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1 **Air-soil Diffusive Exchange of PAHs in an Urban Park of**
2 **Shanghai Based on Polyethylene Passive Sampling:**
3 **Vertical Distribution, Vegetation Influence and Diffusive**
4 **Flux**

5
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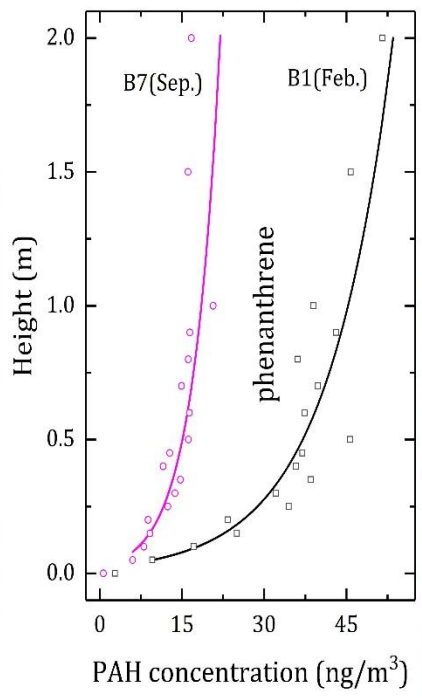
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33 **HIGHLIGHTS**

- 34 ● Gas deposition for phenanthrene and volatilization for benzo[*b*]fluoranthene.
35 ● Impact of vegetation on vertical pattern of PAHs was observed only in growing season.
36 ● Vertical eddy diffusivity in air was calculated by local micrometeorological conditions.
37 ● Diffusive flux of gaseous PAHs in air-soil exchange was compared from two approaches.

38

39

40 **ABSTRACT:**

41 Compared with dry and wet deposition rates, air-soil exchange fluxes cannot be
42 directly measured experimentally. Polyethylene passive sampling was applied to
43 assess transport directions and to measure concentration gradients in order to
44 calculate diffusive fluxes of polycyclic aromatic hydrocarbons (PAHs) across the
45 air-soil interface in an urban park of Shanghai, China. Seven campaigns with high
46 spatial resolution sampling at 18 heights between 0 and 200 cm above the
47 ground were conducted in 2017 - 2018. Air-to-soil deposition was observed, e.g.
48 for phenanthrene, and soil-to-air volatilization for high molecular weight
49 compounds, such as benzo[*b*]fluoranthene. Significant linear correlations
50 between gaseous PAH concentration and log-transformed height were observed.
51 Influence of vegetation on vertical concentration gradients of gaseous PAHs was
52 insignificant in most cases except during the growing season. Local
53 micrometeorological conditions resulted in a directional eddy diffusion in air and
54 then influenced vertical diffusion of gaseous PAHs. Furthermore, the vertical
55 eddy diffusivity was estimated as a function of distance to the air-soil surface.
56 Air-soil exchange fluxes based on the Mackay's fugacity approach were calculated
57 and confirmed by diffusive fluxes within air layer based on vertical concentration
58 gradient of PAHs and eddy/molecular diffusion. Polyethylene passive sampling
59 technology provides a useful tool to investigate air-soil exchange process.

60

61 **KEYWORDS:**

62 LDPE, gas exchange, eddy diffusion, molecular diffusion, fugacity, uncertainty
63 analysis.

64

65 **1. INTRODUCTION**

66 Polycyclic aromatic hydrocarbons (PAHs) are a ubiquitous class of semi-volatile
67 organic pollutants mainly originating from the combustion of organic matter and
68 fuels (Liu et al., 2015), and are widespread all over the world (Casal et al., 2018).
69 Consequently, the earth surface (e.g. soil and vegetation) becomes a critical sink
70 of these hydrophobic pollutants via atmospheric deposition and direct
71 application e.g. by sewage sludge (Feng et al., 2017; Zheng et al., 2015). After
72 primary pollutant sources were reduced or eliminated in air, soils may become
73 secondary sources of accumulated semi-volatile organic compounds depending
74 on air-surface exchange rates (Bao et al., 2015; Degrendele et al., 2016).

75 To assess air-soil exchange, many studies quantified gradients in air and soil
76 fugacities based on model predictions or field measurements. The latter include
77 active air sampling (e.g. low-volume air samplers) (Degrendele et al., 2016; Kurt-
78 Karakus et al., 2006; Tasdemir et al., 2012; Wang et al., 2015) and more recently
79 passive sampling technologies to measure the air fugacity based on polyurethane
80 foam, low-density polyethylene (LDPE) and other materials (Donald and
81 Anderson, 2017; Dumanoglu et al., 2017; Wang et al., 2017; Zhang et al., 2011).
82 Also, concentrations measured in soil samples were converted to fugacities based
83 on air-soil partitioning coefficients and ambient temperatures (Cetin et al.,
84 2017a; Cetin et al., 2019; Qu et al., 2019; Tasdemir et al., 2012; Wang et al., 2017;
85 Zhang et al., 2011). However, large uncertainties exist in estimating air-soil
86 partitioning coefficients and because of soil heterogeneity, e.g. during soil
87 sampling vertical concentration gradients are disturbed leading to
88 misinterpretation on soil fugacity (Ali et al., 2015; Davie-Martin et al., 2015;

89 Donald and Anderson, 2017). Alternative in situ technologies have been
90 developed to directly monitor the soil air, namely the gas phase in soil pores. For
91 example, active fugacity samplers were designed to sample the air close to the
92 soil surface (Cabrerizo et al., 2009; Degrendele et al., 2016; Wang et al., 2015).
93 Donald and Anderson (2017) designed a soil air fugacity sampler based on
94 passive sampling using polyethylene. In addition to assessing the fugacity
95 gradient between soil and air phases, vertical concentration gradients in air can
96 serve as an alternative indicator and reflect the driving force of the air-soil
97 exchange. For example, Kurt-Karakus et al. (2006) mounted active air samplers at
98 the heights of 5, 20, 72, and 200 cm above ground. Lammel et al. (2011) also
99 actively collected air samples from 1.1 and 2.3 m above the soil surface. To not
100 disturb the atmospheric concentration gradients caused by active sampling,
101 Zhang et al. (2011) and Wang et al. (2017) deployed PUF air passive samplers at
102 various heights above the surface.

