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Licensed in Materials Engineering

# Printing of eco‐friendly solution based zinc‐tin oxide for device

Dissertation to obtain the master degree in Materials Engineering

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#### Printing of eco-friendly solution based zinc-tin oxide for device applications

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# Abstract

The research of amorphous metal oxide semiconductors for printed electronics applications, such as transparent and flexible devices, has been increasing to allow the low-cost production, good performance and large area upscaling of these materials.

The most commonly used solution-based semiconductors rely on toxic solvents and are indium-based. To avoid this critical raw material ZTO is the preferred alternative. Nevertheless, replacing the toxic solvent and the chloride-based tin precursor wich also contributes to the toxicity of the solution remains a challenge.

This work focuses on the development of eco-friendly ZTO precursor solutions to produce electronic devices at low temperature and optimization of flexographic printing of these ZTO inks.

ZTO/SiO<sup>2</sup> TFTs were successfully produced at 300 °C with non-toxic solvent, presenting a mobility of 2.98 ± 0.05 cm<sup>2</sup> /V.s and flexoprinted Ag/ZTO/ITO Schottky diodes were successfully produced at low temperature (150 °C + DUV) with a current on/off ratio of 1000. These devices show equivalent performance to the current state-of-the-art of ZTO devices produced with toxic solvents.

Finally, TFTs and Schottky diodes using non-toxic solvent and chloride-free ZTO precursors were successfully demonstrated for the first time.

This work clearly shows that it is possible to produce 100 % eco-friendly inks for high performance devices at low temperature suitable for large area printing.

Key-words: Amorphous metal oxide semiconductors, Printed electronics, Flexographic printing, Non-toxic solvent, TFTs, Schottky diodes, Low temperature

# Resumo

Os semicondutores de óxidos metálicos amorfos têm sido investigados em aplicações de eletrónica impressa, nomeadamente em dispositivos transparentes e flexíveis, permitindo a produção de baixo custo, bons desempenhos dos dispositivos e a produção em larga escala deste tipo de materiais.

Os semicondutores baseados em solução mais usados são normalmente compostos por índio e com solventes tóxicos. De forma a evitar estes materiais críticos ZTO é uma boa alternativa. A substituição destes solventes tóxicos e do precursor de estanho baseado em cloretos permanece um desafio.

Este trabalho foca-se no desenvolvimento de uma solução ecológica precursora de ZTO para a produção de dispositivos eletrónicos a baixa temperatura e a otimização da técnica de impressão de flexografia de tintas de ZTO.

TFTS de ZTO/SiO<sup>2</sup> foram produzidos com sucesso a 300 °C com um solvente não-tóxico, apresentando uma mobilidade de 2.98 ± 0.05 cm<sup>2</sup>/V.s e díodos de Schottky impressos por flexografia de Ag/ZTO/ITO foram também produzidos com sucesso a baixa temperatura (150 °C + DUV) com uma razão de corrente on/off de 1000. Estes dispositivos mostraram desempenhos equivalentes ao presente estado de arte de dispositivos de ZTO produzidos com um solvente tóxico.

Finalmente, TFTs e díodos de Schottky baseados em solução com solvente não-tóxico e com precursores de ZTO livres de cloretos foram demonstrados com sucesso pela primeira vez.

Este trabalho mostra claramente que é possível produzir tintas 100 % ecológicas para a produção de dispositivos com altos desempenhos a baixa temperatura ajustável para impressões à larga escala.

Palavras-chave: Semicondutores de óxido de metal amorfo, Eletrónica impressa, Flexografia, Solvente nãotóxico, TFTs, Díodos de Schottky, Baixa temperatura

# List of Abreviations

2- ME- 2- Methoxyethanol AFM- Atomic Force Microscopy AMOS- Amorphous metal oxide semiconductor ATR- Attenuated total reflectance CEMOP- Center of Excellence in Microelectronics and Optoelectronics Processes CENIMAT – Centro de investigação de materiais DSC- Differential scanning calorimetry DUV- Deep ultraviolet FET- Field effect transistor FTIR- Fourier transform infrared spectroscopy HP- Hot plate IGZO- Indium-gallium-zinc oxide IPA- Isopropyl alcohol ITO- Indium-tin oxide IZO- Indium-zinc oxide LCDs- Liquid crystal displays MIS- Metal-Insulator-Semiconductor M-O-M- Metal oxide metal MP- 1- Methoxy-2-Propanol NIR- Near-infrared OLEDs- Organic light-emitting diodes PET- Polyethylene terephthalate PTFE- Polytetrafluorethylene PVD- Physical vapor deposition R2R- Roll-to-roll rpm- Rotations per minute SCS- Solution combustion synthesis TFTs- Thin film transistors TG- Thermogravimetry UV- Ultraviolet VIS- Visible XRD- X-ray diffration ZTO- Zinc-tin oxide

# List of symbols

µ - Mobility (cm<sup>2</sup>/V·s) µFE - Field-effect mobility (cm<sup>2</sup>/V·s) µsAT – Saturation mobility (cm<sup>2</sup>/V·s)  $C_i$  - Gate capacity per unit area (F/cm<sup>2</sup>) Eg- Bad gap energy (eV) I<sub>D</sub>- Drain current (A) L – Channel Length  $(\mu m)$ SS- Subthreshold Slope (V/dec) VG- Gate Voltage (V) VT - Threshold voltage (V) W- Channel Width (µm)

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# Motivation and objectives

Research in electronics has increasingly focused on printed electronics for the production of devices such as thin film transistors (TFTs), diodes, etc. This is an area with great potential because these techniques provide a high quality electronic products that are thin, they are low cost techniques where it is possible to produce devices with good performances. It is also applicable in the development of materials with applications in flexible electronics that can be processed by solution and environmentally friendly. [1] This makes research in this area growing and becoming more important. The global market in this area is constantly growing, being expected to increase to \$45 billion in 2021. [2]

Flexographic printing technique it will be the printing technique more used on this work and it results in thinnest printed layer, and is the fastest printing technique! [2]

The objectives in this dissertation are mostly focus on environmental friendly and low cost procedures. There are four main objectives:

- The fundamental objective is to optimize flexographic printing parameters and ZTO precursor solution to produce solution-based eco-friendly TFTs and Schottky diodes.
- Use low cost techniques to deposit the films. The most used methods for the deposition of films are physical vapor deposition (PVD) techniques because they originate films with high uniformity and reproducibility. But these techniques involve high production costs so others methods are used, like printing techniques.
- Replace the solvent by a more eco-friendly one. The precursor solution will be improved by changing the most commonly used solvent 2-methoxyethanol (2-ME) since is a toxic and cancerogenic solvent. In order to replace 2-ME, a more friendly substitute 1-methoxy-2-propanol (MP) will be used being compatible in large-scale production.[3]
- Decrease the process temperature. One considerable challenge is to decrease the forming temperature of zinc-tin oxide (ZTO) precursor solution thin films and still reach similar properties to the ones produced at higher temperatures, which until now has not been reported.

# 1. Introduction

<span id="page-22-0"></span>In this chapter it will be presented a brief introduction about the topic in order to contextualize the work done.

### <span id="page-22-1"></span>1.1. Amorphous metal oxide semiconductors

Amorphous metal oxide semiconductors (AMOS) have been highly studied as an active layer for TFTs to use in transparent electronics. [4]

These materials show high optical transparency, smooth and amorphous structures, and high electron mobility, which leads to higher current density and circuit speed [5] with thermal and environmental stability.

AMOS has been exploited with the aim of improving the electronic transport in the metal oxide and also of lowering the process temperature, combining this with cheap and flexible substrates. [4]

The most commonly used semiconductors for this type of devices are indium-zinc oxide (IZO) and indium– gallium–zinc-oxide (IGZO) due to their good electrical performance and stability. [3] IGZO devices usually have a high field effect mobility, faster response speed, good flexibility and lower energy consumption. [6] However, indium and gallium are critical raw materials, imposing significant constrains regarding the sustainability of this approach for large-area electronics. Zinc–tin-oxide (ZTO) is a good alternative, providing abundant materials, although ZTO have some instability issues in the tin oxidation state, which can be +4 on  $SnO<sub>2</sub>$  form and +2 on SnO form. For a n-type transparent semiconductor film the  $SnO<sub>2</sub>$  form is preferred and the SnO to form an intrinsic p-type semi-transparent film. However, these two forms of tin oxide are easily assembled so it is challenging to guarantee only the  $SnO<sub>2</sub>$  formation in the ZTO thin film. [7] But even so ZTO already showed similar performance and processing temperature to IGZO, it was reported TFTs produced at 350°C with 2ME-based ZTO with a saturation mobility of 2.3 cm2/V⋅s. [3][8][9] The most used methods for the deposition of metal oxide semiconductors are physical vapor deposition (PVD) techniques, performed in vacuum, in order to achieve high uniformity and reproducibility of the thin films. However, these techniques involve high production costs, which is a barrier especially when considering large scale production. Taking that into account, one of the objective is replace these high cost techniques by solution deposition techniques, like spin-coating, screen printing, inkjet printing and more specifically in this work, flexographic printing arise due to their up-scale compatibility. [9] For these applications it is required to lower the temperature process to be compatible with temperature sensitive flexible substrates, like PET and PEN. [3]

