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# *Psyllium* and *Laminaria* Partnership—An Overview of Possible Food Gel Applications

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Received: 11 September 2019; Accepted: 13 October 2019; Published: 16 October 2019



**Featured Application:** *Laminaria-psyllium* gels with distinct texture and rheological features, designed for a wide range of food applications.

**Abstract:** Seaweeds are a novel source of important nutritional compounds with interesting biological activities that could be processed into added-value products. In this study, two previously developed products obtained by *Laminaria ochroleuca* processing (liquid extract and a purée-like mixture) were processed with *Psyllium* gel to develop functional hydrogels. The optimization of the formulation and the characterization of the *Laminaria-Psyllium* gels in terms of their mechanical features have allowed the proposal of potential food applications. A beneficial interaction was found between *Laminaria* and *Psyllium* in terms of the reinforcement of texture and rheological properties. The obtained outcomes could provide new healthy gelling formulations with attractive properties to alleviate the growing market demand of eco-novel food matrices.

**Keywords:** kombu; edible brown seaweed; gels; *Psyllium*; *Laminaria ochroleuca*; autohydrolysis; mechanical properties

## 1. Introduction

*Laminaria sp.* are industrially used for alginate extraction (17.1–32% *w/w*, dry basis) [1], a hydrocolloid with unique gelling abilities at low temperature and good heat stability, widely used as a thickener, stabilizer, and restructuring agent in the food, cosmetic, pharmaceutical, biomedical, and textile industries [2,3].

Mainly valued for alginate extraction, this natural resource presents other interesting compounds for human nutrition such as fatty acids, proteins, minerals [4], vitamins (A, C, D, B group, E, K, PP) [1], pigments such as carotenoids and polyphenols with proved antioxidant, hypoglycemic, antitumoral, and antimicrobial activities [5,6]. Due to their high content in biologically active compounds, seaweeds, and especially *Phaeophyceae* (brown algae), have great potential to act as a functional food and as a food ingredient [7,8].

*Psyllium (Plantago ovata)* is an annual plant found mainly in India and Pakistan, but also on Madeira Island (Portugal) due to its subtropical climate. Used in traditional medicine for centuries, *Psyllium* husk has gained more attention by the scientific community since its health allegation was approved by the Food and Drug Administration in 2012 [9] regarding its benefits in reducing the risk

of coronary heart disease. These features come from its high soluble fiber composition, which has the capacity of absorbing up to 15 g of water per g of *Psyllium* [10].

*Psyllium* husk is composed of 85% arabinoxylan, a neutral highly branched polysaccharide with about 35% of non-reducing terminal residues and a 15% non-polysaccharide fraction [11]. It has applications as an edible coating [12] and drug delivery [13].

Innovation in food and feed production is a reality and a tendency for the upcoming years, whether this is the use of poorly exploited raw materials, the reformulation of foods based on green and sustainable technologies, or “back to tradition”. Many of the food and feed industries’ products are gels (e.g., yoghurt, puddings, confectionery products, pasta, pet food, among others) or present gelling agents in their formulation (cream cheese, sauces, . . . ), making gel-systems a growing market.

Among the edible biopolymers, the use of plant seed mucilages namely chia and flax seed [14,15] are one of today’s trends. Aside from the sustainability issue, these ingredients also have health benefits such as the regulation of colonic microbiota, reduction of hyperlipidemia, anti-inflammatory effect, control of glycemic response, and control of satiation [16]. Due to their technological properties, these biopolymers are often used in the food industry as texturizing agents.

To our knowledge, the gel forming ability of the combination of *L. ochroleuca*-*Psyllium* has not been previously investigated. Therefore, the interaction between *Psyllium* husk gel and the edible brown seaweed *Laminaria ochroleuca*, either in its liquid fraction or its purée-like mixtures, with *Psyllium* will be studied, taking advantage of both these poorly exploited natural resources. The aim of the present work is to lay the foundation for a systematic textural and rheological description of *Laminaria*-*Psyllium* gels, focusing on their mechanical features intended for future food applications.

## 2. Materials and Methods

### 2.1. Raw Materials

Dehydrated *Laminaria ochroleuca* (Algas Atlánticas Algamar, S.L., Pontevedra, Spain) was milled and sieved to two different particle size powders (0.25–2 mm and <0.25 mm diameters). *Psyllium* husk of Indian origin (Solgar, lot 107028-01, Leonia, NJ, USA) was purchased at the local market, milled (Pulverisette 14 Premium, Fritsch, Idar-Oberstein, Germany) and sieved to 160–315 µm.

For comparison purposes, commercial references were used, namely baby food (purée), guacamole (purée), fruit jam (gel), jelly gum (rubbery solid), pâté (liver paste), and pet food (moistened feed, pâté like).