103 Many studies focused only on assessing transport direction of air-soil diffusive
104 exchange based on fugacity ratio of air and soil (Cetin et al., 2017a; Cetin et al.,
105 2017b; Degrendele et al., 2016; Tian et al., 2016; Wang et al., 2015; Zhu et al.,
106 2017). However, only in few studies exchange fluxes were calculated based on
107 fugacity gradients between soil and air, following the Mackay fugacity approach
108 (Koblizkova et al., 2009; Pokhrel et al., 2018; Tasdemir et al., 2012; Wang et al.,
109 2017; Zhang et al., 2011) or Fick's law of diffusion (Donald and Anderson, 2017).
110 If the influence of vegetation on air-soil is negligible, the diffusive flux within air
111 layer overlying the soil surface would be equal to the air-soil diffusive exchange
112 rates. This prompted us to deploy more passive samplers with high spatial
113 resolution in the topsoil and air at various heights above ground, in order to

114 estimate and compare the diffusive fluxes from the two methods. Low density
115 polyethylene (LDPE) passive samplers were placed in heights of 5 - 200 cm above
116 the soil surface during 2017 - 2018. The objectives of this study were to (a)
117 investigate vertical distribution of gaseous PAHs within air phase and determine
118 their transport direction in air-soil exchange process, (b) assess influence of
119 vegetation on the vertical distribution of gaseous PAHs, (c) quantify eddy
120 diffusivities and estimate diffusive fluxes of gaseous PAHs at different heights
121 above the surface based on local micrometeorological conditions, and (d)
122 compare diffusive fluxes within air phase and in the air-soil exchange interface.

123

124 **2. MATERIALS AND METHODS**

125 **2.1 Study site.**

126 The monitoring site was located in a green space in the city center of Shanghai,
127 China, ~60 meters from the Inner-ring Elevated Road (see Figure S1). This field
128 was selected to investigate the main atmospheric transport pathway for
129 anthropogenic emissions from the densely populated and highly urbanized area
130 in the Eastern China. The green space is a typical receptor of urban pollutant
131 emissions, in particular traffic-related emissions. During sampling, local
132 temperature ranged from 3-40°C and average wind speed was $<1.5 \text{ m s}^{-1}$ with the
133 maximum of 5.5 m s^{-1} .

134

135 **2.2 Samplers and sampling design.**

136 Commercial 51 μm -thick low-density polyethylene (LDPE) was cut into strips of
137 $40 \text{ cm} \times 10 \text{ cm}$. The strips were cleaned and spiked with performance reference
138 compounds (PRCs, more details in the SI) (Liu et al., 2016). The sharpest

139 variations in meteorological variables with height occur within the surface
140 roughness layer (< 100 cm above ground in this study). Air samplers were
141 deployed as an array along a gradient of 17 heights from 5 cm to 200 cm above
142 ground, and were fixed non-uniformly at 5, 10, 15, 20, 25, 30, 35, 40, 45, 50, 60,
143 70, 80, 90, 100, 150, and 200 cm above ground (see Figure S2). To avoid
144 individual samplers blocking each other, they were sequentially distributed in
145 four directions. Meanwhile, three soil samplers (50 cm × 13 cm × 2 cm) were
146 deployed immediately adjacent to the air samplers (< 1 m²). The soil sampler (a
147 LDPE strip was protected by a metal grid) was vertically inserted into topsoil
148 (<15 cm deep), and stainless steel boxes were placed over the LDPE passive
149 samplers. Seven sampling campaigns of 2 - 8 weeks were performed from March
150 of 2017 to September of 2018. Detailed information on air/soil samplers and
151 temporal coverage is provided in the SI (see Figures S2-S3 and Table S1).

152

153 **2.3 Sample analysis and QA/AC.**

154 After sampling, LDPE samples were wiped clean by pure water and hexane to
155 remove the particles and soil dust absorbed on the surface of LDPE strips, and
156 analyzed for PAH concentrations. The LDPE samples were treated according to a
157 method published previously (Liu et al., 2016). Briefly, all LDPE sheets were
158 spiked with surrogate standards (five deuterated PAHs) to assess recovery, and
159 then extracted overnight in dichloromethane and hexane, sequentially. After
160 extracts were concentrated, an injection standard (deuterated fluorene) was
161 added before GCMS analysis. No additional sample cleanup was performed for air
162 samples, while soil samples were passed through a silica gel and sodium sulfate
163 column for cleanup. 16 US EPA PAHs were quantified using an Agilent GC 7890A

164 with a HP-5 MS fused silica capillary column (30 m × 0.25 mm i.d., 0.25 μm film
165 thickness) equipped with an Agilent 5975C MSD in EI+ selected ion monitoring
166 (SIM) mode. PAH concentrations in LDPE samples were corrected for surrogate
167 standard recoveries and blank-subtracted using field blanks in the same
168 sampling batch. PAH concentrations in field blanks were < 10% of measured
169 concentrations in samples for 3 - 4 rings PAHs, and < 20% for 5 - 6 rings PAHs,
170 except naphthalene (30%) and dibenz [*a,h*] anthracene (47%) (see Table S2).
171 PRC concentrations in field and lab blanks from each batch were used to quantify
172 initial concentrations of PRCs in deployed samples. Measured concentrations in
173 LDPE were used to calculate (equivalent) gaseous concentrations of PAHs in air
174 (or soil) based on LDPE-air partition coefficients corrected to ambient
175 temperature (Liu et al., 2016; McDonough et al., 2014). The calculation is based
176 on the assumption that the uptake rate of target PAHs from ambient environment
177 to the LDPE sheet is equal to the loss rate of PRCs from the LDPE sheet (Booij and
178 Smedes, 2010). Equilibrium percentages of target PAHs were calculated from the
179 quantities of PRCs remaining in each sample. More analytical details and
180 calculations are shown in the Supporting Information (see Tables S2 and S3).

181 Procedural blanks and field blanks were performed during each campaign to
182 assess the potential contamination for quality assurance. Limits of detection of
183 PAHs were < 1 ng m⁻³ in air and soil air except naphthalene and acenaphthene.
184 Average recoveries of surrogate standards were 45 - 81% and average relative
185 standard deviation percentages of target PAHs were 10 - 25%. Sampling rates
186 were 16 m³ day⁻¹ in air and 4.7 m³ day⁻¹ in soil, respectively. More information on
187 quality assurance and quality control is in the Supporting Information (Tables
188 S1-2 and S4).

189

190 **2.4 Air-soil diffusive exchange flux of PAHs.**

191 The mass flux ($F_{a/s}$, in $\text{ng m}^{-2} \text{ day}^{-1}$) is driven by the difference in target's chemical
192 activity between air and soil matrixes, and the Mackay's surface soil fugacity
193 approach was employed for the calculation (Mackay, 2001), see Equation 1.

194

$$195 \quad F_{a/s} = \frac{(C_{pe,a} - C_{pe,s})}{K_{PE-a}} \times v_{a/s} \quad (1)$$

196

197 where $C_{pe,a}$ and $C_{pe,s}$ are the equilibrium concentrations of PAHs in PE matrix
198 deployed in air and soil, respectively, and K_{PE-a} is the air temperature-corrected
199 PE-air partition coefficient. Details on estimating overall mass transfer velocity in
200 air-soil diffusive exchange of PAHs ($v_{a/s}$, in m day^{-1}) from the volume fractions of
201 soil water and soil air are provided in the Supporting Information, and the
202 calculated $v_{a/s}$ and $F_{a/s}$ values are presented in Table S5.

