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1 **Material flow for the intentional use of mercury in China**

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5 **Abstract**

6 Intentional use of mercury (Hg) is an important contributor to the release of Hg into the
7 environment. This study presents the first inventory of material flow for intentional use of Hg
8 in China. The total amount of Hg used in China increased from 803±95 tons in 2005 to its
9 peak level of 1272±110 tons in 2011. Vinyl chloride monomer (VCM) production is the
10 largest user of Hg, accounting for over 60% of the total demand. As regulations on Hg content
11 in products are tightening globally against the background of the Minamata Convention, the
12 total demand will decrease. Medical devices will likely still use a significant amount of Hg
13 and become the second largest user of Hg if no proactive measures are taken. Significant
14 knowledge gaps exist in China for catalyst recycling sector. Although more than half of the
15 Hg used is recycled, this sector has not drawn enough attention. There are also more than 200
16 tons of Hg that had unknown fates in 2011; very little information exists related to this issue.
17 Among the final environmental fates, landfill is the largest receiver of Hg, followed by air,
18 water and soil.

19 **Keywords:** material flow, Hg release, mass balance, environmental fate

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20 **1. Introduction**

21 Mercury (Hg) is a toxic heavy metal and is particularly hazardous to fetus development and
22 IQ development in infants ¹. Hg exposure can cause health problems not only at acute
23 exposure to high concentrations, as in Minamata disease ², but also at chronic low doses ³. Hg
24 and its compounds have a long history of human use. Hg is the only metal that is liquid at
25 room temperature (20°C); it is also a good electrical conductor and is highly resistant to
26 corrosion. Hg is used in many modern industrial applications, including electrical switches,
27 thermometers, dental amalgams, lighting (Hg vapor in fluorescent lamps), flow meters,
28 batteries, catalysis, explosives, and gold recovery.

29 Due to the toxicity and widespread use of Hg in various industries, many efforts have been
30 made to find alternatives to Hg, and regulations are also tightening gradually to reduce the
31 reliance on Hg. For example, European Union issued an Hg export ban in 2008 which was
32 implemented in 2011 ⁴. United States issued a similar ban on Hg export in 2008 which was
33 implemented in 2013 ⁵. The international community has also worked intensively on
34 achieving a global legal binding instrument on Hg. Finally, on 19 January 2013, the world
35 governments agreed to the text of a global legally binding instrument on Hg, which resulted in
36 the “Minamata Convention on Mercury” ⁶. All of these activities send out a clear signal that
37 Hg, as part of a product or unintentional emission, will be reduced and strictly regulated.
38 China is an important participant in the convention and is also the largest Hg emitter and
39 industrial user ⁷. However, until now, there has been no systematic study of Hg material flow
40 in the Chinese industries. Following the implementation of the “Minamata Convention on
41 Mercury”, China is required to implement tightening regulations and administration of
42 intentional Hg usage. There is a great regulatory and administrative demand for an estimate of
43 Hg material flow in the most important industries involving the intentional use of Hg, either
44 as catalyst or as part of a product.

45 The pollution from Hg usually comes from both the intentional use of Hg and the
46 unintentional release of Hg because Hg usually co-exists in raw materials, such as coal and
47 ores. There have been many studies on estimating the Hg emission from major industries
48 (mainly unintentional emission, such as coal combustion and non-ferrous metal smelting)⁸⁻¹².
49 All of these studies focus mainly on unintentional Hg emission to air. However, Horowitz¹³
50 reported that the intentional use of Hg in products and processes (“commercial Hg”) has
51 produced a large and previously unquantified anthropogenic source of Hg in the global
52 environment during the industrial era, with major implications for Hg accumulation in
53 environmental reservoirs¹³. China is the largest user of Hg in the world, China is the only
54 country in the world that uses coal as a raw material to produce polyvinyl chloride monomer
55 (VCM); a Hg-containing catalyst is required in the process. However, there have never been
56 studies to quantify Hg material flow in the Chinese society using the life cycle approach. The
57 current estimate of the emission of Hg in China, therefore, cannot meet the administrative
58 purposes because intentional use of Hg in industries is the most important driver for mining
59 Hg mineral and illegal recycling of Hg-containing waste. It is important to understand the
60 importance and level of uncertainties of the intentional use of Hg for a good management of
61 Hg flow. The goal of this study is to identify the knowledge gaps of the current Hg flow and
62 guide the administration of Hg flow in a product’s whole lifecycle. This study will therefore
63 focus on Hg fate in the lifecycle of product (or catalysts).

64 This study is based primarily on data contained in publications, statistics yearbook, inquiry
65 with relevant personnel and others. This study is not intended to be a detailed investigation of
66 any specific source of Hg discharge into the environment; rather, this study examines the
67 contribution of each major source to total domestic Hg flow. This study is aimed at
68 identifying the largest uncertainties and supplying tools for decision making.