### <span id="page-22-2"></span>1.2. Toward lower process temperature

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New methods and techniques have been studied, like solution combustion synthesis (SCS) and deep ultraviolet (DUV) treatment, to obtain high-quality films at low temperature on low cost flexible substrates. [9]

A particular case of sol-gel synthesis, SCS allows the growth of metal oxide films from solution using low temperatures. SCS is based on a local exothermic redox reaction within the film. This exothermic reaction will provide energy to continue the reaction and form the oxide using a low thermal annealing. [5][4][10] SCS require metal nitrates, which act as a source of metal and oxidizing agents, and an organic fuel, which acts as a reducing agent. The most commonly used fuels are urea, acetylacetone and glycine. [3] When chlorides are used as the source of metal ions, a combustion aid is needed. In this case ammonium nitrate is often used. The mostly used solvents in these solutions for electronic applications are 2-ME and acetonitrile. [3] The application of this method has already proven successful in the production of oxide materials for electronic devices in dielectric and semiconductor thin films. [3]

The SCS of ZTO can be represented by the combination of the following reactions:

$$
3Zn(NO_3)_2.6H_2O + 5CO(NH_2)_2 \rightarrow 3ZnO + 5CO_2 + 8N_2 + 28H_2O (1)
$$
  

$$
SnCl_2 + \frac{1}{3}CO(NH_2)_2 + NH_4NO_3 + \frac{1}{2}O_2 \rightarrow SnO_2 + \frac{1}{3}CO_2 + 2HCl + \frac{4}{3}N_2 + \frac{5}{3}H_2O (2)
$$

From these reactions the oxides of zinc and tin are formed, being merged to form ZTO. [3]

In order to reduce the process temperature and improve film quality, several techniques are used, in addition to the SCS, like ultraviolet (UV) irradiation. This additional energy provided by the UV helps to decrease the presence of oxygen vacancies, enhancing the condensation and film densification of the metal-oxygen-metal (M-O-M). [5]

For deposition of solution-based oxides, spin-coating is the most common technique to deposit the thin films. However, this process has some drawbacks, like the non-compatibility with roll-to-roll (R2R) techniques and waste of material, where just 5 % is used. For this reason additive deposition techniques have been selected for the films deposition.

# <span id="page-23-0"></span>1.3. Printing techniques: Flexography Printing and Inkjet Printing

Nowadays, the aims are increasingly focused on electronic manufacturing in large areas for industrial printing processes, with high productivity and reproducibility, on flexible and low-cost substrates. Inkjet printing is an excellent technique for writing patterns with high resolution, circumventing the need for the use of conventional methods of photolithography and vacuum deposition. [11] It includes the ability to dispense uniform droplets in the picolitre range with a high degree of precision. [12] Nevertheless, inkjet printing technique is used in non-contact discontinuous processes while the flexography printing technique is a continuous R2R process, as exemplified i[n Figure 1.1](#page-23-2) and it is industrially scalable.

More specifically, flexography technology consists of an elastic printing plate or roller patterned with reliefs. It is composed of an anilox where ink is placed, this controls the amount of ink transferred to the mask, with the help of doctor blade that removes excess of ink. The ink that has been passed into the mask is transferred to the substrate by contact force. This force depends on the distance that the anilox, the mask and the substrate meet each other. [13]



<span id="page-23-2"></span>*Figure 1.1- Flexography Printing Scheme [4]*

In addition to the advantages already mentioned, this technique also offers good line edge definition, and a have high range 2-500 cp in the ink viscosity applicability. [5][13]

These printing techniques have been widely used to produce electronic devices, like TFTs and diodes. [5]

# <span id="page-23-1"></span>1.4. Thin Film Transistors: TFTs

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TFT is a field effect transistor (FET). TFTs are usually used as on / off switches with applications on sensor devices, active matrix backplanes of flat panel monitors where it transforms the pixels of liquid crystal displays (LCDs) and organic light emitting devices (OLEDs). The amplification effect, small size, low cost and large area fabrication are some of the advantages that make this type of devices so interesting and with lots of potential. [6] [14][15][16]

TFTs are composed of 3 electrodes: the gate, the drain and the source as showed in [Figure 1.2.](#page-24-0) The gate acts as a switch, where a voltage is applied and allows current to flow between the other two electrodes. From this voltage applied between the source and the drain, an electric field is generated and a conducting channel is formed. [1] The current passing between the two electrodes, source and drain, is controlled by varying the potential gate. [17][15]



<span id="page-24-0"></span>*Figure 1.2- (a) 2D schematic of the straight section of a TFT* [16] *and (b) 3D scheme of a TFT [1]*

These devices are based on a metal-insulator-semiconductor (MIS) structure. [14] Composed of several layers: the substrate (silicon, glass or flexible substrates) [1][18]; the gate electrode, a layer of an insulator (or dielectric) that place between the gate and a layer of a semiconductor. A semiconductor is a material that has its conductivity above an insulator but below a conductor, while an insulator blocks current passing and in a conductor the charges propagate easily, in a semiconductor the current that passes can be adjusted through electric fields and concentrations of impurities, hence these materials are widely used in electronic applications. Above the semiconductor are the contacts (source and drain).[1][15] These contacts are a layer of a metal, for example aluminum. [14]

Observing the energy band diagram in [Figure 1.3](#page-24-1) it is possible analyze the behavior and functioning of an ideal TFT. This behavior depends on the voltage,  $V_G$ , applied to the gate. [15]



<span id="page-24-1"></span>*Figure 1.3- Energy band diagram of an ideal TFT for different bias conditions [13]*

In the ideal case, the V<sub>G</sub> can be negative or positive. If this is negative (V<sub>G</sub><0), Figure 3 (b), it will repel the mobile electrons from the dielectric/semiconductor interface, forming a depletion region, in this case the flowing current (I<sub>D</sub>) is too low, the transistor is in off-state. For  $V<sub>G</sub>$  >0, Figure 3 (c), the electrons are accumulated at the dielectric/semiconductor interface thus making a downward bandbending, in this case there is a considerable current to pass in the transistor  $(I<sub>D</sub>)$ , the transistor is in on-state.

In the real case, the voltage value corresponding to the significant charge accumulation at the dielectric/semiconductor interface is not 0. This voltage value is called threshold voltage ( $V_T$ ). In this way the transistor only reaches the on-state when the voltage applied between the gate and the source  $(V_G)$  is higher than  $V_T$ , provided that positive drain voltage (V<sub>DS</sub>) is applied. When this happens there is a current passing between the two electrodes, source and drain  $(I_D)$ . [1][15][17][19][13]

The  $V_T$  can be positive or negative. For a n-type TFT If the  $V_T$  is negative the transistor is in depletion mode, if it is positive the transistor is in enhancement mode which is more commonly used because it is not necessary to apply voltage to reach the off-state. However both modes can be used with different applications. [20]

By analyzing the output curve of a transistor in Figure  $1.4$  (a) it is possible to detect two types of progress with the variation of  $V_{DS}$ . A first region that is linear (for small  $V_{DS}$  values) and a second region from a certain point tends to be constant (higher V<sub>DS</sub> values). This point is called the pinch off voltage which is when the channel begins to strangle at the end of the drain. These two zones are designated linear region and saturation region, respectively.



<span id="page-25-1"></span>*Figure 1.4- Typical (a) output and (b) transfer curves of a n-type oxide TFT* [15]

The behavior of a transistor on a triode region is described by:

$$
I_{DS} = \frac{W}{L} \cdot C_i \cdot \mu_{FE} \cdot \left[ (V_{GS} - V_T) V_{DS} - \frac{1}{2} V_{DS}^2 \right] \qquad , when \ V_{DS} < V_{GS} - V_T \tag{1.1}
$$

where W and L are the measures of a channel, width and length respectively, C<sub>i</sub> is the gate capacity per unit area,  $\mu_{FF}$  is the field-effect mobility. For very low V<sub>DS</sub>, the quadratic term can be neglected, yielding a linear relation between  $I_{DS}$  and  $V_{DS}$ . In this case, the accumulated charges are considered to be uniformly distributed throughout the channel. [1][15]

In the saturation mode the behavior is described by:

$$
I_{DS} = \frac{W}{2L} \cdot C_i \cdot \mu_{SAT} (V_{GS} - V_T)^2 \qquad , when \ V_{DS} > V_{GS} - V_T \qquad (1.2)
$$

where  $\mu$ <sub>SAT</sub> is the saturation mobility. [1][15]

From the transfer curve (Figure 4(b)) it is possible to extract some parameters: On/Off ratio,  $V_T$ , turn-on voltage (V<sub>ON</sub>), mobility ( $\mu$ ) and subthreshold slope (S).

On/Off ratio- It is the ratio between the maximum current (IoN) and the current when the transistor is in the off-state  $(I<sub>OFF</sub>)$ . [1][15]

Opening voltage, Von - Voltage from which the transistor reaches the on-state.