203

204 **2.5 Vertical diffusive flux of PAHs within the air layer.**

205 Both random molecular motion and turbulence promote mixing within the air
206 layer. The vertical diffusive flux (F_{Dual} , in $\text{ng m}^{-2} \text{ day}^{-1}$) consists of two parts,
207 namely the vertical eddy diffusive flux and the molecular diffusive flux (F_{Eddy} and
208 F_{mol} , in $\text{ng m}^{-2} \text{ day}^{-1}$). Based on Fick's first law, the mass fluxes are driven by a
209 vertical gradient of PAH concentrations (dC/dH), and related to molecular
210 diffusivity (D_A , in $\text{m}^2 \text{ day}^{-1}$) and vertical eddy diffusivity (E_z , in $\text{m}^2 \text{ day}^{-1}$) of PAHs
211 in air, as shown in Equation 2. E_z was estimated only from measured heat fluxes;
212 the contribution of momentum flux was ignored as the horizontal wind vector

213 was small in this study ($< 1.5 \text{ m s}^{-1}$) (Kurt-Karakus et al., 2006; Schwarzenbach et
 214 al., 2002). Typically, D_A is around $0.05 \text{ cm}^2 \text{ s}^{-1}$, while in this study E_z was in the
 215 range of $0.01 - 0.6 \text{ cm}^2 \text{ s}^{-1}$ in time-integrated average. Details on estimating the
 216 vertical eddy diffusivity and vertical diffusive flux within air phase are provided
 217 in the Supporting Information, and the calculated E_z , concentration gradient, and
 218 F_{Dual} values are listed in Table S6-S8.

219

$$220 \quad F_{\text{Dual}} = F_{\text{Eddy}} + F_{\text{mol.}} = (E_z + D_A) \frac{dC}{dH} \quad (2)$$

221

222 2.6 Uncertainty analysis on the calculated fluxes.

223 Error propagation analysis was applied to estimate uncertainty of the air-soil
 224 diffusive exchange flux of PAHs ($F_{a/s}$), errors in PE-air partition coefficients ($K_{\text{PE-}}$
 225 a), overall mass transfer velocities ($v_{a/s}$), concentrations of PAHs in PE strips
 226 deployed in air and soil ($C_{\text{PE},a}$ and $C_{\text{PE},s}$) were considered, as shown in Equation 3.
 227 Five parameters, including vertical eddy diffusivity (E_z), molecular diffusivity in
 228 air (D_A), height difference (h), concentration of PAHs at different heights (C_{up} and
 229 C_{down}), were included to estimate uncertainty of vertical diffusive fluxes within
 230 the air layer (F_{Dual}) in Equation 4.

231

$$232 \quad \frac{\delta F_{a/s}}{F_{a/s}} = \sqrt{\left(\frac{\delta K_{\text{PE-}a}}{K_{\text{PE-}a}}\right)^2 + \frac{(\delta C_{\text{PE},a}^2 + \delta C_{\text{PE},s}^2)}{(C_{\text{PE},a} - C_{\text{PE},s})^2} + \left(\frac{\delta v_{a/s}}{v_{a/s}}\right)^2} \quad (3)$$

$$233 \quad \frac{\delta F_{\text{Dual}}}{F_{\text{Dual}}} = \sqrt{\frac{(\delta E_z^2 + \delta D_A^2)}{(E_z + D_A)^2} + \frac{(\delta C_{\text{up}}^2 + \delta C_{\text{down}}^2)}{(C_{\text{up}} - C_{\text{down}})^2} + \left(\frac{\delta h}{h}\right)^2} \quad (4)$$

234

235 3. RESULTS AND DISCUSSION

236 3.1 Vertical concentration profiles of gaseous PAHs in air and transport 237 direction of air-soil exchange.

238 The vertical concentrations of gaseous PAHs on the air side exhibit distinct
239 concentration gradients in space, and directly reveal the transport direction of
240 PAHs. The vertical concentration distributions of PAHs in air are shown
241 exemplary for campaign ADH04 (2017/6/14 - 2017/6/30) in Figure 1. Gaseous
242 concentrations of phenanthrene (see Figure 1.a) increased with the height, from
243 16 ng m^{-3} at 5 cm to 45 ng m^{-3} at 50 cm above ground, where it remained stable
244 ($47 - 51 \text{ ng m}^{-3}$) within the 1 - 2 m. This indicates gas deposition of phenanthrene
245 from an upper air layer to a lower one close to ground. Turbulence also varied
246 with height, which affected both uptake of PAHs and release of PRCs, and in
247 general, the sampling rate increased with height (see Figure 1.f). Gaseous
248 concentrations of phenanthrene have been corrected based on equilibrium
249 percentage of PRCs observed (see Equations S1-S2). The equivalent gaseous
250 concentrations of phenanthrene in topsoil ranged from $1.62 - 2.1 \text{ ng m}^{-3}$, which is
251 significantly ($p < 0.01$) lower than in air ($16 - 51 \text{ ng m}^{-3}$). Consequently, transport
252 direction of phenanthrene is confirmed as air-to-soil deposition.

253 A similar deposition behavior was observed for fluoranthene as shown in
254 Figure 1.b. In contrast, some 4 - 6 rings PAHs showed an unusual volatilization
255 from soil to air, e.g., benzo[*b*]fluoranthene and benzo[*ghi*]perylene (Figure 1.c, d,
256 and e); the spatial trend of chrysene in air was unclear, a possible explanation of
257 which is that chrysene reached or approached equilibrium in air-soil exchange
258 process. The downward deposition of lower molecular weight PAHs (especially
259 phenanthrene) indicates a significant emission in Shanghai urban space and on-

260 going accumulation of the gaseous PAHs in urban topsoils, whereas the upward
261 volatilization of higher molecular weight PAHs (e.g., 5 - 6 rings PAHs) is different
262 with the general expectation (deposition) and reported observations (Cetin et al.,
263 2017a). We ever suspected the LDPE soil samplers and air samplers at lower
264 heights were more vulnerable to be associated with fine soil dust. In the
265 improved experimental process, the LDPE trips were cleaned up by deionized
266 water and swabbed with a hexane-soaked wiper to ensure that only absorbed
267 molecules into the LDPE samplers would be quantified. Interestingly, passive
268 samplers did 'see' the unusual process. The observed volatilization is possibly
269 attributed to local high legacy pollution or a long-term accumulation from dry
270 and wet deposition (ever-increasing soil fugacity is even greater than the air
271 fugacity) in topsoils and the soil acting as a secondary source of the air PAHs.