69 **2. Methods**

70 UNEP supplies a toolkit to quantify the Hg release into environment, but this toolkit uses
71 simple one-step emission factor which doesn't consider the Hg flow pathways in a country ¹⁴.
72 In order to quantify the material flow, a mass balance-based material flow model which
73 describes the flow pathways in details need to be developed. This method was first used by
74 the Swedish Chemicals Agency (KEMI) to estimate the Hg release from Hg related products,
75 such as batteries and fluorescent lamps ¹⁵. Horowitz used a two tier approach to estimate the
76 global Hg flow by using a similar method ¹³. This study adapts the same method but with
77 modifications to suit the Chinese situation. The model mainly consists of the following two
78 equations:

79 The total Hg usage (F) in a specific sector i is

80
$$F_i = P_i \times C_i \text{ (eq. 1)}$$

81 Where P is the total production volume per year;

82 C is the Hg content (or usage) per unit produced.

83 The main industrial sectors that intentionally use Hg in China include medical devices,
84 fluorescent lamps, batteries, dental amalgam and vinyl chloride monomers (VCMs) (Fig. 1,
85 Table S1).

86 The total mass of Hg (M) that ends up in final fate compartment k is

87
$$M_k = \sum_{i=1}^n (F_i \times \sum_{j=1}^n (D_{ij} \times E_{ijk})) \text{ (eq. 2)}$$

88 Where M_k calculates how much of Hg that is used in a certain year will finally end up in the
89 fate compartment k through its lifetime;

90 F_i is the estimate of Hg use by sector i ;

91 D_{ij} is the distribution factor for path j of sector i ; the main distribution pathways
92 include landfills, waste incineration, breakage during usage, recycling and unknown.

93 E_{ijk} is the emission factors to fate k from path j of sector i ; the major emission
94 endpoints are air, soil, water and landfill, with reclaimed by recycling being another important
95 fate for used catalyst in the VCM industry. The final environmental fates of Hg in Tier 2 are
96 just the primary distributions after treatment in Tier 1; reemission and release through time
97 afterwards are not counted in this study, for example the reemission of Hg from landfill to air
98 is not counted in this study.

99 The data used in the study are primarily from journal publications, industrial statistics
100 yearbooks and interviews with industrial personnel and administration authorities. Detailed
101 data are usually difficult to obtain in China; where data were unavailable because they were
102 either not collected or not reported, certain calculations were made based on estimations and
103 assumptions. Table 1 summarizes the main data sources. The distribution in Hg-containing
104 products are estimated and then combined with release factors to different environment
105 compartments. The model is extended to be able to set different factors on multiple years
106 from 2005 to 2020. Tables S2-S8 in the Supporting Information (SI) describe in detail how
107 the different factors and values are decided and estimated.

108 Table 1. Data sources for product production sources

Data sources	
Medical devices	Chinese Medical Statistical Yearbook
Lightning devices	Annual market analysis report published on China Light & Lighting by the Chinese Association for Lighting Devices
Batteries	Annual market review report published by the China Battery

Enterprise Alliance

VCM China Plastic Industry Statistics Yearbook, the percentage of carbide-based process information is supplied by the China Chemical Industry Environmental Protection Association (CCIEPA)

Dental amalgam Special Policy Study on Hg Management in China

109

110 Due to fact that most of the data used in the estimate are not accurate, they may come from
111 different sources and may just be a range of possible distribution. Some of the values in this
112 study are therefore given a possible range when data exists. Oracle Crystal Ball[®] ¹⁶ is used to
113 estimate the uncertainties of the estimate by using Monte Carlo simulation ¹⁷, the reported
114 data in this study is indicated with plus-minus sign to show the range with 95% certainty level.

115 **3. Results and discussion**

116 ***3.1 Hg demand development and forecast***

117 Fig. 2a shows the estimated annual Hg demand in China and how the demand is supplied. The
118 largest users of Hg in 2005 were the VCM industry (378±40 tons), thermometers (143±79
119 tons), button sized batteries (132±26 tons), Zn-Mn batteries (50±10 tons),
120 sphygmomanometers (48±15 tons), tubular or ring fluorescent lamps (28±3 tons) and compact
121 fluorescent lamps (CFLs) (14±2 tons). The total demand in 2005 was approximately 803±95
122 tons. The demand for Hg in China reached its peak in 2011 (1272±110 tons). The largest user
123 is still the VCM industry (878± 61 tons), followed by thermometers in second place (152±83
124 tons) and sphygmomanometers in third place (97±10 tons). The total demand then started to
125 decline from 2011 to the current level of 903±115 tons in 2014. This decline is largely driven
126 by replacing high Hg-containing catalyst by low Hg-containing catalyst and international
127 pressure on reducing the Hg content in products. The biggest reduction rates of Hg use in

128 products are from lamps and batteries though biggest reduction amount is from VCM industry.
129 Both the EU and the United States, as the largest export market for Chinese products, started
130 to regulate Hg in products many years ago ^{18, 19}. Furthermore the United Nations Environment
131 Program (UNEP) decided to develop a global legally binding instrument on Hg in 2009 ⁶.
132 Because China is absolutely the largest producer and exporter of both fluorescent lamps and
133 batteries, as the international standards tightens, the Chinese producers also must improve
134 their techniques to continue to supply products to these markets. For example, China reduced
135 the Hg use in compact fluorescent lamp from 4.5 mg Hg per lamp in 2011 to 2.5 mg Hg per
136 lamp ²⁰.

137 At the individual level, the different categories of products followed different trends through
138 time and also have different perspectives in the coming future.