Mobility  $(\mu)$  - Measure the efficiency of carrier transport in a material. Can be obtained from the transconductance (g<sub>m</sub>) with low V<sub>D</sub> or with high V<sub>D</sub>. Field-effect mobility (μ<sub>*FE*</sub>) or Saturation mobility (μ*sat*) respectively.  $\sim$   $\sim$ 

$$
\mu_{FE} = \frac{g_m}{\frac{W}{L}c_i \cdot v_{DS}} \tag{1.3} \mu_{SAT} = \frac{\frac{C_0 \sqrt{1DS}}{3V}c_i}{\frac{W}{2L}c_i} \tag{1.4} \tag{1.1}[15]
$$

Subthreshold slope (SS)- It is the voltage that is required to apply to the current increase in a decade.

$$
SS = \left(\left|\frac{\partial \log l_{DS}}{\partial V_{GS}}\right|_{\max}\right)^{-1} \tag{1.5}\tag{1.5}
$$

On the present state of the art ZTO TFTs were already produced with 2ME showed saturation of mobility of  $1.9 - 3.8$  cm<sup>2</sup> V<sup>-1</sup> s<sup>-1</sup>). [9]

#### <span id="page-25-0"></span>1.5. Schottky diodes

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Schottky diode, barrier diode or low voltage diode are the names that can be given to the device which was discovered for the first time by a scientist named Walter.H.Schottky. [21]

It is a device that is composed of a metal-semiconductor junction as showed in [Figure 1.5](#page-26-0) (b), creating a schottky barrier unlike the pn junction diodes which are constituted by a p-type and n-type material, forming a pn junction. The power drop is smaller in schottky diodes than in pn junction diodes. [21] [22][23][24]

Shottky barrier is a potencial energy barrier that is formed at the interface zone at the metal-semiconductor junction. When the metal join the n-type semiconductor, the free electrons of the semiconductor move to the metal in order to establish the equilibrium state [\(Figure 1.5\(](#page-26-0)c)). In this way, positive ions appear on the n-type semiconductor side, near the interface, and negative ions on the metal side, thus forming a depletion region. For electrons to flow across the diode this potential barrier must be overcome. [25]



<span id="page-26-0"></span>*Figure 1.5- (a) Schottky diode symbol, (b) 2D scheme of a schottky diode and (c) Schottky barrier scheme [23]*

This barrier controls the current conduction and its capacitance behavior. These diodes are majority-carrier devices where the storage effect of minority carriers is insignificant, this contributing to their fast recovery, and offer ultra-high switching speed. [18]

[Figure 1.6](#page-26-1) represents a schematic of the electron energy relations of a metal with a high working function and a n-type semiconductor that are in separate systems. The work function is the difference between the fermi level and the vacuum level (energy level of the electrons that are outside the material).



<span id="page-26-1"></span>*Figure 1.6- The electron energy relations of a metal and an n-type semiconductor [23]*

However, the work function of the metal is larger than the work function of the semiconductor so, when these systems come in contact, charges from the semiconductor go to the metal thus establishing a thermal equilibrium. The Fermi levels align and a barrier is created which has to be crossed. This is the operation of a schottky diode [18][25]

It is necessary apply a certain voltage to overcome this barrier, this voltage is called a voltage drop, when the current passes in the diode. This voltage drop is the amount of voltage that is wasted to turn on the diode. In a Schottky diode this voltage drop varies from 0.15V to 0.45V, it is lower than in a pn junction diode (0.6V to 1.7V) [\(Figure 1.7\)](#page-27-0). This provides better system efficiency and higher switching speed on Schottky diodes. [25] [26]



<span id="page-27-0"></span>*Figure 1.7- Variation of the voltage drop between the pn junction diode and a schottky diode*

To characterize this type of diodes can be determined several parameters like the opening voltage ( $V_T$ ) and the current on/off ratio.

On the present state of the art Schottky diodes were alreadybproduced with IGZO by sputtering showed an on/off ratio of  $2.3 \times 10^2$ . [27]

In this dissertation the main objective is to optimize flexographic printing parameters of ZTO precursor solution to produce TFTs and diode devices. Additionally, the precursor solution will be improved by changing the most commonly used solvent 2-methoxyethanol (2-ME) since is a toxic and cancerogenic solvent. In order to replace 2-methoxyethanol (2-ME), a more eco-friendly substitute 1-methoxy-2 propanol (MP) will be used being compatible in large-scale production.

One considerable challenge is to decrease the forming temperature of ZTO thin films and still reach similar properties to the ones produced at higher temperatures, which until now has not been reported.

This work is a crucial part of the European project Supersmart, involving the development and upscale of environmentally friendly semiconductor metal oxide inks and their flexographic printing on different substrates (conventional, polymeric and cellulose-based).

# 2. Materials and methods

<span id="page-28-0"></span>In order to produce TFTs and Schottky diodes from an eco-friendly ZTO precursor solution it was important to study the different parameters for this solution optimization and take into account important parameters: the used solvent, the used solutions concentrations, the molar proportions and the stability of the solutions. The process temperature was also reduced, using the DUV exposure. With the objective of large scale production on flexible substrates the flexography technique was performed for the deposition of the ZTO semiconductor layer. Being possible to produce low cost and eco-friendly devices in large scale and with a low thermal budget.

# <span id="page-28-1"></span>2.1. Precursor Solutions Preparation and Characterization

The precursor solution of ZTO was produced using two different methods: With and without combustion. The ZTO solution with combustion was produced from urea (CH<sub>4</sub>N<sub>2</sub>O, Fisher Scientific, >99.5%) which acts as a fuel for the combustion reaction to occur, the ammonium nitrate (NH4NO3, Roth, ≥ 98%), which act as a source of metal and oxidizing agents, the zinc nitrate (N2O6Zn∙6H2O, Sigma-Aldrich, 98%) and tin chloride (Cl2Sn∙2H2O, Sigma-Aldrich, ≥98%), dissolved in 1-methoxy-2-propanol ( MP, C4H10O2, Roth, ≥99%). The ZTO solution without combustion was produced only from the zinc nitrate and the tin chloride dissolved in MP. Different ratios of zinc nitrate to tin chloride (2:1 and 1:1) were performed. Another parameter that varied was solute concentrations, different values (0.1; 0.2; 0.6; 1; 1.2; 1.6 M) were tested. To measure the viscosity of all the solutions a BROOKFIELD CAP 2000+ viscosimeter was used.

To the ZTO precursor solution with combustion the urea is dissolved in the solvent and then the ammonium nitrate and zinc nitrate are added and the mixture is allowed to stir until dissolved. To the ZTO precursor solution without combustion just zinc nitrate is dissolved in solvent.

For both methods, tin chloride was added to the solvent and stirred until it dissolves, after that the solution is mixed with the zinc nitrate solution<sup>1</sup> and left magnetically stirring for 12 h.

Cl-free ZTO precursor solutions were produced where the tin chloride was replaced with tin(II) acetate (C4H6O4Sn, Sigma-Aldrich) or with tin acetylacetonate ( C10H14O4Sn, Sigma-Aldrich), in this case the precursor solution was prepared at Fraunhofer IISB, a partner in the SuperSmart project.

Thermal characterization of precursor solutions were performed by thermogravimetry and differential scanning calorimetry (TG-DSC). TG-DSC analysis were performed under air atmosphere up to 550 °C with a 10°C/min heating rate in an aluminum crucible with a punctured lid using a simultaneous thermal analyzer, Netzsch (TG-DSC - STA 449 F3 Jupiter).

### <span id="page-28-2"></span>2.2. Thin Film Deposition and Characterization

.

Prior to deposition the rigid substrates (glass, glass with ITO and Silicon wafers) were cleaned in a ultrasound bath at 60°C, during 10 minutes in acetone, 10 minutes in 2-isopropanol and finally passed in water and dried with N2. For the flexible substrates (poliamide Kapton and Polyethylene terephthalate (PET) with ITO) the procedure was similar but without the acetone bath.

After that it is necessary improve the wettability of the surfaces. In rigid substrates, a 15 minutes UV/Ozone surface activation step was performed for a distance lamp of 5 cm using a PSD-UV Novascan system and in flexible substrates was necessary make an  $O<sub>2</sub>$  plasma treatment for 1 min. at a pressure of 0.16 mbar to the substrates using a Diener Plasma Surface Technology to make the surfaces more hydrophilic.

Then a spin-coating technique (Laurell Technologies) was used to deposit the ZTO ink on the rigid substrates with an area of 2.5×2.5 to make the thin film analysis. A polytetrafluorethylene (PTFE) filters, with 4.5 µm mesh was used to filter the ink before the deposition. Two layers of the semiconductor were deposited at 2000 rpm during 35 s for each layer.

To reduce the thermal budget, different temperatures were used to produce the ZTO thin films. After deposition of the first layer, the substrate was quickly placed in a hot plate (HP) at 300°C for half an hour to anneal, where the reaction occur and the film is densified, after this time is necessary another UV treatment during 5 minutes to reinforce the hydrophilic surface and then the second layer is deposited proceeding in the same way.

