	2.87047		1.57228		4.36314	2.20128					8.64423		19.3695		3.13347
STD3	3/7/96	13:23	2.20152							1.75166			2.4709	11.0166	1.95971					3.03971	
STD4	3/7/96	13:26		1.61195				8.9989								0.925		12.9843			
STD6	3/7/96	13:29																			
BLANK-STD 1	3/7/96	13:31	0	0.012	0.003	0.0003	0.0008	0.0062	0.0006	-0.007	0.008	0	-0.0002	0.022	0.001	0.02	-0.007	0.022	0.0003	0.0001	0.0006
USGS T107	3/7/96	13:34	0.225	0.003	12.8	0.0166	0.0139	0.0252	0.0318	0.046	2.29	0.0517	0.0165	22.7	0.031	0.03	0.009	4.13	0.0641	0.0014	0.0888
USGS T117	3/7/96	13:37	0.068	0.002	23.4	0.0022	0.0038	0.0256	0.0064	0.496	11.2	0.2428	0.0129	21.9	0.007	0.26	-0.013	6.31	0.2792	0.0024	0.2028
8-STR.BLANK	3/7/96	13:40	0.033	0.008	0.032	0.0005	0.0003	0.0056	-0.0003	-0.003	0.019	0.0006	-0.0017	0.011	0.003	0.01	-0.019	0.021	0.0002	-0.0022	0
8-STR.BLANK2	3/7/96	13:43	-0.004	0.014	0.013	0.0005	0	0.0071	0.0003	-0.002	0.013	0.0002	0.0013	-0.012	0.004	0.04	-0.019	0.012	0	-0.0014	0.0006
8-STR.BLANK3	3/7/96	13:45	-0.016	-0.004	0.005	0.0006	-0.0002	0.0113	-0.0012	-0.005	0.009	-0.0002	-0.0027	-0.001	0.001	0.03	-0.016	0.001	0	-0.0011	-0.0005
8-BEADBLANK11	3/7/96	13:47	2.72	-0.002	0.732	0.001	0.0007	0.0137	0.0009	0.013	0.146	0.0078	0.0002	0.133	-0.001	0.07	-0.005	0.726	0.0041	0.0118	0.0013
8-BEADBLANK12	3/7/96	13:50	1.54	0	0.459	-0.0002	0	0.0115	-0.0003	0.011	0.079	0.0045	-0.0013	0.074	-0.001	0.03	-0.031	0.474	0.0026	0.0055	0.0006
8-BEADBLANK13	3/7/96	13:53	1.76	0.004	0.508	-0.0006	-0.0002	0.012	0.0009	0.01	0.088	0.0054	-0.0017	0.083	-0.004	0.07	-0.004	0.532	0.0029	0.0068	0.0012
8-BEADBLANK14	3/7/96	13:55	1.59	0.005	0.451	-0.0002	0.001	0.0153	0.0003	0.005	0.079	0.0039	-0.0011	0.082	0.001	0.08	-0.005	0.499	0.0027	0.0068	-0.0001
8-IIB-1 8-14cm	3/7/96	13:58	1.69	0.075	0.941	0.0028	0.0008	0.0139	0.3992	2.75	0.167	0.1261	0.0028	0.137	0.001	0.49	0.161	0.83	0.0098	0.0252	0.7463
8-IIB-1 14-26cm	3/7/96	14:00	1.54	0.061	0.781	0.0005	-0.0002	0.0126	0.2649	1.13	0.171	0.0578	0.0007	0.151	-0.002	0.19	0.085	0.667	0.0067	0.0203	0.5797
8-IIB-1 26-35cm	3/7/96	14:03	1.61	0.142	1.02	0.0019	0.0012	0.0189	0.4125	3.62	0.215	0.1197	0.0054	0.157	-0.001	0.52	0.188	0.988	0.0106	0.0298	0.4673
8-IIB-1 35-41cm	3/7/96	14:06	2.12	0.176	1.17	0.0032	0.0005	0.0175	0.2964	3.1	0.229	0.1164	0.0041	0.196	0.004	0.46	0.123	0.959	0.0116	0.0295	0.5005
8-IIC-2 30-40cm	3/7/96	14:09	1.58	0.015	0.65	0.0013	0.0002	0.0133	0.1015	0.314	0.109	0.037	0.0024	0.14	0.001	0.1	0.01	0.546	0.0042	0.0093	0.1354
8-IIC-2 0-10cm	3/7/96	14:12	2.35	0.062	1.72	0.0033	-0.0007	0.0211	1.077	3.7	0.352	0.3179	0.0044	0.182	-0.001	0.75	0.114	1.59	0.0143	0.0343	0.9279
8-IIB-GW5 7/6/95	3/7/96	14:14	0.158	0.991	141	0.0044	0.01	0.1625	0.0249	229	15.6	23.09	0.3262	43.5	0.004	1.8	0.036	31.3	0.5723	0.0216	0.095
8-IIA-GW5 7/6/95	3/7/96	14:18	1.88	2.79	367	0.0118	0.0525	0.2291	1.061	223	63.2	48.6	0.3428	68.8	0.025	0.88	0.217	34.4	1.208	0.0863	2.373
8-STR.BLANK1	3/7/96	14:21	-0.012	0.028	0.102	-0.0008	-0.0013	0.0119	0	0.583	0.022	0.0132	0.0003	0.033	-0.002	-0.05	-0.029	0.017	0.0006	0.0005	-0.0002
FLUSH 40%-HCL	3/7/96	14:23	-0.179	-0.003	0.038	-0.0052	-0.0144	0.1113	0.0045	0.526	-0.105	-0.0026	0.007	0.231	0	0.17	0.041	0.03	0.0046	0.0213	-0.0052
FLUSH 40%-HCL	3/7/96	14:28	0.034	0.005	0.015	0.0002	0.001	0.0106	0.0003	0.059	0.012	0.0009	0.0005	0	0.003	0.04	-0.014	-0.001	0	-0.0011	0.0012
FLUSH 40%-HCL	3/7/96	14:30	-0.004	0.021	0.008	-0.0008	-0.0007	0.0149	0.0006	0.072	0.014	0.0002	-0.0014	0.007	0.002	0.07	-0.01	-0.002	0.0002	0.0003	0.0018
USGS T107	3/7/96	14:33	0.238	0.029	14.8	0.019	0.0143	0.0527	0.0339	0.055	2.52	0.0584	0.0157	24.9	0.038	0.13	0.023	4.7	0.0708	0.006	0.1034

Sample duplicates (bead samples): Calculations worksheet
(all concentrations adjusted for sample mass and dilution factor)

Sample Name	Sample dup1			Sample dup2			Sample dup3			Sample dup4			Sample dup5			Sample dup6		
	IC-3 18-32mm	IC-3 18-32mm (2)	% CHG	IB-1 18-32mm	IB-1 18-32mm (2)	% CHG	IA-3 18-32mm	IA-3 18-32mm (2)	% CHG	IA-3 23-33mm	IA-3 23-33mm (2)	% CHG	IA-3 13.6-21.6mm	IA-3 13.6-21.6mm (2)	% CHG	IB-3 25-41mm	IB-3 25-41mm (2)	% CHG
Analysis date	1/26/96	1/26/96		1/26/96	1/26/96		1/26/96	1/26/96		1/26/96	1/26/96		1/26/96	1/26/96		1/26/96	1/26/96	
Analysis time	1024	1028		1038	1041		1130	1133		1147	1153		1211	1214		1237	1240	
Al	23.8	29.4	21.4	34.9	16.8	69.9	18.4	25.5	46.2	53.0	50.8	4.4	36.9	36.1	7.8	41.5	12.8	105.4
As	4.3	3.8	11.2	BDL	BDL	BDL	0.4	0.5	26.0	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL
Cd	16.8	14.1	17.6	8.8	9.8	10.7	12.3	13.3	7.9	23.7	26.7	14.5	7.3	8.6	17.4	11.3	7.4	41.4
Co	BDL	0.10	BDL	0.1	0.1	4.0	BDL	DL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL
Cr	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL
Cu	10.0	10.0	0.2	14.2	14.6	2.3	16.8	18.4	8.6	11.0	10.9	0.9	1.2	1.1	13.2	1.2	1.1	11.5
Fe	270	244	9.9	13.1	15.3	15.5	35.2	39.8	11.3	142	138	2.6	13.7	11.3	20.8	70	5.7	21.1
Mg	4.8	5.2	6.7	14.8	14.1	4.9	5.4	6.3	16.3	5.9	5.2	21.1	1.7	1.8	0.9	2.3723	1.1540	69.1
Mn	7.9	7.4	5.6	16.8	16.8	0.0	6.6	7.3	9.5	3.5	2.9	20.5	0.4	0.4	6.5	0.5	0.4	18.6
Mo	BDL	BDL	BDL	BDL	BDL	BDL	0.0	0.0	1.0	0.2	0.2	5.0	BDL	BDL	BDL	BDL	BDL	BDL
Na	4.3	4.7	4.6	14.7	15.3	4.0	7.2	7.7	6.8	40	4.9	21.6	1.4	1.7	24.0	2.4	1.7	34.5
Ni	0.3	0.2	53.4	BDL	BDL	BDL	0	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL
P	29.6	27.1	8.6	4.4	4.9	11.9	10.0	8.5	16.6	5.6	2.3	4.4	5.9	5.9	15.2	2.3	2.0	12.1
Pb	1.4	1.2	17.2	BDL	0.5	BDL	1.0	1.0	3.4	1.7	1.5	12.8	BDL	BDL	BDL	BDL	BDL	BDL
Si	21.2	19.8	6.4	9.1	6.9	15.4	11.5	14.5	23.1	24.4	24.1	1.1	6.3	6.9	9.9	9.1	5.4	50.2
Sr	0.2	0.2	10.9	0.1	0.1	11.9	0.1	0.1	8.1	0.1	0.1	14.7	0.1	0.1	5.6	0.1	0.1	33.8
Tl	0.2	0.2	8.6	0.2	0.1	10.9	0.2	0.3	14.7	0.2	0.2	17.0	0.2	0.2	0.0	0.2	0.1	47.0
Zn	22.6	19.8	13.2	27.7	28.0	1.2	12.3	13.8	11.8	6.7	6.0	11.5	1.3	1.3	4.9	1.4	1.2	13.4

Statistics on compilation of all sample duplicates (beads)

	STD.		MEAN ± STDDEV.	# OF DATA	STD. 95% CONF. INT. OF MEAN	
	MEAN	DEV.			ERR.	INT. OF MEAN
Al	33.42	27.32	66.80	17	1.68	26.36
As	14.84	8.22	23.78	6	1.54	17.88
Cd	14.03	16.42	24.48	17	8.61	15.23
Co	28.27	NA	NA	1	NA	NA
Cr	NA	NA	NA	8	NA	NA
Cu	28.24	11.87	51.31	3	5.69	27.48
Fe	11.97	38.49	32.45	17	1.21	14.33
Mg	10.43	6.47	16.81	17	8.30	11.18
Mn	21.28	21.76	43.34	17	1.28	24.08
Mo	6.43	6.82	16.48	17	8.35	10.12
Na	11.87	8.42	19.48	7	1.28	15.15
Ni	16.48	12.98	29.43	17	8.76	18.15
P	66.80	45.68	131.29	2	2.38	130
Pb	8.47	4.78	13.77	13	8.36	8.36
Si	8.77	11.38	21.84	8	1.41	12.83
Sr	16.84	13.81	29.88	17	8.77	17.54
Tl	13.88	8.24	19.82	17	8.48	12.82
Zn	28.31	13.84	32.15	17	6.81	19.96
	11.14	7.38	18.24	17	8.42	11.84