272 In this study, strong linear correlations were obtained by fitting a relationship
273 of gaseous concentration as a function of log-transformed height. Fitted results
274 (solid line) are presented in Figure 1 ($R > 0.68$), which were significantly
275 correlated ($p < 0.01$) for most PAHs except chrysene. Kurt-Karakus et al. (2006)
276 reported a volatilization gradient of DDT in a historically treated agricultural soil
277 in Canada and a similar logarithmic function between height and gas
278 concentration. Wang et al. (2017) observed a deposition gradient of
279 organochlorine pesticides in the pasture of the central Tibetan Plateau, but used
280 a power function to fit the relationship. The curve is related to wind speed
281 (Majewski et al., 1990), or to be more precise, to vertical turbulence. The slopes
282 derived here are further used to quantify vertical concentration gradient at
283 different heights and estimate diffusive flux of PAHs within the air layer (see
284 below). Although different functions were used to fit, interestingly, mathematical

285 expressions are similar based on derivative of the two functions (dC/dH) to
286 describe vertical concentration gradient, e.g., $12.97 H^{-1}$ for phenanthrene in this
287 study and $4.97 H^{-0.87}$ for DDT from Wang et al (2017).

288

289 **3.2 Influence of vegetation during the growing season on vertical** 290 **concentration distribution of gaseous PAHs.**

291 Vegetation has been considered as a globally important compartment in
292 scavenging and storing atmospheric pollutants and transferring them into soil
293 through falling litter (Odabasi et al., 2015), or back into the air by re-
294 volatilization (Bao et al., 2016; Terzaghi et al., 2017). Many studies reported
295 urban vegetation as an air pollutant collector based on large scale modeling
296 (Simonich and Hites, 1994; Terzaghi et al., 2017). Furthermore, a lack of
297 experimental investigation has been recognized (Setala et al., 2013). The study
298 here provides an empirical-based evidence on influence of urban vegetation on
299 gaseous PAHs.

300 Vertical distributions of gaseous PAHs before (February), during (March) and
301 after (May - September) the growing season are compared in Figure 2. Gaseous
302 concentrations of phenanthrene, fluoranthene and benzo[*b*]fluoranthene were
303 normalized to a common condition of 25 °C and 1 atm for comparison over time,
304 due to gaseous concentrations of PAHs varying with air temperature and
305 pressure. Field vegetation grew up from ~5 cm height above ground before the
306 growing season to ~30 cm height after the growing season. Overall,
307 concentrations of PAHs decreased over time, e.g. phenanthrene from 50 ng m^{-3} in
308 February to 15 ng m^{-3} in September. This time trend of gaseous PAHs is similar as
309 air pollution in Shanghai (see Figure S3). Clearly, PAH concentrations are linked

310 to overall air quality, however these vertical patterns of PAHs are similar in
311 different campaigns except B2 (in March), especially for phenanthrene at the
312 heights of 10 - 35 cm in Figure 2.a. The differences of PAH patterns before, during
313 and after growing seasons are attributed to absorption by vegetation during the
314 growing season. For phenanthrene and fluoranthene deposition in the growing
315 season (B2) resulted in markedly lower concentrations in the vegetation layer
316 than in the upper air layer. However, the adsorption by vegetation was not
317 observed after the growing season.

318 Klingberg et al. (2017) observed substantially lower gaseous PAH levels in
319 parks than adjacent areas near traffic. Ghasemian et al. (2017) reported that the
320 dense canopy improved the near-road air quality. However, the effect remains
321 controversial. Setala's result (2013) suggested a minor capability of urban
322 vegetation to remove air pollutants in northern climates. Moreover, Viippola et al.
323 (2016) and Yli-Pelkonen (2018) reported higher gaseous PAH levels under tree
324 canopies than adjacent open area in southern Finland and northeast China,
325 associating with the trapping of polluted air under canopies and re-volatilization
326 from soil back to the air. Our results support the removal effect of air pollutants
327 by vegetation, but the influence was observable only during the growing season
328 (emergence of new and clean biomass), and the influence faded away after the
329 growing season.

330

331 **3.3. Air-soil diffusive exchange of PAHs.**

332 Air-soil diffusive exchange is an interphase transfer process, which is usually
333 characterized by the Whitman two-resistance mass transfer coefficient approach
334 (Schwarzenbach et al., 2002). Quantification of the diffusive flux in an air layer

335 close to the surface by micrometeorological approach is an alternative method. In
336 this study, the micrometeorological approach focused on an air layer with a
337 specific thickness (100 cm), in which air turbulence or eddy diffusion as well as
338 random molecular motion together drive pollutant transfer. In the thin layer
339 close to the air-soil interface, eddies tend to be dampened and movement to and
340 from the interface occurs only by molecular diffusion (Mackay, 2001). Both
341 methods were applied to estimate comprehensively on the air-soil diffusive
342 exchange flux in this study.

343

344 **3.3.1 Estimation of vertical eddy diffusivity within the air layer close to the** 345 **air-soil surface.**

346 Molecular diffusivity is commonly isotropic, while eddy diffusivity in the
347 vertical direction is usually much lower than horizontal diffusion
348 (Schwarzenbach et al., 2002). Vertical eddy diffusion can strongly influence PAH
349 transfer in air-soil exchange process. Vertical eddy diffusivity can be calculated
350 from heat and momentum fluxes, more specifically, from temperature profile and
351 horizontal wind vector, respectively (Kurt-Karakus et al., 2006). Due to a lower
352 wind speed in this case, the vertical eddy diffusivity was linked only to the
353 vertical temperature profile close to the surface.

354 Taking temperature profile within specific three days (including sunny, rainy
355 and cloudy days) for example, the vertical eddy diffusivities (E_z) at the heights of
356 0 - 5 cm, 5 - 20 cm, 20 - 60 cm, and 60 - 100 cm were calculated from the
357 temperature profile mentioned above, and results are presented in Figure 3. First
358 of all, spatial and temporal variations of air temperature were notable. Especially
359 during sunny days, air temperature varied drastically over time due to solar

360 radiation, and changed with height above the surface. Moreover, air temperature
361 was much higher than soil temperature during sunny days, while the both
362 temperatures were similar on a rainy day and at nighttime. In addition,
363 vegetation layer was believed to have a significant influence on vertical
364 temperature profile (Daemei et al., 2018). During the deployment period of this
365 study, the height of vegetation layer reached up to ~20 cm above the surface. It
366 was covered by a dense grass layer, meanwhile the temperature difference
367 between soil and air above vegetation in the sunny day was > 10 °C.
368 Consequently, for monitoring real-time temperature profiles in subsequent work,
369 three temperature sensors were deployed in air above vegetation, vegetation
370 layer and soil surface, respectively.