To decrease the process temperature the annealing was made in a different way. Following deposition the

<sup>&</sup>lt;sup>1</sup>These solutions have to come together quickly because the tin chloride solution is not very stable when isolated, so it is necessary to join this solution to the zinc nitrate solution immediately before it starts to react and form hydroxides.

substrate was placed on a heating plate at 120°C for 5 minutes to evaporate the solvent and then an annealing was made at 150°C with deep ultraviolet radiation (DUV) during 30 minutes for a distance lamp of 2 cm using a PSD Pro Series Digital UV Ozone Novascan system.

For the flexible substrates, after a plasma treatment, a flexography technique was performed to print the ZTO ink. After deposition the substrate was placed on a HP at 120 °C during 10 minutes to evaporate the solvent and following the low temperature annealing was used as mentioned before.

The roughness of the surfaces of substrates was analyzed by atomic force microscopy (AFM, Asylum MFP3D) and the resistance of conductive substrates surfaces was measured with a multimeter.

To characterize the films Fourier Transform Infra- Red (FTIR) spectroscopy was used to confirm the absence of organic substances. FTIR data were recorded using an Attenuated Total Reflectance (ATR) sampling accessory (Smart iTR) equipped with a single bounce diamond crystal on a Thermo Nicolet 6700 Spectrometer. The spectra were acquired with a 45° incident angle in the range of 1800−540 cm-1 and with a 4 cm<sup>-1</sup> resolution.

In the Perkin Elmer lambda 950 UV/VIS/NIR spectrophotometer the transparency of the films was evaluated where transmittance spectra were obtained from 190 to 800 nm. To measure the thickness of the films spectroscopic ellipsometry (a Jobin Yvon Uvisel system) measurements were made over an energy range of 1.5−5.0 eV with an incident angle of 70°. To determine structural characteristics of the films was performed by an X'Pert PRO PANalytical powder diffractometer using with Cu Kα line radiation (λ = 1.540598 Å) with angle of incidence of the X-ray beam fixed at 0.9°.

### <span id="page-29-0"></span>2.3. Electronic Device Fabrication and Characterization

The conditions of the production of the ink were optimized for each application. To produce TFTs the ZTO precursor solution used was without combustion, with (2:1) ratio, with 0.1 M concentration. However for printed Schottky diodes production a ZTO precursor solution with combustion, with a (1:1) ratio and a 1 M concentration was used. The annealing treatment also varied, for TFTs a HP at 300 °C during 30 minutes treatment was performed while for diodes low temperature treatment with DUV exposure was used as mentioned in section 2.2.

After the deposition of the semiconductor it is necessary to make the S/D contacts deposition. For TFTs the material deposited for these contacts is aluminum. This deposition is made by thermal evaporation (homemade equipment) while to deposit silver (Ag) contacts on the Schottky diode (the Ag are used due to its work function that is higher than the aluminum) an electron beam evaporation (homemade equipment) and Inkjet Printing Pixdro LP50 were both used. Before the deposition of the contacts by inkjet the substrates need a surface activation step as mentioned in section 2.1 to improve their wettability. After the deposition of contacts an annealing is done at 150°C for 30 minutes to improve the interface between the semiconductor and the contacts.

Finally the devices are characterized in a Cascade Microtech Keysiht B1500A semiconductor parameter analyzer, and the data were treated on OriginPro 9.0.

[Figure 2.1](#page-29-1) represents a scheme of the experimental procedure mentioned before.



<span id="page-29-1"></span>*Figure 2.1- Scheme of the procedure to produce TFTs and Schottky diodes* 

# 3. Results and Discussion

<span id="page-30-0"></span>In this chapter the results obtained will be presented and analyzed. It will be discussed the characterization of the films and the electrical characterization of the devices: TFTs and Schottky diodes.

# <span id="page-30-1"></span>3.1. Effect of ZTO ink proportion, precursor and concentration on stability and quality of the ink.

In order to optimize the solutions for devices production it was necessary to analyze their short and long term stability. It is critical that the solutions remain stable for large scale applications.

### <span id="page-30-2"></span>3.1.1. Stability of solutions

.

Tin chloride and zinc nitrate solutions in MP were prepared separately to analyze their stability. A solution of tin acetate in MP was also prepared to try to replace tin chloride and make the solution 100% ecofriendly, not only by replacing the solvent but also eliminating the chlorides that also contribute to the toxicity of the solution.

As can be seen from [Figure 3.1,](#page-30-3) zinc nitrate solutions remain stable, keeping transparent, but the tin chloride were whitish, which indicates the deposition of precursor in the bottom, meaning that a reaction was initiated and hydroxides were formed. The tin acetate solution turned yellow indicating that the tin acetate did not dissolved in MP so it is not a good solution.



<span id="page-30-3"></span>*Figure 3.1- (a) Tin chloride solutions, (b) Zinc nitrate solution in MP and (c) tin acetate in MP solution*

It was noticed that the tin chloride solution was transparent immediately after the dissolution, turning white only a few moments later. Based on this, that solution was added to the zinc nitrate solution immediately to avoid the formation of hydroxides. To make this test a ZTO precursor solutions with and without combustion were made, with  $(1:1)$  and  $(2:1)$  ratio, respectively, with 0.1 M concentration (Figure [3.2\)](#page-30-4).

<span id="page-30-4"></span>

*Figure 3.2- (a) Solution with combustion (1:1) and (b) Solution without combustion (2:1)*

It is observed in [Figure 3.2](#page-30-4) that the solution with combustion in the ratio  $(1:1)$  and the solution without combustion in the ratio (2:1) were both stable with a 0.1 M concentration.

#### <span id="page-31-0"></span>3.1.2. Aging of the solutions

Aging of solutions were evaluated to see if they remain stable over time and results are shown in [Table 3.1.](#page-31-3)

<span id="page-31-3"></span>



Based on the [Table 3.1](#page-31-3) it can be observed that these solutions must be stable over time, even after one month as these remain transparent, which indicates that no reaction occurred.

#### <span id="page-31-1"></span>3.1.3. Viscosity of solutions

.

The viscosity of the solutions was changed by the increasing the ZTO ink concentration with a molar proportion (1:1) with and without combustion. The results obtained are represented in [Figure 3.3.](#page-31-2)



<span id="page-31-2"></span> *Figure 3.3- Variation of viscosity with the concentration of solution of ZTO inks prepared with and without urea*

There is an exponential increase of the viscosity with the concentration for both solutions. The solution with combustion has a higher viscosity than the solutions without combustion, this was expected because the solution with combustion has more salts dissolved (urea and ammonium nitrate).

#### <span id="page-32-0"></span>3.1.4. Thermal characterization of solutions

 The DSC-TG technique were performed to investigate the decomposition behavior of the metal oxide precursors. This thermal analysis was made for ZTO with combustion, (1:1) ratio, with 0.2 M concentration. The obtained results are shown in [Figure 3.4.](#page-32-3)



<span id="page-32-3"></span>*Figure 3.4- TG-DSC analysis of a ZTO precursor solution with combustion (1:1), with 0.2 M concentration*

In [Figure 3.4](#page-32-3) two endothermic peaks are observed at 80 °C and 128 °C where there is an abrupt mass loss indicate the solvent evaporation occurs. At 170 °C there is an exothermic peak where the combustion reaction of the organic fuel with the metal nitrates occurs, when energy is released. At 260 °C an endothermic peak is noted accompanied by a mass loss, this is attributed to the degradation of residual organics. This analysis indicates that 200 °C is the temperature required for the ZTO formation reaction to occur.

### <span id="page-32-1"></span>3.2. ZTO thin films characterization

.

Several techniques were used to characterize the films. To determine film thickness, composition, transparency and crystallinity.

#### <span id="page-32-2"></span>3.2.1. Fourier Transform Infra- Red (FTIR) spectroscopy

To confirm the occurrence of the reaction, the formation of zinc tin oxide and the absence of organic substances, FTIR analysis of the films was performed. A film of solution with combustion (1:1) with 0.1M concentration was analyzed, this film had a 250 °C and 300 °C annealing treatment for 30 minutes and a film of solution with combustion (1:1) with 0.2M concentration that had a 150 °C + DUV annealing

treatment for 30 minutes was also analyzed. All the films were deposited on silicon substrates. The results obtained are presented i[n Figure 3.5.](#page-33-2)



<span id="page-33-2"></span><span id="page-33-1"></span> *Figure 3.5- FTIR spectra for ZTO thin films produced from different precursor solutions.*

The results are identical between regardless of different annealing treatments (250 °C, a 300 °C and 150 °C + DUV). In all of them there is no longer organic substances, because these would present a peak round to 3500  $cm<sup>-1</sup>$ , and in this zone of the graphic no peak is observed. [28]

The peak observed at 1100 cm<sup>-1</sup> represents the silicon-oxygen bond, this peak is related to the native SiO<sub>2</sub> layer on the silicon substrate that reacts with the oxygen from the air. [29] At about 750 cm<sup>-1</sup> there is a peak which is relative to Sn-O vibration and a peak at  $600<sup>-1</sup>$  which represents the Zn-O vibration. [30] So it can be concluded that the reaction was successful even at low temperature.