### 2.2. Formulations Development

#### 2.2.1. *Laminaria ochroleuca* Sample Production

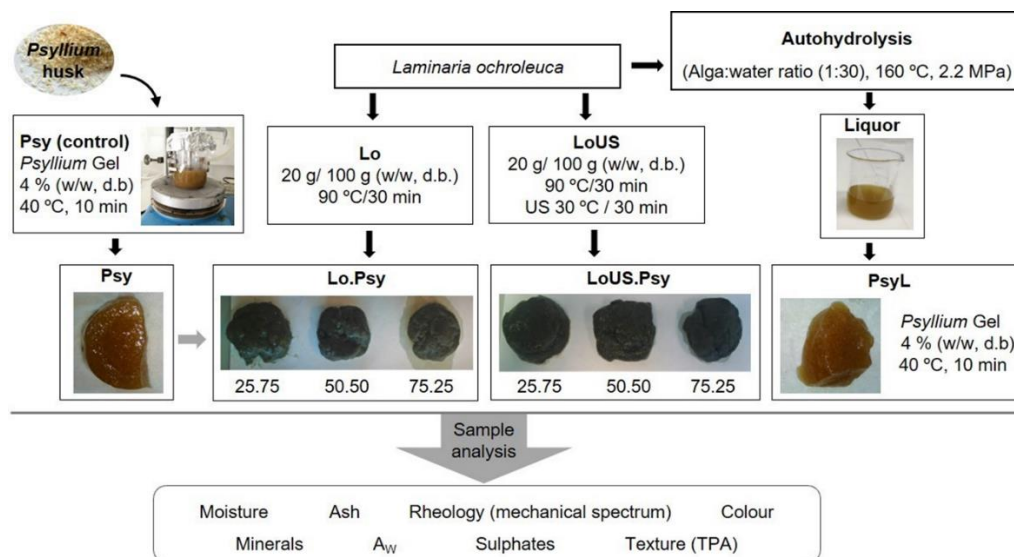
Focusing on the full valorization of *L. ochroleuca*, the 2 mm fraction of the alga was subjected to autohydrolysis (AH) in a pressurized reactor (Parr Instruments series 4848, Moline, IL, USA) [14] and the liquid extract (liquor) obtained was stored at −18 °C until further use. When using the <0.25 mm alga fraction, purée-like mixtures were prepared following the procedure already described by the authors [17], with (LoUS) and without ultrasonic treatment (Lo). The alga purées were matured for 24 h at 4 °C before preparing the *Laminaria*-*Psyllium* gels.

#### 2.2.2. Preparation of the *Laminaria*-*Psyllium* Gels

The control sample (Psy) was prepared by adding 4% (*w/w*, d.b) *Psyllium* husk (160–315 µm range of particle size) to distilled water at 40 °C under mechanical stirring (10 min, 200 rpm) [18]. Another *Psyllium* gel was prepared in a similar manner as the control by using the *Laminaria* autohydrolysis liquid extract (PsyL).

*Laminaria*-*Psyllium* gels were prepared by mixing (Eurostar Power-b, Ika-Werke, Staufen im Breisgau, Germany) for 5 min/200 rpm at room temperature, the control sample and the alga purée

mixture at the ratios 25:75, 50:50, and 75:25 in order to obtain visually different structures with potential different applications. Two batches of 160 g of each gel (*Laminaria-Psyllium*: 25:75, 50:50, 75:25) were prepared, poured into sealed plastic containers ( $\varnothing = 10$  cm) and stored at 4 °C for 24 h to mature. The schematic procedure of the sample preparation and analysis is presented in Figure 1.



**Figure 1.** General schematic procedure of the gel preparation and analysis (Lo—*Laminaria ochroleuca*; Psy—*Psyllium* gel prepared with water (control); US—ultrasound treatment; PsyL—*Psyllium* gel prepared with autohydrolysis liquor).

### 2.3. Physicochemical Measurements

Moisture and ash contents of the raw materials and gels were determined by gravimetric methods by placing the samples in an oven ( $105 \pm 2$  °C) until constant weight, or in the furnace (575 °C, 6 h) respectively.

Sulfate content of the developed gel samples and *Psyllium* husk was obtained by the ionic chromatography method (mobile phase: 3.2 mM sodium carbonate/1 mM sodium bicarbonate at 0.70 mL/min) as previously reported [19].

Water activity ( $a_w$ ) measurements of all gel samples were performed using a LabMaster-aw (Novasina, Pfäffikon, Switzerland) in triplicate.

Mineral content (Ca, K, Na, Fe) of *Psyllium* husk was analyzed by inductively coupled plasma optical emission spectrometry (Optima 4300 DV, Perkin Elmer, Waltham, MA, USA) after microwave-assisted (Saville, Eden Prairie, MN, USA) acid digestion (80 °C, 6 h).

### 2.4. Color Measurements

The color evaluation was performed using a CR-400 colorimeter (Minolta, Tokyo, Japan) with standard illuminant C. Tristimulus color coordinates (CIELAB system) were used to measure the degree of lightness ( $L^*$ ), which ranged from 0 (black) to 100 (white), redness ( $a^*$ ) ranging from  $-60$  (green) to  $+60$  (red), and yellowness ( $b^*$ ) ranging from blue ( $-60$ ) to yellow ( $+60$ ). Total color difference ( $\Delta E^*$ ) between the samples and the control was calculated according to Equation (1).

$$\Delta E^* = (\Delta L^* + \Delta a^* + \Delta b^*)^{1/2} \quad (1)$$

For calibration purposes, a white standard was used ( $L^* = 97.21$ ;  $a^* = 0.14$ ;  $b^* = 1.99$ ). Measurements were conducted at  $20 \pm 1$  °C and replicated at least five times.