371 Such temperature profile led inevitably to a great heterogeneity in vertical
372 eddy diffusivity based on the micrometeorological approach, as shown in Figure
373 3.b. When temperature rose constantly in the morning on a sunny day, air masses
374 warmed up, resulting in an updraft and a negative value of E_z . After temperature
375 reached a peak value at noon, it decreased continually until night and air masses
376 cooled down causing a downdraft and a positive value of E_z . Therefore, vertical
377 eddy diffusion is a vector with a direction (updraft or downdraft). Figure 3.b
378 demonstrates a remarkable variation of eddy diffusivity over time, in magnitude
379 and direction. We calculated the average time-integrated eddy diffusivities of
380 updraft and downdraft, respectively, similar as the E_z values in Figure 4.a and b.
381 Generally speaking, the magnitude of E_z varied with distance to the air-soil
382 surface. The closer to the surface, the smaller E_z value was, and even lower than
383 molecular diffusivity of PAHs in air. The vertical pattern of eddy diffusivity is in
384 line with the above statement that eddy diffusion tends to be damped in a thin

385 layer close to the surface.

386 As a consequence, solar radiation and vegetation cover control temperature
387 vertical profile and finally influence vertical eddy diffusivity in local micro-
388 environment close to ground.

389

390 **3.3.2 Estimation of diffusive flux of gaseous PAHs within the air layer based** 391 **on the micrometeorological approach.**

392 Random molecular motion and vertical eddy motion promote vertical
393 transferring of gaseous PAHs within the air layer close to ground. The diffusive
394 flux, hence, depends on a vertical gradient of PAH concentration (dC/dH), vertical
395 eddy diffusivity (E_z) and molecular diffusivity (D_A) of PAHs in air. The slopes
396 ($dC_{air}/d \ln(H)$) in Figure 1 and Table S7 were used to calculate the vertical
397 concentration gradient (dC/dH) at different heights above the surface (see Figure
398 4.a and b). Molecular diffusivity in air is constant at specific condition, e.g. $D_{Ph,A,A}$
399 = $0.053 \text{ cm}^2 \text{ s}^{-1}$ for phenanthrene at $25 \text{ }^\circ\text{C}$ and 1 atm , while vertical eddy
400 diffusivity in air varies with the height above the air-soil surface, as estimated
401 above and shown in Figure 4.a and b. Additionally, the vertical eddy diffusivity is
402 an instantaneous vector with an upward (negative value) or downward (positive)
403 direction. If the direction of vertical eddy diffusivity was the same as the
404 concentration gradient, gaseous PAHs transfer from high to low concentrations. If
405 not, there is no net PAH deposition during an upward eddy flux. In order to
406 estimate the eddy diffusive flux, we calculated the updraft and downdraft eddy
407 diffusivities in time-integrated average during each sampling campaign,
408 respectively (see Table S6). Eddy diffusive fluxes and molecular diffusive fluxes at
409 the heights $< 1 \text{ m}$ above the surface were calculated (see Table S8) and results of

410 phenanthrene and benzo[*b*]fluoranthene as examples of deposition and
411 volatilization are shown in Figure 4.

412 Since phenanthrene deposited from air to soil, the positive eddy diffusivities
413 (downward) were used for calculation, as presented in Figure 4.a. The values of
414 E_z close to the surface (height < 0.1 m) were lower than molecular diffusivity of
415 phenanthrene in air ($0.053 \text{ cm}^2 \text{ s}^{-1}$). With the increase in height, the E_z values
416 increased up to $\sim 0.61 \text{ cm}^2 \text{ s}^{-1}$ at the height of 1 m, 10 times greater than the
417 molecular diffusivity in air. As for volatilization of benzo[*b*]fluoranthene from soil
418 to air, the negative eddy diffusivities (upward) were used for calculation.

419 Vertical eddy diffusive flux (F_{Eddy}) and total diffusive flux ($F_{\text{Total}} = F_{\text{Eddy}} + F_{\text{molecular}}$)
420 were estimated based on average updraft/downdraft eddy diffusivity, molecular
421 diffusivity and vertical concentration gradient. The results of phenanthrene and
422 benzo[*b*]fluoranthene are shown in Figure 4.c and d, respectively. In the region
423 close to the air-soil surface (e.g., < 0.1 m), molecular diffusion dominated the
424 mass transfer and percentages of eddy flux to total diffusive flux were < 50% for
425 the two PAHs, as well as uncertainties of the both fluxes were greater than in
426 higher air layer. With increasing distance to the surface, the percentage raised
427 to > 80% at the height > 0.5 m (above vegetation layer).

428

429 **3.3.3 Estimation of air-soil diffusive exchange flux based on the Mackay's** 430 **fugacity approach.**

431 To maintain mass balance, the air-soil exchange flux should be equal or similar to
432 the diffusive flux within the air layer. The air-soil exchange fluxes of gaseous
433 PAHs were estimated based on the Mackay's fugacity approach with the Whitman
434 two-resistance model (Mackay, 2001; Schwarzenbach et al., 2002). The exchange

435 flux is related to the concentration difference between two compartments and
436 overall mass transfer velocity of PAHs in air-soil exchange process. The transfer
437 velocity is linked primarily to PAH molecular diffusivities in the atmosphere, soil
438 air and soil water, the boundary layer thicknesses in atmosphere and soil, soil
439 porosity, volume fraction of air and water in soil, soil temperature and Henry's
440 law constant of PAHs. The air-soil exchange transfer velocities and mass fluxes of
441 PAHs are listed in Table S5, and more calculation details in the SI. The mass
442 transfer velocities of PAHs were estimated as 76 - 106 cm day⁻¹ in this study.

443 When calculating the air-soil exchange fluxes based on gradients in air and soil
444 fugacities, Tasdemir et al. (2012) placed air sampler on the roof of a building,
445 Donald et al. (2017) deployed passive sampler at the height of ~1.5 m above
446 ground, and Wang et al. (2017) chose the air at 0 - 3 cm above ground to calculate
447 the flux. Since gaseous PAHs levels tended to stable over 50 cm above the surface
448 in this study, air concentrations at the heights of 100 cm and 5 cm were used to
449 calculate the air-soil exchange fluxes (namely F₁₀₀ and F₅, listed in Tables S5),
450 respectively. Taking data from the sampling campaign ADH04 for example, the
451 air-soil exchange fluxes and diffusive fluxes within the air layer were compared in
452 Table 1. For example, for phenanthrene and fluoranthene, the diffusive fluxes
453 within the air layer were 63.2 ng m⁻² day⁻¹ and 3.62 ng m⁻² day⁻¹, respectively and
454 thus slightly greater than the calculated air-soil exchange fluxes (13.3 - 43.6 ng
455 m⁻² day⁻¹ and 0.19 - 1.62 ng m⁻² day⁻¹, respectively). While in the soil-to-air
456 volatilization process (e.g. chrysene, benzo[*b*]fluoranthene, and
457 benzo[*ghi*]perylene), the air-soil exchange fluxes were slightly higher than their
458 diffusive fluxes within air layer. But broadly speaking, the two calculated fluxes
459 are comparable based on the two methods, namely the micrometeorological

460 approach and the Mackay fugacity approach.