#### <span id="page-33-0"></span>3.2.2. Optical characterization

#### UV/VIS spectrophotometry

.

The use of transparent electronics is increasing and is more and more important to be able to make transparent devices. Optical characterization of the films was performed using a UV/VIS spectrophotometer. Solution combustion synthesis based ZTO (1:1) films with a 300 °C and a 150 °C + DUV annealing treatment for 30 minutes were deposited on a glass substrate and analyzed. The obtained data are shown in [Figure 3.6.](#page-33-3)



<span id="page-33-3"></span>*Figure 3.6- Transmittance spectra of combustion ZTO (1:1) thin films*

#### Printing of eco-friendly solution based zin-tin oxide for device applications

From the graphic it is possible observe that both the films are transparent with 90 % transmittance in the visible light range. [31] It is also possible observe that the radiation is transmitted below 300 nm which means that the ZTO film absorbs the UV radiation.

#### Spectroscopic ellipsometry

In order to determine the thickness and the band gap energy ( $E_g$ ) values of the ZTO films, ellipsometry technique was performed using the Tauc Lorentz2 model. The films produced from different combustion ZTO (1:1) precursor solutions by varying solution concentration, number of layers and annealing treatment were analyzed. The results obtained of thickness and of  $E_g$  can be found i[n Table 3.2.](#page-34-2)

<span id="page-34-2"></span>



Observing the variation of the thickness values it is possible observe that with the decrease of temperature the thickness increases slightly because with higher temperatures the film becomes denser and more condensed. The band gap energy remains approximately constant with temperature.

#### <span id="page-34-0"></span>3.2.3. Structural characterization

.

To get a film with good features like the ones mentioned earlier, it must be amorphous. Previous published results show that ZTO films that took 300 °C annealing treatment are amorphous (see Annex 2A). [9] To confirm if the film deposited was amorphous after a DUV treatment the XRD technique was used. The analyzed film was made from the ZTO precursor solution with combustion (1:1) with 0.6M concentration and with a 150 °C + DUV annealing treatment during 30 minutes, deposited on silicon substrate. The obtained results can be found in th[e Figure 3.7.](#page-34-1)



<span id="page-34-1"></span> *Figure 3.7- XRD spectrum of ZTO thin film*

Observing the graphic no significant peak is noticed in the curve. It can be conclude that the film is amorphous as desired. With the annealing treatment and the DUV the reaction occurred and the zinc tin oxide was formed. Otherwise there were peaks in the spectrum as can be seen in Annex 2B.

#### <span id="page-35-0"></span>3.3. Electrical characterization of ZTO devices

TFTs and Schottky diodes were produced with solution based ZTO and their electrical behavior was characterized.

#### <span id="page-35-1"></span>3.3.1. TFTs

.

The TFTs were fabricated on Si/SiO<sub>2</sub> substrates by depositing 2 layers of ZTO solution with 0.1M concentration using (1:1) and (2:1) proportions and using two methods, with combustion and without combustion precursor. The annealing treatment was 30 minutes at 300 °C. The deposited contacts are aluminum. The V<sub>D</sub> used was 20 V and the V<sub>G</sub> was from -10 V to 20 V. The conditions that shows good results were (2:1) ZTO without combustion and (1:1) ZTO with combustion. The transfer or input curves and output curves of TFTs were analyzed. The obtained results can be observed in [Figure 3.8](#page-35-2) an[d Figure 3.9.](#page-35-3)



<span id="page-35-2"></span>*Figure 3.8*- *IV transfer curves of TFTs produced by ZTO precursor solution 0.1 M concentration (a) without combustion (2:1) (inset photograph of the TFTs produced) and (b) with combustion (1:1) on Si/SiO<sup>2</sup> wafer.*



<span id="page-35-3"></span>*Figure 3.9- IV Output curve of TFTs produced by ZTO precursor solution 0.1M concentration, on Si/SiO<sup>2</sup> wafer.*

From these curves it can be concluded that the devices have a typical TFT behavior because for negative  $V_G$  the passing current between the drain and source ( $I_{DS}$ ) is too low and from a certain voltage (near to zero) the current that passes increase. From transfer curves it is possible determine the characteristic device parameters;  $V_{on}$ , SS (equation 1.5),  $\mu$ s $\Lambda$ T (equation 1.4) and on/off ratio. These values are presented in [Table 3.3.](#page-36-0) From the output curve it is possible distinguish the two regimes: triode region and saturation region. The condition with the solution with combustion (2:1) was also tested but the devices did not show good performance (see Annex 3A).

<span id="page-36-0"></span>*Table 3.3- Characteristic parameters of a TFT produced by ZTO precursor solution 0.1M concentration without combustion (2:1) and with combustion (1:1)*

<b>Conditions</b>	$V_{ON}(V)$	SS (V/dec)	$\mu$ SAT (CM <sup>2</sup> /V.S)	On/off Ratio
Without Combustion (2:1)	$-1 + 0.6$	$0.36 + 0.11$	$2.98 + 0.05$	$1.38 \times 10^{7}$
With Combustion (1:1)	$0.14 + 1.9$	$0.36 \pm 0.12$	$1.24 + 0.32$	7.58×10 <sup>6</sup>

These values prove that the TFTs have a good behavior, they have an on voltage near zero and a high on/off ratio. Ideally the SS values must be low, between 0.07 V and 0.1 V, for fast switching. These values are above the indicated but still low. When the SS values are too high it may indicate a charge leak. [18] The saturation mobility is also not to low which improves device behavior.

Is also noted that in devices made from the solution with combustion (1:1) there is a higher hysteresis and the devices made from solution without combustion (2:1) are more stable due to the fact that in the solution with combustion there is an excess of fuel due to the presence or urea, causing more film defects. These devices show equivalent performance to the current state-of-the-art of ZTO devices produced with 2-ME which is very positive, eliminating the toxic solvent, and remain the good results. [9]

To evaluate the stability of these devices they were tested three months later to see if their behavior was maintained and if they are stable over time.

This test was performed on the TFTs with the best behavior: solution without combustion (2:1), 0.1M concentration, 2 layers and with an annealing treatment of 30 minutes at 300 °C between layers. The used  $V_D$  was 20 V and the  $V_G$  was from -10 V to 20 V. [Figure 3.10](#page-36-2) shows the obtained results and [Table 3.4](#page-36-1) shows the characteristic parameters of these TFTs



<span id="page-36-2"></span>*Figure 3.10- IV transfer curves of TFTs produced by ZTO precursor solution without combustion (2:1), 0.1M concentration, on Si/SiO<sup>2</sup> wafer, after 3 months* 

<span id="page-36-1"></span>*Table 3.4- Characteristic parameters of a TFT produced by ZTO precursor solution 0.1M concentration, without combustion (2:1), after 3 months* 

<b>Conditions</b>	Von (V)	SS (V/dec)	$\mu$ sat (cm <sup>2</sup> /V.s)	On/off Ratio
Without combustion $(2:1)$ , 0.1M, 2 layers, $300^{\circ}$ C	$-2.5 + 1$	$2.52 + 0.54$	$3.0 + 0.1$	$1.19 \times 10'$

It is noticed that V<sub>ON</sub> suffered a slight shift to more negative potential values and the SS increases but these are minor differences because the mobility and the on/off ratio remain high which proves that these devices remain stable over time.

The annealing treatment of 30 minutes at 250 °C between layers was also tested, the results were not so good as can be seen in Annex 3B, showing a low mobility.

In the context of the SuperSmart project a Cl-free solution was tested for devices fabrication. In this solution tin chloride was replaced by tin acetylacetonate. This eliminates the chlorides from solution which makes it less toxic. The TFTs were produced using this solution, with 0.05M concentration, with 4 layers due their smaller concentration, with an annealing treatment of 30 minutes at 300 °C between layers. [Figure 3.11](#page-37-2) an[d Table 3.5](#page-37-1) show the obtained results.



<span id="page-37-2"></span>*Figure 3.11- IV transfer curves of TFTs produced with Cl-free ZTO precursor solution without combustion (a) (1:1) and (b) (2:1), 0.05M concentration, on Si/SiO<sup>2</sup> wafer.* 

<span id="page-37-1"></span>*Table 3.5- Characteristic parameters of TFTs produced with Cl-free ZTO precursor solution, 0.05M concentration, on silicon oxide*



Using a non-toxic solvent and a Cl-free ZTO precursor solution TFTs with good performances were successfully produced. With the (1:1) ratio the on voltage is very negative but it has a good on/off ratio and good mobility. With the (2:1) ratio has also a good performance and the on voltage is near zero so this is a good condition to make TFTs. However these devices have a difficult stabilization as can be seen in Annex 3C.

#### <span id="page-37-0"></span>3.3.2. Schottky diodes

.