### 2.5. Dynamic Viscoelasticity

Dynamic oscillatory rheology measurements were used to monitor the viscoelastic characteristics of the gels. Small amplitude oscillatory shear measurements were conducted in triplicate in a controlled stress rheometer (RheoStress 600, Haake, Vreden, Germany) using serrated parallel plate geometry (diameter 35 mm) and a 1.5 mm gap. Surface geometry was covered with paraffin oil to prevent moisture loss. Samples were rested in the rheometer device for 5 min (20 °C) before rheological testing to temperature equilibration. Initially, stress sweep tests were run at 1 Hz, with the shear stress of the input signal varying from 0.1 to 100 Pa to find the linear viscoelastic region. The mechanical spectra of all samples were assessed through frequency sweep tests performed from 0.1 to 10 Hz (20 °C, 10 Pa) within the linear viscoelastic region previously defined. Experimental storage ( $G'$ ) and loss ( $G''$ ) moduli (Pa) data versus frequency ( $f$ , Hz) were fitted using well-known power models reported elsewhere [20], where  $\alpha'$ ,  $\alpha''$ ,  $b'$ , and  $b''$  are the corresponding fitting parameters (Equations (2) and (3)).

$$G' (f) = \alpha' f^{b'} \quad (2)$$

$$G'' (f) = \alpha'' f^{b''} \quad (3)$$

### 2.6. Texture Profile Analysis (TPA)

Texture profile analysis of all developed gels was performed in a TA-XT2 texturometer (Stable Microsystems, Godalming, UK) with a P0.25S plunger that penetrated 15 mm into the sample at 1 mm/s. From the force vs. time texturogram, three parameters were selected to characterize the materials: firmness, as the maximum rupture force (N); adhesiveness, represented by the negative area of the graph, that translates the recessive force of the probe (N·s); and cohesiveness corresponding to the ratio of the material's response to a second deformation. At least five measurements were made in each sample. Commercial food and feed products were also measured for comparative purposes.

### 2.7. Statistical Analysis

Statistical analysis of the experimental data was performed using RStudio (Version 1.2.1335—© 2009–2019 RStudio, Inc., Boston, MA, USA), using variance analysis (one-way ANOVA), and the Tukey test, *Post Hoc* comparison at a significance level of 95% ( $p < 0.05$ ). A Pearson correlation analysis was also conducted ( $p < 0.05$ ) to determine the relationships between the color, texture parameters, and the amount of alga of the samples. The curve fitting of the rheological data was performed in Excel (version 365, Microsoft). All results are presented as the mean  $\pm$  standard deviation (s.d.).

## 3. Results and Discussion

### 3.1. Physicochemical Characterization of Samples

*Psyllium* gel prepared with autohydrolysis (AH) liquor (PsyL) was visually more fluid than the *Psyllium* gel prepared with water (control). This is confirmed by texture and rheology measurements, so this issue will be discussed in Sections 3.3 and 3.4.

As seen in Figure 1, the *Laminaria-Psyllium* 75.25 gel samples presented syneresis (water in the filter paper, not measured) more evident in the sample not submitted to ultrasonic treatment. According to the previous study in which the purée-like mixtures were optimized [17], the authors did not find syneresis in the systems. This could be due to the alga high sodium content that bonds to *Psyllium*, and at the 25% ratio was not able to retain all the water in the system, causing the *Laminaria-Psyllium* 75.25 gels to contract and release part of the water previously enclosed. This phenomenon did not occur in samples with higher *Psyllium* gel ratios. A recent study by Figueroa and co-workers [21] reported the positive influence of *Psyllium* on the absence of syneresis of fruit jellies enriched with dietary fiber.

In Table 1, a chemical characterization of *Psyllium* husk and the developed gels is presented.

**Table 1.** Moisture, ash, and sulfate content of *Laminaria ochroleuca*, *Psyllium* husk, and the developed gels.

Samples	Moisture	Ash	Sulphates
	(%)	(%, d.b.)	(%, d.b.)
<i>Laminaria ochroleuca</i> [17]	9.20 ± 0.07	35.01 ± 0.31	2.21 ± 0.10
<i>Psyllium</i> husk (160–315 µm)	9.03 ± 0.31 h	2.98 ± 0.05 g	0.09 ± 0.00 e
Psy (control)	95.83 ± 0.08 a	2.91 ± 0.18 g	0.20 ± 0.00 e
PsyL	93.58 ± 0.06 b	18.01 ± 1.83 f	0.98 ± 0.00 d
Lo.Psy_25.75	91.73 ± 0.03 c	24.75 ± 0.51 d	1.75 ± 0.01 c
Lo.Psy_50.50	87.50 ± 0.06 e	33.08 ± 0.24 b	2.20 ± 0.07 a,b
Lo.Psy_75.25	84.48 ± 0.22 g	36.02 ± 0.23 a	1.98 ± 0.13 b,c
LoUS.Psy_25.75	91.66 ± 0.12 c	21.91 ± 0.71 e	1.78 ± 0.02 c
LoUS.Psy_50.50	87.78 ± 0.13 d	30.34 ± 0.41 c	2.14 ± 0.11 a,b
LoUS.Psy_75.25	85.18 ± 0.12 f	34.71 ± 0.37 a,b	2.25 ± 0.01 a

Psy (control), *Psyllium* gels prepared in water; PsyL, *Psyllium* gels prepared in autohydrolysis liquor; *Laminaria-Psyllium* gels: Lo.Psy\_25.75, Lo.Psy\_50.50, and LoPsy\_75.25; *Laminaria* with ultrasonic treatment-*Psyllium* gels: LoUS.Psy\_25.75, LoUS.Psy\_50.50, and LoUSPsy\_75.25. Data are presented as the mean ± sd. Different letters in a column show significantly different data values at the  $p < 0.05$  level.