461 In terms of air-soil exchange flux, the values of F_{100} are closer to the diffusive
462 fluxes within air layer in comparison to the F_5 values, but not indicating that F_{100}
463 is recommended. First of all, because of the assumed air boundary thickness of
464 4.75 mm in the Mackay's fugacity approach (2001), the air concentration should
465 be measured in the air layer as possible as close to the air-soil surface (e.g. 5 cm),
466 and so F_5 would be recommended rather than F_{100} . Second, both of them would
467 be underestimated, but F_{100} is closer to the diffusive flux within air layer due to
468 the greater concentration difference between two compartments. Overall mass
469 transfer velocities ($v_{a/s}$) of gaseous PAHs in air-soil exchange process were
470 estimated based only on their molecular diffusion here. In fact, day-night change
471 (solar radiation) leads to soil temperature variation, which results in different
472 atmospheric boundary layers and different eddy diffusivities of pollutants in air.
473 The calculated transfer velocity, hence, might be underestimated because of the
474 neglect of eddy diffusion in air.

475 As for air-soil exchange diffusive fluxes in the whole study, observed
476 deposition fluxes were $9.8 \text{ ng m}^{-2} \text{ day}^{-1}$ for fluorene, $31.4 \text{ ng m}^{-2} \text{ day}^{-1}$ for
477 phenanthrene, and $2.5 \text{ ng m}^{-2} \text{ day}^{-1}$ for fluoranthene on average. The other PAHs
478 presented a volatilization flux (e.g. $-0.31 \text{ ng m}^{-2} \text{ day}^{-1}$ for benzo[*b*]fluoranthene)
479 or approached equilibrium (e.g. pyrene and perylene, see Table S5).

480

481 **4. CONCLUSIONS**

482 In this study, both air-to-soil deposition and soil-to-air volatilization were
483 observed for different PAHs in an urban micro-environment based on their
484 vertical concentration gradients. Concentration-height relationships were

485 linearized by plotting log-transformed height and gaseous concentrations of
486 PAHs. Influence of vegetation on vertical distribution of PAHs was observable
487 only in the growing season, but insignificant in other seasons. Concentration
488 gradient and micrometeorological condition controlled diffusion of gaseous
489 PAHs. Eddy diffusion dominated the transfer of gaseous PAHs in higher air layer,
490 and was dampened in the layer close to the air-soil surface. Eddy diffusivity was
491 directional and depended mainly on changing vertical profiles of air and soil
492 temperatures. The total diffusive flux within the air layer close to the surface
493 (including vertical eddy diffusive flux and molecular diffusive flux) and air-soil
494 exchange flux based on the Mackay's fugacity approach were comparable to a
495 large extent.

496 This work demonstrated that monitoring vertical concentrations of gaseous
497 PAHs based on LDPE passive sampling technology is vital to assess transport
498 direction and quantify diffusive fluxes in air-soil exchange process. Although
499 there may be differences in air/soil fugacities between passive and active
500 sampling technologies, passive sampling is a logical choice for better observing
501 vertical concentration distribution with a high spatial resolution and without air
502 disturbance of sampling. Due to its low cost, no power requirement, and easy
503 deployment in air and soil compartments, passive sampling provides a useful
504 tool for monitoring air-soil diffusive exchange process, but field correction for
505 non-equilibrium compounds is needed and yet considerable uncertainties in field
506 measurement are unavoidable.

507

508 **ASSOCIATED CONTENT**

509 **Supporting Information**

510 Detailed information on sampling design and flux calculation can be found along
511 with calculated overall mass transfer velocity, eddy diffusivity, and diffusive flux
512 in the estimated site.

513

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517 **Notes**

518 The authors declare no competing financial interest.

519

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657

658 **FIGURES CAPTIONS**

659 **Figure 1.** Vertical gas phase concentration profiles of selected PAHs (a - e) and
660 sampling rate (f) at height 5 - 200 cm above ground. (Deployment period:
661 June 14 - 30, 2017)

662 **Figure 2.** Effect of vegetation on the vertical gas phase concentration distribution
663 of (a) phenanthrene, (b) fluoranthene and (c) benzo[b]fluoranthene
664 from topsoil to the air layer close to ground in February (B1), March
665 (B2), May (B3), June (B4), July (B5), and September (B7).

666 **Figure 3.** Vertical pattern of temperature and eddy diffusivity from topsoil to 1 m
667 height air.

668 **Figure 4.** Pattern of vertical eddy diffusivity, concentration gradient, and diffusive
669 fluxes (b) of phenanthrene (PhA) and benzo[b]fluoranthene (BbF).

670

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672

673 Table 1. PAH mass flux ^a (ng m⁻² day⁻¹) within the air layer close to the ground
 674 and in air-soil exchange process.
 675