Schottky diodes were produced on commercial ITO covered glass substrates. The devices that show best results were made from solution with combustion (1:1), 0.6M concentration, with 2 layers and with a annealing treatment of 150 °C + DUV during 30 minutes between layers. The other results can be observed in Annex 4A. The used  $V_G$  was from -0.8 V to 1 V. The top Ag contact were deposited by thermal evaporation via shadow mask. In this case Ag are used for the contacts due its work function that is higher than in the aluminium and lead to better diodes.

[Figure 3.12](#page-38-0) shows the obtained results.



<span id="page-38-0"></span>*Figure 3.12- IV curves of diodes produced by ZTO precursor solution with combustion (1:1), 0.6M concentration, on glass with ITO, with contacts deposited by thermal evaporation, (a) linear scale (inset photograph of the Schottky diodes produced) and (b) logarithmic scale*

The characteristic rectifying behavior of such a diode is verified, where to negatives voltages the passing current is too low and from a certain voltage (V<sub>ON</sub>) the current increase. This device presenting a V<sub>T</sub> of 0.7 V and a current on/off ratio of 800.

On the same substrate, Ag contacts were printed by inkjet determine the influence of the top contact deposition technique. The used  $V<sub>G</sub>$  was from -0.5 V to 0.5 V[. Figure 3.13](#page-38-1) shows the obtained results.



<span id="page-38-1"></span>*Figure 3.13- IV curves of diodes produced by ZTO precursor solution with combustion (1:1), 0.6M concentration, on glass with ITO, with contacts deposited by Inkjet. (a) Linear scale and (b) logarithmic scale.* 

A smaller voltage was applied on this diodes due to the contacts area which on these devices is smaller. The rectifying behavior of diode remains which demonstrates that the contacts can be deposited by Inkjet. However the obtained curves by inkjet are less smooth because of the interface between the contact and the semiconductor, when the contact is deposited by ebeam this interface have more quality due to this technique is performed in vaccum instead of inkjet technique that is performed in an environment that does not favor the quality of the interface. But anyway this device presenting a  $V<sub>T</sub>$  of 0.3 V and a current on/off ratio of 1000 which are interesting results that proves that the inkjet printing is a good alternative to the ebeam to deposit the contacts.

To verify if ZTO precursor solutions remains stable over time a two weeks aged (1:1) solution combustion was tested due the best results. 2 layers were deposited with a 150 °C + DUV annealing treatment for 30 minutes between layers. The used  $V<sub>G</sub>$  was from -0.5 V to 0.8 V. The Ag contacts were deposited by inkjet. [Figure 3.14](#page-39-0) show the obtained results.



<span id="page-39-0"></span>*Figure 3.14- IV curves of diodes produced by ZTO precursor solution with combustion (1:1), 0.6M concentration, on glass with ITO, with contacts deposited by Inkjet, 2 weeks later. (a) Linear scale and (b) logarithmic scale.* 

It is possible observe the behavior of the diode, however presenting a  $V_T$  of 0.36 and a current on/off ratio of 10.3. It can be noticed a clear decrease in its performance, the on/off ratio decreases two orders of magnitude. It can be conclude that the solution loses some quality over time. The 3 week test also was made, presenting a worst performance than 2 week. The results can be observed in Annex 4B.

Diodes were also produced using a non-toxic Cl-free ZTO precursor solution, with 0.6M concentration, with 2 layers. Two annealing treatments were tested: 30 minutes at 150 °C + DUV between layers and 30 minutes at 250 °C in HP between layers. The Ag contacts were deposited by thermal evaporation. Figure [3.15](#page-39-1) and Table 3.6 shows the obtained results. The used VG was from -1 V to 1 V or to 2 V.



<span id="page-39-1"></span>*Figure 3.15- IV curves of Schottky diodes produced by ZTO precursor solution, where tin chloride was replaced by tin acetylacetonate, 0.6M concentration, on glass with ITO, contacts deposited by Ebeam. (a.1) and (a.2) with an annealing treatment at 250 °C in HP in linear and logarithmic scale respectively and (b.1) and (b.2) With an annealing treatment at 150 °C + DUV in linear and logarithmic scale respectively.* 

<span id="page-40-2"></span>*Table 3.6- Characteristic parameters of a TFT produced by ZTO precursor solution, where tin chloride was replaced by tin acetylacetonate, 0.6M concentration, on glass with ITO* 



It can be conclude that both devices show a diode behavior. It can be compared the obtained results with the annealing in HP and with lower temperature plus DUV. It is verified that with HP annealing the results are better due their higher on/off ratio.

Using a non-toxic solvent and a Cl-free ZTO precursor solution Schottky diodes were successfully produced. It can be conclude that this solution can be a good alternative for these type of devices.

### <span id="page-40-0"></span>3.4. Flexographic Printing of ZTO thin films

With the objective of make Schottky diodes on flexible substrates another deposition technique was used: Flexography printing. For this it is necessary study the substrates that will be used and optimize this technique to later characterize the devices.

#### <span id="page-40-1"></span>3.4.1. Characterization of flexible substrates

#### Substrates Roughness

.

The substrates used to test flexographic printing of ZTO inks were Kapton 75 µm, Kapton 50 µm and PET with ITO with 200 µm of thickness. The surfaces of this substrates must be analyzed to check its roughness. The less rough the surface, better will be the interface between the substrate and the semiconductor and consequently the device performance. This analysis was made by AFM and the data were analyzed using Gwiddion program.

The first substrate used was the Kapton of 75  $\mu$ m where two sides of the sheet were analyzed, the inside and the outside to check which was better to deposit. The results are represented in [Table 3.7.](#page-40-3)



<span id="page-40-3"></span>*Table 3.7- Difference of roughness on the sides of kapton*

It turns out that the less rough surface is the outside surface, concluding that this is the best side to deposit the semiconductor.

Then the surface was studied before and after the annealing treatment of 1 hour at 150 °C on the outside of the kapton 75  $\mu$ m sheet. The results are represented i[n Table 3.8.](#page-41-0)



<span id="page-41-0"></span>*Table 3.8- Difference of roughness on kapton with and without annealing*

It can be noticed an increase of roughness after the annealing treatment but this is not significant. Then the three types of substrates were compared: Kapton 50 µm, kapton 75 µm and PET with ITO. The results are represented in graph of th[e Figure 3.16.](#page-41-1)



<span id="page-41-1"></span>*Figure 3.16- Variation of roughness on different used substrates on flexographic printing: Kapton 50µm, PET with ITO and Kapton 75µm* 

Comparing all the substrates can be conclude that the less rough were the Kapton 50  $\mu$ m, in the outside, and the PET with ITO so these are the ones that will be used to make the printing.

### Conductive layers Resistances of the substrates

.

A very important factor for the good behavior of a device is the resistance of the substrate. If this resistance is too high it will be difficult for the device to behave desirable.

The commercial substrates used in this work (PET with ITO, conductive paper, Kapton with IZO and glass with ITO) to produce ZTO diodes have a conductive layer so it was necessary to determine their resistance. The resistance was measured before and after the annealing treatment to determine its influence. The obtained results can be found in [Figure 3.17.](#page-42-2)



<span id="page-42-2"></span> *Figure 3.17- Resistances of substrates before and after annealing treatment of 30 minutes at 150 °C + DUV*

On the PET with ITO the resistance of the ITO is very high which can affect the performance of devices produced on this substrate, as the resistance is very high the voltage required to pass current has to be higher. It can also be noted that on PET with ITO the resistance variation is high across the substrate. The substrate that has the lowest resistance is the glass with ITO, however this is a rigid substrate, which cannot be used for flexographic printing.

The conductive paper substrate seems to be a good option due its low resistance.

It can be noted that the annealing treatment increases substrate resistance.

### <span id="page-42-0"></span>3.4.2. Optimization of flexographic printing parameters

To deposit the semiconductor layer by flexographic technique it was necessary to optimize several parameters to make prints with best quality, uniform and with good edge, like the printing pressure between the rolls, the surface plasma treatment effect, the type of substrate, the concentration of the ink, volume and number of layers.

#### <span id="page-42-1"></span>3.4.2.1. Printing pressure

.

Before printing it is necessary set print parameters. As mentioned earlier, in this technique, the ink deposited on the anilox is transferred to the mask and then to the substrate by contact forces. This force has to be controlled and optimized to get the best quality prints. The pressure between the anilox and the mask was optimized, gradually increasing the pressure until the ink passes to the substrate. The pressure between the mask and the substrate was varied in the opposite way, first using a higher pressure and then decreasing it. The prints done with different pressures were analyzed. These were made on Kapton 50 µm and with an annealing treatment of 10 minutes at 120 °C, the ZTO (1:1) ink used was with combustion and 1 M concentration and the printing speed was 50 m/min. The higher the pressure, the smaller the distance between the rollers will be, the following values presented in [Figure 3.18](#page-43-2) were distance values.



<span id="page-43-2"></span>It can be noticed that the best quality print, more uniform, it is the one made at a distance of 150. This was the used value from there to the Kapton. The same tests were made for ITO covered PET however, due its higher thickness, a distance of 170 was used.