The moisture content of the *Laminaria-Psyllium* gels depends greatly on the proportion of both components, and decreased with the increase in *Laminaria* content. Ash content revealed the reverse trend, with the *Laminaria* proportion being crucial for the final gel mineral content. This is due to the high ash content (35%, d.b.) of this alga, as we previously determined [17]. It is also noteworthy that the ultrasonic treatment applied to the purée-like mixtures had a significant ( $p < 0.05$ ) negative influence on the total ash content of the gel, especially in the 25.75 and 50.50 samples.

Sulfate presence in the *Laminaria-Psyllium* gels is a clear indication of the presence of fucoidan, a sulfated polysaccharide with reported activity against stomach-gastric adenocarcinoma cells and lung carcinoma cells [22].

The most important chemical feature of *Psyllium* husk is its high soluble fiber content (80%, d.b., as previously determined by the authors in the same sample [10]). However, aside from its fiber content, *Psyllium* husk has a mineral composition of 10.3 g K/kg, 0.88 g Na/kg, 1.36 g Ca/kg, and 94.7 mg Fe/kg, (present study), the last two minerals having a higher content than most cooked pulses [23].

Another important feature is *Psyllium*'s ability to retain sodium at physiological important conditions (pH 1.2—stomach; pH 6.8—intestine), being potentially active in reducing the bioavailable fraction of ingested sodium in the body [24].

Considering both the *L. ochroleuca* [17] and *Psyllium* husk's mineral content, one can conclude that the partnership between this alga and *Psyllium* husk could be advantageous for the development of new food products.

Water activity was very high and ranged between 0.999–1.000 for all samples (data not shown) and these are typical values for gels, although some reduction was expected in gels with a high proportion of *Psyllium*.

### 3.2. Color Evaluation of Samples

From the color evaluation results presented in Table 2, the *Psyllium* gels (control and PsyL) only differed in terms of their chromatic parameters, which could be of importance depending on the desired application.

**Table 2.** Color parameters ( $L^*$ ,  $a^*$ ,  $b^*$ ) and  $\Delta E^*$  of *Psyllium* husk, *Laminaria ochroleuca* and its AH liquor, and the gels developed.

Samples	$L^*$	$a^*$	$b^*$	$\Delta E^*$
<i>Psyllium</i> husk (160–315 $\mu\text{m}$ )	$59.39 \pm 0.91$ a	$6.06 \pm 0.18$ a	$24.22 \pm 0.37$ a	-
<i>Laminaria ochroleuca</i> [17]	$56.36 \pm 0.57$	$-2.48 \pm 0.04$	$14.49 \pm 0.18$	-
AH liquor [17]	$31.42 \pm 1.74$	$1.45 \pm 0.26$	$2.88 \pm 1.60$	-
Psy (control)	$33.12 \pm 1.78$ b,c	$1.32 \pm 0.14$ c	$9.85 \pm 1.36$ c	-
PsyL	$35.70 \pm 2.86$ b	$3.78 \pm 1.50$ b	$17.67 \pm 5.54$ b	8.6
Lo.Psy_25.75	$30.08 \pm 0.37$ c,d	$-0.16 \pm 0.22$ d	$12.82 \pm 0.57$ c	4.5
Lo.Psy_50.50	$27.50 \pm 0.77$ d,e	$-0.07 \pm 0.16$ d	$12.37 \pm 1.07$ c	6.3
Lo.Psy_75.25	$25.14 \pm 1.70$ e	$-0.45 \pm 0.09$ d	$10.90 \pm 1.24$ c	8.2
LoUS.Psy_25.75	$29.98 \pm 0.82$ d	$-0.06 \pm 0.23$ d	$14.10 \pm 0.46$ b,c	5.5
LoUS.Psy_50.50	$26.96 \pm 1.17$ d,e	$-0.40 \pm 0.09$ d	$12.38 \pm 1.15$ c	6.9
LoUS.Psy_75.25	$25.87 \pm 0.55$ e	$-0.64 \pm 0.06$ d	$11.60 \pm 1.14$ c	7.7

Psy (control), *Psyllium* gels prepared in water; PsyL, *Psyllium* gels prepared in autohydrolysis liquor; *Laminaria-Psylim* gels: Lo.Psy\_25.75, Lo.Psy\_50.50, and LoPsy\_75.25; *Laminaria* with ultrasonic treatment-*Psylim* gels: LoUS.Psy\_25.75, LoUS.Psy\_50.50, and LoUSPsy\_75.25. Data are presented as the mean  $\pm$  sd. Different letters in the same column correspond to significant differences ( $p < 0.05$ ).

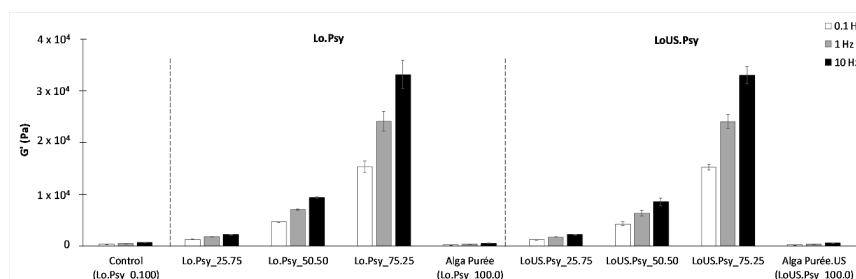
As expected, the lightness ( $L^*$ ) of the gel samples decreased with the incorporation of *Laminaria*, although significant differences were only registered between the *Laminaria-Psylim* 25.75 and *Laminaria-Psylim* 75.25 samples. It is noteworthy that the color parameters of the gel samples with the highest *Laminaria* content were like those obtained in the purée-like mixtures [14]. Moreover, the color parameters  $L^*$  (Lo.Psy:  $r = -0.928$ ,  $p = 1.3 \times 10^{-9}$ ; LoUS.Psy:  $r = -0.920$ ,  $p = 9.3 \times 10^{-9}$ ) and  $a^*$  (Lo.Psy:  $r = -0.849$ ;  $p = 1.2 \times 10^{-6}$ ; LoUS.Psy:  $r = -0.913$ ,  $p = 2.0 \times 10^{-8}$ ) were strongly negatively correlated with *Laminaria* content in the gel formulations.