139 *Hg demand in products*

140 It is shown in Fig. 2a that the battery industry is the first industry that started to rapidly reduce
141 its Hg use. In 2009, Chinese authorities issued two new product standards ^{21, 22} for both Zn-
142 Mn batteries and button cells, aiming at reducing hazardous metals in battery products. The
143 Hg content in batteries was subsequently reduced significantly because battery producers
144 were forced to switch to alkaline Zn-Mn batteries, which require much less Hg than
145 conventional Zn-Mn batteries (Table S3). The international markets have also substantially
146 tightened the restrictions, e.g., the EU has prohibited all batteries or accumulators that contain
147 more than 0.0005% of Hg by weight ²³ since 2006. The Chinese battery industry is heavily
148 export orientated and has actually already been producing products suitable for international
149 markets. Given the condition that both the technology and the capacity already exist in China,
150 the Hg demand by battery industry will very likely continue to decrease.

151 The use of Hg in the fluorescent lamp industry also declined, although not as rapidly as in the
152 battery industry. The total demand increased slowly from 2005 (47 ± 7 tons for all types lamps)
153 until 2007 (75 ± 10 tons) and then decreased to the current level (32 ± 4 tons) in 2014. However,
154 considering the total production volume of lamps almost doubled during this period, the
155 efforts made by the lamp industry are also significant. In particular, the demand for CFLs is
156 increasing very rapidly: the production volume of CFLs doubled between 2005 and 2008 and
157 then increased another fold between 2008 and 2014. In 2012, the Chinese authorities took a
158 further step to set a timetable for reducing the use of Hg in the fluorescent lamp industry by
159 issuing a ‘Roadmap to gradually reduce the Hg content in fluorescent lamps’²⁰. The light
160 emitting diode (LED), as new lighting device, has exhibited rapid expansion of its market
161 share in the lighting market in the past decade²⁴; it is predicted that LED light bulbs will
162 account for 84% of the market share by 2030²⁵. Considering the lowering of the Hg content
163 in lamps and the replacement of fluorescent lamps with LED lights, the Hg demand in
164 fluorescent lamps will very likely continue to decrease.

165 The Hg demand for medical devices has been continuously increasing over time, from
166 191 ± 94 tons in 2005 to 249 ± 93 tons in 2011, largely driven by the strong domestic demand
167 for medical devices since the new millennium when China started its health care insurance
168 reform²⁶. Although there are alternative techniques to replace Hg-containing medical devices
169 with electronic/mechanical devices²⁷, unlike the other two products mentioned above,
170 alternative devices cost much more, e.g., replacements for Hg-filled thermometers of
171 comparable accuracy usually cost 10 times the price of Hg-containing devices. The export of
172 Hg-containing medical devices to the developing world ($\sim 50\%$ for thermometers and $\sim 30\%$
173 for sphygmomanometers as of 2011) has been quite consistent due to their low cost. The
174 World Health Organization (WHO) initiated a campaign called Hg-Free Healthcare, which
175 aims to phase out Hg thermometers and blood pressure measuring devices by 2020²⁸. China

176 has not yet set its agenda to replace Hg-containing medical devices. However, to achieve this
177 goal, there must be strong political will from the government and proper subsidy plans to
178 promote the use of Hg-free medical devices.

179 Hg demand in dental amalgams has been quite stable, ~ 6 tons per year ²⁹. The main fates of
180 the Hg in dental amalgams are wastewater, landfill (burial) and cremation. There have been
181 advances in finding various resin-based alternative dental fillings materials ^{30, 31} for decades,
182 although such materials are not problem free either; the Hg dental amalgam remains quite
183 ubiquitous in China. In addition, there is no national plan to phase out dental amalgam fillings
184 so far. We therefore predict that the Hg demand from dental amalgam fillings will be
185 maintained at the current level until further action from authorities is implemented.

186 *Hg demand in industrial processes*

187 The largest user of Hg in China consistently has been the VCM industry. In China, more than
188 half of all the VCM production capacity is based on the carbide process. VCMs are
189 synthesized from acetylene and hydrochloric acid; acetylene comes from calcium carbide,
190 which is synthesized from limestone and coke at a temperature exceeding 2000°C ³². This
191 production process requires HgCl₂ carried by activated carbon as a catalyst; the HgCl₂ content
192 varies from 4.5% to 12% (weight) (Table S3). Most of the carbide process-based VCM
193 manufacturing plants are located close to the coal producing regions in inland China; in
194 contrast, most of the oil-based VCM productions are located along the coastal provinces,
195 where imported oil arrives at ports. As the oil prices continued to increase in the past decades,
196 the proportion of VCMs produced by the carbide process out of the total VCM outputs
197 increased from ~ 50% in 2005 to ~ 75% in 2011 and then maintained its level until recently
198 (Table S2). The total production volume of VCM from carbide processes almost tripled
199 during 2005 and 2014. The annual Hg used by VCM production was 378±40 tons in 2005 and

200 reached its peak level of 878 ± 61 tons in 2011; thereafter, the production started to decrease
201 due to the introduction of low Hg-containing catalysts (Fig. 2a). China has a plan to entirely
202 replace the high Hg-containing catalyst with a low Hg-containing catalyst by 2015³³. Despite
203 some delay, it is highly possible to achieve the replacement quite soon, according to oral
204 communications with the CCIEPA. The performance of the low Hg-containing catalysts has
205 matched that of the high Hg-containing catalysts. Invention of new Hg-free catalysts is also
206 being promoted. Several Chinese researchers have succeeded in laboratory tests of a gold-
207 based catalyst for acetylene hydrochlorination^{34, 35}. However, it will take some time before
208 the Hg-free catalyst can be applied in real industrial production. The Hg uses in the VCM
209 industry are therefore going to continue for at least one decade. Much of the used catalyst will
210 end up in recycling; if a high percentage of recovery can be achieved, then the release of Hg
211 to the environment can be limited. This recycling issue will be explained in detail in the next
212 section.