These distance values depend on the flexographic equipment, so these are specific for this particular equipment.

#### <span id="page-43-0"></span>3.4.2.2. Surface plasma treatment

When printing it is very important that the solution to spread well on the substrate, forming uniform films. It is important that the substrates are hydrophilic to allow proper spreading of the solution. For this it is necessary to perform an  $O_2$  plasma pretreatment on the substrates surface to make these more hydrophilic. [32][33] Tests were made to check this effect, prints were made with and without  $O<sub>2</sub>$  plasma treatment to compare the prints quality. These tests were made on Kapton 75 µm with an annealing treatment of 10 minutes at 120°C, the used solution was with combustion (1:1), 1M concentration.



<span id="page-43-3"></span>*Figure 3.19- Effect of plasma treatment on prints. (a) Print without plasma treatment and (b) print with plasma treatment. (c) Kapton under the O<sup>2</sup> plasma treatment*

There is a clear difference between the images with [\(Figure 3.19](#page-43-3) (b)) and without (Figure 3.19 (a)) the plasma treatment. On the surface without treatment it appears that the film is not uniform, presenting failures, areas without ink, particularly in the peripheries but also in the middle. While the print on surface with treatment is very uniform, the solution spread evenly over the surface. It can be conclude that the plasma pretreatment is fundamental to obtain a quality print.

#### <span id="page-43-1"></span>3.4.2.3. Ink and substrate characteristics

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Several tests were made to know which the best conditions to achieve quality prints. Several ink concentrations were tested, different types of substrate, different number of layers and different volumes.

Analysis of a combination of the different conditions was made to find what would be the best choice. The prints were made at a speed of 50 m/min and the used ink was the solution with combustion (1:1). In [Table 3.9](#page-44-0) are represented the several prints on Kapton 50 um with different combinations of conditions. Two concentrations were tested: 0.6M with 2 layers and 1M with 1 layer. With 1 layer less material is spent so this option is preferable.

<span id="page-44-0"></span>*Table 3.9- Prints on kapton 50 µm*



The 1M with 1 layer condition leads to good quality prints, uniform and defined, both for a volume of 3 cm<sup>3</sup> and 4 cm<sup>3</sup>. However in the 0.6M with 2 layers condition it can be noticed a deviation between the first and the second layer, the prints did not overlap as they should. This effect can be due to two phenomena: there may have been bubbles when the substrate was placed and may have this effect; or it may be due to the high liquid-liquid surface tension that is causing the shift in the second layer. The print of a single layer is then preferred in this type of substrates.

The prints on PET substrates were tested at concentration of 1.5M, the results can be seen in Annex 5A. Then the same analysis was made but changing the substrate. Printing on PET with ITO was tested using the same conditions tested on Kapton adding the condition of 0.6M concentration with 1 layer [\(Table 3.10\)](#page-44-1).



<span id="page-44-1"></span>*Table 3.10- Prints on PET with ITO*

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On this substrate it can be noted again that the 2 layers condition do not work, excessive material accumulation occurs, especially with 4 cm<sup>3</sup>, and again a deviation of prints is noted. However, the condition of 1 layer has very good results both for 0.6M and 1M, the prints are uniform and well defined. For the 1M condition some material accumulation is noted in some areas (yellowish zones). In condition of 0.6M with 1 layer the prints are barely noticeable because of low thickness of the film due to the lower solution concentration. The 1.5M concentration was also tested but the results were not so good, as can be seen in Annex 5B, they are not uniform neither defined, presenting excessive ink accumulation in some places.

Another substrate that was analyzed was the Kapton 50 µm where a conductive IZO layer was deposited by rf sputtering. The used ink was 1M concentration with 1 layer [\(Table 3.11\)](#page-45-1).

<span id="page-45-1"></span>



These prints are uniform and have a good definition. It can be conclude that would be a good alternative to make the devices however to deposit the IZO layer a more costly and time consuming method is used and the goal is to use printing techniques.

To demonstrate the applicability of the flexographic printed ZTO ink a conductive paper substrate provided by Arjowiggins Creative Paper Ltd. was tested. Prints were made on this substrate with the solution with combustion (1:1) with 0.6M concentration and also the Cl-free solution with 0.6M concentration. However these prints, on this substrate, cannot be viewed under the optical microscope. It is possible observe it with the naked eye, but it is not possible determine the quality of print as can be seen in [Figure 3.20.](#page-45-2)





*Figure 3.20- Prints on paper with Ag nanowiresintegrated. (a) ZTO precursor solution 0.6M with combustion (1:1) and (b) ZTO precursor solution 0.6M (1:1) where tin chloride was replace by tin acetylacetonate.*

#### <span id="page-45-2"></span><span id="page-45-0"></span>3.4.3. Electrical characterization of printed diodes

<span id="page-45-3"></span>.

Two different techniques were used to deposit the silver electrodes: thermal evaporation and inkjet. Printed diodes were produced using PET with ITO substrates and an Ag top contact deposited by thermal evaporation and inkjet.

For contact deposition by electron beam assisted thermal evaporation it was necessary to design a specific mask to allow alignment of the contacts with the printed ZTO patterns [\(Figure 3.21\)](#page-45-3). Thermal evaporation technique leads to contacts with good definition.



*Figure 3.21- Used mask to deposit the contacts by electron beam assisted thermal evaporation.* 

These diodes were produced with the solution with combustion (1:1), with 1 layer and an annealing treatment of 30 minutes at 150°C + DUV. Two concentrations were tested: 0.6M and 1M. The used VG was from -1 V to 1 V. The obtained results can be found i[n Figure 3.22](#page-46-1) an[d Table 3.12.](#page-46-0)



<span id="page-46-1"></span>*Figure 3.22- IV curves of diodes produced by ZTO precursor solution (a.1) and (a.2) 0.6 M in linear and logarithmic scale respectively and (b.1) and (b.2) 1M in linear and logarithmic scale respectively, with combustion (1:1), on PET with ITO, with contacts deposited by Ebeam*

<span id="page-46-0"></span>*Table 3.12- Characteristic parameters of a Schottky diode produced by ZTO precursor solution, with combustion (1:1), on PET with ITO and contacts deposited by Ebeam.*

<b>Conditions</b>		On/off Ratio <sup>1</sup>
0.6M		26
1M	N 47	

Although there is a low on/off ratio in these devices with the 1M concentration and especially with 0.6 M, the results are positive, it is possible observe the behavior of the diode, printed by flexographic printing.

Diode were also produced with Ag contacts deposited by inkjet can be observed in [Figure 3.23.](#page-47-1) Here the alignment of the Ag contacts was done manually for each device.



*Figure 3.23- Alignment of contacts with the prints (contacts by Inkjet).* 

<span id="page-47-1"></span>These diodes were produced with the same ZTO solution as before; with combustion (1:1), with 1 layer and an annealing treatment of 30 minutes at 150°C + DUV. Two concentrations were tested: 0.6M and 1M. The used V<sub>G</sub> was from-0.2 V to 0.5 V. The obtained results can be found i[n Figure 3.24](#page-47-2) an[d Table 3.13.](#page-47-0)



<span id="page-47-2"></span>*Figure 3.24- IV curves of diodes produced by ZTO precursor solution (a.1) and (a.2) 0.6M in linear and logarithmic scale respectively and (b.1) and (b.2) 1M in linear and logarithmic scale respectively, with combustion (1:1), on PET with ITO, with contacts deposited by Inkjet* 

<span id="page-47-0"></span>*Table 3.13- Characteristic parameters of a Schottky diode produced by ZTO precursor solution, with combustion (1:1), on PET with ITO and contacts deposited by Inkjet.*

<b>Conditions</b>	$V_T(V)$	On/off Ratio
0.6M	0.15	วว ว
1M	ก วว	1000

The device made with the solution with 1M concentration present a good behavior, it has a higher on/off ratio and current is blocked when the applied voltage was negative, as intended. In this device is noted a high hysteresis. The device made with the solution with 0.6M concentration has poor performance. The 1 M concentration has better performance so to improve the hysteresis observed the annealing time was increased to 1 hour. The used V<sub>G</sub> was from -0.1 V to 0.5 V. [Figure 3.25](#page-48-1) an[d Table 3.14](#page-48-0) show the obtained results.



<span id="page-48-1"></span>*Figure 3.25- IV curves of diodes produced by ZTO precursor solution, 1M, with combustion (1:1), on PET with ITO, with contacts deposited by Inkjet. (a.1) and (a.2) annealing time of 1h in linear and logarithmic scale respectively and (b.1) and (b.2) annealing time of 1h30min in linear and logarithmic scale respectively.* 

<span id="page-48-0"></span>*Table 3.14- Characteristic parameters of a Schottky diode produced by ZTO precursor solution, with combustion (1:1), 1M, on PET with ITO and contacts deposited by Inkjet. Variation of annealing time.*



After this test it is noted that the hysteresis remained and did not improve the device behavior. For this reason it can be conclude that, at least for low voltages, the annealing time has no significant influence on the device behavior. The annealing time was also tested on prints, the results can be seen in Annex 5C.