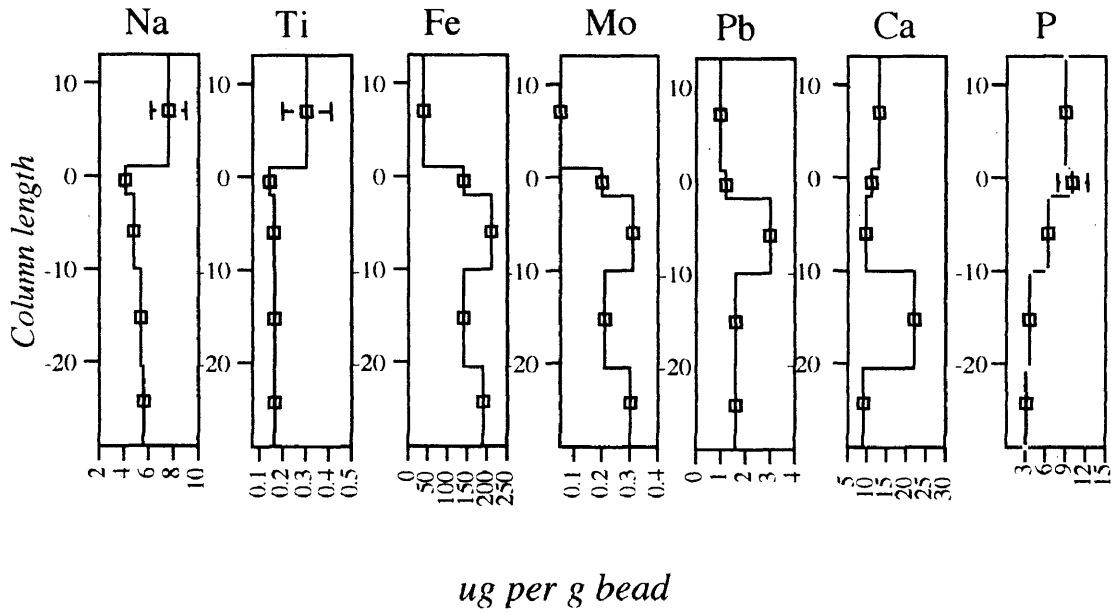
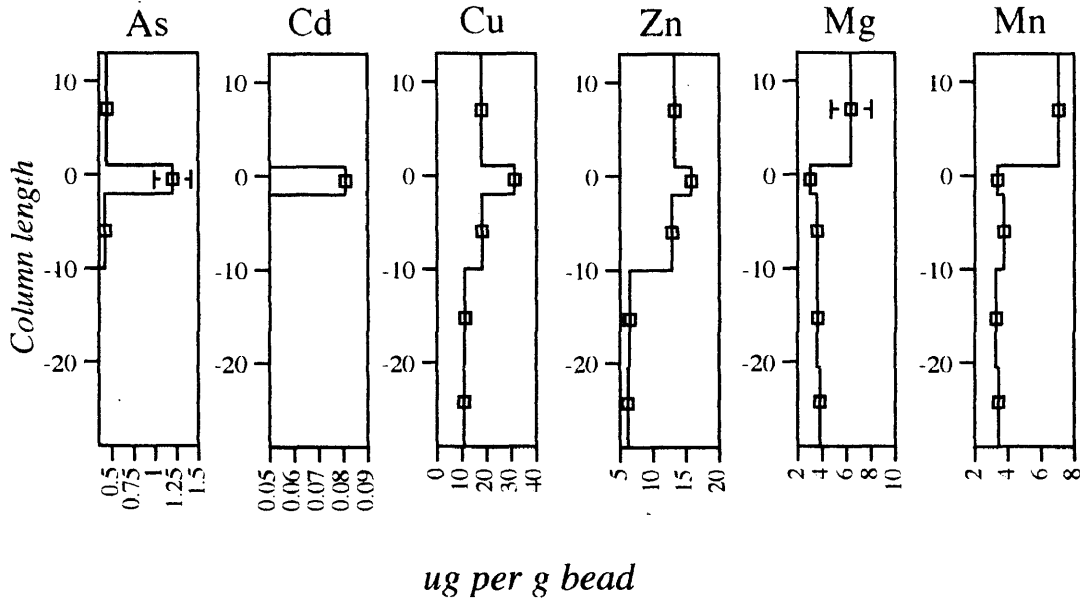
CONCENTRATIONS ALONG BEAD COLUMNS (adjusted to ug/g BEAD)																						
Sample Name	Date	Mass analyzed (g)	Dilution factor	SW/Subs. boundary	Colors on bead tube	Al	As	Cd	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Si	Sr	Ti	Zn	
BEADBLANK1	1/26/96	1.0022	5		(0cm-top);	20	BDL	1.1	BDL	BDL	0.78	BDL	BDL	BDL	BDL	BDL	BDL	2.1	BDL	0.0324	0.024	
BEADBLANK2	1/26/96	0.9939	5		v=very; l=light	11	BDL	1.3	BDL	BDL	0.58	BDL	BDL	BDL	BDL	BDL	BDL	1.9	BDL	0.0332	BDL	
BEADBLANK3	1/26/96	0.9999	5			15	BDL	3.4	BDL	BDL	0.18	0.85	BDL	BDL	0.88	BDL	BDL	3.9	BDL	0.0715	BDL	
BEADBLANK6	2/6/96	1.0076	5			83	BDL	15	BDL	BDL	0.59	3.2	0.160	BDL	2.5	BDL	1.4	BDL	14	0.077	0.213	BDL
BEADBLANK7	2/6/96	1.0012	5			81	BDL	10	BDL	BDL	0.76	2.5	0.119	BDL	1.9	BDL	1.3	BDL	11	0.053	0.150	BDL
BEADBLANK4	2/6/96	1.0023	5			84	BDL	11	BDL	BDL	0.69	3.0	0.129	BDL	1.9	BDL	1.5	BDL	13	0.055	0.165	BDL
BEADBLANK8	2/15/96	0.9961	10			11	BDL	2.4	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	4.8	BDL	BDL	0.0020
BEADBLANK9	2/15/96	1.0071	10			11	BDL	2.8	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	5.9	BDL	BDL	0.0139
BEADBLANK10	2/15/96	1.0016	5			25	BDL	3.9	BDL	BDL	BDL	1.2	0.057	BDL	0.58	BDL	BDL	BDL	4.6	BDL	0.080	0.0120
BEADBLANK11	3/7/96	0.9861	10			28	BDL	7.4	BDL	BDL	BDL	1.5	0.079	BDL	1.35	BDL	BDL	BDL	7.4	BDL	0.120	BDL
BEADBLANK12	3/7/96	1.0065	10			15	BDL	4.6	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	4.7	BDL	0.055	BDL
BEADBLANK13	3/7/96	1.0062	10			17	BDL	5.1	BDL	BDL	BDL	BDL	0.054	BDL	BDL	BDL	BDL	BDL	5.3	BDL	0.068	BDL
BEADBLANK14	3/7/96	0.9742	10			16	BDL	4.6	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	BDL	5.1	BDL	0.070	BDL
IA-1 0-13 cm	2/6/96	1.0076	5	13cm	0-7 green	32	BDL	BDL	BDL	4.54	10	BDL	2.73	BDL	3.0	BDL	2.6	BDL	BDL	BDL	BDL	6.03
IA-1 13-21cm	2/6/96	1.0055	5		7-13 white	16	1.0	BDL	BDL	10.7	78	BDL	2.25	0.10	2.4	BDL	9.9	1.4	BDL	0.082	BDL	6.58
IA-1 21-31cm	2/6/96	1.0047	5		13-21 med. red	18	BDL	BDL	BDL	6.59	15	BDL	3.67	BDL	4.7	BDL	4.4	BDL	BDL	BDL	BDL	7.33
IA-1 31-41cm	2/6/96	1.0009	5		21-31 white, l. red 31-41 white	13	BDL	BDL	BDL	4.37	7.6	BDL	2.98	BDL	4.5	BDL	1.7	BDL	BDL	BDL	BDL	4.93
IA-2 0-12cm	1/26/96	1.0052	5	13 cm	0-13 green	37	0.44	13	0.045	17.8	39	6.4	7.05	0.05	7.6	BDL	9.0	0.97	16	0.128	0.304	13.3
IA-2 12-15cm	1/26/96	1.0042	5		13-15 light red	60	1.2	11	0.081	31.2	140	3.0	3.36	0.20	4.1	BDL	10	1.2	41	0.108	0.142	15.8
IA-2 15-23cm	1/26/96	1.0081	5		15-42 red	103	0.42	9.6	0.054	18.2	210	3.6	3.77	0.31	4.8	BDL	6.4	3.0	59	0.085	0.161	12.9
IA-2 23-33.5cm	1/26/96	0.9994	5			52	0.21	22	BDL	10.9	140	3.6	3.25	0.21	5.4	BDL	3.6	1.6	24	0.117	0.164	6.38
IA-2 33.5-42cm	1/26/96	1.0009	5			48	0.22	9.0	BDL	10.7	190	3.8	3.41	0.30	5.6	BDL	3.1	1.6	16	0.071	0.164	6.13
IA-3 0-10cm	1/26/96	1.0033	5	14 cm	0-10 green	22	0.49	11	0.061	17.8	35	6.6	8.18	BDL	9.7	BDL	10	1.1	13	0.112	0.236	17.9
IA-3 10-20cm	1/26/96	1.0057	5		10-19 light red	27	0.65	8.5	0.063	13.2	75	6.7	7.67	0.09	10	0.65	4.3	1.1	15	0.072	0.159	13.4
IA-3 20-31cm	1/26/96	0.9986	5		19-42 white	29	BDL	8.8	0.066	7.43	5.7	10	9.85	BDL	10	BDL	3.0	BDL	6.2	0.064	0.139	15.5
IA-3 31-42cm	1/26/96	0.9967	5			29	BDL	17	BDL	5.40	5.6	6.6	5.43	BDL	6.1	0.70	2.2	BDL	8.1	0.076	0.189	8.89
IB-1 0-8 cm	1/26/96	0.9972	5			79	BDL	9.5	0.073	14.3	14	16	16.8	BDL	15	BDL	4.6	BDL	12	0.081	0.239	27.8
IB-1 8-12 cm	1/26/96	1.0068	5	12 cm	0-5 green	66	0.70	10	0.053	6.72	50	5.1	6.45	0.06	4.1	BDL	10	1.0	13	0.122	0.258	12.2
IB-1 12-16cm	1/26/96	1.0025	5		5-9 v. light green	13	BDL	7.2	BDL	1.45	3.8	2.0	1.25	BDL	2.5	BDL	2.1	BDL	4.5	0.054	0.124	3.89
IB-1 16-21cm	1/26/96	0.9926	5		9-11.5 light red	17	BDL	5.1	BDL	0.981	1.8	1.4	0.645	BDL	1.6	BDL	1.9	BDL	4.1	0.038	0.104	2.37
IB-1 22-32cm	1/26/96	0.9977	5		11.5-41.5 white	44	BDL	7.6	BDL	2.30	2.6	2.5	1.03	BDL	2.3	BDL	2.1	BDL	6.7	0.050	0.170	3.29
IB-1 32-41.5	1/26/96	1.0017	5			29	BDL	8.7	BDL	3.04	7.2	3.0	1.82	BDL	3.4	BDL	2.3	BDL	6.9	0.066	0.185	3.28
IB-2 0-7.5cm	2/6/96	1.0016	5	12 cm	0-12 green	BDL	0.53	BDL	BDL	12.1	29	BDL	3.73	BDL	2.8	BDL	4.9	0.71	BDL	0.107	0.198	12.1
IB-2 7.5-12cm	2/6/96	0.9965	5		12-42 white	BDL	0.56	BDL	BDL	12.9	33	BDL	3.85	BDL	3.0	BDL	6.0	0.94	BDL	0.121	0.244	10.7
IB-2 12-16cm	2/6/96	1.0054	5			BDL	BDL	BDL	BDL	4.37	6.7	BDL	3.29	BDL	4.2	BDL	1.9	BDL	BDL	BDL	BDL	5.66
IB-2 16-21.5cm	2/6/96	0.9948	5			BDL	BDL	BDL	BDL	3.38	3.3	BDL	3.21	BDL	4.0	BDL	1.7	BDL	BDL	BDL	BDL	4.82
IB-2 22-32cm	2/6/96	1.0039	5			BDL	BDL	BDL	BDL	1.97	2.3	BDL	1.99	BDL	3.6	BDL	1.2	BDL	BDL	BDL	BDL	3.49

Sample Name	Date Analyzed	Mass analyzed (g)	Dilution factor	SW/Subs. boundary	Colors on bead tube	Al	Ag	Ca	Cd	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Si	Sr	Ti	Zn
IB-2 32-42cm	2/6/96	0.9956	5			BDL	BDL	BDL	BDL	2.90	6.5	BDL	1.88	BDL	3.9	BDL	1.7	BDL	BDL	BDL	BDL	3.36
IB-3 0-8cm	2/6/96	1.0073	5	11 cm	0-8 green	BDL	BDL	BDL	BDL	7.89	19	6.5	7.26	BDL	8.1	BDL	3.7	BDL	BDL	BDL	BDL	12.3
IB-3 8-11cm	2/6/96	1.0068	5		8-10 light red	BDL	BDL	BDL	BDL	6.61	33	7.4	8.16	BDL	11	BDL	2.9	BDL	BDL	BDL	BDL	13.9
IB-3 11-21 cm	2/6/96	1.0000	5		10-41 white	BDL	BDL	BDL	BDL	5.88	5.7	12	10.1	BDL	12	BDL	2.5	BDL	BDL	BDL	0.238	16.1
IB-3 21-31cm	2/6/96	0.9944	5			BDL	BDL	BDL	BDL	5.12	4.7	6.0	6.18	BDL	9.0	BDL	2.0	BDL	BDL	BDL	BDL	9.64
IB-3 31-41cm	2/6/96	0.9966	5			BDL	BDL	BDL	BDL	4.00	4.9	5.1	4.92	BDL	5.4	BDL	2.2	BDL	BDL	BDL	BDL	7.75
IC-1 0-8cm	1/26/96	0.9973	5	8 cm	0-8 green with red	21	0.48	9.1	BDL	5.55	18	2.6	1.98	BDL	2.0	0.25	3.6	0.71	8.4	0.075	0.238	6.79
IC-1 8-11 cm	1/26/96	0.9958	5		streak on tube	21	0.58	11	BDL	2.65	30	4.2	4.81	BDL	3.7	0.73	5.0	1.1	7.4	0.094	0.174	8.04
IC-1 11-30 cm	1/26/96	1.0028	5		8-11 v. light red	18	BDL	8.0	BDL	4.49	7.6	4.1	3.52	BDL	4.9	0.34	2.3	0.48	4.4	0.060	0.137	9.84
IC-1 30-42cm	1/26/96	0.9982	5		11-30 white	23	2.3	13	0.123	18.0	300	3.8	6.46	0.43	4.4	1.6	40	1.2	18.1	0.268	0.179	21.9
					30-42 med. red																	
IC-2 0-14cm	1/26/96	0.9999	5	14 cm	0-14 green	16	0.32	12	BDL	7.12	19	2.9	3.04	BDL	2.0	0.55	4.0	0.78	9.3	0.098	0.228	10.5
IC-2 14-18cm	1/26/96	0.9973	5		14-18 l. green/ white	54	0.62	13	0.049	8.89	27	5.2	3.56	BDL	3.8	0.23	4.3	1.6	13	0.117	0.309	8.11
IC-2 18-32 cm	1/26/96	0.9976	5		18-32 med. red	27	4.0	15	0.108	10.0	260	5.0	7.64	0.35	4.6	0.21	30	1.3	20	0.230	0.188	21.22
IC-2 32-41cm	1/26/96	1.0031	5		32-40 white/ v.l. red	14	BDL	6.8	BDL	2.75	7.0	2.1	1.69	BDL	2.8	0.99	2.4	BDL	3.8	0.048	0.119	3.29
IC-3 0-9.5cm	1/26/96	0.9964	5	9.5 cm	0-9.5 green	25	0.315	10	BDL	7.59	20	2.7	2.53	BDL	3.3	0.52	3.8	0.94	9.4	0.081	0.210	8.02
IC-3 9.5-21cm	1/26/96	1.0043	5		9.5-21 v. l. red	23	BDL	7.1	BDL	4.54	14	2.1	1.30	BDL	2.2	0.20	2.2	BDL	6.1	0.050	0.178	2.61
IC-3 21-32cm	1/26/96	1.0002	5		21-32 med. red	56	BDL	6.2	BDL	5.54	66	2.8	1.03	0.1	1.9	0.29	2.3	BDL	15	0.047	0.253	3.84
IC-3 32-42cm	1/26/96	1.0083	5		32-42 light red	30	BDL	6.0	BDL	4.72	27	2.6	1.62	BDL	2.7	0.32	1.8	BDL	7.2	0.042	0.171	3.40
IIA-1 0-10.5cm	2/6/96	0.9928	5	15.5 cm	0-10 dark green	BDL	BDL	BDL	BDL	4.22	10	BDL	1.30	BDL	BDL	BDL	3.0	BDL	BDL	BDL	BDL	3.61
IIA-1 10.5-15.5cm	2/6/96	1.0030	5		10-15.5 green	BDL	BDL	BDL	BDL	2.91	9.2	BDL	0.880	BDL	BDL	BDL	2.9	BDL	BDL	BDL	BDL	2.12
IIA-1 15.5-22cm	2/6/96	1.0004	5		15.5-20.5 l. green	BDL	BDL	BDL	BDL	3.11	6.3	BDL	0.612	BDL	BDL	BDL	2.4	BDL	BDL	BDL	BDL	1.79
IIA-1 22.5-33cm	2/6/96	0.9999	5		20.5-42 white	BDL	BDL	BDL	BDL	2.98	5.5	BDL	0.616	BDL	BDL	BDL	2.5	BDL	BDL	BDL	BDL	4.46
IIA-1 33-42cm	2/6/96	0.9983	5			BDL	BDL	BDL	BDL	1.12	3.4	BDL	0.410	BDL	BDL	BDL	2.3	BDL	BDL	BDL	BDL	2.17
IIA-2 0-10.5cm	2/6/96	0.9974	5	13.5 cm	0-13.5 green	BDL	BDL	BDL	BDL	4.75	17	BDL	1.82	BDL	BDL	BDL	5.0	BDL	BDL	BDL	BDL	4.12
IIA-2 11-13.5cm	2/6/96	0.9997	5		13.5-42 white	BDL	BDL	BDL	BDL	2.35	9.9	BDL	0.780	BDL	BDL	BDL	3.4	BDL	BDL	BDL	BDL	2.54
IIA-2 13.5-21.5 cm	2/6/96	0.9988	5			BDL	BDL	BDL	BDL	1.82	7.6	BDL	0.648	BDL	BDL	BDL	2.8	BDL	BDL	BDL	BDL	1.98
IIA-2 22-32 cm	2/6/96	1.0092	5			BDL	BDL	BDL	BDL	0.846	3.7	BDL	0.365	BDL	BDL	BDL	2.1	BDL	BDL	BDL	BDL	1.08
IIA-2 32-42cm	2/6/96	0.9929	5			BDL	BDL	BDL	BDL	0.766	4.4	BDL	0.340	BDL	BDL	BDL	2.0	BDL	BDL	BDL	BDL	0.96
IIA-3 0-8.5cm	1/26/96	1.0013	5	8.5 cm	0-8.5 green	103	0.27	11	BDL	6.59	23	4.1	1.44	BDL	1.6	BDL	4.7	0.69	20	0.094	0.367	5.61
IIA-3 8.5-13.5cm	1/26/96	1.0083	5		8.5-13.5 red	35	2.1	20	BDL	4.84	270	2.2	1.81	0.38	1.5	BDL	20	0.85	21	0.274	0.178	5.18
IIA-3 13.5-21.5cm	1/26/96	1.0005	5		13.5-41 white	37	BDL	8.0	BDL	1.16	13	1.7	0.432	BDL	1.6	BDL	4.2	BDL	6.6	0.069	0.158	1.31
IIA-3 21.5-29.5cm	1/26/96	1.0009	5			31	BDL	6.5	BDL	0.941	3.8	1.5	0.295	BDL	1.5	BDL	2.1	BDL	5.5	0.048	0.139	1.92
IIA-3 30-41cm	1/26/96	1.0043	5			24	BDL	6.7	BDL	1.33	3.7	1.3	0.325	BDL	1.4	BDL	2.4	BDL	5.6	0.050	0.146	1.52
IIIB-1 0-8cm	1/26/96	0.9996	5	8cm	0-8 green	16	BDL	10	BDL	4.85	16	1.9	1.62	BDL	1.6	0.19	4.1	0.66	8.4	0.083	0.192	5.13
IIIB-1 8-14cm	3/7/96	0.9967	5		8-14 v. light red	17	0.75	9.4	BDL	4.01	27.59	1.68	1.27	BDL	1.37	BDL	4.92	1.62	8.3	0.098	0.253	7.49
IIIB-1 14-26cm	3/7/96	1.0014	5		14-26 white	15	0.6	7.8	BDL	2.65	11.28	1.71	0.58	BDL	1.51	BDL	1.90	0.85	6.7	0.067	0.203	5.79