The  $\Delta E^*$  between the *Laminaria-Psylim* gels and the control ranged from 4.5 to 8.2, increasing with the increase in the *Laminaria* content in the system. These values indicate that the consumer would distinguish between the two samples compared [25].

It should be pointed out that the gels proposed here are not finished products. In this sense, the color of the final food gel product could be optimized by considering these findings and according to the desired final color.

### 3.3. Effect of *Laminaria-Psylim* Ratio on the Dynamic Viscoelasticity

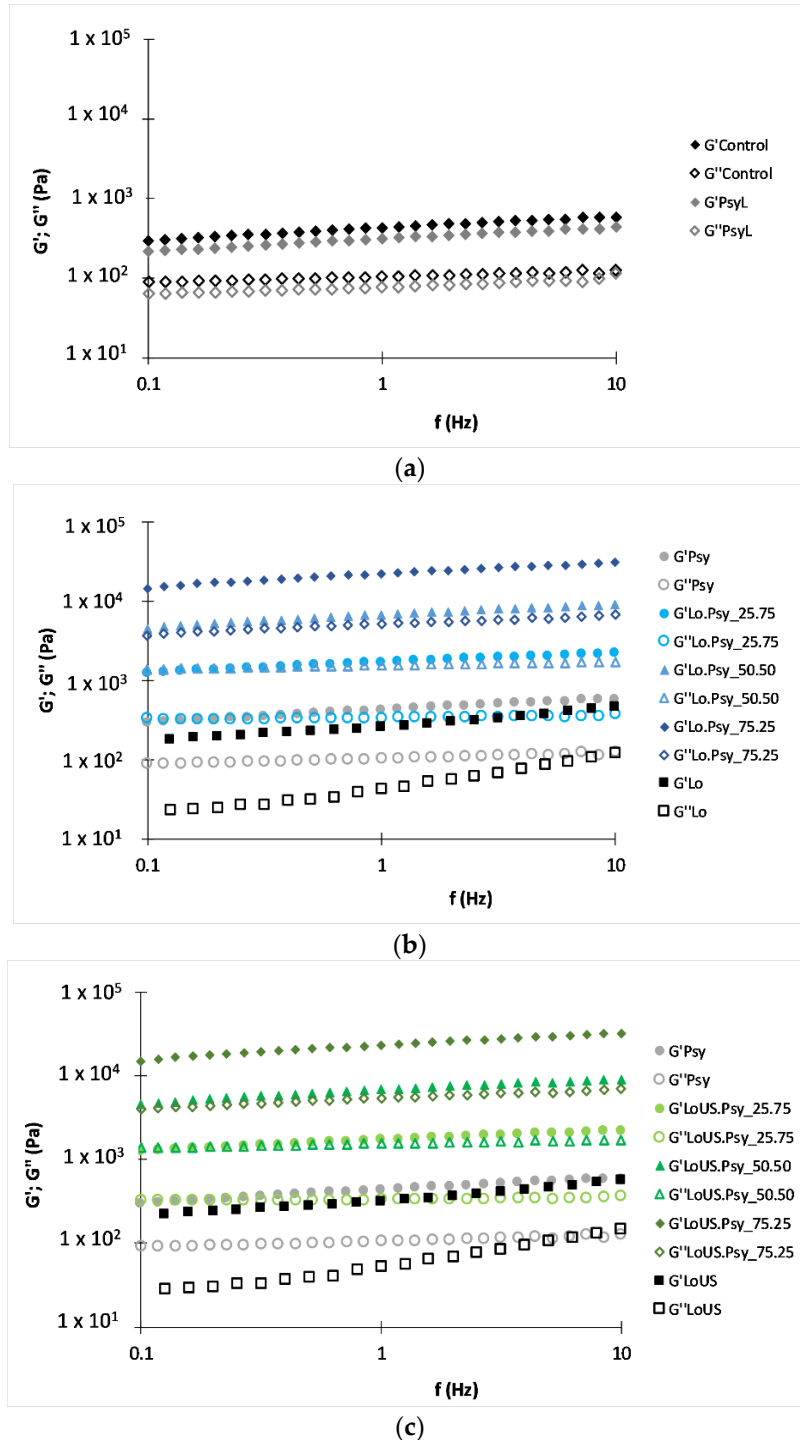
Figure 2 shows the elastic behavior of *Laminaria-Psylim* gel systems. As the alga fraction increased, a structuring effect was observed in the gels. This behavior was markedly noticed in the highest alga concentration (Lo.Psy\_75.25 and LoUS.Psy\_75.25). This interaction is probably due to physical entanglements between the polymers present in both the alga and *Psylim*, causing the reinforcement of the gel. However, in the absence of the *Psylim* gel, the alga purées showed a huge decrease in  $G'$  to values similar to the control (Psy), but more frequency dependent ( $b' = 0.209$ ) [17].