213 *Uncertainty of the estimate*

214 The Hg content in each product is actually not a certain value, as different data sources may
215 provide very different values; we therefore used a range of Hg contents in the product and
216 estimated the uncertainty of our estimate. It is shown that the Hg content in
217 sphygmomanometers and the Hg content in catalysts (Fig. S1) are the largest contributors to
218 the uncertainty, both contributed more 30% of the total uncertainty. The Hg content in lamps
219 is of less importance. It is therefore important in the future to perform a more thorough
220 investigation of the Hg content in medical devices and catalysts.

221 *3.2 Hg supply estimate and forecast*

222 There are two major supply sources for Hg in China: primary mining and recycling (Fig. 2b).
223 China used to import Hg before 2005, but since 2005, no official import and export of Hg has

224 been approved, according to Chemical Registration Center of Ministry of Environmental
225 Protection of China.

226 Primary Hg mining has been the dominant Hg supplier in China until recently. The main
227 mining areas in China are concentrated in Guizhou and Shaanxi Provinces ^{36, 37} where there
228 have also been reports of artisanal Hg smelting activities ^{38, 39}. According to the local
229 authorities, small scale new mines are recently opening again in Guizhou; however, its real
230 scale and activity levels are not clear. The supply of Hg from primary mining stabilized
231 between 600-800 tons per year and started to drop in recent years. The mineral resources have
232 largely been depleted in almost all of the Hg mines in China; as a result, the output from
233 primary mining will decrease rapidly. Currently, there is only one Sb-Hg mine located in
234 Xunyang, Shaanxi, that still produces Hg at a relatively large scale.

235 There are different sources of Hg output data available in China (Table S5). The China
236 Nonferrous Metals Monthly (CNMM) provides monthly statistics of non-ferrous metal output
237 volumes in China, including Hg. The data indicated substantial inconsistencies in the output
238 volume: the total output jumped suddenly from 225 tons in 2008 to 1231 tons in 2009. It is
239 highly possible that the CNMM adjusted its statistics method and included new sources of
240 information from 2009. The China Nonferrous Metals Industry Association (CNIA) also has
241 another set of statistics showing a more consistent development. The total demand estimated
242 by this study showed lower values than the data of both CNMM and CNIA. There are two
243 major reasons for this observation. One reason is that both CNMM and CNIA used a bottom-
244 up method. They acquired information from each individual enterprise that uses Hg by
245 questionnaires. The main data reported to them are trading data, which has a potential risk of
246 double counting some of the Hg that are bought today and may be again sold to others some
247 time later. In contrast, our study used a top-down method by calculating the demand based on
248 how much product they produce. Thus, the results in our study are actually the real demand

249 instead of the real production. Another reason is that some Hg may be lost during the
250 transportation, storage and chemical production processes; this could also explain part of the
251 difference between our study and statistics from CNMM and CNIA.

252 Reclaimed Hg from used catalyst is another important supplier of Hg in China. The recycling
253 raw materials are mainly two types: exhausted catalyst, which contains ~ 5% of Hg, and
254 activated carbon for eliminating Hg from exhausted gas, for which the Hg contents varies
255 considerably. One mass balance study showed that approximately 75% of Hg will end up in
256 either exhausted catalyst or activated carbon ⁴⁰ and will further go to recycling. According to
257 oral communication with relevant personnel in the industry, the recycling rate is estimated to
258 be approximately 80-90% (Table S8). Another important fate of Hg in large VCM plants are
259 HCl acid; HCl vapor in exhausted gas is usually collected by an acid plant to make HCl acid,
260 which is sold on the market as a product ⁴⁰. No information is currently available on this part
261 of Hg in the HCl acid. There is also a small volume of Hg that comes from Zn smelters with
262 Hg reclamation towers ⁴¹; the reclaimed Hg can also be sold in the market.

263 *3.3 Hg emission and release*

264 Products will be distributed to different routes after they are worn out, as described by the
265 distribution factor in tier 1 (Fig. 1). Hg will then be emitted, released, or disposed, as
266 described by the emission factor in tier 2 (Fig. 1). Emission factors for each process area are
267 defined to determine exactly how much Hg will end up in which environmental compartment.