Sample Name	Date	SW/Subs. boundary	Colors on bead tube	Al	Ag	Ca	Cd	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Si	Sr	Ti	Zn
IIB-1 26-35cm	3/7/96		26-41 light red	16	1.41	10.14	BDL	4.10	35.96	2.14	1.19	BDL	1.56	BDL	5.17	1.87	9.8	0.105	0.296	4.65
IIB-1 35-41cm	3/7/96			21	1.77	11.76	BDL	2.96	31.15	2.30	1.17	BDL	2.0	BDL	4.62	1.24	9.6	0.117	0.296	5.03
IIB-2 0-7cm	2/6/96	7 cm	0-6 green	BDL	BDL	BDL	BDL	5.23	16	BDL	1.75	BDL	BDL	BDL	4.4	BDL	BDL	BDL	BDL	4.79
IIB-2 7-12cm	2/6/96		6-8 light green	BDL	BDL	BDL	BDL	5.30	21	BDL	1.83	BDL	BDL	0.68	4.6	BDL	BDL	BDL	BDL	4.51
IIB-2 12-19.5cm	2/6/96		8-40.5 white	BDL	BDL	BDL	BDL	3.85	21	BDL	1.34	BDL	BDL	1.36	3.7	BDL	BDL	BDL	BDL	3.33
IIB-2 20-30cm	2/6/96			BDL	BDL	BDL	BDL	2.13	16	BDL	0.923	BDL	BDL	1.35	2.7	BDL	BDL	BDL	BDL	2.29
IIB-2 30-40.5cm	2/6/96			BDL	BDL	BDL	BDL	0.922	3.1	BDL	0.334	BDL	BDL	BDL	1.6	BDL	BDL	BDL	BDL	1.26
IIB-3 0-7cm	1/26/96	10.5 cm	0-7 green	22	0.29	10	BDL	3.96	14	2.7	1.49	BDL	3.4	BDL	4.3	0.53	9.8	0.079	0.190	4.26
IIB-3 7-15cm	1/26/96		7-9 v. light red	23	0.84	12	BDL	4.09	21	2.2	1.81	BDL	2.6	BDL	4.1	1.2	10	0.102	0.272	14.1
IIB-3 15-25cm	1/26/96		9-41 white	16	BDL	7.8	BDL	1.44	7.0	1.5	0.382	BDL	2.1	BDL	2.9	BDL	6.0	0.058	0.170	2.80
IIB-3 25-41cm	1/26/96			27	BDL	9.3	BDL	1.18	6.4	1.8	0.423	BDL	2.0	BDL	2.2	BDL	7.3	0.062	0.191	1.27
IIC-1 0-11cm	1/26/96	15 cm	0-11 green	50	0.27	13	BDL	4.84	16	3.0	1.79	BDL	2.5	BDL	4.1	0.67	14	0.095	0.276	5.44
IIC-1 11-20cm	1/26/96		11-30 light red	29	1.0	12	BDL	5.82	63	2.5	2.26	0.09	1.9	BDL	6.5	2.2	15	0.106	0.375	4.47
IIC-1 20-30cm	1/26/96		30-42 white	53	0.90	14	BDL	1.69	97	2.6	1.30	0.13	2.0	BDL	10	0.61	14	0.141	0.250	3.60
IIC-1 30-42cm	1/26/96			17	BDL	8.0	BDL	0.85	3.5	1.4	0.407	BDL	2.0	BDL	2	BDL	6.2	0.054	0.144	4.08
IIC-2 0-4.5cm	2/15/96			19	0.32	12	BDL	5.80	18	2.2	2.10	BDL	1.8	BDL	3.9	0.52	10	0.093	0.162	5.49
IIC-2 4.5-8.5cm	2/15/96			28	0.24	12	BDL	4.74	15	2.2	1.81	BDL	2.4	BDL	2.8	0.46	10	0.088	0.172	4.06
IIC-2 8.5-11cm	2/15/96	8.5 cm	0-8.5 green	20	BDL	11	BDL	3.35	11	1.7	1.50	BDL	2.6	BDL	2.0	BDL	9.2	0.670	0.126	3.38
IIC-2 11-21cm	2/15/96		8.5-40 white	20	BDL	9.9	BDL	2.46	7.7	1.5	0.843	BDL	2.4	BDL	1.5	BDL	7.9	0.062	0.122	2.41
IIC-2 21-30cm	2/15/96			12	BDL	6.9	BDL	1.29	4.0	0.90	0.390	BDL	BDL	BDL	BDL	BDL	4.8	BDL	BDL	1.49
IIC-2 30-40cm	3/7/96			16	BDL	6.5	BDL	1.01	3.1	1.1	0.37	BDL	1.4	BDL	BDL	BDL	5.4	0.14	0.090	1.34
IIC-3 0-8.5cm	2/15/96	8.5cm	0-8.5 cm green	111	0.42	13	BDL	8.12	25	5.7	3.07	BDL	1.6	BDL	5.0	0.99	27	0.115	0.362	9.39
IIC-3 8.5-10cm	2/15/96		8.5-10 cm white	14	BDL	7.5	BDL	2.62	8.9	1.2	0.980	BDL	1.3	BDL	3.4	BDL	6.0	BDL	0.08	2.45
IIC-3 10-20 cm	2/15/96		10-20 cm med. red	15	8.1	15	BDL	5.36	230	1.1	1.41	0.34	1.0	BDL		BDL	21	BDL	0.07	7.41
IIC-3 20-30cm	2/15/96		20-30 cm l. red	23	BDL	7.7	BDL	1.13	14	1.1	0.388	BDL	BDL	BDL	3.6	BDL	6.8	0.150	0.08	1.58
IIC-3 30-34 cm	2/15/96		30-34 cm white	18	BDL	7.3	BDL	0.825	21	1.2	0.396	BDL	BDL	BDL	7.1	BDL	5.9	0.053	0.08	2.86
IIC-3 34-41cm	2/15/96		34-41 cm red	13	3.1	26	BDL	1.68	210	1.5	1.90	0.30	BDL	BDL		BDL	10	0.410	0.06	14.3
IHA-1 0-5.5 cm	1/26/96	8 cm	0-5.5 green	27	BDL	15	BDL	7.68	35	2.5	2.29	0.07	2.3	BDL	3.4	0.77	15	0.118	0.226	9.11
IHA-1 5.5-8.5cm	1/26/96		5.5-8 bright red	27	2.1	18	BDL	5.16	330	1.8	1.06	0.50	2.0	BDL	6.5	1.0	31	0.209	0.173	12.2
IHA-1 8-14cm	1/26/96		8-42 white	33	BDL	11	BDL	0.983	3.6	1.6	0.209	BDL	2.3	BDL	BDL	BDL	9.7	0.071	0.132	2.39
IHA-1 14-25cm	1/26/96			22	BDL	10	BDL	0.948	3.5	1.4	0.196	BDL	2.0	BDL	BDL	BDL	8.5	0.066	0.118	2.38
IHA-1 25-42cm	1/26/96			23	BDL	11	BDL	0.687	3.1	1.4	0.207	BDL	2.0	BDL	BDL	BDL	8.6	0.068	0.121	2.13
IHA-2 0-9.5cm	2/15/96	9.5cm	0-9.5 green	15	BDL	10	BDL	6.30	25	1.8	1.50	BDL	1.9	BDL		BDL	9.1	0.087	0.138	5.49
IHA-2 9.5-14.5cm	2/15/96		9.5-14.5 bright red	13	0.59	9.7	BDL	3.68	190	1.6	0.593	0.30	1.7	BDL		BDL	24	0.110	0.09	5.74
IHA-2 14.5-32 cm	2/15/96		14.5-42 white	18	BDL	7.0	BDL	0.437	4.8	1.5	0.183	BDL	2.0	BDL		BDL	5.0	BDL	0.06	0.963
IHA-2 32-42cm	2/15/96			9	BDL	4.8	BDL	0.269	2.1	1.4	0.153	BDL	2.1	BDL		BDL	3.4	BDL	BDL	0.724