**Figure 2.** Elastic modulus at 0.1 Hz, 1 Hz, and 10 Hz of *Laminaria-Psylim* gels with (LoUS.Psy) and without ultrasonic treatment (Lo.Psy), control, and *Laminaria* purées (Alga Purée; Alga Purée.US) [17].

This synergistic interaction was also found in other polymeric systems, namely between the locust bean gum (LBG) and xanthan gum, where the latter did not form gel, rather a shear-thinning solution, but combined with LBG to form a strong gel structure [26].

The mechanical spectra of the developed gels are depicted in Figure 3.



**Figure 3.** Mechanical spectra of the *Psyllium* gels prepared in water (control) and autohydrolysis liquor (PsyL). (a) *Laminaria*-*Psyllium* gels (Lo.Psy\_25.75, Lo.Psy\_50.50, Lo.Psy\_75.25); (b) *Laminaria* with ultrasonic treatment-*Psyllium* gels (LoUS.Psy\_25.75, LoUS.Psy\_50.50, LoUS.Psy\_75.25); (c)  $G'$ , closed symbol;  $G''$ , open symbol.

To quantify the impact of the different combinations of *Laminaria* and *Psyllium* on the viscoelastic moduli, the variation of  $G'$  and  $G''$  with gel composition was obtained from the respective mechanical spectra (Table 3).

**Table 3.** Power law parameters ( $\alpha'$ ,  $\alpha''$ ,  $b'$ , and  $b''$ ) of the gel samples with *Laminaria ochroleuca* and *Psyllium* husk, control, and PsyL.

Samples	$G'$		$G''$	
	$\alpha'$	$b'$	$\alpha''$	$b''$
<b>Psy (Control)</b>	436.8 ± 15.4	0.153 ± 0.002	104.8 ± 7.0	0.071 ± 0.011
<b>PsyL</b>	323.0 ± 18.2	0.149 ± 0.005	79.6 ± 2.4	0.090 ± 0.005
<b>Lo.Psy_25.75</b>	1689.3 ± 20.8	0.124 ± 0.000	344.2 ± 2.7	0.022 ± 0.000
<b>Lo.Psy_50.50</b>	6819.9 ± 103.8	0.148 ± 0.002	1584.5 ± 28.3	0.049 ± 0.002
<b>Lo.Psy_75.25</b>	0.5 ± 1816.6	0.158 ± 0.002	5485.0 ± 434.5	0.119 ± 0.001
<b>LoUS.Psy_25.75</b>	1681.6 ± 3.5	0.121 ± 0.000	331.9 ± 2.8	0.019 ± 0.000
<b>LoUS.Psy_50.50</b>	6239.3 ± 543.3	0.146 ± 0.002	1442.5 ± 122.0	0.049 ± 0.001
<b>LoUS.Psy_75.25</b>	23,533.5 ± 1267.8	0.159 ± 0.001	5530.1 ± 331.8	0.116 ± 0.002

The goodness of fitting ( $R^2$ ) ranged from 0.995–0.997 for  $G'$  and from 0.936–0.997 for  $G''$ .

The mechanical spectra Psy (control) and PsyL exhibited similar viscoelastic performance typical of well-structured weak gels, with slight frequency dependence (Figure 3a). This result is consistent with the rheological study by Haque et al. [27], in which it is also reported that *Psyllium* husk forms gel even at low temperature, this being an important feature to consider in food design. Moreover, the developed gels were more stable at higher frequencies than the ones produced with 10–15% chia flour at 90 °C (Ramos et al., 2016), reinforcing *Psyllium*'s potential as a valuable and sustainable biopolymer.