268 Currently in China, there is no centralized sorting and recycling system for municipal solid
269 waste ⁴². The used Hg containing products are therefore not sorted, they are usually disposed
270 together with all other municipal solid wastes. The main disposal methods of products are
271 either landfills or incinerators. Since 2000, China has been progressively increasing the
272 percentage of wastes that are incinerated to cope with the lacking of landfill capacity,

273 especially in eastern China where the population density is high, and land resources are quite
274 scarce ⁴³. China is planning to incinerate 50% of all its solid waste by 2020 compared with
275 only about 20% in 2010, but considering the decrease of the Hg content in most products due
276 to the demand from Minamata Convention, the total emission from incineration will still
277 decrease (Fig. 3). Worn-out medical devices as medical wastes are usually generated in
278 hospitals, and the Chinese hospitals have been encouraged to collect and recycle some worn-
279 out devices, especially sphygmomanometers (Table S7).

280 Due to the dominance of the VCM industry in Hg consumption, reclaimed Hg is the most
281 common fate of Hg (Fig. S2). In addition, due to the increasing percentage of wastes that are
282 incinerated, the landfills, which was the second most common fate of Hg in 2005, has become
283 the third in 2011, after unknown fate related to Hg in HCl in VCM industry. Hg sold in HCl
284 acid is one of the major knowledge gaps of Hg fate. Landfills represent an important fate for
285 Hg in products; the total amount of Hg that ends up in landfills has been quite stable (> 100
286 tons per year). However, considering the increasing use of incinerators, increasing amounts of
287 Hg will probably end up in the air depending on the atmospheric emission control used in the
288 incinerators. About 53 tons of Hg that are used in the year of 2010 is estimated to be emitted
289 to air ultimately, the amount is similar to the scale of contribution from zinc smelting (63 tons
290 in 2010) which is about 10% of estimated total unintentional emission of Hg to air in 2010
291 (538 tons) ⁴⁴. Hg emission from intentional use of Hg is therefore also an important
292 contributor to total atmospheric Hg emission.

293 Fig. 3 shows the development of Hg release into the environment in three different years:
294 2005, 2011 and 2020. By 2020, most of the products will not generate an excessive release of
295 Hg, except for VCM related activities. The recycling sector will become the largest and most
296 significant release source of Hg in 2020. This prediction is largely because it is difficult to
297 phase-out Hg containing in the VCM industry catalysts in the short term. Currently, very little

298 is known about the Hg recycling sector in China, and this sector has not drawn any significant
299 attention in China either.

300 The uncertainties of release to the environment are quite high because few onsite
301 measurements are conducted for the different types of treatments. Different products also
302 show explicitly different release patterns. The largest contributor to air emission is from the
303 recycling industry, dental amalgams and thermometers. Hg release to soil mainly comes from
304 broken thermometers and florescent lamps. The Hg discharge to water mainly comes from the
305 recycling industry and dental amalgams, with most of the discharge occurring through
306 discharge of waste water. Landfills, as shown in Fig. 3, are consistently the most significant
307 receiver of Hg. The main reason for this observation is as mentioned in the above section:
308 most of the solid waste in China ends up in landfills, and landfills also receive the sludge from
309 wastewater treatments and solid residues (as hazardous waste) from incinerators, both of
310 which contain significant amounts of Hg. The re-emission of Hg from landfills has been well
311 studied, and many research studies have proven that landfills represent an important source of
312 secondary Hg emission^{45, 46}.

313 ***3.4 Implications***

314 The material flow for intentional use of Hg in China is presented in Fig. 4. This study
315 identified several important knowledge gaps that may have important implications for China
316 Hg management and policy making.

317 Recycling of Hg from waste catalyst in the VCM industry is the most important fate of Hg in
318 China. The recycling sector is also the most important contributor of Hg emissions to air and
319 release to water. With the information we have so far, the recycling sector seems to use quite
320 a simple technique, and the potential release of Hg from the recycling sector is high. When
321 the other sectors are actively reducing Hg, this recycling sector may become the most

322 significant contributor of Hg release into environment in the near future. However, this sector
323 has not raised adequate attention in China. The recycling sector is not on the national agenda
324 on reducing Hg pollution, and this could be an overlooked problem.

325 Hg with unknown fate in China is surprisingly high. The most important explanation for this
326 observation is the HCl acid produced by acid plants in VCM production. Carbide process-
327 based VCM production, as a unique Chinese situation, has not been well studied. Little is
328 known about the fate of acid. It is therefore important to establish a system that can trace the
329 flows of Hg in secondary products in China.

330 Primary Hg mining in China is in general declining due to the depletion of Hg minerals in
331 China. Recent development in this sector is that many mines start to produce Hg as by-
332 product of other metals. For example, the most common Hg-associated ores are antimony and
333 zinc. Many companies are therefore registered under the name of other metals and produce
334 Hg as the price of Hg becomes higher. Many of these new producers may be overlooked when
335 conducting the national primary Hg mining survey. The actual output may be higher than the
336 survey data.

337 Landfills are the most important receivers of Hg in the environment, as is well known by both
338 researchers and authorities. It is not very easy to control the secondary emission of Hg from
339 landfill to air. Currently in China, the system of waste sorting, collection and recycling is still
340 in its infancy. Hopefully, China will implement stricter regulations on the waste sector and
341 enable Hg-containing products to be sorted, treated as hazardous waste and stabilized before
342 being sent to landfills. Such regulations could ensure that less Hg ends up in municipal solid
343 waste landfills.