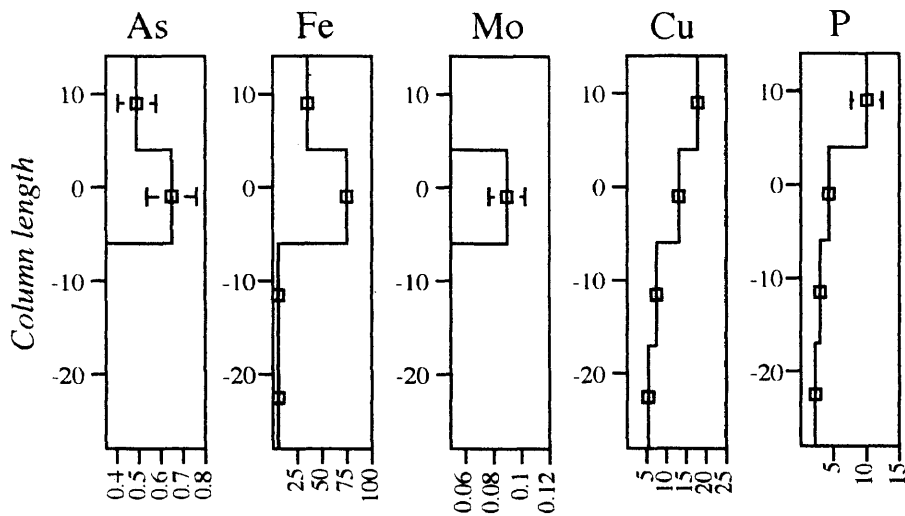
Sample Name	Date	Mass	Dilution	SW/Subs.	Colors on																	
	Analyzed	analyzed (g)	factor	boundary	head tube	Al	As	Ca	Cd	Cu	Fe	Mg	Mn	Mo	Na	Ni	P	Pb	Si	Sr	Ti	Zn
IIIA-3 0-7.5cm	1/26/96	0.9990	5	11 cm	0-7 dark green	19	BDL	12	BDL	5.61	16	2.2	1.52	BDL	1.5	BDL	3.2	0.61	10	0.088	0.170	5.22
IIIA-3 7.5-11cm	1/26/96	1.0014	5		7-11 light green	20	BDL	12	BDL	4.10	11	1.8	0.865	BDL	2.7	3.60	2.1	BDL	9.5	0.075	0.142	4.73
IIIA-3 11-15cm	2/15/96	0.9996	5		11-15 red	12	0.72	7.8	BDL	4.77	60	0.86	0.401	0.09	1.0	BDL		0.55	8.7	0.084	0.07	4.26
IIIA-3 15-21.5cm	2/15/96	1.0016	5		15-42 white	18	BDL	5.3	BDL	1.65	4.7	1.2	0.211	BDL	1.4	BDL	2.2	BDL	4.2	0.043	0.100	4.98
IIIA-3 22-31.5cm	2/15/96	1.0074	5			21	BDL	8.2	BDL	1.13	1.9	1.3	0.218	BDL	1.6	BDL		BDL	6.3	0.054	0.109	3.04
IIIA-3 31.5-42cm	2/15/96	1.0016	5			17	BDL	6.8	BDL	0.882	1.8	0.92	0.129	BDL	1.1	BDL		BDL	5.6	0.046	0.08	1.47
IIIB-1 0-9 cm	2/15/96	0.9965	10	12.5 cm	0-9.5 green	15	BDL	11	BDL	5.74	20	2.0	1.66	BDL	1.4	BDL	3	BDL	9.8	0.086	0.127	6.82
IIIB-1 9-12cm	2/15/96	1.0021	5		9.5-11 red	28	3.2	31	BDL	9.72	240	3.0	4.05	0.35	2.4	BDL	30	1.2	21	0.616	0.258	7.21
IIIB-1 12-20cm	2/15/96	0.9946	5		11-41 white	22	BDL	9.4	BDL	1.23	3.0	1.5	0.324	BDL	2.4	BDL	BDL	0.25	7.5	0.070	0.108	1.10
IIIB-1 20.5-30.5 cm	2/15/96	1.0041	10			170	0.68	31	BDL	16.4	58	10	5.18	BDL	3.0	BDL	10	1.6	50	0.237	0.835	13.2
IIIB-1 31-41cm	2/15/96	1.0004	10			13	BDL	7.3	BDL	2.64	7.6	1.4	0.547	BDL	0.95	BDL	2	BDL	6.4	0.084	0.119	1.96
IIIB-2 0-8cm	1/26/96	0.9948	5	0 cm	0-8 bright red	55	2.7	24	BDL	14.1	700	2.5	0.860	1.1	3.0	0.32	20	0.42	66	0.310	0.188	18.8
IIIB-2 8-18cm	1/26/96	0.9992	5	(top 12 cm	8-29 white	25	BDL	11	BDL	0.693	4.9	1.6	0.165	BDL	1.8	0.28	BDL	BDL	9.0	0.061	0.148	1.13
IIIB-2 18-29cm	1/26/96	1.0012	5	chopped off; all subsurface)		19	BDL	9.1	BDL	0.518	3.3	1.3	0.152	BDL	1.6	0.56	BDL	BDL	6.5	0.050	0.103	0.940
IIIB-3 0-4cm	1/26/96	1.0077	5	9 cm	0-2.5 red	27	1.4	18	BDL	8.63	120	2.1	1.39	0.18	2.5	BDL	10	1.1	17	0.241	0.171	8.16
IIIB-3 4-9cm	1/26/96	1.0013	5		2.5-41 white	24	BDL	11	BDL	3.02	7.0	1.7	0.382	BDL	2.4	BDL	1.5	0.56	9.0	0.090	0.167	4.26
IIIB-3 9-21cm	1/26/96	0.9926	5			20	BDL	10	BDL	1.93	5.6	1.4	0.355	BDL	2.1	BDL	1.5	0.46	7.8	0.078	0.128	4.64
IIIB-3 21-30cm	1/26/96	0.9982	5			28	BDL	11	BDL	3.05	5.7	1.8	0.264	BDL	1.7	BDL	1.4	0.59	9.1	0.079	0.173	3.44
IIIB-3 30-41cm	1/26/96	1.0026	5			117	BDL	12	BDL	1.70	6.2	4.2	0.417	BDL	2.1	BDL	1.2	0.56	14	0.089	0.271	2.33
IIIC-1 0-8cm	1/26/96	1.0014	5	9.5 cm	0-8 green w/ red	30	BDL	16	BDL	3.58	13	3.3	1.59	BDL	7.2	BDL	1.9	0.47	13	0.108	0.195	4.93
IIIC-1 8-11cm	1/26/96	0.9977	5		streak on tube	25	1.9	20	BDL	4.60	200	2.1	2.47	0.31	3.0	BDL	20	0.81	21	0.273	0.165	10.1
IIIC-1 11-28cm	1/26/96	1.0051	5		8-11 red	20	BDL	11	BDL	0.869	3.1	1.5	0.372	BDL	4.6	BDL	1.1	BDL	7.5	0.065	0.106	2.79
IIIC-1 28-42cm	1/26/96	1.0056	5		11-42 white	17	BDL	10	BDL	0.727	2.2	1.5	0.470	BDL	6.5	BDL	0.8	BDL	6.7	0.063	0.104	1.84
IIIC-2 0-10	3/7/96	1.0016	10	30 cm	0-21 green	23	BDL	17	BDL	10.8	3.7	3.5	3.17	BDL	1.8	BDL	7.5	1.14	16	0.140	0.340	9.26
IIIC-2 10-21 cm	2/15/96	1.0026	5		21-32 bright red	23	0.84	18	BDL	15.6	71	3.2	2.52	0.10	1.5	BDL	9.5	1.4	18	0.160	0.349	10.9
IIIC-2 21-32 cm	2/15/96	1.0003	5		39-42 v. light red	21	BDL	14	BDL	13.5	220	1.6	0.824	0.34	1.3	BDL	4.7	BDL	31	0.122	0.145	12.2
IIIC-2 32-39cm	2/15/96	0.9977	5			19	BDL	9.6	BDL	1.33	21	1.2	0.289	BDL	1.5	BDL	1.5	BDL	8.9	0.060	0.096	2.29
IIIC-2 39-42cm	2/15/96	0.8606	5			17	BDL	9.0	BDL	1.29	36	1.0	0.290	0.07	1.1	BDL	1.6	BDL	9.4	0.060	0.080	2.24
IIIC-3 0-11cm	1/26/96	1.0027	5	14 cm	0-11 green	31	0.24	17	BDL	5.92	23	2.6	1.55	BDL	2.4	BDL	3.1	0.55	14	0.119	0.234	5.57
IIIC-3 11-17cm	1/26/96	0.9989	5		11-17 light red	22	0.48	12	BDL	3.11	35	1.5	0.537	0.06	2.1	BDL	2.7	0.67	11	0.082	0.145	3.23
IIIC-3 17-31cm	1/26/96	1.0038	5		17-42 white	22	BDL	10	BDL	1.05	3.4	1.3	0.286	BDL	1.6	0.18	0.9	BDL	8.3	0.061	0.137	3.51
IIIC-3 31-42cm	1/26/96	1.0021	5			30	BDL	11	BDL	0.636	2.6	1.7	0.239	BDL	2.1	0.13	BDL	BDL	9.1	0.068	0.148	2.91
IIIC-2 10-21 ALGAE COATINGS					32-39 white	46	1.9	42	0.122	37.9	130	9.0	8.05	0.20	2.7	BDL	20	3.7	40	0.372	0.909	29.8

Bead Column IA-2

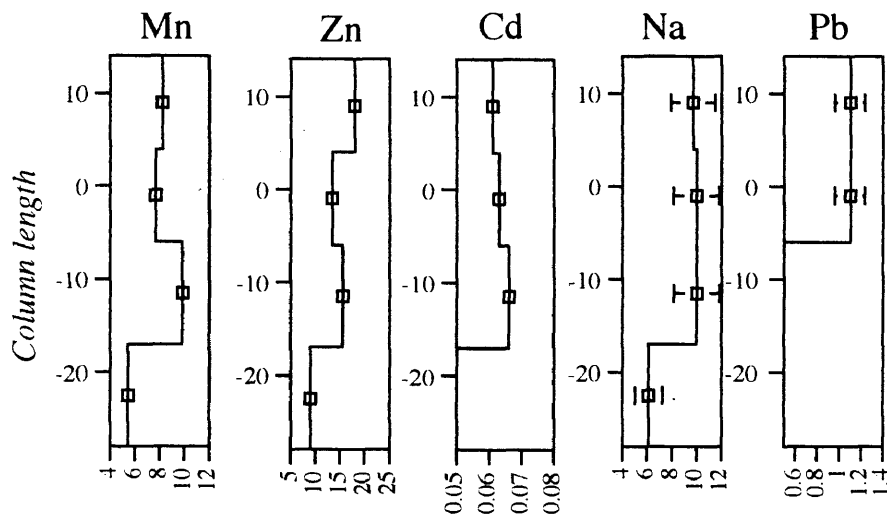


BDL/ No significant trends: Al, Si, Sr

Bead Column IA-3



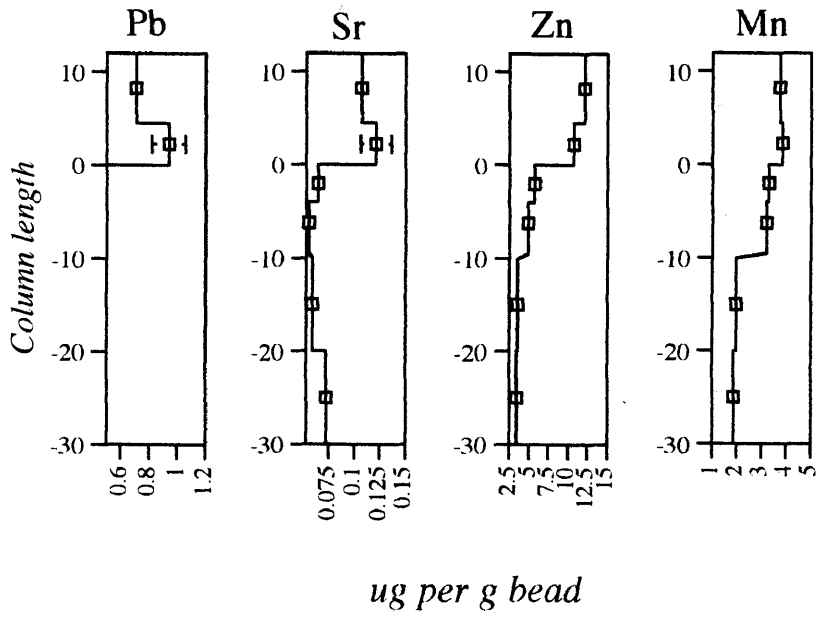
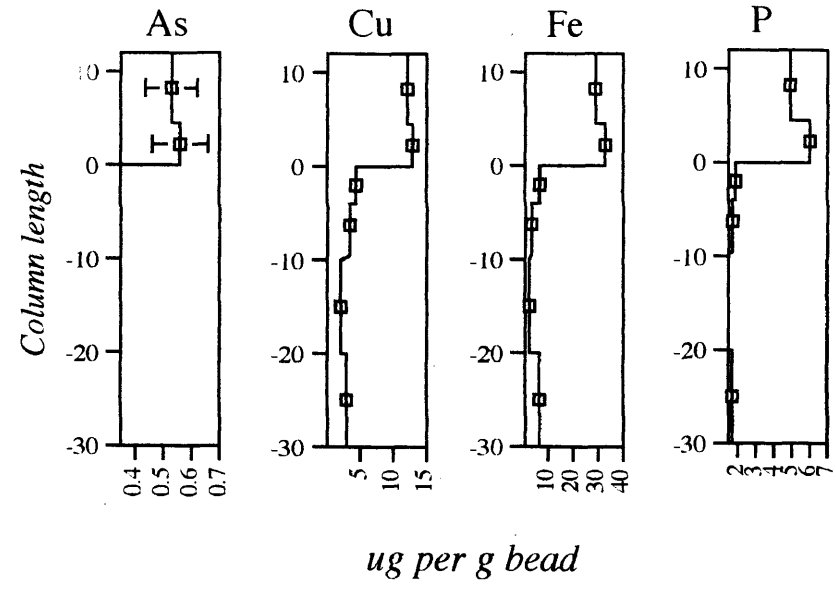
ug per g bead



ug per g bead

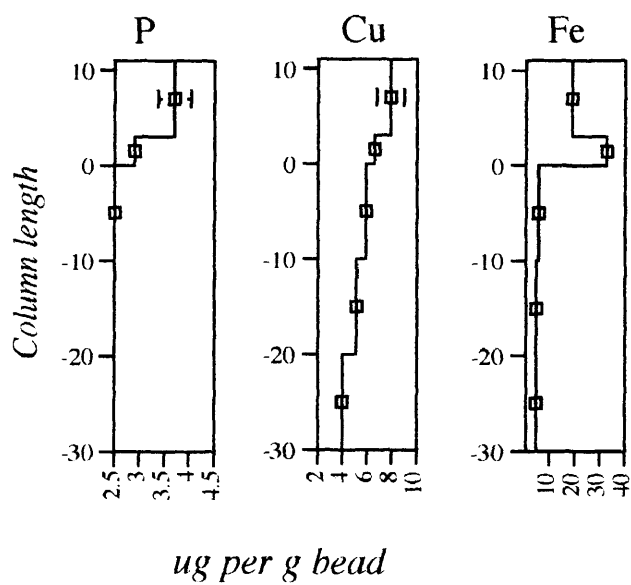
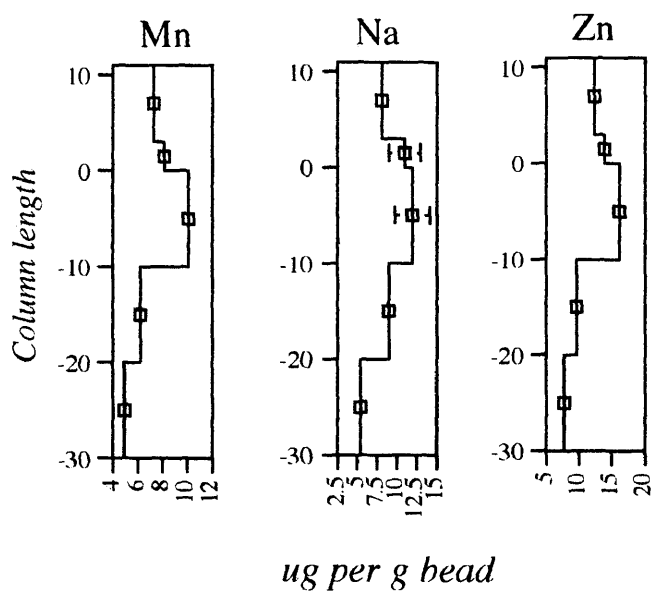
*BDL or insignificant trend:
Al, Ca, Mg, Si, Ti*