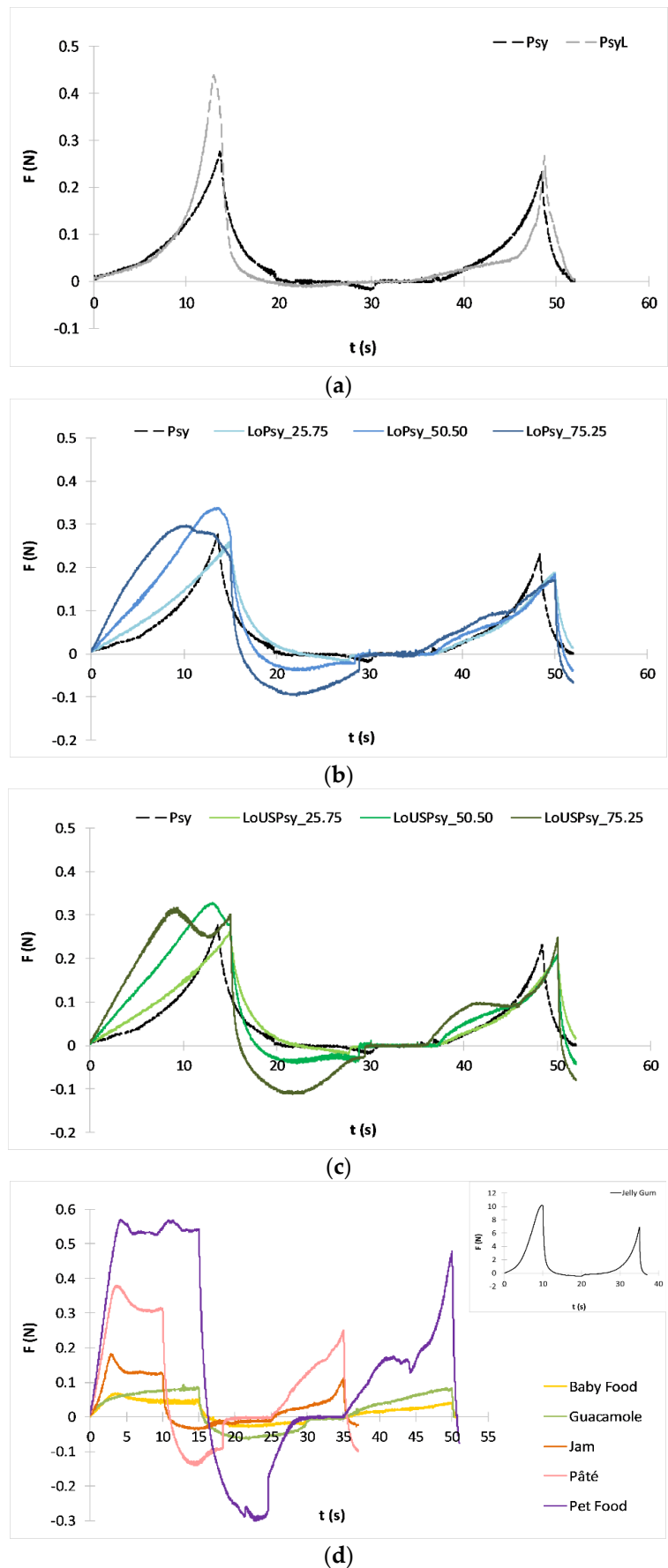
As mentioned earlier, PsyL appeared to be more fluid than the control (Psy), which was confirmed by the mechanical spectra, with both viscoelastic moduli of the control being higher than those of PsyL, and by the decrease of  $\alpha'$  (Table 3). Autohydrolysis promotes the solubilization of minerals [17] and the depolymerization of polysaccharides, namely alginate, fucoidan, and laminarin present in brown algae [28], rendering a liquid extract with an acidic pH ( $\approx 5$ , [19]). Since *Laminaria* liquid extract is a multicomponent matrix, its effect on the *Psyllium* gel properties are more complex, depending not only on the solution pH, but also on the type of ions in the solution, and even on the presence of peptides and other molecules resulting from the depolymerization of the polysaccharides. These polymer fractions can interfere with the gel matrix and exert an antagonism, leading to the reduction of gel links and de-structuring the material.

The ultrasonic pre-treatment applied to *L. ochroleuca* did not affect the rheological behavior of the *Laminaria-Psylim* gels. The weak-gel like behavior was maintained in all *Laminaria-Psylim* gel samples, and was more noticeable as the alga fraction increased, as can be observed by the increase in the  $b'$  parameter of the resulting power law (Table 3). As the incorporation of the alga solid fraction increased, the dependence of the material on frequency increased; however the value of the viscoelastic moduli increased probably due to the reduction of the number of links, but also stronger ones, which may be due to interactions between the alginate from the alga and fiber from the *Psylim* husk reinforced by calcium ions [29].

### 3.4. Texture Properties of the *Laminaria-Psylim* Gel Systems

Texture is a major quality parameter, and is crucial for consumer acceptance. Figure 4 presents the texture profiles of the samples and the commercial references.





**Figure 4.** Texture profiles of Psy (control) and PsyL (a), *Laminaria-Psyllium* gels (b), *Laminaria-Psyllium* gels subjected to ultrasonic treatment (c), and commercial products (d). The differences in the texturograms reflect the different sizes and shapes of the commercial products.

Each food system (e.g., jelly, mayonnaise, and baby food) presents texture properties that are specific, easily recognizable, and desired by the consumers. Despite the information given by the texture profile, the use of this methodology is very limited in the scientific literature. To our knowledge, no published studies refer to the texture features of commercial gelled products, therefore we show here some texture profiles of different commercial food products to illustrate the data for these systems to be used as a target for further product development. Based on these texture profiles (Figure 4d), three groups of samples can be discriminated: one composed of *Pâté*, *Pet Food* and *Jam*; another with *Baby Food* and *Guacamole*; and a third with *Jelly Gum*, whose graph is shown separately. In the first group of samples, the rupture point occurred at a low break distance, and the force reached a plateau until the probe retracted from the sample. This mechanical behavior was similar to what we found in LoUS.Psy\_75.25 (Figure 4c) and has also been reported by Genovese and co-workers [30] for pectin gels. The control and PsyL texture profiles (Figure 4a) were similar to the one obtained by Figueroa et al. [21] for fruit jellies with *Psyllium*, although with much lower magnitude.

From the texture profiles depicted, the firmness, adhesiveness, and cohesiveness values were calculated (Table 4).

**Table 4.** Texture parameters (firmness, adhesiveness and cohesiveness) of the gels developed and commercial references.