344 The predicted future export of Hg in products will be mainly through medical devices. As
345 noted in this study, to promote the production of Hg-free medical devices, there must be

346 strong political will from the government to prohibit the Hg-containing devices along with
347 proper subsidies as incentives for people to change their habits. Take the solar panels and
348 LED lamps for example, many countries have been quite successful in using subsidies to
349 proactively promote such products. Such an approach can be applied for medical devices.

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470 List of figures

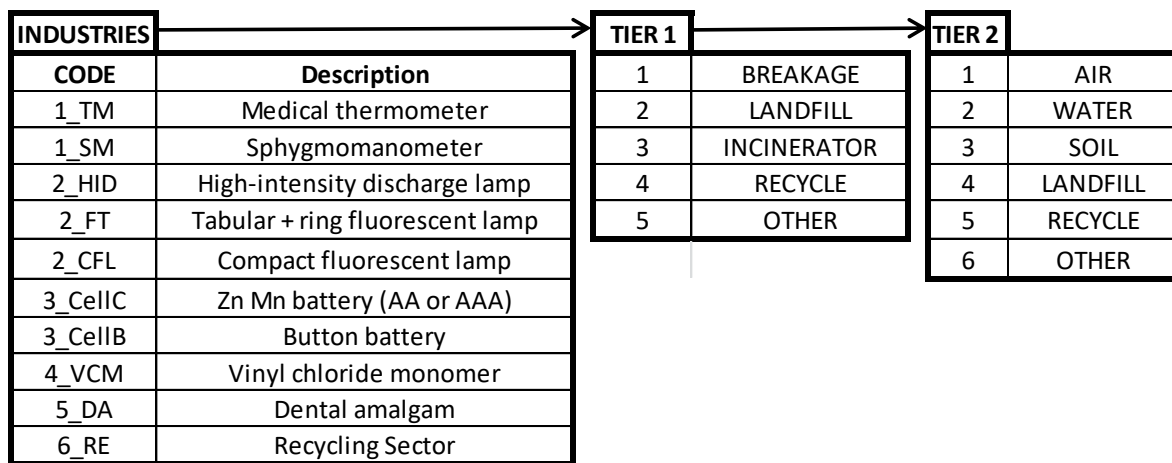
471 Figure 1. Two-tier Hg material flow pathways