Bead Column IB-2



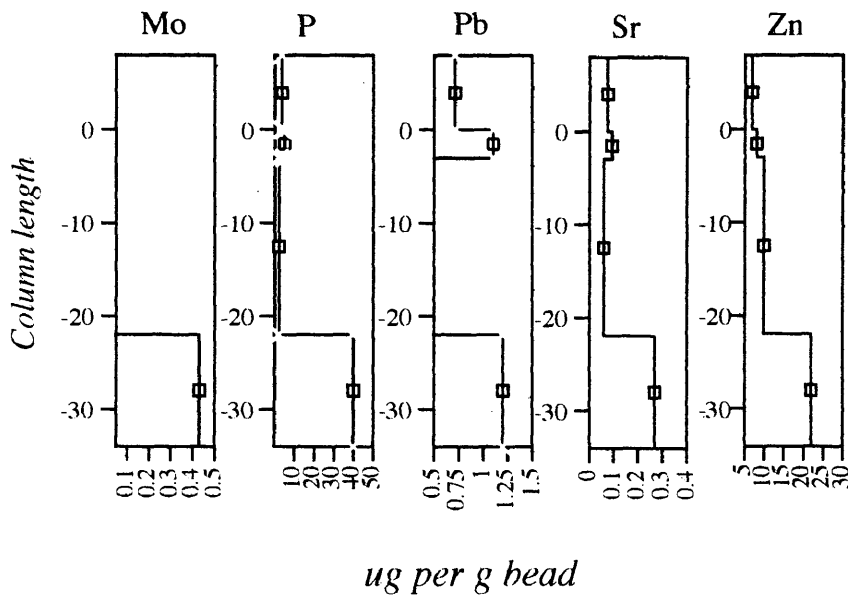
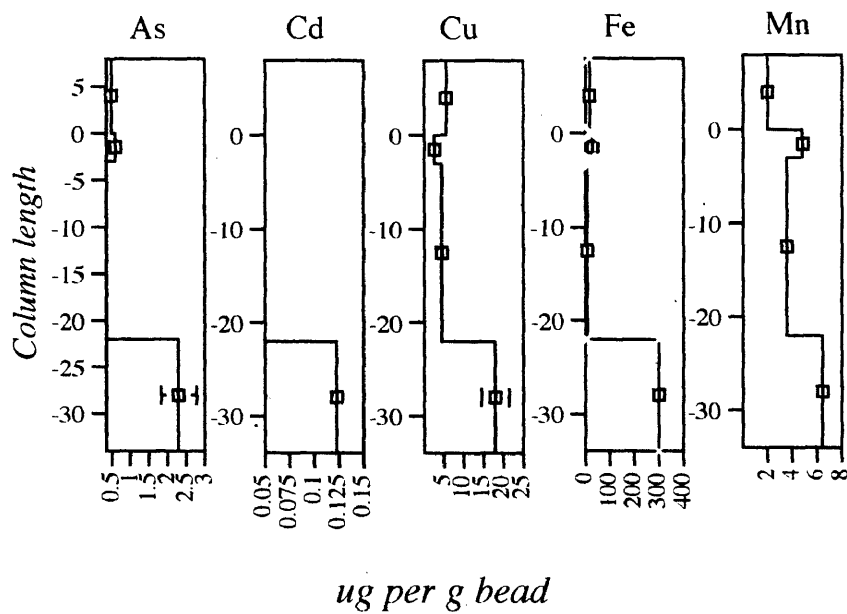
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Bead Column IB-3



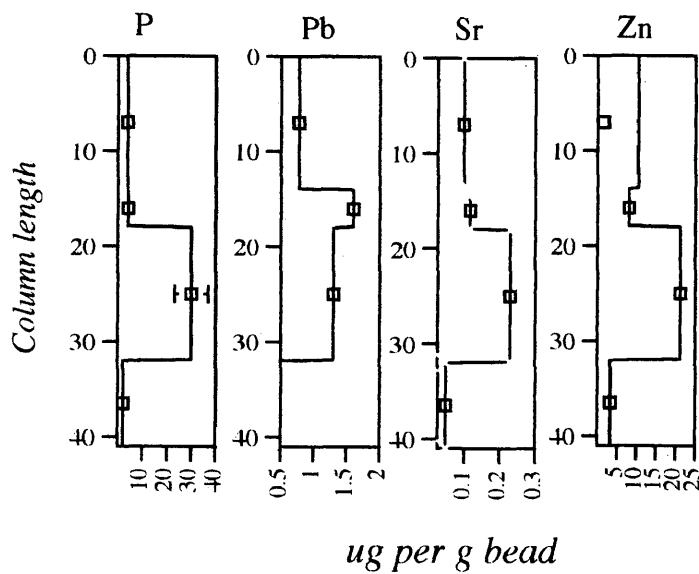
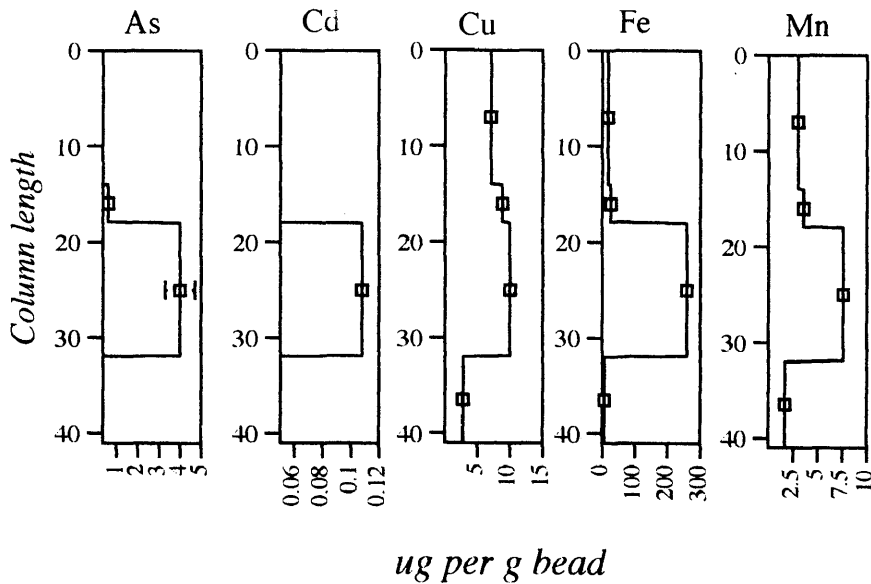
*BDL/ No significant trends:
Al, Ca, Cd, Mo, Pb, Si, Sr,
Ti*

Bead Column IC-1



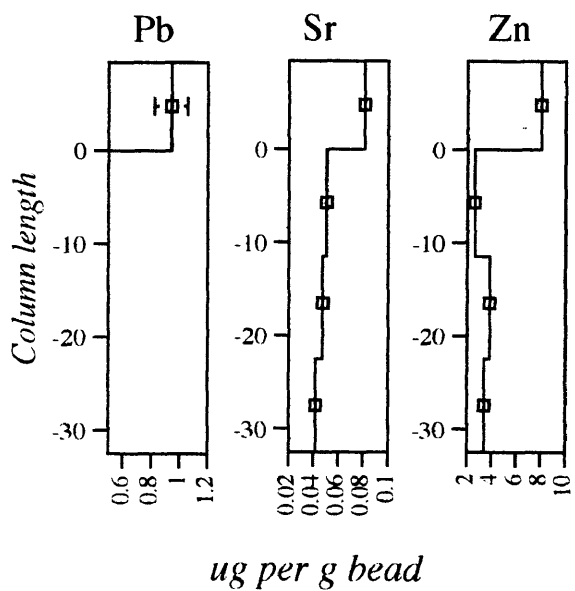
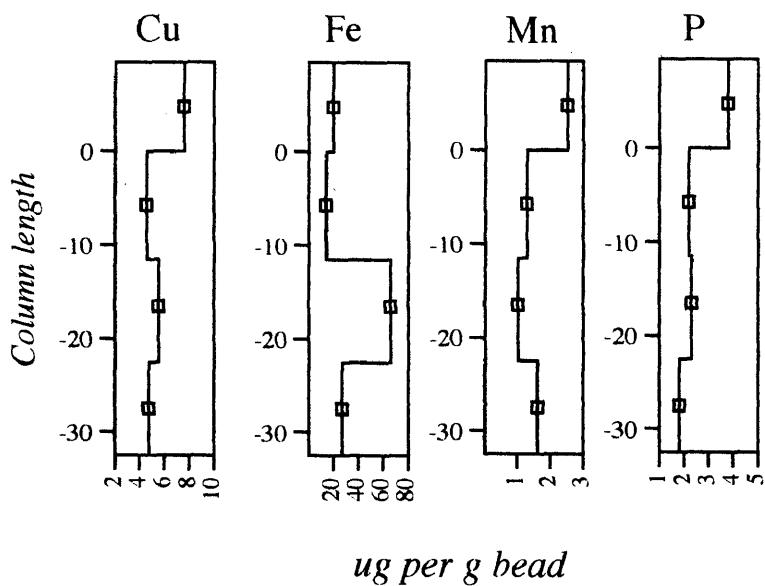
*BDL/ No significant trend:
Al, Ca, Mg, Na, Si, Ti*

Bead column IC-2



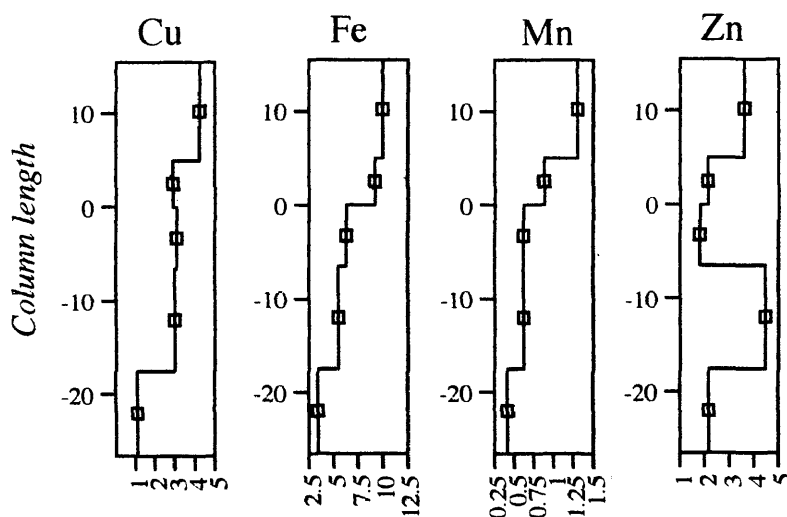
BDL/ No significant trend:
Al, Ca, Mo, Na, Si, Ti, Mo

Bead column IC-3



*BDL/ No significant trend:
Al, Ca, Cd, Mg, Mo, Na, Si, Ti*

Bead Column IIA-1

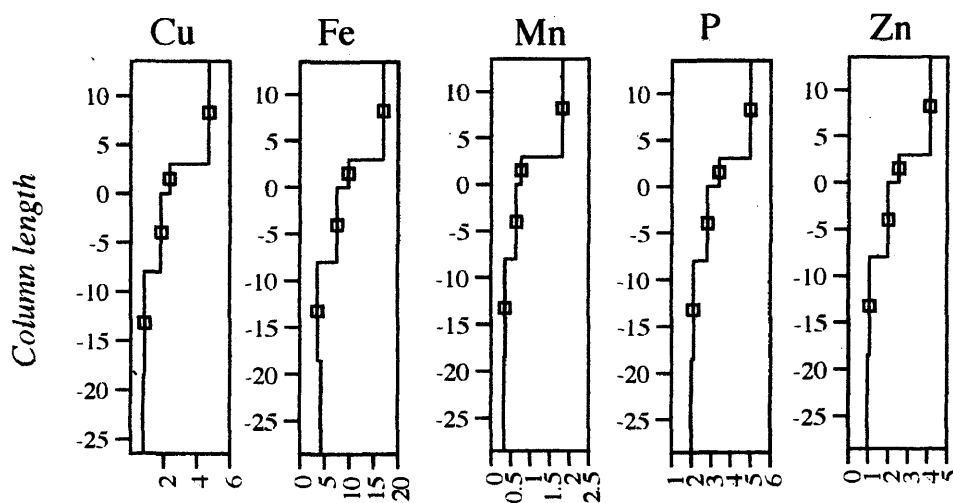


ug per g bead

BDL/ No significant trend:

Al, As, Ca, Cd, Mg, Mo, Na, P, Pb, Si, Sr, Ti

Bead Column IIA-2

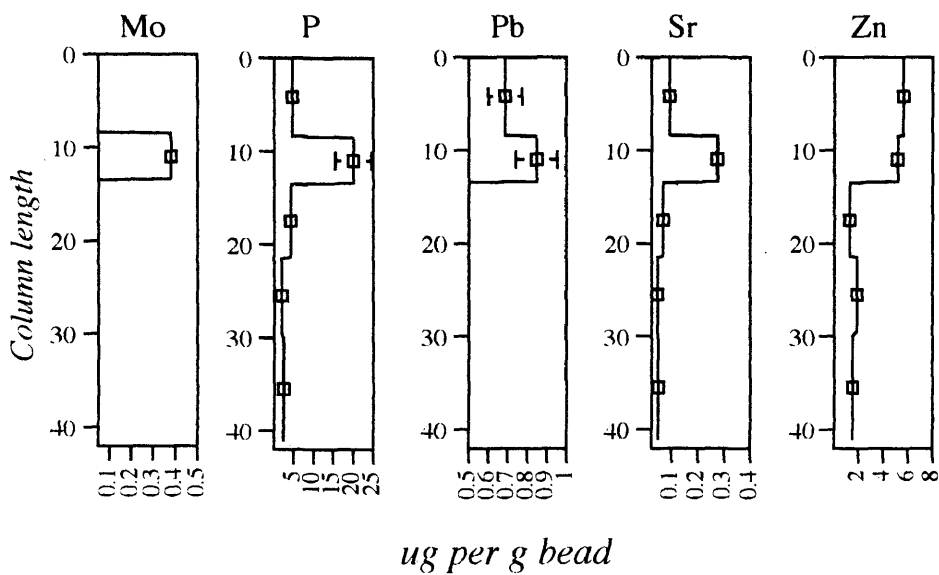
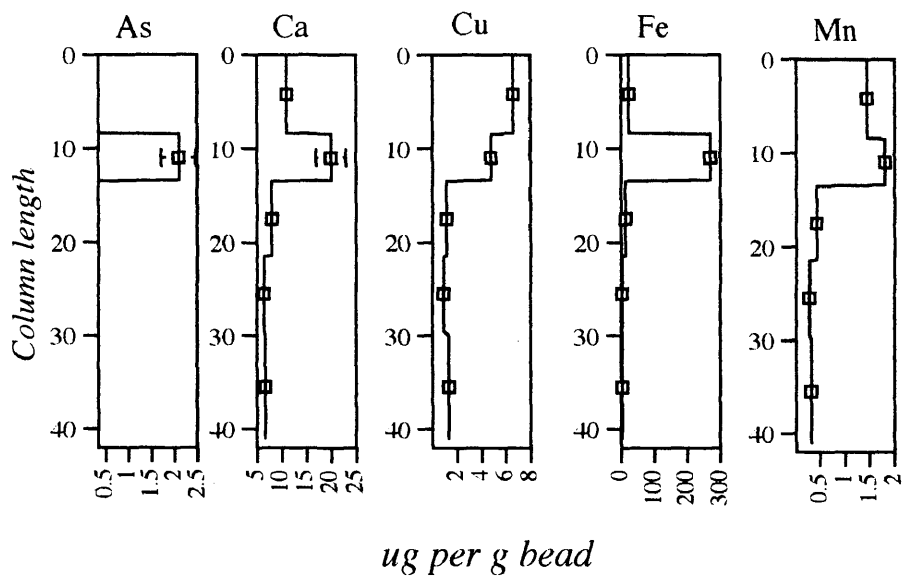


ug per g bead

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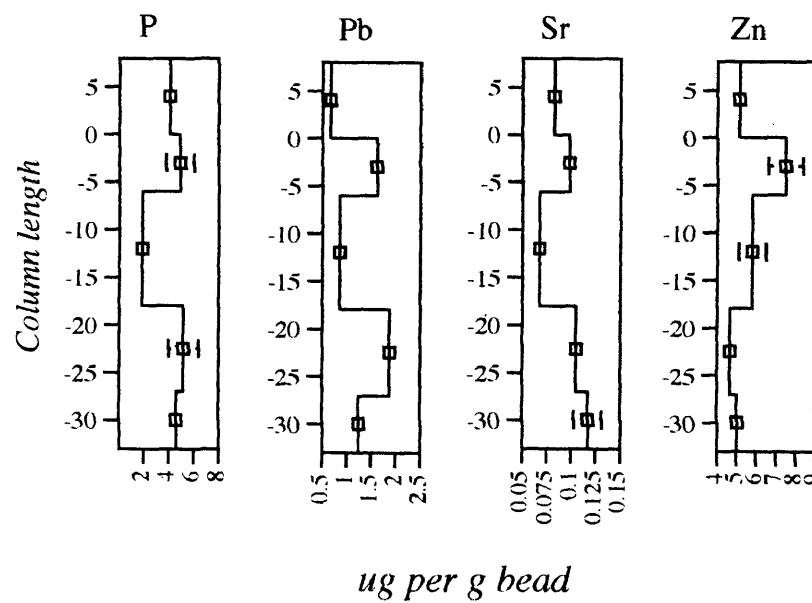
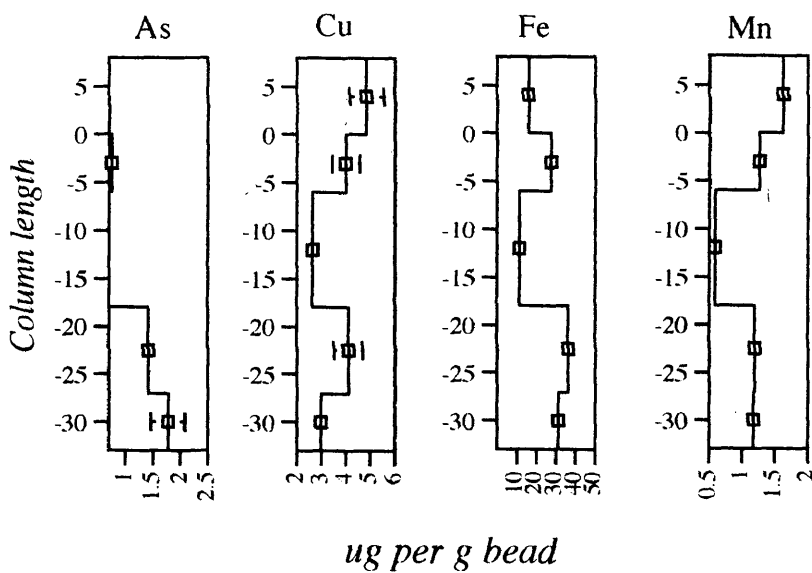
Al, As, Ca, Cd, Mg, Mo, Pb, Si, Sr, Ti

Bead column IIA-3



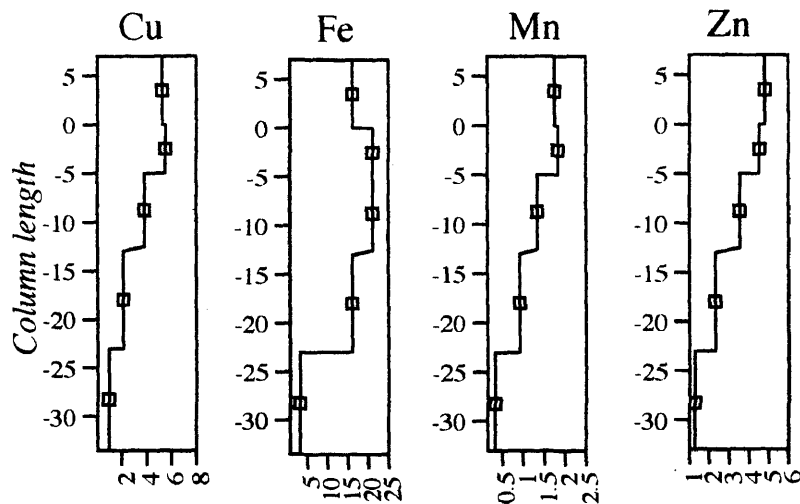
BDL/ No significant trend:
Al, Cd, Mg, Si, Ti

Bead column IIB-1



*BDL/ No significant trend:
Al, Ca, Cd, Mg, Na, Si, Ti*

Bead Column IIB-2

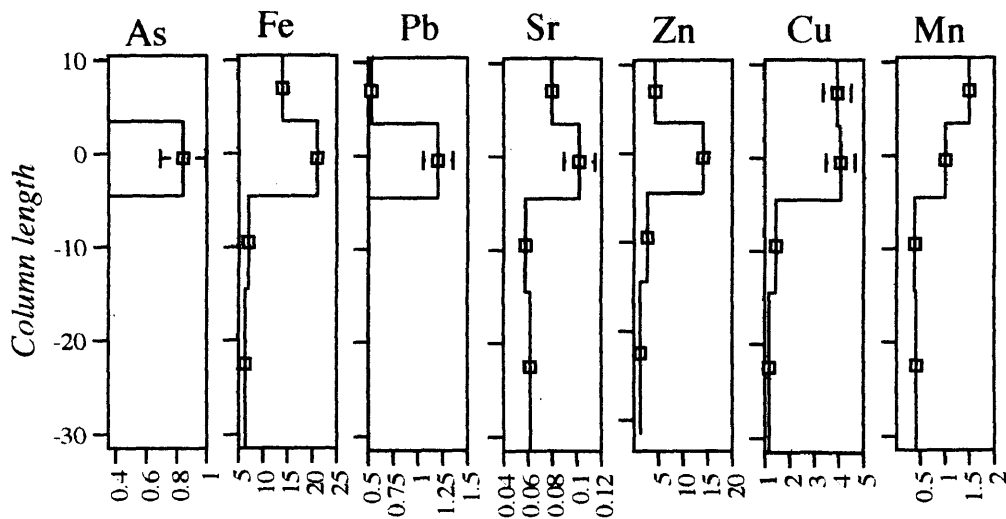


ug per g bead

BDL/ No significant trend:

Al, As, Ca, Cd, Mg, Mo, Na, P, Pb, Si, Sr, Ti

Bead column IIB-3

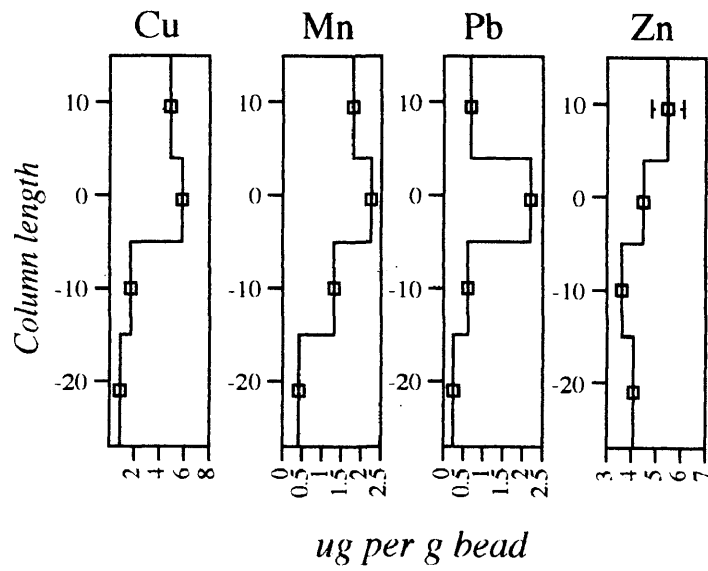
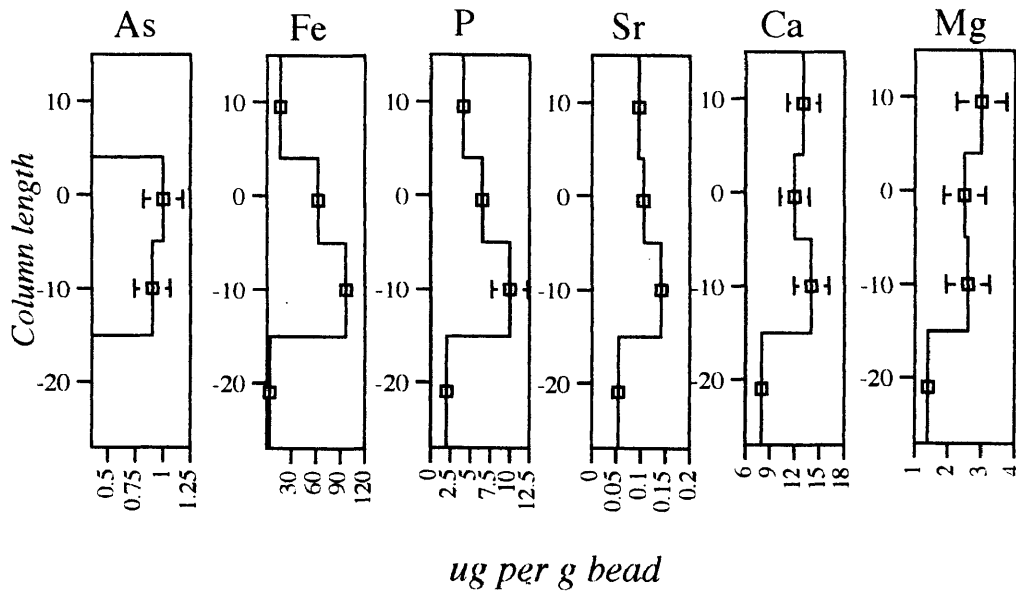


ug per g bead

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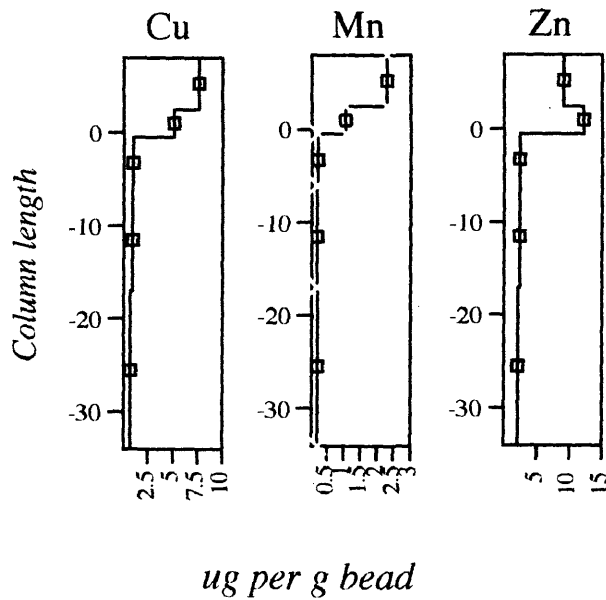
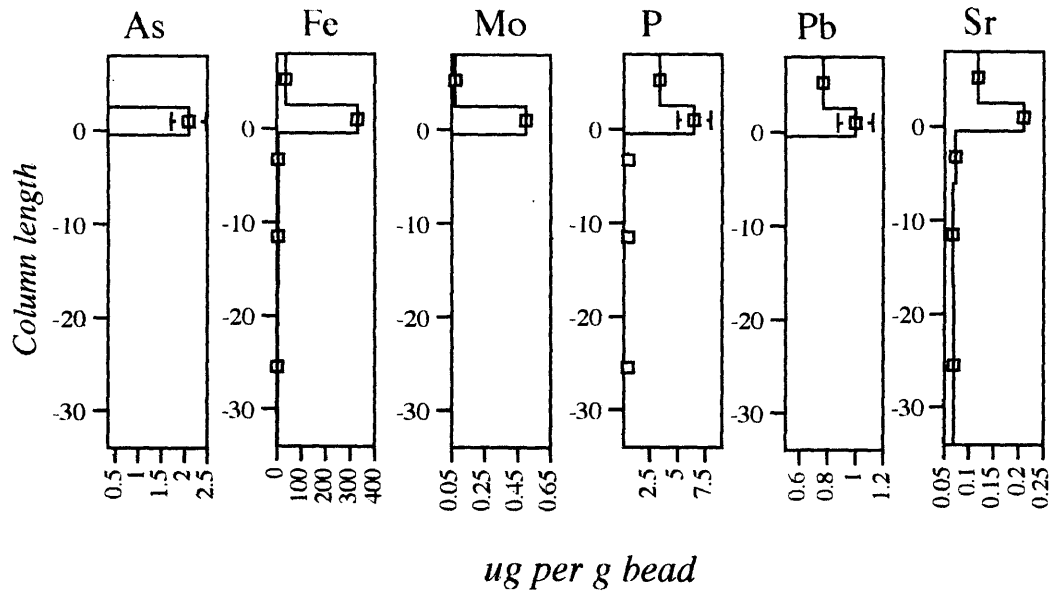
Al, Ca, Cd, Mg, Na, P, Si, Ti