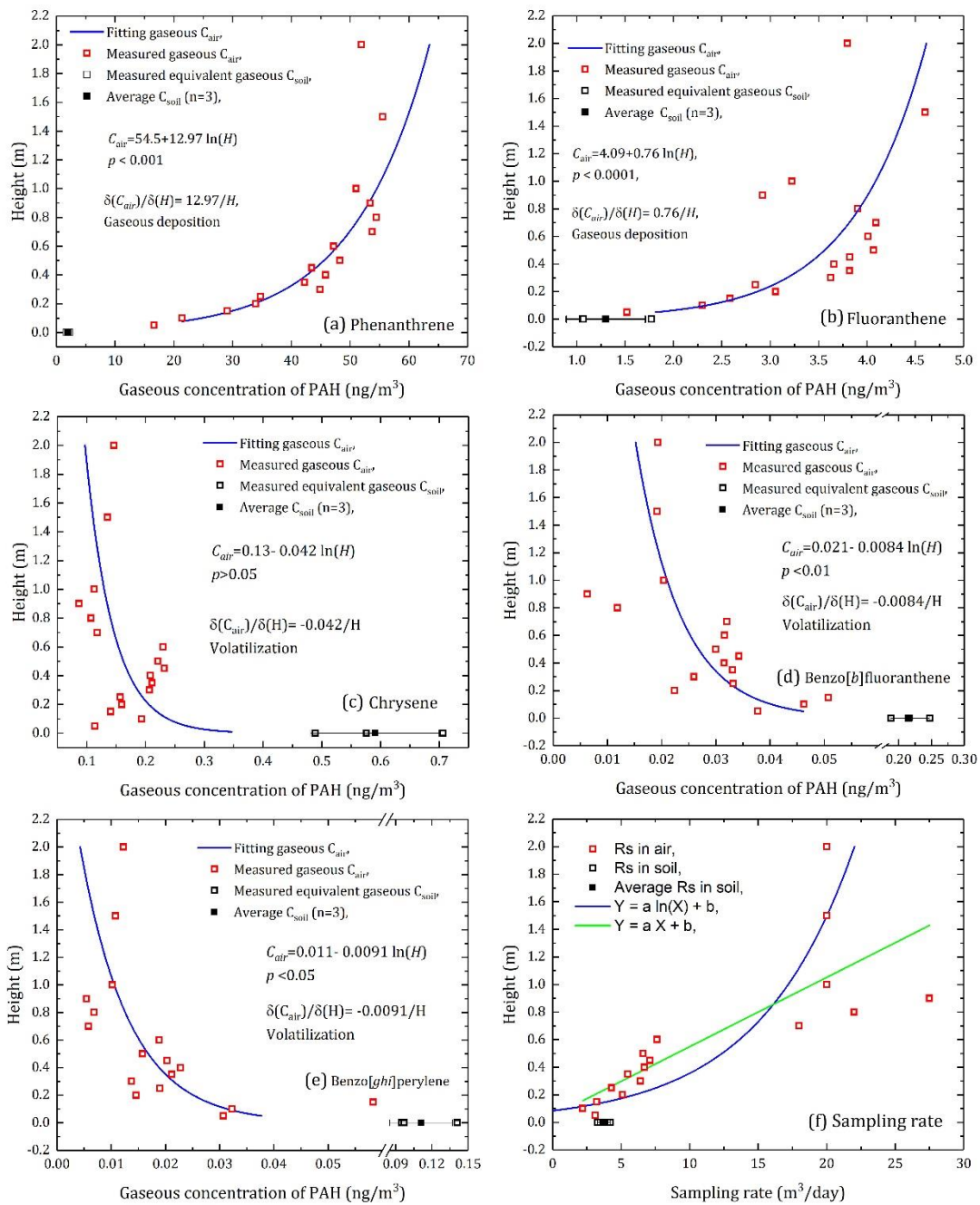
| PAHs | Diffusive flux within air layer | Air-soil exchange flux | |
|-------------------------------|------------------------------------|------------------------|------------|
| | | F_{100} ^b | F_5 |
| Phenanthrene | 63.2±13.6 | 43.6±31.6 | 13.3±9.5 |
| Fluoranthene | 3.62±0.76 | 1.62±1.25 | 0.19±0.38 |
| Chrysene | -0.17±0.03 | -0.39±0.27 | -0.39±0.27 |
| Benzo[<i>b</i>]fluoranthene | -0.03±0.01 | -0.15±0.10 | -0.14±0.10 |
| Benzo[<i>ghi</i>]perylene | -0.04±0.01 | -0.08±0.05 | -0.06±0.04 |

676 ^a: data from the sampling campaign ADH04. Positive values indicate net deposition, while
 677 negative values reflect volatilization from soil to air.

678 ^b: F_{100} and F_5 denote fluxes calculated based on the difference of concentrations between the
 679 soil surface and the air at a height of 100 cm above ground, and 5 cm above the ground.

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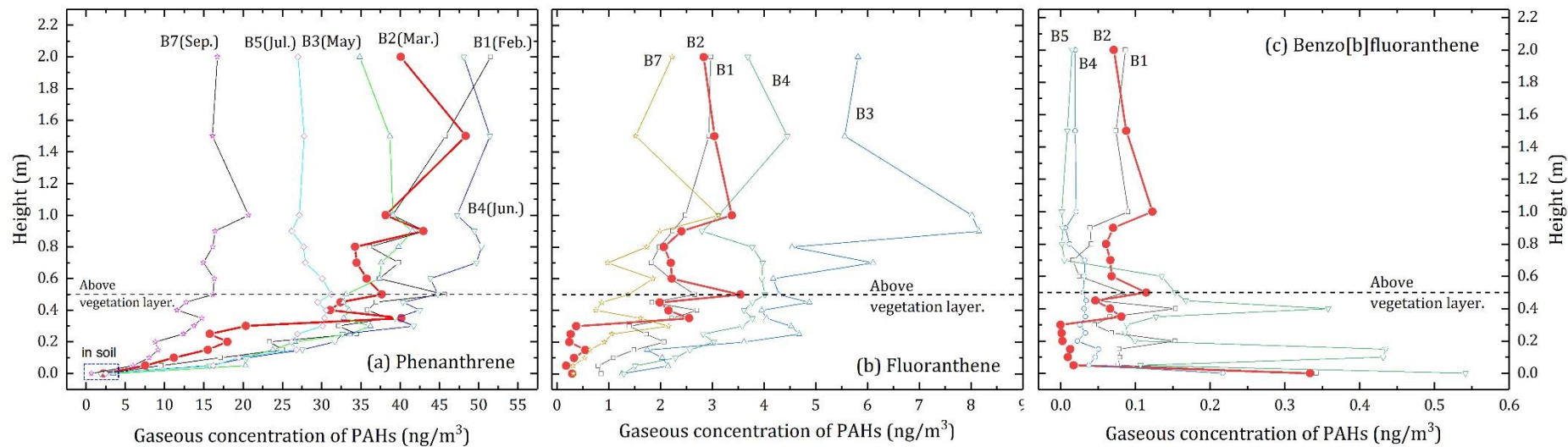
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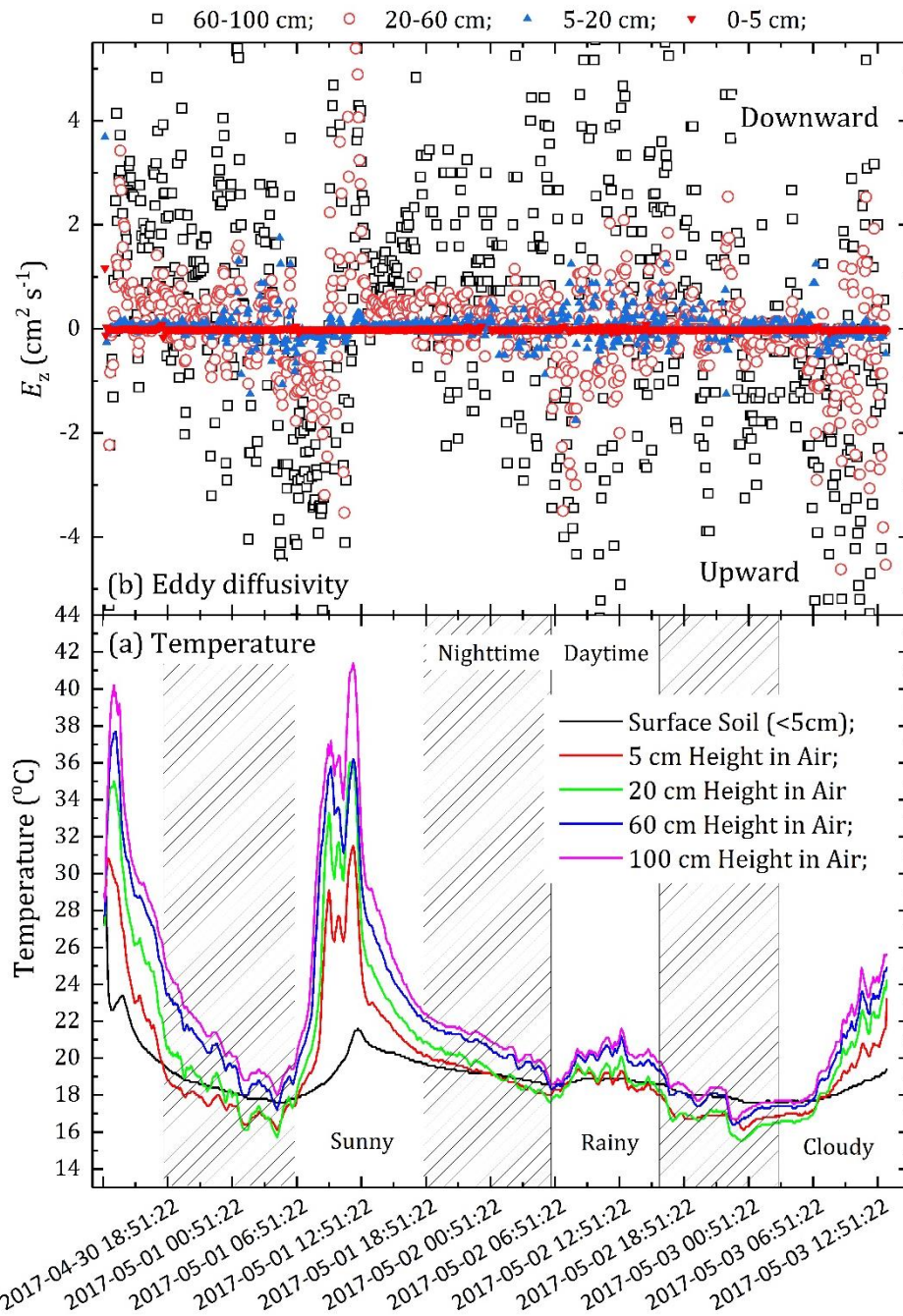
684 Figure 1. Vertical gas phase concentration profiles of selected PAHs (a - e) and
 685 sampling rates (f) at height 5 - 200 cm above ground. (Deployment period:
 686 June 14 - 30, 2017)