472 Figure 2. Estimated annual Hg demand (a) and supply (b) in China

473 Figure 3. Hg release into the environment in China

474 Figure 4. Hg material flow in China (2011 as an example)

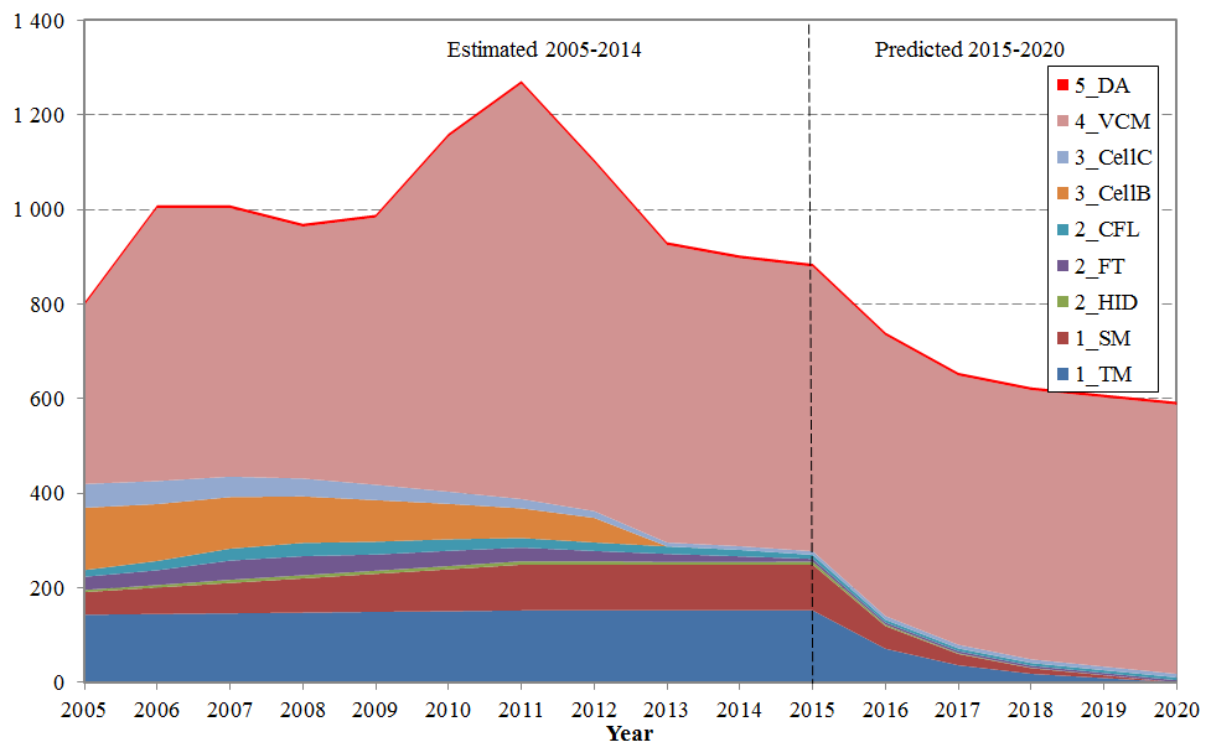
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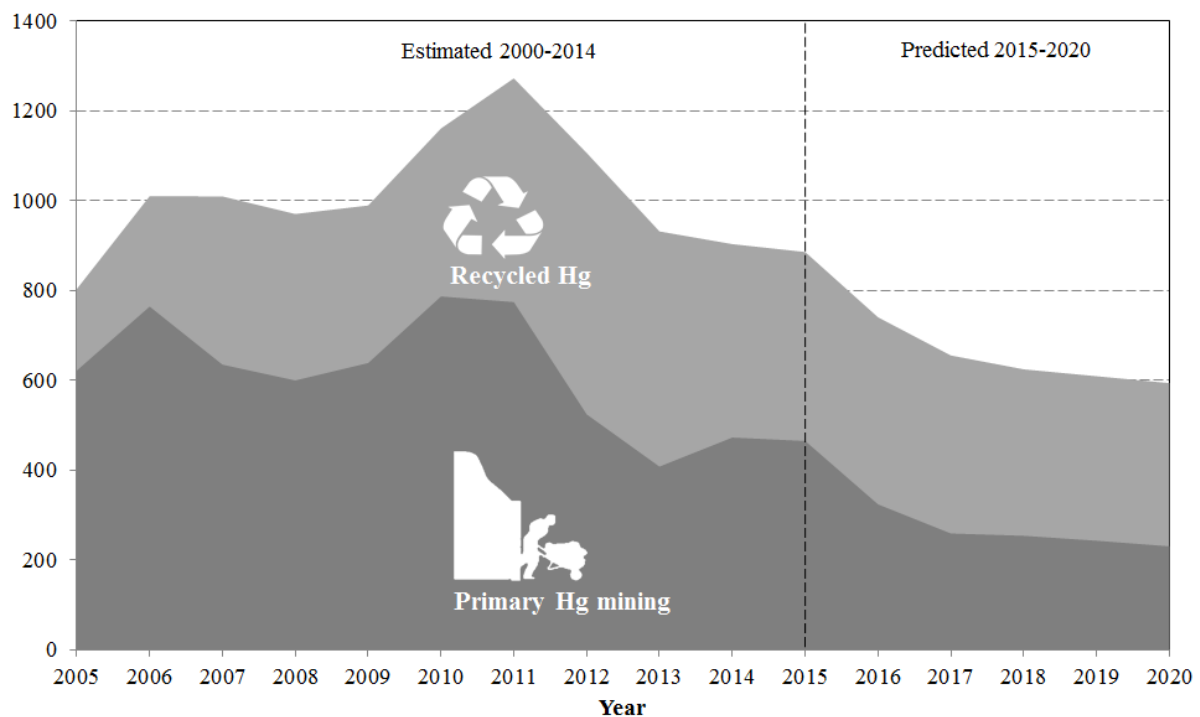
476
477 Figure 1. Two-tier Hg material flow pathways

478

a, annual Hg demand (tons)



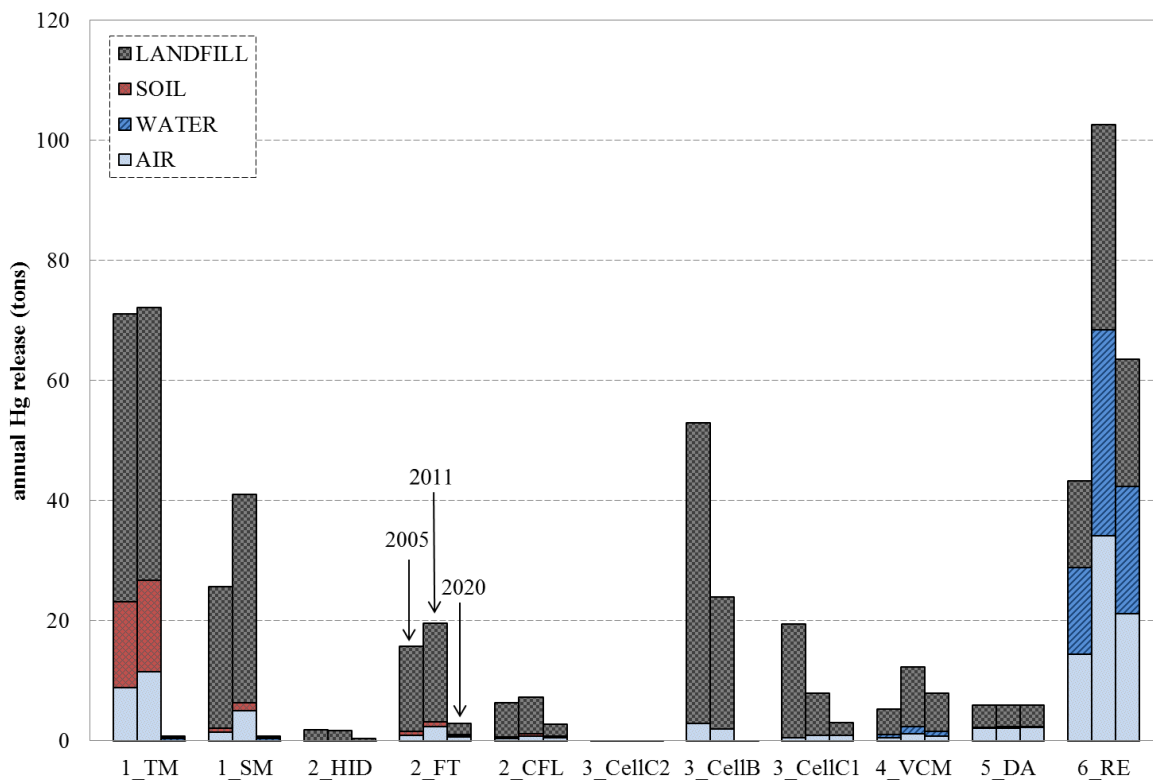
b, annual Hg supply in China (tons)