Bead Column IIC-1



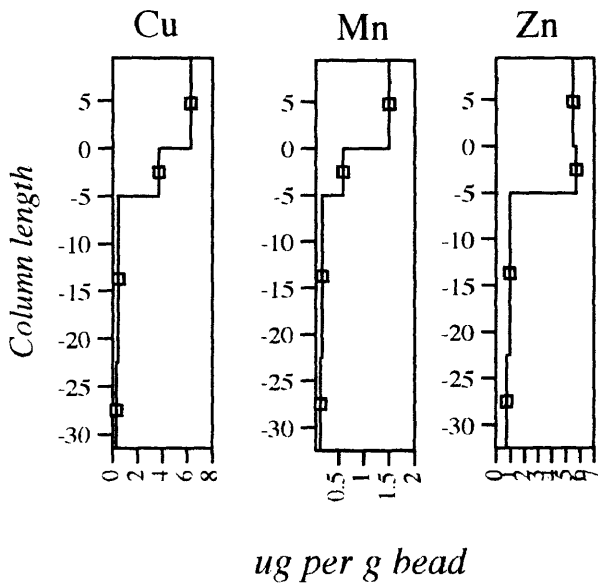
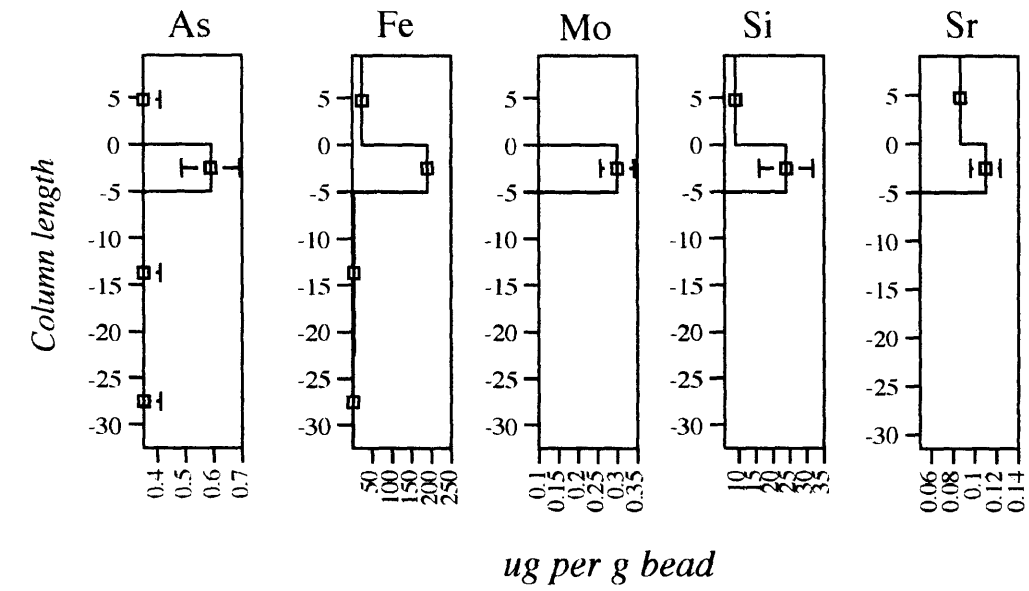
*BDL/ No significant trend:
Al, Cd, Mo, Na, Si, Ti*

Bead column IIIA-1



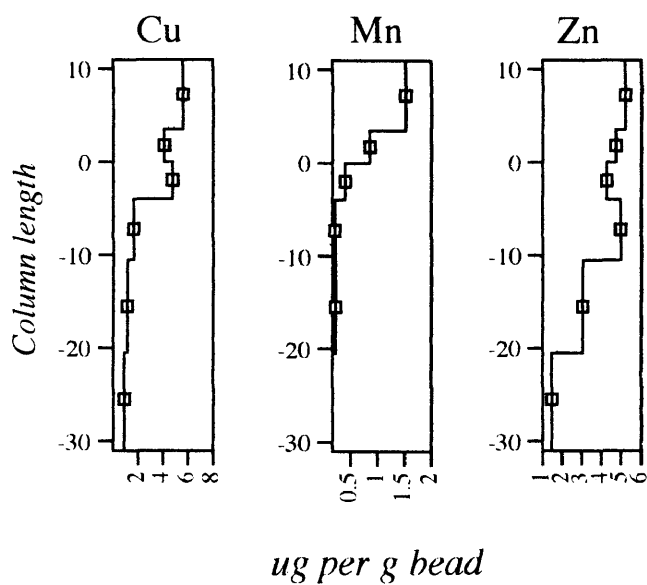
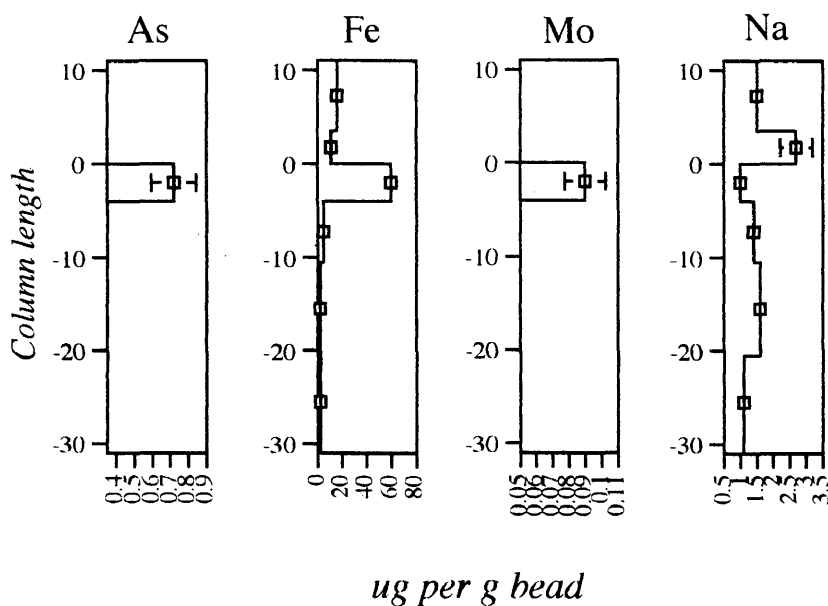
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Bead column IIIA-2



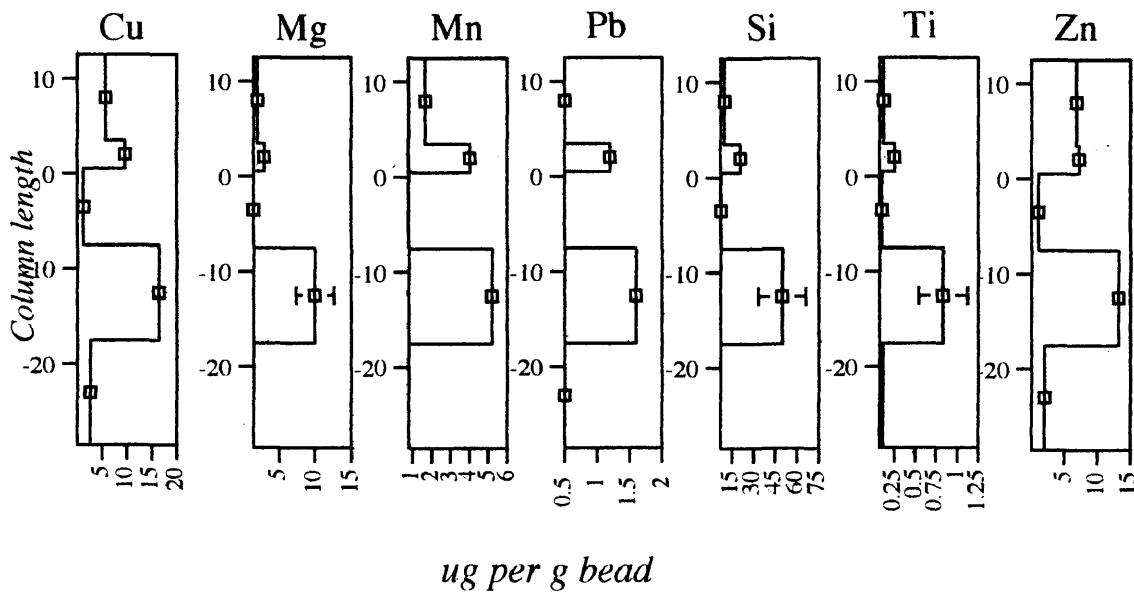
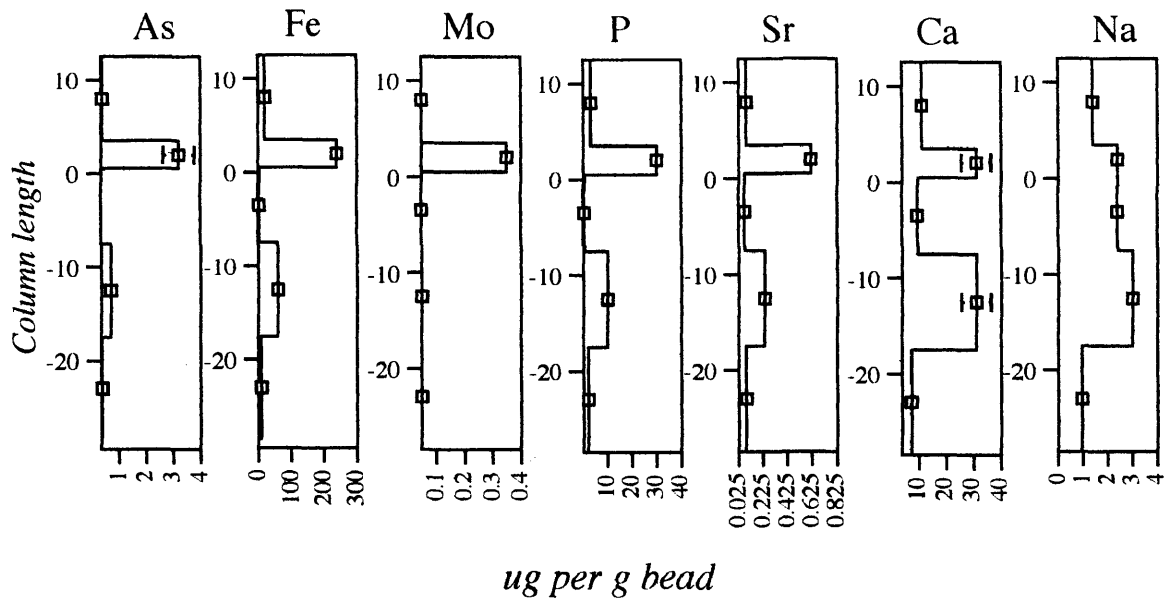
*BDL/ No significant trends:
Al, Ca, Cd, Mg, Na, P, Pb, Ti*

Bead column IIIA-3



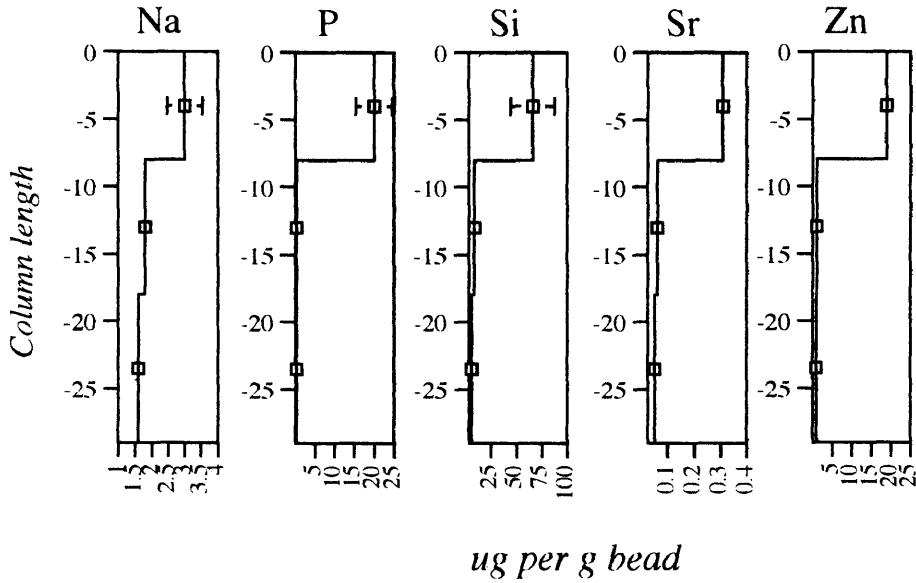
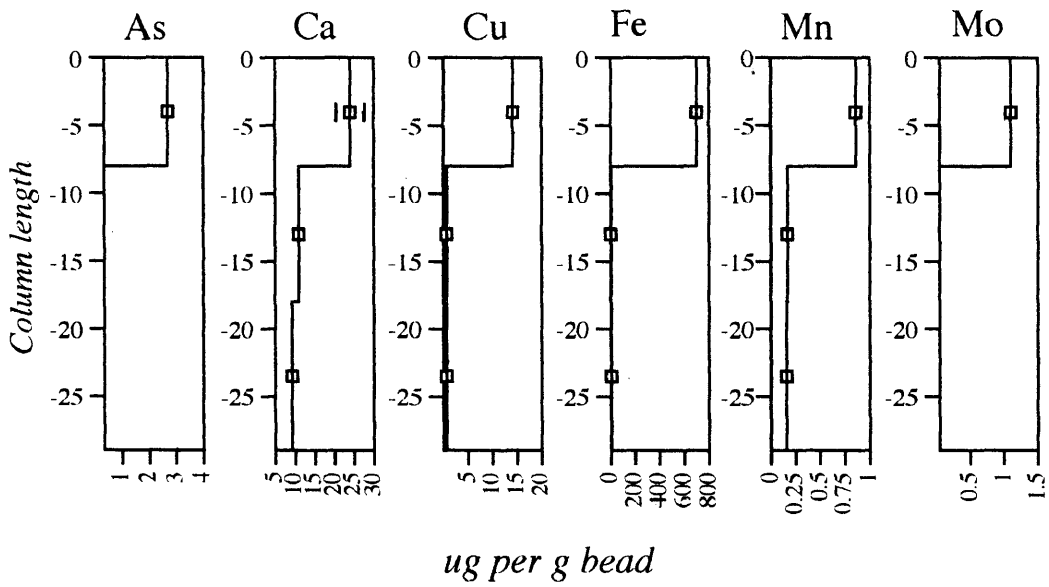
*BDL/ No significant trend:
Al, Ca, Cd, Mg, P, Pb, Si, Ti*