Samples	Firmness (N)	Adhesiveness (N.s)	Cohesiveness	
Developed gel samples	Psy (control)	0.296 ± 0.036 c,d,e	−0.034 ± 0.006 a	0.581 ± 0.051 b
	PsyL	0.419 ± 0.031 b,c	−0.032 ± 0.003 a	0.453 ± 0.024 c,d,e
	Lo.Psy_25.75	0.259 ± 0.028 c,d,e	−0.040 ± 0.012 a	0.517 ± 0.037 b,c
	Lo.Psy_50.50	0.316 ± 0.020 c,d,e	−0.274 ± 0.038 a	0.362 ± 0.044 e,f,g
	Lo.Psy_75.25	0.279 ± 0.039 c,d,e	−0.977 ± 0.075 c	0.403 ± 0.026 d,e,f
	LoUS.Psy_25.75	0.261 ± 0.007 c,d,e	−0.069 ± 0.017 a	0.538 ± 0.009 b,c
	LoUS.Psy_50.50	0.333 ± 0.026 b,c,d	−0.268 ± 0.042 a	0.354 ± 0.028 f,g
Commercial products	LoUS.Psy_75.25	0.319 ± 0.024 c,d,e	−0.978 ± 0.047 c	0.391 ± 0.030 d,e,f,g
	Baby Food	0.065 ± 0.008 d,e	−0.325 ± 0.034 a,b	0.475 ± 0.023 c,d
	Jelly Gum	10.314 ± 0.386 a	−3.596 ± 0.562 e	0.330 ± 0.028 f,g
	Guacamole	0.091 ± 0.011 e	−0.638 ± 0.106 b,c	0.748 ± 0.075 a
	Jam	0.182 ± 0.003 c,d,e	−0.298 ± 0.015 a	0.310 ± 0.015 g
	Pâté	0.364 ± 0.032 b,c	−0.854 ± 0.081 c	0.464 ± 0.038 c,d
	Pet Food	0.587 ± 0.066 b	−2.173 ± 0.148 d	0.371 ± 0.024 e,f,g

Psy (control), *Psyllium* gels prepared in water; PsyL, *Psyllium* gels prepared in autohydrolysis liquor; *Laminaria-Psyllium* gels: Lo.Psy\_25.75, Lo.Psy\_50.50, and LoPsy\_75.25; *Laminaria* with ultrasonic treatment-*Psyllium* gels: LoUS.Psy\_25.75, LoUS.Psy\_50.50, and LoUSPsy\_75.25. Data are presented as the mean ± sd. Different letters in the same column correspond to significant differences ( $p < 0.05$ ).

In general, it can be said that the firmness of the *Laminaria-Psyllium* system is independent of the level of *Laminaria* incorporation and the ultrasonic (US) pre-treatment. On the other hand, there are strong negative correlations between the adhesiveness ( $r = -0.888$ ,  $p < 0.05$ ) and cohesiveness ( $r = -0.844$ ,  $p < 0.05$ ) parameters and *Laminaria* concentration, and again, the US pre-treatment did not make any change. The fact that adhesiveness can be adjusted by keeping the firmness values constant could be useful in product development.

In a previous study by the authors [17], it was found that the texture parameters, especially adhesiveness, of the alga purée-like mixtures varied according to the ultrasonic treatment. However, that difference was not maintained after the addition of *Psyllium*. Since PsyL did not show adhesiveness values that could explain this result, one can conclude that there is a synergistic effect of *Laminaria* and *Psyllium* in this texture parameter, stronger than mere addition. Based on these results and looking at the texture profiles of the samples (Figure 4a–c), one can conclude that the texture properties of *Laminaria-Psyllium* gels are governed by the alga, what is in agreement with the dynamic rheology results. Texture profiles (Figure 4) confirm the relevance of alga in the system.

The texture parameters of the commercial products varied greatly. Comparing the texture parameters of the gelled systems developed with those of the references (Table 4), the developed *Laminaria-Psyllium* gels presented similar values to the commercial products. Although the systems developed are not finished products, but binary systems consisting of alga purée and *Psyllium* gel, this is a good starting point for the development of enriched gelled food products. These *Laminaria* and *Psyllium* whole materials can be used as alternatives to the biopolymers that are usually added to build up structure in foods. This is in line with the food trends regarding the use of natural products over processed ones.

It should be pointed out that in this approach, we developed the *Laminaria-Psyllium* gels using water, but the gels could also be produced in the same way using the AH liquor extract, thus taking advantage of the soluble compounds present there, namely phenolic compounds with antioxidant capacity [17]. Due to the differences in texture and rheological features between the control and *Psyl*, it is plausible to assume that *Laminaria-Psyllium* gels prepared in AH liquor would present distinct mechanical properties.

#### 4. Conclusions

Food gels were developed with *Laminaria* and *Psyllium* husk, adding value to both these natural resources. *Psyllium* husk interacted positively with this alga, reinforcing the viscoelastic behavior of the obtained gels. These novel functional gels with minimal processing can be a starting point to the development of several food and feed applications.

Food application studies are in progress to evaluate the demonstrated potential of this *Laminaria-Psyllium* partnership.

**Author Contributions:** P.F., M.D.T., and A.R. conceived and planned the experiments. P.F. and M.D.T. participated in the sample preparation and analysis, data analysis and interpretation of the results, and writing of the manuscript with input from all authors. I.S., A.R., and H.D. supervised the research work, contributed to the discussion of the data, and revised the manuscript. All authors contributed with suggestions and comments for the final version of the manuscript.

**Funding:** The authors are grateful to the Spanish Ministry of Economy and Competitiveness CTM2015-68503-R). M.D.T. thanks the Spanish Ministry of Economy and Competitiveness for her postdoctoral grant (IJC1-2016-27535). P.F. acknowledges her PhD grant from the Universidade de Lisboa (C0144M) and ERASMUS+ grant. The authors are also grateful to Fundação para a Ciência e a Tecnologia (Portugal) through the research unit UID/AGR/04129/2013 (LEAF) financial support.

**Conflicts of Interest:** The authors declare no conflict of interest.

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