479
 480 Figure 2. Estimated annual Hg demand (a) and supply (b) in China (1_TM: medical
 481 thermometer, 1_SM: sphygmomanometer, 2_HID: high-intensity discharge lamp, 2_FT:
 482 tubular + ring fluorescent lamp, 2_CFL: compact fluorescent lamp, 3_CellC: Zn-Mn battery

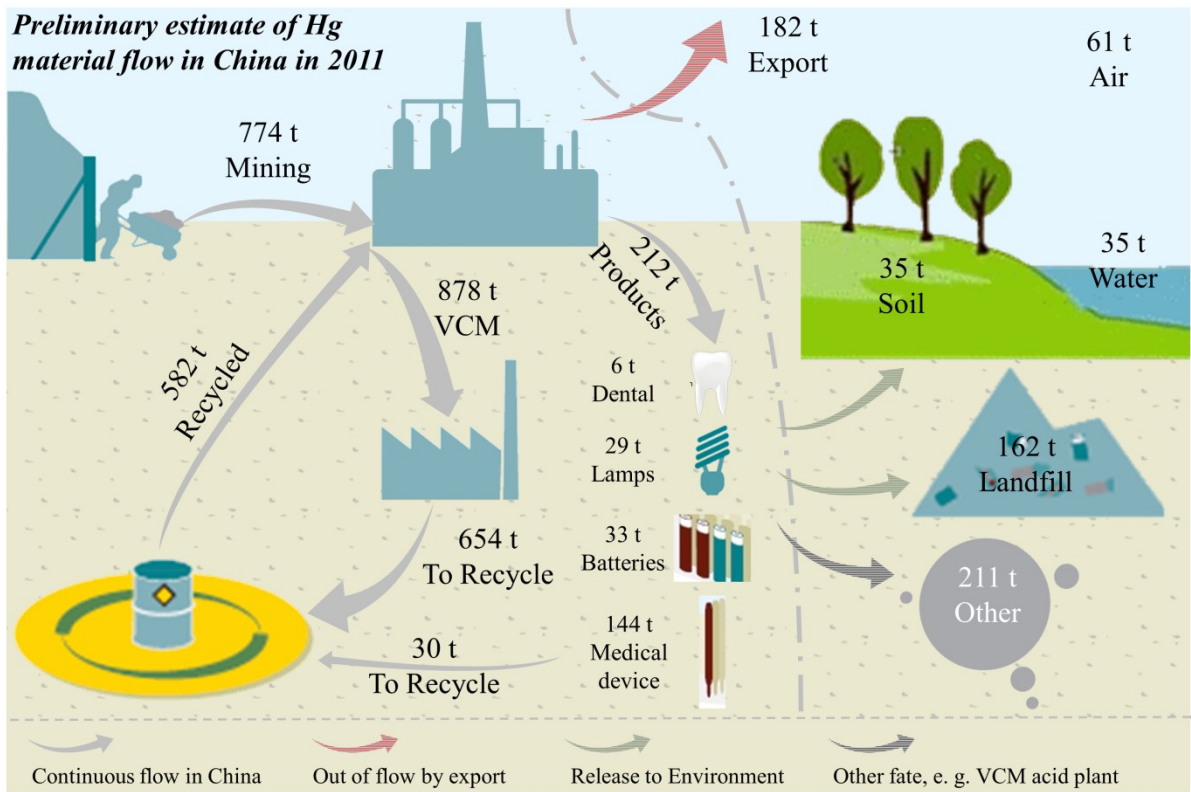
483 (AA or AAA size), 3_CellB: Button cells, 4_VCM: catalyst used VCM production, 5_DA:
 484 dental amalgam)

485



486
 487 Figure 3. Hg release into the environment in China (1_TM: medical thermometer, 1_SM:
 488 sphygmomanometer, 2_HID: high-intensity discharge lamp, 2_FT: tubular + ring fluorescent
 489 lamp, 2_CFL: compact fluorescent lamp, 3_CellC: Zn-Mn battery (AA or AAA size),
 490 3_CellB: Button cells, 4_VCM: catalyst used VCM production, 5_DA: dental amalgam,
 491 6_RE: Hg recycling) (Note: the years 2005, 2011 and 2020 are the years when the product is
 492 produced; however, due to the lifespan of the products being different, Hg in different
 493 products will enter the environment in different years).

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Figure 4. Hg material flow in China (2011 as an example)