Bead Column IIB-1



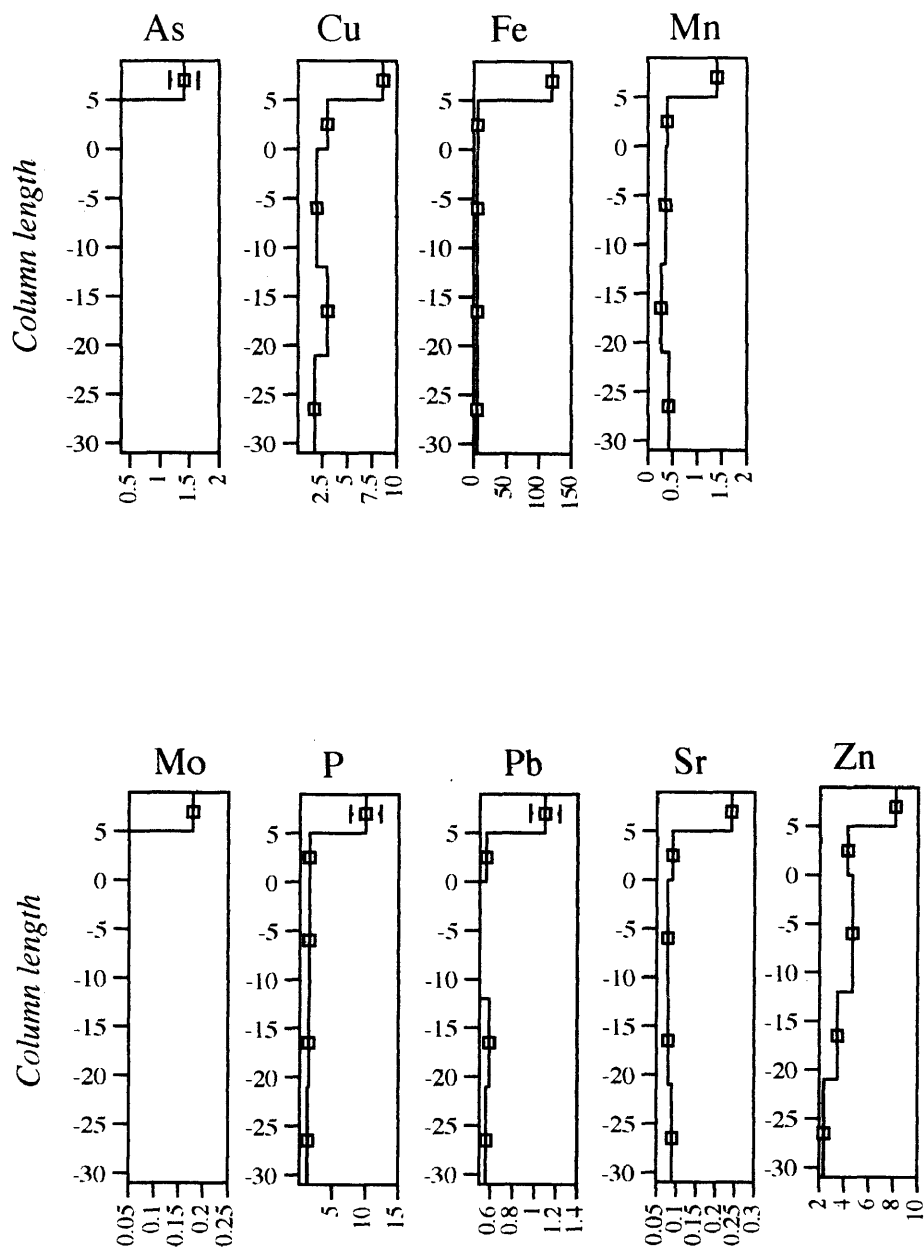
*BDL/ No significant trends:
Al, Cd*

Bead Column IIIB-2



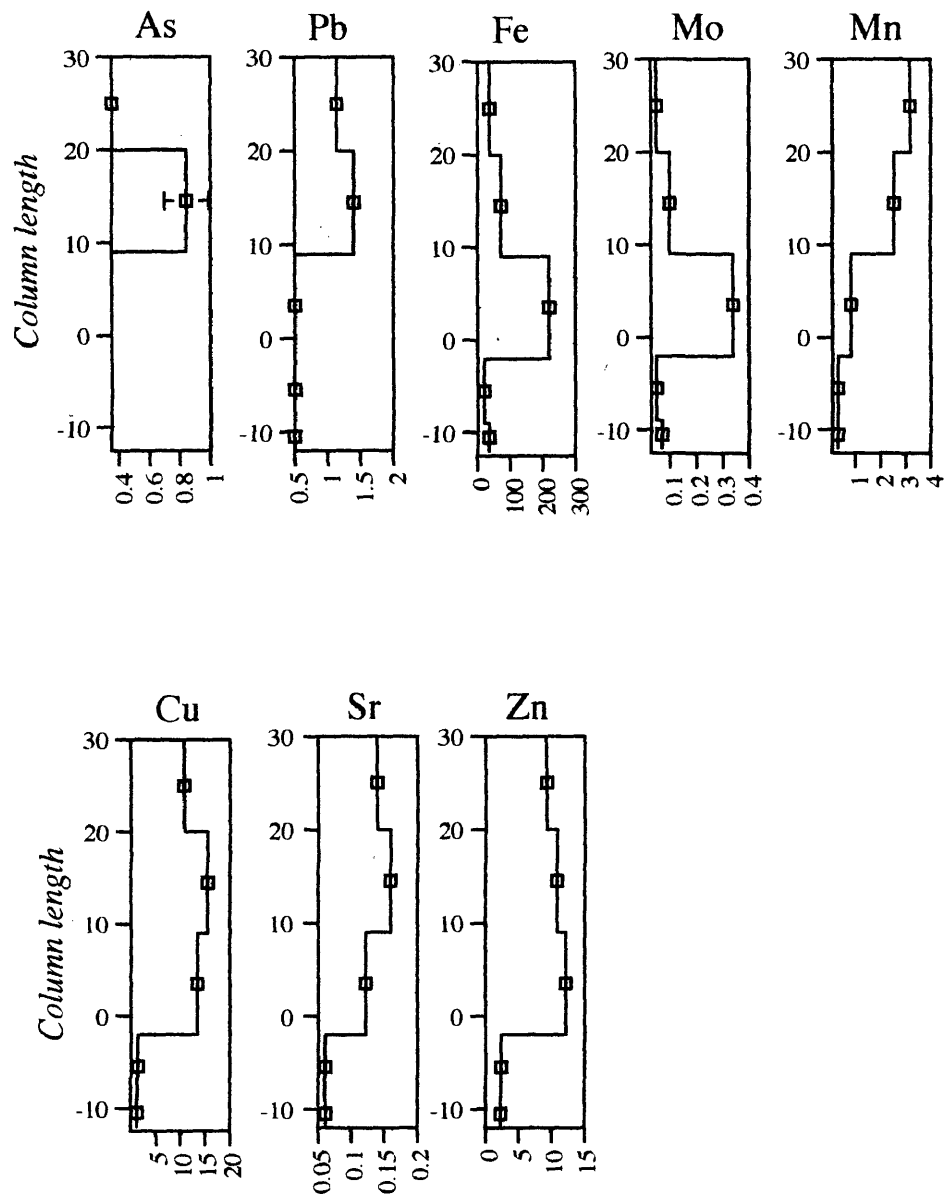
*BDL/ No significant trend:
Al, Cd, Mg, Pb, Ti*

Bead Column IIB-3



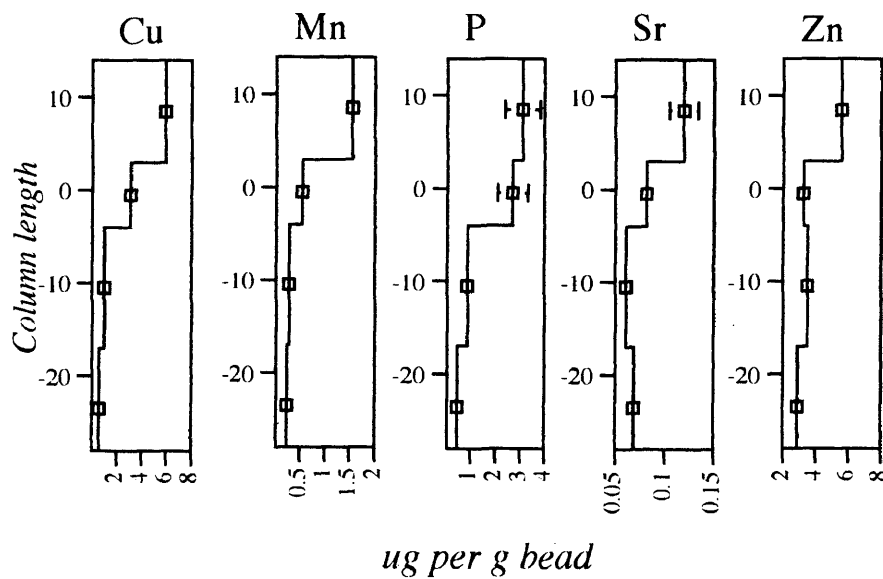
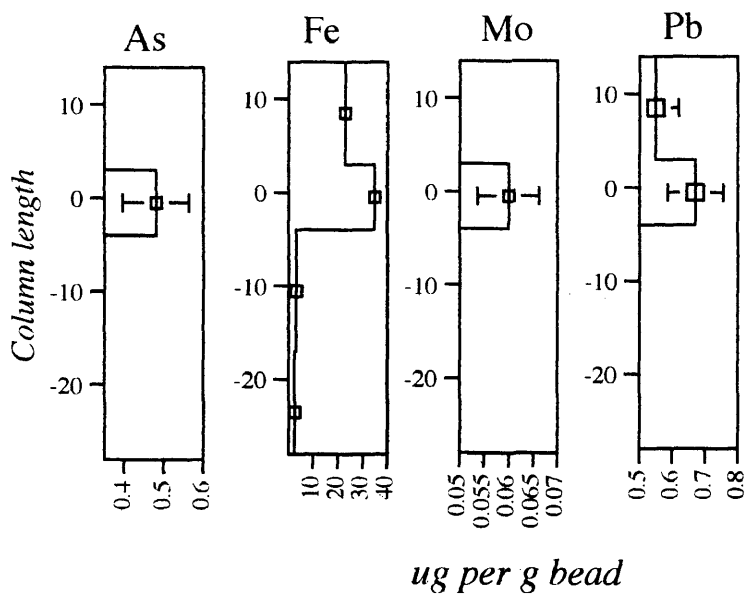
*BDL/ No significant trend:
Al, Ca, Cd, Mg, Na, Si, Ti*

Bead Column IIC-2



*BDL/ No significant trend:
Al, Ca, Cd, Mg, Na, Si, Ti*

Bead Column IIC-3



*BDL/ No significant trends:
Al, Ca, Mg, Na, Si, Ti*

Comparison of solid and dissolved phase metal ratios in the surface water:

(mean values)	IN SURFACE WATER (mg/L) (Dissolved phase)				ON SURFACE WATER BEADS (ug/g bead) (Solid phase)			
	SITE I	SITE II	SITE III	ALL SITES	SITE I	SITE II	SITE III	ALL SITES
Ca/Cu	352	392	337	358	1.2	2.2	2	1.8
Ca/Fe	293	288	313	296	0.46	0.65	0.7	0.58
Ca/Mg	4.6	4.6	4.6	4.6	2.5	4.1	5.81	4.3
Ca/Mn	61	55	54	58	2.7	6.4	8.13	6
Ca/Na	4.8	2	2	3.6	3	5.5	6.9	5.3
Ca/Zn	74	82	84	78	1.02	2.2	2.1	1.8
Fe/Cu	1.92	1.5	1.6	1.7	2.4	3.2	3.4	2.94
Fe/Mg	0.02	0.02	0.02	0.02	5.2	5.7	10.2	7.06
Fe/Mn	0.36	0.21	0.29	0.31	5.8	9.9	13.3	9.44
Fe/Na	0.01	0.01	0.01	0.01	6.2	8.2	8.7	7.4
Fe/Zn	0.43	0.29	0.41	0.39	2.2	3.5	3.4	2.99
Cu/Mg	0.01	0.01	0.01	0.01	2.1	1.8	2.9	2.3
Mg/Mn	13	12	12	13	1	1.7	1.5	1.4
Zn/Mn	0.83	0.67	0.63	0.71	2.5	2.9	3.8	2.99
Na/Cu	164	196	155	170	0.5	0.44	0.31	0.41
Na/Zn	37	41	39	39	0.4	0.44	0.31	0.39
Cu/Zn	0.21	0.20	0.22	0.21	0.9	1.1	1.02	1.01
Zn/Mg	0.06	0.06	0.06	0.06	2.4	1.8	2.8	2.4
Cu/Mn	0.17	0.14	0.17	0.16	2.3	3.2	3.9	3.1
Na/Mg	2.11	2.30	2.10	2.2	1.1	0.8	0.8	0.9
Mn/Na	0.19	0.04	0.04	0.12	1.1	1.0	0.96	1.01

This table was used to find the sequence of preferential precipitation in the surface water, which is found to be Fe>Cu>Zn>Mn>Na>Mg>Ca.