Devices on Kapton with IZO were also tested. The same solutions was used: solution with combustion (1:1), 1M concentration with 1 layer, with an annealing treatment of 30 minutes at 150 °C + DUV with the contacts deposited by Inkjet. The used VG was from -0.2 V to 0.3 V. The obtained results can be found in [Figure 3.26](#page-49-1) and [Table 3.15.](#page-49-0)



<span id="page-49-1"></span>*Figure 3.26- IV curves of diodes produced by ZTO precursor solution 1M with combustion (1:1), on Kapton with IZO, with contacts deposited by Inkjet.* 

<span id="page-49-0"></span>*Table 3.15- Characteristic parameters of a Schottky diode produced by ZTO precursor solution, with combustion (1:1), 1M, on Kapton with IZO and contacts deposited by Inkjet.*



Observing the graphic it can be noted the diode behavior, however the on/off ratio is too low.

This type of substrates (PET with ITO and Kapton with IZO) have an irregular surface and high resistance which may influence the behavior of the devices.

Nevertheless, TFTs and Schottky diodes were successfully produced using environmentally friendly solution processed ZTO at low temperature. TFTs and Schottky diodes were also demonstrated using non-toxic Clfree ZTO solutions showing very promising results for the future.

The ZTO inks were printed on different substrates, including conductive paper, by flexographic printing and Schottky diodes could be produced with inkjet printed Ag electrodes, which very promising for printed oxide electronics applications.

Comparing with the literature it can be conclude that the results of TFTs solution based are equivalent to the published ones made from ZTO with ME solvent (Annex 6). [3][9]

The same is happen with the Schottky diodes produced, the results are similar to those presented in the literature, made from IGZO. [27][34]

The results achieved in this study are relevant in the state of the art. Is the first time that fully printed ZTO Schottky diodes using an eco-friendly solution with a low thermal budget are produced as shown in Figure [3.27.](#page-49-2)



<span id="page-49-2"></span>

*Figure 3.27- Schottky diodes on PET with ITO*

# 4. Conclusion and Future perspectives

<span id="page-50-0"></span>The main objectives of this work were the optimization of environmentally friendly ZTO precursor solutions to produce electronic devices like TFTs and Schottky diodes and optimization of flexographic printing of ZTO inks. These objective were achieved since successful production of these devices and flexoprinting of ZTO inks in different substrates was demonstrated.

For the TFTs fabrication the best condition is use the solution without combustion (2:1), with 0.1 M concentration, with 2 layers deposited by spin coating and with an annealing treatment of 30 minutes at 300 °C between layers. However with the solution with combustion (1:1), on the same conditions, the obtained results were also very good, so this can be also a good alternative for the TFTs.

To fabricate Schottky diodes on flexible substrates the best obtained results were with the solution with combustion (1:1), with 1 M concentration, with 1 layer deposited by flexography printing and with a pre annealing of 10 minutes at 120 °C and an annealing treatment of 30 minutes at 150 °C + DUV.

The best TFTs and Schottky diode fabricated during this study had very good results. TFTs were produced presenting a mobility of 2.98  $\pm$  0.05 cm/V.s, an opening voltage of -1  $\pm$  0.6 V, a subthreshold slope of 0.36  $\pm$  0.11 V/dec, an on/off ratio of 1.38  $\times$  10<sup>7</sup> and good stability over time.

The printed diodes presented also good performances, reaching an on/off ratio of 1000 and opening voltage of 0.32 V at low temperature.

It can be conclude that the contacts deposition by Inkjet printing is a good alternative to the thermal evaporation technique, Inkjet is a simpler and cheaper technique where is possible obtain good results.

The flexography printing technique was optimized. It can be concluded that this is a good option to deposit the semiconductor, it is an additive technique, which involve less material costs and allows easy deposition on flexible substrates, resulting in functional devices.

The roughness of the substrates greatly influence the obtained results. The obtained results for the Schottky diodes were not the best results, with low on/off ratios, due this problem but essentially due the resistance of the conductive layers of the substrates, namely on PET with ITO. These resistances were very high and vary along the surface.

However the fabricated devices showed a rectifier diode behavior that can be improve for further development of printed electronics.

This work was successfully done, was reached very good results with a non-toxic ink, what never been done before. This is an area under constant research where the objectives achieved in this work enable even better results in the future. This dissertation clearly shows that it is possible to produce 100 % eco-friendly inks for high performance devices at low temperature suitable for large area printing.

In the future it would be interesting to:

- Make devices using the conductive paper substrate due its low resistance.
- Reduce the process temperature in the fabrication of TFTs, using the same condition of diodes: 150 °C + DUV.
- Printing of TFTs by flexography to achieve fully printed ZTO devices
- Further explore the tin acetylacetonate solution, make devices from it on flexible substrates to achieve 100% eco-friendly ZTO devices.
- UV exposure after printing, to replace pre-annealing of 10 minutes at 120 °C and improve films quality.

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# 6. Annexes

### <span id="page-54-1"></span><span id="page-54-0"></span>Annex 1

In [Figure 6.1](#page-54-3) are represented the solution of tin chloride with urea and ammonium nitrate in MP that is not stable reacting and forming the hydroxides.



*Figure 6.1- Solution with tin chloride, urea and ammonium nitrate* 

### <span id="page-54-3"></span><span id="page-54-2"></span>Annex 2

.

Intensity (a.u.)  $(1)$  $(II)$  $(III)$ 20 30 40  $10$ 50 60 70  $2\theta$  (degrees)

A: Previous published results show that ZTO films that took a 300 °C annealing treatment are amorphous, as can be seen i[n Figure 6.2.](#page-54-4)

<span id="page-54-4"></span>*Figure 6.2- XRD spectrum of ZTO thin films that took a 300 °C annealing treatment*

B: On [Figure 6.3](#page-55-1) is represented a x-ray diffractogram of zinc nitrate and tin chloride reagents used to form the amorphous film of ZTO.



<span id="page-55-1"></span>*Figure 6.3- XRD of zin nitrate and tin chloride reagents* 

#### <span id="page-55-0"></span>Annex 3

.

 $\overline{a}$ 

A: [Figure 6.4](#page-55-2) represents the IV curve of a TFT made from ZTO precursor solution with combustion (2:1), where the device show a poor TFT behavior.



<span id="page-55-2"></span>*Figure 6.4- IV curve of a device made from ZTO precursor solution with combustion (2:1), 0.1M concentration* 

B: The TFTs were made at low temperature. It was used the same conditions, the solution without combustion (2:1), 0.1M concentration, but the annealing treatment was at 250 °C. [\(Figure 6.5.](#page-56-1))



<span id="page-56-1"></span>*Figure 6.5- IV curve of a device made from ZTO precursor solution without combustion (2:1), 0.1M concentration at 250 °C* 

C: It can be analyzed the stabilization of the TFTs behavior, made from the solution with tin acetylacetonate in [Figure 6.6.](#page-56-2)



<span id="page-56-2"></span>*Figure 6.6- Stabilization of the IV curves made from the solution with tin acetylacetonate (a) in (1:1) ratio and (b) (2:1) ratio.* 

It is possible to see that devices require some time until the curve stabilizes.

#### <span id="page-56-0"></span>Annex 4

.

A: [Figure 6.7](#page-57-0) represents the IV curve of a diode from ZTO solution without combustion (2:1), 0.6M, which did not show rectifying behavior.



<span id="page-57-0"></span>*Figure 6.7- IV curve of a diode from ZTO solution without combustion (2:1), 0.6M*

B: [Figure 6.8](#page-57-1) represents the IV curve of a diode from ZTO solution with combustion (1:1), 0.6M, after 3 weeks.



<span id="page-57-1"></span>*Figure 6.8- IV curve of a diode from ZTO solution with combustion (1:1), 0.6M, 3 weeks later*

### <span id="page-58-0"></span>Annex 5

.

A: [Figure 6.9](#page-58-1) represents flexoprinting of ZTO ink on PET, with the solution with combustion (1:1), 1.5M concentration. The films are very uniform and with very good definition.



*Figure 6.9- Prints on PET, with 1.5M concentration* 

<span id="page-58-1"></span>B: Flexoprinting of ZTO ink with 1.5 M concentration on PET with ITO but films did not show good uniformity, as showed in [Figure 6.10.](#page-58-2)



*Figure 6.10- Prints on PET with ITO, 1.5M concentration* 

<span id="page-58-2"></span>C: The annealing time was analyzed also on prints of ZTO ink, 1M with (1:1) proportion, as showed in Figure [6.11.](#page-58-3)



<span id="page-58-3"></span>*Figure 6.11- Variation of annealing time (a) 30 min, (b) 1h and (c) 1h30min*

### <span id="page-59-0"></span>Annex 6

.

In [Table 6.1](#page-59-1) are represented the results of TFTs made with 2-ME.

<span id="page-59-1"></span>*Table 6.1- Previous published results of TFT made with 2-ME.*