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688

689 Figure 2. Effect of vegetation on the vertical gas phase concentration distribution of (a) phenanthrene, (b) fluoranthene and (c)
690 benzo[b]fluoranthene from topsoil to the air layer close to ground in February (B1), March (B2), May (B3), June (B4), July (B5),
691 and September (B7).



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Figure 3. Vertical pattern of temperature and eddy diffusivity from topsoil to 1 m height air.

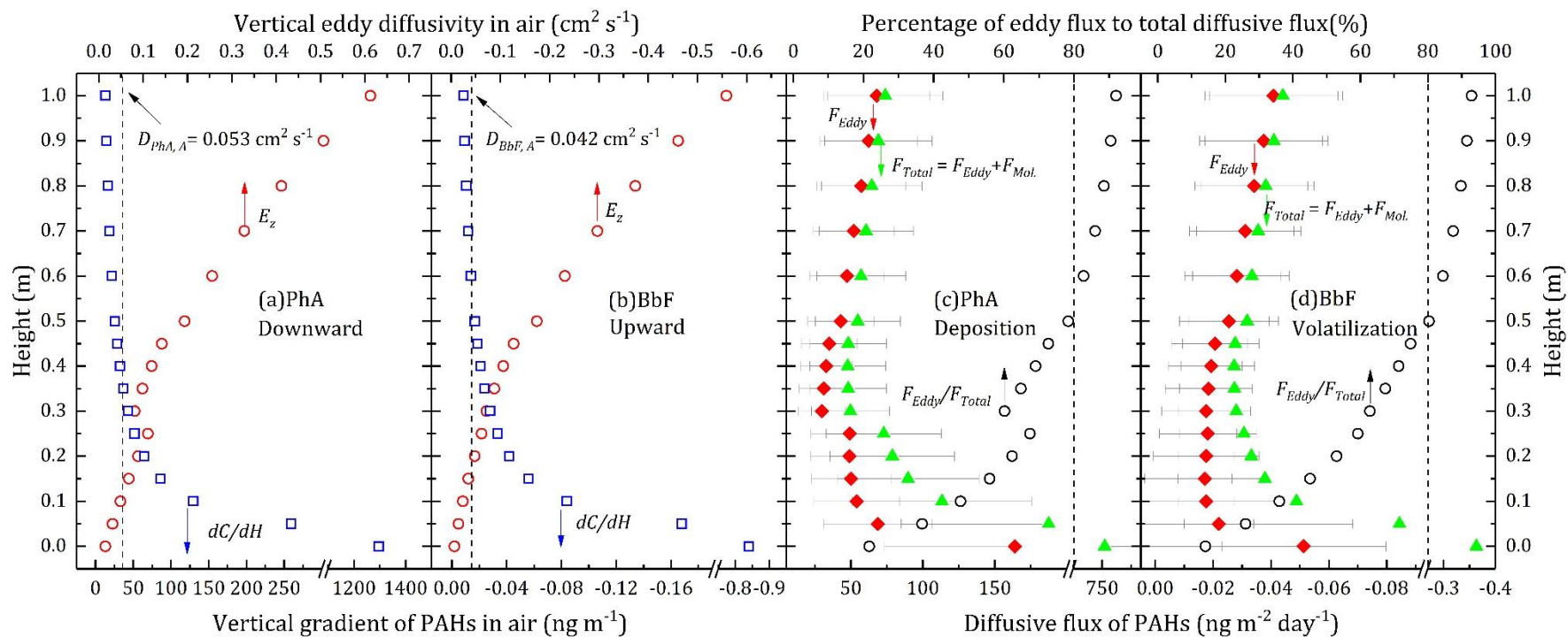


Figure 4. Pattern of vertical eddy diffusivity, concentration gradient, and air diffusive fluxes of phenanthrene (PhA) and benzo[b]fluoranthene (BbF).

